

Room-Temperature ALD of Al₂O₃, TiO₂, SiO₂ and SiN_x Enabled by Energy-Enhanced ALD Techniques

S. E. Potts,^{*,1} H. B. Profijt,¹ R. Roelofs,¹ E. M. J. Braeken,¹
H. C. M. Knoop,¹ S. Haukka,² V. Pore² and W. M. M. Kessels¹

¹Department of Applied Physics, Eindhoven University of Technology, Eindhoven, The Netherlands.

²ASM, Väinö Auerin Katu 12 A, 00560 Helsinki, Finland.

*s.e.potts@tue.nl

Atomic layer deposition (ALD) is known to give high quality films at relatively low temperatures (≤ 200 °C) [1]; however, as ALD is driven by surface reactions, there is the potential to obtain saturating ALD at the lowest extreme of room temperature. For room-temperature ALD (RT-ALD), thermal ALD processes are generally not suitable due to a lack of thermal energy, especially with respect to the ligand removal step when using H₂O or NH₃, making the deposition times and purge times impractically long. Energy-enhanced ALD techniques [2], such as plasma-enhanced [3] or ozone-based ALD, can help to overcome these problems. However, despite employing energy-enhanced co-reactants, RT-ALD is not always possible. Using Al₂O₃, TiO₂, SiO₂ and SiN_x as examples, we discuss here the mechanisms and fundamentals of RT-ALD processes.

To test the viability of RT-ALD processes, Al(CH₃)₃ (TMA), Ti(OⁱPr)₄ (TTIP), SiH₂(NEt₂)₂ (BDEAS) and SiH₂(NH^tBu)₂ (BTBAS) were employed as high-vapour-pressure metalorganic precursors to Al₂O₃, TiO₂, SiO₂ and SiN_x, respectively (Fig. 1). For oxides, an O₂ plasma or O₃ gas was used as the co-reactant [4]. All plasma-enhanced RT-ALD processes gave growth at room temperature (Fig. 1a-c), which was a result of the high reactivity of the O₂ plasma. For the O₃-based processes, growth was only observed for Al₂O₃ (1.1 Å/cycle). In the case of TiO₂, the lack of growth with O₃ was a result of insufficient thermal energy, which could also be the case for SiO₂. However, the heteroleptic nature of the BDEAS precursor was also an important factor, as O₃ exhibits a low reactivity towards Si–H surface groups [5] under ALD conditions. For SiN_x, growth was obtained using a remote plasma (Fig. 1d), but the films contained significant quantities of C and O.

It is evident that the viability of an RT-ALD process is not straightforward. It is essential that both the metalorganic precursor and the co-reactant be sufficiently reactive with the surface groups left after the preceding respective RT-ALD half-cycle. High-vapour-pressure precursors and short purge times are desirable. Using these results, the practicalities and possibilities for RT-ALD will be demonstrated and discussed.

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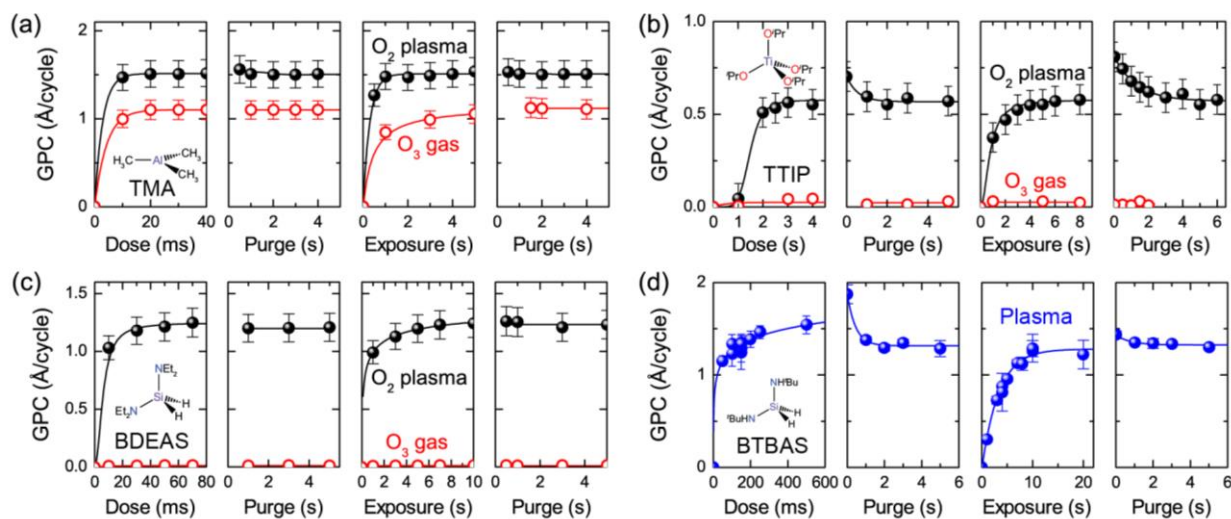


Fig. 1. RT-ALD saturation curves for (a) Al₂O₃, (b) TiO₂, (c) SiO₂ and (d) SiN_x as measured by *in situ* spectroscopic ellipsometry. All plasma-enhanced RT-ALD processes gave saturating growth-per-cycle (GPC) values of (a) 1.5, (b) 0.6 and (c) 1.2 Å/cycle for Al₂O₃, TiO₂ and SiO₂, respectively. Growth was only obtained with O₃ where TMA was the precursor (1.1 Å/cycle). RT-ALD of SiN_x afforded a GPC of 1.3 Å/cycle.