1	Summer greenhouse gas fluxes in different types of hemiboreal lakes
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18 Abstract

19	Lakes are considered important regulators of atmospheric greenhouse gases (GHG). We
20	estimated late summer open water GHG fluxes in nine hemiboreal lakes in Estonia classified under
21	different lake types according to the European Water Framework Directive (WFD). We also used the
22	WFD typology to provide an improved estimate of the total GHG emission from all Estonian lakes
23	with a gross surface area of 2204 km ² representing 45,227 km ² of hemiboreal landscapes (the territory
24	of Estonia). The results demonstrate largely variable CO ₂ fluxes among the lake types with most
25	active emissions from Alkalitrophic (Alk), Stratified Alkalitrophic (StratAlk), Dark Soft and with
26	predominant binding in Coastal, Very Large, and Light Soft lakes. The CO ₂ fluxes correlated strongly
27	with dissolved CO ₂ saturation (DCO ₂) values at the surface. Highest CH ₄ emissions were measured
28	from the Coastal lake type, followed by Light Soft, StratAlk and Alk types; Coastal, Light Soft and
29	StratAlk were emitting CH_4 partly as bubbles. The only emitter of N_2O was the Alk type. We
30	measured weak binding of N ₂ O in Dark Soft and Coastal lakes, while in all other studied lake types,
31	the N_2O fluxes were too small to be quantified. Diversely from the common viewpoint of lakes as net
32	sources of both CO ₂ and CH ₄ , it turns out from our results that at least in late summer, Estonian lakes
33	are net sinks of both CO ₂ alone and the sum of CO ₂ and CH ₄ . This is mainly caused by the
34	predominant CO ₂ sink function of Lake Peipsi forming ³ / ₄ of the total lake area and showing negative
35	net emissions even after considering the Global Warming Potential (GWP) of other GHGs. Still, by
36	converting CH_4 data into CO_2 equivalents, the combined emission of all Estonian lakes (8 T C day ⁻¹)
37	is turned strongly positive: 2720 T CO ₂ equivalents per day.

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Keywords

41 Estonian GHG emission, CO₂, CH₄, N₂O, EU Water Framework Directive lake types, floating
42 chamber FTIR measurement

43 1. Introduction

44 In the world fighting against climate warming, there are still many open questions about 45 greenhouse gas (GHG) emissions from natural sources. One of the less explored areas among others is the emission of the three important GHGs – carbon dioxide (CO₂), methane (CH₄), and nitrous oxide 46 47 (N_2O) – from lakes. There are around 117 million lakes on the globe and even though they represent 48 only a tiny fraction of the Earth's nonglaciated land area (Verpoorter et al., 2014), they are an 49 important part of global carbon (C) and nitrogen (N) cycles. Lakes receive C and N from terrestrial 50 ecosystems and fix these elements also from the atmosphere. Both elements are sequestered and/or 51 released from lakes at various rates and in various forms depending on multiple lake characteristics 52 and processes that ultimately determine the GHG balance (Galloway et al., 2003; Tranvik et al., 2009; Kortelainen et al., 2013; Wang et al., 2021). GHGs have various sources and sinks within lakes: CO₂ 53 54 is received from the catchment under the form of dissolved or particulate inorganic carbon, from 55 atmosphere, and it is released in the course of organic matter decomposition through within-lake 56 pelagic and benthic respiration (Del Giorgio et al., 1999). Depending on the partial pressure in the lake's surface layer, CO_2 may either outgas from the lake or be absorbed from the atmosphere through 57 the water-air interface. Water carbonate and humic substance concentrations are important in 58 59 predicting the dissolved CO₂ (DCO₂) levels in lakes (Karlsson et al., 2007; Marcé et al., 2015; Khan 60 et al., 2022). Diversely, CH₄ is mostly created within lakes and never absorbed from the atmosphere, as this gas is almost always supersaturated in lakes relative to the atmosphere (Bastviken et al., 2004). 61 62 The major part of CH₄ is produced by methanogenic microbes in the anoxic zone of organic-rich 63 sediments (Bastviken et al., 2004). Still, as the surface layer of lakes is often supersaturated with CH_4 , 64 it has been hypothesized that efficient source processes should also exist in the oxic pelagic zone (Bogard et al., 2014). The surface CH₄ pool, however, may also be supplied by an advective transport 65 from the shallow water macrophyte beds which support considerably high rates of conventional 66 67 benthic methanogenesis due to their very high production and burial rates of organic matter (Juutinen 68 et al., 2003; Kankaala et al., 2005). In addition to diffusional flux from the sediments, CH₄

69 supersaturation in sediments may lead to bubble formation and subsequent release, which is especially

70	frequent under shallow water column due to low hydrostatic pressure, and it may also be released to
71	the atmosphere through the stems of vascular plants (Whiting and Chanton, 1996; Bastviken et al.,
72	2004). The third aforementioned gas, N ₂ O, is created in lakes mostly during microbial nitrification or
73	ammonification as a by-product and in denitrification as an intermediate product (Knowles, 1982).
74	Consequently, its source may be either ammonia excreted or degraded from organic matter or nitrates
75	acquired from the atmosphere or catchment runoff, the latter also via groundwater.
76	All the three gases are also quickly consumed in lakes – CO ₂ is fixed by various guilds of

77 autotrophs all over the water column (Lapierre et al., 2017) while CH₄ is actively consumed by either 78 anoxic or oxic methanotrophs in redoxclines or the oxic zone, respectively (Bodelier and Steenbergh, 79 2014). Methane captured in bubbles or released through plant stems, nevertheless, escapes 80 methanotrophy (Langenegger et al., 2019). For N₂O, the only known sink is the last step of 81 denitrification pathway, which in case of classical denitrification takes place in reduced environment 82 in the absence of oxygen and at depleted nitrates (Chapuis-Lardy et al., 2007, Richardson et al., 83 2009). However, recent research has found that an aerobic N_2O reduction may be widespread in soils and oceans, carried out by non-denitrifying microbes possessing a previously unknown type of N_2O 84 reductase gene and commonly lacking enzymes for the preceding steps in the denitrification reaction 85 86 chain (Sanford et al., 2012; Rees et al., 2021).

87 While lakes are increasingly viewed as hotspots for CO_2 and CH_4 emissions, their 88 contribution to N₂O emissions is considered negligible compared to terrestrial ecosystems (Tranvik et 89 al., 2009). Even more, certain types of lakes may at least periodically be important consumers of N₂O 90 (Webb et al., 2019) and may have historically been so even more before the era of intense agricultural 91 use of nitrogen fertilizers. The conditions potentially favouring N₂O reduction – abundant labile 92 organic matter coupled with the temporary shortage of oxidised nitrogen – can be easily met in these 93 waterbodies (Knowles, 1982).

94 The global estimates of GHG emissions from lakes build on the extrapolation of average rates
95 of emissions related to lake size or other widely measured characteristics and the density distributions
96 of lakes along the gradients of these characteristics (Downing, 2009). The best global estimates to

97	date have taken account of both the size of the lakes and some available predictor variables for
98	productivity such as nutrients and surface water chlorophyll a (Chl a) content (DelSontro et al., 2018).
99	However, there is a range of other potentially important predictors of lake GHG fluxes like depth,
100	spatial coverage of macrophytes, the stability of stratification, water pH, concentrations of dissolved
101	salts, organic and inorganic carbon, and volcanism, to name a few (Bastviken et al., 2004; Kankaala et
102	al., 2013; Andrade et al., 2016, 2019a; Davidson et al., 2018). Data on these predictors, while lacking
103	at the global scale, may however be available at a regional scale. Several studies have revealed
104	patterns related to environmental gradients and/or some categorization of waterbodies (e.g. lakes and
105	ditches in Netherlands; ponds, reservoirs, and lakes in Canada) (Schrier-Uijl et al., 2011; Tadonléké et
106	al., 2011; DelSontro et al., 2016). For instance, trophic state, alkalinity, humic substances,
107	stratification, and temperature have all emerged as good predictors for pelagic emissions in boreal
108	and/or temperate lakes in North America and Europe (Schrier-Uijl et al., 2011; Marcé et al., 2015;
109	Weyhenmeyer et al., 2015; Yang et al., 2015; DelSontro et al., 2016; Webb et al., 2019). Some studies
110	from Sweden and Finland have further provided regional estimates for lakes based on water chemistry
111	and/or some physical characteristics like lake size and water retention time (Kortelainen et al., 2006;
112	Kankaala et al., 2013; Weyhenmeyer et al., 2015). Tank et al. (2009) estimated the CO ₂ fluxes of the
113	Mackenzie River Delta area lakes (Canadian Arctic) based on three lake types describing the degree
114	of isolation of lakes from the watershed. However, we are not aware, for regional extrapolations, of
115	any utilization of elaborate composite categorizations (typologies) which are adjusted for monitoring
116	purposes and group lakes based on combination of several factors, despite such systems are widely
117	used for the management of anthropogenic disturbances and are in many cases very finely tuned
118	(Moss et al., 2003; Nõges et al., 2007). National or regional lake typologies based on water chemistry
119	and morphometric features (e.g. as the one established by the European Union Water Framework
120	Directive (WFD)) would allow to improve the estimates of aquatic GHG emissions. Knowing the
121	conditions in different lake types and the type-specific areal net GHG emissions enables (1) to raise
122	strong hypotheses in order to find out the biogeochemical processes responsible for the emissions in
123	particular lakes; (2) to design type-specific mitigation measures based on these processes; (3) to

124 prioritise the management measures depending on the areal proportion of different lake types in a 125 region, e.g., it may be easier to manage large emissions from lakes with a small area than vice versa. 126 The aims of the present study were (1) to create the first inventory of lake type specific summer GHG fluxes in the pelagic zone of hemiboreal Estonian lakes according to the WFD lake 127 128 types (Table 1), (2) to study the connections between GHG emissions and lake physico-chemical and 129 biological parameters, and (3) to compile a regional assessment of GHG emissions from lakes 130 considering the areal distribution of lakes between the different lake types. We mapped GHG fluxes 131 of nine lakes representing the different type classes and hypothesised largely variable type specific 132 GHG emissions. We further extrapolated from our results an estimate of late summer GHG fluxes 133 from all Estonian lakes with a gross surface area of 2204 km². 134

135 **2.** S

Study area characterization

136 **2.1** Lake typology

137 All studied lakes (Fig. 1, A - I) were chosen to be typical representatives of the eight Estonian lake types according to the WFD typology, which is based on lake area, alkalinity, conductivity, 138 chloride content, thermal stratification, and colour (Table 1). The two largest lakes in Estonia, Peipsi 139 140 (Peipus, 3555 km², the fifth largest lake in Europe) and Võrtsjärv (270 km²), form individual types 141 referred to as V-Large and Large (Fig. 1, A - B). They were allocated to separate lake types in the WFD compliant lake typology (ME, 2020) because strong wind induced mixing makes them 142 incomparable with smaller lakes in the region, whereas stronger sediment resuspension in the 143 144 shallower Võrtsjärv causes higher turbidity and light limitation clearly distinguishing it from the 145 deeper Peipsi.

The remaining 2534 small lakes are grouped into six types (Ott, 2006; ME, 2020). For our
study purposes, we created a ninth lake type – the stratified alkalitrophic lakes (StratAlk) – to account
for the potential differences in GHG emissions. To make the text easier to follow, the nine lakes in
this study are further referred to by their abbreviated type names (Table 1 and Fig. 2).

150 A general overview of the most important lake parameters is given in Table 2. All lakes were

rather shallow with a mean depth below 10 m. Measurements in all lakes were carried out within a 2-

month period from July to September 2014 (Table 3), and measurements of CO₂ and CH₄ fluxes

- above the lakes with all supportive physicochemical measurements were made once or twice during
- the same period.
- 155 2.2 Alkalitrophic lakes (Alk)

The study lake, Äntu Sinijärv, is a source lake fed by karstic ground waters and represents highly alkalitrophic (>240 HCO₃⁻ mg L⁻¹) lakes in Estonia (Fig. 1C). Lakes of this type are characteristic of the Pandivere Upland area (only a few elsewhere) and with a total area of ca 200 ha they make up only 0.09% of the total area of lakes in Estonia (EDR – *environmental data register of Estonia*). With a mean light attenuation coefficient (K_d) of 0.16 ± 0.02 m⁻¹ for the photosynthetically active spectral region measured in 1995–96, Äntu Sinijärv was the most transparent lake in Estonia (Nõges and Nõges, 1998; Nõges, 2000).

163

2.3 Stratified alkalitrophic lakes (StratAlk)

In the WFD compliant lake typology (ME 2020), these lakes are categorised under stratified lakes with medium alkalinity (StratMedAlk). Considering the different character of many dimictic stratified lakes with high alkalinity (>240 HCO₃⁻⁻ mg L⁻¹), mainly located in South-East Estonia, we created a new type and selected Lake Karijärv (Fig. 1D) as the representative for those 114 lakes. Karijärv is a highly alkalitrophic stratified light-coloured eutrophic system in generally good ecological state. During the thermal stratification from June to September, anoxia is formed at the bottom despite rather high water transparency in the epilimnion (Secchi depth ca 4 m).

171 2.4 Non-stratified lakes with medium alkalinity (MedAlk)

172This is the most abundant type of small lakes in Estonia, comprising approximately 1/5 of the173total number and 4% of the total area of lakes in this country. Lakes under this type are relatively

- 174 shallow, with medium water retention times and may exhibit only episodic thermal stratification.
- 175 Most of the lakes are eutrophic or hypertrophic. Lake Ülemiste, selected as the study site (Fig. 1G),
- 176 provides drinking water for Estonian capital city Tallinn (0.43 mln people).

177 2.5 Stratified lakes with medium alkalinity (StratMedAlk)

178 The 201 lakes of this type, represented in our study by Lake Saadjäry (Fig. 1H), form 2.3% of 179 the total lake area making it the second most abundant small lake type in Estonia. Being deeper, these 180 lakes provide more diverse habitats than the non-stratified lakes but are also more sensitive to human impacts due to the potential build-up of bottom anoxia, leading to sediment phosphorus release and 181

- 182 thus a positive feedback to eutrophication (Ott, 2010).
- 183 2.6 **Dark-coloured soft-water lakes (DarkSoft)**

184 By area, about 1.5% of Estonian lakes are dark-coloured soft-water lakes (890 lakes). These lakes are shallow and acidic containing large amounts of humic matter that contributes to radiative 185 heating in summer (Ott, 2010). Strong light attenuation (up to 11 m^{-1}) restricts the euphotic layer and 186 187 limits photosynthesis (Reinart et al., 2000). In our study, this lake type is represented by Valguta Mustjärv, a very shallow 20-ha lake that has partly recovered from a heavy nutrient loading in the 188 189 1980s (Fig. 1E).

190

2.7 Light-coloured soft-water lakes (LightSoft)

191 The study of Ott and Kõiv (1999) shows that the 310 light-coloured soft-water lakes comprise a little more that 0.8% of all Estonian lakes by area. Those lakes have originally been oligotrophic or 192 semi-dystrophic, mainly characterised by low productivity, small catchment area implying slow water 193 194 exchange, low buffering capacity, and weak stress tolerance (Ott, 2010). Because of high water 195 transparency, these lakes do not develop stable stratification. If impacted, e.g. by eutrophication, the 196 ecosystems of light-coloured soft-water lakes become strongly destabilized, characterised by frequent 197 algal blooms and onset of stratification. In the current study, this lake type is represented by Lake 198 Erastvere (Fig. 1F).

199 2.8 **Coastal lakes (Coastal)**

The 221 coastal lakes constitute approximately 1.6% of the total lake area in Estonia. Their 200 201 functioning depends strongly on the irregular marine water inflow, which creates highly variable 202 conditions and unstable biota. Estonian coastal lakes are shallow with high pH, transparent water (to 203 the bottom) and high water temperature in summer. The main primary producers are charophytes and

not phytoplankton. Since those lakes are shallow and clear-water, there is no temperature

stratification. In our study, this category is represented by Lake Mullutu Suurlaht (Fig. 1I).

206 3. Methods

207 3.1 Greenhouse gas flux experiments

The gas flux was measured on two days (with the exception of Large) within the summer 208 209 stratification period, from July to September 2014 (Table 3) in situ using floating chamber combined 210 with a computer-operated portable Fourier transform infrared (FTIR) spectrometer Gasmet Dx-4030 211 (Gasmet Technologies Oy), also described in Rõõm et al. (2014). Measurements took place at one precise location of each lake (Table 3), chosen to represent the open-water area. The deepest area of 212 213 the lake was chosen except for Large and V-Large. In Large, the measurement site was located close to the long-term monitoring point which, according to Nõges and Tuvikene (2012), is representative 214 for 90% of the lake area. In V-Large, measurements were made at a station in the Mustvee bay, 215 approximately 1 km from the western shore. This station represents the characteristic depth for the 216 217 open-water area and is located at the downwind side of the open lake areas for most of the time. Transparent rectangular polycarbonate floating chamber with a volume of 31 L was used for 218 this study. The chamber was surrounded by an isolation foam tube to achieve chamber buoyancy. 219 Foam frames were equipped with stabilizing weights (around 6 kg) to withstand wave action during 220 221 windy conditions common especially in large lakes. The chamber and FTIR spectrometer were connected with teflon tubes and the air was pumped through the 0.4-L sample cell of the spectrometer 222 with a constant speed of 2 L min⁻¹. Rotation of the air was found to be sufficient and no fans were 223 224 added to the chamber. The temperature in the chamber did not differ more than ± 5 °C from the 225 ambient air temperature, guaranteeing a less than 2% concentration difference in the chamber caused by temperature. The spectrometer was calibrated with pure gaseous nitrogen (AGA, 5.0 N₂) for the 226 227 baseline correction once before every measuring day and, if needed, repeatedly during the day. The 228 results (average of 10 measured spectra per minute) were collected automatically within 1-min 229 interval for at least 3 min for achieving the gas concentrations of ambient air and for at least 8 min from the chamber above the water. For each site, according to the consistency of the measured results, 230

- two or more consecutive measurements were performed (following the method described by
- Tremblay et al., 2005). The minimal time span spent on a single experimental site typically ranged
- between 45 to 60 min. The concentrations of CO₂, CH₄, and N₂O as well as water vapour were
- automatically calculated for each measured spectrum by applying the calibration datasets of IR-
- spectra (in FTIR gas analyser database) and background calibration spectra. According to the manual,
- the detection limit of the spectrometer Gasmet Dx-4030 for N_2O is 0.02 ppm, for CH₄ 0.06 ppm, and
- for CO₂ 11 ppm, and the linearity deviation is <2% of the measuring range. Typical volumetric
- concentrations of the three gases in ambient air (Table 3) were, accordingly, 0.28-0.31, 1.73-2.11, and
- 239 369–382 ppm.
- 240 3.2 Measurement of environmental variables
- **241 3.2.1**

Manual measurement and sampling

The set of variables measured at each floating chamber sampling occasion (Table 3) included 242 243 air pressure, local wind speed and direction, temperatures in the open air and floating chamber, 244 temperature profile of 0.5 m resolution down from the water surface layer (10-15 cm) to the sediment top layer (at 2–3 cm depth), water and sediment pH, concentration of dissolved oxygen (DO) in water 245 as well as the depth of the water column. Air pressure was measured automatically by Gasmet Dx-246 247 4030. Air temperature, local wind speed and direction were measured 1 m above the water surface 248 with a HD2303.0 anemometer (DeltaOhm S.r.L.) by averaging the results of 5 equally distributed 1min measurements within a period of at least 30 min. Temperature in the floating chamber was 249 250 measured with a thermometer DM-9231A (Transfer Multisort Elektronik Sp. z o.o.). For field 251 measurements of DO, pH and temperature in water and sediments we used a handheld 252 multiparametric sonde (YSI ProPlus). Water samples for laboratory analyses of dissolved organic carbon (DOC), phosphates (PO₄-253 254 P), nitrates (NO₃-N), sulphates, chlorides, silicium, ammonium (NH₄-N), total phosphorus (TP) and 255 nitrogen (TN), carbonate alkalinity (HCO₃⁻), Chl a, and phytoplankton were taken from all studied 256 lakes once during the sensor deployment period. DOC concentration was measured in Whatman GF/F

257 filtrate according to Toming et al. (2013). Total phosphorus (TP) was determined with C. Zeiss

spectrophotometer, according to Estonian national standard EVS-EN 1189, total nitrogen (TN) with

259 Bran + Luebbe autoanalyser, according to EVS-EN ISO 13395, HCO₃⁻ was determined 260 colorimetrically using 0.02% methyl orange test. For Chl a, 0.1–1 L of water was passed through Whatman GF/F glass microfiber filter and concentrations were measured spectrophotometrically in 261 262 96% ethanol extracts at a wavelength of 665 nm (Edler, 1979). The phytoplankton was fixed with 263 Lugol's solution and maintained in darkness until microscopic analysis. Phytoplankton cells were 264 counted and measured with an inverted microscope (Ceti Versus, Kontich-Antwerp, Belgium) at 100× or 400× magnification using Utermöhl (1958) technique. Samples were counted until at least 400 265 counting units (filaments, cells, colonies), which gives a counting error of $\pm 10\%$ for the total biomass. 266

267

3.2.2 Sensor deployments and monitoring stations

All lakes were equipped with a high frequency monitoring platform or small lake buoy 268 (OMC-7012 data-buoy) for a 6 to 12 full days period to register changes of water temperature, DO, 269 270 and DCO_2 in every 10 to 30 min over the study period. Before sensor deployment, water temperature 271 (T, °C), DO, and electrical conductivity profiles were measured with handheld multiparametric sonde (YSI ProPlus) to determine the location of the metalimnion from water temperature profile according 272 to Wetzel (2001). Continuous monitoring of DO concentration and water temperature was performed 273 274 by an automated station equipped with a multiparametric sonde (Yellow Springs Instruments (YSI) 275 66,002-4) at one-meter depth. Additional sensors for DO/temperature (Ponsel OPTOD) and for dissolved CO_2 partial pressure (p CO_2) (AMT Analysenmesstechnik GmbH) were deployed at up to 276 277 four depths. In all lakes, the upper sensors were placed at 0.5 m depth and the position of other 278 sensors was decided depending on stratification to adequately characterize all different layers. In the 279 deepest lake (StratMedalk), a chain of 12 HOBO Pendant temperature loggers was used reaching from 0.5 to 20 m depth. Detailed information on high frequency data collection is given in Laas et al., 2016. 280 Automated stations were placed near the GHG flux measurement points. In DarkSoft, we 281 282 could not capture parallel data on CO₂ and GHG fluxes because of malfunctioning of devices. Thus, 283 for this lake, we chose close dates with similar weather conditions and used the recordings of DCO₂ from these days for the comparison of fluxes with DCO₂. In MedAlk, parallel measurements of both 284

- 285 gases at all depths succeeded for one full day only. In StratMedAlk with a maximum depth of 25 m 286 we could measure temperature down to 20 m and CO₂ only down to 10 m depth due to the limited 287 length of the cables. The lake was stratified with the largest drop in pH (0.4-0.5 units) and DO (5.3-6.6 mg L⁻¹) between 6 and 7 m on July 19th and between 8 and 9 m on July 26th. Below oxycline, DO 288 remained relatively stable and fluctuated by less than 2 mg L⁻¹ in every 5 m depth interval while pH 289 290 decreased towards bottom by ≤ 0.2 units per 5 m. Temperature drop was the largest (~8°C) between 8 291 and 15 m depth and only 0.4 °C between 15 and 20 m. For those reasons we assume that the limited 292 length of sensor cables did not affect our conclusions. 293 3.3 Theory and calculation methods
- 294 **3.3.1** Greenhouse gas flux calculations

(1)

295 Temperature and air pressure corrections were made automatically by Gasmet Dx-4030
296 software. Gas fluxes were calculated from linear regression of the gas concentrations in the chamber
297 headspace versus time according to Eq. (1):

 $Flux = \frac{Slope * V}{S}$

299

where *Flux* denotes GHG flux in ppm(v) C m⁻² day⁻¹, *Slope* – slope of the GHG concentration 300 in the chamber headspace versus time in ppm(v) day⁻¹, V – volume of air trapped in the chamber (m³), 301 and S – surface area of the floating chamber (m^2). The fluxes were rejected if the Pearson 302 303 determination coefficient (R^2) was below 0.9 for CO₂, below 0.85 for CH₄, and below 0.6 for N₂O. Since the concentrations of CH_4 and N_2O were about 10–1000 times smaller than those of CO_2 , we 304 found, similarly to Tremblay et al. (2005), that the less strict R^2 value would be justified. When the 305 flux is approaching zero, the difference between measurements will be close to the precision of the 306 307 instrument which results in higher share of random fluctuation in data and lower R^2 . Exception was made for the repeatedly measured similar low efflux concentrations of CO_2 in case of highly 308 persistent CH₄ efflux with an R^2 higher than 0.95. 309 In case of ebullition, the additional *Slope* for ebullition was calculated directly from visible 310

311 concentration 'jumps' on the plots versus total time of floating experiments at a particular measuring

312 site (Fig.3), i.e. in case of a single ebullition event for two or more floating chamber experiments at a

particular measuring site, the time for ebullitive flux *Slope* calculation was gained as a sum of all

314 "on-water" experiment times. To avoid considering the ebullition due to the possible disturbance

315 created by setting up the experiment, all plots with ebullition occurring during the first three minutes

- 316 of floating were rejected and remeasured.
- 317 **3.3.2**

3.3.2 Calculation of dissolved CO₂ values

All CO₂ sensors had a fixed measuring depth, therefore we could do the depth correction for each measurement time interval once for all. We assumed a constant atmospheric pCO₂ of 400 μ atm (http://co2now.org/), which was taken as the equilibrium value for the air-water interface for the calculations of dissolved CO₂ saturation (DCO₂) values (Laas et al., 2016) as during the 24 h period CO₂ concentrations in air are changing. For the comparison with the flux measurement, the exact DCO₂ values were calculated according to measured air CO₂ concentrations (Table 3).

324 **3.3.3** Estimation of GHG emissions of the Estonian lakes

325 We used our collected data to estimate the daily summer GHG emissions of the studied lakes. 326 We extrapolated our results further by combining the experimental data from our study with the typology of Estonian lakes (Ott, 2006; ME, 2020) to present a rough estimate of GHG emissions of all 327 328 Estonian lakes. We assumed that lake type is affecting the summer emission of GHG from the open 329 water area. The typology utilizes physical and chemical parameters which all display large variation 330 in the study area: carbonate content of water (DIC), water humic substance (DOC) content, the presence of summer stratification, size, depth, and connection with the brackish water Baltic Sea. 331 332 While macrophytes are not used as a classificator in the typology, the extent of macrophyte coverage 333 is nevertheless largely predicted by the lake type in the study area (Alahuhta et al., 2018), mostly because of being related to depth and water optical parameters. Carbonates, humic substances, 334 stratification, and macrophytes have all been shown to affect lake GHG emissions (e.g. Tank et al., 335 336 2009; Weyhenmeyer et al., 2015; Yang et al., 2015; Andrade et al., 2019b; Webb et al., 2019; 337 Andrade et al., 2020, 2021). Accordingly, our assumption is well supported by published evidence.

Pelagic GHG emission of a whole lake was calculated by multiplying the average GHG emissions in mg m⁻² day⁻¹ by the lake surface area. To compare the warming effect on climate of these three GHGs, it is reasonable to convert all the emissions into CO 2 equivalents (denoted in tables 4–5 as CO 2e) by weighing the amount of the different gases with their Global Warming Potential (GWP). We used the GWP coefficient of 1 for CO₂, 28 for CH₄, and 265 for N₂O based on the lifetime of 100 vears without feedback (Myhre et al., 2013).

344 To estimate the GHG emissions from all 2536 Estonian lakes, average type specific emissions of CO₂ and CH₄ in mg m⁻² day⁻¹ were multiplied by the total area of lakes belonging to that type. We 345 decided to leave N₂O out from the final balance, as we considered the precision of our instrument 346 347 insufficient for measuring prevailing very small fluxes in the lakes. The environmental data register of Estonia (EDR) was used for calculating the type specific total areas of lakes. For the lakes with 348 unidentified type in the EDR (1345 of total 2536), if possible, we assigned a type by using known 349 physico-chemical parameters (Information System of Environmental Monitoring) in combination with 350 351 soil and landscape type published in Estonian Geoportal (GP). In special cases, the Estonian name of the lake was used as a hint for the lake type, e.g. lakes with names including 'allik-' (spring), 'laugas-352 ' or 'lauka-' (marsh pool), and '-laht' or 'lahe-' (bay) in certain landscape areas were designated to 353 Alk, DarkSoft, and Coastal type, respectively. The remaining area of lakes with unidentified type 354 355 (1.7% of total lake area and 14% of the pooled area of small lakes, *i.e.*, excluding Large and V-Large), was divided among types proportionally with the areal division of the small lakes with known 356 types excluding Coastal. The area of Narva Reservoir (106 km²) was added to Lake Peipsi (V-Large) 357 358 since it is fed by waters from this lake whereas only the Estonian parts of both water bodies (shared 359 with Russia) were added to the calculations.

360 3.4 Statistical analysis

361 The fluxes of CO₂ and CH₄ were compared with physical, chemical, and biological variables
362 (DCO₂, DO, pH, water temperature, lake size, mean and maximum depth, depth at sampling site,
363 metalimnion depth, TP, TN, N:P ratio, Chl *a*, DOC, HCO₃⁻, Kd, Secchi depth, phytoplankton biomass,
364 air temperature and pressure, wind speed) by means of pairwise linear (Pearson) correlation. As

365 Coastal deviated in water salinity and specific conductance from all the other lakes by about an order

of magnitude, we used Spearman rank correlation to study the relationships of these two variables and

their water column gradients with the fluxes of CO_2 and CH_4 .

368

369 **4. Results**

370

4.1 Environmental parameters

371 The maximum depth of the studied lakes ranged from 1 to 25 m with three lakes being more than 10 m deep (Table 2). During our field campaign, the three deepest lakes displayed stable 372 stratification with anoxic hypolimnion while Alk, albeit shallow, showed well developed thermal 373 374 stratification with fully oxygenated hypolimnion. The depth profiles of temperature, salinity, pH. DCO₂, and DO are presented on Fig. 4. TP levels were indicative to rather high eutrophication in all 375 lakes except Alk, while the opposite was true for TN (Table 2), and all inorganic nutrients NO₃-N. 376 NH₄-N, and PO₄-P remained below 1 mg L⁻¹ in most of the lakes (see Table S1 for more detail). The 377 378 hypolimnia of LightSoft and StratAlk were also rich in reduced nitrogen (Table S1). Water sulphates content varied from 2 mg L^{-1} to 32 mg L^{-1} with the exception of Coastal, which had sulphates level as 379 high as 266 mg L^{-1} at the time of our experiments (Table S1). 380

381 4.2 Fluxes of CO₂, CH₄, and N₂O

Based on the R^2 of flux calculations (linear regression of the gas concentrations in the 382 chamber headspace versus time), we omitted two nonlinear declines of N₂O from DarkSoft and 383 StratAlk with $R^2 < 0.6$, where the readings of N₂O in the chamber air decreased by 0.04-0.05 ppm(v) 384 during the first 5 minutes of chamber incubation and levelled off subsequently. For the rest of the 385 measurements with low R^2 values, the change in N₂O concentrations was less than 0.04 ppm(v) 386 during the incubation time. We consider this to be too close to the detection limit of our instrument for 387 quantitative conclusions, and we interpret these measurements hereafter as very small fluxes that we 388 cannot confirm to differ from zero (Fig. 5). 389

Our results demonstrate a broad variety of CO₂ fluxes among the lakes (Figs 5 and 6, Table 4)
with the most active emissions from Alk, StratAlk, and DarkSoft and with negative fluxes in Coastal,

392 V-Large, and LightSoft. As expected, the fluxes correlated strongly with surface DCO₂ values in lakes

- 393(Table S2). The strong correlation with Secchi depth must be taken cautiously since among the two394most transparent waterbodies, Alk was the most active CO_2 emitter and Coastal the most active
- 395 binder.

By CH₄ emissions, Coastal was the most active, followed by LightSoft, StratAlk, and Alk 396 397 (Figs 5 and 6, Table 4). The first three lake types were also emitting bubbles with the same order of 398 activity. CH₄ fluxes displayed strong correlation with CO₂ saturation differences between the lakes' 399 surface and metalimnion (Table S2). CH₄ emissions also correlated with the salinity and/or specific 400 conductance depth gradients (increases towards bottom) in water (Table S2). Both CO₂ and CH₄ 401 fluxes correlated weakly with pH and TN; in case of CO₂, both correlations were negative. We observed measurable N_2O emission only in Alk (Fig. 5, Table 4). It is worth mentioning 402 that Alk displayed very high NO₃-N levels (3.13 mg L⁻¹ compared to 0.03 ± 0.04 mg L⁻¹ in other 403 lakes) as well as the lowest TP content (9 μ g L⁻¹ against 91 \pm 68 μ g L⁻¹) and the highest N/P mass 404 405 ratio (38 against 12 ± 6). A weak uptake of N₂O was recorded in Coastal and DarkSoft lake types (Fig. 5, Table 4). In all the other cases, N₂O fluxes were negligible. 406 4.3 **Total GHG fluxes of Estonian lakes** 407 408 Regarding summer GHG fluxes, Estonian lakes were net sinks of CO₂, and a negative net

balance was observed also for the sum of CO₂ and CH₄ (Table 5). The main reason here is the areal
dominance of the V-Large Lake Peipsi which individually remained a net sink even after considering
the GWP coefficients. Still, after converting the CH₄ emissions from the small lakes into CO₂
equivalents (CO_{2e}), the net GHG emission from Estonian lakes turned strongly positive.

We included the N₂O emissions in the regional balance calculations only for the three lake types for which we had reliable quantitative estimates. Considering the N₂O fluxes, the GHG balance increased by 11% for Alk lakes and decreased by 10% and 0.6% for DarkSoft and Coastal lakes,

416 respectively (Table 5).

417

418 **5. Discussion**

419 5.1 General patterns

We found largely variable GHG emissions in lakes located in close vicinity in a spatially 420 421 confined area (Estonia) but belonging to different lake types *sensu* WFD lake typology. The average (\pm standard deviation) CO₂ emission of 1070 \pm 2275 mg m⁻² d⁻¹ in our study is comparable with the 422 previous recordings from lakes of boreal and temperate regions (Huttunen et al., 2003; Schrier-Ujil et 423 al., 2011: Tadonleke et al., 2011). However, the lakes larger than 1 km² in our study had notably 424 425 lower emission rates compared to Swedish and Norwegian lakes sized between 1 and 100 km² (Yang et al., 2015) and the Finnish large lakes with size $> 100 \text{ km}^2$ (Rantakari and Kortelainen, 2005). Our 426 average CH₄ emission of 114 ± 207 mg m⁻² d⁻¹ was notably high, likely because of the very high 427 emissions from LightSoft and Coastal lakes, as our median CH₄ emission of 33 mg m⁻² d⁻¹ is in good 428 accordance with other studies from the boreal region (Huttunen et al., 2003; Rasilo et al., 2015; Yang 429 et al., 2015). Our measured N₂O range is comparable with the other studies from the region (Huttunen 430 431 et al., 2003; Yang et al., 2015).

432 The well-known fact is that CH_4 is mostly released by sediments. Judging upon the CO_2 profiles (Fig. 4), lake sediments were likely also an important CO_2 source in our study in summer 433 (Laas et al., 2016; Rantakari and Kortelainen, 2005), with the exception of Alk where the main source 434 of high CO₂ is predominant calcareous ground water feeding (Laas et al., 2016). Lakes highest in 435 436 carbonates (Alk, StratAlk) or DOC (DarkSoft) were the strongest emitters of CO₂. As expected, the correlation between DCO₂ in the surface layer and CO₂ fluxes was very strong (Table S2), suggesting 437 DCO_2 as a very good predictor for the flux. Laas et al. (2016) has demonstrated clear negative 438 439 correlation between the gradient of DCO_2 and the trophic state of the lake. Thus, DCO_2 and the 440 associated emission of CO₂ in the studied lakes were likely affected by an interaction of many factors: the carbonate content of water (e.g. Li et al., 2021), sediment and water column catabolic activity (e.g. 441 Jonsson et al., 2003), phytoplankton and macrophyte photosynthesis (Rõõm et al., 2014; Andrade et 442 al., 2021; Trolle et al., 2012), and the aeration of water by mixing (Laas et al., 2016; Andrade et al., 443 444 2019b). The logical sequence of fluxes activity according to the trophic state and carbonate content in our studied lake types would be as Alk – StratAlk – StartMedAlk – MedAlk – LightSoft. In other four 445

lake types, CO₂ fluxes seem to be triggered by other characteristics, for example by high water

column turbulence in Large and V-Large lake types or by high concentrations of organic carbon inDarkSoft and Coastal.

With the exception of Coastal, shallow unstratified lakes were rather low CH₄ emitters 449 450 compared to stratified lakes. One possible explanation might be better oxygenation of sediment 451 surface in several large and shallow unstratified lakes (Fig. 4). This does not always imply more 452 efficient methanotrophy. Notably, as methanotrophy has been shown to require certain undersaturation of oxygen (Thottathil et al., 2019) and is often limited by substrate (Duc et al., 2010), 453 454 one important factor reducing methane emissions in these well-oxygenated lakes may have been the 455 competition of aerobic sediment respiration with methanogenesis for the organic matter in surface sediments. The observed positive correlation between the DCO₂ difference between the surface and 456 457 metalimnion and the emissions of CH₄ could result from the organic carbon content in sediments affecting both variables. Opposite to CO_2 , CH_4 emissions were slightly positively related to TN 458 459 content in water. All major CH₄-emitting lake types – Coastal, LightSoft, and StratAlk – are nutrientand organic-rich with fast metabolism, likely to build up sediment layers with high organic matter 460 content. This is in accordance with the previous studies relating the increased CH₄ emission to 461 462 eutrophication (Huttunen et al., 2003; Bastviken et al., 2004).

463 The lack of measurable N₂O emission in most lakes in our study is likely following the seasonal dynamics of N cycling. Lake water N₂O concentrations in late summer may be at their 464 lowest due to (1) the high primary production and (2) the strongest stratification at this time. 465 466 Nitrification and denitrification are considered the main sources for N₂O in lakes (Seitzinger and 467 Kroeze, 1998). Both in unstratified lakes and in the epilimnia of stratified lakes, the late summer N availability may be insufficient for either nitrification or denitrification to occur as phytoplankton 468 outcompetes bacteria for nitrous substrates (Webb et al., 2019). Indeed, we saw very low nitrate levels 469 in the mixed layers of all lakes except the strongly P limited Alk lake type, suggesting N limitation as 470 471 a reason for the possible N₂O undersaturation. Thermal stratification of lakes has also proven to be an important factor in their N₂O metabolism. In the start of summer stratification in dimictic lakes, there 472

473 may be enough oxygen and nitrates available to curb the N_2O reduction, which is the most oxygen-

sensitive step of denitrification (Richardson et al., 2009), leading to the build-up of N₂O in the
hypolimnion. As the stratification continues, oxygen will be gradually used up and the bottom water
becomes increasingly reduced. By late summer, near-bottom anoxia may have lasted long enough to
deplete the nitrate pool and favour N₂O reduction. Indeed, many studies have found a strong seasonal
gradient in summer N₂O saturation in lakes and reservoirs with the N₂O peak in June and the lowest
and often negative flux values occurring in late summer (Knowles et al., 1981; Jacinthe et al., 2012;
Beaulieu et al., 2014; McCarthy et al., 2016).

481

5.2 Individual features of lakes

The studied Alk lake was intense CO₂ and moderate CH₄ emitter. As described in Laas et al. 482 (2016), the main feature that makes Alk distinct is the predominant ground water feeding of this lake 483 type while the watershed is located in carbonate-rich karstic area. Carbonate-rich ground water (> 300 484 mg L⁻¹ of HCO₃⁻, Information System of Environmental Monitoring) releases CO₂ via intense calcite 485 486 precipitation (Marcé et al., 2015; Laas et al., 2016) that forms the most likely source of the outstanding CO₂ emissions in Alk which were comparable to the average emission of 6703 mg m⁻² d⁻¹ 487 from an 1- year-old reservoir in the boreal Canada (Tadonleke et al., 2011) and surpassed 3- to- 30-488 fold the average emissions measured for the other 15 lakes and 7 - 35 years old reservoirs in that 489 490 study, as well as the other reports for boreal lakes (Huttunen et al., 2003; Yang et al., 2015). The low TP water of the lake is highly transparent revealing almost no planktonic life forms in the water 491 column (Table 2), and most of the DCO₂ gradient is caused by diffusional efflux. These lakes behave 492 as " CO_2 -chimneys" and contribute to the Estonian net CO_2 emissions equally to the other emitting 493 494 lake types despite of a magnitude smaller net area. In our study, the diffuse CH₄ emission in Alk varied from moderate to high. The most probable source is the constantly undisturbed sediment, as the 495 lake is small and sheltered from winds, and the water column is very stable. Also, the abundant *Chara* 496 497 community covering most of the bottom may have reinforced CH_4 diffusional efflux to the water 498 column. Notable N₂O emission distinguishes Alk from the other studied lakes. This feature may be 499 related to the water NO₃-N levels that were by at least one order of magnitude higher than in the other

20

500 lakes. In fact, midsummer undersaturation of N₂O has been observed in lakes with water NO₃-N

501 content below 1 mg L⁻¹ (Jacinthe et al., 2012), a threshold which was surpassed only by Alk in our 502 study. Nitrate content of water has been shown to predict N₂O emissions also in several other studies (McCrackin and Elser, 2010; Webb et al., 2019). Even more, well oxygenated lakes as Alk have been 503 turned out to be net N₂O sources even at low nitrate levels as such lakes may lack a relevant 504 505 functional sink for N₂O (Salk and Ostrom, 2019). Due to the apparent P limitation, also the NH₄-N 506 concentration in the water is relatively high. Thus, the N₂O emitted from Alk may have been 507 produced both by aerobic nitrification in the whole water column or by denitrification at the sediment 508 chemocline.

509 The watershed of the studied StratAlk lake includes mainly sandy and loamy areas. The lake has springs but also an inflowing groundwater fed stream. Because of higher phosphorus content and 510 phytoplankton activity, the lake is more eutrophic than Alk (hypertrophic for TP). Still, the content of 511 512 carbonates is only slightly smaller compared to Alk and the CO₂ emission levels are clearly distinct 513 from StratMedAlk to which type these lakes are usually categorised. As for Alk, the most likely source may have been calcite precipitation (Marcé et al., 2015). Stable stratification most likely 514 prevented the release of CO_2 from the anoxic hypolimnion (Andrade et al., 2020). At the same time, 515 516 more active photosynthesis in the epilimnion, indicated by Chl a, pH, and O₂ saturation profiles, may 517 have been the cause for lower DCO₂ and CO₂ emissions in StratAlk compared to Alk despite the comparable water carbonate levels in these lakes. In StratAlk, the CH₄ emission was the third highest. 518 519 Clear evidence of ebullition was observed and measured during the second experiment while the 520 diffusional flux of CH_4 was moderate. Interestingly, the bottom layer was fully depleted of NO_3 -N 521 indicating the most prominent denitrification in the hypolimnion among the stratified lakes in our study. The extent to which such depletion may affect the emission of N_2O from the lake probably 522 depends on the conditions in the epilimnion and/or the chemocline mixing rates (Heiskanen et al., 523 524 2014). While it seems that the flux of N_2O from the epilimnion to depleted hypolimnion can turn a lake into a net sink for N₂O, in other cases, despite the undersaturation in the hypolimnion, the 525 epilimnion may still act as a N₂O source (Knowles et al., 1981; Webb et al., 2019; but see Mengis et 526

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527	al., 1997). In the present case, no epilimnetic N ₂ O production could be expected as the epilimnion of
528	StratAlk was depleted of both NO_3^- and NH_4^+ . Moreover, the sharp initial decline in N_2O
529	concentrations recorded during our chamber measurements may indicate a fast uptake of N_2O by this
530	lake, but higher precision methods would be needed to confirm this hypothesis.
531	Both CO ₂ and CH ₄ fluxes of studied MedAlk lake were slightly positive and higher on the
532	second measurement with higher wind speed. The mean values of 108 and 11.7 ppm m ⁻² d ⁻¹ for CO_2
533	and CH ₄ fluxes, respectively, remained below average compared with the other studies from the
534	Nordic region. For example, average CO_2 and CH_4 emissions of 771.3 and 31.9 ppm m ⁻² d ⁻¹ ,
535	respectively, were recorded for oligotrophic and mesotrophic Norwegian and Swedish lakes with
536	surface area of $1 - 100 \text{ km}^2$ (Yang et al., 2015), and 767 and 33.3 ppm m ⁻² d ⁻¹ of CO ₂ and CH ₄ ,
537	respectively, for eutrophic Finnish lakes and reservoirs (Huttunen et al., 2003). High nutrient content
538	supports high phytoplankton biomass and Chl a, but the photosynthetic activity seems to be still the
539	minor component of the CO ₂ flux balance compared to respiration, and as a result, the net flux
540	remains positive. The low CH ₄ emission may be related to the large and shallow bathymetry of the
541	lake which apparently causes the low stability of the water column: DO dropped at the sediment
542	surface during our first sampling at the wind speed of 5.1 m s ⁻¹ , while at the second sampling with the
543	higher wind speed of 10.3 m s ⁻¹ , high DO zone reached the muddy surface sediment layer (Fig. 4).
544	Thus, the oxygenation of the sediment may enable high CH ₄ consumption during its diffusion from
545	sediments (Thottathil et al., 2019), as well as suppress methanogenesis in the upper sediment layer
546	consisting the highest quality organic matter (Jarrell, 1985; Yang et al., 2017).
547	In the deep and stratified StratMedAlk, the nutrient content and phytoplankton biomass are
548	considerably lower compared to MedAlk, while DO is below the saturation level even in the surface
549	layer. Although CO ₂ emission stayed below the average values measured in Scandinavian studies
550	(Huttunen et al., 2003; Yang et al., 2015), it still exceeded the emissions from MedAlk despite similar
551	alkalinity and DOC. The emission of CH4 resembled closely the Scandinavian lakes (Huttunen et al.,
552	2003; Yang et al., 2015). Low DO and Chl <i>a</i> indicate strong heterotrophy in epilimnion as a possible
553	source of CO ₂ . Based on the measured DCO ₂ gradient and the absence of DO below thermocline we

can assume substantial storage of both CO_2 and CH_4 in the hypolimnion (Fig. 4). As the lake is relatively large, wind-induced fluctuations in the depth of thermocline and related leakage of CO_2 and CH_4 from hypolimnion may thus also be possible (Huotari et al., 2011; Heiskanen et al., 2014).

The DarkSoft lake was the second highest CO₂ emitter of the study whereas the CH₄ emission 557 558 was the second lowest. With its hypertrophic nature, it is not the most common dystrophic lake, but 559 still quite representative for the post-soviet landscapes of Estonia (Ott and Kõiv, 1999). Compared to 560 the other lakes, the pH was remarkably variable during the measurements suggesting that the lack of stabilising carbonate buffer was influencing the results, and at windy conditions pH stayed moderate 561 while in windless conditions the high respiration activity increased the acidity of the surface water 562 563 layer quite rapidly due to its very low carbonate content. Here, the phytoplankton biomass is strongly light-limited in the dark brown water (Table 2), and the low DO levels give evidence of strong 564 heterotrophy (Cremona et al., 2016). Hence, the decomposition of high concentrations of DOC, 565 primarily the humic components in allochthonous organic matter, is apparently the main contributor 566 567 to the CO_2 flux. This kind of lakes form the second type of "CO₂-chimneys" in our study, where high fluxes are caused by high organic C respiration rates (Cole et al., 1994; Jonsson et al., 2003). Low 568 CH₄ emission may be explained by water column methanotrophic activity observed also in other 569 humic boreal lakes (van Grinsven et al., 2021). Furthermore, as DarkSoft is one of the two lakes with 570 measurable N₂O uptake in our study, the simultaneous consumption of CH₄ and N₂O may indicate 571 active anaerobic CH₄ oxidation. This process would take advantage of the abundant humic substances, 572 573 which can promote the extracellular electron transfer and thus couple the anaerobic oxidation of CH₄ 574 with the reduction of N_2O (Valenzuela et al., 2020).

The hypertrophic and low carbonate content LightSoft system was the second highest CH_4 source with a very high proportion of ebullitive flux, and at least a part-time CO_2 sink. Very high cyanobacterial biomass, Chl a level, and DO supersaturation above the metalimnion during the first measurement campaign, suggested that the CO_2 uptake can be likely attributed to high photosynthetic activity. The small CO_2 release at the second measurement episode could be explained by strong winds (up to 10 m s⁻¹) on 25–26 August 2014 causing deepening of the thermocline and bringing up

less oxygenated water (but still not considerably alleviating the lack of CO₂ in the epilimnion). The

- 582 hypertrophic lake sediment was evidently the strong bubbling source of CH₄, whereas the wind might
- 583 have increased the emissions during the second measurement.

The Large type (Võrtsjärv) was characterised by notably small fluxes of both CO₂ and CH₄. 584 585 As demonstrated by Rõõm et al. (2014), the lake may also become a weak CO_2 sink in summer. The 586 shallowness and the large area opened to the wind balancing the partial pressures of gases between the 587 atmosphere and water, as well as the good balance between gross primary production and respiration 588 (Laas et al., 2012) results in a net CO_2 flux close to zero (Cremona et al. 2016). Also, the low CH_4 589 emission despite organic-rich sediment is likely due to the well oxygenated sediment surface enabling 590 high methanotrophy. In large lakes, wave action could induce the lateral transport of GHG from the open littoral zone to the pelagic area (Hofmann et al., 2010), and this lake is characterized by rather 591 592 extensive zone of submerged macrophytes (18 % of the lake area); however, this likely does not influence its pelagic GHG balance substantially, as CO₂ and CH₄ emissions from this zone are similar 593 594 to the pelagic area (Rõõm et al., 2014).

The V-Large type (Peipsi) is also characterised by a large open area but since the lake is 595 deeper, it is less turbid compared to Large. Still, the whole water column is nearly saturated with DO. 596 597 Similarly to the Large, the V-Large acts as a weak CO_2 sink and CH_4 source, whereas the 598 photosynthetic activity is likely exceeding the total community respiration. The weak CH₄ emission may be related to the sandy organic-poor sediment and good oxidising conditions. The lower 599 600 emissions of CO_2 in both Large and V-Large compared to the Finnish large lakes may rise from the 601 latter being mostly oligotrophic (Rantakari and Kortelainen, 2005), as lower trophic state predicts 602 lower CO₂ emissions (Trolle et al., 2012).

The Coastal system was the strongest CH_4 source and at the same time, the strongest CO_2 sink among the studied lake types. By its GHG emissions, it reminded more a subtropical macrophyte lake than a boreal lake (*e.g.* Colina et al., 2021). The lake is very shallow, fully saturated with DO and transparent to the bottom. Dense *Chara* cover on the bottom is characteristic of lakes with low TP and high TN. The lake is connected to the sea by Nasva River. Depending on the tide and currents, the

lake's salinity differed almost by a factor of two between the measurement campaigns and is generally 608 609 more similar to that of the Baltic Sea than of all the other lakes in our study. In the Coastal type, CO_2 610 is concentrated in bottom waters, although the lake is typically shallow with a short water retention time. One reason for this gradient may be active methanotrophy inside the *Chara* stands (Kuivila et 611 612 al., 1988; Liebner et al., 2011). Since both Chl a and total phytoplankton biomass were moderate and 613 the bottom concentration of DCO₂ was clearly the highest among shallow unstratified lakes, the 614 strong CO_2 uptake could be explained by intense photosynthesis by *Chara* (Laas et al., 2016). The highest CH₄ emissions, both by diffusion and ebullition, gave evidence of a high sediment 615 616 decomposition rate in the warm environment (Duc et al., 2010; DelSontro et al., 2016), while the 617 diffusion could be amplified by *Chara* similarly to the Alk type. Our finding does not fully agree with DelSontro et al (2016) who found that temperature increases CH₄ emissions only at high TP. 618 Regarding N₂O uptake in this lake, higher water sulphate concentrations may facilitate sulphur 619 620 cycling in the sediment and the growth of heterotrophic or autotrophic sulphur-oxidizing bacteria that 621 can utilize N_2O as electron acceptor in the bottom redoxcline (Shao et al., 2010). Abundant CH₄ may 622 support nitrate/nitrite-dependent anaerobic methane oxidation, competing for oxidised N species in the sediments (Ettwig et al., 2010; Cui et al., 2015). 623 624 5.3 **Estimated GHG balance for all Estonian lakes**

625 The estimated emissions of CO_2 and CH_4 from the open water area of all Estonian lakes in summer period were -69 and 36 T day⁻¹, which equals to -19 and 27 T C (tons of carbon) day⁻¹, 626 respectively (Table 5). Accounting for the higher GWP of CH₄, the climate impact of the total 627 628 emission of both gases equalled to 2720 T CO₂ equivalents per day (Table 5). Interestingly, the two 629 largest lakes are strongly underrepresented in their climate impact compared to their areal share: while the largest Lake Peipsi (V-Large) with the areal share of 76% contributed to CO_2 exchange by a 630 negative flux with the magnitude of 49% of all the CO_2 exchange, its contribution to climate impact, 631 632 while still negative, formed only 2% of all the fluxes. The second largest, Lake Võrtsjärv (Large) with 633 the areal share of 12% contributed 6% to the CO₂ exchange and 8% to the climate impact of

emissions. One factor behind the disproportionately small climate impact was the low 7%

 $635 \qquad \text{contribution of both large lakes to the CH_4 efflux.}$

636 The most intriguing in the GHG balance of Estonian lakes is the ability of V-Large (Peipsi) to bind CO₂ in late summer. This is strikingly different from the estimates by Rantakari and Kortelainen 637 (2005) for the Finnish large lakes (mean summer emission 337.7 ppm m⁻² d⁻¹ in lakes sized 100 – 638 4400 km², compared to -79.8 ppm m⁻² d⁻¹ in Lake Peipsi). It suggests that the annual net emission of 639 640 Peipsi might also be negative since the period of autumn CO₂ emission typically caused by die-off of 641 the planktonic community had still not begun in mid-September and high CO₂ uptake could be expected again in early spring just after an emission peak typical at the ice break-up (Rõõm et al., 642 2014). While being only a small CO₂ sink per m^2 , due to its large surface area. Lake Peipsi had still a 643 huge influence on the total net flux from Estonian lakes, turning the total daily CO₂ emission negative. 644 By the CO₂ sink function, Peipsi was followed by the Coastal type having 48 times smaller total net 645 646 area compared to Peipsi but only 6 times smaller total CO₂ uptake.

647 Among the CO_2 net sources, the high areal emission from the Alk type was counterbalanced by the small quantity of these lakes, and the most intense emitters at the country level were small 648 marsh pools (DarkSoft) followed by stratified alkalitrophic lakes (StratAlk), the second rarest type by 649 number of lakes but with a large surface area. Considering CH₄ emissions, the Coastal type was a real 650 651 outlier both per square metre and at the country level, showing values of the same magnitude as the CO₂ fluxes of most other lake types. The next largest CH₄ sources at the country level were LightSoft 652 and StratAlk, both forming a small share of the total lake area but still considerably influencing the 653 654 total flux. On the other hand, the considerable contributions of the Large and V-Large to the country 655 budget of lakes could be attributed to their large surface areas.

Despite of excluding N₂O from the GHG balance of all Estonian lakes, the fluxes below our detection limit may still have an important regional effect due to the high GWP of N₂O. For instance, a close-to-zero flux of ± 0.12 ppm (N₂O) m⁻² d⁻¹, too small to be confirmed by our method, would affect the total exchange of GHG of V-Large by ± 92 T (CO₂ equivalent) day⁻¹. Thus, more precise

quantification of the N₂O fluxes, especially from large lakes, would improve the accuracy of the

661 regional estimates.

Similar to most of the regional studies, we have based our estimate on the pelagic zone
measurements without accounting for the littoral zone, which may emit large amounts of GHG
depending on the type and amount of vegetation and sediment characteristics (Kankaala et al., 2005;
Rõõm et al., 2014). In our study region, helophytes have been shown to support high GHG emissions
(Rõõm et al., 2014) and may add to the pelagic GHG balance assessed by us, making our present
estimate more conservative.

668 Both the measurements in boreal lakes globally (Vachon et al., 2017) and the earlier 669 measurements by our group (Rõõm et al., 2014 and unpublished data) reveal strong CO₂ emissions 670 during early spring and late autumn due to the release of winter under-ice and summer hypolimnic accumulations, respectively. Therefore, the annual CO_2 emissions from many lake types could be 671 even higher than the summer estimates in the present study. This release, especially the autumn 672 673 release, is larger from deeper waterbodies (Ducharme-Riel et al., 2015). It has been estimated to account on average for about 24 % of the annual flux in some Canadian lakes, and 29 - 46 % in large 674 Finnish lakes in different years (Rantakari and Kortelainen, 2005; Ducharme-Riel et al., 2015). 675 676 However, these releases may not suffice to counterbalance the cessation of gas exchange with 677 atmosphere during ice cover: median efflux values from large Finnish lakes show that daily summer estimates can be on average two times higher than the daily annual estimates where seasonal 678 679 differences are accounted for (from data presented by Rantakari and Kortelainen, 2005). Since the 680 strong spring CO_2 emission is followed by the most active uptake during the period of fast planktonic biomass burst in most of the boreal lakes, and from the fact that the largest Lake Peipsi was still a 681 carbon sink in late summer, prediction of the net annual CO₂ balance of Estonian lakes would require 682 data of finer seasonal resolution than presently available. It is probable that on an annual basis, uptake 683 684 would balance the emissions and turn the net CO_2 flux if not negative, then at least only moderately 685 positive. In the case of CH₄, the strong positive temperature dependence of fluxes has been demonstrated (Kelly et al., 1981, Schulz et al., 1997, Yvon-Durocher et al., 2011, Rõõm et al., 2014), 686

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687 and, therefore, the average annual daily emission of CH₄ from the most important emitter, Coastal 688 lake type, can only be smaller compared to midsummer emissions. For stratified lakes, the autumn 689 overturn is an important part of the yearly CH_4 cycle (Kankaala et al., 2006; Fernández et al., 2014), thus potentially increasing the annual balance for these lake types. However, due to winter 690 691 methanotrophy, the annual daily emission may still be notably lower compared to the daily efflux at 692 the summer stratification period (Kankaala et al., 2006). While N₂O has been observed to build up 693 under ice (Cavaliere and Baulch, 2018), no significant release fluxes are expected at ice melt in spring 694 based on measurements from the Finnish lakes (Huttunen et al., 2003).

695 6. Conclusions

696 The studied lakes, representing all lake types in Estonia and most lake types in the whole hemiboreal region, showed a large variety of their GHG fluxes. The strongest CO₂ sources were lakes 697 698 with predominantly allochthonous supplies of either mineral (carbonate rock) or organic (peat) carbon 699 - the alkalitrophic lake with spring water supply (Alk) and the dystrophic lake with high humic substance content (DarkSoft), followed by stratified alkalitrophic and medium alkalinity lakes 700 701 (StratAlk and StratMedAlk). Three of nine types (Coastal, LightSoft, and V-Large) acted as net CO₂ 702 sinks. The fluxes correlated strongly with surface DCO₂ values in lakes. Highest CH₄ emissions were measured in the Coastal lake type, followed by LightSoft, StratAlk, and Alk, most of them also 703 emitting bubbles of CH₄. The Alk type was also the only late summer N₂O emitter, while Coastal and 704 705 DarkSoft lakes were acting as small sinks for N₂O. In other lakes, N₂O fluxes remained close to the detection limit. 706

The results of our analysis suggest a predominant sink effect of Estonian lakes for CO₂ in
summer. However, the summer efflux of carbon in GHG was 8 T C day⁻¹, and the greenhouse effect
of this emission, 2720 T CO₂ equivalents per day, remains positive due to the high GWP of methane.
More detailed seasonal and spatial sampling, incorporation of littoral helophyte zones, and
higher precision measurements for N₂O would be the first scientific improvements required to
decrease the uncertainty of the regional estimate for Estonian lakes. Given the high GWP of this gas,

precise N₂O emission estimates for large lakes are advisable to adequately assess its greenhouse effect

in the region.

713

In this study, methanogenic lakes added the most to the regional climate warming effect. Notably the Coastal lake type, despite its small areal share (1.6% of the regional lake area), was the key emitter of GHG (62 % of total CH_4 emission). On the other hand, the two largest lakes with the joint areal share of 1945 km² (88% of the regional lake area) showed very small climate impact and even a cooling effect in case of Lake Peipsi. For climate mitigation efforts in Estonia, one important target would be to protect the largest lakes from further eutrophication to maintain their low methane and nitrous oxide emissions.

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Figure 1. Lakes included in the present study and representing all the EU WFD lake types (and one additional type, StratAlk) in the study area, Estonia. Abbreviated WFD type names are indicated in brackets. A – Lake Peipsi (V-Large) and the portative monitoring buoy; B – Võrtsjärv (Large) with the stationary monitoring buoy; C – Transparent water of Äntu Sinijärv (Alk) with the reflection of the sky on the surface and our measurement sensors under the water; D – Lake Karijärv (StratAlk); E – Valguta Mustjärv (DarkSoft): almost no reflection of the sky even in a sunny day due to the blackish water; F – Lake Erastvere (LightSoft); G – Lake Ülemiste (MedAlk) with the portative monitoring buoy; H – Lake Saadjärv (StratMedAlk) on a windless day; I – Lake Mullutu Suurlaht (Coastal) with the FTIR gas analyser. Photos A-C, E, G-I by Eva-Ingrid Rõõm, photos D and F by Henn Timm.

Figure 2. Location of the studied lakes in Estonia.

Figure 3. Example of simultaneous occurrence of diffusive and ebullitive GHG fluxes at StratAlk on July 30th, 2014. All concentrations are presented in volumetric parts per million units. Empty red and green circles represent the floating chamber aeration before the experiment and filled circles – measurements during the floating experiment. Dashed trendlines show data used for CO_2 (green) and CH_4 (red) diffusive flux calculation. Concentration "jump" between blue circles (vertical arrow) against whole "on-water" experiment time (horizontal arrow) is used as a Slope for CH_4 ebullitive flux calculation (see Eq. (1)).

Figure 4. Measured physical and chemical parameters of the studied lakes during GHG flux evaluations with floating chamber method. The deepest measured value is taken 2-4 cm below sediment surface (except for DCO₂ measured with buoy). For a better overview,

some points (marked with black) are labelled with measured values. Note the differences in x-axis scaling in case of DCO₂ (marked with different green colours).

Figure 5. Results of the floating chamber flux measurements. A – summaries of CO₂ (green) CH₄ diffusive fluxes (red) and the ebullitive flux of CH₄ (blue) on the left scale; B - diffusive CH₄ flux presented with higher resolution on the right scale; C - measured N₂O fluxes. White columns denote very small fluxes which numeric value cannot be reliably confirmed by our method. Diagonally cut columns indicate that values exceed the scales. Column labels show the average flux values for CO₂ and ebullitive CH₄ (A), diffusive CH₄ (B), and N₂O (C). Error bars show the standard deviation (in section A separately for CO₂ and CH₄ ebullitive flux). Lakes are denoted by type as described in Table 1.

Figure 6. Distribution of lake types according to average CO_2 and CH_4 fluxes in ppm $m^{-2} day^{-1}$.

Figure 1



Figure 2



[Type here] Figure 3



[Type here]

Figure 4



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Figure 5



Figure 6



Table 1. Estonian lake types according to the European Water Framework Directive

[Type here]												
Table 1. Estonian lake types according to the European Water Framework Directive												
Type name	Abbreviation	Area km ²	Alkalinity	Conductivity	Cl⁻	Thermal	Colour	Representative in				
			HCO3 mg L	μs cm -	mg L ⁻¹	stratification	scale	the current study				
Alkalitrophic	Alk	<10	>240	> 400	<25	Non-stratified	N/A	Äntu Sinijärv				
Alkalitrophic, Stratified ^a	StratAlk	<10	>240	> 400	<25	Stratified	N/A	Karijärv				
Non-stratified, medium alkalinity	MedAlk	<10	80–240	165–400	<25	Non-stratified	N/A	Ülemiste				
Stratified, medium alkalinity	StratMedAlk	<10	80–240	165-400	<25	Stratified	N/A	Saadjärv				
Dark-coloured soft-water lakes	DarkSoft	<10	<80	<165	<25	Non-stratified	≥100°	Valguta Mustjärv				
Light-coloured soft-water lakes	LightSoft	<10	<80	<165	<25	Non-stratified	<100°	Erastvere				
Lake Võrtsjärv	Large	100-300	80–240	165–400	<25	Non-stratified	<100°	Võrtsjärv				
Lake Peipsi	V-Large	>1000	80–240	165–400	<25	Non-stratified	<100°	Peipsi				

[Type h	ere]
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Table 2. General description of the study lakes. ^{OM} - average values of our measurements, ^{DB} - long-term average values from the database of the Centre for Limnology, Estonia. TP — total phosphorous, TN — total nitrogen, Chl a — chlorophyll a, DOC — dissolved organic carbon, HCO^-_3 — alkalinity, K_d — vertical light-attenuation coefficient. [#] - most samples of Phytoplankton are integrated, with the exception of Saadjärv and Erastvere, where the surface water / three layers (surface/metalimnion/bottom) average samples were used. ^{*} - In case of Lake Peipsi (Peipus), only an areal share of 10489 km² is in Estonia. Phytoplankton taxonomic groups: Bac stands for diatoms, Chryso – chrysophytes, Crypto – cryptophytes, Cy – cyanobacteria, Chloro – chlorophytes, Dino – dinoflagellates.

Variable	Alk (Äntu	StratAlk	Medalk	StratMedalk	DarkSoft	LightSoft	Large	V-Large	Coastal
	Sinijärv)	(Karijärv)	(Ülemiste)	(Saadjärv)	(Valguta	(Erastvere)	(Võrtsjärv)	(Peipsi)*	(Mullutu
					Mustjärv)				Suurlaht)
Trophic status ^{DB}	Alkalitrophic	Eutrophic	Eutrophic	Mesotrophic	Hypertrophic	Hypertrophic	Eutrophic	Eutrophic	Eutrophic
Mixing regime ^{DB}	Polymictic	Dimictic	Polymictic	Dimictic	Polymictic	Dimictic	Polymictic	Polymictic	Polymictic
Area (ha) ^{DB}	2.1	84	944	724.5	20.4	16.3	27000	261100	412.7
Mean depth (m) ^{DB}	6	5.7	2.5	8	<1	3.5	2.8	8.3	<1
Max depth (m) ^{DB}	8	14.5	4.2	25	1	9.7	6	12.9	1.7
TP (μg L ⁻¹) ^{OM}	9	126	48	22	242	137	48	47	60
TN (μg L ⁻¹) ^{OM}	3450	654	723	414	670	1126	910	375	1000
Chl <i>a</i> (μ g L ⁻¹) ^{OM}	1	13.5	24.7	5.62	23.19	125.64	35.71	13.4	9.04
DOC (mg L ⁻¹) ^{OM}	4.72	11.9	13.7	9.9	35.2	12.3	11.8	12.3	18.1

HCO_{3}^{-} (mg L^{-1}) ^{OM}	293	266	201	150	31	99	211	170	109
$K_d (m^{-1})^{DB}$	0.25	-	3.5	0.42	10.34	2.96	2.76	1.6	0.58
Total Biomass of	0.77	3.02	25.07	5.19/ 3.41	1.12	23.60/ 11.85	8.34	7.38	8.67
Phytoplankton (g/m ³) ^{OM #}									
Dominant families of	Bac, Chryso,	Cy, Dino,	Cy, Bac,	Bac, Cy,	Chryso, Cy,	Cy, Chloro,	Cy, Bac,	Cy, Bac,	Dino, Cy,
Phytoplankton ^{OM}	Crypto	Chryso	Chloro	Chloro	Chloro	Dino	Chloro, Dino	Chloro	Chloro
Secchi depth (m) ^{OM}	8 (bottom)	4.1	0.55	4.9	0.15	1.25	0.55	2.9	1.7 (bottom)
Water residence time (y) ^{DB}	7	1	0.33	0.13	0.33	0.5	1	2	0.2
Watershed size (km ²) ^{DB}	1,37	11.1	98.8	28.4	1.34	5.2	3116	47800	238

Table 3. Dates of experiments and weather conditions during measurements. MSt – data range (and average) of daily mean meteorological variables during measurements from the closest meteorological station to the lake, Local – data from the measuring site. Lakes are denoted by type as described in Table 1.

Lake	Sampling	Period of buoy	Dates of flux	CO ₂ air,	CH4 air,	N ₂ O	Wind	Air	Air	Wind	Depth,	Air
	point	measurement	experiments	ppm	ppm	air,	speed, m s ⁻	temperature,	temperature,	speed, m s ⁻	m Local	pressure,
	coordinates					ppm	² MSt	°C MSt	°C Local	² Local		mBar
												Local
Alk	N 59°03'50.6''	15.07-	16 07. 22 07	270	1.96	0.20	1.2-3.4	17 4 10 9 (19 7)	24.2	26	4.4	1009 5
	E 26°14'27.0''	22.07.2014	10.07; 22.07	572	1.86	0.30	(2.4)	17.4-19.8 (18.7)	24.2	2.6	4.4	1008.5
StratAlk	N 58°18'01.2''	23.07-	22.07.20.07	272	2.07	0.20	2.5-3.4			2.5	10.2	1014.5
	E 26°25'00.5''	05.08.2014	23.07; 30.07 37	575	2.07	0.28	(2.9)	22.0-24.7 (23.7)	20.9	2.3	10.2	1014.3
MedAlk	N 59°24'45.3''	16.07-	14.05.00.05	27.4	1.02	0.20	1.6-2.1	20.1.22.2 (21.2)	22.2			1016
	E 24°46'34.3''	23.07.2014	14.07; 22.07	3/4	1.92	0.30	(1.9)	20.1-22.2 (21.2)	22.5	1.1	4.4	1016
StratMedAlk	N 58°32'14.8''	18.08-			. = 2		1.7-4.1					
	E 26°38'42.2''	27.08.2014	19.08; 26.08	369	1.73	0.31	(2.6)	13.0-14.3 (13.5)	17.4	7.2	21.2	996
DarkSoft	N 58°12'27.4''	23.07-	22.07.5.08	275	1.01	0.20	2.5-3.4	(2, 2, 3, 4, 7, (2, 2, 7))	27.5	1.0	1 1	1017
	E 26°08'38.0''	05.08.2014	23.07, 5.08	575	1.91	0.30	(2.9)	22.0-24.7 (23.7)	21.3	1.9	1.1	1017
LightSoft	N 57°58'45.8''	19.08-	10.00 07.00	200	1.05	0.20	1.4-4.4	10 4 15 5 (10 0)	17.1	6.0	7.0	002
	E 26°46'57.7''	29.08.2014	19.08; 26.08	369	1.85	0.30	(2.8)	12.4-15.7 (13.9)	17.1	0.8	7.9	993

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Large	N 58°12'43.0''	23.07-	23.07	367	2.11	0 29	2.5-3.4	22 6-24 7 (23 7)	7 7 7	0.4	11	085
	E 26°06'12.5''	05.08.2014	23.07	20, 2.11			(2.9)	()	27.7	0.4	4.1	
V-Large	N 58°50'27.2''	09.09-	11.00, 16.00	292	1 0 1	0.31	0.6-2.8	9.9-16.6 (13.8)	17.1	2.0	3.6	1025.5
	E 26°58'20.6''	16.09.2014	11.09, 10.09	302	1.01		(1.3)		17.1			
Coastal	N 58°14'30.7''	07.08-	7.09.12.09	272	1.07	0.29	1.9-7.0			7.6	1.0	1011 5
	E 22°22'00.6''	14.08.2014	7.08; 15.08	572	1.7/		(4.5)	17.1-21.9 (20.4)	22.5	7.0	1.0	1011.5

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Table 4. Estimates of daily summer CH₄, CO₂ and N₂O emissions of the studied lakes based on average fluxes recorded on the days of observation. For N₂O, values close to detection limits are given in brackets. Lakes are denoted by type as described in Table 1. * - CO₂ equivalent calculated by applying the 100 years coefficients of Global Warming Potential (GWP) with no feedback: 1 for CO₂, 28 for CH₄, and 265 for N₂O (Myhre et al., 2013).

Lake	Flux of CO ₂ ,	Flux of CH ₄ ,	Flux of	Flux of	Flux of	Flux of	CO₂ flux	CH₄ flux	N₂O flux	Total	Total	Total CO ₂ +CH ₄	Total CO ₂ +CH ₄	Total CO ₂ +CH ₄ +N ₂ O
	ppm m ⁻² d ⁻¹	ppm m ⁻² d ⁻¹	N₂O, ppm	CO ₂ , mg	CH₄, mg	N ₂ O, mg	per lake,	per lake,	per lake,	CO ₂ +CH ₄	CO2+CH4	flux, mol CO _{2e} *	flux, T CO _{2e} *	flux, T CO _{2e} *
			m-a-	m-a-	m-u-	m-a-	kg day -	kg day -	kg day -	GHC lake-1	lako ⁻¹ davr ¹	lake - day -	lake - day -	lake ⁻¹ day ⁻¹
											lake uay			
										day -				
Alk	3717.8	45.9	2.32	6600	29.7	4.12	139	0.6	0.1	3188	38	4237	0.2	0.2
StratAlk	737.8	137.3	(-0.32)	1310	88.8		1100	75		29653	356	155259	6.8	
MedAlk	108.2	11.7	(-0.23)	192	7.6		1813	72		45645	548	165983	7.3	
StratMedAlk	340.8	33.4	(-0.23)	605	21.6		4384	157		109379	1314	372969	16.4	
DarkSoft	1225.4	8.0	-0.53	2175	5.2	-0.95	444	1.1	-0.2	10150	122	11925	0.5	0.5
LightSoft	-51.5	355.1	(-0.15)	-91	229.8		-15	37		1996	24	65030	2.9	
Large	59.3	14.7	(-0.11)	105	9.5		28434	2561		805720	9677	5115633	225.1	
V-Large	-79.8	2.3	(-0.12)	-142	1.5		-370071	3965		-8161746	-98028	-1488676	-65.5	

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Coastal	-631.9	972.4	-0.62	-1122	629.3	-1.10	-4630	2597 -	4.5	56681	681	4427657	194.9	193.7
											i	61		

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Table 5. Estimation of daily summer greenhouse gas (GHG) emissions from all Estonian lakes based on measured GHG emissions and division of the total lake area between types. The N₂O emissions were excluded from the estimate of the total emission due to their marginal importance and lack of reliable quantitative data for most lake types.

Туре	% of total	Total area	Number	Mean area	Total CO ₂	Total CO ₂	Total CH ₄	Total CH ₄	Total N ₂ O	Total CO _{2e} emission ^a	Total CO _{2e} emission ^a
	lake area	of lakes by	of lakes	of lakes by	emission	emission	emission	emission	emission	by type, N ₂ O	by type, N ₂ O
	in Estonia	type, ha	by type	type, ha	by type, T	excluded, T day ⁻¹	included, T day ⁻¹				
	by type				day ⁻¹	C day ⁻¹	day-1	C day ⁻¹	day-1		
Alk	0.1	213	94	2.3	14	4	0.06	0.05	0.01	19	21
StratAlk	1.5	3362	114	29.5	44	12	3.0	2.3		273	
MedAlk	4.3	9536	539	17.7	18	5	0.7	0.5		74	
StratMedAlk	2.3	5025	201	25.0	30	8	1.1	0.8		114	
DarkSoft	1.5	3404	890	3.8	74	20	0.2	0.2	-0.03	88	79
LightSoft	0.8	1742	310	5.6	-2	-1	4.0	3		306	
Large	12.2	26919	1	26919	28	8	2.6	2		224	
V-Large	75.7	167588	2	83794	-238	-65	2.5	1.9		-42	
Coastal	1.6	3524	221	16.0	-40	-11	22.2	16.7	-0.04	1664	1654
Total	100	220354	2536		-69	-19	36	27		2720	

^a - GHG fluxes weighed by Global Warming Potential coefficients for 100 years without feedback: 1 for CO₂, 28 for CH4, and 265 for N₂O (Myhre et al., 2013)

[Type	here]
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Appendix A (Tables S1 – S2)

Supplement 1

Lake	Date	Sampling depth	Chlorides	Sulphates	Si	NO ₃ -N	NH4-N	PO ₄ -P
Alk	22/07/2014	integrated water column	12.4	17.7	2.56	3.13	0.067	0.006
StratAlk	26/08/2014	surface	5.44	13.1	2.5	0.012	0.006	0.008
	26/08/2014	metalimnion	5.17	13.3	2.53	0.005	0.007	0.005
	26/08/2014	bottom	5.3	6.77	5.28	*	0.55	0.263
MedAlk	22/07/2014	integrated water column	8.09	31.9	0.763	*	0.028	0.005
StratMedAlk	26/08/2014	surface	10.2	22.8	0.8	*	0.015	0.011
	26/08/2014	metalimnion	19.3	24.3	1.34	*	0.025	0.005
	26/08/2014	bottom	8.35	22.3	2.77	0.124	0.036	0.005
DarkSoft	22/07/2014	integrated water column	3.18	5.21	1.56	0.127	0.013	0.184
LightSoft	26/08/2014	surface	14	2	0.189	*	0.003	0.003
	26/08/2014	metalimnion	14.1	1.93	0.192	*	0.006	0.005

Table S1. Chloride, sulphate, and inorganic nutrient concentrations (mg L^{-1}) in studied lakes. * - indicates values below detection limit.

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	26/08/2014	bottom	15.8	10.9	0.685	0.027	0.542	0.2
Large	22/07/2014	integrated water column	7.2	12	3.3	*	*	0.011
V-Large	16/09/2014	integrated water column	5.46	12.55	0.456	0.004	0.018	0.015
Coastal	12/08/2014	integrated water column	very high	266	6.52	0.012	0.088	0.005

Supplement 2

Table S2. Correlations between the GHG fluxes in lakes and lake parameters. Spearman correlation coefficients are presented for relationships including salinity or specific conductance and Pearson correlation coefficients for all the other relationships. Only results with $p \le 0.15$ are presented, results with $p \le 0.05$ are marked in bold.

	Fluxes, ppm							
	CO	\mathbf{D}_2	СН	4	CH4 Total			
	Diffi			onal				
	R	р	R	р	R	р		
DCO ₂ Surface, %	0.99	<0.001						
ΔDCO ₂ Surface vs Metalimnion, %			0.79	0.01	0.91	0.001		
pH Surface	-0.60	0.09	0.63	0.07	0.63	0.07		
pH Metalimnion	-0.55	0.13	0.59	0.10	0.54	0.13		
Δt _{water} Surface vs_Metalimnion, °C	0.58	0.11						
ΔSalinity Surface_Metalimnion, ppt					-0.85	0.07		
ΔSalinity Surface_Bottom, ppt					-0.88	0.04		
Δ Specific Conductance Surface_Bottom, μ S cm ⁻¹					-0.85	0.07		
Total Nitrogen (μ g L ⁻¹)	-0.56	0.12			0.596	0.09		

Total Biomass of Phytoplankton (g/m³) Surface-0.520.15Secchi Depth (m)0.710.03