



**PHYSICAL CHEMISTRY 2014**

12<sup>th</sup> International Conference  
on Fundamental and Applied Aspects of  
Physical Chemistry

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The Conference is dedicated to the  
25. Anniversary of the Society of Physical Chemists of Serbia

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September 22-26, 2014  
Belgrade, Serbia

**ISBN 978-86-82475-30-9**

**Title:** PHYSICAL CHEMISTRY 2014 (Proceedings)

**Editors:** Ž. Čupić and S. Anić

**Published by:** Society of Physical Chemists of Serbia, Studenski trg 12-16, 11158, Belgrade, Serbia

**Publisher:** Society of Physical Chemists of Serbia

**For Publisher:** S. Anić, President of Society of Physical Chemists of Serbia

**Printed by:** “Jovan” Printing and Publishing Company; 200 Copies;

**Number of pages:** 6+ 441; **Format:** B5; Printing finished in September 2014.

**Text and Layout:** “Jovan”

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PHYSICAL CHEMISTRY 2014

*12th International Conference on  
Fundamental and Applied Aspects of  
Physical Chemistry*

*Organized by  
The Society of Physical Chemists of  
Serbia*

*in co-operation  
with\_*

*Institute of Catalysis Bulgarian Academy of Sciences*

*Boreskov Institute of Catalysis of Siberian Branch of the Russian Academy  
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## PHOTOCATALYTIC DECOLORISATION OF SELECTED ORGANIC DYES BY MESOPOROUS TiO<sub>2</sub> THIN FILMS

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### ABSTRACT

The mesoporous TiO<sub>2</sub> films were prepared by dip coating technique combined with the evaporation-induced self-assembly method using selected polymer templates. Influence of specific surface area, pore size, (nano)crystal structure and morphology of TiO<sub>2</sub> films on the photocatalytic behaviour was investigated. In addition, particular process parameters were considered in decolorisation of organic dyes, such as: thickness of TiO<sub>2</sub> films, initial concentration of selected pollutants and number of reaction cycles.

### INTRODUCTION

One of the major global problems is water pollution. The large amount of organic dyes used in textile manufacturing processes and other industries represents an increasing environmental danger owing to their toxic, carcinogenic and mutagenic nature [1].

Due to their numerous favorable properties TiO<sub>2</sub> thin films have been commonly used photocatalysts for degrading a wide range of organic pollutants. Several techniques and preparation process parameters can be used in order to obtain photocatalyst with the desired properties.

The evaporation-induced self-assembly method [2-4] using various titania precursors and templating polymers has been used in order to prepare mesoporous TiO<sub>2</sub> films. It is proposed and shown that the type of polymer template has a substantial influence on the porosity of the final material [5].

The aim of this study was to investigate the photocatalytic activity of homogenous and crack-free TiO<sub>2</sub> films with templated mesoporosity, prepared by using two different polymers (Pluronic F127 and PSM02). In this paper photocatalytic decolorisation of methylene blue (MB) and crystal violet (CV) was used in order to test photocatalytic activity of the TiO<sub>2</sub> thin

films as a function of porosity, crystallinity, particle size, and thickness (catalyst dosage).

### EXPERIMENTAL

Templated TiO<sub>2</sub> thin films were prepared by using silicon wafers as substrates. Precursor solutions for dip coating were prepared using TiCl<sub>4</sub> dissolved in dry ethanol under an inert atmosphere. An alcohol-water mixture containing either Pluronic F127 or PSM02 polymer was added and molar ratio between TiCl<sub>4</sub> and F127 was adjusted to 1 : 0.001, while the molar ratio between TiCl<sub>4</sub> and PSM02 was 1 : 0.01. These solutions were stirred at room temperature for 24 hours and the addition of deionized water led to a molar ratio between selected solution constituents TiCl<sub>4</sub> : EtOH : H<sub>2</sub>O = 1 : 30 : 10. The TiO<sub>2</sub> films were deposited onto silicon substrates by dip coating combined with the evaporation-induced self-assembly method. Calcination of the TiO<sub>2</sub> films was carried out at 450 °C for 15 min. The physico-chemical characterization of the catalysts were conducted using textural (BET), structural (XRD) and morphological (SEM/TEM) analyses. Photocatalytic efficiency of mesoporous TiO<sub>2</sub> thin films was tested in the decolorisation reactions of MB and CV over a broad range of initial concentrations and reaction cycles. Changes in concentrations of organic dyes were monitored by using UV/VIS spectra. Catalyst dosage in each photocatalytic test was 28±2 mg. The organic dyes, like methylene blue and crystal violet, were selected as model pollutants in photocatalytic test reaction. In order to establish the equilibrium of potential sorption process, thin films were tested during 12 h. The UV lamp (ROTH Co.) characterized with following working features: 16W, 2.5 mW/cm<sup>2</sup> and λ<sub>max</sub> = 366 nm was used.

### RESULTS AND DISCUSSION

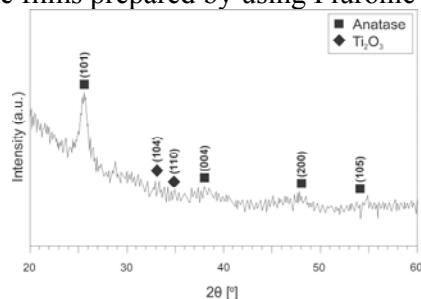
BET surface areas of TiO<sub>2</sub> films obtained by the dip coating technique in the presence of F127 and PSM02 templates after calcination at 450 °C are shown in Table 1. The characteristics of pore system in titania thin films can be related to the applied preparation method, used polymer template and titania precursor, calcination temperature and finally substrate surface properties. The specific surface area may have a significant effect on the photocatalytic efficiency of titania thin films [5]. The TiO<sub>2</sub> film templated with F127 exhibits BET surface area of 48 m<sup>2</sup>/m<sup>2</sup> and pore size up to 10 nm, while titania based thin film with PSM02 poses BET surface area of 20 m<sup>2</sup>/m<sup>2</sup> with pores of 22 nm in size (Table 1). In accordance with developed pore system and larger BET surface area, TiO<sub>2</sub> film templated with F127 has shown higher efficiency in photocatalytic decolorisation of selected organic dyes comparing with titania based film templated with PSM02 (Fig. 3). The XRD analysis (Fig. 1) indicated the presence of TiO<sub>2</sub> in anatase crystal

phase with a small amount of nonstoichiometric  $\text{Ti}_2\text{O}_3$ . The crystallite size of anatase phase, calculated by Scherrer's equation, was found to be 12 nm (Table 1). The XRD pattern of  $\text{TiO}_2$  film was recorded exclusively for the film of an adequate thickness to obtain reliable data ( $\text{TiO}_2$  - F127, thickness of 270 nm). It is expected that the crystal phase composition of  $\text{TiO}_2$  - PSM02 film, after the calcination at the same temperature (450 °C), is also crystalline and characterized with high amount of anatase crystal phase.

**Table 1.** Textural and structural properties of titania thin films

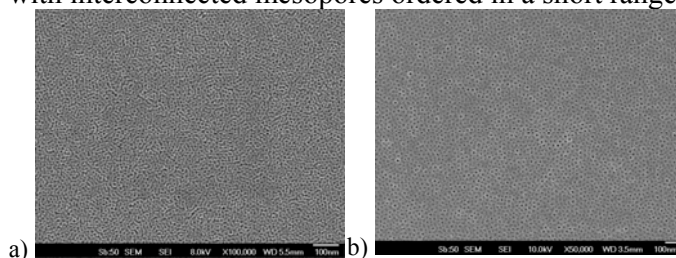
Photocatalyst – polymer	BET surface area ( $\text{m}^2/\text{m}^2$ )	Average pore size (nm)	Crystal phases	Crystallite size (nm)
$\text{TiO}_2$ – F127	48	10	$\text{TiO}_2$ anatase / $\text{Ti}_2\text{O}_3$	12
$\text{TiO}_2$ – PSM02	20	22	/	/

FE-SEM micrographs, the top view, show that  $\text{TiO}_2$  film prepared by using PSM02 polymer as a template, leads to formation of crack-free mesoporous  $\text{TiO}_2$  films with larger pores (20-22 nm) (Fig. 2b) compared to the films prepared by using Pluronic F127 polymer (8-10 nm) (Fig. 2a).



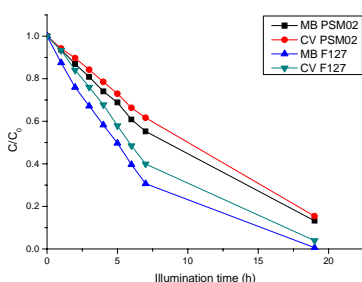
**Figure 1.**  
The XRD pattern of  $\text{TiO}_2$ -F127 thin film with thickness of 270 nm [5]

The images of cross-section of the  $\text{TiO}_2$  films prepared by using PSM02 polymer indicated the elliptical shape of pores, and a lower level of their interconnectivity. On the other hand, the polymer Pluronic F127 used for film preparation resulted with a cubic-like porosity in the titania film surface with interconnected mesopores ordered in a short range.



**Figure 2.**  
FE-SEM images of ordered mesoporous titania thin films templated with different polymer matrices: a) F127,

BET specific surface area, pore size and level of pores connectivity all together affect photocatalytic activity of titania thin films. It can be observed that photocatalytic decolorisation process of selected dyes is the most effective when thicker mesoporous TiO<sub>2</sub> films (higher catalyst dosage) with larger specific surface area (TiO<sub>2</sub>-F127, 48 m<sup>2</sup>/m<sup>2</sup>) was used. Also, under the same experimental conditions, the photocatalytic degradation of MB is faster compared to the degradation of CV (Fig. 3). Most likely, this effect is a consequence of different structures of the organic dyes.



**Figure 3.**  
Decolorisation kinetics of MB and CV using mesoporous titania thin films (TiO<sub>2</sub>-F127 and TiO<sub>2</sub>-PSM02)

## CONCLUSION

The dip coating technique was proved to be effective and relatively simple technique to obtain desired morphology of ordered mesoporous TiO<sub>2</sub> thin films. Complete degradation/decolorisation of the selected organic dyes was achieved. The obtained kinetic data of photodegradation reactions were correlated with the properties of TiO<sub>2</sub> thin films (specific surface area, pore size and shape, and crystallite size).

## ACKNOWLEDGEMENT

The authors wish to thank to the Ministry of Education and Science and Technological Development of the Republic of Serbia for a financial support (Projects: ON 172061 and III 45020).

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