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# Effects of Bioleaching on the Chemical, Mineralogical and Morphological Properties of Natural and Waste-Derived Alkaline Materials

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## Effects of bioleaching on the chemical, mineralogical and morphological properties of natural and waste-derived alkaline materials

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### ABSTRACT

Bioleaching is a potential route for the valorisation of low value natural and waste alkaline materials. It may serve as a pre-treatment stage to mineral carbonation and sorbent synthesis processes by increasing the surface area and altering the mineralogy of the solid material and by generating an alkaline rich (Ca and Mg) aqueous stream. It may also aid the extraction of

high value metals from these materials (e.g. Ni), transforming them into valuable ore reserves. The bioleaching potential of several bacteria (*Bacillus circulans*, *Bacillus licheniformis*, *Bacillus mucilaginosus*, *Sporosarcina ureae*) and fungi (*Aspergillus niger*, *Hemicola grisea*, *Penicillium chrysogenum*) towards the alteration of chemical, mineralogical and morphological properties of pure alkaline materials (wollastonite and olivine) and alkaline waste residues (AOD and BOF steel slags, and MSWI boiler fly ash) at natural pH (neutral to basic) was assessed. Bioleaching was conducted using one-step and two-step methodologies. Increased solubilisation of alkaline earth metals and nickel were verified. Alteration in basicity was accompanied by alteration of mineralogy. AOD slag experienced solubilisation-precipitation mechanism, as evidenced by the decline of primary phases (such as dicalcium-silicate, bredigite and periclase) and the augmentation of secondary phases (e.g. merwinite and calcite). Nickel-bearing minerals of olivine (clinocllore, lizardite, nimite and willemseite) significantly diminished in quantity after bioleaching. Altered mineralogy resulted in morphological changes of the solid materials and, in particular, in increased specific surface areas. The bioleaching effect can be attributed to the production of organic acids (principally gluconic acid) and exopolysaccharides (EPS) by the microorganisms. The similarities between fungal and bacterial mediated bioleaching suggest that biogenic substances contribute mostly to its effects, as opposed to bioaccumulation or other direct action of living cells.

Keywords: Bioleaching, Mineral processing, Waste processing, Alkaline materials, Nickel

## 1. Introduction

The depletion of high grade natural minerals leads to the search for alternative resources for the mineral processing industry. Additionally, the costly treatment and disposal of waste materials is driving the industry to develop sustainable solutions. Alkaline bioleaching rises as a potential valorisation route for the low-value or hazardous waste materials that consist mainly of calcium and magnesium silicates (Willscher and Bosecker, 2003; Chandraprabha and Natarajan, 2009). Bioleaching may serve as a pre-treatment stage to mineral carbonation (Power et al., 2010) and sorbent synthesis processes. It is envisioned that through the extraction and re-precipitation of alkaline components (e.g. Ca, Mg, Si), bioleaching can increase material porosity and alter the mineralogy. This effect may enhance the mineral carbonation reactions by overcoming inherent rate and conversion limitations (Santos and Van Gerven, 2011). Moreover, the increased porosity of biodegraded solids, which can result in the formation of micro- and meso-porous structures, and the re-precipitation of altered crystalline phases onto the solid surface, can also inherently increase the sorption capacity of certain materials, thus enabling their valorisation as sorbents (Chiang et al., 2012). Extracted components from bioleaching processes can be recuperated or utilized for the synthesis of new materials. Solubilised calcium, magnesium and silicon are precursors to many inorganic compounds, including sorbents (e.g. zeolites, hydroxylapatite, and layered double hydroxides (Kuwahara et al., 2010)) and carbon sinks (e.g. precipitated calcium carbonate (PCC)). Bioleaching is also an attractive route for extraction of valuable metals (e.g. copper, nickel) from low grade ores (e.g. ultramafic rocks), given the relatively low energy requirements compared to traditional hydrometallurgical processes (namely high-pressure acid leaching (HPAL)).

Microorganisms capable of resisting alkaline environments, including several types of bacteria and fungi, produce organic acids and exopolysaccharides (EPS) that accelerate the dissolution of siliceous materials (Vandevivere et al., 1994). Acidolysis has been suggested as the principal mechanism in bioleaching with *Aspergillus niger* (Mulligan and Kamali, 1993), while production of EPS has been attributed to the performance of *Bacillus mucilaginosus* (Welch et al., 1999). Application of bioleaching in industrial processes is analogous to the acceleration of naturally occurring biodegradation and biodeterioration, which are attributable to mechanical and aesthetic wearing of stony and cementitious construction materials (Warscheid, 2000; De Belie, 2010; Wiktor et al., 2011). Bioleaching can lead to the extraction of valuable components from raw materials (Groudeva et al., 2007), and the extraction of hazardous components present in waste materials, which reduces the toxicity enabling reutilization or valorisation (Wu and Ting, 2006; Yang et al., 2010). It can also change the mineralogy of the material, thus altering its reactivity and crystallinity/amorphylicity (De Windt and Devillers, 2010). Biodegradation can also increase the surface area and porosity of the material (Wang et al., 2011), and produce (nano)particles (Kulkarni et al., 2008).

The alkaline bioleaching potential of the following microorganisms has been reported in literature: *Aspergillus niger* (Wu and Ting, 2006; De Windt and Devillers, 2010; Ball and Banik, 2011), *Humicola grisea* (Kulkarni et al., 2008), *Penicillium chrysogenum* (Groudeva et al., 2007), *Bacillus circulans* (Vrvić et al., 1989; Ball et al., 2010), *Bacillus licheniformis* (Mohanty et al., 1990), *Bacillus mucilaginosus* (Liu et al., 2006; Yang et al., 2010), and *Sporosarcina ureae* (Chandraprabha and Natarajan, 2009). These studies concerned the bioleaching of the following alkaline materials: oil shale (Vrvić et al., 1989), chromite ore (Ball et al., 2010; Ball and Banik, 2011), magnesite ore (Mohanty et al., 1990), municipal solid waste incineration fly ash (Wu and Ting, 2006; Yang et al., 2010), Portland cement (De

Windt and Devillers, 2010), borosilicate glass (Kulkarni et al., 2008), rich-in-carbonates copper ore (Groudeva et al., 2007), mica and feldspar (Liu et al., 2006).

The aforementioned microorganisms have been reported in the cited literature to survive in alkaline pH, in some cases in the presence of potentially harmful substances contained in waste materials, and to leach silica from siliceous minerals. The published literature also suggests that other desired effects of alkaline bioleaching are possible, such as mineralogical and morphological changes; however the data are insufficient to provide clear direction for implementation, due to the diversity of microorganisms and materials tested and to varying degrees of investigation depth and rigour.

The main objective of this research is to assess the technical feasibility of the desired effects of alkaline bioleaching. The work focused on screening of the seven aforementioned microbial organisms towards their susceptibility of leaching and/or altering five waste-derived and naturally available alkaline materials: Basic Oxygen Furnace (BOF) steel slag, Argon Oxygen Decarburization (AOD) stainless steel slag, municipal solid waste incinerator boiler fly ash (BFA), wollastonite ( $\text{CaSiO}_3$ ) and olivine ( $(\text{Mg,Fe})_2\text{SiO}_4$ ). The following activities were performed:

- Demonstrate the technical feasibility of alkaline bioleaching to achieve the desired effects: solubilisation of material components, mineral alterations and morphological transformations.
- Determine the most suitable microorganism(s) and material(s) towards producing the desired effects, in view of applying alkaline bioleaching as a pre-treatment stage to mineral carbonation, sorbent synthesis and/or metal extraction processes.
- Elucidate dominant mechanisms of alkaline bioleaching (e.g. leaching and adsorptive actions of organic acids and extracellular polysaccharides (EPS) and bioaccumulation).

## 2. Materials and Methods

### 2.1 Microbial strains, growth media and bioleaching materials

Three fungal and four bacterial strains were obtained from culture collections. The fungal strains utilized were: *Aspergillus niger* (DSM-872), *Penicillium chrysogenum* (DSM-848), and *Humicola grisea* (DSM-2691), all obtained from the Deutsche Sammlung von Mikroorganismen und Zellkulturen GmbH (DSMZ, Germany). The bacterial strains utilized were: *Bacillus circulans* (DSM-11), *Bacillus licheniformis* (DSM-13), *Bacillus mucilaginosus*, and *Sporosarcina ureae* (DSM-2281). The *B. mucilaginosus* was obtained from the China Center of Industrial Culture Collection (CICC, China), while the other strains were acquired from DSMZ. *S. ureae* was maintained on nutrient agar (NAU) (0.5% peptone, 0.3% meat extract, 1.5% agar) with 0.001%  $\text{MnSO}_4 \cdot \text{H}_2\text{O}$  and 2% urea. *A. niger*, *P. chrysogenum* and *H. grisea* were maintained on potato dextrose agar (PDA) (infusion from potatoes, 2% glucose and 1.5% agar). The *Bacillus* species were maintained on nutrient agar (NA) with 0.001%  $\text{MnSO}_4 \cdot \text{H}_2\text{O}$  for sporulation enhancement. Growth media components were obtained from Sigma Aldrich (Belgium). All strains were incubated at 30 °C (bacterial strains for 24 hours, fungal strain for 5 days) and preserved at 4 °C. Stock cultures were frozen in 10% glycerol by quick-freezing with solid  $\text{CO}_2$ /ethanol and were preserved at -34 °C.

Wollastonite material (Casiflux F125) was acquired from Ankerpoort (Belgium). Olivine material (GL40) was acquired from N.Z.M. GRIT N.V. (Belgium). Representative samples of boron-free Argon Oxygen Decarburization (AOD) slag, Basic Oxygen Furnace (BOF) slag, and Municipal Solid Waste Incineration Boiler Fly Ash (MSWI,BFA) were obtained from Belgian industries.



## 2.2 *Bioleaching procedures*

All growth media, solid samples and glass materials were sterilized for 15 minutes at 120 °C prior to use, and every handling was performed under sterile conditions.

### 2.2.1 *One-step bioleaching*

Inoculation solutions were prepared by inoculating sterile physiological solution using a 5 mm inoculation loop and diluting them serially to prepare  $10^{-1}$  to  $10^{-5}$  solutions. One millilitre of the dilution that yielded 30-300 colony forming units (CFU) upon spread plating was used for inoculation of the 'bioleaching solution'. This solution consisted of 100 ml of the respective nutrient broth of the microbial strain (same composition as nutrient agar, with exception of agar) with 10 g of solid material placed in 0.25 L Erlenmeyer flasks. These flasks were shaken at 150 rpm on an incubated shaking table at 30 °C for two weeks. Control experiments were performed in the same way without microbial inoculation. At the end of the bioleaching period, the solids were collected from the aqueous phase by centrifuging (Jouan C412, 3000 rpm for 20 minutes) and overnight drying at 105 °C.

### 2.2.2 *Two-step bioleaching*

The microbial strains were inoculated into and cultivated in 500 ml of their respective nutrient broths by magnetically stirring in 1 L Erlenmeyer flasks at 30 °C in an incubation chamber. After incubation for seven days, cells were harvested by centrifuging (3000 rpm for 20 minutes) and the total biomass was resuspended in the 'bioleaching solution'. This solution consisted of sterile 500 mL nutrient broth and 50 g of solid material in 1 L Erlenmeyer flasks. Control experiments were performed in the same way without adding microbial biomass. The bioleaching reaction was carried out at 30°C and flasks were continuously stirred on a

magnetic stirrer for a period of four weeks. At the end of the bioleaching period, the solids were collected from the aqueous phase by centrifuging and overnight drying at 105 °C.

### *2.3 Sampling and Analytical methods*

Samples were taken weekly from the bioleached suspensions and relevant parameters including pH, cell counts and elemental concentrations were determined. The pH was analyzed with a pH meter from Mettler Toledo. Biomass concentration was determined with the spread plate method using a Drigalski spatula and by counting CFU. Samples of 100 mL were taken and replaced by fresh nutrient broth. Before analytical analysis, aqueous samples were centrifuged for 15 minutes at 15,000 g (Biofuge Stratos) and filtered through a Whatmann paper filter.

Aqueous magnesium and calcium concentrations were measured with a DIONEX DX-100 ion chromatograph (IC) supplied with an IonPac CS12A 4x250 mm column and conductivity detector with 20 mM CH<sub>3</sub>SO<sub>2</sub>OH (methane sulfonic acid) as eluent; the column was supported by an IonPac CG12A 4x50 mm Guard column and CSRS 300 4 mm suppressor. Aqueous silicon and nickel concentrations were measured by Inductively Coupled Plasma Mass Spectrometry (ICP-MS) (Thermo Electron X Series) in 2% HNO<sub>3</sub> (nitric acid) solutions. Detection of organic acids was performed by high-performance liquid chromatography (HPLC, Agilent Technologies 1200 series).

The chemical composition of solid materials was determined by quantitative X-ray Fluorescence (XRF, Panalytical PW2400). The mineral composition was determined by X-ray Diffraction (XRD, Philips PW1830) equipped with a graphite monochromator and a gas proportional detector, using Cu K $\alpha$  radiation at 30 mA and 45 kV, step size of 0.02° 2 $\theta$  and counting time 2 s per step, over 5 to 70° 2 $\theta$  range. Mineral identification was done in

Diffraction Plus EVA (Bruker) software and quantitative phase analysis was performed by the full profile Rietveld method implemented in Topas Academic v4.1 software. Specific surface area ( $S_{\text{BET}}$ ) was measured by nitrogen adsorption (Micromeritics TriStar 3000) using the BET (Brunauer–Emmett–Teller) model. Loss on ignition (LOI) was measured by thermogravimetric analysis (TGA, TA Instruments Q500). Fourier transform-infrared (FTIR) spectra were recorded on a Perkin Elmer Frontier spectrometer with attenuated total reflection (ATR) accessory in the region of 4000–650  $\text{cm}^{-1}$  at a resolution of 1  $\text{cm}^{-1}$ .

### **3. Results and Discussion**

#### *3.1 Materials characterization*

The elemental composition of the five alkaline materials used in bioleaching experiments is shown in Table 1. Four of the materials are predominantly calcium based, with the exception of olivine which is rich in magnesium. AOD slag, wollastonite and olivine contain large amounts of silicon. BOF slag contains substantial amounts of Ca and Fe, due to its predominant free lime and calcium ferrite composition. Municipal solid waste incineration boiler fly ash has the most distinct composition, containing significant quantities of Cl, K and S, indicating high content of salts. In terms of metals and heavy metals, the waste-derived materials contain more varied contents than the pure materials. Most prominent are the Cr contents of the slags, and the Pb and Zn content of BFA. Wollastonite is nearly absent of heavy metals, while olivine contains notable quantities of chromium and nickel. Heavy metal and salt content could be potentially detrimental to the microorganisms due to their toxicity.

The morphology of the solid material particles under SEM imaging is shown in Fig. 1. Of these materials, BFA and AOD slag are used in the experiment as received since they are

powdery materials. BFA has considerably finer particles than AOD slag (note image magnifications and scales). BOF slag is produced as a monolith material; therefore the sample utilized for bioleaching was crushed and sieved to < 1 mm particle size. Similarly, the natural minerals are milled materials, with nominal particle sizes of 57  $\mu\text{m}$  and 200  $\mu\text{m}$  for wollastonite and olivine, respectively.

### 3.2 Microbial screening

The selected microorganisms (pictured in Fig. 2) were first tested for their bioleaching potential by performing a series of one-step bioleaching experiments. A total of 24 experiments were performed, combining six microorganisms (*B. mucilaginosus* was unavailable at this stage) with the five alkaline materials of interest. The objective of this research phase was to determine the two most promising microorganisms and materials towards alkaline bioleaching, which would be studied in more detail in the subsequent research phase. The selection criteria are pH and solubilisation performance ratios, determined by dividing the number of experiments achieving substantial positive results towards alteration of pH, and increased Ca or Mg solubilisation, over the total number of experiments performed with each material.

Solution pH, and solubilisation of calcium and magnesium were monitored during the tests and the bioleaching performance towards each solid material is summarized in Table 2. Values shown are the selection criteria ratios, and the last column is the sum of the two ratios, which is used as a qualitative indication of which solid materials are most susceptible towards alkaline bioleaching, and therefore to be studied in more detail.

AOD slag and olivine achieve the highest marks in their respective categories (waste and natural materials). The pH of AOD slag consistently increased during bioleaching, possible

indicating the degradation of silicate minerals towards their more alkaline oxides. Similarly consistent increases in solubilised calcium were observed. With respect to the other two waste materials, it is thought that the soluble salt and heavy metal contents of BFA are detrimental towards the microorganisms and bioleaching performance could be compromised. In the case of BOF slag, its high basicity (12.7, significantly higher than that of the other materials which ranged from 7.8 to 10.8) due to large free lime content, and its predominant ferrite mineral content, appear to be inhibitory to satisfactory bioleaching performance. In the case of natural minerals, neither wollastonite nor olivine showed increased solubilisation after bioleaching, however pH alteration was consistent and significant for olivine. Moreover this material offers greater potential of valorisation due to its valuable metals contents (Cr and Ni); therefore olivine was chosen for further study.

The average performance of each microorganism towards bioleaching was analyzed in similar fashion from the 24 one-step experiments performed, and is presented in Table 3. In contrast to the materials selection, the average performance of the microorganisms was difficult to differentiate from the ratio summations. Only *B. licheniformis* performed more poorly than the others. Out of the three fungi tested, *A. niger* was chosen for continuation in the subsequent research phase as its pH alteration and increased solubilisation performance were more balanced (inferring greater potential for mineralogical alteration), and this microorganism has been more often studied in literature towards bioleaching (both alkaline and acidic) than the other two fungi. With respect to the two satisfactorily performing bacteria, *S. ureae* was chosen for further study as in the next experimental phase another bacillus, *B. mucilaginosus*, was also used (in this way avoiding the use of two bacteria from the same genus, which may work on similar bioleaching mechanisms). By no means does it mean the four omitted microorganism are unsuitable for alkaline bioleaching; further work with them in the future may be valuable.

### 3.3 Two-step bioleaching

Two-step bioleaching experiments were performed with the chosen microorganisms (*Aspergillus niger* (AN), *Bacillus mucilaginosus* (BM) and *Sporosarcina ureae* (SU)) on the selected alkaline materials (AOD slag and olivine). Two-step bioleaching methodology was selected over one-step bioleaching to ensure the microorganisms had reached substantial levels of growth and generation of extra-cellular biogenic substances (e.g. organic acids and exopolysaccharides (EPS)) prior to introduction of the solid materials. Yang et al. (2008) found two-step bioleaching more favourable than one-step bioleaching for the extraction of heavy metals from MSWI fly ash at high solids loading (> 20 g/L), due to the generation of greater amounts of biogenic substances prior to solids addition.

#### 3.3.1 Solution analysis

Fig. 3 shows the evolution of the pH of bioleaching and reference solutions over time (four weeks). The pH of AOD slag solution was generally higher than olivine throughout due to its higher basicity. AOD slag pH increased during bioleaching by up to 1.5 units, suggesting mineralogical changes; pH of reference samples was approximately constant for the same duration. The pH of olivine solutions differed in the first two weeks, both for the comparison of bioleaching/reference samples and the different microbial solutions. While bioleaching pH increased over the first two weeks for all microorganisms, the pH of reference samples lingered lower until week three. Over time olivine pH tended to stabilize regardless of sample towards a pH value of approximately 9. This suggests that solubilisation of some alkaline components of olivine is slow, and is intensified by bioleaching. It is also worth noting that the AN+olivine solution (Fig. 3d) had significantly lower pH compared to other

solutions. This is likely a consequence of larger amounts of organic acids produced by the fungus compared to the bacteria. It was verified by HPLC that AN produces significant quantities of gluconic acid; this is in accordance with findings of Wu and Ting (2006), Yang et al. (2008), and Amiri et al. (2011), who found gluconic acid to be the main lixiviant of *Aspergillus niger*. The same trend is not seen with AOD slag (Fig. 3a) since it has much higher acid neutralization capacity near its natural pH value than olivine.

Fig. 4 shows the solubilisation over time of calcium (for AOD slag) and magnesium (for olivine) from bioleached and reference materials. AN achieves the highest levels of solubilisation for both materials, with Ca amounts one order of magnitude higher than Mg. Reference sample solubilisation in PDA is also higher than for other samples, likely due to its lower pH (about one unit, see Fig. 3); however, bioleaching increases solubilisation over four weeks by about four times. BM and SU do not increase solubilisation of AOD slag compared to the reference sample (Fig. 4b and Fig. 4c), despite the pH effect previously mentioned. It is possible that the altered minerals produced have lower solubility than the original minerals (e.g. calcite and portlandite). It was also verified by HPLC that organic acid production from the bacteria is low (compared to AN); organic acid is a likely carrier of solubilised Ca and its presence in AN solutions may be inferred from lower pH (by approximately one unit) of these solutions (Fig. 3), in addition to verification by HPLC. In the case of olivine, on the other hand, there is evidence of increased solubilisation of Mg by BM and SU (Fig. 4e and Fig. 4f), possibly as a result of greater solubility of altered minerals or greater production of organic acids by the bacteria in the olivine's less alkaline environment. Relatively small quantities of acetic, butyric, citric, formic and lactic acids could be identified by HPLC from bacterial solutions obtained at near neutral pH; these acids have been observed by De Windt and Devillers (2010) to contribute to biogenic degradation of Portland cement. An observation to make is that the pH of bioleached and reference olivine was the same after four weeks (Fig.

3); therefore the amount of extra Mg solubilised by bioleaching may not be 'free' in the solution (which could imply a higher solution pH), but adsorbed onto soluble biogenic compounds (and thus not affecting pH).

Silicon solubilisation behaves different than the alkaline earth metals (Fig. 5), in particular for the case of AOD slag. The concentrations of Si are lower in the bioleaching solution than in the reference solutions (where silica concentrations are near the predicted equilibrium of  $\text{Si}(\text{OH})_4$  of 100 ppm  $\text{SiO}_2$  (Burkin, 2001)). Mechanistically this may be explained by the bioaccumulation of Si, either in the cellular walls or in the produced EPS. These insoluble materials would result in lowering of Si concentration in the analyzed (centrifuged) solutions. Bansal et al. (2007) suggested that fungal enzymes may specifically hydrolyze the silicates present to form silicic acid, which on condensation by certain other fungal enzymes results in the synthesis of silica nanoparticles. The effect of bioleaching on Si solubilisation from olivine is opposite, with moderately higher concentrations in the bioleached solutions. The greater contribution of organic acids to Si leaching, produced in larger quantities at moderate pH (as was previously suggested to explain Mg solubilisation), may explain this behaviour. This could also explain the slightly higher Si concentrations in the case of AN solutions (Fig. 5a and Fig. 5d), compared to BM and SU. The solubilisation of Mg, however, is significantly higher than that of Si, when it is expected that these two components of olivine dissolve stoichiometrically due to the orthosilicate nature of the mineral (Burkin, 2001); therefore biogenically induced re-precipitation of Si also appears to occur in this case.

Lastly, a positive effect on the leaching of nickel from olivine was observed. Fig. 6 presents the solubilisation trends of this metal. The nickel concentrations substantially increase compared to reference solutions (which have near zero values), with AN and SU solutions reaching values double that of BM. Olivine is known to be a nickel-bearing ore,



however its relatively low nickel content (0.2-0.4 wt% compared to 1-2.6 wt% of typical lateritic nickel ores (Brand et al., 1998)) and high alkalinity have made it economically unattractive for industrial hydrometallurgical processes (primarily based on high pressure acid leaching). Bioleaching may prove to be a feasible and sustainable solution for these lower grade ores, abundantly available on the Earth's crust (3% of lithosphere (Ollier, 1984)). According to the nickel content of the olivine material utilized (0.4 wt%), the leaching extents achieved in the present study represent up to 16% nickel extraction. Improvement of these values may require continuous rather than batch leaching (e.g. heap or column leaching) to overcome solubility limits, and reduction in solids particle size (here 200  $\mu\text{m}$  nominal size, considerably coarse compared to mineral processing standards).

### 3.3.2 Solids analysis

Bioleached solid materials were analyzed by XRD to assess alteration of mineralogy and the creation or solubilisation of mineral phases. Analysis results of AOD diffractograms by Rietveld refinement are presented in Table 4. Shown are the main mineral phases that suffer significant alteration in content compared to reference samples (at least 1 wt% change ( $\Delta\text{wt}\%$ ), or 20% change of original content (% change)). The content of gamma polymorph of dicalcium silicate ( $\gamma\text{-C}_2\text{S}$ ,  $\text{Ca}_2\text{SiO}_4$ ) consistently decreases (by 5 to 13 wt%) under the three bioleaching conditions. In contrast, the concentration of merwinite ( $\text{Ca}_3\text{Mg}(\text{SiO}_4)_2$ ) consistently and significantly increases (by 5 to 15 wt%) for all samples. These proportionally similar values may indicate a solubilisation-precipitation mechanism of mineral alteration. Fig. 7 displays the morphology of the bioleached solid particles; besides surface erosion, crystalline precipitation products can also be visualized, once compared to the rather uniform morphology of the original material (Fig. 1). Magnesium, present in merwinite, may originate from the concurrent dissolution of bredigite ( $\text{Ca}_{14}\text{Mg}_2(\text{SiO}_4)_8$ ) and åkermanite ( $\text{Ca}_2\text{MgSi}_2\text{O}_7$ )

phases. It is unclear if changes in periclase (MgO) can be attributed to Mg solubilisation (only moderate in the case of AN, not shown previously), or if it is an artefact of the semi-quantitative analysis. The production of calcite (CaCO<sub>3</sub>), however, appears to be better founded. Calcite likely is a product of the carbonation of degradation products of calcium silicates (i.e. Ca(OH)<sub>2</sub>, which was nearly absent (< 1 wt%) from all samples). The CO<sub>2</sub> needed for this reaction may come from the atmosphere, or may very well come from the metabolic microbial activity, as a product of utilization of organic energy sources (Hammes and Verstraete, 2002; Rendek et al., 2006). Calcite production from SU was negligible (0.3 wt%), in line with the smaller mineral alterations produced with this bacteria, compared to the overall results of the other microorganisms.

XRD diffractograms of fresh and bioleached olivine samples are displayed in Fig. 8. The major mineral phases of olivine are forsterite (Mg<sub>2</sub>SiO<sub>4</sub>) and forsterite ferroan ((Mg,Fe)<sub>2</sub>SiO<sub>4</sub>). The intensity of their peaks shifts moderately after bioleaching, potentially indicating morphological changes that alter preferential crystallographic directions. Fresh olivine also contains four identified silicates that are typically present in lateritic nickel ores (Brand et al., 1998), and which comprise approximately 8 wt% of the fresh sample: clinocllore ((Mg,Fe)<sub>5</sub>Al(Si<sub>3</sub>Al)O<sub>10</sub>(OH)<sub>8</sub>), lizardite (Mg<sub>3</sub>Si<sub>2</sub>O<sub>5</sub>(OH)<sub>4</sub>), nimite ((Ni,Mg,Fe)<sub>5</sub>Al(Si<sub>3</sub>Al)O<sub>10</sub>(OH)<sub>8</sub>), and willemseite ((Ni,Mg)<sub>3</sub>Si<sub>4</sub>O<sub>10</sub>(OH)<sub>2</sub>). The main peaks of these minerals are marked for fresh olivine in Fig. 8. These phases disappear nearly completely after bioleaching; the content of these minerals in the three bioleached samples reduces to below 3 wt% as determined by quantitative XRD. This is in agreement with nickel solubilisation observed (presented in Section 3.3.1).

Mineral degradation is also evident by the change in material specific surface area measured by nitrogen adsorption. BET isotherms of fresh and bioleached AOD slag and olivine are shown in Fig. 9 along with S<sub>BET</sub> values. For both materials there is a substantial

increase in surface area, with bioleached AOD slag having the greatest change. The increased surface area is predominantly due to the generation of micropores (evidenced by the sharp adsorption increase at low relative pressure ( $P/P_0 < 0.01$ )) (Tan et al., 2007), which may be mainly related to precipitation of altered minerals from solubilised components. AN produced the greatest increase in surface area for AOD slag, while BM delivered slightly superior results for olivine. The hysteresis loops occurring in the pressure range  $0.45 < P/P_0 < 1.0$  are due to the formation of the macropores (Tan et al., 2007), indicating bioleaching induced degradation of particle surfaces, creating cracks and crevices. Increased specific surface areas and porosity are encouraging results, as these factors can potentially improve the sorption capacity and reactivity (towards mineral carbonation or chemical leaching, for example) of certain materials.

In order to characterize the residual biomass on the recuperated dried solids, TGA and FTIR analyses were performed. Fig. 10 shows the thermogravimetric data of fresh and bioleached samples of AOD slag and olivine. The mass drop in the range of 200-400 °C can be predominantly attributed to decomposition of EPS materials (Ismail and Nampoothiri, 2010; Wang et al., 2010); it is noted that the drop is more significant for AOD slag than for olivine, and for AN than for the other microorganisms. This is in line with the previously proposed mechanism for reduction of Si leaching in the case of AOD slag, due to attachment of Si to insoluble biogenic materials. Further heating of the samples induces the decomposition of carbonates (up to 800 °C).

FTIR also suggests greater EPS content on AOD slag than on olivine (Fig. 11). The intense broad stretching peaks around  $3,307\text{ cm}^{-1}$ , which can be attributed to hydroxyl groups, are characteristic of polysaccharides, while the strong absorption at  $1,652\text{ cm}^{-1}$  and  $1,448\text{ cm}^{-1}$  can be related to biogenic proteins and peptides (Wang et al., 2010). Remaining peaks below  $1,000\text{ cm}^{-1}$  are inherent of the silicate materials.

#### 4. Conclusions

The results presented here suggest that alkaline bioleaching can be a potentially valuable process for the treatment of alkaline materials prior to valorisation (e.g. as carbon sinks or as sorbents) or for utilization of these abundant materials as metal ores (e.g. nickel from olivine). Bioleaching induced positive results with respect to increased solubilisation of alkaline earth metals and nickel, and the alteration of basicity, mineralogy and morphology of solid materials. The similarities between fungal and bacterial mediated bioleaching suggest that biogenic substances such as organic acids and exopolysaccharides contribute mostly to these effects, as opposed to bioaccumulation or other direct action of living cells. This aspect supports the use of two-step bioleaching as opposed to one-step, since the biogenic substances can be produced under optimal conditions prior to the introduction of potentially hazardous solid materials. Further work is required to better understand the bioleaching mechanisms. Routes for optimization of the biochemical processes include: (i) adoption of continuous leaching (e.g. heap or column) to overcome solubility limits; (ii) genetic amelioration of the utilized microorganisms, to produce strains more resistant to alkaline environments and to the presence of heavy metals, and that generate greater quantities of the key lixiviant biogenic substances; and (iii) identification of other suitable strains isolated from alkaline environments (e.g. mining and metallurgical waste sites and biodeteriorated cementitious and stony structures).

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List of Figures:

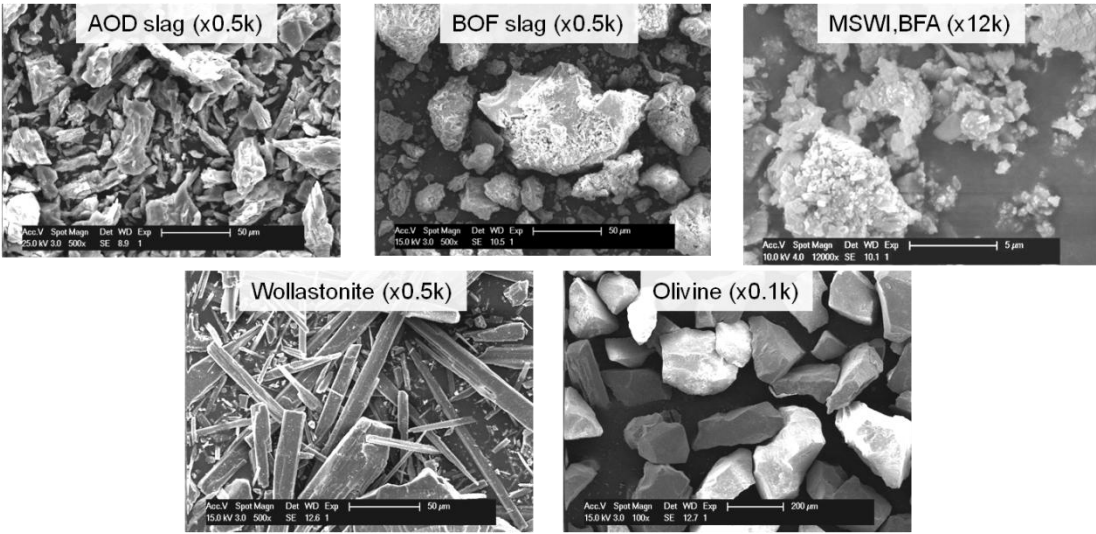


Fig. 1. Morphology of fresh solid materials utilized for bioleaching inspected by SEM.

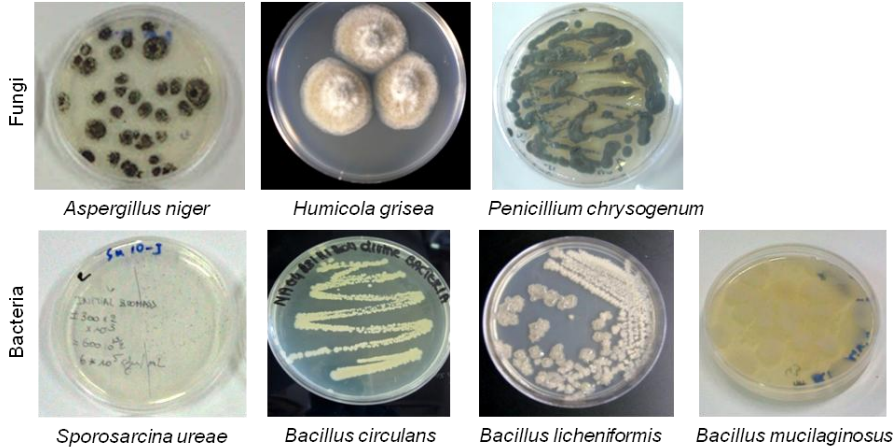


Fig. 2. Microbial organism utilized for bioleaching on agar plate.

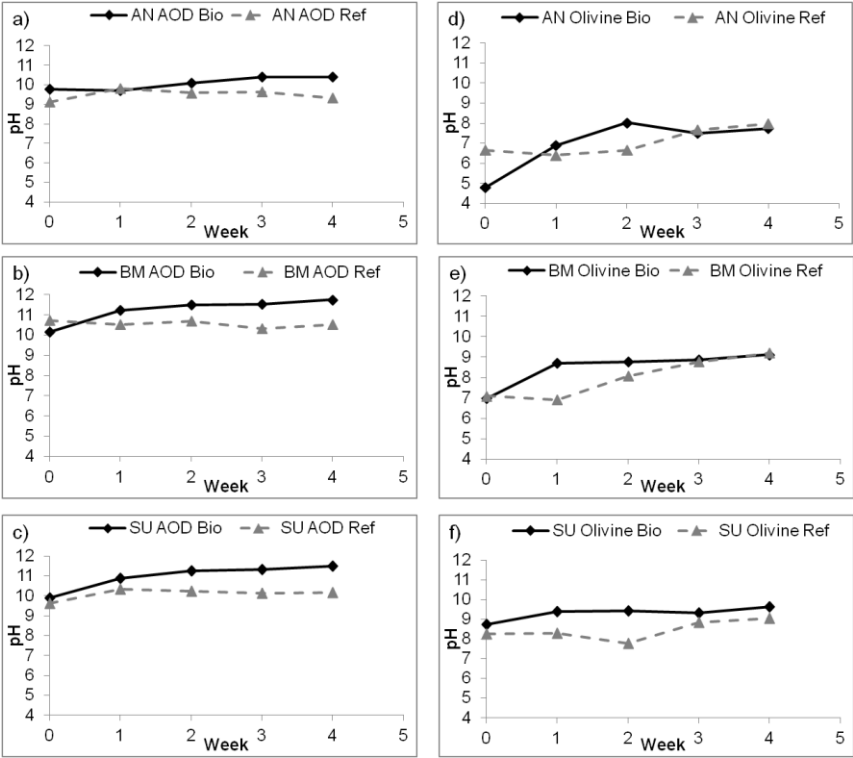


Fig. 3. Solution pH from reference and bioleached (by AN, BM, and SU) AOD slag and olivine over time.

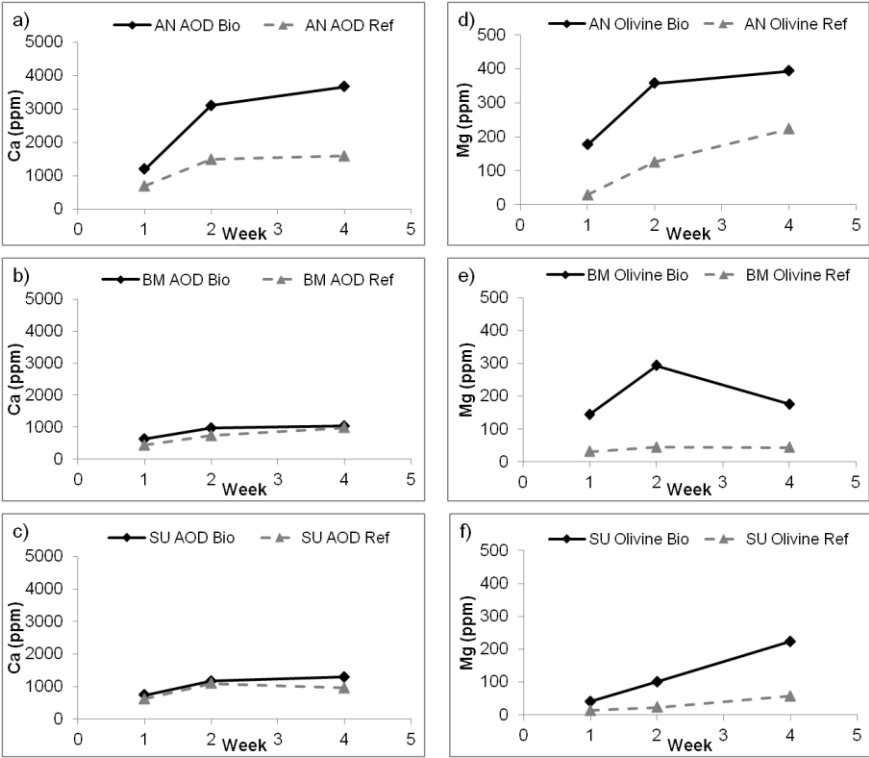


Fig. 4. Solubilised calcium (a,b,c) and magnesium (d,e,f) concentrations from reference and bioleached (by AN, BM, and SU) AOD slag and olivine over time.

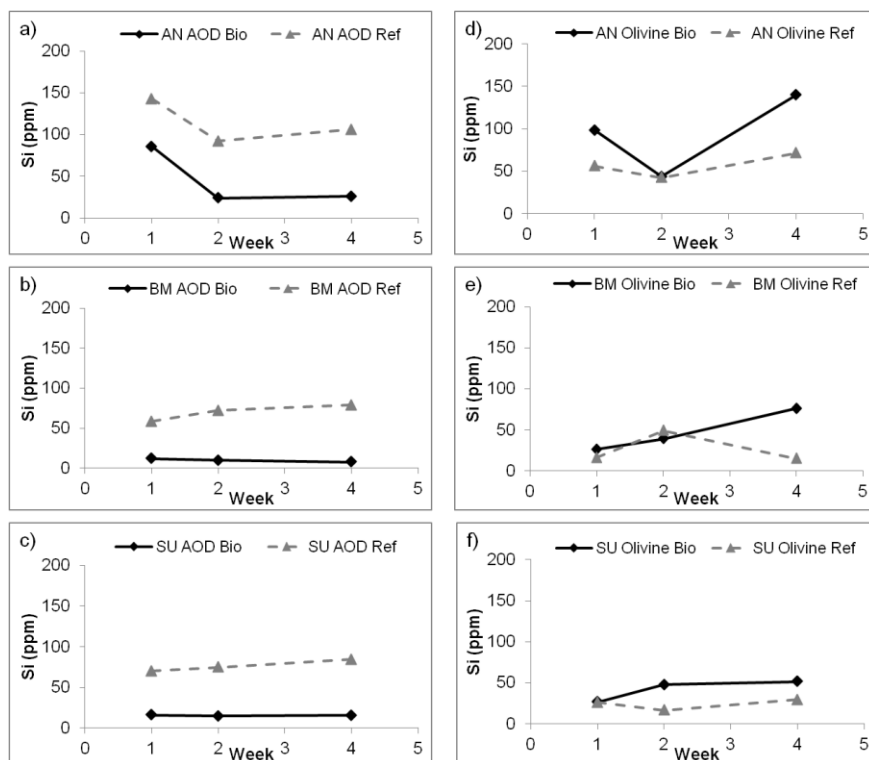


Fig. 5. Solubilised silica concentrations from reference and bioleached (by AN, BM and SU) AOD slag and olivine over time.

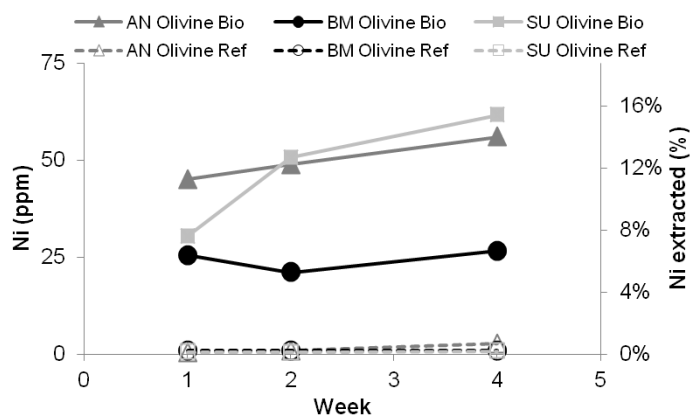


Fig. 6. Solubilised nickel concentrations and percentage extractions from reference and bioleached (by AN, BM and SU) olivine over time.

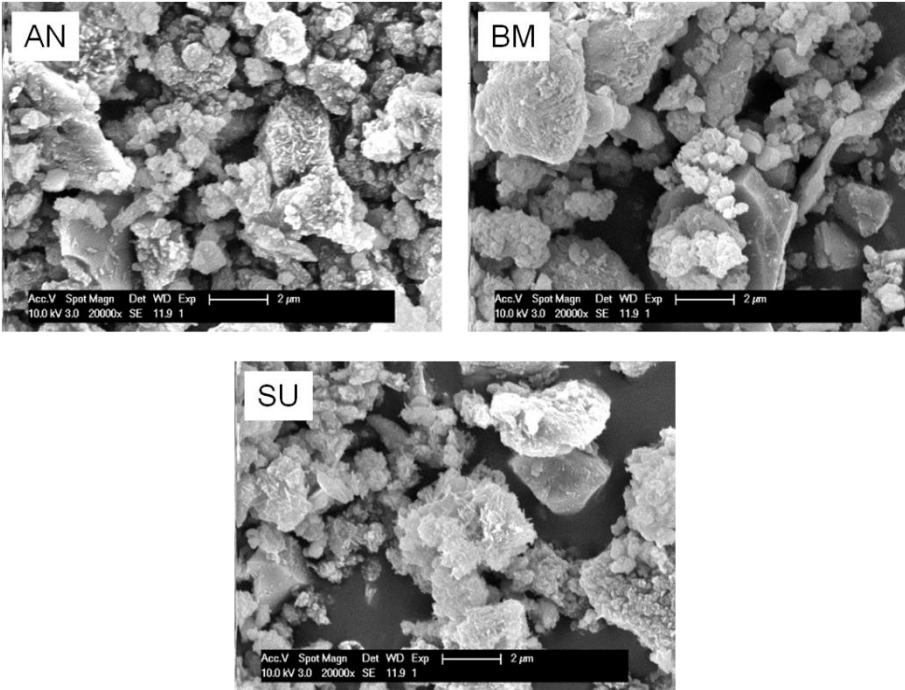


Fig. 7. Morphology of bioleached (by AN, BM and SU) AOD inspected by SEM (20,000x magnification).

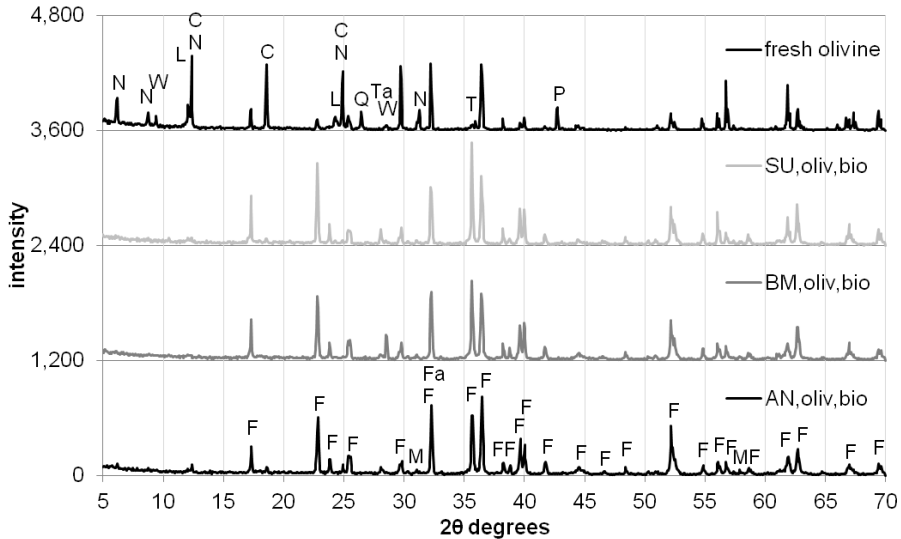


Fig. 8. Crystallographic composition of fresh and bioleached (by AN, BM and SU) olivine determined by XRD; C = clinocllore, F = forsterite, Fa = fayalite, L = lizardite, M = magnetite, N = nimite, P = periclase, Q = quartz, T = tirodite, Ta = talc, W = willemseite.

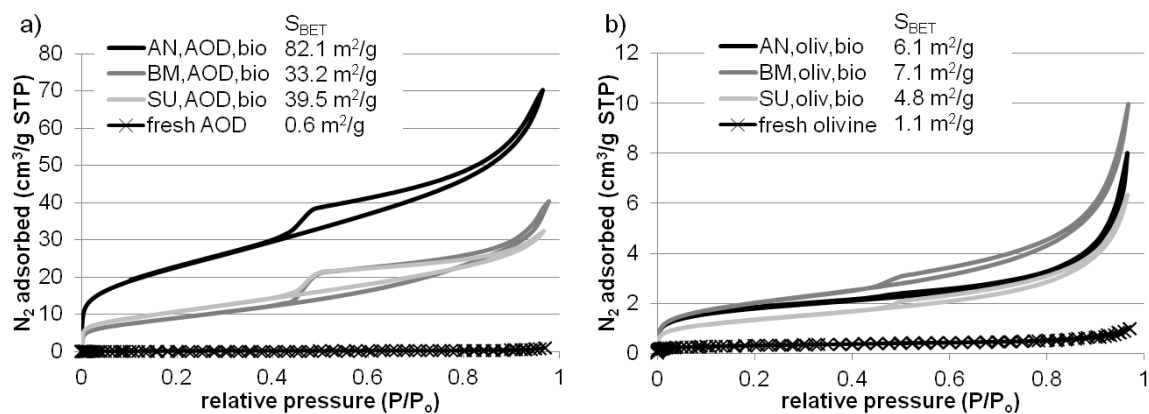


Fig. 9. Nitrogen adsorption isotherms of fresh and bioleached (by AN, BM and SU) AOD slag (a) and olivine (b).

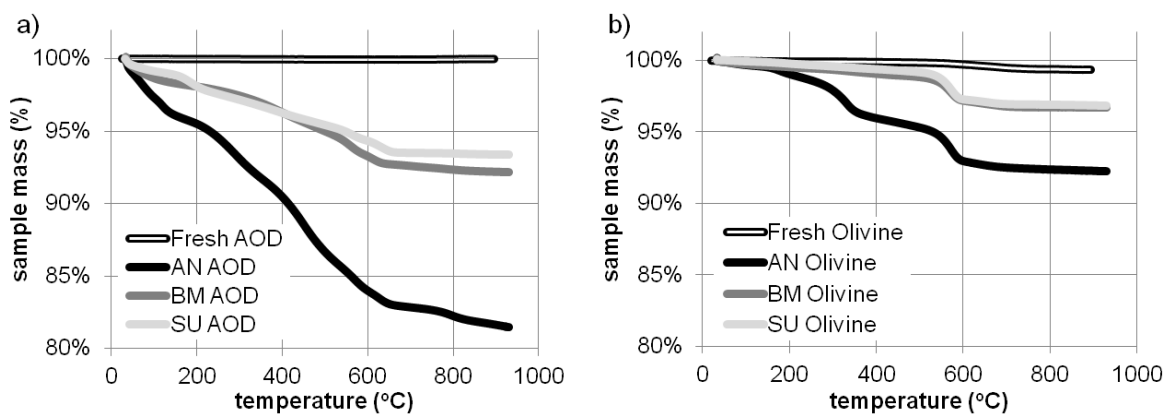


Fig. 10. Loss on ignition (LOI) of fresh and bioleached (by AN, BM and SU) AOD slag (a) and olivine (b) determined by TGA.

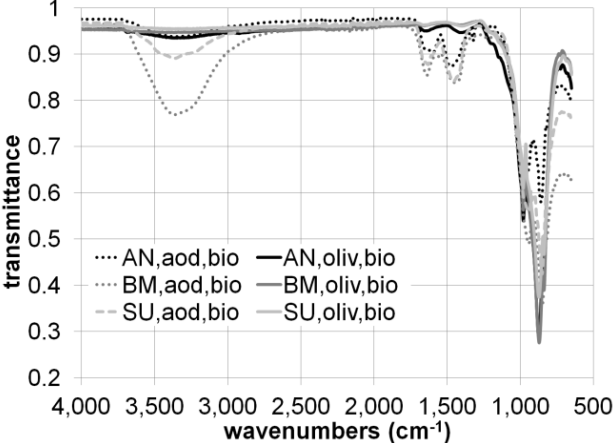


Fig. 11. FTIR spectrum of bioleached (by AN, BM and SU) AOD slag and olivine.



**List of Tables:**

Table 1

Elemental composition (wt%) of bioleached solid materials determined by XRF.

Element	AOD slag	BOF slag	MSWI,BFA	Wollastonite	Olivine
Al	0.5	1.0	2.8	0.4	0.6
Ca	40.6	38.5	21.4	36.6	-
Cl	0.02	0.03	10.4	0.02	0.06
Cr	0.5	0.2	0.1	0.01	0.4
Fe	0.2	23.1	1.0	0.4	6.6
K	-	-	4.9	0.1	0.04
Mg	4.5	0.6	1.1	0.4	27.0
Mn	0.3	3.1	0.08	0.03	0.09
Ni	-	-	0.01	-	0.4
Pb	-	-	0.2	-	-
S	0.2	0.06	9.8	0.02	-
Si	15.2	5.6	4.0	21.7	20.3
Ti	0.2	0.4	1.2	0.03	0.01
Zn	-	-	1.0	-	-
Balance (H,C,O,trace)	37.8	27.4	42.0	40.3	44.5

Table 2

Bioleaching performance and decision matrix for alkaline materials; values equal to ratio of positive tests over total number of tests.

Materials	Type	pH	Solubilisation	Sum <sup>a</sup>
Olivine	Natural	1	0	1.0
Wollastonite	Natural	0.5	0	0.5
AOD	Waste	0.8	1	1.8
BOF	Waste	0	0.5	0.5
Fly ash	Waste	0.25	0.25	0.5

<sup>a</sup> sum of scores in Solubilisation and pH criterion.

Table 3

Bioleaching performance and decision matrix for microorganisms; values equal to ratio of positive tests over total number of tests.

Microorganism	Type	pH	Solubilisation	Sum <sup>a</sup>
<i>Bacillus circulans</i>	Bacteria	0.3	0.7	1.0
<i>Bacillus licheniformis</i>	Bacteria	0.6	0.2	0.8
<i>Sporosarcina ureae</i>	Bacteria	0.7	0.3	1.0
<i>Aspergillus niger</i>	Fungus	0.6	0.4	1.0
<i>Humicola grisea</i>	Fungus	0.3	0.7	1.0
<i>Penicillium chrysogenum</i>	Fungus	0.7	0.3	1.0

<sup>a</sup> sum of scores in Solubilisation and pH criterion.

Table 4

Analysis of mineral alterations of bioleached (by AN, BM and SU) AOD slag determined by XRD and Rietveld refinement.

Mineral	AN,AOD		BM,AOD		SU,AOD	
	$\Delta$ wt%	% change	$\Delta$ wt%	% change	$\Delta$ wt%	% change
$\gamma$ -C2S, $\text{Ca}_2\text{SiO}_4$	-5.2	-22	-13.1	-40	-7.2	-23
Periclase, MgO	-3.6	-51	ns	ns	+3.6	+68
Bredigite, $\text{Ca}_{14}\text{Mg}_2(\text{SiO}_4)_8$	-8.5	-50	-1.7	-9	ns	ns
Åkermanite, $\text{Ca}_2\text{MgSi}_2\text{O}_7$	ns	ns	-1.4	-37	-1.6	-55
Merwinite, $\text{Ca}_3\text{Mg}(\text{SiO}_4)_2$	+11.1	+98	+14.7	+166	+5.0	+71
Calcite, $\text{CaCO}_3$	+3.2	+28	+3.1	+86	ns	ns

ns = not significant (less than 1  $\Delta$ wt% or 20 % change)