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Organic electrochemical transistors as novel biosensing platforms to study the electrical response of whole blood and plasma†

Valentina Preziosi,^a Mario Barra,^b Giovanna Tomaiuolo,^a
Pasquale D'Angelo,^c Simone Luigi Marasso,^{cd} Alessio Verna,^d
Matteo Cocuzza,^{cd} Antonio Cassinese^{be} and Stefano Guido^{afg}

In this paper, for the first time to the best of our knowledge, organic electrochemical transistors are employed to investigate the electrical response of human blood, plasma and alternative buffer solutions that inhibit red blood cell (RBC) aggregation. Our focus is on selecting a suitable electrolytic platform and the related operating conditions, where the RBC effect on the OECT response can be observed separately from the strong ionic environment of plasma in whole blood. The transient response of whole blood to pulse experiments is characterized by two time constants, which can be related to blood viscosity and to the capacitive coupling between the ionic and electronic components of the overall system. The role of capacitive effects, likely due to enhanced double-layer formation by negatively charged RBCs, is also confirmed by the increase of transconductance which was found in RBC suspensions as compared to the suspending buffer. Overall, the complex behavior found in these experiments provides new insights for the development of innovative blood-based sensing devices for biomedical applications.

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1. Introduction

Recently, organic thin-film transistors (OTFTs) have been attracting increasing attention thanks to their potential as sensors with many advantages, such as real-time analysis, low cost, biocompatibility, flexibility, and ease of integration in lab-on-a-chip devices.^{1,2} Organic electrochemical transistors (OECTs) – a special class of organic thin-film transistors – are particularly interesting for bioelectronic applications,³ being based on the exploitation of conductive polymers such as poly(3,4-ethylenedioxythiophene) polystyrene sulfonate (PEDOT:PSS), with mixed ionic-electronic

conduction.⁴ The OECT operation mechanism has been deeply discussed with reference to several kinds of applications.^{5–8} More in general, the OECT response with its very high transconductance enables operations at ultra-low voltages in liquids, making them particularly suitable to also investigate cellular systems and living matter.⁹ OECTs work as transducers and amplifiers from ionic signals, present in the electrolytic solution, to electronic ones. This feature allows the detection of low concentrations of ionic species, going well beyond the performances of the currently-available commercial sensors.^{10–13}

Such a highly sensitive tool and miniaturized devices could be coupled to provide robust platforms representing *in vitro* physio-pathologically relevant microenvironments. These could be able to mimic cell and organ level organizational structures, thanks to their user-defined design, small length scales and optimal control and sensing of mechanical and tissue environments. Taking advantage of microfluidics principles, organ-on-chip devices are integrated platforms consisting of microfluidic channels populated with several cell types to create minimal functional organ units for drug testing and modelling of human physiology and disease.^{14–16} Beyond specific cell lines, blood, *i.e.* the fluid tissue that delivers oxygen and nutrients to tissues and cells and that is responsible of the immunological response, is the most exploited biological fluid for biomedical applications. Blood is a complex fluid consisting in a suspension of cells, erythrocytes (red blood cells

^a Department of Chemical, Materials and Production Engineering - University Federico II, P.le Tecchio 80, I-80125 Naples, Italy.

E-mail: valentina.preziosi@unina.it

^b CNR-SPIN, c/o Department of Physics "Ettore Pancini", P.le Tecchio, 80, I-80125 Napoli, Italy. E-mail: mario.barra@spin.cnr.it

^c IMEM-CNR, Parco Area delle Scienze 37/A, I-43124 Parma, Italy

^d Chi-Lab, Department of Applied Science and Technology, Politecnico di Torino, C.so Duca degli Abruzzi 24, 10129 Torino, Italy

^e Department of Physics "Ettore Pancini", University Federico II, P.le Tecchio 80, I-80125 Naples, Italy

^f National Interuniversity Consortium for Materials Science and Technology (INSTM), 50121 Firenze, Italy

^g CEINGE, Advanced Biotechnologies, 80145 Napoli, Italy

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– RBCs), leucocytes (white blood cells – WBCs) and platelets, immersed in a low viscous fluid called plasma.¹⁷

In this scenario, OECTs can be particularly precious, unravelling the issue of blood handling and sensing in integrated microfluidic devices. In particular, RBCs, being the major constituent of blood, are deformable cells that have a great impact on the physiological flow of blood in microvasculature.^{18,19} Their abnormal shape and concentration, indeed, can lead to organ and tissue disfunction and death.^{20–23} Moreover, plasma, being a water solution of cations like Na^+ , K^+ , Ca^{2+} and Mg^{2+} , anions like Cl^- and HCO_3^- , and high molecular weight substances like proteins and lipoproteins, strongly impacts the electrical response of blood.^{24,25} For a long time, there has been a lot of interest about the electrical behavior of human blood, in terms, for example, of the conductivity (or its reciprocal, the resistivity), given its strict correlation with blood physiology and the repercussion of blood electrical variations on biomedical devices. In 1873, Maxwell²⁶ published a book containing a theoretical dissertation about the conductivity of a suspension of non-conducting spheres in a conducting medium. Some years later, Hober determined for the first time the different resistivity of blood, at low and high frequencies, due to the presence of a membrane around RBCs.

Later on, first Fricke²⁷ and then Velick and Gorin,²⁸ extended Maxwell's work and applied it to blood, assuming RBCs as non-conducting ellipsoids and plasma as a conductive medium. Other works followed that correlated blood conductivity with the presence of erythrocytes and plasma leading to the conclusion that, even if plasma is a complex fluid full of ionic species, the presence of red cells plays a key role in the electrical response of blood.^{29,30} In particular, some studies addressed the change in human blood electrical properties in static conditions due to the haematocrit (*Hct*), *i.e.* the ratio of the volume fraction of the blood cells and the total volume, and to the red blood cell random orientation. Other studies instead focused on the effect of RBCs' parallel orientation under flow confirming that, even if RBCs display a low conductivity with respect to plasma, their impact cannot be neglected.²⁹ Despite all the studies on the electrical properties of human blood, the effect of RBCs in terms of electrical conductivity is still controversial, making the exploitation of blood-based electrical devices still an open issue.

In the present work, the problem of the response of OECTs when human blood, plasma and an alternative buffer solution containing RBCs in suspension are used, is addressed. Building on previous reports by our group on the use of OECTs as sensing devices^{31–37} and on the analysis of blood flow in blood-on-chip devices,^{38,39} the main goal of this paper is to exploit OECTs as a reliable platform to distinguish the effect of RBCs from those related to the plasma, which is characterized by a high ionic concentration.

A major challenge associated with running OECTs with blood and RBC suspensions is related to the particulate nature of blood and to the peculiar properties of RBCs, such as deformability and a cell's tendency to aggregate under static or weak flow conditions, as well as to the number of electrolytes present in plasma. Despite the key role that blood could play in

integrated devices, no experimental study on the use of OECTs for blood testing has been published so far, to the best of our knowledge. Our approach is based on the exploitation of OECTs with whole blood and RBC suspensions at different values of hematocrit to separate the effects of RBC and plasma. Different responses have been found in this way.

2. Materials and methods

Materials and sample preparation

RBC suspensions. Venous blood samples were withdrawn from healthy volunteers and used within 4 hours of collection. Blood was stored in vials and rotated on a blood tube rotator to avoid cell sedimentation just before the experiment. For whole blood experiments, blood was measured at hematocrit values *Hct* (*i.e.* RBC volume fraction) around 40%. In addition to whole blood, suspensions at 25% and 45% of *Hct* were prepared by diluting RBCs in a 1 : 100 Anticoagulant Citrate Dextrose (ACD) and Albumin solution. To this end, blood was centrifuged (Hettich Universal 32 R) at 1600 revolutions per minute for 15 min and at 20 °C to separate the RBCs from other cellular components and plasma. ACD was purchased from Haemonetics (0.6% citric acid, 1.1% anhydrous dextrose, 2.3% sodium citrate, 96% water) and used to avoid clotting. Fetal bovine serum (FBS-Albumin) was purchased from Merck and it was dissolved in ACD (4 g in 10 ml ACD) by mixing the solution in a beaker on a magnetic stirrer for about 20 min at room temperature.

Conductance measurements of ACD and ACD plus albumin solutions, performed by using a conductivity meter (PC 2700, Eutech Instruments), gave values about 14 mS for ACD and 10 mS for ACD plus albumin. These values are similar to the typical ones measured for whole blood.⁴⁰

Microscopy imaging

Optical images of whole blood and of RBC suspensions at 25% and 45% were acquired by using an optical inverted microscope (Zeiss Axiovert 100) connected to a videocamera (Phantom 4.3) in bright field using a 40× oil immersion objective (Zeiss).

OECT fabrication

The experimental apparatus used in this work has been described in detail elsewhere^{31,32} and it will be just briefly reviewed here. Device fabrication followed the already reported procedure in ref. 31 and 41 (Fig. 1). Electrodes, fabricated by e-beam evaporation (ULVAC EBX-14D), were obtained by depositing Ti/Au film (thickness 10 nm/100 nm) on a Si wafer (100) finished with 1 μm of thermal oxide. A 200 nm thick PEDOT:PSS layer (Clevios PH 1000, doped with 5 volume % ethylene glycol, 0.1 volume % dodecyl benzene sulfonic acid, and 1 wt% of GOPS (3-glycidylxypropyl)trimethoxysilane) was obtained by spin coating on the source and drain electrodes and subsequently patterned *via* photolithography (Microchemicals AZ9260 resist) and etching with O_2 plasma. A PDMS chamber, with an internal volume of 150 μl, was aligned with the

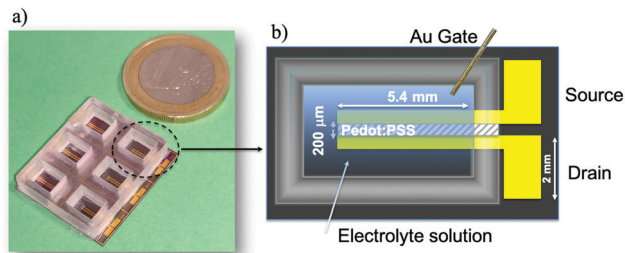


Fig. 1 (a) Picture and (b) sketch of the OECT device.

PEDOT:PSS channel and irreversibly bonded on the surface (Fig. 1a). The final device geometry (see Fig. 1b) is characterized by a width to length (W/L) ratio equal to 30 ($W = 6$ mm, $L = 200$ μm). In particular, drain and source electrodes are 250 μm in width and 7 mm in length both having a contact pad of 1 mm \times 2 mm.

OECT characterization. OECT devices were characterized by using a homemade probe station connected to a two-channel multimeter (Keithley 2612A) and controlled by Labview software. As the gate electrode, a gold wire (99.99% Au supplied by Franco Corradi sas) with a diameter of 1 mm was immersed into the electrolyte for the OECT characterization. Hence, in the investigated configuration, the gate electrode was not co-planar with source and drain contacts. Additional measurements were also performed using a platinum wire (99.9% supplied by Franco Corradi sas, with a diameter of 0.6 mm). During the measurements, attention was paid in order to immerse the same equivalent area (about 8 mm²) for both gold and platinum electrodes.

During the overall time required to complete the OECT measurement protocol (about 30 minutes), we can assume that the RBCs were sedimented. However, our experimental findings suggest that the cell sedimentation does not play a significant role on the OECT response, as will be discussed in the Results section. After each measurement, fresh blood was replaced and all the experiments were carried out at room temperature.

3. Results and discussion

Microscopy imaging

Fresh venous human blood withdrawn from healthy volunteers was initially observed using an optical microscope to check its health state. In static conditions, healthy RBCs assume a biconcave shape with a diameter around 8 μm and in absence of any anticoagulant they tend to aggregate and form microstructures called rouleaux.⁴² An optical image of whole blood over time is shown in Fig. 2 where the enhanced presence of rouleaux occurring in the order of seconds from the first acquisition is more and more evident. RBCs have also been observed in ACD plus albumin solutions at two different cell concentrations, 25 and 45%, where cells do not aggregate owing to the presence of the anticoagulant (ACD). Cell viability was also monitored after the OECT experiments (about 2 hours after collection) and no cell alteration was observed (Fig. S1, ESI[†]).

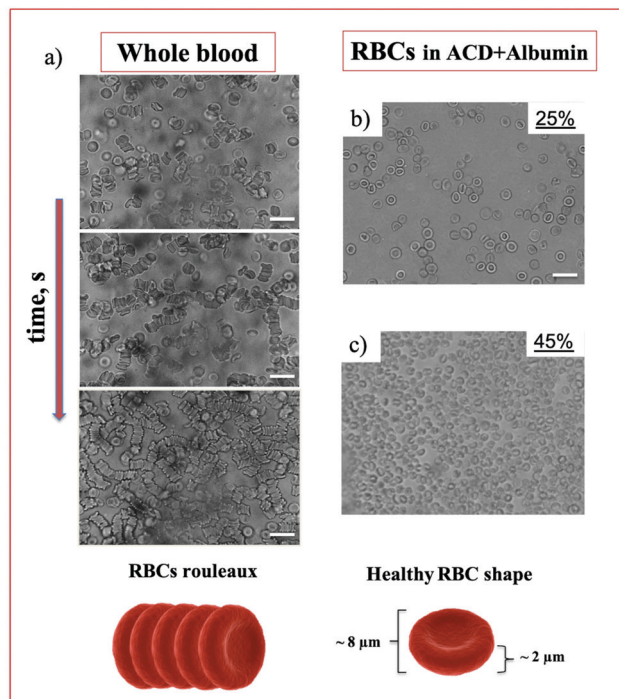


Fig. 2 Optical images of whole blood just after withdrawal and after a few seconds (about 30 s from one image to another) to show the formation of rouleaux (left panel), and of RBCs at 25% and 45% suspended in ACD plus albumin solution. Scale bar is 20 μm . Schematic drawings of RBCs rouleaux and cell shape in healthy conditions are shown.

OECT response in blood and plasma

In the first part of this work, fresh whole blood and plasma were analyzed in terms of the related OECT response. Plasma, in particular, was obtained by centrifuging blood from the same donor as described in the Experimental section. The PEDOT:PSS active channel was in contact with about 150 μl of the solution under investigation thanks to a Polydimethylsiloxane (PDMS) well, acting as a fluid reservoir.

Gold gate electrodes were basically considered to study the OECT response. After gate immersion into the reservoir, a positive voltage between the gate and source, V_{GS} , was applied in order to drive the PEDOT:PSS-based OECT operation in the depletion-mode regime.¹⁰

In particular, in the first set of experiments the OECT response was visualized by applying sequences of V_{GS} pulses, with a duration of 60 s and progressively-increasing amplitudes going from 0 to 0.6 V (step 0.1 V) at a fixed drain–source voltage, $V_{\text{DS}} = -0.1$ V (Fig. 3a). The drain–source current (I_{DS}) and the corresponding gate–source current (I_{GS}) were thus recorded as a function of time. Hence, the OECT current modulation capability could be expressed by the ratio, $\Delta I_{\text{DS}}/I_0$, derived from the pulsed measurements and described in detail elsewhere³¹ (Fig. 3b).

Measurements in Fig. 3 confirm that, even in this complex media, OECTs operate correctly as depletion-mode transistors under the application of positive V_{GS} (*i.e.*, I_{DS} is lowered in absolute value). It should also be mentioned that, in this case,

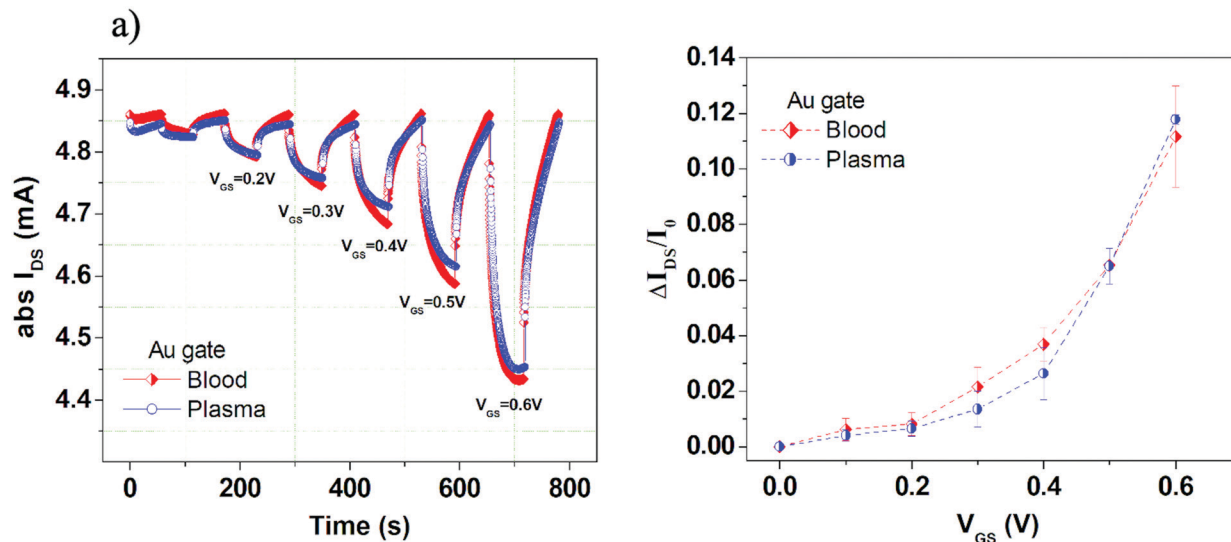


Fig. 3 OECT electrical response. (a) Pulse reporting I_{DS} in absolute values as a function of time at V_{GS} from 0 to 0.6 V for plasma and whole blood using a gold gate. (b) OECT modulation ratio ($\Delta I_{DS}/I_0$) as a function of V_{GS} for plasma and whole blood. The $\Delta I_{DS}/I_0$ values were obtained as an average of three consecutive measurements, with the error bars representing the standard deviation, in order to assess the device operation stability throughout the measurements.

the measurements were exploited up to $V_{GS} = 0.6$ V, due to the occurrence of device response degradation at larger V_{GS} (namely, the lack of the correct depletion-mode operation). While comparing the OECT responses in blood and plasma with gold electrodes, we notice that only small differences are appreciable mainly for V_{GS} between 0.2 and 0.5 V, with slightly enhanced current modulation values in blood. The strong similarity between the responses in these two fluids can be likely ascribed to the very large ionic concentrations in plasma.

In order to get additional insights about the OECT response in plasma and blood, the $I_{DS}(t)$ transient pulses, achieved with the gold electrode and for the different V_{GS} pulses, were further analyzed. Hence, these experimental curves were normalized with respect to the initial value and shifted at $t = 0$ s. The obtained datasets are reported in Fig. 4a for whole blood and in Fig. 4b for plasma. Then, following the approach introduced in ref. 31, these data were fitted by using the double-exponential equation:

$$I_{DS}(t, V_{GS}) = I_{SS} + A_1 \cdot e^{-\left(\frac{t}{\tau_1}\right)} + A_2 \cdot e^{-\left(\frac{t}{\tau_2}\right)} \quad (1)$$

where I_{SS} is the current value at the end of the 60 s pulse, A_1 and A_2 are two pre-exponential factors and τ_1 and τ_2 are two time constants. The fitting parameters τ_1 and τ_2 of whole blood and plasma represent mainly the OECT response for long and short times, respectively, and are reported in Fig. 4c.

For better clarity, a single $I_{DS}(t)$ pulse, compared with the related fitting curve, is shown (Fig. S2a) in the ESI,[†] where the overall behavior of A_1 and A_2 achieved for the whole set of transient measurements is also presented (Fig. S2b and c, ESI[†]).

Appreciable differences between blood and plasma can be observed in Fig. S2b and c (ESI[†]) for the A_1 pre-factor with

respect to the A_2 one. In the case of τ_1 and τ_2 , for both time constants, blood experimental data provide higher values in the V_{GS} range between 0.2 and 0.5 V (Fig. 4c and d). As a whole, these data suggest that the OECT response in blood is able to have larger I_{DS} modulation but its dynamics as a function of time is slightly slower if compared with the case of plasma.

As a further confirmation of this interpretation, in Fig. 4e we report the estimated current modulation at the end of the 60 s pulse conditions (*i.e.* derived from the I_{SS} fitting parameter in eqn (1)). Here, the enhanced current modulation capability in blood is still more evident than what was observed in Fig. 3b.

As discussed in ref. 31, the larger values of τ_1 should be ascribed to the enhanced viscous character of whole blood.⁴³ This type of observation is in qualitative agreement with a recently-proposed model where the diffusion of cations throughout the electrolytes was correlated to the solution viscosity, in analogy with the Einstein-Smoluchowski theory.⁴⁴ On the other hand, the short time constant τ_2 should be more related to the capacitive coupling between the ionic and electronic components of the overall system. In our experiments, in particular, by considering the size of the gate immersed area and the overall dimensions of the PEDOT:PSS channel with the related volumetric capacitance ($C_V \sim 40$ F cm⁻³), we can infer that the OECT response is basically ruled by the electric double-layer (EDL) capacitance (C_G) at the gate-electrolyte interface (*i.e.* $C_G \ll C_V$).⁴⁵ In this scenario, the increased values of τ_2 (with V_{GS} between 0.2 and 0.5 V) for blood may be associated to larger values of C_G as a possible effect of the presence of RBCs and related motion toward the gate surface.

As reported in the literature,^{46,47} a RBC's surface is negatively charged and hence, in an electrolytic solution, an electrical double layer is present also around the RBC. Under the effect of an external field (*i.e.* under positive V_G during the

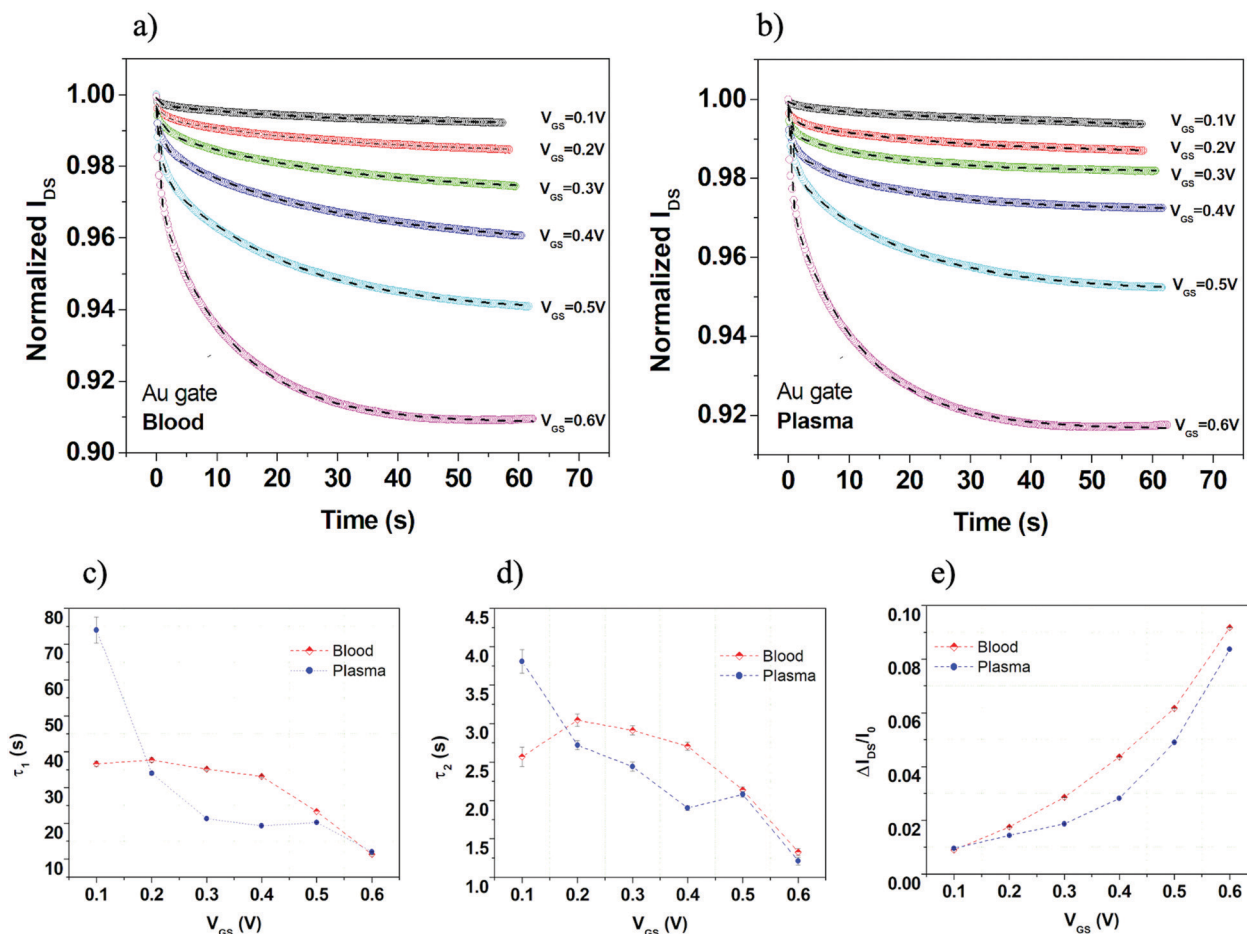


Fig. 4 (a) Set of I_{DS} vs. time curves (normalized to the initial value and time shifted at $t = 0$ s) recorded at different V_{GS} for blood and (b) plasma using a gold gate. (c) τ_1 and (d) τ_2 time constants as a function of V_{GS} extracted by fitting experimental $I_{DS}(t)$ curves with eqn (1) for whole blood and plasma. (e) OECT modulation ratio values estimated by the fitting procedure in the steady state conditions as a function of V_{GS} for plasma and whole blood.

OECT operation) RBCs, with the surrounding charge layers and the related diffusive components, are attracted toward the gate electrode. As a whole, this effect should produce a local increase of the ionic concentrations close to the gate surface, providing the final increase of C_G . Although this is a tentative explanation requiring additional and more specific experiments to be definitely proved, it is to be highlighted how the hypothesized C_G capacitance increase in blood vs. plasma is compatible also with the observation of the corresponding larger gate current I_{GATE} values (see the data below and refer to Fig. S3b, ESI[†]). Given the polarizable nature of the gold electrode, in this case, a direct relation between the gate current and electrolyte-gate capacitive couplings is indeed expected.

Complementary experiments were also performed by using a platinum wire as the gate electrode. In particular, Fig. S3 (ESI[†]) shows $I_{DS}(t)$ pulsed measurements achieved in this case for whole blood and plasma. As shown, the OECT responses with the platinum gate electrode are even more similar and no significant differences can be observed in terms of $\Delta I_{DS}/I_0$ values which are, however, larger (by a factor close to 2) than those measured with the gold electrode. These findings highlight the less polarizable nature of platinum when compared

with gold, in agreement with other recent works where the occurrence of faradaic reactions at the platinum surface was indeed exploited to develop specific sensing devices.^{48–50} In particular, the excellent electrocatalytic activity of Pt (even in the form of nanoparticles incorporated in the PEDOT:PSS channel) toward H_2O_2 oxidation was exploited to demonstrate the efficient detection of neurotransmitters.⁵¹ Overall, it is essential to remember that the OECT response can be strongly affected by the physiochemical nature of the gate.^{52,53}

OECT response in ACD plus albumin solutions

A second set of experiments was focused on the specific role of RBCs in the OECT response. For this reason, alternative electrolytic solutions able to preserve the initial healthy conditions of the erythrocytes were used. Hence, RBCs were separated from whole blood by centrifugation and suspended again in a buffer solution made of ACD and albumin⁵⁴ (see the Material and methods section).

ACD is commonly employed as an anticoagulant to prevent the formation of cell aggregates and albumin, being the most abundant protein in blood plasma, operates, at an appropriate concentration, as a nutrient for RBCs. ACD and albumin

supplemented ACD solutions have been used by the authors in other studies^{18,19,55} and have been preliminarily tested here as electrolytes in the OECT (Fig. S4, ESI†). As shown, ACD and ACD plus albumin solutions show a very similar OECT response in terms of the current modulation capability, with the measured $\Delta I_{DS}/I_0$ values remaining remarkably smaller than those observed in plasma and blood (by about a factor of 5 at $V_{GS} = 0.6$ V). Fig. S4b (ESI†) also shows the I_{GATE} values achieved for ACD and ACD plus albumin solutions in comparison again with blood and plasma. This trend reproduces very well that observed for the $\Delta I_{DS}/I_0$ values with the I_{GATE} being strongly reduced for the ACD and ACD plus albumin solutions.

Once characterized in terms of their baseline OECT modulation signal, ACD plus albumin solutions were used as a buffer to suspend the RBCs. Two different RBC concentrations, at 25% and 45% (the latter corresponding to the hematocrit in whole blood), were considered here.

In this regard, Fig. S5a (ESI†) shows the $\Delta I_{DS}/I_0$ modulation curves achieved for the whole blood, the basic ACD plus albumin solution and for the same type of solution containing 25% and 45% of RBCs. These data outline unambiguously how the OECT response is affected by the presence of RBCs. In particular, the responses recorded in the solutions containing the two RBC concentrations, 25 and 45% (the latter corresponding to whole blood *Hct*), exhibit an intermediate character in comparison with the whole blood and the basic suspending medium. The I_{DS} modulation capability was somewhat enhanced for the larger RBC concentration, at increasing V_{GS} (Fig. S5b, ESI†).

In order to further analyze the OECT response in these fluids, transfer curves at $V_{DS} = 0.1$ V were also recorded. In this case, the investigated V_{GS} range was extended up to 0.8 V without observing any degradation signature in the depletion-mode OECT response.

Normalized transfer curves, reporting the change in the device current I_{DS} normalized to the value measured at $V_{GS} = 0$ V, are

shown in Fig. 5a. Coherently with the features observed in the $\Delta I_{DS}/I_0$ curves (Fig. S5a, ESI†), the responses recorded in the ACD plus albumin solutions containing the two RBC concentrations exhibit again an intermediate character. A more careful inspection of these measurements highlights also a different V_{GS} dependence. This specific behavior was better emphasized by analyzing the OECT transconductance ($g_m = \frac{\partial I_{DS}}{\partial V_{GS}}$), which is a

very important parameter simply defined as the derivative of the channel current I_{DS} with respect to the gate voltage V_{GS} . In particular, such a parameter describes the OECT signal amplification performances.^{56,57} In Fig. 5b, transconductance (g_m) is plotted as a function of V_{GS} for all the solutions investigated and very different trends can be observed. Indeed, g_m rises at increasing V_{GS} for all fluids, but the specific V_{GS} value where this behavior starts and the presence of a peak, with the related magnitude, are characteristic of any solution.

In particular, three different regions can be identified:

- The low V_{GS} range (approximately up to $V_{GS} = 0.3$ V) where all g_m curves assume values of few hundreds of μ S, with a slight V_{GS} dependence. Here, the solutions containing RBCs seem to display smaller g_m even in comparison with the basic ACD plus albumin suspending medium;

- The intermediate V_{GS} region (approximately between 0.3 and 0.6 V) where the g_m rising behavior is much more evident. In this case, in comparison with the ACD plus albumin suspending medium, the presence of a major concentration of RBCs tends to reduce the V_{GS} threshold above which this trend is more pronounced.

- The high V_{GS} range (between 0.6 and 0.8 V), where, except for the basic ACD plus albumin suspending medium, g_m exhibits a peak for all the solutions. Here, the ACD plus albumin solutions with 25% and 45% RBC concentrations differ for both the peak amplitude and the V_{GS} value where this occurs.

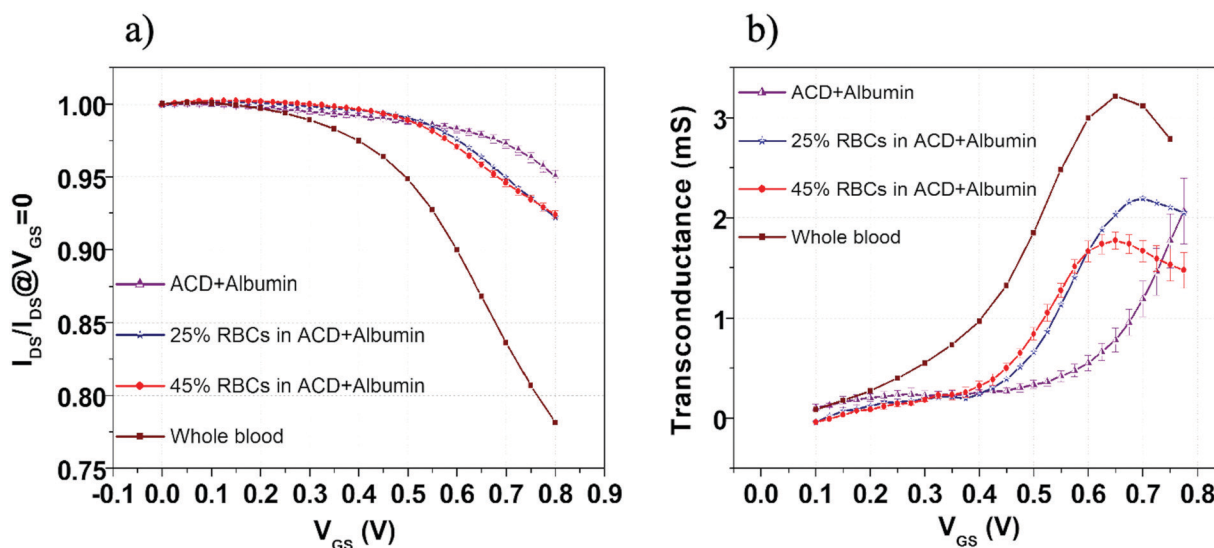


Fig. 5 (a) Normalized transfer curves I_{DS}/I_{DS} at $V_{GS} = 0$ V and (b) transconductance as a function of V_{GS} for ACD plus albumin, plasma, RBCs at 45% in ACD plus albumin and whole blood. V_{GS} is pulsed from 0 to 0.8 V and V_{DS} was set at 0.1 V.

Thus, the g_m behavior in the intermediate V_{GS} region, with higher *trans*-conductance for the ACD plus albumin solution at 45% RBC concentration with respect to that at 25% RBC, seems to suggest again an increase of the capacitive coupling ($g_m \propto C_G$ since $C_G \ll C_V$) between the gold gate electrode and the electrolytic medium. This occurrence agrees well with the abovementioned considerations about the OECT transient response in blood (*i.e.* larger values of the short time constant τ_2), pointing again at the role of negatively-charged RBC in enhancing the C_G value in the same V_G range.

On the other hand, when V_{GS} values increase (between 0.6 V and 0.8 V), the OECT response reaches a different regime where the major RBC concentration in ACD plus albumin solutions provides a reduced I_{DS} modulation capability (*i.e.* the g_m peak amplitude is lowered and occurs at lower V_{GS}). This condition underlines a weaker penetration of cationic species into the PEDOT:PSS film and is compatible with a field effect-like operation mode localized at the PEDOT:PSS surface.⁵⁸

The physical interpretation of this phenomenon is not straightforward but it could be tentatively explained in terms of an increasing repulsion effect among the negatively-charged RBCs close to the gold surface or to the detrimental occurrence of RBC sticking effects at the same surface. Both occurrences, indeed, could degrade the electric double layer formation and make the progressive enhancement of the gate signal ineffective in promoting a further cationic uptake by the active channel. It is worth mentioning, in any case, that while it was previously shown that the position and magnitude of the g_m peak in the OECT response can be tailored as a function of the device geometrical layout and/or the biasing conditions, here we found that these features can be clearly associated also with the nature of the analyzed electrolyte.^{59,60}

As a final comment suggested by the data summarized in Fig. 5, we want to highlight how the transconductance curves recorded for blood and ACD plus albumin solution at 45% RBC concentration display a very different behavior, even in the presence of the same RBC concentration.

As discussed above, this finding is obviously related to the different suspending medium, but it is also a demonstration that the RBC sedimentation does not affect ion diffusion towards the PEDOT:PSS channel. On the other hand, if RBC sedimentation on the PEDOT:PS surface was the dominant feature affecting the operation of OECT here investigated, higher I_{DS} modulation values would be expected for plasma in comparison with the whole blood. This occurrence was clearly not observed during our experiments.

3. Conclusions

In this work, OECT based devices are exploited for the first time to study the electrical response of whole blood and RBC suspensions. The response of OECTs in blood and plasma was analyzed by using both gold and platinum gate electrodes. While the OECT behavior with the platinum gate was found to be completely dominated by the strong ionic concentration

related to plasma, with gold electrodes we were able to identify distinctive features of the OECT steady state and transient responses in blood and plasma. In a second set of experiments, aimed at separating the effect of plasma and RBCs by suspending RBCs, at different concentrations, in a buffer (ACD) supplemented with albumin, we observed a clear difference in the response of OECTs. In particular, the latter was analyzed in terms of the transconductance (g_m) parameter extracted from transfer curve measurements. The role of negative charges distributed on the RBC surface has been tentatively invoked to explain the main findings reported here. Finally, RBC morphology was monitored just after the withdrawal and during the electrical experiments and no alterations were observed. Indeed, microscopy observations show that, after the application of voltage to blood or RBC suspensions, no cell lysis or evident structural modifications can be observed, thus proving that OECTs are suitable tools for blood-based devices. In conclusion, despite the fact that the use of OECTs for blood sensing is far from straightforward due to the intrinsic complexity of blood and its components, these results could lead to the development of a new class of integrated devices suitable for drug testing and clinical diagnostics.

Author contributions

V. P., M. B. and G. T. conceived the project. M. B. and V. P. performed OECT measurements. V. P. and G. T. performed conductivity measurements and acquired microscopy images. A. V. and M. C. designed the OECT and realized the electrode photolithographic masks. S. M., A. V. and P. D. realized OECT devices and performed some of the OECT experiments. S. G. and A. C. contributed to the discussion and reviewed the manuscript. All the authors discussed results. V. P. and M. B. wrote the manuscript.

Conflicts of interest

The authors declare no competing financial interests.

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