

Influence of Carbon Nature on the Photocatalytic Properties of Carbon/TiO₂ Hybrids for Elimination of Volatile Organic Compounds in Gas Phase

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Adsorbent-Photocatalyst Hybrids (APHs) based on graphitized carbons and TiO₂, for the degradation of volatile organic compounds in gas phase were studied. Commercial Starbons with different hydrophobicity and micro-mesoporosity, activated and graphitized mesoporous carbons were selected as adsorbents to prepare the carbon/TiO₂ composites. The influence of carbon nature and the way in which titania nanoparticles are incorporated on the carbonaceous materials for the photoactivity of the composites has been analyzed. The physicochemical properties of the raw materials and carbon/TiO₂ hybrids were analyzed by N₂ adsorption-desorption, XRD, UV-Vis spectroscopy, contact angle, SEM and HRTEM-EELS. The adsorption capacity of the photocatalysts towards the volatile organic compound (formaldehyde and trichloroethylene) at dynamic conditions was also evaluated. The results obtained in this work show the importance of the preparation route for the distribution of the TiO₂ nanoparticles on the porous carbon and the influence of the photocatalytic properties. The samples prepared via mechanical mixing exhibit higher photocatalytic activity than the ones prepared via incipient wet impregnation. This is because TiO₂ particles are located on the surface of the carbon material in the former case, whereas a fraction of TiO₂ nanoparticles are placed inside the pores of the carbon support when prepared via incipient impregnation, thus preventing TiO₂ excitation by UV-A photons. This work shows that the hydrophobicity of the carbon matrix is a key feature for the photoactivity of the composites. Hydrophobic carbon/TiO₂ composites showed total removal of the pollutant studied at 1100 mL min⁻¹ total gas flow in a one single pass photoreactor.

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