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Aquaponics Using Saline Groundwater:

Effect of adding microelements to fish wastewater on the growth of Swiss chard (*Beta vulgaris* L. spp. *cicla*)

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1 **Abstract**

2 Saline soil and saline groundwater reduce agricultural productivity on drylands. We are
3 developing a new aquaponics system to improve food productivity on such lands while
4 effectively utilizing saline groundwater. In this study, cultivation of Swiss chard (*Beta vulgaris*
5 *L. spp. cicla* cv. Seiyou Shirokuki) was carried out using fish wastewater with a high salt
6 concentration (1150 mg L⁻¹ NaCl). The levels of microelements (e.g., Fe, Mn, Zn, and Cu) in
7 the fish wastewater were very low, so we added microelements at 100% (W100), 50% (W50),
8 25% (W25), and 0% (W0) of the levels in the standard hydroponics solution to the fish
9 wastewater and investigated the effects on growth of Swiss chard. At the first harvest, yields
10 in all wastewater treatments were as high or higher than in the control. At the second harvest,
11 yields in W100, W50, and W25 were not significantly different from the control, while in W0
12 the yield was significantly lower and chlorosis was evident. At the third harvest, the yield in
13 all wastewater treatments was less than in the control, and chlorosis symptoms were observed
14 in W25 and W0. Since leaf Mn and Zn concentrations in W25 and W0 had decreased to below
15 the critical values for those microelements, Mn and Zn deficiency might have contributed to
16 the observed chlorosis and yield loss. For the cultivation of Swiss chard with fish wastewater,
17 sufficient yield (i.e., comparable to or better than the control) without chlorosis was obtained
18 when microelements were added at 50% of the level of the control solution. In addition, since
19 sufficient yield was obtained even in W0 at the first harvest, it is suggested that longer-term
20 cultivation and higher yield could be achieved by applying 50% microelements after the first
21 harvest.

22

23 **Keywords:** Salinity, Wastewater, Yield, Microelement, Productivity, Dryland

24 **1. Introduction**

25 According to the United Nations, the world's population is estimated to reach
26 approximately nine billion in the year 2050 (Pitman and Läuchli, 2002). In order to secure the
27 food consumption needs of this increased population, food production urgently needs to be
28 developed and expanded. Although the technology required for increasing food production has
29 progressed remarkably, the production volume per land area has almost plateaued. In addition,
30 it is reported that the dryland area will increase by 23% by the year 2100, and will account for
31 56% of the total land area of the earth (Huang et al., 2016). It is believed that 78% of this
32 expansion of dryland will occur in developing countries, which could exacerbate rising poverty
33 levels and land degradation in dryland areas (Huang et al., 2016). As a result, there is a need
34 for innovative and sustainable food production technologies in drylands that occupy a large
35 land area. Agriculture in drylands can be problematic, as those places often do not have
36 adequate water resources. In an effort to increase food production in those conditions, farmers
37 often unintentionally use excessive amounts of saline water to irrigate their land, and/or overly
38 fertilize low fertility soils, both of which causes salt accumulation in the soil and results in
39 declining crop yields (Saysel and Barlas, 2001). On the other hand, drylands are rich in solar
40 radiation, which promotes photosynthesis in plants and increases agricultural productivity.
41 Thus, it is possible that drylands can become highly productive for agriculture if water
42 resources are secured and irrigation is carried out properly.

43 In the 1960s, aquaponics (a combined system of aquatic animal production and
44 hydroponic plant growth) was introduced as a new food production system to achieve effective
45 water use (Enduta et al., 2011). In aquaponics, freshwater fish are cultured in an aquaculture
46 system, then the plant is hydroponically cultivated using fish wastewater from that system,
47 which contains various nutrients derived from fish feed and manure. This removes nutrients
48 from the fish wastewater while simultaneously purifying the water.

49 In general, aquaponics uses low saline or non-saline water, so it is possible to cultivate
50 high cash crops hydroponically, including glycophytes (salt sensitive) such as lettuce and herbs.
51 Those plants, however, cannot be grown in conditions where the salt concentration of
52 groundwater is high, such as drylands. Therefore, we have been aiming to develop a new non-
53 circulate “aquaculture–hydroponics–field cultivation type” aquaponic system that uses saline
54 groundwater and is thus suitable for dryland conditions.

55 In a standard aquaponic system, when the salt concentration in the wastewater is
56 increased by sodium (Na) contained in fish feed and manure, the yield of hydroponic crops is
57 reduced, and a large amount of water must be discarded (Hambrey consulting, 2013). An
58 advantage of aquaculture is that it does not consume water (water will be lost only by
59 evaporation). In our aquaponics system, Na and other nutrients from fish wastewater are
60 removed by hydroponically grown plants, and that purified water can be used effectively for
61 conventional field farming without wasting it. There have been many reports of aquaponics
62 focusing on aquaculture, but few studies have focused on the nutrient dynamics of plants
63 (Goddek et al., 2015). The aquaponics technology we aim to develop is diverse, such as
64 aquaculture, hydroponics, and field cultivation. In this study, we aim to contribute to the
65 development of aquaponics technology in drylands, as well as broadening the knowledge on
66 the nutrient dynamics of hydroponic culture, especially in aquaponics using saline water.

67 The electrical conductivity of groundwater in Los Planes, Baja California Sur,
68 Mexico, which is categorized as dryland, is approximately 3 to 5 dS m⁻¹ and its sodium (Na)
69 concentration is as high as 690 to 1150 mg L⁻¹ (unpublished results). In areas with such high
70 sodium levels, it is difficult to cultivate agricultural plants, most of which are categorized as
71 glycophytes. Instead, salt-tolerant or salt-loving plants are often cultivated. Salt-loving plants
72 are those whose growth is promoted by Na in the medium (Yamada et al., 2016a, 2016b) and
73 include Swiss chard (*Beta vulgaris* L. spp. *cicla* cv. Seiyou Shirokuki), table beet (*Beta vulgaris*

74 *L. spp. vulgaris* cv. Detroit Dark Red), and dwarf glasswort (*Salicornia bigelovii* Torr.). These
75 plant species not only utilize Na for growth but also absorb and accumulate high concentrations
76 of Na in the plant body (Yamada et al., 2016a). Therefore, even high-Na groundwater could
77 potentially produce high yield in an aquaponic system if salt-loving plants are cultivated. In
78 this study, we elucidated the cultivation conditions for growing Swiss chard in our aquaponic
79 system. Swiss chard is a plant of the Amaranthaceae family and is widely produced and
80 consumed throughout the world, including in Mexico. It has been found that sodium can be
81 used for the growth of Swiss chard as an alternative to potassium (Kaburagi et al., 2015;
82 Yamada et al, 2016b). Therefore, a medium that is lacking potassium but is rich in sodium,
83 which is provided by fish wastewater with high saline content, may be also suitable for the
84 cultivation of Swiss chard.

85 When fish wastewater is used directly for hydroponics in an aquaponic system, the
86 mineral concentrations in the water must be sufficient for plant cultivation. Fish wastewater
87 contains sufficient N for plant growth, which is derived from fish feed and manure. Our
88 previous studies have confirmed that 280 mg L⁻¹ of NO₃-N does not inhibit the growth of
89 Swiss chard (unpublished results). On the other hand, fish wastewater is often lacking essential
90 microelements for plants (Graber and Junge, 2009; Roosta and Hamidpour, 2011). Analysis of
91 mineral concentrations in the fish wastewater used in this study revealed the similar pattern.
92 The concentration of phosphorous (P) was 1/2 that of the standard hydroponics nutrient
93 solution, zinc (Zn) was 1/5, manganese (Mn) was 1/20, and iron (Fe) was about 1/90 (Table 1).

94 **Table 1** Comparison of mineral concentration in the standard hydroponics nutrient solution
 95 and the fish wastewater

Mineral	Standard nutrient solution*	Fish wastewater
(mg L ⁻¹)		
NO ₃ -N	42.0	241.9
P	12.4	6.8
K	78.0	89.5
Ca	40.8	359.1
Mg	48.6	40.8
Fe	2.0	0.02
Mn	0.5	0.02
Zn	0.1	0.02
Cu	0.01	0.04

96 *Kaburagi et al. (2014)

97

98 Fe is involved in the formation of chlorophyll, which is necessary for light absorption
 99 during photosynthesis (Jeong and Guerinot, 2009). Mn is involved in processes such as
 100 photosynthesis, protein synthesis, and redox reactions to remove oxygen radicals (Ducic and
 101 Polle, 2005; Shao et al., 2017). However, these microelements are not readily translocated from
 102 older leaves to other growing plant parts. Therefore, if the medium is lacking microelements,
 103 deficiency symptoms appear in photosynthetically active leaves and plant growth is inhibited
 104 (Hell and Stephan, 2003).

105 In this study, we added different concentrations of microelements to saline fish
 106 wastewater and investigated the effects on the growth of Swiss chard. We also evaluated the
 107 food safety (NO₃ content) of Swiss chard leaves grown using fish wastewater.

108 2. Materials and Methods

109 2.1. Aquaculture

110 Tilapia (*Oreochromis niloticus*) was stocked and reared in a 1500-L tank designed
111 as a closed recirculating aquaculture system. Tilapia is tolerant to salinity, low oxygen, and
112 high nitrate nitrogen (NO₃-N) in the water and can adapt to a wide range of pH and temperature
113 conditions (Stickney, 2017; Goddek et al., 2015). In this study, groundwater was used for the
114 fish farming and adjusted to 1150 mg L⁻¹ of sodium chloride (NaCl) to make the salt
115 concentration similar to that of the groundwater in Baja California, Mexico. Feeding was
116 carried out twice a day. Water was collected every 7 days to analyze NO₃-N concentration.
117 The wastewater was used for hydroponic cultivation when the NO₃-N concentration reached
118 approximately 250 mg L⁻¹. This research was conducted in a greenhouse at Tottori University
119 from March to May 2016.

120

121 2.2. Plant cultivation

122 Seeds of Swiss chard (*Beta vulgaris* L. spp. *cicla* cv. Seiyou Shirokuki) were sown
123 in vermiculite. When the first true leaves appeared, seedlings were transplanted and grown
124 hydroponically in 30-L plastic containers (6 plants per container). Plants were subjected to five
125 different treatments based on either groundwater or fish wastewater (Table 2).

126

127 **Table 2** Mineral concentrations in the treatment solutions at the start of cultivation

	Macroelements (mg L ⁻¹)					Microelements (mg L ⁻¹)			
	NO ₃ -N	P	K	Ca	Mg	Fe	Mn	Zn	Cu
Control	44.1	12.7	80.1	55.5	45.8	1.78	0.44	0.12	0.01
W100	238.0	6.7	93.9	344.8	40.0	1.62	0.46	0.12	0.04
W50	260.6	6.8	93.6	350.8	40.9	0.86	0.25	0.07	0.05
W25	241.5	6.4	93.1	356.5	41.0	0.42	0.13	0.05	0.04
W0	248.0	6.9	100.9	357.5	41.1	0.02	0.02	0.02	0.03

128 Groundwater was used for the control solution and adjusted to 1150 mg L⁻¹ of NaCl as well as the fish
129 wastewater

130

131 The treatments were Control (groundwater with standard hydroponics nutrient solution
132 (Kaburagi et al., 2014)); W100 (fish wastewater with 100% of the control microelement level);
133 W50 (fish wastewater with 50% of the control microelement level), W25 (fish wastewater with
134 25% of the control microelement level), and W0 (fish wastewater without micronutrient
135 supplementation). Each treatment comprised four replicates. The first harvest was carried out
136 27 days after transplanting. We harvested leaves that were about 20 cm in length (market-
137 transaction size) and left the smaller, younger leaves for the next harvest. The second harvest
138 was carried out 10 days after the first harvest, and the third harvest was 11 days after the second
139 harvest. Leaves from six plants in each bat were weighed to determine fresh weight (FW) and
140 then oven-dried at 70°C for 48 h. The nitrogen concentration in the control solution was 56 mg
141 L⁻¹, which was 1/5 of the wastewater. This was renewed weekly so that nitrogen deficiency
142 would not become a growth-inhibiting factor throughout the cultivation period. The pH of the
143 control solution was adjusted to 5.5. The wastewater treatment solutions were not renewed
144 during the cultivation period, and the pH was adjusted to 5.5 only at the transplanting.

145

146 2.3. *Chlorophyll*

147 At the time of harvesting, the chlorophyll content of the largest developed leaves of
148 each individual plant was measured with a portable chlorophyll meter (SPAD-502, Konica
149 Minolta, Tokyo, Japan).

150

151 2.4. *Determination of inorganic ion concentrations*

152 Dried leaves from each harvest were ground, decomposed by sulfuric acid (H₂SO₄)
153 and hydrogen peroxide (H₂O₂) at 200°C, and diluted with deionized water. Sodium (Na),
154 potassium (K), calcium (Ca), magnesium (Mg), iron (Fe), manganese (Mn), zinc (Zn), copper
155 (Cu) concentrations in decomposed leaves and in each treatment solution were quantitatively

156 analyzed at the time of each harvest by inductively coupled plasma atomic emission
157 spectroscopy (ICP-AES, Spectro Ciros CCD, Spectro, Kleve, Germany).

158

159 *2.5. Determination of NO₃-N and P concentrations*

160 Nitrate (NO₃) and P were extracted from dried leaves by water at 80°C for 2 h. The
161 Cataldo method (Cataldo et al., 1975) was used to measure the NO₃ concentrations in leaves
162 and treatment solutions, and then converted to the concentration on FW basis. The P
163 concentrations in leaves and treatment solutions were determined at the time of each harvest
164 by the molybdenum blue method (Murphy and Riley, 1962) using a spectrophotometer (V-
165 630BIO, JASCO, Tokyo, Japan).

166

167 *2.6. Statistical analysis*

168 Statistical analyses for each treatment at each harvest time were carried out using
169 version 7.0b of the GraphPad Prism software program (GraphPad Software, Inc., La Jolla, CA,
170 USA). All data were presented as means ± standard error (SE). Significant differences ($p <$
171 0.05) were assessed by one-way ANOVA, followed by Bonferroni's multiple-comparison test
172 if the ANOVA revealed a significant difference.

173

3. Results

174

3.1. Plant growth and total yield

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Leaf FW at the first harvest was significantly higher in all of the wastewater treatments

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than in the control (Fig. 1a). Even at the second harvest, FWs of W100, W50, and W25 were

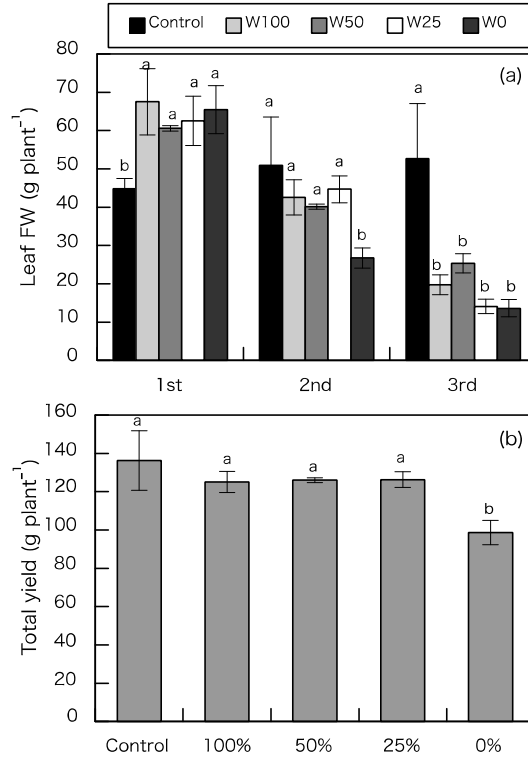


Fig. 1 (a) Leaf fresh weight (FW) from five different treatments at three harvest times. (b) FW-based total yield across all three harvests. Bars labeled with different letters at each harvest time (a) and each treatment (b) differ significantly (one-way ANOVA, $P < 0.05$; Bonferroni's multiple-comparison test). Values are mean \pm SE ($n = 4$).

177

not significantly different from the control, but FW of W0 was significantly lower than for the

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other treatments. At the third harvest, the leaf FWs in all wastewater treatments was less than

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50% of the control value. On the other hand, the total yield across all three harvests did not

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differ significantly between the control and wastewater treatments except for W0, which was

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significantly lower (70% of the control value; Fig. 1b).

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3.2. Chlorophyll in leaves (SPAD value)

184 The SPAD values of the leaves in W100 and W50 were the same as or higher than
185 the control value at all three harvest times (Fig. 2).

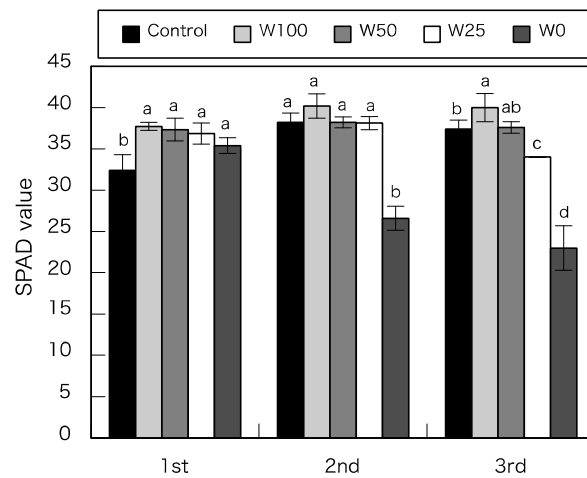


Fig. 2 SPAD values of leaves in five different treatments at three harvest times. Bars labeled with different letters at each harvest time differ significantly (one-way ANOVA, $P < 0.05$; Bonferroni's multiple-comparison test). Values are mean \pm SE ($n = 4$).

186
187 In W25, the SPAD value equaled or exceeded the control value until the third harvest,
188 when it was significantly lower than that of the control and the plants showed chlorosis
189 symptoms. In W0, the SPAD value was not significantly different from the control at the first
190 harvest. However, it was significantly lower (70% of the control level) at the second harvest,
191 and 60% of control at the third harvest, both of which exhibited chlorosis.

192

193 3.3. Fe, Mn, Zn, and Cu concentrations in leaves

194 The Fe concentration in leaves was not significantly different between treatments at
195 the first and second harvests. At the third harvest, Fe concentration in leaves of W50, W25 and
196 W0 were significantly lower than the control. However, the Fe concentration was constant in
197 all treatments throughout the three harvest times, even in the W25 and W0 which exhibited
198 chlorosis (Fig. 3).

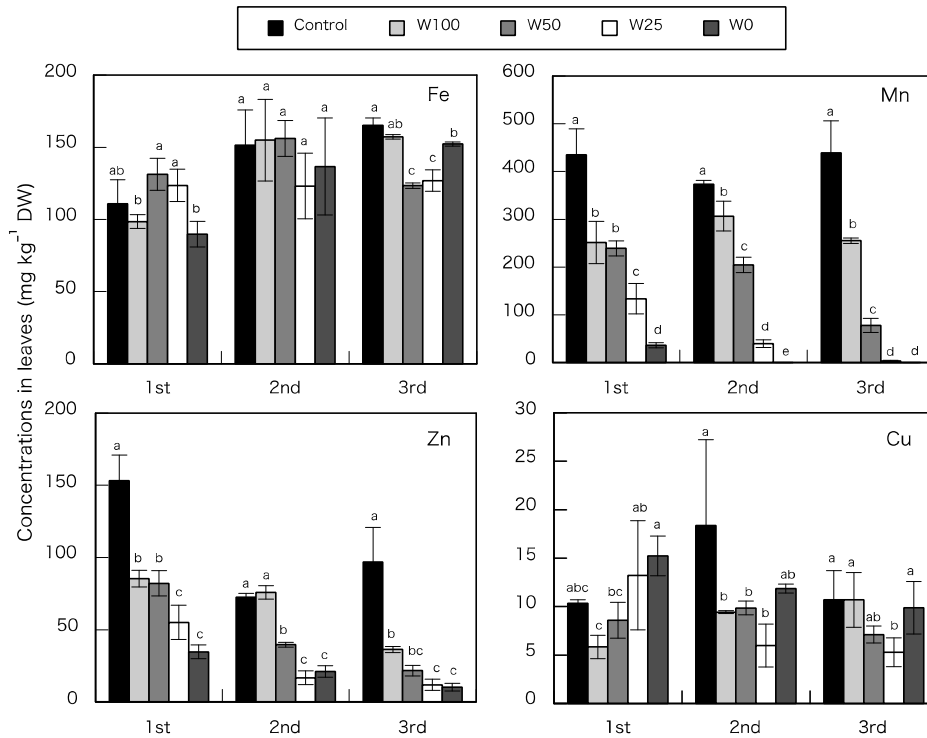


Fig. 3 Fe, Mn, Zn, and Cu concentrations in leaves (DW basis) from five different treatments at three harvest times. Bars labeled with different letters at each harvest time differ significantly (one-way ANOVA, $P < 0.05$; Bonferroni's multiple-comparison test). Values are mean \pm SE ($n = 4$).

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At all harvest times, the Mn concentration in leaves from the W50, W25, and W0 decreased as the treatment micronutrient concentration decreased, while the W100 maintained constant concentration. In W0, the Mn concentration in leaves decreased markedly to 1 mg kg⁻¹ DW and the chlorosis symptoms were exhibited by the second harvest; the same pattern was observed in W25 at the third harvest. As with Mn, the Zn concentration in leaves decreased as the treatment micronutrient concentration decreased. The Zn concentration in leaves of all wastewater treatments was significantly lower than the control at the third harvest. The Cu concentration in leaves did not differ markedly among harvests or treatments. The Cu concentration in leaves remained constant even in W0 where the decrease in growth was significant.

3.4. NO₃, P, K, Ca, Mg, and Na concentrations in leaves

212 The NO₃ concentration in leaves (FW basis) was not significantly different between
213 treatments at the first or second harvest times. At the third harvest, NO₃ was significantly lower
214 in W0 than in W100 or W50 (Fig. 4).

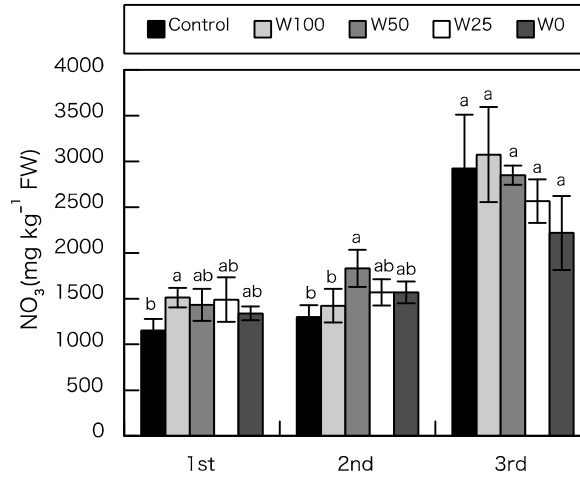


Fig. 4 NO₃ concentrations in leaves (FW basis) from five different treatments at three harvest times. Bars labeled with different letters at each harvest time differ significantly (one-way ANOVA, $P < 0.05$; Bonfferoni's multiple-comparison test). Values are mean \pm SE ($n = 4$).

215
216 The P concentration in leaves from all wastewater treatments was significantly lower
217 than the control at all three harvest times (Fig. 5). However, the P concentration in leaves from
218 all wastewater treatments at the second and third harvests significantly decreased to 1/2 and
219 1/3 of the control levels, respectively.

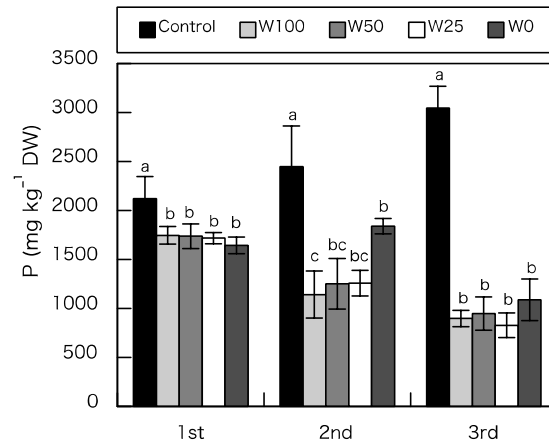


Fig. 5 P concentrations in leaves (DW basis) from five different treatments at three harvest times. Bars labeled with different letters at each harvest time differ significantly (one-way ANOVA, $P < 0.05$; Bonfferoni's multiple-comparison test). Values are mean \pm SE ($n = 4$).

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221 Leaf concentrations of K, Ca, Mg, and Na are not mentioned since the trends were

222 unclear and not significant.

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4. Discussion

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As a water source for aquaponics, fish wastewater contains adequate amounts of $\text{NO}_3\text{-N}$ derived from fish feed and manure. The fish wastewater used in this study contained 250 mg L^{-1} of $\text{NO}_3\text{-N}$ at the start of hydroponic cultivation, which is approximately 6 times the concentration in the standard hydroponics solution (Table 2). Nitrogen is an essential element for plant growth, and NO_3 is readily absorbed as a counterion of cations such as K, Ca, and Mg. NO_3 taken up into the plant is synthesized into amino acids and proteins (Tischner, 2000). In this study, at the first harvest, all wastewater treatments showed growth exceeding that of the control (Fig. 1a), likely due to the high levels of $\text{NO}_3\text{-N}$. At the third harvest, the FW decreased significantly in all wastewater treatments compared to the control, while the total yields of W100, W50, and W25 across the three harvest times were not significantly different from the control (Fig. 1b). Together, these results demonstrate that the optimum amount of microelement supplementation in fish wastewater is 50% of the control level, which was sufficient to prevent chlorosis and to obtain the same yield as in the control. Furthermore, because yields at the first harvest exceeded the control level, even in W0, we speculate that long-term cultivation with high yield and good plant color could be achieved by the addition of 50% microelements after the first harvest.

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In general, the requirement for Fe is very high among all the microelements, and it is essential to chlorophyll biosynthesis. Therefore when Fe is deficient, chlorosis symptoms occur in young leaves. In this study, the initial Fe concentration in fish wastewater (W0) was about 1/90 of that in the control (Table 2). Interestingly, however, the Fe concentration in Swiss chard leaves was not significantly different between treatments at any harvest time, and was sufficiently maintained even in leaves from W25 and W0, which showed chlorotic symptoms at later harvest times (Fig. 3). This result suggested that Swiss chard has either a high efficiency

248 of Fe absorption or a low requirement for Fe. Similarly, there were no significant differences
249 in Cu concentration in leaves between treatments at each harvest date since according to Table
250 2, there were no substantial differences in Cu concentration among the treatments.

251 On the other hand, at each harvest time, the Mn and Zn concentrations in leaves
252 remarkably decreased as treatment micronutrient concentration decreased (Fig. 3). The Mn
253 concentration in healthy plant leaves is dozens to hundreds of milligrams per kilogram dry
254 matter, and it is reported that deficiency occurs when the leaf Mn concentration is 10–20 mg
255 kg^{-1} or less (Mengel and Kirkby, 2001). Most Mn contained in the chloroplasts is associated
256 with PSII (Anderson et al., 1964). Ohki (1981) reported that net photosynthesis and chlorophyll
257 content were significantly decreased when Mn was deficient. Also, a clear relationship between
258 the intensity of Mn deficiency symptoms and Mn concentration was found in sugar beet, which
259 is in the same family (Amaranthaceae) as Swiss chard (Farley and Draycott, 1973). In this
260 study, in W0 (at first harvest) and W25 (at second harvest), the SPAD value and the leaf FW
261 did not differ from the control when the leaf Mn concentration was only about 40 mg kg^{-1} . At
262 the third harvest, on the other hand, chlorosis symptoms were observed and the leaf FW was
263 greatly reduced in W0 and W25 when the leaf Mn concentration was 10 mg kg^{-1} or less. Zn
264 plays important roles in plant metabolism such as maintaining the conformation (structure) of
265 various enzymes (Barker and Pilbeam, 2015). Other authors suggest a relationship between Zn
266 and auxin, a plant hormone promoting elongation, which might explain the suppression of plant
267 growth observed when Zn is deficient (Salisbury and Ross, 1992). In general, Zn deficiency in
268 plant leaves appears at 15 mg kg^{-1} or less, which is considered as the critical value (Barker and
269 Pilbeam, 2015; Jones, 2012). In this study, the leaf Zn concentration was less than 15 mg kg^{-1}
270 in W25 and W0 at the third harvest, when the decrease in FW was remarkable. From these
271 results, we conclude that deficiency of Mn and Zn limited the growth of Swiss chard in fish
272 wastewater. Thus, the addition of microelements, especially Mn and Zn, is necessary to

273 maintain yield during long-term cultivation. However, confirming the effects of specific
274 microelements (Mn and Zn) would require additional experiments (e.g., manipulating each
275 microelement separately).

276 P plays an important role as a major constituent of phosphate compounds such as
277 DNA and biomembranes and is involved in energy generation, enzyme activation and
278 inactivation, etc. (Vance et al., 2003). An adequate amount of P in plant leaves is generally
279 about 2000 to 4000 mg kg⁻¹ on a dry-matter basis, and P deficiency occurs at 1000 to 2000 mg
280 kg⁻¹ or less (Mengel and Kirkby, 2001). The P concentration in the fish wastewater used in this
281 study was about 1/2 that of the control solution (Table 2). Thus, the P concentration in the
282 leaves of all wastewater treatments reached the deficiency level at the third harvest (Fig. 5).
283 However, with the exception of W0, the P concentration in the leaves of all wastewater
284 treatments was above the deficiency level at the first two harvests, and the leaf FW was
285 comparable to the control (Fig. 1a and Fig. 5). From these results, we considered it is possible
286 to cultivate Swiss chard for at least two harvests using fish wastewater without adding P, which
287 indicates that fish wastewater can be a valuable source of P nutrition. However, the P
288 concentration in all wastewater treatment solutions at the third harvest was below the detection
289 limit (Table 1S) and the P concentration in leaves was also at the deficiency level (Fig. 5). In
290 the case of longer-term cultivation using fish wastewater, addition of P may be necessary.

291 Due to the depletion of phosphate ore in recent years, P fertilizers are becoming
292 increasingly expensive (Abelson, 1999; Vance et al., 2003). In addition, fertilizers generally
293 account for between five to ten percent of the production costs in conventional hydroponic
294 cultivation, and the fossil fuels used for fertilizer production may have a negative
295 environmental impact (Hochmuth and Hanlon 2010). Therefore, reducing both the cost of
296 fertilizer as well as the amount of fertilizer required for crop cultivation using wastewater in
297 aquaponic systems is necessary for the future of sustainable food production.

298 The Swiss chard leaves was not negatively impacted by the concentrations of K, Ca
299 and Mg in the fish wastewater at each of the harvest times (data was not shown and mentioned
300 in the results section as the trends were not significant). These nutrients were not additionally
301 applied. This suggests that Swiss chard is suitable for cultivation not only in saline water, but
302 also in less nutrient rich mediums, such as fish wastewater.

303 The initial NO₃-N concentration in fish wastewater was about 6 times that of the
304 control solution (Table 2). When nitrogen absorbed by the plant body becomes excessive, the
305 leaves become succulent and dark green in color, which can increase disease and pest damage
306 (Jones, 2012). Also, if excess NO₃ is taken into the human body, it is quickly metabolized to
307 NO₂, which oxidizes the iron of hemoglobin in the blood and may hinder oxygen transport
308 (IARC, 2010). In our study, the SPAD values (an indicator of leaf color) in the wastewater
309 treatments were no more than 15% higher than the control in all of the wastewater treatments
310 except W0 (Fig. 2). In addition, the NO₃ concentration in leaves (FW basis) was lower than the
311 upper limit (3500 mg kg⁻¹ FW) of the regulated value for the spinach, which is in the same
312 family (Amaranthaceae) as Swiss chard, reported by the UK's Food Standard Agency (Red
313 Tractor Farm Assurance, 2016) (Fig. 4). However, the NO₃ concentration was markedly higher
314 at the third harvest time than at either of the first two, and some values are approaching the
315 3500 mg kg⁻¹ FW limit. This suggests that longer growth times might need to be monitored for
316 further increases in leaf NO₃.

317

318 **5. Conclusion**

319 We found that sufficient yield of Swiss chard (i.e., comparable to or better than the
320 control) could be obtained by adding 50% of the microelement level of the control solution
321 when utilizing fish wastewater for aquaponics. Also, since sufficient yield of Swiss chard was
322 obtained in all treatments at the first harvest, even with no addition of microelements, it appears

323 that long-term cultivation and high yield could be obtained by supplying 50% microelements
324 after the first harvest. Although the depletion of phosphorus ore has become a problem in recent
325 years, the amount of P originally contained in fish wastewater was enough to obtain sufficient
326 yield of Swiss chard for at least two harvests. However, to enable long-term cultivation, the
327 effectiveness of supplying P as well as microelements, especially Mn and Zn, should be
328 examined in the future. The NO₃ concentration in Swiss chard leaves was below the regulated
329 value for spinach in the UK even though the concentration of NO₃-N in fish wastewater was
330 about 6 times that in the control solution.

331

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