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Lightning Impulse and AC Breakdown Characteristics of SF₆ and its Alternatives

P. Ranjan¹, Q. Han¹, F. O. Bahdad¹, A. Alabani¹, L. Chen^{1*}, I. Idrissu², and L. van der Zel³

¹Department of Electrical and Electronic Engineering, The University of Manchester, Manchester, M13 9PL, UK

²National Grid Electricity Transmission plc, 1-3 Strand, London, WC2N 5EH, UK

³Power Delivery and Utilization, Electric Power Research Institute, NC, 28262-7097, USA

*lujia.chen@manchester.ac.uk

Abstract- SF₆ alternatives with significantly lower environmental impact have been extensively researched. This paper investigates two electrode configurations and test pressures to mimic the conditions as found in MV/HV equipment to systematically compare the breakdown performance of SF₆, C₃H₂F₄ and a mixture of 30% CF₃I / 70% CO₂ under AC and LI stresses. SF₆ possesses comparatively better breakdown performance than the other alternatives investigated in this study. C₃H₂F₄ shows “first breakdown” behavior and only the stable data series is considered for evaluation of 50% breakdown voltage, U_{50} . The lower boiling points of C₃H₂F₄ and CF₃I limit its potential application to medium-voltage equipment with a low operating pressure. Due to pyrolysis, there is formation of soot and iodine post-breakdown in C₃H₂F₄ and 30% CF₃I / 70% CO₂ respectively. This will reduce their dielectric performance and could pose health hazard to maintenance personnel.

I. INTRODUCTION

High global warming potential (GWP) of SF₆ and the drive towards Net Zero have resulted in an increasing interest for insulating gases with low GWP. Different studies were carried out to phase out the usage of SF₆ either as a newbuild or a retro-fill solution [1] in a functional SF₆ equipment.

CIGRE WG D1.51 [2] studied the feasibility of natural origin gases such as air and CO₂. However, the comparatively low dielectric strength of these gases, about one third of SF₆, makes it a less viable option especially for high-voltage (HV) equipment. Most of the new alternatives have the drawback of high boiling point, which can be mitigated through the use of CO₂ or N₂ as a buffer gas to avoid liquefaction.

3MTM developed NovecTM insulating gases, namely, C₃F₇CN (C4-FN or NovecTM 4710) and C₅F₁₀O (C5-FK or NovecTM 5110) which have been widely researched with newbuild equipment installed in the network [3]. C₃H₂F₄, also known as HFO1234ze(E) is researched as a potential SF₆ alternatives but generates soot formation [4] due to pyrolysis under electrical discharges. CF₃I is another candidate investigated as a mixture with CO₂ or N₂ [5]. The aforementioned studies provide a quantitative study on individual gases or mixtures but do not compare these gases under identical experimental conditions.

As part of CIGRE WG D1.67 [6], a round-robin investigation was carried out on the insulating properties of different alternatives such as C₃H₂F₄, CF₃I, C₃F₇CN, C₅F₁₀O against SF₆. Researchers at The University of Manchester focused on breakdown performance of C₃H₂F₄ and 30% CF₃I /

70% CO₂ against SF₆ under AC and LI voltage stresses. Experimental setup and the breakdown results are detailed in Sections II and III respectively. The visual inspection of gas degradation post-breakdown is discussed in Section IV.

II. EXPERIMENTAL SETUP

A hemispherical-rod to plane configuration was used with a gap of 10 mm and a hemispherical rod diameter of 7 mm. This is to mimic the field uniformity of contacts in 11 kV ring main unit. A sphere to plane configuration with a gap of 15 mm and a sphere diameter of 40 mm was used to represent the field uniformity in HV equipment. The field utilization factor, f , of the hemispherical-rod to plane and sphere to plane electrode configurations is 0.297 and 0.628 respectively. The total height of the roughness (R_t) of the HV electrodes, i.e., hemispherical-rod and sphere, were 50 μ m and 20 μ m, respectively. Gas pressures of 1.3 and 6 bar were tested to mimic the typical operating pressures of medium-voltage (MV) and HV gas-insulated equipment, respectively. All pressures in this paper are absolute pressure in bar. Table I shows different gases tested with their corresponding test parameters. Table II shows the dew temperature of tested gases with the corresponding critical pressure reduced electric field strength (E/p in kV/mm/bar). C₃H₂F₄ liquifies at room temperature under 6 bar and its dielectric performance was evaluated at 1.3 bar only.

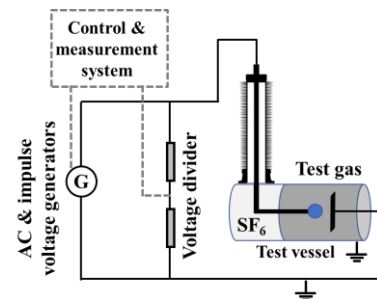


Fig. 1. Schematic of experimental setup

Spacers of 10 and 15 mm thickness were fabricated to accurately set the inter-electrode gap spacing prior to any experiment. A stainless-steel pressure vessel was connected to the HIGHVOLT 800 kV AC test set or BHT 800 kV impulse generator as shown in Fig. 1. The voltage dividers were used to measure the breakdown voltages. The bushing side of the

vessel is always filled with SF₆. Vessel was filled with the test gas after (a) vacuuming, (b) filling CO₂ to remove residual moisture, and (c) re-vacuuming the vessel down to 1 mbar. In the case of a single test gas, a direct filling procedure is adopted. For gas mixture, the manometric method is used where 30% of the total required pressure was first filled with CF₃I and then topped up with CO₂.

TABLE I EXPERIMENTAL PARAMETERS

Configuration	f	Gas type	Voltage stress	Pressure (bar)
Hemispherical-rod-to-plane	0.297	SF ₆ , C ₃ H ₂ F ₄ , 30% CF ₃ I / 70% CO ₂	±LI, AC	1.3
Sphere-to-plane	0.628	SF ₆ , 30% CF ₃ I / 70% CO ₂	±LI, AC	6

TABLE II CALCULATED DEW POINT AND E/p OF THE TESTED GASES

Test gas	Dew temperature (°C)		E/p* (kV/mm/bar)
	1.3 bar	6 bar	
SF ₆	-62.6	-24.6	8.9
C ₃ H ₂ F ₄	-14.1	30.8	7.1
30% CF ₃ I / 70% CO ₂	-41.6	-4.4	5.5

*Values for 1 bar pressure under uniform electric field

Procedures detailed in [7] were followed for the tests in this paper with a time interval of 2 minutes between successive voltage applications. Each test series consists of 100 voltage applications. For LI tests, up-and-down method with a step voltage of 2.5% of the first breakdown was used to evaluate U_{50} , and standard deviation, σ using equations provided in [8]. For AC tests, a continuous rising method was used with a fast ramp rate of 5 kV/s for the first 40% and a slow ramp rate of 0.5 kV/s for the last 60% of the prospective breakdown voltage. The mean value of the 100 breakdowns was designated as U_{50} for AC tests.

III. RESULTS AND DISCUSSION

Fig. 2 shows the applied voltage sequence for all three gases using the hemispherical-rod to plane electrode. For C₃H₂F₄, the data trend can be seen to have less variation when compared to that for SF₆ and 30% CF₃I / 70% CO₂, especially under +LI. This is due to the soot formation after the first breakdown [4] occurred for both LI polarities. The first breakdown behavior for C₃H₂F₄ is illustrated in Fig. 3 where the data series is more stable after the first withstand following the first breakdown and multiple voltage collapses all chopped on the front. This unique first breakdown for C₃H₂F₄ was prominent under +LI and can be double of the U_{50} value. For the calculation of U_{50} and σ of C₃H₂F₄, the final stabilized 100 voltage applications were used as shown in Fig. 2 and as highlighted in yellow in Fig. 3.

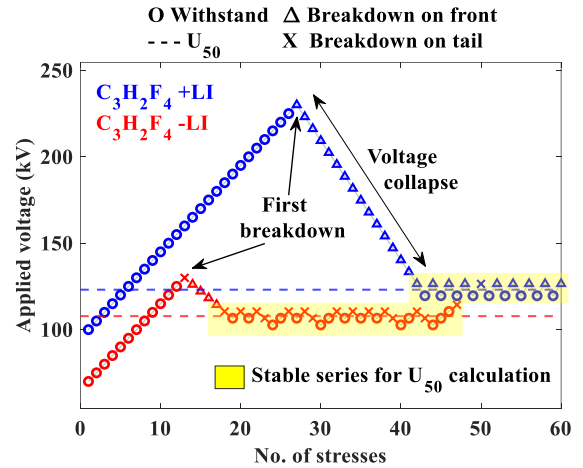


Fig. 3. Examples of LI test sequence for C₃H₂F₄ using 25 mm dia. hemisphere to plane configuration with a gap of 10 mm and under a pressure of 2 bar [4]

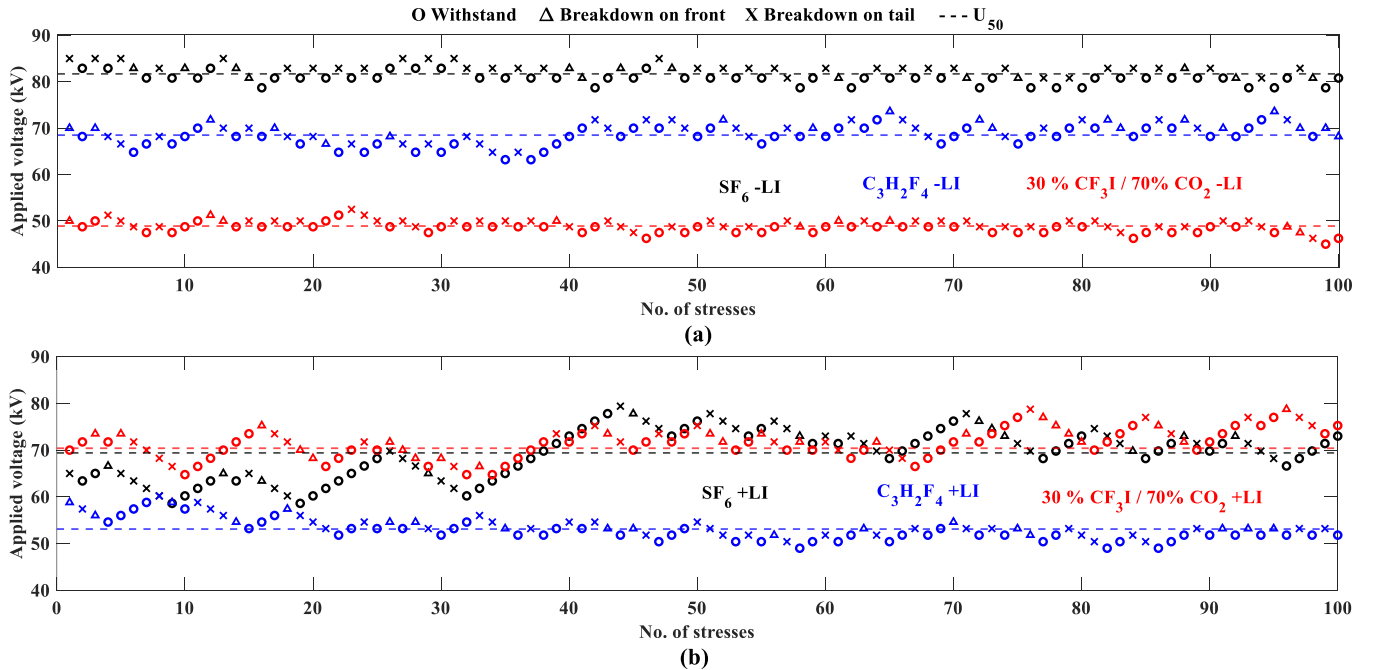


Fig. 2. LI test sequences for different gases under (a) -LI and (b) +LI, using a hemispherical-rod to plane configuration with a fixed gap of 10 mm and a pressure of 1.3 bar

For -LI, the data sequence is comparatively stable with lower σ than +LI as shown in Fig. 4. For $C_3H_2F_4$ and SF_6 , higher U_{50} values were observed for -LI on account of a positively headed space charge at the HV electrode which reduces the electric field [4]. However, 30% CF_3I / 70% CO_2 exhibit higher U_{50} in +LI than -LI. Similar polarity effect was reported in [5] using coaxial geometry. The opposite polarity effect was observed when the field is non-uniform and tested at a comparable pressure [9].

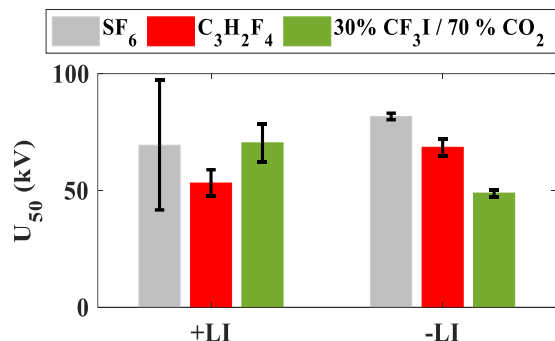


Fig. 4. Plot of U_{50} and σ for different gases using a hemispherical-rod to plane configuration with a gap of 10 mm and under a pressure of 1.3 bar

Impulse Ratio Calculations

Impulse ratio is defined as the ratio of withstand/breakdown LI voltage to the corresponding AC_{peak} withstand/breakdown voltage. This is an engineering value that helps with design margin consideration for gas insulated systems [2] to fulfill the specified withstand voltages. The highlighted “yellow” region in Fig. 5 shows the limits specified in IEC 62271-1 [10]. Data points show the impulse ratios from the experimentally obtained values as depicted in Table III.

When the experimental obtained ratios are higher than the IEC defined impulse ratio range, the insulation system design is dominated by AC. Vice versa, LI voltage is more important if the experimentally obtained ratio is below the IEC impulse ratio range. For 1.3 bar, -LI values are within the IEC impulse ratio limit while for +LI, it is lower except that of 30% CF_3I / 70% CO_2 . For 6 bar, all the experimental ratios are lower than that of IEC ones. These values indicate that the insulation system design for MV (1.3 bar) or HV (6 bar) applications will be decided by LI withstand tests for the tested gases. This also agrees with [2], where dry air, N_2 , CO_2 and SF_6 insulated systems until 7 bar, are dominated by LI voltages.

TABLE III. U_{50} FOR TESTS OBTAINED IN THE PRESENT WORK

Hemispherical-rod to plane / 10 mm / 1.3 bar					
Test gas	+LI (kV)	-LI (kV)	AC_{peak} (kV)	Impulse Ratio	
				+LI	-LI
SF_6	69.4	81.7	52.3	1.33	1.56
$C_3H_2F_4$	53.1	68.5	47.9	1.11	1.43
30% CF_3I / 70% CO_2	70.4	48.9	29.9	2.35	1.64
Sphere to plane / 15 mm / 6 bar					
SF_6	489.6	408.5	368.3	1.33	1.11
30% CF_3I / 70% CO_2	332.7	279.3	271.4	1.23	1.03

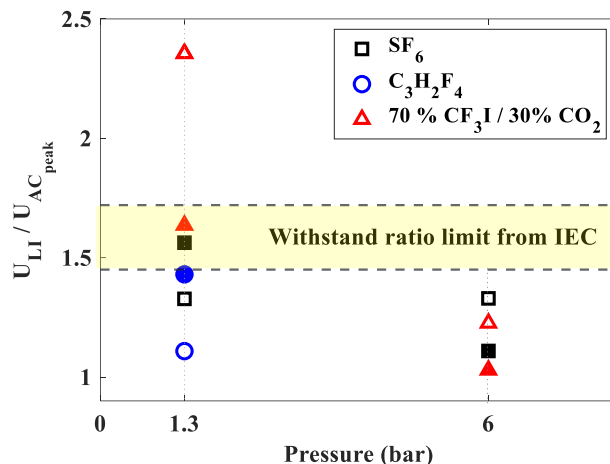


Fig. 5. Ratio of U_{50} under LI and AC voltage stresses for the investigated test conditions in the present work (unfilled markers: +LI; filled markers: -LI)

IV. DECOMPOSITION AFTER EXTENSIVE AC BREAKDOWNS

For SF_6 , there was no obvious decomposition observed unlike $C_3H_2F_4$ and 30% CF_3I / 70% CO_2 . The byproducts after the tests and during the cleaning of electrodes/vessel were photographed as detailed in subsequent paragraphs.

Soot Formation in $C_3H_2F_4$

Fig. 6 shows the photographs of test electrodes after 100 AC breakdowns using $C_3H_2F_4$ under 1.3 bar. The dark colored soot deposited on the electrodes was identified as fluorinated carbon due to pyrolysis. Experiments carried out in [4] showed soot formation which did not appear to react with the metal electrode surface and can be easily wiped off. The formation of soot can be minimized with the addition of O_2 . Considering the boiling point of $C_3H_2F_4$, it can be used as an insulation medium in non-switching MV equipment, but one must take into account the soot formation and its potential adverse effect on insulation performance.

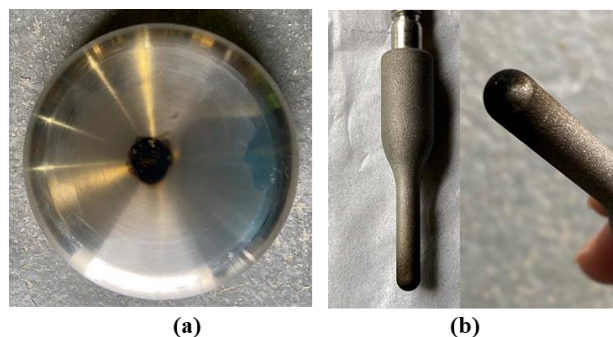


Fig. 6. Photos taken after 100 AC breakdown of $C_3H_2F_4$ under 1.3 bar: (a) plane electrode and (b) hemispherical-rod electrode

Iodine Formation in CF_3I

In the case of 30% CF_3I / 70% CO_2 , extended AC breakdowns under 6 bar caused the colorless gas to turn

V. CONCLUSIONS

orange/red which was not observed in other test gases. Note that the discharge energy of the AC generator used was ~10 times higher than the LI generator used in this study. For 30% CF_3I / 70% CO_2 under 6 bar, the colorless CF_3I gas in the test vessel changed to orange/red color as shown in Fig. 7(a). It is the characteristic color of iodine vapor and was reported in [11] after electrical discharges. The electrodes shortly after opening the test vessel are shown in Figs. 7(b) and 7(c). The plane electrode showed different colors of red, brown, and purple. Sphere and plane electrodes appear moist when first taken out of the vessel. This may be due to hygroscopic nature of iodide salts or the degradation product on the surface of the electrode reacting with moisture in the open-air.

The electrodes and test chamber were cleaned with the isopropanol and cleaning paper. Different colors were visually observed on the cleaning paper. The yellow color on the paper may be due to the free iodine in the test vessel. The red/purple color on the plane electrode and the cleaning paper indicates the precipitation of iodine on the electrode surface. This was confirmed by Fig. 7(d) where brown colored solid precipitates can be seen on the sphere electrode after being exposed to open-air for several days.

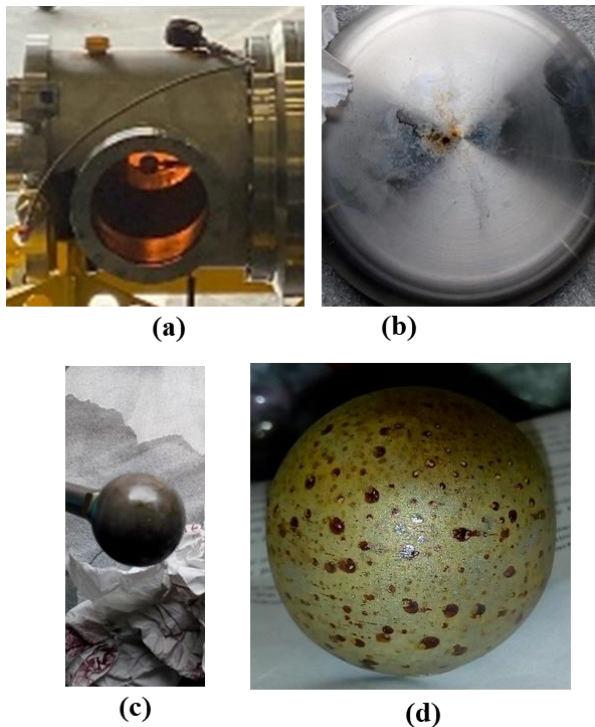


Fig. 7. Photos taken after 100 AC breakdowns of 30% CF_3I / 70% CO_2 under 6 bar: (a) test chamber with orange/red colored gas, (b) plane electrode, (c) 40 mm sphere electrode; (d) different colored deposits found on the cleaning papers and (e) sphere electrode after several days in open-air environment

Iodine decomposition of 30% CF_3I / 70% CO_2 mixture poses considerable challenge for its use in MV/HV equipment. The gas handling for this two-component gas needs to be carried out with care avoiding ingress of moisture and/or O_2 . The chamber must be cleaned with appropriate personal protective equipment such as hand gloves and full-face respirator, and preferably to be carried out in a well-ventilated area.

For breakdown performance evaluation under LI tests, a stable data series should be assessed after the pre-conditioning. For all three tested gases, -LI test data had comparatively less voltage levels in up-and-down test than +LI. The U_{50} values will vary depending on the gas medium, field configurations and especially the polarity of applied voltage. This indicates that gases must be evaluated for both polarities to determine the insulation system design parameter.

The comparatively higher boiling point limits the application of $\text{C}_3\text{H}_2\text{F}_4$ and CF_3I to MV non-switching equipment. The formation of by-products such as fluorinated carbon soot and iodine post-breakdowns present operational risk of surface flashover occurrence at significantly reduced voltage withstand level due to solid precipitation along the insulator.

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