

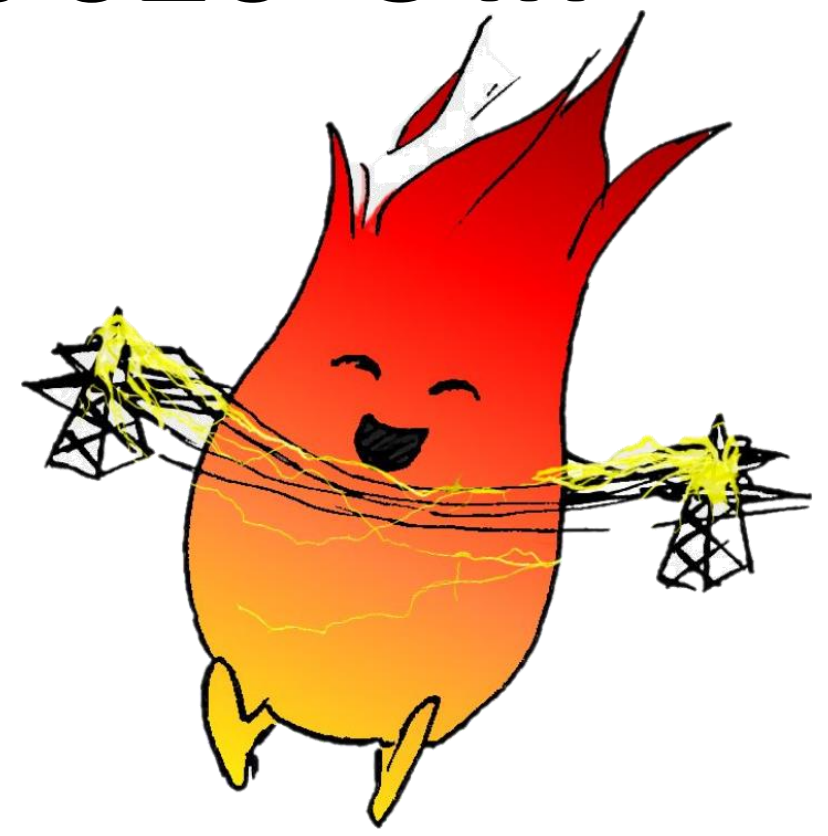
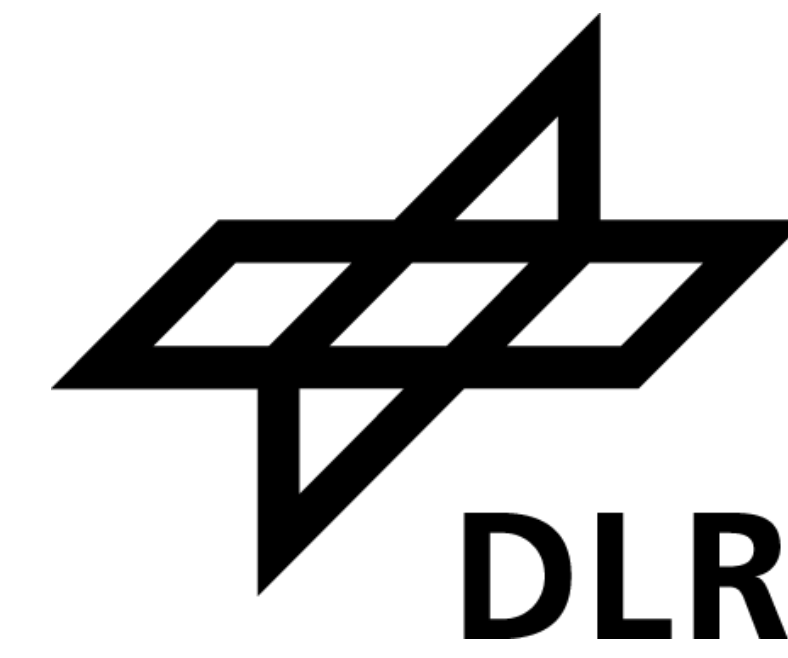
Concentrating Solar Power at higher limits: Studies on molten Nitrate Salts at 620°C in a laboratory pilot-scale hot tank

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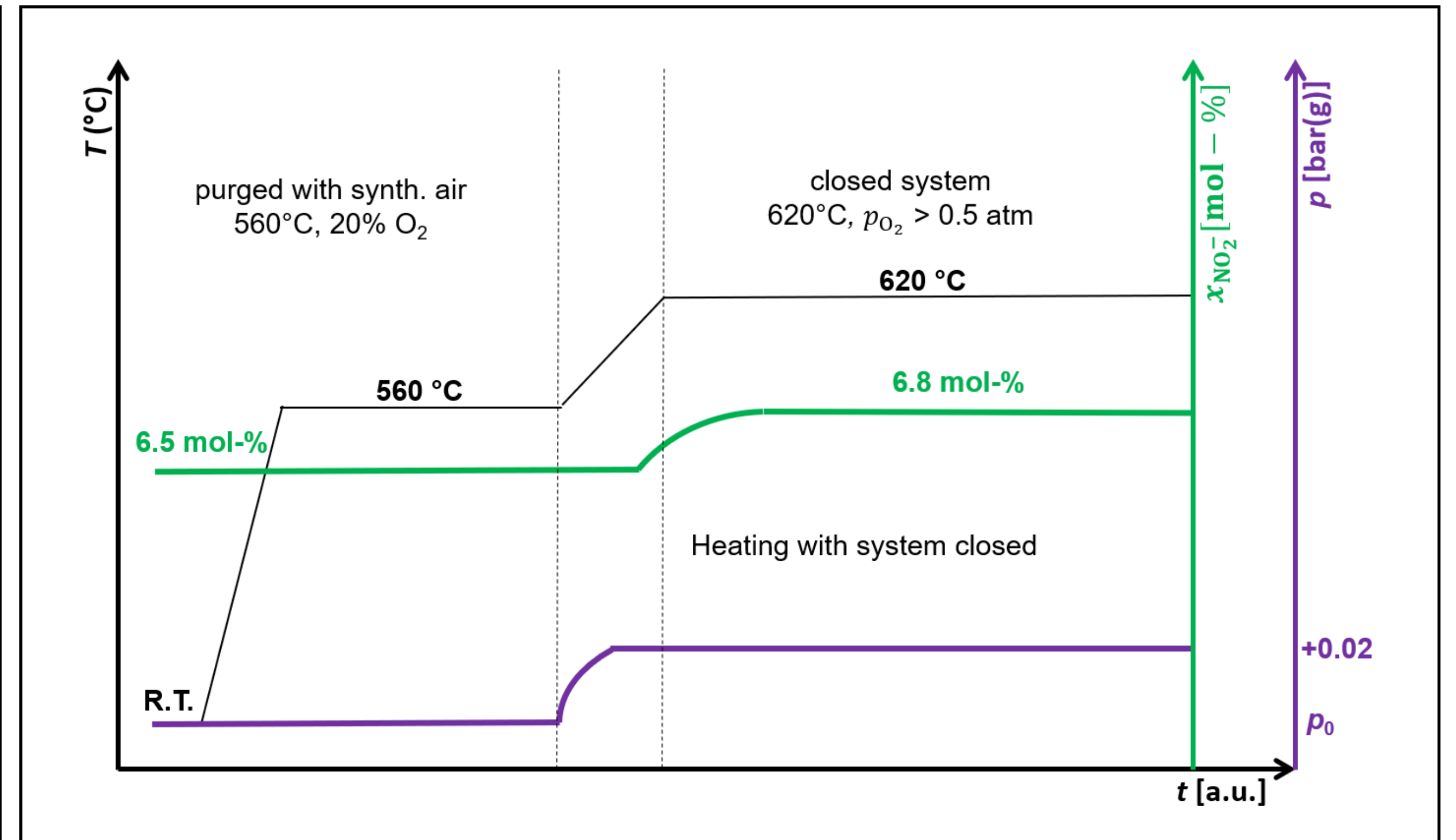
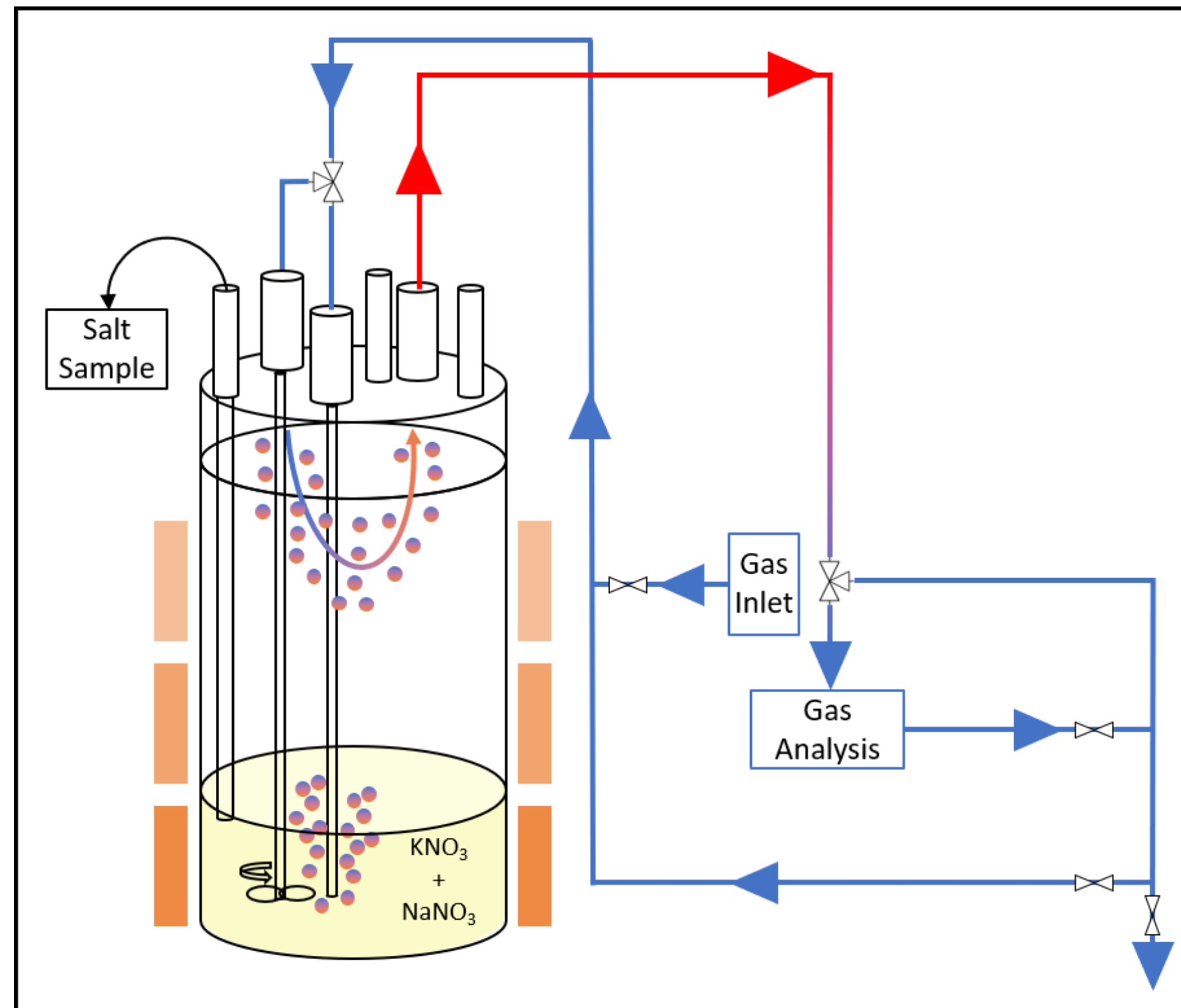
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Motivation

- Molten Solar Salt (60 wt-% NaNO₃, 40 wt-% KNO₃) is the state-of-the-art media in modern concentrated solar power plants (CSP) at up to 565 °C^[1]
- Volumetric storage capacity, heat-to-electric conversion efficiencies mostly rely on the applied hot tank temperature^[2,3]
- Thermal stability in terms of decomposition of the nitrate ions limits the maximum operating temperature
- Stability maintained up to 600 °C in a lab-scale by sealing the storage system^[4]
- No experiments on Solar Salt with temperatures beyond 565 °C in a larger scale (> 100 kg) known

Experimental Set-Up

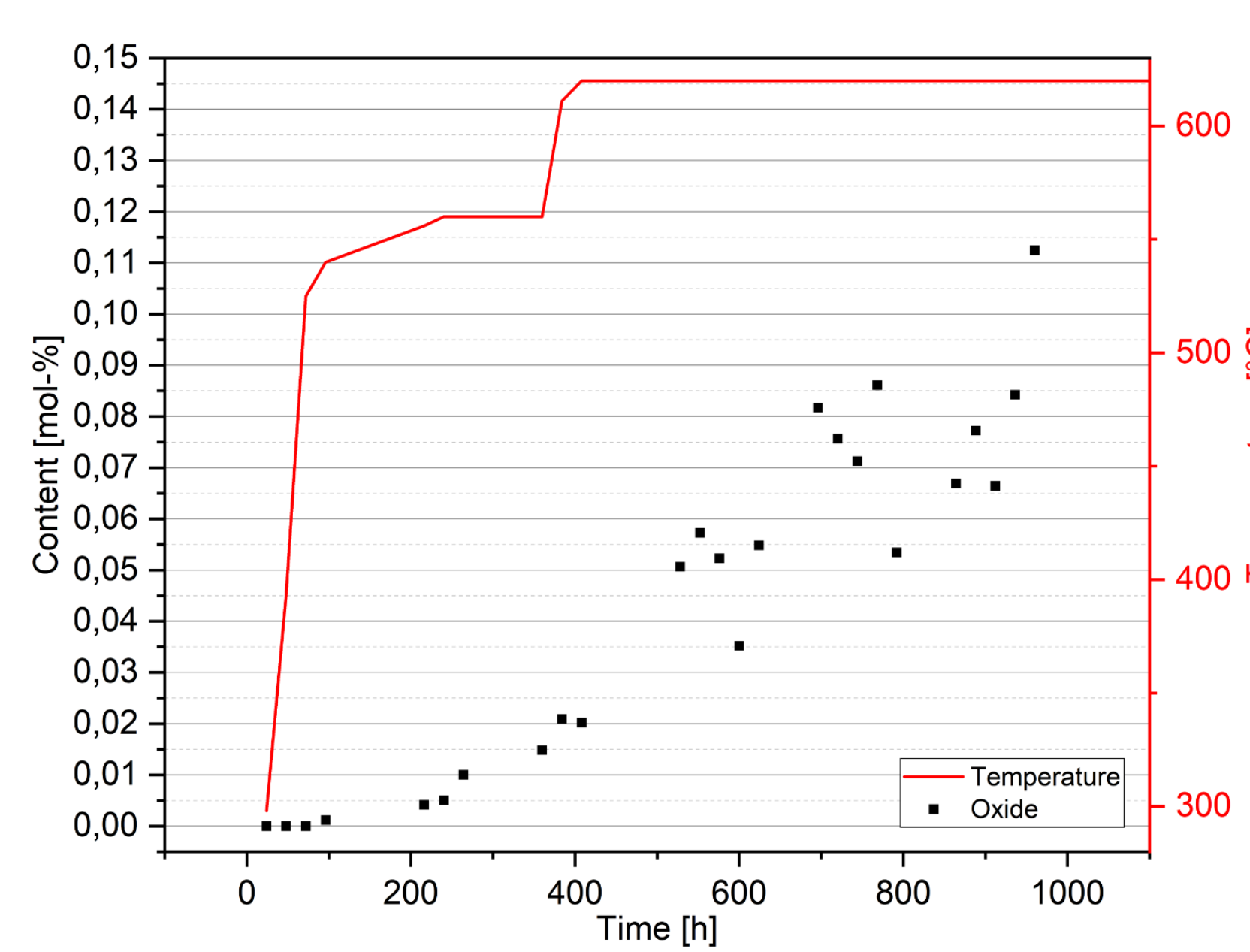
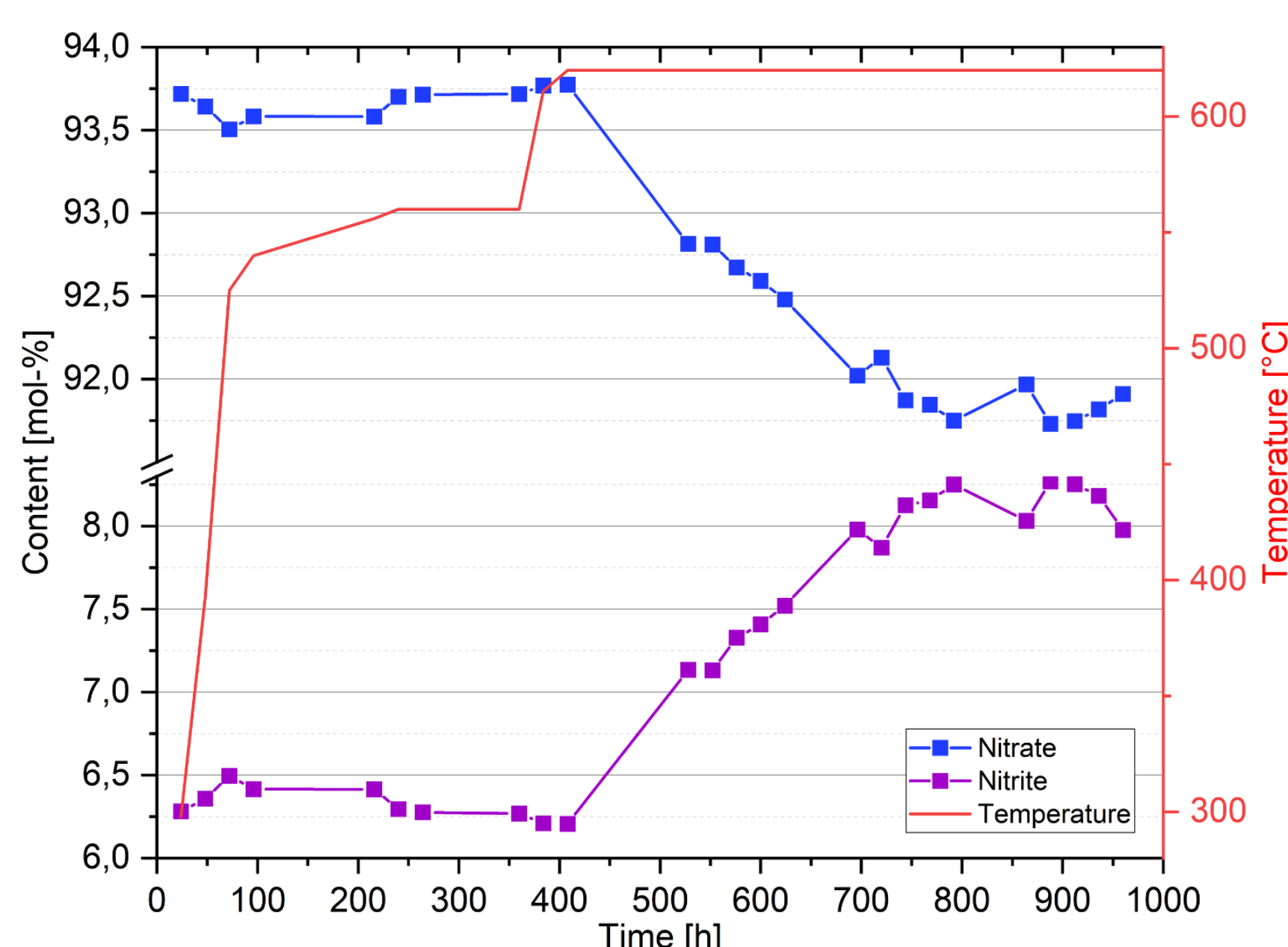


- 100 kg technical-scale hot tank with in situ gas analysis, temperature and pressure control
- $T_{\max} = 620 \text{ °C}$, $p_{\max} = 0.02 \text{ bar(g)}$, $t \approx 2000 \text{ h}$
- Starting salt composition: Solar Salt (60 wt-% NaNO₃, 40 wt-% KNO₃), 6.5 mol-% NaNO₂

- Post-analysis of salt samples with ion chromatography (IC) and titration → Ion composition of the molten salt
- Continuous gas analysis measurement → Monitoring of decomposition gases (NO, NO₂, O₂)

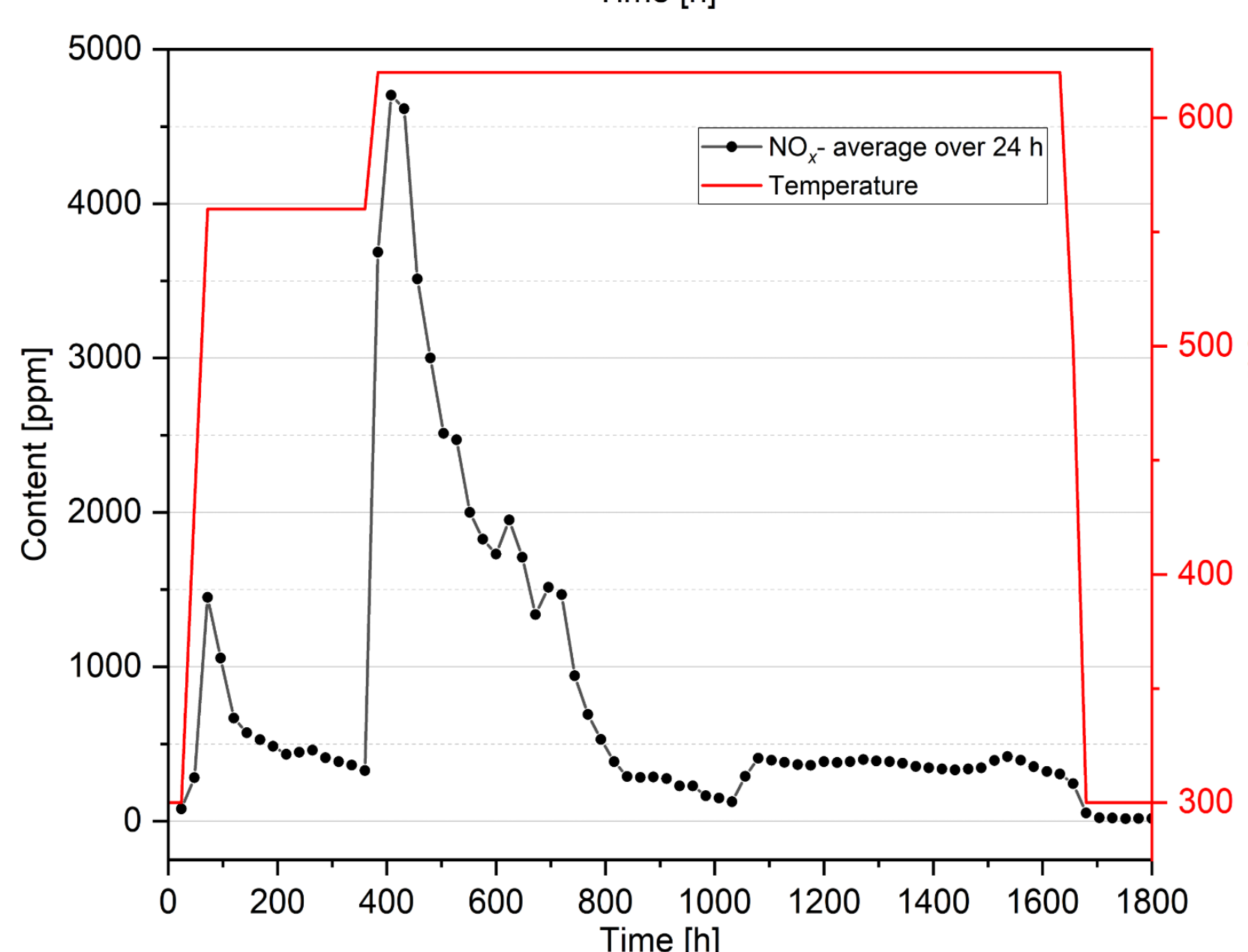
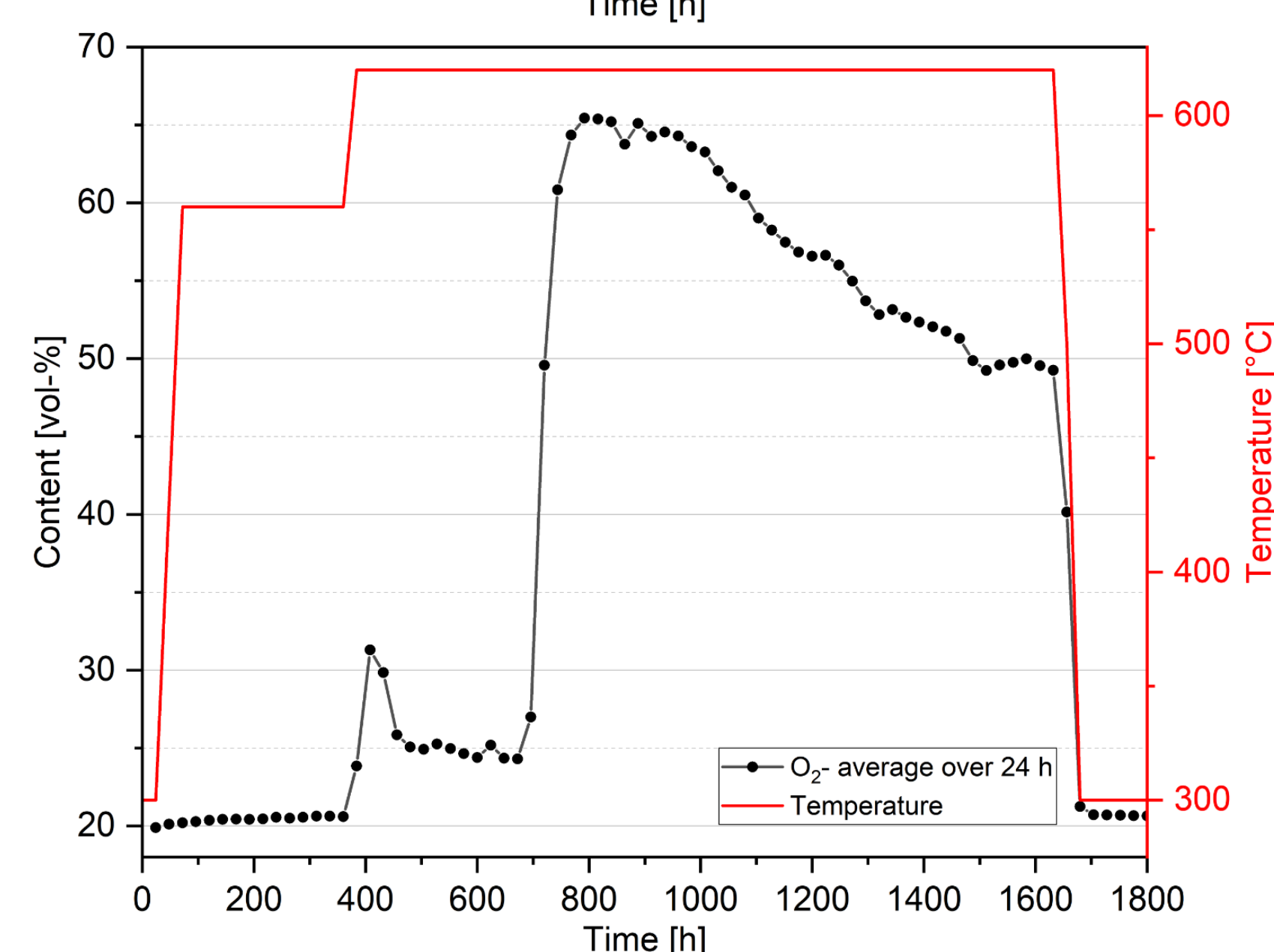
Results

Ion composition with regard to nitrate/nitrite at 300 – 620 °C obtained from IC measurements.



Oxide ion content at 300 – 620 °C obtained via titration with HCl_(aq) according to Warder.

Oxygen content of the gas phase in the furnace and gas system during the experiment in vol-%.



NO_x content of the gas phase in the furnace and gas system during the experiment in ppm.

Summary & Conclusion

- Operation of 100 kg hot tank VeraHTS A at $T_{\max} = 620 \text{ °C}$ successful
- Stable nitrate / nitrite content (92 mol-% / 8 mol-%) → chemical equilibrium reached after about 800 h
- Lower nitrate / higher nitrite content compared to 200 g lab-scale
- Stable oxide content at around 0.12 ppm
- Raising the temperature of molten Solar Salt beyond the state-of-the-art level while maintaining salt stability is possible

References

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