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THE
ELECTRICAL PROPERTIES
OF
GOLD AND TANTALUM
THIN FILMS
AFTER
ARGON ION IMPLANTATION

R. G. R. ROBINSON, B.Sc. DUNELM 1943

540905

Foreward by I.H. Wilson (Supervisor)

*This thesis has been assembled by myself from papers found at his home after the tragic death of Graham Robinson, known to us all as Gray. His experimental work was nearly complete and he had finished the chapters on experimental procedure and the results for his thesis. The other main document available to us was his transfer report and parts of this are reproduced in the appropriate chapters. The work falls into two parts - argon bombardment of gold films, and argon bombardment of tantalum films. The transfer report refers only to the former and so comments by myself (in italics) will be included where necessary. Extracts from other material will be included where appropriate. The main results of the work on gold films has already been published.*¹

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ABSTRACT

Evaporated gold and tantalum films have been bombarded with argon ions. In the case of gold films this resulted in an increase of the sheet-resistance by sputter etching to a maximum of $40 \text{ k}\Omega/\square$. The strain-gauge coefficient of resistance (γ) (i.e. the fractional change in resistance per unit strain) was measured for films with a wide range of sheet-resistance, and was found to be almost invariant with an average of 2.6. This contrasts greatly with the published values of γ of up to 100 for thin island-structure evaporated films of similar sheet-resistance.

The temperature coefficient of strain-gauge factor (β) was found to be similar in magnitude but opposite in sign to the temperature coefficient of resistivity (α), which was measured as $+12 \times 10^4 / ^\circ\text{C}$.

The measured values of γ , β and α agree well with values calculated assuming metallic conduction modified by reduction of the electron mean-free-path. We, therefore, conclude that a connected metallic layer still exists at very high values of sheet-resistance.

ABSTRACT (continued)

In the case of tantalum films (that contain 30 atomic per cent oxygen) conduction was found to be by a combination of metallic and activated tunnelling. In the latter case there is some evidence for an increase in importance of this mechanism with oxygen concentration and for the existence of at least two activation energies.

After bombardment with low doses of argon the resistivity (ρ) and also α shifted markedly towards values expected of very pure tantalum films, probably as a result of radiation enhanced diffusion and preferential sputtering of oxygen combined with re-arrangement of the film to form large precipitates of b.c.c. tantalum.

There was also a significant increase in γ (from an average of 3 up to 5.2) at similar doses, possibly as a result of changes in the microstructure increasing the importance of strain induced changes in the metallic conduction paths in the film.

The metallic phase appears to be metastable as ρ increases with time (up to two years) at room temperature, probably due to reaction with oxygen near the film/substrate interface.

For higher doses ρ drops (but not in a way explainable by sputter etching) α changes from a large positive value to a small negative one, and γ drops towards a value of 2 which is the predicted value for Ta. on glass.

ABSTRACT (continued)

These results indicate that a single phase, stable, low sputtering rate compound is formed, probably by reaction with the glass substrate. This compound has a mixture of activated tunnelling and metallic conduction with the strain gauge factor apparently determined by the metallic component of conduction.

Some preliminary attempts at electrical depth profiling of the bombarded tantalum films by anodization are reported and the results support the models proposed above.

1. Introduction

This work is part of a programme aimed at understanding the effects of ion bombardment on the electrical properties of thin metal films.

Measurement of strain-gauge coefficient of resistance, henceforth called strain-gauge factor, was undertaken because it provides information on conduction mechanisms that are dominant in the implanted thin films, and because work on active ion implantation of active metal films has indicated that this technique might be used to produce useful devices^{2,3}.

Argon bombardment of gold films was chosen for the first study because of the inert nature of ion and target, and because the sputtering characteristics are reasonably well understood. In addition evaporated gold films have shown a very large difference in strain-gauge factor (γ) (two orders of magnitude) between structures of the connected type (metallic conduction) and discontinuous island structures (tunnelling conduction).^{4,5}

The strain gauge factor is defined by; $\gamma \equiv (\delta R/R)/(\delta l/l)$ where R is the resistance and l is the length of the conductor.

Continuous 200 Å evaporated gold films on glass substrates have been bombarded with 40 keV argon ions to increase the sheet-resistance by several orders of magnitude, into the region where

thin evaporated films would be discontinuous. The values of γ and α (the temperature coefficient of resistance or T.C.R.) and also the temperature coefficient of the strain-gauge factor (β), were measured.

In the second study the same (inert) ion was used but a chemically reactive film, tantalum, was used in order to determine the effects on strain-gauge factor of structural changes observed by other workers at Surrey^{6, 7, 8} on similar films. A large amount of data was accumulated on the properties of the 'as deposited' films (resistivity, T.C.R., activation energy for conduction, strain-gauge factor and stability), as well as determination of the effects of argon bombardment on these properties.

A first attempt was made at determining electrical properties as a function of depth by anodic oxide profiling. Although the latter was not completely successful it provided the ground-work that lead to a very successful technique being developed by other workers at Surrey.⁸

2. Theory

2.1 Introduction

This section is a compilation of the relationships devised by Gray. These are used in the results section for comparison with experimental results.

2.2 The effect of sputtering on the conductivity of thin films

Consider 1 cm³ of pure gold. Its mass is 19.3 g, and its relative mass is $\frac{19.3}{197}$ mole.

The number of atoms contained in 1 cm³, is, by Avogadro's Number,

$$\frac{19.3}{197} \times 6.025 \times 10^{23}, = 59 \times 10^{21} \text{ atoms.}$$

Thus the number of gold atoms per unit atomic area in a 1 cm length is, on average,

$$\sqrt[3]{59 \times 10^{21}} = 3.893 \times 10^7 ;$$

per unit atomic area and the average number in the thickness of a 240 Å film is

$$240 \times 10^{-8} \times 3.893 \times 10^7 = 93.5 \text{ atoms.}$$

The mean linear distance between atoms is 2.57 Å .

2.2.1 Ion Bombardment

A monatomic layer of gold 1 cm² in area would contain,

$$(3.893 \times 10^7)^2 = 15.15 \times 10^{14} \text{ atoms.}$$

Let the number of conducting atoms affected by the ion bombardment be S atoms/ion. [Note: this is similar to the sputter rate which, for gold, is 10 atoms/ion with argon ions at 40 keV.]

The number of ions required to remove, by sputtering, the number of atoms equivalent to a monatomic layer of gold is, therefore,

$$\frac{15.15 \times 10^{14}}{S} \text{ ions.cm}^{\frac{1}{2}} ;$$

and this would involve an electric charge of

$$\frac{15.15 \times 10^{14}}{S} \times 1.60 \times 10^{-19} \text{ coulomb/cm}^2$$

$$= \frac{242}{S} \mu\text{C.cm}^{-2} . \quad \text{--- 2.1}$$

The charge to remove atoms equivalent to a film of mean thickness $240\text{-}\text{\AA}$ would be

$$\frac{22620}{S} \mu\text{C}\cdot\text{cm}^{-2}$$

2.2.2 Electrical Resistance

The resistivity of gold in bulk is

$$= 2.285 \times 10^{-6} \Omega \text{ cm, (at } 30^{\circ}\text{C),}$$

and the resistance of a slab, of thickness t , length l and breadth b is

$$R = \rho \frac{l}{bt} \Omega$$

When the slab is of square format $l = b$, and the resistance

$$R = \frac{\rho}{t} \Omega/\square.$$

The Conductance of such a slab, of thickness 240 \AA , is

$$\begin{aligned} G &= \frac{1}{R} = \frac{240 \times 10^{-8}}{2.285 \times 10^{-6}} \\ &= 1.05 \text{ siemen/square (at } 30^{\circ}\text{C).} \end{aligned}$$

Under argon-ion bombardment the film thickness (and hence the film conductance) will be reduced.

Consider such a film after ion bombardment with charge C coulomb. The mean film thickness remaining will be

$$\frac{t \left(\frac{22620}{S} - C \right)}{\frac{22620}{S}} = \frac{t (22620 - SC)}{22620} \text{ cm.}$$

The conductance would be

$$G = \frac{t (22620 - SC)}{22620 \rho_f} \text{ S/cm} \text{ ----- 2.2}$$

which is shown in figure 4.3 for three values of S.

The resistance of the film would be

$$R = \frac{22620 \rho_f}{t (22620 - SC)} \text{ } \Omega/\square \text{ ----- 2.3}$$

2.2.3 Tantalum films (unknown density)

$$\text{Atoms/cm}^3 = \frac{D}{(Wt)} \times N_o \quad \text{Atoms/cm} = \sqrt[3]{\frac{D N_o}{(Wt)}}$$

$$\text{linear spacing} = \sqrt[3]{\frac{Wt}{D \times N_o}} \quad \text{cm}$$

$$\text{Atoms/cm}^2 \text{ monolayer} = \left[\frac{D N_o}{(Wt)} \right]^{2/3}$$

$$\text{Ions needed to remove a monolayer} = \frac{I}{S} \left[\frac{D N_o}{(Wt)} \right]^{2/3}$$

Ions needed to remove thickness t

$$= \frac{I}{S} \left[\frac{D N_o}{(Wt)} \right]^{2/3} \times \frac{t}{\left[\frac{Wt}{D N_o} \right]^{1/3}}$$

$$= \frac{I}{S} \left[\frac{D N_o}{(Wt)} \right] \times t \quad \text{Ions/cm}^2$$

$$N = \frac{D}{S} \times \left[\frac{N_o}{AtWt} \right] \times t$$

$$N = \frac{D}{S} \times \frac{6.025 \times 10^{23}}{180.95} \times 750 \times 10^{-8} \text{ Ions/cm}^2$$

$$= \frac{D}{S} \times 0.250 \times 10^{17} \text{ Ions/cm}^2$$

By experiment this = 3×10^{17} Ions/cm²
determined by extrapolation of the resistance versus dose curve

$$\therefore 0.25 \frac{D}{S} = 3 \quad \therefore \frac{D}{S} = \underline{12}$$

$$N = \frac{D}{S} \times 0.25 \times 10^{17} = 3 \times 10^{17} \text{ Ions/cm}^2$$

Film sheet resistance during bombardment

$$R_S = \frac{\rho}{t} \times \frac{N}{(N - n)} \quad \text{----- 2.4}$$

2.3 Strain Sensitivity γ .

Under tensile elastic deformation the change in resistance of a thin film is given by

$$\frac{dR}{R} = \frac{d\ell}{\ell} = \frac{db}{b} - \frac{dt}{t} + \frac{d\rho}{\rho}$$

Poisson's ratio for unconstrained dimensions would give

$$db = dt = -\sigma \frac{d\ell}{\ell}, \text{ and}$$

Parker and Krinsky⁴ have shown that this leads to a Strain-Gauge Coefficient of Resistance of

$$\gamma = \frac{dR/R}{d\ell/\ell} = 1 + 2\sigma + \frac{d\rho/\rho}{d\ell/\ell} \quad \text{----- 2.5}$$

The dimensions (apart from the thickness) of thin films, however, may be assumed to be the same as those of the substrate, so that the width b is constrained by the contraction experienced by the substrate; that is

$$\frac{db}{b} = -\sigma_s \frac{d\ell}{\ell},$$

a circumstance which gives rise to a differential lateral tensile strength of amount

$$\sim (\sigma_f - \sigma_s) \frac{d\ell}{\ell};$$

which, in turn, causes a further lateral strain in the direction of the film thickness and a stress reaction in the direction of the applied strain.

Consider the three principal strains in an isotropic medium (the film);

$$\epsilon_1 = \frac{P_1}{E} - \sigma_f \frac{(P_2 + P_3)}{E} = \frac{P_1 - \sigma_f (P_2 + P_3)}{E};$$

$$\epsilon_2 = \frac{P_2}{E} - \sigma_f \frac{(P_1 + P_3)}{E} = \frac{P_2 - \sigma_f (P_1 + P_3)}{E};$$

$$\epsilon_3 = \frac{P_3}{E} - \sigma_f \frac{(P_1 + P_2)}{E};$$

where ϵ_1 = strain in the direction of the length ℓ and the applied stress p_1 ,

ϵ_2 = strain in the direction of the width b and the differential lateral stress p_2 ,

ϵ_3 = strain in the direction of the thickness t
and the lateral stress p_3 , ($=0$),

E = Young's modulus for the film material,

E_s = Young's modulus for the substrate,

P_s = the stress applied to the substrate by the
bending rig.

By combining the above three equations and applying the known end-conditions, ϵ_3 may be obtained in terms of ϵ_1 and Poisson's ratios for film and substrate,

$$\epsilon_3 = \frac{\sigma_f (1 - \sigma_s)}{(1 - \sigma_f)} \epsilon_1 ;$$

or
$$\frac{dt}{t} = - \sigma_f \frac{(1 - \sigma_s)}{(1 - \sigma_f)} \frac{d\ell}{\ell} .$$

This leads to values for the strain-gauge coefficients of resistance.

$$\gamma = \frac{dR/R}{d\ell/\ell} = 1 + \sigma_s + \sigma_f \frac{(1 - \sigma_s)}{(1 - \sigma_f)} + \frac{dp/p}{d\ell/\ell} \quad \text{--- 2.6}$$

This is similar to the result of Verma¹⁰

2.3.1 Longitudinal Gauge Factor γ_b

Change of resistance is measured in the direction ($\pm\ell$) of the applied strain.

$$\text{The resistance } R_\ell = \rho \frac{\ell}{bt} ,$$

so that
$$\frac{\partial R}{R\ell} = \frac{\partial \rho}{\rho} - \frac{\partial t}{t} + \frac{\partial \ell}{\ell} - \frac{\partial b}{b} .$$

Using the relationship derived above, the longitudinal strain-gauge coefficient of resistance is;

$$\gamma_l = \frac{\partial R/R_l}{\partial l/l} = (1 + \sigma_s) + \sigma_f \frac{(1 - \sigma_s)}{(1 - \sigma_f)} + \frac{\partial \rho/\rho}{\partial l/l}$$

- - - - - 2.7

2.3.2 Transverse Gauge Factor γ_b

In this case change of resistance is measured in the direction ($\pm b$) at right-angles to the direction ($\pm l$) of the applied strain.

$$\text{The resistance } R_b = \rho \frac{b}{l t},$$

so that

$$\frac{\partial R}{R_b} = \frac{\partial \rho}{\rho} - \frac{\partial t}{t} - \frac{\partial l}{l} + \frac{\partial b}{b}.$$

again, using the same relationships, the transverse strain-gauge coefficient of resistance is formulated as,

$$\gamma_b = \frac{\partial R/R_b}{\partial l/l} = -(1 + \sigma_s) + \sigma_f \frac{(1 - \sigma_s)}{(1 - \sigma_f)} + \frac{\partial \rho/\rho}{\partial l/l}$$

- - - - - 2.8

It is assumed that the films are isotropic, and that,

$$\rho_b = \rho_l = \rho.$$

2.4 The temperature coefficient of strain-gauge factor β

The temperature coefficient of resistance (T.C.R.) is defined as

$$\alpha = \frac{1}{R} \frac{\partial R}{\partial T}.$$

Writing the temperature coefficient of strain-gauge factor

as

$$\beta = \frac{1}{\gamma} \frac{\partial \gamma}{\partial T} ;$$

we have

$$\beta = \frac{1}{\frac{\partial R/R}{\partial \ell/\ell}} \frac{\partial}{\partial T} \left(\frac{\partial R/R}{\partial \ell/\ell} \right) ;$$

and

$$\beta = \frac{1}{\partial R/\partial \ell} \frac{\partial^2 R}{\partial \ell \partial T} + \frac{1}{\ell} \frac{\partial \ell}{\partial T} - \frac{1}{R} \frac{\partial R}{\partial T} .$$

The first term is a temperature coefficient of the rate of change of resistance with length - which should be invariant: the second term is the coefficient of linear expansion, which may be neglected as being two orders of magnitude below the third term: the third term is the T.C.R. This result has been obtained by Witt and Coutts¹¹,

$$\beta = - \alpha \quad \text{-----} \quad 2.9$$

2.5 Theoretical value of the temperature coefficient of resistance α

We will derive a value of α assuming metallic conduction in the film that is limited by the reduction in electron mean free path caused by the proximity of the film surfaces.

It has been suggested¹² that the average free path length is given by

$$\bar{\lambda} = \frac{3t}{4} + \frac{t}{2} \ln \left(\frac{\lambda_B}{t} \right)$$

where t is the film thickness and λ_B is the bulk mean free path.

The relative conductivity is then given by

$$\frac{\sigma}{\sigma_B} = \frac{\bar{\lambda}}{\lambda_B} = \frac{3t}{4\lambda_B} + \frac{t}{2\lambda_B} \ln \left(\frac{\lambda_B}{t} \right)$$

where σ_B is the bulk conductivity. This expression has to be modified to account for diffuse scattering at boundaries^{13,14}, giving the relation

$$\frac{\sigma}{\sigma_B} = \frac{3t}{4\lambda_B} (1 + \frac{2}{p}) \left\{ \ln \left[\frac{\lambda_B}{t} \right] + 0.4228 \right\} \quad \text{--- 2.10}$$

where p is the fraction of incident electrons specularly reflected and $p^2 \ll 1$. Using an extension of Matthiessen's rule, the resistivity of the film ρ_f is the sum of components due to lattice scattering (ρ_L), scattering by defects and impurities (ρ_D) and surface scattering (ρ_s). The bulk resistivity ρ_B is just composed of the first two components since ρ_s is assumed to be negligible.

The T.C.R. is defined as

$$\alpha = \frac{1}{R} \frac{dR}{dT} = \frac{1}{\rho} \frac{d\rho}{dT}$$

As a first approximation one can say that ρ_D and ρ_s do not depend strongly on temperature so that for a continuous, homogeneous film;

$$\alpha_f = \frac{1}{\rho_f} \frac{d\rho_f}{dT} = \frac{1}{\rho_f} \frac{d\rho_L}{dT}$$

and

$$\alpha_B = \frac{1}{\rho_B} \frac{d\rho_B}{dT} = \frac{1}{\rho_B} \frac{d\rho_L}{dT}$$

Then

$$\frac{\alpha_f}{\alpha_B} = \frac{\rho_B}{\rho_f} = \frac{\sigma}{\sigma_B}$$

Using Heavens'¹² approximate solution for eqn. 2.10 for $t \ll \lambda_B$,

$$\alpha_f = \alpha_B \left\{ \ln \left[\frac{\lambda_B}{t} \right] + 0.4228 \right\}^{-1}$$

Now the sheet resistivity $R_s = \rho_f/t$ so that

$$\alpha_f = \alpha_B \left\{ \ln \left[\frac{\lambda_B R_s}{\rho_f} \right] + 0.4228 \right\}^{-1} \quad \text{----- 2.11}$$

This model includes a large number of assumptions, and it may be more realistic to take into account scattering at grain boundaries.

2.6 Statistical Methods

The measurements made in the course of this strain-gauge evaluation include unpredictable errors, and although an attempt was made to minimise their occurrence the graphs show their presence only too clearly.

It is assumed that these errors are random in occurrence and magnitude, and that they have a Normal probability distribution. It follows, therefore, that Standard Deviations may be calculated for suitable quantities such that the probability of the quantity lying within the range

- mean value ± 1 S.D. is 0.68,
- or mean value ± 1.96 S.D. is 0.95,
- or mean value ± 2.58 S.D. is 0.99.

It is usually satisfactory to take the range ± 1.96 S.D. as defining the 95% confidence limits.

3. Experimental Techniques

3.1 Thin Films

Films were vacuum evaporated onto soda-glass substrates approximately 1 mm thick. High temperature annealing was not employed. It was therefore hoped that no problems would be encountered from other metal ions from the substrate. diffusion of sodium and / For gold films 3" x 1" glass was used whilst for tantalum both 1" x 1" squares and 3" x 0.2" strips were used. A few tantalum films were laid onto mica and some onto alumina substrate: these were used for sheet resistance and TCR measurements but no significant difference was noted as a result of their use.

3.1.1 Substrate Preparation

The substrate was cleaned by gentle abrasion in hot detergent solution. By rinsing in hot water and hot distilled water, and with a final rinse in de-ionised water. Drying was carried out on a lamina-flow bench, and only those substrates that dried without cloudiness to give clear specula reflections were used for film deposition.

3.1.2 Gold Films

Gold was evaporated from a tungsten wire spiral electrically heated in a vacuum of between 1×10^{-6} and 2×10^{-6} torr. The substrate temperature was that of the equipment ambient.

Films were laid down having a thickness within the range 200 Å to 300 Å. This was measured after evaporation by Talystep Indicator. But thickness was monitored during some evaporation runs using a quartz crystal thickness monitor oscillating at about 250 KHz, whilst for other runs the resistance of a test film was monitored, and evaporation was stopped when this had reached a resistance of 10 Ω.

3.1.3 Tantalum Films

Tantalum was evaporated by electron-beam heating in an ultra-high vacuum system pumped by an oil diffusion pump and fitted with a

liquid nitrogen cold trap. A titanium sublimation pump was available, enabling a pressure of 5×10^{-9} torr to be reached at the start of each evaporation run. During evaporation the pressure (and temperature) rose and after about five minutes had reached an indicated level of about 10^{-6} , at which point the run was usually interrupted for cooling.

Film thickness measurements made after unloading the evaporator indicated consistently a rate of deposition of tantalum of 50 \AA per minute, and this rate was used to estimate film thickness during the subsequent evaporation runs.

3.1.4 Electrode Pads

Each film had gold electrodes formed at either end to define the film length and to enable electrical connections to be made. The gold pads were evaporated in vacuo through masks, after a preliminary flash evaporation of a titanium "glue" layer. The resulting electrodes were robust enough to allow connecting leads to be soft-soldered into place.

The masks used for the gold and tantalum films are shown in Figures 3.1 and 3.2 respectively.

3.1.5 Ion Bombardment

The ion bombardment was carried out in a 500 keV heavy ion accelerator with sector-magnet mass-analysis and closed loop voltage control using the mass defining slits. Uniformity of dose was achieved by electrostatic scanning of the beam on a raster equivalent to double the area of the beam defining aperture. Uniformity is estimated at better than 1%, beam convergence is estimated to be less than 0.2° . The ion dose was monitored by measuring the charge collected by the target. An electrostatic field was used to suppress secondary electrons. The target chamber was pumped by liquid nitrogen trapped diffusion pump with a cold finger (liquid nitrogen) mounted close to the target. Typical operating pressure was 5×10^{-7} torr.

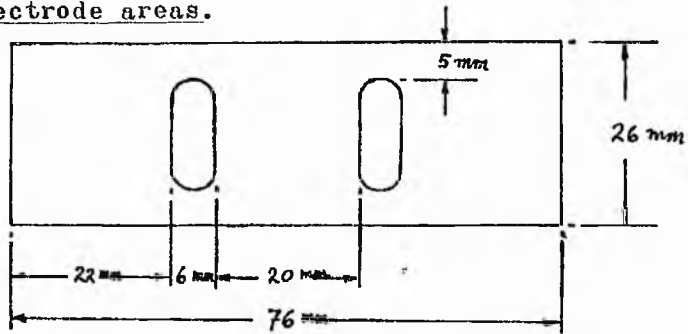
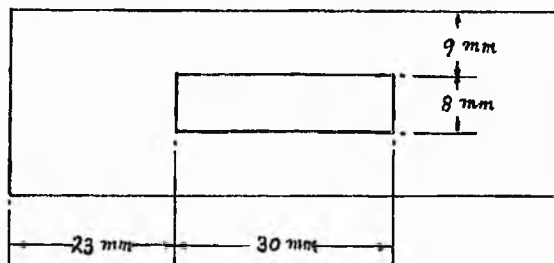
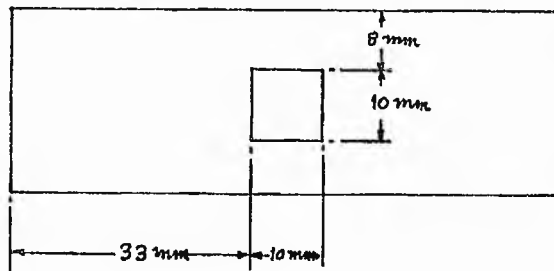
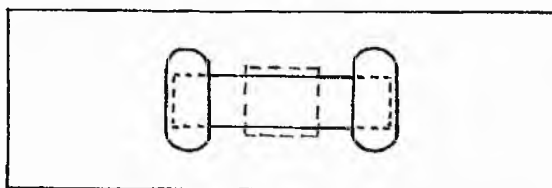
3.2 Measurements

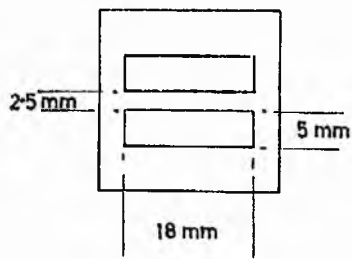
All electrical measurements were conducted with the specimens at atmospheric pressure.

The effect of the unimplanted ends on the resistivity, TCR and strain gauge factor of implanted gauges was corrected for by using formulae derived by M. R. Moulding (transfer report 1977) and reproduced here as appendix 3.

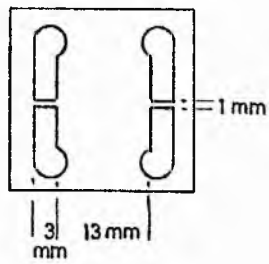
3.2.1 Resistance Measurement

An equal-arm Wheatstone Bridge circuit was used for measuring film resistance, the DC supply being 2 V for most films so as to limit the power dissipation to 2 mW/cm^2 and so avoid thermal errors. For high resistance films the bridge ratio arms were unequal and the supply could be increased to 10 V.

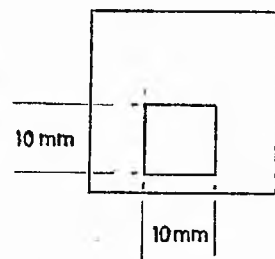
Mask for electrode areas.Mask for strain-gauge film.Ion bombardment mask.Specimen strain-gauge.



Tantalum films



Gold electrodes



Bombardment window

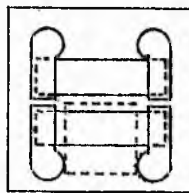
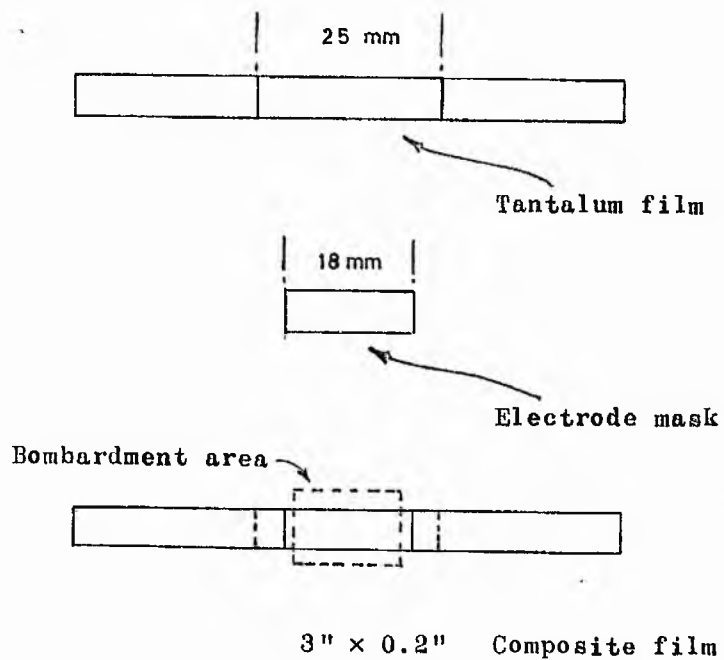
1" x 1"
Composite film

FIG. 3.2

MASKS FOR TANTALUM FILMS

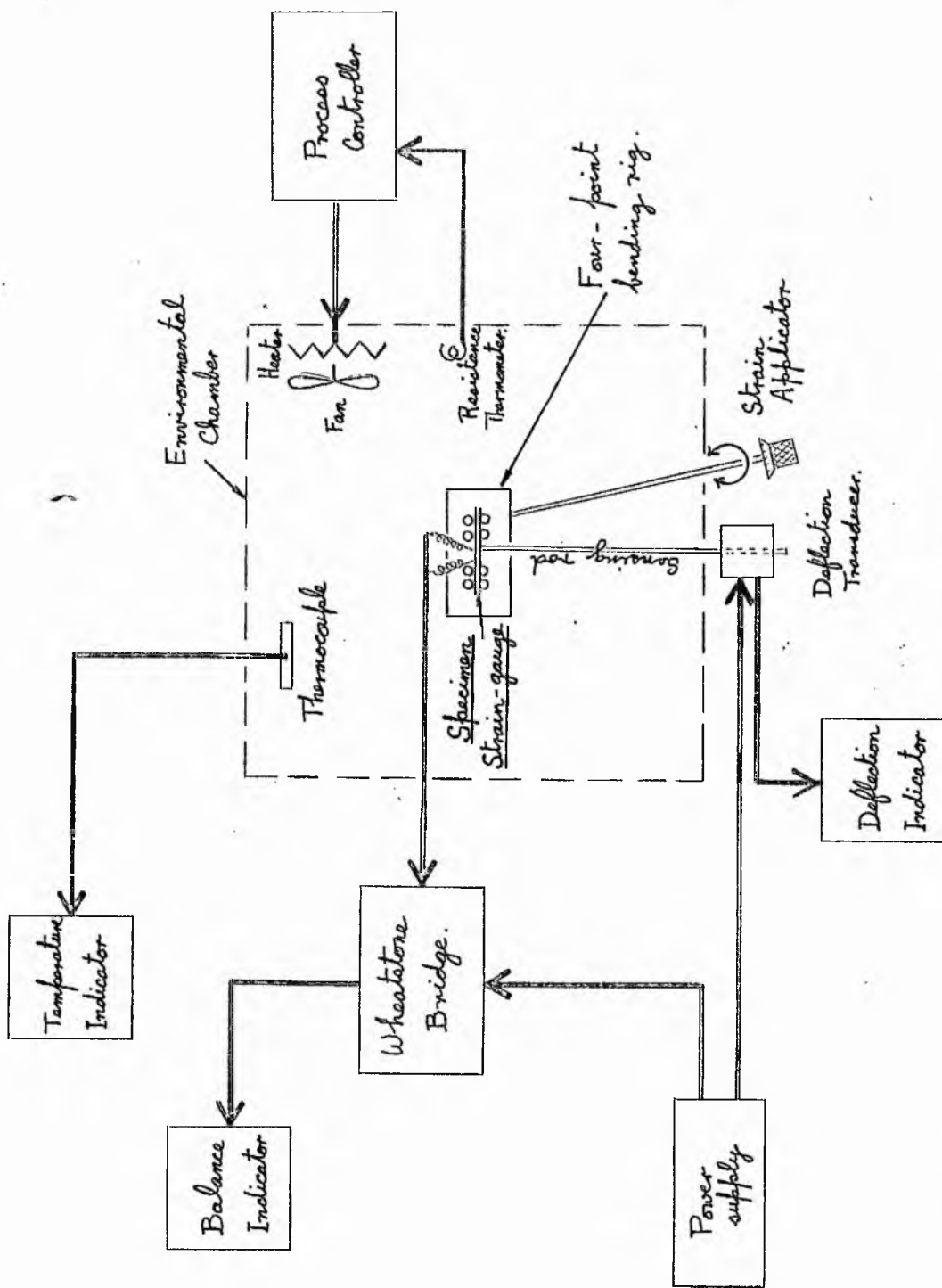
For the measurement of the gold films a centre-zero micro-ammeter was used as balance detector ($\pm 100 \mu\text{A}$ range), and a series of spot measurements was made for each run. In the case of the tantalum films an x-y plotter was available and this allowed a continuous record to be made of the measurements.

Before use the films were annealed at room temperature for a minimum time of one week, and also at 120°C for two hours.

3.2.2 Temperature Control and Measurement

In order to avoid errors due to temperature variation during the measurement of strain-gauge factor the temperature must be held constant. The use of glass as the substrate limits the maximum safe strain to about 250 PPM and the change of resistance this gives is, in the case of gold films, the equivalent of a temperature change of less than one half degree celsius. Thus, for there to be no more than a 5% error in strain-gauge factor from this cause the temperature must not vary by more than 0.02°C during straining. This is difficult to achieve over protracted periods but seems possible for the duration of one strain run.

The four-point straining rig was positioned inside a fan-blown environmental chamber in such a way that strain could be applied, and measured, from the outside. The schematic arrangement of this equipment is shown in Figure 3.3. Temperature could be tightly controlled by a closed-loop servo over the range 20°C to 120°C using a filamentary air-heating element and a filamentary tungsten resistor as the sensor. The temperature was measured with an iron-constantan thermocouple soldered to a copper vane of 2 sq.cm. placed in the air stream of the chamber, and with the cold junction kept in melting ice. The thermal



SCHEMATIC ARRANGEMENT OF THE EQUIPMENT

FIG. 3.3

EMF was read on a moving coil indicator. Temperature control was possible, but to a reduced accuracy, over the extended range -20°C to $+150^{\circ}\text{C}$ with the aid of auxiliary cooling or heating.

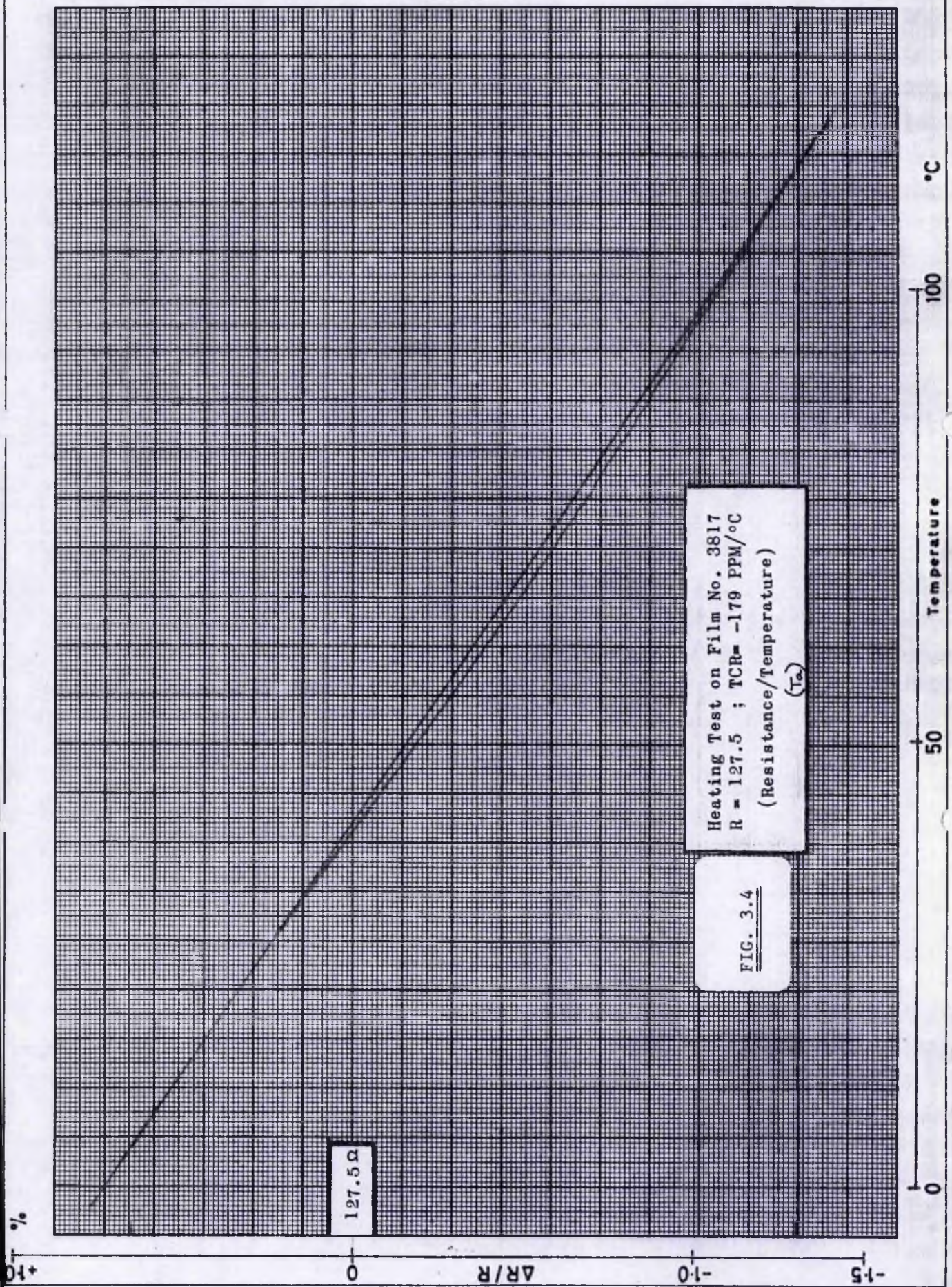
The environmental chamber was also used for making TCR measurements.

3.2.3 Temperature Coefficient of Resistance

In the case of the gold films precise measurements of resistance were made at about twelve spot temperatures between room temperature and 120°C both for rising and for falling temperature. The results were corrected for end effects in the case of bombarded films by a simple computer routine, and were displayed as a graph plot for a visual check on high temperature stability: a simple line fitting procedure was used to obtain the TCR and the sheet resistance at 30°C .

Most of the tantalum films were measured with the aid of an x-y plotter, vertical deflection being the out-of-balance output of the Wheatstone Bridge and horizontal deflection the thermocouple EMF. The procedure adopted here was to heat the chamber to about 120°C , to start the x-y plot, and to allow the chamber to cool slowly over a period of about an hour. Half way through the period water cooling was started so as to maintain a more or less constant cooling rate down to room temperature. The resistance scale was calibrated each time by altering the bridge balance point and marking the change on the plot, and the temperature scale was calibrated once and for all with the thermojunction in steam. A best-fit straight line was drawn through the plot, and this was subsequently corrected for end effects to give the TCR and sheet resistance at 30°C . Figure 3.4 shows a typical TCR run for a tantalum film.

A few tantalum films were measured over the extended temperature range -190°C to $+110^{\circ}\text{C}$. These films were mounted within an aluminium



block narrow enough to pass through the neck of an ordinary picnic dewar flask and also through an aperture into the environmental chamber; see Figure 3.5. The flask was half filled with liquid nitrogen and the block was carefully lowered into it so as to cool the films to -195° C. The block was then raised so that the temperature would rise slowly, to reach 0° C after about four hours. Temperature was again measured by thermocouple and moving coil indicator, and resistance by Wheatstone Bridge. Spot measurements were made at approximately 10° C intervals, and when the temperature had reached about $+5^{\circ}$ C the block was transferred to the environmental chamber, and heating was applied in steps to $+110^{\circ}$ C. The results were corrected for end effect and were plotted (R v.T) and ($\log R$ v. $1/T$). Typical outputs of data for these plots are shown in Table 3.1 and plotted in Figure 3.6.

3.2.4 Strain-Gauge Factor

The substrates were loaded into the four-point bending rig positioned inside the environmental chamber, see Figure 3.7. Strain was calculated from a measurement of substrate deflection, made mid-way between the two fixed posts of the bender. Using a differential transformer transducer. With the maximum safe strain for 1 mm thick glass substrate applied by the rig ($\pm 250 \times 10^{-6}$) the deflection obtained was 0.1 mm and the transducer output was ± 118 millivolt.

For gold films the transducer output was measured on a centre-zero millivoltmeter having a backing-off facility. A number of spot readings of resistance and deflection were taken as the strain was applied; then the results were corrected for end effects, were displayed in graphical form, and a straight line established through them to give the strain-gauge factor. All these calculations and the plots were

FIG. 3.5

and

FIG. 3.7

missing.

TABLE 3.1 Argon-Ion Bombardment of Tantalum Films

		NO. 1203. (Before bombardment)			
'A'	20.500	36.200	51.300	65.5CC	81.000
	124.568	124.423	124.191	124.059	123.896
'B'	20.500	36.200	51.300	65.500	81.000
	124.094	123.893	123.723	123.545	123.382
FILM LENGTH =	13.0mm		BREADTH =	5.0mm	IMPLANTED LENGTH =
					0.0mm
FILM THICKNESS = $8CC \cdot \lambda$					
ICN COSE = $C \cdot \mu C \cdot cm^{-1}$					
RES AT 0 DEG C = 0.1247E 03					
S.E. OF RESIST = 0.8328E-02					
RES AT 30 DEG C = 0.1243E 03					
S.E. OF RESIST = 0.4671E-02					
HORIZ SCALE = 0.2000E-03					
COORDS OF AXES = C.2414E-02 C.4810E 01					
HORIZ SCALE = 0.2000E-03					
COORDS OF AXES = C.2000E-02					
COORDS OF AXES = C.2414E-02 C.4810E 01					
RES AT 0 DEG C = 0.1241E 03					
S.E. OF RESIST = 0.3729E-02					
RES AT 30 DEG C = 0.4762E 02 Ω/p					
ICN COSE = $67CC \cdot \mu C \cdot cm^{-1}$					
(42 $\cdot 10^{16}$ Ions $\cdot cm^{-2}$)					
RES AT 0 DEG C = 0.1241E 03					
S.E. OF RESIST = 0.3729E-02					
RES AT 30 DEG C = 0.4762E 02 Ω/p					
ICN COSE = $67CC \cdot \mu C \cdot cm^{-1}$					
(42 $\cdot 10^{16}$ Ions $\cdot cm^{-2}$)					

		NO. 1203. (After bombardment)			
'A'	14.200	30.500	44.500	59.200	74.000
	47.800	32.000	123.652	123.482	123.319
	124.034	123.836	123.612	123.809	123.809
	14.200	30.500	44.500	59.200	74.000
'B'	47.800	32.000	117.142	117.124	117.110
	117.187	117.178	117.142	117.124	117.110
	117.160	117.193			
FILM LENGTH =	13.0mm		BREADTH =	5.0mm	IMPLANTED LENGTH =
					10.0mm
RES AT 0 DEG C = 0.4766E C2 Ω/p					
ICN COSE = $-0.8944E-C4$					
RES AT 30 DEG C = 0.4766E C2 Ω/p					
ICN COSE = $-0.9289E-C4$					
ACTIVATION ENERGY = C.873CE-C3 eV					
ACTIVATION ENERGY = C.9049E-C3 eV					
RES AT 0 DEG C = 0.4762E 02 Ω/p					
ICN COSE = $8CC \cdot \lambda$					
$67CC \cdot \mu C \cdot cm^{-1}$					
(42 $\cdot 10^{16}$ Ions $\cdot cm^{-2}$)					
RES AT 30 DEG C = 0.4762E 02 Ω/p					
ICN COSE = $-C.9375E-C4$					
RES AT 0 DEG C = 0.8855E 02					
S.E. OF RESIST = 0.3455E-02					
RES AT 30 DEG C = 0.4429E 02 Ω/p					
ICN COSE = $C.1764E-C4$ (IMPLANTED)					
ACTIVATION ENERGY = C.8855E-C3 eV					
ACTIVATION ENERGY = $[-0.1754E-C3$ (IMPLANTED)]					

ARRHENIUS PLOT

$\times 10^3$ ARRH 1201

1.57672

1.57478

1.57351

1.57257

1.57126
OHMS

1.56956
LOG (RESISTANCE)

1.56919

1.56787

1.56694

1.56623

1.56500

1.56467

1.56300

$\cdot 10^2$

1.18155

1.17556
OHMS

1.17760
LOG (RESISTANCE)

1.17540

1.17415

1.17168

NON-BOMBARDED FILMS, 850 Å
SHEET RESISTANCE AT 30°C
A-FILM = 45.4 Ω/□; B-FILM = 45.5 Ω/□
TCR = -105 PPM; -132 PPM
E = .0010 eV; .0013 eV

B-FILM AFTER ION BOMBARDMENT
ION DOSE = 2.00×10^{15} IONS · CM⁻² A⁻¹
SHEET RESISTANCE = 790 Ω/□ AT 30°C
TCR = -91 PPM; E = 0.0008 eV

1.05500
0.88800
0.74000
0.58400
0.45000
0.29200
0.15100
 $\times 10^2$

RECIPROCAL TEMPERATURE (T⁻¹) (marked °C)

1.04000
0.90000
0.80000
0.75000
0.68200
0.58100
0.44800
0.30100
0.20100
0.09300
 $\times 10^2$

RECIPROCAL TEMPERATURE (T⁻¹) (marked °C)

FIG. 3.6

carried out using a digital computer. Each strain run was completed within a time span of about two minutes so as to minimise temperature errors (see Section 3.2.2), and the results from four runs in tension and four in compression were averaged to obtain the final strain-gauge factor. A typical computer plot for a gold strain-gauge is shown in Figure 3.8.

For the tantalum films the x-y plotter was available, and this was used to display Wheatstone Bridge out-of-balance vertically and transducer output horizontally. A typical strain run for a tantalum film is shown in Figure 3.9., where the film was strained from tension to compression and back again as a check on temperature drift. The resistance scale was established by altering the bridge balance point, the slope of the best straight line drawn through the plot being used to calculate the strain-gauge factor after end corrections had been applied. The correction for the unbombarded section of bombarded gauges necessary in order to determine the true value of χ was done using formulae similar to those derived by M. R. Moulding (transfer report 1977) Appendix .

3.3 Anodising

Some of the tantalum films were examined in layers by progressive anodic oxidation. A strong solution of acetic acid was used as the electrolyte (50% glacial acetic acid in water), and a platinum foil of 2.5 sq. cm. was the counter electrode. A difficulty encountered in this work is that of insulating the submerged electrode and defining the anodised portion of the film. This was overcome here by the use of a generous layer of apiezon wax applied over the electrode and its lead so as partially to cover the tantalum film and to leave the required window exposed.

The anodic potential was increased in steps of approximately 2 V from a source current limited to 25 μ A. At each step the film was withdrawn from the electrolyte, washed in several changes of distilled water, dried, and placed in the environmental chamber for the resistance and TCR

505.800

43 2

TENSION

505.750

4321

505.700

4 2 1

505.650

423 1

505.600

432 1

505.550

4 32

505.500

-0.05

0.00

0.05

505.450

v n 13

COMPRESSION

505.420

433

505.400

432

505.390

432

505.350

324

505.320

43

505.300

32

505.280

431

505.250

-0.05

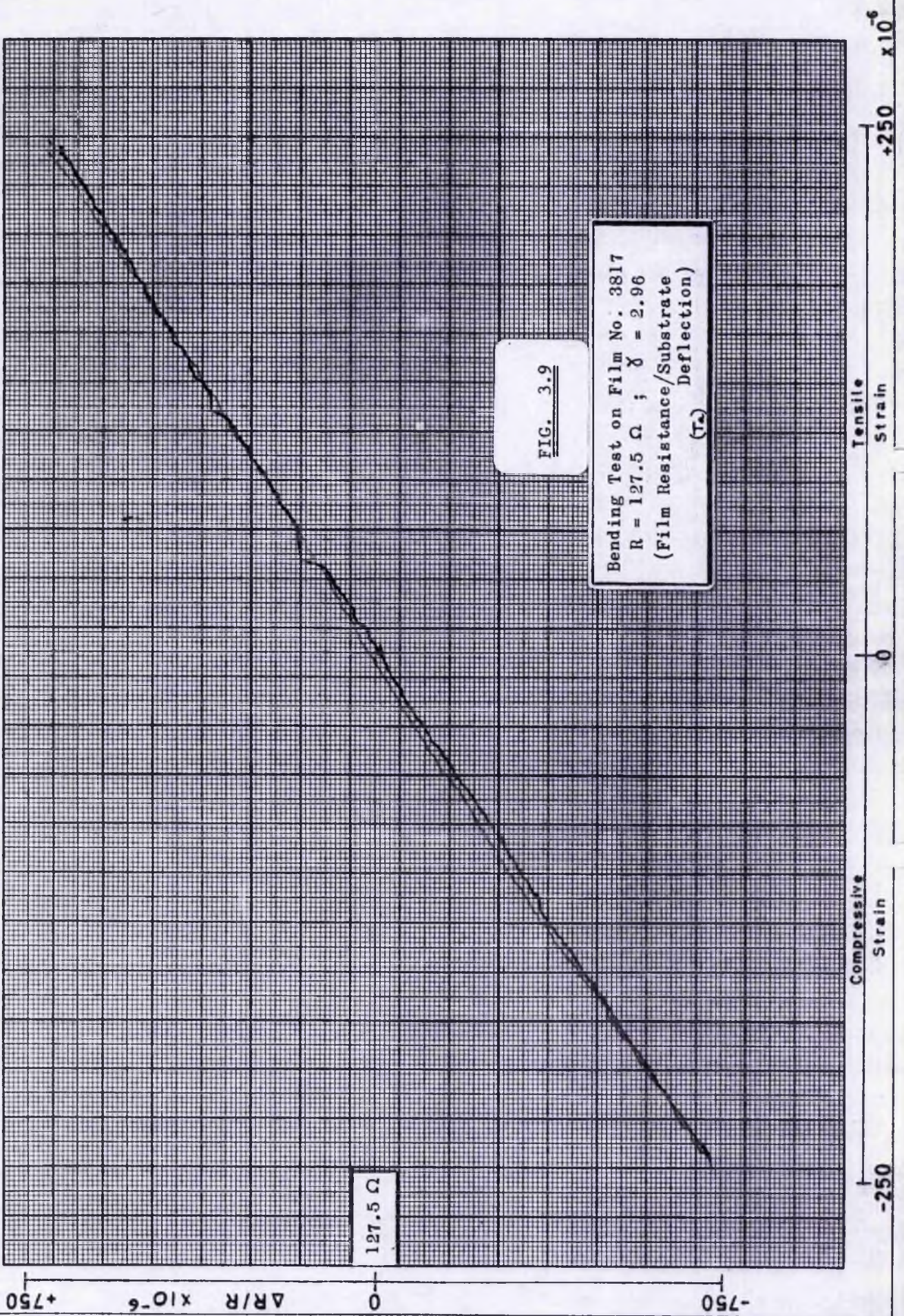
0.00

0.05

FIG. 3.8

Strain measurement of an Argon-bombarded Gold Film.
 (Electrical Resistance/
 Substrate Deflection)

AG 20-VII-77



to be measured. Anodising was continued at increasing potential until the film became open circuit. The results of a typical run are listed in Table 3.2.

3.4. Vacuum Annealing

A number of tantalum films were annealed isothermally in vacuo at temperatures up to 300°C.

The films were loaded onto a sliding platform. This was positioned at the cool end of a metals research vacuum furnace and kept there until the pressure was reduced below 10^{-4} Torr. The platform was then pushed into the hot region of the furnace for one hour, by which time the pressure was usually better than 5×10^{-5} Torr. The platform was then withdrawn to cool for half an hour, and then unloaded.

At each step in the annealing, the sheet resistance of the films was measured. This was done using a four-point probe because of the impracticability of using soft-soldered joints. The probe spacing was 1.0 mm and a current of 1 mA was used for measuring the low resistance films and 0.1 mA for those with high resistance.

The temperature was increased in steps of 50°C from 150°C to 300°C.

TABLE 3.2 Anodisation of 800-Å Tantalum Film

Plain Film		Argon-Ion Bombarded					
VOLTAGE	J	CONDUCTANCE (MHO)	TCR (PPM)	VCLTAGE	L	CCONDUCTANCE (MHO)	TCR (PPM)
1203 A	4	0.2112E-01	-0.1111E-03	000 V	4	0.2287E-01	-0.6148E-05
1203 A	4	0.2093E-01	-0.1086E-03	002 V	4	0.2235E-01	0.1559E-06
1203 A	4	0.2077E-01	-0.1029E-03	003 V	4	0.2206E-01	-0.1538E-05
1203 A	4	0.2033E-01	-0.1037E-03	005 V	4	0.2143E-01	0.3628E-05
1203 A	4	0.1983E-01	-0.1061E-03	007 V	4	0.2082E-01	-0.2834E-05
1203 A	4	0.1901E-01	-0.1084E-03	010 V	4	0.1988E-01	-0.1044E-04
1203 A	4	0.1833E-01	-0.1149E-03	013 V	4	0.1895E-01	-0.9216E-05
1203 A	4	0.1752E-01	-0.1188E-03	016 V	4	0.1776E-01	-0.1600E-04
1203 A	4	0.1700E-01	-0.1194E-03	019 V	4	0.1708E-01	-0.2034E-04
1203 A	4	0.1627E-01	-0.1276E-03	022 V	4	0.1616E-01	-0.2731E-04
1203 A	4	0.1520E-01	-0.1313E-03	026 V	4	0.1499E-01	-0.3408E-04
1203 A	4	0.1404E-01	-0.1390E-03	030 V	4	0.1384E-01	-0.4595E-04
1203 A	8	0.1288E-01	-0.1379E-03	032 V	8	0.1262E-01	-0.5172E-04
1203 A	4	0.1279E-01	-0.1500E-03	033 V	4	0.1254E-01	-0.5624E-04
1203 A	4	0.1261E-01	-0.1489E-03	034 V	4	0.1237E-01	-0.5692E-04
1203 A	4	0.1231E-01	-0.1484E-03	035 V	4	0.1208E-01	-0.5731E-04
1203 A	4	0.1210E-01	-0.1468E-03	036 V	4	0.1187E-01	-0.5744E-04
1203 A	4	0.1167E-01	-0.1429E-03	038 V	4	0.1147E-01	-0.5937E-04
1203 A	4	0.1110E-01	-0.1450E-03	041 V	4	0.1084E-01	-0.6588E-04
1203 A	4	0.1033E-01	-0.1342E-03	045 V	4	0.1000E-01	-0.6581E-04
1203 A	4	0.9633E-02	-0.1376E-03	047 V	4	0.9219E-02	-0.8059E-04
1203 A	4	0.9169E-02	-0.1219E-03	048 V	4	0.8268E-02	-0.8395E-04
1203 A	4	0.9155E-02	-0.1407E-03	049 V	4	0.8186E-02	-0.9044E-04
1203 A	4	0.9144E-02	-0.1454E-03	054 V	4	0.8092E-02	-0.8855E-04
1203 A	4	0.8971E-02	-0.1393E-03	060 V	4	0.7885E-02	-0.8401E-04
1203 A	4	0.7571E-02	-0.6637E-04	070 V	4	0.5438E-02	-0.9207E-04
1203 A	4	0.5149E-02	0.3421E-04	080 V	4	0.3787E-02	-0.9410E-04
1203 A	4	0.5118E-02	-0.9852E-04	090 V	4	0.3390E-02	-0.1109E-03
1203 A	4	0.4901E-02	-0.6283E-04	100 V	0	0.0000E 00	0.0000E 00
1203 A	4	0.4766E-02	-0.9463E-04	109 V	0	0.0000E 00	0.6000E-00
1203 A	4	0.3402E-02	0.1945E-04	118 V	0	0.0000E 00	0.0000E 00
1203 A	0	0.0000E 00	0.0000E 00	126 V	0	0.0000E 00	0.0000E 00

TABLE 3.2 Anodisation of 850-Å Tantalum Film

Plain Film

Argon-Ion Bombarded

	VOLTAGE	J	CONDUCTANCE (MHO)	TCR (PPM)		VCLTAGE	L	CONDUCTANCE (MHO)	TCR (PPM)
1201 A	000 V	4	0.2294E-01	-0.1504E-03	1201 B	000 V	4	0.1204E-02	-0.9597E-04
1201 A	002 V	4	0.2276E-01	-0.1494E-03	1201 B	002 V	4	0.1138E-02	-0.1080E-03
1201 A	003 V	4	0.2258E-01	-0.1480E-03	1201 B	003 V	4	0.1102E-02	-0.1024E-03
1201 A	005 V	4	0.2217E-01	-0.1396E-03	1201 B	005 V	4	0.1023E-02	-0.1027E-03
1201 A	007 V	4	0.2175E-01	-0.1440E-03	1201 B	007 V	4	0.9270E-03	-0.1150E-03
1201 A	010 V	4	0.2110E-01	-0.1494E-03	1201 B	010 V	4	0.8076E-03	-0.1021E-03
1201 A	013 V	4	0.2045E-01	-0.1501E-03	1201 B	013 V	4	0.6729E-03	-0.1012E-03
1201 A	016 V	4	0.1963E-01	-0.1491E-03	1201 B	016 V	4	0.5040E-02	-0.1035E-03
1201 A	019 V	4	0.1911E-01	-0.1510E-03	1201 B	019 V	4	0.4085E-02	-0.9736E-04
1201 A	022 V	4	0.1839E-01	-0.1539E-03	1201 B	022 V	4	0.3339E-03	-0.1069E-03
1201 A	026 V	4	0.1737E-01	-0.1503E-03	1201 B	026 V	4	0.2358E-03	-0.1009E-03
1201 A	030 V	4	0.1624E-01	-0.1512E-03	1201 B	030 V	4	0.1454E-02	-0.1069E-03
1201 A	032 V	8	0.1514E-01	-0.1586E-03	1201 B	032 V	8	0.5718E-04	-0.3870E-03
1201 A	033 V	4	0.1506E-01	-0.1691E-03	1201 B	033 V	4	0.5221E-04	-0.8717E-04
1201 A	034 V	4	0.1487E-01	-0.1682E-03	1201 B	034 V	4	0.4720E-04	-0.3235E-03
1201 A	035 V	4	0.1455E-01	-0.1671E-03	1201 B	035 V	4	0.4254E-04	-0.2807E-03
1201 A	036 V	4	0.1432E-01	-0.1630E-03	1201 B	036 V	4	0.3969E-04	-0.9823E-04
1201 A	038 V	4	0.1384E-01	-0.1622E-03	1201 B	038 V	4	0.3495E-04	-0.5617E-04
1201 A	041 V	4	0.1310E-01	-0.1610E-03	1201 B	041 V	4	0.2819E-04	-0.9552E-04
1201 A	045 V	4	0.1222E-01	-0.1491E-03	1201 B	045 V	4	0.1527E-04	-0.5762E-04
1201 A	047 V	4	0.1152E-01	-0.1531E-03	1201 B	047 V	4	0.1150E-04	-0.8748E-04
1201 A	048 V	4	0.1108E-01	-0.1400E-03	1201 B	048 V	4	0.7942E-05	0.5008E-03
1201 A	049 V	4	0.1106E-01	-0.1545E-03	1201 B	049 V	4	0.7975E-05	-0.3390E-03
1201 A	054 V	4	0.1105E-01	-0.1570E-03	1201 B	054 V	0	0.0000E-00	0.3361E-03
1201 A	060 V	4	0.1095E-01	-0.1555E-03	1201 B	060 V	4	0.7258E-05	0.0000E-00
1201 A	070 V	4	0.9310E-02	-0.8996E-04	1201 B	070 V	4	0.6378E-05	-0.1156E-03
1201 A	080 V	4	0.6633E-02	0.2108E-04	1201 B	080 V	0	0.0000E-00	-0.7629E-04
1201 A	090 V	4	0.6588E-02	-0.1091E-03	1201 B	090 V	0	0.0000E-00	0.0000E-00
1201 A	100 V	4	0.6345E-02	-0.7496E-04	1201 B	100 V	0	0.0000E-00	0.0000E-00
1201 A	109 V	4	0.6169E-02	-0.1061E-03	1201 B	109 V	0	0.0000E-00	0.0000E-00
1201 A	118 V	4	0.4852E-02	0.4114E-05	1201 B	118 V	0	0.0000E-00	0.0000E-00
1201 A	126 V	0	0.0000E-00	0.0000E-00	1201 B	126 V	0	0.0000E-00	0.0000E-00

4. Results

The electrical properties of the metal films made as described above in section 3, have been measured for different levels of bombardment with argon ions. In this work all ion bombardment was carried out using argon.

The electrical properties investigated are:

Sheet Resistance,

Temperature Coefficient of Resistance,

Strain-Gauge Coefficient of Resistance.

4.1 Gold Films

All the gold films were made by evaporation 'in vacuo' onto glass substrate; the film thickness being controlled (see 3.1.2) so as to produce films of approximately 12 ohms resistance.

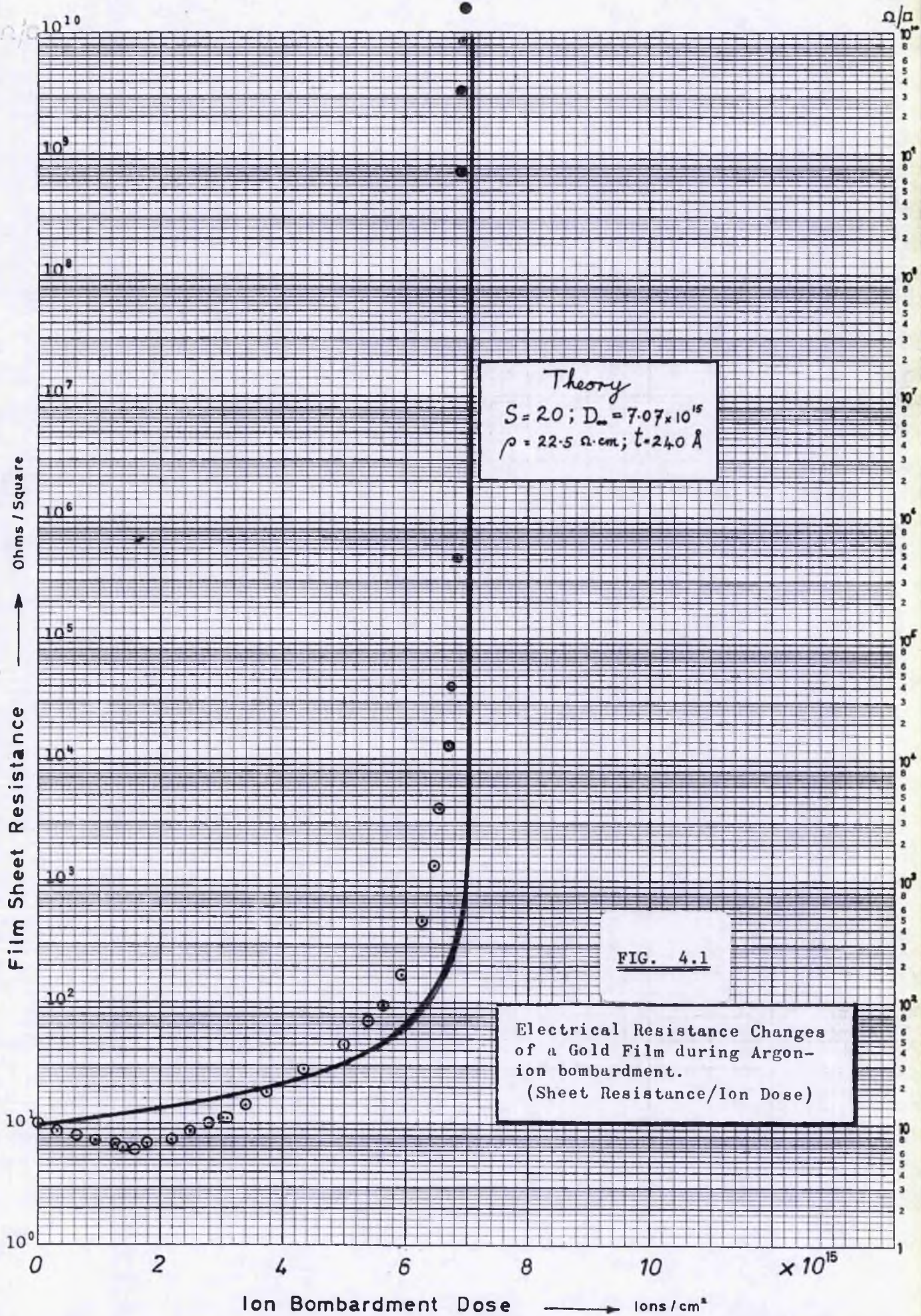
4.1.1 Sheet Resistance

The sheet resistance of the gold film at evaporation was, on average, 4.8 ohm/sq \pm 0.6 ohm/sq, measured at 30° C. Multiplication by the film thickness shows an apparent resistivity of 11.5 microhm-cm. (The resistivity of bulk gold is 2.25 microhm-cm.)

The effect of ion bombardment on the sheet resistance of a single 240 - Å gold film is shown in Figure 4.1. Here the resistance of a particular film is plotted against bombardment dose at different stages during the ion bombardment. The main feature is the dramatic increase in sheet resistance that occurs at a bombardment dose of 6.5×10^{15} ions/cm² (corresponding to 1050 μ C/cm²). It is at this level of bombardment that the film goes open circuit, so that only a small increase in bombardment effects a large change.

Initially on bombardment the sheet resistance drops slightly with increasing ion dose, reaching -40% at 1.5×10^{15} ions/cm². Beyond this dose the resistance rises steadily through two orders of magnitude. For argon-ion bombardment above 6.5×10^{15} ions/cm² the resistance rises rapidly to an open-circuit condition.

The continuous curve shown in Figure 4.1 is a "best fit" line calculated using eqn.2.3 for assumed values of sputter rate and initial resistivity, and is based on the premise that the film is uniformly thinned as the bombardment proceeds. The curve which most closely fits the observed points is that for a sputter rate of 20 atoms/ion and an initial resistivity of 22.5 microhm-cm.



4.1.2 Resistivity

The effective resistivity of the film during argon bombardment may be calculated from the sheet resistance/bombardment dose relationship plotted in Figure 4.1 by assuming the film thickness to vary inversely as the dose. This calculated result is plotted in Figure 4.2 against bombardment dose.

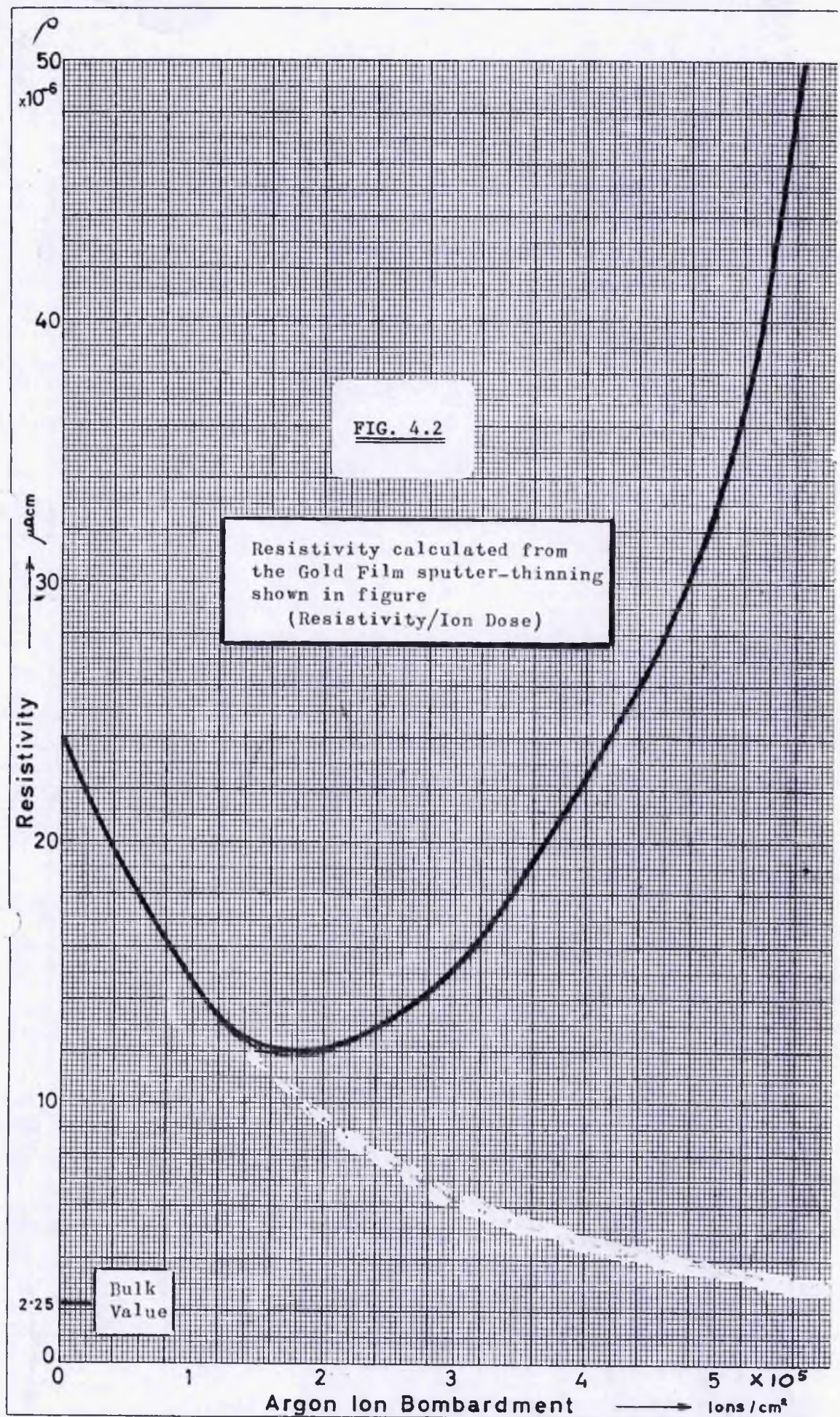
The initial resistivity of this film was $24 \mu\Omega\text{.cm.}$, but dropped gradually to $12 \mu\Omega\text{.cm.}$ at a bombardment dose of 1.9×10^{15} ions/cm². On further bombardment the resistivity progressively increased until the film became effectively open-circuit. In Figure 4.3 is shown the variation of conductivity with ion dose with theoretical curves for three values of sputtering rate. The relationship between volume resistivity and sheet resistance is shown in Figure 4.4.

4.1.3 Annealing

The isothermal annealing of gold films at room temperature is shown in Figures 4.5, and 4.6. Figure 4.5 shows the resistance change for a film bombarded to a resistance of 100 ohms. No significant change occurred over a period of about 200 hours (~1 week), but after this the resistance increased steadily and reached 2000 ohms after 1000 hours. Beyond this period the resistance remained practically constant. During the rapid resistance change the slope of the line (on the log/log scale) is 2, indicating a second-order dependence with time.

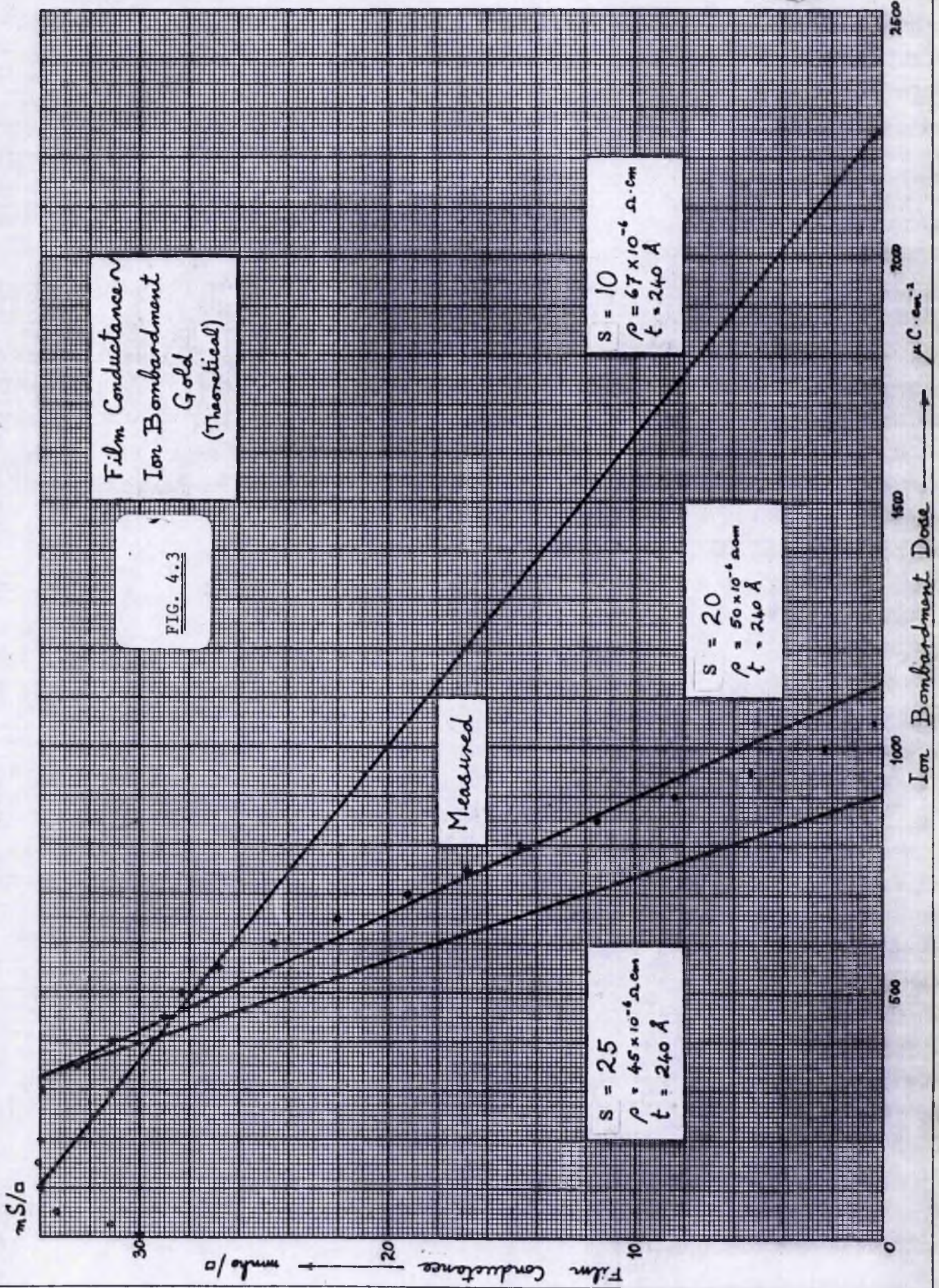
Annealing curves for several more films are given in Figure 4.6. The resistances have been normalised to the values obtaining at the end of the ion bombardment. It is characteristic of all films that the resistance remained constant for about 200 hours, increased during the next 200 or 300 hours, and thereafter remained relatively unchanged for several thousand hours.

The extent of the resistance change has been observed to be as much



Film Conductance
Ion Bombardment
Gold
(Theoretical)

FIG. 4.3



S = 25
ρ = 45 × 10⁻⁶ Ω·cm
t = 240 Å

S = 20
ρ = 50 × 10⁻⁶ Ω·cm
t = 240 Å

S = 10
ρ = 67 × 10⁻⁶ Ω·cm
t = 240 Å

Measured

mS/cm

Film Conductance → mho/cm

500

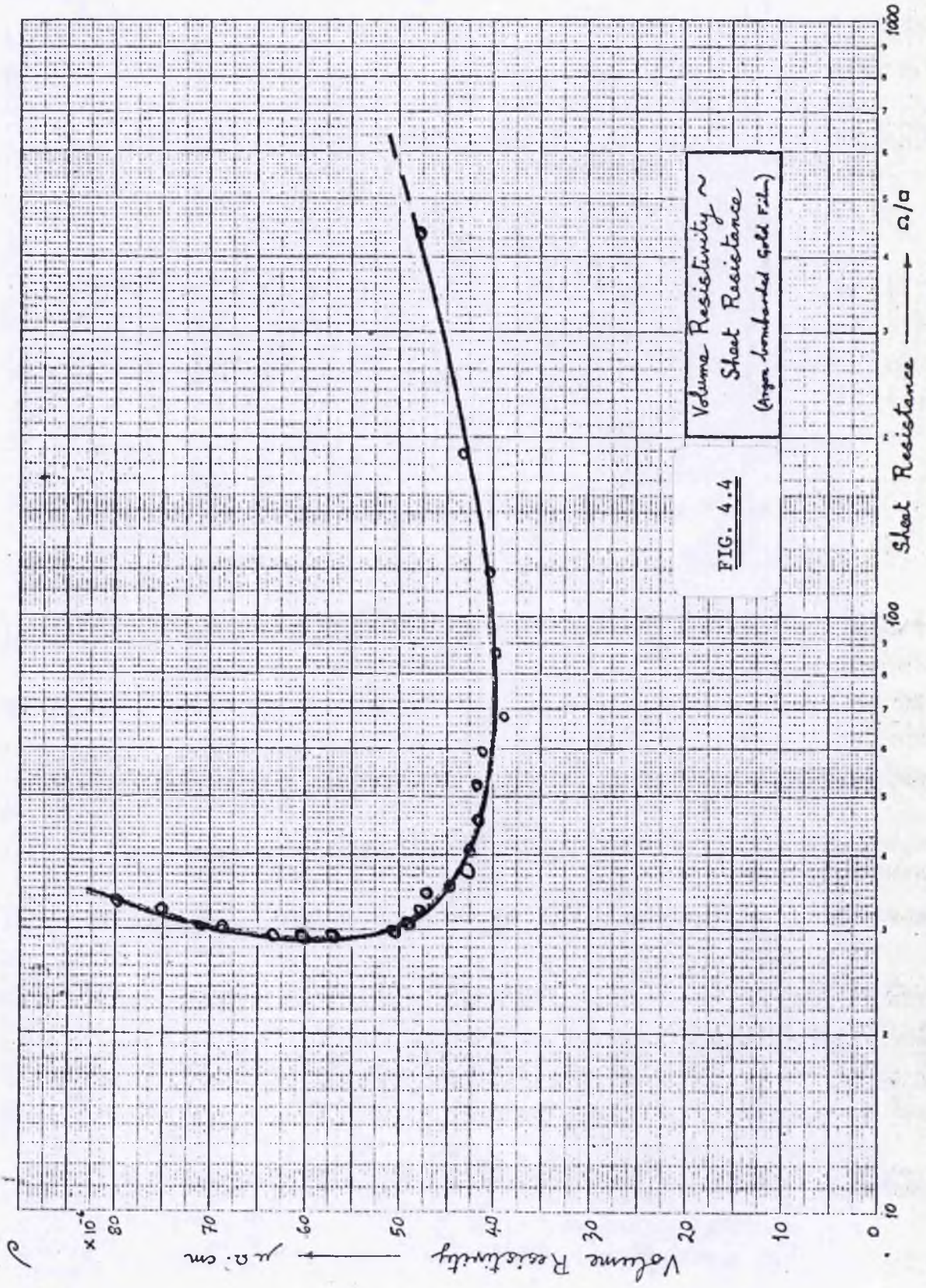
1000

1500

2000

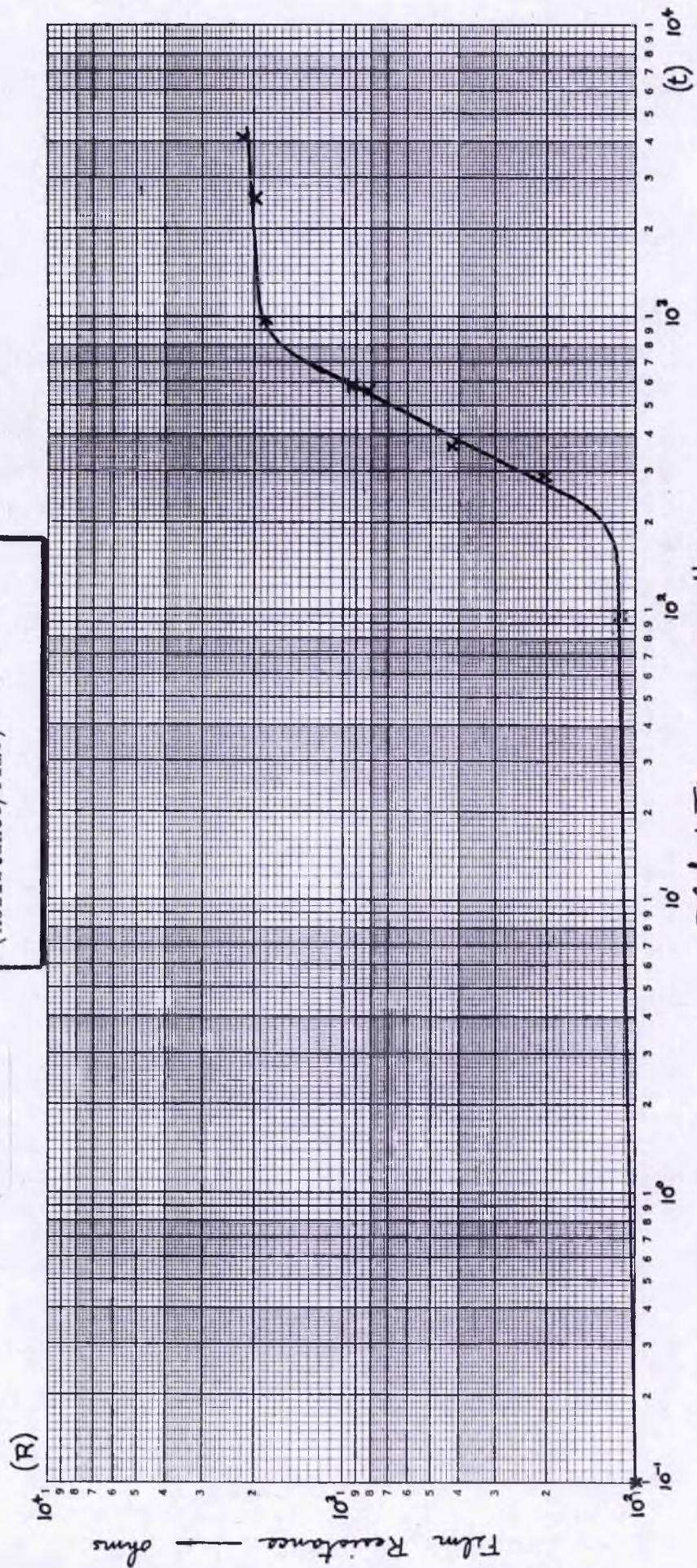
2500

Ion Bombardment Dose → μC·cm⁻²



Room-temperature Annealing of
 an Ion-bombarded Gold Film.
 (Resistance/Time) 0.25×10^4

FIG. 4.5



Elapsed Time — Hours
 after Bombardment.

Film Resistance — ohms

as $\times 40$ and as little as $\times 1.1$, but the high resistance films become open-circuit eventually, after about 5000 hours.

The resistance changes for 12 films are given in Table 4.1.

4.1.4 Temperature Coefficient of Resistance

The temperature coefficient of resistance (TCR) of the sputtered films was measured after annealing for two weeks at room temperature and two hours at 120°C , and is plotted against their sheet resistance in Figure 4.7(a,b). In the main the TCR's all lie between 1000 parts per million per degree C and $1500\text{ PPM}/^{\circ}\text{C}$, though there is some indication of an increase in TCR at low sheet resistance (above 10 ohms/square). The average TCR of ion-bombarded gold films having sheet resistances between 3 ohms/square and 4×10^4 ohms/square is $+1175\text{ PPM}/^{\circ}\text{C} \pm 95\text{ PPM}/^{\circ}\text{C}$.

The theoretical curve is based on equn.2.11 for the TCR of thin films, using $\text{MPF} = 800\text{ \AA}$, resistivity = $50\text{ }\mu\Omega\text{.cm}$ and TCR of bulk gold = $4000\text{ PPM}/^{\circ}\text{C}$.

Values of TCR for 15 specimens are given in Table 4.2. It from this and Fig. 4.7(b), can be seen that for the $35\text{ M}\Omega\text{/square}$ specimen we have a negative value.

4.1.5 Strain-Gauge Coefficient of Resistance

The strain-gauge coefficients of the sputter-etched gold films, measured after annealing, are shown in Figure 4.8 plotted against their sheet resistance. Those having low sheet resistance ($< 6\text{ ohm/sq}$) are for films measured before bombardment, although the sheet resistances extend from less than 3 ohms/square to more than 4×10^4 ohms/square the resulting strain-gauge coefficients show no corresponding variation. The average value of strain-gauge coefficient for all gauges is $\gamma = 2.58 \pm 0.07$.

The straight horizontal line drawn at $\gamma = 2.43$ is developed from the theoretical relationship given in equn. 2.6.

T A B L E 4.1

NO.	AT BOMBARDMENT				RESISTANCE (Ω)	DATE	TIME LAPSE (Hrs)	RESISTANCE (Ω) H	RELATIVE RESISTANCE B/A	DATE	TIME LAPSE (Hrs)	RESISTANCE (Ω) C	RELATIVE RESISTANCE C/A	DATE	TIME LAPSE (Hrs)	RESISTANCE (Ω) D	RELATIVE RESISTANCE D/A
	RESISTANCE (Ω)	DATE	DOSE (μCec)	RESISTANCE (Ω) A													
020	10.8	18.I.73	1220	10 000	9.II.73	528	54 150	5.41	11.VII.73	4224	52 400	5.24	2.III.74	9766	0/C		
021	11.0	18.I.73	1035	3 080	9.II.73	528	37 100	12.1	11.VII.73	4224	38 750	12.6	2.III.74	9766	88 400	28.7	
023	10.75	17.I.73	780	92	2.II.73	354	128	1.41	11.VII.73	4248	127	1.38	2.III.74	9790	142.3	1.6	
024	11.77	17.I.73	980	1 070	2.II.73	384	1568	1.47	11.VII.73	4248	1637	1.55	2.III.74	9790	1630	1.52	
025	14.16	17.I.73	640	32	2.II.73	384	37.7	1.18	11.VII.73	4248	39	1.22	2.III.74	9790	36.7	1.15	
026	13.68	17.I.73	1000	290	9.II.73	552	456.9	1.58	11.VII.73	4248	550	1.9	2.III.74	9790	485	1.07	
035	6.3	1.XII.73	1650	192	29.XII.73	360	211	1.1	2.III.74	2160	205.5	1.07					
038	8.9	5.IX.73	--	98.5	9.IX.73	96	112	1.14	17.IX.73	288	201.8	2.05	20.IX.73	360	429	4.35	
042	11.6	1.XII.73	300	150	28.IX.73	562	834	8.45	29.IX.73	576	944	9.6	15.X.73	960	1844	18.8	
043	46.6	1.XII.73	250	124	20.XII.73	2544	1977	20.1	24.II.74	4128	2219	22.5	2.III.74	4272	2252	22.9	
044	9.5	1.XII.73	1360	153	24.XII.73	552	3069	20.5	2.III.74	2160	5348	35					
047	5.2	1.XII.73	750	800	26.XII.73	600	1131	9.15	2.III.74	2160	1255	10.1					
					27.XII.73	624	3375 (10220)	35.2	2.III.74	2160	13 053 (annealed)	85.3					

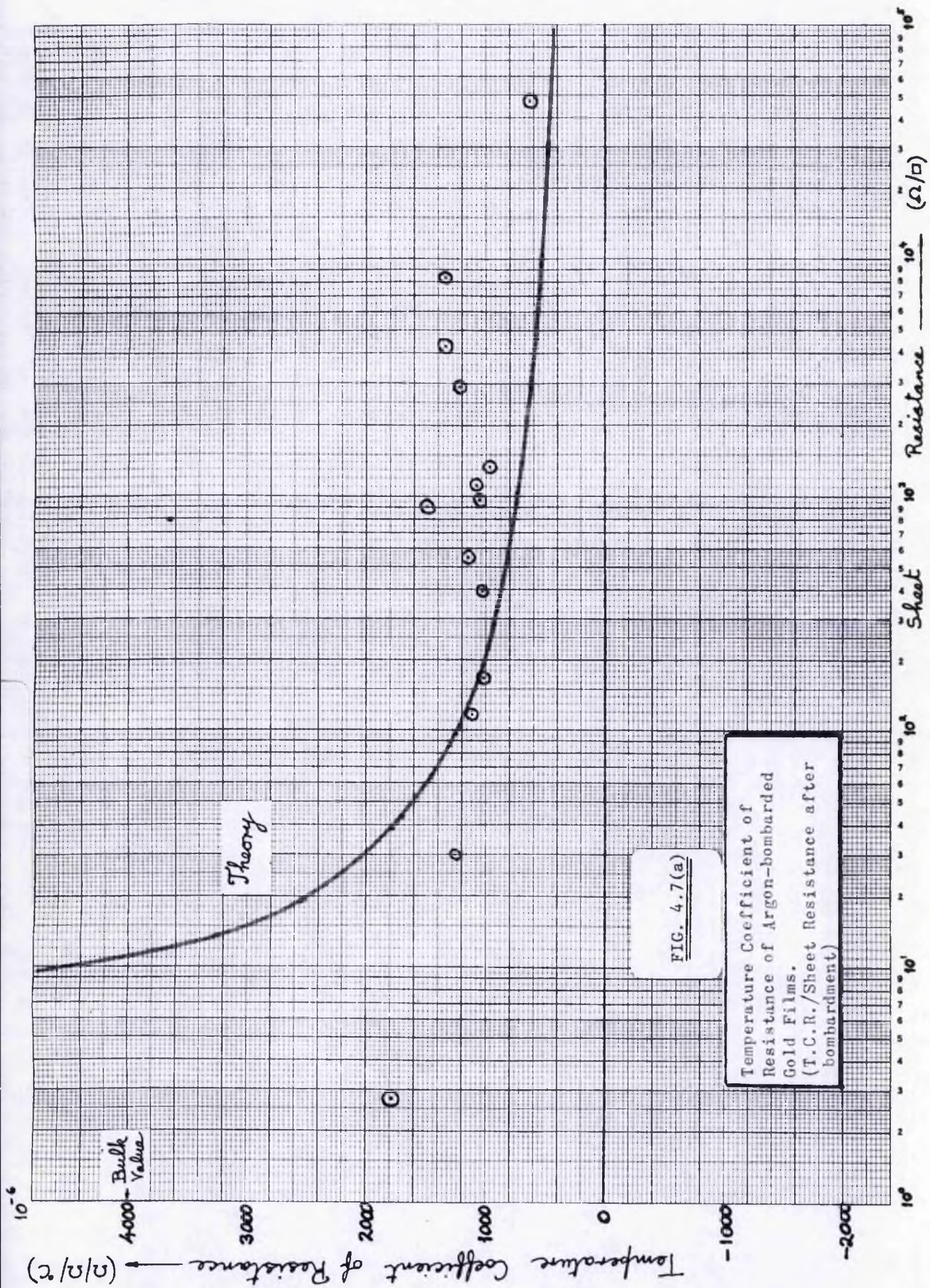
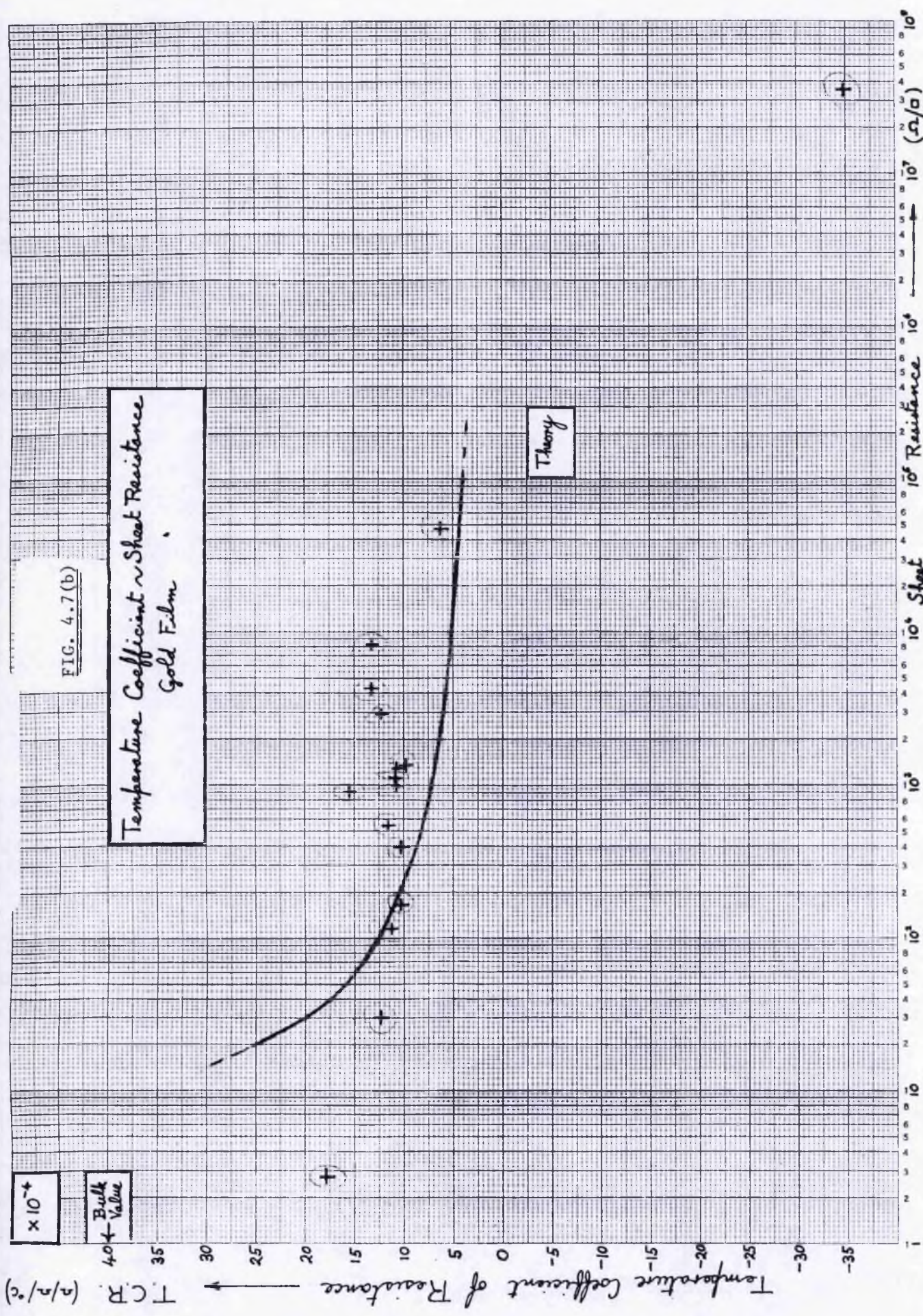


FIG. 4.7(b)

Temperature Coefficient ~ Sheet Resistance
Gold Film



$\times 10^{-4}$
Bulk Value

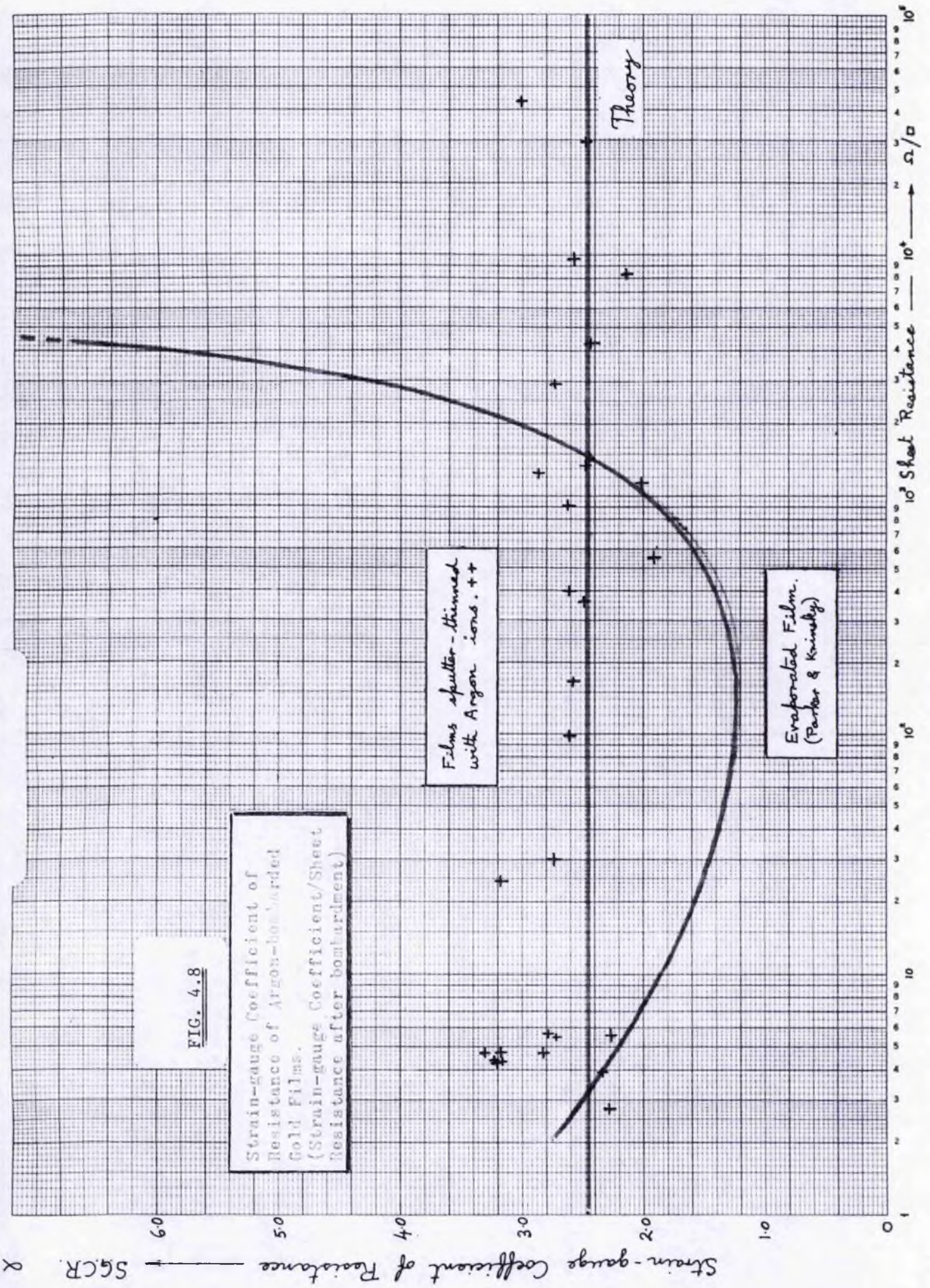
T.C.R. ($1/^\circ\text{C}$)

Theory

Resistance (Ω)

TABLE 4.2

Specimen Number	010	021	023	024	025	020	025	035	038	038	042	043	044	044	047	
Substrate thickness (mm)	.835	.87	.85	.85	.88	.875	.008	.908	.96	.96	.89	.80	.883	.583	.885	
Dombardment window (mm sq)	10	10	10	10	10	10	—	10	14	14	10	10	10	10	10	
Ion dose (C cm ⁻²)	1505	1036	780	980	640	1000	N11	1050	—	—	800	250	1350	1360	750	
Sheet resistance	35.0 MO/□	46.7 KΩ/□	116.5 Ω/□	1325 Ω/□	30.0 Ω/□	395 Ω/□	2.75 Ω/□	169 Ω/□	545 Ω/□	1140 Ω/□	2930 Ω/□	905 Ω/□	4300 Ω/□	8300 Ω/□	980 Ω/□	
T.C.R. (x 10 ⁻⁴)	-34.8	6.19	11.2	9.77	12.5	10.2	17.9	10.2	11.5	10.7	12.2	15.5	13.4	13.3	10.0	
Temperature (°C)	STRAIN-GAUGE COEFFICIENT OF RESISTANCE, (S.G.C.R.)															
-20	—	—	3.71	2.85	—	2.78	2.63	2.58	2.08	—	—	—	—	—	—	—
-10	—	—	—	2.82	2.85	—	2.86	2.47	2.49	—	—	—	—	—	—	—
0	—	—	—	2.97	2.77	—	2.53	2.04	2.04	—	—	—	—	—	—	—
10	—	—	—	3.94	3.04	3.72	2.34	2.71	2.04	2.02	2.81	2.81	2.21	2.19	—	—
20	—	—	—	2.68	2.97	2.60	2.49	2.60	1.93	2.16	3.03	3.02	2.29	2.29	—	—
30	—	—	—	2.48	2.74	2.61	2.28	2.58	1.92	2.05	2.72	2.62	2.43	2.14	—	—
40	—	—	—	2.48	2.90	2.60	2.52	2.56	1.95	2.09	2.50	2.54	2.57	2.18	—	—
50	—	—	—	2.53	2.60	2.74	2.66	2.58	1.93	2.07	2.90	2.80	—	2.10	—	—
60	—	—	—	2.38	2.68	2.61	2.56	2.64	2.0	2.06	—	2.50	—	2.16	—	—
70	—	—	—	2.50	2.52	2.68	2.40	—	1.96	2.03	—	—	—	—	—	—



The curved line is taken from the work of Parker and Krinsky,⁴ and shows the variation in γ with sheet resistance found by them for evaporated thin gold films (without ion bombardment). They found that the strain-gauge factor of this type of film has a minimum value of $\gamma = 1.2$ at $R_s = 150$ ohms/square, that it rises to the bulk value ($\gamma = 4.48$) for low sheet resistances (thicker films), and that it rises steeply towards $\gamma = 100$ for films having R_s greater than about 2×10^5 ohms/square.

4.1.6. Temperature Coefficient of Strain-Gauge Factor

Measurements were made by strain-gauge factor at different temperatures, with the result shown in Figure 4.9. Because no significant variation of γ was found with change in R_s due to ion bombardment, measurements are included of films having had different ion bombardment doses. The very large scatter in the strain-gauge measurements arises from difficulties in controlling the bending of the specimens in the environmental chamber.

A simple line-fitting routine was used to obtain the temperature coefficient of strain-gauge factor of $\rho = -1590$ PPM/ $^{\circ}$ C + -1100 PPM/ $^{\circ}$ C, and an average value for γ of 2.58 at 30° C. This result is statistically significant.

4.1.7 S.E.M. Microscopic Examination

The structure of the bombarded gold films has been examined under the scanning electron microscope at a magnification of X14000. Unbombarded films show a smooth featureless surface; but after bombardment this is found covered with irregularly shaped dark areas (substrate). Figure 4.10 shows micrographs of four films at different bombardment levels.

At a dose of 4×10^{15} ions/cm² (Film No. 25) the micrographs shows thin fissures in the film surface: the resistance of this film had been

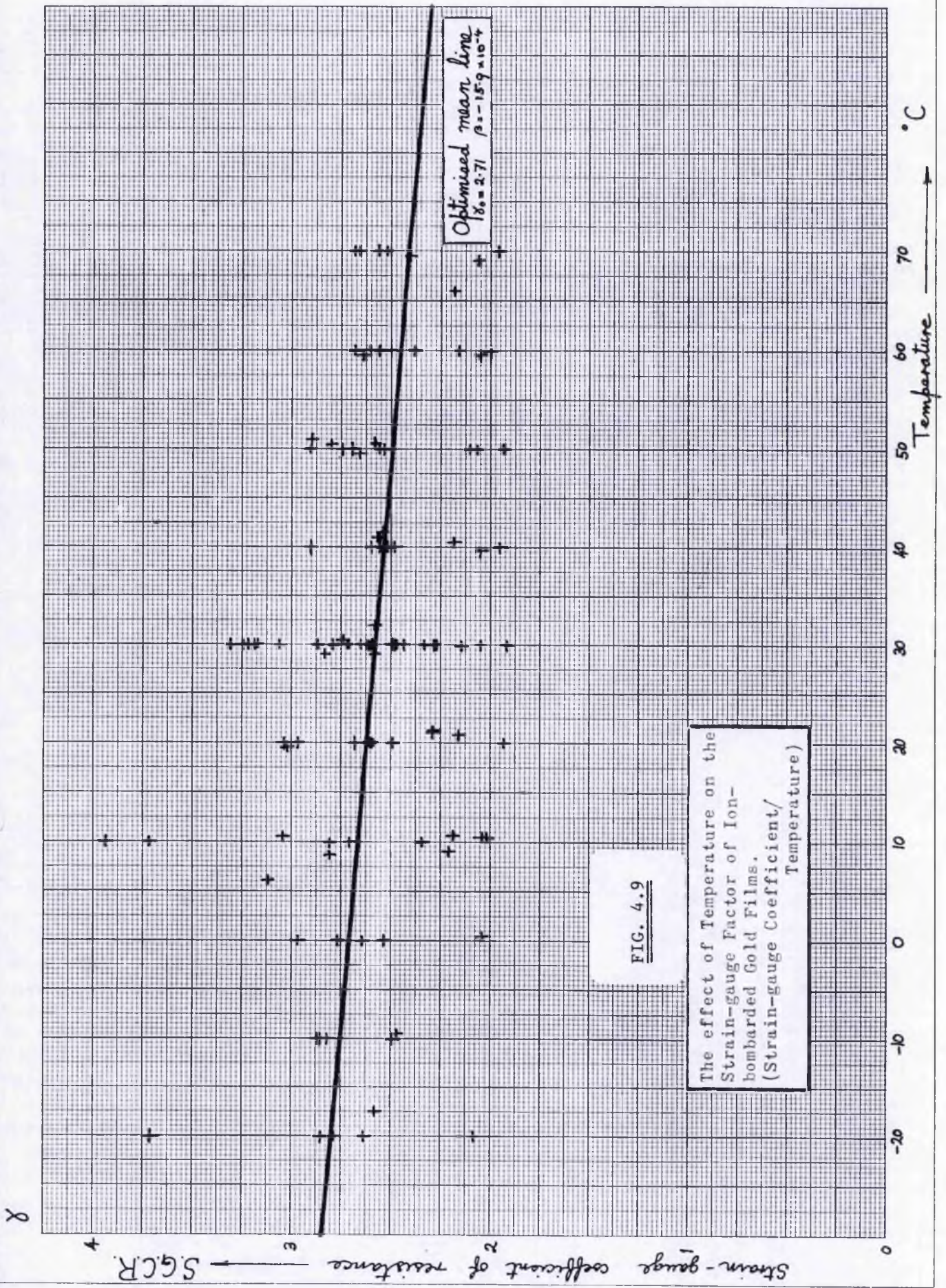
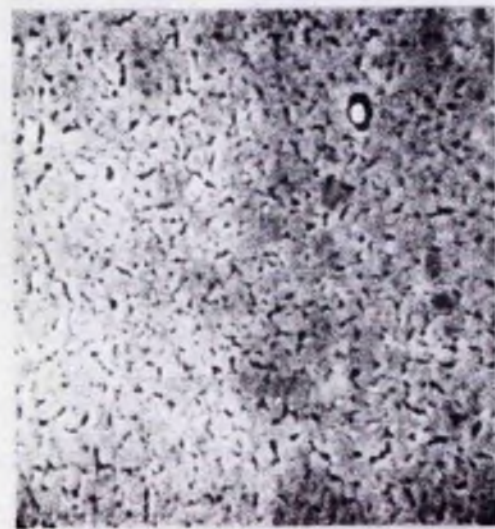


FIG. 4.9

The effect of Temperature on the Strain-gauge Factor of Ion-bombarded Gold Films. (Strain-gauge Coefficient/ Temperature)

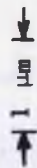


No. 25 $R_s = 30 \Omega/\square$, 4×10^{15} Ions/cm²

Magnification
x 14 k



Tilt 35°



No. 26 $R_s = 290 \Omega/\square$, 6.2×10^{15} Ions/cm²



No. 47 $R_s = 980 \Omega/\square$, 4.7×10^{15} Ions/cm²



No. 21 $R_s = 99 \text{ k}\Omega/\square$, 6.5×10^{15} Ions/cm²

FIG. 4.10

GOLD FILM

S.E.M. MICROGRAPHS

increased by ion bombardment to 30 ohm/sq. A second film (No.47) bombarded to 4.7×10^{15} ions/cm², had a sheet resistance of 980 ohm/sq. The fissures in this had thickened about five times without a significant increase in their length, though the mean spacing of the damage sites was the same as in the first case.

Film No.26 had been bombarded to 6.2×10^{15} ions/cm². Here the areas of damage (dark areas) covered more than half the film surface and the remaining metal had taken on a spongiform appearance. The resistance of this film seems unusually low (being 290 ohm/sq) since film No. 21, bombarded to 6.5×10^{15} ions/cm² had a sheet resistance of 99 kilohm/sq. In this film also the appearance was spongiform, though having a finer texture than that of No. 26.

4.1.8 Microprobe Analysis

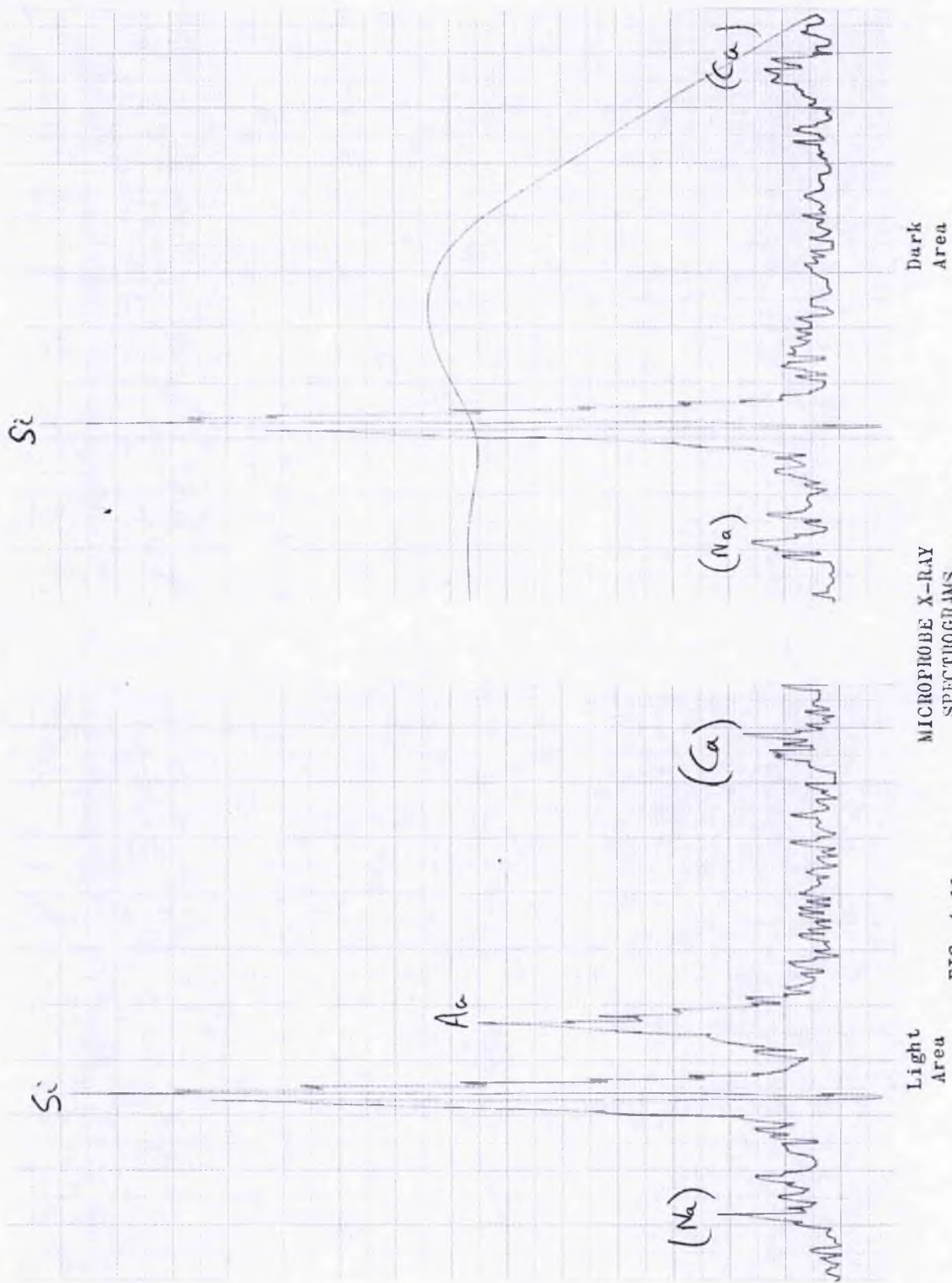
An examination was made of the light and dark areas shown in the micrographs, using x-ray analysis in a C.S.I. microprobe analyser. Film No. 26 was used for this (bombardment was 4.7×10^{15} ions/cm²): the spectrograms are shown in Figure 4.11. The trace on the left was made with the probe on a light area, and it shows a well defined gold response of about one-half that obtained for silicon. On the right is the trace for a dark area, which has no gold peak but shows correspondingly similar values for sodium, silicon and calcium.

This result demonstrates that the light areas of the micrographs are gold film and that the dark areas are holes in the film through which the soda glass substrate is observed.

With the probe directed to an unbombarded part of the film the response peaks for gold and for silicon were of equal amplitude.

4.2 Tantalum Films before Bombardment

The tantalum films used for strain measurement were all deposited



Dark Area

MICROPROBE X-RAY
SPECTROGRAMSFIG. 4.11 ($R_s = 980 \text{ } \Omega/\square$, $4.7 \times 10^{15} \text{ Ions/cm}^2$)

Light Area

onto glass substrates, but a few others - used for sheet resistance and TCR measurement - were deposited onto alumina and a few onto mica substrates, no noticeable difference was observed in their use.

4.2.1 Sheet Resistance

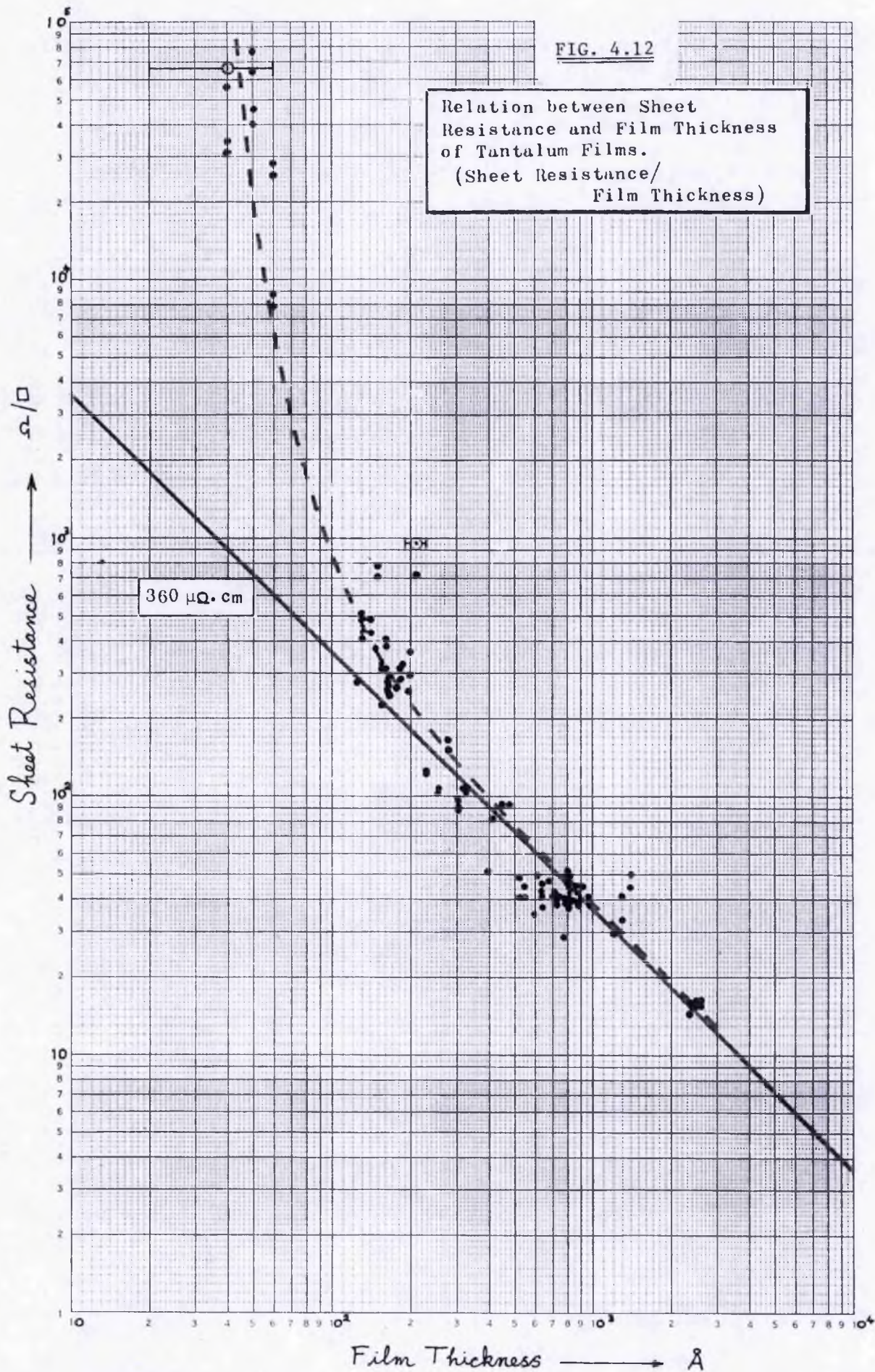
The variation of sheet resistance with film thickness was explored within the thickness range 50 Å to 2600 Å, and was found to extend from $R_s = 80 \text{ K}\Omega/\text{sq}$ to $R_s = 14\Omega/\text{sq}$ respectively. A plot of this result is given in Figure 4.12. The sheet resistance varies inversely with film thickness, being asymptotic to a line $\rho = 360 \mu\Omega.\text{cm}$ for the thicker films and rising steeply above this line at film thicknesses below 100 Å.

The corresponding variation of resistivity with film thickness has been calculated for the mean sheet resistance line (shown dotted in Figure 4.12) and is given in Figure 4.13. This shows that for film thickness above 400 Å the resistivity is sensibly constant, being $380 \mu\Omega.\text{cm}$ for $T = 2500 \text{ Å}$ and $400 \mu\Omega.\text{cm}$ for $T = 400 \text{ Å}$. Below this thickness the resistivity rises rapidly, and reaches $2500 \mu\Omega.\text{cm}$ at $T = 50 \text{ Å}$.

4.2.2. Temperature Coefficient of Resistance

The TCR of evaporated tantalum film measured between room temperature and about 120°C is very small, and is practically constant for films of greater thickness than 200 Å. The average TCR of these films is $-150 \text{ PPM}/^\circ \text{C}$. Films of lesser thickness show a progressively greater negative TCR, 50-Å films having a TCR of $-1800 \text{ PPM}/^\circ \text{C}$. This is shown in Figure 4.14. The transition (around $T = 150 \text{ Å}$) between large negative TCR and a constant, small TCR is quite abrupt.

TCR plotted against sheet resistance is given in Figure 4.15. The region of constant TCR extends from $R_s = 14 \Omega/\text{sq}$ to $R = 9 \text{ K}\Omega/\text{sq}$.



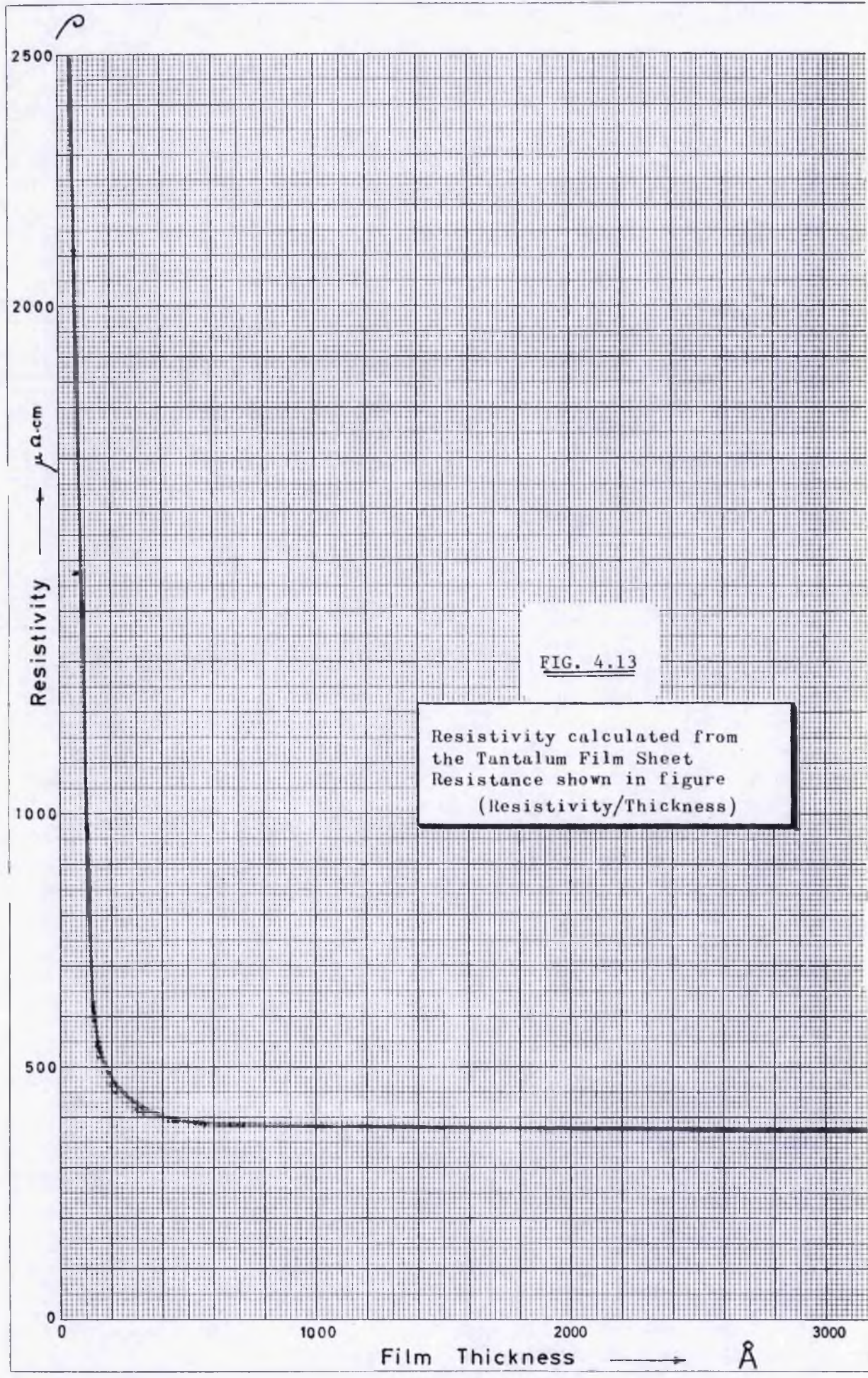
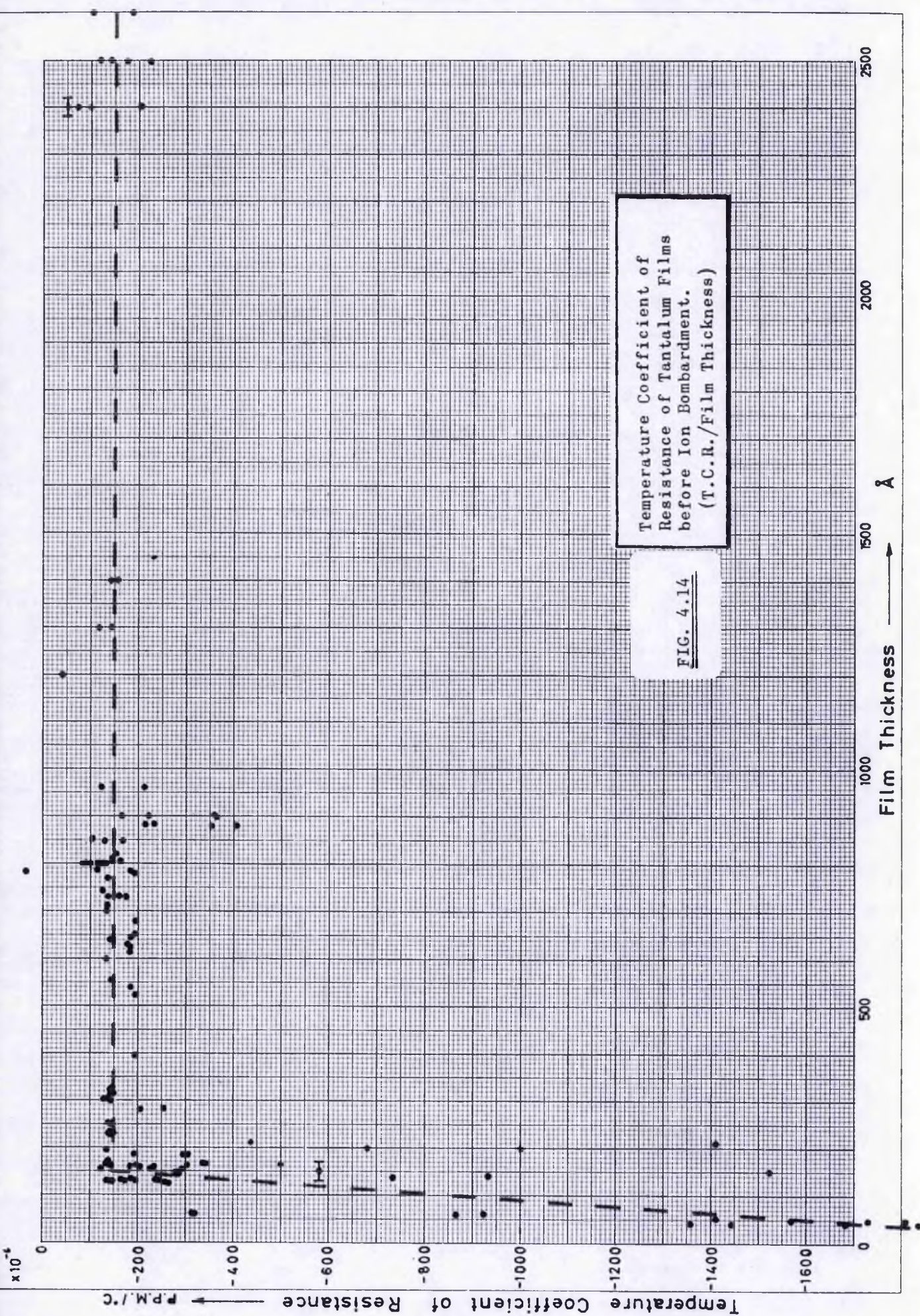


FIG. 4.13

Resistivity calculated from
the Tantalum Film Sheet
Resistance shown in figure
(Resistivity/Thickness)

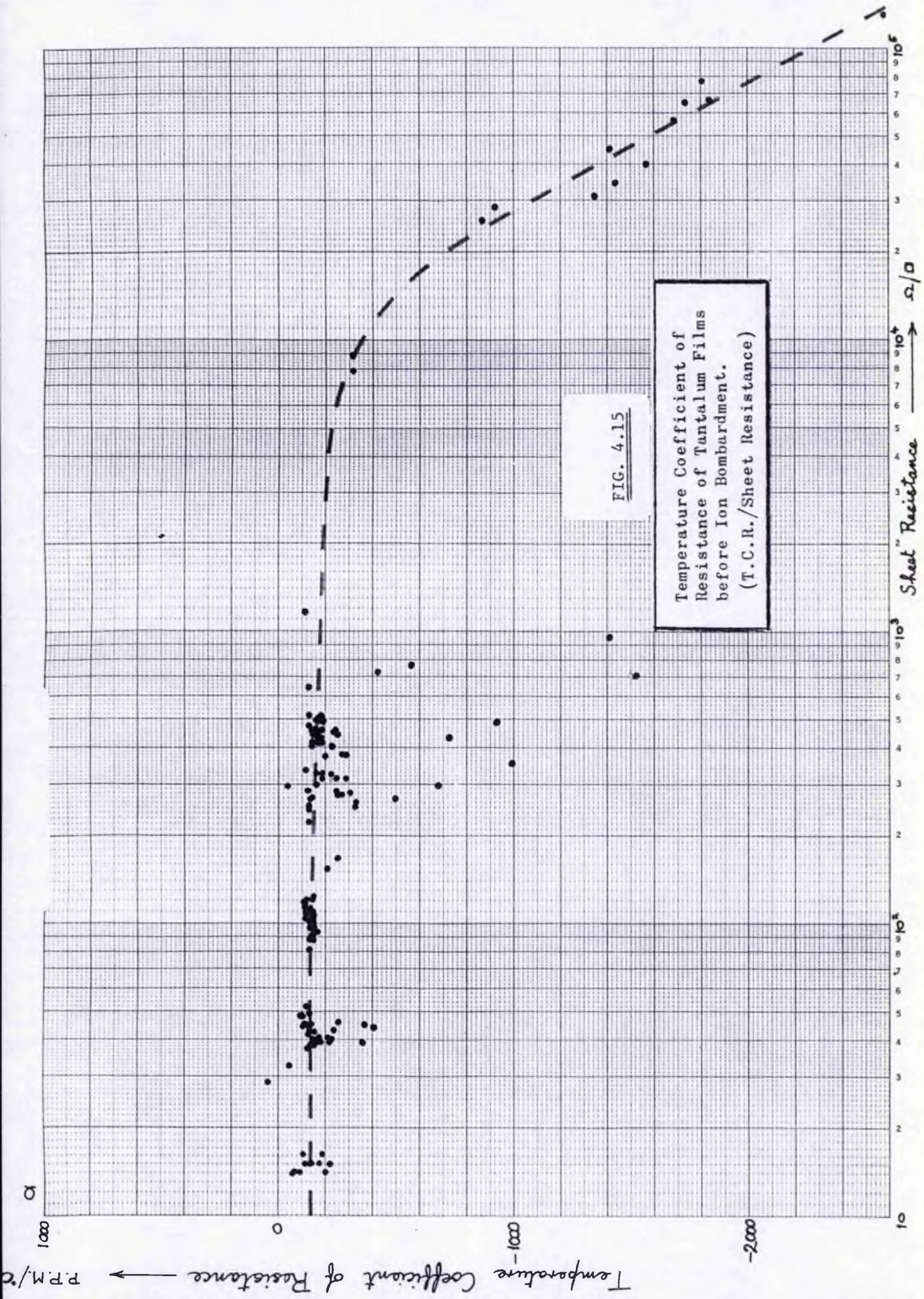
2500
2000
Resistivity
 $\mu\Omega\text{-cm}$
1000
500
0

Film Thickness \rightarrow Å
1000 2000 3000



Temperature Coefficient of Resistance of Tantalum Films before Ion Bombardment. (T.C.R./Film Thickness)

FIG. 4.14



At greater sheet resistance the TCR becomes larger and more negative, but the transition is not abrupt.

A few films with sheet resistance below 1000 Ω/sq have TCR greater than usual, but this seems to be abnormal.

The resistance of one film (No.3706, $T = 730 \text{ \AA}$) has been observed over the temperature range -195°C to $+110^{\circ} \text{C}$, see Figure 4.16. By treating this result as a second order relationship the following expression is obtained:

$$R_s = 40.9(1 - 144 \times 10^{-6} T + 0.28 \times 10^{-6} T^2) \Omega / \text{sq}.$$

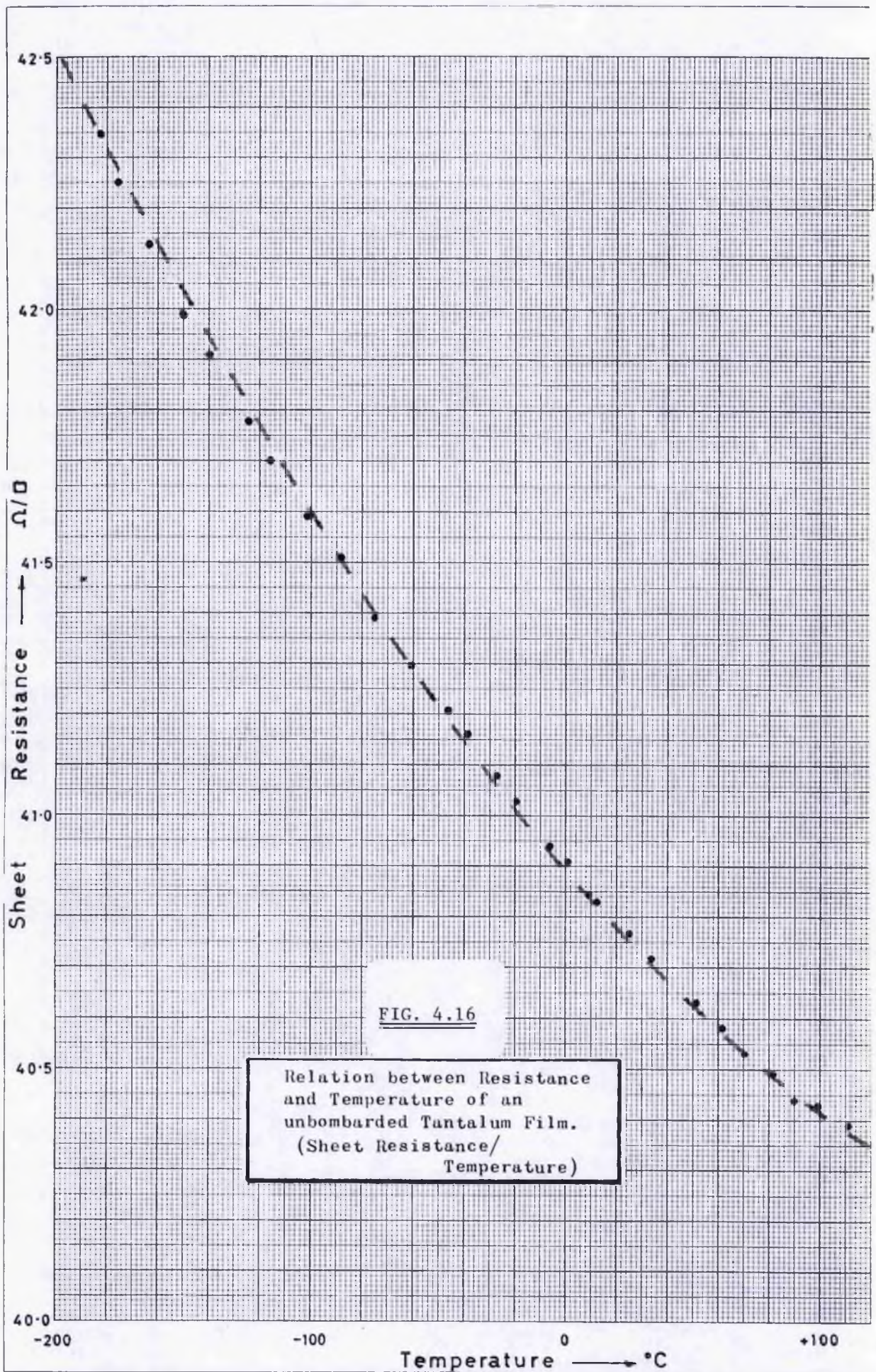
Extrapolation indicates a minimum sheet resistance of 40.1 Ω/sq . occurring at a temperature of 257°C .

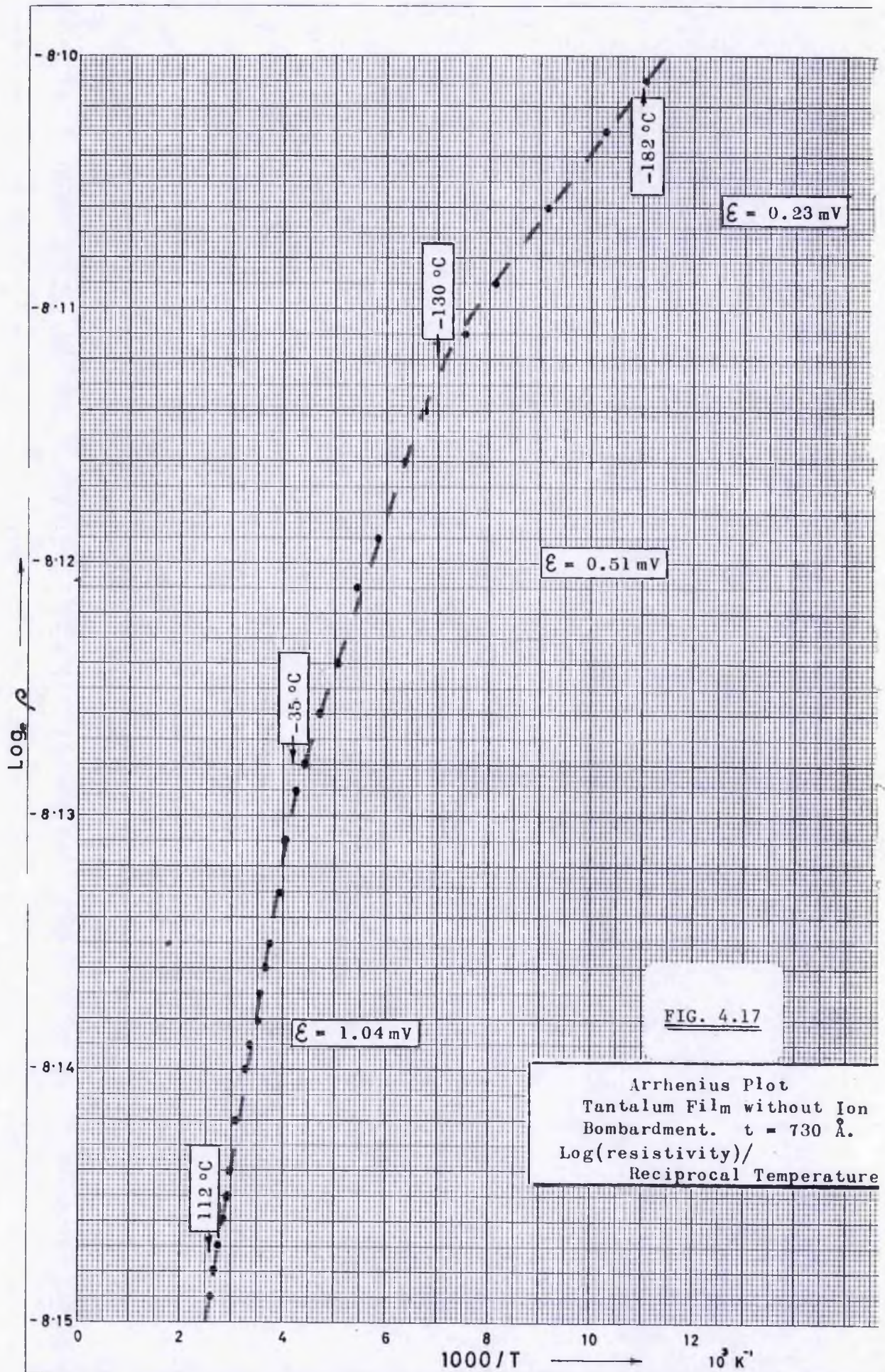
An Arrhenius plot from these resistance/temperature measurements is given in Figure 4.17. Three activation levels are evidence in this; at temperatures below -130°C an activation energy of $0.23 \times 10^{-3} \text{ eV}$; between -130°C and -35°C a level of $0.51 \times 10^{-3} \text{ eV}$; and for temperatures above -35°C a level of $1.04 \times 10^{-3} \text{ eV}$. These activation energies stand approximately in the ratios 1:2:4.

The activation energies of all films are shown plotted against sheet resistance in Figure 4.18. The energies were measured over the temperature range room temperature to about 120°C . Although there is a general trend towards higher activation energy at greater sheet resistance, there is a considerable spread in the energy values, extending over about one order of magnitude.

4.2.3 Strain-Gauge Coefficient of Resistance

Strain-gauge measurements were made on tantalum films evaporated to a nominal thickness of 750 \AA : those used had sheet resistances between 30 Ω/\square and 50 Ω/\square . Under simple four-point bending up to a value of about 250 micro-strain, the strain-gauge effect was linear, with an average coefficient of $\gamma = 3.04$ and standard error 0.15 (5%). No significant change in γ was observed with sheet resistance, see Figure 4.19.





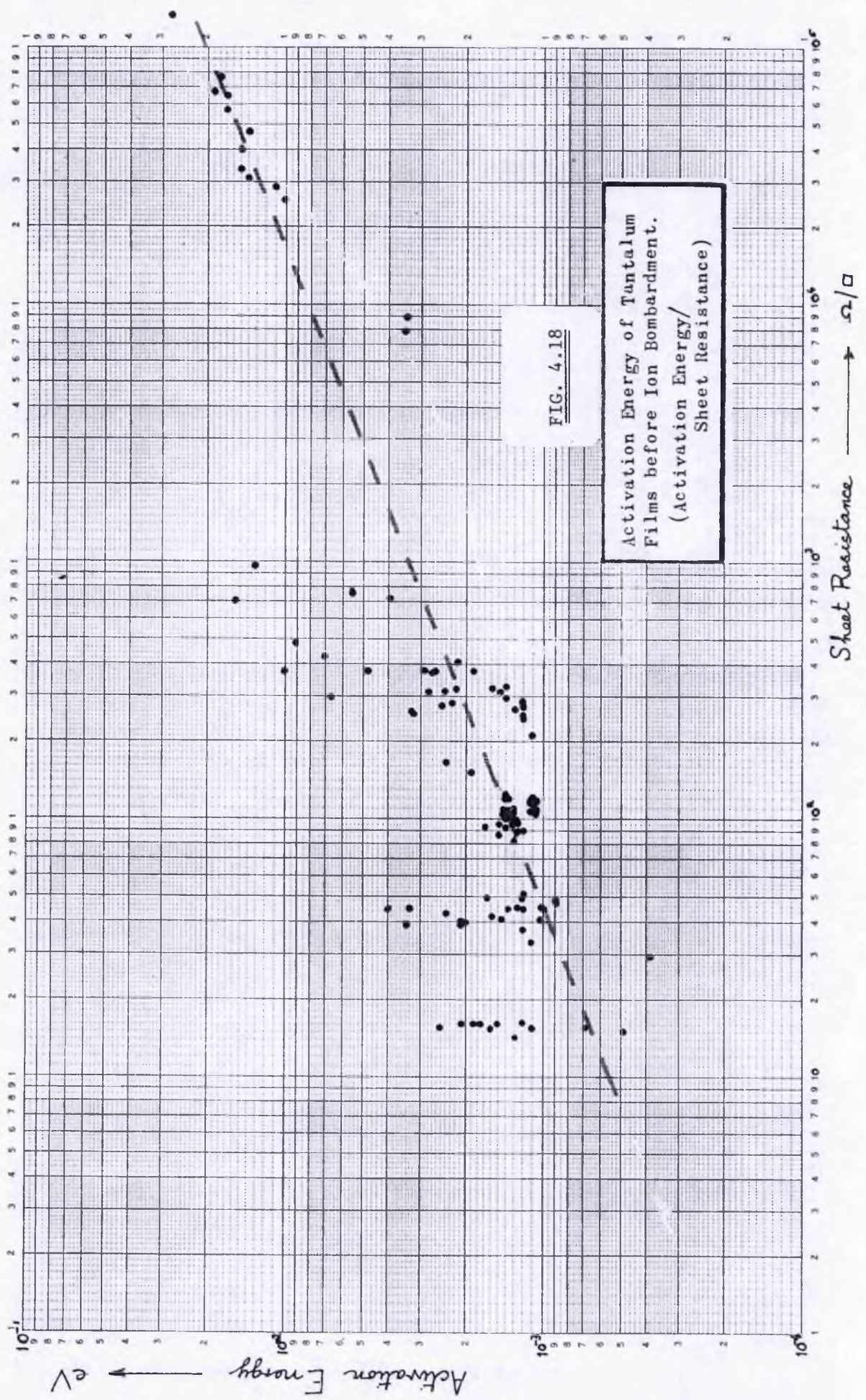
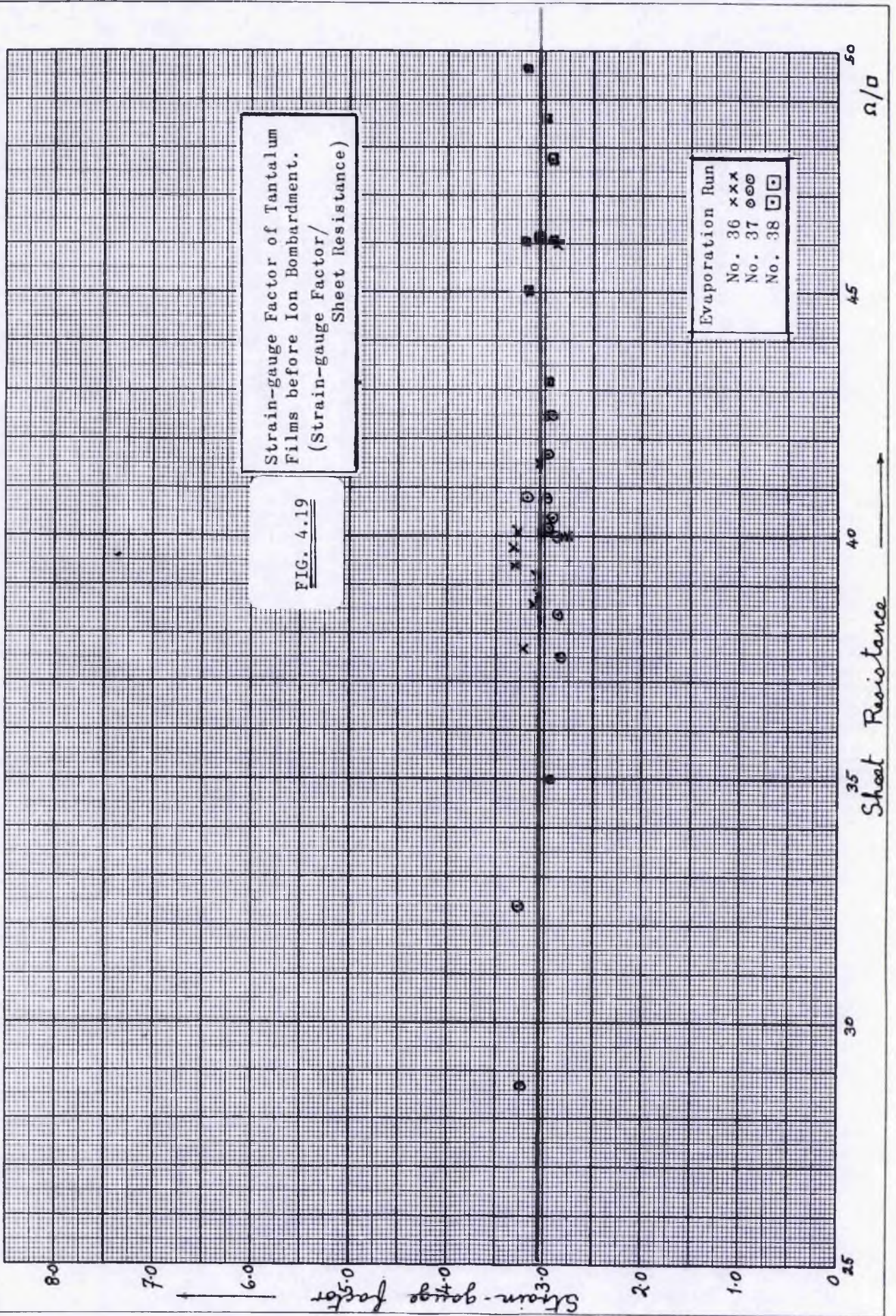


FIG. 4.18

Activation Energy of Tantalum
Films before Ion Bombardment.
(Activation Energy /
Sheet Resistance)



Strain-gauge factor was used as a measure of the repeatability of the evaporated tantalum films. Figure 4.20 shows histograms of γ for three different evaporation runs and also for the combined result. The mean γ obtained in each evaporation run is within 3% of the average value for γ quoted above for all runs combined, and generally is very much closer than this.

A cumulative frequency diagram for the same films is shown in Figure 4.21.

4.3 Bombarded Tantalum Films

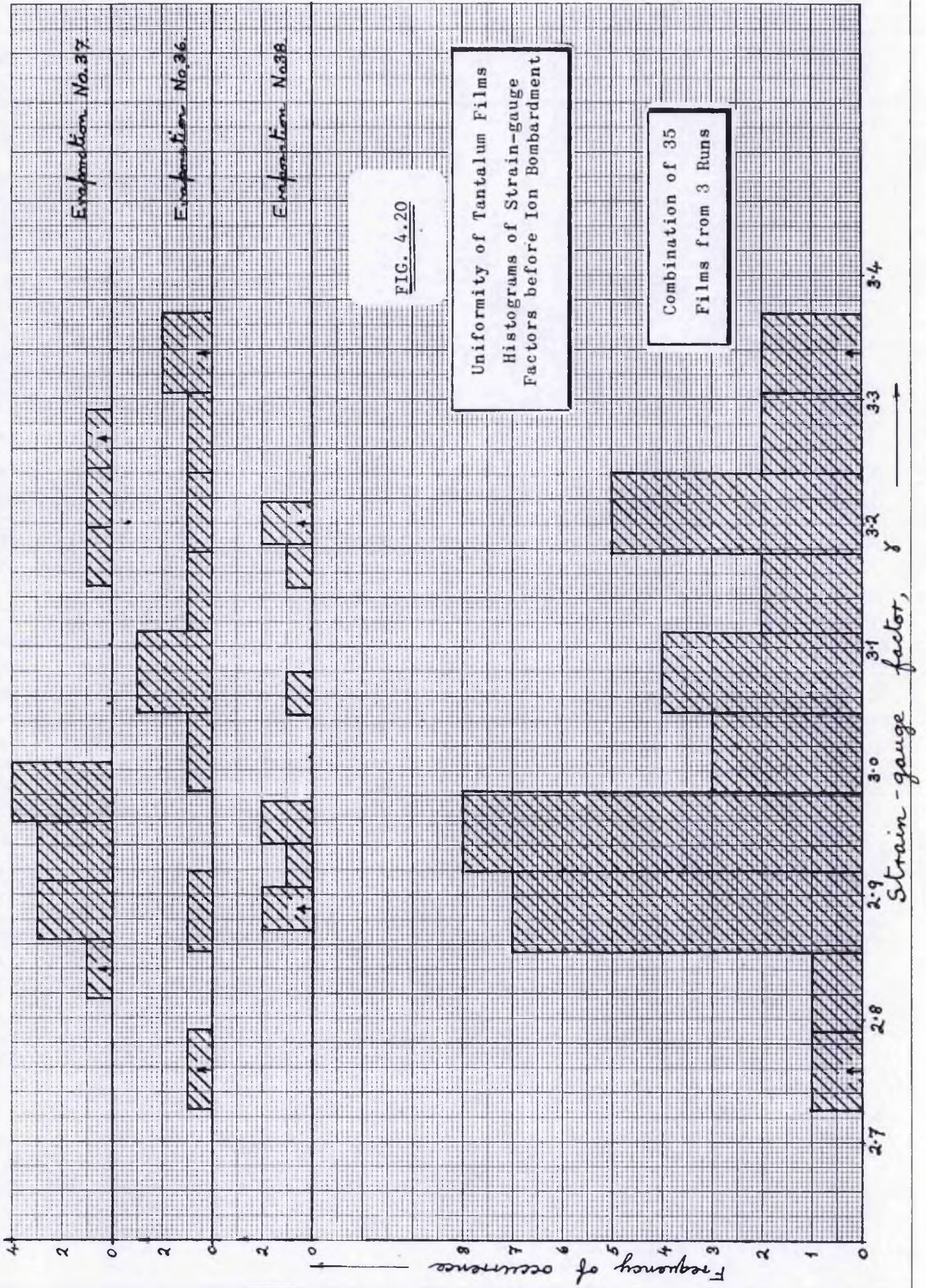
The films previously used for strain-gauge measurements were then subjected to argon-ion bombardment at 50 KeV. The projected range of such ions in β - tantalum is 210 \AA^{16} , so that nearly all the beam energy was absorbed within the film thickness, nominally of 750 \AA . The maximum dose used was 2.8×10^{17} ions/cm².

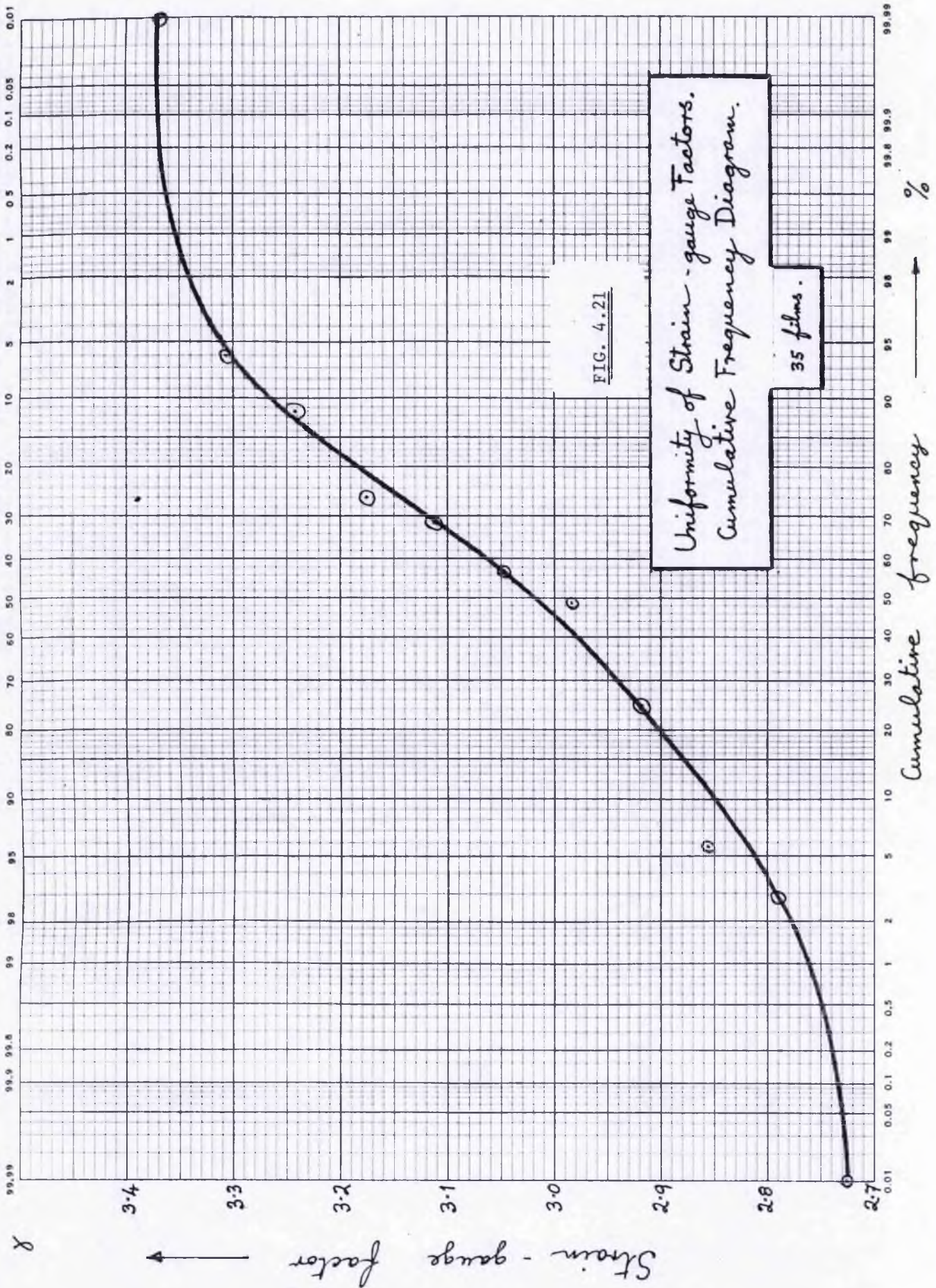
4.3.1 Resistance

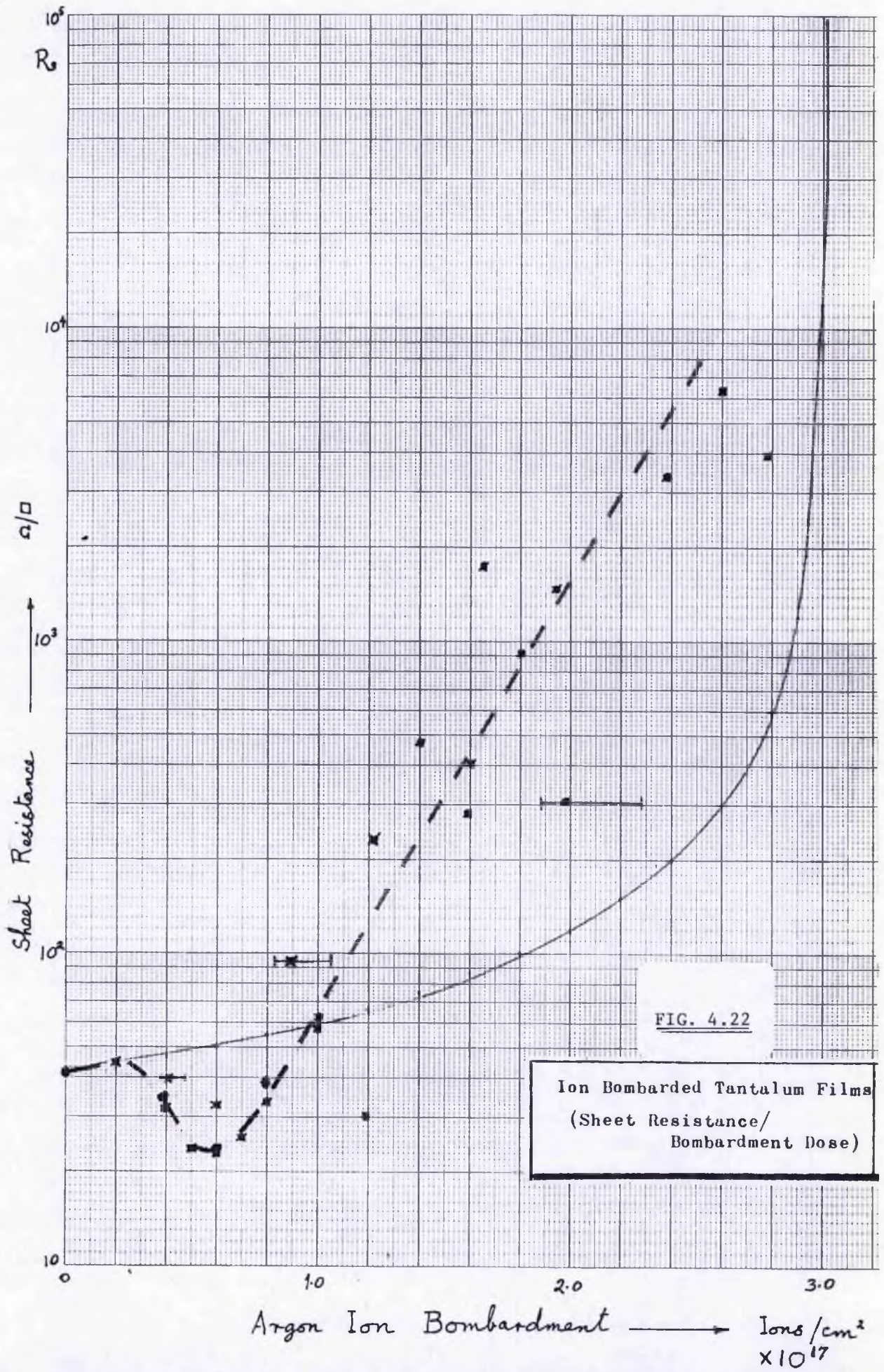
The effect of ion bombardment on the resistance of tantalum films was similar to its effect on gold films. Initially bombardment reduced the resistance but as it proceeded the resistance was increased until eventually the film became open circuit. At a dose of 0.6×10^{17} ions/cm² the sheet resistance was reduced to 50% of its original value ($40 \text{ \Omega}/\square$), and this was followed by a steady rise in resistance that reached $6000 \text{ \Omega}/\square$ at 2.6×10^{17} ions/cm² (see Figure 4.22). All the films bombarded with doses greater than 2.8×10^{17} ions/cm² were found to be open circuit upon measurement.

Shown as a continuous line is the result of simple sputtering theory (equ.2.4).

The minimum ion dose needed to remove a given film completely is related to this value, but a contributory cause of increased resistance must be the oxidation that follows breaking the vacuum. Thus it may be







assumed that a dose of 3.0×10^{17} ions/cm² is required to sputter such films right through. Assuming this and also that sputtering uniformly reduces the film thickness, values for film resistivity were calculated and plotted against ion dose; Figure 4.23. The initial resistivity was $360 \mu\Omega\cdot\text{cm}$ before bombardment, which was reduced to $160 \mu\Omega\cdot\text{cm}$ at 0.6×10^{17} ions/cm² and thereafter increased steadily towards infinity at a dose of 3×10^{17} ions/cm².

The resistivity of bulk tantalum is $13.9 \mu\Omega\cdot\text{cm}$ at 30°C .

4.3.2 Temperature Coefficient of Resistance

Figure 4.24 shows the TCR of tantalum films plotted against ion bombardment dose: TCR was measured between room temperature and $+120^\circ\text{C}$.

Films bombarded with a dose greater than 1.3×10^{17} ions/cm² showed constancy in TCR; the value being close to $-100 \text{ PPM}/^\circ\text{C}$. This was slightly less than the TCR of unbombarded films of $-150 \text{ PPM}/^\circ\text{C}$.

Between 0.2×10^{17} and 1.2×10^{17} ions/cm² bombardment raises the TCR considerably. In this region the curve forms a cusp, centred at 0.6×10^{17} ions/cm², which extends into positive TCR values. The largest TCR value encountered so far is $+420 \text{ PPM}/^\circ\text{C}$.

The resistance of four films having representative values of bombardment within the dose range established above was measured over the temperature range boiling point of nitrogen to the boiling point of water. The graphs of the non-bombarded film have already been reported, see Figures 4.16 and 4.17, but all the results accord well with measurements made only above room temperature. Table 4.3 shows the principal results for these films. Negative TCR is shown both by the non-bombarded and by the heavily bombarded film (with dose = 2.6×10^{17} ions/cm²); the TCR of the bombarded film being slightly the lesser ($-127 \text{ PPM}/^\circ\text{C}$ as against $-160 \text{ PPM}/^\circ\text{C}$ before bombardment.) The two films with bombardment doses within the transition

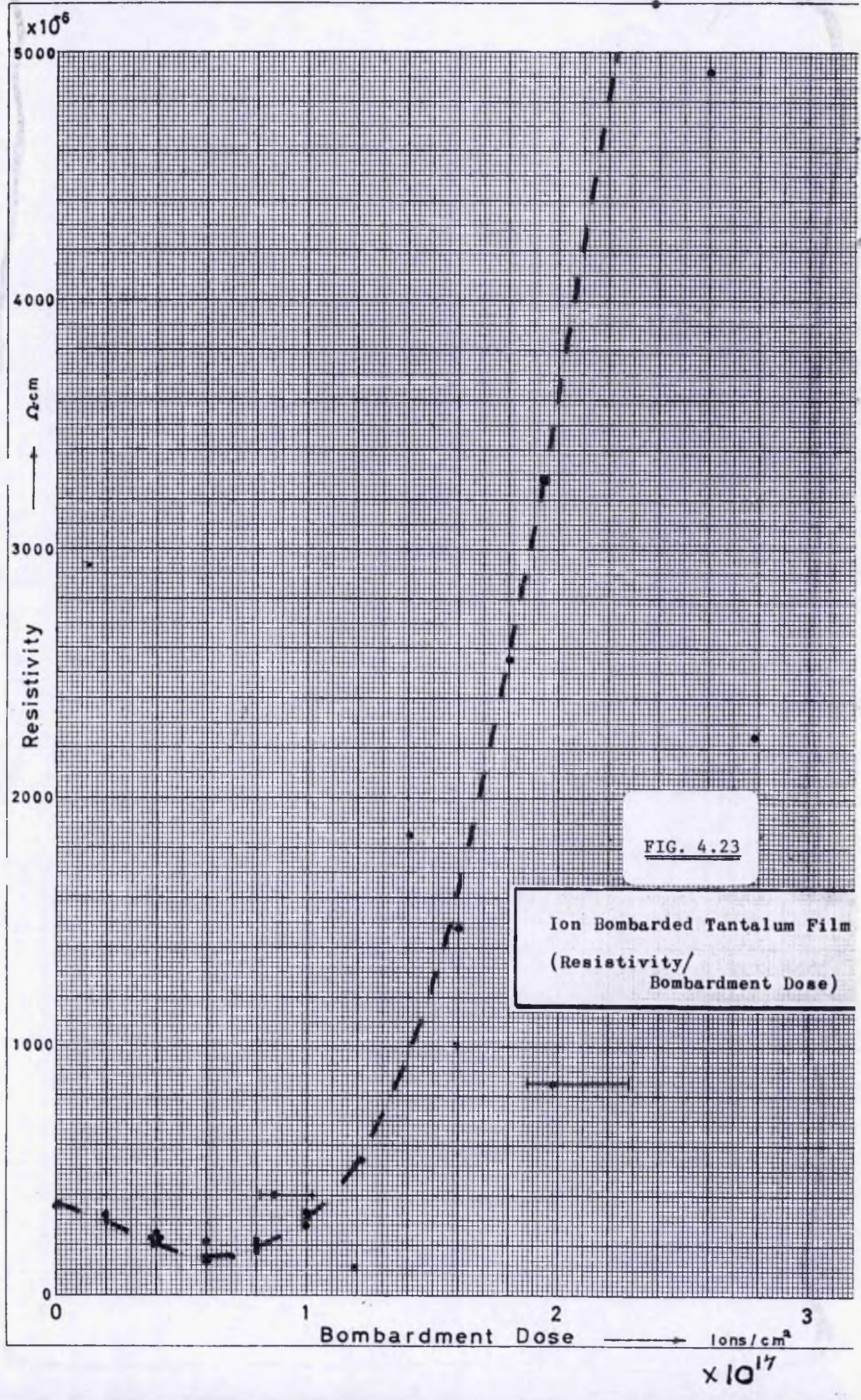
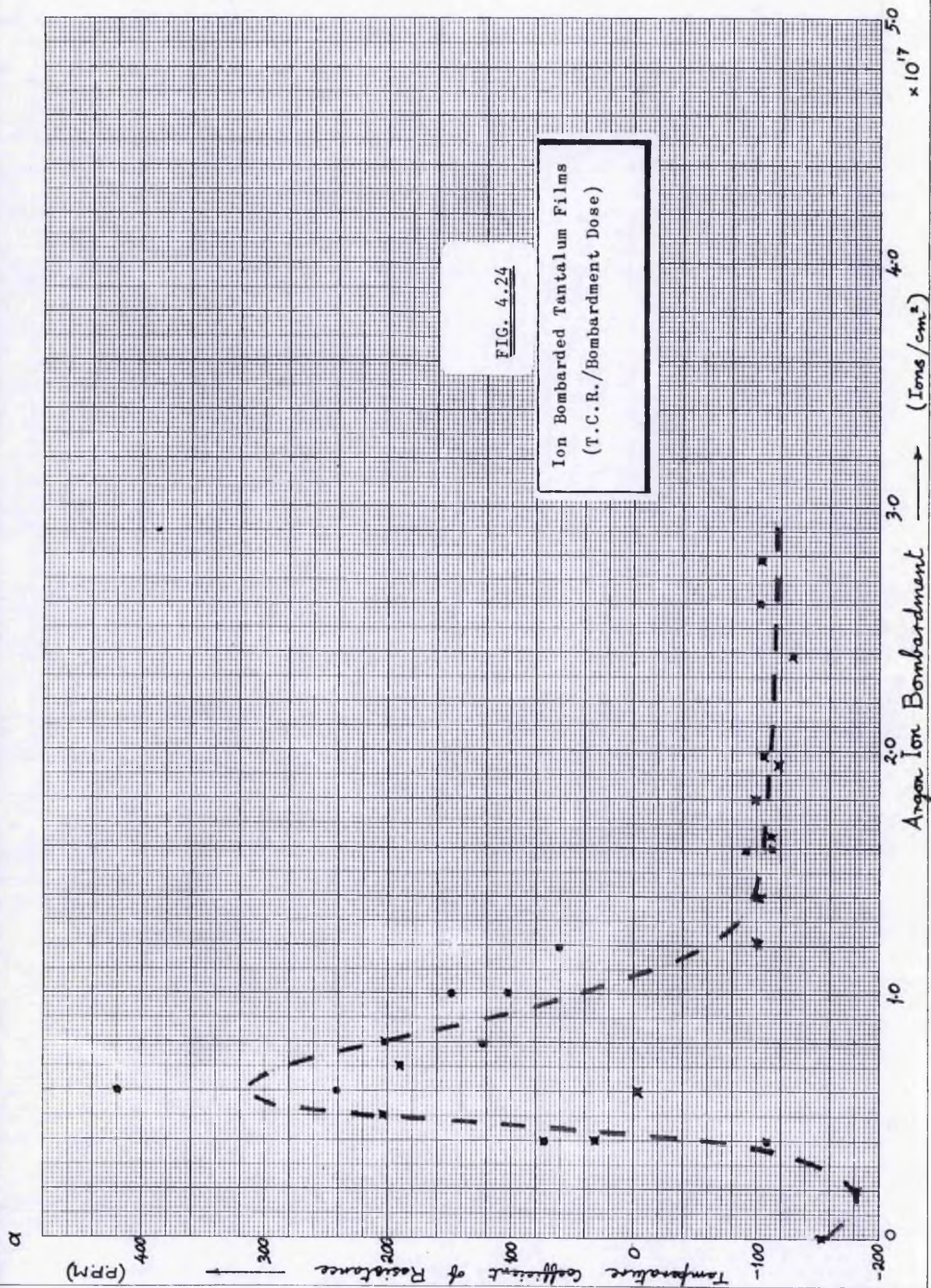


FIG. 4.23

Ion Bombarded Tantalum Film
(Resistivity/
Bombardment Dose)



region both show positive TCR with values similar to those given above in Figure 4.24., and they show considerably linearity in their resistance-temperature plots; the standard error being only about one part in 10^4 in each case. Arrhenius plots were made for these films and were found to have a linear region at each end of the temperature scale

FILM NUMBER	3706	3610	3714	3703
BOMBARDMENT DOSE ($\times 10^{17}$ IONS/CM ²)	NIL	0.6	0.8	2.6
SHEET RESISTANCE (Ω/\square 30° C)	40.8	28.6	36.8	7330
TCR (PPM/°C)	-160	+177	+175	-127
ACTIVATION ENERGY				
T < -130° C (meV)	0.25	-0.24	-0.22	0.20
T > -35° C (meV)	1.02	-1.41	-1.51	0.90

Table 4.3 Tantalum Film Resistance Measurements over the Temperature Range -195° C to + 110° C.

connected by a curved transition. In the case of the non-bombarded film three linear regions were evident (see section 4.2.2. and Figure 4.17). The ratio of the activation energy at high temperature to that at low temperature is about 4:1 for the negative TCR films.

4.3.3. Strain-Gauge Factor

The Strain-Gauge factors found for non-bombarded tantalum films averaged $\gamma = 3.04$. A plot of strain-gauge factor of films bombarded with different ion dose levels is given in Figure 4.25. It shows a transition region between doses of 0.3×10^{17} ions/cm² and 1.3×10^{17} ions/cm² in which γ increased to a maximum value of about 5.2 and then returned to about 3.0. This is followed by a linear region in which γ was reduced as bombardment proceeded, being 2.2 at a dose of 2.8×10^{17} ions/cm².

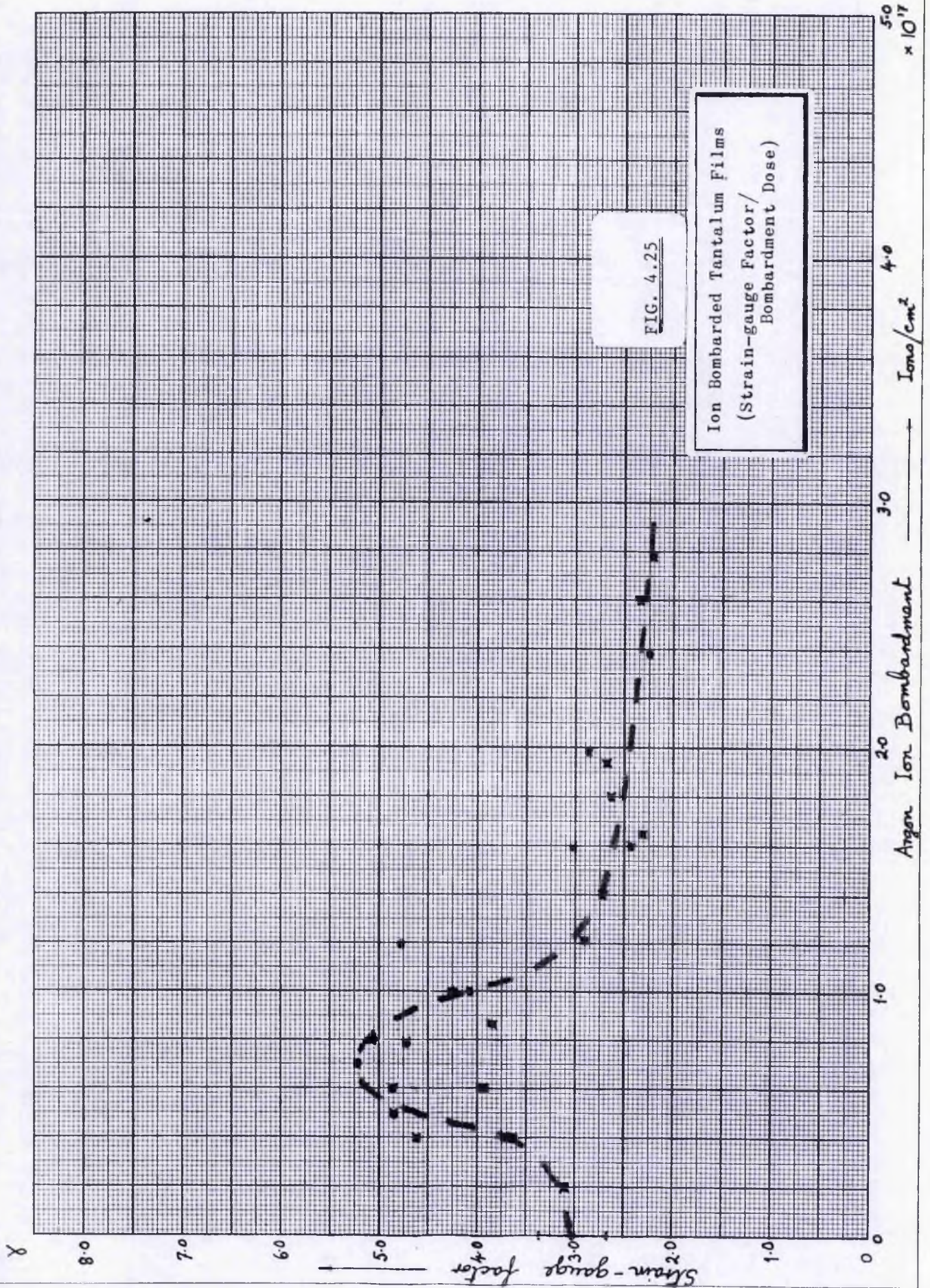
4.3.4. Annealing

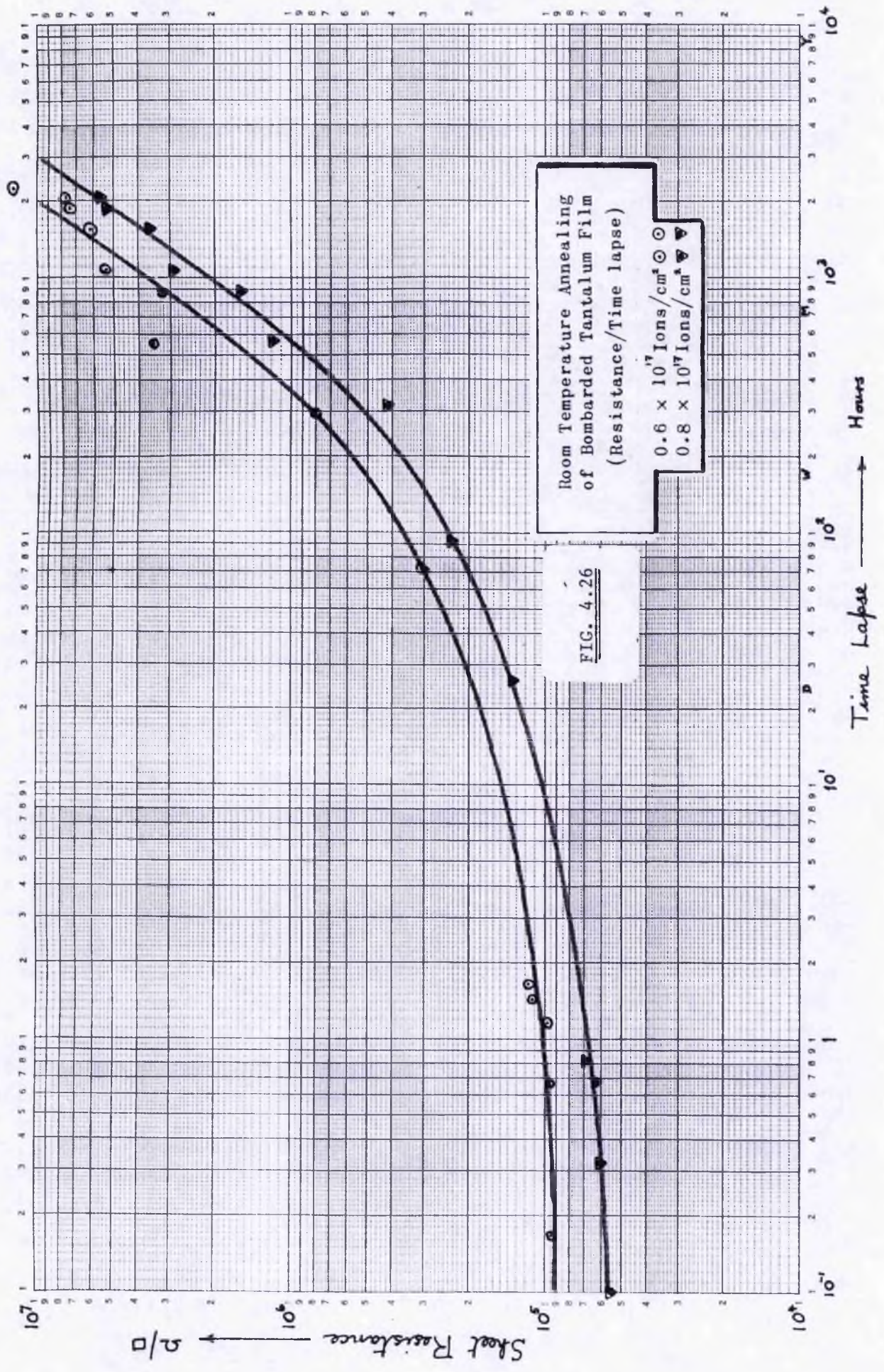
After manufacture all the films were kept for about one week (at room temperature) before any measurements could be made. Additionally this isothermal annealing period was followed by heating to about 130° C for two or three hours so as to stabilise the resistance and prevent drift during the TCR runs. The bombarded films were also heat stabilised at 130° C before use.

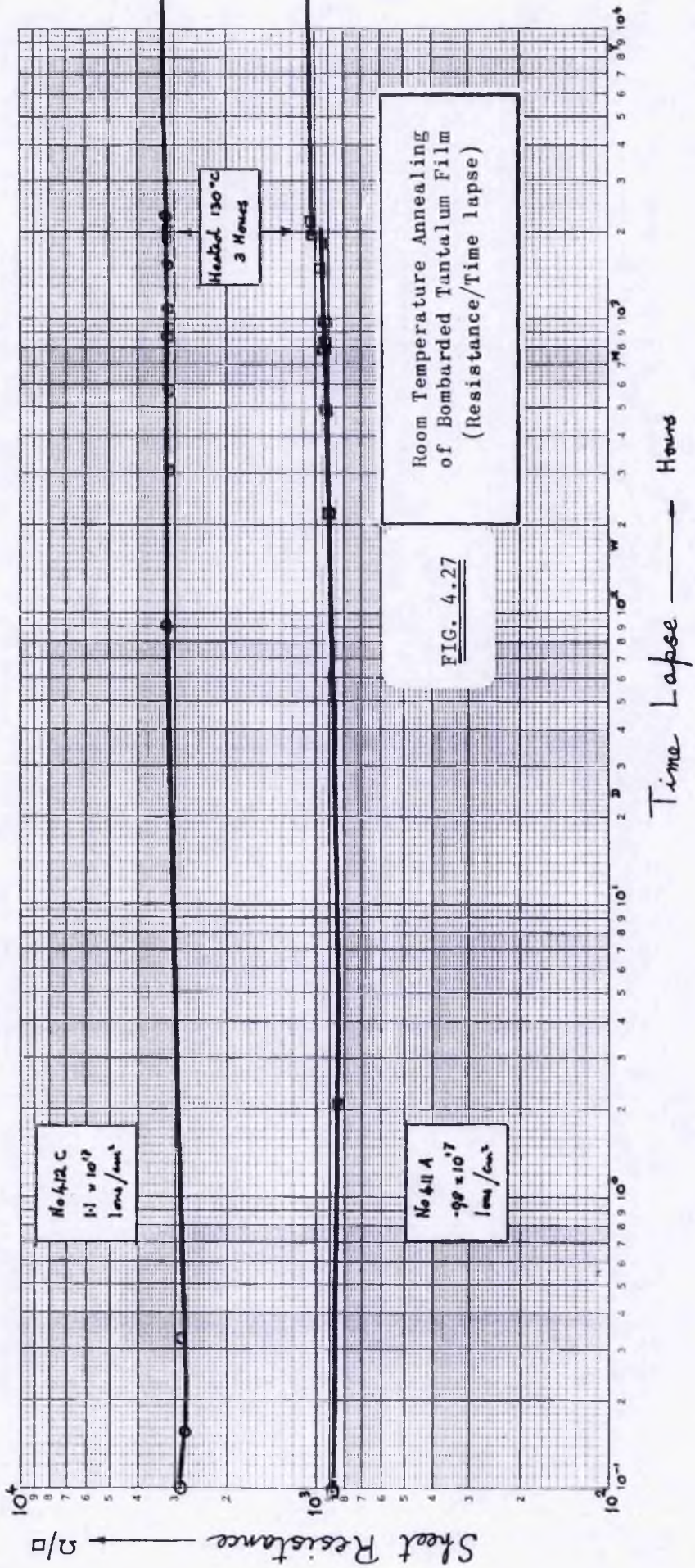
The long-term history of four bombarded films has been followed over a period of two years. This annealing was carried out at room temperature without the heat ageing, except that two of the films were annealed at 130° C after 2000 hours. These results are shown in Figures 4.26 and 4.27.

Figure 4.26 shows changes in the resistance of two films bombarded with doses in the transitional region, i.e. 0.6×10^{17} ions/cm² and 0.8×10^{17} ions/cm² respectively. The resistance continued to rise with time lapse, and after 2000 hours had increased by two orders of magnitude. This was the case for both films.

The change in resistance of the films reported in Figure 4.27 on the other hand, amounted only to a rise of 10% in 2000 hours. These films had been bombarded with 0.98×10^{17} ions/cm² & 1.1×10^{17} ions/cm² respectively.





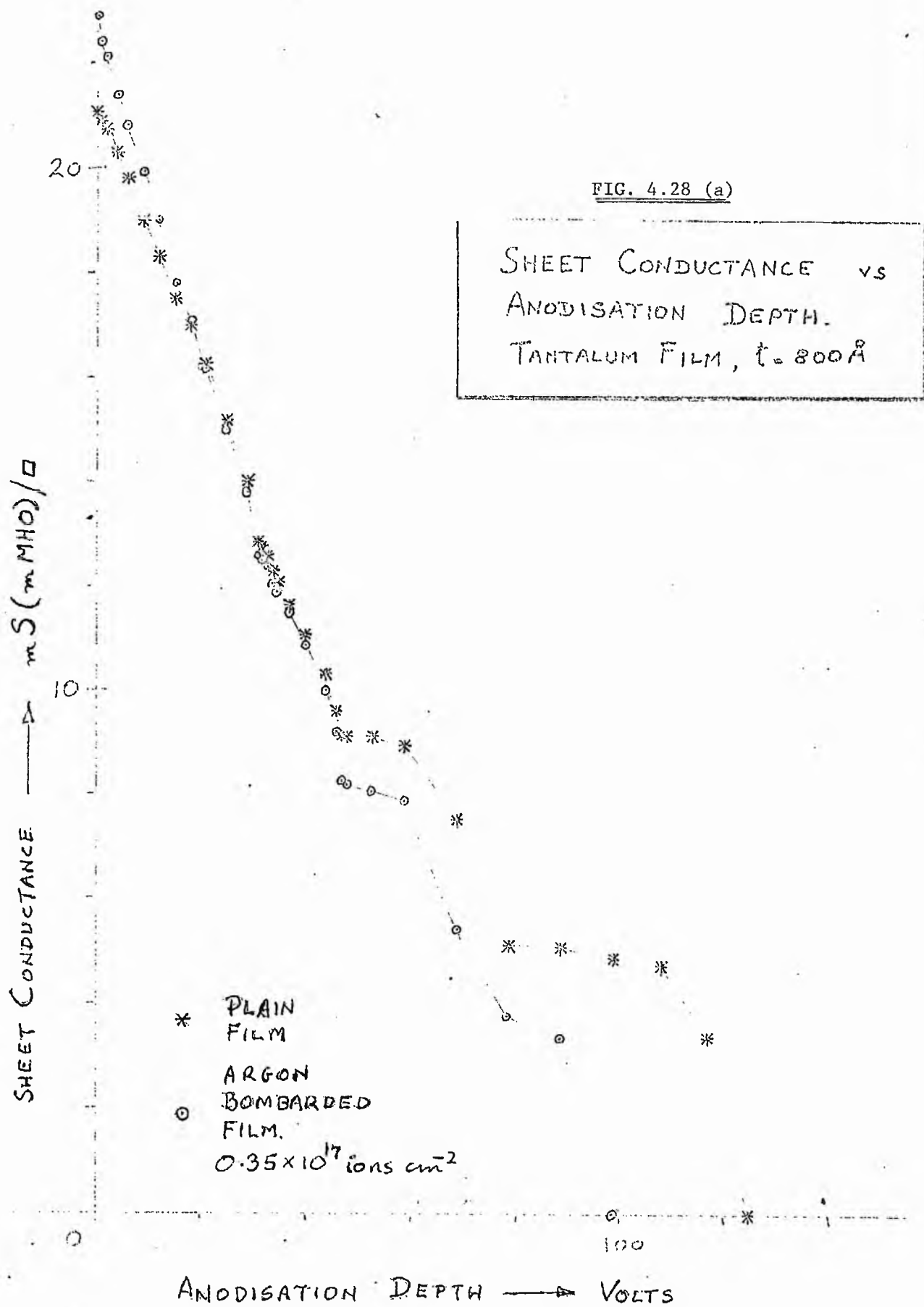


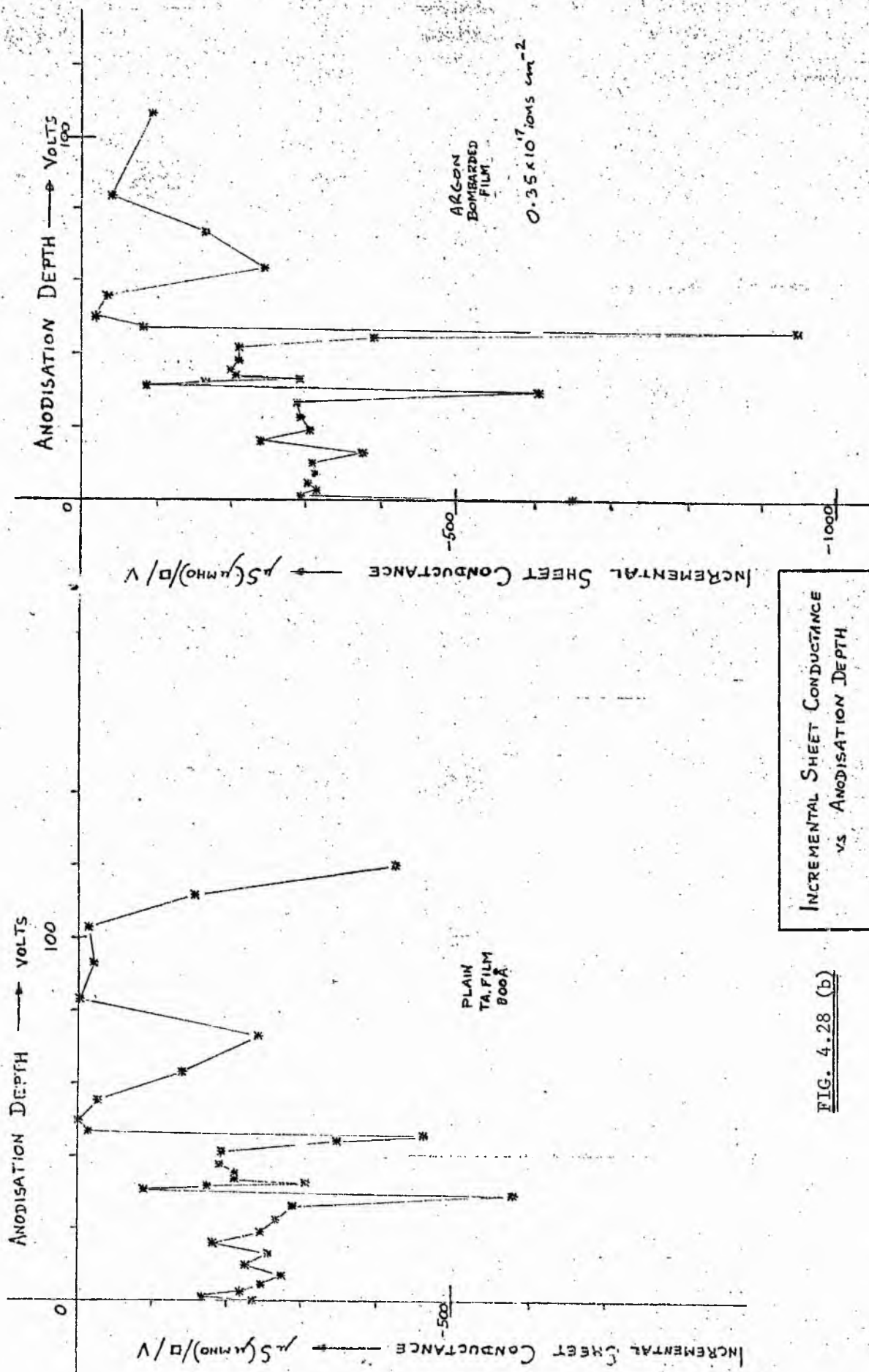
After the heat aging one film had not changed at all and the resistance of the other had increased by a further 7%. At the end of 18000 hours (two years) an additional 4% increase was observed in the resistance of both films.

4.3.5. Depth Profiling by Anodization

Shown in Figure 4.28 a, b and c are the results of incremental anodic profiling of an unbombarded tantalum film and one bombarded to a dose of 0.35×10^{17} ions/cm⁻². This was an early attempt to use this technique and there is a large scatter of values, probably resulting from drifts of resistance between measurements (time between measurement steps could vary from a few minutes to a weekend). One can tell from these measurements (assuming that the anodization rate stayed constant) that the bombarded film was thinner, both films appear to have a region of uniform conductivity near the surface with lower conductivity near the substrate. From the TCR plot it appears that the unbombarded film is fairly uniform, whilst the bombarded film has a region of very positive TCR (remember the curve is an integral one), falling to the original value as the substrate is approached).

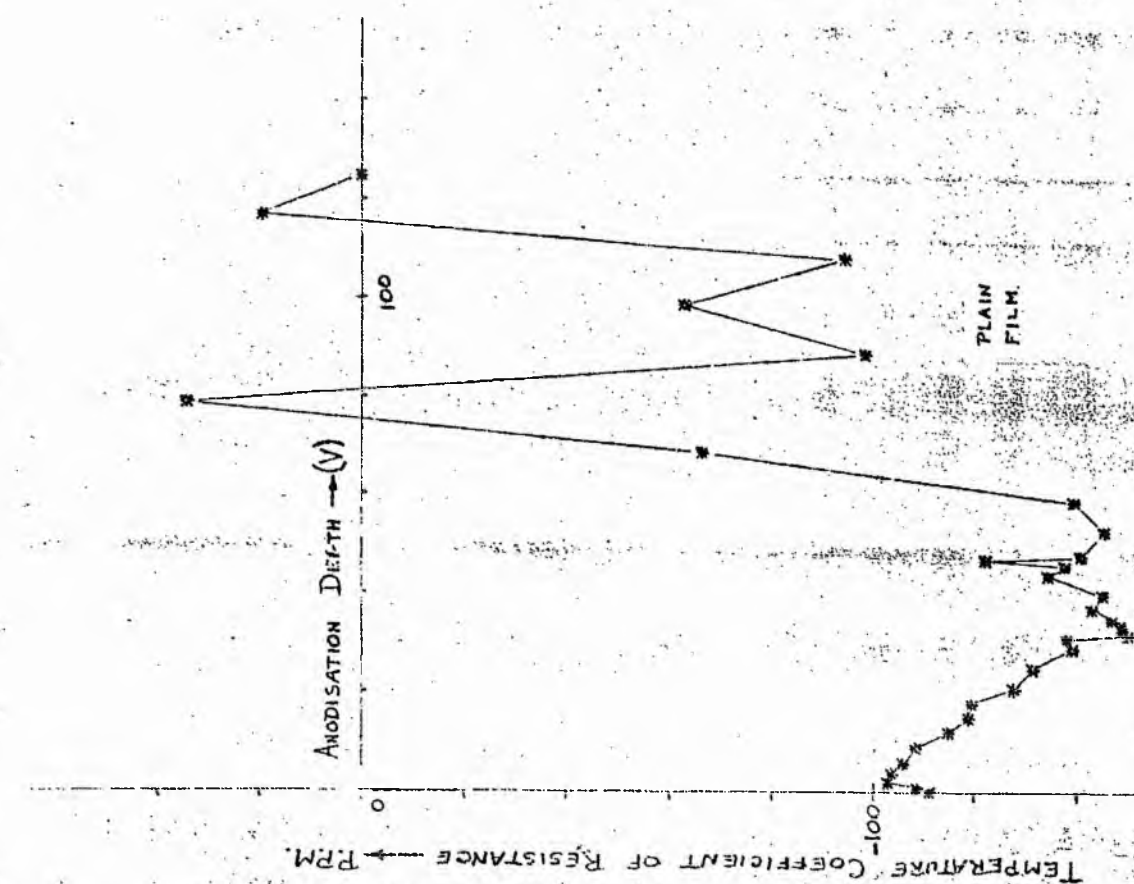
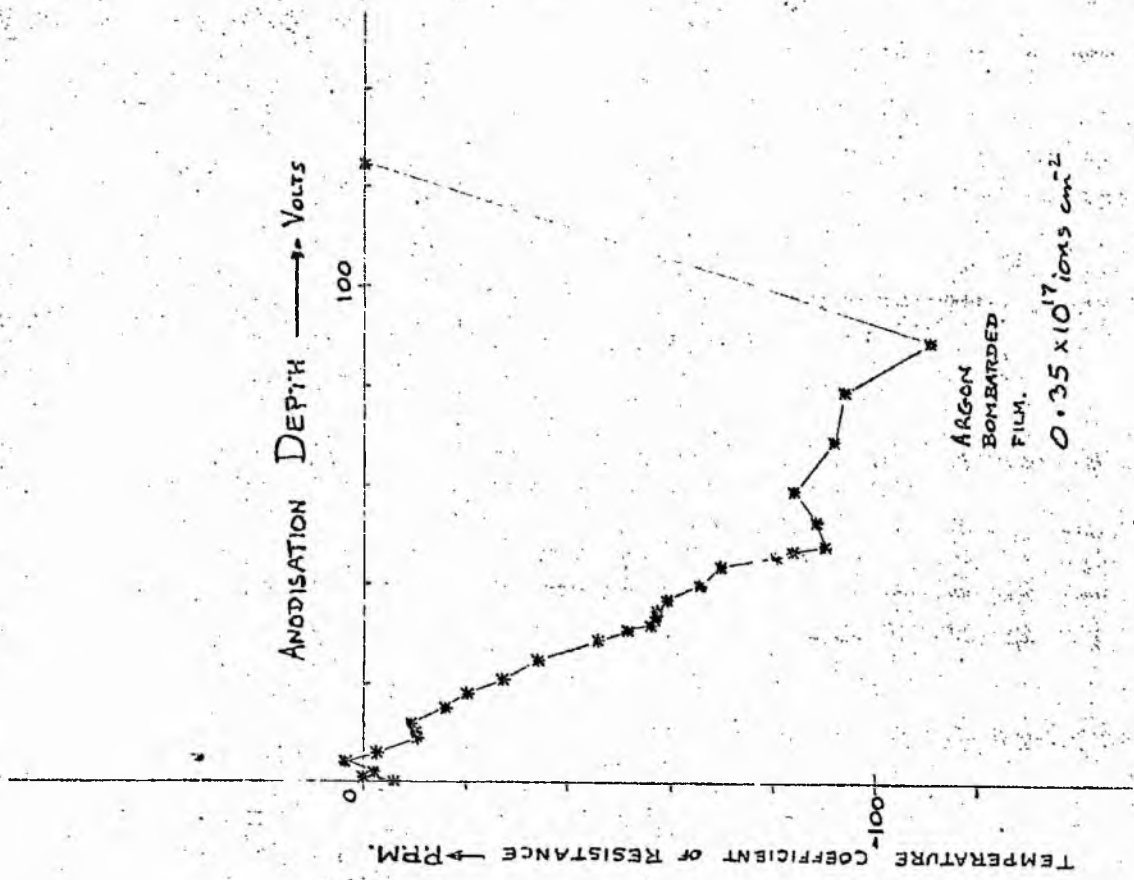
Similar results for a dose of 2×10^{17} ions/cm⁻² are shown in Figure 4.29. The bombarded film is much thinner and of lower conductivity and lower (negative) TCR, but the electrical properties appear to vary little with depth.





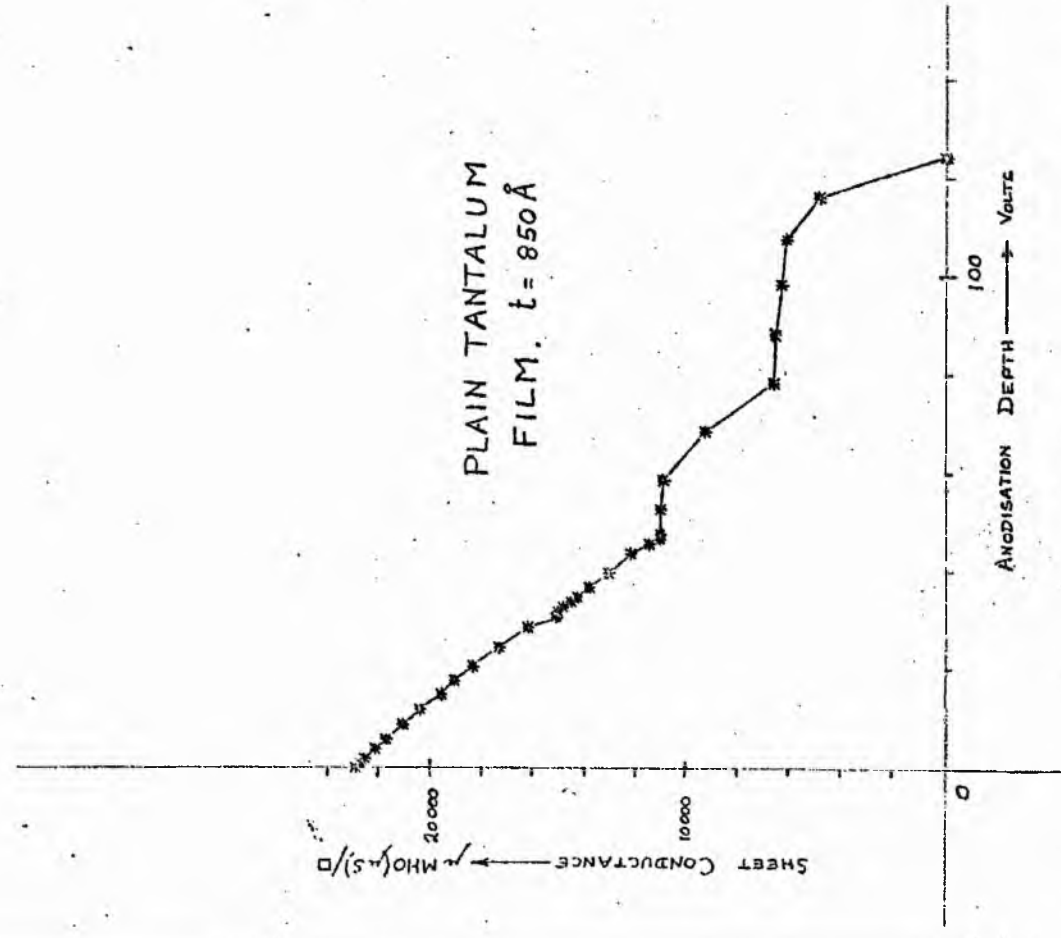
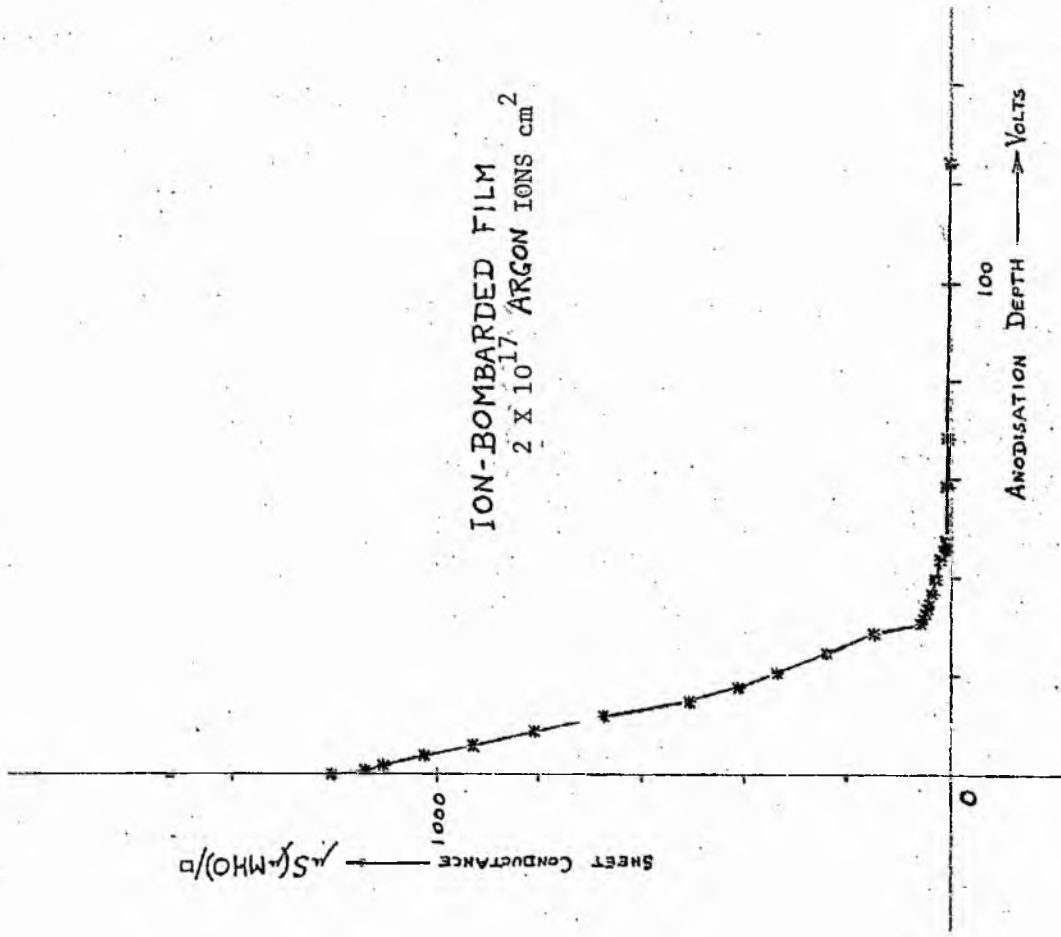
INCREMENTAL SHEET CONDUCTANCE
 vs ANODISATION DEPTH

FIG. 4.28 (b)



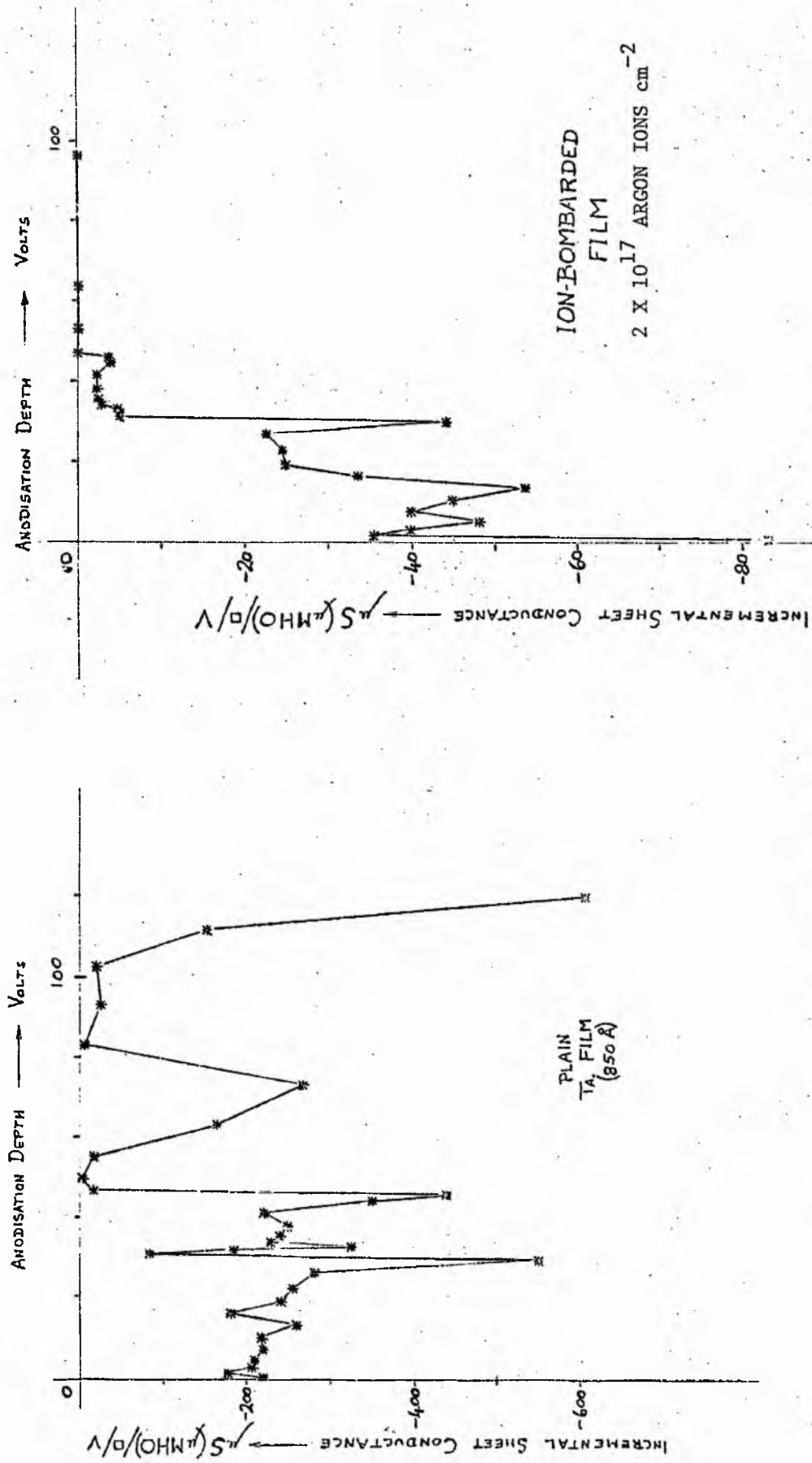
TEMPERATURE COEFFICIENT OF RESISTANCE
VS ANODISATION DEPTH.
TANTALUM FILM. $t = 800 \text{ \AA}$.

FIG. 4.28 (c)



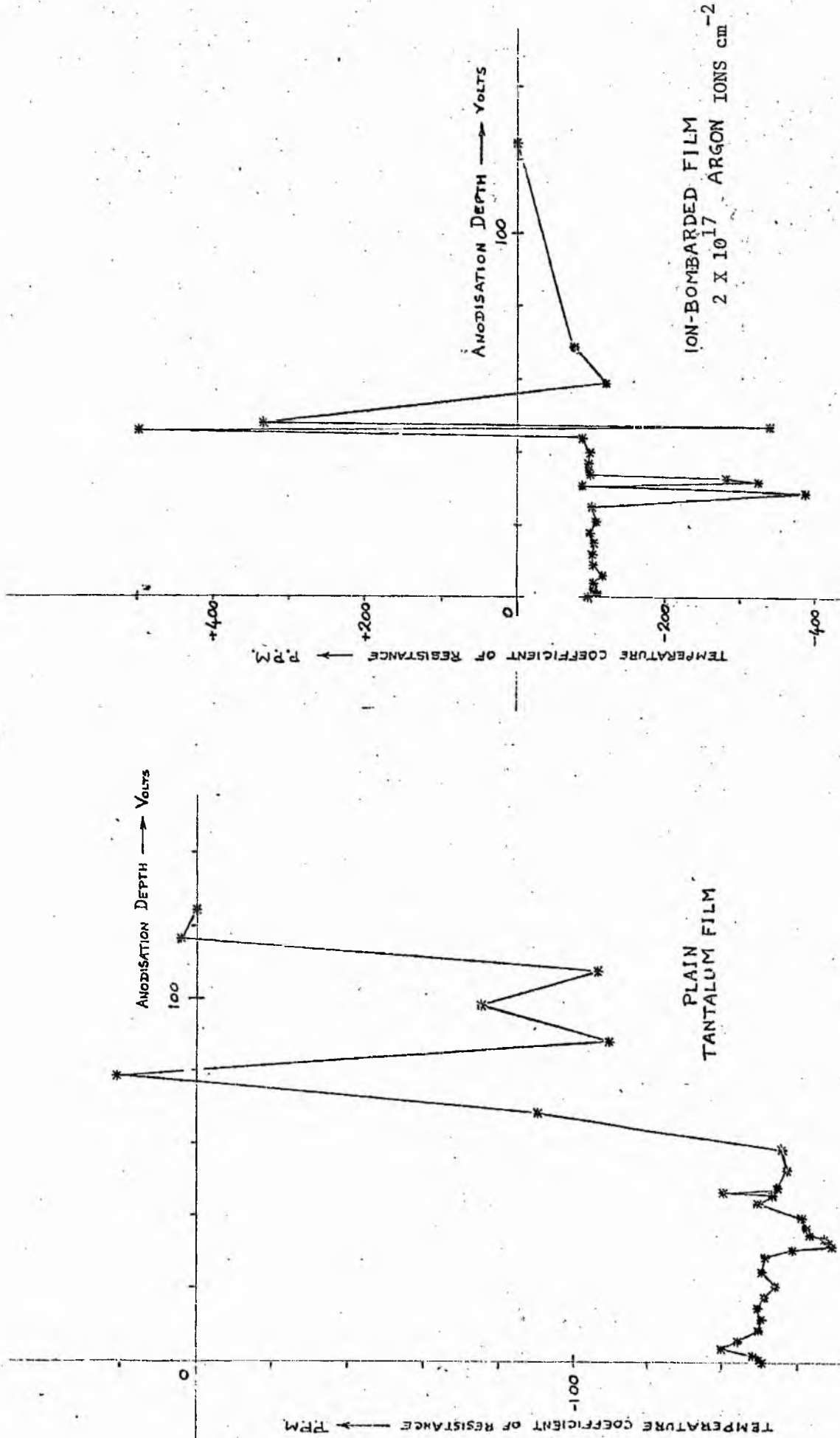
SHEET CONDUCTANCE VS
 ANODISATION DEPTH

FIG. 4.29 (a)



INCREMENTAL SHEET CONDUCTANCE
 VS
 ANODISATION DEPTH

FIG. 4.29 (b)



TEMPERATURE COEFFICIENT OF RESISTANCE
vs
ANODISATION DEPTH

FIG. 4.29 (c)

5. Discussion of the Results

5.1 Gold Films

The films examined in this section are gold films, and range in sheet resistance from $3 \Omega/\square$ to above $4 \times 10^4 \Omega/\square$. Those of low resistance are films used "as deposited" and without ion bombardment, and were nominally of 200 \AA thickness. The high-resistance films were evaporated as 200 \AA low-resistance films and were then bombarded with argon ions until the resistance was increased to the desired value.

This inert ion bombardment of low-resistance films is the most significant difference between the films described here and those of other workers. Parker and Kirnsky⁴; and Nishiura, Yoshida and Kinbara⁵ stopped short the nucleation upon reaching the particular resistance values required, and Stroud² bombarded already high-resistance titanium films ($\sim 400\Omega$) with oxygen ions - a reactive combination. Parker and Krinsky covered their films in vacuo with a thick layer of SiO_2 to prevent oxidation: all other films discussed have been left exposed to the atmosphere.

These differences make it important to examine the conduction mechanism taking place in the films, and this will be done first in the context of the strain-gauge coefficient of resistance (S.G.C.R.)

5.1.1 Strain-Gauge Coefficient (γ)

It has been shown (eqn.2.6) that the strain-gauge factor of a thin film carried on a substrate should be

$$\gamma = 1 + \sigma_s + \frac{\sigma_f (1 - \sigma_s)}{(1 - \sigma_f)} + \frac{d\rho/\rho}{d\ell/\ell}$$

where σ_s and σ_f = Poisson's ratio for substrate and film, respectively,

ρ = the film resistivity,
 l = the longitudinal dimension of the film,
 and $\epsilon = dl/l$ = the applied strain.

Large values of gauge-factor (γ) may only be expected for correspondingly large rates of change of resistivity (ρ) with strain (ϵ). Two possible conduction mechanisms may be postulated for such films - an activated tunnelling mechanism whereby charge carriers cross the dielectric regions in an island structure of metal, or metallic conduction with restriction (due to edge effects) of the electron mean-free-path.

5.1.2 Tunnelling

An implication of this type of electrical conduction is the relation between γ and sheet resistance. Parker and Krinsky⁴ show (by measurement) that for gold γ rises from a minimum value around 1.1, to 100 as the resistance changes between $10^2 \Omega/\square$ and $10^5 \Omega/\square$, and this is attributed to an increase in the separation of metallic islands in the film. Such dependence of γ on R_s is not found to be the case here, however, since γ remained constant for all resistance values measured, i.e. up to $4 \times 10^4 \Omega/\square$. Parker and Krinsky derive the following relation between S.G.C.R. (γ_t) and particle separation (s)

$$\gamma_t = 1.03 \xi_0^{\frac{1}{2}} s$$

where ξ_0 is an effective electron barrier potential lying between 0 and ϕ (the metal work function).

Working backwards, to estimate the particle separation implied by the measured values of γ (average = 2.58 at 30°C), and assuming $\xi_0 = \phi = 4.9$ eV for gold (P and K's choice),

$$\text{Island separation} = \frac{\gamma_t}{1.03 \xi_0^{\frac{1}{2}}} = 1.13 \text{\AA}.$$

This separation is less than the mean atomic spacing for bulk gold ($= 2.57 \text{ \AA}$, see section 2.2), and lends support for the view against tunnelling as the conduction mechanism of the films reported here.

5.1.3 Metallic Conduction

This model is based on the assumption that conduction within the film substantially follows the value of resistivity found in bulk material, but that the proximity of the film surfaces severely limits the mean-free-path (m.f.p.) of the conduction electrons. Parker and Krinsky give the m.f.p. in most metals as lying between 200 - 800 \AA .

We have already derived an expression for the conductivity (equ. 2.10), in terms of resistivity we have;

$$\frac{\rho_B}{\rho} = \frac{3t}{4\lambda_B} (1 + 2p) \left[\ln \left(\frac{\lambda_B}{t} \right) + 0.4228 \right]$$

To a first approximation the conductance may be taken to be proportional to film thickness, so that resistance

$$R = \frac{\text{const}}{t}, \quad \frac{dR}{R} = - \frac{dt}{t}$$

and

$$\begin{aligned} \frac{d\rho}{\rho} &= - \frac{dt}{t} \\ \frac{dt}{t} &= - \sigma_f \frac{(1 - \sigma_s)}{(1 - \sigma_f)} \frac{d\ell}{\ell} \quad (\text{See section 2.3}) \end{aligned}$$

This leads to

$$\frac{1}{\rho} \frac{d\rho}{d\epsilon} = \sigma_f \frac{(1 - \sigma_s)}{(1 - \sigma_f)}$$

Also, from equn.2.6, we have

$$\gamma = 1 + \sigma_s + \sigma_f \frac{(1 - \sigma_s)}{(1 - \sigma_f)} + \frac{1}{\rho} \frac{d\rho}{\epsilon},$$

so that S.G.C.R. becomes

$$\gamma = 1 + \sigma_s + 2\sigma_f \frac{(1 - \sigma_s)}{(1 - \sigma_f)} \text{ - - - - - 5.1}$$

Kaye and Laby⁹ give values for Poisson's ratio,

$$\text{for Gold, } \sigma_f = 0.44,$$

$$\text{and for Glass, } \sigma_s = 0.25 \text{ (average),}$$

so that the predicted value of S.G.C.R. for gold film on glass is

$$\gamma \approx 2.43$$

This is in very good agreement with the observed values, the average of which is 2.58 ± 0.07 at the 95% confidence limits.

5.1.4 The Temperature Coefficient of Resistance (α)

The positive TCR of the bombarded films also indicates that metallic conduction is the predominant mechanism.

An approximate relationship between sheet-resistivity (R_s) and the TCR of the film (α_F) is derived in section 2.5 (equn.2.11);

$$\alpha_F \approx \alpha_B / \left[\text{Ln} \left(\frac{\lambda_B R_s}{\rho_F} \right) + 0.4228 \right]$$

where α_B is the TCR of the bulk material, ρ_F is the resistivity of the film and λ_B the electron mean-free-path in the bulk. This assumes metallic conduction modified by the reduction of electron mean-free-path caused by the proximity of surfaces. It also assumes that the surface and defect scattering is temperature independent and that the film thickness is much less than λ_B .

This function is plotted in Figure 4.7 as the continuous curve for $\lambda_B = 800 \text{ \AA}$, $\rho_F = 50 \text{ } \mu\Omega\text{cm}$ and $\alpha_B = 40 \times 10^{-4} \text{ } ^\circ\text{C}^{-1}$.

The agreement is reasonable considering the approximations that have been made. Certainly the reduction of TCR with increasing sheet resistance is observed.

Over the range of sheet resistance from $30 \Omega/\square$ to $10^4 \Omega/\square$ measurement shows the TCR to be almost invariant, the average value at the 95% confidence level being,

$$\alpha = 11.75 \times 10^{-4} \pm 0.95 \times 10^{-4}$$

5.1.5 The Temperature Coefficient of Strain-Gauge Factor (β)

It has been shown (section 2.4) from the definitions of γ , α and β that the relationship $\beta = -\alpha$ should hold true.

The results for the gold films appear to confirm this prediction experimentally for the first time.

In the present work the variability of S.G.C.R. measurement makes it impracticable to arrive at reliable values of β for individual films, but, on the grounds that the S.G.C.R. is independent of sheet resistance, a simultaneous reduction of all data has been carried out. Table 4.2 and Figure 4.9 show the related S.G.C.R. and temperature values, and the result of the computations. The points on the graph of Figure 4.9 are very scattered, and the correlation coefficient is only 0.26. However, an application of Fisher's z-test¹⁵ indicates that this is a significant correlation at well above the 95% confidence level. On the assumption, therefore, that a linear relationship holds, it is possible to calculate the mean intercept and slope, and this yields values, within the 95% probability limits, of

$$\begin{aligned} \text{Intercept} &= \text{mean S.G.C.R. at } 0^\circ\text{C} = 2.71 \pm 0.07, \text{ and} \\ \text{Slope / Intercept} &= \text{mean temperature coefficient of S.G.C.R.,} \\ \beta &= -16 \times 10^{-4} \pm 11 \times 10^{-4}. \end{aligned}$$

The limits of the temperature coefficient are seen to be extremely wide, showing the uncertainty associated with its true value. However, by using this value of temperature coefficient, the mean S.G.C.R. may be adjusted to 30°C, giving

$$\text{mean } \gamma = 2.58 \pm 0.07 \quad \text{at } 30^{\circ}\text{C.}$$

The mean value for β of -16×10^{-4} compares favourably with the mean T.C.R., α , for these same films, of 11.75×10^{-4} .

5.1.6 The Structure of the Bombarded Films

The lack of change in γ with sheet-resistance and the positive value of α indicate that metallic conduction remains the predominant mechanism.

This result is remarkable when one considers that if one assumes the very high resistivity of say 1000 $\mu\Omega\text{cm}$ for the 40 $\text{k}\Omega/\square$ films this would infer, assuming a uniform film, that the film thickness is of the order of one monolayer. This would only be the case if the thickness change due to sputtering is extremely uniform.

It has been shown¹⁸ that the sputtering rate of gold varies with orientation such that the surface of a polycrystalline specimen soon becomes very irregular, with low sputtering rate grains standing proud of the rest. This would lead to a network type of structure where the gold in a thin film is removed in the areas of high sputtering rate to expose the substrate. The stereoscan micrographs (Figure 4.10) reveal that a network structure is indeed formed. It appears therefore that the resistance increases with dose due to a reduction in the number of interconnecting necks, rather than an overall decrease in film thickness. This would explain the apparently very high sputtering rate. Most material will be removed around the edge of the holes due to the oblique

incidence of the beam until there is no connecting pathway between electrodes, and it appears that with our films the gaps between the islands of gold remaining on the film are too large for tunnelling type conduction.

The mean thickness of the 'necks' must stay of the order of the film thickness until the width decreases rapidly to zero, as γ does not fall to the minimum value of about 1.0 reported for many thin evaporated films where the network structure is first established.

Another explanation of the results could be that gold is implanted into the surface of the glass to form a gold-glass cermet of a similar type to that produced by sputter deposition of a mixture of gold and glass.¹⁹

The mechanism whereby this is done is the exchange of momentum between an argon ion and a gold atom in the film. This leads to the familiar recoil sputtering with gold ejected from the surface, and also to transmission sputtering where the gold is implanted into the glass. It has been predicted theoretically²⁰ and demonstrated experimentally²¹ that the transmission sputtering yield can be considerably greater than the recoil sputtering yield.

5.1.7 The Resistance Ageing Effect

Inspection of the historical record of gold film number 038 (see Figure 4.5) shows that the resistance increased from about 100 Ω to 2000 Ω in the first 1000 hours after ion bombardment. Thereafter the resistance, though having still increased, did so at a reduced rate.

A similar effect is apparent with all films (Figure 4.6), though the resistance change was not always so severe.

Two mechanisms could be proposed to explain the increase and saturation of resistance with time and also the increase in resistance

on annealing. Either oxidation of the gold surface or diffusion of gold atoms to form clusters. Clustering would decrease the number of conducting pathways and therefore increase the resistance of the film. In the case of the films that go open circuit i.e. those with a resistance immediately after ion bombardment of greater than 10,000 Ω , a discontinuous island structure may be formed where the island separation is usually too large for a significant amount of conduction by tunnelling. The one film with a large negative TCR (see Table 4.2) may be of this type, with an island separation that is smaller than usual.

The initial incubation period of about one week, when little change occurs, is indicative of another mechanism that prevents diffusion, or oxidation, for this time, but we have no explanation for this behaviour at present.

As a chemical element gold is in Group 1 of the Periodic Table, Sub-group (b), which comprises Cu, Ag, and Au. Gold may be either uni- or trivalent and, although trivalency characterises most of its compounds, forms auric oxide - unstable and slightly acidic.¹⁷

The films reported here were all kept in closed (but not airtight) boxes, but were exposed to the atmosphere during measurement. When heated above 90°C an irreversible increase in resistance always occurred (see, for example, film 044 in Figure 4.6). Neither this nor the ageing of films at room temperature materially effected the value of S.G.C.R.

5.1.8 The Initial Drop in Resistance for doses below 2×10^{15} ions/cm⁻²

There is a dramatic reduction in the resistivity of the gold films from 24 $\mu\Omega$ cm to 12 $\mu\Omega$ cm (bulk value 2.25 $\mu\Omega$ cm) for a dose of 1.8×10^{15} ions cm⁻² (Figure 4.2). This effect has been observed by other workers^{22, 23} for other films and has been attributed to desorption of gases in the surface of the film. More recent results⁶ indicate that it is more probable that the reduction in resistivity arises from re-arrangement of the film structure; a form of radiation enhanced annealing, plus possible reduction of oxides by preferential sputtering of oxygen.²⁴

5.2 Tantalum Films

5.2.1 The as-deposited Films

Films above 200 Å in thickness appear to have fairly uniform properties with a resistivity of 370 $\mu\Omega$ cm and TCR of -150 PPM/°C (one batch has larger negative values of TCR, possibly due to a poor vacuum during deposition). These values are similar to those measured by other workers using the same deposition system.^{6, 7, 8} From their results it seems probable that the films are b.c.c. tantalum, semi amorphous (grain size <50 Å), with 40 Å of surface oxide and an average of 30 atomic per cent of oxygen dissolved in the rest of the film.

The large negative values of TCR (and high resistivity) for $t \leq 200$ Å indicate that the first 200 Å to be deposited contain larger amounts of oxygen (due to gettering), there is therefore no continuous metallic conduction path and activated tunnelling dominates the conduction process.

The measured activation energies for thicker films (Figure 4.17) are low because one is seeing the combined effects of metallic and activated tunnelling mechanisms. There appears to be evidence for at

least two thermally activated mechanisms one becoming evident above -130°C and the other above -35°C . The actual value of activation energy measured will depend sensitively on the balance between metallic and activated conduction and this will be greatly affected by the microstructure of the films. This would explain the large scatter in measured values of activation energies (Figure 4.18) which can be taken as evidence for differences in microstructure (or oxygen concentration) and the trend to higher values at higher resistivity indicates the reduced contribution from metallic conduction as the oxygen concentration increases as it will do for the thinner films.

Although strain-gauge factor has been measured over a much narrower range of sheet resistance it seems clear that there is very little scatter in the value (3.02 ± 0.15). This indicates that the strain-gauge factor is less sensitive than net activation energy (and TCR) to the microstructure of the films and is probably (like the gold films) determined by relative changes in the metallic component of conduction but not by the absolute magnitude of resistance. The value of strain-gauge factor using equn. 5.1 and the value of Poisson's ratio given for bulk tantalum (0.342)⁹ is; $\gamma = 2.03$. Some factor is acting to increase the strain sensitivity of our tantalum films. A possible mechanism to explain the higher value of γ is proposed in section 5.2.4.

5.2.2 The Effect of Argon Bombardment on Resistivity

The initial drop in resistivity towards values expected of bulk tantalum could be explained by preferential sputtering of oxygen (particularly in view of the vast difference in mass between oxygen and tantalum) and radiation enhanced diffusion of oxygen to the surface and re-arrangement of the film resulting in a more metallic phase being formed. This is supported by the microscope observations of other workers^{6,7}, by the anodic stripping results reported here (Figure 4.28), and by theoretical predictions on preferential sputtering.²⁴

After passing through a minimum at 0.6×10^{17} ions/cm² the resistivity rises with increasing dose, but the linear rate of increase cannot be fitted to a simple sputtering theory, and there is a large scatter in values. The latter could arise from differences in microstructure that affect the balance between conduction mechanisms. The slow rise could be explained if a phase of low sputtering rate is formed by reaction with the substrate, the oxygen content increasing with increasing ion dose.

5.2.3 Changes in TCR as a result of Argon Bombardment

The TCR reaches a peak positive value at the same dose as the minimum in resistivity and the peak value measured (+ 400 PPM per °C) approaches the value one would expect from a very pure tantalum film²⁵. This fully supports the ideas of purification and re-arrangement of the film at low doses as detailed in the last section.

It is curious that although a similar swing to a large positive TCR has been seen for in-situ measurements on oxygen bombarded tantalum films⁶, no such effect has been seen for in-situ measurements for argon bombardment⁶, or for later measurements (taken after removal from the target chamber) on oxygen bombarded films⁸.

It must however be pointed out that the latter measurements were made on annealed (5 hr. at 250°C) films whereas the results presented here are for films which are only heated to 120°C at the start of the TCR run. Thus the apparent discrepancy could arise if the metallic phase leading to low values of resistivity and high positive TCR is a metastable phase, which changes to a more stable (oxide?) phase on annealing at higher temperatures. The wide scatter of values in the region of the TCR peak would be explained if some annealing takes place at room temperature (see section 5.2.5 below) as the time before determination of the TCR varies from sample to sample.

At high doses there is little scatter in TCR values indicating that a stable phase has been formed, and the value of TCR is similar to that measured by other workers.⁶

5.2.4 Changes in Strain-Gauge Factor as a Result of Argon Bombardment

The strain-gauge factor peaks at a similar dose to the peak in TCR and values of γ as high as 5.2 have been observed. Once more a wide scatter in the values is observed. It is not clear why this structure has a strain sensitivity greater than that of 2.03 predicted using the bulk value of Poisson's ratio. The highly metallic properties indicate that a cermet film is not formed in this case. One could postulate that when the film is strained the phase structure is such that the number of metallic conducting pathways is reduced so increasing resistance whilst retaining the dominance of metallic conduction. If this is so then bombardment induced changes in microstructure seems to enhance the effect. This process would have to be reversible as no hysteresis is observed.

At high doses where TCR measurements indicate that a stable phase is formed the gauge factor approaches the predicted value of 2.

5.2.5 Resistance Ageing

The results of ageing tests (Figures 4.26 and 4.27) support fully the models for film structure proposed above. At doses where a supposedly metastable metallic structure is formed the resistance increases by up to two orders in magnitude after two years at room temperature. The most probable mechanism for the disappearance of the metallic phase is by reaction with the oxygen that remains below the metal phase near the film/substrate interface. This model is supported by the results of other workers.²⁶

At higher doses where it is proposed that a stable structure is

formed by reaction with the substrate, an extremely stable structure is indeed formed which is little affected by annealing at 130°C. These models are also supported by the results of anodic profiling.

It seems clear that while the effects of argon bombardment of gold films can be explained in terms of sputter etching, the mechanisms that determined the observed changes in electrical properties of tantalum films are changes in phase structure and chemical effects and reaction such as preferential sputtering of oxygen with the glass substrate.

6. Conclusions

6.1 Argon Bombardment of Gold Films

2, 4, 5, 22, 23

Many workers have examined the strain-gauge properties of gold film prepared by thermal evaporation to have sheet resistances of up to a few thousand Ω/\square . Gauge factors rising to ~ 100 have been measured, coupled with a negative temperature coefficient of resistance (TCR). These factors have been satisfactorily explained on the basis of electrical conduction by tunnelling across the boundaries of islands of metal.

The films examined in this thesis do not, however, exhibit the same behaviour even though the sheet resistances cover much the same range ($3 \Omega/\square$ to $4 \times 10^4 \Omega/\square$). The strain-gauge factor (S.G.C.R.) has been found to be invariant, and to have an average value of 2.58. The TCR has been found to be positive, and with a constant value of about 11.75×10^{-4} . The temperature coefficient of S.G.C.R. has been measured as -16×10^{-4} . All these values are inconsistent with electron tunnelling but they are consistent with metallic electrical conduction, the resistivity being presumed to have the bulk value modified by a reduction of the electron mean-free-path.

This fundamental divergence in results can only be attributed to the method of film preparation. Although the films of the other workers were also prepared by condensation at room temperature from thermally evaporated gold (the same method as was used in the present work), evaporation of the films reported here was not stopped on reaching the desired resistance, but was continued down to resistances of $\sim 5 \Omega/\square$. The films were then bombarded with argon until the resistance had been increased to the desired value. By this means the structure of the film had been made similar in effect to that of bulk metal, a feature which is evidently retained after ion bombardment.

The effect of argon bombardment of thin gold films appears to be the formation of a 'network' structure where metallic conduction is retained.

When the film eventually becomes discontinuous either by further bombardment, or by processes taking place during ageing, the island separation appears to be too great for conduction by tunnelling.

The increasing resistance during bombardment is attributed to a reduction in the number of connecting pathways rather than a reduction in their size.

Exposure to the atmosphere results in an irreversible increase in resistance. The initial 'incubation' during ageing and the ageing mechanism have yet to be fully explained.

The film quality as determined by resistivity is improved for low doses (below one quarter of the dose required to make the film go open-circuit).

6.2 Argon Bombardment of Tantalum Films

The as deposited films have electrical properties consistent with conduction by a combination of metallic and activated tunnelling mechanisms. The latter component appears to increase in importance as the oxygen content of the film increases, and there is evidence for at least two activation processes.

The strain-gauge factor appears to be determined by changes in the metallic component but is higher than predicted from bulk metal values, possibly as a result of strain altering (reversibly) the number of conducting pathways in the film, thereby changing the resistance, but not the conduction mechanism. Bombardment with argon results first in a change in electrical properties towards those expected of very pure tantalum films. This is thought to be the result of preferential sputtering of oxygen, radiation enhanced diffusion of oxygen to the surface and re-arrangement of the film to form large islands of b.c.c. tantalum. The strain-gauge factor is increased to up to 5.2 at this point, possibly because the change in phase structure has increased the effect of the conducting path way mechanism proposed above.

The tantalum phase appears to be metastable and the resistance increases markedly with time at room temperature, probably as a result of reaction with oxygen that exists near the film/substrate interface.

For higher doses of argon it appears that a single phase, stable, low sputtering rate compound is formed, probably by reaction with the glass substrate. This compound has a mixture of activated and metallic conduction with a strain-gauge factor (probably determined by the metallic component) close to that predicted for a tantalum film on a glass substrate.

To summarise the overall result of this work it appears that the phenomenon that dominates the electrical properties of argon ion bombarded

gold films is that of sputter etching, whilst for tantalum (plus oxygen) films changes in phase structure and chemical effects dominate.

7. References

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8. Appendices

- 1 COMPUTER PROGRAMMES
 - 1.1 Computation of slope of line through strain gauge data
 - 1.2 "Free" formal input of decimal data from paper tape
 - 1.3 Subroutine to calculate to "best fit" straight line
 - 1.4 Graph plotting

- 2 EXTRA RESULTS AND CORRELATIONS (OF USE TO OTHER WORKERS)
 - 2.1 Bending ring for 1" strain gauges
 - 2.2 R_S vs. thickness (T_a) showing different evaporation runs
 - 2.3 TCR vs. R_S (T_a) showing different evaporation runs
 - 2.4 TCR vs. thickness (T_a) showing different evaporation runs
 - 2.5 TCR vs. thickness (T_a) showing different evaporation runs
 - 2.6 TCR vs. activation energy (T_a)
 - 2.7 Activation energy vs. thickness (T_a)
 - 2.8 Activation energy vs. thickness (T_a) showing different runs
 - 2.9 Activation energy vs. R_S (T_a) showing different runs
 - 2.10 R_S vs. $(\text{time})^{\frac{1}{2}}$ (T_a)
 - 2.11 R_S vs. $(\text{time})^{\frac{1}{2}}$ (T_a)
 - 2.12 SGF vs. R_S (T_a)
 - 2.13 SGF vs. TCR (T_a)
 - 2.14 Relative resistance vs. R (after 1000 hrs.), Au films.

- 3 END CORRECTIONS FOR IMPLANTED STRAIN GAUGES

PAGE 1 ENG01693

/ JOB 1001 1002 1002 *ENG01693 01

LOG DRIVE	CART SPEC	CART AVAIL	PHY DRIVE
0000	1001	1001	0000
0001	1002	1002	0001

L M00 CONFIG 8K

CAT VERSION D.M. SYSTEM V1 M00

EQUAT(PAPTZ,PAPTC)

/ FOR

NAME GRAAL COMPUTATION OF SLOPE OF LINE THROUGH STRAIN-GAUGE DATA

LIST SOURCE PROGRAM

COCS(CARD,1132 PRINTER)

COCS(PAPER TAPE)

NINE WORD INTEGERS

000	DIMENSION X(20),Y(10),NOTE(36)	GRAAL 1
000	FORMAT('0'T39'DATA'// (T 2,10F8.3))	GRAAL 2
001	FORMAT('0'T2'SPECIMEN'T11,I3,T14', TEMP = 'T24,I2T27'DEG C, 'T35'SL	GRAAL 3
001	SLOPE = 'T43,E10.3,T53', CORR COEFF = 'T69,E10.3)	GRAAL A3
002	FORMAT('0'T 6,36A2/)	GRAAL 4
003	FORMAT('1')	GRAAL 5
000	FORMAT(36A2)	GRAAL 6
	ILP=3	GRAAL 7
	ITR=4	GRAAL 8
	READ(ITR,4000)NOTE	GRAAL 9
	WRITE(ILP,3002)NOTE.	GRAAL 10
	NUM=20	GRAAL 11
	CALL GREAD(NUM,X)	GRAAL 12
	IF(X(1))2,1,3	GRAAL 13
	WRITE(ILP,3003)	GRAAL 14
	CALL EXIT	GRAAL 15
	WRITE(ILP,3000)(X(I),I=1,19,2),(X(I),I=2,20,2)	GRAAL 16
	INO=IFIX(X(1))	GRAAL 17
	ITEMP=IFIX(X(2))	GRAAL 18
	K=0	GRAAL 19
	DO 6 I=4,20,2	GRAAL 20
	IF(X(I-1))5,4,5	GRAAL 21
	IF(X(I))5,6,5	GRAAL 22
	K=K+1	GRAAL 23
	X(K)=X(I-1)	GRAAL 24
	Y(K)=X(I)	GRAAL 25
	CONTINUE	GRAAL 26
	CALL GRYLL(K,X,Y,A,B,C)	GRAAL 27
	WRITE(ILP,3001)INO,ITEMP,A,C	GRAAL 28
	GO TO 1	GRAAL 29
	END	GRAAL 30

FEATURES SUPPORTED

NINE WORD INTEGERS

COCS

REQUIREMENTS FOR GRAAL

COMMON 0 VARIABLES 112 PROGRAM 308

AGE 1 ENG01720

JCB 1001 2001 2001 *ENG01720

JG DRIVE	CART SPEC	CART AVAIL	PHY DRIVE
0000	1001	1001	0000
0001	2001	2001	0002
		1002	0001

M11 CCONFIG 8K

CAT VERSION D.M. SYSTEM V1 M02

FCR FREE FORMAT INPUT

NAME GREAC

LIST SOURCE PROGRAM

ONE WORD INTEGERS

SUBROUTINE GREAC(IPR,N,OUT)

GREAC 1

DIMENSION OUT(1),INPUT(72),KIND(15)

GREAC 2

DATA KIND/' ','/','0','1','2','3','4','5','6','7','8','9','.',',','-'

GREAC 3

1,','/

GREAC 4

) FORMAT(72A1)

GREAC 5

) READ INPUT RECORD.

GREAC 6

READ(IPR,40)(INPUT(I),I=1,72)

GREAC 7

TEST ARRAY SUBSCRIPT RANGE

GREAC 8

IS=1

GREAC 9

IF(N)19,19,1

GREAC 10

IF(N-36)3,3,2

GREAC 11

NN=36

GREAC 12

GC TC 4

GREAC 13

NN=N

GREAC 14

INITIALIZE

GREAC 15

DO 5 I=1,NN

GREAC 16

OUT(I)=0.

GREAC 17

SEPARATE THE DATA WORDS

GREAC 18

DO 18 I=1,NN

GREAC 19

SKIP LEADING BLANKS

GREAC 20

DO 51 K=IS,72

GREAC 21

DO 51 L=2,15

GREAC 22

IF(INPUT(K)-KIND(L))51,52,51

GREAC 23

) CONTINUE

GREAC 24

GC TC 19

GREAC 25

START NEW DATA WORD

GREAC 26

IS=K

GREAC 27

IF=IS+7

GREAC 28

IF(IF-72)7,7,6

GREAC 29

IF=72

GREAC 30

SIGN=1.

GREAC 31

IDP=0

GREAC 32

MM=0

GREAC 33

SEPARATE AND IDENTIFY DATA CHARACTERS

GREAC 34

DC 16 J=IS,IF

GREAC 35

DO 8 K=1,15

GREAC 36

IF(INPUT(J)-KIND(K))8,9,8

GREAC 37

CONTINUE

GREAC 30

GC TC 16

GREAC 31

IF(K-2)16,17,10

GREAC 32

IF(K-13)13,14,11

GREAC 33

IF(K-15)12,17,17

GREAC 34

PAGE 2 ENG01720

12	SIGN=SIGN*(-1.)	GREAD 35
	GC TC 16	GREAC 36
	* BUILD DATA VALUE	GREAD 37
13	OUT(I)=OUT(I)*10.+FLOAT(K-3)	GREAD 38
	IF(MM)16,16,15	GREAD 39
14	IF(IDP)142,142,141	GREAD 40
141	IDP=MM	GREAD 41
142	IDP=IDP+1	GREAC 42
15	MM=MM+1	GREAD 43
16	CONTINUE	GREAC 44
	J=J-1	GREAD 45
17	IS=J+1	GREAC 46
	* INSERT DECIMAL POINT AND SIGN	GREAC 47
	OUT(I)=OUT(I)*SIGN*(10.** (IDP-MM))	GREAD 48
	IF(IS-72)18,18,19	GREAD 49
18	CONTINUE	GREAD 50
19	RETURN	GREAC 51
	END	GREAD 52

FEATURES SUPPORTED
ONE WORD INTEGERS

* CORE REQUIREMENTS FOR GREAD
COMMON 0 VARIABLES 106 PROGRAM 370

RELATIVE ENTRY POINT ADDRESS IS 007C (HEX)

END OF COMPILATION

// DUP

*DELETE		GREAD	2001		
CART ID 2001	DB ADDR	1220	DB CNT	0018	
*STORE	WS	UA	GREAD	2001	2001
CART ID 2001	DB ADDR	1850	DB CNT	001B	

PAGE 1 E0000295

// JOB 1001 1002 1002 *E0000295

LOG DRIVE	CART SPEC	CART AVAIL	PHY DRIVE
0000	1001	1001	0000
0001	1002	1002	0001

//2 M11 CONFIG 8K

CAT VERSION D.M. SYSTEM V1 M02

// FOR STRAIGHT-LINE OPTIMISATION

*NAME GRYLL

*LIST SOURCE PROGRAM

*ONE WORD INTEGERS

SUBROUTINE GRYLL(K,X,Y,N,A,B,C,D,E,F)	GRYLL 1
DIMENSION X(1),Y(1)	GRYLL 2
A=0.	GRYLL 3
B=0.	GRYLL 4
C=0.	GRYLL 5
D=0.	GRYLL 6
E=0.	GRYLL 7
F=0.	GRYLL 8
AGGR=FLOAT(K)	GRYLL 9
IF(K-2)16,16,1	GRYLL 10
INITIALISE STORAGE	GRYLL 11
XTOT=0.	GRYLL 12
YTOT=0.	GRYLL 13
XYTOT=0.	GRYLL 14
X2TOT=0.	GRYLL 15
Y2TOT=0.	GRYLL 16
COVAR=0.	GRYLL 17
VARX=0.	GRYLL 18
VARY=0.	GRYLL 19
SUMX=0.	GRYLL 20
SUMY=0.	GRYLL 21
CENTRALISE AXES	GRYLL 22
DO 2 I=1,K	GRYLL 23
SUMX=SUMX+X(I)	GRYLL 24
SUMY=SUMY+Y(I)	GRYLL 25
XMEAN=SUMX/AGGR	GRYLL 26
YMEAN=SUMY/AGGR	GRYLL 27
DO 3 I=1,K	GRYLL 28
X(I)=X(I)-XMEAN	GRYLL 29
Y(I)=Y(I)-YMEAN	GRYLL 30
COMPUTE TOTALS	GRYLL 31
XTOT=XTOT+X(I)	GRYLL 32
YTOT=YTOT+Y(I)	GRYLL 33
XYTOT=XYTOT+X(I)*Y(I)	GRYLL 34
X2TOT=X2TOT+X(I)*X(I)	GRYLL 35
Y2TOT=Y2TOT+Y(I)*Y(I)	GRYLL 36
CONTINUE	GRYLL 37
COMPUTE SAMPLE MEAN VALUES	GRYLL 38
XBAR=XTOT/AGGR	GRYLL 39
YBAR=YTOT/AGGR	GRYLL 40
XYBAR=XYTOT/AGGR	GRYLL 41
X2BAR=X2TOT/AGGR	GRYLL 42
Y2BAR=Y2TOT/AGGR	GRYLL 43

PAGE 2 E0000295

DO 4 I=1,K	GRYLL 40
COVARIANCE	GRYLL 41
COVAR=COVAR+(X(I)-XBAR)*(Y(I)-YBAR)	GRYLL 42
VARIANCES	GRYLL 43
VARX=VARX+(X(I)-XBAR)*(X(I)-XBAR)	GRYLL 44
VARY=VARY+(Y(I)-YBAR)*(Y(I)-YBAR)	GRYLL 45
CORRELATION COEFFICIENT	GRYLL 46
C=COVAR/SQRT(VARX*VARY)	GRYLL 47
IF(N-2)5,6,6	GRYLL 48
SOLUTION BY X-REGRESSION	GRYLL 49
SLOPE OF LINE	GRYLL 50
A=VARY/COVAR	GRYLL 51
VARIANCE OF DEVIATIONS	GRYLLA51
S=(VARX-COVAR*COVAR/VARY)/FLOAT(K-2)	GRYLLB51
S.E. OF SLOPE	GRYLLC51
D=SQRT(S/VARY)	GRYLLD51
D=D*A*A	GRYLLE51
S.E. OF INTERCEPT	GRYLLF51
E=SQRT(S*(1./AGGR+YBAR*YBAR/VARY))	GRYLLG51
E=-E*A	GRYLLH51
GO TO 7	GRYLL 52
SOLUTION BY Y-REGRESSION	GRYLL 53
SLOPE OF LINE	GRYLL 54
A=COVAR/VARX	GRYLL 55
VARIANCE OF DEVIATIONS	GRYLLA55
S=(VARY-COVAR*COVAR/VARX)/FLOAT(K-2)	GRYLLB55
S.E. OF SLOPE AND INTERCEPT	GRYLLC55
D=SQRT(S/VARX)	GRYLLD55
E=SQRT(S*(1./AGGR+XBAR*XBAR/VARX))	GRYLLE55
INTERCEPT	GRYLL 56
B=YBAR-A*XBAR	GRYLL 57
IF(N-2)14,14,8	GRYLL 58
SOLUTION BY DOUBLE REGRESSION	GRYLL 59
SUM OF SQUARES OF PERPENDICULAR DISTANCES	GRYLL 60
PD2=0.	GRYLL 61
DO 9 I=1,K	GRYLL 62
PD2=PD2+(A*X(I)-Y(I)+B)*(A*X(I)-Y(I)+B)	GRYLL 63
PD2=PD2/(A*A+1.)	GRYLL 64
OPTIMISATION OF MEAN SQUARE DISTANCE	GRYLL 65
PERPENDICULAR DISTANCE OF CENTROID	GRYLL 66
PDCEN=(A*XBAR-YBAR+B)/(1.+A*A)	GRYLL 67
INTERIM VALUES	GRYLL 68
F=F+1.	GRYLLA68
STOR1=2.*AGGR/(1.+A*A)	GRYLL 69
STOR2=B*XBAR-XYBAR	GRYLL 70
STOR3=Y2BAR - X2BAR + B*(B - 2.*YBAR)	GRYLL 71
FIRST DIFFERENTIAL COEFFICIENTS	GRYLL 72
GRDA=STOR1*(STOR2-(A/(1.+A*A))*(STOR3+2.*A*STOR2))	GRYLL 73
GRDB=2.*AGGR*PDCEN	GRYLL 74
SECOND DIFFERENTIAL COEFFICIENTS	GRYLL 75
GRDA2=STOR1*(STOR3-2.*A*(3.+A*A)*STOR2-3.*A*A*X2BAR)/((1.+A*A)*1(1.+A*A))	GRYLL 76
GRDB2=STOR1	GRYLL 77
GRDAB=STOR1*(XBAR-2.*A*PDCEN)	GRYLL 79
INVERSION OF THE HESSIAN MATRIX	GRYLL 80
DETERMINANT	GRYLL 81
DET=GRDA2*GRDB2-GRDAB*GRDAB	GRYLL 82

PAGE 3 E0000295

```

DET=DET*(-1.)
IF(DET)11,14,11
INVERTED MATRIX
1 H1=GRDB2/DET
H2=-GRDAB/DET
H3=H2
H4=GRDA2/DET
NEW VALUES OF SLOPE AND INTERCEPT
AA=A+H1*GRDA+H2*GRDB
BB=B+H3*GRDA+H4*GRDB
NEW VALUE OF SUM OF SQUARES OF PERPENDICULAR DISTANCES
SUM=0.
DO 12 I=1,K
2 SUM=SUM+(AA*X(I)-Y(I)+BB)*(AA*X(I)-Y(I)+BB)
SUM=SUM/(AA*AA+1.)
TEST FOR VALIDITY OF CORRECTION
OPT=(PD2-SUM)/PD2
IF(OPT)131,131,13
APPLY CORRECTIONS
3 A=AA
B=BB
PD2=SUM
IF(1.-(OPT*1.E04))10,14,14
31 F=-F
REPLACE AXES
4 DO 15 I=1,K
X(I)=X(I)+XMEAN
5 Y(I)=Y(I)+YMEAN
B=B+YMEAN-A*XMEAN
6 RETURN
END

```

GRYLL 83
 GRYLL 84
 GRYLL 85
 GRYLL 86
 GRYLL 87
 GRYLL 88
 GRYLL 89
 GRYLL 90
 GRYLL 91
 GRYLL 92
 GRYLL 93
 GRYLL 94
 GRYLL 95
 GRYLL 96
 GRYLL 97
 GRYLL 98
 GRYLL 99
 GRYLL100
 GRYLL101
 GRYLL102
 GRYLL103
 GRYLL104
 GRYLL105
 GRYLLA05
 GRYLL106
 GRYLL107
 GRYLL108
 GRYLL109
 GRYLL110
 GRYLL111
 GRYLL112

FEATURES SUPPORTED
 ONE WORD INTEGERS

MORE REQUIREMENTS FOR GRYLL
 COMMON 0 VARIABLES 86 PROGRAM 1058

RELATIVE ENTRY POINT ADDRESS IS 0062 (HEX)

END OF COMPILATION

PAGE 1 CCSTF675

// JOB 1001 1002 1002 *CCSTF675

LOG DRIVE	CART SPEC	CART AVAIL	PHY DRIVE
0000	1001	1001	0000
0001	1002	1002	0001

V1 MOO CONFIG BK

CCAT VERSION D.M. SYSTEM V1 MOO

GRAPH PLOTTING

// FOR

```
*IOCS(CARD,1132 PRINTER,PLOTTER)
*IOCS(CARD,TYPEWRITER,KEYBOARD,1132 PRINTER,PAPER TAPE,DISK)
*ONE WORD INTEGERS
```

#LIST SOURCE PROGRAM

```
DIMENSION RESIS(16), RUN(16,4) ,TITLE(6)
DATA B/' '/
```

C READ TITLE IF BLANK FINISH

```
7 READ(2,100) TITLE
IF(TITLE(1)-B) 20,21,20
```

C MOVE PEN TO L.H.S. AND PAPER FORWARD

```
20 CALL SCALF(1.0,1.0,0.0,0.0)
CALL FPLLOT(1, 7.6,13.0)
CALL SCALF(1.0,1.0,0.0,0.0)
```

C DRAW 3 LINES TO DIVIDE GRAPHS AND GO BACK TO L.H.S.

```
CALL FPLLOT(1,0.0,-0.85)
CALL FPLLOT(2,8.2,-0.85)
CALL FPLLOT(1,0.0,-6.2)
CALL FPLLOT(2,8.2,-6.2)
CALL FPLLOT(1,0.0,-11.6)
CALL FPLLOT(2,8.2,-11.6)
CALL FPLLOT(1,0.0,0.0)
```

C WRITE TITLE

```
CALL FCHAR(1.0,-.8 ,0.2,0.2,0.0)
WRITE(7,100)TITLE
```

100 FORMAT(6A4)

```
CALL FPLLOT(1,1.0,0.)
DO 16 L=1,2
CALL SCALF(1.0,1.0,0.0,0.0)
```

DO 2 I=1,16

READ(2,102) RESIS(I),(RUN(I,K),K=1,4)

C IF FIRST RESISTANCE IS ZERO NO COMPRESSION DATA

IF(RESIS(1))6,7,6

C IF RESISTANCE IS ZERO STOP READING AND CALCULATE RANGE

6 IF(RESIS(I))2,3,2

2 CONTINUE

C FIND RANGE OF RESISTANCES

102 FORMAT(F9.3,4(1X,F6.4))

3 II=I-1

PAGE 2 CCSTF675

```
RANGE = RESIS(11)- RESIS(1)
IF(L-1)14,13,14
```

```
C TOP IS TOP OF GRAPH AREA
```

```
13 TOP =-1.
GO TO 15
14 TOP =-6.35
15 CALL FCHAR(1.0,-.5+TOP,0.2,0.2,0.0)
IF(L-1)19,17,19
```

```
C WRITE GRAPH DESCRIPTION
```

```
17 WRITE(7,101)
101 FORMAT('TENSION ')
GO TO 18
19 WRITE(7,107)
107 FORMAT('COMPRESSION')
18 CALL SCALF(1.0,1.0,3.0,0.0)
CALL FPLOT(1,1,8,0)
BACK =RANGE-(TOP*RANGE/5.)
```

```
C CHANGE SCALE AND ORIGIN FOR PLOTTING
```

```
CALL SCALF(1/0.02,5./RANGE,-.052,BACK)
```

```
C NOW FOR THE ACTUAL PLOTTING
```

```
DO 4 J=1,11
Y=RESIS(J)-RESIS(1)
CALL FCHAR(-0.08,Y,.1,.1,0)
IF(RESIS(J)-9999.)10,10,11
10 WRITE(7,201) RESIS(J)
201 FORMAT(F9.3)
GO TO 12
11 WRITE(7,203) RESIS(J)
203 FORMAT(F9.2)
12 DO 4 K=1,4
```

```
C IF X VALUE IS 1. DON'T PLOT
```

```
IF(RUN(J,K)-1.)5,4,5
5 CALL FCHAR(RUN(J,K),Y,0.1,0.1,0.0)
WRITE(7,200) K
200 FORMAT(I1)
4 CONTINUE
```

```
C DRAW X-AXIS
```

```
CALL FGRID(0,-.050,0,0.05,2)
X=-.1
DO 9 I=1,3
X=X+0.05
CALL FCHAR(X-.005,-.17*RANGE/5.,.08,.08,0)
9 WRITE(7,202) X
202 FORMAT(F5.2)
```

```
C LINE UP FOR NEXT GRAPH
```

```
CALL FPLOT(1,-.076, BACK)
16 CONTINUE
GO TO 7
```

PAGE 3 CCSTF675

21 CALL EXIT
END

FEATURES SUPPORTED
ONE WORD INTEGERS
IOCS

CORE REQUIREMENTS FOR
COMMON 0 VARIABLES 198 PROGRAM 630

END OF COMPILATION

// DUP

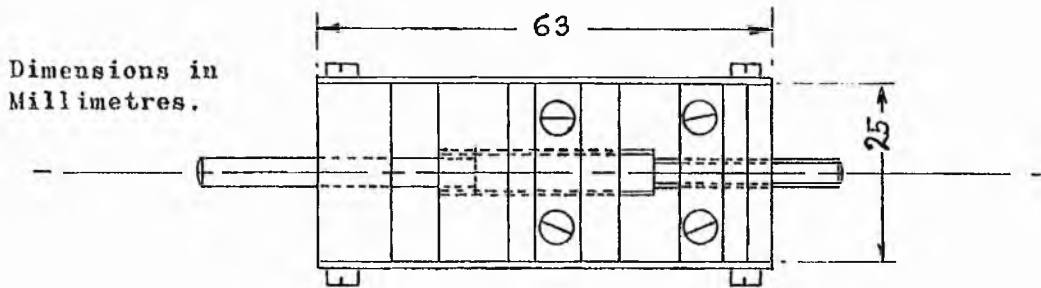
*DELETE IHGRF
CART ID 1002 DB ADDR 440F DB CNT 002C

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CART ID 1002 DB ADDR 440F DB CNT 002C

BENDING RIG FOR 1" STRAIN-GAUGE SUBSTRATES

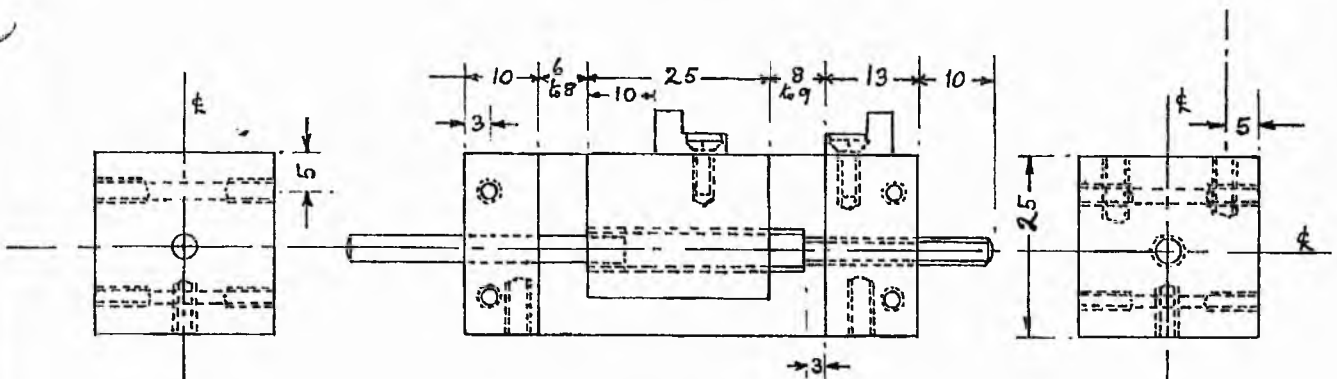
Material:- Mild Steel, Duralumin, Brass, or Stainless Steel, except where stated.

All fixing screws may be be 4 B.A. or 6 B.A.



Dimensions in Millimetres.

PLAN VIEW



DETAIL A

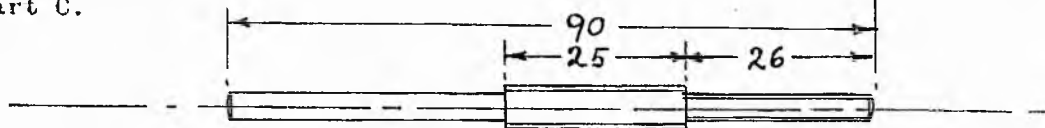
Bearing Block. Reamed as a bearing for part C.

SIDE ELEVATION

Shown with one guide-plate removed.

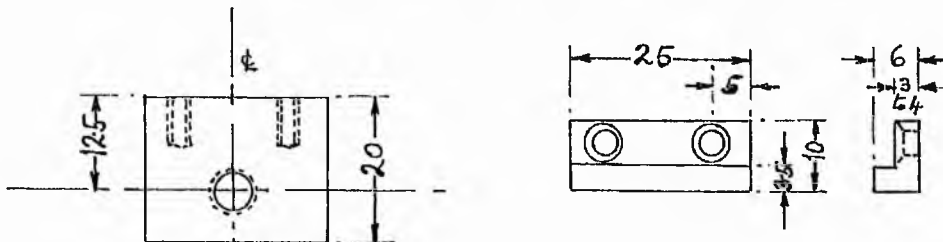
DETAIL B

End Block. Tapped 4 B.A. to fit part C.



DETAIL C

Lead Screw. Material:- Silver Steel 4mm dia., threaded 4 B.A. at one end. The sleeve, 6mm O.D., is hard-soldered in position and threaded 0 B.A.

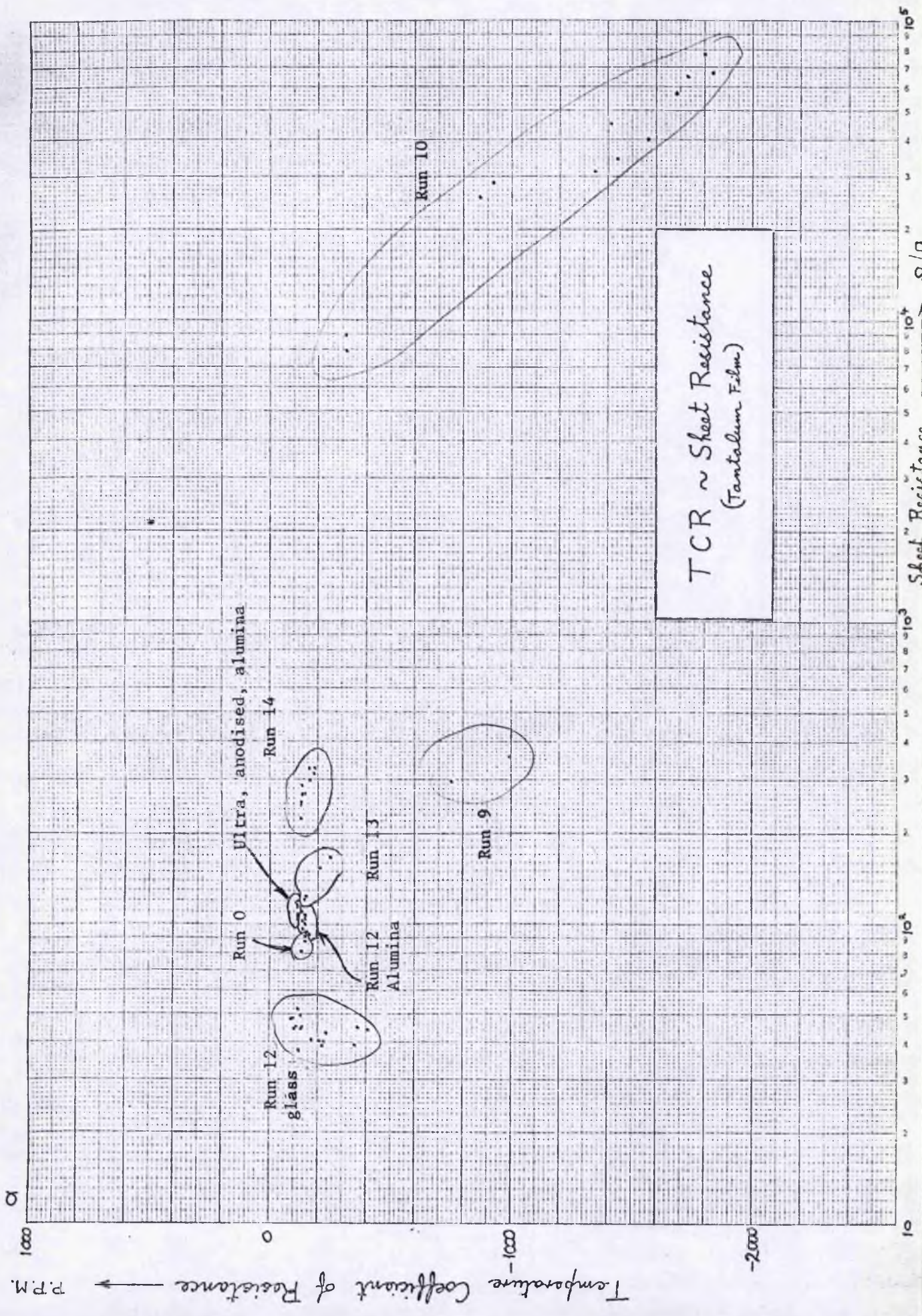


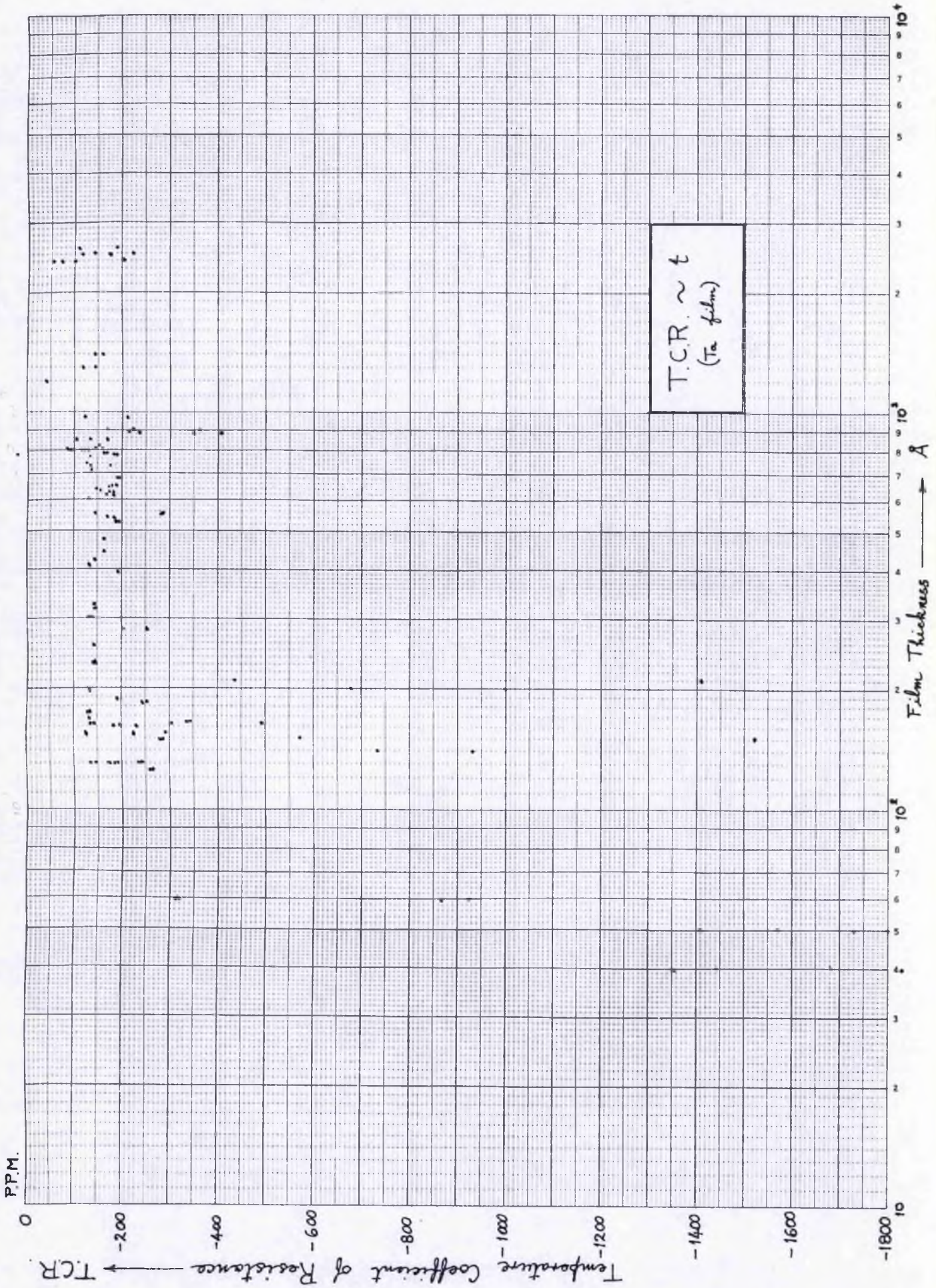
DETAIL D

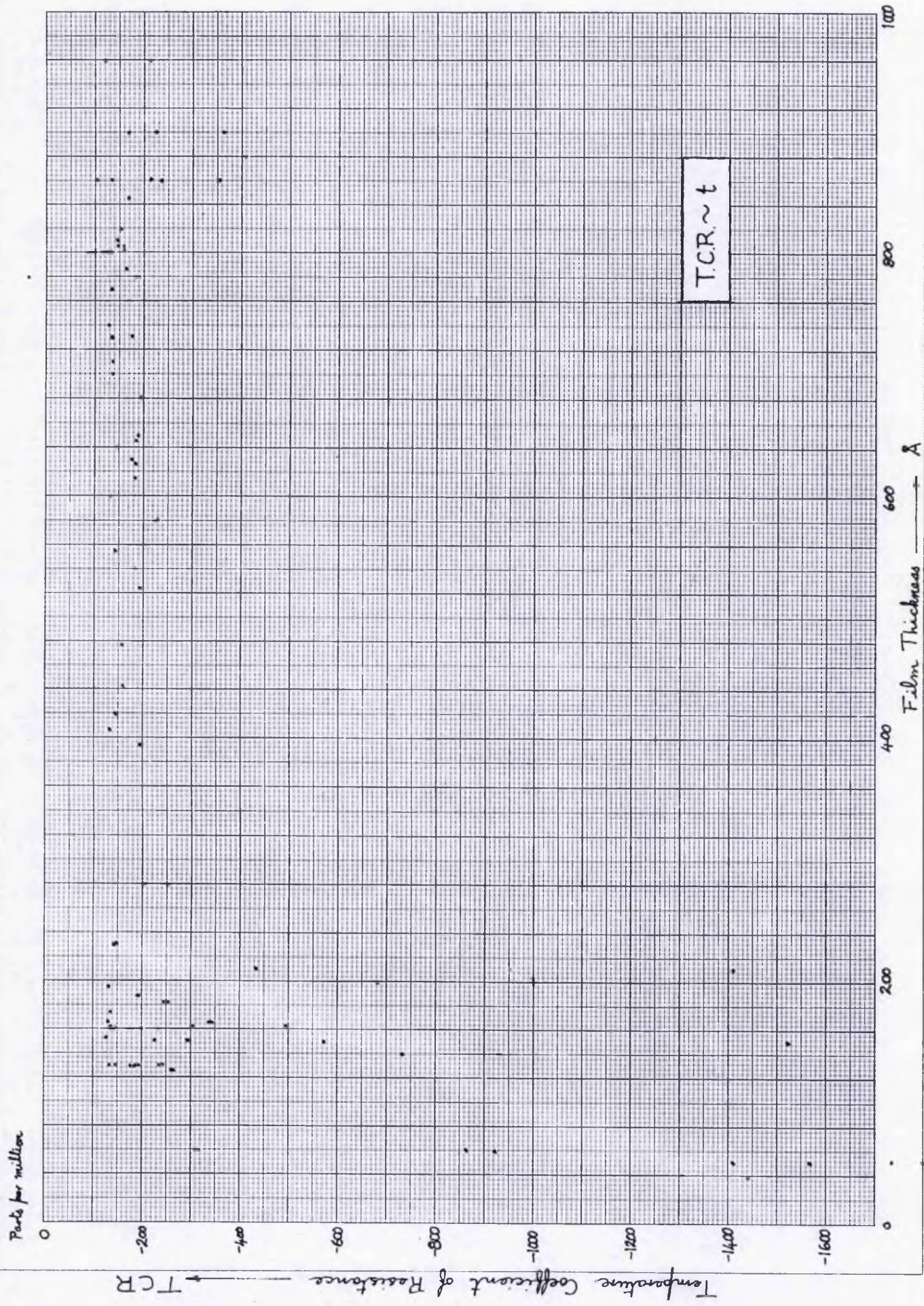
Slide Block. Tapped 0 B.A. to fit part C.

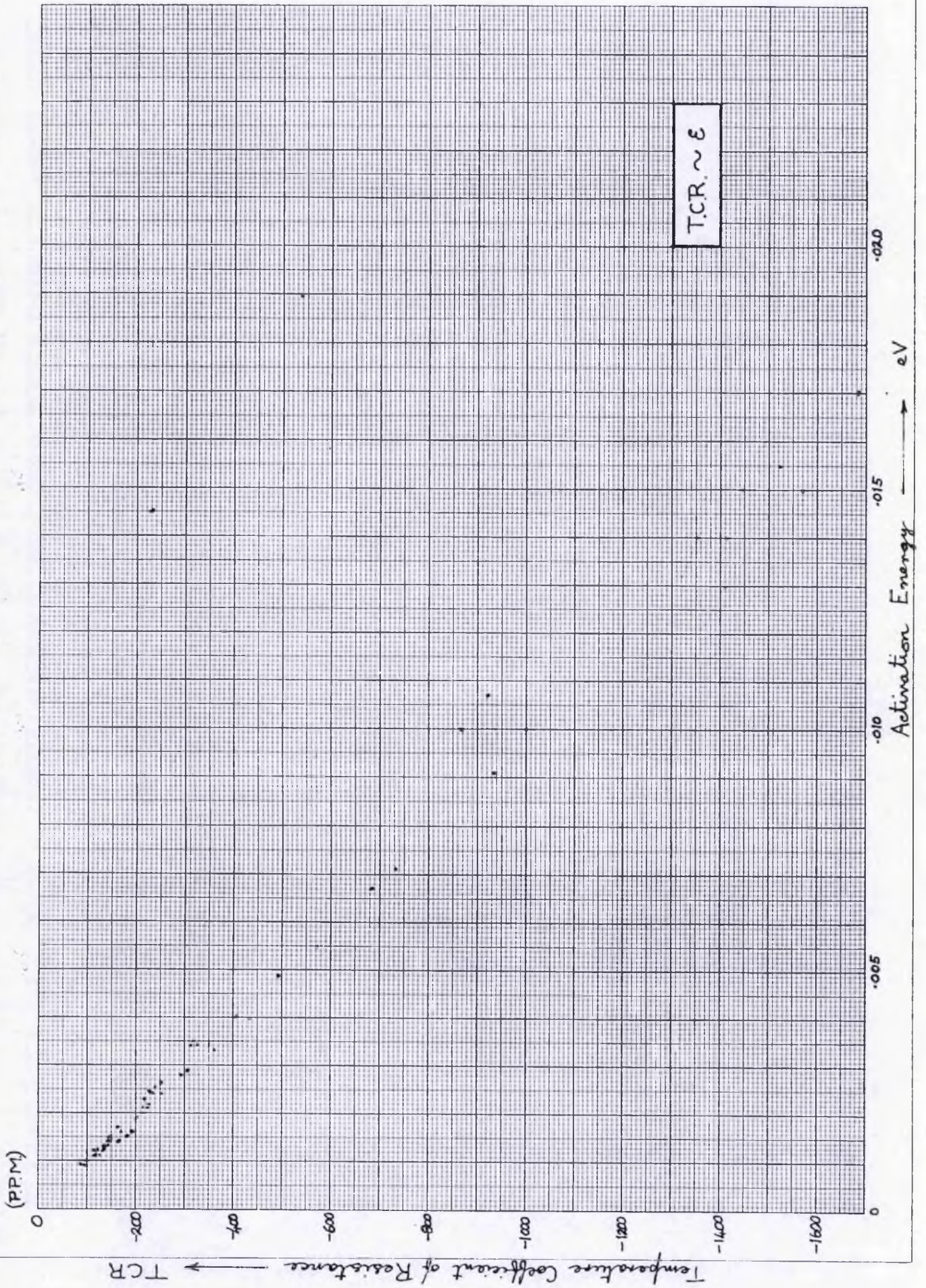
DETAIL E

Work Holder. 2 off.

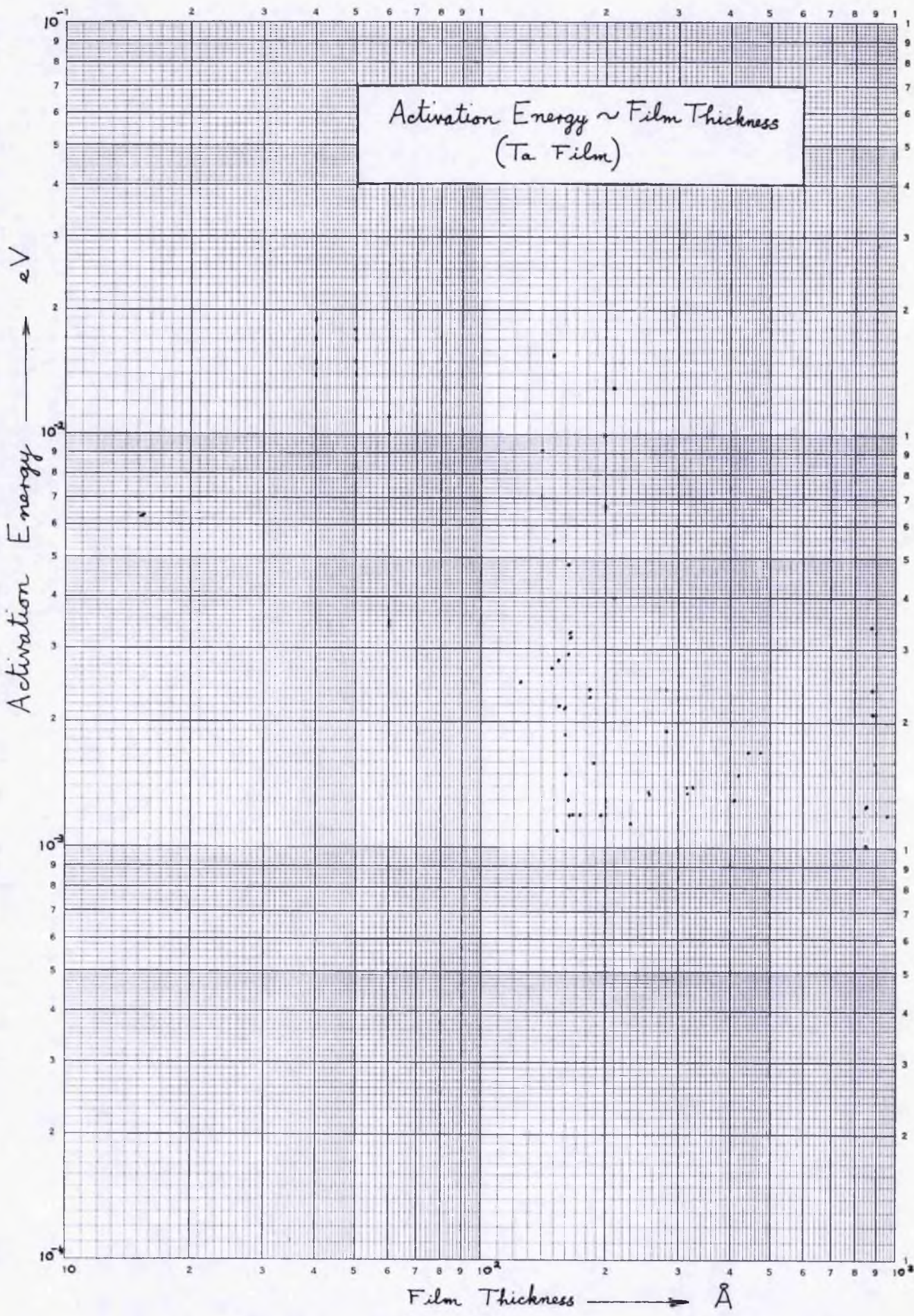




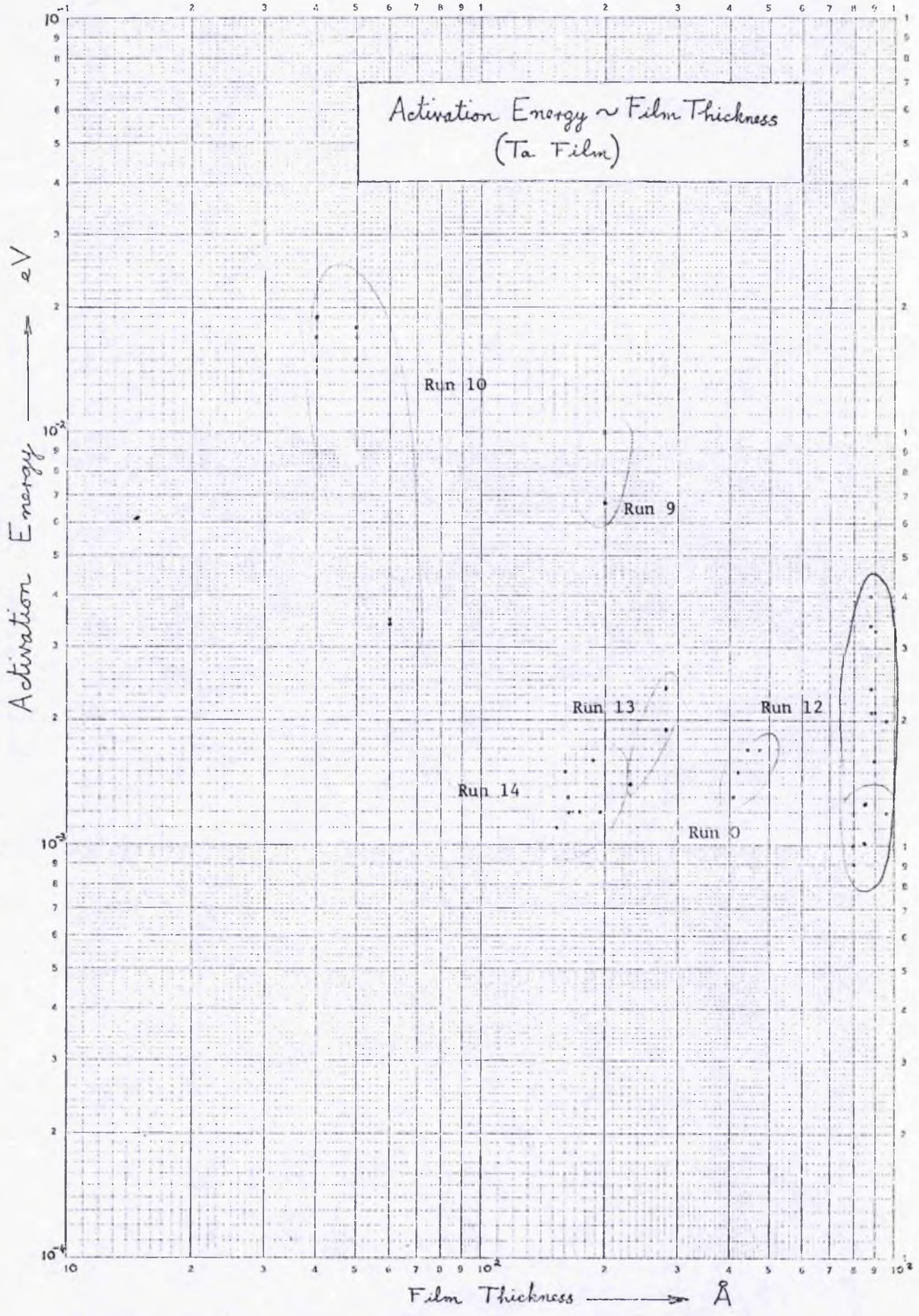


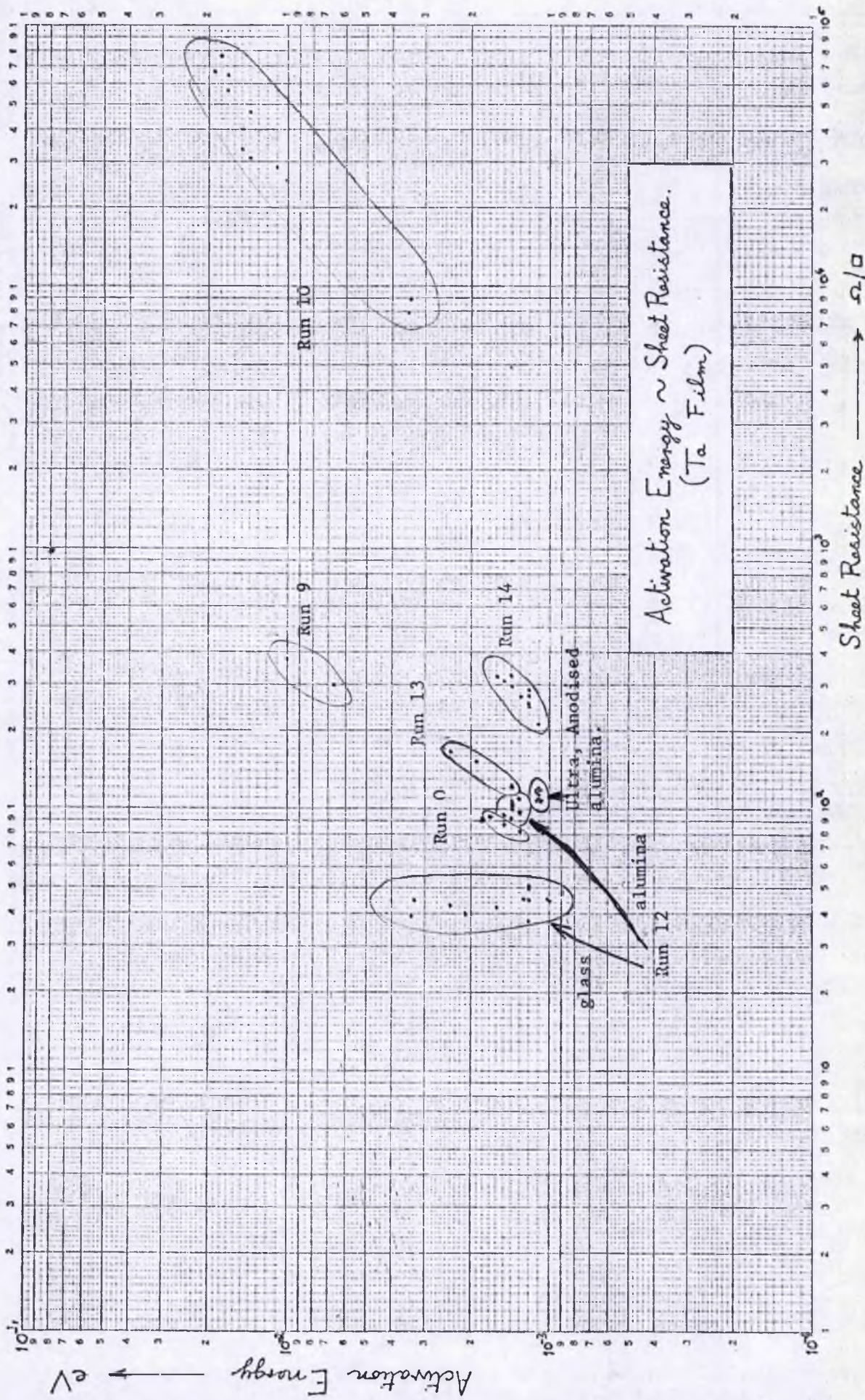


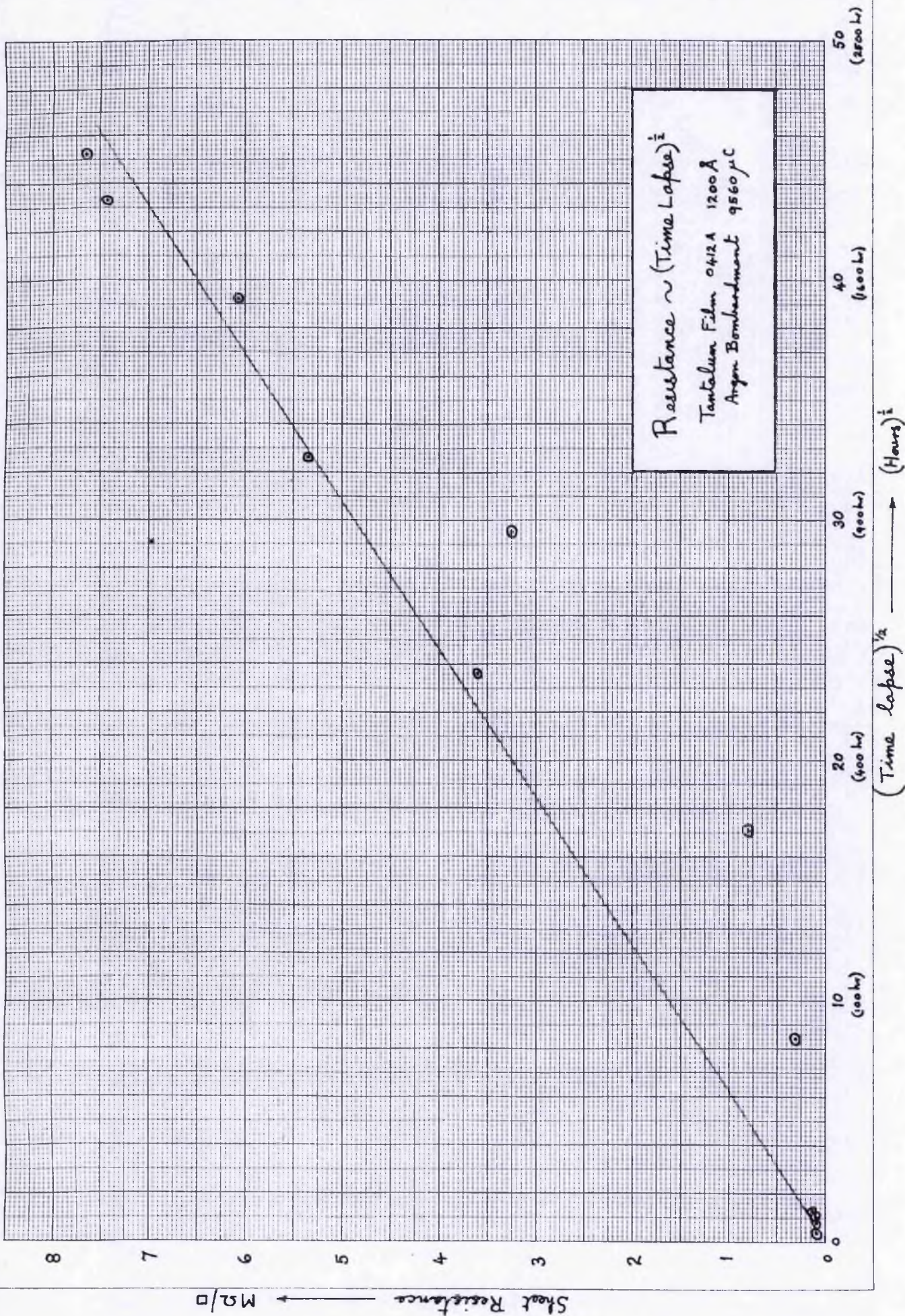
WELL
Graph Data Ref. 5932
Log 3 Cycles x 2 Cycles



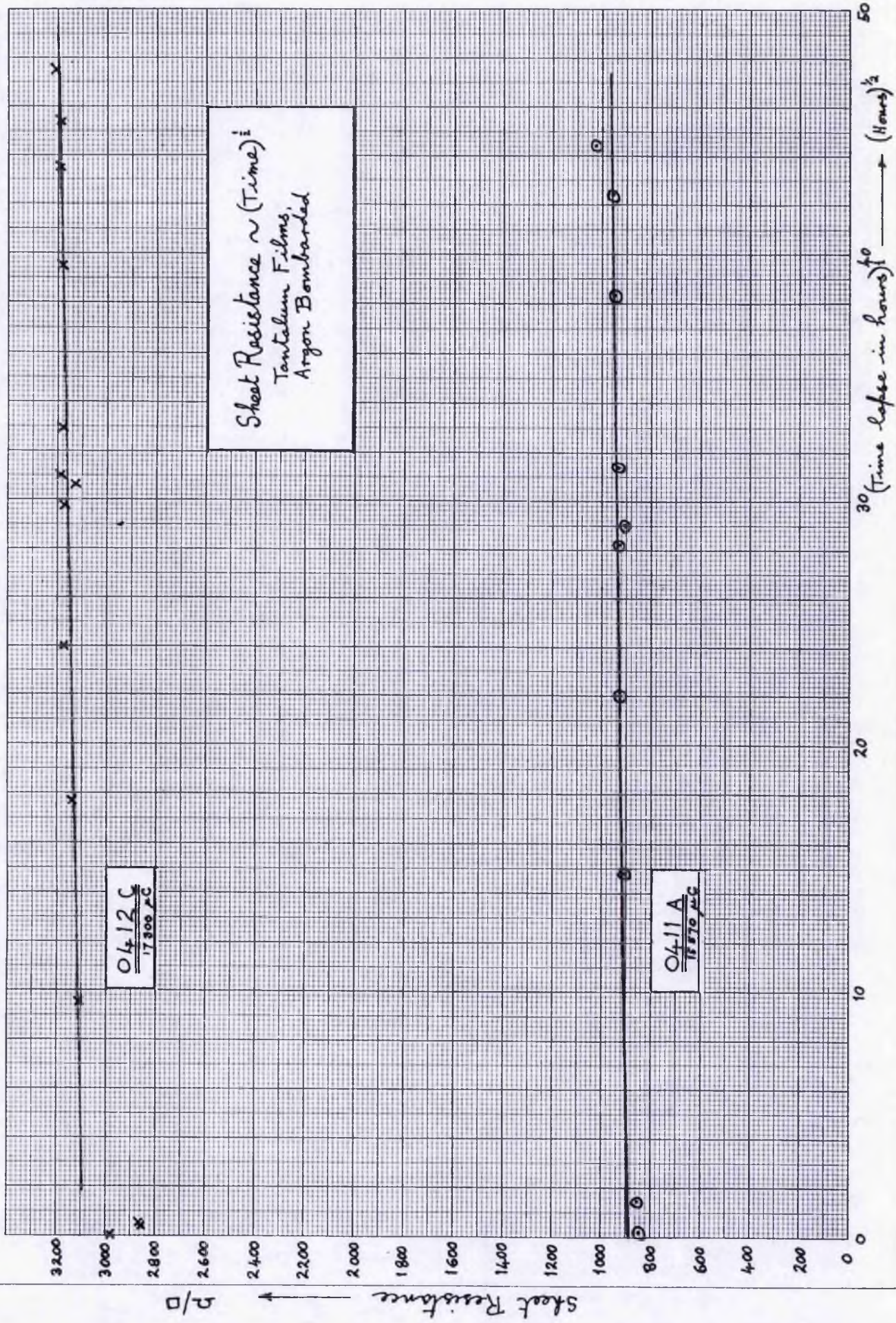
Log 3 Cycles x 2 C
Graph Data Ref. 5032
WELL







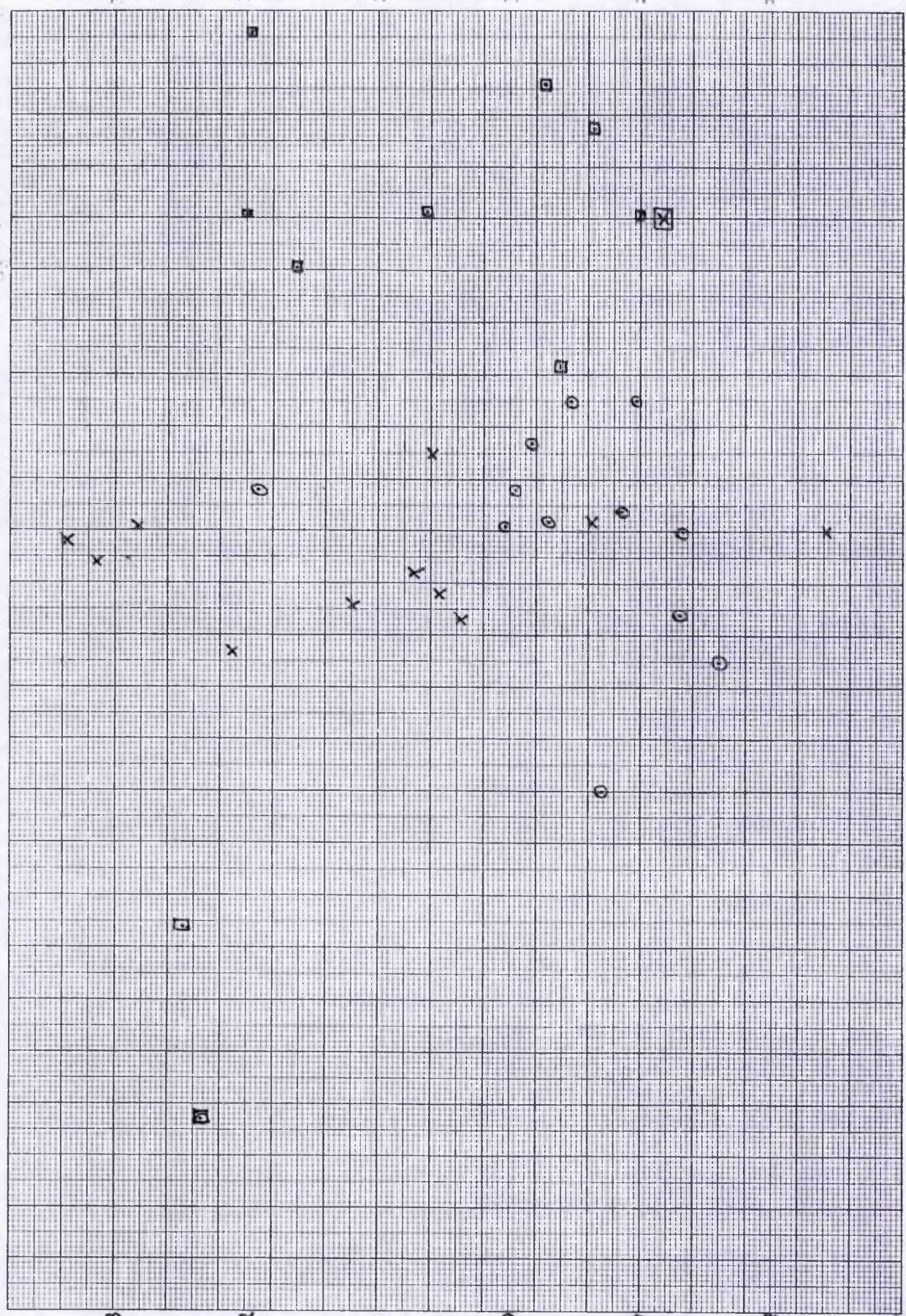
15 V. 25



before bond

40

x



50 Ω/□

45

40

35

30

27.25

Sheet Resistance →

Strain-gauge factor

33

32

31

30

29

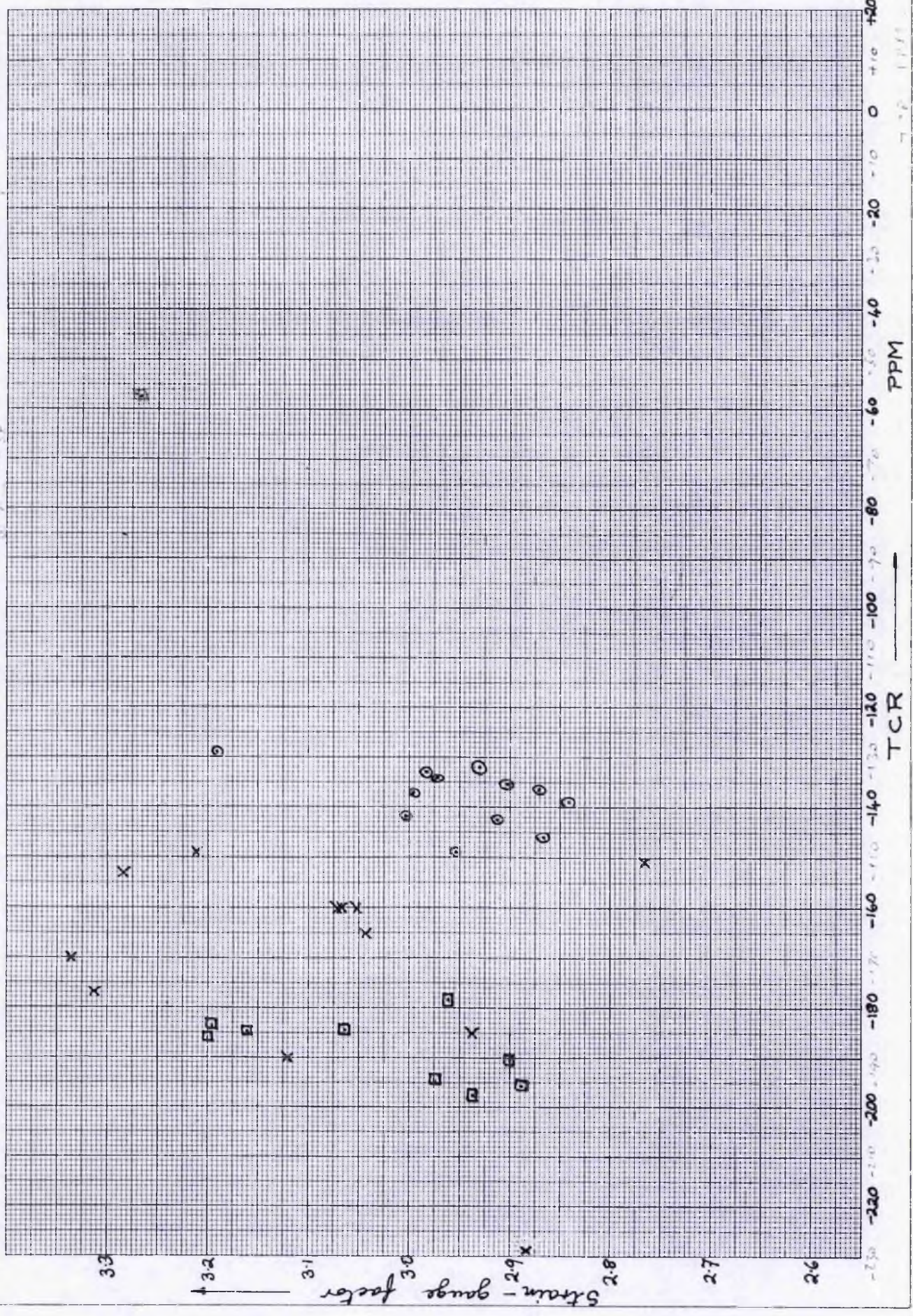
28

27.25

Run 37
 2 Run 36
 8

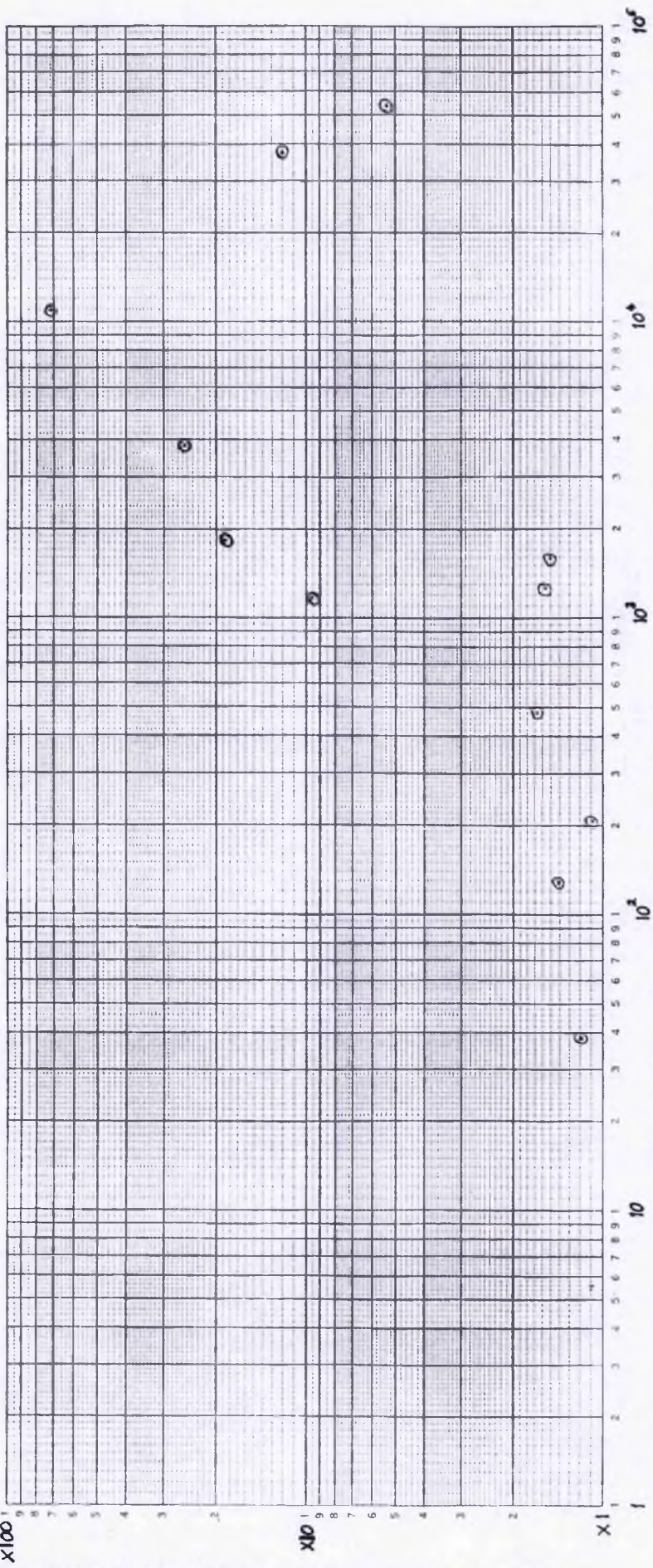
220 200 180 160 140 120 100 80 60 40 20

8



Relative Resistance ~ Resistance after 1000 Hours.
Gold Film.

$$\text{Relative Resistance} = \frac{\text{Resistance after 1000 Hours}}{\text{Resistance at bombardment}}$$



Film Resistance after 1000 Hours → Ω

STRAIN GAUGE FACTOR CORRECTION CALCULATIONS(i) Virgin Gauge

$$\gamma_{\text{REAL LONG}} = \gamma_{\text{APP LONG}} \times \frac{\gamma_{\text{APP SHORT}}}{\gamma_{\text{APP LONG}}} \times \frac{\gamma_{\text{REAL SHORT}}}{\gamma_{\text{APP SHORT}}}$$

where: $\gamma_{\text{REAL LONG}}$ = Real strain gauge factor of virgin film

$\gamma_{\text{APP LONG}}$ = Strain gauge factor of virgin film assuming four point bending

$\frac{\gamma_{\text{APP SHORT}}}{\gamma_{\text{APP LONG}}}$ = Result of strain measurements on virgin film which had gold electrodes extended to bombardment area. Four point bending assumed.

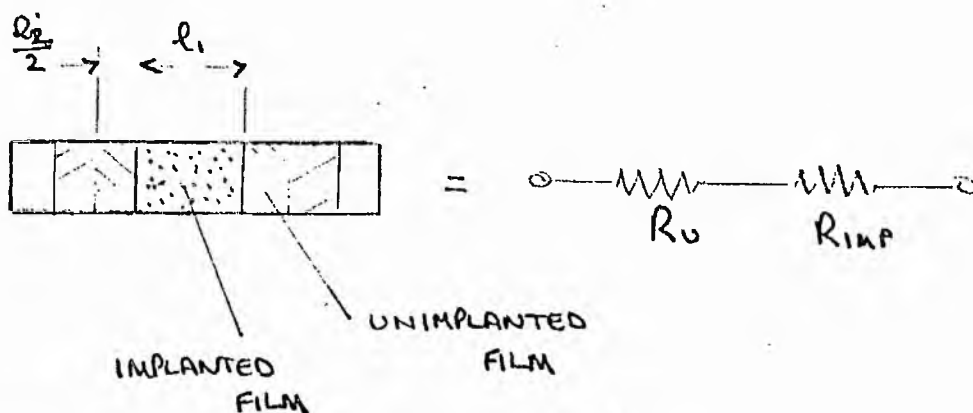
$\frac{\gamma_{\text{REAL SHORT}}}{\gamma_{\text{APP SHORT}}}$ = Result of the calibration of the short geometry with micro-measurements strain gauge.

Evaluating we have:-

$$\gamma_{\text{REAL LONG}} = \gamma_{\text{APP LONG}} \times 1.06 \times 0.766$$

OR

$$\underline{\underline{\gamma_{\text{REAL LONG}} = \gamma_{\text{APP LONG}} \times 0.814}}$$

(ii) Implanted Film

Now,

$$R_T = R_u + R_{IMP} \dots\dots\dots (i)$$

where R_T = total resistance of compound film

R_u = resistance of unimplanted film

R_{IMP} = resistance of implanted film

Let the specimen be strained in the bending rig then:-

$$\Delta R_T = \Delta R_u + \Delta R_{IMP}$$

OR

$$\Delta R_T = R_u \gamma_u \epsilon_2 + R_{IMP} \gamma_{IMP} \epsilon_1 \dots\dots\dots (ii)$$

where	R_u = resistance	} of the unimplanted film
	γ_u = strain gauge factor	
	ϵ_2 = strain	

	R_{IMP} = resistance	} of the implanted film
	γ_{IMP} = strain gauge factor	
	ϵ_1 = strain	

Now if we assume a relationship between the two strain levels:-

$$\epsilon_2 = K \epsilon_1 \dots\dots\dots (iii)$$

substituting (iii) into (ii):-

$$\Delta R_T = R_u \gamma_u K \epsilon_1 + R_{IMP} \gamma_{IMP} \epsilon_1$$

thus:-

$$\frac{\Delta R_T/R_T}{\epsilon_1} = \frac{R_u \gamma_u K + R_{IMP} \gamma_{IMP}}{R_u + R_{IMP}} \dots\dots\dots (iv)$$

Now:-

$$\gamma_{T(app)} = \frac{\Delta R_T/R_T}{\epsilon_{app}} \dots\dots\dots (v)$$

where $\gamma_{T(app)}$ = apparent strain gauge factor of the combined gauge assuming 4 point bending

ϵ_{app} = apparent strain in specimen assuming 4 point bending.

But:-

$$\gamma_{T(app)} = \frac{\Delta R_T/R_T}{\epsilon_1} \times \frac{\epsilon_{real\ short}}{\epsilon_{app\ short}} \dots\dots\dots (vi)$$

where $\frac{\epsilon_{real\ short}}{\epsilon_{app\ short}}$ = result from central region strain calibration
 (= $\frac{1}{0.766}$)

thus substituting (iv) into (vi) we have:-

$$\gamma_{T APP} = \frac{R_u \gamma_u K + R_{IMP} \gamma_{IMP}}{R_u + R_{IMP}} \times \frac{1}{0.766} \dots\dots\dots (vii)$$

Determination of K

Now,

$$l_T = l_1 + l_2 \dots\dots\dots (viii)$$

where l_T = total gauge length

l_1 = length of implanted gauge

l_2 = length of unimplanted gauge

(iv)

Hence:-

$$\Delta l_T = \Delta l_1 + \Delta l_2$$

OR

$$\epsilon_T l_T = l_1 \epsilon_1 + l_2 \epsilon_2 \quad \dots\dots\dots (ix)$$

where ϵ_T = integrated strain over l_T

ϵ_1 = integrated strain over l_1

ϵ_2 = integrated strain over l_2

From equation (ix):-

$$\epsilon_2 = \frac{\epsilon_T l_T - l_1 \epsilon_1}{l_2}$$

thus:-

$$K = \frac{\epsilon_2}{\epsilon_1} = \frac{(\epsilon_T/\epsilon_1)l_T - l_1}{l_2} \quad \dots\dots\dots (x)$$

evaluating:-

$$l_T = 1.3 \text{ cm}$$

$$l_1 = 0.67 \text{ cm}$$

$$l_2 = 0.63 \text{ cm}$$

$$\frac{\epsilon_T}{\epsilon_1} = \frac{\epsilon_{\text{long}}}{\epsilon_{\text{short}}} = \frac{1}{1.06} \quad (\text{from part (i)})$$

therefore,

$$\underline{\underline{K = 0.879}}$$

(v)

Substituting this value of K into equation (vii)

$$Y_{TAPP} = \frac{R_u Y_u^{0.879} + R_{IMP} Y_{IMP}}{(R_u + R_{IMP})} \times \frac{1}{0.766}$$

and

$$Y_{IMP} = \frac{Y_{TAPP} R_T^{0.766} - R_u Y_u^{0.879}}{R_{IMP}}$$

Note

R_u can be determined from:-

$$R_u = \frac{l_2}{l_T} R_{Bl}$$

where R_{Bl} = resistance of gauge before implantation