

Detection of Catalase Activity by Polyacrylamide Gel Electrophoresis (PAGE) in Cell Extracts from *Pseudomonas aeruginosa*

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[Abstract] Bacteria in nature and as pathogens commonly face oxidative stress which causes damage to proteins, lipids and DNA. This damage is produced by the action of reactive oxygen species (ROS) such as hydrogen peroxide (H₂O₂), singlet oxygen, superoxide anion and hydroxyl radical. ROS are generated by antimicrobials, environmental factors (e.g., ultraviolet radiation, osmotic stress), aerobic respiration, and host phagocytes during infective processes. *Pseudomonas aeruginosa*, a versatile bacterium, is a prevalent opportunistic human pathogen which possesses several defense strategies against ROS. Among them, two catalases (KatA and KatB) have been well characterized by their role on the defense against multiple types of stress. In this protocol, KatA and KatB activities are detected by polyacrylamide gel electrophoresis (PAGE). It is also suggested that the detection of KatB is elusive.

Keywords: *Pseudomonas aeruginosa*, Catalase, PAGE, KatA, KatB, H₂O₂, Oxidative stress, ROS

[Background] *P. aeruginosa* is a ubiquitous bacterium that can be found in a free form in terrestrial and aquatic habitats or as an opportunistic human pathogen causing fatal infections in immunocompromised individuals, patients with skin damage or cystic fibrosis. To defend itself from ROS generated by its strong aerobic metabolism, host phagosomal vacuoles and environmental factors, this microorganism possesses several antioxidative strategies. Among them, two monofunctional catalases (KatA and KatB) are responsible for decomposing H₂O₂ to water and O₂. KatA is the main catalase and has unique characteristics: it is unusually stable and essential to H₂O₂ resistance, osmoprotection and virulence (Hassett *et al.*, 2000; Lee *et al.*, 2005). It has been suggested that the stability of KatA is one of the main factors for the high level activity under normal growth conditions, and for this reason, *katA* has been regarded as a constitutively expressed gene in *P. aeruginosa* (Heo *et al.*, 2010). However, it has been reported that KatA activity is induced in the stationary growth phase (up to 10-fold) and by increased levels of H₂O₂ (Brown *et al.*, 1995; Suh *et al.*, 1999; Heo *et al.*, 2010). Moreover, *katA* expression has been demonstrated to be modulated by the global regulator OxyR and Quorum Sensing, whose activation depends on increased levels of H₂O₂ and high cell density, respectively (Hassett *et al.*, 1999; Heo *et al.*, 2010). KatB is only detected in the presence of H₂O₂ or paraquat and is partially involved in resistance to oxidative stress (Brown *et al.*, 1995; Lee *et al.*, 2005).

Solar ultraviolet-A (UVA) radiation is one of the main environmental stress factors for *P. aeruginosa*. Given the oxidative nature of UVA damage, we studied the role of catalases in defense of this microorganism against radiation. We demonstrated that KatA is essential in the optimal response against lethal doses of UVA, both in planktonic cells and biofilms (Costa *et al.*, 2010; Pezzoni *et al.*,

2014). In addition, we reported that low doses of UVA increase KatA and KatB activity and that this regulation occurs at the transcriptional level (Pezzoni *et al.*, 2016). This phenomenon is relevant since it constitutes an adaptive mechanism that prevents cell damage by subsequent exposure to lethal doses of UVA, H₂O₂, or sodium hypochlorite (Pezzoni *et al.*, 2016).

In the course of our studies, it became necessary to do an in-depth analysis of catalase activity. The total catalase activity in cell extracts was quantified by following spectrophotometrically the decomposition of H₂O₂, according to Aebi (1984). However, this assay cannot distinguish between KatA and KatB activities. To analyze individual catalase activity, we implement the method proposed by Wayne and Díaz (1986). In brief, crude cell extracts are loaded onto non-denaturing polyacrylamide gels (PAGE), and both catalases are separated by their differential electrophoretic motility; colorless bands of catalase activity are revealed by incubation of the gel with H₂O₂ and subsequent addition of a ferric chloride-potassium ferricyanide solution. The principle of this method involves the reaction of H₂O₂ with potassium ferricyanide (III) by reducing it to ferrocyanide (II); the peroxide is oxidized to O₂. Ferric chloride reacts with ferrocyanide (II) to form an insoluble blue pigment. Because of the action of catalase on H₂O₂ decomposition, areas where this enzyme is active develop as clear bands in a blue gel (Patnaik *et al.*, 2013). Additional papers were consulted to fine-tune this technique (Brown *et al.*, 1995; Hassett *et al.*, 1999; Elkins *et al.*, 1999). The studies were performed with the prototypical *P. aeruginosa* strain PAO1 and isogenic derivatives PW8190 (*katA::IslacZ/hah*) and PW8769 (*katB::IslacZ/hah*) carrying mutations into *katA* and *katB*, respectively. Mutant strains devoid for KatA or KatB are useful to analyze the role of each enzyme in response to stress and as controls in PAGE catalase assays.

In this protocol, we describe how to detect individual catalase activity by PAGE using cell extracts from *P. aeruginosa*. Because of the particular characteristics of KatA (high abundance and stability), its detection does not present major difficulties. On the contrary, KatB detection is elusive, so that two changes were applied to the conventional technique: a protein extraction reagent was used instead of sonication to prepare the cell extracts, and the electrophoresis was performed at 4 °C. Based on these assays, it was concluded that KatB is an unstable enzyme, a fact that should be taken into account in quantitative or qualitative catalase assays under inducing (oxidative) conditions.

Materials and Reagents

1. Pipette tips
2. 50 ml sterile conical Falcon tubes (Nunc® EZ Flip™, Thermo Fisher Scientific, catalog number: 362696)
3. 1.5 ml sterile Eppendorf centrifuge tubes (Eppendorf, catalog number: 022364111)
4. Sonication device

Note: This was assembled in our laboratory by attaching four plastic tubes (3 cm diameter, 3 cm high) to a plastic box (Figure 1).

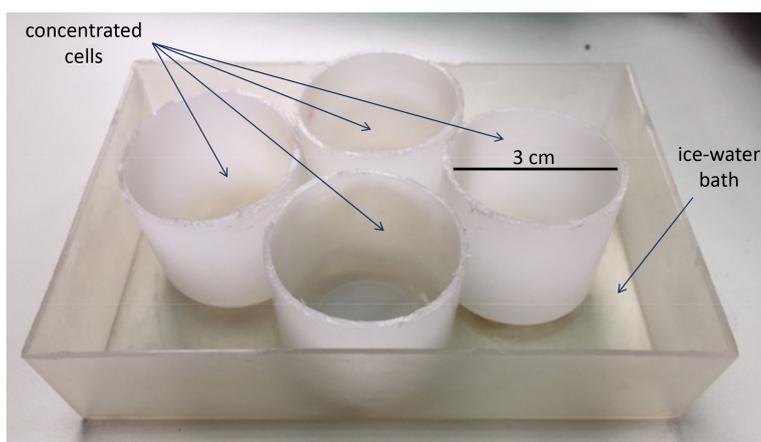


Figure 1. Sonication device

5. Paper towel (WypAll* X 60 Jumbo Roll, KCWW, Kimberly-Clark, catalog number: 30218593)
6. Spatula
7. Pyrex tray (Pyrex® Storage 13 x 18 cm)
8. *P. aeruginosa* strains

Note: PAO1, referred to as the wild-type, and catalase mutants PW8190 and PW8769 were obtained from the Washington Genome Center. Catalase mutants were constructed by insertion of IslacZ/hah transposon into katA (PW8190, hereinafter KatA-less strain) or katB (PW8769, hereinafter KatB-less strain) (Jacobs et al., 2003).

9. Distilled water
10. Tryptone (Oxoid, catalog number: LP0042)
11. Yeast extract (Merck, catalog number: 103753)
12. Sodium chloride (NaCl) (Biopack, catalog number: 1646.08)
13. Albumin from bovine serum (Sigma-Aldrich, catalog number: A4378)
14. Sodium phosphate dibasic (Na₂HPO₄) (Sigma-Aldrich, catalog number: S3264)
15. Sodium phosphate (NaH₂PO₄) (Sigma-Aldrich, catalog number: S0751)
16. Hydrogen peroxide (H₂O₂) 30% (Merck, catalog number: 107210)
17. Bugbuster Protein Extraction Reagent (Merck, Novagen, catalog number: 70584-4)
18. Sodium thiosulfate (Na₂S₂O₃) (Avantor Performance Materials, Macron™, catalog number: 8100-04)
19. Ammonium persulfate ((NH₄)₂S₂O₈) (MP Biomedicals, catalog number: 04802811)
20. TEMED (MP Biomedicals, catalog number: 02195516)
21. Trizma base (Tris[hydroxymethyl]aminomethane) (C₄H₁₁NO₃) (Sigma-Aldrich, catalog number: T1503)
22. Glycine (Sigma-Aldrich, catalog number: G7126)
23. Hydrochloric acid fuming 37% (HCl) (Merck, catalog number: 100317)
24. Acrylamide (Sigma-Aldrich, catalog number: A8887)
25. Bisacrylamide N,N'-methylene-bis-acrylamide (Sigma-Aldrich, catalog number: M7256)

26. EDTA (Merck, Calbiochem, catalog number: 324503)
27. Sodium hydroxide (NaOH) (Avantor Performance Materials, Macron™, catalog number: 7708)
28. Glycerol (Merck, catalog number: 104094)
29. Bromophenol blue (VWR, DBH, catalog number: 20015)
30. Ferric chloride (FeCl₃) (Avantor Performance Materials, Macron™, catalog number: 5029-04)
31. Potassium ferricyanide (K₃Fe (CN)₆) (UCB, catalog number: b1599)
32. LB medium (see Recipes)
33. 4 M NaCl (see Recipes)
34. Saline solution (see Recipes)
35. 50 mM sodium phosphate buffer, pH 7 (see Recipes)
36. 30 mM H₂O₂ (see Recipes)
37. 4 mM H₂O₂ (see Recipes)
38. 10% ammonium persulfate (see Recipes)
39. 1.5 M Tris-HCl buffer pH 8.8 (see Recipes)
40. 1.5 M Tris-HCl buffer pH 6.8 (see Recipes)
41. 30% acrylamide mix solution (acrylamide bisacrylamide ratio 37.5:1) (see Recipes)
42. 6% resolving gel solution (see Recipes)
43. 5% stacking gel solution (see Recipes)
44. 1 M Tris-HCl buffer pH 8 (see Recipes)
45. 0.5 M EDTA pH 8 (see Recipes)
46. Loading sample buffer (see Recipes)
47. Running buffer (see Recipes)
48. Ferric chloride/potassium ferricyanide solution (see Recipes)

Equipment

1. 50, 125 and 150 ml sterile Erlenmeyer flasks (DWK Life Sciences, Duran®, catalog numbers: 21 216 17, 21 216 28, 21 990 27)
2. 2-20 µl, 20-100 µl, 100-1,000 µl Kartell pluripet micropipettes (Kartell LABWARE, catalog numbers: 13000, 13210, 13220) and 1-10 ml Acura® manual micropipette (Socorex, model: Acura® manual 825/Acura® manual 835)
3. 50, 100 and 1,000 ml borosilicate measuring cylinders (VILABO, catalog number: 3501114, 3501115, 3501118)
4. Conventional incubator shaker (New Brunswick Scientific, model: G25)
5. Gyrotory water bath shaker (New Brunswick Scientific, model: G76)
6. Ice maker (Brema, model: TB 551)
7. UV-Vis Spectrophotometer (Biotraza, model: 752)
8. Refrigerated centrifuge (Hanil Scientific, model: Combi 514R)
9. Vibra-Cell sonicator (Sonics & Materials, model: VC500)

10. Electrophoresis cell (Bioamerica, model: DYCZ-24DNBA)
11. Power supply (Bioamerica, model: DYY-6CBA)
12. Freezer ultra-low temperature (Sanyo, model: MDF-U76VC)
13. Autoclave (HIRAYAMA, HICLAVE™, model: HVE-50)
14. Hot air oven sterilizer (Dalvo Instrumentos, model: OHR/T)

Procedure

A. Preparation of cell extracts

Non-inducing conditions

1. Grow *Pseudomonas aeruginosa* strains (PAO1, KatA-less and KatB-less) overnight in 30 ml of LB medium in 150 ml Erlenmeyer flasks at 37 °C with shaking (200 rpm).
2. Centrifuge the cultures (20 ml) in 50 ml Falcon tubes for 10 min, 10,000 $\times g$, 4 °C. Discard the supernatants.
3. Resuspend the cells with 20 ml of cold saline solution and keep them on ice.
4. Centrifuge for 10 min, 10,000 $\times g$, 4 °C. Discard the supernatants.
5. Resuspend the cells in ice-cold 50 mM sodium phosphate buffer, pH 7 up to OD₆₅₀ 1 (about 7 ml).
6. Sonicate 5 ml of concentrated cells in plastic tubes in an ice-water bath (Figure 1) under the following conditions: 2 min in pulsed mode, 18 mm tip diameter, 50% duty cycle, microtip limit 2. Keep the samples on ice.
7. Centrifuge the extracts for 10 min, 10,000 $\times g$, 4 °C. Discard the pellets carrying unlysed cells and cellular debris. Keep the supernatants on ice for a few hours until using them or store at -80 °C.
8. Determine protein concentration of the extracts according to Lowry *et al.* (1951); bovine serum albumin is used as a standard.

Inducing conditions

1. Grow *Pseudomonas aeruginosa* strains (PAO1 and KatB-less) overnight in 30 ml of LB medium in 150 ml Erlenmeyer flasks at 37 °C with shaking (200 rpm).
2. Dilute overnight cultures in 30 ml of fresh LB medium in 125 ml Erlenmeyer flasks to OD₆₅₀ 0.01 and grow in a gyratory water bath shaker at 37 °C until the cultures reach an OD₆₅₀ 0.3. This OD is reached in about 2 h.
3. Divide the cultures into two fractions of 10 ml each in 50 ml Erlenmeyer flasks. Maintain one of them untreated (control) and add 2.26 μ l of 30 mM H₂O₂ to the other fraction every 10 min for 1 h under sterile conditions, while shaking both fractions at 37 °C.
4. Take 5 ml of the treated cultures at the end of 1 h and add 10 μ l of 1 mg/ml sodium thiosulfate to neutralize the effect of H₂O₂; it is not necessary to keep the samples on ice.

5. Centrifuge the neutralized cultures and 5 ml of the control cultures for 10 min, 10,000 x g, 4 °C. Discard the supernatants.
6. Resuspend the cells with 10 ml of ice-cold 50 mM sodium phosphate buffer, pH 7.
7. Centrifuge for 10 min, 10,000 x g, 4 °C. Discard the supernatants.
8. Resuspend the cells in 1 ml of BugBuster Protein Extraction Reagent and keep for 15 min at room temperature.
9. Centrifuge for 10 min, 16,000 x g, 4 °C. Keep the supernatants on ice for a few hours until using them or store at -80 °C.
10. Determine protein concentration of the extracts according to Lowry *et al.* (1951); bovine serum albumin is used as a standard.

The protocol for inducing conditions is schematized in Figure 2.

Notes:

- a. *Antibiotics are not added to culture media.*
- b. *BugBuster Protein Extraction Reagent was employed instead of sonication to detect KatB activity. This reagent is capable of cell wall perforation without denaturing soluble proteins. It provides an alternative to mechanical methods such as French Press or sonication for releasing proteins.*

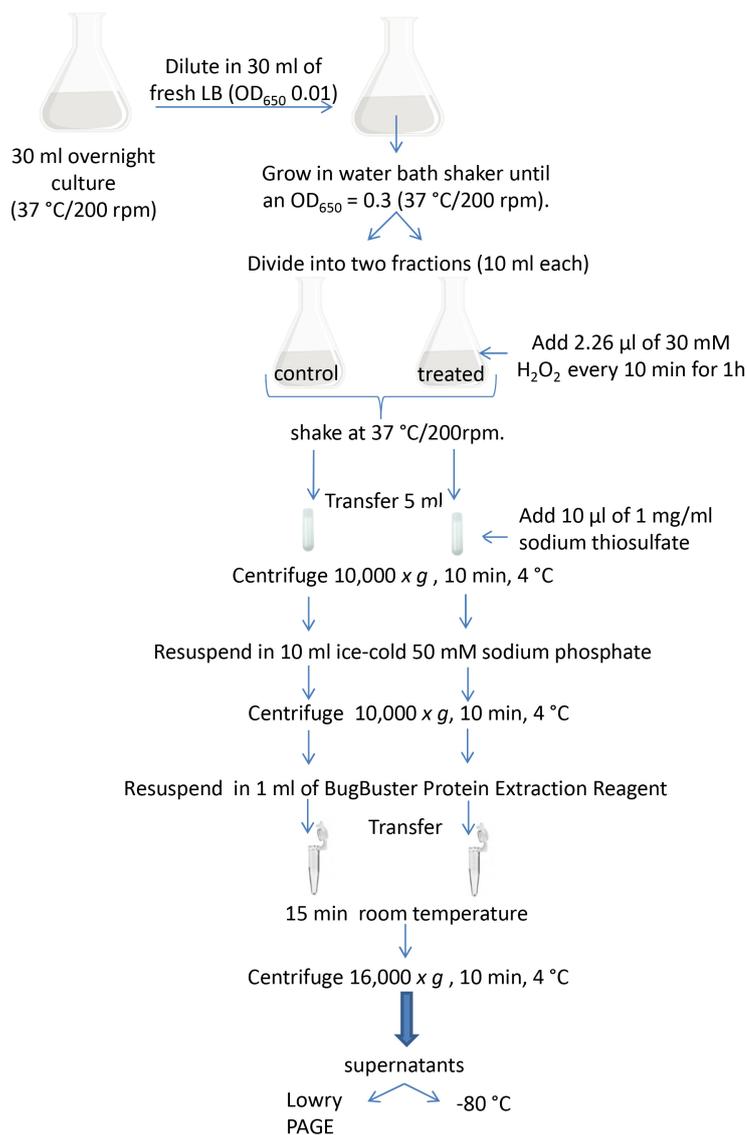


Figure 2. Schematic diagram of the experimental procedure

B. PAGE

1. Set the two glass plates with the special wedge frames in the gel casting stand (see Figure 3).

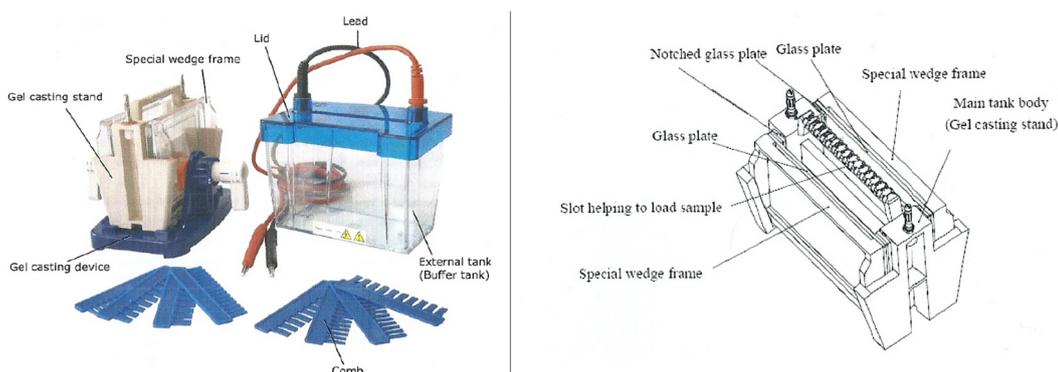


Figure 3. Electrohoresis cell (Bioamerica)

2. Pipet the 6% resolving gel solution into the gap between the glass plates.
3. Wait for 15-30 min until it solidifies.
4. Pipet the 5% stacking gel solution until overflow.
5. Insert the comb without trapping air under the teeth. Wait for 15-30 min until it solidifies.
6. Take the glass plates out of the gel casting stand and set them into the electrohoresis cell.
7. Pour running buffer into the electrohoresis cell until the buffer level is higher than the top of the (shorter) inner gel plate (*i.e.*, until the buffer covers the wells).
8. Mix 50 μ l of the cell extracts with 2 μ l of loading buffer in 1.5 ml Eppendorf tubes and heat them in boiling water for 10 min.
9. Load 10 μ g of protein per sample into each well. Cover the cell with the lid and connect the electrodes to the power supply.
10. Run the electrohoresis at 15 mA at room temperature (non-inducing conditions, KatA activity) or at 4 $^{\circ}$ C (inducing conditions, KatB activity).
Note: Electrohoresis is run at 4 $^{\circ}$ C for detecting KatB activity. This can be done in a cold room or in a refrigerator. However, this is not necessary for the detection of KatA activity.
11. Stop PAGE running when the dye front almost reaches the foot line of the glass plate. The run generally takes 4 h.

C. Gel development

1. Remove the glass plates from the electrohoresis tank and place them on a paper towel. Separate the plates by using a spatula.
2. Soak the gel in distilled water in a Pyrex tray for 5 min at room temperature. Discard the water.
3. Incubate the gel with 100 ml of a solution containing 4 mM H_2O_2 for 10 min at room temperature.
4. Remove the solution and wash the gel with 100 ml of distilled water at room temperature.
5. Soak the gel in 100 ml of a solution containing 1% (w/v) ferric chloride and 1% (w/v) potassium ferricyanide at room temperature.
6. As soon as the gel turns dark green, remove the ferric chloride/potassium ferricyanide solution and rinse with distilled water to prevent overdevelopment.

- Once the dye has been removed, photograph the gel immediately. Storage in distilled water for a few days at 4 °C is possible. Areas of catalase activity show up as clear bands.

Note: It is not necessary to use a shaker or a nutator for the washing and staining steps.

Data analysis

According to this procedure described above, Figure 4 shows representative images of non-denaturing polyacrylamide gels stained for catalase activity employing non-inducing conditions (A) or inducing conditions with H₂O₂ (B).

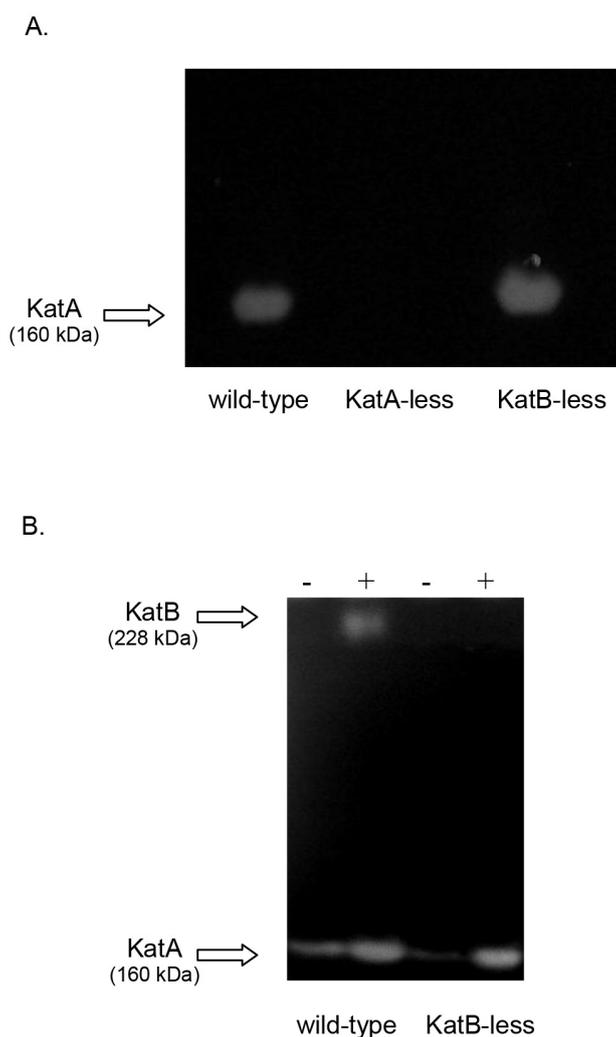


Figure 4. Catalase native PAGE analysis of *P. aeruginosa* wild type (PAO1) and its derivatives KatA-less and KatB-less strains. A. Extracts of PAO1 (wild-type), KatA-less and KatB-less strains grown under non-inducing conditions. The only activity detected corresponds to KatA since KatB is not expressed under this condition. B. Cultures of PAO1 and KatB-less strains, untreated (-) or treated with sublethal concentrations of H₂O₂ (+), were analyzed for catalase activity to demonstrate detection of KatB activity in non-denaturing polyacrylamide gels.

Note: The images were originally reported in Pezzoni et al., (2014). Protective role of extracellular catalase (KatA) against UVA radiation in Pseudomonas aeruginosa biofilms. J Photochem Photobiol B: Biol 131, 53-64.

Recipes

1. LB medium
 - Dissolve:
 - 10 g tryptone
 - 5 g yeast extract
 - 5 g NaCl
 - Bring the volume up to 1,000 ml in distilled water
 - Autoclave at 1 atm for 20 min
2. 4 M NaCl
 - Dissolve 46.7 g NaCl in 200 ml of distilled water
 - Autoclave at 1 atm for 20 min
3. Saline solution
 - Mix 7.5 ml sterile 4 M NaCl with 300 ml of sterile distilled water
4. 50 mM sodium phosphate buffer, pH 7
 - Solution A: dissolve 70.99 g Na_2HPO_4 in 500 ml of distilled water
 - Solution B: dissolve 59.98 g NaH_2PO_4 in 500 ml of distilled water
 - Mix 10.6 ml of solution A with 14.4 ml of solution B and add 475 ml distilled water
5. 30 mM H_2O_2
 - Add 0.340 ml of 30% H_2O_2 to 100 ml 50 mM sodium phosphate buffer, pH 7
 - Prepare fresh for every activity assay
 - The solution can be kept at room temperature during the experiment
6. 4 mM H_2O_2
 - Add 0.045 ml of 30% H_2O_2 to 100 ml distilled water
 - Prepare fresh for every activity assay
7. 10% ammonium persulfate
 - Dissolve 1 g ammonium persulfate in 10 ml distilled water
 - Store at $-20\text{ }^\circ\text{C}$ (shelf life 3 months)
8. 1.5 M Tris-HCl buffer pH 8.8
 - Dissolve 18.5 g Trizma base in 80 ml distilled water
 - Adjust to pH 8.8 with concentrated HCl and make up the volume to 100 ml
9. 1 M Tris-HCl buffer pH 6.8
 - Dissolve 12.114 g Trizma base in 80 ml distilled water
 - Adjust to pH 6.8 with concentrated HCl and make up the volume to 100 ml
10. 30% acrylamide mix solution

Dissolve 60 g acrylamide and 1.6 g bis acrylamide in 200 ml distilled water (acrylamide bisacrylamide ratio 37.5:1)

Notes:

- a. *Avoid directly contacting with polyacrylamide, ferricyanide and gels; they need to be handled with care.*
- b. *The order of adding solutions in the resolving and stacking gels solutions is important to avoid an early polymerization before pouring them between the glass plates.*

11. 6% resolving gel solution

5.4 ml distilled water
 2 ml 30% acrylamide mix
 2.5 ml 1.5 M Tris (pH 8.8)
 0.1 ml 10% ammonium persulfate
 0.008 ml TEMED

12. 5% stacking gel solution

2.1 ml distilled water
 0.5 ml 30% acrylamide mix
 0.38 ml 1.5 M Tris (pH 6.8)
 0.03 ml 10% ammonium persulfate
 0.003 ml TEMED

13. 1 M Tris-HCl buffer pH 8

Dissolve 12.114 g Trizma base in 80 ml distilled water
 Adjust to pH 8 with concentrated HCl and make up the volume to 100 ml

14. 0.5 M EDTA pH 8

Dissolve 47 g EDTA in 200 ml distilled water
 Adjust to pH 8 with concentrated NaOH and make up the volume to 250 ml

15. Loading sample buffer

1 ml 1 M Tris-HCl (pH 8)
 4 ml 0.5 M EDTA (pH 8)
 4 ml glycerol
 25 mg bromophenol blue
 1 ml distilled water
 Store at -20 °C

16. Running buffer

Dissolve 3.03 g Trizma base and 14.4 g glycine in 1,000 ml distilled water
 Store at 4 °C (shelf life 3 months)

17. Ferric chloride/potassium ferricyanide solution

Dissolve 1 g ferric chloride and 1 g potassium ferricyanide in 100 ml distilled water
 Prepare fresh for every activity assay

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