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### Impact of WTRs in a Saturated Bioretention System

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# Impact of WTRs in a Saturated Bioretention System

Troy Membrere and Cara Poor, Ph.D., P.E. Shiley School of Engineering.

# Objectives

To evaluate equilibrium chemistry in bioretention systems with water treatment residuals (WTRs) under saturated conditions.

# Background

- Bioretention is a common green stormwater system used to treat urban stormwater
- However, leaching occurs within the bioretention system soil mixture – mainly phosphorus
- The saturated zone helps remove nitrate (Palmer et al., 2013)
- There is very little research into how WTRs will impact chemistry particularly for saturated conditions

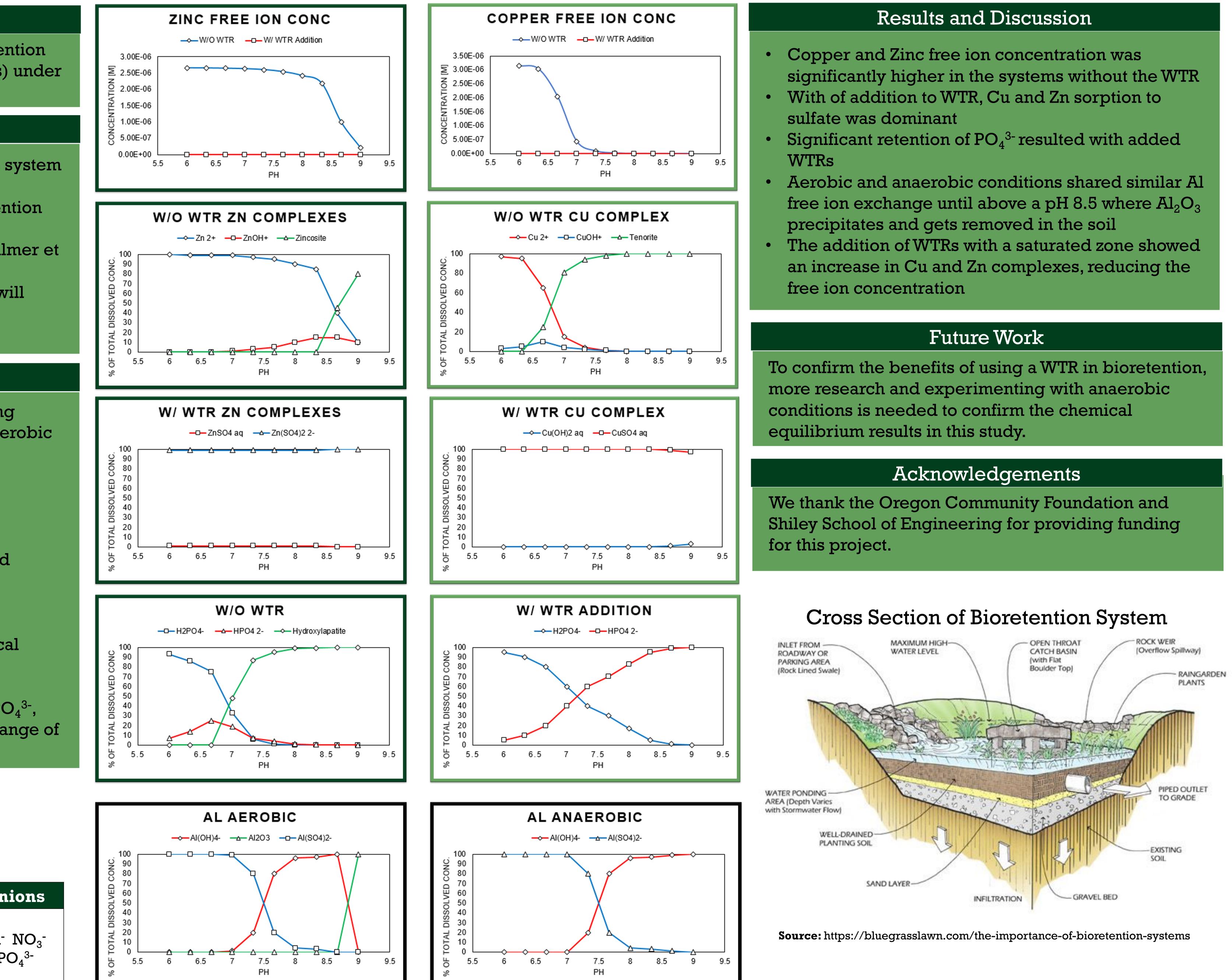
# **Design and Methods**

- MINEQL +, a chemical equilibrium modelling program, was used to create a total of four aerobic and anaerobic aquatic systems
  - Aerobic
  - 2. Anaerobic
  - 3. Aerobic + WTR Addition
  - . Anaerobic + WTR Addition
- The urban stormwater ionic composition and concentrations mimics runoff conditions comparable to the Pacific Northwest
- To simulate the addition of a WTR, alum was introduced into the system using the chemical reaction below

## $2Al^{3+} + 3(SO_4)^{2-} - > Al_2(SO_4)_3$ (s)

• Species and free ion formations of  $Zn, Cu, PO_4^{3-}$ , and Al were evaluated in the model over a range of pН

Aerobic	Anaerobic		
O <sub>2</sub> available Equilibrated with atm @ 21% O <sub>2</sub>	<ul> <li>Anoxic; W/O O<sub>2</sub></li> <li>Partial pressure of O<sub>2</sub> set to 0</li> <li>Oxidation reactions were</li> </ul>	-	
		Cations	An
	removed	$Ca^{2+}$ K <sup>+</sup> Mg <sup>2+</sup> Na <sup>+</sup> NH <sub>4</sub> <sup>+</sup> Zn <sup>2+</sup> Cu <sup>2+</sup> Pb <sup>2+</sup>	Cl- Po



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