

1 Basic protocol for the analysis of thylakoid membrane protein complexes by blue native gel
2 electrophoresis

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16 KEYWORDS

17 BN-PAGE, 2D gel electrophoresis, thylakoid protein complexes, native protein complexes,
18 thylakoid membrane, photosystem

19 20 SUMMARY

21 Basic protocol for the elucidation of plant thylakoid protein complex organization and
22 composition with blue native polyacrylamide gel electrophoresis (BN-PAGE) and 2D-SDS-PAGE is
23 described. The protocol is optimized for *Arabidopsis thaliana*, but with minor modifications can
24 be used for other plant species.

25 26 ABSTRACT

27 Photosynthetic electron transfer chain (ETC) converts solar energy to chemical energy in the form
28 of NADPH and ATP. Four large protein complexes embedded in the thylakoid membrane harvest
29 solar energy to drive electrons from water to NADP⁺ via two photosystems, and use the created
30 proton gradient for production of ATP. Photosystem (PS)II, PSI, cytochrome *b₆f* (Cyt *b₆f*) and
31 ATPase are all multiprotein complexes with distinct orientation and dynamics in the thylakoid
32 membrane. Valuable information about the composition and interactions of the protein
33 complexes in the thylakoid membrane can be obtained by solubilizing the complexes from the
34 membrane integrity by mild detergents followed by native gel electrophoretic separation of the
35 complexes. Blue native polyacrylamide gel electrophoresis (BN-PAGE) is a analytical method used
36 for the separation of protein complexes in their native and functional form. The method can be
37 used for protein complex purification for more detailed structural analysis, but it also provides a
38 tool to dissect the dynamic interactions between the protein complexes. The method was
39 developed for the analysis of mitochondrial respiratory protein complexes, but has since
40 been optimized and improved for the dissection of the thylakoid protein complexes. Here we
41 provide a detailed up-to-date protocol for analysis of labile photosynthetic protein complexes

42 and their interactions in *Arabidopsis thaliana*.

43 INTRODUCTION

44 Large multisubunit protein complexes photosystem (PS) I and PSII, Cyt b_6f and ATPase coordinate
45 the production of NADPH and ATP in photosynthetic light reactions. In higher plant chloroplasts,
46 the complexes are located in the thylakoid membrane, which is a structurally heterogeneous
47 membrane structure, comprising appressed grana and non-appressed stroma thylakoids. Blue
48 native polyacrylamide gel electrophoresis (BN-PAGE) is an extensively used method in the
49 analysis of large multisubunit protein complexes in their native and biologically active form. The
50 method was established for the dissection of mitochondrial membrane protein complexes¹, but
51 been customized for the separation of thylakoid protein complexes^{2,3}. The method is suitable (i)
52 for the purification of individual thylakoid protein complexes for structural analysis, (ii) for
53 determining native interactions between protein complexes and (iii) for the analysis of overall
54 organization of the protein complexes upon changing environmental cues.

55
56 Prior to the separation, protein complexes are isolated from the membrane with carefully chosen
57 nonionic detergents, which are generally mild and preserve the native structure of the protein
58 complexes. Detergents contain hydrophobic and hydrophilic sites and form stable micelles above
59 a certain concentration, called a critical micellar concentration (CMC). Increasing the detergent
60 concentration above the CMC results in disruption of the lipid-lipid interactions and in the
61 solubilization of protein complexes. The choice of detergent depends on the stability of the
62 protein complex of interest and on the solubilization capacity of the detergent. Routinely used
63 detergents include α/β -dodecyl-maltoside and digitonin. Following the solubilization of protein
64 complexes in their native state, insoluble material is removed by centrifugation. In higher plants
65 the thylakoid membrane is highly heterogenic in structure and some detergents (e.g. digitonin)
66 selectively solubilize only a specific fraction of the membrane³. Therefore, to characterize the
67 protein complex organization or the interactions between the protein complexes, it is crucial to
68 always determine the solubilization capacity of the chosen detergent by determining the
69 chlorophyll content and the chlorophyll a/b ratio of supernatant to assess the yield and the
70 represented thylakoid (sub)domain, respectively, of the solubilized fraction. The chlorophyll a/b
71 ratio in intact thylakoids of growth-light acclimated plants is typically around 3, whereas the chl
72 a/b value of thylakoid fractions enriched either with grana or stroma thylakoids falls below (~2.5)
73 or exceeds (~4.5) the value of the total thylakoids, respectively.

74
75 To provide negative charge to the protein complexes, Coomassie brilliant blue (CBB, G-250) dye
76 is added to the solubilized sample. Due to the charge shift, protein complexes migrate towards
77 the anode and are separated on acrylamide (AA) gradient according to their molecular mass and
78 shape. Effective and high resolution separation is achieved by using a linear acrylamide
79 concentration gradient. During the electrophoresis, the protein complexes migrate towards the
80 anode until they reach their size-dependent pore-size limit. The pore-size of polyacrylamide gel
81 depends on (i) the total acrylamide/bis-acrylamide concentration (T) and (ii) on the cross-linker
82 bis-acrylamide monomer concentration (C) relative to the total monomers⁴. After the separation
83 with BN-PAGE, the protein complexes can be further subdivided into their individual protein
84 subunits by second-dimension (2D)-SDS-PAGE. Here we describe a detailed protocol for the
85 analysis of thylakoid membrane protein complexes by BN-PAGE/2D-SDS-PAGE.

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87
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PROTOCOL

89 1. Preparing BN gel¹⁻³

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91 1. Set up the gel caster with 8 x10 cm plates (rectangular glass and notched alumina plate)
92 according to manufacturer's instructions using 0.75 mm spacers.

93

94 2. Place a gradient mixer on a stir plate and connect it with the peristaltic pump by a tubing.
95 Attach a syringe needle to the other end of the tubing and place the needle between the
96 glass and aluminum plate. Place magnetic stirrer to the "heavy" (H)-chamber.

97

98 3. Prepare the 3.5% (v/v) and 12.5% (v/v) acrylamide (AA) solutions in 15 ml conical
99 centrifuge tubes for the separation gel gradient (see recipes in Table 1). To prevent
100 untimely polymerization, keep the centrifuge tubes on ice while preparing the solutions.

101

102 CAUTION: Acrylamide is neurotoxic and carcinogenic, wear protective clothes and gloves.

103

104 4. Add 5% APS and TEMED right before pipeting the solutions to the gradient mixer. Pipet
105 the 12.5% solution to the H-chamber. Remove air bubbles from the channel connecting
106 the "light" (L) and H-chamber by opening the valve connecting the two chambers
107 allowing solution to enter to the L-chamber. Close the valve and pipet the traces of
108 solution back to H-chamber. Finally, pipet the 3.5% solution to the L- chamber.

109

110 5. Switch on the magnetic stirrer (the speed of the stir is not critical, but it should ensure
111 proper mixing of the the heavy and light solutions), open the valves and switch on the
112 peristaltic pump. Allow the gel solutions to flow between the glass and aluminum plate,
113 the flowrate should be roughly 0.5 mL/min. The needle must be above the liquid all the
114 time, it can be attached to the upper part of the glass plate with a tape.

115

116 6. When the H- and L-chambers have emptied, fill them with ultrapurewater and allow it to
117 gently overlay the gel surface. The gel polymerization takes around 1-2 hours at RT.

118

119 7. Prepare the 3% acrylamide solution (see recipe in Table 1) for the stacking gel at RT. Pipet
120 the stacking gel on top of the polymerized separation gel (before casting the stacking gel,
121 remove the water overlaying the gel surface) and place a sample gel comb between the
122 glass and aluminum plate avoiding air bubbles. Allow to polymerize 30-60 min at RT.
123 Remove the comb gently under ultrapure water. Store the gel at +4 °C.

124

125 Pause point. The gel can be stored at +4 °C few days. The gel should be kept in moist
126 condition, the gel surface should not dry.

127 3. Thylakoid solubilization ¹⁻³

128

129 Note: All steps should be performed under very dim light. Keep samples and buffers on ice.

130

131 1. Dilute isolated thylakoids with ice-cold 25BTH20G buffer to a final chlorophyll
132 concentration of 1 mg/mL. For 2D-BN-SDS-PAGE analysis roughly 4-8 µg of chlorophyll/
133 sample is suitable.

134

135 Note: The thylakoids used in the experiments must be isolated from fresh leaves (for
136 protocol of thylakoid isolation, see ³)

137

138 2. Add an equal volume of detergent buffer, i.e. 2% β-DM (w/v) or 2% digitonin (w/v). Mix
139 the detergent to the thylakoid sample gently with the pipet tip and avoid making air
140 bubbles. The final concentration of the detergent is 1% and that of the thylakoids
141 0.5mg/mL. Solubilize the thylakoids for 2 min on ice (β-DM) or 10 minutes at RT with
142 continuous gentle mixing on a rocker/shaker (digitonin).

143

144 Note: Digitonin and β-DM are generally used for the solubilization of thylakoid protein
145 complexes. If other non-ionic detergents are used, the detergent concentration and the
146 solubilization time must be first optimized. Usually the detergent concentration range
147 varies from 0.5%-5% (w/v).

148

149 CAUTION: Digitonin is toxic, wear protective clothes and gloves

150

151 3. Remove the insolubilized material by centrifugation at 18 000 x g for 20 min, at +4 °C.

152

153 4. Transfer the supernatant to a new 1.5 mL tube and add 1/10 (v/v) of CBB buffer to the
154 sample.

155

156 Note: When the overall composition of the thylakoid membrane protein complexes is
157 examined, determining the yield and the represented thylakoid domain of solubilized
158 fraction is recommended. To determine the yield of the solubilized material, take 5 µL of
159 the supernatant to a new tube (before adding CBB) and measure the Chl content and Chl
160 a/b ratio according to ⁵.

161

162 4. BN-PAGE ¹⁻³

163

164 1. Assemble the gel to a vertical electrophoresis system (e.g. Hoefer SE 250). Pour blue
165 cathode buffer (see recipe) to the upper buffer chamber and anode buffer to the lower
166 buffer chamber.

167

168 2. Load thylakoid sample (e.g. 5 µg of chlorophyll) into the wells.

169

170 3. Start the electrophoresis and gradually increase the voltage: 75 V for 30 min, 100 V for 30
171 min, 125 V for 30 min, 150 V for 1 h and 175 V until the complexes have been separated
172 completely. The gel running should be done at +4°C either at cold room or adjusted with
173 a cooling system.

174
175 Note: Change the blue cathode buffer to a clear cathode buffer when the sample front
176 has migrated about one third of the gel.

177
178 4. After the electrophoretic run, scan the gel for image archiving.

179 5. 2D-SDS-PAGE

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181 1. Assemble vertical electrophoresis system (gel size 16 x 20 cm). Use 1 mm spacers.

182

183 2. Prepare standard SDS gel (12% acrylamide, 6M Urea, see recipe in Table 2) with a 2D-
184 comb (single large well for the strip and one standard well for molecular weight marker).

185

186 3. Cut the lane from BN-gel and place it in a (5 mL) tube. Add 2 mL of Laemmli buffer
187 (containing 5 % β -mercaptoethanol) and incubate the strip for 45 min with gentle shaking
188 at RT.

189

190 4. Place the lane with e.g. a spacer on top of the gel avoiding air bubbles.

191

192 5. Pipet 5 μ l of molecular weight marker on a narrow piece of filter paper and place the
193 paper to the standard well.

194

195 6. To seal the BN-gel strip and the marker paper, pour 0.5% agarose (in running buffer) on
196 top of the gel strip and allow to solidify.

197

198 7. Perform electrophoresis according to standard protocols. After the electrophoretic run,
199 visualize the proteins with e.g. Sypro Ruby stain or silver staining according to ⁶.

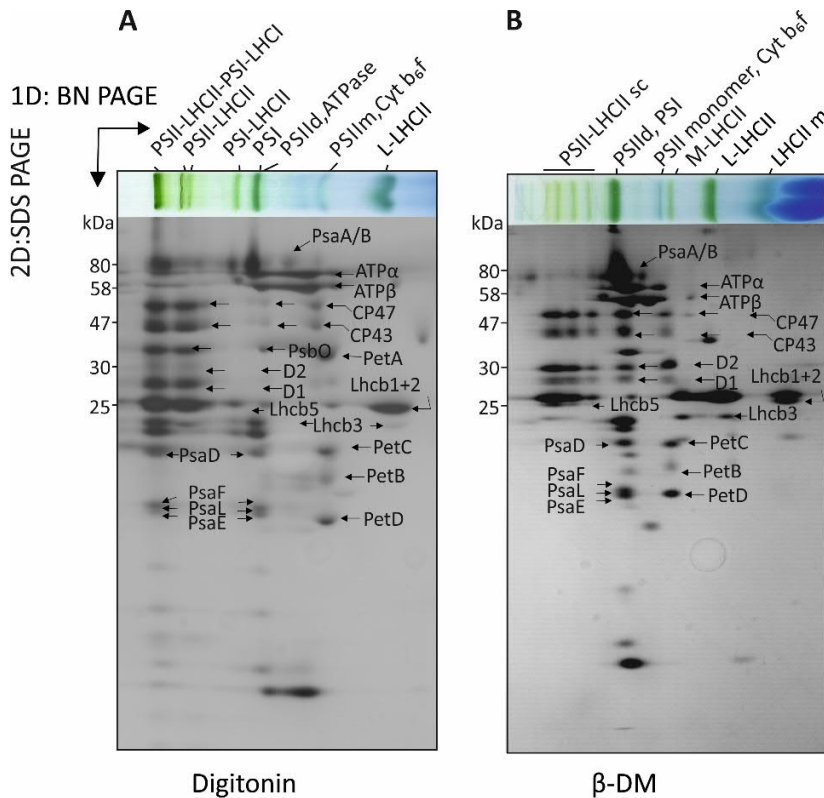
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201 REPRESENTATIVE RESULTS

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203 The representative results of 2D-BN/SDS-PAGE in Figure 1 shows the detailed protein subunit
204 composition of digitonin and β -DM-solubilized thylakoids. The protein complex pattern of
205 digitonin solubilized thylakoids contain the PSII-LHCII-PSI megacomplex, two large PSII-LHCII
206 supercomplexes (sc), PSI-LHCII supercomplex, PSI monomer (m), PSII m/Cyt b_6f , loosely bound
207 (L)-LHCII trimer (Figure 1A). The slightly stronger detergent, β -DM, solubilizes the entire thylakoid
208 membrane, but is unable to preserve weak interactions between protein complexes. Thylakoid
209 solubilization with β -DM typically produces four PSII-LHCII supercomplexes (with different
210 amount of LHCII antenna attached), PSII dimer (d) and PSI m, ATPase, PSII m and Cyt b_6f , M-LHCII,
211 L-LHCII and LHCII monomer (Figure 1B). Unlike β -DM, digitonin produces only minor amount of
212 LHCII monomer. The protein complex pattern may differ upon different light conditions and in

213 different mutant lines, since the protein complex interactions are dynamic and dependent on i.e.
 214 protein phosphorylation.
 215



216
 217 Figure 1. A two-dimensional BN-PAGE/SDS-PAGE of Arabidopsis thylakoid. Thylakoid protein
 218 complexes solubilized with (A) 1% digitonin and (B) 1% β -DM and separated first by 1D-BN-PAGE
 219 (the lanes on top) and subsequently on 2D-SDS-PAGE to demonstrate the individual protein
 220 composition of each complex. Due to the incubation of BN-strips with denaturing Laemmli buffer,
 221 the protein subunits of each complex (in the BN strip) dissociate and are separated in a vertical
 222 line during the 2D-SDS-PAGE. The protein identification is based on mass spectrometry analysis
 223 presented in references ^{7,8}.

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237 TABLES OF MATERIALS AND BUFFER RECIPES

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239 Table of Materials

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Name of Material/ Equipment	Company	Catalog Number	Comments/Description
6-aminocaproic acid (ACA)	Sigma-Aldrich	A2504	
BisTris	Sigma-Aldrich	B4429	
Sucrose	Sigma-Aldrich	S0389	
Acrylamide (AA)	Sigma-Aldrich	A9099	Caution: Neurotoxic!
n-dodecyl- β -D-maltoside	Sigma-Aldrich	D4641	
Tricine	Sigma-Aldrich	T0377	
Tris	Sigma-Aldrich	T1503	
SDS	VWR	442444H	
Urea	VWR	28877.292	
Glycerol	J.T. Baker	7044	
Sodium Fluoride (NaF)	J.T. Baker	3688	
EDTA disodium salt	J.T. Baker	1073	
Digitonin	Calbiochem	300410	Caution: Toxic!
Pefabloc SC	Roche	11585916001	
Serva Coomassie Blue G	Serva	35050	
β -mercaptoethanol	Bio-Rad	1610710	
APS (Ammonium persulfate)	Bio-Rad	161-0700	
TEMED (Tetramethylethylenediamine)	Bio-Rad	1610801	
(N,N'-Methylene)-Bis-Acrylamide	Omnipur	2610	
Glycine	Fisher	G0800	
Prestained Protein Marker, Broad Range (7-175 kDa)	New England Biolabs	P7708	
Falcon, Conical Centrifuge Tubes 15 ml	Corning	352093	
Dual gel caster with 10 x 8 cm plates	Hofer	SE215	
Gradient maker SG5	Hofer		
0.75 mm T-spacers	Hofer	SE2119T-2-.75	
Sample gel comb, 0.75 mm	Hofer	SE211A-10-.75	
Mighty Small SE250 vertical electrophoresis system	Hofer	SE250	
IPC-pump	Ismatec		
Power supply, PowerPac HV	Bio-Rad	164-5097	
Centrifuge	Eppendorf	5424R	
Rocker-Shaker	Biosan	BS-010130-AAI	
PROTEAN II xi Cell	Bio-Rad	1651813	

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244 Table 1. Buffers and solutions for native gel electrophoresis

Buffer	Content	Comments
3.5% (T) Acrylamide (AA) BN Separation gel	48% AA, 1.5% bis-AA: 148 μ L 3xGel Buffer: 700 μ L 75% (w/v) glycerol: 140 μ L Ultrapure H ₂ O: 1092 μ L 5% APS: 15 μ L TEMED 3 μ L	NOTE: Acrylamide is neurotoxic. Add APS and TEMED immediately before use. The recipe is sufficient for casting one small BN gel.
12.5% (T) Acrylamide (AA) BN Separation gel	48% AA, 1.5% bis-AA: 530 μ L 3xGel Buffer: 700 μ L 75% (w/v) glycerol: 560 μ L Ultrapure H ₂ O: 290 μ L 5% APS: 11 μ L TEMED 2 μ L	NOTE: Acrylamide is neurotoxic. Add APS and TEMED immediately before use. The recipe is sufficient for casting one small BN gel.
3 % (T) Acrylamide (AA) BN Stacking gel	20% AA, 5% bis-AA: 180 μ L 3xGel Buffer: 500 μ L Ultrapure H ₂ O: 800 μ L 5% APS: 30 μ L TEMED 3 μ L	NOTE: Acrylamide is neurotoxic. Add APS and TEMED immediately before use. The recipe is sufficient for one small BN gel.
3x Gel buffer	1.5M ACA 150mM BisTris/HCl (pH 7.0)	Store at + 4 °C.
CBB buffer	100 mM BisTris/HCl (pH 7.0) 0.5 M ACA 30% (w/v) sucrose 50 mg/ml Serva Blue G	Store at + 4 °C
Anode buffer	50 mM BisTris/HCl (pH 7.0)	Store at + 4 °C. Buffer can be prepared as 10x stock solution
Cathode buffer	50 mM Tricine 15 mM BisTris 0.01% Serva Blue G	Store at + 4 °C. Buffer can be prepared as 10x stock solution, add the dye to the 1x solution
25BTH20G	25 mM BisTris/HCl (pH 7.0) 20% (w/v) glycerol 0.25 mg/ml Pefabloc (add freshly) 10 mM NaF (add freshly)	Buffer can be prepared as 2X stock solution (store at + 4 °C) , but add Pefabloc and NaF freshly
Detergent buffer	2% β -dodecyl maltoside/Digitonin (w/v) 25 mM BisTris/HCl (pH 7.0) 20% (w/v) glycerol and 0.25 mg/ml Pefabloc (add freshly from the stock solution) 10 mM NaF (add freshly)	Detergents can be prepared as 5-10% stock solutions (in water). If other detergents are used, the final detergent concentration has to be optimized.

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253 Table 2. Buffers for 2D-SDS-PAGE

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Buffer	Content	Comments
Laemmli buffer	138 mM Tris/HCL pH 6.8 6M Urea 22.2 % (v/v) glycerol 4.4 % SDS	Reference: 9
12% Acrylamide, 6M Urea SDS Separation gel	50% AA, 1,33% bis-AA: 10.5 mL 20% SDS: 0.7 mL 1.5 M Tris-HCl (pH 8.8): 8.05mL Urea: 12.6 g MQ-H ₂ O: 6.16 mL 10% APS: 200 μ L TEMED 28 μ L	NOTE: Acrylamide is neurotoxic. The recipe is suitable for casting one big SDS-gel.
6% Acrylamide, 6M Urea SDS Stacking gel	50% AA, 1.33% bis-AA: 1.2 mL 20% SDS: 0.2 mL 0.5 M Tris-HCl (pH 6.8): 2.5 mL Urea: 3.6 g MQ-H ₂ O: 3.45 mL 10% APS: 100 μ L TEMED 10 μ L	NOTE: Acrylamide is neurotoxic.
SDS Running buffer	19 mM Tris 2.5 mM Glycine 0.01 % SDS	

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258 DISCUSSION

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260 The photosynthetic energy conversion machinery is composed of large multisubunit protein
 261 complexes, which are embedded in the thylakoid membrane. This protocol describes a basic
 262 method for analysis of the plant thylakoid protein complexes from *Arabidopsis thaliana* with BN-
 263 PAGE combined with 2D-SDS-PAGE. The protocol is also suitable for the analysis of thylakoid
 264 protein complexes from tobacco and spinach thylakoids, but might require small adjustments.

265

266 For the solubilization of membrane protein complexes, nonionic detergents are commonly used
 267 for their ability to preserve the complexes in their native form. Here, two commonly used
 268 detergents β -DM and digitonin were applied. Dodecyl maltoside solubilizes individual protein
 269 complexes, whereas digitonin can be used for the analysis of larger protein complex assemblies
 270 ¹⁰. The bulky structured digitonin is unable to fit to the tightly appressed grana and therefore
 271 solubilizes only the non-appressed regions of the thylakoid membrane ^{3, 11}. It is therefore suited
 272 for the analysis of stroma thylakoids and grana margins. However, when digitonin is used
 273 together with aminocaproic acid (ACA), the combination solubilizes the entire thylakoid
 274 membrane, including also the appressed grana thylakoids ¹². Through an unknown mechanism,
 275 ACA allows digitonin to have an access to the partition gap between adjacent grana membrane
 276 layers. Importantly, digitonin preserves labile interactions between protein complexes and can
 277 therefore be used for the analysis of labile protein super and megacomplexes, which are most

278 abundant in the non-appressed thylakoid regions⁸. It must be noted that detergents always
279 interfere with some of the labile interactions between the protein complexes and therefore it is
280 not possible to isolate completely intact network of protein complexes. Some of the complexes
281 are degradation products that have been disconnected from larger protein complexes during the
282 solubilization and the electrophoresis. The quality of the BN-PAGE separation depends not only
283 of the sample preparation (protein complex solubilization), but also on the quality of the
284 thylakoid isolation, which must be done from fresh leaves. If special care is not taken, the PSII-
285 LHCII supercomplexes typically degrade during β -DM solubilization.

286
287 BN-PAGE maintains the integrity of the solubilized protein complexes. The separation capacity of
288 the BN gel depends on the acrylamide gradient, and the gradient should be optimized based on
289 the protein complexes of interest. The pore-size of the polyacrylamide gel can be modified by
290 changing the concentration gradient (total acrylamide concentration, T) or by adjusting the bis-
291 acrylamide concentration (C) relative to the total amount of acrylamide monomers⁴. The BN-PA
292 gel gradient used here is optimized and well suited for the analysis of large protein super- and
293 megacomplexes³. After protein complex separation with BN-PAGE, the composition, structure or
294 the spectroscopic properties of each individual protein complex band can be further analyzed.
295 For analysis of the subunit composition of the protein complexes, a denaturing 2D-SDS-PAGE is
296 described here. The subunits of each complex are separated in a vertical line and can be easily
297 identified. It has to be noted that several proteins may be present in a single spot and the subunits
298 in the same vertical line may belong to separate complexes co-migrating in BN-PAGE.

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302 (675006). The protocol is based on reference ³.

303 DISCLOSURES

304 The authors have nothing to disclose.

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