

# Studies on H<sup>+</sup> ions conducting bio-polymer blend electrolyte based on alginate-PVA doped with NH<sub>4</sub>NO<sub>3</sub>

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## ABSTRACT

This study provides insights into the protonation of bio-polymer blend electrolytes (BBEs) that are based on alginate (Alg)-PVA doped with various NH<sub>4</sub>NO<sub>3</sub> compositions, which was prepared using the solution casting method. The physicochemical of BBEs were studied by using electrical impedance spectroscopy (EIS) analysis, thermogravimetric analysis (TGA), scanning electron microscope (SEM), x-ray diffraction (XRD) and Fourier transform infrared (FTIR) spectroscopy. The complexation had occurred between the Alg-PVA functional groups with the H<sup>+</sup>-NH<sub>3</sub>NO<sub>3</sub> through the shifting and changes in the intensity of the bands. The BBEs films showed the enhancement of amorphous and the presence of globules when introduced NH<sub>4</sub>NO<sub>3</sub>, which enhanced the ionic conductivity. The addition of 35 wt.% of NH<sub>4</sub>NO<sub>3</sub> resulted in the highest ionic conductivity value of 5.20×10<sup>-4</sup> S cm<sup>-1</sup> and demonstrated excellent thermal property. It was found that the system's ionic conductivity was generally influenced by the charge carriers based on evaluation of the Nyquist fitting approaches.

**KEYWORDS:** Amorphous conducting polymer, Bio-materials, Ionic conducting mechanism, Nyquist fitting method

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## REFERENCES

- [1] D.E. Fenton, J.M. Parker, P.V. Wright, Complexes of alkali metal ions with poly (ethylene oxide), *Polymer* 14 (11) (1973) 589, [https://doi.org/10.1016/0032-3861\(73\)90146-8](https://doi.org/10.1016/0032-3861(73)90146-8) (Guildf).
- [2] C. Ambika, et al., Effect of dimethyl carbonate (DMC) on the electrochemical and cycling properties of solid polymer electrolytes (PVP-MSA) and its application for proton batteries, *Solid State Ion.* 321 (2018) 106–114, <https://doi.org/10.1016/j.ssi.2018.04.013>. Aug.
- [3] S.N.A.M. Johari, N.A. Tajuddin, H. Hanibah, S.K. Deraman, A review: ionic conductivity of solid polymer electrolyte based polyethylene oxide, *Int. J. Electrochem. Sci.* 16 (2021) 1–15, <https://doi.org/10.20964/2021.10.53>.
- [4] R. Singh, et al., Electrical and structural properties of ionic liquid doped polymer gel electrolyte for dual energy storage devices, *Int. J. Hydrog. Energy* 42 (21) (2017) 14602–14607, <https://doi.org/10.1016/j.ijhydene.2017.04.126>. May