

1,3,5-Trisubstituted and 5-Acyl-1,3-Disubstituted Hydantoin Derivatives via Novel Sequential Three-Component Reaction

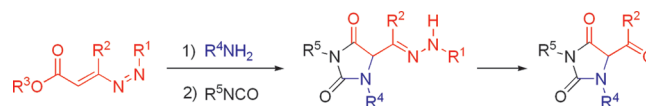
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ABSTRACT



1,2-Diaza-1,3-dienes (DDs) react as Michael acceptors with primary amines to afford α -aminohydrazone derivatives that were in situ coupled with isocyanates. Intramolecular ring closure of the asymmetric urea derivatives so formed allows for a selectively substituted hydantoin ring to be obtained. The hydrazone side chain introduced by the conjugated heterodiene system at the 5-position of the heterocycle represents a valuable functionality for accessing novel 5-acyl derivatives difficult to obtain by other methods.

Hydantoin-based scaffolds have been found to possess significant pharmacological activities. In fact, many derivatives have been identified as anticonvulsant,¹ antimuscarinics,² antiulcers and antiarrhythmics,³ antivirals, antidiabetics,⁴ and serotonin and fibrinogen receptor antagonists.⁵ Moreover, substituted hydantoins are important building blocks for the synthesis of nonnatural amino acids by alkaline degradation.⁶ Therefore, many methods for the rapid acquisition of structurally varied and functionalized hydantoins are desirable. The synthesis of 1,3,5-trisubstituted hydantoins is

usually accomplished by reacting *N*-substituted α -amino acids or their esters with isocyanates, either in solution⁷ or in solid phase.⁸ Other strategies for the synthesis of 1,3,5-hydantoins have been recently reported in the literature and are based on the reaction of *N,N'*-disubstituted ureas with carbon monoxide and aldehydes,⁹ on a Ugi four-component condensation¹⁰ and on the reaction between activated α,β -unsaturated carboxylic acids and asymmetric carbodiimides.¹¹ To the best of our knowledge, there is no report on the synthesis of 1,3,5-trisubstituted hydantoins having hydrazone or acyl function at the 5-position of the ring neither

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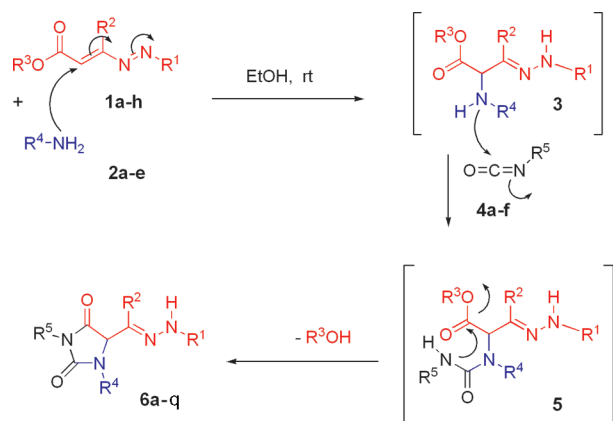
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Scheme 1. One-Pot Synthesis of 1,3,5-Trisubstituted Hydantoin Derivatives **6a–q**



from amino acid derivatives nor from *N,N'*-disubstituted ureidomalonate building blocks.¹² Indeed, by continuing our investigations designed to develop the usefulness of the conjugated azo-ene system of 1,2-diaza-1,3-dienes (DDs) as building blocks in heterocyclic chemistry, the present paper reports a synthetic strategy for regioselective trisubstituted hydantoin derivatives bearing novel and valuable functionality at the C-5 position of the hydantoin ring as a general procedure to achieve 5-acyl derivatives.

The versatility of DDs **1** in the synthesis of useful heterocyclic scaffolds is well documented¹³ and relies on their ability to undergo 1,4-Michael additions. Our approach, for acquiring the title compounds in a one-pot procedure, involves the construction of *N,N'*-disubstituted asymmetric urea moieties linked to suitable DD substrates. Since the conjugated heterodiene system exalts the electrophilic char-

acter of the terminal carbon of **1** making it capable to undergo nucleophilic attack, primary amines **2** constitute useful reagents to perform an aza-Michael addition producing the corresponding α -aminohydrazone derivatives **3**.¹⁴ Subsequent acylation of secondary amines **3** with isocyanates **4** generates the requisite asymmetric ureas **5** to be directed to a ring closure. Indeed, compounds **5** provide spontaneous regioselective heteroring closure owing to the nucleophilic attack of the amidic NH at the terminal ester function of the azo-ene system, affording the hydantoin derivative **6** (Scheme 1) by loss of an alcohol molecule. This one-pot reaction sequence represents a valuable route to variously 1,3,5-trisubstituted hydantoin derivatives **6a–q** containing an electron-withdrawing hydrazone function at C-5 derived from the conjugated azo-ene system of DDs. It can be easily accomplished in EtOH at room temperature, with satisfactory yields (47–76%, Table 1) overcoming the drawback of regiocontrol (i.e., **6c,d,i,p**) especially when weakly asymmetric carbodiimides are used.¹¹

Although aromatic amines (i.e., 4-methoxyaniline) worked well in the Michael addition producing α -aminohydrazone **3**, unfortunately the subsequent coupling with isocyanates (i.e., butylisocyanate) failed probably because of the poor nucleophilicity of the amine nitrogen atom of **3** (only traces of **5** were observed even upon prolonged reaction times).

Since the hydrazone side chain represents a protected carbonyl function, the hydrolytic cleavage of the hydrazide moiety under heterogeneous conditions (Scheme 2) introduces a point of diversity leading to novel 5-acyl disubstituted 1,3-hydantoin scaffolds difficult to obtain from amino acid ester building blocks^{8a} or by other methods.^{9,11,12}

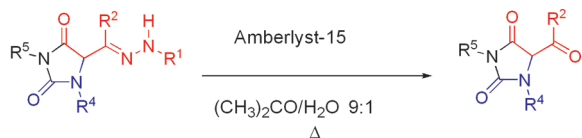
In summary, we have demonstrated the synthetic utility of 1,2-diaza-1,3-dienes in the construction of diversified trisubstituted 1,3,5-hydantoin derivatives with a controlled regioselectivity in the substitution at N-1 and N-3 of the heterocycle

Table 1. Results of the Synthesis of Hydantoin Derivatives **6a–q**

entry	DD 1			amine 2		isocyanate 4		hydantoin 6	yield ^a (%)	
	R ¹	R ²	R ³	R ⁴	R ⁵					
1	1a	CO ₂ Et	Me	Et	2a	<i>n</i> -Bu	4a	Ph	6a	67
2	1b	CO ₂ <i>t</i> -Bu	Me	Et	2b	<i>n</i> -Pr	4a	Ph	6b	76
3	1b	CO ₂ <i>t</i> -Bu	Me	Et	2b	<i>n</i> -Pr	4b	Cyclohexyl	6c	65
4	1c	CO ₂ Me	Me	Et	2b	<i>n</i> -Pr	4c	CH ₂ CO ₂ Et	6d	68
5	1d	CO ₂ Bn	Me	Et	2a	<i>n</i> -Bu	4a	Ph	6e	62
6	1a	CO ₂ Et	Me	Et	2b	<i>n</i> -Pr	4a	Ph	6f	66
7	1c	CO ₂ Me	Me	Et	2c	Allyl	4a	Ph	6g	65
8	1a	CO ₂ Et	Me	Et	2d	Propargyl	4d	3-Cl-Ph	6h	63
9	1a	CO ₂ Et	Me	Et	2e	Benzyl	4b	Cyclohexyl	6i	63
10	1e	CO ₂ <i>t</i> -Bu	Et	Et	2b	<i>n</i> -Pr	4a	Ph	6j	66
11	1a	CO ₂ Et	Me	Et	2b	<i>n</i> -Pr	4d	3-Cl-Ph	6k	63
12	1c	CO ₂ Me	Me	Et	2d	Propargyl	4a	Ph	6l	63
13	1c	CO ₂ Me	Me	Et	2e	Benzyl	4e	4-Cl-Ph	6m	73
14	1f	CO ₂ <i>t</i> -Bu	Ph	Et	2a	<i>n</i> -Bu	4e	4-Cl-Ph	6n	48
15	1g	CO ₂ <i>t</i> -Bu	CH ₂ CO ₂ Et	Et	2b	<i>n</i> -Pr	4a	Ph	6o	65
16	1c	CO ₂ Me	Me	Et	2e	Benzyl	4f	<i>n</i> -Bu	6p	69
17	1h	CO ₂ Me	CO ₂ Et	Et	2e	Benzyl	4e	4-Cl-Ph	6q	47 ^b

^a Yield of pure isolated product. ^b Yield referred to isolated α -aminohydrazone derivative.

Scheme 2



6a R ¹ = CO ₂ Et; R ² = Me; R ⁴ = <i>n</i> -Butyl; R ⁵ = Ph	7a (80%)
6d R ¹ = CO ₂ Me; R ² = Me; R ⁴ = <i>n</i> -Propyl; R ⁵ = CH ₂ CO ₂ Et	7b (77%)
6j R ¹ = CO ₂ <i>t</i> -Bu; R ² = Et; R ⁴ = <i>n</i> -Propyl; R ⁵ = Ph	7c (87%)
6m R ¹ = CO ₂ Me; R ² = Me; R ⁴ = Benzyl; R ⁵ = 4-Cl-Ph	7d (83%)
6n R ¹ = CO ₂ <i>t</i> -Bu; R ² = Ph; R ⁴ = <i>n</i> -Butyl; R ⁵ = 4-Cl-Ph	7e (74%)

with respect to that obtained when weakly asymmetric carbodiimides are coupling with α,β -unsaturated carboxylic acids. The one-pot procedure described here is based on sequential aza-Michael addition/condensation reactions and introduces a valuable hydrazone functionality at the 5-position of the heteroring that allows access to 5-acyl hydantoin. Noteworthy, the acyl residue directly bonded at the C-5 of the hydantoin nucleus is not easily achievable from amino acid esters or with ureidomalonate building blocks.

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Further extension of this sequential three-component pathway is currently being pursued in our laboratories and will be reported in due course.

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Supporting Information Available: Detailed experimental procedures and full characterization for all compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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