

# Environmental Science and Pollution Research

## Analysis of a photobioreactor scaling up for tertiary wastewater treatment: denitrification, phosphorus removal and microalgae production

--Manuscript Draft--

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<b>Abstract:</b>	The present work studies the removal of nutrients (nitrate and phosphate) from a synthetic wastewater simulating a secondary treatment effluent using the microalgae <i>Chlorella vulgaris</i> in autotrophic photobioreactors, together to an analysis of the critical points affecting the scaling-up process from laboratory to pilot scale. Laboratory experiments were done in open agitated 1 L photobioreactors under batch operation mode, while pilot-scale experiments were done using a 150 L closed tubular photobioreactor under continuous operation mode. In both scales, nitrate was the limiting substrate and the effect of its concentration on microalgae performance was studied. From laboratory experiments, an average microalgae productivity of 85 mgVSS L <sup>-1</sup> d <sup>-1</sup> and approximate maximum N-NO <sub>3</sub> <sup>-</sup> and P-PO <sub>4</sub> <sup>3-</sup> removal rates of 8 mg N gVSS <sup>-1</sup> d <sup>-1</sup> , and 2.6 mg P gVSS <sup>-1</sup> d <sup>-1</sup> were found. Regarding pilot scale, the average microalgae productivity slightly decreased (76 mgVSS L <sup>-1</sup> d <sup>-1</sup> ) while the approximate maximum N-NO <sub>3</sub> <sup>-</sup> and P-PO <sub>4</sub> <sup>3-</sup> removal rates slightly were increased (11.7 mg N gVSS <sup>-1</sup> d <sup>-1</sup> and 3.04 mg P gVSS <sup>-1</sup> d <sup>-1</sup> ) with respect to the laboratory scale results. The pilot scale operation worked under lower levels of turbulence and higher dissolved oxygen concentration and light intensity than laboratory experiments; those parameters were difficult to control and they can be identified as the critical points in the differences found on both nutrient removal and microalgae production.
<b>Response to Reviewers:</b>	see attachment
<b>Additional Information:</b>	
<b>Question</b>	<b>Response</b>
§Are you submitting to a Special Issue?	No

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*Environmental Science and Pollution Research*  
Editor

3-July-2018

Dear Editor:

Attached you will find the revised manuscript ESPR-D-18-02643 “*Analysis of a photobioreactor scaling up for tertiary wastewater treatment: denitrification, phosphorus removal and microalgae production*”, by José Villaseñor Camacho, Carmen M. Fernández Marchante, and Luis Rodríguez Romero (corresponding author: [jose.villasenor@uclm.es](mailto:jose.villasenor@uclm.es)), in order to be reviewed for a possible publication as original research paper in *Environmental Science and Pollution research*.

The following items are included in the new submission:

- The “**Revised Manuscript**” (using MS Word).
- The “**Highlighted Revised manuscript**”, that is the same revision manuscript MS Word file, using the track changes mode, where you can easily find the modifications made to the text.
- The “**Responses to reviewers**”: One MS Word document containing the detailed answers to each concrete reviewer’s comments. Each answer indicates the position of the modifications in the revision changes marked manuscript.
- Revised figures **3, 4 and 5**.

Yours sincerely

Dr. José Villaseñor Camacho

## Revision Notes: Response to Reviewers

This document shows detailed responses to the reviewer's comments. The responses indicate also the location (page and lines) of changes made in the highlighted revised manuscript.

### Reviewers' comments:

#### Reviewer #1:

I have difficulty to see why nitrate is used as the sole nitrogen source in this work. It is well-known that microalgae prefer ammonium over nitrate, nitrite and urea, and ammonium is the dominant nitrogen source in secondary effluents from wastewater treatment plant. The author should provide more information to clarify the innovation of this work by investigating the nitrate removal other than ammonium removal.

#### RESPONSE

According to the authors' opinion, and as it has been indicated in the introduction section, *“Nitrification is practically and successfully implemented in the aerobic step most of the classical biological treatments, but denitrification and P removal may not be considered in the plant design (old plants) or may not work properly”*. Because of it, this work is focused on nitrate discharges in secondary effluents, and this is one of the two novelty points. It is well known that most of the nitrogen removal works using microalgae are focused on ammonium removal, while the use of nitrate is not so common. This statement was already included in the manuscript (lines 73-77) and also the novelty point has been clearly indicated in the revised manuscript (lines 106-110).

Furthermore, it is a good idea to compare the performance between laboratory and pilot work and to find out the critical points affecting the scaling-up process from lab to pilot. However, if nothing could be controlled in pilot-scale, how could we compare these two systems?

From the work results, some points have been identified as to be critical, and so it has been concluded that future efforts should be made to improve control of such factors and thus to

allow studies that compare laboratory and pilot scale systems. This statement has been included in the conclusions section (revised manuscript, lines 382-385).

It is recommended to be accepted after revising or clarifying the abovementioned points.

Specific comments:

P5, line 40: add reference for BBM medium

RESPONSE

BBM is a useful medium for microalgae growth (including *C. Vulgaris*). A new reference has been included (revised manuscript line 120, and line 416)

P5, line 45: Why 10% CO<sub>2</sub> was used? Is there any reference?

RESPONSE

10% CO<sub>2</sub> is commonly used to simulate flue gas. Two references have been included in the revised manuscript (line 122 and line 412)

P5, line 52: what is the pH in synthetic BBM medium?

RESPONSE

pH was between 6.6 and 6.8. It was indicated in the revised manuscript (line 133).

P6, line 16: Information on LED lamps (company, model and country)?

RESPONSE

Information has been included in the revised manuscript (line 138).

P7, line 48: what is the initial amount of microalgae?

RESPONSE

It was a seed, approximately 0.15 g<sub>vss</sub> L<sup>-1</sup>. This information has been included in the revised manuscript (line 176).

P8, line 8: flow rate instead of flowrate

RESPONSE

The change has been made.

P8, line 47: why dissolved oxygen concentration was maintained at  $7.0 \pm 1.2$  mg/l?

RESPONSE

It was the saturation level obtained because of the air supply using the laboratory compressor (see Figure 1, part 1). The manuscript has been modified according to this comment (line 199).

Please add error bars in the figures.

RESPONSE

Figures 3, 4 and 5 (where batch experiments were performed by triplicate) have been changed.

**Reviewer #2:**

This study researched the removal of nutrients (nitrate and phosphate) from a synthetic wastewater simulating a secondary treatment effluent using the microalgae *Chlorella vulgaris* in autotrophic photobioreactors, together to an analysis of the critical points affecting the scaling-up process from laboratory to pilot scale. Overall, this study is interested and valuable to reduce the nutrients such nitrogen and phosphorus from wastewater. However, as we know in practice, the phosphorus in wastewater is less than  $10 \text{ mg L}^{-1}$  while the input concentration of phosphorus in this study is very more higher than this value. Thus, it should be introduce the scenes where the phosphorus concentration is such high.

Regarding nutrients removal, the work has been especially focused on N removal, and nitrate has been always the limiting nutrient. The present work also studied P removal although it is true that we used always an excess P concentration (P was not limiting nutrient in our

experiments) as we used a very common growth medium for microalgae (BBM). A brief change has been included in revised manuscript (line 127) regarding this comment.

Second, there are lots of similar works using the microalgae to reduce (or accumulate ) nutrients based on the results of crossrefernces. Thus, the novelty of this studying using microalgae *Chlorella vulgaris* to reduce nutrients concentrations should be highlighted.

## RESPONSE

One of the novelty points is that most of the nitrogen removal works using microalgae are focused on ammonium removal, while the use of nitrate is not so common. Moreover, the second novelty point is focused on the scaling-up process, and the critical operating parameters to perform scale-up have been identified.

The main novelty point in the present work has been included in the revised manuscript (106-110)

Third, experiments of from lab-scale to pilot are valuable, while the running conditions are dissimilar, please give some explanations why the conditions were conducted in different pattern.

Bench-scale studies are usually performed in single-designed PBR although it is known that tubular PBR are preferentially selected for scale-up. This work considered some criteria for scale-up: maintaining the retention time, the nutrients and CO<sub>2</sub> concentrations, and the flow model (complete mixed system). However it is true that it was not possible for us to maintain the same high turbulence level at pilot-scale and, obviously, the weather conditions also influenced in the pilot-scale operation. Authors consider that, if possible, these factors should be better controlled and additional efforts should be made in a future in scaling-up studies. A brief change has been included in the revised manuscript regarding this comment (line 382-385).

As the authors said that "Although N and P removal rates were low compared to classical biological nutrient removal secondary processes, the use of microalgae could be considered

a more sustainable technology for wastewater treatment. ", If I am the coauthor, I will estimate the cost and the running fee, to compare the classical biological nutrient removal secondary process, to check which one is the feasible at economical level.

## RESPONSE

Under the authors' opinion is quite difficult to propose a broad calculation of cost and running fee to compare PBR vs activated sludge BNR systems, The works previously reported and cited in the manuscript (Judd et al 2015; Marbelia et al 2014) give details about the economic and environmental advantages/disadvantages of both systems. Some details have been included in the revised manuscript in order to improve the last paragraph in discussion section: According to Judd et al (2014), because of the lower nutrient removal rates and the high surface (low depth) necessary for microalgae, PBR systems may be 15 times slower than classical activated sludge BNR systems, and overall there is two order of magnitude difference in footprint between them. Against this, the biomass yield is quite lower in PBR and microalgae is considered a valuable product. Additionally, Marbelia et al (2014) describe different scenarios to combine classical wastewater treatment plants with microalgae nutrient removal and they propose approximate values of power consumption and operation costs (lines 355-365).

[Click here to view linked References](#)

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3 **26 Abstract**

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6 27 The present work studies the removal of nutrients (nitrate and phosphate) from a synthetic  
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8 28 wastewater simulating a secondary treatment effluent using the microalgae *Chlorella*  
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10 29 *vulgaris* in autotrophic photobioreactors, together to an analysis of the critical points  
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12 30 affecting the scaling-up process from laboratory to pilot scale. Laboratory experiments  
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14 31 were done in open agitated 1 L photobioreactors under batch operation mode, while pilot-  
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16 32 scale experiments were done using a 150 L closed tubular photobioreactor under  
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18 33 continuous operation mode. In both scales, nitrate was the limiting substrate and the effect  
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20 34 of its concentration on microalgae performance was studied. From laboratory  
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22 35 experiments, an average microalgae productivity of  $85 \text{ mg}_{\text{VSS}} \text{ L}^{-1} \text{ d}^{-1}$  and approximate  
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24 36 maximum  $\text{N-NO}_3^-$  and  $\text{P-PO}_4^{3-}$  removal rates of  $8 \text{ mg N } \text{g}_{\text{VSS}}^{-1} \text{ d}^{-1}$ , and  $2.6 \text{ mg P } \text{g}_{\text{VSS}}^{-1} \text{ d}^{-1}$   
25  
26 37 were found. Regarding pilot scale, the average microalgae productivity slightly  
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28 38 decreased ( $76 \text{ mg}_{\text{VSS}} \text{ L}^{-1} \text{ d}^{-1}$ ) while the approximate maximum  $\text{N-NO}_3^-$  and  $\text{P-PO}_4^{3-}$   
29  
30 39 removal rates slightly were increased ( $11.7 \text{ mg N } \text{g}_{\text{VSS}}^{-1} \text{ d}^{-1}$  and  $3.04 \text{ mg P } \text{g}_{\text{VSS}}^{-1} \text{ d}^{-1}$ ) with  
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32 40 respect to the laboratory scale results. The pilot scale operation worked under lower levels  
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34 41 of turbulence and higher dissolved oxygen concentration and light intensity than  
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36 42 laboratory experiments; those parameters were difficult to control and they can be  
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38 43 identified as the critical points in the differences found on both nutrient removal and  
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40 44 microalgae production.  
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51 **46 Keywords**

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54 47 Photobioreactor, microalgae, nitrate removal, phosphorus removal, scale-up, *Chlorella*  
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56 48 *vulgaris*.  
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**1. Introduction**

Nutrient removal in modern domestic wastewater treatment plants is usually included in the secondary biological step and it is achieved by means of well-known processes such as nitrification, denitrification and biological phosphorus removal (Ekama 2015), the whole of them is called as Biological Nutrient Removal (BNR). Nitrification is practically and successfully implemented in the aerobic step most of the classical biological treatments, but denitrification and P removal may not be considered in the plant design (old plants) or may not work properly in modern plants because of the operation costs or the lack of pre-hydrolyzed easily biodegradable organic matter in the wastewater, which is an important requisite (Zheng et al. 2015). Thus, it is usual that nitrogen removal is not satisfactory and it is present mainly as  $N-NO_3^-$  in the effluent, and also the P concentration may exceed the discharge limit, which caused adverse environmental impacts.

Because of this problem there are currently different tertiary systems for the removal of nitrate and phosphate in secondary effluents. Some of them are based on physicochemical fundamentals but they are associated with significant costs due to the consumption of chemical reagents. An alternative to such physicochemical methods may be the biological nutrient removal using microalgae and photobioreactors. There is currently a lot of information in the scientific literature about the use of microalgae for N and P removal in wastewater (Cai et al. 2013). It can be achieved through an autotrophic photosynthetic process capable to treat a secondary effluent with no biodegradable organic matter. Microalgae may be also used in mixotrophic processes, a combination of heterotrophic and autotrophic biological treatment in which bacteria and microalgae simultaneously remove organic carbon, carbon dioxide from heterotrophic respiration, nitrogen and phosphorus

1 73 (Zheng et al. 2015). Since microalgae prefer ammonium rather than oxidized forms of  
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3 74 nitrogen, the number of scientific papers focused on ammonium removal is higher than  
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5 75 those about nitrate removal, despite the fact that nitrate is the main nitrogen form in  
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7 76 secondary effluents from wastewater treatment plants (Cai et al. 2013). Moreover, the  
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9 77 potential of nitrate-accumulating microalgae for nutrient recovery has not been  
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11 78 adequately investigated to date (Coppens et al. 2014).

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14 79 In addition to nutrients capture, photosynthetic autotrophic processes using microalgae  
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16 80 have been widely used for carbon dioxide removal in exhaust combustion gas and for  
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18 81 microalgae biomass production because of its multiple uses. Unlike sewage sludge from  
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20 82 wastewater treatment plants, microalgae biomass is considered as a valuable raw material  
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22 83 instead of a biowaste. Many applications of microalgae biomass have been proposed and  
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24 84 investigated (Odjadjare et al. 2017): (i) transformation into valuable bioproducts such as  
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26 85 lipids, oil, fatty acids, pigments, vitamins and proteins, (ii) transformation into energy  
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28 86 sources, e.g. biofuels, biogas or biohydrogen, and (iii) animal food manufacture. The cost  
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30 87 of such biomass production could be significantly reduced by using treated sewage as  
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32 88 inorganic nutrients source (Cabanelas et al. 2013).

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35 89 However, most of the research work regarding microalgae nutrient removal have been  
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37 90 made under laboratory scale, while the research focused on scale-up to pilot or full scale  
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39 91 photobioreactors did not receive so much attention. It is very important to study the  
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41 92 differences that could appear in the scale-up step because some parameters such as  
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43 93 temperature, light intensity or turbulence are easily controlled under lab scale but not  
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45 94 under pilot scale (Ación Fernández et al. 2013). Several authors reported scale-up  
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47 95 investigations where both nutrient removal efficiency (Van den Hende et al. 2014) and  
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49 96 microalgae productivity (Lam et al. 2015) decreased in pilot scale photobioreactors.  
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1 97 In this context, the present work shows the results of an experimental study in which  
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3 98 *Chlorella vulgaris* microalgae was used to remove nitrate and phosphate from a synthetic  
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6 99 wastewater simulating a secondary treatment effluent. The study was performed both at  
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8 100 laboratory and pilot scales. Laboratory experiments were done in open agitated  
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11 101 photobioreactors under batch operation mode and using different nitrate concentrations,  
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13 102 while pilot-scale experiments were carried out in a closed tubular photobioreactor (150 L  
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15 103 reaction volume) under continuous operation mode. One of the novelty points is that most  
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18 104 of the nitrogen removal works using microalgae are focused on ammonium removal,  
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20 105 while the use of nitrate is not so common. Moreover, the second novelty point is focused  
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23 106 on the scaling-up process, and the critical operating parameters to perform scale-up have  
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25 107 been identified. Thus, the objectives of the present work were: (i) to assess the N-NO<sub>3</sub><sup>-</sup>  
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27 108 and P-PO<sub>4</sub><sup>3-</sup> removal rates and biomass yields both under laboratory and pilot scale, (ii)  
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29 109 to study the relationship between the differences found in those parameters and the main  
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31 110 operating variables for both scales, and (iii) to identify the most critical operating  
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35 111 parameters in the scale-up process.  
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## 40 113 **2. Materials and Methods**

### 41 42 114 **2.1. Microalgae and growth medium**

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44 115 *Chlorella vulgaris* was obtained from the Culture Collection of Algae in the University  
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47 116 of Las Palmas de Gran Canaria (Spain). The microalgae culture was incubated in closed  
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49 117 flasks using Bold's Basal Medium, BBM, as synthetic culture medium (Frumento et al.  
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52 118 2016), weekly growth cycles at ambient temperature (approximately 21°C) and day/night  
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54 119 alternation. Air with 10% CO<sub>2</sub> simulating flue gas was bubbled during light periods (Judd  
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57 120 et al 2015; Duarte et al 2016).  
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1 121 BBM was chosen as a synthetic medium to simulate the secondary effluent as it contains  
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3 122 nitrate and phosphate as the main inorganic nutrients, although P concentrations used in  
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5 123 BBM were higher than in the usual secondary effluent levels. BBM also presents buffer  
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8 124 capacity for pH control, and it was the liquid medium used both in the laboratory and  
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10 125 pilot-scale experiments. Its composition was the following ( $\text{mg L}^{-1}$ ):  $\text{NaNO}_3$ , 250.0;  
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12 126  $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ , 25.0;  $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ , 75.0;  $\text{K}_2\text{HPO}_4$ , 175.0;  $\text{KH}_2\text{PO}_4$ , 75.0;  $\text{NaCl}$ , 25.0;  
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14 127 EDTA, 50.0;  $\text{KOH}$ , 31.0;  $\text{H}_3\text{BO}_3$ , 11.5;  $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$ , 5.0;  $\text{ZnSO}_4 \cdot 7\text{H}_2\text{O}$ , 8.8;  
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16 128  $\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$ , 1.8;  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ , 1.6. pH was between 6.6 and 6.8.  
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## 20 129 **2.2. Experimental installations**

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23 130 Figure 1 shows a scheme of the experimental installation used for the lab-scale  
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25 131 experiments. It consisted of a photobioreactor with the following parts: (1) a system for  
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27 132 atmospheric air feeding enriched in  $\text{CO}_2$  (2), in order to simulate a combustion exhaust  
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29 133 gas; (3) a thermostated closed chamber with adjustable temperature; (4) UEETK (USA)  
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31 134 28cm LED lamps for artificial lighting and a timer (5) to adjust the duration of light/dark  
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33 135 cycles; (6) a multiple magnetic stirring system and several 1 L glass bottles (7) which  
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35 136 acted as completely mixed batch reactors that contained the microalgae suspensions and  
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37 137 the liquid growth medium, receiving the air/ $\text{CO}_2$  mixture flow through bubbling (8).  
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42 138 Figure 2 shows a scheme (a) and a photograph (b) of the pilot-scale installation. It  
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44 139 consisted of a tubular photobioreactor that contained the following parts: (1) a system for  
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46 140 atmospheric air feeding enriched in  $\text{CO}_2$  similar to the one used in laboratory; (2) a  $\text{CO}_2$   
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48 141 absorption tank (100 L) with a mechanical stirring system that contained the liquid growth  
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50 142 medium; (3) the feeding system of the liquid growth medium saturated in  $\text{CO}_2$  consisting  
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52 143 of a peristaltic pump; (4) a 150 L tubular photobioreactor composed by consecutive tubes  
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55 144 of 1.5 m long and 9 cm diameter; (5) a degasification unit; (6) a peristaltic pump for liquid  
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1 145 flow recirculation; (7) an effluent outlet; and (8) an atmospheric air compressor to  
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3 146 improve the turbulence. It was also equipped with temperature and lighting sensors. The  
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5 147 system worked under continuous operation mode as a completely mixed biological  
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7 148 reactor without biomass recirculation. The whole pilot-scale installation was located into  
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9 149 a greenhouse (Figure 2b) which allowed temperature control by an air-conditioning  
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11 150 device, and maximum light intensity control by a manual adjustable solar radiation  
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13 151 mitigation system. The greenhouse was located next to the Institute of Chemical and  
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15 152 Environmental Technology of the University of Castilla-La Mancha, Ciudad Real  
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20 153 (Spain).

### 23 154 **2.3. Experimental procedure**

25 155 Lab-scale nutrient (nitrate and phosphate) removal experiments were performed under  
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27 156 batch operation mode. Glass bottles were filled with the growth liquid medium.  
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29 157 Depending on the experiments, the growth medium contained different nitrate  
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31 158 concentration. The same initial amount of microalgae was inoculated in all bottles and  
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33 159 then magnetic mixing and light/darkness cycles were connected during 10 days. Air  
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35 160 containing 10% (v/v) CO<sub>2</sub> was bubbled only during light cycles. Temperature was 21°C.  
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37 161 Mixing rate was 10 s<sup>-1</sup> (Approximate Reynolds number of 25·10<sup>3</sup>). Light/darkness cycles  
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39 162 were 12h/12h and light intensity was 100 μmol m<sup>-2</sup> s<sup>-1</sup>. Three experiments were  
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41 163 performed: N1, N2 and N3 that contained initial concentrations of 14.6, 28.2 and 40.8 mg  
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43 164 N-NO<sub>3</sub><sup>-</sup> L<sup>-1</sup>, respectively. Experiments were made by triplicate. The liquid samples were  
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45 165 taken every 10 hours from start to day 3, and then every 24 hours until the end.

52 166 Pilot-scale nutrient removal experiments were made under continuous operation mode  
53  
54 167 during 7 months. The system worked as a tubular completely-mixed biological reactor  
55  
56 168 without biomass recirculation, that is, the hydraulic retention time was the same as the

1 169 cell (microalgae) retention time. The photobioreactor was filled with the BBM liquid  
2  
3 170 growth medium and inoculated with an initial amount of microalgae culture  
4  
5 171 (approximately 0.15 gvss L<sup>-1</sup>) from the laboratory. A batch operation was applied during  
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7  
8 172 the first week, as acclimation step in order to reach enough microalgae concentration, and  
9  
10 173 then a continuous flow of CO<sub>2</sub> saturated liquid growth medium containing 40.8 mg N-  
11  
12 174 NO<sub>3</sub><sup>-</sup> L<sup>-1</sup> was used (as in the N3 laboratory experiment). According to the results  
13  
14 175 previously found in the laboratory experiments, hydraulic retention time were varied  
15  
16 176 throughout the experiment: 5.5 d in the first month, 6.0 d in the second month and 6.5 d  
17  
18 177 in months 3 to 7 (which was the main stationary period to compare with the laboratory  
19  
20 178 results), corresponding to mean flow rates of 27.3, 25.0 and 23.1 L d<sup>-1</sup>, respectively.  
21  
22 179 Atmospheric air was supplied by a compressor to favour liquid flow and mixing and  
23  
24 180 degasification of excess dissolved oxygen (Acién Fernández et al. 2013).

## 30 181 **2.4. Analytical methods**

31  
32 182 All analytical methods followed Standard Methods (A.P.H.A., 1998). Microalgae  
33  
34 183 concentration in the liquid samples was measured as volatile suspended solids (VSS) by  
35  
36 184 weight loss after ignition at 550°C. Nitrogen (N-NO<sub>3</sub><sup>-</sup>) and phosphorus (P-PO<sub>4</sub><sup>3-</sup>)  
37  
38 185 concentrations were measured by colorimetric methods using a DR2700 Hach portable  
39  
40 186 spectrophotometer (Colorado, USA). Dissolved inorganic carbon (IC) was measured by  
41  
42 187 a TOC analyser (Shimadzu TOC-VCSH, Columbia, USA). The pH was measured by a  
43  
44 188 pH-meter (*PCE-228M*). Dissolved Oxygen was measured using a *YSI 5000* dissolved  
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46 189 oxygen probe. Light intensity was measured by a *Collihigh* illuminometer.  
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## 52 191 **3. Results and Discussion**

### 53 192 **3.1. Evolution and control of experimental conditions**

1 193 Experimental conditions were easily controlled in the laboratory experiments.  
2  
3 194 Temperature was maintained in  $21.0 \pm 2^\circ\text{C}$ , pH was  $6.5 \pm 0.5$  and saturation dissolved  
4  
5 195 oxygen concentration was  $7.0 \pm 1.2 \text{ mg L}^{-1}$ . Light intensity was kept constant at  $100 \mu\text{mol}$   
6  
7  
8 196  $\text{m}^{-2}\text{s}^{-1}$ .

9  
10 197 Pilot scale conditions were more difficult to control than in laboratory. Temperature, pH,  
11  
12 198 dissolved oxygen concentration and lighting values monitored throughout the  
13  
14 199 experimental period have been included as supplementary material (online resource,  
15  
16 200 Figure S1). Temperature values varied between  $15$  and  $28^\circ\text{C}$  during the first 100 d,  
17  
18 201 although it were better controlled and maintained (around  $24^\circ\text{C}$ ) during the rest of the  
19  
20 202 experiment; pH varied between  $6.5$  and  $8.0$  by means of the buffer capacity of the BBM  
21  
22 203 growth medium; dissolved oxygen concentration varied between  $6$  and  $12 \text{ mg L}^{-1}$   
23  
24 204 approximately, and, finally, light intensity reached higher values than those of lab-scale  
25  
26 205 tests (in the range  $100\text{-}300 \mu\text{mol m}^{-2} \text{ s}^{-1}$ ) during the first 100 days, although it was better  
27  
28 206 controlled (in values around  $90 \mu\text{mol m}^{-2} \text{ s}^{-1}$ ) throughout the rest of the experimental  
29  
30 207 period. Observed variations in temperature and lighting were mainly due to changes in  
31  
32 208 external climatic conditions, while pH changes may be attributed to variations in carbon  
33  
34 209 availability throughout the experiment (Acién Fernández et al. 2013). Therefore, it seems  
35  
36 210 clear that, in spite of the control systems implemented in the pilot plant used, the  
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38 211 achievement of precise control of the operating conditions at the pilot plant scale is a  
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40 212 complex issue that needs to be adequately addressed in future experiments.  
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### 50 213 **3.2. Nutrient removal and microalgae production in laboratory**

51  
52 214 Figure 3(a) shows the growth of microalgae during the laboratory batch experiments  
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54 215 using different values of the initial nitrate concentration (mean values from 3 replicates).  
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56 216 As previously indicated, batch experiments were performed under excess concentrations  
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1 217 of N and P, being N-NO<sub>3</sub><sup>-</sup> the limiting nutrient at the end of the experiments (as discussed  
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3  
4 218 in Figure 4). Microalgae growth profiles were straight lines indicatives of first order  
5  
6 219 growth kinetics, without a clear influence of the initial nitrate concentration. The  
7  
8 220 maximum growth rate began to decline from day 8. An average microalgae productivity  
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10 221 value of 85 mg<sub>vss</sub> L<sup>-1</sup> d<sup>-1</sup> was calculated during this period. Taking into account also the  
11  
12 222 average biomass concentration, an approximate maximum specific growth rate of 0.18 d<sup>-1</sup>  
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14 223 <sup>1</sup> may be calculated, which corresponds to an approximate hydraulic retention time of 5.5  
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17  
18 224 d in a hypothetical continuous operation mode.  
19  
20 225 Figure 3(b) shows the dissolved inorganic carbon (IC) consumption measured in the  
21  
22 226 closed batch tests carried out in the laboratory (average values from three replicates). The  
23  
24 227 gas mixture air/CO<sub>2</sub> (10% v/v) was bubbled into the microalgae suspension (0.25 g<sub>vss</sub> L<sup>-1</sup>  
25  
26  
27 228 <sup>1</sup>) in BBM medium until it was saturated with CO<sub>2</sub>. Then, the air/CO<sub>2</sub> flow was stopped  
28  
29 229 and the dissolved inorganic carbon (IC) concentration consumption was measured during  
30  
31 230 several hours. It can be observed a constant maximum carbon consumption rate during  
32  
33 231 the first 9 h approximately. It means that an excess of inorganic carbon, as well as nitrogen  
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35 232 and phosphorus, was kept during the first 9 h and therefore it was not the limiting substrate  
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37 233 for microalgae growth. Under such conditions, the maximum microalgae IC consumption  
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39 234 rate in the laboratory photobioreactor may be calculated, obtaining a value of 118 mg C  
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41 235 g<sub>vss</sub><sup>-1</sup> d<sup>-1</sup>. Taking into account the microalgae productivity value previously calculated,  
42  
43 236 an approximate biomass yield of 1.2 g<sub>vss</sub> g<sub>C</sub><sup>-1</sup> was obtained.  
44  
45 237 Figure 4(a) shows the N-NO<sub>3</sub><sup>-</sup> concentration profiles in the three laboratory batch  
46  
47 238 experiments performed with different initial nitrate concentrations. Figure 4(b) shows the  
48  
49 239 values of the biomass specific N-NO<sub>3</sub><sup>-</sup> removal rates (mg N g<sub>vss</sub><sup>-1</sup> d<sup>-1</sup>). Assuming the  
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51 240 typical variability of the experimental results in this type of biological processes (see  
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1 241 dashed lines in Figure 4b), it is possible to observe a Monod-type trend in the removal  
2  
3 242 rates and, thus, it may be approximately established the range in which the specific N  
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5 243 removal rate is maximum. Maximum N removal rate was approximately  $8 \text{ mg N gvss}^{-1} \text{ d}^{-1}$   
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7  
8 244 <sup>1</sup>, and it began to decrease from N concentrations of approximately  $18 \text{ mg L}^{-1}$  (that is,  
9  
10 245 approximately from day 5 in N3 experiment) and, from that concentration, it can be  
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12 246 assumed that N became the limiting nutrient as the liquid medium was always saturated  
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14 247 in  $\text{CO}_2$  and P concentration was quite high (Figure 5). N removal rates during the last  
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16 248 days were very low and some additional days would be necessary to the complete N  
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18 249 depletion.

22 250 Figure 5(a) shows the  $\text{P-PO}_4^{3-}$  concentration profiles in the laboratory batch experiments  
23  
24 251 while Figure 5(b) shows the specific P removal rates ( $\text{mg P gvss}^{-1} \text{ d}^{-1}$ ). The average  
25  
26 252 biomass specific P removal rate observed during the first days, that is, in the period where  
27  
28 253 the maximum N removal rate was kept (no N limitations), was approximately  $2.6 \text{ mg P}$   
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30 254  $\text{gvss}^{-1} \text{ d}^{-1}$ . As it was previously indicated, P was not a limiting nutrient in the present  
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32 255 work. Moreover, the buffer capacity of the BBM growth medium kept pH between 6.0  
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34 256 and 7.0 avoiding P precipitation (Cai et al. 2013).

37 257 From the approximate values of the nutrient removal rates previously calculated, a mass  
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39 258 stoichiometric removal ratio IC/N/P of 100/6.8/2.2 was found. It would indicate a mass  
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41 259 removal N/P ratio of approximately 3.1, which can be considered as a low value compared  
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43 260 to previously reported values ( $\text{N/P} = 7$ ) regarding N and P removal by microalgae (Acién  
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45 261 Fernández et al. 2013; Ruiz et al. 2013); nevertheless, these works usually refer to  
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47 262 ammonium nitrogen capture instead of nitrate.

### 54 263 **3.3. Nutrient removal and microalgae production in the pilot-scale photobioreactor**

1 264 Figure 6 shows (a) the microalgae and (b) the dissolved inorganic carbon (IC)  
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3 265 concentrations, in the effluent of the pilot scale photobioreactor during the 7 months of  
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5 266 continuous operation. According to the results obtained in the laboratory experiments, the  
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8 267 hydraulic retention time (HRT) used in the first month of operation was 5.5 d. However,  
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10 268 as the removal of the limiting nutrient (nitrate) was not completed (see later discussion,  
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12  
13 269 Figure 7), HRT was increased to 6.0 d in the second month and, finally, to 6.5 d during  
14  
15 270 the rest of the continuous operation period. It can be observed in Figure 6a that the effluent  
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17 271 microalgae concentration was approximately stabilized in  $0.5 \text{ g}_{\text{vss}} \text{ L}^{-1}$  (horizontal line),  
18  
19 272 which means an average microalgae productivity of  $76 \text{ mg}_{\text{vss}} \text{ L}^{-1} \text{ d}^{-1}$ , that is  
20  
21 273 approximately 12% lower than the productivity value found in laboratory (Section 3.2).  
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23 274 The IC effluent concentration (Figure 6b) showed significant fluctuations but an  
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25 275 approximate average value of  $85 \text{ mg L}^{-1}$  (horizontal line) may be considered; it  
26  
27 276 corresponded to an approximate removal efficiency of 82% with respect to the IC in the  
28  
29 277 saturated liquid stream flowing into the photobioreactor from the absorption tank. Thus,  
30  
31 278 it was calculated a IC removal rate value of  $121 \text{ mg C g}_{\text{vss}}^{-1} \text{ d}^{-1}$  and a biomass yield of  
32  
33 279  $1.26 \text{ g}_{\text{vss}} \text{ gC}^{-1}$ , which are similar values to those found in the laboratory experiments.  
34  
35 280 The microalgae productivity values obtained in the present work could be compared to  
36  
37 281 other previously reported values. We have selected previous works regarding nitrogen  
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39 282 (mainly ammonium) and phosphate removal in secondary effluents using *Chlorella*  
40  
41 283 *vulgaris*. For instance, Ruiz et al. (2013) reported values between  $40$  and  $170 \text{ g L}^{-1} \text{ d}^{-1}$ ;  
42  
43 284 Gao et al. (2014) reported  $10.3 \text{ g L}^{-1} \text{ d}^{-1}$ ; and Marbelia et al. (2014) reported  $33 \text{ g L}^{-1} \text{ d}^{-1}$   
44  
45 285 using 5 d as HRT. Honda et al. (2012) reviewed microalgae productivity values between  
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47 286  $48$  and  $1500 \text{ mg L}^{-1} \text{ d}^{-1}$  under different experimental conditions. Finally, Arbib et al.  
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1 287 (2015) reported values between 600 and 800 g L<sup>-1</sup> d<sup>-1</sup> using again 5d as HRT. All these  
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3 288 reported results show great variability depending on the specific experimental conditions.  
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7  
8 290 Figure 7 shows the effluent nutrient concentrations (N-NO<sub>3</sub><sup>-</sup> and P-PO<sub>4</sub><sup>3-</sup>) during the  
9  
10 291 operation of the pilot scale photobioreactor (data points) together with the inlet N and P  
11  
12 292 concentrations (horizontal lines). As previously indicated, phosphorus was always in  
13  
14 293 excess, therefore being nitrate the limiting nutrient and the parameter used to decide about  
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16 294 the HRT conditions. Effluent N-NO<sub>3</sub><sup>-</sup> concentrations were in the range 0-25 mg L<sup>-1</sup>  
17  
18 295 throughout the first month of operation (HRT 5.5 d), which indicated that this nutrient  
19  
20 296 was not completely used. On the contrary, after increasing HRT to 6.5 d, N was almost  
21  
22 297 completely consumed and thus N became as limiting nutrient. According to the high P  
23  
24 298 inlet concentration and the lower P capture capacity of microalgae, P removal efficiency  
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26 299 was quite lower than N removal efficiency.  
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28 300 From Figure 7, average effluent N and P concentrations were estimated, being 2.1 and  
29  
30 301 38.3 mg L<sup>-1</sup>, respectively. They corresponded to nutrient removal rates of 5.9 mg N L<sup>-1</sup>d<sup>-1</sup>  
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32 302 <sup>1</sup> and 1.5 mg P L<sup>-1</sup>d<sup>-1</sup>, respectively, and biomass specific removal rates of 11.7 mg N gvss<sup>-1</sup>  
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34 303 <sup>1</sup> d<sup>-1</sup> and 3.04 mg P gvss<sup>-1</sup> d<sup>-1</sup>. According to the IC, N and P removal rates at the final  
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36 304 stationary period, the mass stoichiometric removal ratio IC/N/P was calculated as  
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38 305 100/9.8/2.5. It means a N/P ratio of 3.9 which is higher than that obtained in the laboratory  
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40 306 experiments (3.1) but still lower than those formerly reported in the literature (Acién  
41  
42 307 Fernández et al. 2013; Ruiz et al. 2013).  
43  
44 308 As previously stated, nitrate removal studies by microalgae are not as common as  
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46 309 ammonium removal studies. Honda et al (2012) reported lower growth rate values for *C.*  
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48 310 *vulgaris* using nitrate compared to those using ammonium. Gao et al. (2014) reported that  
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1 311 *C. vulgaris* did not seem to use nitrate as substrate while, in contrast, Aslan and Kapdan  
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3 312 (2006) reported that *C. vulgaris* removed 510 mg N-NO<sub>3</sub><sup>-</sup> and 29 mg P-PO<sub>4</sub><sup>3-</sup> in a 6 d  
4  
5 313 batch period. Finally, Coppens et al. (2014) showed some results for different nitrate-  
6  
7 314 accumulating microalgae (but not for *C. vulgaris*).

#### 10 315 **3.4. Scaling up implications and comparison with classical BNR techniques**

12 316 The scaling up has been carried out in the present work using as main criteria to use a  
13  
14 317 similar HRT in lab and pilot plant experiments. Additionally, air was fed in the same  
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16 318 conditions, an excess of IC and P concentrations was kept and we tried to maintain similar  
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18 319 values of T, pH and lighting. However, different results were obtained under both scales.  
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20 320 The main differences observed in the pilot-scale test with respect to laboratory were a  
21  
22 321 slight decrease in microalgae productivity (which caused that higher HRT were  
23  
24 322 necessary) and a slight increase in N and P removal rates. So, in general, it cannot be said  
25  
26 323 that scaling up of our process caused a clear efficiency decrease as reported in previous  
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28 324 works (Ruiz et al. 2013; Van den Hende et al. 2014).

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30 325 The main differences observed in the operating conditions between the two processes  
31  
32 326 (laboratory and pilot scales) were the liquid flow mode and turbulence levels, the  
33  
34 327 dissolved oxygen concentrations and the lighting level. High dissolved oxygen  
35  
36 328 concentrations in the pilot scale process could have influenced negatively the microalgae  
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38 329 performance as inhibition could appear at oxygen concentrations from 7 mg L<sup>-1</sup> (Acién  
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40 330 Fernández et al. 2013). An air compressor was necessary to avoid dead flow zones in the  
41  
42 331 horizontal tubes which would cause even higher oxygen accumulation. Moreover, the  
43  
44 332 movement induced to microalgae is quite different in both systems reaching much lower  
45  
46 333 turbulence levels in the pilot scale; in fact, Reynolds number was 25·10<sup>3</sup> in laboratory  
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48 334 test while it was lower than 10 in the pilot plant. Arbib et al. (2013) reported that high  
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1 335 turbulence is necessary to avoid biofouling and excessive dissolved oxygen levels.  
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3 336 Biofouling (microalgae accumulation in the internal reactor surface) was detected only in  
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5 337 the pilot scale operation in the present work, which was related to the low turbulence and  
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7  
8 338 the high reactor specific surface ( $\text{m}^2 \text{m}^{-3}$ ). Biofouling in the pilot scale caused organic  
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10 339 waste accumulation that could eventually produce problems such as reported by  
11  
12 340 Grobbelaar (2012), i.e. predators, pathogens and alien microalgae invasion. Regarding  
13  
14 341 the light intensity, since the higher specific surface of the pilot scale photobioreactor, it  
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16 342 could receive more light than the lab-scale reactor; in fact, light intensity was only  
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18 343 correctly controlled in the pilot plant during the last stationary 100 days. In general, it can  
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20 344 be said that all abovementioned factors were difficult to control and would cause the  
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22 345 differences found between the results from laboratory and pilot scales.  
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25 346 Regarding the nutrient removal applications, on one hand, the current microalgae  
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27 347 technology is clearly slower and less effective than classical BNR secondary processes  
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29 348 but, on the other hand, it could be a more sustainable technology. Judd et al. (2015)  
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31 349 reported the advantages and disadvantages of using microalgae instead of the classical  
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33 350 BNR processes. According to these authors, because of the lower nutrient removal rates  
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35 351 and the high surface (low depth) necessary for microalgae, PBR systems may be 15 times  
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37 352 slower than classical activated sludge BNR systems, and overall there is two order of  
38  
39 353 magnitude difference in footprint between them. Against this, the biomass yield is quite  
40  
41 354 lower in PBR and microalgae is considered a valuable product, and they also conclude  
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43 355 that microalgae nutrient removal is less effective but involves lower operation costs and,  
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45 356 additionally, allows  $\text{CO}_2$  capture. Additionally, Marbelia et al. (2014) describe different  
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47 357 scenarios to combine classical wastewater treatment plants with microalgae nutrient  
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49 358 removal and they propose approximate values of power consumption and operation costs,  
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1 359 and they reported that conventional BNR are energy-intensive and involve extra  
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3 360 equipment instruments which may cover 60-80% of the total energy consumption in the  
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6 361 treatment process.

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11 363 **4. Conclusions**

12  
13 364 The results showed in this work showed that the microalgae *Chlorella.vulgaris* is capable  
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15 365 to effectively remove nitrate and phosphate from a synthetic secondary effluent, with no  
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18 366 need of organic carbon and rendering a valuable waste material. Although N and P  
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21 367 removal rates were low compared to classical biological nutrient removal secondary  
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23 368 processes, the use of microalgae could be considered a more sustainable technology for  
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25 369 wastewater treatment.

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27 370 Microalgae productivity values found here were similar to previous reported works. The  
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30 371 work in a 150 L pilot plant showed the difficulties to keep an adequate control of the  
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33 372 process variables. Nevertheless, although microalgae production was lower those that of  
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35 373 the laboratory tests, N and P removal rates were slightly increased in the pilot plant. On  
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38 374 summary, it can be said that scaling up of the process caused some differences with  
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40 375 respect to the laboratory results being them mainly attributed to the differences in  
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43 376 turbulence, dissolved oxygen concentrations and lighting levels. These points have been  
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45 377 identified as to be critical, and so it is considered that future efforts should be made to  
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48 378 improve control of such factors and thus to allow studies that compare laboratory and  
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50 379 pilot scale systems.

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386

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11 454 **Fig.1** Lab-scale photobioreactor  
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15 456 **Fig. 2** Pilot-scale photobioreactor. (a) Scheme. (b) Photograph  
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19 458 **Fig. 3** (a) Microalgae growth and (b) dissolved inorganic carbon consumption in the  
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21 laboratory batch experiments  
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26 461 **Fig. 4** (a) N-NO<sub>3</sub><sup>-</sup> concentration profiles and (b) biomass specific N-NO<sub>3</sub><sup>-</sup> removal rates  
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28 in the laboratory batch experiments  
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33 464 **Fig. 5** (a) P-PO<sub>4</sub><sup>3-</sup> concentration profiles and (b) biomass specific P-PO<sub>4</sub><sup>3-</sup> removal rates  
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35 in the laboratory batch experiments  
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43 dissolved inorganic carbon (IC) concentration  
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48 470 **Fig. 7** Nutrients effluent concentrations (N-NO<sub>3</sub><sup>-</sup> and P-PO<sub>4</sub><sup>3-</sup>) during the operation of the  
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50 pilot scale photobioreactor  
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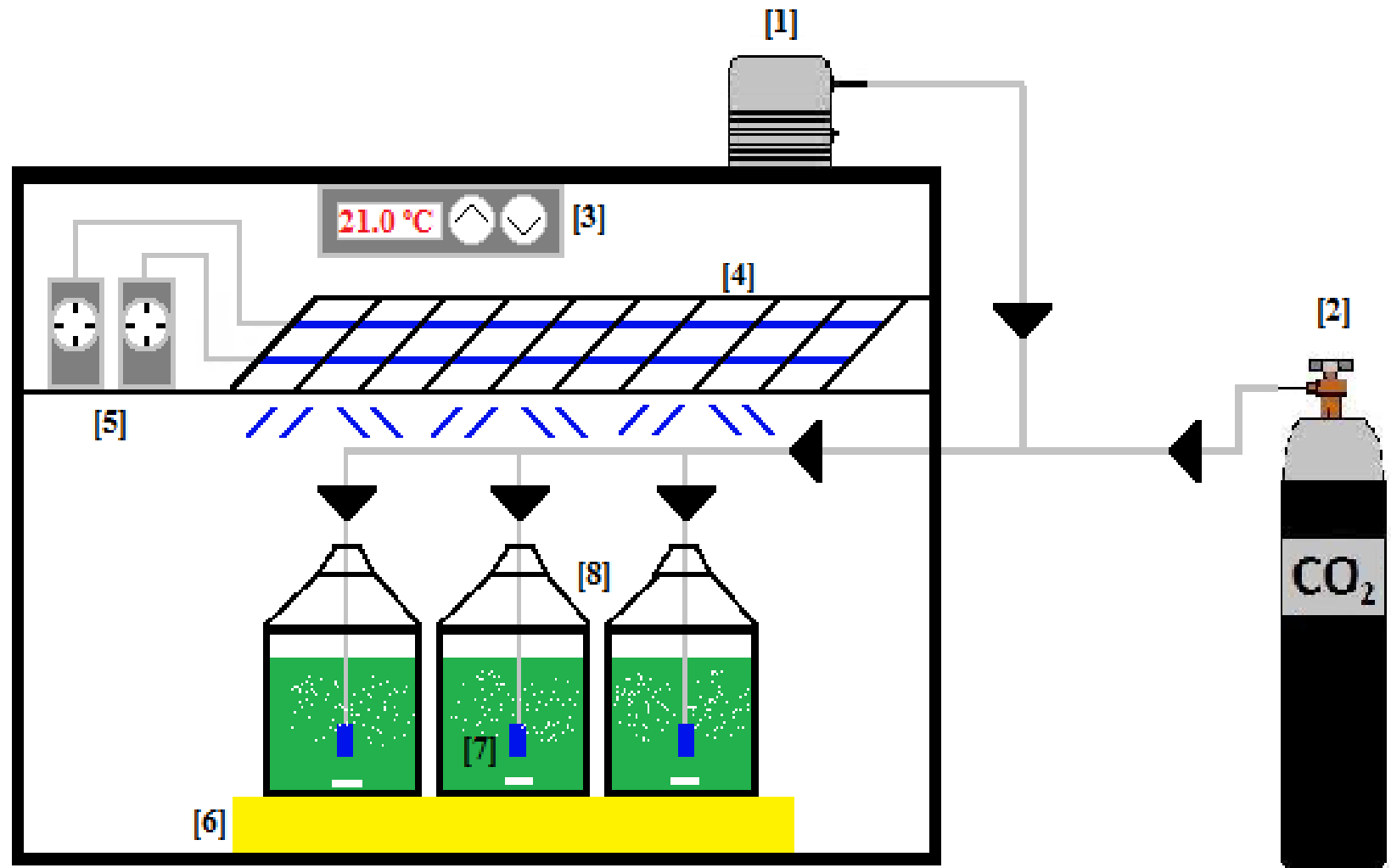


Figure 1

Figure 2

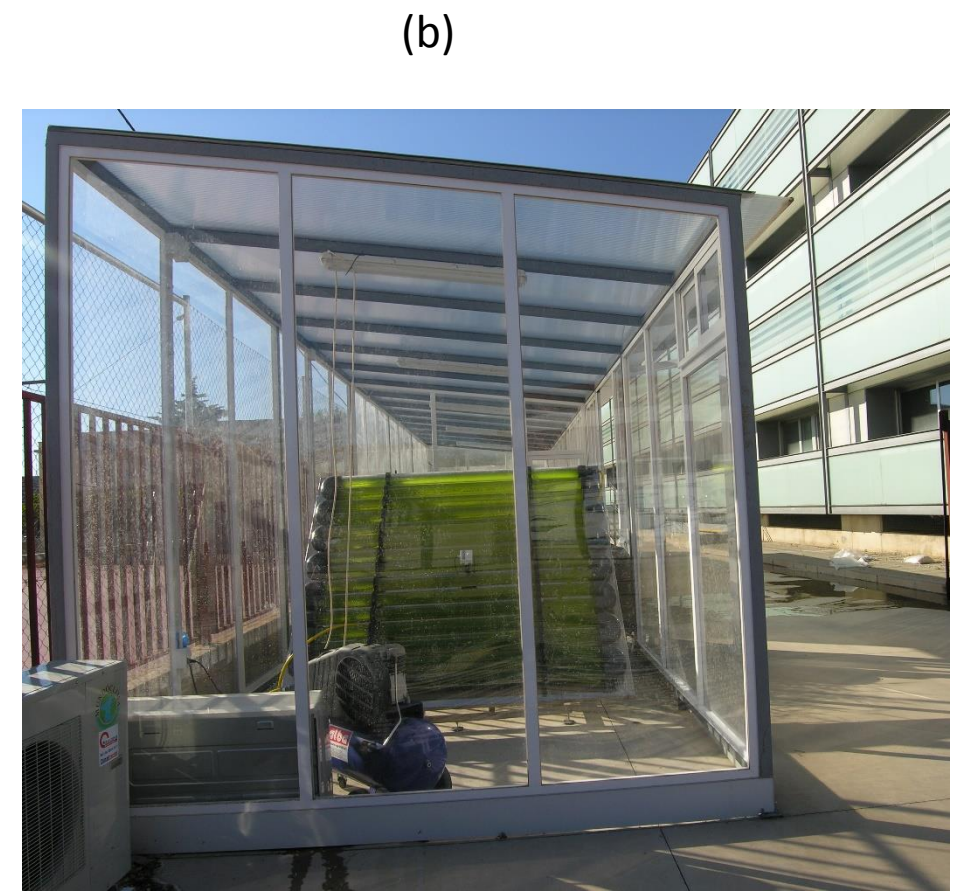
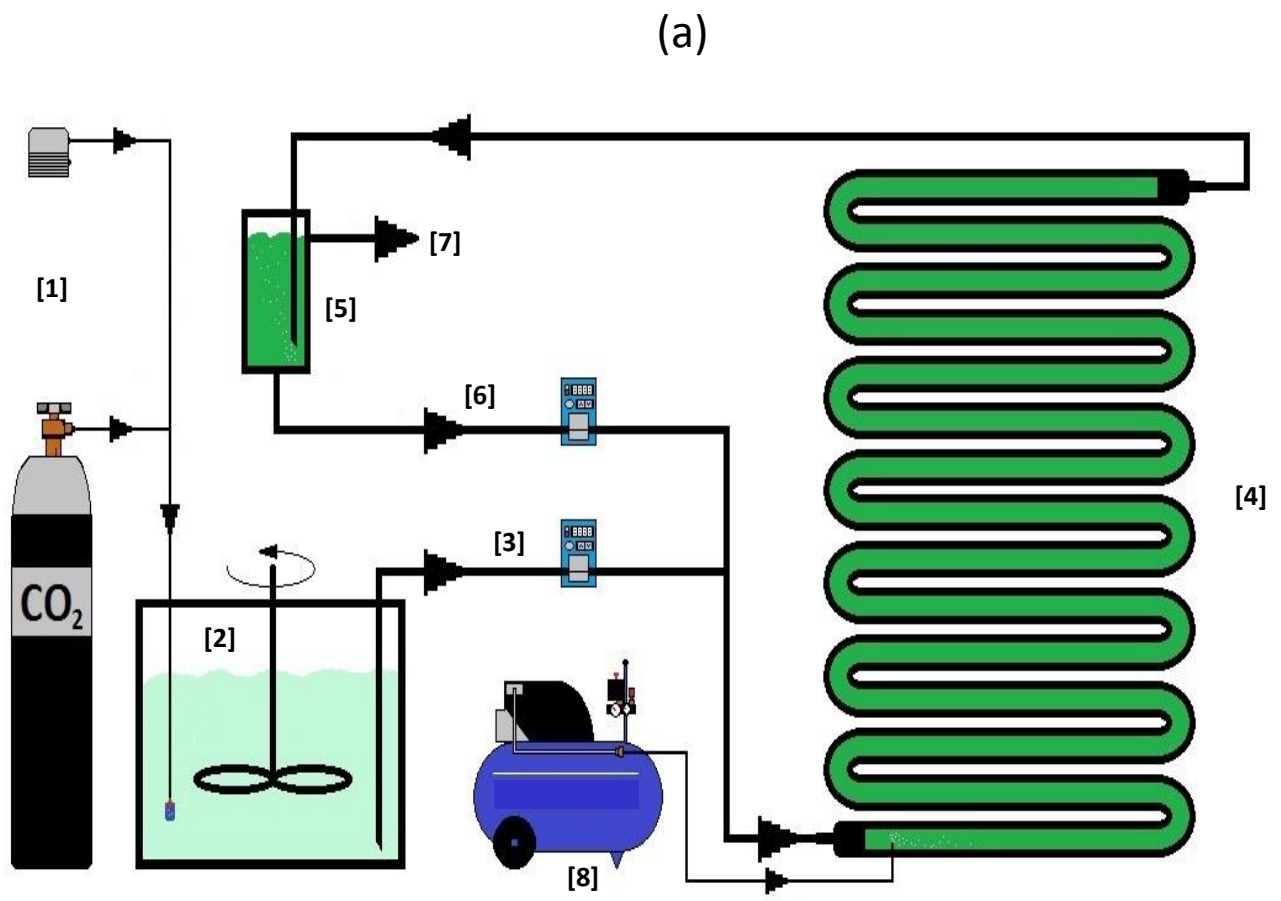


Figure 2

Figure 3

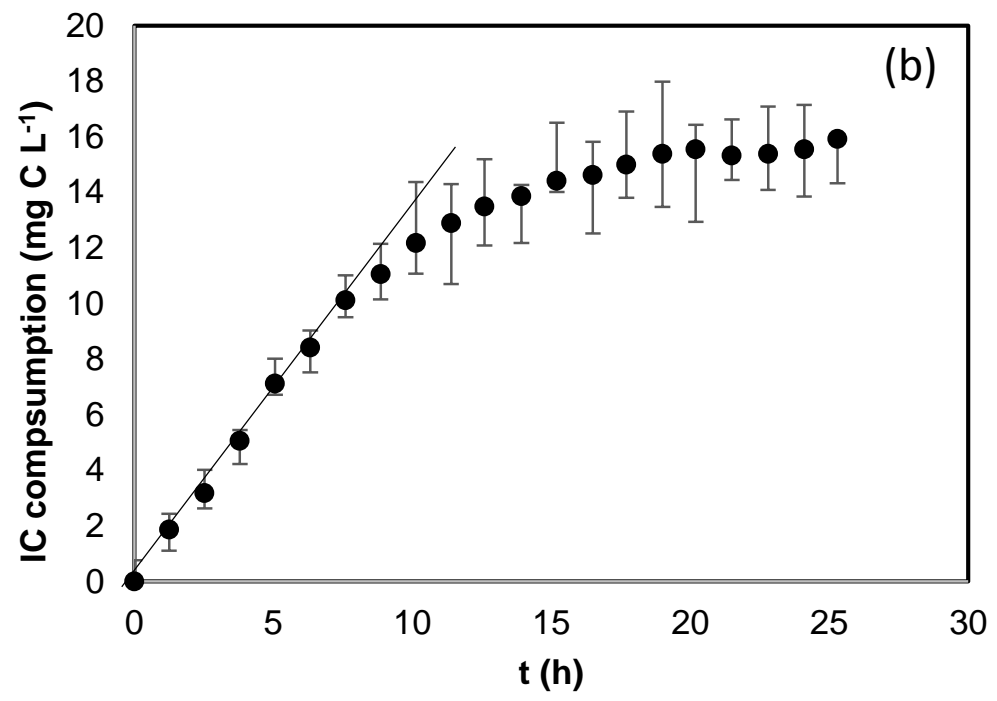
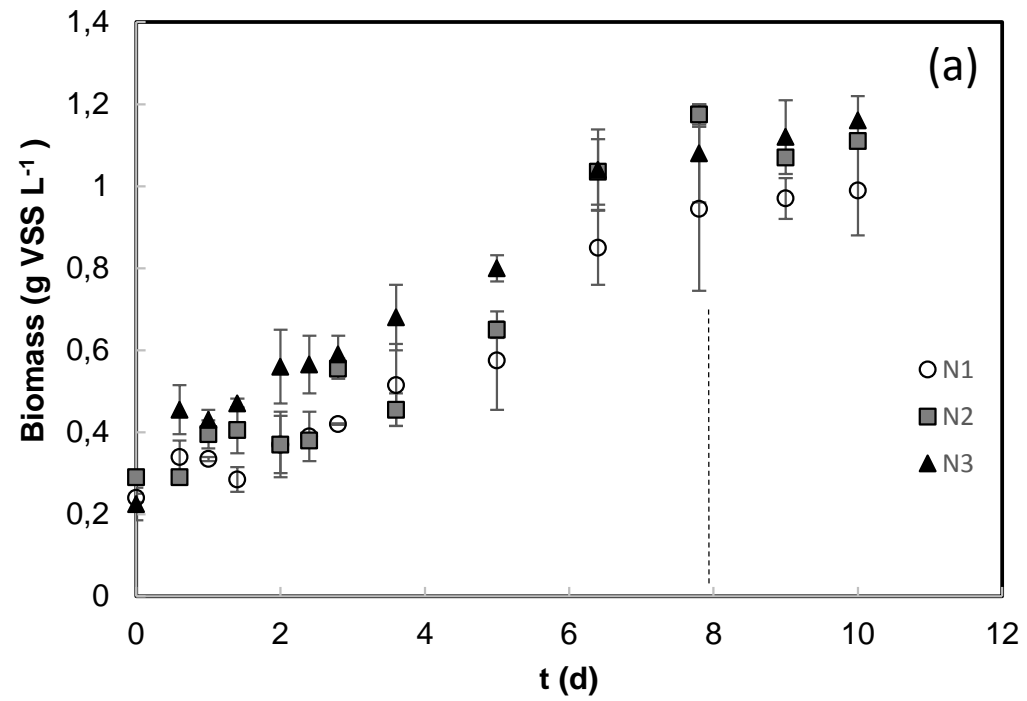


Figure 3

Figure 4

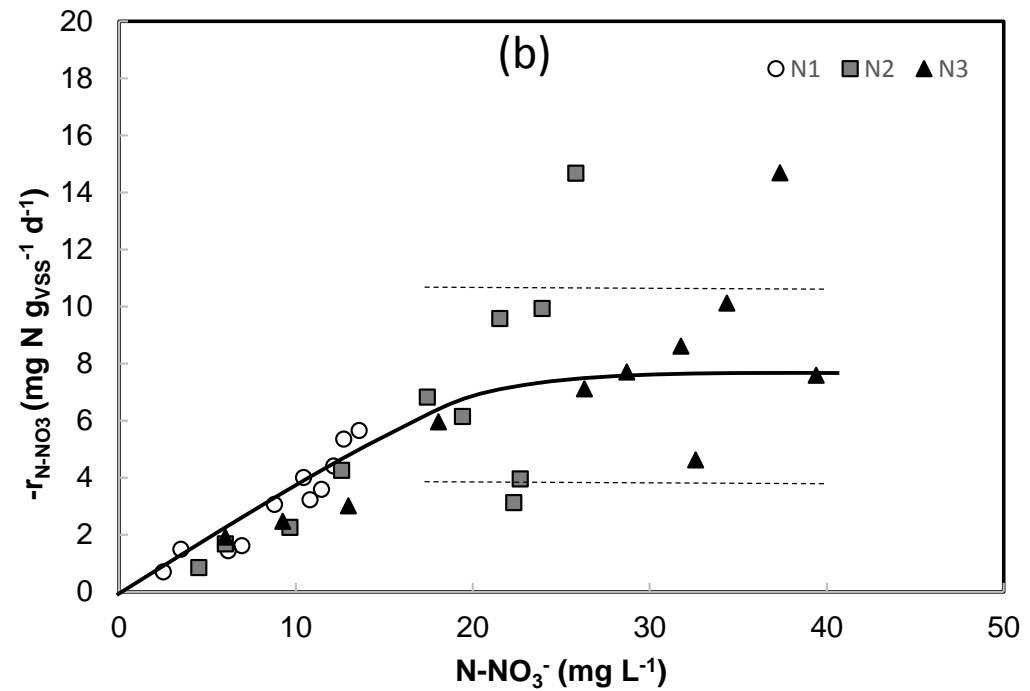
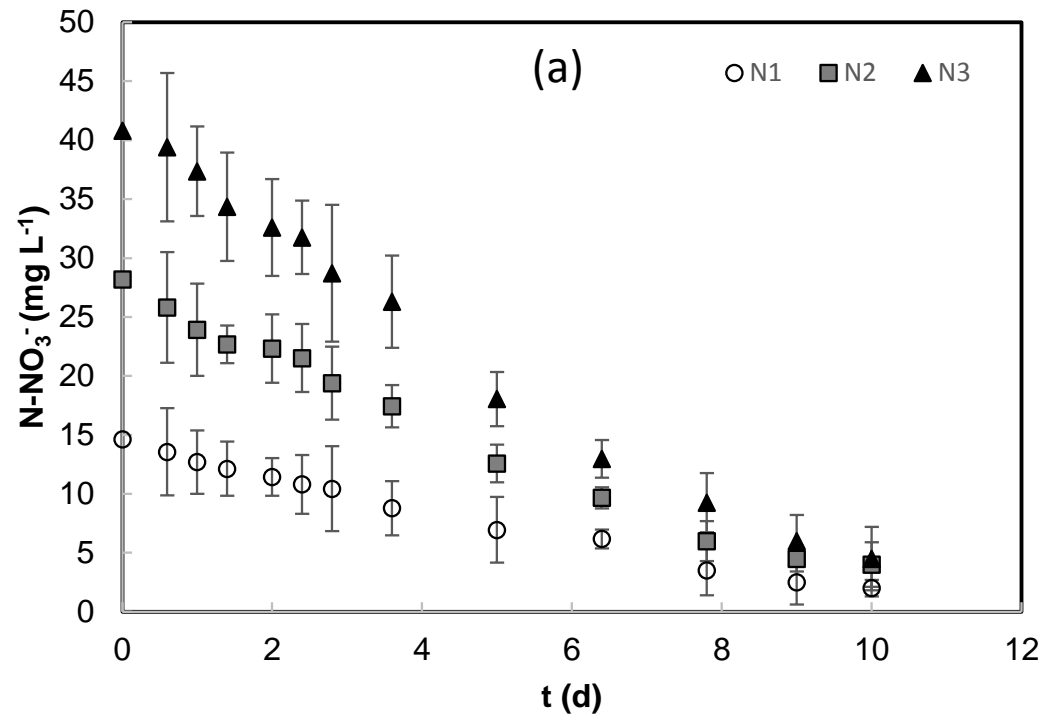


Figure 4



Figure 5

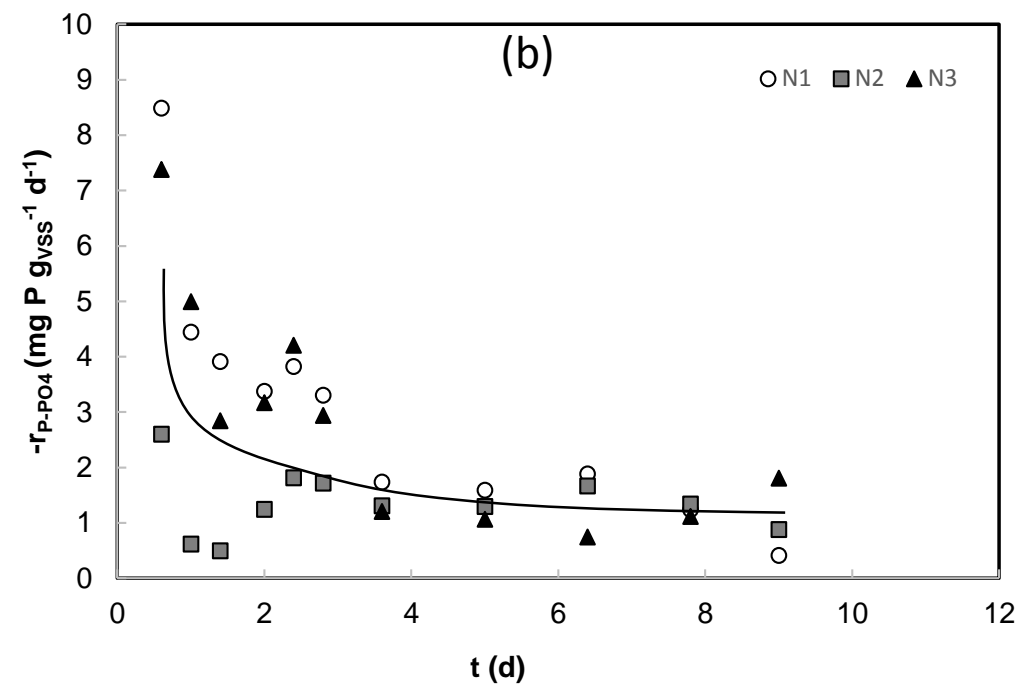
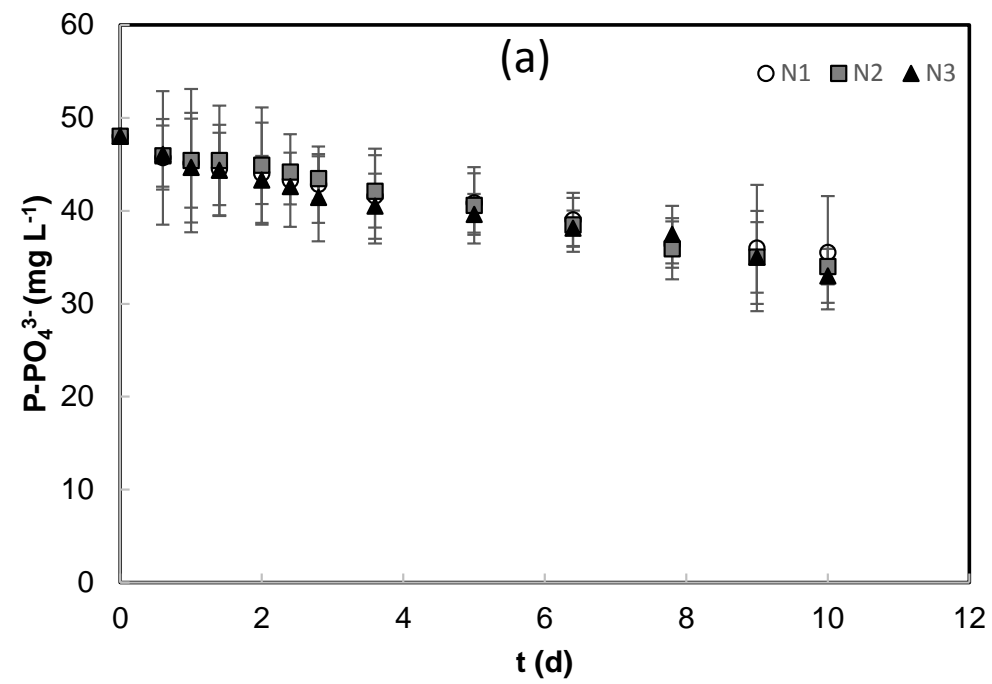


Figure 5

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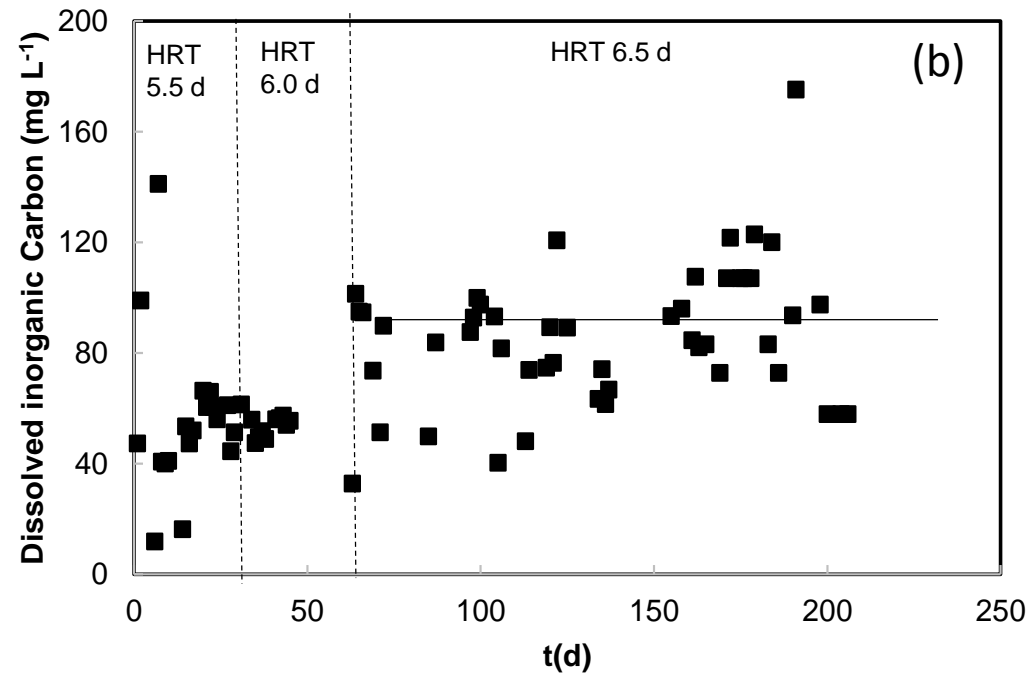
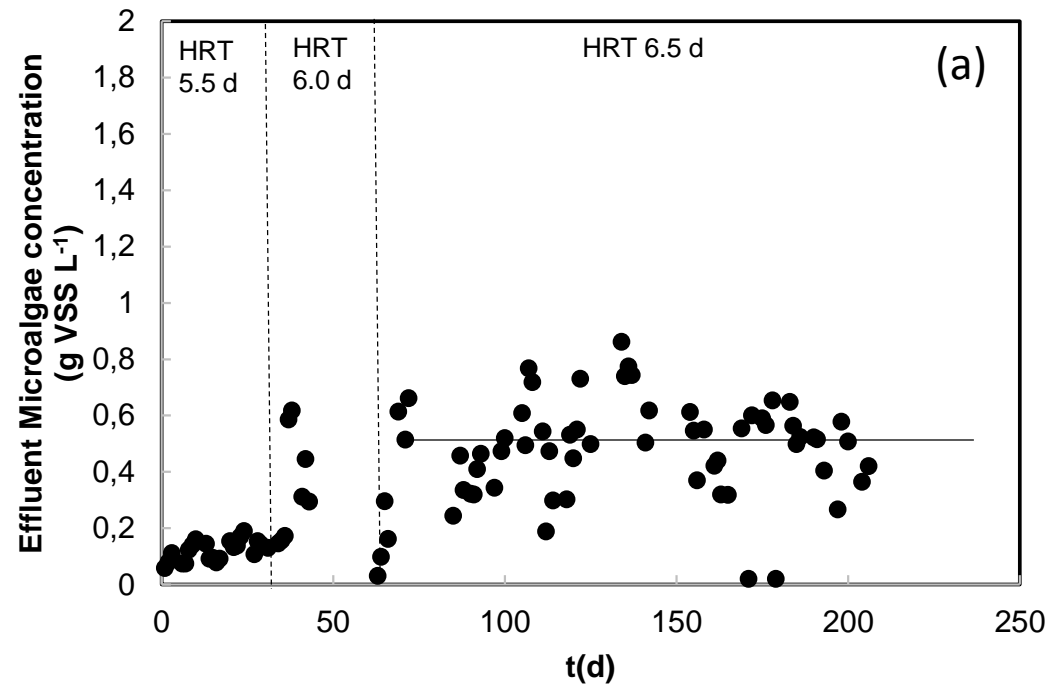


Figure 6

Figure 7

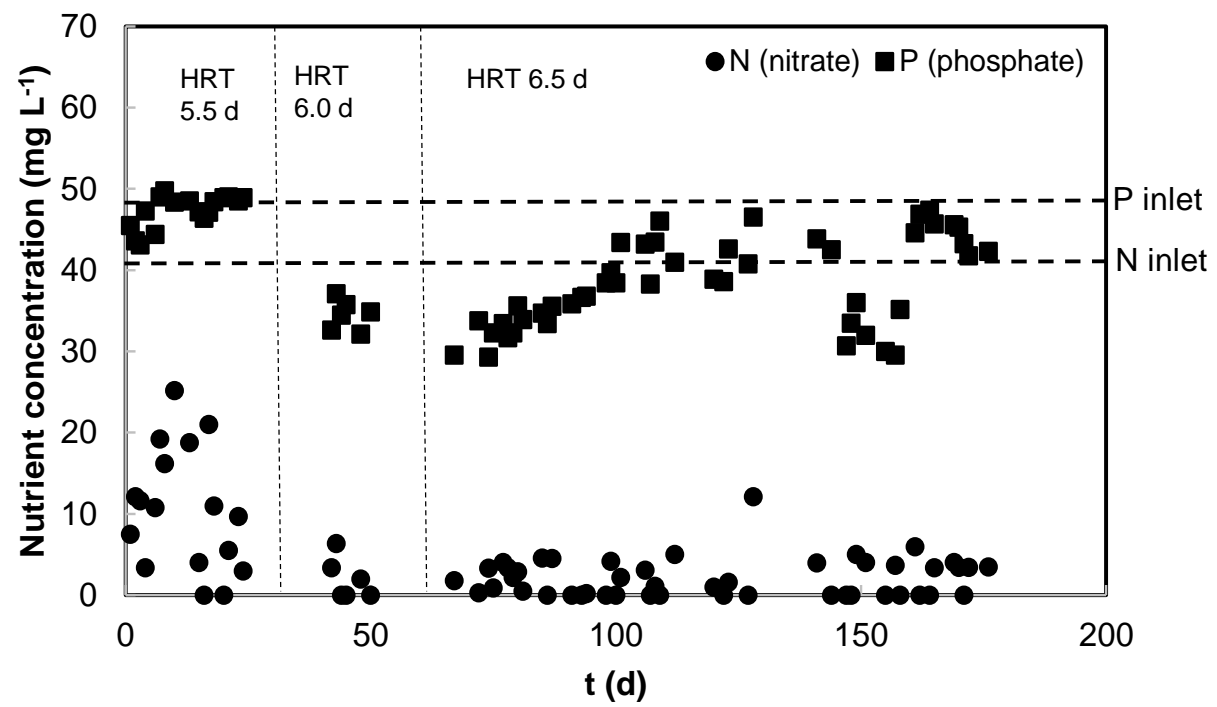


Figure 7



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