



# Modification of ZnO nanoparticles with silanes enables their application as anticancer agents

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To cite this article: Fernanda Marques, Aurel Tăbăcaru, Mariana Bușilă, Teresa Pinheiro & António P. A. Matos (2021) Modification of ZnO nanoparticles with silanes enables their application as anticancer agents, *Annals of Medicine*, 53:sup1, S32-S32, DOI: [10.1080/07853890.2021.1896916](https://doi.org/10.1080/07853890.2021.1896916)

To link to this article: <https://doi.org/10.1080/07853890.2021.1896916>



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Published online: 28 Sep 2021.



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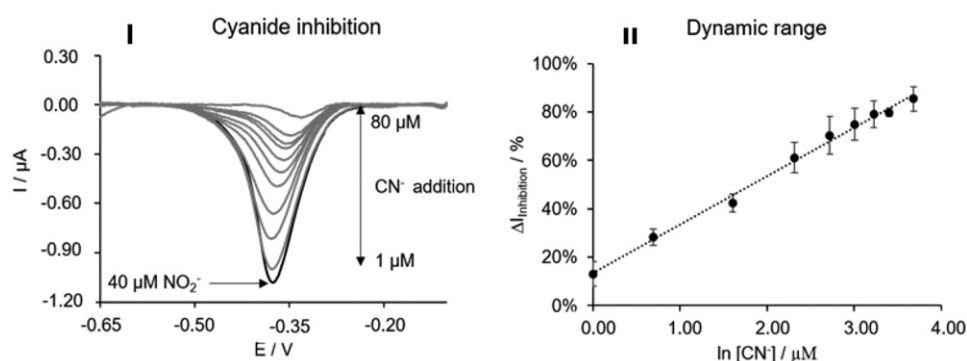
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**Figure 1.** (I) Square wave voltammograms with baseline correction of one of the WE used in the study. Enzymatic inhibition by cyanide among 1–80  $\mu\text{M}$  in the presence of 40  $\mu\text{M}$  of nitrite. (II) Calibration curve ( $y = 0.200x + 0.135$ ;  $R^2 = 0.995$ ) for  $\text{CN}^-$  quantification, with the dynamic range 1 to 40  $\mu\text{M}$ . Results are expressed as triplicate average with the corresponding standard deviation.

**Materials and methods:** Electrochemical measurements were carried out in a conventional electrochemical cell, composed by a three-electrode system. The reference was an Ag/AgCl electrode, and the counter electrode was a Pt wire. The working electrode (WE) was in-house made using a graphite lead with the ccNiR (from bacteria *D. desulfuricans* ATCC 27774; stored in 0.05 M phosphate buffer, pH 7.6) immobilised by drop cast at the WE surface. The electrochemical technique used was square wave voltammetry. Electrochemical cells contained 0.1 M KCl in 0.1 M Tris-HCl buffer (pH 7.6) as supporting electrolyte. Dissolved oxygen was removed by a biochemical system (GOx, catalase and glucose).

**Discussion and conclusions:** In Figure 1 we can observe the decrease in catalytic activity due to the presence of cyanide. The biosensor dynamic range comprises the maximum value imposed by the European Union, 1.92  $\mu\text{M}$  (98/83/EC directive), which does not happen with other cyanide biosensors [3]. Furthermore, a graphite lead cost 0.35€ and each of them can be split, allowing a very low-cost biosensor. Given that enzyme inhibition is reversible [4], the sensor can be used more than one time, if we optimise the enzyme immobilisation method.

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## Acknowledgments

The authors thank the support from REQUIMTE and the Applied Molecular Biosciences Unit-UCIBIO, financed by national funds from FCT/MCTES (UID/Multi/04378/2013) and co-financed by the ERDF under the PT2020 Partnership Agreement (POCI-01-0145-FEDER-007728). The authors also acknowledge financial support from Fundação para a Ciência e Tecnologia (PD/BD/109687/2015).

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DOI: 10.1080/07853890.2021.1896915

## Modification of ZnO nanoparticles with silanes enables their application as anticancer agents

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#### ABSTRACT

**Introduction:** Nanoparticles (NPs) are a wide class of materials that include particulate substances sized less than 100 nm. Inorganic ZnO NPs have found applications in several industrial fields such as the optical, electronic, pharmaceutical and cosmetics [1]. However, in many specific fields the applications are limited, since the particles tend to aggregate/agglomerate due to the hydrophilic nature of the surface. For potential clinical applications, surface modification of ZnO plays a crucial role in the biocompatibility of ZnO NPs [2]. Using organosilane modifying agents, improved hydrophobicity of the resulting ZnO NPs, induced by the non-polar terminal groups, can be achieved, and thus, very small and highly dispersed particles can be obtained. ZnO silanes were found to have antibacterial activity against several pathogens [3]. The NP size highly influenced the antibacterial activity, which increase with decreasing size. The ZnO NPs may induce bacterial cell membrane damage, resulting in bacterial cell death. Motivated by these findings and others reporting anticancer activity for pristine ZnO NPs [4], the aim of this study was to investigate the anticancer activity and mechanism of cell death of silane-modified ZnO nanoparticles against A2780 ovarian cancer cells and evaluate if for the cancer cells a correlation between size and activity was also observed.

**Materials and Methods:** The silane modified ZnO NPs were prepared by the addition of (3-glycidyloxypropyl) trimethoxysilane (GPTMS) as a surface modifier at 0% (G0) and 10% (G3) molar ratio of Si/Zn. The obtained NPs were characterised by high resolution transmission electron microscopy (HRTEM), X-ray diffraction and UV-Vis spectrometry. The cytotoxic activity in ovarian cancer cells were assessed by the MTT colorimetric assay. The morphological cellular alterations were visualised by electron microscopy (TEM).

**Results:** The silane modified ZnO NPs exhibited significant cytotoxic activity against A2780 ovarian cancer cells. The NP size, *ca.* 13 nm for G0 and *ca.* 3 nm for G3, highly influenced the cytotoxic activity, which increased with decreasing particle size,  $IC_{50}$ :  $\sim 100 \mu\text{g/mL}$ (G0) and  $\sim 30 \mu\text{g/mL}$ (G3). The ZnO silanes affected the cellular integrity by the induction of organelle damage evidenced by TEM.

**Discussion and conclusions:** Although preliminary, results indicate that ZnO silanes are interesting platforms to explore as anticancer agents. Studies on the ultrastructural level (TEM/SEM) are needed to understand their cytotoxic mechanism and to give clues on their potential targets.

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#### Acknowledgements

The authors acknowledge funding from the Fundação para a Ciência e Tecnologia (FCT) (UID/MULTI/04349/2013, UID/BIO/04565/2013) and LISBOA-01-0145-FEDER-007317.

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DOI: 10.1080/07853890.2021.1896916

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