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Apteniols A–F, oxyneolignans from the leaves of Aptenia cordifolia

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Abstract—Investigation of the organic extract of *Aptenia cordifolia* leaves revealed six new oxyneolignans named apteniols A–F. The structures were determined by means of spectroscopic methods. The C_6C_3 units are linked by an oxygen atom at C4–C4' or C4–C2' and they are dihydrophenylpropanoid acid units. Their effects on germination and growth of *Lactuca sativa* L. have been studied in the range concentration 10^{-4} – 10^{-7} M.

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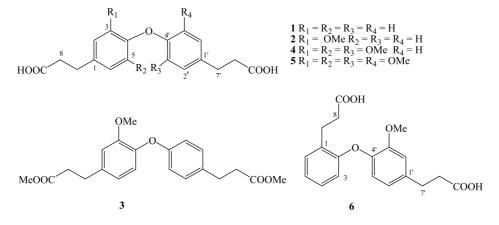
1. Introduction

Aptenia cordifolia, belonging to the Aizoaceae family, is a perennial herb native to South Africa and has now largely spread throughout Europe. The most commonly grown plant, usually grown under the cultivar name of 'Red Apple', is considered by some botanists to be actually a hybrid between *Aptenia cordifolia* and the closely related *Platythyra Aptenia haeckeliana*.¹ Previous chemical studies of the aerial part of *A. cordifolia* evidenced the presence of flavonoids.² As part of our research on bioactive natural products isolated from spontaneous plants present in Italy and their use as natural herbicide models, we recently reported that some metabolites isolated from *Brassica fruticulosa* and *Chenopodium album* inhibited the germination and growth of some mono and dicotyledons.^{3,4}

Continuing the phytochemical study of common weeds widely distributed in the Mediterranean area, we have investigated *A. cordifolia*. The present paper reports the isolation and structure elucidation of six new oxyneolignans from the hydroalcoholic infusion of fresh leaves of *A. cordifolia*. The C_6C_3 units are linked by an oxygen atom at C4–C4' or C4–C2' and they are dihydrophenylpropanoid acid units. The phytotoxic activity of the compounds have been evaluated on the dicotyledon *Lactuca sativa*.

2. Results and discussion

Repeated column chromatographies and preparative HPLC of the methanol/water extract of leaves of the plants, after acetone precipitation, afforded compounds 1-6. The



Keywords: Aptenia cordifolia; Apteniols A–F; Oxyneolignans; Spectroscopic analysis; *Lactuca sativa.* * Corresponding author. Tel.: + 39 081 674162; fax: + 39 081 674393; e-mail: dellagre@unina.it

Position	1	2	3 ^b	4 ^b	5 ^b	6
2	6.92d (8.5)	6.72d (1.5)	6.70d (1.5)	6.44s	6.42s	
3	6.63d (8.5)	. ,	· /			7.21 ^c
4						7.23 ^c
5	6.63d (8.5)	6.82d (8.2)	6.83d (8.0)			7.12m
6	6.92d (8.5)	6.70dd (8.2, 1.5)	6.68dd (8.0, 1.5)	6.44s	6.42s	7.21 ^c
7	2.84t (8.1)	2.89t (8.0)	2.88t (7.5)	2.90t (7.5)	2.88t (8.0)	2.89t (7.6)
8	2.66t (8.1)	2.66t (8.0)	2.59t (7.5)	2.67t (7.5)	2.61t (8.0)	2.44t (7.6)
2'	6.92d (8.5)	7.05d (8.5)	7.05d (9.0)	6.72s	6.42s	6.82d (2.0)
3'	6.63d (8.5)	6.74d (8.5)	6.75d (9.0)			
5'	6.63d (8.5)	6.74d (8.5)	6.75d (9.0)	6.82d (8.5)		6.67d (8.0)
6'	6.92d (8.5)	7.05d (8.5)	7.05d (9.0)	6.70d (8.5)	6.42s	6.63dd (8.0, 2.0)
7'	2.84t (8.1)	2.89t (8.0)	2.88t (7.5)	2.90t (7.5)	2.88t (8.0)	2.82t (7.8)
8'	2.66t (8.1)	2.66t (8.0)	2.61t (7.5)	2.67t (7.5)	2.61t (8.0)	2.41t (7.8)
3-OMe		3.86s	3.87s	3.87s	3.88s	
5-OMe				3.87s	3.88s	
3'-OMe				3.87s	3.88s	3.87s
5'-OMe					3.88s	
9-OMe			3.67s			
9'-OMe			3.67s			

Table 1. ¹H NMR Data of 1–6 (CD₃OD) at 500 MHz^a

 $^{a}_{L}J$ values (in Hz) in parentheses.

^b Recorded in CDCl₃ 500 MHz.

^c Multiplicity was not determined due to overlapping.

structure of these compounds was elucidated by their spectral data.

Apteniol A (1) showed a molecular peak at m/z 337 [M + Na]⁺ in the MALDI-MS spectrum suggesting, along with the elemental analysis, a molecular formula C₁₈H₁₈O₅. The UV spectrum revealed band at 278 nm. The ¹H NMR and the ¹³C NMR spectra indicated a highly symmetric molecule. In the ¹H NMR spectrum (Table 1) eight aromatic protons were present as two *ortho*-coupled protons and eight methylene protons as two triplets in aliphatic region. The ¹³C NMR spectrum (Table 2) showed only seven carbon signals. The ¹H and ¹³C resonances of 1 were assigned by combination of COSY, DEPT, HMQC and HMBC experiments. The DEPT spectrum showed two methylenes,

and two methines. According to the structure in the HMBC spectrum both the H-7/H-7' and H-8/H-8' protons were correlated to the C-9/C-9' at δ 180.6 and C-1/C-1' at δ 129.6; furthermore the H-3/H-3' protons were correlated to C-1/C-1', and C-4/C-4' carbons at δ 157.4.

The molecular ion at m/z 390 $[M+2Na]^+$ in the MALDI-MS spectrum along with the elemental analysis, defined the molecular formula $C_{19}H_{20}O_6$ of apteniol B (2). Its ¹H NMR spectra (Table 1) indicated the presence of four *ortho*-coupled protons, two doublet protons, and one double doublet proton in the aromatic region; a methyl singlet of a methoxyl group, and two triplets in the aliphatic region. In the ¹³C NMR spectrum (Table 2) fourteen carbons were evident. The DEPT spectrum indicated the presence of a

Table 2. ¹³C NMR Data of 1–6 (CD₃OD) at 125 MHz

Position	1	2	3 ^a	4 ^a	5 ^a	6
1	129.6	132.1	132.7	133.1	132.0	135.7
2	130.7	111.0	110.2	104.8	104.9	113.5
3	116.6	146.4	146.7	146.9	147.0	148.6
4	157.4	144.0	144.2	131.9	145.2	145.8
5	116.6	114.4	114.6	146.9	147.0	116.4
6	130.7	120.8	121.0	104.8	104.9	122.0
7	35.7	30.3	30.9	30.7	31.2	34.0
8	42.9	35.9	36.3	35.7	36.1	41.9
9	180.6	178.9	173.8	177.8	178.4	182.2
1'	129.6	132.1	130.0	131.2	132.0	127.1
2'	130.7	129.4	129.6	110.8	104.9	144.2
3'	116.6	115.4	115.5	146.9	147.0	129.2
4'	157.4	154.1	154.4	144.0	145.2	129.2
5'	116.6	115.4	115.5	114.3	147.0	126.6
6'	130.7	129.4	129.6	120.7	104.9	129.2
7′	35.7	30.3	30.9	30.2	31.2	34.4
8′	42.9	35.8	36.3	35.7	36.1	41.6
9′	180.6	178.9	173.8	177.8	178.4	182.2
3-OMe		55.8	56.1	56.1	56.3	56.5
5-OMe				56.1	56.3	
3'-OMe				55.7	56.3	
5'-OMe					56.3	
9-OMe			51.9			
9'-OMe			51.9			

^a Recorded in CDCl_{3.}

methyl, three methylenes, and five methines. The HMQC experiment allowed the assignment of the protons to the corresponding carbons. The HMBC spectrum showed crosspeaks of both the H-2 and H-5 protons with the C-3 and C-4 carbons and the H-6 with the C-1 and C-7. Furthermore the H-2'/H-6' protons gave cross peaks with the C-4' and C-7' carbons, the H-7' and the H-8' protons gave cross peaks with the C-1' and C-9' carbons. These data led to the structure **2** as depicted. The assignment of the methoxyl at C-3 was confirmed by NOE between the methoxyl and the H-2 proton.

Apteniol C (**3**) had the molecular formula $C_{21}H_{24}O_6$ as deduced from the molecular peak at m/z 372 in the MALDI-MS spectrum. The general features of its IR and NMR spectra (Tables 1 and 2) closely resembled those of **2**, except for the presence of two methoxyl signals in the NMR spectra. The structure of compound **3** was also deduced from the NOE observed between the methoxyls at δ 3.67 and the H-8/H-8' protons in the NOESY experiment. Compound **3** was also obtained by treatment of Apteniol B with ethereal CH₂N₂.

Apteniol D (4) had the molecular formula $C_{21}H_{24}O_8$, as established by the ion at m/z 427 [M+Na]⁺ in the MALDI-MS spectrum and elemental analysis. The ¹H NMR spectrum (Table 1) showed the presence of one aromatic ring with three coupled protons in a ABX system, an aromatic ring with two protons *meta* coupled, two methylenes as triplets and a methyl singlet of three methoxyls. In the ¹³C NMR spectrum (Table 2) fifteen carbon signals were present, and the DEPT experiment evidenced two methyls, two methylenes, and four methines. The ¹H and ¹³C resonances of apteniol C were assigned by

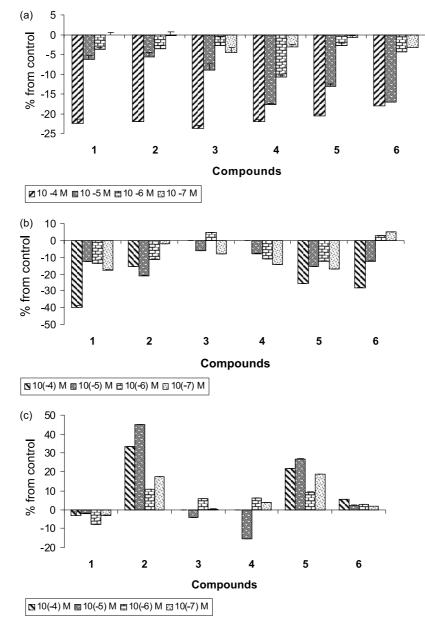


Figure 1. (A) Effect of compounds 1–6 on germination of *Lactuca sativa* L. Value presented as percentage differences from control. (B) Effect of compounds 1–6 on root length of *Lactuca sativa* L. Value presented as percentage differences from control. (C) Effect of compounds 1–6 on shoot length of *Lactuca sativa* L. Value presented as percentage differences from control. (C) Effect of compounds 1–6 on shoot length of *Lactuca sativa* L. Value presented as percentage differences from control.

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combination of COSY, DEPT, HMQC and HMBC experiments. The HMBC spectrum of compound **4** showed cross-peaks of both the H-2 and H-6 with C-4 and C-7, and both the H-2' and the H-5' protons with the C-1' and C-4' carbons. NOEs between the signal at δ 3.87 and the H-2, H-6 and H-2' protons allowed to assign the methoxyls at C-3, C-5, and C-5' positions.

Apteniol E (5) showed a molecular peak at m/z 480 [M+ 2Na]⁺ in the MALDI-MS spectrum suggesting, along with the elemental analysis, a molecular formula C₂₂H₂₆O₉. The UV spectrum revealed bands at 322, 279 and 244 nm. The ¹H NMR spectrum (Table 1) showed only one singlet aromatic signal, two triplets in aliphatic region, and a methyl signals. In the ¹³C NMR spectrum (Table 2) only eight carbon signals were present indicating a highly symmetric molecule. The ¹H and ¹³C resonances of 5 were assigned by combination of COSY, DEPT, HMOC and HMBC experiments. The DEPT spectrum showed one methyl, two methylenes, and one methine. According to the structure in the HMBC spectrum both the H-7/H-7' and the H-8/H-8' protons were correlated to the C-9/C-9' at δ 178.4 and the C-1/C-1' at δ 132.0; furthermore the H-2, H-6/H-2', H-6' protons were correlated to the C-1/C-1' and C-4/C-4' carbons. The NOE of the signal at δ 3.88 with H-2/H-6 and H-2'/H-6' allowed assignment of the methoxyls at the C-3, C-3['], C-5, and C-5['] positions.

Apteniol F (6) had the same molecular formula as 2 as deduced from a molecular peak at m/z 343 $[M-H]^+$ in the MALDI-MS spectrum, and the elemental analysis. The ¹H NMR spectrum (Table 1) showed the presence of one aromatic ring with three coupled protons in a ABX system, an aromatic ring with four adjacent protons, two methylenes as triplets and a methyl singlet of a methoxyl group. In the ¹³C NMR spectrum (Table 2) sixteen carbon signals were present, and the DEPT experiment showed one methyl, four methylenes, and five methines. The ¹H and ¹³C resonances of apteniol F were assigned by a combination of COSY, DEPT, HMQC and HMBC experiments. The HMBC spectrum of compound 6 showed cross-peaks of H-7 with C-2, C-6 and C-9, H-8 with C-1, and both the H-2' and the H-5' protons with the C-1' and C-4' carbons. NOEs between the signal at δ 3.87 and H-2' allowed assignment of the methoxyl at the C-3' position.

All of the compounds isolated from *A. cordifolia* were tested for their phytotoxicity on the seeds of *Lactuca sativa*.⁵ This species was selected as representative of main dicotyledon commercial crops.⁵ It has been used extensively as a test organism because of its fast germination and high sensitivity, and allows comparison of bioassay results for many different compounds.^{6,7} Aqueous solution of oxyneolignans **1–6**, ranging between 10^{-4} and 10^{-7} M, were tested on germination, root length and shoot length of treated lettuce seeds (Fig. 1). All the tested compounds reduced the germination by 25% at 10^{-4} M, and dose dependence effects were observed. They reduced the root elongation by 20% at 10^{-4} M, but compound **1** was the most active (40%) when compared to the control and compound **3** was almost completely inactive. Amongst compounds **1–6**, only oxyneolignan **4** reduced shoot elongation by 15% at the highest concentration and no important effects were observed for compounds 1, 3, and 6, while compounds 2 and 5 showed stimulatory effects within 20–50%.

Lignans, found widely throughout the plant kingdom, exhibit interesting antimicrobic, antiviral, herbicidal, or antifeedant activities that are thought to participate in plant defence mechanisms against biotic stresses.^{8–11} Cancer protective effects of dietary lignans have been also demonstrated.^{12,13} In contrast, neolignans are distributed in the limited plant families, as Pinaceae and Cupressaceae.^{14,15} The presence of dihydrocinnamic acid residues, in oxyneolignans isolated from *A. cordifolia* is peculiar because, to the best of our knowledge, the more common compounds isolated contain a reduced carboxylic group as alcohol or alkane.

3. Experimental

3.1. General procedures

NMR spectra were recorded at 25 °C on a Varian UNITY INOVA-500 spectrometer, operating at 500 and 125 MHz for ¹H and ¹³C, respectively. Matrix assisted laser desorption ionization (MALDI) mass spectra were recorded using a Voyager-DE MALDI-TOF mass spectrometer. UVvis spectra were recorded in methanol on a Perkin-Elmer Lambda 7 spectrophotometer. Liquid chromatography over silica gel (230-400 mesh) was performed at medium pressure. Preparative HPLC was run on a Agilent 1100 equipped with an UV detector and using SiO₂ (LiChrospher Silica $10 \,\mu\text{m}$, $250 \times 10 \,\text{mm}$ i.d., Merck), RP-18 (LiChrospher 10 μ m, 250 \times 10 mm i.d., Merck) columns. Analytical TLC was performed on precoated Merck aluminum sheet (DC-Alufolien Kielselgel 60 F_{254} , 0.2 mm) or RP-18 F₂₅₄ plates with 0.2 mm film thickness. The spots were visualized by UV light or by spraying with H_2SO_4 -AcOH- H_2O (1:20:4). The plates were then heated for 5 min at 110 °C. Prep. TLC was performed on Merck Kiesegel 60 F₂₅₄ plates, with 0.5 or 1 mm film thickness.

3.2. Plant material

Leaves of *Aptenia cordifolia* were collected in Italy (Campania) during the summer (August) and identified by Professor Pollio of the Dipartimento di Biologia Vegetale of University Federico II of Napoli. A voucher specimen (HERBNAPY680) has been deposited in the herbarium at the University Federico II.

3.3. Extraction and isolation of metabolites

Fresh leaves (12 kg) of the plants were powdered and extracted with H₂O–CH₃OH (9/1) at room temperature (25 °C) for 7 days. To an aqueous suspension (800 ml) of the H₂O/CH₃OH extract (450 g), cold (CH₃)₂CO (1.0 l) was added, and the mixture was placed on a stirring plate overnight at -18 °C. The (CH₃)₂CO addition led to heavy precipitation consisting mostly of proteinaceous material, which was removed by centrifugation. The (CH₃)₂CO was removed and the aqueous solution was chromatographed on Amberlite XAD-2 with H₂O, CH₃OH and (CH₃)₂CO to give 3 fractions. The fraction eluted with CH₃OH (45.0 g) was rechromatographed on silica gel column to give 13 fractions. Fraction 5 (5.5 g) eluted with CH₂Cl₂–CH₃OH (4/1) was rechromatographed on SiO₂ flash column eluting with CH₂Cl₂–AcOEt–(CH₃)₂CO gradient to afford 9 subfractions. Subfraction 1 (91 mg) eluted with CH₂Cl₂– (CH₃)₂CO (4/1) was purified by preparative TLC [petrol ether–(CH₃)₂CO (3/1)], to give **3** (8 mg).

Fraction eluted with (CH₃)₂CO (50.0 g) was rechromatographed on silica gel column to give 11 fractions. Fraction 6 (5.0 g) eluted with CH₂Cl₂-CH₃OH (1/1) was rechromatographed on SiO₂ flash column eluting with CH₂Cl₂-(CH₃)₂CO-CH₃OH gradient to afford 15 subfractions. Subfraction 10 (498 mg) eluted with CH₂Cl₂-CH₃OH (4/1) was rechromatographed on SiO₂ flash column eluting with AcOEt-(CH₃)₂CO-CH₃OH gradient to afford the fractions A-M. Fraction B (52 mg) eluted with AcOEt- $(CH_3)_2CO$ (9/1) was purified by reversed-phase HPLC column (LiChrosphere RP-18 10 µm, 250×10 mm i.d., Merck) $[H_2O-CH_3OH (1/1)]$, to give 6 (4 mg). Subfraction 13 (2.3 g) eluted with $(CH_3)_2CO-CH_3OH$ (1/1) was rechromatographed on SiO₂ flash column, eluting with CH₂Cl₂-(CH₃)₂CO-CH₃OH gradient to afford the fractions A-I. Fraction C (46 mg) eluted with CH₂Cl₂-(CH₃)₂CO (4/1) was purified by preparative TLC [Petrol ether 40-60- $(CH_3)_2CO$ (3/1)], to give 5 (4 mg). Fraction D (30 mg) eluted with CH₂Cl₂-(CH₃)₂CO (3/1) was purified by reversed-phase HPLC column [H₂O-CH₃OH (1/1)], to give 4 (9 mg) and 2 (11 mg). Fraction 8 (16.0 g) eluted with CH₃OH was rechromatographed on SiO₂ flash column eluting with (CH₃)₂CO-CH₃OH gradient to afford 7 subfractions. Subfraction 6 (292 mg) eluted with $(CH_3)_2CO-CH_3OH$ (1/1) was rechromatographed on SiO₂ flash column eluting with $(CH_3)_2CO-CH_3OH(7/3)$ to afford the fractions A-G. Fraction B (76 mg) was purified by preparative TLC [(CH₃)₂CO-CH₃OH-H₂O (35/14/1)], to give 1 (8 mg).

3.3.1. 4,4'-Oxyneolign-9,9'-dioic acid (1). UV λ_{max} (CH₃OH) nm: 278 (log ε 3.2). MALDI-MS m/z (%): 337 (70), 277 (10), 255 (100). Anal. Calcd for C₁₈H₁₈O₅: C, 68.78, H, 5.77. Found: C, 69.00, H, 5.79. ¹H and ¹³C NMR data are listed in Tables 1 and 2.

3.3.2. 3-Methoxy-4,4'-oxyneolign-9,9'-dioic acid (2). UV λ_{max} (CH₃OH) nm: 280 (log ε 2.2). MALDI-MS *m/z* (%): 390 (15), 344 (10), 329 (20), 300 (30), 179 (100). Anal. Calcd for C₁₉H₂₀O₆: C, 66.27, H 5.85. Found: C, 66.45, H, 5.95. ¹H and ¹³C NMR data are listed in Tables 1 and 2.

3.3.3. Dimethyl 3-methoxy-4,4'-oxyneolign-9,9'-dioate (3). UV λ_{max} (CH₃OH) nm: 280 (log ε 3.1). MALDI-MS *m*/*z* (%): 372 (18), 254 (30), 209 (100), 163 (40). Anal. Calcd for C₂₁H₂₄O₆: C, 67.73, H, 6.50. Found: C, 67.91, H, 6.48. ¹H and ¹³C NMR data are listed in Tables 1 and 2.

3.3.4. 3,3',5-Trimethoxy-4,4'-oxyneolign-9,9'-dioic acid (4). UV λ_{max} (CH₃OH) nm: 321 (log ε 0.9), 280 (2.6), 242 (2.5). MALDI-MS *m*/*z* (%): 427 (10), 409 (20), 331 (10), 226 (20), 273 (100). Anal. Calcd for C₂₁H₂₄O₈: C, 76.18, H, 4.79. Found: C, 76.10, H, 4.50. ¹H and ¹³C NMR data are listed in Tables 1 and 2.

3.3.5. 3,3',5,5'-Tetramethoxy-4,4'-oxyneolign-9,9'-dioic acid (5). UV λ_{max} (CH₃OH) nm: 322 (log ε 1.5), 279 (3.0), 244 (2.8). MALDI-MS m/z (%): 507 (20), 344 (20), 240 (100). Anal. Calcd for C₂₂H₂₆O₉: C, 60.82, H, 6.03. Found: C, 61.00, H, 6.05. ¹H and ¹³C NMR data are listed in Tables 1 and 2.

3.3.6. 3'-Methoxy-2,4'-oxyneolign-9,9'-dioic acid (6). UV λ_{max} (CH₃OH) nm: 280 (log ε 1.4). MALDI-MS *m*/*z* (%): 343 (20), 272 (30), 179 (100), 196 (35). Anal. Calcd for $C_{19}H_{20}O_6$: C, 66.27, H, 5.85. Found: C, 66.91, H, 5.90. ¹H and ¹³C NMR data are listed in Tables 1 and 2.

3.4. Bioassays

Seeds of Lactuca sativa L. (cv. Cavolo di Napoli, collected during 2003, were obtained from Ingegnoli Spa (Milan, Italy). All undersized or damaged seeds were discarded and the assay seeds were selected for uniformity. For the bioassays we used Petri dishes of 50 mm diameter with one sheet of Whatman No. 1 filter paper as support. In four replicate experiments, germination and growth were conducted in aqueous solutions at controlled pH. Test solutions (10⁻⁴ M) were prepared using MES (2-[Nmorpholino]ethanesulfonic acid, 10 mM, pH 6) and the rest $(10^{-5}-10^{-7} \text{ M})$ were obtained by dilution. Parallel controls were performed. After adding 25 seeds and 2.5 ml test solutions, Petri dishes were sealed with Parafilm[®] to ensure closed-system models. Seeds were placed in a growth chamber KBW Binder 240 at 25 °C in the dark. Germination percentage was determined daily for 5 days (no more germination occurred after this time). After growth, plants were frozen at -20 °C to avoid subsequent growth until the measurement process. Data are reported as percentage differences from control in the graphics and tables. Thus, zero represents the control, positive values represent stimulation of the parameter studied and negative values represent inhibition.

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References and notes

- Von Willert, D. J.; Thomas, D. A.; Lobin, W.; Curdts, E. Oecologia 1977, 29, 67–76.
- Elgindi, M. R.; Elgindi, O. D.; Mabry, T. J.; Fiorentino, L. Asian J. Chem. 1999, 11, 1525–1527.
- 3. Cutillo, F.; D'Abrosca, B.; DellaGreca, M.; Fiorentino, A.; Zarrelli, A. J. Agric. Food. Chem. 2003, 51, 6165–6172.
- DellaGreca, M.; Di Marino, C.; Zarrelli, A.; D'Abrosca, B. J. Nat. Prod. 2004, 67, 1492–1495.
- Macias, F. A.; Castellano, D.; Molinillo, J. M. G. J. Agric. Food. Chem. 2000, 48, 2512–2521.
- 6. Leather, G. R. J. Chem. Ecol. 1983, 9, 983-989.

- 7. Yu, J. Q.; Matsui, Y. J. Chem. Ecol. 1994, 20, 21-31.
- 8. Bedows, E.; Hatfield, G. M. J. Nat. Prod. 1982, 45, 725-728.
- Rimando, A. M.; Dayan, F. E.; Mikell, J. R.; Moraes, R. M. *Nat. Toxins* 1999, 7, 39–43.
- Russel, G. B.; Singh, P.; Fenemore, P. G. Aust. J. Biol. Sci. 1976, 29, 99–103.
- Oliva, A.; Moraes, R. M.; Watson, S. B.; Duke, S. O.; Dayan, F. E. Pestic. Biochem. Physiol. 2002, 72, 45–54.
- Thompson, L. U.; Seidl, M. M.; Rickard, S. E.; Orcheson, L. J.; Fong, H. H. *Nutr. Cancer* **1996**, *26*, 159–165.
- 13. Thompson, L. U.; Rickard, S. E.; Orcheson, L. J.; Seidl, M. M. *Carcinogenesis* **1996**, *17*, 1373–1376.
- Gottlieb, O. R.; Yoshida, M. In Natural products of woody plants; Rowe, J. W., Ed.; Springer: Berlin, 1989; pp 441–511.
- 15. Castro, M. A.; Gordaliza, M.; Del Corral, M. J.; San Feliciano, A. *Phytochemistry* **1996**, *41*, 995–1011.