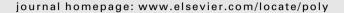


Contents lists available at ScienceDirect

Polyhedron





Synthesis, characterization and application of monocarboxy-phthalocyanine-single walled carbon nanotube conjugates in electrocatalysis

Tawanda Mugadza, Tebello Nyokong*

Chemistry Department, Rhodes University, Grahamstown 6140, South Africa

ARTICLE INFO

Article history: Received 8 December 2010 Accepted 11 April 2011 Available online 19 April 2011

Keywords:
Single walled carbon nanotubes
Phenyl-amine
Cobalt(II)-tris(benzyl-mercapto)mono(carboxyphenoxy)-phthalocyanine
Amitrole
Diuron
Voltammetry
Chronoamperometry

ABSTRACT

In this paper we report on the synthesis, characterization and use of monocarboxy-phthalocyanine-single walled carbon nanotube conjugates in the electrocatalysts of amitrole and diuron. UV-Vis, FTIR and XRD spectroscopies were used in the characterization of constit(II)-tris(benzyl-mercapto)-mono(carboxyphenoxy)-phthalocyanine conjugates (CoMCPc-PA-SWSSYT(linked)), while AFM was used to show changes in surface morphologies of the modified electrodes. Social collammetry and chronoamperometry were used for the electrocatalytic oxidation of amitrole and diuron on the modified glassy carbon electrode. The catalytic rate constants for amitrole and diuron were found to be 1.83×10^6 and 1.99×10^6 M $^{-1}$ s $^{-1}$, respectively. The linear range for both was 1.0×10^{-5} – 2.0×10^{-4} M, with sensitivities of 5.10 and 3.70 A mol $^{-1}$ L cm $^{-2}$ for amitrole and diuron, respectively. The limits of detection were estimated to be 0.14 and 0.20 μ M for amitrole and diuron, respectively, using the 3 δ notation.

© 2011 Elsevier Ltd. All rights reserved.

1. Introduction

Metallophthalocyanines (MPcs) are 18π electron complexes that are being used extensively in electro-catalysis [1–4]. Electrode surface modifications with MPcs enhance electron transfer reactions and this could be achieved through adsorption [1], electrode-position [5] and electropolymerization [6]. These electrode modifiers lower the potentials of redox processes at the electrode|solution interface [7]. Functionalizing MPcs with appropriate groups like carboxylic, hydroxyl and amino groups allows for further derivatization, for example with single walled carbon nanotubes (SWCNTs) carrying extendable functional groups.

Carbon nanotubes (CNTs) are chemically functionalized to provide a higher degree of fine tuning of their chemical and physical properties, including further derivatization and improved solubility [8–11]. Their conductivity and reactivity is a consequence of their electronic properties and chemical functionalization, and this plays an important role in electrochemistry [12–16]. Sidewall functionalization with amines has been reported before [17,18]. Both terminal and sidewall functionalization provide sites for further derivatization with other molecules like phthalocyanines [8].

E-mail address: t.nyokong@ru.ac.za (T. Nyokong).

Chemically linking MPcs to CNTs [19–22] is known to improve electro-catalysis.

In this paper we report on the phenyl-amine functionalization of SWCNT (PA-SWCNT, structure shown in Scheme 1) according to a literature method [18] and chemically linking them to cobalt(II)-tris(benzyl-mercapto)-mono(carboxyphenoxy)-phthalocyanine (CoMCPc), a low symmetry molecule (structure shown in Scheme 2) to give CoMCPc-PA-SWCNT (possible structure shown in Scheme 3). Symmetrically substituted MPcs have been linked to CNTs before [19-22], but linking to unsymmetrically monofunctionalized phthalocyanines produces nano-structures of well defined size that provide good alignment on the electrode surface. There are very limited studies of SWCNT linked to low symmetry phthalocyanines due to the difficulties in the synthesis of these complexes [17,23,24]. The synthesis of CoMCPc is reported for the first time in this work. The large surface areas provided by modification of electrodes with nano-structures provide more sites for the electron transfer processes to occur. The presence of the aryl thio groups in CoMCPc lowers the redox potentials of the cobalt center because of their electron donating nature [25].

The performance of these complexes as electrocatalysis was studied for the electrochemical characterization of diuron (3-(3,4-dichlorophenyl)-1,1-dimethylurea) and amitrole (3-amino-1,2,4-triazole). Amitrole and diuron are non-selective herbicides that are persistent in soils and water, with the former inhibiting the carotenoid biosynthesis [26]. Electrochemical methods have been employed for the detection of diuron [27,28], however

^{*} Corresponding author. Address: Chemistry Department, Rhodes University, P.O. Box 94, Grahamstown 6140, South Africa. Tel.: +27 46 6038260; fax: +27 46 6225109.