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Antispasmodic Saponins from Bulbs of Red Onion, *Allium cepa* L. Var. Tropea

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A phytochemical analysis of the polar extract from the red bulbs of *Allium cepa* L. var. Tropea, typical of Calabria, a southern region of Italy, was performed extensively for the first time, leading to the isolation of four new furostanol saponins, named tropeoside A1/A2 (1a/1b) and tropeoside B1/B2 (3a/3b), along with the respective 22-*O*-methyl derivatives (2a/2b and 4a/4b), almost certainly extraction artifacts. High concentrations of ascalonicoside A1/A2 (5a/5b) and ascalonicoside B (6), previously isolated from *Allium ascalonicum* Hort., were also found. This is the first report of furostanol saponins in this *A. cepa* variety. The chemical structures of the new compounds were established through a combination of extensive nuclear magnetic resonance, mass spectrometry and chemical analyses. High concentrations of quercetin, quercetin 4¹-glucoside, taxifolin, taxifolin 7-glucoside, and phenylalanine were also isolated. The new saponins were found to possess antispasmodic activity in the guinea pig isolated ileum; such an effect might contribute to explaining the traditional use of onion in the treatment of disturbances of the gastrointestinal tract.

KEYWORDS: Alliaceae; red onion; saponins; tropeosides A and B; flavonoids; antispasmodic activity

INTRODUCTION

Onion (*Allium cepa* L.) is one of the oldest cultivated plants, and it is now used both as a food and for medical purposes. In fact, onion is a rich source of a number of phytonutrients, which make it an important element of the Mediterranean diet. It is also useful for the treatment or prevention of a number of diseases, including cancer, coronary heart disease, obesity, hypercholesterolemia, type 2 diabetes, hypertension, cataract, and disturbances of the gastrointestinal tract (e.g., colic pain, flatulent colic, and dyspepsia) (1).

The history of the onion is well documented and can be traced to its origin in a wide area from India to Israel, where its production started in 3000 B.C., and then it was introduced in Europe by the Phoenicians around 2000 years ago. In ancient Egypt, the onion was believed to be a sacred food, and it was also widely consumed by the Romans and Greeks, who liked its taste and knew of its curative properties (2). Accordingly, Pliny the Elder, in his *Naturalis historia*, lists 30 ailments that can be treated by onions and pointed out that any dishes containing the onions are, to a certain extent, curative as well as more nourishing and tasty. Epidemiological studies have consistently shown an inverse association between consumption of vegetables (onion included) and the risk of human diseases. Notably, a population-based case-control study showed that the consumption of *Allium* vegetables was associated with a reduced risk of prostate cancer (3). These observations prompted us to continue the phytochemical investigation into this prolific source of interesting compounds.

The red onion, *A. cepa* L. var. Tropea, is a typical Italian variety, cultivated in some areas of southern Italy (Calabria region). This variety, characterized by both white and purple flowers, is known for its distinctive bulb, lengthened or oval, red, and sweet. Analysis of the MeOH extract from the bulbs of Tropea onions revealed the presence of high concentrations of flavonoids and saponins. Among these, four saponins, named tropeosides A1/A2 (**1a/1b**) and B1/B2 (**3a/3b**), appeared to be new compounds.

Plant-derived saponins present a broad spectrum of biological uses, such as anti-cancer, anti-inflammatory, ion channel blocking, immune stimulating, antifungal, antithrombotic, and hypocholesterolemic properties (4, 5). Due to the traditional use of onion in the treatment of gastrointestinal tract disturbances, the isolated saponins have been tested for the first time to evaluate an eventual antispasmodic activity on the guinea pig isolated ileum, and the results will be discussed herein.

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MATERIALS AND METHODS

General Methods. Optical rotations were determined on a Perkin-Elmer 192 polarimeter equipped with a sodium lamp (589 nm) and a 10-cm microcell. ¹H and ¹³C NMR spectra were recorded at 500 and 125 MHz, respectively, on a Bruker AMX-500 spectrometer. Chemical shifts were referred to the residual solvent signal (CD₃OD: $\delta_{\rm H}$ 3.31, $\delta_{\rm C}$ 49.0). The multiplicities of ¹³C NMR resonances were determined by DEPT experiments. ¹H connectivities were determined using COSY and HOHAHA experiments; the 2D HOHAHA experiments were performed in the phase-sensitive mode (TPPI) using the MLEV-17 (mixing time of 125 ms) sequence. One-bond heteronuclear ¹H-¹³C connectivities were determined with a 2D HSQC pulse sequence with an interpulse delay set for $^1\!J_{\rm CH}$ of 130 Hz. Two- and three-bond heteronuclear ¹H-¹³C connectivities were determined with 2D HMBC experiments, optimized for 2-3JCH of 8 Hz. Nuclear Overhauser effect (NOE) measurements were performed by 2D ROESY experiments. High-resolution FAB mass spectra (glycerol matrix) were measured on a Prospect Fisons mass spectrometer. Medium-pressure liquid chromatography (MPLC) was performed on a Büchi 861 apparatus using LiChroprep RP-18 (40-63 μ m) columns. HPLC in isocratic mode was performed on a Varian apparatus equipped with an RI-3 refractive index detector using a Waters semipreparative 300×7.5 mm i.d. μ -Bondapack C₁₈ column (injection amount = 180 μ L, flow rate = 2.5 mL/min) and an analytical 300 \times 3.9 mm i.d. μ -Bondapack C₁₈ column (injection amount = 100 μ L, flow rate = 0.8 mL/min).

Plant Material, Extraction, and Isolation. Cultivated bulbs of A. cepa L. var. Tropea were collected in Tropea (Calabria, Italy) in April 2003. A voucher specimen (no. 2003ACT1) has been deposited at the Dipartimento di Scienze e Tecnologie Agroalimentari, Ambientali e Microbiologiche, Campobasso. Fresh bulbs (5.05 kg) were finely handcut and then exhaustively extracted, at room temperature, first with CHCl₃/MeOH (9:1) and then with pure MeOH. Each solvent extraction was conducted for several days and was repeated four times using 4 L of solvent, under stirring. The MeOH extract (241 g) was partitioned between water and BuOH to remove sugar compounds and water, and the organic layer was then filtered and concentrated in vacuo to afford a crude extract (44.7 g), which was chromatographed by MPLC on an RP-18 column, using a linear gradient from H₂O to MeOH. Preliminary NMR studies revealed that fractions eluted from H₂O/MeOH (7:3) to H₂O/MeOH (4:6) contained flavonoids in high concentrations and fractions eluted from H₂O/MeOH (3:7) to H₂O/MeOH (1:9) contained saponins as major metabolites.

Fractions eluted in H₂O/MeOH (7:3) (231.6 mg) and H₂O/MeOH (6:4) (551.3 mg) yielded the known taxifolin 7-glucoside (29.5 mg) and taxifolin (490.4 mg), respectively. This last fraction also contained the amino acid phenylalanine in relatively high amounts (50.1 mg). Fractions eluted in H₂O/MeOH (1:1) (215.3 mg) and H₂O/MeOH (4: 6) (501.6 mg) yielded the known quercetin 4^{I} -glucoside (140.5 mg) and the flavonol quercetin (511.8 mg), respectively.

The fraction eluted in H₂O/MeOH (3:7) (271.2 mg) yielded the known saponin ascalonicoside A1/A2 (**5a/5b**, 235.9 mg) (**Figure 1**), whereas ascalonicoside B (**6**, 250.6 mg) was found in the fraction eluted in H₂O/MeOH (2:8) (435.1 mg) (6). The last fraction, H₂O/MeOH (1: 9), 375.3 mg, was further purified by HPLC on a semipreparative C₁₈ column with a mobile phase of H₂O/MeOH (3:7), affording the mixtures of new saponins tropeoside A1/A2 (**1a/1b**, 67.4 mg, $t_R = 11.0$ min) and tropeoside B1/B2 (**3a/3b**, 106.4 mg, $t_R = 14.6$ min). Each mixture was separately purified on an analytical C₁₈ column, mobile phase H₂O/MeOH (35:65), affording pure tropeoside A1 (**1a**, 26.4 mg, $t_R = 5.7$ min), tropeoside A1 (**1b**, 38.0 mg, $t_R = 8.0$ min), tropeoside B1 (**3a**, 46.1 mg, $t_R = 7.2$ min), and tropeoside B2 (**3b**, 60.3 mg, $t_R = 9.5$ min).

Tropeoside A1 (1a): yield, 26.4 mg; colorless amorphous solid. Furost-5(6)-en-3 β ,22 α -diol 1 β -*O*- β -D-galactopyranosyl 26-*O*- α -L-rhamnopyranoside: [α]_D²⁵ -44.3° (*c* 0.1, MeOH); HRFABMS (negative ion) of the equilibrated mixture (*m*/*z*) [M – H]⁻ calcd for C₃₉H₆₄O₁₄, 756.4278; found, 755.4205; ¹H NMR data, see **Tables 1** and **3**; ¹³C NMR data, see **Tables 2** and **3**.

Tropeoside A2 (Ib): yield, 38.0 mg; colorless amorphous solid. Furost-5(6)-en- 3β ,22 β -diol 1 β -O- β -D-galactopyranosyl 26-O- α -L-rham-



Figure 1. Saponins of A. cepa L. var. Tropea.

nopyranoside: $[\alpha]_D^{25}$ -44.3° (*c* 0.1, MeOH); HRFABMS (negative ion) of the equilibrated mixture (*m*/*z*) [M – H][–] calcd for C₃₉H₆₄O₁₄, 756.4278; found, 755.4205; ¹H NMR, data see **Tables 1** and **3**; ¹³C NMR data, see **Tables 2** and **3**.

Tropeoside B1 (3a): yield, 46.1 mg; colorless amorphous solid. Furost-5(6)-en-3 β ,22 α -diol 1 β -*O*- β -D-xylopyranosyl 26-*O*- α -L-rhamnopyranoside: [α]_D²⁵ -43.2° (*c* 0.1, MeOH); HRFABMS (negative ion) of the equilibrated mixture (*m*/*z*) [M – H]⁻ calcd for C₃₈H₆₂O₁₃, 726.4173; found, 725.4090; ¹H NMR data, see **Tables 1** and **3**; ¹³C NMR data, see **Tables 2** and **3**.

Tropeoside B2 (3b): yield, 60.3 mg; colorless amorphous solid. Furost-5(6)-en-3 β ,22 β -diol 1 β -*O*- β -D-xylopyranosyl 26-*O*- α -L-rhamnopyranoside: [α]_D²⁵ -43.2° (*c* 0.1, MeOH); HRFABMS (negative ion) of the equilibrated mixture (*m*/*z*) [M – H]⁻ calcd for C₃₈H₆₂O₁₃, 726.4173; found, 725.4090; ¹H NMR data, see **Tables 1** and **3**; ¹³C NMR data, see **Tables 2** and **3**.

Determination of Sugar Absolute Configurations. A solution of each isolated compound (1 mg) in 1 N HCl (0.25 mL) was stirred at 80 °C for 4 h. On cooling, the solution was concentrated in a stream of N2. The residue was dissolved in 1-(trimethylsilyl)imidazole (Trisil-Z) and pyridine (0.1 mL), and the solution was stirred at 60 °C for 5 min. After the solution had been dried with a stream of N₂, the residue was separated by water and CH₂Cl₂ (1 mL, 1:1). The CH₂Cl₂ layer was analyzed by GC on a 25 m \times 0.32 mm \times 25 m l-Chirasil-Val column (Alltech, Deerfield, IL). Temperatures for the injector and detector were 200 °C, and the temperature gradient system for the oven was as follows: 100 °C for 1 min, raised to 180 °C at 5 °C/min. Peaks of the hydrolysate of 1a/1b were detected at 12.89 (L-rhamnose) and 13.77 min (D-galactose), in the ratio of 1:1. Peaks of the hydrolysate of 3a/ 3b were detected at 10.97 (D-xylose) and 12.90 min (L-rhamnose), in the ratio of 1:1. Retention times for authentic samples after being treated simultaneously with Trisil-Z were 12.78 (D-rhamnose) and 12.89 min (L-rhamnose); 13.77 (D-galactose) and 13.84 min (L-galactose); and 10.97 (D-xylose) and 11.05 min (L-xylose). Co-injection of each hydrolysate with standard D-galactose, L-rhamnose, and D-xylose gave single peaks.

Table 1. ¹H NMR Data of the Aglycon Portion of Tropeosides A1 (1a), A2 (1b), B1 (3a), and B2 (3b) (Spectra Measured in CD₃OD at 500 MHz)

position	δ_{H} (int, mult, J in Hz)				
	1a	1b	3a	3b	
1	3.36 (1H, dd, 7.2, 3.5)	3.36 (1H, dd, 7.2, 3.5)	3.36 (1H, dd, 7.2, 3.5)	3.36 (1H, dd, 7.2, 3.5)	
2a	2.08 ^a	2.06 ^a	2.09 ^a	2.07 ^a	
b	1.70 ^a	1.70 ^a	1.70 ^a	1.70 ^a	
3	3.31 ^a	3.30 ^a	3.30 ^a	3.29 ^a	
4a	2.24 (1H, dd, 11.5, 7.3)	2.23 (1H, dd, 11.5, 7.3)	2.27 (1H, dd, 11.5, 7.3)	2.26 (1H, dd, 11.5, 7.3)	
b	2.20 (1H, dd, 11.5, 3.5)	2.20 (1H, dd, 11.5, 3.5)	2.21 (1H, dd, 11.5, 3.5)	2.21 (1H, dd, 11.5, 3.5)	
6	5.55 (1H, dd, 5.5, 2.5)	5.55 (1H, dd, 5.5, 2.5)	5.56 (1H, dd, 5.5, 2.5)	5.56 (1H, dd, 5.5, 2.5)	
7a	1.98 ^a	1.98 ^a	1.98 ^a	1.98 ^a	
b	1.96 ^a	1.96 ^a	1.98 ^a	1.98 ^a	
8	1.52 (1H, m)	1.52 (1H, m)	1.50 (1H, m)	1.50 (1H, m)	
9	1.31 ^a	1.31 ^a	1.29 ^a	1.29 ^a	
11a	2.56 (1H, dd, 10.5, 2.5)	2.55 (1H, dd, 10.5, 2.5)	2.55 (1H, dd, 10.5, 2.5)	2.54 (1H, dd, 10.5, 2.5)	
b	1.41 (1H, m)	1.42 (1H, m)	1.43 (1H, m)	1.44 (1H, m)	
12a	1.70 ^a	1.70 ^a	1.70 ^a	1.70 ^a	
b	1.19 ^a	1.20 ^a	1.20 ^a	1.21 ^a	
14	1.17 ^a	1.17 ^a	1.16 ^a	1.16 ^a	
15a	1.98 ^a	1.98 ^a	1.98 ^a	1.98 ^a	
b	1.31 ^a	1.31 ^a	1.31 ^a	1.31 ^a	
16	4.60 (1H, q, 5.5)	4.38 (1H, q, 5.5)	4.60 (1H, q, 5.5)	4.39 (1H, q, 5.5)	
17	1.72 ^a	1.67 ^a	1.72 ^a	1.67 ^a	
18	0.80 (3H, s)	0.79 (3H, s)	0.80 (3H, s)	0.79 (3H, s)	
19	1.09 (3H, s)	1.10 (3H, s)	1.09 (3H, s)	1.08 (3H, s)	
20	2.07 ^a	2.09 ^a	2.08 ^a	2.10 ^a	
21	0.99 (3H, d, 6.6)	0.96 (3H, d, 6.6)	0.99 (3H, d, 6.6)	0.96 (3H, d, 6.6)	
23a	1.72 ^a	1.75 à	1.72 ^a	1.75 ^a	
b	1.63 ^a	1.61 ^a	1.63 ^a	1.61 ^a	
24a	1.40 ^a	1.36 ^a	1.40 ^a	1.36 ^a	
b	1.31 ^a	1.32 ^a	1.31 ^a	1.32 ^a	
25	1.71 ^a	1.71 ^a	1.71 ^a	1.71 ^a	
26a	3.87 (1H, dd, 8.5, 6.9)	3.87 (1H, dd, 8.5, 6.9)	3.85 (1H, dd, 8.5, 6.9)	3.85 (1H, dd, 8.5, 6.9)	
b	3.30 ^a	3.30 ^a	3.30 ^a	3.30 ^a	
27	0.97 (3H, d, 6.6)	0.97 (3H, d, 6.6)	0.97 (3H, d, 6.6)	0.97 (3H, d, 6.6)	

^aOverlapped with other signals.

Biological Assav. Male guinea pigs (250–350 g.) (Harlan, Italy) were killed by asphysiation with CO2, and segments (1-1.5 cm) of ileum were removed, flushed of luminal contents, and placed in Krebs solution (119 mM NaCl, 4.75 mM KCl, 1.2 mM KH₂PO₄, 25 mM NaHCO₃, 2.5 mM CaCl₂, 1.5 mM MgSO₄, and 11 mM glucose). The segments were prepared as previously described (7). In brief, the segments were set up in such a way as to record contractions mainly from the longitudinal axis, in an organ bath containing 20 mL of Krebs equilibrated with 95% O2 and 5% CO2 at 37 °C. The tissues were connected to an isotonic transducer (load = 0.5 g), connected to a Gemini recording apparatus (Ugo Basile, Comerio, Italy). Ileal segments were allowed to equilibrate for 60 min before experiments. To evaluate the antispasmodic activity of the tested compounds, contractions were elicited by acetylcholine (10^{-7} M) or histamine (10^{-7}) . After stable control contractions evoked by acetylcholine (or histamine) had been recorded, the responses were observed in the presence of increasing concentrations $(10^{-8}-10^{-5} \text{ M})$ of the tested compounds (or papaverine, used as a reference compound). The contact time for each concentration was 30 min. Ascalonicoside and tropeosides were dissolved in DMSO. DMSO (<0.01%) did not modify acetylcholine- or histamine-induced contractions. Results are expressed as mean ± SEM. Nonlinear regression analysis for all concentration response curves was performed. Data were analyzed by ANOVA. The results were considered to be significant if the probability of error was <5%.

RESULTS AND DISCUSSION

Bulbs of red onion, *A. cepa* L. var. Tropea, were exhaustively extracted with CHCl₃/MeOH, 9:1, and MeOH. The MeOH extracts were partitioned between butanol and water, and the butanol-soluble portions were separated by sequential chromatographic techniques, affording, as major metabolites, furostanol saponins (total saponin content = 130.75 mg/kg) and flavonoids (total flavonoid concentration = 232.12 mg/kg). As

major saponin metabolites, we have isolated four new furostanol saponins (**Figure 1**), named tropeosides A1/A2 (**1a/1b**) and B1/B2 (**3a/3b**), the known ascalonicoside A1/A2 (**5a/5b**), and ascalonicoside B (**6**), previously isolated from *Allium ascalonicum* Hort. (*6*). Concerning the flavonoids, quercetin, quercetin 4^{I} -glucoside, both already described from *A. cepa* (8), taxifolin, and taxifolin 7-glucoside have been also found.

Tropeoside A1 (1a, Figure 1) was isolated as a white amorphous solid and showed in the negative FAB MS spectrum a pseudomolecular ion peak at m/z 755 [M - H]⁻. Highresolution measurements on the pseudomolecular ion peak indicated the molecular formula C₃₉H₆₄O₁₄. Diagnostic resonances in the ¹H NMR spectrum of **1a** (CD₃OD, **Tables 1** and 3) were those attributed to two tertiary methyls (δ 0.80 and 1.09), three secondary methyls (δ 0.97, 0.99, and 1.25), and one olefinic (δ 5.55), and two anomeric (δ 4.30 and 5.30) protons. The ¹³C NMR spectrum (Tables 2 and 3) showed 39 resonance lines, supporting the molecular formula deduced from MS, 27 of them attributed to the aglycon and 12 to two monosaccharides. Analysis of 2D COSY, HOHAHA, HSQC, and HMBC spectra of 1a allowed assignment of all the signals belonging to the aglycon moiety, which was identified as a trihydroxylated steroid of the furostanol type. In particular, ring A constitutes an isolated spin system possessing two hydroxyl groups at positions 1 and 3 (Table 1). The NOE effect between H-1 and H-3, observed in a ROESY spectrum (Figure 2), indicated the cis orientation of the two OH groups of ring A. The third hydroxyl group of the aglycon was easily located on the hemiacetal carbon at C-22 (δ 110.0). The 25R stereochemistry of the side chain was deduced by the resonances of protons and carbons at C-25, C-26, and C-27 and by the vicinal

Table 2. ¹³C NMR Data of the Aglycon Portion of Tropeosides A1 (1a), A2 (1b), B1 (3a), and B2 (3b) (Spectra Measured in CD_3OD at 125 MHz)

	$\delta_{ extsf{C}}$ (mult)						
position	1a	1b	3a	3b			
1	84.6 (CH)	84.6 (CH)	84.6 (CH)	84.6 (CH)			
2	37.5 (CH ₂)	37.5 (CH ₂)	37.3 (CH ₂)	37.3 (CH ₂)			
3	69.4 (CH)	69.4 (CH)	69.1 (CH)	69.1 (CH)			
4	43.3 (CH ₂)	43.3 (CH ₂)	43.4 (CH ₂)	43.4 (CH ₂)			
5	139.4 (C)	139.4 (C)	139.6 (C)	139.6 (C)			
6	125.8 (CH)	125.8 (CH)	125.9 (CH)	125.9 (CH)			
7	32.5 (CH ₂)	32.3 (CH ₂)	32.7 (CH ₂)	32.5 (CH ₂)			
8	33.9 (CH)	33.9 (CH)	34.1 (CH)	34.1 (CH)			
9	51.5 (CH)	51.5 (CH)	51.5 (CH)	51.5 (CH)			
10	43.2 (C)	43.2 (C)	43.4 (C)	43.4 (C)			
11	24.5 (CH ₂)	24.5 (CH ₂)	24.7 (CH ₂)	24.7 (CH ₂)			
12	40.9 (CH ₂)	40.9 (CH ₂)	41.1 (CH ₂)	41.0 (CH ₂)			
13	41.1 (C)	41.1 (C)	41.2 (C)	41.2 (C)			
14	57.9 (CH)	57.9 (CH)	58.0 (CH)	58.0 (CH)			
15	32.7 (CH ₂)	32.7 (CH ₂)	32.9 (CH ₂)	32.9 (CH ₂)			
16	82.1 (CH)	82.1 (CH)	82.3 (CH)	82.3 (CH)			
17	63.8 (CH)	63.7 (CH)	63.9 (CH)	63.8 (CH)			
18	17.3 (CH₃)	17.2 (CH ₃)	17.1 (CH₃)	17.0 (CH ₃)			
19	15.1 (CH₃)	15.0 (CH ₃)	15.2 (CH₃)	15.1 (CH ₃)			
20	40.8 (CH)	40.6 (CH)	41.1 (CH)	40.8 (CH)			
21	17.0 (CH ₃)	17.3 (CH ₃)	16.4 (CH ₃)	16.6 (CH ₃)			
22	110.0 (C)	113.0 (C)	111.0 (C)	114.0 (C)			
23	37.4 (CH ₂)	37.2 (CH ₂)	37.3 (CH ₂)	37.0 (CH ₂)			
24	31.3 (CH ₂)	31.1 (CH ₂)	31.4 (CH ₂)	31.3 (CH ₂)			
25	34.2 (CH)	34.2 (CH)	34.1 (CH)	34.1 (CH)			
26	75.9 (CH ₂)	76.0 (CH ₂)	76.0 (CH ₂)	76.2 (CH ₂)			
27	18.2 (CH ₃)	18.2 (CH ₃)	17.5 (CH ₃)	17.5 (CH ₃)			

couplings between H-25 and the two H-26, in comparison with literature data (9). The ROESY correlations (**Figure 2**) H-11/ H_3 -19, H-11/ H_3 -18, H-9/H-14, H-14/H-16, H-12/H-17, and H-17/ H_3 -21 completed the relative stereochemistry of **1a** with the usual B/C trans, C/D trans, D/E cis, and C-20 α stereochemistries (10, 11). On the basis of those data, the stereochemistry of the aglycon chiral centers has been assigned as in **Figure 1**, except for the stereochemistry at C-22, which will be discussed below.

Concerning the sugar portions of the molecule, the first step of the analysis was the association of the two anomeric carbons (δ 100.7 and 101.3) with the relevant anomeric proton signals (δ 4.30 and 5.30, respectively), through the HSQC experiment.



Figure 2. Selected HMBC ($H\rightarrow C$) and ROESY ($H\leftrightarrow H$) correlations observed for compound 1a.

The nature of the monosaccharides was determined by analysis of 2D COSY, HOHAHA, HSQC, and HMBC spectra. Starting from the anomeric proton at δ 4.30 (H-1 Gal), we identified the sequence of a hexopyranose unit. The large coupling constants observed in a 2D HOHAHA subspectrum for H-1 Gal/ H-2 Gal and H-2 Gal/H-3 Gal, and the relatively small coupling constant of H-3 Gal/H-4 Gal, together with the ROESY correlation H-3 Gal/H-5 Gal, indicated its β -galactopyranose nature. With the same type of analysis the sugar linked at C-26 was identified as a 6-deoxyhexopyranose. Within its spin system, the axial-axial H-3 Rha/H-4 Rha and H-4 Rha/H-5 Rha couplings and the equatorial-axial H-2 Rha/H-3 Rha relationship led to the assignment of a rhamnose. Its α -anomeric configuration, suggested by the very small J value of H-1 Rha, resonating as a broad singlet, was confirmed by the chemical shift of C-3 Rha and C-5 Rha (Table 3), in accordance with data reported in the literature (10, 12). The β -galactose was placed at C-1 of tropeoside A1 by interpretation of the key HMBC correlation peak (Figure 2) between H-1 Gal (δ 4.30) and C-1 (δ 84.6), this junction being also confirmed by the strong ROESY peak (Figure 2) between H-1 Gal and H-1 (δ 3.36). The α -rhamnose had to be linked at position 26 on the furostanol structure of the aglycon. In fact, we observed the HMBC cross-peak (Figure 2) between C-26 (δ 75.9) and H-1 Rha (δ 5.30).

To confirm the nature of the sugar units and to determine their absolute configuration, **1a** was subjected to acid hydrolysis (1 N HCl), followed by trimethylsilylation and GC analysis on a chiral column in comparison with both series of galactose and rhamnose. By this procedure the sugars were identified as D-galactose and L-rhamnose. This procedure was applied to all of the new isolated compounds.

Table 3. ¹H and ¹³C NMR Data of the Sugar Portion of 1a/1b (Data Extracted from 1a) and 3a/3b (Data Extracted from 3a) (Spectra Measured in CD₃OD at 500 and 125 MHz)

	1a			3a	
position	$\delta_{\rm H}$ (int, mult, J in Hz)	$\delta_{\rm C}$ (mult)	position	δ_{H} (int, mult, <i>J</i> in Hz)	δ_{C} (mult)
Gal			Xyl		
1	4.30 (1H, d, 7.8)	100.7 (CH)	1	4.27 (1H, d, 7.5)	101.0 (CH)
2	3.70 (1H, dd, 7.8, 6.8)	74.0 (CH)	2	3.68 (1H, dd, 7.5, 6.8)	74.1 (CH)
3	3.68 (1H, dd, 6.8, 2.5)	75.2 (CH)	3	3.88 (1H, dd, 7.0, 6.8)	75.6 (CH)
4	3. 87 (1H, dd, 3.2, 2.5)	76.6 (CH)	4	3.65 (1H, ddd, 2.0, 7.0, 8.5)	69.6 (CH)
5a	3.44 ^a	74.0 (CH)	5a	3.50 (1H, dd, 2.0, 11.5)	66.0 (CH ₂)
5b			5b	3.95 (1H, dd, 8.5, 11.5)	(
6a	3.47 ^a	62.2 (CH ₂)			
6b	3.65 ^a				
Rha			Rha		
1	5.30 (1H, bs)	101.3 (CH)	1	5.30 (1H, bs)	101.5 (CH)
2	3.89 (1H, bs)	72.2 (CH)	2	3.90 (1H, bs)	72.3 (CH)
3	3.72 (1H, d, 6.5)	70.6 (CH)	3	3.72 (1H, d, 6.5)	70.7 (CH)
4	3.40 (1H, dd, 6.5, 6.0)	72.0 (CH)	4	3.39 (1H, dd, 6.5, 6.0)	72.1 (CH)
5	4.10 (1H, dq, 6.6, 6.0)	69.0 (CH)	5	4.10 (1H, dq, 6.6, 6.0)	67.4 (CH)
6	1.25 (3H, d, 6.6)	18.2 (CH ₃)	6	1.25 (3H, d, 6.6)	18.3 (CH ₃)

^a Overlapped with other signals.

High-resolution FABMS indicated that tropeoside A2 (1b) was isomeric with tropeoside A1. The ¹H and ¹³C NMR spectra of 1b (Tables 1-3) were almost identical with that of 1a, but slightly different in the resonances of the atoms located near C-22. This suggested that the two compounds were epimers having the opposite configuration at the hemiacetal carbon (C-22). This was confirmed by the observation that both 1a and 1b, after being kept one night in aqueous solution at room temperature, gave rise to an equilibrium between the two forms (about 40% of **1a** and 60% of **1b**). We have tentatively assigned the 22 α orientation to tropeoside A1 and 22 β orientation to tropeoside A2 on the basis of the downfield shift in the ¹H NMR resonances of H₃-21 and H-16 of 1a, in comparison with those of **1b** (H₃-21 δ 0.99 instead of 0.96; H-16 δ 4.60 instead of 4.38), due to the deshielding effect of the cis-oriented OH-22 group (13).

The structure elucidation of tropeoside B1 (3a) was aided by comparison with the MS and NMR data obtained for 1a. The high-resolution FABMS gave the molecular formula of 3a as $C_{38}H_{62}O_{13}$, which differs from **1a** in being 30 amu less. The NMR profiles of 3a (Tables 1-3) appeared to be superimposable with those of **1a** for the signals relative to the aglycon and to the α -rhamnose. Differences were related to the second sugar unit, which by MS and 2D NMR analyses appeared to be based on a pentopyranose structure. The large coupling constants observed in a 2D HOHAHA subspectrum for H-1 Xyl/H-2 Xyl, H-2 Xyl/H-3 Xyl, and H-3/H-4 Xyl indicated a β -xylopyranosyl nature. This was confirmed by acid hydrolysis, followed by silylation and analysis on a chiral column, which afforded D-xylose and L-rhamnose. Diagnostic HMBC correlations of H-1^I with C-1 and of H-1^{II} with C-26 located xylose and rhamnose at C-1 and C-26, respectively, thus fully defining the structure of 3a.

Mass spectrometric analysis of tropeoside B2 (3b) showed the same molecular formula as that of 3a. In analogy, the NMR profiles (Tables 1–3) also revealed the same structure for both compounds, indicating that the difference had to be limited to the stereochemistry of a chiral carbon. As observed for 1a/1b, the two compounds 3a/3b were in equilibrium if left in solution overnight. We therefore determined the structure of 3b as the epimer of 3a at the hemiacetal carbon. The assignments of the ¹H and ¹³C NMR spectra, obtained through 2D NMR spectra, are reported in Tables 1–3. On the basis of the same arguments as for 1a/1b, we tentatively assigned the configurations 22α to 3a and 22β to 3b.

Compounds 2a/2b and 4a/4b showed ¹H and ¹³C NMR spectra superimposable with those of 1a/1b and 3a/3b, respectively. The only difference was the presence of an additional methoxy group signal in each spectrum of 2a/2b and 4a/4b [¹H NMR, δ 3.12 (3H, s); ¹³C NMR, δ 47.2]. This allowed us to identify compounds 2a/2b and 4a/4b as the 22-*O*-methyl derivatives of 1a/1b and 3a/3b, respectively, from which they originate during the extraction and purification steps. This was confirmed by repeating the extraction procedure in acetone and observing the absence of 2a/2b and 4a/4b in the extract and their successive formation during the purification of tropeosides A1/A2 and B1/B2 by HPLC.

Finally, we have also isolated in high amounts the trisaccharide saponins ascalonicosides A1/A2 (5a/5b) and B (6), recently reported by some of the authors from shallot (6).

Because one of the remarkable traditional uses of onion is the treatment of intestinal spasm, in the present study we have investigated the possible antispasmodic effect of the isolated saponins on the contractions elicited by acetylcholine and



Figure 3. Effect of tropeoside A1/A2, tropeoside B1/B2, and ascalonicoside A1/A2 on the contractions evoked by acetylcholine (10^{-7} M) (**A**) or histamine (10^{-7} M) (**B**) in the guinea pig ileum. Each point represents the mean of four to five experiments. Vertical lines show SEM. *, *P* < 0.05 versus corresponding control.

histamine in the isolated guinea pig ileum. The pharmacological results are shown in **Figure 3**. Tropeosides A1/A2 and B1/B2 reduced, in a concentration-dependent manner, the contractions evoked by both acetylcholine (**Figure 3A**) and histamine (**Figure 3B**). The effect was significant at the concentration of 10^{-5} M (~50% inhibition). By contrast, ascalonicoside A1/A2 did not show any significant effect. Papaverine, used as a reference compound, inhibited both acetylcholine-induced (percent inhibition, 10^{-8} M, 4 ± 3 ; 10^{-7} M, 16 ± 4 ; 10^{-6} M, 32 ± 5 ; 10^{-5} M, 68 ± 5 ; 10^{-4} M, 89 ± 6 , n = 4) and histamine-induced (percent inhibition, 10^{-8} M, 3 ± 4 ; 10^{-7} M, 14 ± 6 ; 10^{-6} M, 35 ± 5 ; 10^{-5} M, 71 ± 4 ; 10^{-4} M, 90 ± 4 , n = 4) contractions.

Comparison of the chemical structures of the tested compounds in terms of bioactivity suggested that the presence of a further sugar in ascalonicoside A1/A2 results in the loss of the antispasmodic activity.

With regard to the flavonoids, the finding of high concentrations agree with the observation that onions are characterized

generally by a higher content of flavonoids if compared to garlic and leek (14, 15). From the biological point of view, flavonoids are considered to be important factors of the overall antioxidant activity of dietary plants (16, 17). In particular, guercetin has shown an anti-HIV property (18) and the ability to protect lowdensity lipoprotein (LDL) cholesterol from oxidation, thereby reducing the risk of cardiovascular (19) and neurodegenerative diseases, such as Alzheimer's and Parkinson's. In addition, flavonoids including quercetin and taxifolin also affect the gastrointestinal tract by possessing antiulcer, antispasmodic, and antidiarrheal activities (20). For all of these reasons, the finding of high concentrations of saponins and flavonols in bulbs of Tropea red onion makes it a healthy food in the Mediterranean diet. In addition, because these onions are often consumed uncooked, due to their crispness and sweetness, these phytonutrients are totally preserved.

In conclusion, the present study reports the isolation of high concentrations of flavonoids and saponins from a common food, the red onion. Among them, we have characterized two new saponins that exhibited antispasmodic activity. This result may be important from ethnopharmacological, pharmaceutical, or nutritional points of view. From an ethnopharmacological point of view, the antispasmodic activity of these compounds, together with the previously reported antispasmodic activity of flavonoids (7), might contribute to the justification of the traditional use of onion in the treatment of intestinal spasms. From a pharmaceutical point of view, the antispasmodic activity may open the way to study in vivo the effect of the isolated compounds in models of intestinal hyperactivity. Last but not least, from a nutritional point of view, the presence in onion of compounds with antispasmodic activity may provide input to perform epidemiological studies aiming at investigating the relationship between onion intake and chronic digestive illnesses, characterized by hypercontractility of intestinal smooth muscles (e.g., irritable bowel syndrome).

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