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# **Low-grade metamorphism around the down-dip limit of seismogenic subduction zones: Example from an ancient accretionary complex in the Shimanto Belt, Japan**

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## ABSTRACT

Reactions involving clay minerals during low-grade metamorphism at the depth of an ancient accretionary complex in the Shimanto Belt, Kyushu, Japan, were studied by integrated transmission electron microscopy-energy dispersive X-ray spectroscopy and X-ray diffraction analyses of the bulk rock and clay fraction. The analyzed metasediment (the Kitagawa unit) contain an incipient sub-horizontal slaty cleavage. Illite crystallinity data and mica *b* dimensions indicate that the conditions of metamorphic deformation were anchizone–epizone grade and intermediate pressure. Cleavage formation was linked to two reactions involving clay minerals: (1) the recrystallization of  $1M$ –dominant matrix mica, inherited from the original sedimentary fabric, into thick, defect-free  $2M_1$  packets along cleavage planes; and (2) the formation of chlorite from  $7 \text{ \AA}$  berthierine. Balanced equations among the clay phases, based on compositional data and their relative abundance, suggest that the decomposition of matrix mica resulted in the formation of paragenetic mica and chlorite along the cleavage planes, without significant elemental outflux. Although a modal increase in phyllosilicates is not indicated by the data, the growth of chlorite and mica along cleavage planes may have a large influence on the rheological properties of a décollement and may be related to the occurrence of the seismic–aseismic transition at  $\sim 350 \text{ }^\circ\text{C}$ .

**Key words:** accretionary prism; low-grade metamorphism; mica; chlorite; X-ray diffraction; transmission electron microscopy

## 1. Introduction

The mechanical properties of sediments entering a subduction zone are progressively modified by dewatering, lithification, and deformation (Cowan, 1982; Byrne, 1984; Taira et al., 1992; Moore et al., 2001; Moore and Saffer, 2001; Hashimoto et al., 2006; Kimura et al., 2007). Such modification along a plate boundary fault (i.e., a décollement) has a major influence on several aspects of accretionary prism dynamics, such as wedge geometry, internal structure, and seismicity (Wang and Hu, 2006; Hyndman, 2007; Kimura et al., 2007). In general, a décollement can be divided into three segments in terms of its mechanical regime: shallow and deep aseismic portions, and an intermediate seismic region (Hyndman et al., 1997; Hyndman, 2007). The upper of the two seismic–aseismic transitions occurs at temperatures of ~150 °C and is believed to be related to dewatering of subducted sediments due to tectonic loading and diagenetic dehydration reactions (e.g., smectite-illite transition and opal-quartz transition), which allow increase in effective normal stress of the décollement (Moore and Saffer, 2001; Spinelli and Saffer, 2004). Moreover, active carbonate, clay and quartz cementations around this temperature condition may cause sliding instability of the sediments (Moore and Saffer, 2001). On the other hand, onset of the aseismic behavior above ~350 °C is inferred to be related to the mechanical properties of subducted sediments subjected to low-grade metamorphism (Hyndman et al., 1997; Hyndman, 2007).

Although the nature of deep crustal structure in subduction zones remains debated, several seismic reflection surveys (particularly those undertaken at the Cascadia margin) suggest that the décollement widens with increasing depth to be several kilometers thick, forming reflection banding in a seismic profile (e.g., Nedimovic et al., 2003; Calvert et al., 2006). A candidate

material for such reflectors is plastically deformed metasediments affected by low-grade metamorphism, with a structure similar to that of mylonite zones. Therefore, phase transitions in subducted sediments, particularly those for phyllosilicates, are likely to influence aseismic faulting along deep décollements (Hyndman et al., 1997; Hyndman, 2007). Nevertheless, the brittle–ductile transition is rarely exposed at the surface, as most accretionary prism rocks do not reach the transition or are overprinted by late-stage higher-grade metamorphism. Consequently, little is known of the prograde evolution of mineral assemblages and their reaction pathways around the down-dip part of the seismogenic zone, which may control the mechanical transition from seismic to aseismic behavior.

The Kitagawa unit, exposed on the easternmost side of the Kyushu Shimanto Belt, Japan, provides an opportunity to trace the progress of low-grade metamorphism within a subduction zone, because this unit records the highest metamorphic grade of the rocks within the Shimanto Belt (320–330 °C; Kondo et al., 2005) and comprises a complete succession ranging from shale/sandstone to intensely cleaved metasediment (Raimbourg et al., 2009). A structural analysis of the unit indicates that vertical loading, probably in a setting around the plate interface, produced the ubiquitous subhorizontal cleavage defined by networks of recrystallized chlorite and mica (Raimbourg et al., 2009).

Illite crystallinity (IC) measurements on micas of the Kitagawa unit (Mukoyoshi et al., 2007, 2009) yield lower values at a site located close to the Nobeoka Thrust ( $0.24 \Delta^\circ$ ) compared with a site located more than 5 km from the thrust ( $0.31 \Delta^\circ$ ). Because the intensity of deformation in this unit shows an increase toward the Nobeoka Thrust (Raimbourg et al., 2009), the observed trend in IC values is inferred to reflect the extent of mica growth, although additional analyses

are needed to characterize the complete sequence of metamorphic processes that occurs during the formation of a mature metapelite.

The aim of the present study is to clarify several aspects of low-grade metamorphism of the Kitagawa unit, including its metamorphic setting, evolution of the mineral assemblage and bulk chemistry, and the reaction pathways of clay minerals. To this end, we subjected bulk samples and the clay fraction (<2  $\mu\text{m}$ ) to quantitative analyses by X-ray diffraction (XRD) and high-resolution transmission electron microscopy (HR-TEM), for samples that record varying degrees of deformation. Based on these data, we discuss the influence of low-grade metamorphism on the aseismic behavior of deep décollements.

## **2. Geological setting**

The Shimanto Belt is an ancient accretionary complex that extends for 800 km along the Pacific side of southwest Japan (Fig. 1a). The belt is divided into northern and southern sub-belts, separated by the Aki Tectonic Line in the Shikoku and Kii regions, and by the Nobeoka Thrust in Kyushu. In eastern Kyushu, the Kitagawa Group was exhumed north of the Nobeoka Thrust, emplaced over the southern Hyuga Group (Fig. 1b). Metamorphic mineral assemblages suggest that the northern belt was metamorphosed in the greenschist facies (Toriumi and Teruya, 1988), whereas the southern belt was metamorphosed in the prehnite–pumpellyite to zeolite facies.

The Kitagawa unit, a turbiditic sequence of alternating sandstone- and mudstone-rich layers that contain Eocene radiolarian fossils (Ogawaguchi et al., 1984), records two stages of deformation: an early stage of horizontal contraction, characterized by folding and thrusting, was overprinted by the development of a pervasive slaty cleavage during vertical shortening

(Raimbourg et al., 2009). The slaty cleavage is more pronounced in the deeper levels of the Kitagawa unit; i.e., closer to the Nobeoka Thrust.

Vitrinite reflectance values (%Rm) obtained for the Kitagawa unit range from 5.33 to 5.63 (Kondo et al., 2005), corresponding to maximum paleotemperatures of 320–330 °C based on the method proposed by Sweeney and Burnham (1990). These temperatures are similar to the estimated formation temperature of metamorphic chlorite (Raimbourg et al., 2009).

The Hyuga Group, located south of the Nobeoka Thrust, possesses a shale-dominant *mélange* texture. Vitrinite reflectance data indicate temperatures of 250–270 °C (%Rm = 2.58–2.75; Kondo et al., 2005), which are 50–80 °C lower than those obtained for the Kitagawa unit. This difference in temperature suggests that the Nobeoka Thrust was active as an out-of sequence thrust (Ohmori et al., 1997), responsible for exhuming the unit from deep within the accretionary complex (Raimbourg et al., 2009). Faulting is thought to have occurred during ~40–48 Ma, as inferred from illite K–Ar and zircon fission track ages (Hara and Kimura, 2008).

### **3. Experimental procedure**

Samples with various densities of cleavage planes were collected from the Kitagawa unit and were classified into the three groups proposed by Raimbourg et al. (2009): non-cleaved samples (NM) do not contain slaty cleavage; slightly cleaved samples (SIM) possess both cleavage and bedding; and strongly cleaved samples (StM) contain an intensively developed cleavage with little sign of the original bedding fabric. A total of 13 samples (6 NM samples, 3 SIM samples, and 4 StM samples) were analyzed (Table 1).

Representative portions of each sample were ground using a vibrational mill for 2 minutes under dry conditions for bulk XRD analyses. Each powder, mixed with high-purity corundum (AX-5H, Hinomoto Kenmazai Co., Japan) as an internal standard, was mounted on an XRD glass holder in a side-load manner to minimize any preferred alignment of phyllosilicates. The patterns were recorded using a Rigaku Rint-2000 with monochromatized  $\text{CuK}\alpha$  radiation at 40 kV and 30 mA, with  $1^\circ$  divergence and anti-scattering slits, and a 0.3 mm receiving slit. The  $2\theta$  range of  $2^\circ$ – $65^\circ$  was scanned in step scan mode for 4 s, at steps of  $0.04^\circ$   $2\theta$ . The RockJock program, developed by Prof. D.D. Eberl (Eberl, 2003), was used for quantitative analyses of rock composition. The XRD patterns were fitted using reference patterns for quartz, albite, calcite, pyrite,  $1M$  and  $2M_1$  micas, and trioctahedral chlorite, as stored in the program.

To obtain the clay size fraction ( $<2 \mu\text{m}$ ), rock chips were gently crushed by hand and dispersed ultrasonically in distilled water. The clay fraction was then separated by centrifugation and used for oriented analyses of clay minerals in air-dried (AD) and ethylene-glycolated (EG) states. The Newmod program (Reynolds, 1985) was used to interpret the XRD patterns. IC was measured based on the width of the  $10 \text{ \AA}$  peak at half height, under the AD condition (Kübler, 1968). The  $b$  dimension of mica was determined using the d-value of the mica 060 peak calibrated by the 221 peak of quartz at  $1.541 \text{ \AA}$  (Merriman and Peacor, 1999).

The TEM was operated at 200 kV to observe fine structures in minerals (JEM-2010, JEOL, Japan). Some TEM images were digitized using a CCD camera and processed to remove noisy contrast due to amorphous materials on the specimen surface, using a Wiener-filter (Marks, 1996; Kilaas, 1998) developed by K. Ishizuka (HREM Research, Inc.), as implemented in Gatan Digital Micrograph version 3.10.0. The clay fraction powders were dispersed on a holey carbon



film supported by a Cu-grid for chemical analysis by energy dispersive X-ray spectrometer (EDX) in the TEM (EX-24025JGT, Jeol). To minimize the loss of alkali elements, which is significant in the analysis of clay minerals (van der Pluijm et al., 1988), beam diameter was extended to 200 nm and the counting duration was reduced to 30 seconds. Each grain was planarly exposed such that its  $c^*$  axis was rotated by  $20^\circ$  from the beam axis toward the EDX detector. Quantitative chemical analyses were performed with  $k$ -factors for O, Na, Mg, Al, Si, K, Ca, Ti, Mn, and Fe determined from standard samples of biotite, albite, olivine, and hornblende. The atomic ratio was obtained by correcting for the effect of X-ray absorption by the specimen under the criterion of charge neutrality (Van Cappellen and Doukhan, 1994), using a program developed by one of the authors (T. Kogure).

## 4. Results

### *4.1. Deformation structures within sediment of the Kitagawa unit*

Microstructures and deformation stages of the Kitagawa unit have been described in detail by Raimbourg et al. (2009), and are briefly summarized here. The protolith of the low-grade metamorphic rocks is a turbiditic unit that was buried deep within the Shimanto paleo-accretionary prism, down to the base of the inner wedge. The protolith consists of alternating sandstone- and mudstone-rich layers with thicknesses of 2 mm to 10 cm. The sediment consists mainly of detrital quartz and minor plagioclase clasts embedded in a fine-grained matrix. The sandstone-rich layers contain a larger proportion of clasts than do the mudstone-rich layers. A small amount of detrital mica grains is scattered throughout the rock, with the grains being larger in the sandstone-rich layers (Fig. 2a, b, and b'). The microcrystalline

matrix is too fine-grained for its mineralogical composition to be determined optically.

Macroscopically, the deformation is characterized by the development of a subhorizontal slaty cleavage that crosscuts and disrupts the original sedimentary fabrics. Structural and magnetic (AMS) analyses indicate that the kinematics of the associated deformation is mainly coaxial shortening, with a compressive axis oriented close to vertical (Raimbourg et al., 2009). At the microscopic scale, the cleavage consists of a dense network of mica and chlorite surrounding quartz and feldspar grains (Fig. 2c and c'). Both of these phyllosilicates have a shape-preferred orientation that defines the foliation, as indicated by their simultaneous extinction when rotating the microscope stage under cross-polarized light (Fig. 2c'). Micas are large (up to 100  $\mu\text{m}$  long), while chlorite grains consist mostly of very thin, entangled fibers.

#### ***4.2. XRD analyses of the clay fraction***

The Kitagawa unit contains a simple and uniform mineral assemblage through the succession, comprising quartz, plagioclase (albite), and phyllosilicates with accessory pyrite, Fe oxides, and rare calcite. However, as shown below, the abundance, composition, and crystallographic features of the phyllosilicates are progressively modified at different stages of cleavage formation. Figure 3 shows typical XRD patterns of the  $<2 \mu\text{m}$  size fraction for NM, SIM, and StM samples. The most common phase is dioctahedral mica. Table 1 summarizes the IC values of the analyzed samples. IC values for NM samples show a wide scatter from 0.20 to 0.41  $\Delta^\circ$ , while those for SIM samples are 0.22–0.29  $\Delta^\circ$ , and those for StM samples are within a limited range around 0.2  $\Delta^\circ$ . The lowest of the IC values obtained for NM samples ( $\sim 0.2 \Delta^\circ$ ) probably reflect the presence of detrital mica, whereas a sharpening of the 001 reflection through SIM and

StM indicates the thickening of mica grains, which is ascribed mainly to recrystallized mica along cleavage planes. These IC values indicate that the low-grade metamorphism of this unit corresponds to the anchizone–epizone grade (the values that define the boundary between the diagenetic zone and the anchizone, and between the anchizone and epizone, are 0.42 and 0.25  $\Delta^\circ$ , respectively; Merriman and Peacor, 1999). Although the present IC values are not interlaboratory standardized, comparable values have been reported from samples obtained near the Nobeoka Thrust (0.24  $\Delta^\circ$ ; Mukoyoshi et al., 2009) and from the northern part of the unit (0.31  $\Delta^\circ$ ; Mukoyoshi et al., 2007), for which the IC indexes were measured using the JIC (Japanese Illite Crystallinity Standard) scale, which was developed for paleothermal analyses of accretionary complexes (Hara and Kimura, 2003). JIC values are converted to the widely accepted CIS (Crystallinity Index Standard; Warr and Rice, 1994) by the following relation (Hara and Kimura, 2003):  $IC (CIS) = 1.55 \times IC (JIC) - 0.07$ . For instance, the value of 0.24  $\Delta^\circ$  (JIC) obtained at the southernmost part of the unit, within intensely cleaved rock (Mukoyoshi et al., 2009), is converted to 0.3  $\Delta^\circ$  (CIS), corresponding to the metamorphic conditions of the boundary between the low and high anchizone (Merriman and Peacor, 1999). Metapelites of this grade commonly contain an incipient cleavage (Merriman and Peacor, 1999), which is consistent with the rocks of the present study area.

The *b* length of mica has been used as a semi-quantitative barometer of crystallization conditions (Sassi and Scolari, 1974; Guidotti and Sassi, 1986). An increase (decrease) in *b* length is caused mainly by a greater (smaller) extent of phengite substitution (Guidotti et al., 1989). In the present study, the *b* values determined from mica 060 reflections range from 9.008 to 9.033 Å (Table 1). The relatively deformed samples yield lower *b* values (Table 1), suggesting a

reduction of phengitic content in recrystallized mica toward more aluminous muscovite.

The relative abundance of different polytypes in micas was qualitatively determined by analyzing random mounts of the clay fraction using XRD (Fig. 4). According to Caillère et al. (1982) and Dalla Torre et al. (1994),  $I(112)$  of  $1M$  and  $I(025)$  of  $2M_1$  are good indices for estimating the relative abundance of these polytypes because of their low sensitivity to the effect of mica preferred orientation or overlap with the feldspar peak. The intensity relations between the two reflections indicate a progressive transition from  $1M$  to  $2M_1$  from NM through SIM and StM (Fig. 4).

In addition to mica, chlorite occurs as a clay phase. Although chlorite does not appear in some of the NM samples, the 14 Å and 7 Å peaks become more distinctive with increasing metamorphic grade (Fig. 3). This observation indicates the progressive growth of chlorite within the slaty cleavage. However, the intensity of the 7 Å reflection relative to that of 14 Å is much higher than that of a theoretical chlorite XRD pattern, using a structural formula determined by the present EDX analyses (see Section 4.5.). This discrepancy is probably caused either by peak overlap with another 7 Å phase or the interstratification of 7 Å layers within the chlorite stacking; the latter is more probable, given the broad and diffuse 001 reflection (Reynolds, 1988). Such a broad 001 peak is more common in NM than in SIM and StM samples (Table 1).

A candidate material for the 7 Å phase in metapelites is kaolin minerals. We tested for the occurrence of kaolin by performing an acid treatment test (2 hours immersed in 1 N HCl at 60 °C). The complete disappearance of the 7 Å peak after the test suggests that the mineral is more likely to be Fe-bearing berthierine, which is highly reactive with acid. The occurrence of chlorite–berthierine, as well as the fine structures of micas, was further examined by TEM (see

Section 4.4.).

### ***4.3. Quantitative bulk-rock composition determined by RockJock***

The nature of prograde changes in the bulk-rock composition was quantitatively analyzed using RockJock (Eberl, 2003), which can be used to estimate the relative proportions of different polytypes in illite–muscovite (Aldega and Eberl, 2005). Accessory minerals (e.g., Fe oxides and rutile) occur in trace amounts (<0.5 wt%) and were excluded from the analyses. Table 1 lists the obtained relative abundances (wt%). Using this program, the total weight percent is a measure of analytical quality, because the content of each mineral is estimated independently with respect to the internal standard of corundum (Eberl, 2003). In the present study, the deviation from a total of 100% is within  $\pm 2\%$ , except for one sample (Table 1), indicating the high quality of the analyses. The data for the sample that yielded a low total (HN24) were recalculated to 100%, to enable a comparison with the other samples.

Figure 5a shows a clear trend of  $1M$ -to- $2M_1$  conversion in micas with progressive deformation; NM samples show a wide range of  $1M/(1M + 2M_1)$  values, ranging from 0.3 to 1.4, but values are below 0.1 in StM samples. This finding is consistent with XRD analyses of the clay fraction (Fig. 3). This reaction is concomitant with the growth of chlorite (Fig. 5b). Although chlorite is scarcely distributed and is sometimes absent in NM, it makes up approximately 40% of the mica content in StM samples. Such chlorite growth is also qualitatively compatible with the results of XRD analyses of the  $<2 \mu\text{m}$  fraction (Fig. 3).

Variations in the total clay content, which have a strong influence on rheology (e.g., Brown et al., 2003), are shown as the summation of  $1M$  mica,  $2M_1$  mica, and chlorite (Fig. 5c). The clay

content does not show a large increase with cleavage formation (showing an average change of 1 wt% from NM to StM; Table 1), but the large variation in NM samples is reduced to a uniform composition with cleavage development. The initial variation in NM samples probably reflects variations in grain size (i.e., the ratio of clay- to silt-size grains) within the bulk samples, which are subsequently homogenized in terms of rock composition and fabric by the recrystallization and neocrystallization of phyllosilicates along cleavage planes, thereby diminishing the original sedimentary-related variations.

#### ***4.4. Microtextural and crystallographic analyses of phyllosilicates by TEM***

The analyzed metapelite contains basically two forms of micas (Raimbourg et al., 2009): matrix mica inherited from the original sedimentary fabric, and recrystallized mica that defines the cleavage. TEM images showed that the matrix mica has variable packet thickness, from <30 nm to >100 nm, with several strain features such as voids and layer terminations (white arrows in Fig. 6a). In places, we observed relatively ordered, selected-area electron diffraction (SAED) patterns with dominantly  $1M$  stacking, although accompanied by packets showing two-layer ( $2M_1$ ) periodicities (Fig. 6b). Stacking disorders are visible, even in crystals that yield ordered ED patterns (Fig. 6b).

Recrystallized mica that occurs along cleavage planes has a relatively homogeneous thickness (mostly >100 nm) and has fewer defects than the matrix mica (Fig. 6c). Although  $1M$  was occasionally present,  $2M_1$  was more commonly observed.

Chlorite occurs as well-defined packets (>100 nm thick) and is oriented parallel to the cleavage (Fig. 7a). Lattice fringe images reveal the presence of 7 Å berthierine (Fig. 7b and c),

which is generally intergrown with 14 Å chlorite layers at the monolayer level rather than forming discrete packets. In some cases, berthierine constitutes the larger fraction of a single packet (Fig. 7b). The interstratification of berthierine and chlorite is commonly observed in diagenetic to low-grade metamorphic rocks (Lee et al., 1984; Ahn and Peacor, 1985; López-Munguira and Nieto, 2000) and in synthetic samples (Velde, 1973; James et al., 1976). The interstratified phases are considered to represent a transient state toward a mature chlorite. In Fig. 7d, two berthierine layers (TOTO) are transformed to a single chlorite layer (TOT and one brucite-like B layer) by tetrahedral inversion (white bracket in the figure), strongly suggesting a solid-state conversion, as discussed in detail by Xu and Veblen (1996).

#### ***4.5. TEM–EDX chemical analyses***

The chemical compositions of chlorite in the present samples were measured by Raimbourg et al. (2009) using electron microprobe analyzer (EMPA), and were employed to estimate the formation temperature of chlorite based on the chemical thermometer developed by Vidal and Parra (2000). However, the grain size of clays is generally too small to be measured satisfactorily by EMPA; thus, we re-examined the composition of mica and chlorite by EDX in TEM (Tables 2 and 3). The Si content of micas in NM samples ranges from 3.0 to 3.5 if normalized to  $O_{10}(OH)_2$ , but the range of values is smaller in SIM and StM samples (3.1–3.3), which have a high interlayer charge; i.e.,  $Na + K = \sim 1.0$  (Fig. 8a). The interlayer composition is characterized by a low Na content ( $Na/(Na + K) < 0.15$ ) for all deformation stages (Table 2). In addition to a low interlayer deficiency, SIM and StM samples have low Mg + Fe contents (Fig. 8b) and a small Fe/(Fe + Mg) ratio (Fig. 8c) compared with NM. The depletion in mafic elements is consistent

with the low phengitic content inferred from the lower *b* dimensions in SIM and StM samples (Table 1).

Figure 9 shows the chemical compositions of chlorite determined by TEM–EDX, along with data measured by EMPA (Raimbourg et al., 2009). The two techniques yield comparable results, indicating that element contamination during EMPA is not significant. There is almost no compositional variation among the samples of different metamorphic grade.

## 5. Discussion

### *5.1. Clay mineral reactions in the Kitagawa unit during anchizone–epizone grade metamorphism*

The exposure of a series of lithologies ranging from shale/sandstone with well-preserved sedimentary fabrics to moderately or intensely deformed slates enables a discussion on the change in mineralogy that occurs during cleavage formation in the Kitagawa unit. As shown above, clay minerals in the unit are restricted to two types: dioctahedral micas and trioctahedral chlorite. Although many factors (e.g., temperature, pressure, bulk rock composition, and fluid–rock ratio) may affect the mineral assemblage and its evolution in low-grade metamorphic rocks (Merriman and Peacor, 1999), Merriman (2002) showed that the clay mineralogy of a slate belt in the British Isles is strongly dependent on the regional geotectonic setting. Many different species of clay minerals can be observed in extensional basins, including the aluminous kaolin–pyrophyllite and sodium mica–paragonite series, as well as dioctahedral K-mica and the trioctahedral 2:1 series. This variation in clay mineralogy is due to intense seawater circulation or the intrusion of volcanic rocks, especially during sedimentation (Merriman, 2002). In contrast,

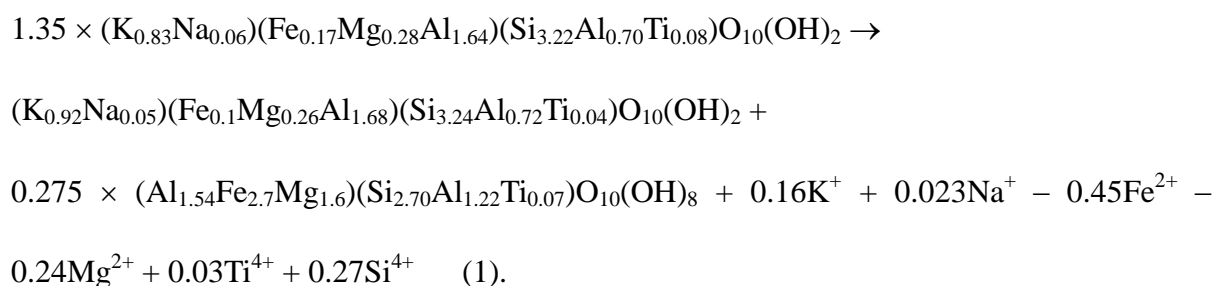


the restricted circulation of seawater or absence of high-Na/K fluids may reduce such diversity in convergent basins (Merriman, 2002). This may explain the poor diversity of clay minerals in the present samples, as well as the absence of kaolin and pyrophyllite, and the low sodium content in the mica interlayer.

Despite the invariant K-dominant interlayer composition, the crystallographic features of mica are modified by deformation-induced recrystallization. Bulk-rock XRD analyses and microscopic observations by TEM suggest that the matrix mica, which is  $1M$ , is converted to  $2M_1$  packets along cleavage planes. In addition, strain-related structures are healed to form thick, defect-free packets, which is also indicated by the low IC values obtained for deformed samples. Such crystallographic maturation has been widely recognized in anchizone-grade metapelites that show the development of incipient slaty cleavage (Lee et al., 1986; Li et al., 1994).

Unlike the characteristic conversion of micas, the origin of chlorite is somewhat problematic. Chlorite in metapelites is commonly derived from a prograde reaction that originates from trioctahedral smectite via corrensite (e.g., Hillier, 1993; Schmidt and Livi, 1999). However, the formation of Fe–Mg-rich smectite within an accretionary prism is generally localized to specific lithologies such as incorporated oceanic crust (e.g., Kameda et al., in review), which is not found in the Kitagawa unit. Previous studies have noted that chlorite forms via the conversion of dioctahedral smectite to illite (Hower et al., 1976; Ahn and Peacor, 1985), because of the typical depletion of mafic elements in illite relative to the starting smectite (Drief and Nieto, 2000). In the present case, however, the chlorite content is generally low prior to deformation (Fig. 5b) and it is locally absent, despite the completion of smectite-to-illite conversion, which typically occurs at  $<150$  °C. This reaction may have contributed to some of the chlorite in the NM samples, but

the progressive growth of chlorite from the SIM to StM samples (Fig. 11b) requires another process, possibly the decomposition of matrix mica of phengitic composition to form paragenetic mica and chlorite along cleavage planes. The likelihood of this reaction is evaluated with reference to the following balanced equation. We roughly consider that the increment of chlorite from NM to StM samples (by 10 wt% on average) and the total mica content in StM samples (22 wt%) reflect the effects of recrystallization. Based on the chemical composition determined by EDX (Tables 2 and 3), the molar weight of mica and chlorite per structural formula in StM samples is 395 and 636, respectively. Then, the molar ratio of the final product between the total mica content in StM samples (22 wt%) versus the increase in chlorite content from NM to StM samples (10 wt%) is estimated to be  $22/395:10/636 = 1:0.275$ . Assuming the amount of  $\text{Al}_2\text{O}_3$  to be constant throughout the reaction, the molar ratio of the starting material (mica in NM samples) required to complete the reaction is derived from the following equation:



This equation indicates that about 30 wt% ( $= 22 \text{ wt\%} \times 1.35$ ) of the starting material is needed, which is compatible with the average mica content in NM samples (31 wt%). The mass balance results, based on Al conservation, support the validity of reaction (1). However, this reaction requires the presence of  $\text{Mg}^{2+} + \text{Fe}^{2+}$ . The origin of these elements is uncertain, but may have been other mafic minerals such as detrital biotite, pyrite, or Fe-oxides.

When this reaction proceeds, berthierine may occur as a precursor. As mentioned above, the

XRD patterns of the oriented clay preparations show a weak, broad 14 Å reflection relative to pure chlorite (001), which is ascribed to interstratification between berthierine and chlorite. Hillier and Velde (1992) used NEWMOD (Reynolds, 1985) to calculate the one-dimensional XRD pattern of interstratified chlorite–berthierine with a variable mixture ratio, and obtained a near-linear correlation between berthierine content (5%–35%) and width ( $^{\circ}2\theta$ ) at half the height of the 14 Å peak, independent of chemistry and crystallite thickness. Our experimental patterns show a sharpening of the 14 Å peak from NM samples (0.25–0.37 of width at half peak height; average, 0.282) to SIM (0.265) and StM (0.192) (Table 1). Based on fig. 7 in Hillier and Velde (1992), NM is estimated to contain >5% berthierine. The reduction in berthierine content is interpreted to reflect post-precipitated conversion from berthierine layers to mature chlorite in a solid-state manner, as observed by TEM (Fig. 7d).

### ***5.2. Element flux during cleavage formation***

The mass flux of elements during slate formation has been extensively studied in recent decades. For example, studies of kinematic strain (e.g., Write and Platt, 1982; Beutner and Charles, 1985; Brandon et al., 1991) have shown a significant volume loss (30%–60%) in slates, but Erslev (1998) noted that such estimates are based on a misinterpretation of complicated, heterogeneous strain and metamorphic reactions, and emphasized the near-isochemical nature of the slates, as also indicated by geochemical signatures (Shaw, 1954; Ague, 1991; Wintsch et al., 1991). The extent of element flux may be recorded in the resultant mineral assemblage, but few studies have provided quantitative data on slate composition. Ward and Gomez-Fernandez (2003) applied the Rietveld method to Spanish roofing slates, and assessed the applicability of

quantitative analyses with respect to the bulk chemical composition, although the authors did not quantify the degree of modal change during slate formation.

Lee et al. (1986) performed a semi-quantitative XRD analysis of mudstone/slate samples and compared the relative intensity of the constituent minerals, indicating an approximately homogeneous rock composition across the lithologic transition. The quantitative data of the present study indicate a near-constant modal composition, considering the average values for samples from each deformation stage (Table 1). For example, the content of quartz, which is one of the main mobile elements, deviates by just ~4 wt% from 40 wt%; thus, it is unlikely that volume outflux occurred on the order of several tens of weight percent.

Van de Kamp (2008) suggested that measurable amounts of SiO<sub>2</sub> (~20 wt%) are released by the illite-to-muscovite conversion. Equation (1) indicates that the production of 1.0 mole of cleavage-parallel mica leads to 0.26 moles of SiO<sub>2</sub> release, corresponding to 1 wt% at most (= 22 wt% × 0.26 × 60/395). Thus, in the present case, negligible amounts of SiO<sub>2</sub> were released via reactions involving clay minerals.

Raimbourg et al. (2009) performed XRF analyses of bulk samples (the same samples as those analyzed in the present study), revealing that the concentration of immobile TiO<sub>2</sub> is inversely related to the concentration of mobile SiO<sub>2</sub>. The authors proposed that any outflow of SiO<sub>2</sub> would have been caused primarily by quartz dissolution. However, in terms of the relation between SiO<sub>2</sub> content and deformation, their data show a high degree of scatter. This result, combined with the constant modal composition of samples of variable metamorphic grade, suggests that SiO<sub>2</sub> outflow occurred prior to cleavage formation, during diagenesis at shallow levels (e.g., Renard et al., 1999; Yang, 2000; Kawabata et al., 2007); thus, high fluxes of

elements would not have occurred during anchizone metamorphism.

### ***5.3. Geotectonic setting of the Kitagawa unit and implications for the weak and aseismic nature of the deep décollement***

Previous studies have estimated the peak paleotemperature of the Kitagawa unit. Based on vitrinite reflectance values, Kondo et al. (2005) estimated thermal maturation at 320–330 °C. Application of the chlorite geothermometer developed by Vidal and Parra (2000) yielded an average temperature of ~272 °C (Raimbourg et al., 2009), although the data show a wide scatter corresponding to temperatures of 200–400 °C. Mukoyoshi et al. (2007) and Underwood et al. (1993) derived regression equations for IC–temperature relations applicable to paleothermal analysis of the Shimanto accretionary complex. The IC value of 0.24  $\Delta^\circ$  obtained near the Nobeoka Thrust (Mukoyoshi et al., 2009) corresponds to temperatures of 304±30 °C and 330 °C, respectively. The use of our IC value for StM samples (0.2  $\Delta^\circ$ ) yields slightly higher temperatures of 312±30 °C and 347 °C. Considering these estimates, a metamorphic temperature of 300–350 °C is an appropriate estimate for the Kitagawa unit.

Although there are few direct constraints on the peak burial depth, Raimbourg et al. (2009) estimated a minimum burial depth of ~6 km based on a high geothermal gradient of 50 °C/km in the Eocene, influenced by the subduction of a young, hot oceanic plate (Sakaguchi, 1999). This high regional geotherm may be overestimated.

The *b* dimensions of micas in the analyzed samples show a narrow range of 9.01–9.03 Å (9.02 Å on average; Table 1), and StM samples yield values on the lower side of this range (~9.01 Å). According to the regression relation between cell parameters and the composition of

$2M_1$  mica (Guidotti et al., 1989), a  $b$  dimension of  $\sim 9.01$  Å is estimated to correspond to a Mg + Fe content of 0.4 in the case of an interlayer composition of  $\text{Na}/(\text{Na} + \text{K}) < 0.15$ , which is consistent with the mica composition determined independently by EDX (Table 2). The consistency of the two parameters indicates that the  $b$  dimension is a reliable measure of metamorphic conditions, as suggested by Abad et al. (2003). The  $b$  dimensions obtained in the present study, which range from 9.01 to 9.03 (Table 1), are within the values of 9.0 and 9.04 assigned by Guidotti and Sassi (1986) to the lower/upper limit of the intermediate-pressure facies series for a geothermal gradient of 25–35 °C/km. Although the thermal structure of a subduction zone may show some longitudinal variations rather than being isothermal (e.g., Nankai, Watanabe et al., 1970; Yamano et al., 1984, 2003), the Kitagawa unit may have experienced burial in an area with a geothermal gradient of 25–35 °C/km, at least at the time of mica formation. At this time, the minimum burial depth of the Kitagawa unit would have been at least 8.5 km, further supporting the proposal by Raimbourg et al. (2009) that the rock has been transported far below the frontal contractional setting of the prism (4–5 km thick in the case of modern complexes), where it was subjected to vertical compaction (Raimbourg et al., 2009).

It is uncertain how closely the Kitagawa unit approached the ancient décollement, but temperatures of 300 °C or more suggest a vertical separation of less than 2 km (for a geothermal gradient of 25–30 °C/km), if the unit was located just above the area where the 350 °C isotherm intersects the décollement. This distance (between the Kitagawa unit and the décollement) increases (decreases) as the location of the unit is shifted landward (seaward).

Seismic reflection surveys of the Cascadia margin (Nedimovic et al., 2003; Calvert et al.,

2006) suggest that a deep aseismic décollement has a banded reflection structure, extending regionally at depths of 5–8 km from the top of the subducting slab. These studies proposed that the structure corresponding to this reflector is a ductile shear zone composed of foliated metasediment. It is likely that subhorizontal slaty cleavage within the Kitagawa unit would produce such a seismic structure.

Previous sliding experiments on crustal rock (granite) indicate that the seismic–aseismic transition occurs at ~350 °C, possibly related to pressure solution creep of quartz (Blanpied et al., 1991, 1995). This observation may explain the down-dip limit of seismogenic subduction zones (Hyndman et al., 1997). However, the sediments analyzed in the present study are rich in phyllosilicates, meaning that it is uncertain whether the rheology of the subducted sediments is controlled by quartz deformation. In the case of the Kitagawa unit, we consider that chlorite growth would have had a marked effect on the bulk-rock rheology, because of the ready activation of (001) glide, giving rise to plastic deformation even below 300 °C (Bons, 1988). The stable (i.e., aseismic) and low-friction behavior of chlorite (Ikari et al., 2009) may be related to such intracrystalline deformation. Although the chlorite content in the present study is <12 wt% within intensely cleaved samples, Collettini et al. (2009) experimentally showed that even a small amount of phyllosilicates (in their case, 5% talc) may control the mechanical behavior of a fault zone if they form a dense network, thereby maintaining the long-term shear strength at a reasonably low level. In the present case, the occurrence of a well-organized network of micas and chlorite may explain the aseismic behavior of the deep décollement.

## 6. Conclusion

Our mineralogical study on the Kitagawa unit of the Shimanto accretionary prism showed that the growth of paragenetic mica and chlorite during anchizonal-epizonal low-grade metamorphism contributed to the development of a pervasive slaty cleavage. The reactions proceeded without large elemental flux or modal fluctuations in phyllosilicate content. The mechanical transition from seismic to aseismic behavior along the deep décollement is attributable in part to the progressive development of subhorizontal mica and chlorite networks, similar to those observed at the surface.

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## Figure captions

Table 1. Crystal-chemical parameters and quantitative rock compositions determined by the program RockJock

Table 2. Chemical compositions of micas normalized to  $O_{10}(OH)_2$

Table 3. Chemical compositions of chlorite normalized to  $O_{10}(OH)_8$

Fig. 1. Geological map of the Kitagawa unit in the northern Shimanto Belt, Kyushu, Japan. The gray level (from light to dark) assigned to the Kitagawa unit corresponds to the intensity of deformation (disruption of bedding and the development of a metamorphic foliation).

Fig. 2. Microstructural evolution from the incipient development of a horizontal foliation (a, b, and b') to intensely deformed domains (c and c'). Cleavage planes are defined by networks of chlorite and mica aligned parallel to the foliation (c, c'). The pairs of images (b–b' and c–c') show the same site under plane- and cross-polarized light, respectively.

Fig. 3. Typical XRD patterns for oriented preparations of the  $<2 \mu\text{m}$  fraction in the EG state. Chl=chlorite.

Fig. 4. Typical XRD patterns for random preparations of the  $<2 \mu\text{m}$  fraction. Black arrows indicate the 112 and 025 peaks for the  $1M$  and  $2M_1$  polytypes, respectively.

Fig. 5. Evolution of the  $1M$ -to- $2M_1$  ratio of micas ( $1M/(1M + 2M_1)$ ) (a), chlorite/mica (b), and total clay content (micas + chlorite) as a function of metamorphic deformation. The weight percent amount of each constituent was determined by RockJock (Eberl, 2003).

Fig. 6. (a) TEM image of a mica packet in an NM sample containing pervasive defect structures. White arrows indicate the sites of layer terminations. (b) Rotationally filtered high-resolution

TEM (HRTEM) image of a thick mica packet in an NM sample, showing a periodic two-layer ED pattern (inset). Pairs of layers marked by black arrows are arranged in the  $1M$  stacking manner, within areas of  $2M_1$  stacking. (c) Thick, well-defined packets ( $>100$  nm) of mica in an StM sample, showing a periodic two-layer ED pattern (inset).

Fig. 7. (a) Thick packet of chlorite in an StM sample. (b) Lattice fringe image of a packet dominated by  $7 \text{ \AA}$  berthierine (white brackets). The black bracket shows a chlorite packet; arrows indicate isolated chlorite layers intergrown with berthierine packets. (c) Rotationally filtered HRTEM image of a chlorite packet containing  $7 \text{ \AA}$  berthierine layers (white arrows); a magnified image is shown in (d). TOT = tetrahedral–octahedral–tetrahedral 2:1 unit layer. B = brucite-like interlayer sheet. Tetrahedral inversion (white bracket) causes a layer transition from berthierine (top) to chlorite (bottom).

Fig. 8. Chemical compositions of micas plotted as a function of Si. (a) Interlayer cations, total (Na + K). (b) Mg + Fe. (c) Fe/(Mg + Fe) ratio. Data are normalized to  $O_{10}(OH)_2$ .

Fig. 9. Atomic ratios of Al–Fe–Mg in chlorite. Also shown are data obtained by EMPA (Raimbourg et al., 2009).

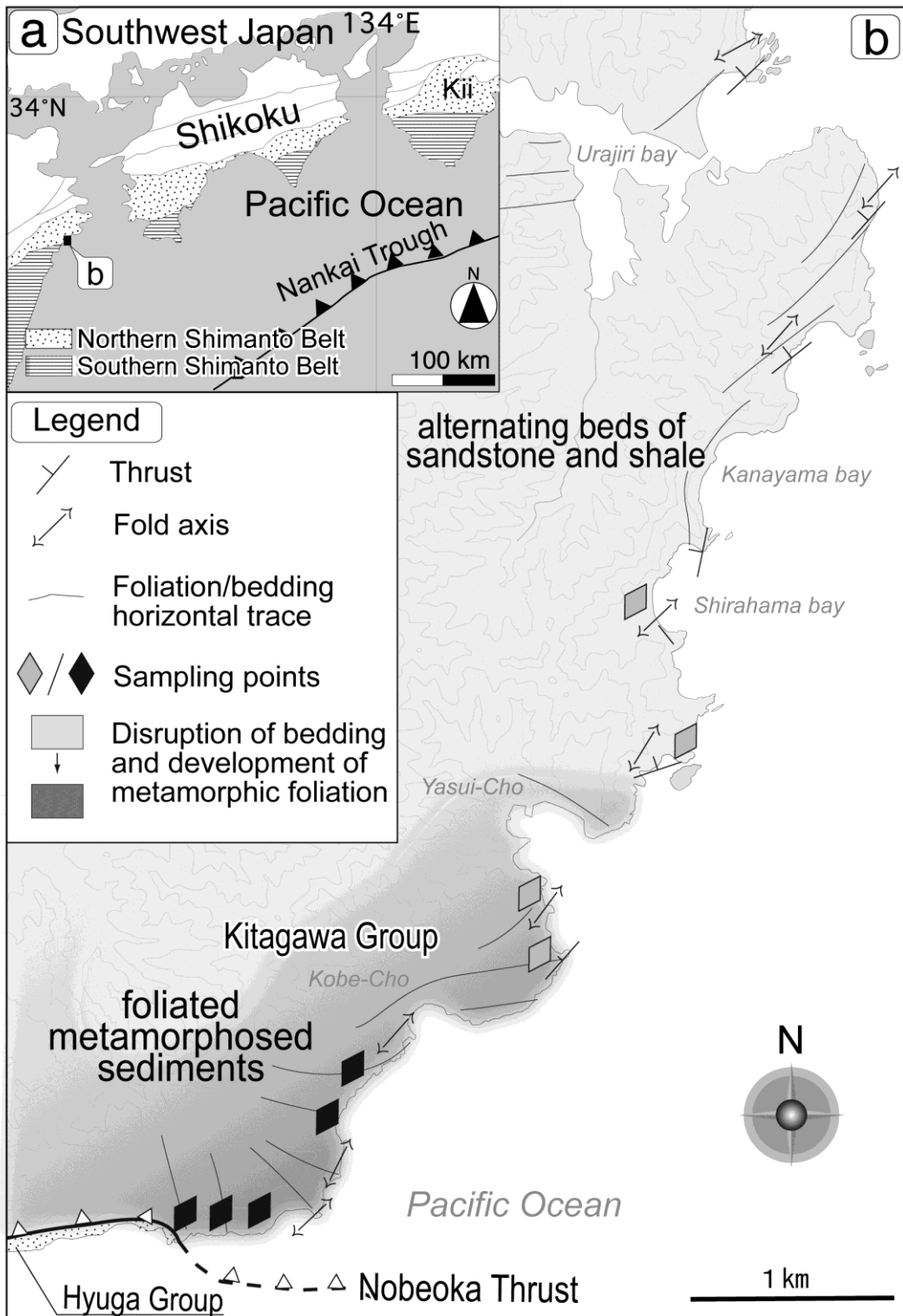


Figure 1

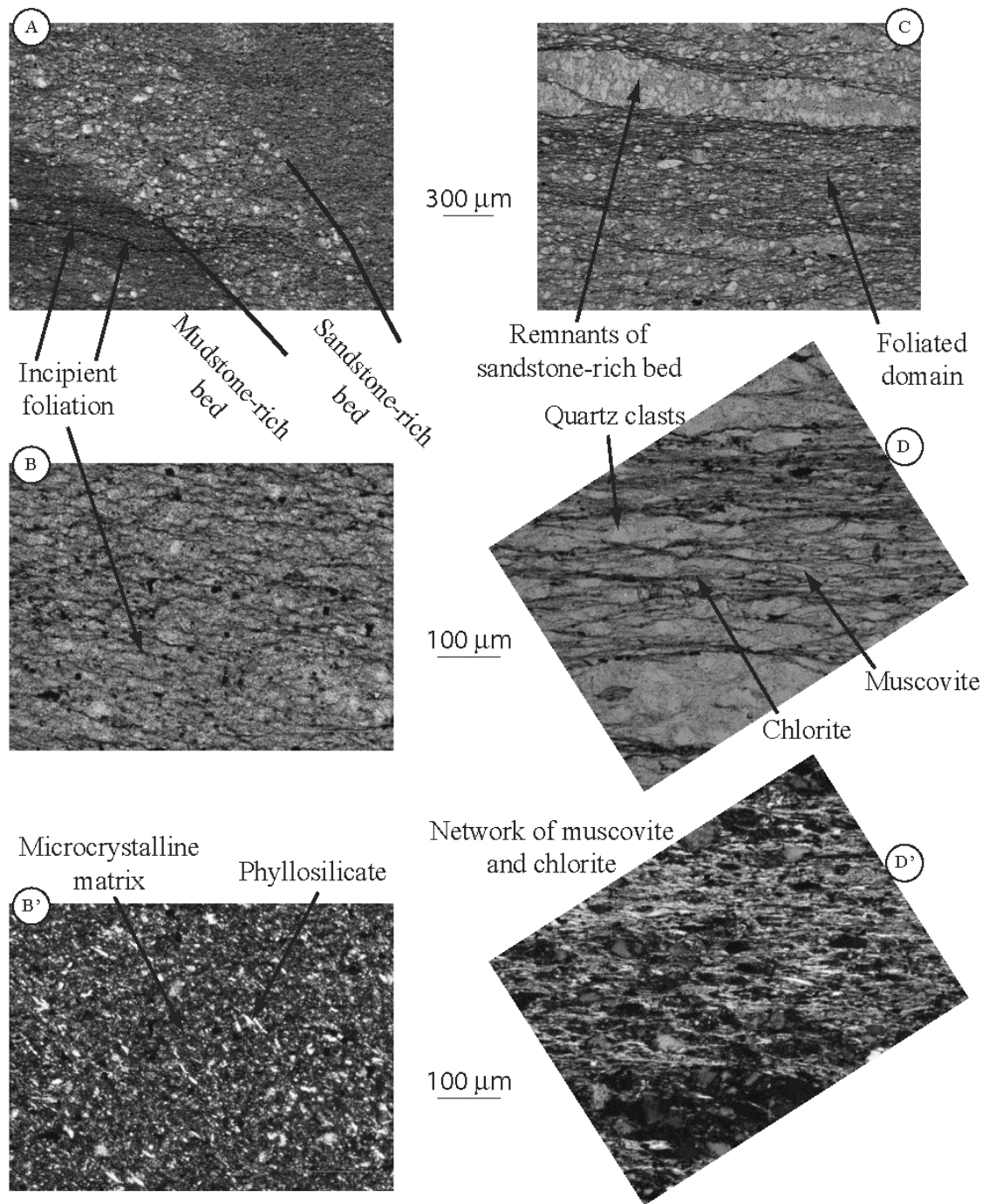


Figure 2

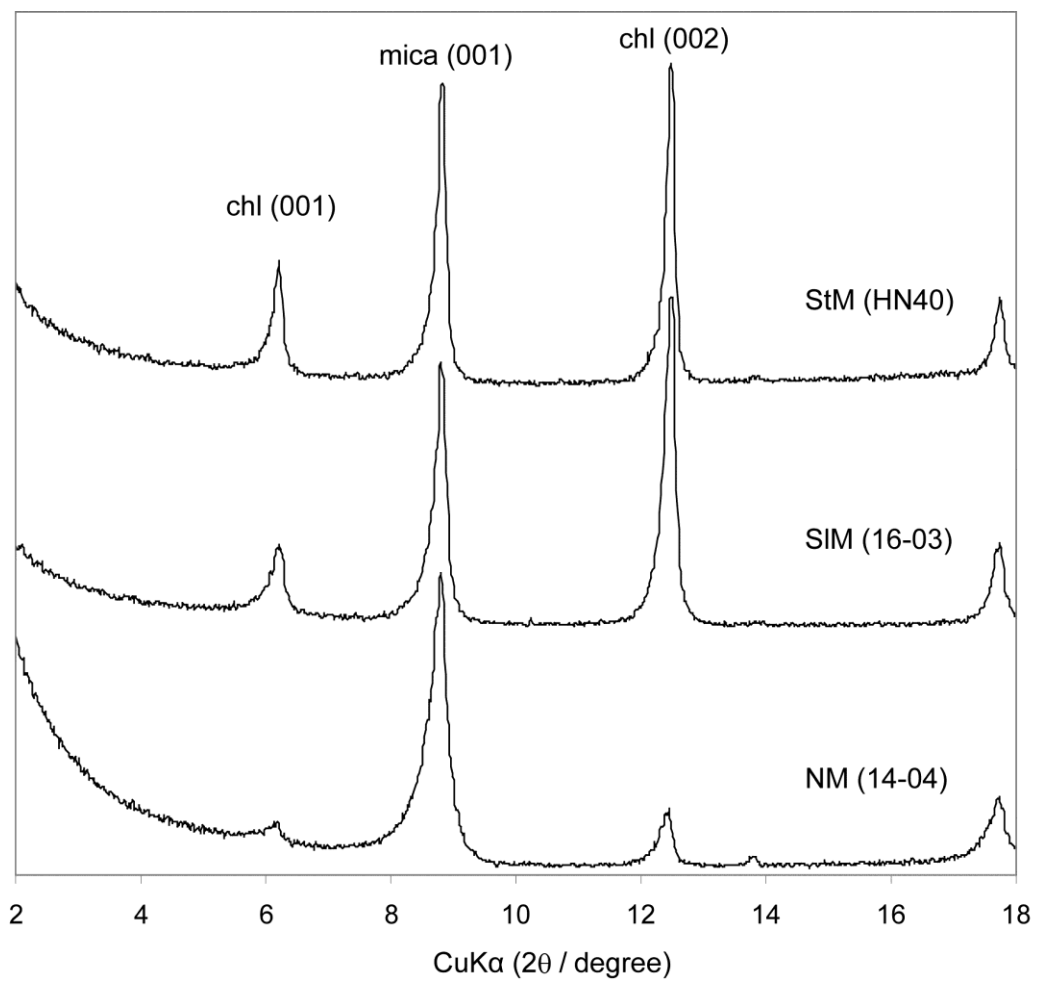


Figure 3

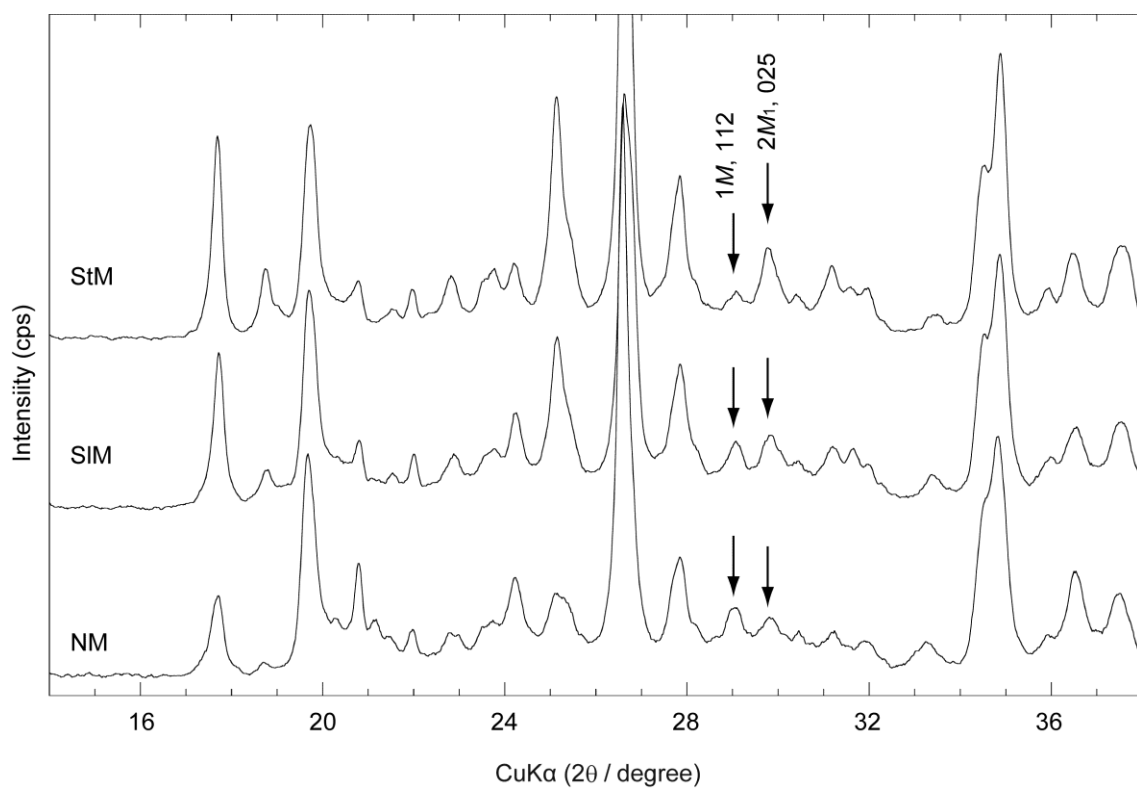
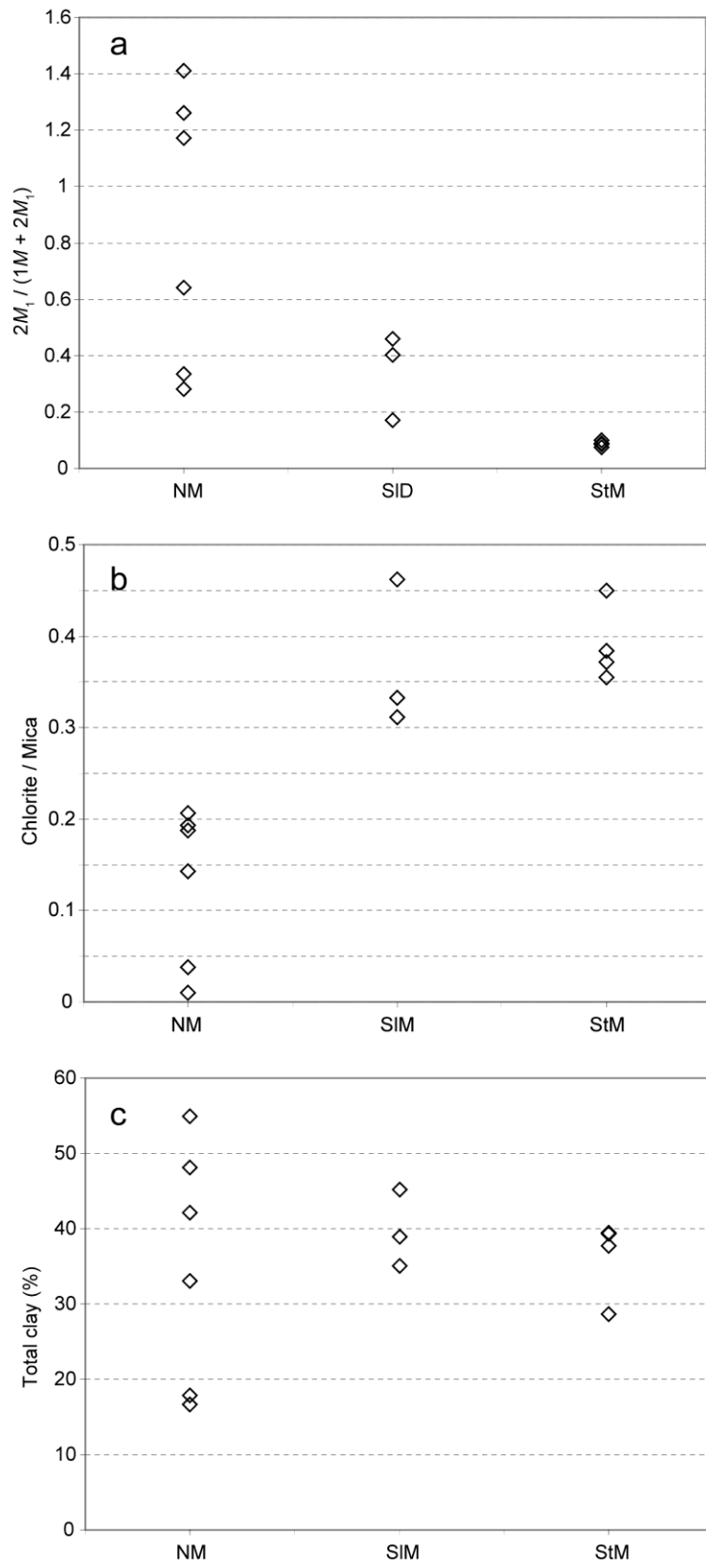


Figure 4





Figure

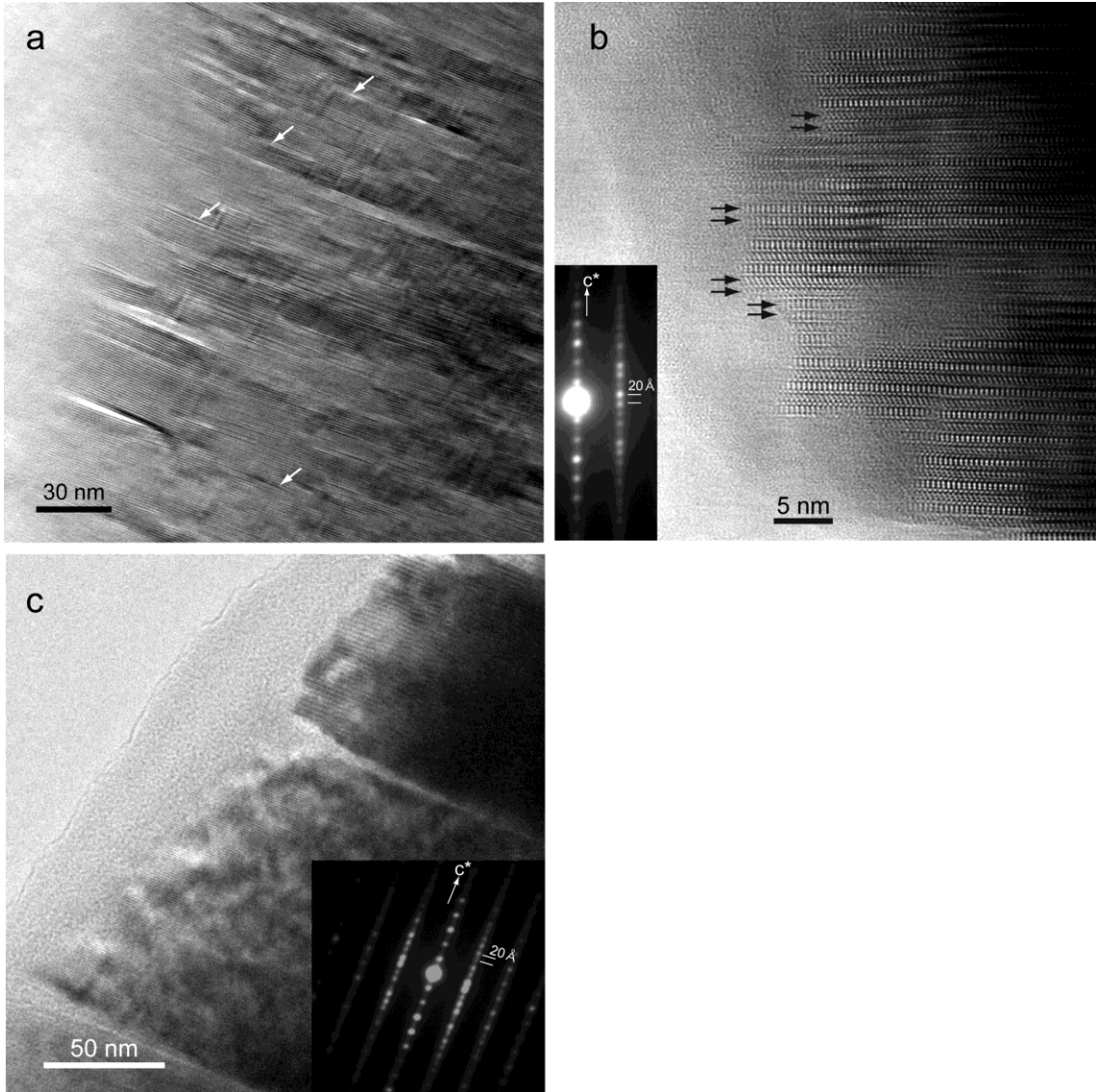


Figure 6

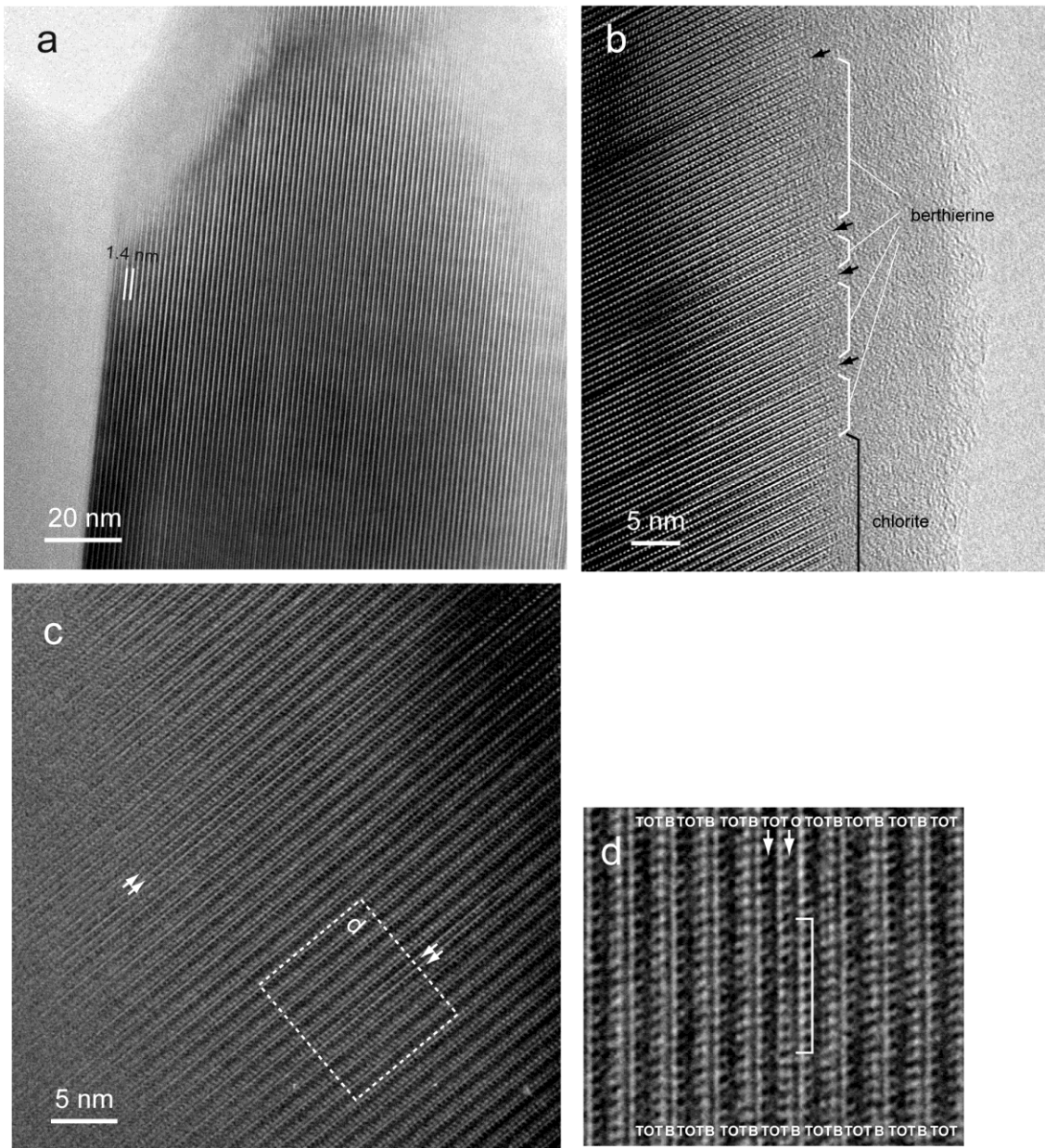


Figure 7

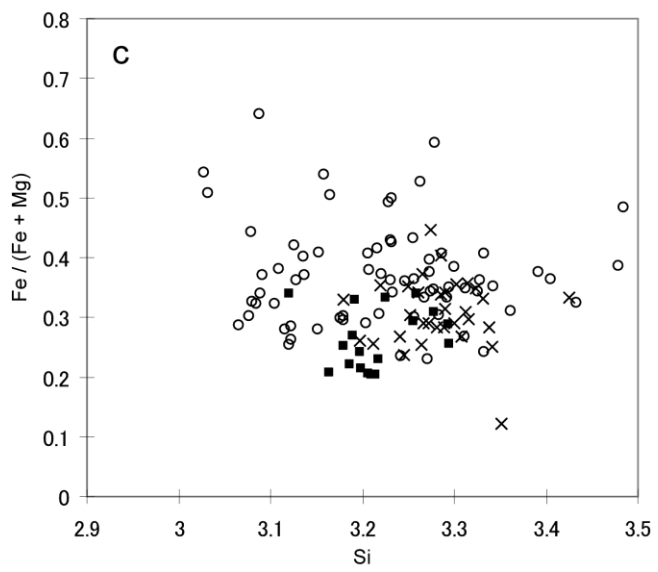
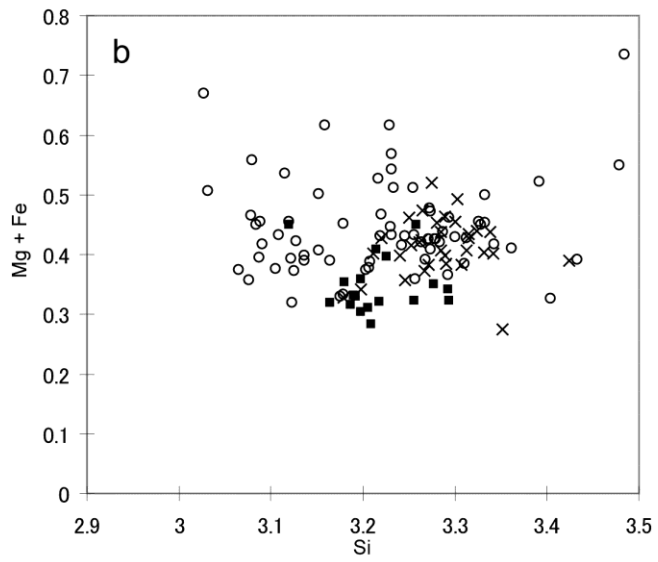
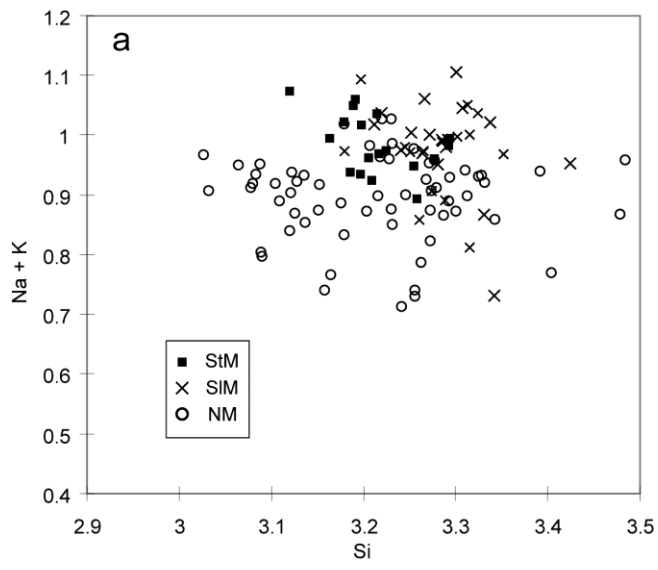


Figure 8

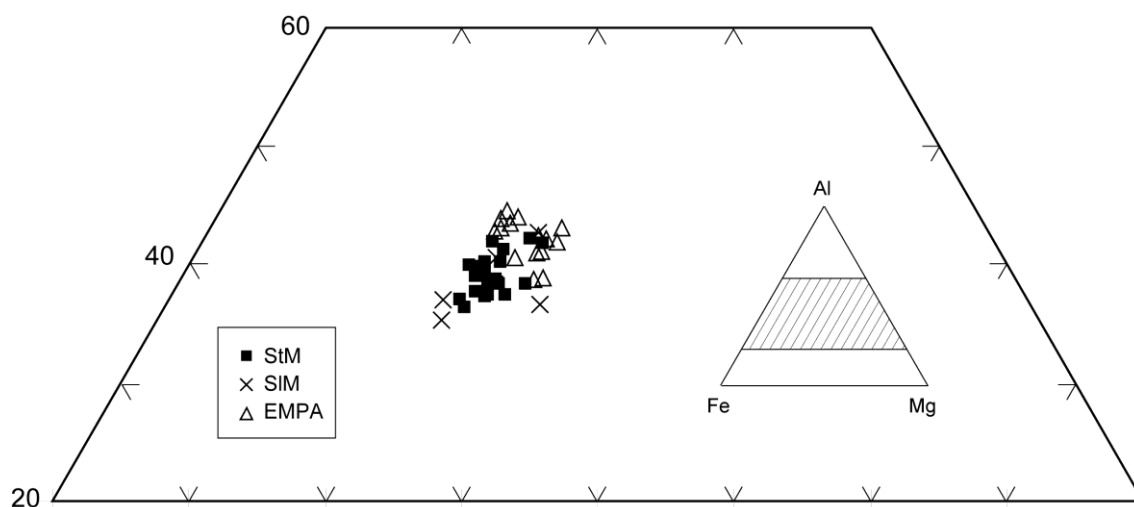


Figure 9

	NM							SIM				StM				
	13_06	14_04	HN18	HN20	HN23	HN24	avg.	16_03	16_05	HN26	avg.	16_07	HN34	HN40	HN41	avg.
IC ( $\Delta^\circ$ )	0.208	0.410	0.191	0.243	0.245	0.345	0.274	0.238	0.228	0.294	0.253	n.d.	0.212	0.193	0.205	0.203
<i>b</i> (Å)	9.032	9.020	9.033	9.028	9.029	9.022	9.027	9.008	9.023	9.022	9.018	n.d.	9.011	9.016	9.011	9.013
Chl(001)**	-	0.371	0.256	0.248	0.251	0.359	0.282	0.262	0.235	0.297	0.265	n.d.	0.187	0.180	0.209	0.192
Quartz	46.2	30.2	37.4	63.0	58.0	30.2	44.2	48.2	35.9	30.8	38.3	37.7	38.0	38.5	56.3	42.6
Albite	19.7	22.1	20.0	19.2	24.8	14.7	20.1	15.5	25.9	22.1	21.2	23.7	22.3	22.5	15.5	21.0
Calcite	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.7	0.0	0.0	0.2
Pyrite	0.0	0.4	0.6	1.1	0.0	0.1	0.4	0.4	0.0	0.1	0.2	0.1	0.1	0.2	0.2	0.2
Mica 1 <i>M</i>	8.2	27.1	7.9	5.3	7.8	24.3	13.4	2.7	7.7	9.5	6.6	2.3	2.2	1.8	1.4	1.9
Mica 2 <i>M</i> <sub>1</sub>	24.5	19.2	28.1	8.2	6.7	19.2	17.7	16.1	19.1	20.7	18.6	22.6	22.0	20.0	16.6	20.3
Chlorite	0.4	1.8	6.0	3.2	3.4	11.5	4.4	16.2	12.1	15.1	14.5	13.4	15.1	17.8	10.7	14.2
Total	99.0	100.8	100.0	100.0	100.6	100*		99.1	100.6	98.1		99.8	100.3	100.6	100.7	

calculated to be 100 % total. Calculated total by RockJock is 92.1 %.

Width of chlorite 001 peak at half height ( $^\circ 2\theta$ )

= not determined

Table 1

	NM				SIM			StM		
	14_04	HN18	HN24	avg.	16_05	HN25	avg.	HN34	HN40	avg.
n	33	17	20	70	17	16	33	17	7	25
Si	3.20	3.26	3.23	3.22	3.21	3.28	3.24	3.25	3.21	3.24
<sup>IV</sup> Al	1.66	1.60	1.64	1.64	1.49	1.63	1.56	1.66	1.72	1.68
<sup>VI</sup> Al	0.72	0.71	0.66	0.70	0.76	0.67	0.72	0.71	0.74	0.72
Fe	0.14	0.20	0.19	0.17	0.13	0.14	0.13	0.11	0.07	0.10
Mg	0.29	0.29	0.25	0.28	0.29	0.28	0.28	0.27	0.24	0.26
Ti	0.08	0.03	0.11	0.08	0.03	0.05	0.04	0.04	0.06	0.04
K	0.83	0.89	0.83	0.84	0.91	0.92	0.91	0.92	0.91	0.92
Na	0.07	0.05	0.42	0.06	0.05	0.05	0.05	0.05	0.04	0.05
Σinter	0.90	0.94	0.87	0.90	0.96	0.97	0.96	0.98	0.95	0.97

Table 2.

	SIM		StM	
	16_05	16_07	HN34	avg.
n	5	17	7	24
Si	2.84	2.73	2.68	2.71
<sup>IV</sup> Al	1.53	1.53	1.55	1.54
<sup>VI</sup> Al	1.12	1.20	1.26	1.22
Fe	2.67	2.69	2.71	2.70
Mg	1.59	1.61	1.60	1.60
Ti	0.03	0.07	0.06	0.07

Table 3.