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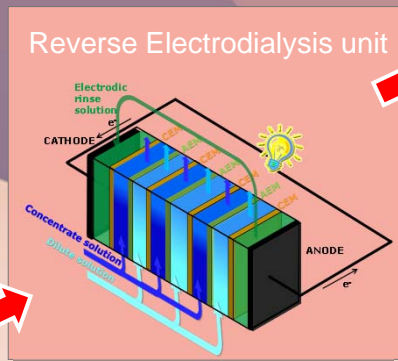
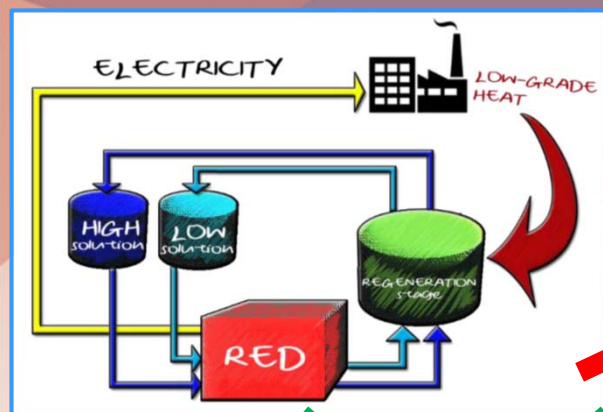
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INTRODUCTION: The urgent need to reduce dependency on fossil fuels and concerns of the negative impact of many dated and unsustainable industrial processes on the environment are topics of crucial importance nowadays. These have led the interest towards heat recovery technologies for the utilisation of waste heat from industrial processes to continuously grow. Only some heat engines are able to operate at low temperature level (<100°C), but none of them is able to achieve satisfactory value of efficiency. In this regard, a closed cycle based on the use of **Reverse Electrolysis** is theoretically able to operate at very low temperature will be presented in the following.

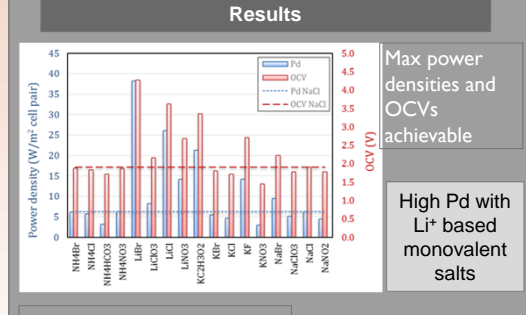
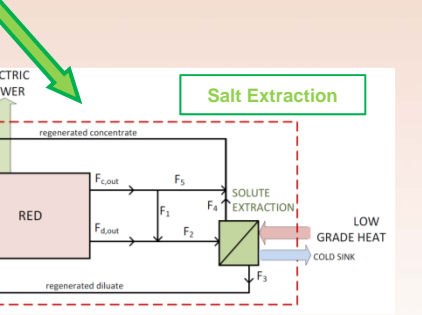
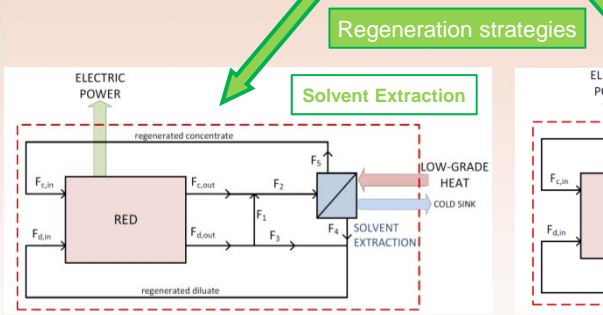


RED optimization via modeling

Performance of the RED unit with different salts

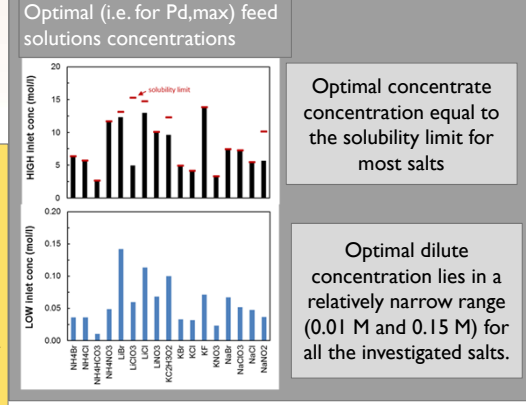
RED model assumptions:

- all solutions' and cell pairs' variables are evaluated at the average conditions between inlet-outlet of the feed channels;
- membrane properties (permselectivity, resistance) are kept constant (i.e. independent of the solute employed);
- solvent transport through membranes is neglected;
- polarization phenomena are neglected;
- ideal current distribution (i.e. no parasitic currents)



The amount of salt transferred in the RED stack from the concentrate solution to the diluted one is rebalanced by the preliminary addition of stream F_1 to F_2 . Then stream F_3 is subjected to the necessary solvent excess extraction which results into two exiting streams, i.e. one solute rich F_5 and one solute free F_4 , which regenerate the two input streams necessary for RED system operation.

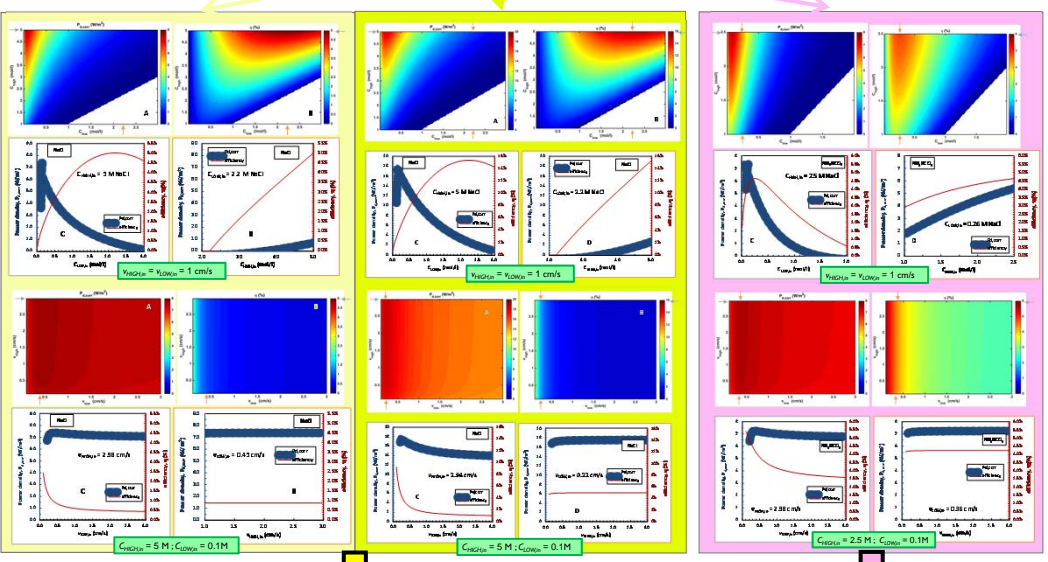
The amount of salt transferred in the RED stack from the concentrate solution to the dilute one is extracted from the dilute stream F_2 in the salt extraction unit. This results in two exiting streams, i.e. F_5 containing only the solute and F_4 , a solvent rich stream, which regenerates the two input streams necessary for RED system operation.



CASE STUDIES			
	Case study I	Case study II	Case study III
Case study typology	State of the art	Perspectives	Perspectives
Waste heat source temperature (°C)	90	90	60
RED stage			
Salt	NaCl	NaCl	NH ₄ HCO ₃
ABM permselectivity α_{ABM} (-)	0.65	0.90	0.90
CEM permselectivity α_{CEM} (-)	0.90	0.90	0.90
ABM electrical resistance R_{ABM} (Ω cm)	2.96	0.50	0.50
CEM electrical resistance R_{CEM} (Ω cm)	1.55	0.50	0.50
Temperature (°C)	25	25	25
Regeneration stage			
Typology	Solvent extraction	Solvent Extraction	Salt Extraction
Technology	MED	MED	Thermal degradation
Energy consumption	40 kWh/m ³ of extracted solvent	25 kWh/m ³ of extracted solvent	100 kJ/mol of extracted salt [1] and 4°C of thermal heating for F_5
Sensitivity analysis	Parameters and variation range		
C_{Dilute} (mol/l)	1.0±5.0	1.0±5.0	1.0±2.5
C_{Conc} (mol/l)	0.01±3.0	0.01±3.0	0.01±2.0
α_{ABM} (cm/s)	0.1±3.0	0.1±3.0	0.1±3.0
α_{CEM} (cm/s)	0.1±3.0	0.1±3.0	0.1±3.0

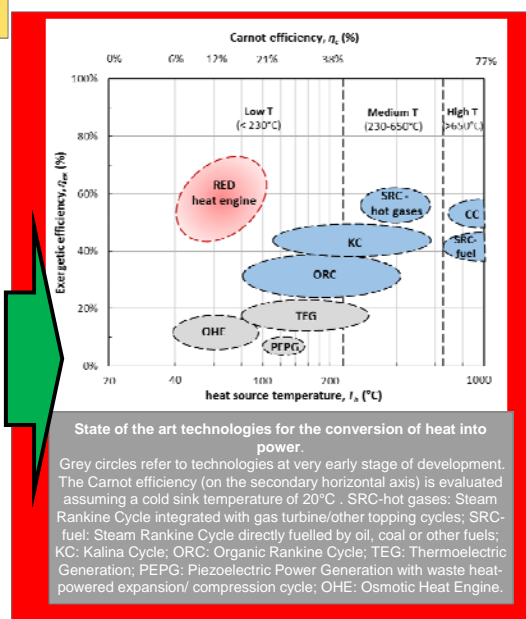
Cycle Efficiencies

- Energetic efficiency $\eta = \frac{P_{d,corr}}{Q_{in}}$
- Carnot efficiency $\eta_c = 1 - \frac{T_{cold}}{T_{hot}}$
- Exergetic efficiency $\eta_{ex} = \frac{\eta}{\eta_c}$



With NaCl-water solutions efficiencies up to about 15% corresponding to an exergetic efficiency of about 85% can be achieved

With NH₄HCO₃-water solutions max $P_{d,corr}$ and η are achieved at very close values of the operating parameters



RED could be promising as possible future converter of low-grade heat into electric energy