

1                   **OLIVE POMACE VERSUS NATURAL GAS FOR METHANOL**  
2                   **PRODUCTION: A LIFE CYCLE ASSESSMENT**

3  
4    María Puig-Gamero, María Magdalena Parascanu, Paula Sánchez, Luz Sanchez-Silva\*

5                   Department of Chemical Engineering, University of Castilla –La Mancha

6                   Avda. Camilo José Cela 12, 13071 Ciudad Real, Spain

7                   \*Corresponding author phone: +34 926 29 53 00 ext. 6307

8                   e-mail: [marialuz.sanchez@uclm.es](mailto:marialuz.sanchez@uclm.es)

9  
10  
11  
12  
13  
14  
15  
16  
17  
18  
19  
20  
21  
22  
23  
24  
25   **Abstract:**

26           Increasing demand for methanol production and global competition for the use of  
27 natural resources are key issues in finding new and environmentally routes for methanol  
28 production. In this work, life cycle assessment was performed using software SimaPro v9  
29 to analyse the environmental impact of methanol production process from olive pomace  
30 and compare with natural gas route. The main stages considered in the methanol  
31 production from olive pomace were: olive production, olive oil extraction and methanol  
32 production. In addition, the methanol production in turn can be divided in three main  
33 processes, olive pomace gasification, syngas purification and methanol production which  
34 were also evaluated individually. Finally, the global environmental impacts associated  
35 with the methanol production from olive pomace was compared with a conventional  
36 methanol production from natural gas. This assessment determined that the production of  
37 methanol from the olive pomace had a greater environmental impact for all the categories  
38 studied except the one related to the shortage of fossil fuels. These results were directly  
39 related to the technical performance of the processes.

40

41

42 **Keywords:** Gasification; olive pomace; natural gas; methanol; LCA; SimaPro v9.

43

44

45

46

47

48

49 **1. Introduction**

50           Nowadays, methanol is one of the highly synthesised chemicals around the globe,  
51 being a key molecule in both our daily lives and the global economy (Demirbas 2007).  
52 This alcohol can be used directly as a clean fuel or can be mixed with other specific fuels.  
53 From a thermodynamic point of view, the use of methanol in internal combustion engines  
54 could have many benefits, such as a considerable increase in power and energy efficiency  
55 (Amigun et al. 2010). Furthermore, it is essential for the synthesis of other chemical  
56 products such as paints, solvents, formaldehyde, gasoline and polymers, among others.

57           Worldwide, there are currently around 90 industrial plants which have a production  
58 capacity of about 110 million ton (Methanex). Each day, nearly 0.2 Mt of methanol are  
59 used as a chemical feedstock and transportation fuel. In addition, the global methanol  
60 production has increased significantly in recent years due to its widespread use in the  
61 chemical and process industries and the development of olefins and fuel cells industries  
62 in Europe (Pérez-Fortes et al. 2016). However, European methanol demand is becoming  
63 increasingly dependent on imports to feed its market (ICIS).

64           Conventionally, the methanol is principally produced from fossil fuel, mainly  
65 through natural gas reforming with steam. The main problem associated with the  
66 methanol production from fossil resources are the emissions of large amounts of  
67 greenhouse gases (GHG). Moreover, these emissions of methanol production depend on  
68 process configuration and feedstock (in the range from 462 to 2965 kg CO<sub>2</sub> eq/t methanol)  
69 (Kajaste et al. 2018).

70           However, the last climate change study published by the Intergovernmental Panel  
71 on Climate Change (IPCC) (IPCC) established that use of fossil fuels must be reduced  
72 by half in under 15 years and eliminated completely in 30 years to reduce emissions of  
73 greenhouse gases. Thus, due to the environmental policies are increasingly rigorous and  
74 moreover, most of these fuels are imported from outside the EU, the interest in

75 researching new sustainable alternatives has increased. Among the non-fossil production  
76 alternatives, methanol obtained through biomass gasification is considered as a suitable  
77 candidate to replace conventional methanol from natural gas, helping to reduce climate  
78 change and diminishing such a dependence.

79 Olive pomace biomass is a large resource in Mediterranean countries, especially in  
80 Spain, since it is the world's leading olive oil producer and exporter, accounting for about  
81 half of total global production (Commission 2019). In addition, previous works have  
82 demonstrated that olive pomace is a suitable feedstock for methanol production. In this  
83 regard, methanol production from syngas obtained through biomass gasification was  
84 simulated using Aspen Plus<sup>®</sup> (Puig-Gamero et al. 2018), being the aim to simulate,  
85 validate and optimize, but above all, demonstrate its technical feasibility. However,  
86 important economic or environmental aspects were ignored. In a second study, the  
87 economic viability of this plant was carried out demonstrating that the methanol  
88 production through olive pomace gasification was highly probable would be profitable  
89 (Puig-Gamero et al. 2020). Nonetheless, its suitability needs to be confirmed by carrying  
90 out economic and techno-environmental analyses upon which decision can be made.  
91 Hence, this paper focused on environmental assessment, since this is essential to achieve  
92 a sustainable future.

93 Currently, one of the most used tools for evaluating the environmental impacts  
94 associated with a product, process or activity is the Life Cycle Assessment (LCA). LCA  
95 is a standardized methodology in accordance with ISO 14040 and 14044 standards  
96 (Standardization 2006a; Standardization 2006b). Hence, this analysis involves the  
97 complete cycle of the product, process or activity, including the extraction and feedstock  
98 processing, production, transportation and allocation, use, recycling and final disposal of  
99 the product. In this sense, LCA helps to identify weaknesses and features of the product

100 that could be improved to reduce environmental impacts and use fewer resources in the  
101 life cycle stages.

102 Several studies on the LCA associated with the methanol production have been  
103 recently reported. Different authors (Renó et al. (2011) [11], Tangviroon et al. (2014) [12]  
104 and Yadav et al. (2020) [13]) studied the LCA of the methanol production from biomass,  
105 concretely, sugarcane bagasse, soybean oil and wood, respectively. For its part, Li et al.,  
106 compared the LCA of methanol production from coke oven gas, coal and natural gas (Li  
107 et al. 2018). Gao et al., analysed the coal-to-methanol production and identified that this  
108 process produced significant emissions of GHG (Gao et al. 2018). Lerner et al., studied  
109 the best conditions to reduce the GHG for methanol production from natural gas (Lerner  
110 et al. 2018). Moreover, the comparison of different raw materials (coal, natural gas, flue  
111 gas, corn, and wood biomass) was studied by Kajaste et al. (Kajaste et al. 2018).

112 On the other hand, there are studies focused on LCA associated with olive pomace  
113 valorization. In this sense, Parascanu et al., carried out the environmental assessment of  
114 olive pomace valorization through pyrolysis, combustion and gasification (Parascanu et  
115 al. 2018a; Parascanu et al. 2018b). Duman et al., studied the LCA of olive pomace  
116 utilization in Turkey. For that purpose, they analysed different scenarios involving  
117 producing fuel pellets, fodder additives and composting (Duman et al. 2020). In addition,  
118 Uceda-Rodríguez et al., evaluated the environmental benefits associated with the addition  
119 of olive pomace for the manufacture of lightweight aggregates (Uceda-Rodríguez et al.  
120 2020).

121 However, to the best of the authors' knowledge, there are not environmental  
122 assessment of the olive pomace valorization through gasification for its subsequent use  
123 in the production of methanol. Hence, the main aim of this study was to compare in terms  
124 of environmental impact, the methanol production from natural gas (conventional

125 methanol production) and from syngas obtained by olive pomace gasification.  
126 Additionally, the environmental impact performed for each process involved in methanol  
127 production from olive pomace was also evaluated.

## 128 **2. Methodology**

129 The LCA study was carried out according to the ISO 14040 standard  
130 (Standardization 2006b), which recommends four main steps: goal and scope definition,  
131 life cycle inventory analysis, environmental impact assessment and interpretation of the  
132 results.

### 133 2.1 Goal and scope definition

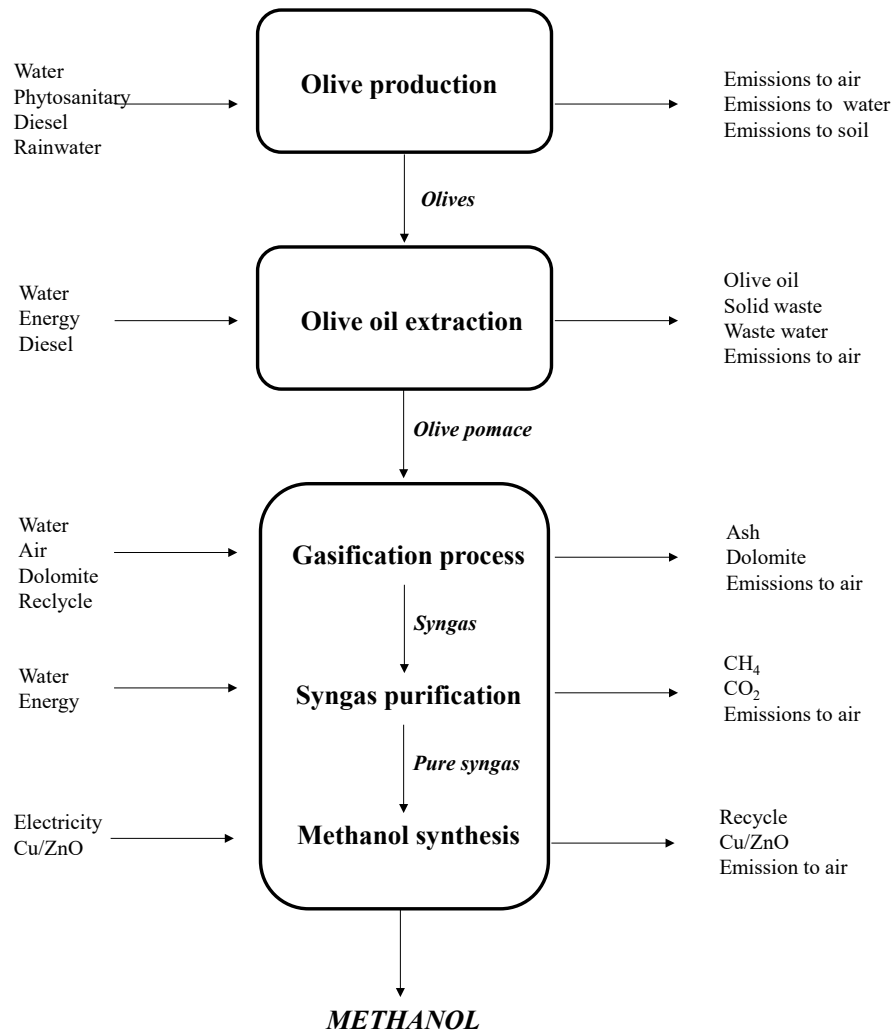
134 The aim of this study was to compare in terms of environmental impact, the  
135 methanol production from natural gas (conventional methanol production) and from  
136 syngas obtained by olive pomace gasification. For that purpose, the LCA was carried out  
137 using SimaPro v9 software. Besides, it was performed in accordance with the cradle to  
138 gate approach, including all the phases involved from raw materials extraction to  
139 methanol production. Moreover, the functional unit was defined as 1 kg of methanol.

140 Figs 1 and 2 show the diagram of the system boundaries for the methanol production  
141 from olive pomace and natural gas, respectively. As can be seen, the main stages  
142 considered in conventional method were the natural gas extraction, importation and  
143 methanol production, while for methanol production from biomass the main stages  
144 considered were: olive production, olive oil extraction and methanol synthesis. In  
145 addition, the three main processes involved in the methanol production were also studied.

146

147 In order to carry out the LCA, several hypotheses have been considered in order to  
148 avoid overlapping in the decision-making process:

- 149 - Planting and tree growth have been omitted due to the long time in which there is  
150 no production.
- 151 - The transport of olive to the olive oil mill and the transport of fertilizers have been  
152 considered.
- 153 - The acquisition and the maintenance of the agriculture machinery are not  
154 considered.
- 155 - The acquisition and the maintenance of machinery used to extract the natural gas  
156 and gas pipeline are not deemed.
- 157 - The equipment involved in the methanol plant are excluded from the assessment.
- 158 - The olive oil mill and the methanol plant are located at the same place. Thus, the  
159 biomass transport in this last case has not been considered.
- 160 - Desulfurization process of conventional method for producing methanol was not  
161 taken in account.

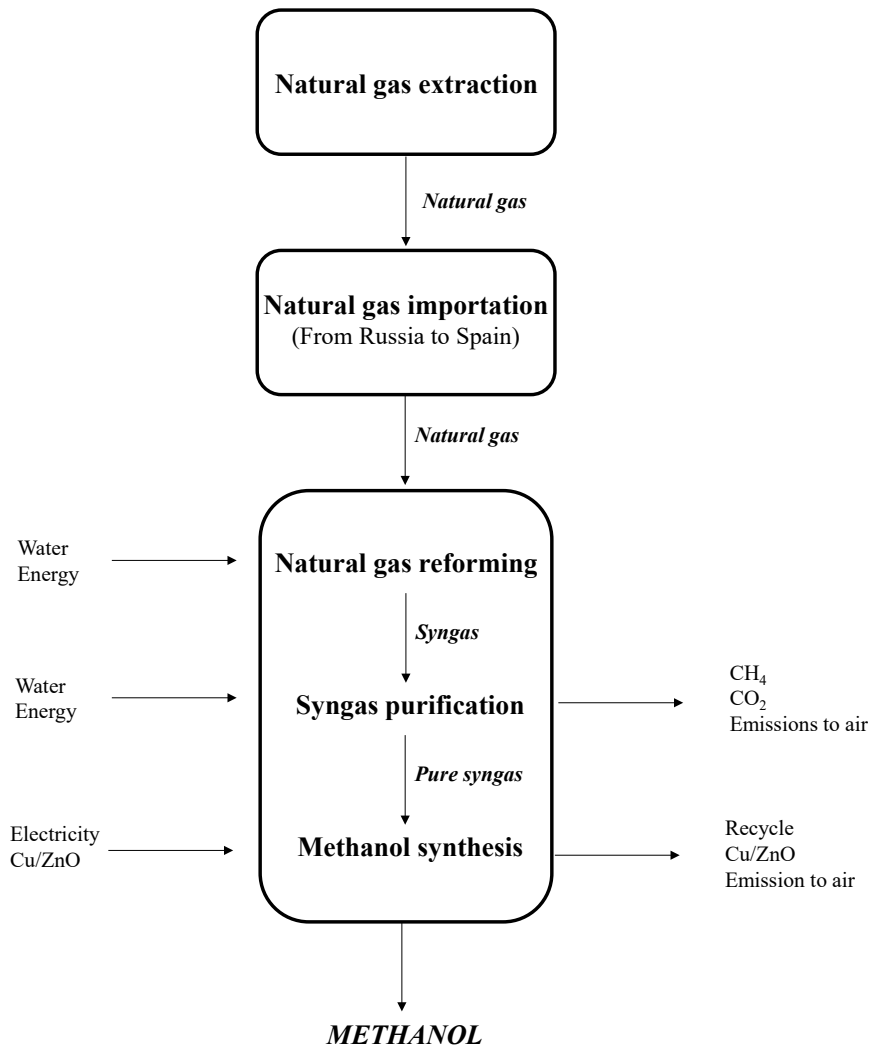


162

163

**Fig 1.** System boundary of methanol production from olive pomace gasification.





164

165

**Fig 2.** System boundary of methanol production from natural gas.

166

167 2.2 Life cycle inventory analysis

168 2.2.1 Methanol production from olive pomace gasification

169 The inventory data, such as the direct inputs and outputs of each stage considered  
 170 in the boundary system, were collected from a real olive mill plant, the Aspen  
 171 Plus® software and the Ecoinvent database (Ecoinvent, 2017). The olive production and  
 172 olive extraction stage were explained in detail in previous works (Parascanu et al. 2018a;  
 173 Parascanu et al. 2018b).

174 a) Olive production stage

175 Table 1 summarizes the inventory data used which were provided by the real olive  
176 mill plant *Aceites García de la Cruz* located in Toledo (Spain). These data were processed  
177 to determine the corresponding values for the functional unit. On the other hand, the air,  
178 water and soil emissions associated with the olives production stage were taken from the  
179 Ecoinvent 3.4 database (Ecoinvent, 2017). Table 2 shows the main input and output of  
180 the agricultural stage referred to functional unit (1 kg of methanol).

181 b) Olive oil extraction stage

182 Once the olives were harvested, they were transported to the oil mill to produce the  
183 olive oil. In this case, all collected data were obtained from *Aceites García de la*  
184 *Cruz* olive oil mill (Table 3). Table 4 lists the main input and output of the olive oil  
185 extraction stage referred to functional unit (1 kg of methanol).

186 Finally, economical allocation was used for the olive pomace, since the physical  
187 allocation is not appropriate due to the byproduct is much higher than the main product  
188 and, thus, this assumption can result in important estimation errors. In this sense, to  
189 prevent unfairness, the price of the product and by-products has been used to calculate  
190 the allocation keys (Parascanu et al. 2018b). In this way, the economic allocation factor  
191 for the olive oil was 97.2 % (the average price for 1 kg of extra olive oil is 3.65€ /kg and  
192 the average yield is 18.5 %), for the olive pomace was 1.7 % (the average price for olive  
193 pomace is 15€ /ton and the average yield is 73.5 %) and for the olive stone was 1.1%  
194 (the average price of olive stone is 90€ /ton and the average yield is 8 %) (Parascanu et  
195 al. 2018b).

196

197 **Table 1.** Information considered for olive production stage (provided by *Aceites García*  
198 *de la Cruz* olive oil mill).

---

<b>Final product</b>	Olives
<b>Harvest year</b>	2015/2016
<b>Growing area (ha)</b>	40
<b>Conversion factor (kg olives/ha)</b>	718
<b>Means of transport</b>	Truck
<b>Irrigation system</b>	Rainwater
<b>Rainwater collected (m<sup>3</sup>/ha)</b>	2140
<b>Fertilizers transport (km)</b>	285
<b>Phytosanitary treatment (L)</b>	4500, twice a year
<b>Total diesel consumed (L/ha)</b>	24.2

---

199

200

201

202

203

204

205

206

207

208

209

210

211

212

213

214

215

216

217 **Table 2.** Inputs and outputs for the olive production stage, considering to the functional  
 218 unit, 1 kg of methanol produced from olive pomace gasification.

<i>Inputs*</i>		
Water	m <sup>3</sup>	6.53E-05
Fertilizers	kg	7.51E-6
Diesel	kg	7.46E-06
<i>Outputs*</i>		
Olives	kg	2.99
<i>Emissions to air**</i>		
NH <sub>3</sub>	kg	3.95E-03
CO <sub>2</sub>	kg	5.22E-06
N <sub>2</sub> O	kg	6.89E-09
NO <sub>x</sub>	kg	9.09E-12
H <sub>2</sub> O	kg	1.20E-14
<i>Emissions to water**</i>		
Cr	kg	1.34E-05
Cu	kg	4.19E-06
Pb	kg	1.04E-06
Hg	kg	3.77E-09
Ni	kg	1.15E-06
NO <sub>3</sub>	kg	3.02E-02
P	kg	5.09E-05
H <sub>2</sub> O	m <sup>3</sup>	5.99E-02
Zn	kg	1.05E-05
<i>Emissions to soil**</i>		
Cd	kg	1.98E-06
Cr	kg	7.31E-06
Cu	kg	3.50E-06
Dimethoate	kg	8.95E-07
Pb	kg	7.64E-06
Hg	kg	1.61E-09
Ni	kg	2.70E-06
Zn	kg	3.59E-06

\* Olive mill plant data; \*\* Ecoinvent database

219

220

221

222

223

224 **Table 3.** Information considered for olive oil extraction stage (provided by *Aceites*  
 225 *García de la Cruz* olive oil mill).

<b>Final product</b>	Olive oil
<b>Distance between plot and olive oil mill (km)</b>	19
<b>Means of transport</b>	Truck
<b>By-products</b>	Olive pomace, olive stone, solid waste (leaves, dust and stones)
<b>Factor conversion (kg/ha):</b>	
<b>Olive oil</b>	126.5
<b>Olive pomace</b>	534
<b>Olive stone</b>	57.5
<b>Solid waste</b>	9.4
<b>Operating time (h)</b>	2208
<b>Electrical energy consumed (kW)</b>	78
<b>Water consumption (m<sup>3</sup>)</b>	887
<b>Total diesel consumed (L/ha)</b>	24.2

226

227

228 c) Methanol production stage (from olive pomace)

229 The olive pomace was used as a raw material in this stage. Moreover, Aspen Plus®  
 230 v9.0 was employed to simulate the process of methanol synthesis from biomass which  
 231 has been evaluated environmentally herein. The design specifications for modelling the  
 232 process, the optimization and the validation have been published in a previous study  
 233 (Puig-Gamero et al. 2018). The processes involved in the production of methanol from  
 234 biomass are: gasification process, syngas cleaning and methanol synthesis. Firstly, the  
 235 biomass was gasified at 900 °C in a double chamber gasifier using steam as gasifying  
 236 agent and a steam/biomass mass ratio of 0.9. Moreover, dolomite was used as the catalyst  
 237 to decompose the tar produced. Then, the syngas produced was fed to the pressure swing  
 238 adsorption (PSA) in which the gas is cleaning and adjusting to achieve a H<sub>2</sub>/CO ratio  
 239 close to 2.4-2.5. Finally, the syngas with the correct ratio is fed to methanol synthesis.

240 The operating conditions for methanol synthesis were 220 °C and 55 atm. In addition, due  
 241 to the low conversions obtained in this process, Cu/ZnO was used as catalyst. Finally, to  
 242 improve the system performance, the waste stream of methanol synthesis was recycled to  
 243 the combustion chamber. Figs 3, 4 and 5 show the Aspen Plus® flowsheet of gasification  
 244 process, syngas cleaning and methanol synthesis, respectively. In addition, Tables 5, 6  
 245 and 7 summarize the main block used in each process. Finally, Table 8 lists the main input  
 246 and output of each process involved in the methanol production from olive pomace  
 247 (referred to FU).

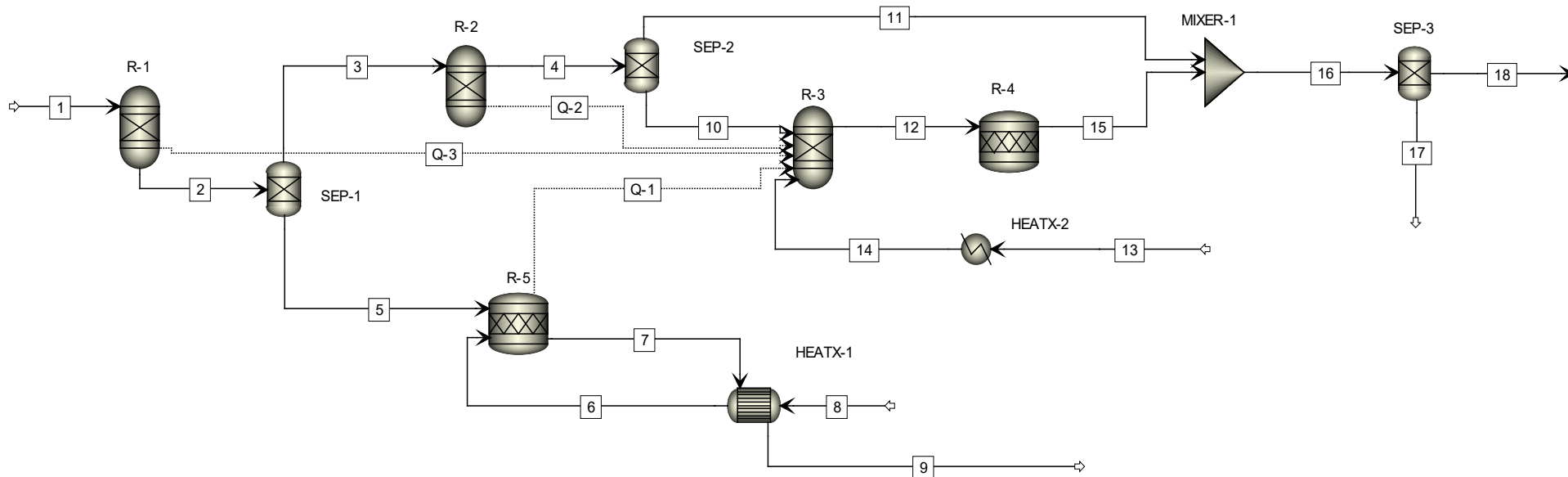
248 **Table 4.** Inputs and outputs of the olive oil extraction stage, considering the functional  
 249 unit of 1 kg of methanol produced from the olive pomace gasification.

<i>Inputs*</i>		
Olives	kg	2.99E+00
Water	m <sup>3</sup>	1.52E-02
Electrical energy	MJ	2.39E+00
Diesel	L	1.52E-01
<i>Outputs*</i>		
Olive oil	kg	5.34E-01
Olive pomace	kg	2.23E+00
Olive stone	kg	2.08E-01
Waste	kg	3.92E-02
<i>Emissions to air*</i>		
NO <sub>x</sub>	kg	1.20E-04
SO <sub>2</sub>	kg	1.94E-05
CO	kg	1.70E-03
PM (particulate matter)	kg	6.25E-05

250 \*Olive mill plant data

251

252



253

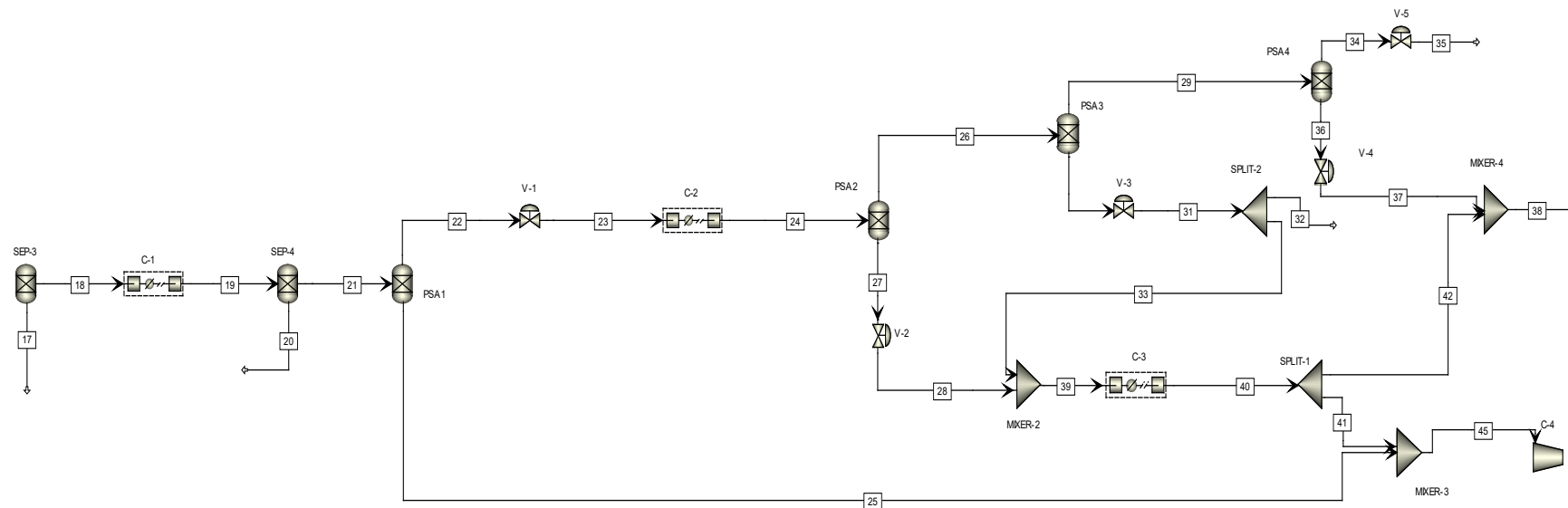
254

255

**Fig 3.** Aspen Plus® flowsheet simulation of the gasification process.

256

257



258

259

260

261

**Fig 4.** Aspen Plus® flowsheet simulation of the syngas cleaning.



262

263

264

265

266

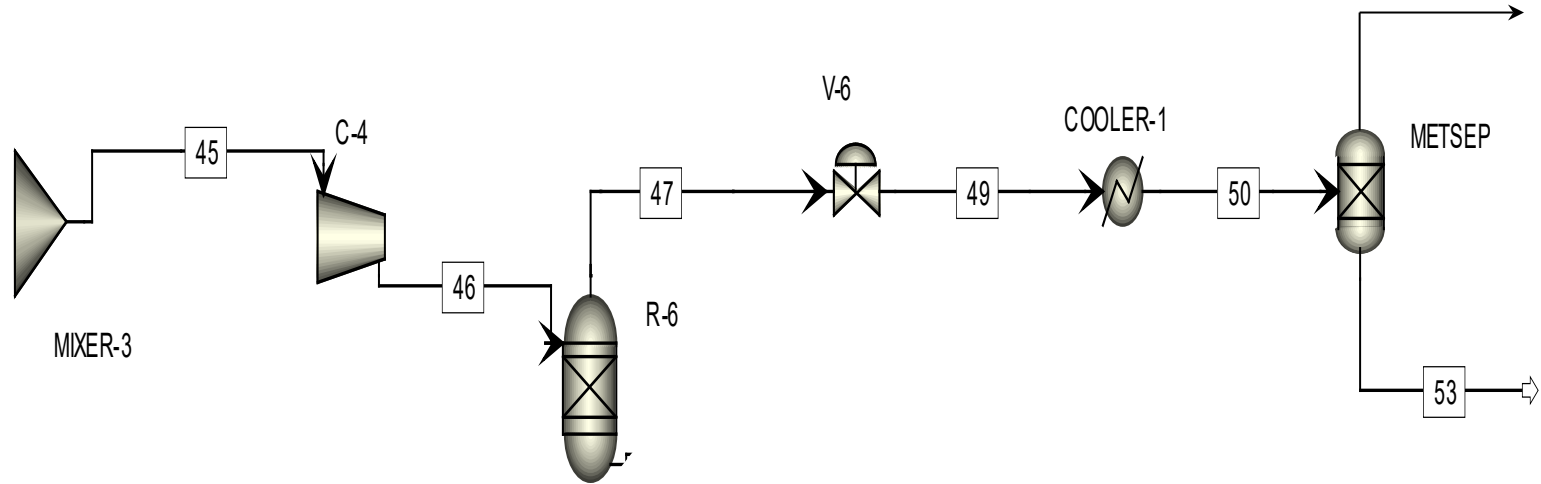
267

268

269

270

271



**Fig 5.** Aspen Plus® flowsheet simulation of methanol synthesis.

**Table 5.** Blocks description used in the gasification model.

<i>NAME</i>	<i>TYPE</i>	<i>DESCRIPTION</i>
<i>R-1</i>	RYIELD	Biomass pyrolysis reactor, it decomposed the biomass into its compounds and ash. It operated at 1 atm and 900 °C.
<i>SEP-1</i>	SEP	Separator of the amount of char necessary to achieve the gasification temperature.
<i>R-2</i>	RGIBBS	It models chemical equilibrium minimizing the Gibbs free energy. It was used to produce CO <sub>2</sub> , CO, CH <sub>4</sub> , H <sub>2</sub> S and NH <sub>3</sub> . It operated at 1 atm and 900 °C.
<i>SEP-2</i>	SEP	Separator of H <sub>2</sub> S and NH <sub>3</sub> from C, H <sub>2</sub> , CO <sub>2</sub> , CO and CH <sub>4</sub> .
<i>R-3</i>	RGIBBS	Gasifier. It operated at 1 atm and 900 °C.
<i>R-4</i>	RSTOIC	This reactor allowed to introduce a catalyst. It was used to model the tar reforming using Dolomite as a catalyst. It operated at 1 atm and 900 °C.
<i>SEP-3</i>	SEP	It separated the active char from the syngas.
<i>R-5</i>	RSTOIC	Char combustion reactor. It operated at 1 atm and 900 °C.
<i>HEATX-1</i>	HEATX	Exchange heat between the outlet stream from R-5 and the air inlet stream which was warmed up to 150 °C
<i>HEATX-2</i>	HEATER	Heater to warm up the gasifying agent (water steam) to 150 °C.

275

**Table 6.** Blocks description used in the syngas cleaning model.

<i>NAME</i>	<i>TYPE</i>	<i>DESCRIPTION</i>
<i>C-1, C-2 and C-3</i>	MULTISTAGE COMPRESSORS	The multistage compressors were used to compress to 30 atm and cool down to 35 °C the unclean gas, the PSA2 inlet stream and the CO and CO <sub>2</sub> mixture, respectively.
<i>SEP-4</i>	SEP	Separator of the water condensed and the syngas.
<i>PSA1, PSA2, PSA3 and PSA4</i>	SEP	Separator to adsorb and separate at 30 and 35 atm rich H <sub>2</sub> , rich CO, rich CO <sub>2</sub> and rich CH <sub>4</sub> , respectively.

276

277

**Table 7.** Blocks description used in the methanol synthesis.

<i>NAME</i>	<i>TYPE</i>	<i>DESCRIPTION</i>
<i>C-4</i>	COMPRESSOR	It was used to compress and to purify the syngas.
<i>R-6</i>	REQUIL	Methanol synthesis reactor.
<i>COOLER-1</i>	COOLER	It was used to cool down to 25 °C the methanol produced to separate it.
<i>METSEP</i>	SEP	Separator of the crude methanol and gas-phase and impurities.

278

279

280

281

282

283

284  
285

**Table 8.** Inputs and outputs of the methanol production plant from olive pomace gasification (FU=1kg of methanol) (from software Aspen Plus®).

<b>OLIVE POMACE GASIFICATION</b>		
<i>Inputs</i>		
Olive pomace	kg	2.23E+00
Air	kg	3.28E+00
Water	kg	2.01E+00
Recycling gas *	kg	8.89E-01
Dolomite	kg	2.03E+00
<i>Outputs</i>		
Syngas	kg	3.99E+00
Ash	kg	1.69E-01
Dolomite (waste)	kg	2.03E+00
<i>Emissions to air</i>		
Heat	MJ	2.23E-04
O <sub>2</sub>	kg	1.91E-01
CO <sub>2</sub>	kg	1.34E+00
H <sub>2</sub> O	kg	9.97E-01
N <sub>2</sub>	kg	1.71E+00
<b>SYNGAS CLEANING</b>		
<i>Inputs</i>		
Syngas	kg	3.99E+00
Energy	MJ	1.90E+01
<i>Outputs</i>		
Clean syngas	kg	1.93E+00
CH <sub>4</sub>	kg	6.29E-02
CO <sub>2</sub>	kg	3.14E-01
<i>Emissions to air</i>		
H <sub>2</sub> O	kg	9.77E-01
Heat	MJ	2.93E+01
CO	kg	4.91E-01
CO <sub>2</sub>	kg	1.71E-01
CH <sub>4</sub>	kg	1.01E-03
H <sub>2</sub> S	kg	6.03E-03
NH <sub>3</sub>	kg	4.00E-02

286  
287  
288  
289  
290

291

**Table 8.** Continuation.

<b>METHANOL SYNTHESIS</b>		
<i>Inputs</i>		
Clean syngas	kg	1.93E+00
Energy	MJ	3.57E-01
Cu/Zn	kg	1.75E+00
<i>Outputs</i>		
CH <sub>3</sub> OH	kg	1.00E+00
Recycling gas	kg	9.22E-01
Cu/Zn (waste)	kg	1.75E+00
<i>Emissions to air</i>		
H <sub>2</sub> O	kg	4.81E-03
Heat	MJ	4.39E+00

292

\*Waste stream of methanol synthesis was recirculated to gasification process.

### 293 2.2.2 Methanol production from natural gas

294 As above mentioned, the stages involved in the methanol production from natural  
 295 gas are natural gas extraction, importation and methanol production (Fig 2). The  
 296 inventory data of the two first stages were collected directly from Ecoinvent database  
 297 (Ecoinvent, 2017), while the last stage was obtained from the Aspen Plus<sup>®</sup> software.

#### 298 a) Natural gas extraction

299 In this work, importation of natural gas from Russian to Spain was considered since  
 300 Spanish natural gas reserves are negligible, while Russian is one of the main suppliers to  
 301 the European natural gas market (Mikulska 2020).

302 Natural gas production, according to data collected from the Ecoinvent database  
 303 (reference), occurs through a series of stages: exploration, production, processing and  
 304 underground storage of natural gas. Gas production ends when gas is fed into the transport  
 305 pipeline to the country where it will be consumed. This process includes leaks produced

306 during production and the processing of crude oil. The water produced was considered to  
307 be discharged into surface waters.

308 b) Importation of natural gas

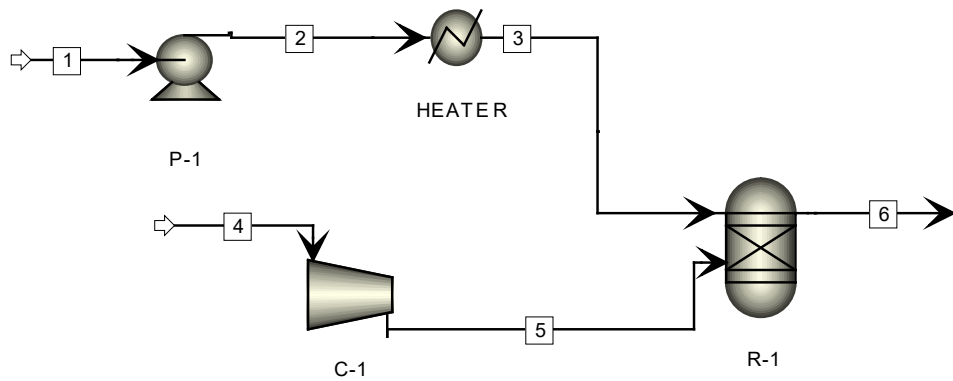
309 The inventory data of natural gas importation stage was also obtained from  
310 Ecoinvent database (Ecoinvent, 2017). These data describe the transport to export the  
311 natural gas from Russia to Spain in gas pipeline, including in this transport the losses and  
312 gas emissions during seasonal storage. The estimated average distance was 7000 km.

313 c) Methanol production stage (from natural gas)

314 A methanol production plant from natural gas was simulated using Aspen Plus<sup>®</sup> to  
315 obtain the inventory data of this stage. In the same way than methanol production from  
316 olive pomace, this stage can be divided in three main processes: natural gas reforming,  
317 syngas cleaning and methanol synthesis. In order to simplify the explanation of the  
318 simulation, the three main parts will be explained step by step. The fluid dynamic package  
319 selected was Peng-Robinson with Boston Mathias function, being the appropriate for high  
320 temperature processes (Pala et al. 2017).

321 - Natural gas reforming

322 Fig 6 shows the Aspen Plus<sup>®</sup> flowsheet of natural gas reforming process. In this  
323 work, an equilibrium model based on a Gibbs free energy minimization was used to model  
324 a steam reforming reactor. Firstly, the natural gas and steam was comprised at 30 bar and  
325 fed to block R-1. In this process, the natural gas reacted with steam at 900 °C, moreover,  
326 in this reactor CH<sub>4</sub>, CO<sub>2</sub>, CO, H<sub>2</sub>O and H<sub>2</sub> were fixed as the main reaction products. Then,  
327 the outlet syngas was introduced to PSA system. Table 9 summarizes the main blocks  
328 used in steam reforming process.



329

330

**Fig 6.** Aspen Plus® flowsheet simulation of the natural gas reforming.

331

**Table 9.** Blocks description used in the gasification model.

<i>NAME</i>	<i>TYPE</i>	<i>DESCRIPTION</i>
<i>P-1</i>	PUMP	It was used to compress to 30 atm the water.
<i>HEATER</i>	HEATER	Heater to warm up the water to 280 °C.
<i>C-1</i>	COMPRESSOR	It was used to compress to 30 atm the natural gas.
<i>R-1</i>	RGIBBS	It was used to model the natural gas reforming and operated at 30 atm and 900 °C.

332

333 - Syngas cleaning: Pressure swing adsorption (PSA)

334

The aim of this part was to clean the outlet gas of steam reforming to obtain a high-

335

quality syngas which will be used for the methanol synthesis. The outlet gas (stream 6)

336

was introduced into the PSA system, which consisted for four units, as shown in Fig 7.

337

All of them were simulated in a simplified way, by ideal separator, but working at realistic

338

temperature and pressure (35 °C and 30 atm, respectively) (Gutiérrez Ortiz et al. 2013).

339

Moreover, the percentage of component recovered in the PSA system was obtained from

340 the literature (Gutiérrez Ortiz et al. 2013). Table 10 summarizes the main blocks used in  
 341 this stage.

342 **Table 10.** Blocks description used in the syngas cleaning model.

<i>NAME</i>	<i>TYPE</i>	<i>DESCRIPTION</i>
<i>C-2 and C-3</i>	MULTISTAGE COMPRESSORS	The multistage compressors were used to compress to 30 atm and cool down to 35 °C the PSA2 inlet stream and the CO and H <sub>2</sub> mixture, respectively.
<i>SEP-4</i>	SEP	Separator of the water condensed and the syngas.
<i>PSA1, PSA2, PSA3 and PSA4</i>	SEP	Separator to adsorb and separate at 30 °C and 35 atm rich H <sub>2</sub> , rich CO, rich CO <sub>2</sub> and rich CH <sub>4</sub> , respectively.

343

344 - Methanol synthesis

345 Finally, the syngas was introduced to methanol synthesis reactor. The blocks used  
 346 to simulate the methanol synthesis are summarized in Table 11 and Fig 8 shows the  
 347 methanol synthesis flowsheet. In this work, the pressure and temperature of methanol  
 348 synthesis were 220 °C and 55 atm.

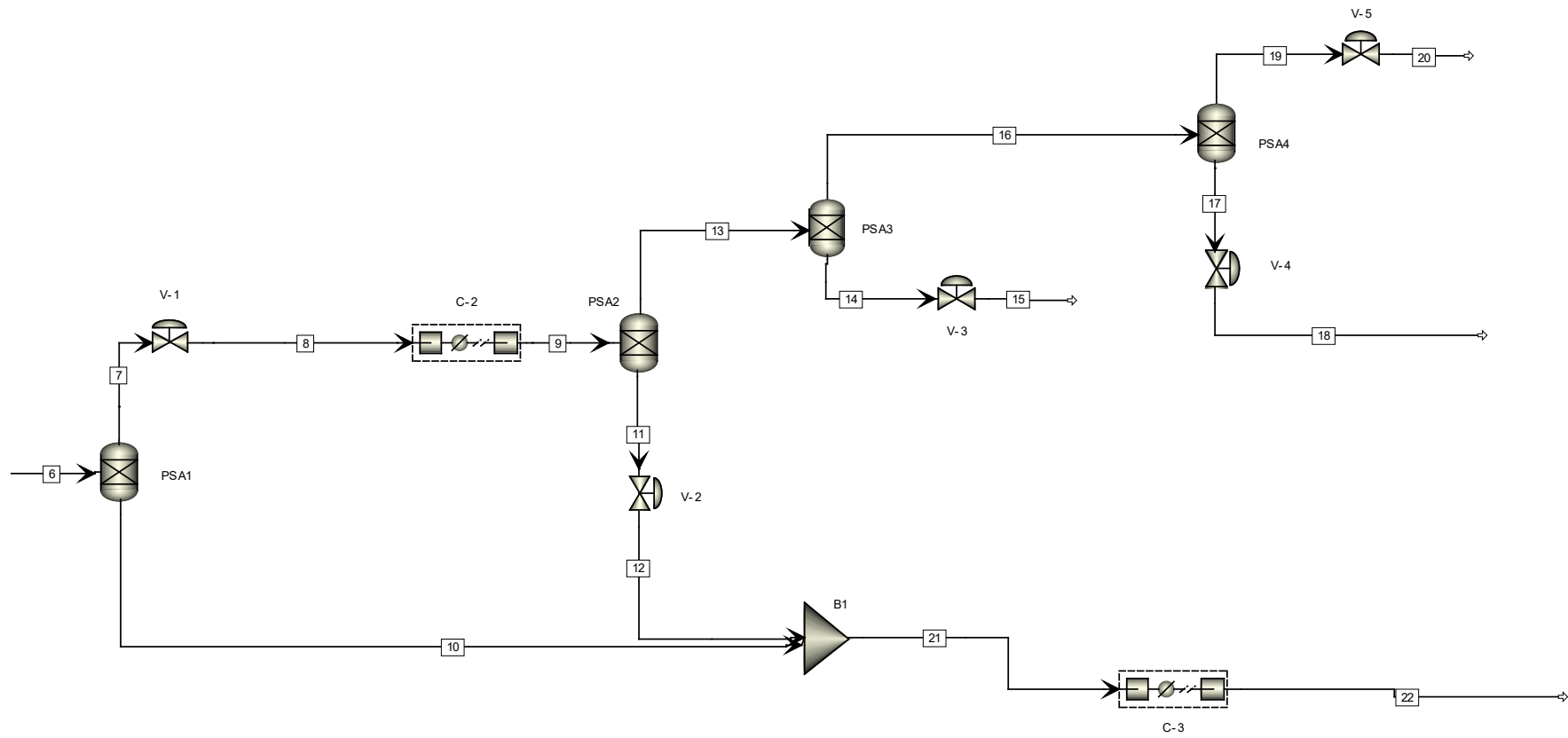
349 The syngas was introduced in the R-2 to accomplish the methanol production. The  
 350 catalyst selected to the synthesis was Cu/ZnO, achieving conversions of 35 % and 17 %  
 351 for CO and CO<sub>2</sub>, respectively (Trop et al. 2014). Then, reaction product was depressurized  
 352 to 1 atm and cooled down to 25 °C to condense and separate the crude methanol from the  
 353 gas-phase in the METSEP flash. Finally, stream 28, which contained unconverted H<sub>2</sub>,  
 354 was compressed and recycled to methanol reactor again. Finally, Table 12 lists the main  
 355 input and output of each process involved in the methanol production from olive pomace  
 356 (referred to FU).



**Table 11.** Blocks description used in methanol synthesis.

<i>NAME</i>	<i>TYPE</i>	<i>DESCRIPTION</i>
<i>C-4</i>	COMPRESSOR	It was used to compress and to purify the syngas.
<i>R-6</i>	REQUIL	Methanol synthesis reactor.
<i>COOLER-1</i>	COOLER	It was used to cool down to 25 °C the methanol produced to separate it.
<i>METSEP</i>	SEP	Separator of the crude methanol and gas-phase and impurities.

359



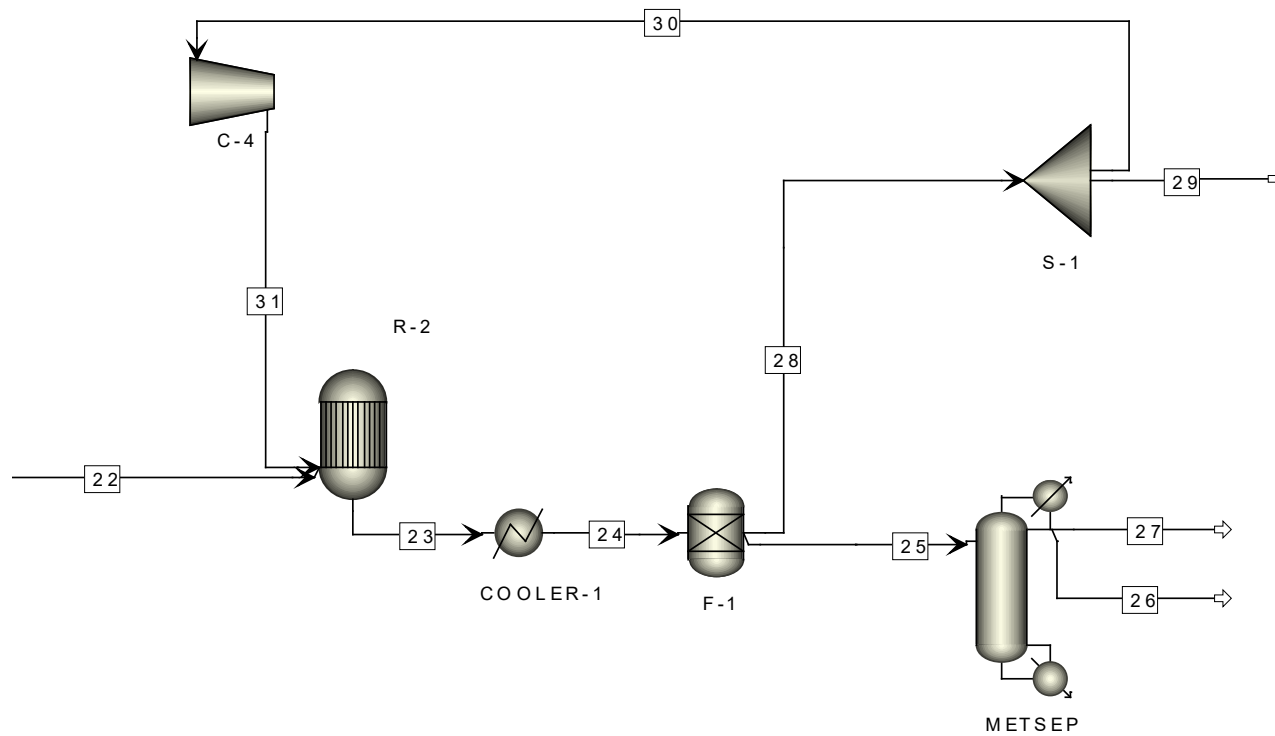
360

361

**Fig 7.** Aspen Plus® flowsheet simulation of the syngas cleaning.

362

363



364

365

**Fig 8.** Aspen Plus® flowsheet simulation of methanol synthesis.

366  
367  
368

**Table 12.** Inputs and outputs of the methanol production plant from natural gas (FU= 1kg of methanol) (from software Aspen Plus®).

<b>NATURAL GAS REFORMING</b>		
<i>Inputs</i>		
Natural gas	m <sup>3</sup>	1.17E+00
Water	kg	1.05E+00
Energy	MJ	1.78E+01
<i>Outputs</i>		
Syngas	kg	1.98E+00
<b>SYNGAS CLEANING</b>		
<i>Inputs</i>		
Syngas	kg	1.98E+00
Energy	MJ	7.50E+00
<i>Outputs</i>		
Clean syngas	kg	1.22E+00
CH <sub>4</sub>	kg	1.81E-01
CO <sub>2</sub>	kg	5.10E-01
<b>Emissions to air</b>		
Heat	MJ	1.36E+01
CO	kg	1.85E-02
CO <sub>2</sub>	kg	5.46E-02
<b>METHANOL SYNTHESIS</b>		
<i>Inputs</i>		
Clean syngas	kg	1.22E+00
Energy	MJ	7.62E+00
<i>Outputs</i>		
CH <sub>3</sub> OH	kg	1.00E+00
<b>Emissions to air</b>		
Heat	MJ	1.55E+01
CO	kg	3.86E-02
CO <sub>2</sub>	kg	1.11E-03
CH <sub>4</sub>	kg	2.01E-03
H <sub>2</sub>	kg	1.76E-01

369

370 2.3 Impact assessment methodology

371 In this study, the LCA was performed using the software SimaPro v9, which follows  
372 the recommendations of the ISO14040 series (Standardization 2006a; Standardization  
373 2006b). Moreover, the methodology used in this research was the Mid-point ReCiPe 2016

374 due to the fact that the midpoint results have less statistical uncertainty than the endpoint  
375 results (Goedkoop et al. 2008). In this sense, the results for the environmental  
376 performance associated with the methanol production process from natural gas and olive  
377 pomace were calculated for eleven midpoint indicators: global warming potential (GWP);  
378 ozone depletion potential (ODP); photochemical oxidation formation potential - humans  
379 (HOFP); photochemical oxidation formation potential - ecosystems (EOFP); terrestrial  
380 acidification potential (TAP); freshwater eutrophication potential (FEP); marine  
381 eutrophication potential (MEP); human toxicity potential - cancer (HTPc); human toxicity  
382 potential - non-cancer (HTPnc), fossil fuel potential (FFP), water consumption potential  
383 (WCP).

### 384 **3. Results and discussion**

385 This section presents and discusses the main results obtained from the LCA.  
386 Therefore, subsection 3.1 focuses on environmental assessment of methanol production  
387 from olive pomace gasification and subsection 3.2 compares the global environmental  
388 impacts associated with the methanol production from olive pomace and from natural gas.

#### 389 3.1 Environmental assessment of the methanol production from olive pomace 390 valorization

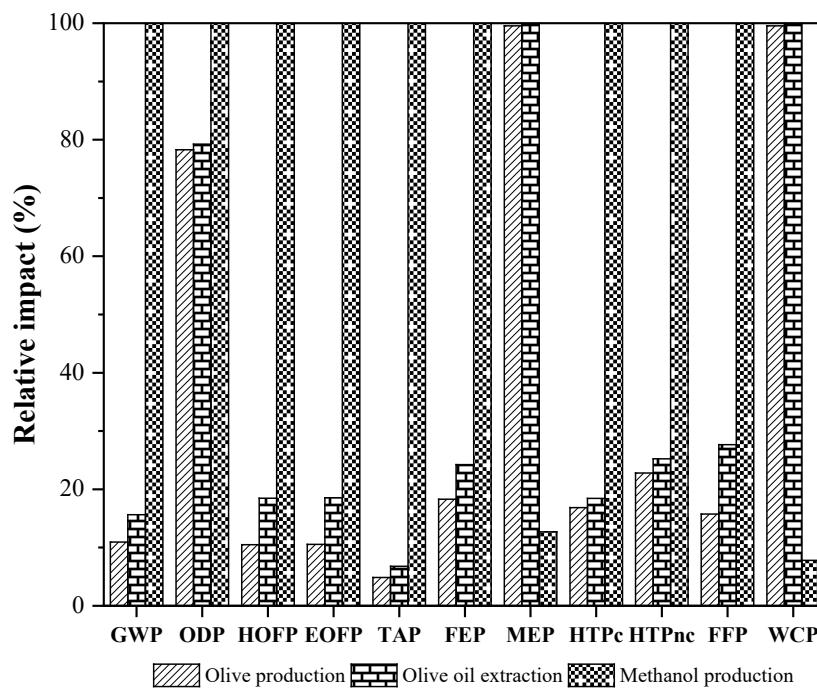
391 Fig 9 shows the assessed environmental performance of the methanol production  
392 from olive pomace valorization, considering all the evaluated stages (olives production,  
393 olive oil extraction and methanol production) at the mid-point level.

394 The results obtained showed that the methanol production exhibited the highest  
395 impact for the most of impact categories (except MEP and WCP), followed by olive oil  
396 extraction. According to Li et al., and Ai et al., (Ai et al. 2006; Li et al. 2018) who studied  
397 the LCA of methanol production from different routes, this fact can be attributed to the

398 high energy consumption and greenhouse emissions of the methanol production which  
399 are responsible for the impact values of almost all the selected categories. At this respect,  
400 the higher GHG emission ( $\text{CH}_4$  and  $\text{CO}_2$ ) observed for the methanol production in  
401 comparison with olive production and olive oil extraction stages (Tables 2, 4 and 8) could  
402 explain the higher GWP, HOFP and ODP impacts of the first one (Fig 9). Although  $\text{N}_2\text{O}$   
403 was emitted in olive production stage, which is about 300 times worse than  $\text{CO}_2$  in terms  
404 of the greenhouse effect, its existence in traces contributed to a small GWP in comparison  
405 with the larger  $\text{CO}_2$  emissions. In addition, the methanol production also presented the  
406 highest value in the HTPc and HTPnc categories, which includes all the direct toxic  
407 effects of human emissions (Parascanu et al. 2018b). This fact can be related to the ash  
408 generated and the emissions released into the air (Table 2). Moreover, the inorganic air  
409 pollutants, fertilizers and heavy metals (Cd, Pb, Hg, Zn) linked to the first stage can also  
410 cause toxic effects (Brentrup et al. 2004). The high  $\text{NH}_3$  emissions of methanol  
411 production (Oreggioni et al. 2017) directly affected the impact category of TAP. On the  
412 other hand, the FFP impact category was directly associated with the diesel and energy  
413 consumption (Tables 2, 4 and 8) and indirectly with the demand for natural gas, crude  
414 oil or coal required for the background processes (Parascanu et al. 2018b).

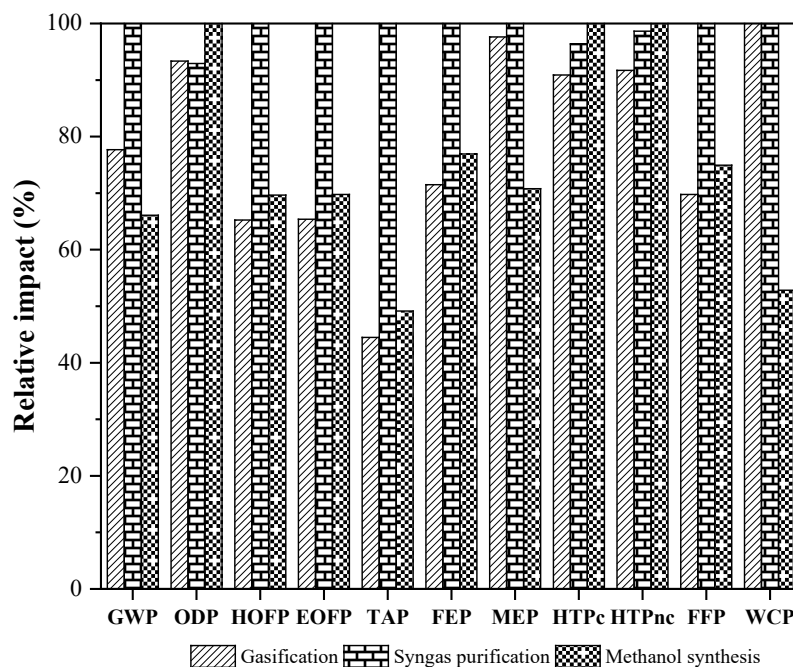
415 A difference trend was observed in the case of the MEP and WCP categories. In  
416 this case, the olive oil extraction and olive growing were more affected, mainly due to  
417 high values of wastewater generated (Tables 2 and 4) and the large amount of water  
418 required for olive production and olives washing in the olive oil mill for olive oil  
419 extraction in comparison to methanol production. In addition, the emissions of nitrogen  
420 and sulphur compounds ( $\text{NH}_3$ ,  $\text{NO}_x$ ,  $\text{NO}_3$  and  $\text{SO}_2$ ) during olive production stage (Table  
421 2) were indirectly responsible for the MEP impact (Goedkoop et al. 2009).

422 On the other hand, as aforementioned, the methanol production from olive pomace  
 423 can be divided in three main processes, gasification process, syngas purification and  
 424 methanol synthesis which were also evaluated environmentally. Fig 10 shows the impact  
 425 values at the mid-point level using the ReCiPe methodology for each process involved in  
 426 the methanol production from olive pomace. Syngas purification process was major  
 427 contributor in most of impact categories assesses since in this stage is where the most of  
 428 GHG obtained in the gasification process were released into the atmosphere and  
 429 moreover, the required energy was considerably higher. While the gasification process is  
 430 autothermal and, therefore, no energy input is required, the purification of the syngas was  
 431 carried out by means of pressure cycles, and due to the high number of adsorbers, requires  
 432 large amounts of energy that, compared to the energy required for methanol synthesis,  
 433 the last one can be considered negligible.



434  
 435 **Fig 9.** Comparison of the environmental impacts of olive production, olive oil  
 436 extraction and methanol production stages.  
 437

438 As commented before, the GWP, ODP and HOFP categories were directly  
 439 influenced for the CO<sub>2</sub> and CH<sub>4</sub> emissions. Syngas purification showed the highest value  
 440 for GWP and HOFP followed by gasification. In the case of gasification process, GWP  
 441 impact was associated with combustion chamber where a high amount of CO<sub>2</sub> was  
 442 released due to char combustion. While, the methanol synthesis had the least influence  
 443 on GWP category due to the waste stream were recycled, thus, the impact could be  
 444 associated with the background processes to obtain the required energy. However, this  
 445 last process showed the highest impact value for ODP, which could be associated with  
 446 the higher heat emissions of this stage (Table 8). Concerning MEP and WCP categories,  
 447 the high value obtained by gasification and purification process can also be attributed to  
 448 the high water consumption as agent gasifying and as refrigerant, respectively. Finally,  
 449 the HTPc and HTPnc categories presented similar values for three process, being  
 450 associated with the emissions to air.



451

452 **Fig 10.** Comparison of the environmental impacts of three processes involved in the

453

methanol production from olive pomace.



454

455

456

457 3.2 Comparison between methanol production from olive pomace and natural gas

458 Finally, the comparison between methanol production from olive pomace and  
459 natural gas, considering all stages was carried out, since it was the main aim of this work.

460 In this sense, Fig 11 displays for the two routes of methanol production the impact values  
461 at the mid-point level using the ReCiPe methodology. Table 13 shows the aggregated

462 according to ReCiPe Mid-point methodology for each methanol production route. As can  
463 be observed the olive pomace route presented at the mid-point level higher values for all

464 the impact categories than the natural gas one. The reason that olive pomace route leads  
465 to higher environmental contribution was probably due to the low methanol efficiency

466 (0.4 kg methanol and 1.2 kg methanol per kg of olive pomace and natural gas,  
467 respectively) and thus, high CO<sub>2</sub> emissions are produced per kg of methanol (FU) during

468 olive pomace route. But, it is also due to the environmental impacts of the previous stages  
469 (olive production and olive oil extraction in comparison to extraction and importation of

470 natural gas). It was identified that methanol yield of natural gas route was three times  
471 greater than olive pomace route. Therefore, the latter route presented outstanding

472 disadvantage when the data was normalized considering the FU. Moreover, it should be  
473 noted that the desulfurization process for natural gas route, which can be associated with

474 large amount of required energy, was not taken in account.

475 Regarding Table 13, both routes followed the same order of impact magnitude:

476

477 - Olive pomace to methanol: GWP > FFP > HTPnc > WCP > TAP > HTPc > HOFp  
478 = EOFp > FEP > MEP > ODP

479 - Natural gas to methanol: GWP > FFP > HTPnc > WCP > TAP > HTPc = HOFP  
480 = EOFP > FEP > MEP > ODP

481

482 The GWP category had the highest value compared to the other impact categories  
483 for both routes which was caused mainly by GHG emissions, the energy consumption  
484 and heat releases (Tables 2, 4, 8 and 11). In the case of FFP category which was lower  
485 for olive pomace route was mostly affected by the energy necessary to carry out the  
486 processes (background system). Moreover, the extraction of natural gas was the main  
487 contributor to this category for methanol production from natural gas route. The HTPc  
488 and HTPnc was higher for the olive pomace route which can be related to the ash  
489 generated and the emissions released into the air (Table 2), but also fertilizers and heavy  
490 metals (Cd, Pb, Hg, Zn) related to olive production. For both routes, the HTPc presented  
491 lower values than HTPnc category.

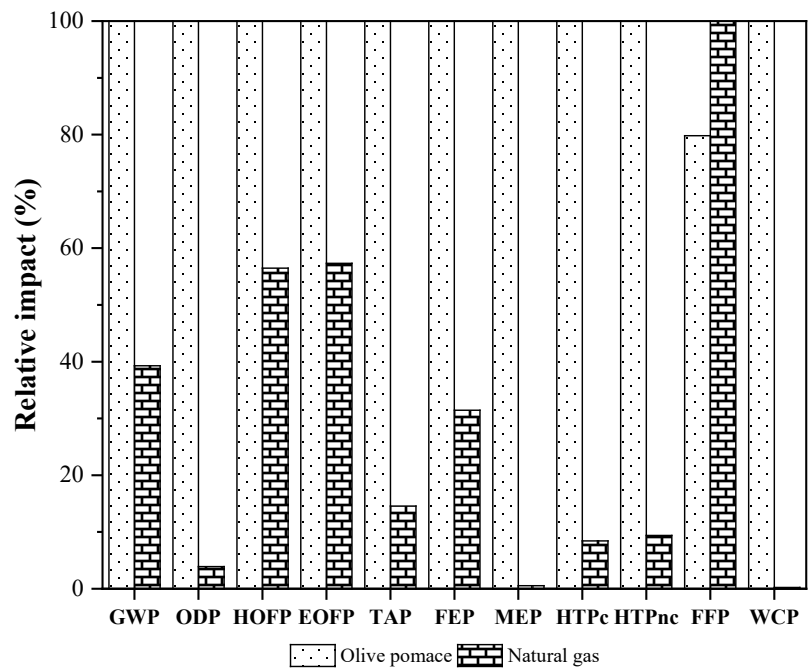
492 Furthermore, it can be seen in Fig 11 that WCP and MEP categories was  
493 insignificant in the methanol route, which was directly associated with the high amount  
494 of water consumed in olive pomace production and olive oil extraction. The high nitrogen  
495 and sulphur compound emissions of methanol production in olive pomace route  
496 (Oreggioni et al. 2017) directly affected the impact category of TAP.

497 Finally, the FEP, MEP and ODP were associated with the emissions that are  
498 produced during the energy production in the background system and the releases of  
499 nitrogen and sulphur compounds, being higher in the case of olive pomace route.

500 According to the results, methanol yield from olive pomace is disadvantaged due  
501 to its lower H<sub>2</sub> content in comparison with natural gas. In this regard, the co-gasification  
502 of olive pomace with other types of feedstock could produce synergistic effects during

503 the process, thereby further improving gas yield and quality. Thus, it could be a good  
 504 alternative to enhance the gasification process and, thus, the methanol yield, leading to  
 505 environmental improvements.

506



507

508 **Fig 11.** Environmental impact for methanol production form olive pomace and natural  
 509 gas, associated with the functional unit 1kg of methanol (ReCiPe mid-point).

510

511

512

513

514

515

516

517

518

519 **Table 13.** Impact assessment results of methanol production routes referred to FU  
 520 considering the cradle-gate approach.

<b>Impact</b>	<b>Unit</b>	<b>Olive pomace to methanol</b>	<b>Natural gas to methanol</b>
<b>GWP</b>	kg CO2 eq	19.50	8.74
<b>ODP</b>	kg CFC11 eq	6.82E-05	5.74E-06
<b>HOFP</b>	kg NOx eq	0.03	0.02
<b>EOFP</b>	kg Nox eq	0.03	0.02
<b>TAP</b>	kg SO2 eq	0.30	0.05
<b>FEP</b>	kg P eq	5.46E-04	2.44E-04
<b>MEP</b>	kg 1,4-DCB	2.63E-04	2.31E-05
<b>HTPc</b>	kg 1,4-DCB	0.13	0.02
<b>HTPnc</b>	kg 1,4-DCB	1.07	0.15
<b>FFP</b>	kg oil eq	3.21	4.03
<b>WCP</b>	m <sup>3</sup>	1.05	0.06

521

522 **4. Conclusions**

523 In this study a comparison between the methanol production from olive pomace  
 524 gasification and natural gas was carried out in terms of environmental impact. The results  
 525 of the analysis showed that, in the valorization of olive pomace, the stage of methanol  
 526 production had the most significant impact in almost all the categories studied, which was  
 527 associated with the higher energy required and greenhouse gas emissions produced from  
 528 last stage. Moreover, the environmental burdens of the main processes involved in the  
 529 methanol production from olive pomace; biomass gasification, syngas purification and  
 530 methanol synthesis were also individually evaluated. The results revealed that the greatest  
 531 environmental impact was obtained by syngas cleaning stage, since the most of GHG  
 532 releases were emitted in this process and, thus, the required energy was also higher.

533 Finally, the global environmental effect of both methanol production processes was  
 534 compared. The results of this comparison determined that the production of methanol

535 from the olive pomace had a greater environmental impact for all the categories studied  
536 except the one related to the shortage of fossil fuels. This fact was directly related to the  
537 technical performance of the processes and the functional unit deemed. Thus, due to  
538 methanol yield from olive pomace route was disadvantaged mainly by differences in  
539 composition, the co-gasification of olive pomace with others raw materials could be a  
540 good alternative to improve the methanol yield and compete with natural gas.

#### 541 **Acknowledgments**

542 The authors would like to acknowledge the Spanish government for their financial  
543 support (Grant No. FPU15/02653) and the “*Aceites Garcia de la Cruz*” olive oil mill.

544

#### 545 **Data availability**

546 All data generated and analysed during this study are included in this published article.

#### 547 **References**

548 Ai C, Ni W, Li Z (2006) Life cycle assessment of the coke oven gas utilization system  
549 Coal Conversion 29:25-31

550 Amigun B, Gorgens J, Knoetze H (2010) Biomethanol production from gasification of  
551 non-woody plant in South Africa: Optimum scale and economic performance  
552 Energy policy 38:312-322

553 Brentrup F, Küsters J, Kuhlmann H, Lammel J (2004) Environmental impact assessment  
554 of agricultural production systems using the life cycle assessment methodology:  
555 I. Theoretical concept of a LCA method tailored to crop production European  
556 Journal of Agronomy 20:247-264

557 Commission E (2019) Short-term outlook for EU agricultural markets.  
558 [https://ec.europa.eu/info/food-farming-fisheries/farming/facts-and-](https://ec.europa.eu/info/food-farming-fisheries/farming/facts-and-figures/markets/outlook/short-term_en)  
559 [figures/markets/outlook/short-term\\_en](https://ec.europa.eu/info/food-farming-fisheries/farming/facts-and-figures/markets/outlook/short-term_en). Accessed 17/03/2020

560 Demirbas A (2007) Progress and recent trends in biofuels Progress in Energy and  
561 Combustion Science 33:1-18 doi:<https://doi.org/10.1016/j.peccs.2006.06.001>

562 Duman AK, Özgen GÖ, Üçtuğ FG (2020) Environmental life cycle assessment of olive  
563 pomace utilization in Turkey Sustainable Production and Consumption 22:126-  
564 137

565 Gao D, Qiu X, Zhang Y, Liu P (2018) Life cycle analysis of coal based methanol-to-  
566 olefins processes in China Computers & Chemical Engineering 109:112-118

567 Goedkoop M, De Schryver A, Oele M, Durksz S, de Roest D (2008) Introduction to LCA  
568 with SimaPro 7 PRé Consultants, The Netherlands

569 Goedkoop M, Heijungs R, Huijbregts M, De Schryver A, Struijs J, Van Zelm R (2009)  
570 ReCiPe 2008 A life cycle impact assessment method which comprises harmonised  
571 category indicators at the midpoint and the endpoint level 1:1-126

572 Gutiérrez Ortiz FJ, Serrera A, Galera S, Ollero P (2013) Methanol synthesis from syngas  
573 obtained by supercritical water reforming of glycerol Fuel 105:739-751  
574 doi:<https://doi.org/10.1016/j.fuel.2012.09.073>

575 ICIS Independent Commodity Information Service. 2020

576 IPCC Special report: global warming of 1.5°C Accessed 20/05/2019

577 Kajaste R, Hurme M, Oinas P (2018) Methanol-Managing greenhouse gas emissions in  
578 the production chain by optimizing the resource base AIMS Energy 6:1074-1102

579 Lerner A, Brear MJ, Lacey JS, Gordon RL, Webley PA (2018) Life cycle analysis (LCA)  
580 of low emission methanol and di-methyl ether (DME) derived from natural gas  
581 Fuel 220:871-878

582 Li J, Ma X, Liu H, Zhang X (2018) Life cycle assessment and economic analysis of  
583 methanol production from coke oven gas compared with coal and natural gas  
584 routes Journal of Cleaner Production 185:299-308  
585 Methanex <https://www.methanex.com>. 20/06/2019

586 Mikulska A (2020) Gazprom and Russian Natural Gas Policy in the First Two Decades  
587 of the 21st Century Orbis 64:403-420  
588 doi:<https://doi.org/10.1016/j.orbis.2020.05.004>

589 Oreggioni GD et al. (2017) Environmental assessment of biomass gasification combined  
590 heat and power plants with absorptive and adsorptive carbon capture units in  
591 Norway International Journal of Greenhouse Gas Control 57:162-172

592 Pala LPR, Wang Q, Kolb G, Hessel V (2017) Steam gasification of biomass with  
593 subsequent syngas adjustment using shift reaction for syngas production: An  
594 Aspen Plus model Renewable Energy 101:484-492  
595 doi:<https://doi.org/10.1016/j.renene.2016.08.069>

596 Parascanu M, Gamero MP, Sánchez P, Soreanu G, Valverde J, Sanchez-Silva L (2018a)  
597 Life cycle assessment of olive pomace valorisation through pyrolysis Renewable  
598 Energy 122:589-601

599 Parascanu MM, Sánchez P, Soreanu G, Valverde JL, Sanchez-Silva L (2018b)  
600 Environmental assessment of olive pomace valorization through two different  
601 thermochemical processes for energy production Journal of Cleaner Production  
602 186:771-781 doi:<https://doi.org/10.1016/j.jclepro.2018.03.169>

603 PCR (2014) Product group: UN CPC 21537: Virgin Olive Oil and its Fractions. Version  
604 2.01.

605 Pérez-Fortes M, Schöneberger JC, Boulamanti A, Tzimas E (2016) Methanol synthesis  
606 using captured CO<sub>2</sub> as raw material: Techno-economic and environmental

607 assessment Applied Energy 161:718-732

608 doi:<https://doi.org/10.1016/j.apenergy.2015.07.067>

609 Puig-Gamero M, Argudo-Santamaria J, Valverde JL, Sánchez P, Sanchez-Silva L (2018)

610 Three integrated process simulation using aspen plus®: Pine gasification, syngas

611 cleaning and methanol synthesis Energy Conversion and Management 177:416-

612 427 doi:<https://doi.org/10.1016/j.enconman.2018.09.088>

613 Puig-Gamero M, Trapero JR, Sánchez P, Sanchez- Silva L (2020) Is methanol synthesis

614 from co-gasification of olive pomace and petcoke economically feasible? Fuel

615 278:118284 doi:<https://doi.org/10.1016/j.fuel.2020.118284>

616 Standardization IOF (2006a) Environmental Management: Life Cycle Assessment;

617 Principles and Framework. vol 2006. ISO,

618 Standardization IOF (2006b) Environmental management: Life cycle assessment;

619 requirements and guidelines. ISO Geneva,

620 Trop P, Anicic B, Goricanec D (2014) Production of methanol from a mixture of torrefied

621 biomass and coal Energy 77:125-132

622 doi:<https://doi.org/10.1016/j.energy.2014.05.045>

623 Uceda-Rodríguez M, López-García AB, Moreno-Maroto JM, Cobo-Ceacero CJ, Cotes-

624 Palomino MT, García CM (2020) Evaluation of the environmental benefits

625 associated with the addition of olive pomace in the manufacture of lightweight

626 aggregates Materials 13 doi:10.3390/ma13102351

627 **Ethical Approval**

628 Not applicable

629 **Consent to Participate**

630 Not applicable



631 **Consent to Publish**

632 Not applicable

633 **Authors Contributions**

634 **María Puig-Gamero:** Conceptualization, Data curation, Formal analysis, Investigation,  
635 Writing. **María Magdalena Parascanu:** Conceptualization, Data curation, Formal  
636 analysis, Investigation. **Luz Sanchez- Silva:** Conceptualization, Funding acquisition,  
637 Investigation, Project administration. **Paula Sánchez:** Conceptualization, Formal  
638 analysis, Funding acquisition, Project administration.

639 **Competing interests**

640 The authors declare that there is no conflict of interest.