Strain-modulated optical response in 2D MoSe₂ made by Na-assisted CVD on glass

Cite as: Appl. Phys. Lett. **120**, 213104 (2022); https://doi.org/10.1063/5.0090034 Submitted: 02 March 2022 • Accepted: 11 April 2022 • Published Online: 26 May 2022

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Lock-in Amplifiers up to 600 MHz





Appl. Phys. Lett. **120**, 213104 (2022); https://doi.org/10.1063/5.0090034 © 2022 Author(s).

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ABSTRACT

Extended investigations on 2D transition metal dichalcogenides (TMDCs) have opened sound possibilities to apply these materials in several technological fields such as sensing. To this end, fully reproducible methods for the wafer-scale production of crystalline and uniform 2D TMDCs are in demand. In this work, atomically thin MoSe₂ was grown by atmospheric-pressure chemical vapor deposition using the Na-assisted process with Se powder and Mo foil precursors on a glass substrate. The samples were extensively characterized via Raman and photoluminescence spectroscopy, atomic force microscopy, transmission electron microscopy, and x-ray photoelectron spectroscopy. The MoSe₂ samples consist of sub-millimeter, monolayer single-crystals with 2H phase configuration. Being monolayer and crystalline, the samples exhibit well-defined and intense photoluminescence. CVD-grown 2D MoSe₂ was integrated into a device with strain-tunable optical properties and tested. Under tensile strain (in the range of 0.2%-0.4%), the spectral emission responded to an in-plane strain with marked peak shifts toward lower energies for increasing levels of strain (~3 and ~2 nm shift for the main PL component at 0.2% and 0.4%, respectively), indicating a reduction of the bandgap.

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Two-dimensional (2D) transition metal dichalcogenides (TMDCs) are a class of layered nanomaterials suitable for a wide range of electronic devices due to a variety of thickness-dependent electronic and optical properties.¹⁻⁶ Structurally, a TMDC layer is composed of a plane of transition metal atoms (e.g., Mo and W) in between two planes of chalcogenide atoms (e.g., S, Se, and Te) held together by strong covalent bonds. Weak interlayer van der Waals forces bind these layers along the z-direction, forming a layered crystal. Monolayer MoSe₂ can be applied as a photoactive layer in optoelectronics due to its intrinsic direct bandgap of ~ 1.5 eV, which opportunely match the visible region of the solar spectrum (400–700 nm). Several kinds of photovoltaic cells^{7–10} and photodetectors¹¹⁻¹³ based on 2D MoSe₂ have been recently reported. Various techniques were proposed to change/modulate the physical properties of 2D MoSe₂, including doping, defect, and strain engineering, thus extending the range of application of this material. Strain engineering, in particular, allows fine-tuning of the electronic and optical properties of the 2D crystal by physically modifying its lattice and energy bands structure.¹⁴ First-principles calculations showed that

the bandgap of monolayer MoSe₂ should decrease by uniaxial strain (either compressive or tensile strain, 2%-10% range tested).¹⁵ Cheng et al. applied progressive strain (\pm 5%, from compression to tension) in monolayer MoSe₂ (produced by mechanical exfoliation) and measured the photoluminescence (PL): The optical bandgap was found to gradually decrease when the strain changed from compression to tension, and a direct-to-indirect transition occurred at a compressive strain of -1.3%. No changes were observed in the Raman spectrum under tensile strain, whereas evident changes occurred for the compressive strain.¹⁶ PL studies on mechanically exfoliated MoSe2 under uniaxial strain showed a nearly linear relationship between the strain and excitonic bandgap.^{17,18} Interestingly, fluctuations in the optical bandgap energy were reported in monolayer CVD MoSe₂, which could be caused by the random distribution of local tensile strain, due to surface roughness and different thermal expansion coefficients between MoSe₂ and the substrate.¹⁹ These effects could be potentially exploited for the fabrication of MoSe₂-based sensors and detectors. At present, only strain sensors (based on piezo-resistive effect) made with solution-processed MoSe₂

nanosheets were reported,^{20,21} while no reports on the integration of CVD $MoSe_2$ in a strain sensor configuration exist.

As other TMDCs, 2D MoSe₂ can be grown via CVD on Si/SiO₂ substrates by evaporation of two solid precursors (i.e., Se and MoO3 or MoO₂ powder) at suitable temperatures.^{22–25} In general, CVD methods often entail reproducibility issues associated with the fine-tuning and full control of the process parameters (i.e., pressure, temperatures, and gas flow as well as precursors and substrate conditions). Other than that, the lower chemical reactivity and higher melting point of Se with respect to S make it more challenging to grow 2D MoSe₂ than MoS₂.²¹ A CVD process to grow wafer-scale 2D MoSe₂ has yet to be established,^{23,27,28} but recent research on salt-assisted approaches showed a pathway to reduce the melting point of the Mo-containing precursor, while increasing the reaction rate and reducing the nucleation density.² In particular, the use of Na as a catalyst has been recently proposed to increase the growth rate and deposited area of several TMDCs³⁰ (including MoSe₂³¹). To this end, NaCl was mixed with the metal powder and used as a precursor in the growth of submillimeter MoSe₂ single-crystals.³² By a different approach, Na-containing glass was used as an "active" substrate (which released Na ions during the CVD) in the growth of MoS₂.^{33–35} This approach was then applied to MoSe₂, leading to mm-sized, monolayer single crystals on a 6-centimeter- long sodalime glass template (900 °C, ambient pressure).³⁶ Similarly, Chen et al. reported mm-sized, monolayer MoSe₂ single-crystals directly on molten soda-lime-silica glass (at 1050°C, ambient pressure).³⁷ Overall,

glass substrates with high amount of Na (~9.5 wt. %) appeared to promote the growth of uniform monolayer crystal.³⁸ Here, we have grown atomic-thin MoSe₂ by an innovative alkali metal-assisted approach using Se powder and Mo foil as precursors, and Na-containing glass as a substrate [Fig. 1(a)]. The approach led to monolayer MoSe₂ singlecrystals (sizes above 100 μ m) with a predominant 2H phase, showing intense PL emission. We optimized our "pick and place" method [Fig. 1(b), described in the Experimental section in the supplementary material] to transfer MoSe₂ crystals from glass to a flexible polydimethylsiloxane (PDMS) substrate. To this end, the PDMS substrate was exposed to water vapor to maximize the adhesion of the 2D crystal by capillary forces. When testing the CVD MoSe₂ crystals under uniaxial tensile strain, the PL emission was found to steadily shift toward lower energies for increasing levels of strain.³⁹

2D MoSe₂ samples were grown via atmospheric pressure CVD (AP-CVD), as described in the Experimental section in the supplementary material. By microscopic analysis, 5–150 μ m-sized polygonal single-crystals were grown [Fig. 2(a)]. They are most often isolated but coalesce into a film in some instances. This is likely due to concurrent nucleation sites, whose existence might be linked to the particle-like spots often found in the center of the crystals. Small particles surrounding the crystal edges are also visible [as indicated by the black arrows in the inset of Fig. 2(a)]. The AFM analysis shows that the MoSe₂ crystals are generally ~1 nm thick [Fig. 2(b)], which is a value in line with monolayer thickness (also considering that the glass surface might coarsen during CVD,



FIG. 1. (a) Schematic of the Na-assisted CVD process. (b) Sketch describing the pick and place method: (1) a viscoelastic PDMS support is exposed to water vapor for tens and lowered on top of the MoSe₂ crystal on glass. (2) After peeling the support from the glass substrate, the capillary forces allow one to "pick" the MoSe₂ crystal, which adheres to the PDMS surface. (3) By microscopic inspection, the PDMS/MoSe₂ stack is positioned on the desired location to "place" the crystal on the target substrate by differential adhesion. (4) The PDMS support is removed. This technique allows the dry-transfer of any 2D materials and is compatible with several types of substrates such as TEM grids.



FIG. 2. (a) Optical image of the monolayer $MoSe_2$ crystals on glass, showing sporadic thicker nucleation spots. The inset shows a magnified region of the sample, highlighting small particles on the crystal edges (black arrows). (b) AFM image of the MoSe₂ crystal and thickness measurement. (c) Raman spectra (normalized to the A_{1g} mode intensity) of the monolayer MoSe₂ crystal, and a few-layer nucleation spot. (d) PL spectrum of a monolayer MoSe₂ crystal.

making the measurement less accurate). The samples were further evaluated by Raman spectroscopy. MoSe2 usually shows two characteristic peaks related to out-of-plane (A_{1g}) and in-plane (E_{2g}^{-1}) vibration modes. Depending on the number of layers (i.e., passing from monolayer to bulk), these modes vary in position: from 240.5 to 242.5 cm^{-1} and from 287.2 to 286 cm⁻¹, respectively.⁴⁰ Figure 2(c) shows the vibrational modes in the range of interest. A main \bar{A}_{1g} mode at $240.5\,\text{cm}^{-1}$ stands out, which is indicative of monolayer thickness for the single-crystals. In the sporadic regions with secondary nucleation (often hosting a small particle), the A_{1g} mode blue-shifts to 241.7 cm⁻¹, pointing at a few-layer thickness. A very small peak at $\sim 250 \text{ cm}^{-1}$ [Fig. 2(c)] can be seen, which could be related to minor disorder such as Se vacancies or other crystallographic defects.⁴¹ A close-up analysis of the A_{1g} , E_{2g}^{-1} and B_{2g}^{-1} modes is given in Fig. S1. The PL spectrum of the MoSe₂ crystal [Fig. 2(d)] could be fitted with two components: a first one related to the neutral exciton A^0 and a second one at a lower energy (generally related to a trion).¹⁹ This is different from mechanically exfoliated MoSe₂, which usually shows a single, symmetric PL peak.^{17,18} The main component appears at \sim 784 nm and is slightly red-shifted with respect to mechanically exfoliated samples (790 nm).⁴⁰ The second component is located at \sim 807 nm. In terms of optical absorbance, the two expected excitonic peaks appear (A and B, Fig. S2). Their positions (~785 and 690 nm) are also redshifted

with respect to the literature (~794 and 700 nm).⁴² The red-shifts observed in PL and optical absorbance peaks could be connected to a minor strain in the MoSe₂ crystals, induced by the heating/cooling process of the glass substrate.¹⁶ The MoSe₂ crystals on glass were analyzed by x-ray photoelectron spectroscopy (XPS) to ascertain the elemental composition [Fig. S3(a)]. Figure S3(b) shows the high-resolution spectrum of Mo3d, which consists of two main peaks with binding energies of ~228.5 and ~231.7 eV (corresponding to Mo3d_{5/2} and Mo3d_{3/2}, respectively). Figure S3(c) shows the high-resolution spectrum of Se3d with two peaks at ~54.1 and ~55 eV corresponding to divalent Se ions (Se 3d_{5/2} and 3d_{3/2}, respectively). Interestingly, the position and width of the main peaks in the Mo3d and Se3d spectra confirm the presence of the sole 2H phase (no 1T phase). These characteristics are in line with the literature on CVD MoSe₂.^{12,22,43,44}

A detailed TEM analysis was employed to ascertain the atomic structures of the MoSe₂ crystals. Figure 3(a) shows triangular-shaped MoSe₂ crystals transferred on TEM grids. Figure 3(b) shows a high-angle annular dark-field imaging (HAADF) scanning transmission electron microscopy (STEM) image of their inner structure. The fast Fourier transformation (FFT) shown in the inset confirms the single-crystal form of MoSe₂. A few defects can be found in the crystals, as shown within the white square in Fig. 3(c). These triangular voids can appear in either



FIG. 3. (a) HAADF STEM image highlighting a triangular MoSe₂ crystal with lateral size of \sim 6.5 μ m. (b) HAADF STEM image of the crystal in (a) and corresponding FFT (inset), confirming the crystalline structure of MoSe₂. (c) HAADF STEM image showing two defects in the single crystal (dark triangular regions). The region within the white square is depicted in detail in (d) which shows a HAADF STEM image of a triangle-shaped defect (monolayer) in bilayer MoSe₂ with 2H stacking. (e) HAADF STEM image of a triangle-shaped defect in monolayer MoSe₂, next to a bilayer with 3R stacking. (f) Higher magnification image of (e), showing a hole in monolayer MoSe₂.

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monolayer [Figs. 3(d) and 3(e)] or bilayer [with 2H stacking, Fig. 3(f)] regions of the crystal.²⁷ In the CVD of 2D TMDCs, molten salts as NaCl are known to increase the reaction rate by reducing the melting point of the metal precursors and increasing the mass flux of vaporized precursors.³⁰ By our CVD approach (i.e., using a Mo foil and a glass substrate), small amounts of Na are released from the glass and interact with the Mo foil, inducing corrosion: This, in turn, would increase the reactivity and volatilization of Mo atoms.^{45–47} The Mo foil precursor, in lieu of the Mo oxide powder, thus, assures a uniform Mo supply, fostering a large-scale growth as previously observed for MoS₂.^{35,48} Overall, in contrast with common two-powder precursor approaches, our growth method offers a much higher control on the 2D crystal formation, which ends up in atomic-thick, polygonal (e.g., triangular and hexagonal) crystals.

Following the transfer of monolayer MoSe₂ crystals to a flexible PDMS substrate, we investigated their optoelectronic properties under uniaxial tensile strain using a custom-made setup [Fig. 4(a)]. Figure 4(b) shows the MoSe₂ crystal under test. Raman and PL measurements were acquired in a strain range 0%-0.4% (estimated as reported in Ref. 49). The Raman spectra remain unaltered regardless of the applied strain: In all cases, the A_{1g} peak vibration is centered at 240.5 cm⁻¹, as expected for monolayer MoSe₂ [Fig. 4(c)].¹⁶ The situation is different for the PL, as shown in Fig. 4(d). The PL spectra of MoSe₂ on PDMS are broader than those acquired on glass, mostly due to a significant FWHF increase in the second component (\sim 57 vs \sim 21 nm). This could be due to defects induced in the crystal by the dry-transfer process.⁵⁰ After transfer to PDMS, the first component remains in the same position (\sim 784 nm), while the second red-shifts to \sim 807 nm (vs 797 nm). The second component appears sensitive to the mechanical deformation: This could be explained by a stress relaxation occurring after transferring the crystals from the native glass to PDMS. At a 0.2%



FIG. 4. (a) Photograph of the custom-made setup used to apply external strain to the MoSe₂ crystals. (b) Optical microscope image of the as-transferred MoSe₂ crystal on PDMS. (c) Raman and (d) PL plots of the MoSe₂ crystal acquired at different strain levels.

strain, a red-shift in both components is observed with the second one experiencing again the larger shift (+3, +16 nm, respectively). At a 0.4% strain, a blue-shift in the components take place (-1, -7 nm, respectively). Finally, when recovering to a flat state, a further slight blue-shift occurs (-2, -2 nm): Overall, the first component goes back to the initial value, while the second stays red-shifted by 7 nm.

The observed shifts in PL peak positions can be attributed to tension-induced change in the lattice parameter, which modifies the electronic band structure and bandgap energy. In mechanically exfoliated, monolayer MoSe₂, a 1% of strain was reported to reduce the bandgap energy by $\sim 30-40$ meV.^{17,18} Assuming a linear behavior, a 0.2% strain would translate in a reduction of 6-8 meV, which match the 6.7 meV reduction we measured in the main PL component (equivalent to the 3 nm shift).^{16,18} In summary, our innovative Naassisted route (using Se powder and Mo foil precursors) led to the growth of monolayer MoSe₂ single-crystals with sizes above $100 \,\mu m$ on glass. We transfer the crystals to desired locations on flexible PDMS by a pick and place method without compromising their structural integrity and optoelectronic properties. In this way, we could test the photoluminescence of the crystals under uniaxial tensile strain, highlighting the peak shift at different strain levels. These results lay the ground in view of 2D MoSe₂-based strain sensing devices.

See the supplementary material for experimental procedures, detailed Raman analysis, optical absorbance, and XPS of the 2D MoSe₂.

We acknowledge the financial support of the project "GEMIS— Graphene-enhanced Electro Magnetic Interference Shielding" with Reference No. POCI-01-0247-FEDER-045939, co-funded by COMPETE 2020—Operational Programme for Competitiveness and Internationalization and FCT—Science and Technology Foundation, under the Portugal 2020 Partnership Agreement, through the European Regional Development Fund. This work was supported by FCT, through IDMEC-Mechanical Engineering Institute, under LAETA-Associate Laboratory of Energy, Transports and Aeronautics, Project No. UIDB/50022/2020 and via the Strategic Funding UIDB/04650/2020. We thank Dr. S. Sadewasser and Mr. B. Baumgartner for their assistance in preliminary experiments.

AUTHOR DECLARATIONS

Conflict of Interest

The authors have no conflicts to disclose.

DATA AVAILABILITY

The data that support the findings of this study are available from the corresponding author upon reasonable request.

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