

Using transition metal tetraphosphonates as precursor of phosphorus-containing electrocatalysts

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Coordination polymers (CPs) are widely studied due to their applicability in many fields.[1] Among them, metal phosphonates (MPs) are attractive materials due to their versatile structural diversity and functionality, with interesting properties as proton conductors and electrocatalyst precursors [2,3].

In this work, we report the synthesis and crystal structures of several MPs derived from the combination of hexamethylenediamine-*N,N,N',N'*-tetrakis(methylenephosphonic acid) (HDTMP) with different transition metals (M^{2+} = Mn, Fe, Co, and Ni). The resulting solids, $M[(HO_3PCH_2)_2N(CH_2)_6N(CH_2PO_3H)_2] \cdot 2H_2O$, show pillared-layered structures with capabilities of ammonia adsorption (Co²⁺ and Ni²⁺ derivatives). The ammonia-containing solids are crystalline, with a composition $M[(HO_3PCH_2)_2N(CH_2)_6N(CH_2PO_3H)_2(H_2O)_2](NH_3)_4(H_2O)_{12}$. The catalytic activities toward Oxygen Evolution Reaction (OER), Oxygen Reduction Reaction (ORR) and Hydrogen Evolution Reaction (HER) of the corresponding (5% H₂-Ar)-pyrolyzed materials, as well as the crystal structure of non-pyrolyzed precursor solids, will be discussed.

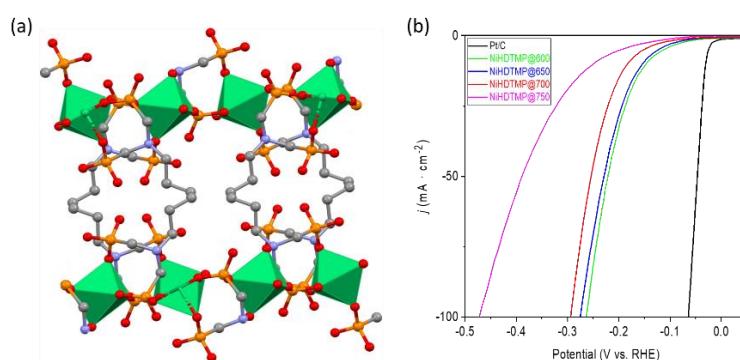


Figure 1. (a) Crystal structure of NiHDTMP and (b) LSV curves of H₂-pyrolyzed derivatives of NiHDTMP toward HER.

References/Referencias

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