



ASME

ES-SHTC 2022

16th International Conference
on Energy Sustainability
Summer Heat Transfer Conference

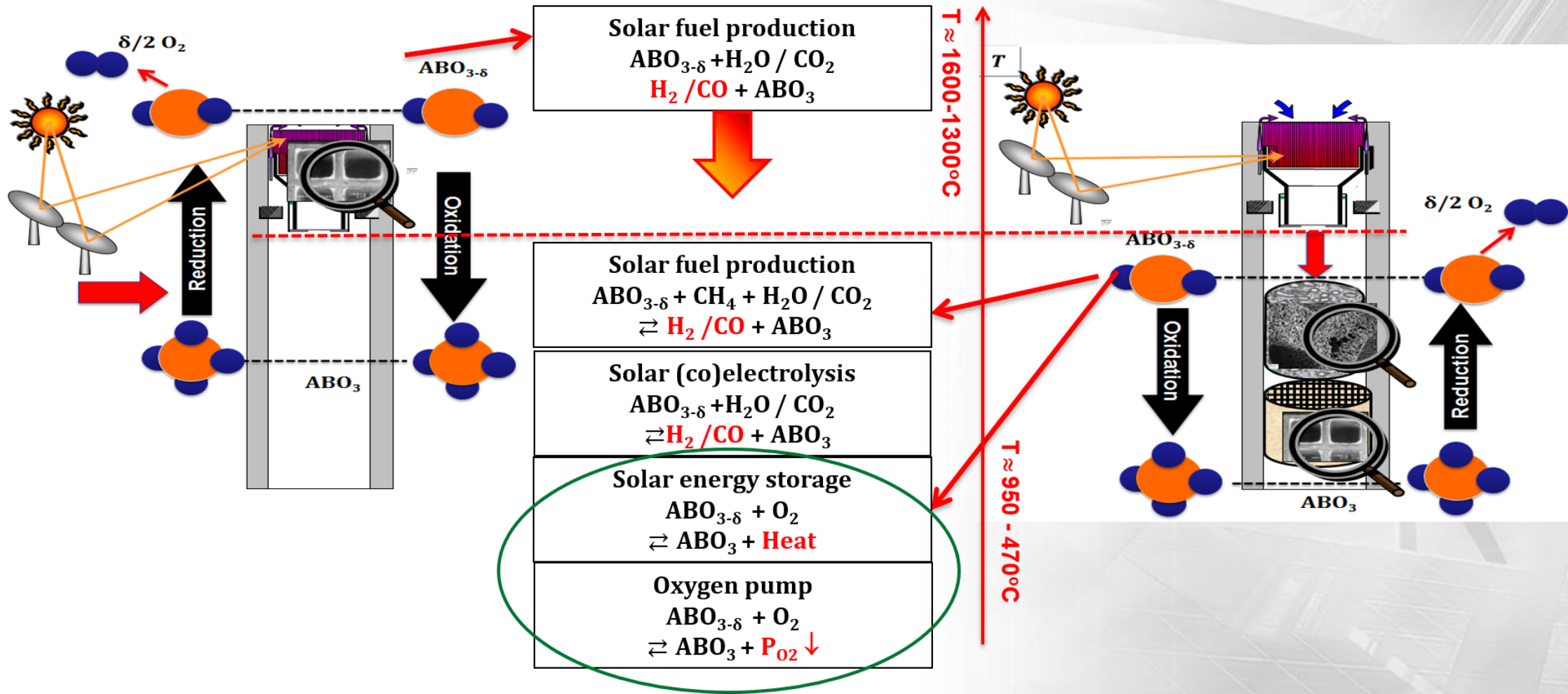
CONFERENCE JULY 11–13

Monolithic Porous Structures of Ca-Mn-Based Perovskites for Thermochemical Energy Conversion

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Motivation – Perovskites in thermochemical cycles





Motivation - CaMnO_3 as storage material

- $\Delta H = 161 \text{ kJ/mol}^1$
- Significant reduction in air $>760^\circ\text{C}$
- Cheap, abundant, non-toxic components
- Scalable production (solid state synthesis)
- Is considered as candidate for thermochemical energy storage
 - $>0.8 \text{ kJ/kg}\cdot\text{K}$ of storage capacity for Sr-doped CaMnO_3^2

¹ B. Bulfin et al., *J. Mater. Chem. A*, 2017,5, 7912-7919

² P. Thiel et al., *Journal of Applied Physics*, 2013, 114, 243707

Motivation – Open porous structures in TCES

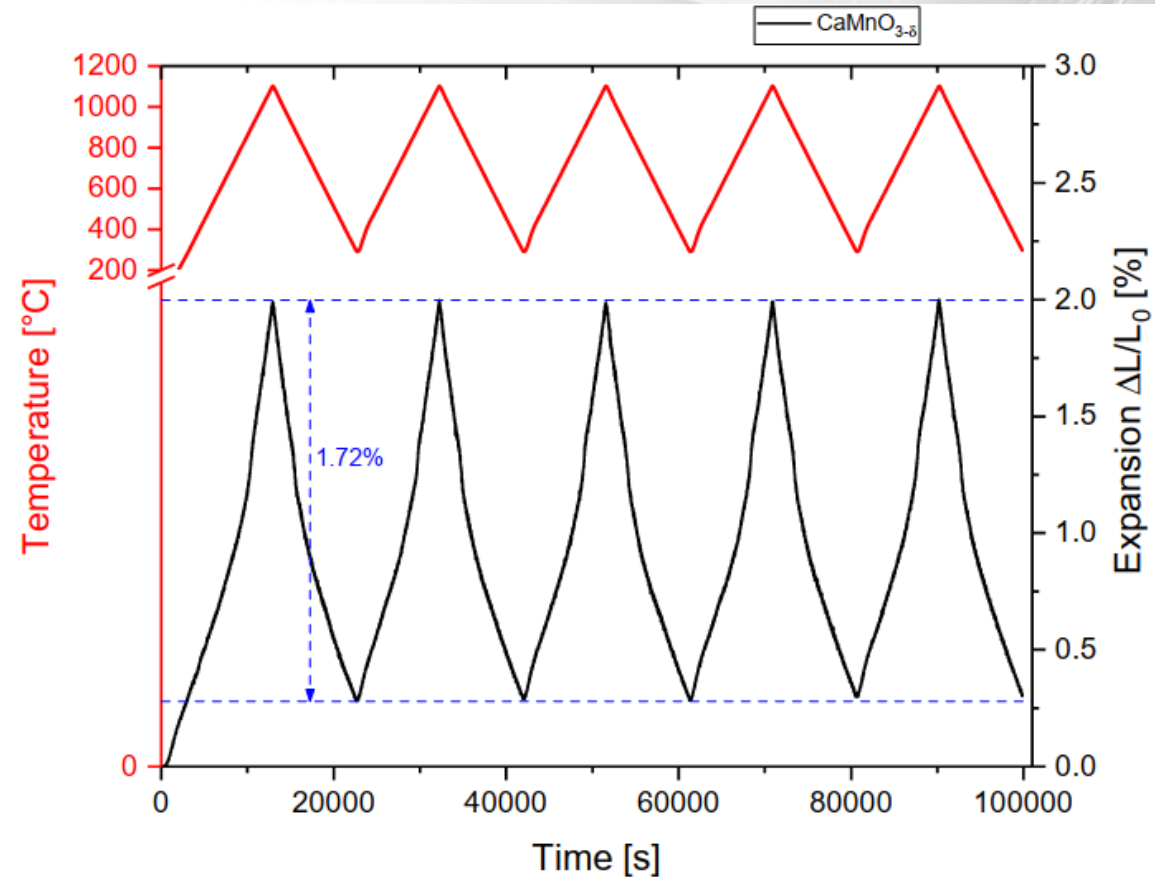
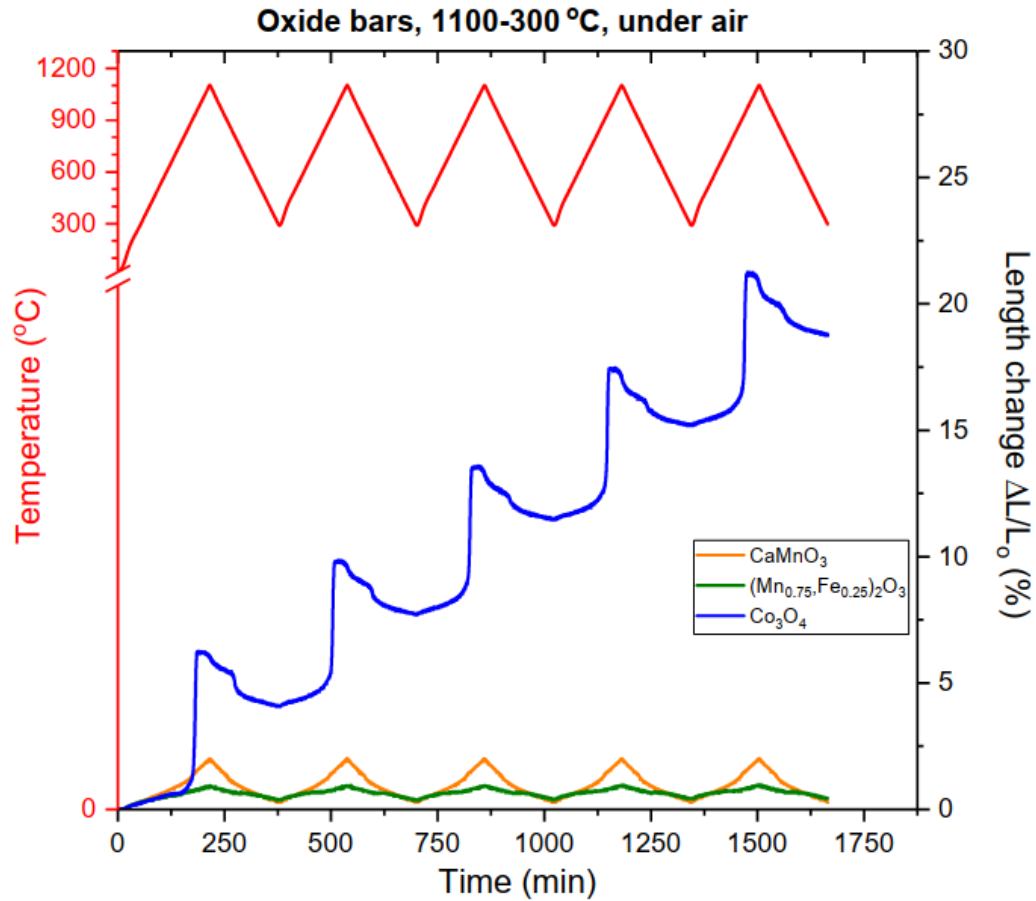


- Allows for higher mass flows of HTF compared to packed bed reactors
- No moving hot solids
- 50 kWh_{th} reactor built in previous project
- Co₃O₄ coated on Cordierite honeycombs
- Monolithic structures from stoichiometric oxides were not stable
- Thermal expansion / contraction causing too much stress

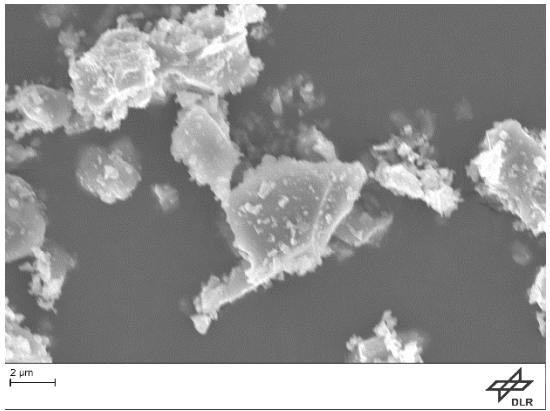
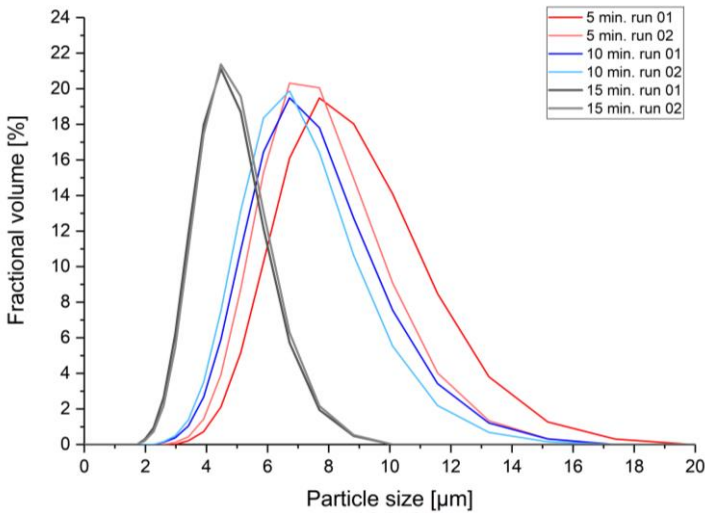
S. Tescari, A. Singh, C. Agrafiotis, L. de Oliveira, S. Breuer, B. Schlögl-Knothe, M. Roeb, C. Sattler, Experimental evaluation of a pilot-scale thermochemical storage system for a concentrated solar power plant, Applied Energy, Volume 189, 2017



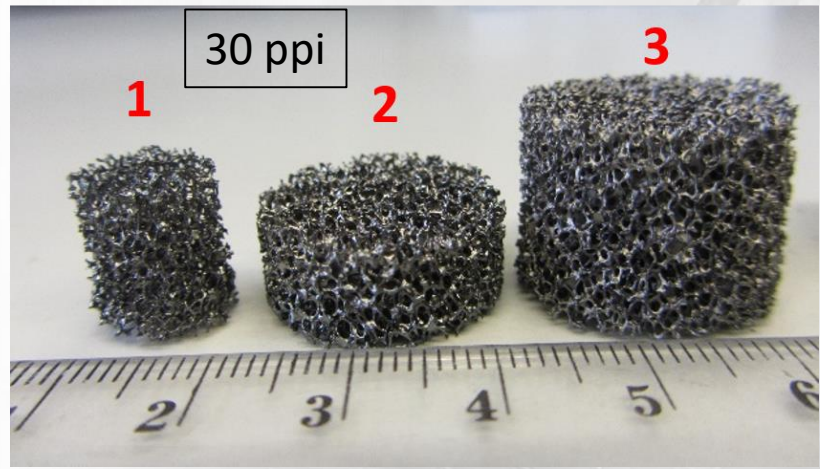
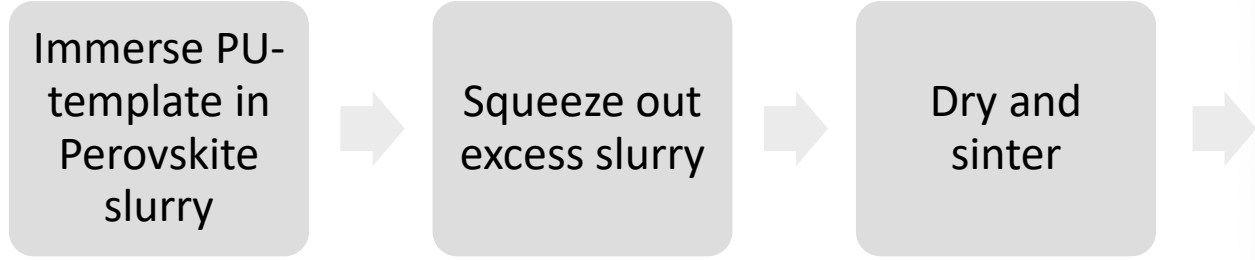
Cyclic thermal expansion



Fabrication of monolithic foams via replica method



Ingredient	wt.-%
Perovskite powder	59
Binder	6
Dispersing Agent	1
H ₂ O	34

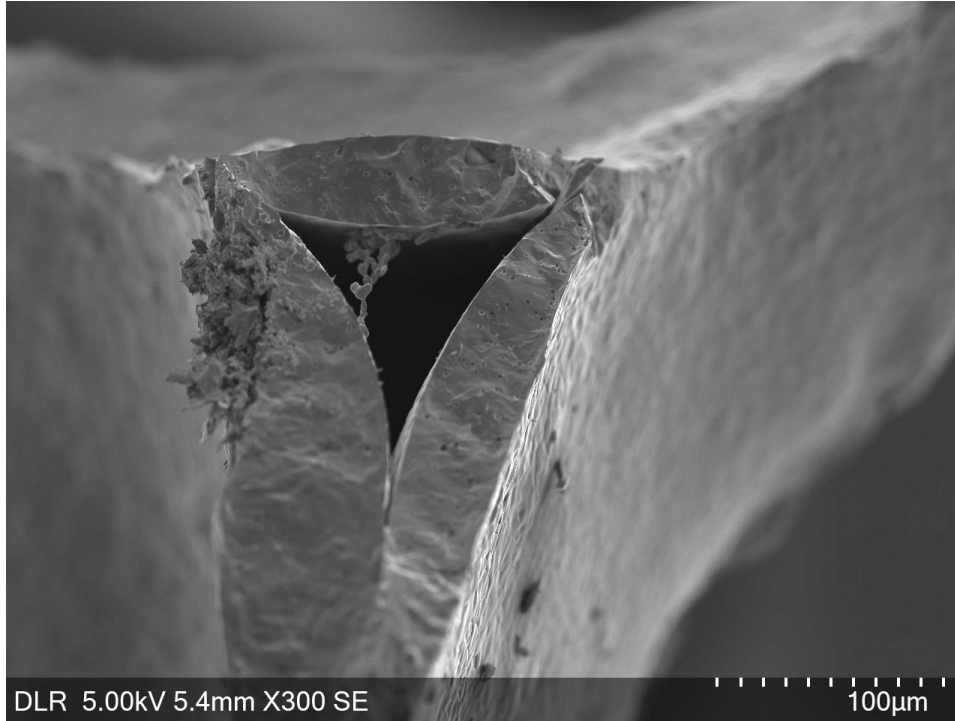


1: Ø1x1.2 cm; 2: Ø2x1 cm; 3: Ø2x2 cm

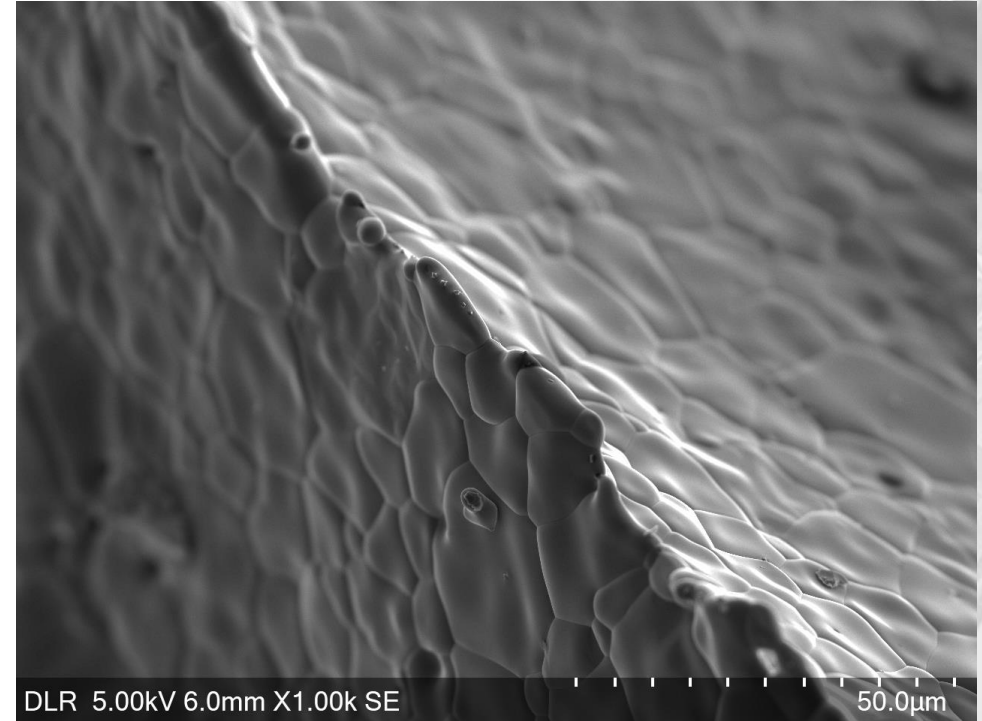
Pein et al. "Reticulated Porous Perovskite Structures for Thermochemical Solar Energy Storage", *Advanced Energy Materials* (2022), p. 2102882. DOI: <https://doi.org/10.1002/aenm.202102882>.



Microstructure of CaMnO_3 -specimen



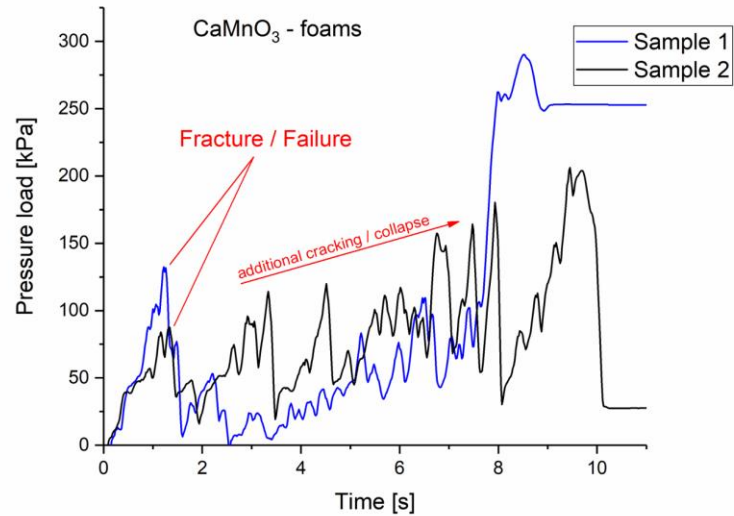
- Struts are $\approx 100\mu\text{m}$ thick
- Struts appear hollow to some degree
- Smooth surface



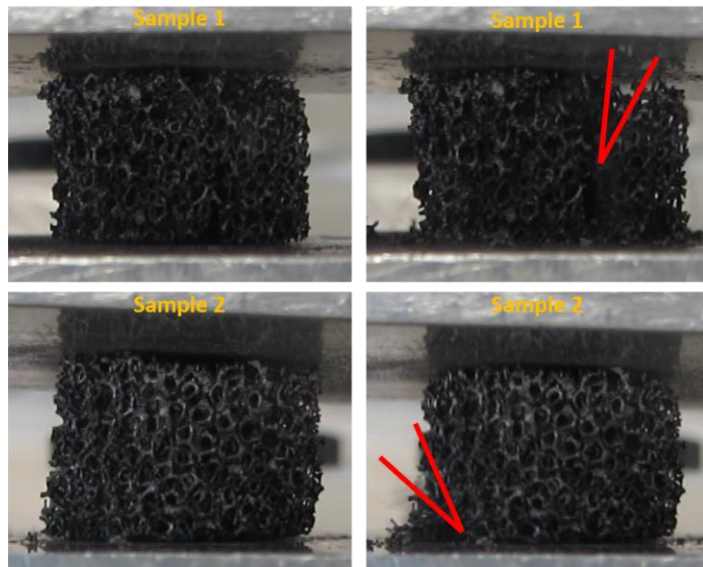
- Triangular form of struts
- Sharp edges



Stability of CaMnO₃-specimen



- Samples withstand pressure load of approx. 1 bar
- Easy to handle and stackable
- No deformation upon cycling
- Scalability tested up to 10x10x5cm block

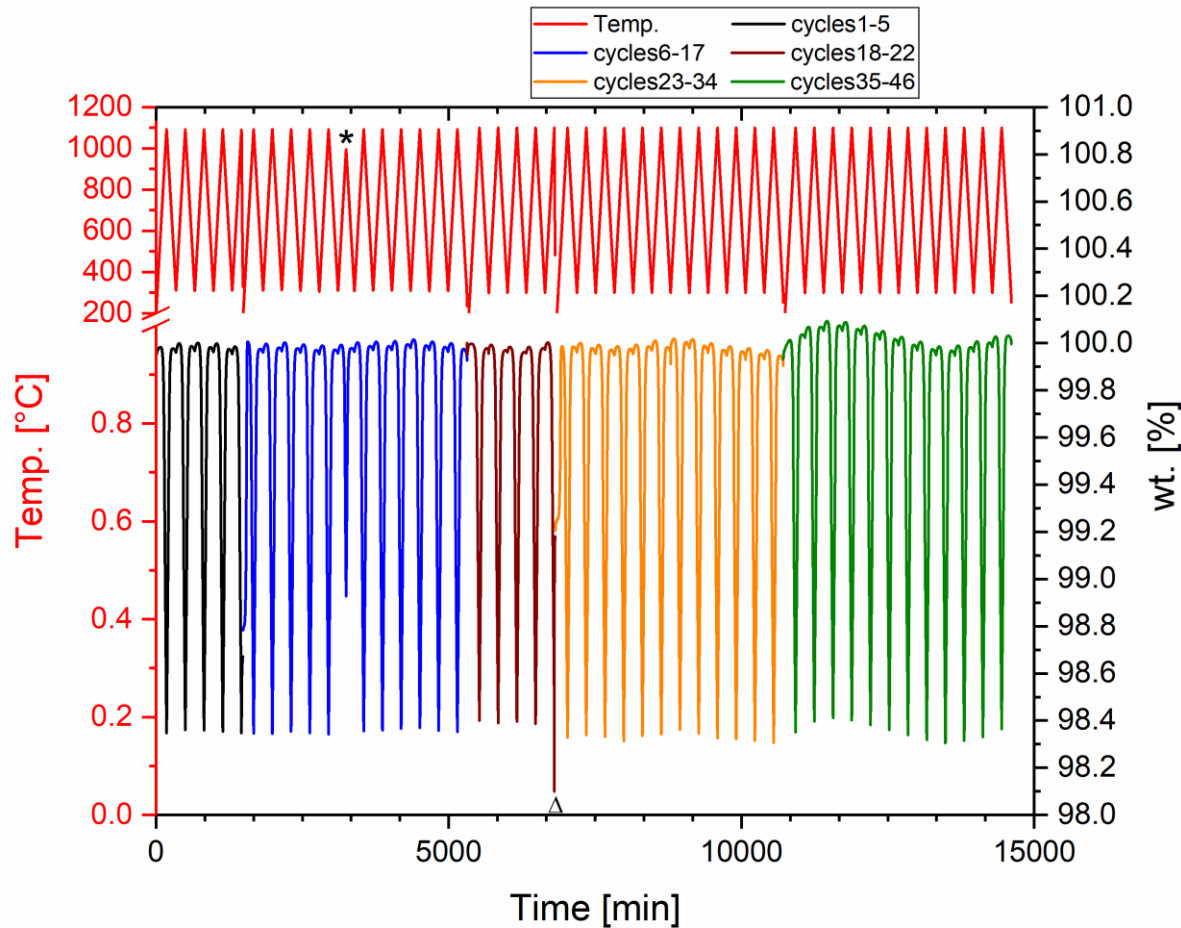


Pein et al. "Reticulated Porous Perovskite Structures for Thermochemical Solar Energy Storage", *Advanced Energy Materials* (2022), p. 2102882. DOI: <https://doi.org/10.1002/aenm.202102882>.





Cyclic stability



before

after

- Stable thermodynamics over 46 cycles 300-1100°C
- No deformation of the foam sample

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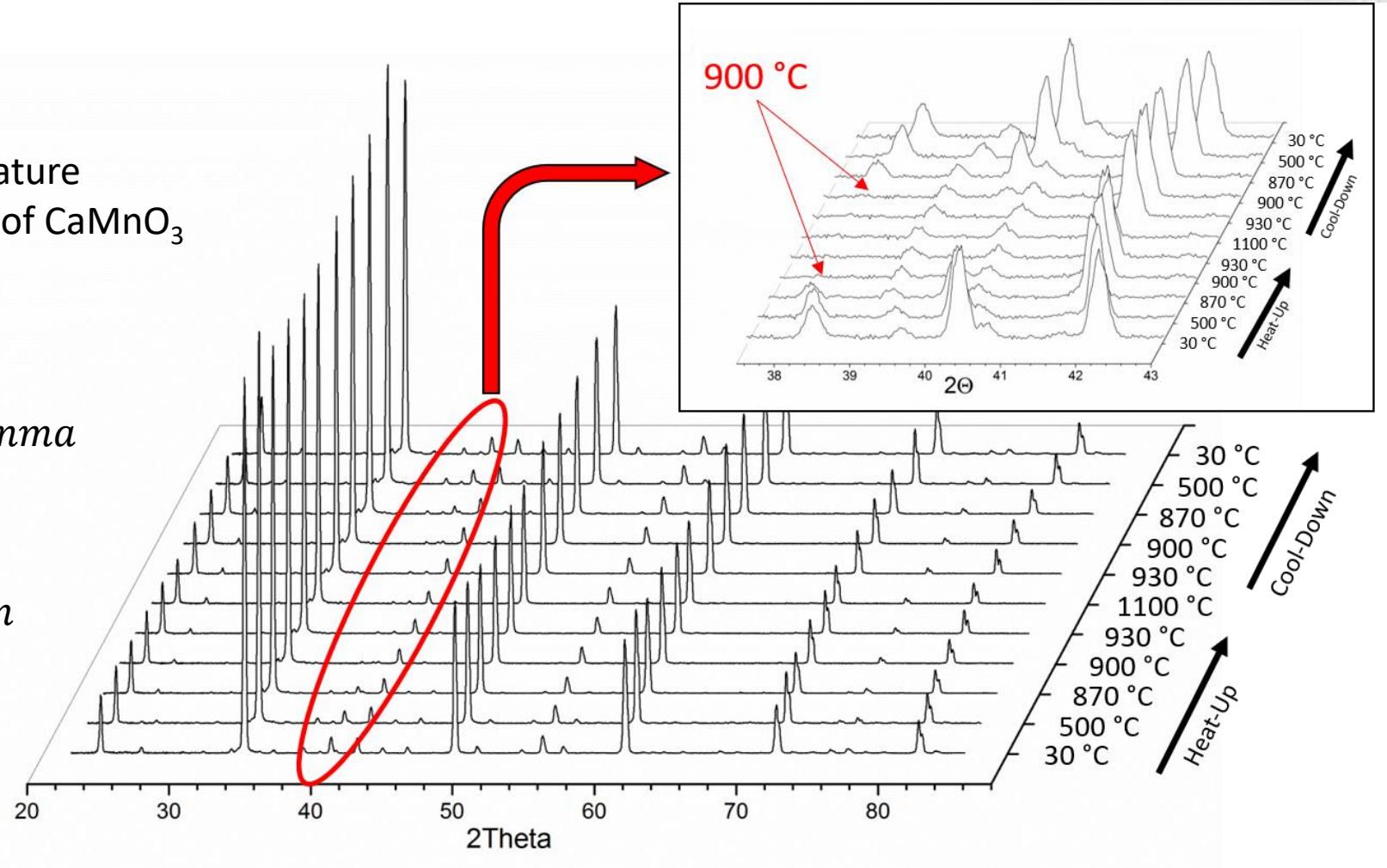
High temperature phase transition

High temperature powder XRD of CaMnO_3

Orthorhombic - $Pnma$



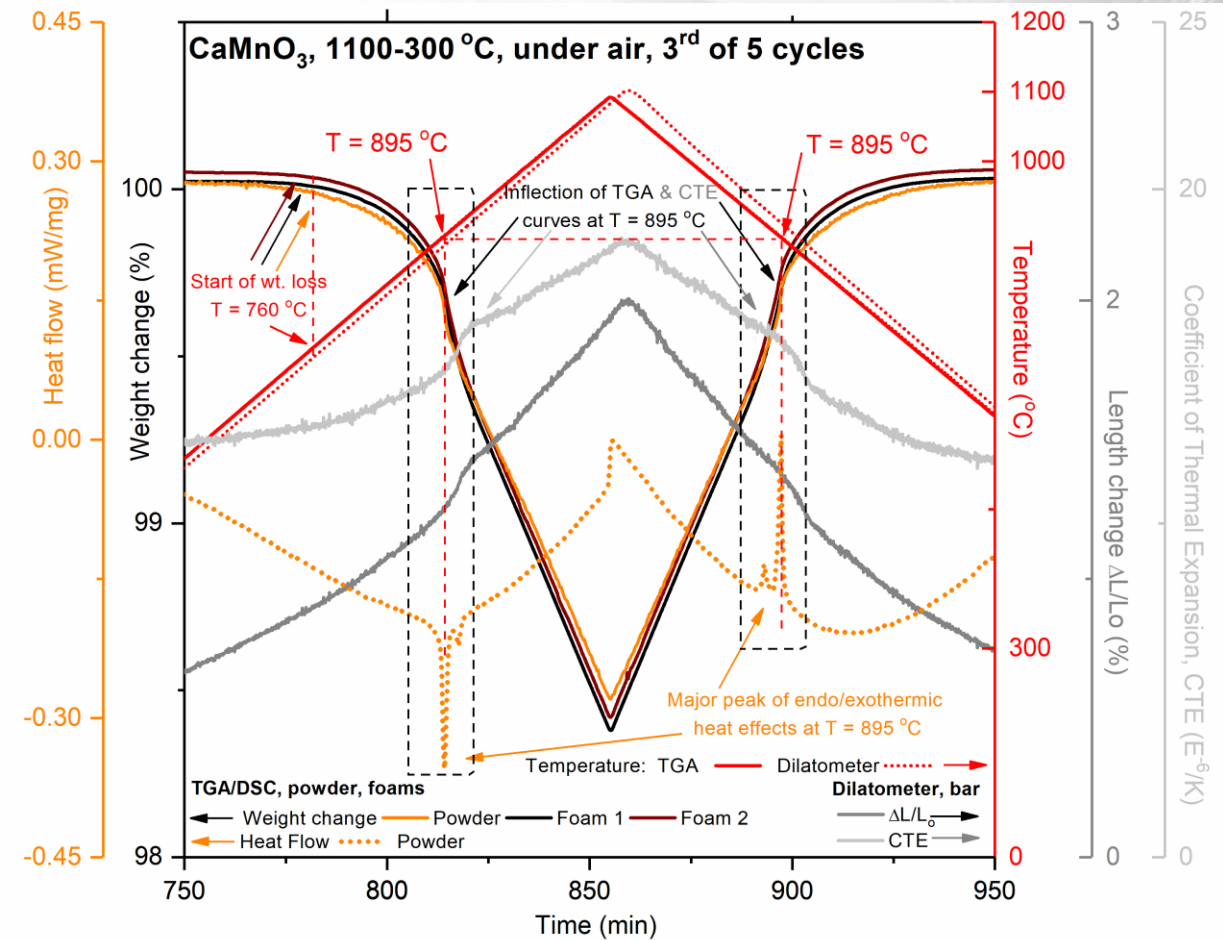
Cubic - $Pm\bar{3}m$



Source: Pein et al. "Reticulated Porous Perovskite Structures for Thermochemical Solar Energy Storage", *Advanced Energy Materials* (2022), p. 2102882. DOI: <https://doi.org/10.1002/aenm.202102882>.

TGA/DSC phase transition

- Phase transition (PT) fully reversible in air
- Heat effects and weight loss/gain at T_{pt}
- PT leads to lattice expansion
- Cubic phase exhibits greater thermal expansion coefficient $\alpha_{cubic} > \alpha_{orthorhombic}$
- Cubic phase has lower formation energy of oxygen vacancies
- Foam and powder exhibit identical thermodynamic behavior



Source: Pein et al. "Reticulated Porous Perovskite Structures for Thermochemical Solar Energy Storage", *Advanced Energy Materials* (2022), p. 2102882. DOI: <https://doi.org/10.1002/aenm.202102882>.

Outlook: Influence of A-site Sr-substitution in $\text{Ca}_{1-x}\text{Sr}_x\text{MnO}_3$

Increasing Sr-content

- Phase transition *orthorhombic* \leftrightarrow *cubic* shifts towards lower temperatures
- Heat effects of phase transition get smaller and eventually vanish ($x \geq 0.2$)
- Oxygen vacancy formation energy reduced
- Faster oxidation kinetics
- Small amounts of Sr-content stabilize thermal expansion/contraction at reduced $p(\text{O}_2)$





Outlook 2: Oxygen Pumping with foam specimen

- Sr-substituted CaMnO₃-Specimen with 0%, 5% and 10% Sr-content tested in lab scale demonstrator
- Oxygen purposefully pumped from reference chamber in Temp.-swing operation
- ≈ 2-fold increase in $\Delta\delta$ of reference material CeO₂
- Experimental campaign ongoing

O₂-pumping reactor:

T_{Red:} 800°C

T_{ox:} 700°C

Reference reactor:

T_{Red:} 1500°C

T_{ox:} 1000°C

Operating total pressure: 10⁻⁵ bar



Conclusion

- Mechanically stable open porous monolithic foams from CaMnO_3
- Foams exhibit identical thermodynamic behavior as powder
 - No ageing observed over 46 cycles
- Fully reversible thermal expansion and contraction between 300-1100°C under air
- A-site Sr-substitution: production successfully transferred
 - Lowers T_{pt} and reduces heat effects
 - Reduced mechanical stress during thermal cycling around T_{pt}
- Foam specimen proven to work in oxygen pumping demonstrator reactor



Thank you for your attention

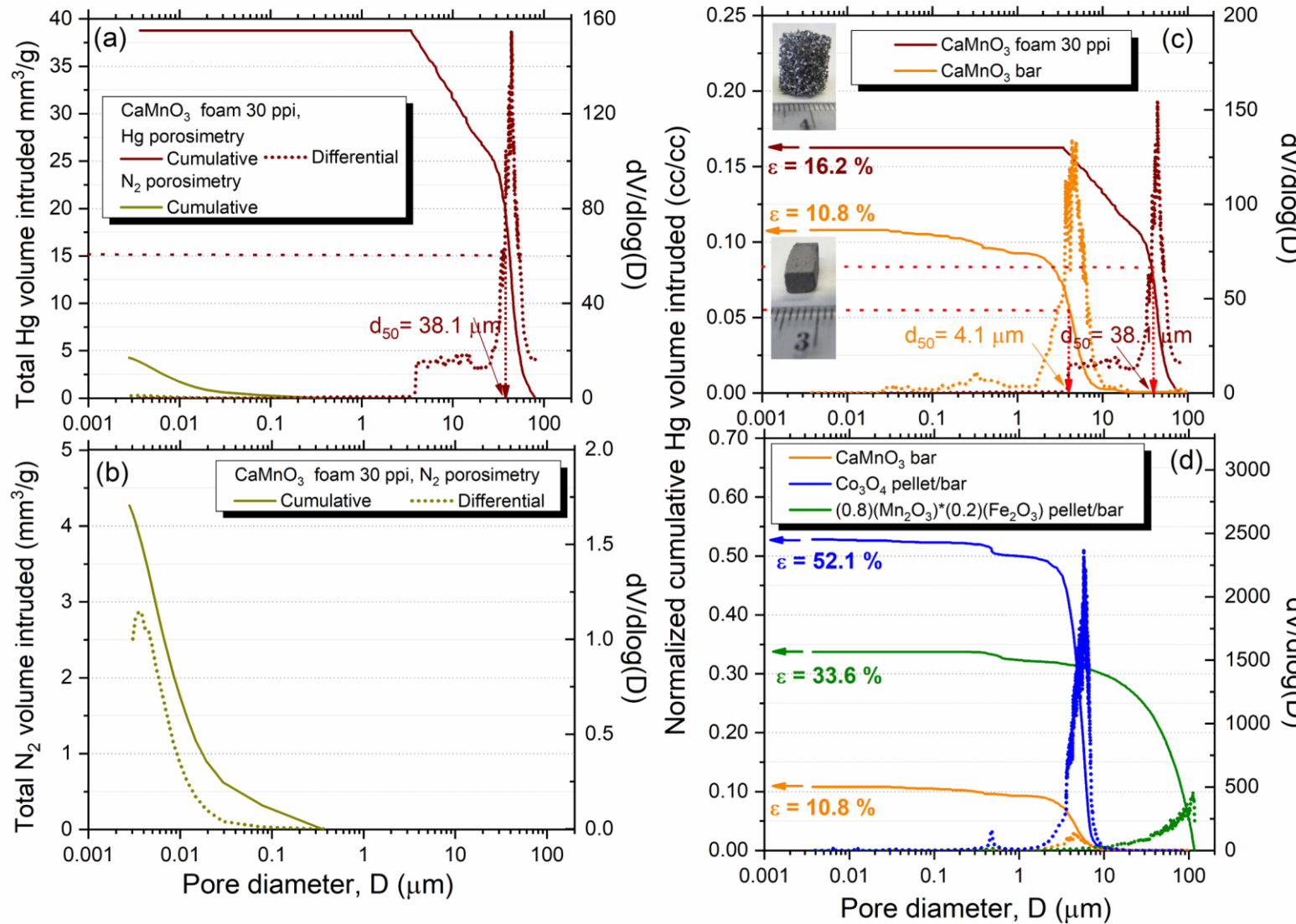
Further information:

- Pein et al. “Reticulated Porous Perovskite Structures for Thermochemical Solar Energy Storage”, *Advanced Energy Materials* (2022), p. 2102882. DOI: <https://doi.org/10.1002/aenm.202102882>.

Thanks to all colleagues for contributing to this work



Porosity of CaMnO₃-specimen



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O₂-pumping setup

