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Highly Ordered Nanoscale Film Morphologies of Block Copolymers Governed by Nonlinear Topologies

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ABSTRACT: Among many properties of cyclic block copolymers, the notable domain spacing (dspacing) reduction offers nonlinear topology as an effective tool for developing block copolymers for nanolithography. However, the current consensus regarding the topology-morphology correlation is ambiguous and in need of more studies. Here we present the morphological investigation on nanoscale films of cyclic tadpole-shaped poly(*n*-decyl and glycidyl ether-block-2-(2-(2methoxyethoxy)ethoxy)ethyl glycidyl ether)s and their linear counterpart via synchrotron grazingincidence X-ray scattering. All copolymers form phase-separated nanostructures, in which only the nonlinear copolymers form highly ordered and unidirectional nanostructures. Additionally, dspacings of cyclic and tadpole-shaped block copolymers are 49.3%-53.7% and 25.0%-32.5% shorter than that of their linear counterpart, respectively, exhibiting greater or comparable d-spacing reductions against the experimentally and theoretically achieved values from the literature. Overall, this study demonstrates that cyclic and tadpole topologies can be utilized in developing materials with miniaturized dimensions, high structural ordering, and unidirectional orientation for various nanotechnology applications.

Pioneering work on cyclic polymers dates back to 1946 when cyclic poly(dimethylsiloxane) was reported as the first synthetic cyclic polymer.¹⁻³ In 1949, Zimm and Stockmayer theoretically estimated an ideal, flexible cyclic polymer to exhibit a shorter radius of gyration than its linear analogue of the same molecular weight.⁴ Then, 1962 onwards, studies on cyclic deoxyribonucleic acids,⁵⁻⁸ cyclic proteins,^{9,10} and cyclic polysaccharides¹¹ inspired new synthetic methods for producing various cyclic homopolymers with high yield and purity.¹²⁻²⁷ Many reports have discussed the absence of chain ends as the primary reason for the unique physical properties of cyclic polymers, such as higher glass transition temperature, higher melting temperature, and lower viscosity than the linear counterparts.^{18-22,26-39}

In addition to homopolymers, many synthetic strategies producing various cyclic block copolymers have been reported.^{17-19,29,30,40-45} However, there is only a handful of reports exploring the topological influence on phase-separated morphological features in bulk state.^{40,46-49} Based on those reports, the consensus is that cyclic block copolymers form the same type of nanostructure as their linear counterpart, but with a 5%–16% shorter domain spacing (*d*-spacing).⁴⁶⁻⁴⁸ Interestingly, these experimentally observed *d*-spacing reductions are lesser than the theoretical predictions of 30%–37%.⁴⁹⁻⁵¹ Moreover, some studies reported controversial results where different nanostructures were observed between cyclic and linear iterations of poly(styrene-*block*-isoprene)⁵² and poly(styrene-*block*-butadiene).⁴⁶ Compared to bulk state, there are fewer investigations on nanoscale film morphologies. One report discusses hexagonally packed cylinders formed by both cyclic and linear poly(styrene-*block*-ethylene oxide)s (PS-*b*-PEO), where cyclic PS-*b*-PEO exhibits a 33% shorter *d*-spacing than that of linear PS-*b*-PEO.⁵³ This *d*-spacing reduction agrees with the theoretical prediction, ⁴⁹⁻⁵¹ however, we have previously achieved 51.3%–72.8% *d*-spacing reduction and characterized different morphological types.⁵⁴ The report, however, is based on bicyclic topology, not cyclic. Nonetheless, the fact that there are only two reports investigating nanoscale film morphologies

of cyclic block copolymers indicates the need for additional studies. Overall, synthesis of cyclic block copolymers has progressed significantly, and the correlations among cyclic topology, chemical composition, physical properties, and morphology need further investigation. Particularly, morphology remains as the least explored aspect of cyclic block copolymers.

In this study, we report the first quantitative investigation of nanoscale film morphologies of poly(n-decyl glycidyl ether-block-2-(2-(2-methoxyethoxy)ethoxy)ethyl glycidyl ether)s (PDGE-b-PTEGGE) with cyclic and tadpole topologies—cy-BCP, tp-A-BCP, and tp-B-BCP (Figure 1; Table S1 in Supporting Information)—using synchrotron grazing-incidence X-ray scattering (GIXS). This quantitative study provides detailed structural parameters and discusses the correlation between molecular topology and morphological features. All the block copolymers are synthesized with a narrow dispersity (D) of 1.04, and they phase-separate into nanostructures driven by the immiscibility between PDGE and PTEGGE blocks. Despite nearly equivalent volume fractions (degree of polymerization for each block is approximately 50), all the copolymers form either distorted hexagonal cylindrical or lamellar structures, a unique behavior deviating from the aforementioned consensus and conventional volume fraction rule. For each copolymer, the d-spacing varies significantly with topology. cy-BCP exhibits a d-spacing 49.3%-53.7% shorter than the linear counterpart (l-BCP), in which the d-spacing reduction is considerably greater than experimental and theoretical results from the literature. Additionally, tp-A-BCP and tp-B-BCP exhibit 25.0%-32.5% shorter *d*-spacings than *l*-BCP. Overall, the cyclic and tadpole block copolymers form nanostructures with higher degrees of structural integrity and orientation than *l*-BCP. All the structural details are discussed in correlation with the molecular topology effects.



Figure 1. Chemical structures of cyclic and tadpole-shaped poly(*n*-decyl glycidyl ether-*block*-2-(2-(2-methoxy)ethoxy)ethyl glycidyl ether)s and their linear counterpart.

cy-BCP, *tp*-*A*-BCP, *tp*-*B*-BCP, and *l*-BCP were synthesized and characterized, as described in the literature.^{43,55} Their molecular characteristics are summarized in Table S1. All topological copolymers exhibit glass transitions below -36.0 °C and crystal melting points below 23.5 °C (Figure S1 and Table S2). In this study, nanoscale films of each block copolymer were prepared through spin-coating onto silicon substrates with a 0.5 wt% polymer solution in tetrahydrofuran, which was filtered through a syringe equipped with a polytetrafluoroethylene filter membrane (0.2 μ m pores). The film thicknesses were found to be 100–120 nm after drying in vacuum at room temperature for 24 h. Individual films were stored in vacuum at room temperature prior to grazing-incidence small- and wide-angle X-ray scattering (GISAXS, GIWAXS) measurements, which were also conducted at room temperature. Because phase transitions occur below 23.5 °C, the prepared copolymer films were considered to be thermally annealed through the drying process and storage period.



Figure 2. Representative GISAXS data of *cy*-BCP films (100–120 nm) measured at SDD = 2909.8 mm; $\alpha_i = 0.1276^\circ$ at room temperature using a synchrotron X-ray beam ($\lambda = 0.12095$ nm). (a) 2D scattering image in angle space. (b) Reconstructed 2D scattering image based on the structural parameters. (c) 2D scattering image in (a) reduced to scattering vector space. (d) 1D out-of-plane scattering profile along the meridian line at $2\theta_f = 0.485^\circ$ from (a), where the red line is the sum of calculated intensities of short-range ordered (green line) and long-range ordered (blue line) cylindrical domains (the scattering peaks generated by the reflected and transmitted X-ray beam are indicated as "R" and "T", respectively). (e) 1D in-plane scattering profile along the equatorial line at $\alpha_f = 0.229^\circ$ from (a). In (d) and (e), open circles represent the measured scattering intensities and solid red lines are the calculated intensities based on the GIXS formula of hexagonal cylindrical structure model (see Supporting Information). (f) 1D azimuthal scattering profile from the azimuth at q = 0.564 nm⁻¹ from (c) where the open circles are the measured data and the solid lines were obtained *via* data deconvolution: the blue and green solid lines are the scattering peaks of cylindrical domains in hexagonal packing order, the dashed magenta line corresponds to Yoneda peak, the dashed orange line is residual intensities from the reflected X-ray beam, and the red solid line is the total sum of all deconvoluted peaks.

A representative two-dimensional (2D) GIWAXS image of the cy-BCP nanoscale films (100-120 nm thick) is shown in Figures S2a and b. The GIWAXS data revealed two isotropic scattering peaks: one near 3.32° (2.09 nm d-spacing) and another at approximately 14.59° (0.476 nm d-spacing). The two peaks are commonly referred to as amorphous halos, which are also observed for tp-A-BCP, tp-B-BCP, and l-BCP (2D images not shown). The in-plane scattering profiles extracted from the measured data are compared in Figure S2c, in which the peak near 3° correlates to the mean interdistance between the polymer chains, and the peak near 15° corresponds to the mean intramolecular and intermolecular interdistances between *n*-decyl and 2-(2-(2methoxyethoxy)ethoxy)ethyl bristles. Collectively, the GIWAXS measurements confirmed that the PDGE and PTEGGE blocks of all copolymers are amorphous in the nanoscale films at room temperature.

Figure 2a shows a representative GISAXS image of the *cy*-BCP films (100–120 nm) at room temperature; its *q*-space image is shown in Figure 2c. The image clearly reveals a hexagonal array of peaks, in which each peak appears as a pair split in the α_f axis direction. The peak splitting phenomenon occurs specifically for in-plane oriented (*i.e.* horizontal) nanostructures, and it implies that, based on GIXS optics, the above and below peaks are generated by the reflected and transmitted X-ray beams, respectively. In addition, from the specular reflection position, the relative scattering vector lengths of the peaks at $\alpha_f = 0.344^\circ$, 0.904°, and 1.460° occurring along the meridian line at $2\theta_f = 0.485^\circ$ are found to be 1, $\sqrt{3}$, and 2, respectively. These are typical scattering characteristics of horizontal hexagonally packed cylindrical structures. Considering this, we quantitatively analyzed the data in Figure 2a with the GIXS formula of the hexagonally packed cylinder model (details provided in Supporting Information and Figure S3), in which two-phase cylinders (core and shell) are surrounded by a matrix phase, derived for this study. The out-of-plane and the in-plane scattering profiles extracted along the median line at $2\theta_f = 0.485^\circ$ and equatorial line at $\alpha_f = 0.229^\circ$, respectively,

could be fitted reasonably well with the hexagonally packed cylinder model (Figures 2d and 2e). The analysis on the azimuthal scattering profile (Figure 2f) extracted at q = 0.564 nm⁻¹ from the q-space image (Figure 2c) quantified the cylinder orientation with respect to the substrate. The structural parameters determined from the aforementioned analyses are summarized in Table 1. Based on the obtained structural parameters, a 2D GISAXS image was reconstructed through the GIXS formula (Figure 2b), which agreed well with the measured data and confirmed successful data analysis.



Figure 3. Schematic representations of nanostructures (cross-sectional view) inside topological block copolymer films. (a) Short-range order (60 vol%) and (b) long-range order (40 vol%) horizontal hexagonally-packed cylindrical structure inside *cy*-BCP film; for (a) and (b), the blue cylinder phase is assigned to PTEGGE block and the red matrix phase is assigned to PDGE block of *cy*-BCP. (c) Horizontal hexagonally-packed cylindrical structure inside *tp-A*-BCP film where the blue cylinder phase is assigned to linear PTEGGE block and the red matrix phase is assigned to cyclic PDGE block of *tp-A*-BCP. (d) Horizontal lamellar structure inside *tp-B*-BCP film where red sublayers l_1 and l_2 are formed by the linear PDGE block, and the blue sublayer l_3 is formed by the cyclic PTEGGE block of *tp-B*-BCP.

Nanoscale film ⁻ morphology	Topological polymer							
	cy-BCP		tp-A-BCP	<i>tp-B</i> -BCP	<i>l</i> -BCP			
	C)	~~	~0	\sim	\sim		
Hexagonal	horizontal		horizontal					
cylindrical	short-range-	long-range-						
structure	order (sro) $11.80(0.50)r$	order ($1ro$)	19.00 (1.22)					
L_z (nm)	11.80 (0.50)	12.17 (0.50)	18.00 (1.22)					
$L_{\rm y}$ ° (nm)	8.25 (0.36)	8.45 (0.20)	11.10 (0.86)					
γ^{c}	1.43	1.50	1.62					
$r_{\rm cz}^{a}$ (nm)	4.00 (0.40)	4.40 (0.50)	3.10 (0.70)					
$r_{\rm cy} e({\rm nm})$	2.50 (0.30)	2.90 (0.20)	1.90 (0.70)					
$t_{\rm sz}^{f}({\rm nm})$	0.70 (0.30)	0.50 (0.05)	4.10 (1.00)					
$t_{\rm sy} ^g ({\rm nm})$	0.80 (0.20)	0.40 (0.01)	2.50 (0.50)					
ε^{h}	1.42	1.49	1.64					
g^{i}	0.06	0.01	0.08					
ϕ^{j} (vol%)	60	40	100					
$\overline{\varphi}^{k}$ (deg.)	0		0					
$\sigma_{\varphi}{}^{l}$ (deg.)	1.63		1.60					
$O_{\rm s}$ ^m	0.994		0.994					
Lamellar				horizontal	horizontal	vertical		
structure					nonzontar			
D_{L} " (nm)				17.20 (1.26)	24.00 (2.16)	25.50 (2.79)		
l_1^{o} (nm)				2.60 (0.80)	9.40 (1.30)	9.00 (1.00)		
$l_2 p$ (nm)				2.90 (0.40)	2.50 (0.80)	3.60 (1.70)		
l_3^q (nm)				8.80 (0.89)	9.60 (1.53)	9.30 (1.97)		
g				0.04	0.33	0.35		
$\overline{\varphi}$ (deg.)				0	0	90		
$\sigma_{\varphi}(\deg.)$				2.00	6.75	6.25		
$O_{\rm s}$				0.974	0.902	-0.210		
ϕ (vol%)				100	50	50		

Table 1. Morphological parameters of nanoscale films (100–120 nm thick) of topological polymers

^{*a*}Mean interdistance between the arrays of the in-plane oriented cylindrical domains. ^{*b*}Mean center-to-center distance of the cylindrical domains lain in the film plane. ^{*c*}Ratio between L_z and L_y (= L_z/L_y). ^{*d*}Core radius of elliptical cylindrical domain along the *z*-axis which is parallel to the out-of-plane of the film. ^{*e*}Core radius of elliptical cylindrical domain along the *y*-axis which is parallel to the in-plane of the film. ^{*f*}Shell thickness of elliptical cylindrical domain along the *z*-axis which is parallel to the out-of-plane of the film. ^{*f*}Shell thickness of elliptical cylindrical domain along the *y*-axis which is parallel to the out-of-plane of the film. ^{*f*}Shell thickness of elliptical cylindrical domain along the *y*-axis which is parallel to the in-plane of the film. ^{*h*}Elliptical axial ratio (= polar radius/equatorial radius). ^{*f*}Paracrystal distortion factor of nanostructure (i.e., hexagonal cylinder structure or lamellar structure) along the *z*-axis which is parallel to the out-of-plane of the film. ^{*f*}Wolume fraction in percent. ^{*k*}Mean value of the polar angle φ (i.e., orientation angle) between the orientation vector *n* set along a direction normal to the {001} plane of horizontal hexagonal cylindrical structure or the in-plane of lamellar structure, and the out-of-plane direction of the film. ^{*l*}Standard deviation for the polar angle φ . ^{*m*}Second-order orientation factor of nanostructure (i.e., hexagonal cylindrical structure or lamellar structure). ^{*n*}Long period of lamellar structure. ^{*o*}Thickness of the l_1 layer. ^{*p*}Thickness of the l_2 layer. ^{*q*}Thickness of the l_3 layer. ^{*r*}Standard deviation.

Interestingly, the quantitative analysis indicated two different levels of order existing within the cy-BCP film, the majority (60 vol%) of which exhibited a short-range order, and the rest (40 vol%) exhibited a long-range order (Figures 3a and b). The short-range order domain consisted of cylinders with elliptical cross-sections with a semi-major radius $R_{\text{sro,z}}$ of 4.70 nm [= 4.00 nm (core radius: r_{cz}) + 0.70 nm (shell thickness: t_{sz})] along the out-of-plane direction of the film (*i.e. z*-axis) and a semiminor radius $R_{\text{sro,y}}$ of 3.30 nm [= 2.50 nm (r_{cy}) + 0.80 nm (t_{sy})] along the in-plane of the film (*i.e.* yaxis). Owing to the elliptical nature, the cylinders are positioned at non-conventional d-spacings, where L_z and L_y are 11.80 nm and 8.25 nm along the z- and y-axes, respectively. The lattice distortion factor g for the hexagonal arrangement of cylinders in the short-range order domain was 0.06. The long-range order domain consisted of elliptical cylinders with $R_{\rm lro,z}$ of 4.90 nm and $R_{\rm lro,y}$ of 3.30 nm. These elliptical cylinders were hexagonally packed with *d*-spacings of $L_z = 12.17$ nm and $L_y = 8.45$ nm, and the corresponding hexagonal lattice distortion factor g (g-factor) was 0.01. Such a low gfactor value indicates that the cylinders were exceptionally ordered in this domain. The sum of the short- and long-range order domains of the hexagonally packed cylinders was found to have a secondorder orientation factor O_s of 0.994 and a mean polar angle $\overline{\varphi}$ of 0° (φ is the angle between the orientation vector \mathbf{n} set normal to the {001} plane of the horizontal hexagonal cylinders as shown in Figure S4, and the out-of-plane direction of the film; $\sigma_{\varphi} = 1.63^{\circ}$, standard deviation for φ). Based on the structural parameters, the volume fraction of the cylinders was estimated to be 50.0% for the shortrange order domain and 49.4% for the long-range order domains. Considering that the PDGE block in cy-BCP has a slightly higher volume fraction than that of the PTEGGE block, we could assign the cylinders and matrix comprising of the PTEGGE and PDGE blocks, respectively.

These are significant morphological features compared to *l*-BCP, which forms a mixture of horizontal and vertical lamellar structures (Figures S5, S6 and Table 1). The observed *d*-spacing reduction of *cy*-BCP was 49.3%–53.7% by comparing the L_z values of both the short- and long-range order domains against long periods (D_L) of *l*-BCP. This resulted in a greater reduction than the theoretical predictions⁴⁹⁻⁵¹ and experimental observations.^{46-48,53} In addition, *cy*-BCP cylindrical structure exhibits a considerably greater structural integrity and ordering, as indicated by the noticeable differences in the *g*-factor values (g_{cy} -BCP = 0.01, 0.06; $g_{l-BCP} = 0.33, 0.35$). The visual assessment (Figure 2a shows up to third-order peaks) and statistical evaluation (g_{cy} -BCP = 0.01, 0.06) of the cylindrical structure are noteworthy because it is uncommon to observe a well-defined polymer nanostructure to this degree inside nanoscale films. Furthermore, cylindrical structures formed by *cy*-BCP present a unidirectional orientation with O_s approaching unity, whereas *l*-BCP lamellae lack orientation control.



Figure 4. Representative GISAXS data of *tp-A*-BCP films (100–120 nm thick) measured at SDD = 2909.8 nm; $\alpha_i = 0.1361^\circ$ at room temperature using a synchrotron X-ray beam ($\lambda = 0.12095$ nm). (a) 2D scattering image in angle space. (b) Reconstructed 2D scattering image based on the structural parameters. (c) 2D scattering image in (a) reduced to scattering vector space. (d) 1D out-of-plane scattering profile along the meridian line at $2\theta_f = 0.101^\circ$ from (a), where the scattering peaks generated by the reflected and transmitted X-ray beam are indicated as "R" and "T", respectively. (e) 1D in-plane scattering profile along the equatorial line at $\alpha_f = 0.192^\circ$ from the scattering image in (a). In (d) and (e), open circles represent the measured scattering intensities and solid red lines are the calculated intensities based on the GIXS formula of hexagonal cylindrical structure model (see Supporting Information). (f) 1D azimuthal scattering profile from the azimuth at q = 0.417 nm⁻¹ from (c) where the open circles are the measured data and the solid lines were obtained *via* data deconvolution: the blue and green solid lines are the scattering peaks of cylindrical domains in hexagonal packing order, the dashed magenta line corresponds to Yoneda peak, the dashed orange line is residual intensities from the reflected X-ray beam, and the red solid line is the total sum of all deconvoluted peaks.

The GISAXS pattern of *tp-A*-BCP film in Figure 4a resembles that of *cy*-BCP with the hexagonal array of peaks and peak splitting. Using the hexagonally packed cylinder model, the outof-plane and in-plane scattering profiles extracted along the median and equatorial lines at $2\theta_f = 0.101^\circ$ and $\alpha_f = 0.192^\circ$, respectively, could be fitted reasonably well (Figures 4d and e). The azimuthal scattering profile (Figure 4f) extracted at q = 0.417 nm⁻¹ from Figure 4c was fitted to parameterize cylinder orientation. The structural parameters thus obtained are summarized in Table 1. The reconstructed 2D GISAXS image (Figure 4b) complied well with the measured data and confirmed successful data analysis. *tp-A*-BCP formed elliptical cylinders (Figure 3c) with the semi-major (R_z) and semi-minor (R_y) radii of 7.20 nm and 4.40 nm, respectively. The hexagonal arrangement of cylinders were defined by $L_z = 18.00$ nm, $L_y = 11.00$ nm, g = 0.08, and $O_s = 0.994$ ($\overline{\varphi} = 0^\circ$ and $\sigma_{\varphi} =$ 1.60°). Further, the cylinders revealed a volume fraction of 49.8%, implying that the linear PTEGGE block and the cyclic PDGE block formed the cylinders and the matrix, respectively.

The self-assembly of *tp-A*-BCP into a well-defined cylindrical structure with an excellent unidirectional orientation is a similar behavior exhibited by *cy*-BCP. However, the cylinders formed by *tp-A*-BCP are larger, positioned further apart, and relatively less ordered than those of *cy*-BCP (g_{cy} -BCP = 0.01, 0.06; g_{tp-A} -BCP = 0.08). Moreover, the 25.0%–29.4% *d*-spacing reduction compared to *l*-BCP corresponds with the theoretical predictions,⁴⁹⁻⁵¹ but this result is not directly applicable since the predictions are based on the cyclic topology.



Figure 5. Representative GISAXS data of *tp-B*-BCP films (100–120 nm thick) measured at SDD = 2909.8 mm; $\alpha_i = 0.1353^\circ$ at room temperature using a synchrotron X-ray beam ($\lambda = 0.12095$ nm). (a) 2D scattering image in angle space. (b) Reconstructed 2D scattering image based on the structural parameters. (c) 2D scattering image in (a) reduced to scattering vector space. (d) 1D out-of-plane scattering profile along the meridian line at $2\theta_f = 0.117^\circ$ from (a), where the scattering peaks generated by the reflected and transmitted X-ray beam are indicated as "R" and "T", respectively. (e) 1D in-plane scattering profile along the equatorial line at $\alpha_f = 0.202^\circ$ from (a). In (d) and (e), open circles represent the measured scattering intensities and solid red lines are the calculated intensities based on the GIXS formula of lamellar structure model (see Supporting Information). (f) 1D azimuthal scattering profile from the azimuth at q = 0.430 nm⁻¹ from (c) where the open circles are the measured data and the solid lines were obtained *via* data deconvolution: the blue solid line is the first order scattering peak of lamellar structure, the dashed magenta line corresponds to Yoneda peak, the dashed orange line is residual intensities from the reflected X-ray beam, and the red solid line is the total sum of all deconvoluted peaks.

The tp-B-BCP film produced a 2D GISAXS pattern (Figure 5a) different from those of tp-A-

BCP and *cy*-BCP, with three pairs of peaks appearing along the meridian line at $2\theta_f = 0^\circ$. Moreover,

relative scattering vector lengths of peaks at $\alpha_{\rm f} = 0.460^\circ$, 0.868° , and 1.267° were 1, 2, and 3, respectively, from the specular reflection position. These are typical scattering characteristics of a horizontal lamellar structure. Moreover, the out-of-plane and in-plane scattering profiles extracted along the meridian and equatorial lines at $2\theta_{\rm f} = 0.117^\circ$ and $\alpha_{\rm f} = 0.202^\circ$, respectively, were satisfactorily fitted with the GIXS formula of the lamellar structure model prepared for this study (Figures 5d and e; see GIXS formula detail in Supporting Information and Figure S3). The lamellar orientation was evaluated from the azimuthal scattering profile (Figure 5f) extracted at $q = 0.430 \text{ nm}^{-1}$ from Figure 5c. The obtained structural parameters are summarized in Table 1, and they were utilized for the scattering image reconstruction (Figure 5b), which agrees well with the measured data. The horizontal lamellar structure (Figure 3d) has a long period $D_{\rm L}$ of 17.2 nm; sublayers l_1 , l_2 , and l_3 of 2.60 nm, 2.90 nm, and 8.80 nm, respectively; lamellar lattice *g*-factor of 0.04; and $O_{\rm s}$ of 0.974 ($\overline{\varphi} = 0^\circ$ and $\sigma_{\varphi} = 2.00^\circ$). For *tp-B*-BCP, the PTEGGE block had a slightly larger volume fraction than the PDGE block. Hence, the l_3 sublayer can be assigned by the cyclic PTEGGE block phase, whereas the sum of the l_1 and l_2 sublayers can be assigned to the linear PDGE block phase.

Interestingly, among all the topological copolymers used in this study, only *tp-B*-BCP exhibited a similar behavior to the aforementioned predictions⁴⁹⁻⁵¹ with additional features, despite possessing a tadpole topology. It achieved a 28.3%–32.5% *d*-spacing reduction compared to *l*-BCP in the same morphological type (lamellar), and showed a unidirectional horizontal orientation unlike the mixed orientation of *l*-BCP lamellae. Additionally, the lamellar ordering was significantly higher than *l*-BCP, as indicated by the statistical evaluation ($g_{tp-B-BCP} = 0.04$; $g_{l-BCP} = 0.33$, 0.35) and visual assessment (Figure 5a shows up to third-order peaks). The ordering of the *tp-B*-BCP lamellae was also higher than cylindrical structures of *tp-A*-BCP and *cy*-BCP (short-range order domain) but lower than that of the long-range order domain in the *cy*-BCP film.

Considering that the PDGE and PTEGGE blocks in all the copolymers possessed similar degrees of polymerization (~50) and volume fractions, the drastic differences in their morphological features among the linear, cyclic, and tadpole topologies were highly noteworthy. *tp-B*-BCP is the only copolymer to form the same nanostructure type (i.e. lamellar structure) as *l*-BCP. This suggests that in tp-B-BCP, which consists of linear PDGE block and cyclic PTEGGE block, the cyclized PTEGGE block does not significantly change its phase diagram when compared to *l*-BCP. In contrast, cy-BCP and tp-A-BCP form cylindrical structures, and both of their PDGE blocks are either fully cyclized or are a part of a macrocycle. Hence, the nonlinear topology of PDGE seemed to shift the phase diagram in comparison to *l*-BCP. The stark contrast in the sensitivity to the topological alteration between the PDGE and PTEGGE blocks may stem from their chemical structures. Although the polyether based backbone is shared by both blocks, PDGE and PTEGGE blocks differ in their side chain bristles. The *n*-decyl bristle in the PDGE blocks is a linear, saturated hydrocarbon with a sufficient length to exhibit a relatively greater stiffness (*i.e.* longer persistence length) than the flexible 2-(2-(2-methoxy)ethoxy)ethyl bristles of the PTEGGE blocks. Therefore, compared to the PTEGGE block, the PDGE block would experience a significant change in its chain conformation caused by the nonlinear topology, resulting in a significant shift in the phase diagram to yield cylindrical structures.

However, the cylindrical structures achieved by *cy*-BCP and *tp-A*-BCP are uniquely defined by their elliptical cylinders with elliptical axial ratios ε much greater than 1 (ε_{cy} -BCP = 1.42, 1.49; ε_{tp} - $_{A-BCP} = 1.64$). In addition, their L_z/L_y ratios $\gamma(\gamma_{cy}$ -BCP = 1.43, 1.50; $\gamma_{tp-A-BCP} = 1.62$) deviate significantly from $\sqrt{3}/2$, which is γ value for conventional cylindrical structures. These results indicate that both the shape and arrangement of the cylinders conform to non-conventional dimensions to retain the cylindrical structure with near-equivalent volume fractions between the cylinders and matrix. Overall, the cyclic topology and related complex variations (*i.e.* tadpole and bicyclic), in conjunction to block configuration, critically influence the nanostructure type.

GISAXS characterization suggested that the cyclic and tadpole topologies increase the structural ordering as evidenced by significantly reduced *g*-factors for all topological copolymers compared to *l*-BCP. In *cy*-BCP, it can be theorized that the cyclic topology drastically improves structural ordering by the following rationale. Cyclic topology reduces the degree of freedom to the copolymer, yielding a more compact chain conformation. Consequently, both intra and intermolecular interactions among the same side chains for both blocks increase to generate a significant enthalpic gain, which promotes the formation of a highly ordered nanostructure. It appears that the enthalpic gain is sufficient to overcome the entropic loss of forming the long-range order domain with an exceptionally low *g*-factor of 0.01. For *tp*-*A*-BCP and *tp*-*B*-BCP, it is also speculated that the compact chain conformation of cyclized block extends its positive enthalpic influence to the linear block to establish the nanostructure ordering that corresponds with *cy*-BCP. These insights are, however, solely based on GISAXS characterization results, and further theoretical and experimental investigations are required to evaluate the extent of thermodynamic impact of a cyclic topology in detail.

In summary, the nanoscale film morphologies of cyclic and tadpole block copolymers, cy-BCP, tp-A-BCP, and tp-B-BCP, and their linear counterpart l-BCP, were investigated in detail for the first time using synchrotron GIXS analysis. The quantitative GIXS analysis provided structural details of the nanoscale film morphologies. Furthermore, the analysis provided critical correlations between the film morphology and molecular topology. All topological block copolymers revealed phase-separated nanostructures in which the nanostructure type was found to be highly dependent on the molecular topology, *i.e.* cy-BCP and tp-A-BCP form distorted hexagonally packed cylindrical structures, whereas tp-B-BCP and l-BCP form lamellar structures. Furthermore, the magnitude of nanostructure domain spacing was significantly dependent on the topology, *i.e.* cy-BCP < tp-B-BCP

 \approx *tp-A*-BCP << *l*-BCP. Notably, *cy*-BCP achieved 49.3%–53.7% reduction in domain spacing than *l*-BCP. This reduction in domain spacing was greater than the theoretical predictions and previous experimental reports. *tp-A*-BCP and *tp-B*-BCP achieved domain spacing reductions of 25.0%–32.5% compared to *l*-BCP, which is comparable to both theoretical predictions and previous reports. Most importantly, *cy*-BCP, *tp-A*-BCP, and *tp-B*-BCP formed nanostructures with exceptionally high structural ordering and unidirectional horizontal orientation, unlike *l*-BCP, which exhibited poor ordering and low orientation control. Overall, this study demonstrated that both the cyclic and tadpole block copolymer approaches are powerful strategies for producing highly ordered, unidirectional, and miniaturized domain-based nanostructures.

Associated Content

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi Sample preparation, synchrotron grazing-incidence X-ray scattering, grazing-incidence X-ray scattering data analysis, Table S1-S3, Figures S1-S5.

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T.S. supervised the project. B.J.R., Y.S., and T.I. designed the experiments, solved the technical issues, and checked the experimental results. All authors contributed to developing the overall scope, interpreting the results and preparing the manuscript.

Notes

The authors declare no competing interests.

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Table of Contents graphic

Highly Ordered Nanoscale Film Morphologies of Block Copolymers Governed by Nonlinear Topologies

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