



*Citation for published version:*

Mazinani, S, Al-Shimmery, A, Chew, YMJ & Mattia, D 2022, '3D printed nanofiltration composite membranes with reduced concentration polarisation', *Journal of Membrane Science*, vol. 644, 120137.  
<https://doi.org/10.1016/j.memsci.2021.120137>

*DOI:*

[10.1016/j.memsci.2021.120137](https://doi.org/10.1016/j.memsci.2021.120137)

*Publication date:*

2022

*Document Version*

Peer reviewed version

[Link to publication](#)

*Publisher Rights*

CC BY-NC-ND

**University of Bath**

**Alternative formats**

If you require this document in an alternative format, please contact:  
[openaccess@bath.ac.uk](mailto:openaccess@bath.ac.uk)

**General rights**

Copyright and moral rights for the publications made accessible in the public portal are retained by the authors and/or other copyright owners and it is a condition of accessing publications that users recognise and abide by the legal requirements associated with these rights.

**Take down policy**

If you believe that this document breaches copyright please contact us providing details, and we will remove access to the work immediately and investigate your claim.

# 3D printed nanofiltration composite membranes with reduced concentration polarization

Saeed Mazinani, Abouther Al-Shimmery, Y.M. John Chew, Davide Mattia  
Department of Chemical Engineering, Centre for Advanced Separations Engineering,  
University of Bath, Claverton Down, Bath BA2 7AY, UK

## Abstract

3D printed nanofiltration (NF) composite membranes with surface patterns minimising the impact of concentration polarisation (CP) are presented here for the first time. The membranes consist of a NF polydopamine-coated polyvinylidene fluoride (PVDF/PDA) selective layer on a 3D printed asymmetric wavy (patterned) support. The result is a wavy composite membrane with pure water permeance of  $14 \pm 2$  LMH bar<sup>-1</sup> and molecular weight cut-off of ~550 Da, measured using a crossflow NF setup at a transmembrane pressure of 2 bar for Reynold number ( $Re$ ) of 700, using a range of dyes (mass balance > 97% for all tests). The CP behaviour of the composite membranes was assessed by filtration of Congo red (CR) dye solution (0.01 g L<sup>-1</sup>), showing that the wavy pattern significantly reduced the impact of CP compared to the flat membranes, with a nearly tripling of the mass transfer coefficient and a 57% decline of the CP factor. Computational fluid dynamics showed that these significant performance improvements were due to improved hydrodynamics, with the maximum surface shear stress induced by the wavy structure (1.35 Pa) an order of magnitude higher than that of the flat membranes (0.18 Pa) at  $Re = 700$ . These results demonstrate that 3D printing is a viable technology route to reducing concentration polarisation in membrane nanofiltration applications.

**Keywords:** 3D printing, CFD simulation, Wavy composite membrane, 3D asymmetric support, Concentration polarization.

## 31 1. Introduction

32 Concentration polarization (CP) occurs when non-interacting solutes build up near the  
33 membrane-feed interface, resulting in a higher solute concentration than that in the bulk solution  
34 [1]. This build up can induce water flux decline, reduce quality of produced water, increase  
35 power consumption and operating costs in membrane-based water treatment [2, 3], including  
36 reverse osmosis (RO) and nanofiltration (NF) processes [4, 5]. Numerous attempts have been  
37 made to reduce the impact of CP through modification of the membranes, e.g. surface coating  
38 [6], grafting [7] and interfacial polymerization [8]. However, these measures have shown  
39 limited success since they also negatively impact membrane flux and, moreover, the  
40 uncertainty around their long-term stability and scalability has resulted in their low uptake [9].  
41 Therefore, there is intense interest in methods that would effectively reduce CP without  
42 negatively affecting permeance and ensuring stable performance over time.

43 The patterning of membranes' surfaces has been extensively explored as a potential route to  
44 minimise CP and fouling by altering the membrane's surface topography [10, 11]. Micrometre-  
45 scale surface patterns can effectively reduce the build-up of CP layer by promoting fluid shear  
46 stress on the membrane surface, while at the same time enhance permeance by increasing the  
47 effective filtration area [12, 13]. It was demonstrated that a patterned membrane can lead to a  
48 reduction in energy consumption of up to 88% compared to a flat one with the same rejection  
49 [14]. However, the actual use of patterned membranes has been limited due to insufficient  
50 fidelity and flexibility of current patterning methods, e.g. stamping, moulding or  
51 nanoimprinting, as they negatively affect permeance and durability of the membranes [15].

52 Additive manufacturing, known as three-dimensional (3D) printing can produce well-defined  
53 patterned membranes with more complex patterns in a single process that are impossible to  
54 manufacture otherwise [16]. One of the most versatile 3D printing technique that is used in  
55 membrane fabrication for liquid-based filtration is multi-jet printing (MJP). MJP offers a high  
56 resolution, in the order of few micrometres, and is able to create complex parts via the  
57 controlled deposition of light-curable resins [17]. During the fabrication process using this  
58 technique, the 3D printer utilizes two polymeric materials, one to build the structure and the  
59 other as support, which is subsequently removed. Although there are many research studies on  
60 the advancement of 3D printers, only few of them have used 3D printing to fabricate patterned  
61 membranes, and all of which employed symmetric substrates [18].

62 Poly(vinylidene fluoride) (PVDF) membranes are widely used in water and wastewater  
63 treatment applications due to their high mechanical and chemical stability along with good

64 processability [19]. Commercially available in microfiltration (MF) and ultrafiltration (UF)  
65 ranges, PVDF membranes, due to their hydrophobicity, are generally hydrophilized for better  
66 performance [20]. PVDF UF composite membranes have been prepared via coating of  
67 hydrophilic polydopamine (PDA) over MF and loose UF PVDF membranes as support [21-  
68 23]. There is, however, no study on preparation of PVDF NF composite membranes in the  
69 literature.

70 This work demonstrates the fabrication via 3D printing of NF composite membranes with a  
71 wavy pattern, resulting in reduced CP build-up. The asymmetric wavy supports were first  
72 carefully designed to enable the fabrication of thin supports with higher intrinsic permeability,  
73 and better mechanical stability compared to symmetric supports [18]. The structural design  
74 parameters of the wavy supports were optimised integrating computational fluid dynamics  
75 (CFD) simulations with material characterization. A multi-jet 3D printer was then used to print  
76 asymmetric supports. NF selective layers were prepared by coating a polydopamine (PDA)  
77 layer over as-prepared UF PVDF membranes. The NF PVDF/PDA selective layers were then  
78 applied onto the asymmetric supports via vacuum filtration to fabricate 3D printed NF  
79 composite membranes. The MWCO of the composite membranes was determined via dye  
80 rejection tests using a cross-flow NF setup at 2 bar. The dye CP behaviour of the composite  
81 membranes was assessed by filtration of Congo red dye solution through determination of time-  
82 dependent maximum solute concentration, and calculation of mass transfer coefficient and CP  
83 factor. The results were compared with a 3D NF flat composite membrane as a reference.

## 84 **2. Materials and methods**

### 85 **2.1. Materials**

86 A commercial 3D printer (ProJet 3500 HD Max printer, *3D Systems*, USA) was used to  
87 fabricate the asymmetric membrane supports, using proprietary urethane acrylate (VisiJet®  
88 M3-X, *3D Systems*, USA) and a support wax (VisiJet® S300, *3D Systems*, USA). PVDF (Kynar  
89 761, Mw = 634,000 g mol<sup>-1</sup>, *Arkema*), N-Methyl-2-pyrrolidone (NMP, purity > 99%, *Acros*),  
90 dopamine hydrochloride (DPA, purity > 98%, *Sigma-Aldrich*) and Trizma hydrochloride  
91 (TRIS HCl, purity > 99%, *Sigma-Aldrich*) were used to prepare selective layer. Reactive black  
92 5 (RB5, MW = 991 g mol<sup>-1</sup>, *Sigma-Aldrich*), Congo red (CR, MW = 697 g mol<sup>-1</sup>, *Sigma-*  
93 *Aldrich*), acid red 1 (AR1, MW = 509 g mol<sup>-1</sup>, *Sigma-Aldrich*) and methyl orange (MO, MW  
94 = 327 g mol<sup>-1</sup>, *Sigma-Aldrich*) were used as model dyes in the cross-flow nanofiltration  
95 experiments. Sodium sulphate (Na<sub>2</sub>SO<sub>4</sub>, purity > 99%, *Sigma-Aldrich*) was used to prepare the

96 salt solution for the concentration polarisation studies. All the solutions used in this work were  
97 prepared using deionized water (DI, Purelab<sup>®</sup>, ElgaVeolia).

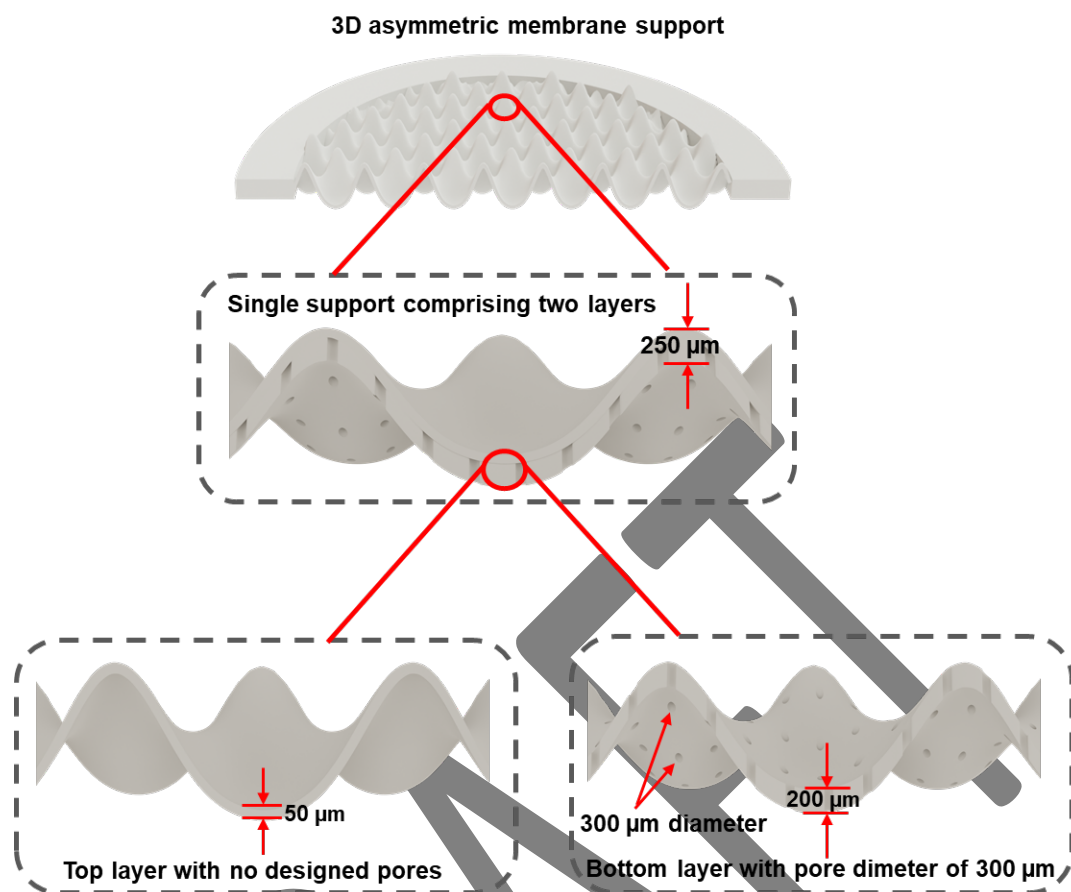
## 98 **2.2. Preparation of the 3D printed nanofiltration composite membranes**

99 The fabrication procedure for the 3D printed nanofiltration (NF) composite membranes  
100 comprises three main steps: (i) Fabrication of the asymmetric supports using 3D printing; (ii)  
101 preparation of the NF selective layer; and (iii) application of the NF selective layer onto the 3D  
102 printed supports.

### 103 **2.2.1. Fabrication of the asymmetric supports using 3D printing**

104 The asymmetric support consists of a single 3D printed object, comprising two sections, a top  
105 dense layer (50  $\mu\text{m}$  thickness) and a bottom one (200  $\mu\text{m}$  thickness) with straight-through  
106 cylindrical pores (**Figure 1**). The asymmetric design actually improves the mechanical stability  
107 of the support and enables the fabrication of the supports with lower overall thickness and  
108 larger pore diameter in the bottom layer compared to the symmetric case [18]. This led to an  
109 increase in the support's intrinsic permeability and eased removal of the wax support material  
110 [24]. The asymmetric support was designed using OpenSCAD, using a procedure reported  
111 previously [18].

112



113

114 **Figure 1.** The 3D asymmetric support consists of a single 3D printed object, comprising two  
 115 sections, a top dense one (50 μm thickness) and a bottom one (200 μm thickness) with straight-  
 116 through cylindrical pores of 300 μm.

117

118 The pores were designed with 300 μm diameter and 200 μm interpore distance using Autodesk  
 119 Inventor Professional 2016 and superimposed on the wavy surface (37 mm diameter) of the  
 120 bottom section. Thereafter, the two sections were merged into a single digital object and a rim  
 121 (6.5 mm wide) was added around the circular wavy area to generate an asymmetric support  
 122 with 50 mm overall diameter. The final computer-aided design (CAD) file was subsequently  
 123 converted into the stereolithography (STL) format using OpenSCAD and transferred to the 3D  
 124 printer using 3D Sprint software (2.0, 3D systems, USA). The flat supports were designed using  
 125 the same procedure. The final post-treatment step was to remove the wax from inside the pores  
 126 of the subsection using EZ Rinse-C oil [18].

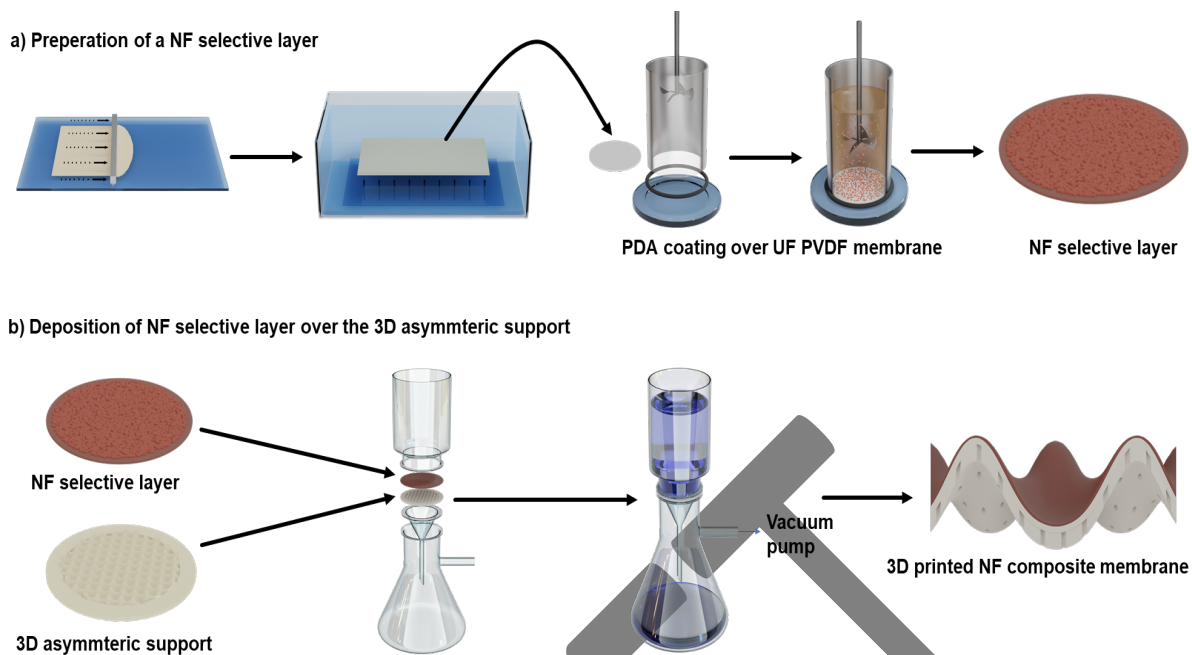
### 127 2.2.2. Preparation of NF selective layer

128 The NF selective layer was obtained by coating PDA over an UF PVDF membrane. To prepare  
 129 a dope solution, PVDF (17.5 wt%) was first dissolved in NMP and then stirred using a roller

130 mixer (120 VAC, *Cole-Parmer*) for two days. The dope solution was then left overnight to  
131 degas. The bubble-free solution was cast on top of a clean glass plate at room temperature using  
132 a casting knife with 50  $\mu\text{m}$  gap height under  $\sim 35\%$  relative humidity (**Figure 2(a)**). After 15 s  
133 of exposure to air, the glass plate was immersed in a DI water bath at room temperature to  
134 complete the phase inversion process. The DPA solution ( $2\text{ g L}^{-1}$ , pH $\sim 8$ , 300 ml) preparation  
135 involved first dissolution in DPA in TRIS HCl solution (pH = 8.5, 10 mM, 200 ml), followed  
136 by pH adjustment to 8.5 by addition of sodium hydroxide solution (NaOH, 10 mM, 100 ml).  
137 The as-prepared PVDF membrane was cut 6 cm diameter discs and placed into a customised  
138 membrane holder so that only its front side was exposed to the coating solution. Subsequently,  
139 the fresh DPA was poured into the membrane holder and continuously stirred using an  
140 overhead stirrer for 24 h at room temperature. The DPA solution was then replaced with fresh  
141 solution and the membrane was exposed to the fresh solution for another 24 h. The interaction  
142 between PVDF and DPA forms a thin PDA layer on the PVDF surface through oxidant self-  
143 polymerization [25, 26], resulting in a NF selective layer (**Figure 2(a)**). Thereafter, the NF  
144 selective layer was removed from the membrane holder and rinsed with water for 30 mins to  
145 remove unreacted PDA particles. The resultant membrane was stored in DI water before further  
146 use.

### 147 **2.2.3. Preparation of 3D printed NF composite membranes**

148 The NF PVDF/PDA selective layer was first applied onto the 3D support, followed by vacuum  
149 filtration with DI water for at least 3 h to ensure adhesion of the selective layer on the 3D  
150 printed support, resulting in the formation of wavy or flat 3D composite membranes (**Figure**  
151 **2(b)**).



153 **Figure 2.** Schematic of the preparation of wavy 3D composite membranes: (a) preparation of  
 154 NF selective layer by coating a PDA layer over a UF PVDF membrane, (b) NF selective layer  
 155 deposition over the 3D support via vacuum filtration.

156

### 157 2.3. Characterisation of 3D composite membrane

158 A scanning electron microscope (SEM, JEOL SEM6480LV, Japan) along with a digital  
 159 microscope (VHX-6000, Keyence, Japan) were used to investigate the morphology of the 3D  
 160 support, selective layer, and 3D composite membrane. Membrane samples were first dried  
 161 under vacuum ( $\sim 40$  °C) overnight and then sputter-coated with 10 nm of chromium (Edwards  
 162 Sputter Coater S150B, Mechatech Systems, UK) for SEM imaging.

163 To characterise chemical bonds on the surface of the selective layer, an ATR-FTIR  
 164 spectrometer (Frontier, PerkinElmer, Germany) was used. The spectra were collected as a  
 165 result of 16 running scans at a resolution of  $4\text{ cm}^{-1}$  within the  $600\text{ cm}^{-1}$  to  $4000\text{ cm}^{-1}$   
 166 wavelength range.

167 The surface chemical composition of the selective layer was analysed by X-ray photoelectron  
 168 spectroscopy (XPS, K-alpha, Thermo Fisher Scientific) using Al K $\alpha$  as the radiation source  
 169 over an area of around  $400\text{ }\mu\text{m}$ . Data was collected at pass energies of 150 eV and 40 eV for  
 170 survey (1 eV step size) and high (0.1 eV step size) resolution scans, respectively.



171 The water contact angle of the support, selective layer, and composite membrane was  
172 determined using a contact angle goniometer (OCA20, *DataPhysics Instruments*, Germany) in  
173 sessile mode at room temperature, using 5  $\mu\text{l}$  droplets. The reported values are the average of  
174 at least 10 measurements.

175 The surface roughness of the 3D supports was assessed using a digital microscope (VHX-6000,  
176 *Keyence*, Japan) over scan areas of  $100 \times 100 \mu\text{m}$ . The surface roughness of the selective layer  
177 was analysed using atomic force microscopy (AFM, *Bruker Multimode IIIA*, USA) over scan  
178 areas of  $5 \times 5 \mu\text{m}$  in the tapping mode (time/line = 1 s, samples/line = 256).

179 A Zetasizer Nano (ZS, Malvern Instruments Ltd.) was used to measure the surface zeta  
180 potential of membrane samples at neutral pH = 7.0. The reported values are the average of at  
181 least three measurements.

#### 182 **2.4. Filtration performance and CP characterisation**

183 The filtration performance and CP behaviour of the composite membranes was evaluated using  
184 a cross-flow NF setup (**Figure S2**) described previously [18]. All experiments were carried out  
185 at crossflow velocity of  $0.1 \text{ ms}^{-1}$  (corresponding to  $Re = 700$ ) at TMP of 2 bar for 30 minutes,  
186 following a pre-compaction step at 3 bar for 24 h using pure water. The permeance ( $K$ , LMH  
187  $\text{bar}^{-1}$ ) of the membranes was calculated via the following:

$$K = \frac{V}{\Delta t \cdot A \cdot \Delta p} \quad (1)$$

188 where  $V$  is the permeate volume ( $\text{m}^3$ ),  $\Delta t$  is time (hr),  $A$  is the effective membrane area ( $\text{m}^2$ ),  
189 and  $\Delta p$  is the transmembrane pressure (bar). The effective filtration areas for the flat and wavy  
190 asymmetric supports are 1074 and 1288  $\text{mm}^2$ , respectively, calculated using 3D Sprint (*3D*  
191 *Systems*, USA) software [24].

192 The dye rejection of the membranes was calculated using equation (2):

$$R (\%) = \frac{C_F - C_P}{C_F} \quad (2)$$

193 where  $C_F$  and  $C_P$  represent the solute concentration in the feed and permeate, respectively.

194 Dye concentration was measured using a UV-Vis spectrophotometer (*Cary, Agilent*, USA).

195 The mass balance in dye rejection tests was calculated using equation (3):

$$\text{mass balance (\%)} = \frac{C_p V_p + C_r V_r}{C_F V_F} \times 100 \quad (3)$$

216 where  $C_r$  is the retentate concentration ( $\text{g L}^{-1}$ ),  $V_p$ ,  $V_r$ ,  $V_F$  represent the permeate, retentate and  
 217 feed solutions volumes (L), respectively.

218 To assess the CP behaviour of the membranes, Congo red dye solution ( $0.01 \text{ g L}^{-1}$ ) was  
 219 circulated through the setup and the membrane flux  $J_v$  (LMH) was measured. The mass transfer  
 220 coefficient  $k$  ( $\text{m s}^{-1}$ ) was calculated from equation (4) using the stagnant film model [27]:

$$k = \frac{J_v}{\ln \left( \frac{C_m - C_p}{C_F - C_p} \right)} \quad (4)$$

201 where  $J_v$  is the permeation flux (LMH),  $C_m$  is the maximum solute concentration at the  
 202 membrane surface ( $\text{g L}^{-1}$ ),  $C_p$  is the permeate concentration ( $\text{g L}^{-1}$ ) and  $C_F$  is the feed  
 203 concentration ( $\text{g L}^{-1}$ ).  $C_F$  and  $C_p$  were calculated from experimental measurements while the  
 204 concentration of dye at the membrane surface,  $C_m$ , was estimated using computational fluid  
 205 dynamics (next section).

206 The CP factor,  $\Gamma$ , was determined using equation (5) [10]:

$$\Gamma = \frac{C_m - C_F}{C_F} \quad (5)$$

207

## 208 **2.5. CFD simulations**

209 To study flow and concentration fields over the surface of flat and wavy composite membranes,  
 210 the conservation of mass and momentum equations along with mass transport equations for  
 211 steady-state flow in the laminar flow regime, were numerically solved using Laminar Flow and  
 212 Transport of Diluted Species interfaces of COMSOL Multiphysics (COMSOL 5.6, *Comsol*  
 213 *Inc.*, USA) software. Two-dimensional (2D) simulation domains (**Figure S1**) were created to  
 214 reproduce the cross section of the filtration cell with overall size of  $4 \text{ mm} \times 50 \text{ mm}$  (height  $\times$   
 215 length) [18]. The wavy structure was set at amplitude  $\alpha = 0.70 \text{ mm}$ , and wavelength  $\lambda = 3 \text{ mm}$   
 216 following optimisation. Further details of the simulation are reported in the **Supporting**  
 217 **Information**.

218 CFD modelling was also used to simulate the permeance decline as a function of time for the  
 219 dye solution. This was then used to determine the local concentration of the dye at the  
 220 membrane surface,  $C_m$ , which, in turn, allowed calculating the mass transfer coefficient in

221 equation (4). The simulated flux decline was compared to experimental profiles, with an  
222 excellent match, discussed below. The CFD model has been previously validated using BSA  
223 by the authors [18] and independently [28].

224

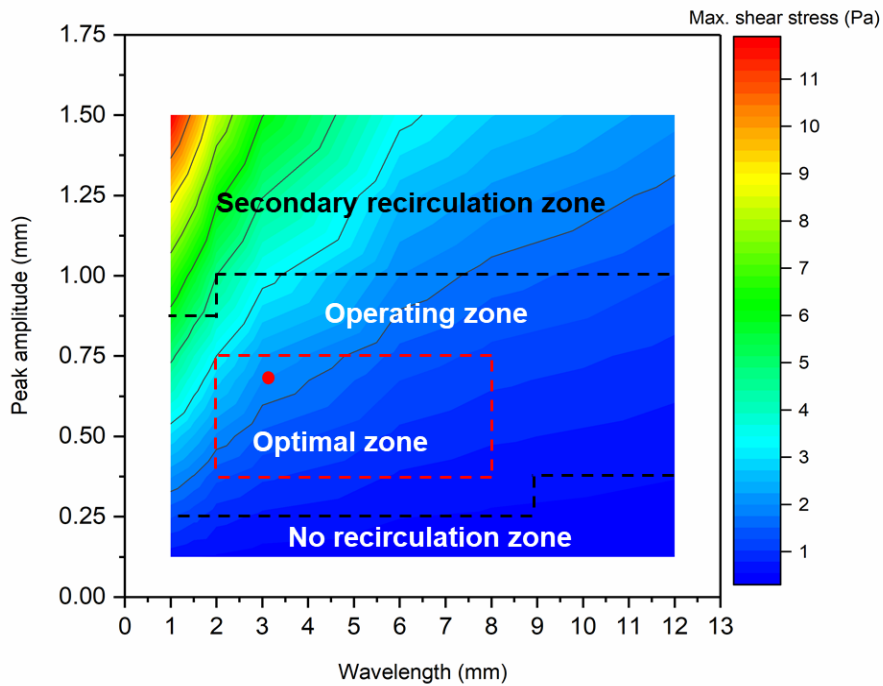
### 225 **3. Results and discussion**

#### 226 **3.1. Design and optimisation of 3D printed composite membranes**

227 The optimisation process of the 3D printed wavy composite membranes has been described  
228 previously [18]. A combination of CFD simulations along with material characterisation was  
229 employed to optimise the wavy structure parameters including amplitude ( $\alpha$ ), and wavelength  
230 ( $\lambda$ ). Several wavy supports with certain amplitude and wavelength were first fabricated using  
231 the 3D printer and their morphology was then characterised using SEM and digital microscope  
232 to identify a range for amplitude ( $\alpha$ , 0-1.50 mm) and wavelength ( $\lambda$ , 0-12 mm) compatible with  
233 the printing process. The minimum thickness of the wavy support is dictated by its mechanical  
234 properties where below a certain thickness the support can no longer withstand any significant  
235 pressure. For symmetric supports the minimum thickness was found to be 500  $\mu\text{m}$  whereas  
236 asymmetric design allowed reducing the minimum thickness to 250  $\mu\text{m}$ , thereby reducing  
237 support resistance and material consumption. Moreover, the asymmetric support having no  
238 pores in the top prevents the selective layer from breaking during vacuum filtration which as  
239 observed for symmetric wavy supports.

240 The fluid flow around the wavy surface in terms of surface shear stress and vortex pattern was  
241 analysed using CFD simulations. **Figure 3** shows a contour plot with different shear stress  
242 regions caused by the wavy structure based on printable values of amplitude and wavelength  
243 at  $Re = 700$ .

244



245

246 **Figure 3.** Contour plot with different shear stress regions caused by the wavy structure based  
 247 on printable values of  $\alpha$  and  $\lambda$  at  $Re = 700$  using a 3D printed asymmetric support. The red dot  
 248 represents the optimal fabrication parameters used in this work.

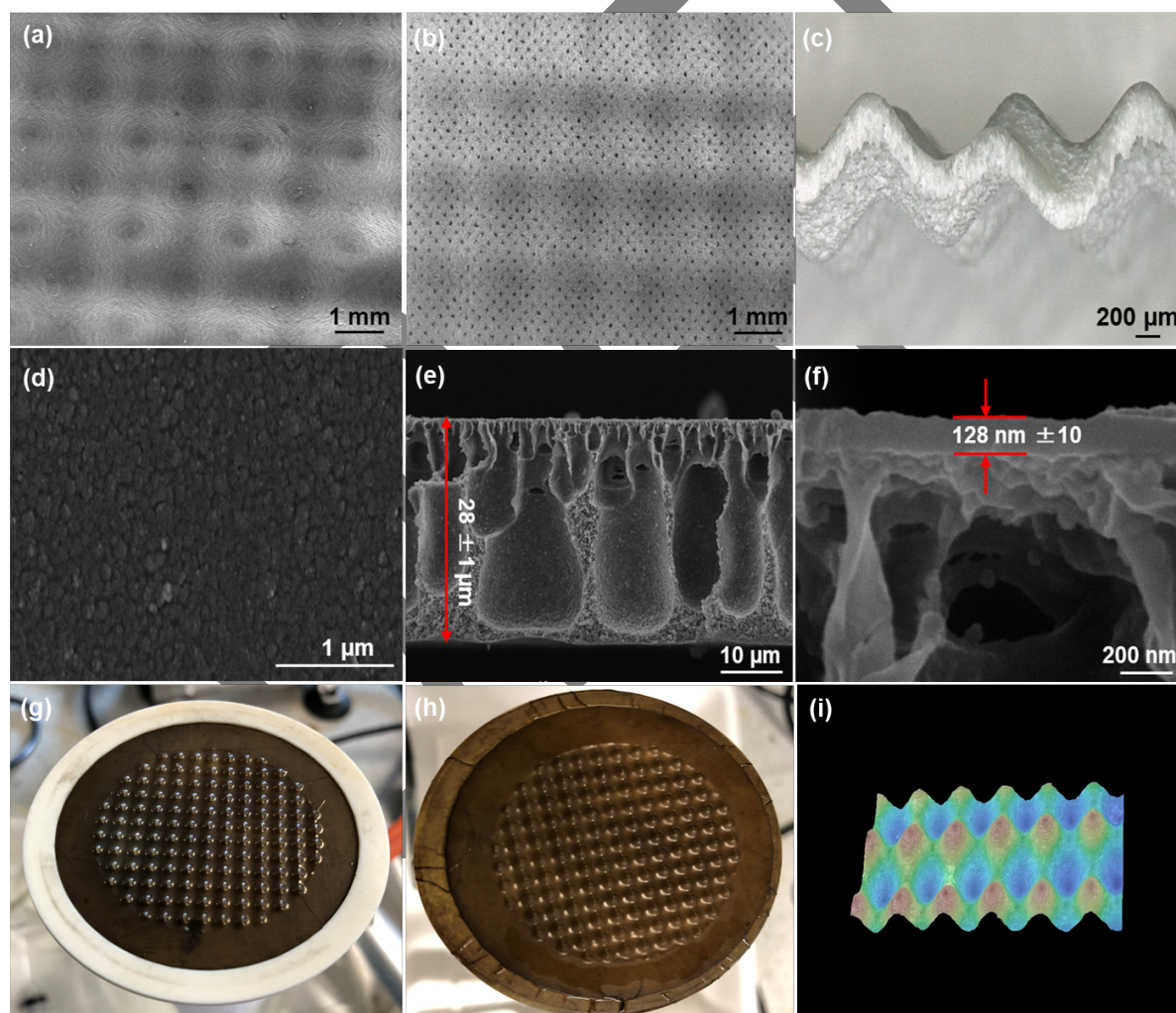
249 Wavy structures with large amplitudes and short wavelengths induce the highest fluid shear  
 250 stress while small amplitudes and long wavelengths cause lowest shear stress. High shear stress  
 251 and vortex formation at the membrane's surface enhance mass transfer coefficient and  
 252 therefore reduce concentration polarisation [29]. Within the operating zone, materials  
 253 considerations further limited the printable range for amplitude and wavelength: For  $\alpha > 0.5$   
 254 mm, the selective layer prepared from polyethersulfone (PES) would be pierced-through during  
 255 the vacuum-driven adhesion step. To improve the selective layer's conformability for larger  
 256  $\alpha$  values, PVDF was selected to use as the main polymer for preparation of the selective layer  
 257 due to its outstanding mechanical properties [30]. This systematic optimisation expanded the  
 258 'optimal zone', by raising the upper limit for  $\alpha$  from 0.63 mm to 0.75 mm. In turn, higher  $\alpha$   
 259 values result in higher shear stress, further reducing fouling. This optimisation process led to  
 260 the selection of  $\alpha = 0.70$  mm and  $\lambda = 3$  mm to induce maximum shear stress (represented by  
 261 the red dot in **Figure 3**) over the membrane surface. By using PVDF as the selective layer, it  
 262 was possible to increase the amplitude  $\alpha$  by 40% from 0.5 mm to 0.7 mm, resulting in  
 263 approximately doubling the shear stress from 0.7 Pa to 1.35 Pa, compared to 0.18 Pa for the  
 264 flat 3D printed membrane ( $\alpha = \lambda = 0$ ). An extensive discussion of the interplay between the

265 different parameters of the 3D printer, the selective layer and the hydrodynamic landscape has  
266 been reported elsewhere [18].

267

### 268 3.2. Characterization of the 3D printed support, selective layer, and 3D composite 269 membrane

270 SEM micrographs of the 3D printed top dense section (**Figure 4a**) and bottom section with  
271 cylindrical pores with diameter of 300  $\mu\text{m}$  (**Figure 4b**) show a highly regular structure with no  
272 defects and following the wavy structure design (cfr. **Figure 1**). Although the top layer is dense,  
273 i.e. there were no pores designed into this, there is nonetheless some porosity due to the  
274 resolution of the 3D printer itself, as previously observed for polysulfone membranes [31].

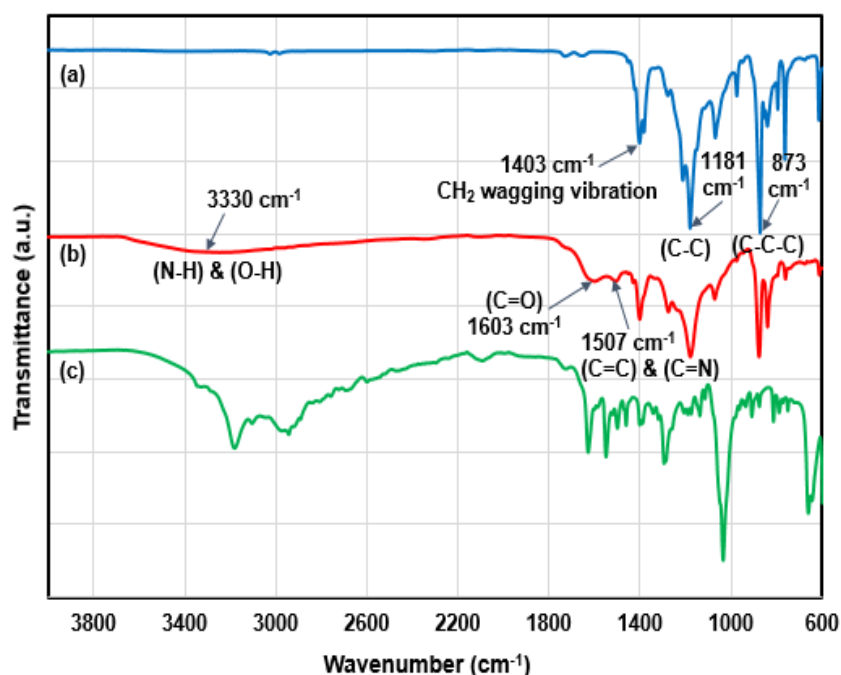


275

276 **Figure 4.** SEM micrographs of (a) top dense section, (b) bottom section (with pore diameter  
277 of 300  $\mu\text{m}$ ) of the asymmetric 3D printed wavy support; (c) digital micrograph of the cross-  
278 section of 3D printed wavy support; SEM micrographs of (d) surface, and cross section of the  
279 PVDF/PDA selective layer at (e)  $\times 55\text{K}$  and (f)  $\times 75\text{K}$  magnification; optical micrographs of 3D

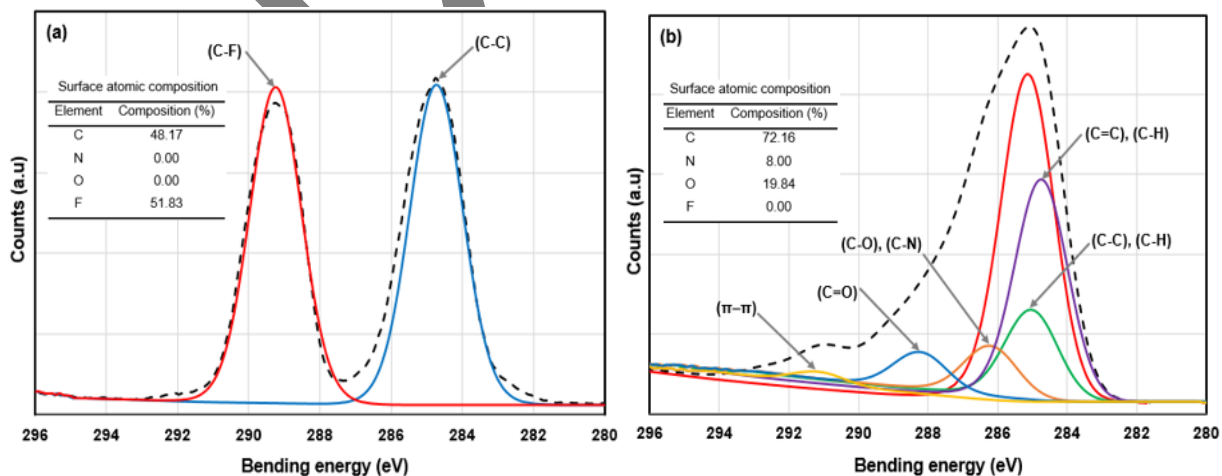
280 wavy composite membrane from (g) top and (h) side, showing a conformal layer of the  
281 PVDF/PDA onto the wavy support; (i) digital micrograph of 3D wavy composite membrane.  
282 **Figure 4 (c)** shows the cross section of the wavy support with no gaps between the two layers  
283 confirming the asymmetric support was successfully printed as a single continuous object with  
284 lower total thickness ( $\sim 250 \mu\text{m}$ ) and higher intrinsic permeability ( $9 \pm 1 \times 10^{-12} \text{ m}^2$ ) compared  
285 to the symmetric supports [18]. The SEM micrographs of the surface and cross section of the  
286 PVDF/PDA selective layer are shown in **Figure 4 (d-f)**. The densely packed PDA particles can  
287 be clearly observed on the surface of the selective layer (**Figure 4(d)**). The deposition of PDA  
288 can lead to the formation of numerous nanostructured PDA aggregates formed from the non-  
289 covalent interactions between PDA oligomers [32]. The cross-section of the selective layer  
290 (**Figure 4e**) presents a typical asymmetric structure (overall thickness of  $28 \pm 1 \mu\text{m}$ ) with a  
291 micro-void porous sublayer and a thin dense skin layer (average thickness of  $128 \pm 10 \text{ nm}$ )  
292 formed from PDA (**Figure 4f**). A conformal adhesion between the PVDF/PDA selective layer  
293 and membrane support was confirmed using the optical micrographs and topographical image  
294 of the wavy 3D composite membrane (**Figure 4 (g-i)**).

295 The FTIR spectra of the (a) pristine PVDF (b) PVDF/PDA selective layer and (c) pristine PDA  
296 are reported in **Figure 5**. In the PVDF spectra, the characteristic peaks at  $1403 \text{ cm}^{-1}$  and  
297  $1181 \text{ cm}^{-1}$  are attributed to  $\text{CH}_2$  wagging vibration and C–C band, respectively, while the  
298 sharp peak at  $873 \text{ cm}^{-1}$  is assigned to C–C–C asymmetrical stretching vibration [33]. In the  
299 PVDF spectra, the characteristic peaks at  $1403 \text{ cm}^{-1}$  and  $1181 \text{ cm}^{-1}$  are attributed to  
300  $\text{CH}_2$  wagging vibration and C–C band, respectively, while the sharp peak at  $873 \text{ cm}^{-1}$  is  
301 assigned to C–C–C asymmetrical stretching vibration [33]. In the spectra of the coated PVDF  
302 membrane with PDA, three new peaks can be observed: A broad peak between  $3100 \text{ cm}^{-1}$  and  
303  $3600 \text{ cm}^{-1}$  corresponds to N–H/O–H stretching vibrations of PDA, while the two small peaks  
304 at  $1603 \text{ cm}^{-1}$  and  $1507 \text{ cm}^{-1}$  are attributed to C=O and C=C/C=N bands, respectively,  
305 confirming a strong bond between PVDF and PDA [30].



306  
 307 **Figure 5.** ATR-IR spectra of (a) pristine PVDF, (b) PVDF/PDA selective layer, and (c)  
 308 pristine PDA.

309  
 310 The C1s XPS spectra along with surface atomic composition of pristine PVDF and PVDF/PDA  
 311 membranes are shown in **Figure 6 (a)** and **(b)**, respectively. For the former, only C and F  
 312 elements are present on the surface of pristine PVDF membrane, as expected. After coating  
 313 with PDA, the atomic composition of C significantly increased, with signals appears for N and  
 314 O elements appeared, and no detection of F, confirming the formation of a uniform coating.  
 315 This data agrees well with the literature [30]. This was further confirmed looking at the C1s  
 316 XPS spectra of pristine PVDF and PVDF/PDA membrane. For the pristine PVDF, only two  
 317 peaks at binding energies of 284.6 eV (C-C) and 289.1 eV (C-F) are observed [34].



318 **Figure 6.** C1s X-ray photoelectron spectra of (a) pristine PVDF membrane and (b) PVDF/PDA  
 319 membrane; the inset tables summarise the surface atomic composition.

320

321 In the C1s XPS spectra of the PVDF/PDA membrane, the C-F band has disappeared and new  
 322 peaks appeared at binding energies of 284.6 eV (C=C/C-H), 285 eV (C-C/C-H), 286.3 eV (C-  
 323 C/C-O), 288.1 eV (C=O), and 291.2 eV ( $\pi$ - $\pi$  stacking) [30]. These new bands confirm the  
 324 presence of PDA on the PVDF surface as PDA can adhere to polymeric surface through amino,  
 325 imino, hydroxyl and catechol functional groups [35]. The O1s spectra of PVDF/PDA  
 326 membrane can be found in **Figure S3**.

327

328 The properties of the 3D printed support, pristine PVDF, PVDF/PDA selective layer and 3D  
 329 wavy composite membrane are reported in **Table 1**. The water contact angle of the pristine  
 330 PVDF dropped from 93° to 51° after coating with PDA. This is attributed to the hydrophilic  
 331 amine and catechol groups of PDA [23], further confirming the successful coating of PDA onto  
 332 the PVDF surface and increased its hydrophilicity. Compared to the pristine PVDF membrane  
 333 (-70.9 ± 3 mV), the PVDF/PDA selective layer showed a higher negatively charge (-89.1 ± 5  
 334 mV) due to the existing functional groups in the PDA. The 3D printed support has the highest  
 335 roughness value ( $R_a = 190 \pm 15$  nm) which can be related to MultiJet Printing (MJP) technology  
 336 used by the 3D printer which deposits the building material layer by layer on the top of each  
 337 other, resulting in a rough surface [17]. The roughness of the 3D wavy composite membrane  
 338 is the same as that of the PVDF/PDA selective layer, confirming the formation of a continuous,  
 339 uniform coating onto the 3D printed support.

340

341 **Table 1.** Physical properties of the 3D support, pristine PVDF, PVDF/PDA selective layer and  
 342 3D wavy composite membrane

Membrane	Contact angle (°)	Surface Zeta Potential (mV) <sup>(a)</sup>	Roughness parameters (nm)	
			$R_a$	$R_{RMS}$
3D support	85 ± 2	-----	251 ± 15	-----
Pristine PVDF	93 ± 2	-70.9 ± 3	22.7 ± 0.6	29.7 ± 0.6
PVDF/PDA selective layer	51 ± 3	-89.1 ± 5	90.1 ± 8.0	121.7 ± 9.7
3D wavy composite membrane	51 ± 3	-89.1 ± 5	90.1 ± 8.0	121.7 ± 9.7

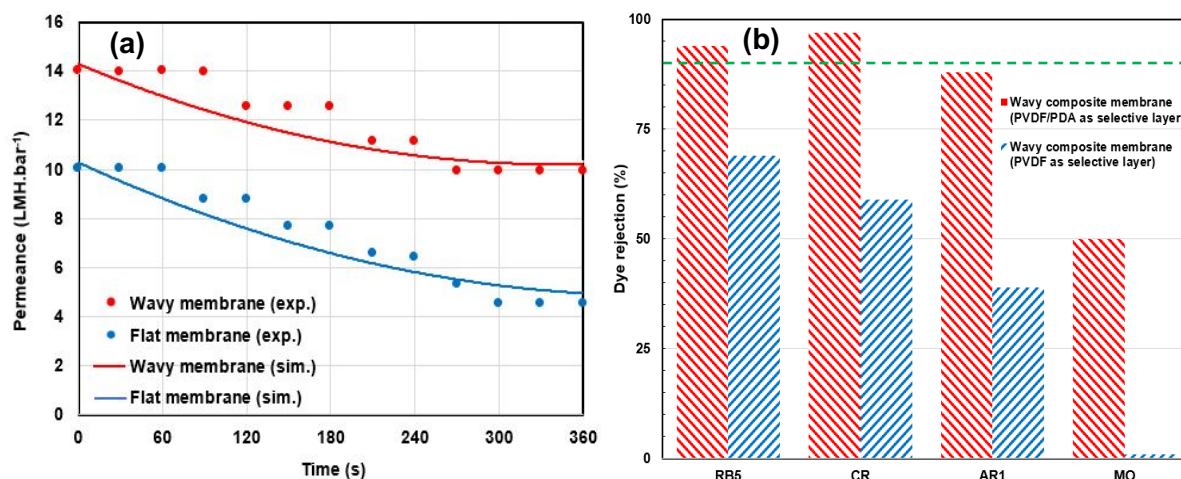
343 <sup>(a)</sup>The surface zeta potential was measured at pH = 7.0.



### 344 3.3. Filtration performance and CP characterisation

345 The 3D wavy composite membranes prepared using pristine PVDF had a pure water permeance  
346 (PWP) of  $41 \pm 1.5$  LMH bar<sup>-1</sup> and 67% rejection for RB5 (MW = 991 Da), showing UF  
347 performance. The PWP for wavy and flat composite membranes prepared from PVDF/PDA  
348 selective layer are  $14 \pm 2$  LMH bar<sup>-1</sup> and  $10 \pm 2$  LMH bar<sup>-1</sup>, respectively (**Figure 7a**, at time  
349 0). The higher PWP for wavy membranes is attributed to two factors: first, a ~20% larger  
350 surface area compared to the flat one, even though they have the same footprint (50 mm). This  
351 can be considered as one additional advantage of manufacturing patterned membranes using  
352 3D printing compared to conventional techniques, where the size and the number of pores does  
353 not reduce due to the printer's high resolution [36]. Second, a reduced thickness of the laminar  
354 sub-layer near the membrane surface due to turbulence/eddies caused by the wavy structure, as  
355 evidenced by the reduced fouling build-up (**Figure 8**). When the feed was switched to dye  
356 solution, permeance decreased due to concentration polarisation, reaching steady state after ~5  
357 minutes of permeation, with the wavy membrane showing a lower flux decline than the flat  
358 one (**Figure 7a**). This behaviour is consistent with BSA fouling tests on flat and wavy 3D  
359 printed UF composite membranes [18]. The NF performance of the wavy composite  
360 membranes was evaluated via rejection of a series of dyes, including RB5, CR, AR1 and MO,  
361 with different molecular weights at 2 bar and feed concentration of 0.01 g L<sup>-1</sup> (**Figure 7b**).  
362 RB5 and CR dyes exhibited a high rejection of > 94%, AR1 had a slightly lower rejection of  
363 88%, and MO had a low rejection of 50%. The mass balance for all rejection tests was above  
364 97%, signalling that the adsorption of the dyes onto the membranes was negligible. The main  
365 rejection mechanisms in NF membranes are size (steric) exclusion and Donnan (charge)  
366 exclusion [37, 38]. As both the composite membranes (**Table 1**) and the dyes tested have  
367 negative charge, it can be assumed that the dominant factor is size sieving. Since RB5, CR and  
368 AR1 have larger molecular sizes compared to MO, high resistance of the former dyes results  
369 in their high rejection in the filtration tests. The flat composite membranes prepared from  
370 PVDF/PDA selective layer showed similar dye rejection to the wavy composite membranes  
371 due to the fact that the rejection performance is attributed to the selective layer only [39]. The  
372 MWCO value of the flat and wavy composite membrane is determined to be around ~550 Da.  
373 Calibration curves for the dye rejection measurements are presented in **Figure S4**.

374



375

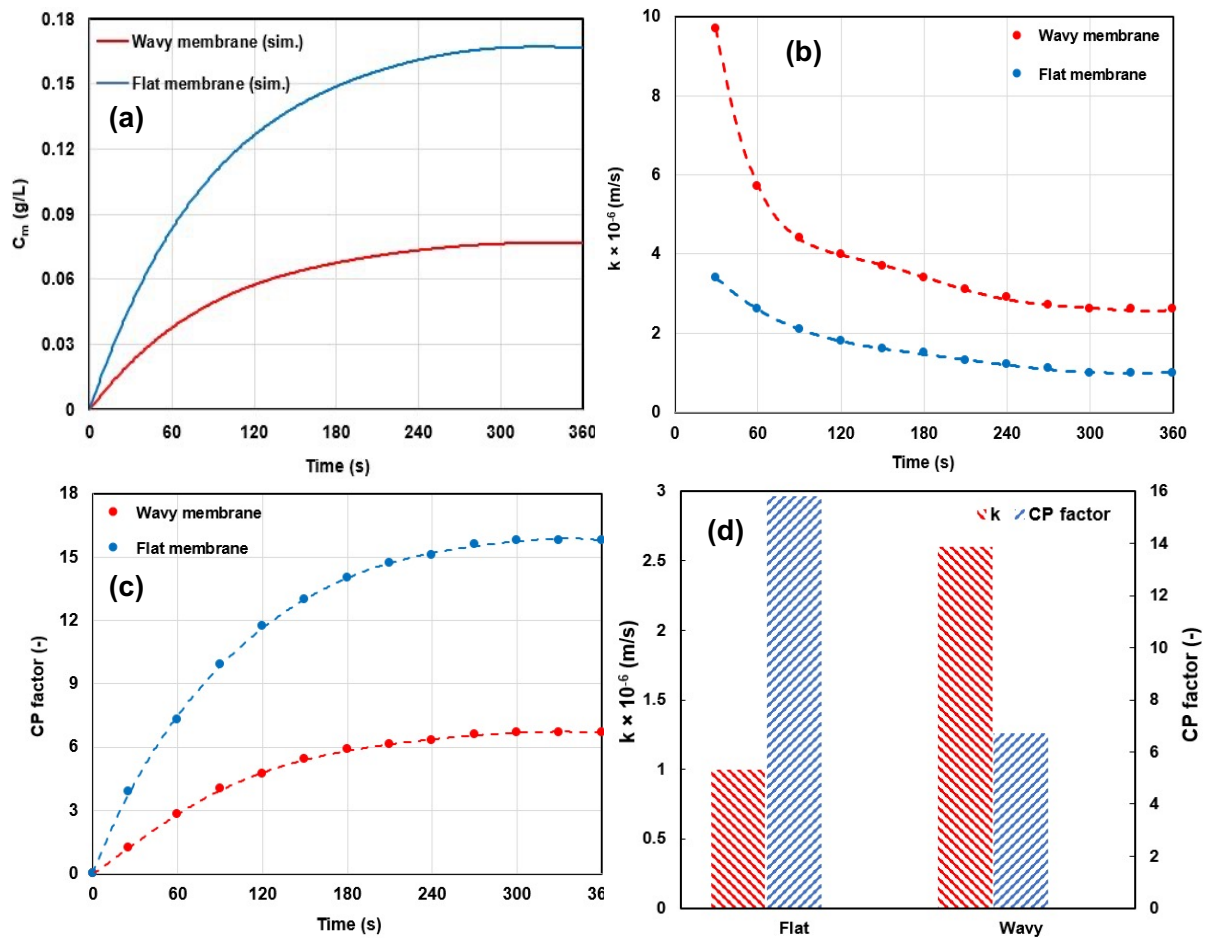
376 **Figure 7.** (a) Dye solution permeance decline for the first 5 minutes of permeation for wavy  
 377 and flat membranes: dots report experimental data points, whereas the continuous line comes  
 378 from CFD simulations; (b) Dye rejection performance of the wavy composite membranes  
 379 prepared from pristine PVDF and PVDF/PDA as selective layers at 2 bar and inlet  
 380 concentrations of 0.01 g L<sup>-1</sup>. RB5: reactive black 5; CR: Congo red; AR1: acid red 1; MO:  
 381 methyl orange. The mass balance for all the filtration experiments was > 97% (green dotted  
 382 line refers to 90% rejection).

383

384 Concentration polarisation is governed by two simultaneous processes: (1) convective transport  
 385 of particles towards the membrane along with the permeate flow and (2) back-transport of  
 386 particles from the concentrated layer into the bulk phase [40]. As such, the extent of CP can be  
 387 quantified by using the solute concentration at the membrane surface, the mass transfer  
 388 coefficient  $k$  and the CP factor [41]. A lower solute concentration at the membrane surface  
 389 means a higher  $k$  value and, therefore, lower CP. To calculate  $k$  (equation (4)), the solute  
 390 concentration at the membrane surface ( $C_m$ ) is needed, but this cannot be easily measured  
 391 experimentally [42]. On the other hand, CFD simulations have been effectively used to  
 392 calculate solute concentrations at the membrane surface, with high accuracy and reliability [43-  
 393 45]. CFD simulation results can be very accurate if the conservation of mass and momentum  
 394 along with mass transport equations are solved under laminar flow conditions [42, 46]. In this  
 395 work, the CFD simulations were performed at TMP of 2 bar and  $Re = 700$  to determine the  
 396 maximum CR (solute) concentration at the surface of flat and wavy composite membranes.  
 397 Results confirm that the CFD permeance curves accurately model experimental data (**Figure**  
 398 **7a**). Further details of the model development can be found in the **Supporting Information**.  
 399 The steady state fluxes of CR dye solution (0.01 g L<sup>-1</sup>) at 2 bar for the wavy and flat membranes

400 are 20 LMH and 10 LMH, respectively (**Table S1**). The maximum CR concentration ( $C_m$ ),  
401 mass transfer coefficient  $k$  and CP factor as a function of filtration time for flat and wavy  
402 composite membranes are shown in **Figure 8 (a – c)**.  $C_m$  increases with time and reached a  
403 steady state after 300 s, with values of  $0.17 \text{ g L}^{-1}$  on the flat surface and  $0.08 \text{ g L}^{-1}$  on the wavy  
404 surface. The steady state is given by a balance between solute flow to the membrane surface,  
405 solute flux through the membrane and backflow from the surface to the bulk. A lower  $C_m$  value  
406 indicates that a higher mass transfer coefficient was achieved for the wavy membrane  
407 compared to the flat one (**Figure 8 (b)**). The CP factor for the wavy membrane is consistently  
408 lower than that of the corresponding time on the flat membrane in all instances (**Figure 8 (c)**).  
409 A quantitative comparison between wavy and flat composite membranes in terms of  $k$  and CP  
410 factor is reported in **Figure 8 (d)**, showing that the final (at  $t = 300 \text{ s}$ ) mass transfer coefficient  
411 of the wavy membrane ( $2.7 \times 10^{-6} \text{ m/s}$ ) is nearly triple that of the flat membrane ( $1.0 \times 10^{-6}$   
412  $\text{m/s}$ ), with the CP factor being 57% lower on the wavy surface. The data and calculations for  
413 these values are reported in **Table S1**. As CP depends on various parameters such as feed  
414 solution, feed concentration, membrane type, crossflow velocity and transmembrane pressure  
415 [1], comparison with the literature is challenging. In the case of patterned membranes, this is  
416 made more complex by considerations about which area to use, i.e. the membrane's footprint  
417 (in the present case the wavy and flat membranes having the same) or the effective filtration  
418 area (in the present case the wavy membrane has  $\sim 20\%$  larger effective filtration area than the  
419 flat one). The approaches vary in the literature, but most consider the footprint as this coincides  
420 with the membrane's module [10, 18, 41]. On this basis, the increase in mass transfer  
421 coefficient and reduction in CP factor obtained in this work are very similar to values obtained  
422 via CFD for a patterned membrane with a cambered structure, reporting a x3 increase for mass  
423 transfers and a 50% reduction in CP factor for Rhodamine 6G [41]. A higher permeance  
424 coupled with a significant reduction in CP was also observed for micropatterned membranes  
425 using NaCl [10], whereas no increase in permeance was associated with reduced colloidal  
426 fouling for a nanoimprinted UF membrane [15]. It is worth mentioning that the osmotic  
427 pressure difference ( $\Delta\pi$ ) of the Congo red solution, due to the low CR concentration, is  
428 negligible (**Supporting Information**).

429



430

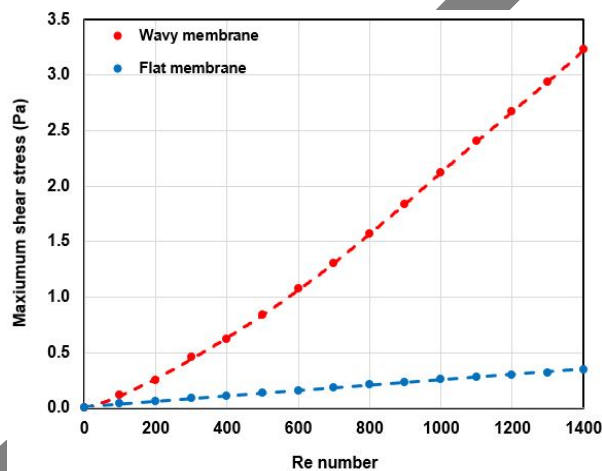
431 **Figure 8.** Time-dependent (a) maximum solute concentration at the membrane's surface; (b)  
 432 mass transfer coefficient; (c) CP factor along with (d) quantitative comparison of wavy and flat  
 433 composite membranes at TMP of 2 bar under  $Re = 700$ .

434

435 This divergent behaviour can be explained by analysing the shear stress over the wavy and flat  
 436 composite membranes determined from CFD simulations (**Figure 9**), as the hydrodynamic  
 437 conditions around the membrane surface can have a significant effect on CP. In fact, there is  
 438 strong evidence in the literature that shear stress plays an important role in suppressing the  
 439 formation of concentration polarization formation [41, 47-49]. It can be observed from **Figure**  
 440 **9** that the maximum shear stress increase by an order of magnitude, from 0.18 Pa on the flat  
 441 surface to 1.35 Pa on the wavy surface. It is worth mentioning that  $Re = 700$  was chosen so that  
 442 the shear stress on the surface of the wavy membrane is within the range (0 - 1.5 Pa) used in  
 443 NF membranes for industrial wastewater treatment [50]. Higher  $Re$  values would have further  
 444 enhanced the effects of the wavy structure. The wavy structure induces a non-uniform shear  
 445 stress distribution on the membrane surface where high shear stress/rate values occur at the

446 peaks and low shear stress/rate values occur in the valleys of the wavy structure. The average  
447 shear stress on the wavy membrane is also larger than the flat counterpart. In the presence of a  
448 higher shear stress gradient, there is a net migration of dye in the direction of a lower shear  
449 rate. The presence of flow recirculating regions in the valleys also makes the laminar sub-layer  
450 near the membrane thinner, thus reducing the overall flow resistance at the membrane surface.  
451 The shear stress values presented in **Figure 9** represent the maximum shear stress at the peaks  
452 for the wavy structures.

453



454

455 **Figure 9.** The maximum surface shear stress vs. Re number for wavy and flat composite  
456 membranes obtained from CVF simulations.

457

458 The energy requirement during filtration was characterised using the pressure drop incurred  
459 along the membrane. Any difference between the flat and wavy membranes was below the  
460 accuracy of the pressure gauges used experimentally ( $\pm 0.01$  bar). CFD simulations reported a  
461 pressure drop per unit length of  $\sim 0$  and  $2 \times 10^{-5}$  Pa $\cdot$ m $^{-1}$ , for the flat and wavy membranes,  
462 respectively. This negligible difference is to be expected given the laminar flow condition,  
463 i.e.  $Re = 700$ , employed in this study, and suggests that any additional energy consumption for  
464 wavy membranes is outweighed by its improved performance in permeance and concentration  
465 polarisation. It is, however, reasonable to expect the pressure drop caused by the wavy structure  
466 to be higher at turbulent flow conditions, thus diminishing somewhat its more favourable  
467 filtration properties. On the other hand, the reduction in concentration polarisation build up  
468 would increase the operating time between cleaning cycles, leading to an overall reduced  
469 downtime and resource requirement, as shown for UF membranes [14]. The authors have

470 shown that wavy UF membranes retain up to 87% of their initial pure water permeance after  
471 10 complete fouling cycles.

472 These results show that the impact of CP in NF membranes could be effectively reduced by  
473 using 3D printed composite membranes designed to control hydrodynamics (i.e., shear stress  
474 and flow patterns) over the membrane surface.

475

#### 476 **4. Conclusion**

477 This work reports on the successful fabrication of NF composite membranes comprising 3D  
478 printed asymmetric wavy and flat support and NF PVDF/PDA selective layers prepared by  
479 coating PDA over the UF PVDF membranes. The dye rejection results showed a high CR dye  
480 rejection (> 95%) for both membranes. The MWCO of the composite membranes was  
481 determined to be ~550 Da. The wavy structure resulted in an increase of 40% in pure water  
482 permeance due to a 20% larger surface area for the same footprint and reduced CP. The wavy  
483 structure nearly tripled the mass transfer coefficient and reduced concentration polarization by  
484 57%. This major reduction was related to the higher shear stress on the membrane's surface,  
485 with an order of magnitude increase passing from a flat to a wavy structure. As concentration  
486 polarisation remains a major challenge in nanofiltration, these results demonstrate that 3D  
487 printing represents a viable technology route to improving the performance of nanofiltration  
488 membranes in a wide range of applications.

489

#### 490 **Authorship contributions**

491 The manuscript was written through contributions of all authors.

#### 492 **Acknowledgement**

493 The authors thank UK EPSRC for support (grant EP/M01486X/1). All data used in the  
494 manuscript is reported in the SI document.

495 **References**

- 496 1. She, Q., et al., *Membrane fouling in osmotically driven membrane processes: A review*. Journal  
 497 of Membrane Science, 2016. **499**: p. 201-233.
- 498 2. Curcio, E., et al., *Membrane technologies for seawater desalination and brackish water*  
 499 *treatment*, in *Advances in Membrane Technologies for Water Treatment*. 2015, Elsevier. p.  
 500 411-441.
- 501 3. Antony, A., et al., *Scale formation and control in high pressure membrane water treatment*  
 502 *systems: a review*. Journal of membrane science, 2011. **383**(1-2): p. 1-16.
- 503 4. Shirazi, S., C.-J. Lin, and D. Chen, *Inorganic fouling of pressure-driven membrane processes—A*  
 504 *critical review*. Desalination, 2010. **250**(1): p. 236-248.
- 505 5. Tang, C.Y., T. Chong, and A.G. Fane, *Colloidal interactions and fouling of NF and RO*  
 506 *membranes: a review*. Advances in colloid and interface science, 2011. **164**(1-2): p. 126-143.
- 507 6. Rana, D. and T. Matsuura, *Surface modifications for antifouling membranes*. Chemical reviews,  
 508 2010. **110**(4): p. 2448-2471.
- 509 7. Kochkodan, V., D.J. Johnson, and N. Hilal, *Polymeric membranes: Surface modification for*  
 510 *minimizing (bio) colloidal fouling*. Advances in colloid and interface science, 2014. **206**: p. 116-  
 511 140.
- 512 8. An, Q.-F., et al., *Study on a novel nanofiltration membrane prepared by interfacial*  
 513 *polymerization with zwitterionic amine monomers*. Journal of membrane science, 2013. **431**:  
 514 p. 171-179.
- 515 9. Warsinger, D.M., et al., *A review of polymeric membranes and processes for potable water*  
 516 *reuse*. Progress in polymer science, 2018. **81**: p. 209-237.
- 517 10. ElSherbiny, I.M., A.S. Khalil, and M. Ulbricht, *Surface micro-patterning as a promising platform*  
 518 *towards novel polyamide thin-film composite membranes of superior performance*. Journal of  
 519 Membrane Science, 2017. **529**: p. 11-22.
- 520 11. Jeffree, M.A., et al., *Gel Layer Limited Haemofiltration Rates can be Increased by Vortex*  
 521 *Mixing*. Clinical and Experimental Dialysis and Apheresis, 1981. **5**(4): p. 373-380.
- 522 12. Heinz, O., et al., *Surface-patterning of polymeric membranes: fabrication and performance*.  
 523 Current opinion in chemical engineering, 2018. **20**: p. 1-12.
- 524 13. Choi, D.-C., et al., *Effect of pattern shape on the initial deposition of particles in the aqueous*  
 525 *phase on patterned membranes during crossflow filtration*. Environmental Science &  
 526 Technology Letters, 2017. **4**(2): p. 66-70.
- 527 14. Scott, K., et al., *Intensified membrane filtration with corrugated membranes*. Journal of  
 528 Membrane Science, 2000. **173**(1): p. 1-16.
- 529 15. Maruf, S.H., et al., *Use of nanoimprinted surface patterns to mitigate colloidal deposition on*  
 530 *ultrafiltration membranes*. Journal of membrane science, 2013. **428**: p. 598-607.
- 531 16. Tijing, L.D., et al., *3D printing for membrane separation, desalination and water treatment*.  
 532 Applied Materials Today, 2020. **18**: p. 100486.
- 533 17. Low, Z.-X., et al., *Perspective on 3D printing of separation membranes and comparison to*  
 534 *related unconventional fabrication techniques*. Journal of Membrane Science, 2017. **523**: p.  
 535 596-613.
- 536 18. Mazinani, S., et al., *3D printed fouling-resistant composite membranes*. ACS applied materials  
 537 & interfaces, 2019. **11**(29): p. 26373-26383.
- 538 19. Liu, F., et al., *Progress in the production and modification of PVDF membranes*. Journal of  
 539 membrane science, 2011. **375**(1-2): p. 1-27.
- 540 20. Kang, G.-d. and Y.-m. Cao, *Application and modification of poly (vinylidene fluoride)(PVDF)*  
 541 *membranes—a review*. Journal of membrane science, 2014. **463**: p. 145-165.
- 542 21. Xi, Z.-Y., et al., *A facile method of surface modification for hydrophobic polymer membranes*  
 543 *based on the adhesive behavior of poly (DOPA) and poly (dopamine)*. Journal of Membrane  
 544 Science, 2009. **327**(1-2): p. 244-253.

- 545 22. Jiang, J.-H., et al., *Improved hydrodynamic permeability and antifouling properties of poly*  
546 *(vinylidene fluoride) membranes using polydopamine nanoparticles as additives.* Journal of  
547 membrane science, 2014. **457**: p. 73-81.
- 548 23. Xiang, Y., F. Liu, and L. Xue, *Under seawater superoleophobic PVDF membrane inspired by*  
549 *polydopamine for efficient oil/seawater separation.* Journal of membrane science, 2015. **476**:  
550 p. 321-329.
- 551 24. Al-Shimmery, A., et al., *3D printed composite membranes with enhanced anti-fouling*  
552 *behaviour.* Journal of membrane science, 2019. **574**: p. 76-85.
- 553 25. Lee, H., et al., *Mussel-inspired surface chemistry for multifunctional coatings.* science, 2007.  
554 **318**(5849): p. 426-430.
- 555 26. Yao, L., et al., *Codeposition of polydopamine and zwitterionic polymer on membrane surface*  
556 *with enhanced stability and antibiofouling property.* Langmuir, 2018. **35**(5): p. 1430-1439.
- 557 27. Zydney, A.L., *Stagnant film model for concentration polarization in membrane systems.*  
558 Journal of Membrane Science, 1997. **130**(1): p. 275-281.
- 559 28. Baitalow, K., et al., *A mini-module with built-in spacers for high-throughput ultrafiltration.*  
560 Journal of Membrane Science, 2021. **637**: p. 119602.
- 561 29. Lee, Y.K., et al., *Flow analysis and fouling on the patterned membrane surface.* Journal of  
562 Membrane Science, 2013. **427**: p. 320-325.
- 563 30. Kotsilkova, R., et al., *Tensile and surface mechanical properties of polyethersulphone (pes) and*  
564 *polyvinylidene fluoride (PVDF) membranes.* Journal of Theoretical and Applied Mechanics,  
565 2018. **48**(3): p. 85-99.
- 566 31. Yuan, S., et al., *Super-hydrophobic 3D printed polysulfone membranes with a switchable*  
567 *wettability by self-assembled candle soot for efficient gravity-driven oil/water separation.*  
568 Journal of Materials Chemistry A, 2017. **5**(48): p. 25401-25409.
- 569 32. Wang, J., et al., *High flux electroneutral loose nanofiltration membranes based on rapid*  
570 *deposition of polydopamine/polyethyleneimine.* Journal of Materials Chemistry A, 2017. **5**(28):  
571 p. 14847-14857.
- 572 33. Liu, F., et al., *Surface immobilization of polymer brushes onto porous poly (vinylidene fluoride)*  
573 *membrane by electron beam to improve the hydrophilicity and fouling resistance.* Polymer,  
574 2007. **48**(10): p. 2910-2918.
- 575 34. Ma, Z., et al., *Functional surface modification of PVDF membrane for chemical pulse cleaning.*  
576 Journal of membrane science, 2017. **524**: p. 389-399.
- 577 35. Zhao, J., et al., *Manipulating the interfacial interactions of composite membranes via a mussel-*  
578 *inspired approach for enhanced separation selectivity.* Journal of Materials Chemistry A, 2015.  
579 **3**(39): p. 19980-19988.
- 580 36. Maruf, S.H., et al., *Critical flux of surface-patterned ultrafiltration membranes during cross-*  
581 *flow filtration of colloidal particles.* Journal of membrane science, 2014. **471**: p. 65-71.
- 582 37. Nagy, E., *Basic equations of mass transport through a membrane layer.* 2018: Elsevier.
- 583 38. Koyuncu, I. and D. Topaçik, *Effect of organic ion on the separation of salts by nanofiltration*  
584 *membranes.* Journal of Membrane Science, 2002. **195**(2): p. 247-263.
- 585 39. Baker, R.W., *Membrane technology and applications.* 2012: John Wiley & Sons.
- 586 40. Kromkamp, J., et al., *Shear-induced diffusion model for microfiltration of polydisperse*  
587 *suspensions.* Desalination, 2002. **146**(1): p. 63-68.
- 588 41. Shang, W., et al., *Effective suppression of concentration polarization by nanofiltration*  
589 *membrane surface pattern manipulation: Numerical modeling based on LIF visualization.*  
590 Journal of Membrane Science, 2021. **622**: p. 119021.
- 591 42. Santos, J., et al., *Investigation of flow patterns and mass transfer in membrane module*  
592 *channels filled with flow-aligned spacers using computational fluid dynamics (CFD).* Journal of  
593 Membrane Science, 2007. **305**(1-2): p. 103-117.



- 594 43. Amokrane, M., et al., *A study of flow field and concentration polarization evolution in*  
595 *membrane channels with two-dimensional spacers during water desalination.* Journal of  
596 *membrane science*, 2015. **477**: p. 139-150.
- 597 44. Ahmad, A.L., et al., *Integrated CFD simulation of concentration polarization in narrow*  
598 *membrane channel.* Computers & chemical engineering, 2005. **29**(10): p. 2087-2095.
- 599 45. de Pinho, M.N., V. Semião, and V.t. Geraldes, *Integrated modeling of transport processes in*  
600 *fluid/nanofiltration membrane systems.* Journal of Membrane Science, 2002. **206**(1-2): p. 189-  
601 200.
- 602 46. Schlichting, H. and K. Gersten, *Boundary-layer theory.* 2016: Springer.
- 603 47. Ding, A., et al., *Impact of aeration shear stress on permeate flux and fouling layer properties*  
604 *in a low pressure membrane bioreactor for the treatment of grey water.* Journal of Membrane  
605 Science, 2016. **510**: p. 382-390.
- 606 48. Liu, X., et al., *CFD modelling of uneven flows behaviour in flat-sheet membrane bioreactors:*  
607 *From bubble generation to shear stress distribution.* Journal of Membrane Science, 2019. **570-**  
608 **571**: p. 146-155.
- 609 49. Chan, C.C.V., P.R. Bérubé, and E.R. Hall, *Relationship between types of surface shear stress*  
610 *profiles and membrane fouling.* Water Research, 2011. **45**(19): p. 6403-6416.
- 611 50. Du, X., et al., *Shear stress in a pressure-driven membrane system and its impact on membrane*  
612 *fouling from a hydrodynamic condition perspective: a review.* Journal of Chemical Technology  
613 & Biotechnology, 2017. **92**(3): p. 463-478.

614

DRAFT