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# 1 Particle growth with photochemical age from new particle 2 formation to haze in the winter of Beijing, China

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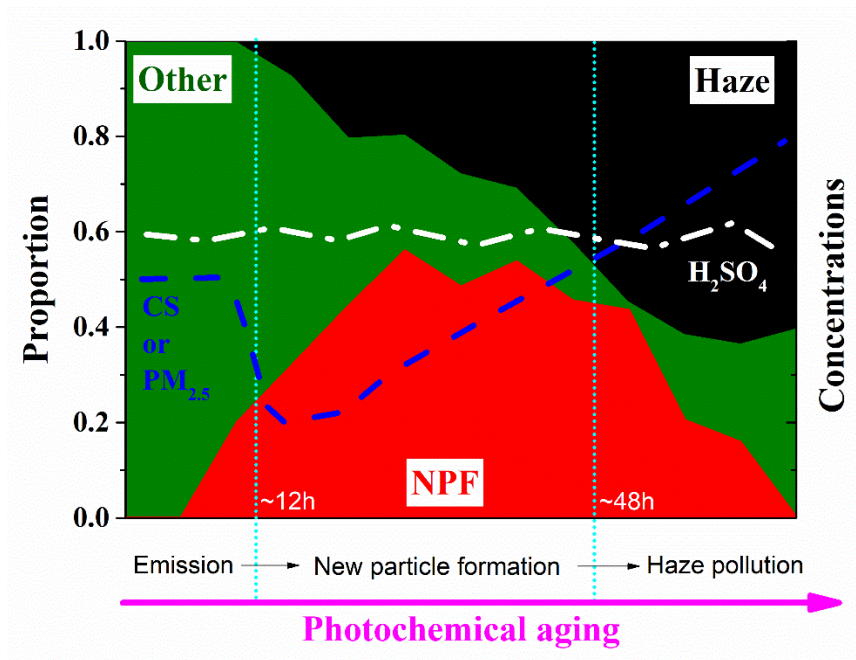
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22 TOC:



24 HIGH LIGHTS

25 The evolution of pollution episodes as a function of photochemical age was studied.

26 Persistent nucleation with photochemical age ranging from 12-48 h was observed.

27 CS and PM<sub>2.5</sub> increased faster with photochemical age on NPF days than other days.

28 Keywords

29 photochemical aging; new particle formation; haze; pollution evolution; condensation sink

30 ABSTRACT.

31 Secondary aerosol formation in the aging process of primary emission is the main reason for haze pollution  
32 in eastern China. Pollution evolution with photochemical age was studied for the first time at a  
33 comprehensive field observation station during winter in Beijing. The photochemical age was used as an  
34 estimate of the timescale attributed to the aging process and was estimated from the ratio of toluene to  
35 benzene in this study. A low photochemical age indicates a fresh emission. The photochemical age of air  
36 masses during new particle formation (NPF) days was lower than that on haze days. In general, the  
37 strongest NPF events, along with a peak of the formation rate of 1.5 nm ( $J_{1.5}$ ) and 3 nm particles ( $J_3$ ), were  
38 observed when the photochemical age was between 12 and 24 h while rarely took place with  
39 photochemical ages less than 12 h. When photochemical age was larger than 48 h, haze occurred and NPF  
40 was suppressed. The sources and sinks of nanoparticles had distinct relation with the photochemical age.  
41 Our results show that the condensation sink (CS) showed a valley with photochemical ages ranging from  
42 12 to 24 h, while  $H_2SO_4$  concentration showed no obvious trend with the photochemical age. The high  
43 concentrations of precursor vapours within an air mass lead to persistent nucleation with photochemical  
44 age ranging from 12-48 h in winter. Coincidentally, the fast increase of  $PM_{2.5}$  mass was also observed during  
45 this range of photochemical age. Noteworthy, CS increased with the photochemical age on NPF days only,  
46 which is the likely reason for the observation that the  $PM_{2.5}$  mass increased faster with photochemical age  
47 on NPF days compared with other days. The evolution of particles with the photochemical age provides  
48 new insights into understanding how particles originating from NPF transform to haze pollution.

## 49 **1 Introduction**

50 Atmospheric aerosol particles, originating from both primary emissions and atmospheric oxidation  
51 followed by gas-to-particle conversion, have adverse effects on human health and visibility (Kaiser, 2005;  
52 Lelieveld et al., 2015; Wang et al., 2015b), and affect the climate (IPCC, 2013). In an urban environment,

53 secondary atmospheric processes were argued to be the main source of aerosol particles in terms of their  
54 number (Kulmala et al., 2016) and mass (Sun et al., 2016). Such secondary processes include the initial  
55 new particle formation (NPF) and condensation of gas phase compounds to aerosol surfaces, which results  
56 in the growth of the particle size (Kulmala et al., 2014).

57 In China, the rapid economic development and urbanization have led to high emissions of pollutants from  
58 coal combustion, motor vehicle exhaust and industrial activities, which has resulted in highly complex air  
59 pollution. High concentrations of trace gases, such as SO<sub>2</sub>, NO<sub>x</sub>, NH<sub>3</sub> and VOCs (Liu et al., 2013; Wang  
60 et al., 2015a; Ye et al., 2011; Zou et al., 2015), have caused high concentrations of secondary inorganic  
61 and organic species in fine particles (PM<sub>2.5</sub>) during haze formation (Huang et al., 2012; Yang et al., 2011;  
62 Zhao et al., 2013). However, the frequency of NPF events in high aerosol-loading environments were  
63 higher than those in low aerosol-loading environments in China (Peng et al., 2014). Compared with rural  
64 or urban sites in western countries, the observed particle formation rates were one to two orders of  
65 magnitude higher and the particle growth rates (GRs) were also somewhat higher for NPF in China (Chu  
66 et al., 2019; Shen et al., 2016; Yao et al., 2018). The extent by which the high concentrations of pollutants  
67 or the highly complex cocktail of air pollution influences NPF remains highly uncertain (Chu et al., 2019;  
68 Kulmala, 2015; Kulmala et al., 2017). Meanwhile, the contribution of NPF to haze formation is still  
69 controversial. With high concentrations of condensable vapours, newly-formed particles have the  
70 potential to grow quickly, which results in an increase of PM volume or mass concentrations. The survival  
71 of growing particles to larger sizes could then directly affect air quality, visibility and climate (Chu et al.,  
72 2019). Guo et al. (2014) reported NPF followed by a continuous growth and appearance of haze pollution  
73 in Beijing and proposed that NPF probably led to haze formation. The controversy about the contribution  
74 of NPF to haze requires analysing the connections between these two phenomena using suitable  
75 parameters.

76 NPF and haze are dominated by secondary processes in terms of the particle number and mass  
77 concentration (Kulmala et al., 2016; Sun et al., 2016). Oxidation of gaseous precursors is usually needed  
78 for the gas-to-particle conversion in both NPF and haze formation, in which photooxidation plays an  
79 important role. The status of the oxidation process will be dependent on both the concentrations of  
80 oxidants in the atmosphere and the photochemical age of the air mass, which is the length of the aging  
81 process: a low photochemical age indicates fresh emissions while a high photochemical age indicates aged  
82 air mass (Parrish et al., 1992). Particle formation, particle mass growth, visibility decrease and many other  
83 processes are expected to occur during different aging periods of the pollutants after their emission, since  
84 the limiting factors for these processes are different. However, according to our knowledge, there is no  
85 study that investigates NPF and haze evolution with photochemical age. During atmospheric observations,  
86 the air mass is constantly changing, which makes it difficult to study chemical transformation in isolation  
87 from the effect of air mass transportation in the pollution evolution. Analysing pollution evolution with  
88 photochemical age may provide insights into how gaseous pollutants and particles chemically transform  
89 with an average emission during different transportation conditions.

90 A continuous comprehensive atmospheric field observation site was established in urban Beijing in 2018  
91 and has been operating since then (Lu et al., 2019; Zhou et al., 2020). In this study, we investigate the  
92 photochemical age and related processes during the winter 2018 in Beijing. We investigated changes in  
93 many key factors affecting both NPF and haze, such as particle number and mass concentration,  
94 condensation sink (CS, which describes the condensing vapor sink caused by the particle population) and  
95 sulfuric acid ( $\text{H}_2\text{SO}_4$ ) concentration, as a function of the photochemical age of air, aiming to draw a  
96 general picture on the evolution of pollution episodes in winter in Beijing. Based on this general picture,  
97 we connect photochemical aging to major atmospheric phenomena, such as NPF and haze formation as  
98 well as their intensity. We also investigate the relationship between NPF and haze formation.

## 99 2 Research methods

### 100 2.1 Comprehensive measurement station in Beijing

101 The results presented in this study are collected at a comprehensive measurement station in urban Beijing  
102 (N 33.94°, E 116.30°) between 23<sup>rd</sup> of January and 31<sup>st</sup> March 2018. The observation station is located on  
103 the western campus of Beijing University of Chemical Technology (BUCT), which is 400 m to the west  
104 of the West Third Ring Road and surrounded by residential and commercial areas (Lu et al., 2019; Zhou  
105 et al., 2020). Particle number concentration and size distribution were monitored with a suite of  
106 instruments, including a Particle Size Magnifier (PSM) to measure small particles in the range of 1.5 –  
107 2.5 nm (Vanhanen et al., 2011), a diethylene glycol (DEG) SMPS to measure 1.5 – 6 nm particles (Cai et  
108 al., 2017; Jiang et al., 2011), a Particle Size Distribution (PSD) to detect particles from 3 nm to 10 µm  
109 (Liu et al., 2016), and a neutral cluster and air ion spectrometer (NAIS) to monitor the 0.8 – 20 nm ions  
110 and 2 – 20 nm particles (Mirme and Mirme, 2013). More detailed information for particle size distribution  
111 measurements and calibrations can be found in section 1 of the supporting Information (SI). These data  
112 were mainly used to calculate the CS, particle formation and growth rates (Kulmala et al., 2012), and to  
113 identify NPF events, which are characterized by the burst of nucleation mode (sub– 25 nm) particles and  
114 subsequent growth and are classified with the method introduced by Dal Maso et al. (2005) The detail  
115 information about the calculation methods and the criteria for identifying an NPF event can be found in  
116 sections 2, 3 and 4 of the SI.

117 Air quality was monitored with various instruments, including gas analyzers (model 42i-TL, 43i-TLE,  
118 48i-TLE, and 49i, Thermo Scientific) to measure mixing ratios of trace gases (NO<sub>x</sub>, SO<sub>2</sub>, CO, and O<sub>3</sub>),  
119 and an online single photo ionization time-of-flight mass spectrometer (SPI-ToF-MS 3000R, Hexin Mass  
120 Spectrometry) (Gao et al., 2013) to measure VOCs, which cannot distinguish isomers but only identify

121 VOC species with unit mass resolution. Meteorological parameters (temperature, relative humidity,  
122 ultraviolet radiation (UVB), pressure, wind speed and direction) and visibility were measured by an  
123 automatic weather station (QML201C and PWD22, Vaisala Inc.). PM<sub>2.5</sub> mass concentration used in this  
124 study was the average value of the four surrounding national observation stations operated by the China  
125 National Environmental Monitoring Center (Website: <http://www.cnemc.cn>), while the composition of  
126 PM<sub>2.5</sub> was measured with an time-of-flight Aerosol Chemical Speciation Monitor (ToF-ACSM, Aerodyne  
127 Inc.) (Frohlich et al., 2013). These data were used to identify haze events, which were mainly  
128 characterized as the visibility less than 10 km and the relative humidity less than 80% (see more detail in  
129 section 4 of the SI), and also to analyse the pollution characteristics. Besides, the sulfuric acid (H<sub>2</sub>SO<sub>4</sub>)  
130 and highly oxygenated organic molecules (HOM) in the gas phase were measured by a chemical ionization  
131 long time-of-flight mass spectrometer (LToF-CIMS, Aerodyne Research, Inc.) equipped with a nitrate  
132 chemical ionization source (see more detail in section 5 of the SI).

## 133 **2.2 Photochemical age of air mass**

134 Hydroxyl radical (OH) is the most important oxidant in the atmosphere. Assuming that two different  
135 volatile organic compounds (VOCs) are simultaneously introduced into an air parcel and react with OH  
136 radical at different rates following pseudo-first-order kinetics and neglecting the influences of dilution,  
137 the ratio of the concentrations of the two VOCs was used to estimate the photochemical aging of the  
138 emission, i.e. the so-called the photochemical age in this study (Parrish et al., 1992). Alkanes in the  
139 atmosphere are primarily removed by oxidation in bimolecular reactions with OH, and the reaction can  
140 be assumed pseudo-first-order because the OH concentration is not significantly affected by the reaction  
141 with the alkane but is determined by a steady state governed by much faster production and removal  
142 reactions<sup>22</sup>. Similar assumptions can also be applied to some aromatics, such as toluene and benzene.  
143 Since toluene reacts faster with OH radical than benzene, the ratio of toluene to benzene will decrease due



144 to aging. Toluene and benzene have widespread sources such as traffic, industry and solvent use (Gao et  
145 al., 2018 and references therein). In Beijing, traffic might be the most important contributor to these VOC  
146 emissions, in which toluene and benzene are emitted simultaneously (Gao et al., 2018). Since there is  
147 abundance of toluene and benzene in the atmosphere and isomers have no significant influence on their  
148 concentration measurement, the ratio between toluene and benzene is used to estimate the photochemical  
149 age in this study. They are selected as the representative VOCs of the local emission source at our  
150 observation site that is surrounded by heavy traffic. NPF and haze take usually place in a regional scale.  
151 In North China Plain, traffic was estimated to be one of the main emission sources of air pollution(Liu et  
152 al., 2013). Besides traffic, coal burning is another important emission source (Sun et al., 2016) which also  
153 emits toluene and benzene (Gao et al., 2018 and references therein). Therefore, toluene and benzene could  
154 be used as the representative VOCs of the regional emission sources.

155 Generally, the photochemical age  $t_a$  was derived from the following equation in this study:

$$156 \quad t_a = \frac{LN\left[\frac{C_{\text{toluene}}(t_0)}{C_{\text{benzene}}}\right] - LN\left[\frac{C_{\text{toluene}}(t)}{C_{\text{benzene}}}\right]}{(k_{\text{toluene}+\text{OH}} - k_{\text{benzene}+\text{OH}}) \times \text{OH}} \quad (1)$$

157 where  $\frac{C_{\text{toluene}}}{C_{\text{benzene}}}(t_0)$  and  $\frac{C_{\text{toluene}}}{C_{\text{benzene}}}(t)$  are the mole concentration ratios of toluene to benzene at the initial  
158 emission time and the observation time, respectively  $k_{\text{toluene}+\text{OH}}$  and  $k_{\text{benzene}+\text{OH}}$  are the rate constants  
159 for reactions of toluene and benzene with OH radical, respectively. OH concentrations varies a lot ( $< 1 \times$   
160  $10^6 - 1.7 \times 10^7$  molecules  $\text{cm}^{-3}$ ) in different seasons and under different weather conditions (Lu et al., 2013;  
161 Tan et al., 2018; Yang et al., 2017). To be comparable with prior publications (Chu et al., 2016; Liu et al.,  
162 2018; Mao et al., 2009), the average OH concentration in the winter of Beijing was assumed to be  $1.5 \times$   
163  $10^6$  molecules  $\text{cm}^{-3}$  in this study. Varied concentrations of OH will speed up or slow down the pollution  
164 evolution with time, but will not change the picture of pollution evolution with photochemical age since

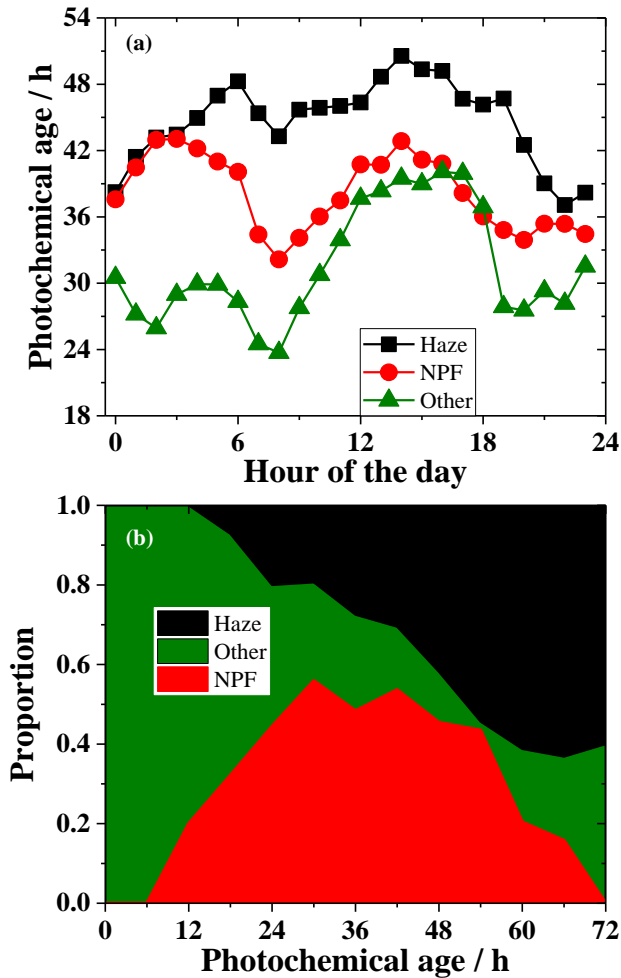
165 using photochemical age actually converts the pollution evolution with time to pollution evolution with  
166 OH exposure. In this study, the 99th percentile of observed ratios of toluene to benzene (i.e. 2.11) was  
167 assumed to be the mole concentration ratio between toluene and benzene at the initial emission time. This  
168 ratio is consistent with the value that can be calculated from the emission ratio (ER) of toluene and benzene  
169 in Beijing (Yuan et al., 2012), and it is also close to the slope of the upper “edge” in the scatterplot (Figure  
170 S5 in section 6 of the SI). We have to point out that mixing with other air masses (fresher emission, more  
171 aged air mass, or air mass with significant different emission sources) during the dilution or transport will  
172 influence the photochemical age. A detailed introduction of the calculation method and its uncertainty  
173 analysis can be found in section 6 of the SI. Basically, aging two air masses separately and then mixing  
174 will result in a little slower aging process compare to mixing two air masses firstly and then aging the  
175 mixture for an pollutant-mass-weighted average photochemical age of the two air masses.

### 176 **3 Results and discussions**

#### 177 **3.1 Variation of photochemical age on different pollution events**

178 During the observation period, days were classified into three groups (Figure S3 in section 4 of the SI):  
179 NPF days (35 days), haze days (23 days) and other days when no NPF nor haze was observed (10 days).  
180 In this study, full 24 hours of data were used in analyzing the haze days as haze events usually lasted for  
181 a few days, whereas the data between 7:00 and 17:00 were used in analyzing the NPF days (except that  
182 the full 24 hours of data were used in Figure 1) as all the NPF events took place in this time window (Dada  
183 et al. 2018, Zhou et al. 2020, and references therein). The diurnal variations of photochemical age of air  
184 mass on these three types of days are displayed in Figure 1a. Overall, the photochemical age of the air  
185 mass on NPF days was lower (by 22% on average) than that on haze days (the p-value for hypothesis test  
186 in statistics is less than 0.05, which means there is a <5% chance that the results could be random), as

187 shown in Figure S6 in section 6 of the SI. This observation can be attributed to the fact that a few hours  
188 of photochemical reactions are able to accumulate enough precursors for NPF to occur, while a longer  
189 period of atmospheric aging is needed to accumulate high enough concentration of particle mass for haze.  
190 The photochemical age increased by about 9-12 hours during a 4-hour period from 8:00 to 12:00 in the  
191 morning of all the three types of days excluding haze days. In other words, the air underwent 2-3 hours  
192 photochemical aging each hour during these periods, which may indicate that the actual concentration of  
193 OH was about 2-3 times the assumed  $1.5 \times 10^6$  molecules  $\text{cm}^{-3}$ . The increase of photochemical age would  
194 also be accelerated in the case of mixing with a more aged air mass at the observation site. However, this  
195 was unlikely as there was no particular wind direction or wind speed change in the morning. The  
196 photochemical age of the air mass on NPF days was not statistically significantly different from the 'other'  
197 days in the daytime (p-value < 0.05), but it was higher in the evening. Fresh local emissions associated  
198 with a small photochemical age seem not to favour the occurrence of NPF in the winter of Beijing, as  
199 NPF rarely took place when the photochemical age was less than 12 h (Figure 1b). As photochemical age  
200 increased, the probability of NPF occurrence increased as well, however, if the photochemical age was  
201 too large, NPF was suppressed and haze was likely to take place.



202

203 **Figure 1:** Median diurnal variations of the photochemical age on different groups of days (a) and  
 204 probability of NPF and haze occurrence as a function of photochemical age (b) during winter Beijing. The  
 205 full 24 hours of data were used in the analysis for this figure.

206

### 207 3.2 Effects of photochemical age on particle formation and growth

208 The median values of some pollutant concentrations and NPF parameters in each bin of photochemical  
 209 age are shown in Figure 2 to display the average case for the evolution of pollution with photochemical  
 210 age. As indicated by the median aerosol number size distribution in Figure 2a, an obvious formation of  
 211 sub-10 nm particles was observed when the photochemical age of the air mass ranged between 12 h and  
 212 48 h, and the PM<sub>2.5</sub> mass concentration also showed an obvious growth over this photochemical age range

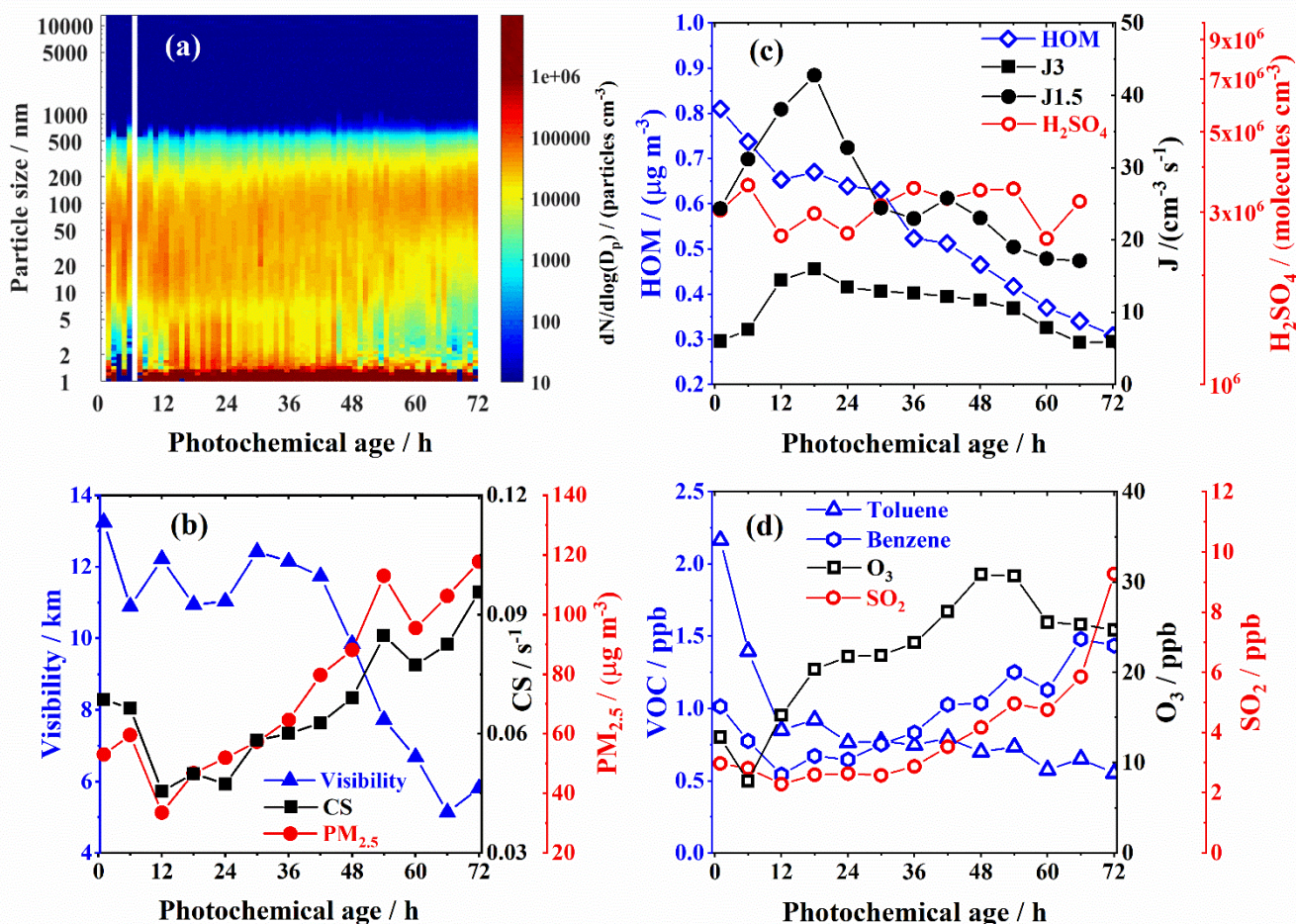
213 (Figure 2b). When photochemical age was less than 12 h, i.e. primary emissions dominated the pollution,  
214 the frequency of NPF (Figure 1b) and the number concentrations of sub-10 nm particles (Figure 2a) were  
215 low, while the PM<sub>2.5</sub> mass concentration was usually relatively high and resulted in relatively high CS  
216 compare to that with photochemical age ranging from 12 h to 24 h (Figure 2b). NPF seemed to be the  
217 strongest for a photochemical age ranging from 12 h to 24 h, and became thereafter weaker, which  
218 indicates that NPF is favoured by relatively fresh air in Beijing winter. Both  $J_{1.5}$  and  $J_3$ , which are the  
219 growth flux through a certain particle diameter, i.e. 1.5 nm and 3 nm (see more detail in section 3 of the  
220 SI), showed a peak and CS showed a valley for the photochemical age ranging from 12 h to 24 h (Figure  
221 2b and c). At photochemical ages larger than 48 h, NPF rarely took place. Meanwhile, the PM<sub>2.5</sub> mass  
222 concentration exceeded 75  $\mu\text{g m}^{-3}$ , which is the secondary air quality standard in China<sup>1</sup>, and visibility  
223 rapidly dropped to 10 km indicating haze, as shown in Figure 2b.

224 In Figure 2c, the variations of gas phase HOM and H<sub>2</sub>SO<sub>4</sub> concentrations with photochemical age are  
225 shown. There was no obvious change in the H<sub>2</sub>SO<sub>4</sub> concentrations with photochemical age, while HOM  
226 showed a decreasing trend. As shown by previous studies (Lu et al., 2019; Petäjä et al., 2009), the H<sub>2</sub>SO<sub>4</sub>  
227 concentration is proportional to OH radical and SO<sub>2</sub> concentrations, while being inversely proportional to  
228 CS. The stable H<sub>2</sub>SO<sub>4</sub> concentrations with photochemical age might be due to a balance of increasing SO<sub>2</sub>  
229 concentrations and CS during the accumulation of pollutants, as shown in Figure 2d and b, respectively.  
230 H<sub>2</sub>SO<sub>4</sub> is a key precursor of nucleation for NPF in megacities when stabilized by strong base molecules  
231 such as ammonia or amines<sup>17</sup>. The independence of the H<sub>2</sub>SO<sub>4</sub> concentration from the photochemical age  
232 indicates that there are abundant H<sub>2</sub>SO<sub>4</sub> precursors for NPF to occur over a wide range of photochemical  
233 ages or rather that regardless of the photochemical age. H<sub>2</sub>SO<sub>4</sub> seems not to be the limiting factor for NPF

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<sup>1</sup> GB 3095—2012, [http://kjs.mee.gov.cn/hjbhbz/bzwb/dqhjbh/dqhjzlbz/201203/t20120302\\_224165.shtml](http://kjs.mee.gov.cn/hjbhbz/bzwb/dqhjbh/dqhjzlbz/201203/t20120302_224165.shtml), 2019/08/19

234 to occur. The high concentrations of H<sub>2</sub>SO<sub>4</sub> within the air mass with a relatively large photochemical age  
235 led to persistent atmospheric nucleation, which can be seen from the fact that  $J_{1.5}$  did not drop below 10  
236 cm<sup>-3</sup> s<sup>-1</sup> throughout the whole range of photochemical ages in Figure 2c. Although not all newly-formed  
237 clusters survive to initiate an NPF event due to their loss to pre-existing particles (high CS), coagulation  
238 of clusters and condensation of vapours, such as H<sub>2</sub>SO<sub>4</sub> and HOMs, onto aerosol surface will continuously  
239 contribute to the particle growth. Similar as H<sub>2</sub>SO<sub>4</sub>, which was proportional to SO<sub>2</sub> concentrations and  
240 inversely proportional to CS (Lu et al., 2019; Petäjä et al., 2009), the HOM concentration is likely to be  
241 proportional to the concentrations of its VOC precursors, while being inversely proportional to CS. The  
242 decreasing trend of HOM concentration with an increasing photochemical age is likely due to the  
243 increased CS and decreased concentrations of some of the high-reactivity VOC precursors, such as toluene  
244 in Figure 2d, which could be oxidized to generate HOM (Molteni et al., 2018). A growing number of  
245 studies reported that HOMs are key contributors to the particle growth (Bianchi et al., 2019 and references  
246 therein), however, their pollution characterization and sources need more dedicated studies especially in  
247 urban environments.



248

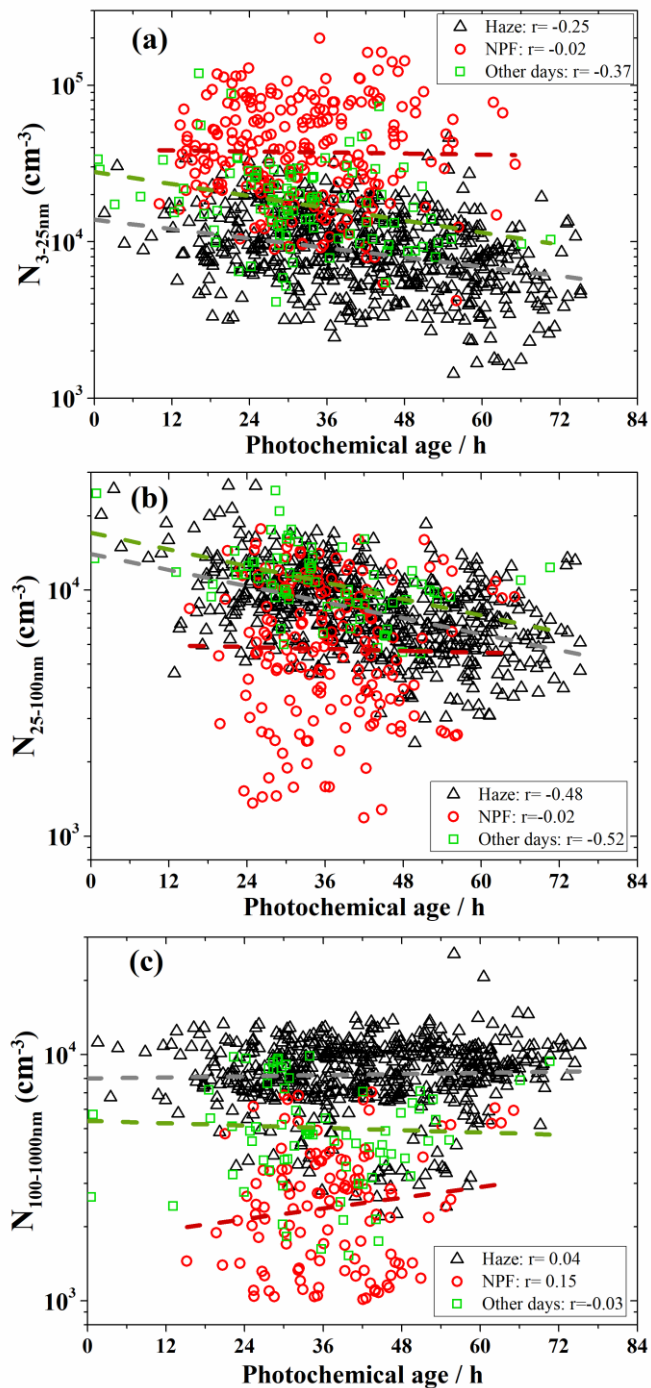
249 **Figure 2:** Variation of aerosol size distribution (a), visibility, CS, and PM<sub>2.5</sub> (c), HOM concentrations, the  
 250 formation rates of 1.5 nm ( $J_{1.5}$ ) and 3 nm particles ( $J_3$ ), and H<sub>2</sub>SO<sub>4</sub> concentrations (c), and trace gas  
 251 (toluene, benzene, O<sub>3</sub>, and SO<sub>2</sub>) concentrations (d) with photochemical age from Equation (1).

252

253 Figure 3 shows the number concentrations of nucleation mode (3-25 nm), Aitken mode (25-100 nm) and  
 254 accumulation mode (100-1000 nm) aerosol particles, CS and PM<sub>2.5</sub> as a function of photochemical age on  
 255 different days. Sub-100 nm particle number concentrations showed a decreasing trend with photochemical  
 256 age on non-NPF days, while no such trend was observed on NPF days (Figure 3a and b). There is an  
 257 increasing trend of accumulation mode particles with photochemical age on the NPF days but not on the  
 258 non-NPF days (Figure 3c). These results indicate that NPF contributed not only to the nucleation mode  
 259 particle number concentration, but also to the Aitken mode and accumulation mode particle number

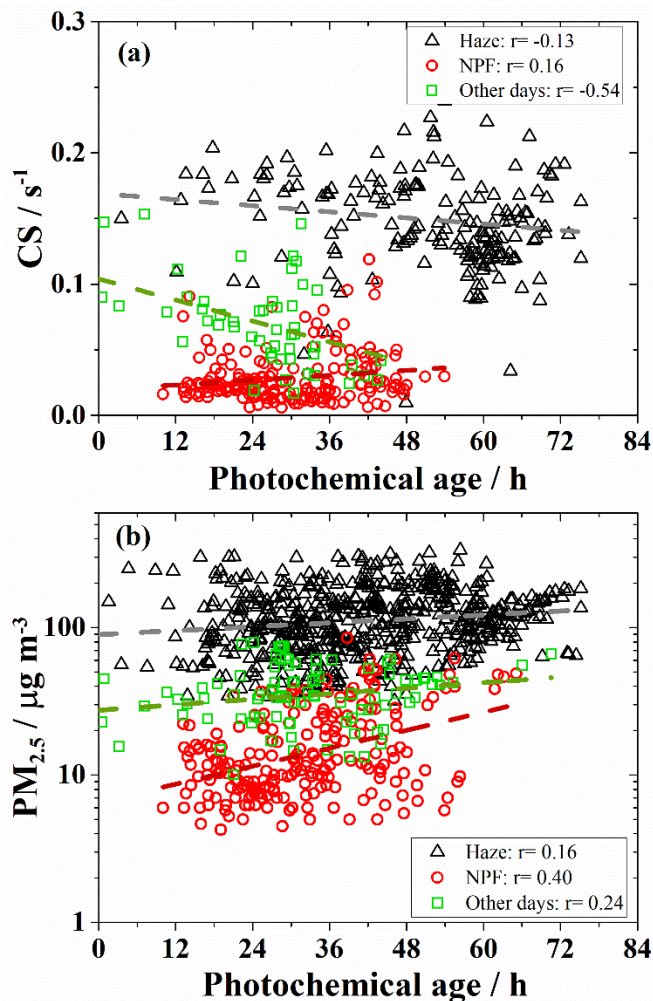
260 concentrations. On days with no NPF, CS decreased with the photochemical age, while it increased with  
261 the photochemical age on NPF days (Figure 4a). On the other hand, increases of PM<sub>2.5</sub> mass  
262 concentrations were observed on all the three types of days with the development of photochemical age,  
263 regardless of whether haze or NPF event occurred or not. In general, PM<sub>2.5</sub> mass concentrations were  
264 lower on NPF days, as shown in Figure 4b and previous studies (Chu et al., 2019 and references therein).  
265 The lower PM<sub>2.5</sub> mass concentrations are expected to be associated with weaker sources of gas pollutants,  
266 including the condensing precursors. However, the increase rate of PM<sub>2.5</sub> mass accumulation with  
267 photochemical age on NPF days ( $0.5 \mu\text{g m}^{-3} \text{h}^{-1}$ ) turned out to be faster than that on non-NPF days (about  
268  $0.2 \mu\text{g m}^{-3} \text{h}^{-1}$ ), as shown in Figure 4b. With photochemistry activity differences already considered in the  
269 photochemical age, this phenomenon might result from the increase of CS with photochemical age on  
270 NPF days. Adopting this point of view, we hypothesize that NPF could have an important contribution to  
271 the increase of PM<sub>2.5</sub> mass concentration in the winter of Beijing participating in haze formation.





272

273 **Figure 3:** Trends of hourly averages of number concentrations of Nucleation mode (3-25 nm) (a),  
 274 Aitken mode (25-100 nm) (b) and accumulation mode (100-1000 nm) (c) aerosols with photochemical age in  
 275 NPF and haze days. The dark grey, dark red and dark green dashed lines in the pictures are the linear  
 276 fitting of the data points for haze days, NPF days and other days with no haze and no NPF, respectively.  
 277 Pearson's  $r$  is displayed for each fitting.  
 278



279

280 **Figure 4:** Trends of hourly averages of condensation sinks (a) and  $\text{PM}_{2.5}$  mass concentrations (b) with  
 281 photochemical age in NPF and haze days. The dark grey, dark red and dark green dashed lines in the  
 282 pictures are the linear fitting of the data points for haze days, NPF days and other days with no haze and  
 283 no NPF, respectively. Pearson's  $r$  is displayed for each fitting.

284

#### 285 4. Conclusions

286 We draw a general picture of the evolution of pollution episodes as a function of photochemical age in  
 287 winter of Beijing based on a comprehensive measurement station for the first time. A few hours of  
 288 photochemical reactions are able to accumulate enough precursors for NPF to occur, while a longer period  
 289 of atmospheric aging is needed to accumulate high enough concentration of particle mass for haze. Fresh

290 local emissions associated with a small photochemical age and haze pollution associated with a large  
291 photochemical age do not favour the occurrence of NPF in the winter of Beijing. In an average situation,  
292 the frequency of NPF was low when the photochemical age was lower than 12 h, while the strongest  
293 nucleation (highest  $J_{1.5}$  and  $J_3$ ) took place with photochemical ages ranging from 12 h to 24 h, and then  
294 became weaker as photochemical age increased further. With photochemical age larger than 48 h, haze  
295 was always present and NPF disappeared.  $\text{H}_2\text{SO}_4$  concentration showed no obvious trend with  
296 photochemical age due to the balance of increasing  $\text{SO}_2$  concentrations and CS with photochemical age  
297 in the atmosphere. The high concentrations of precursors in the air mass lead to a persistent nucleation  
298 with photochemical age ranging from 12 h to 48 h in the winter of Beijing. Coincidentally, the fast increase  
299 of  $\text{PM}_{2.5}$  and CS was also observed during this range of photochemical ages. These results highlighted  
300 that nucleation and subsequent growth of nucleated particles, when strong enough, can contribute to CS,  
301  $\text{PM}_{2.5}$  and even haze.

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### 305 **Author Contributions**

306 BC, YL and MK designed the research. BC, LD, YL, LY, YW, WD, JC, KD, YZ, CD, YF, RY, HL, XCH,  
307 ZF, and CY carried out the observation and analyzed the data from different instruments. BC, LD, XC,  
308 PS, J., Kangasluoma, FB, JJ, J., Kujansuu, VMK, TP, HH and MK analyzed the observation results. BC  
309 prepared the manuscript with contributions from all co-authors.

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