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# CO<sub>2</sub> abatement and renewable chemicals production by using microbial electrosynthesis within the framework of CO<sub>2</sub> biorefineries



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# INTRODUCTION

Microbial electrosynthesis (MES) relies on **microorganisms** as catalysts for the anoxic **reduction** of organic compounds at the cathode,  $CO_2$  in this study, at the expense of oxidizing reactions performed at the anode. Due to the oxidized nature of the carbon molecule, **external energy** must be **supplied** for the reaction to be favourable. In a nutshell, MES can decrease surplus  $CO_2$  in the environment while **synthesizing value-added compounds**.

# **OBJECTIVES\***

- (i) Evaluation of the current state of CO<sub>2</sub> emissions.
- (ii) Study of materials and microorganisms.
- (iii) Research on the economic viability and scale-up.
- (iv) Review of real production schemes.
- \* Accomplished via a literary review of general and specific scientific papers.

### GENERAL SCHEME ANODE CATHODE PRODUCTION OF WASTEWATER BIOGAS **FLUE GAS** BIOPLASTICS WATER INPUT **VOLATILE FATTY TREATMENT PURIFICATION** DESULPHURISATION PRODUCTION Electricity cost (EUR ton<sub>product</sub><sup>-1)</sup>: ACIDS $2H_{2}O \rightarrow O_{2} + 4H^{+} + 4e^{-}$ $2HSO_3^- + 2H_2O \rightarrow 2SO_4^{2-} +$ Acetate is generated at the the $CO_2 + 8e^- + 8H^+ \rightarrow CH_A +$ Reduction (i) Non-renewable: 406 $6H^{+} + 4e^{-}$ $2CO_2 + 8e^- + 8H^+ \rightarrow CH_3COO^- +$ cathode (5 g $L^{-1}$ ) and Chemical Oxygen O<sub>2</sub> finds application in + 2H<sub>2</sub>O (ii) Renewable: **162-243** extracted in a central $H^{+} + 2H_{2}O$ Demand (COD) the metal industry using SO<sub>4</sub><sup>2-</sup> is used as cleansing Upgrading the methane chamber. Later, mixed enriched cultures. and foaming agent. yield (0.16 g $m^{-2}$ $d^{-1}$ , synthesized acetate is Issues: Acetate: 1330 g m<sup>-2</sup> d<sup>-1</sup>, 99% CE 113% CE) of the obtained used by Ralstonia eutopha OER overpotential; Ethanol: 8.46 g m<sup>-2</sup> d<sup>-1</sup>, 26% CE Advantages: biogas from anaerobic *ex situ* to produce 600 mV are required for Butyrate: 160 g m<sup>-2</sup> d<sup>-1</sup>, 99% CE (i) No oxygen generation. polyhydroxybutyrate (PHB) digestion CHLORINE carbon electrodes to Butanol: 3.12 g m<sup>-2</sup> d<sup>-1</sup>, 12 % CE (ii) Lower potentials $(0.46 \text{ g L}^{-1}, 71\% \text{ CE}).$ methanogenic cultures. display intrinsic OER in a Caproate: 46 g m<sup>-2</sup> d<sup>-1</sup>, 13 % CE **PRODUCTION** required. current density of 5 mA (iii) Cheaper electrodes PHB is an environmentally $2Cl^{-} \rightarrow Cl_{2} + 2e^{-}$ Filters are required as cm<sup>-2</sup>. Simultaneous Prices (EUR kg<sup>-1</sup>): friendly and completely the inlet of biogas (ii) Generation of oxygen, wastewater treatment (i) Acetic acid: **0.6** Cl<sub>2</sub> is used in wastewater biodegradable product usually contains toxic threatening the required (anode) and chemicals (ii) Caproic acid: 1.6 treatment and as a that finds application in gases (H<sub>2</sub>S, NH<sub>3</sub> and anoxic conditions for the production (cathode). (iii) Butyric acid: 3.5 bleaching agent. the packaging industry. sulphur oxides). cathode. Methane WHY ···· Acetate MES? Ethanol Butyrate Solar panels with 40% & etc. efficiency sunlight to power conversion and 20% efficiency with a derived power of 2 V, require a current of 100 A $m^{-2}$ . Thus, if 50% $H_2Q$ conversion electron CO<sub>2</sub> or Atmospheric $CO_2$ efficiency from butyrate organics levels (ppm): Anaerobic Abiotic anode to butanol is assumed; Biocathode (i) Pre-industrial: 278 1.512 tonnes hectare<sup>-1</sup> (ii) 2021: **417** CATHODE ANODE while produced, tonnes of algae biomass Carbonaceous metallic or carbo-metallic CO<sub>2</sub> capture cost (EUR Carbonaceous materials mixed with weight hectare<sup>-1</sup> materials. metals (Ti, Pt). ton CO<sub>2</sub>-1): **184** \* annum<sup>-1</sup> are generated, Biofilm improvement with chitosan. Photo-anodes (TiO<sub>2</sub> nanowires). from which biodiesel is Porous cathodes for major biofilm Bioanodes (Shewanella, Geobacter). \*For the steel and cement after obtained adhesion. cumbersome process. industry CATHODIC MICROORGANISMS From the genus: Clostridium, Moorella and Sporomusa. Metabolic pathway: Wood-Ljungdahl. Electron transfer: direct (NADH delivering enzyme, coenzyme Q, ubiquinone) or

# CONCLUSIONS AND FUTURE OUTLOOK

Strong points: (i) requirement for little amount of water and, sometimes not even freshwater, (ii) it does not need arable land, rendering it free from the debate of food vs. fuel, (iii) ease in the biofuel obtention process, and (iv) the possibility of (renewable) energy storage.

**Economic viability**: cost of electricity and reagents: **4,430** EUR year<sup>-1</sup> and benefits: **3,356** EUR year<sup>-1</sup> for 5.37 tonnes of acetate year<sup>-1</sup> (3.94 tonnes  $CO_2$  year<sup>-1</sup>). Economically non-viable under the current state-of-the-art (**Technology Readiness Level 4**).

Future research: (i) good CO<sub>2</sub> availability, (ii) high couolombic efficiency with low overpotentials, (iii) increase production rates and specificity, and (iv) effective and economic downstream processing.

# SCALE-UP Electric power CO, stream

## KEY REFERENCES

indirect (H<sub>2</sub>, neutral red, methyl viologen, thionin).

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