

Thin film MoS₂ nanocrystal based ultraviolet photodetector

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Abstract: We report on the development of UV range photodetector based on molybdenum disulfide nanocrystals (MoS₂-NCs). The inorganic MoS₂-NCs are produced by pulsed laser ablation technique in deionized water and the colloidal MoS₂-NCs are characterized by transmission electron microscopy, Raman spectroscopy, X-ray diffraction and UV/VIS absorption measurements. The photoresponse studies indicate that the fabricated MoS₂-NCs photodetector (MoS₂-NCs PD) operates well within 300–400 nm UV range, with diminishing response at visible wavelengths, due to the MoS₂-NCs absorption characteristics. The structural and the optical properties of laser generated MoS₂-NCs suggest promising applications in the field of photonics and optoelectronics.

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1. Introduction

Semiconductor nanocrystals have been under intense investigation for the past decade due to the potential applications in new nanomaterials based devices. MoS₂ is a newly emerging transition-metal dichalcogenide (TMD) semiconductor material which is rallied up as finite-energy-bandgap alternative to graphene in advanced electronic and photonic device applications [1, 2]. The presence of its natural bandgap (~1.2 eV indirect bandgap in multilayer/bulk form, ~1.85 eV direct bandgap in monolayer form) makes it advantageous compared to graphene-based CMOS-like logic device applications [1, 5]. It is also shown that MoS₂ TMDs resist oxidation in moist air environments at temperatures up to 85°C which makes them more durable in device fabrication compared to Silicon and Germanium nanostructures [3, 6]. Si-based transistors are outperformed at the scaling limit by TMDs owing to the absence of dangling bonds and low dimensionality [1, 5].

Due to their 2D ultra-thin atomic layer structure, MoS₂/WS₂ TMDs are shown to exhibit unique physical, optical and electrical properties [4, 7]. Their fullerene like nature [8] and their excellent mechanical properties such as friction-reduction and self-lubrication [9] along with their unique optoelectronic properties (UV light absorption [10], band-gap tunability [1, 2]) make MoS₂/WS₂ nanostructures applicable in heterogeneous catalysis, hydrogen storage, lithium-magnesium ion batteries [11, 12], various bio-medical applications [9, 11, 13] and as well as, various electronic and optoelectronic device applications (transistor, photo-transistor and solar cell applications) [2, 7, 4]. Since they possess interesting optoelectronic properties that are tunable by physical layer thickness, various electronic and photonic devices are fabricated based on MoS₂ TMDs, such as hybrid bulk heterojunction solar cells (BHJs) [4], single-layer transistors [2], single-layer phototransistors [7] and back-gated bi-layer field-effect transistors [1]. Despite the fact that, single-layer MoS₂ has a large direct bandgap of 1.8 eV and a low electron mobility of 0.5-3 cm²V⁻¹s⁻¹, Radisavljevic et al. showed that, in a single-layer transistor based on MoS₂ TMDs, it is possible to achieve an electron mobility of 200 cm²V⁻¹s⁻¹ and an on/off ratio of 10⁸ using hafnium oxide gate dielectric material [2]. This work has been very good indicator of MoS₂ TMDs potential to complement graphene in electronics applications [2]. Recently, Liu et al. showed that, for MoS₂ TMD based transistors, it is further possible to achieve an electron mobility of 517 cm²V⁻¹s⁻¹ using a multilayer MoS₂ flake as the channel material [14], compared to the single-layer one used by Radisavljevic [2]. In addition to transistor applications, it is shown in the literature that, MoS₂ TMDs are also potential candidates in solar cell applications [4, 15]. In 1997 Gourmelon et al. came up with the idea of using MoS₂ thin films in solar cell applications [15] and later on, Shanmugam et al. fabricated a hybrid bulk heterojunction (BHJ) solar cell based on MoS₂/TiO₂ nanocomposites and showed that MoS₂ TMDs are also applicable in solar cell applications [4]. All these research results demonstrate the great potential that MoS₂ TMDs have in terms of nanoelectronics and nanophotonics applications.

There are various ways of synthesizing MoS₂ TMD nanostructures including solution-based exfoliation [16], CVD based synthesis [17], thermal decomposition [18], powder sublimation [19] and electrochemical/chemical synthesis [3]. Another promising method for

obtaining MoS₂ TMD nanostructures is the laser ablation method [11]. Laser ablation is a versatile nanoparticle synthesis technique which can be applied at industrial scales. However, there are very limited reports on the applications of MoS₂-NCs synthesized by laser ablation. Wu et al showed that MoS₂ TMD nanoparticles that are obtained through laser ablation are fullerene-like and have good solubility and biocompatible nature which opens the doors for MoS₂ TMD nanostructure applications in various biomedical areas [11].

Here, we present for the first time the development of UV photodetector based on MoS₂-NCs. The pulsed laser ablation technique in deionized water allows generating crystalline inorganic MoS₂-NCs. The colloidal MoS₂-NCs were characterized by transmission electron microscopy, X-ray diffraction and UV/VIS absorption measurements. The photocurrent measurements demonstrate that the fabricated MoS₂-NCs PD operates in UV range from 300 to 400 nm with diminishing response at the visible range.

2. Experiments and results

Colloidal MoS₂-NCs solution was generated by using pulsed laser ablation technique in deionized water using bulk MoS₂ powder (99,99% Sigma-Aldrich). The commercial nanosecond pulsed ND:YLF laser (Empower Q-Switched Laser, Spectra Physics) operated at 527 nm with pulse duration of 100 ns, average output power of 16 W at a pulse repetition rate of 1 kHz corresponding to a pulse energy of 16 mJ were used. The laser beam was focused on the bulk MoS₂ powder target placed in a glass vial containing 10 ml of deionized water using a plano-convex lens with a focal length of 50 mm. The height of liquids layer over the bulk MoS₂ powder target was about 5 mm. The pulsed laser ablation process was carried out for 60 min. To obtain colloidal MoS₂-NCs solution, the solution was continuously stirred by a magnetic stirrer at 800 rpm. MoS₂-NCs PD fabrication was performed on highly p-type (10-18 mohm-cm Boron doped) Si substrate. The substrate was cleaned through standard solvent degreasing and cleaning procedures involving acetone, isopropanol and deionized water. A germanium wetting layer (3nm) followed by 25-nm-thick Ag film was thermally evaporated (VAKSIS-MIDAS) on Si substrate. This was followed by deposition of 5 nm Al₂O₃ (trimethylaluminum and H₂O as the precursor gases 50 cycles at 250°C) on the Ag film by atomic layer deposition (ALD Cambridge Nanotech Savannah 100). Colloidal MoS₂-NCs solution was initially treated in a Branson 2510 type ultrasonic bath with an operating frequency of 40 kHz. In order to form a thin film of MoS₂-NCs on top of the Al₂O₃/Ag/Ge/Si stack, 1.2 grams of polyvinyl alcohol was added to a 20 ml MoS₂-NCs/deionized water solution that was prepared through laser ablation. The solution was stirred vigorously at 90°C for 48 hours in order to dissolve the added polymer and obtain homogeneous solution. The sample surface was then coated with the prepared MoS₂-NCs solution using dip-coating method. This procedure was followed by a second atomic layer deposition of 5 nm Al₂O₃ on top of MoS₂-NCs/Al₂O₃/Ag/Ge/Si structure using the same ALD recipe. 100-nm-thick Ag was evaporated to form the top and bottom electrical contact pads. An Al layer was used for efficient photo-carrier collection and was kept very thin (<10 nm) to allow for light penetration. As supporting information, we should add the transmission vs wavelength for 10 nm Aluminum layer. Illustration of the device structure and an SEM image of a completed photodetector is shown in Fig. 3. The morphology of bulk MoS₂ powder and MoS₂-NCs were performed by using a FEI-Quanta 200 FEG SEM instrument at an accelerating voltage of 20 kV. The morphology and the elemental analysis of the MoS₂-NCs were performed by using a FEI - Tecnai G²F30 TEM instrument. The optical absorption spectra of the colloidal MoS₂-NCs solution were obtained with a Varian Cary 5000 UV/Vis/NIR spectrophotometer. Raman spectrum of the MoS₂-NCs was performed by Witec Alpha 300S Micro Raman spectrometer with an excitation wavelength of 532 nm (laser power: 10 mW).

We first present the synthesis of colloidal MoS₂-NCs in deionized water. 1 mg of 2H-MoS₂ powder was used as starting material. The powder was first added to 10 ml of deionized water. During the laser ablation process, the formation of colloidal nanoparticle solution

(CNS) in liquid media was observed. In order to obtain well-dispersed nanoparticles solution, the CNS was followed by ultrasonic post-treatment for 100 minutes. The color of the final CNS product became light-orange. To further understand and better characterize the structure and the composition of MoS₂ nanoparticles, detailed scanning transmission electron microscope (STEM) study was performed with High-Angle Annular Dark Field (HAADF). Representative TEM images of the CNS containing MoS₂ nanoparticles obtained by nanosecond pulsed laser ablation are shown in Fig. 1(a) and Fig. 1(b). The images show that nanosecond pulsed laser ablation of a bulk MoS₂ powder target and then ultrasonic treatment of product produced sub-50 nm sized spherical MoS₂ nanoparticles in deionized water. The elemental compositions of the MoS₂ nanoparticles (the inset of Fig. 1(a)) were obtained by EDAX analysis. The peaks related to carbon (C), oxygen (O) and copper (Cu) are associated with the TEM grid used. The presence of the Mo and S peak in the EDAX spectrum confirms that MoS₂ nanoparticles were successfully generated by laser ablation technique with no significant impurities observed. The nanocrystal size distribution is given in the inset of Fig. 1(b). Nanocrystal sizes range from 4.5 nm to 50 nm and most of the nanocrystals are within 10 nm-25 nm range.

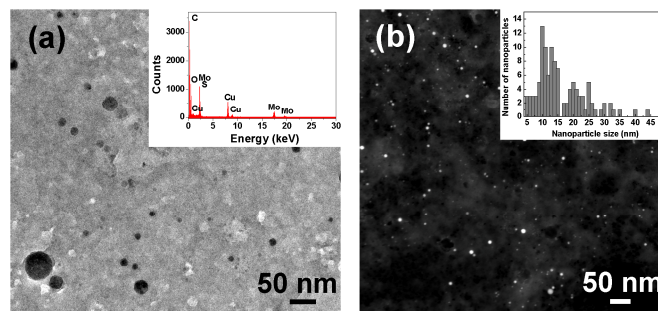


Fig. 1. (a) Representative TEM image of MoS₂-NCs after ultrasonic treatment and the EDAX spectrum in the inset, (b) representative STEM HAADF image of ultrasonic-treated MoS₂-NCs and the size histogram in the inset.

The optical properties of MoS₂-NCs have been characterized by UV/VIS absorption spectroscopy technique. Figure 2(a) shows the normalized optical absorption spectrum of MoS₂-NCs. The optical absorption spectrum of MoS₂-NCs shows a minimum optical absorption feature at 500 nm, strong rising absorption edge shifts towards UV region. The Raman spectrum of MoS₂-NCs given in Fig. 2(b) shows two sharp peaks centered at 381.7 cm⁻¹ and 407.5 cm⁻¹. These are attributed to the in-plane E_{2g} and out-of-plane A_{1g} vibration of MoS₂-NCs, respectively [11], providing further evidence that the obtained nanocrystals are MoS₂-NCs.

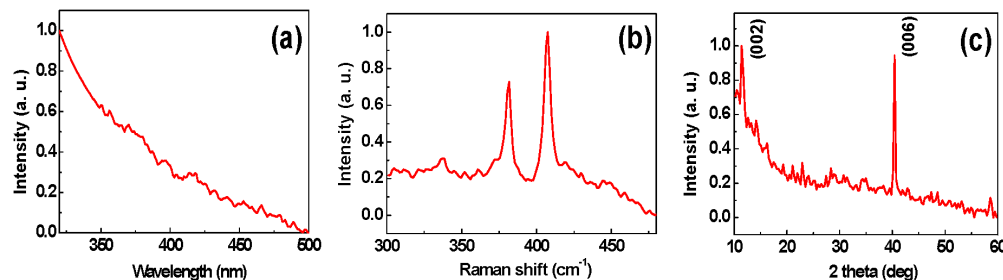


Fig. 2. UV-Vis absorption spectrum of ultrasonic-treated MoS₂-NCs in deionized water (a), Raman spectra of MoS₂-NCs (b), XRD profile of MoS₂-NCs (c).

XRD measurements analyses were also carried out to determine the crystalline structure and the composition of MoS₂-NCs. XRD pattern of ultrasonic-treated MoS₂ NCs is shown in Fig. 2(c). Two main sharp and narrow diffraction peaks at 11.45° (002) and 40.46° (006) are observed. Results indicate that good quality MoS₂-NCs with hexagonal structure were successfully generated by laser ablation technique with no significant impurities [11].

The diagram, SEM image, the electrical and photoresponsivity measurements of the fabricated MoS₂-NCs PD based on MoS₂-NCs obtained through laser ablation are given as in the following. Figure 3(a) shows an illustration of the fabricated MoS₂-NCs PD and Fig. 3(b) shows a top-view SEM image of a completed MoS₂-NCs PD. Figure 4(a) plots electrical current-voltage measurements exhibiting a characteristic exponentially increasing current with applied voltage typical of an insulator-semiconductor-insulator device. We have observed very low dark current density of 3.9 $\mu\text{A}/\text{cm}^2$ at 10V bias from the fabricated MoS₂-NCs PD limited by tunneling through the insulator layers.

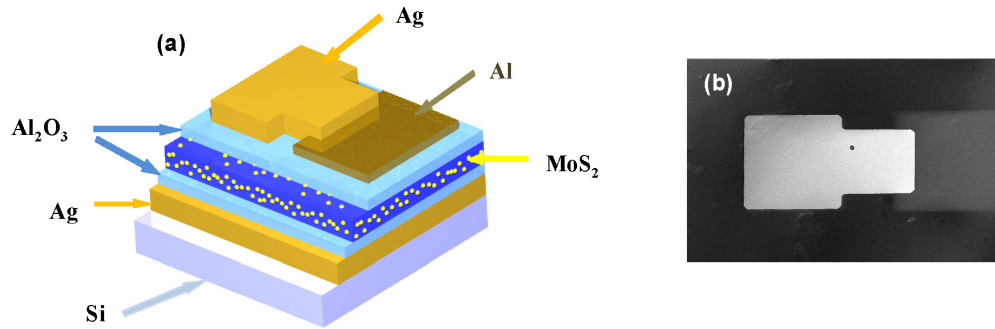


Fig. 3. Diagram of the fabricated MoS₂-NC PD (a) and SEM image of top view of the fabricated MoS₂-NC PD (b).

Photogenerated current (I_{photo}) is measured with mechanically chopped, monochromated white light source (1/8m grating 600 lines/mm) and a lock in amplifier (SRS 830). The optical beam is normally incident on the device as shown in Fig. 3(a). Incident light power (P_{in}) is recorded with a calibrated Silicon photodetector. Spectral responsivity is calculated as the ratio of measured I_{photo} to P_{in} at each wavelength. Figure 4(b) plots measured relative responsivity vs wavelength of the photodetector. The responsivity at 340 nm illumination is 2.5 times that of 400 nm which agrees well with measured MoS₂-NCs absorption characteristics (Fig. 2(a)).

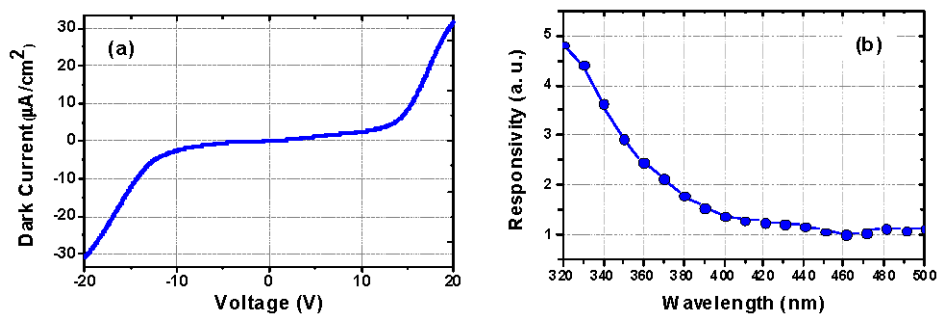


Fig. 4. Dark current-voltage measurements for the fabricated MoS₂-NCs PD (a) and the photoresponsivity vs wavelength data (b).

3. Conclusion

MoS₂-NCs have been obtained through laser ablation in liquid and we have demonstrated that thin film of MoS₂-NCs that have been obtained through laser ablation could be used as a

material for the fabrication of ultraviolet photodetectors. The synthesized MoS₂-NCs gave characteristic Raman peaks at 381.7 cm⁻¹ and 407.5 cm⁻¹ that are attributed to the in-plane E_{2g}¹ and out-of-plane A_{1g} vibration modes of MoS₂ and the crystals gave strong UV absorption towards the UV region. The fabricated MoS₂-NCs PD gave a dark current value of 3.9 μA/cm² under 10 V external bias and the fabricated MoS₂-NCs PD generated photoresponse between 300 to 400 nm UV range and diminishing photoresponse at the visible range. A good agreement is observed between the MoS₂-NCs absorption characteristics and the photoresponsivity data. We believe that our demonstration will open way to using MoS₂-NCs in new optoelectronic device applications. Photodetectors that are fabricated based on such MoS₂-NCs could offer new advantages over Silicon and Germanium based photodetectors due to the low oxidation tendency of MoS₂-NCs [3, 6].

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