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Tingzhen Ming[1,2], Hanbing Xiong[1], Tianhao Shi[1], Yongjia Wu[1,*], Caixia Wang[1], Yuangao Wen[1], Wei Li[3], Renaud de Richter[4], Nan Zhou[5]

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1. School of Civil Engineering and Architecture, Wuhan University of Technology, Wuhan 430070, China

2. School of Architectural Engineering, Huanggang Normal University, No. 146 Xingang Second Road, Huanggang 438000 China

3. Institute for Materials and Processes, School of Engineering, The University of Edinburgh, Edinburgh EH9 3FB, Scotland, UK

4. Tour-Solaire.Fr, 8 Impasse des Papillons, F34090 Montpellier, France

5. Energy Technologies Area, Lawrence Berkeley National Laboratory, 1 Cyclotron Road, Berkeley, CA 94720 USA

Corresponding author:

Yongjia Wu, Email: yjwu2019@whut.edu.cn

Abstract: The greenhouse effect is exacerbated as greenhouse gas concentrations rise. Capturing and degrading methane in the atmosphere can effectively slow the trend of global temperature rise. The solar chimney power plant integrated with photocatalytic reactor (SCPP-PCR) is a promising concept for both clean electricity generation and large-scale atmospheric methane removal. In this paper, this concept was for the first time quantitatively verified by integrating a photocatalytic reactor in the collector of SCPP to realize the above two targets. A systematic numerical model was proposed to evaluate the performance of degradation of methane and the electricity generation of the SCPP-PCR. The results revealed that the methane purification rate decreased with increasing turbine rotational speed, but the photocatalytic efficiency improved. In this research, the start of the PCR was set at the entrance of the collector, and it was cost-

effective to lay 40 meters in the radial direction with an investment of $4,587 (around 0.37% of the total investment of the system). The system could degrade 30,595.47 g of atmospheric methane and reduce CO_2 emissions by 245.38 kg in the daytime in Wuhan. It was revealed that SCPP-PCR could be crucial for reducing greenhouse gas and limiting climate change.

Keywords: Solar chimney; Photocatalytic reactor; Greenhouse gas removal; Methane photocatalysis; CO_2 emissions reduction

Nomenclature

n	Turbine rotational speed, rpm	S_Φ	Momentum loss term
B, B_1, B_2	Experimental parameter value	S_i	Additional rate owing to the discrete phase
$C_{1\varepsilon}$, $C_{2\varepsilon}$	Constants for turbulent model	m_1	Mass fraction of methane at the SC entrance
Q_m	Mass flow rate, kg·s^{-1}	m_2	Mass fraction of methane at the SC exit
c_1	CH_4 concentration, mol·m^{-3}	x, y, z	Cartesian space coordinates
c_2	O_2 concentration, mol·m^{-3}		
$\varepsilon_{methane}$	Photocatalytic efficiency	*Greek symbols*	
$\dot{m}_{methane}$	Purification rate, g·s^{-1}	v	Kinetic viscosity, m^2·s^{-1}
G	Solar radiation, W·m^{-2}	β	Coefficient of thermal expansion, K^{-1}
$\vec{J_i}$	Diffusion flux of species i, mol·s^{-1}·m^{-3}	ρ	Glass density, kg·m^{-3}
J_1	Methane concentration at system entrance, mol·m^{-3}	τ	Shear stress, N·m^{-2}
J_2	Methane concentration at system exit, mol·m^{-3}	k	Karman Constant
M	Total blade moments of the turbine, N·m^{-1}	η	Efficiency of the turbine
q	Heat flux, W·m^{-2}	*abbreviation*	
R_a	Rayleigh number	*SCPP*	Solar chimney power plant
P_t	Turbine output power, kW	*PCR*	Photocatalytic reactor
r_{AI}	Reaction rate of CH_4, mol W^{-1}·s^{-1}	*UNEP*	United Nations Environment Program
Δp	Drop across the turbine, Pa	*GWP*	Global warming potential

1. Introduction

Methane, being the second-largest greenhouse gas, contributed significantly to

global warming and climate change. The United Nations Environment Program (UNEP) stated that CH_4 emissions must be controlled to slow down global warming and climate change.[1] According to a UNEP report,[2] if the CH_4 emission could be effectively restricted, the global temperature could be reduced by 0.4 ~ 0.5°C in 2050. Kretschmer et al[3] claimed that there were 114.6 billion tons of methane carbon dissolved in seawater, if the temperature of the deep ocean kept raising, methane carbon became unstable and decomposed. It was estimated that 473 tons of CH_4 would be escaped into the atmosphere within 100 years. In addition, considerable methane was emitted as a result of agriculture activities and the use of fossil fuels, accounting for nearly 60% of the total emission.[4] Although methane levels in the atmosphere were substantially lower than carbon dioxide (1.886 ppm vs 415 ppm), CH_4 had a much larger global warming potential. Jackson et al[5] claimed the global warming potential (GWP) of the CH_4 was 84 times that of CO_2 in the first 20 years after it was produced, and it was still 28 times that of CO_2 in the next 100 years. As a result, the total oxidation of atmospheric CH_4 to CO_2 removes 97% of the warming effect over a century time period, leading to significant climatic benefits.[5-7] Recently, several strategies for removing atmospheric CH_4 were described and compared in detail.[8]

Photocatalysis technology offered a promising way to remove the methane in the atmosphere. Electron-hole pairs would generate on the surface of the photocatalyst material under UV lamp, resulting in carboxyl active groups. The carboxyl groups could effectively oxidize the contaminants in the environment to non-toxic and innocuous substances.[9,10] Wada et al[11] conducted experimental research on methane, ethane, and propane photocatalytic reactions, and the results showed that methane was converted into carbon dioxide and a small amount of formaldehyde under UV irradiation. The amount of formaldehyde generated was primarily dependent on ultraviolet lamp irradiation intensity, time, and catalyst activity. Wada et al[12] found that the methane would be actively catalyzed and oxidized under UV irradiation with a wavelength less than 310 nm, producing methanol, CO_2, and H_2O. However, the production of water vapor would hinder the oxidation process of methane. When TiO_2 acted as the photocatalyst, Graetzel et al[13] claimed that the methane was effectively photocatalyzed

to CO_2 and H_2O.

$$CH_4 + 2O_2 \xrightarrow{TiO_2} CO_2 + 2H_2O$$

However, previous photocatalytic methane research was conducted solely in the laboratory but not in the field, because the environmental conditions, such as wind speed and direction, sun intensity, and humidity of the air, had impacts on the photocatalytic effect.[14]

The Solar Chimney Power Plant (SCPP) is a green power generation system consisting of a chimney, canopy, turbine and storage layer. During the day, the sunlight passed through the canopy and heated the storage layer. The density of hot air in the SCPP was lower than that in the external environment, thus a strong flow was created in the system due to the stack effect.[15]

In 1982, the first SCPP was erected in Spain[16] and successfully ran for more than 7 years, demonstrating the possibility of generating electricity using the solar chimney (SC). Following then, researchers put many efforts to study SCPP. Bernardes et al[17,18] developed a numerical model and found that the chimney height, turbine construction, collector diameter, and material were the key parameters for SC design. Further, the entropy generation and the efficiency of the system were calculated.[19] Guo et al[20,21] established an unsteady CFD model for the thermodynamic analysis of the SCPP considering the thickness of storage layer, and obtained the optimal turbine pressure drop ratio by numerical simulation.[22] Since then, many research groups around the world studied the smaller prototypes experimentally. A SC with a 12.3 m tall tower and a 12.5 m radius canopy was constructed and tested in Belo Horizonte.[23] Kasaeian et al[24] developed a small SCPP and found the temperature inversion phenomena at the inlet of chimney. Koonsrisuk et al[25] used CFD model to study the optimal geometry parameters of the SCs. Zhou et al[26] built a theoretical model to analyzing the heat transfer and fluid behaviors in a sloped-collector SCPP. Due to the low solar utilization efficiency of SCPP, researchers had developed several novel SCPPs with higher energy conversion efficiency or new functionalities. Maia et al[27-29] utilized small SCPP to dry agricultural products and developed theoretical models for thermodynamic analysis, demonstrating

the practicality of this technology. Zuo et al[30] suggested the WSSCPPCSD technology and revealed that the system boosted the freshwater production and the power output significantly due to the special turbine at the chimney exit. In 2019, Jamali et al[31] proposed to combine Semi-transparent Photovoltaic (STPV) with SCPP to reduce the temperature of STPV. The convection of the canopy helped to cool the STPV and increase the power production efficiency of the system owing to the chimney effect. Some other innovative studies on SCPP[32-34] and accumulation of solar energy[35,36] were also worth noting.

de Richter et al[37] first time proposed the concept of integrating SCPP with photocatalytic reactors (PCR) for large-scale removal of non-CO_2 greenhouse gases from the atmosphere at the GHGR Conference in Oxford (UK) and a SCPP Conference (DE). He claimed that the SCPP-PCR could weaken the interference of environmental factors on the photocatalytic reaction, ensuring the continuous removal of atmospheric methane. The photocatalytic process of methane was mostly governed by the light intensity and concentration of methane, while temperature played a minor role.[38] Although the SCPP-PCR appeared to be a promising technique for the removal of methane from the atmosphere,[39,40] this had yet to be quantitatively evaluated or verified. In 2021, Ming et al[41] numerical studied the methane removal performance by the SCPP-PCR for the first time, where the TiO_2 was used to fill the photocatalytic reaction zone on honeycomb monolith. The effects of channel diameter, layer length, porosity, and solar radiation on the catalytic efficiency and purification rate were studied, providing invaluable information for the development of the SCPP-PCR prototype. However, due to the large press drop in the photocatalytic reaction zone, the flow rate inside the system was reduced by the filling porous PCR within the collector, which in turn impaired the output power. Thereby, it is hardly to generate electricity via wind turbine installed in the SCPP.

Therefore, the purpose of this paper mainly includes the following two aspects: 1. Designing a simple structure of PCR to realize both the photocatalytic removal of non-CO_2 greenhouse gases and reducing CO_2 emissions through SCPP, so as to quantitatively verify the feasibility of the proposed SCPP-PCR concept by de Richter

et al.[37] 2. Further analyzing the relationship between the power output performance and the amount of methane removal under varying parameters in the SCPP-PCR. Herein, the photocatalytic zone was painted on the ground, and the effects of different solar radiation and rotational speeds of turbine on methane catalytic efficiency, purification rate, turbine output power, and CO_2 reduction were systematically studied by a three-dimensional numerical model.

2. Theoretical models

2.1. Physics model

The model used in this study has the same geometric specifications as the Spanish SCPP prototype. The radius of the collector made of a transparent or translucent material is 120.0 m, the entrance is 2.0 m high, and the bottom of the chimney is 6.0 m high. The surface of the canopy, which has a slope of 1.9 degrees, aiding in the airflow throughout the system and reduces energy loss. The height of the chimney made of concrete is 200 m. The turbine is mounted 7.0 m above the ground. The turbine has four blades and uses CLARK-Y airfoils. The diameter of the impeller and the wheel is 9.8 m and 2.1 m, respectively. And the space between the blade tip and the chimney wall is 0.1 m.[42]

The structure of photocatalytic reactors is the key element for the methane purification effect of the SCPP-PCR. The most extensively-used structures are the plate,[43] honeycomb,[44] and annular column.[45] These reactors have different mass transfer rates, reaction rates, and specific surface areas. as shown in Table 1. The atmospheric methane concentration is low (about 1886 ppb). The SCPP can provide a vast reaction area to assure methane removal without compromising its power-generating impact. As a result, the plate photocatalytic reactor is adopted, and TiO_2, an effective and economical photocatalyst, is painted on the ground. The working principle of the SCPP-PCR is depicted in Figure 1.

TABLE 1. The characteristics of different photocatalytic reactors

Types of	Mass transfer rate	Reaction rate	Specific surface

photocatalytic reactors			area
Plate	High	Medium	Low
Honeycomb	Medium	Low	High
Annular column	Low	High	Medium

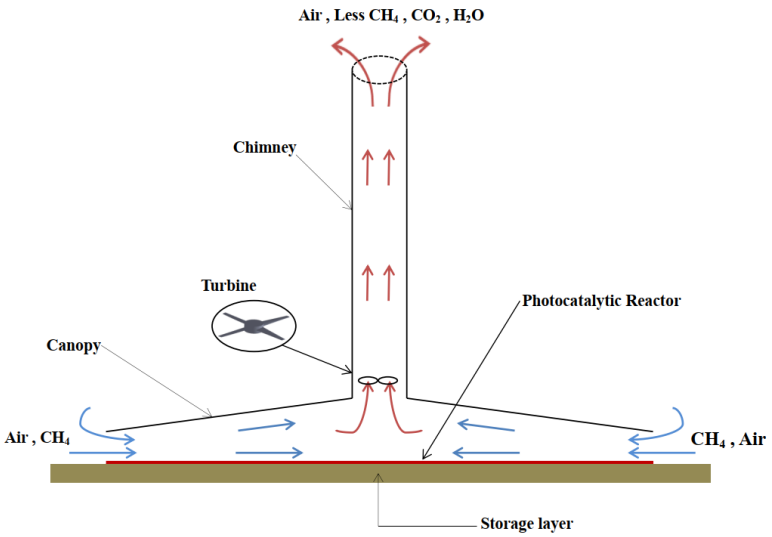


FIGURE 1. Schematic diagram of the SCPP-PCR system.

2.2. Numerical model

The following assumptions are made in the numerical modeling of the SCPP-PCR.

(1) The system works in the steady-state.

(2) The heat transfer within the photocatalyst layer is not considered.

(3) The incompressible ideal gas model is used to simulate the gas density.[46]

(4) The methane photocatalytic produces carbon dioxide and water, no other products.

(5) The energy loss at the chimney-to-collector connection is not considered.

The fluid flow inside the SCPP-PCR is generated by natural convection, which is caused by the ambient air heated by the solar radiation. The strength of the buoyancy-induced flow can be judged by the Rayleigh number:

$$R_a = \frac{g\beta\Delta T H^3}{av} \quad (1)$$

where g, β, a and H are the gravitational acceleration (9.81 m^2/s), the thermal expansion coefficient, the thermal diffusivity and the collector height, respectively. ΔT is the maximum temperature rise within the SCPP-PCR and v is the kinematic viscosity, respectively. A rough calculation finds that Ra $>10^{10}$ in the chimney, indicating that the fluid flow reaches the turbulent state in the system. In fact, the *k-ε* model is the most commonly used turbulence model in the field of solar chimneys. The RNG k-ε model can better deal with flows owning a large swirl or curvature of streamline.[47] Therefore, the RNG *k–ε* model is adopted in this study. The governing equations, including the continuity equation, the momentum equations, the energy equation, the RNG *k-ε* equations, and the transport equations, are given as follows:

Continuity equation

$$\frac{\partial(\rho u_i)}{\partial x_i} = 0 \tag{2}$$

Momentum equation

$$\frac{\partial(\rho u_i u_j)}{\partial x} = \rho g_i - \frac{\partial p}{\partial x_i} + \frac{\partial \tau_{ij}}{\partial x_j} \tag{3}$$

Energy equation

$$\frac{\partial(\rho c_p u_j T)}{\partial x_j} = \frac{\partial}{\partial x_j}\left(\lambda \frac{\partial T}{\partial x_j}\right) + \tau_{ij}\frac{\partial u_i}{\partial x_j} + \beta T\left(\frac{\partial p}{\partial t} + u_j \frac{\partial p}{\partial x_j}\right) \tag{4}$$

Equation for the turbulent kinetic energy (k)

$$\frac{\partial}{\partial x_i}(\rho k u_i) = \frac{\partial}{\partial x_j}\left(\alpha_k \mu_{eff} \frac{\partial k}{\partial x_j}\right) + G_k + \rho\varepsilon \tag{5}$$

Equation for the energy dissipation (ε)

$$\frac{\partial}{\partial x_i}(\rho \varepsilon u_i) = \frac{\partial}{\partial x_j}\left(\alpha_\varepsilon \mu_{eff} \frac{\partial \varepsilon}{\partial x_j}\right) + C_{1\varepsilon}\frac{\varepsilon}{k}G_k - C_{2\varepsilon}\rho\frac{\varepsilon^2}{k} \tag{6}$$

Component transport equation

$$\nabla \cdot (\rho \vec{v} Y_i) = -\nabla \cdot \vec{J_i} + R_i + S_i \tag{7}$$

where μ_{eff} ($\mu_{eff} = \mu + \mu_t$) represents the effective kinematic viscosity, G_k represents the generation of turbulence kinetic energy owing to buoyancy and can be defined as $G_k = -\rho\overline{u_i' u_j'}\frac{\partial u_j}{\partial x_i}$, α_k and α_ε are the turbulent Prandtl numbers for k and ε, respectively. Here, $\alpha_k = \alpha_\varepsilon = 1.30$. $C_{1\varepsilon}$ and $C_{2\varepsilon}$ are two constants for the turbulent model with$C_{1\varepsilon} =$ 1.44, $C_{2\varepsilon}$ = 1.92. $\vec{J_i}$ represents the diffusion flux of

species i, which is given by $\vec{J_i} = -\rho D_{i,m} + R_i$, where R_i represents the amount of component i produced or consumed in a chemical reaction, S_i represents the additional rate owing to the discrete phase, and Y_M indicates the variable dilatation incompressible turbulence contribution to the total dissipation rate.

The multiple reference frame (MRF) model is used to solve the turbine region. The MRF model provides a cost-effective solution for computation domains with moving parts. In the MRF model, the computational domain is broken down into several subdomains. Each subdomain owns specific motion mode, either stationary, rotating, or translational. The governing equations of the flow field in each subdomain are solved independently. Then the flow field information in the nearby subdomains is exchanged at the interface by converting the velocity to absolute velocity. The velocity in the rotating coordinate system is given by

$$\vec{v_r} = \vec{v} - \vec{\omega} \times \vec{r} \tag{8}$$

The governing equations of flow in the turbine region are given as follows:

Continuity equation

$$\nabla \cdot (\rho \vec{v_r}) = 0 \tag{9}$$

Momentum equations

$$\nabla \cdot (\rho \vec{v_r} \vec{v_r}) + \rho(2\vec{\omega} \times \vec{v_r} + \vec{\omega} \times \vec{\omega} \times \vec{r}) + \rho \frac{\partial \vec{\omega}}{\partial t} \times \vec{r} = \nabla \cdot (\mu \nabla \vec{v_r}) + \mathrm{S}_{\vec{v_r}} \tag{10}$$

where $\vec{v}$ is the absolute velocity, $\vec{\omega}$ is the angular velocity vector, and $\vec{r}$ is the position vector.

The output power (P_t) and the efficiency of the turbine (η) is calculated by Eq. (11)-(12).

$$P_t = \frac{2\pi n M}{60} \tag{11}$$

$$\eta = \frac{P_t}{\Delta p Q_v} \tag{12}$$

where n represents the rotation speed, M represents the total blade moments, Q_v represents the volume flow rate of the system, Δp represents the pressure drop of turbine.

Andreas et al[48] devised a set of experiments to derive an effective expression for the photocatalyzed total oxidation of methane:

$$r_{AI} = B\frac{B_1 c_1}{1+B_1 c_1}\frac{B_2 c_2}{1+B_2 c_2} \quad (13)$$

where r_{AI} is the reaction rate of CH_4, c_1 is the concentration of CH_4, c_2 is the concentration of O_2, B, B_1 and B_2 are measured by experiments and the parameter value are 5.37×10^{-7}, 2.42, and 4.60, respectively.

The photocatalytic efficiency and rate of methane is given by Eq. (14)-(15).

$$\varepsilon_{methane} = \frac{J_1 - J_2}{J_1} \quad (14)$$

$$\dot{m}_{methane} = Q_m(m_1 - m_2) \quad (15)$$

where J_1 and J_2 are the inlet and outlet methane concentrations, respectively, Q_m represents the mass flow rate of the system, m_1 and m_2 are the mass fraction of methane at the entrance and exit, respectively.

3. Method and Validity

3.1. Meshing and independence verification

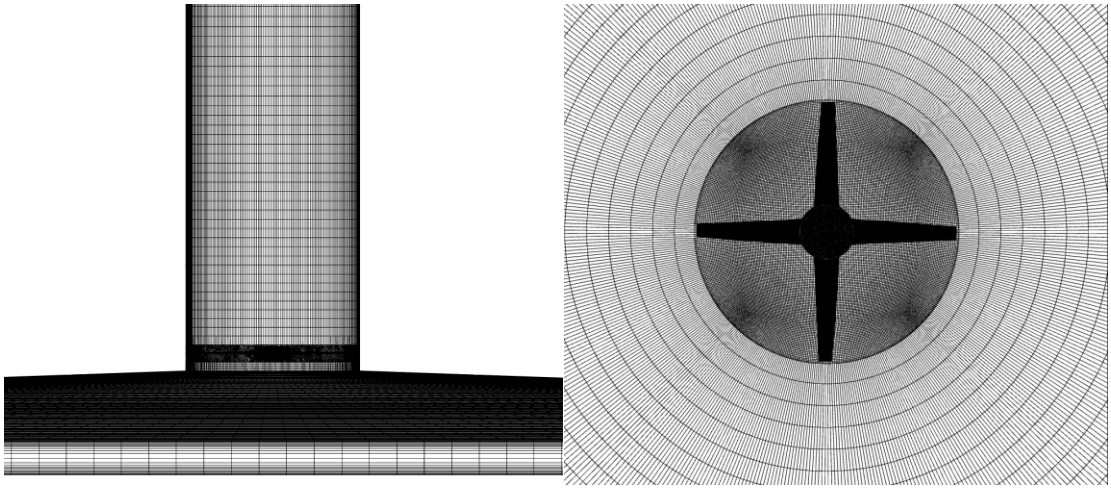
In the pre-processing, ICEM-CFD 19.2 is utilized for the meshing of SCPP-PCR. The computational domain is split into 3 sections: the canopy region, the turbine region, and the chimney region. A hybrid unstructured/structured grid system is implemented. In the turbine region, the chord length and blade installation angle vary with positions. As a result, in the turbine region, the unstructured grids are used, while the structured grids are generated in other areas. The grids near the blade surface are densified to improve the accuracy. Three grid systems with grid numbers of 6.65, 7.58, and 8.57 million, respectively, are used to check the grid independence. The volume flow rates at the outlet of the system are 698.457, 700.2578, and 706.3555 m^3/s, respectively, under the same environmental conditions. as shown in Table 2. A deviation of less than 1.0% demonstrates grid independence. Finally, the model with a total grid number of 7.58 million is adopted for the modeling. Table 3 shows the detail of grid information and Figure 2 depicts the grid diagram.

TABLE 2. Information of the grid independence

Grid numbers	6,654,174	7,584,953	8,576,342
Volume flow rates at the outlet of the system (m^3/s)	698.457	700.2578	706.3555

TABLE 3. Specific information of the grid

Region	Type	Grids($\times 10^6$)	Minimum mesh quality
Canopy	Structured	2.403	0.75
Turbine	Unstructured	1.862	0.3
Chimney	Structured	3.721	0.9



(a) (b)

FIGURE 2. The meshing of SCPP-PCR:
(a) Front view of the grid and (b) Top view of the grid

3.2. Boundary conditions and numerical solution

The ANSYS FLUENT 19.2 is utilized for the numerical solution. Table 4 shows the boundary conditions in the computational domain. In the SCPP-PCR, the entrance and exit of the SCPP-PCR are set as pressure inlet and pressure outlet, respectively. The inlet temperature is assumed to be the same as the ambient temperature (293.15 K). The relative pressures at the entrance and exit are both preset at 0 Pa.[49] When the ambient wind speed does not exceed 2 m/s, the collector surface is considered as a convective boundary condition, with a convective coefficient of 10 W/(m^2·K).[50] The wall of the chimney is regarded as an adiabatic boundary. In the turbine domain, a multiple reference frame (MRF) model is used, with two interfaces combining the rotating and stationary regions. The wall boundaries are set as no-slip walls. The solar radiation is adjusted to the ground surface heat flux after accounting for the heat loss. According to solar radiation data from the desert in northwest China,[51] the heat flux of the ground is

600W/m^2 under the canopy, accounting for G = 857 W/m^2. Assuming that the SCPP-PCR is established in Wuhan, China, the solar radiations for case studies are set to be 285.7, 428.5, 571.4, 714.2, and 857 W/m^2, respectively. The corresponding heat flux of the ground are 200, 300, 400, 500, and 600 W/m^2.

Table 4. The boundary conditions.

Location	Boundary type	Value
Collector inlet	Pressure inlet	p = 0 Pa, T = 293 K
Chimney outlet	Pressure outlet	p = 0 Pa
Ground surface	Heat flux	q = 200 - 600 W/m^2
Collector canopy surface	convection	T = 293 K, h = 10 W/(m^2·K)
Chimney wall	Adiabatic wall	q = 0 W/m^2
Turbine rotational speed		ω = 60-200 rpm

The Boussinesq approximation[52] and incompressible ideal gas model[46] are commonly used density models to solve natural convection. According to Boussinesq approximation, the air density is made up of the sum of the reference density term (ρ_0) and the density change term ($\Delta\rho$) owing to temperature, but this does not apply to multicomponent gases. In this paper, the incompressible ideal gas model is used to is used to simulate the air density in SCPP-PCR due to the photocatalytic reaction of methane. The density calculation formula is: $\rho = \frac{p_{op}}{RT}$ and is only related to temperature. The RNG $k-\varepsilon$ model is selected taking into consideration the buoyancy effect.

For photocatalytic reactions, the chemical reactions occur solely on the surface of photocatalysts (TiO_2). To characterize the rate of chemical reaction, a finite-rate reaction model is adopted. The rate of component convective diffusion to the surface and surface consumption formation is used to compute the concentration of each component at the reaction surface. Only a layer of grid near the photocatalyst surface is involved in the photocatalytic surface reaction. Convective transfer of methane along the vertical direction of the reactor surface is relatively small compared to diffusive transfer because the fluid velocity near the ground is very small. The double precision solver is utilized in numerical calculation. The PRESTO! discretization scheme is used

for pressure term. The second-order upwind scheme is employed for the other terms. The SIMPLEC algorithm is adopted for pressure-velocity coupling. The solution is thought converged when the residuals of each equations are less than 10^{-5}.

3.3. Validation of numerical simulations

According to experimental data supplied by Haaf,[53] when G = 800 W/m^2 and n = 100 rpm, the outlet velocity of the chimney is 9 m/s, the output power of the turbine is 36 kW, and the maximum temperature rise in the system is 17.5 K. In the validation procedure, the environmental conditions of the numerical model are set the same to the experiment of Haff, the results show that the system outlet velocity is 8.97 m/s, the turbine output power is 37.86 kW, and the maximum temperature rise in the system is 18.7 K. As is shown in Table 5, the relative error between the experimental and modeling results are less than 6.9%. The temperature rise and output power of the system are higher than the experimental result because the heat loss through the wall of the SC is neglected.[54] Overall, the numerical results match well with the experimental data, confirming the accuracy of the numerical simulation approach.

TABLE 5. Comparison of numerical simulation data with experimental data

Parameters	Experimental data	Numerical simulation date	Tolerance
Maximum temperature rise (K)	17.5	18.7	6.9%
Chimney outlet velocity (m/s)	9	8.97	0.3%
Output power of turbine (kW)	36	37.86	5.2%

4. Results and discussion

The temperature, velocity, and pressure fields inside SCPP have not varied appreciably as a layer of photocatalyst is coated on the ground in this paper. The fluid field analysis is not presented because there have been some published studies on this.[55] This paper focuses on the methane distribution, the capacity to degrade atmospheric methane, and the photoreactor reaction rate of the SCPP-PCR. Furthermore, the atmospheric CH_4 degradation potential is compared to that of SCPP integrated with

honeycomb photoreactor,[41] and CO_2 emission reduction is briefly analyzed.

4.1. Atmospheric methane degradation

Figure 3 shows contours of the CH_4 concentration in the axial plane of the system when G=857 W/m^2 with different turbine rotational speeds. The methane concentration in the atmosphere is 1886 ppb. Because of the stack effect caused by the SCPP, the methane is continually sucked into the system. The oxidation reaction occurs when the CH_4 contacts with the titanium dioxide painted on the ground under solar radiation. The CH_4 is continually degraded on the ground in the radial direction of the collection. The air velocity in the collector is very small, thus the mass convective process is relatively weak. The methane in the mainstream has to diffuse down to be degraded on the photocatalyst surface. The methane concentration under the wheel is zero. The rotation of the turbine continues to impact the distribution of methane inside the system after it passes through the turbine. But the methane distribution becomes more uniform as the flow moves upward. The concentration of the methane at the exit is 269.2 ppb, about 14.2% of that in the atmosphere.

When the speeds of the turbine rotational are 100, 120, and 140 rpm, respectively, the corresponding methane concentration at the output are 226.0, 144.8, and 141.5 ppb, respectively. The methane content at the exit drops with the increase of turbine rotational speed. Simultaneously, the turbine rotation speed has a substantial impact on the distribution of methane in the chimney. The slower the rotation speed, the faster the methane is mixed uniformly.

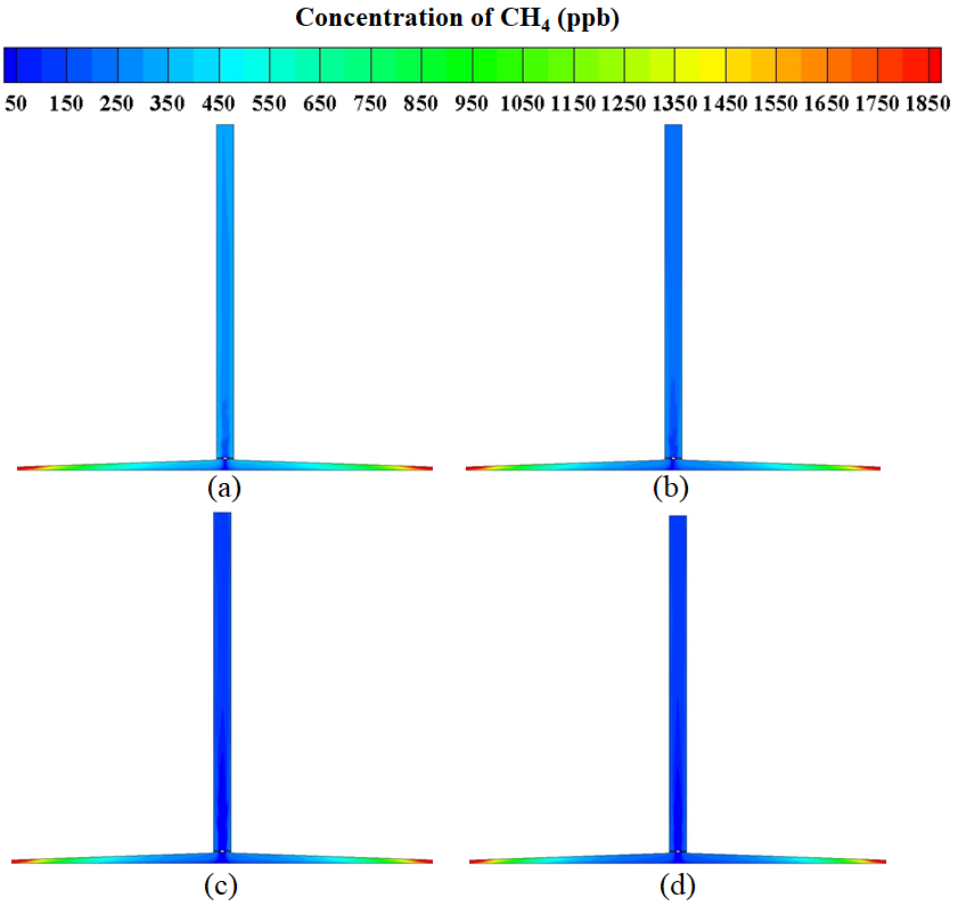


FIGURE 3. Contours of the CH_4 concentration in the axial plane of the system with G = 857 W/m^2. (a) n = 80 rpm, (b) n = 100 rpm, (c) n = 120 rpm, (d) n = 140 rpm.

Figure 4 shows contours of the CH_4 concentration 0.01 m above the ground painted with photocatalyst when G = 857 W/m^2. The concentration gradient in the radial direction of the collector is considerable due to the photocatalytic reaction. It is found that the methane concentration drops faster at the reactor entrance of the reactor. The reaction rate of methane is relatively fast in this region where the concentration of methane is high and the flow velocity there is very slow.

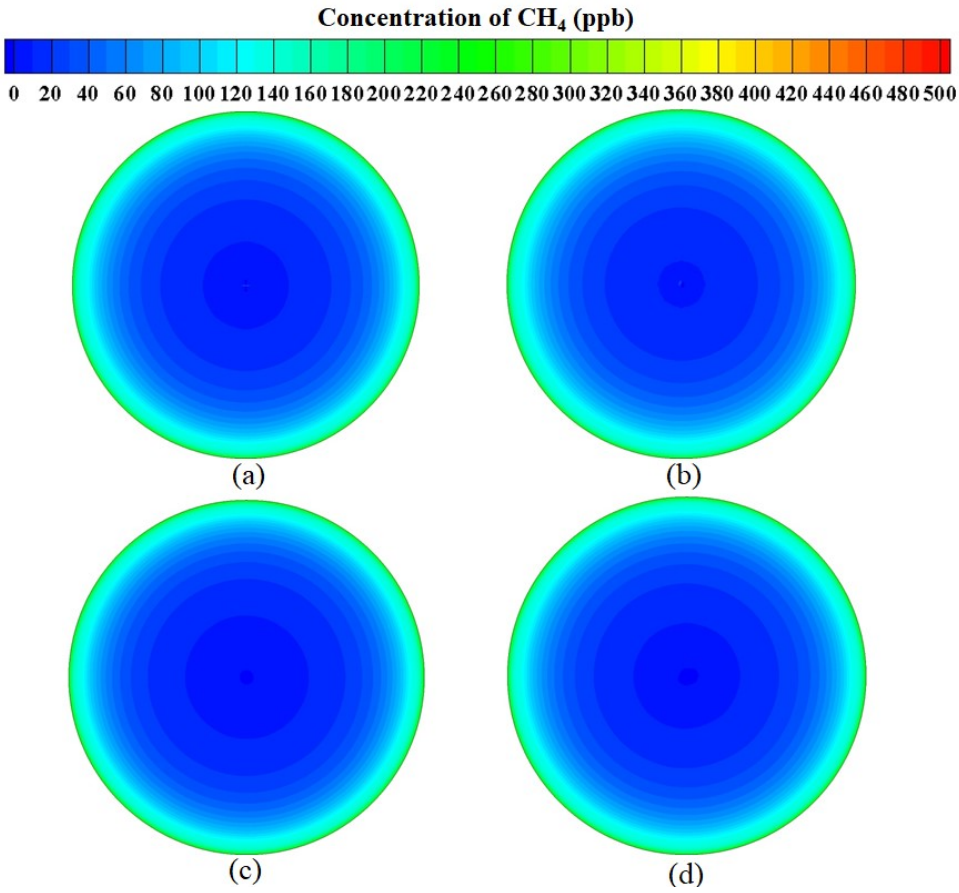


FIGURE 4. Contours of the CH_4 concentration 0.01 m above the ground with G = 857 W/m^2. (a) n = 80 rpm, (b) n = 100 rpm, (c) n = 120 rpm, (d) n = 140 rpm.

The photocatalytic efficiency and purification rate of the methane are used to characterize the CH_4 removal performance of the SCPP-PCR. Figure 5 shows the effect of turbine rotational speed on the photocatalytic efficiency of CH_4. The rate of electron and hole creation is impacted by the intensity of the light, which in turn influences the rate of photocatalytic reaction.[56] When the rate of electron and hole creation exceeds that of the photocatalytic reaction, the rate of the reaction is independent of light intensity and is primarily influenced by mass transfer rate. In the SCPP-PCR, the complex cooperative process of mass transfer is influenced by both solar radiation and turbine rotation. When G = 428.5 W/m^2, the airflow velocity in the system is moderate, and mass transfer rate on the surface of the photocatalyst is influenced by the turbine speed. The mass transfer rate between 100 and 120 rpm is greatly increased, as well as the methane photocatalytic efficiency.

Figure 6 shows the influence of turbine rotational speed on the purification rate of

methane. As the rotating speed of the turbine increases, the rate of methane purification drops, photocatalytic efficiency improves, but the mass flow rate of SCPP-PCR is decreased. It can be noticed that the mass flow rate of the system has a more significant impact on the methane purification rate. And the quantity of methane degradation per unit time is relatively high under high solar radiation. For example, the maximum purification rate of CH_4 of the SCPP-PCR is 0.905 g/s when G = 857 W/m^2.

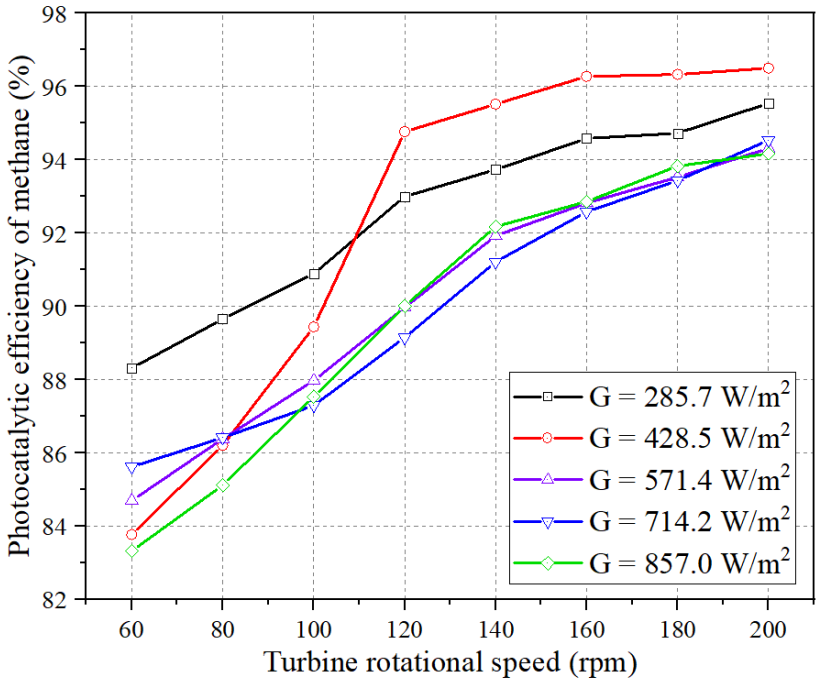


FIGURE 5. Effect of turbine rotational speed on photocatalytic efficiency of methane

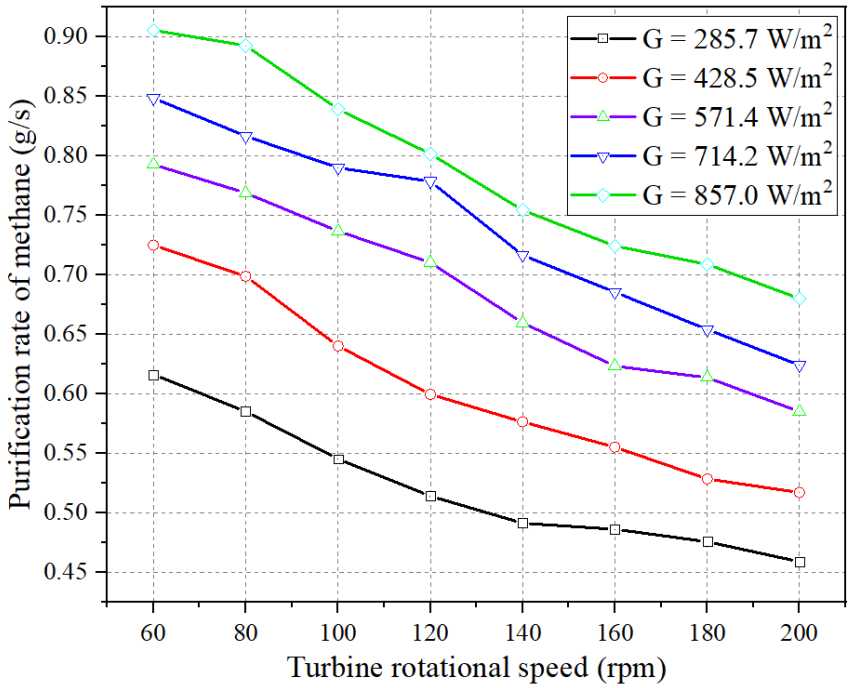


FIGURE 6. Effect of turbine rotational speed on purification rate of methane

4.2 Reaction rate in photoreactor

Furthermore, considering the initial investment cost of PCR, it is critical to investigate the reaction rate inside the photocatalytic reactor to determine the titanium dioxide laying length. The reaction rates on the surface of catalyst with various turbine rotational speeds with G = 857 W/m^2, as illustrated in Figure 7. The reaction rate reduces drastically within 1 m of the reactor entrance and then keeps at a low level. Though the rotational speed varies, the reaction rate remains almost constant within 8.06 m of the entry, peaking at 44.39×10^{-9} mol/(m^2·s) at the entrance, and 3.44×10^{-9} mol/(m^2·s) at 8.06 m. The lower the turbine rotational speed, the higher the reaction rate after 8.06 m in the same position. This is due to the fact that the concentration of methane is high and the contact time with the photocatalyst is lengthy in the first 8.06 m at the reactor entrance, therefore the reaction rate is basically the same. After 8.06 m, the lower the turbine rotational speed, the higher the mass flow rate of the system, i.e., the more methane enters the system per unit time. In this case, in the methane concentration is slightly higher, which results in a slightly higher reaction rate.

Figure 8 shows the CH_4 reaction rate on the surface of the catalyst with different solar radiation under a turbine rotational speed of 100 rpm. It shows that the higher the

solar radiation, the faster the reaction rate is at the same position. The reaction rate of CH_4 declined dramatically within 1 m of the reactor entrance, despite differences in the solar radiation intensity. The maximum reaction rates at the inlet are 44.39, 36.99, 29.59, 22.19, and 17.49×10^{-9} mol/(m^2·s), respectively, when solar radiation is 285.7, 428.5, 571.4, 714.2, and 857 W/m^2, respectively. The photocatalytic reaction rate is lower than 1×10^{-9} mol/(m^2·s) at the position 40 m away from the entrance. If the start of the PCR is placed at the entrance of the system, it is cost-effective to paint the photocatalyst for the first 40 m in the radial direction, giving the photocatalytic reaction zone a total area of 25,132.74 m^2.

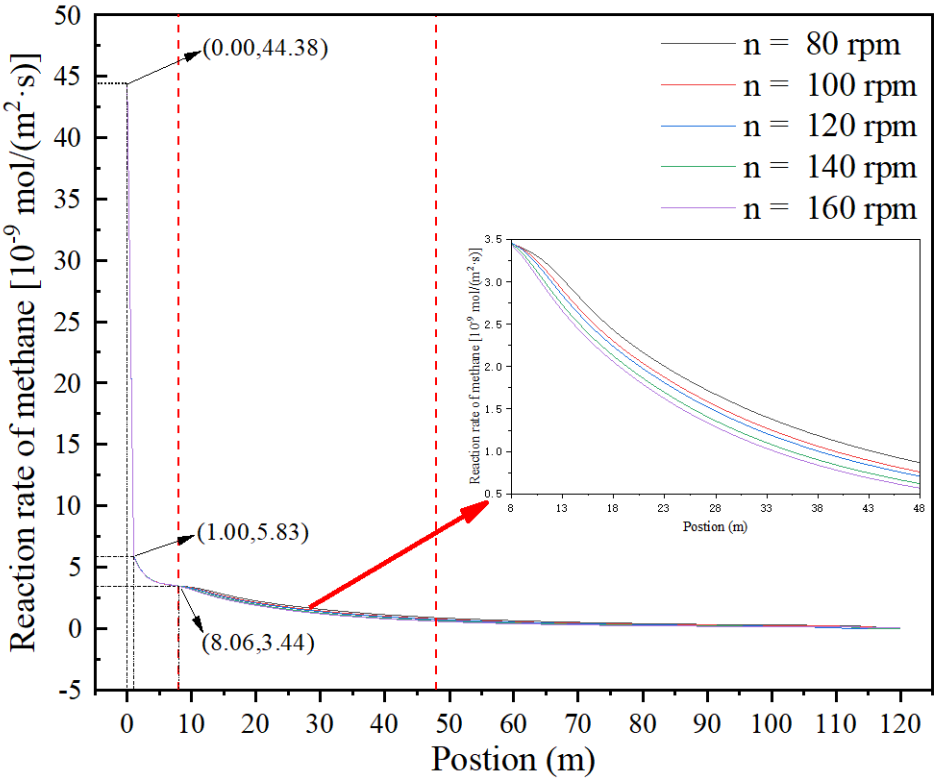


FIGURE 7. Photocatalytic reaction rate on the surface of catalyst with various turbine rotational speeds under G = 857 W/m^2

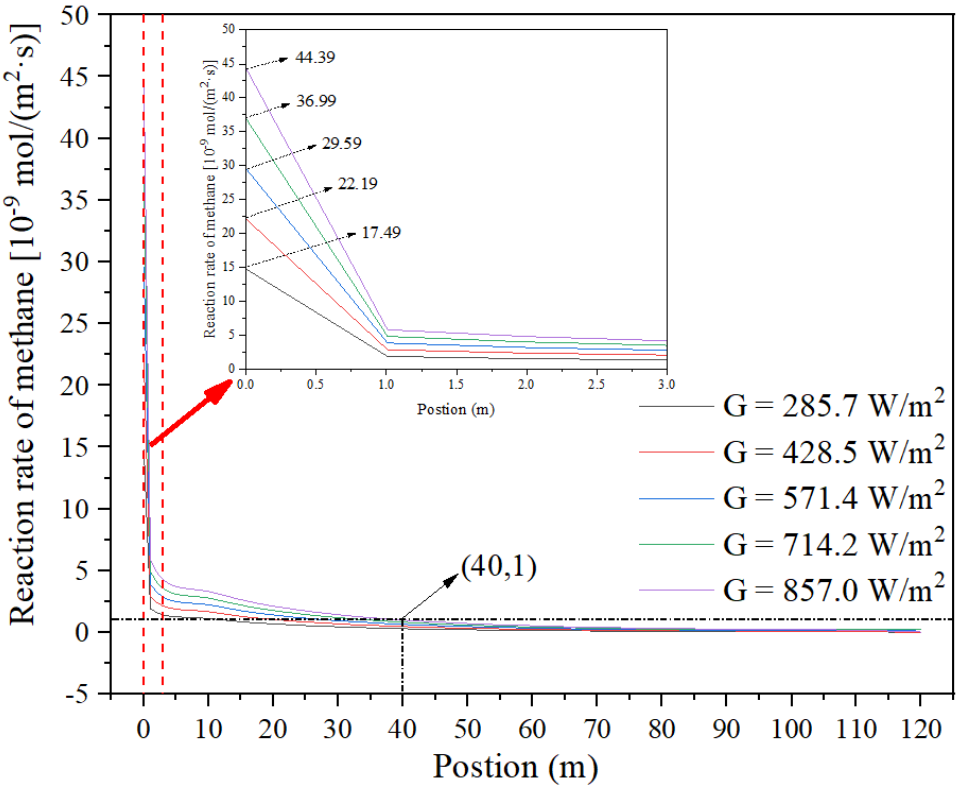


FIGURE 8. Photocatalytic reaction rate on the surface of catalyst with various solar radiation under n = 100 rpm

4.3. Comparison of different types of photoreactor

The types of photoreactors have significant influence on the ability of the system to degrade atmospheric methane. According to the numerical simulation results,[41] in the case of no load, the photocatalytic efficiency of methane can be improved to more than 95% by extending the length of the reaction zone, despite the use of honeycomb reactors with variable pore diameters. Therefore, the comparison of methane photocatalytic effects between SCPP with plate photoreactor or the honeycomb photoreactor (d_{pore}=4 mm) system is shown in Table 6, under G=857 W/m^2.

The methane will stay on the catalyst surface longer if the turbine rotation speed or the length of the honeycomb photocatalytic reactor is increased, resulting in more complete catalysis. Though increasing the length of the honeycomb improves methane catalytic efficiency, the long reactor creates a large pressure drop, reducing the mass flow rate of the SCPP-PCR and resulting in a poor methane purification rate. For example, when the length of the honeycomb reactor is 8 m, the catalytic efficiency of

methane is 93.46 %, and the pressure drop is 149.99 Pa, resulting a purification rate of 0.68 g/s. Though the catalytic efficiency of the plate photoreactor is increased to97.03% when the reactor length is extended to 10 m, the methane purification rate reduced to 0.66 g/s as the mass flow of the system is significantly reduced.

The methane photocatalytic process is normally carried out in numerous phases.[57] The honeycomb photoreactor can achieve full oxidation of methane by prolonging the reaction time. However, the large internal pressure drop in the honeycomb photoreactor restricts the power output of the SCPP. In this paper, both the complete oxidation of methane and power generation are of great importance to reduce CO_2 emissions. It is thought that a SCPP integrated with a plate photoreactor is more ideal, as it can not only purify methane but also create energy.

TABLE 6. Degradation of atmospheric methane by coupling different types of photoreactors

Plate			Honeycomb			
Turbine rotational speed (rpm)	Photocatalytic efficiency (%)	Purification rate (g/s)	Length of the PCR (m)	Photocatalytic efficiency (%)	Purification rate (g/s)	Pressure of the PCR (Pa)
60	83.32	0.91	3	59.46	0.52	80.45
80	85.12	0.89	4	71.26	0.59	97.48
100	87.53	0.84	5	80.11	0.63	110.53
120	90.01	0.80	6	86.23	0.66	123.92
140	92.18	0.75	7	90.38	0.67	138.99
160	92.86	0.72	8	93.46	0.68	149.99
180	93.83	0.71	9	95.59	0.67	160.29
200	94.17	0.68	10	97.03	0.66	169.87

4.4. CO_2 emission reduction technology

Hydropower, thermal power, and nuclear power were the primary sources of electricity at the end of the 1980s.[58] The position of the three primary kinds differed depending on countries. At the moment, thermal power generation dominates the world, accounting for more than 70% of overall power generation. Thermal power is created by burning fuel, which emits tremendous CO_2. Muangthai et al[59] investigated the relation of power generation and carbon dioxide emission of thermal power plant in Thailand, and the link between calculated unit power generation and carbon dioxide

emission is presented in Figure 9. They found that 0.6 kg of carbon dioxide is released from a thermal power plant for every 1 kWh of electricity produced.

The SCPP is a green system that generates electricity from solar energy without emitting CO_2. Figure 10 depicts the influence of turbine rotation speed on turbine output power and CO_2 emission reduction under various solar radiation conditions. They both follow the same pattern, rising first and then falling. With the turbine rotational speed increasing, the mass flow rate of the system falls, resulting in a drop in the torque of the blade. The output power of a turbine is the product of the turbine rotational speed and the torque, which reaches a peak value at certain turbine rotational speed. The increase in solar radiation improves the output power of SCPP-PCR, easing the burden on the thermal power plant and reducing CO_2 emissions. For example, turbine output power reaches a maximum of 80.49 kW and carbon dioxide emission reduction is 48.29 kg/h when solar radiation is 857W/m^2 and turbine rotation speed is 200 rpm. Electricity produced from non-fossil fuels utilizing such a clean power generation system reduces CO_2 emissions significantly.

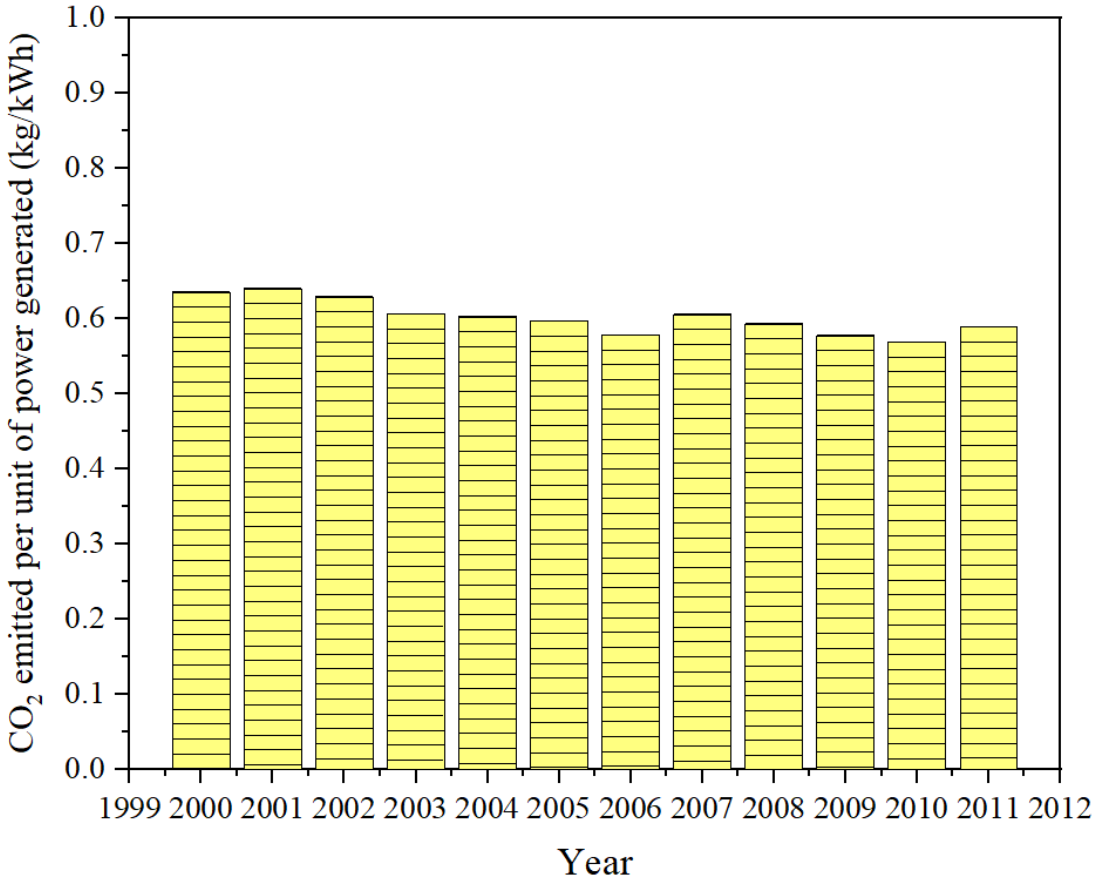


FIGURE 9. CO_2 emissions from thermal power plant unit power generation

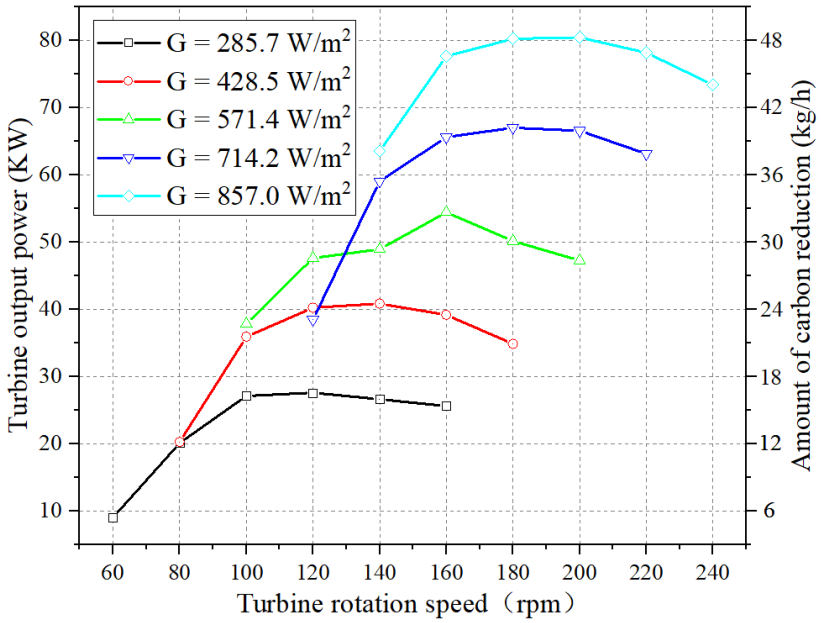


FIGURE 10. The variation curves of turbine output power and CO_2 reduction

5. Discussion

This study is inspired by the limits of present methods to tackle global warming. The investment of the SCPP-PCR and the methane degradation capability are estimated as follows.

It costs around $1.25 million to build the SCPP, including the costs for glass, concrete, power equipment (turbines, gearboxes, and power generators), and installation.[60] P25 is a white nano-sized titanium dioxide powder with a density of 3,900 kg/m^3 and a particle size of 25 nm in average. It is a cost-effective photocatalyst with a long lifetime, high stability, and difficult deactivation properties.[61] In the SCPP system, the P25 is painted on the ground. To achieve the best photocatalytic effect, the PCR is expected to consume 50 g photocatalyst per square meter.[37] For a SCPP-PCR with a reaction zone of 25,132.74 m^2, a total of 1.39 tons of photocatalyst paint is used. The current price of P25 powder is $3,300 per ton,[37] therefore the total investment of the PCR is about $4,587, approximately 0.37% of the total investment of the SCPP-PCR system.

The calculation is done based on the solar radiation data of Wuhan on July 17,

2011.[62] The solar radiation from 6:00 a.m. to 6:00 p.m., with the corresponding values are 155, 535, 820, 883, 720, and 385 W/m^2, respectively, are input data for the analysis. Table 7 shows the volume flow rate, methane purification rate, output power of turbine, CO_2 reduction of the SCPP-PCR when n = 100 rpm.

TABLE 7. Calculation results in Wuhan on July 17, 2011

Time (Local Beijing time) (July 17, 2011)	Solar radiation (2 hours average) (W/m^2)	Volume flow rate （m^3/s）	Methane purification rate(g/s)	The output power of turbine (KW)	The CO_2 reduction(kg/h)
6:00 am–8:00 am	155	397.49	0.44	16.48	9.89
8:00 am–10:00 am	535	682.87	0.70	37.83	22.70
10:00 am–12:00 am	820	841.79	0.84	38.27	22.96
12:00 am–2:00 pm	883	847.12	0.87	39.76	23.86
2:00 pm–4:00pm	720	784.04	0.79	38.47	23.08
4:00 pm–6:00pm	385	575.88	0.61	33.67	20.20

The simulations demonstrate that the SCPP-PCR can degrade methane in the atmosphere on a large scale. The system can reduce 245.38 kg CO_2 emissions generated from thermal power in the daytime. Furthermore, the system can process 29,730,199.61m^3 of air and purify 30,595.47 g of methane in one day, which can significantly mitigate the greenhouse effect. The methane purification rate can be further enhanced by developing a night operation strategy. For example, the SCPP can work continuously by employing a heat storage layer, and the ultraviolet irradiation for the photocatalysis can be provided by the electricity generated by the SCPP itself. In this way, the methane purification rate might be doubled.[39]

6. Conclusion

In this study, the photocatalytic performance and the carbon dioxide reduction of the SCPP-PCR are analyzed by conducting three-dimensional steady-state CFD simulations. The feasibility of the SCPP-PCR for both clean electricity generation and

large-scale methane removal is analyzed. The following are conclusions:

(1) The solar chimney power plant integrated with plate photoreactor is more optimal than a honeycomb photoreactor when neglecting the influence of the wind environment on the system. The honeycomb photoreactor provides higher methane conversion efficiency, but correspondingly high internal resistance. Meanwhile, the plate photoreactor provides both considerable methane purification rate and produce power.

(2) The reaction rate reduces drastically within 1 m of the entrance of the PCR and then slows down gradually. It is cost-effective to paint the photocatalyst at the entrance of the SCPP for the first 40 m along the radial direction.

(3) According to the simulation results, the photocatalytic efficiency increases with the turbine rotational speed and the maximum efficiency reaches 96.5%. However, the methane purification rate decreases with the turbine rotational speed and the maximum rate is 0.91 g/s. The system can process 29,730,199.61m^3 of air, purify 30,595.47 g of methane, and reduce CO_2 emissions by 245.38 kg in a day at Wuhan, China.

Acknowledgements

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