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Characterisation and Oenological Interest of Nine Commercial Bentonites

Original Citation:

Availability:

This version is available at: 11577/3200171 since: 2016-10-12T14:29:56Z

Publisher:

Published version:

DOI:

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INTRODUCTION

Bentonites are hydrous aluminium silicates belonging to the montmorillonite group (Fig. 1). The brute formula is: $Si_4(Al(2-x)R_x)(O_{10}, H_2O)(Cex, nH_2O)$ or $Si_4(Al(2-x)R_x)(H_2O)_n$ [1].

Information on elemental composition and technical properties of commercial bentonites can be not accessible in the literature or provided by the manufacturers. Bentonite addition may influence the elemental composition of the wine and affect its sensory properties, legal quality and health safety [2].

The aim of this work was to study the release and removal of elements from commercial bentonites to wine by using inductively coupled plasma optical emission spectrometry (ICP-OES). Fining trials were performed using a Chardonnay standard wine and nine types of bentonite added each at three different rates. The structure and morphology of the bentonites were determined by BET method and X-ray diffraction, respectively. The technological efficiency of bentonites was evaluated by measuring the amount of protein removed, the residual protein instability of the wine, and their settling capacity.

The elemental composition of the bentonites studied showed concentrations of Fe and Al above the OIV specifications, while As was below the OIV limit in only one bentonite sample [1].

The element analysis classified the commercial bentonites as sodium, calcium and sodium-activated calcium which is not agreement with the reported by the manufacturers to all case. It was also demonstrated that the use of commercial bentonites in white wine affect the elemental composition of the wine.

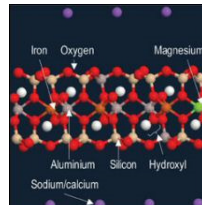


Fig. 1 The crystal lattice structure of Montmorillonite.

MATERIALS & METHODS

Bentonites and wine samples: Fining trials were performed using nine types of bentonite (Fig. 2) added each at three different rates (50, 100 and 200 g/hL) to a Chardonnay standard wine vintage 2012 from Casablanca valley, Chile. All the bentonites were water hydrated at 5% (p/V) for 24 hours before use.

Elements analysis: The determination of elements in bentonite, wine sample and treated wine was done by using inductively coupled plasma optical emission spectrometry (ICP-OES) and specifically by atomic fluorescence spectrometry to As case.

Physical properties of bentonites: The surface properties of the bentonites were studied with Brunauer-Emmett-Teller (BET) model adsorption with liquid N using a surface analyzer (Micromeritics ASAP 2000, USA) and assuming a cross-sectional area of 0.162 nm² for nitrogen. Before the adsorption measurements were taken, the samples were outgassed under a vacuum of 0.001 mbar at 120 °C. The morphology of the material was studied by X-ray diffraction (XRD) using a nickel-filtered Cu K α_1 radiation ($\lambda=1.5418 \text{ \AA}$) in the 2θ range 10-70° through a SIEMENS D5000 diffractometer.

Protein content and heat protein instability: The protein content was determined by Bradford's method and the protein stability according to heat test at 80 °C for 2 h and at 4 °C for 2 h, with $\Delta NTU < 2$ taken to mean stability.

RESULTS & DISCUSSION

Elements analysis: A total of fifteen elements were detected. The addition of bentonite resulted in an increased concentration of As, Cd, Fe, Mn, Mo, Na, Ni, and Zn in wine, while for Cu and K reductions proportional to the dosage used were observed.

Physical properties of bentonites: An average pore size of 20 Å was observed in all bentonites, while their surface areas presented high distribution, with the sodium-activated calcium bentonite showing the highest (Table 1). X-ray diffraction profiles highlighted the presence of different levels of diffraction patterns, with cristobalite and quartz being the most represented (Fig. 3).

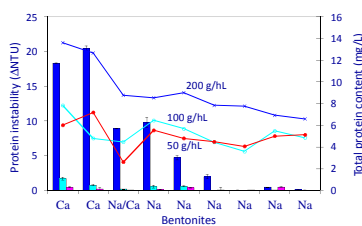


Fig. 2 Capacity of protein removal and protein stability of different commercial bentonites classified according to elements analysis and using different doses.

Protein content and heat protein instability: Among the nine bentonites studied, a great variability in terms of protein removal (Fig. 2) and settling capacity (Fig. 5) was observed, with only four bentonites being able to stabilize the wine at 50 g/hL (Fig. 2).

Table 1 Physical properties of different commercial bentonites

| SAMPLE* | Area BET (m ² /g) | Pore volume (cc/g) | Half pore width (Å) |
|---------|------------------------------|--------------------|---------------------|
| Na | 23.483 | 0.043 | 18.972 |
| Na/Ca | 60.965 | 0.112 | 20.758 |
| Na/Ca | 63.801 | 0.078 | 20.758 |
| Na | 27.367 | 0.077 | 19.845 |
| Na | 32.771 | 0.070 | 20.758 |
| Na | 26.565 | 0.070 | 18.972 |
| Na | 57.625 | 0.129 | 19.845 |
| Na | 28.107 | 0.069 | 20.758 |
| Na | 44.820 | 0.073 | 18.972 |

* Classification of the bentonites according to the manufacturer information

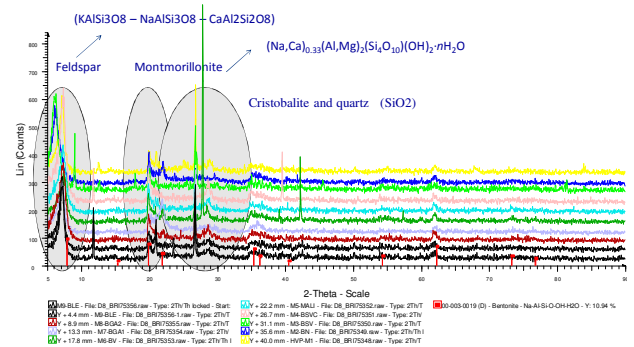


Fig. 3 Power x-ray diffraction of different commercial bentonites

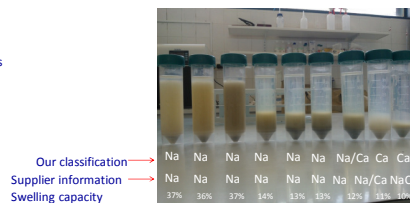


Fig. 4 Swelling property of different commercial bentonites

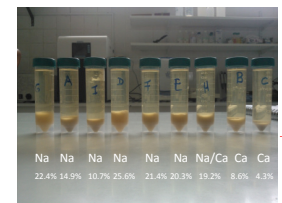


Fig. 5 Wine lees obtained by using 200 g/hL of different commercial bentonites

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ACKNOWLEDGEMENTS

This work was supported by the Chilean National Commission for Research, Science and Technology (CONICYT-Chile) through Grant FONDECYT INICIACION N° 11121594.