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**Textural and petrologic study of the subvolcanic mafic and  
ultramafic ejecta of Piton de la Fournaise as a key to unravel the  
shallow magmatic crystallization processes at La Réunion Island  
(Indian Ocean)**

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*"Se le cose le vuoi veramente  
accadono".*

*A Mirco*

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## I. Abstract

The Réunion Island consists of two juxtaposed volcanic massifs, the oldest Piton des Neiges (in the north-west) and the youngest, presently active Piton de la Fournaise (in the south-east). A third, poorly known buried volcano, named Les Alizés was mainly defined through exploratory drilling and constitutes the basement of Piton de la Fournaise (Lénat et al., 2012). The construction of Piton de la Fournaise *sensu stricto* started at about 400-450 ka. Its lavas are mainly transitional basalts defining a differentiation trend towards mugearites, intermediate in character between the alkaline and tholeiitic series, both aphyric ("cotectic basalts") and olivine-rich ("oceanites"). They slightly differ from Les Alizés aphyric basalts ("abnormal group") that belong to the more alkalic (K<sub>2</sub>O-rich) series, with higher MgO and lower CaO and SiO<sub>2</sub> contents. The moderate chemical heterogeneity has been since long attributed to a combination of slight mantle source heterogeneity and influence of clinopyroxene fractionation at mantle/underplating depth. Magmas stored at crustal depths and erupted at the summit area are quite homogeneous and result from the hybridization process of "tholeiitic" and "alkaline" end members (Di Muro et al., 2014; Boudoire et al., 2018; Vlastelic et al., 2018).

The present Ph.D. work deals with a petrological study of mafic and ultramafic ejecta of the "Bellecombe Ash Member" (BAM,  $\approx$  5.5-3.0 ka), in order to unravel the pre-eruptive conditions of basalt crystallization and possibly matching the various recognized petrographic groups of the subvolcanic igneous rocks with the whole extrusives (Piton de la Fournaise, Piton des Neiges, Les Alizés) combining mineralogy, petrographic texture and geochemistry. Large caldera forming events such as that of the Bellecombe Ash Member, the largest explosive activity produced by Piton de la Fournaise involve, as ejecta, a large fraction of the crustal magmatic plumbing system and allowed to i) access to a rare snapshot of the magmatic processes (cumulate vs. liquid composition; volatile exsolution and percolation; melt and rock assimilation) occurring below one of the most active basaltic volcanoes of the world, and ii) study the magmas driving these rare but very hazardous geological events. The petrological study of more than one hundred samples, representative of the ejecta within the Bellecombe Ash Member, emphasizes the presence of many subvolcanic mafic and ultramafic igneous rocks: dunites, wherlites, ophitic gabbros, sub-ophitic gabbros, poikilitic gabbros, doleritic gabbros, micro-monzogabbros and porphyrogabbros. Cumulate cognates *sensu stricto* are present, often with quenched basaltic glass entrapped interstitially and in minerals (melt inclusions), with no disequilibrium textures recognized in the subvolcanic crystal frameworks. Other subvolcanic clasts (e.g. doleritic gabbros) may simply represent slowly-cooled equivalents (with no crystal-liquid fractionation) of the erupted cotectic basalts. Chemistry of both minerals (olivine, clinopyroxene, feldspar, oxides and phlogopite) and interstitial glass from the subvolcanic ejecta has been performed by EMPA. It is worth to note

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that porphyrogabbros emphasize a wide range of feldspars, as interstitial crystals between the large olivine phenocrysts, represented by plagioclase, anorthoclase and sanidine, defining a complete miscibility feldspar curve. Olivines of each mafic and ultramafic lithotype ejecta show different forsterite contents and the whole compositional range is from Fo<sub>69</sub> (sub-ophitic gabbros) to Fo<sub>90</sub> (dunite), with an overall range in CaO contents from 0.05 to 0.4 wt.%. A fluid inclusion chemical and thermobarometric study (CO<sub>2</sub>, H<sub>2</sub>O and noble gases) was made on selected olivine, clinopyroxene, and feldspar crystals. Noble gases and CO<sub>2</sub> content in fluid inclusions of the studied ejecta reveal variable concentrations and elemental ratios (e.g. He/Ar\*), indicating, for the whole mafic and ultramafic suite of the Piton de la Fournaise magmatic plumbing system a wide range of depths, from the underplating zone to the submarine base of the edifice and various extent of magmatic degassing. The above barometric data have been also confirmed by fluid inclusion study by Raman spectroscopy analyses. A comprehensive geochemical comparison among the analysed mafic and ultramafic ejecta (major and trace elements by ICP-OES-MS) and the whole compositional spectrum of the Piton de la Fournaise and Les Alizés volcanic rocks available from literature was also carried out. Major and trace elements geochemistry allowed to confirm the three fractional crystallization trends already defined by the literature, namely gabbro- wherlite- and dunite-fractionations which well agree with the erupted basalts in the Réunion Island in the last 500 ka. A wide range of textures in these ejecta results from variable P-T gradients in the subvolcanic environment of crystallization. In particular the CO<sub>2</sub>-H<sub>2</sub>O fluid inclusion and thermobarometric studies carried out by Raman spectroscopy on selected single crystals (olivine and clinopyroxene) permitted to better understand the depths of crystallization of the samples. The study of the mafic and ultramafic subvolcanic igneous rocks of Piton de la Fournaise entrapped in the BAM eruption allow to target the transcrustal plumbing system still active during large caldera forming events. Compositions of the interstitial glass and minerals in some subvolcanic ejecta has been also compared with small volumes of evolved liquids (trachytes) detected as pumiceous components in the products of the Bellecombe Ash Member caldera-forming eruption and the old Piton des Neiges volcano. In general, the textural and petrologic study of the subvolcanic ejecta of Piton de la Fournaise provides new insights on the magmatic plumbing system and the subvolcanic crystallization processes at La Réunion Island. Some peculiarities on the processes of crystallization in the magmatic plumbing system were pointed out, with respect previous knowledges. Some ejecta defined as porphyrogabbros could represent oceanite magmas stored at crustal levels where, the interstitial basaltic melts supporting the abundant olivine crystals gave rise to a sanidine- and phlogopite-bearing micro-cryptocrystalline groundmass. This micro-cryptocrystalline groundmass could be really considered cogenetic with trachytic liquid, which may be erupted as

glassy pumice if the undercooling would be high enough to preclude groundmass crystallization. The overall textural and geochemical data of the investigated mafic-ultramafic ejecta, do not only reflect different mode of subvolcanic crystallization (cumulate *sensu stricto* vs. slowly-cooled equivalents of the steady state basalts which are represented by the doleritic gabbro and some of the subophitic gabbro ejecta) but also (i) the interaction between basaltic liquids and cumulates already formed at crustal level of the Piton de la Fournaise plumbing system and (ii) the possibility that small pockets of liquids may also evolve to trachytic compositions. The compositional range from basalt to basaltic trachyandesite of the interstitial glass found in the intrusive clasts, indicate that the melt was entrapped within the crystal frameworks and quenched during the eruption. This likely suggests the clast were disrupted from the wall rocks and transported very fast to the surface, mostly as cognates. If not entrapped interstitially during *in situ* crystallization, silicate liquid could be also expelled from the crystal frameworks, forming separated melt pockets or larger body, e.g. slightly differentiated sills.

Fluid inclusion barometry (Raman analyses) and noble gases and CO<sub>2</sub> from fluid inclusion crushing extraction and mass spectrometer analyses indicate the dunites are the intrusive ejecta from the deepest levels, approximately at the underplating levels of formation (~13 km b.s.l.). Wherlites should have crystallized close to the submarine base of the edifice (~6 km b.s.l.), whereas the mafic subvolcanic samples are even shallower. Nevertheless, a large composite intrusive ejecta (B108) emphasizes the coexistence of mesoscale ultramafic-mafic patchy microdomains (wherlite to plagioclase-wherlite to troctolite to gabbro) at intermediate crustal level of crystallization, below 0.3 GPa according to the liquidus curves of the La Reunion magmas (Fisk et al., 1988). Textures in this composite ejecta can be the result of basaltic melts interacting with wherlitic cumulates at relatively low pressure, leading to the instability of clinopyroxene and high dissolution of olivines (now highly rounded), with a high loss of temperature of the system. In this way at the same pressure, but lower temperature, plagioclase is the *liquidus phase* and therefore wherlitic cumulates interacting with basaltic melts at relatively low pressure can produce plagioclase-wherlite and troctolite patchy microdomains at the mesoscale. Similar interactions between batches of basaltic magmas and cumulates already present in the magmatic plumbing system are also highlighted by other wherlites (i.e. B28 sample) coupled with olivine dissolution/reaction leading also to clinopyroxene. Finally, the presence of loose quartz xenocrysts in products at the Langevin Caldera, possibly with aqueous inclusions with low-salinity and low CO<sub>2</sub> content and an enrichment of some trace elements (e.g. Li, Al, Sr) at their rims could suggest the involvement of the hydrothermal system of Piton del la Fournaise during explosive eruptions of the volcano.



## 1 Introduction

Réunion Island is a large and dominantly effusive basaltic volcanic oceanic system in the southernmost part of the Mascarene Basin (Indian Ocean), 800 km east of Madagascar, located at the youngest end of the hot spot track, which produced the Deccan Trapps. The island consists of two juxtaposed volcanic massifs, the oldest Piton des Neiges (in the north-west) and the youngest, presently active Piton de la Fournaise (in the south-east). A third, poorly known buried volcano, named Les Alizés was mainly defined through exploratory drilling and constitutes the basement of Piton de la Fournaise. The construction of Piton de la Fournaise *sensu stricto* started at about 400-450 ka. Piton de la Fournaise is one of the most active intra-plate volcanoes of the world. The main eruptive centre corresponds to the poly lobate U-shaped Enclos Fouqué caldera. The horseshoe-shaped geometry was interpreted as the result of a huge flank landslide, occurring with the emplacement of the Bellecombe Ash Member. Bellecombe Ash Member is the main tephra identified all around the Enclos Fouqué caldera, corresponding to a sequence of ash, crystal and lithic-rich deposits, (Bachèlery, 1981; Mohamed-Abchir, 1996; Ort et al., 2014), erupted between  $4880 \pm 35$  yr BP and  $2340 \pm 30$  yr BP (Morandi et al., 2016; Bachèlery et al., 2015). This deposit is thought to be emplaced during violent phreatic/phreatomagmatic events related to the flashing of an extensive hydrothermal system during the collapse of the Enclos Fouqué caldera (Bachèlery, 1981; Fontaine et al., 2002; Upton et al., 2000).

The possibility of studying deposits of the Bellecombe Ash Member allow to access to a rare snapshot of the magmatic processes (cumulate vs. liquid composition; volatile exsolution and percolation; melt and rock assimilation) occurring below one of the most active basaltic volcanoes of the world. As the large explosive event of the Bellecombe Ash Member could have disrupted the subvolcanic plumbing system of both Piton de la Fournaise and Les Alizés, the whole crystallization/fractionation trends of the studied mafic and ultramafic ejecta will be compared with those of the main extrusive products of the La Réunion Island oceanic system inferred by the literature.

## 2 Geological and volcanological setting

The Indian Ocean started to open around 175 Ma (Schettino and Scotese, 2005). The early stages of Gondwanan break-up separated Africa from a block composed of Madagascar, India, Australia and Antarctica (Coffin and Rabinowitz, 1987; Smith and Sandwell, 1997). Around 83 Ma the Mascarene Basin opened, marking the separation of Madagascar and India (Bernard and Munschy, 2000; Dymant, 1991a). The Mascarene basin rift became extinct at  $\approx 61$  Ma, when the spreading jumped to a new location, between the Seychelles and India, to become the Central Indian Ridge. The Seychelles and Madagascar are microcontinents isolated by episodes of sea floor spreading. The other

plateaus, Laccadives, Maldive, Chagos, Mascarene, Mauritius and La Réunion islands are widely regarded as volcanic constructions of the La Réunion hotspot can be traced back to the emission of the Deccan Trapps at the K-T boundary in the late Cretaceous era (Duncan, 1990; Duncan et al., 1989; Georgen et al., 2001; Storey, 1995; Berthod et al., 2020; Fig. 2.1). When the hotspot was near the mid-ocean spreading ridge, it produced complex structural and geochemical interactions (Ito et al., 2003; Morgan, 1978; Sleep, 2002). Mascarene Basin has been the subject of numerous studies because it is a key area to understand the geodynamics of the early stages of the Indian Ocean. Based on radiometric dating of subaerial rocks, current effusive rate and volumetric estimates of the island, La Réunion is thought to be active since at least 5 Ma (Gillot et al., 1994; Lénat et al., 2001; Berthod et al., 2020) with the edification of two successive shield volcanoes.

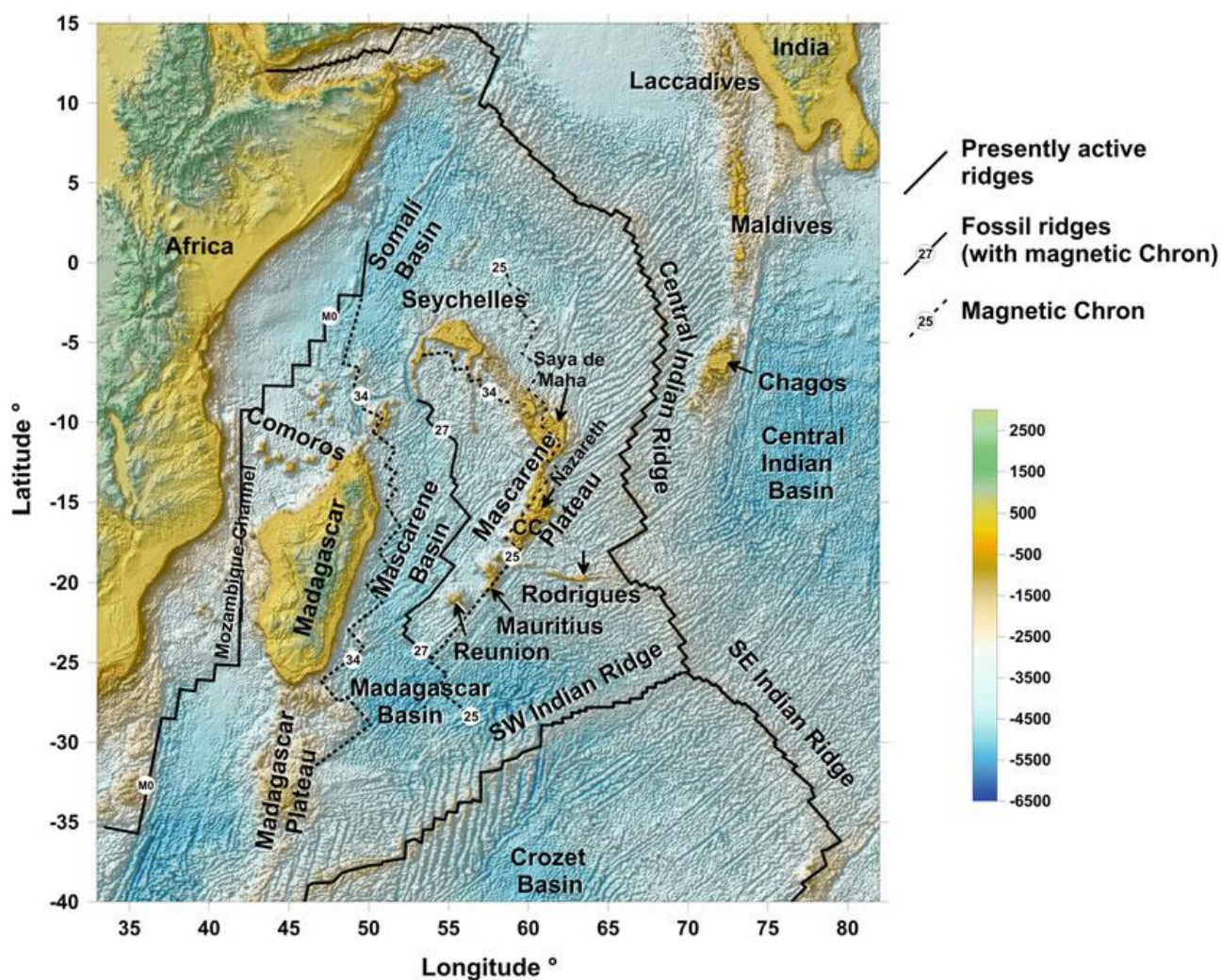


Fig. 2.1 - Geodynamic setting of La Réunion. The magnetic chronos are taken from Müller et al. (1997).

## 2.1 La Réunion

La Réunion (21.0°S; 55.5°E) lies in the middle of an approximately 350 km-wide zone bounded by the Mauritius transform fault to the SE and by the Mahanoro-Wilshaw transform fault to the NW. Another structural boundary, to the SW, is a trough identified as an extinct spreading centre. La Réunion is a huge volcanic oceanic system in the southernmost part of the Mascarene Basin, in the Indian Ocean (Fig. 2.1.1). The island is a large cone of about 51500 km<sup>3</sup>, is elliptical in shape (50x70 km) with a NW-SE orientation, and rises on the ocean floor (about 4200 m below sea level) to 3070 m above sea level and its base diameter is of about 210 km.

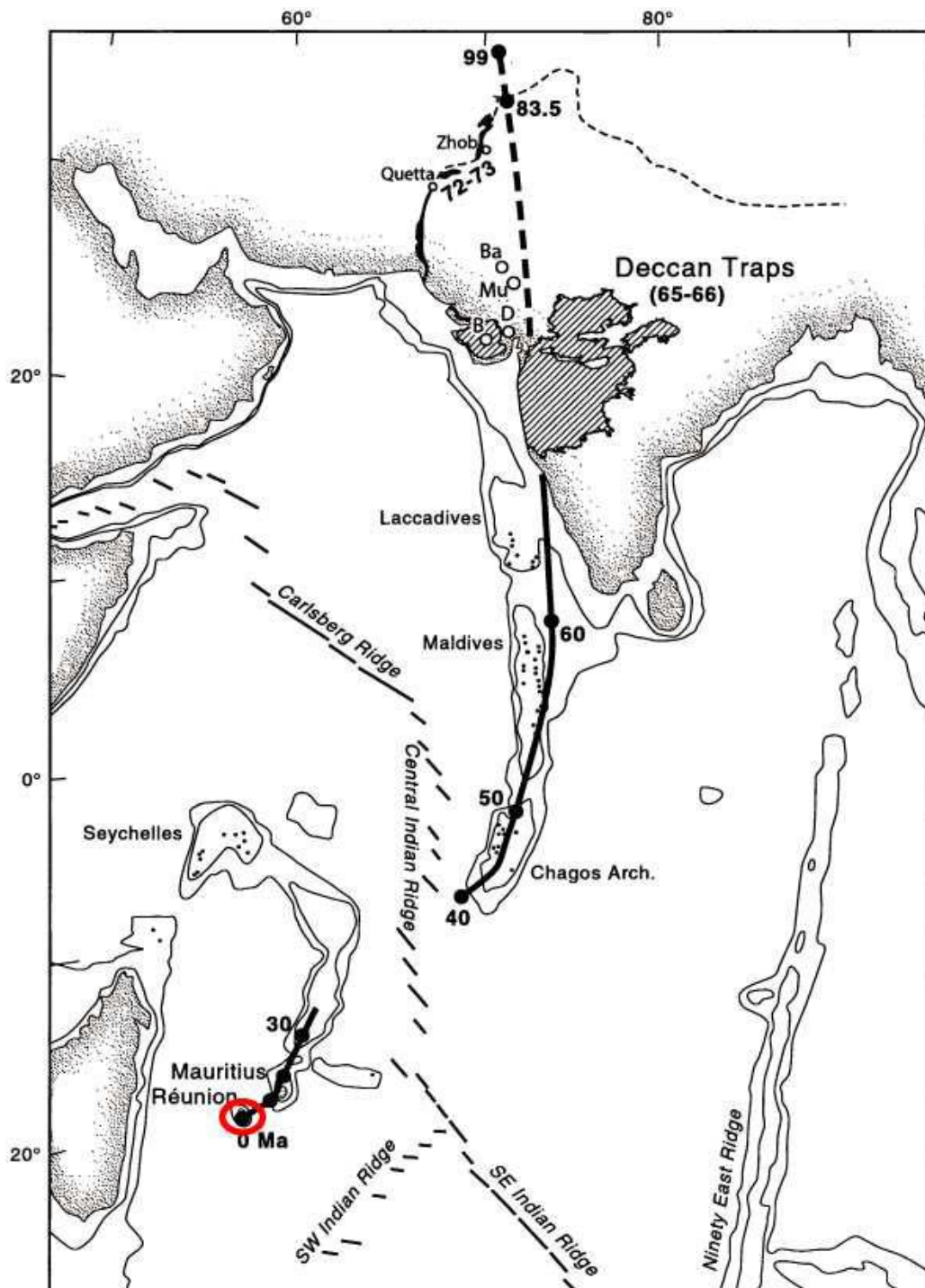


Fig.2.1.1 - The heavy line in the map indicates the model 0-99 Ma track of the Réunion hotspot, assuming a stationary hotspot and that plate reconstructions are accurate. The age is in Ma. The Réunion Island is indicated in the map with a red circle (Kent et al., 2002; Duncan, 1990).

The island of Réunion is situated 700 km east of Madagascar and is composed of two juxtaposed volcanic massifs, the oldest Piton des Neiges (in the north-west) and the youngest, presently active Piton de la Fournaise (Piton de la Fournaise, in the south-east; Fig. 2.1.2), respectively 3070 and 2600 m above sea-level. A third, poorly known buried volcano, named Les Alizés, has been recognized below

Piton de la Fournaise on the eastern submarine flank and was mainly defined through exploratory drilling. It constitutes the basement of Piton de la Fournaise (Gailler and Lénat, 2010; Lénat et al., 2001a; Malengreuet et al., 1999; Smietana, 2011; Smietana et al., 2010; Fig. 2.1.3).

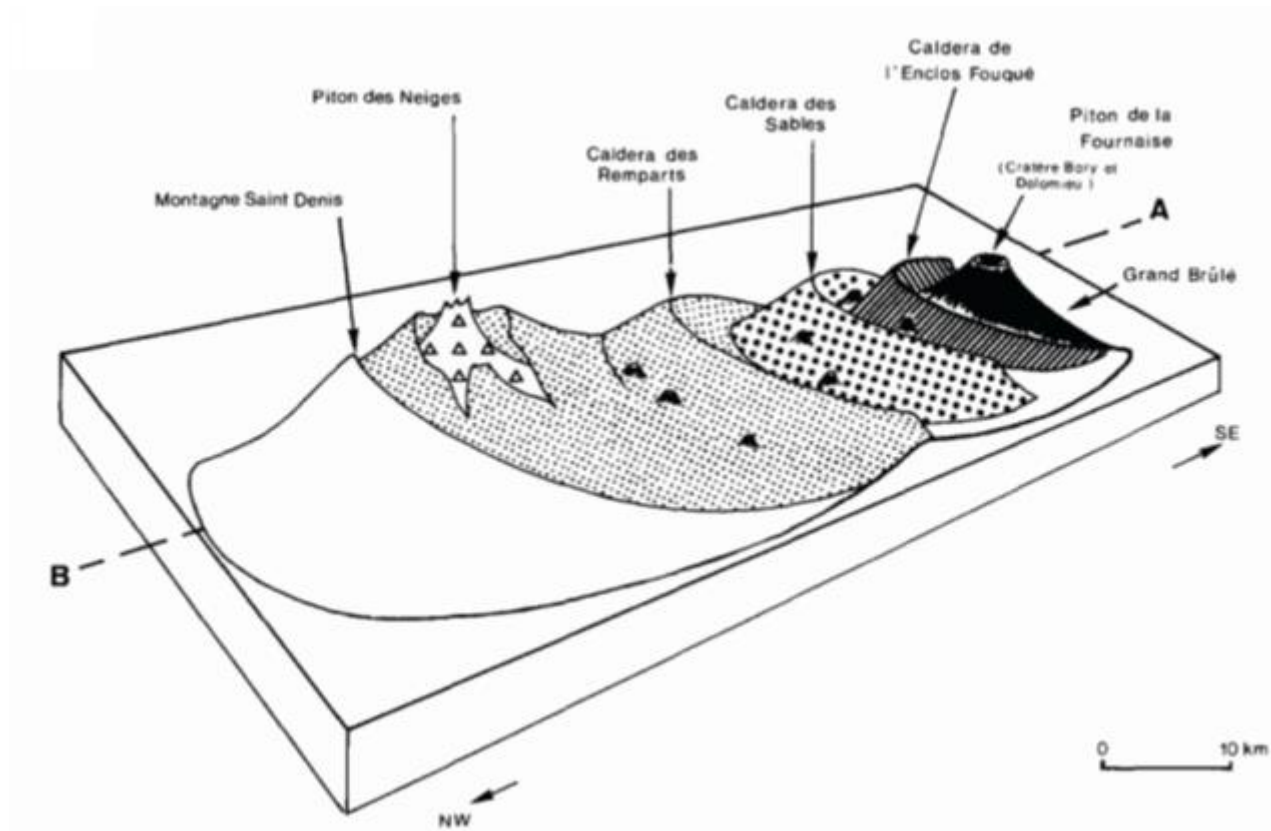


Fig. 2.1.2 - Evolutionary diagram of construction of La Réunion Island with the succession of nested collapses (from NW to SE) over time (Gillot et al., 1994; Smietana, 2011).

The whole edifice appears as a flattened cone, 220–240 km in diameter and 7 km in height, with a very large submarine base, about 210 km in diameter (Saint-Ange et al., 2011, 2013; Le Friant et al., 2011; Sisavath et al., 2011, 2012; Babonneau et al., 2013; Michon et al., 2016). Its contact with the quasi-flat sediments of the ocean bottom is clear, at depths of around -4000 m. In the transition zone between the lower flanks of the edifice and the abyssal plain, there are almost continuous sediment waves, concentric to the edifice, except where major canyons reach the abyssal plain (Le Friant et al., 2011; Oehler et al., 2008). These concentric features are classically attributed to turbidity currents (Le Friant et al., 2011; Wynn et al., 2000), but they could correspond to the folding of ductile pelagic sediments in response to compressive strain created at the front of advancing debris avalanches (Oehler et al., 2008). The volcanic edifice does not correspond to a perfectly shaped cone, although the base of the whole island is almost circular (Le Friant et al., 2011). The submarine cone is composed of four main fan-shaped bulges which are separated by major canyon systems (Labazuy,

1996; Oehler et al., 2008). In La Réunion, products of flank destabilization and erosion are widely distributed on the submarine slopes (Gailler and Lénat, 2010; Saint-Ange et al., 2013). These deposits are related to spreading, which implies slow deformation, and rapid mass movement, like debris avalanches or sediment transfer by shallow landsliding and turbidity currents (Oehler et al., 2008; Le Friant et al., 2011). A widespread extension of volcanoclastic products emphasize the importance of remobilization process on La Réunion Island. The part of the submarine flanks of Piton de la Fournaise that are directly built by lava-flows are only few portions. Among volcanoclastic deposits, large volume of debris avalanche deposits has been identified on the submarine flanks of Piton de la Fournaise (Oehler et al., 2004; Lénat et al., 2009; Le Friant et al., 2011; Saint-Ange et al., 2013). These deposits reveal the occurrence of old large flank collapse affecting Piton de la Fournaise.

The geological maps of La Réunion island (Oehler et al., 2008; Lebas, 2012) show surface rocks resulting from both constructional and destruction processes. On the submarine surface, few constructional features (cones, eroded piles of lava flows) have been recognized. Most of these features are located close to the coast in continuation of the on land southwest and northeast rift zones of Piton de la Fournaise. Pristine, small monogenic cones and flat-topped cones on the north and northeast flanks indicate small eruptions outside of the main volcano centres during the construction of La Réunion (Gailler and Lénat, 2010; Oehler et al., 2008). Debris avalanche deposits are the most extensive and voluminous formations in the submarine domain (Oehler et al., 2008). They have built four huge submarine bulges to the south, west, north and east of the island, forming fans 20-30 km wide at the coastline and 100-150 km wide at their ends, 70-80 km offshore. Submarine flanks have relatively gentle slopes ranging between 2° and 5° on average (Saint-Ange et al., 2011; Michon et al., 2016). In La Réunion, products of flank destabilization and erosion are widely distributed on the submarine slopes; this is the reason why the flanks of La Réunion are less steep than other volcanic islands, like Hawaii (10° to 15°; Michon et al., 2016). These deposits are related to slow deformation (spreading) and rapid mass movements, like debris avalanches or sediment transfer by shallow landslides and turbidity current (Oehler et al., 2008; Le Friant et al., 2011; Michon et al., 2016). The importance of remobilization processes on La Reunion Island results in a widespread extension of volcanoclastic products and a very wide submarine base of the volcanic edifice compared to the size of the island itself (Babonneau et al., 2016). The portion of the submarine flanks built by lava flows accumulations are rare. They essentially correspond to the submarine extensions of the subaerial NE and SE rift zones. The smooth surface that characterizes the shallowest part (from sea level to 1800 m deep) of the NE rift zone submarine extension is in a good morphologic continuity with the subaerial slopes. However, the incision of deepest relief

together obtained on pillow lavas dredged 1400 and 2100 m depth on this area (Smietana, 2011) clearly indicate that the NE rift zone is built on an older edifice, probably Les Alizés volcano (Lénat et al., 2001a).

La Réunion is built on Upper Cretaceous to Paleocene oceanic crust. Although the onset of La Réunion volcanism is unknown; it is probably younger than 10 Ma and therefore a thick sequence of oceanic sediments should already have blanketed the basaltic crust when the hotspot volcanism started. Seamounts were built up after formation of the oceanic crust and some are probably buried beneath La Réunion (Lénat, 2016).

Seismic studies show that there is virtually no or very weak lithospheric flexure beneath La Réunion, whereas a large body of underplating material has been demonstrated beneath the southwest part of the island (Charvis et al., 1999; Voogd et al., 1999; Gallart et al., 1999). This absence of flexure is supported by the gravity data (Lénat, 2016). The absence of a large flexure contrasts with most similar oceanic volcanoes and suggests a dynamic equilibrium where positive buoyancy of the plume would contribute to the dynamic support of the load of La Réunion. The scale models suggest that the most important factor controlling the buoyant flow of the plume material is the presence of mechanical discontinuities at the lithospheric scale. On the base of this geometry of the ocean floor, it has been estimated that the subaerial part of the edifice of La Réunion represents only 4% of the total volume of the whole volcanic system (Michon et al., 2016).

With different geophysical approach, the interior of the submarine flanks of La Réunion have been studied, showing negative Bouguer anomalies, which more or less mimics the four bathymetric bulges described above (Oehler et al., 2008).

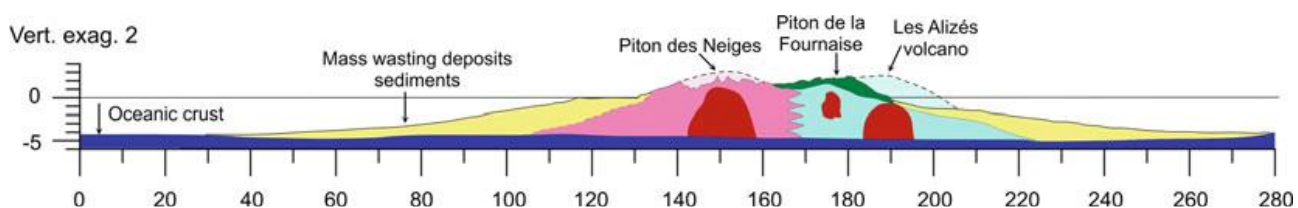


Fig. 2.1.3 - A summary view of the internal structure of La Réunion (Bachelery et al., 2015).

## 2.2 Piton des Neiges

Piton des Neiges, whose oldest recognized outcropping lava flows are 2.2 Ma old basalts (McDougall, 1971; Quidelleur et al., 2010), is located in the north-west of the island and is the largest of the three volcanoes. It has been dormant for less than 30 ka (Deniel et al., 1992; Gillot and Nativel, 1982; McDougall, 1971; Salvany et al., 2012). After a period of apparent quiescence, Piton des Neiges then erupted differentiated alkaline lavas between 350 and 29 ka.

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The evolution of Piton des Neiges volcano over more than 2 Ma, involving volcanic and volcano-tectonic activity and high erosion rates, has resulted in a complex geology exposed in deep valleys. The internal structure of Piton des Neiges volcano has been studied with several approaches. The main feature of the interior of the volcano is the presence of a huge intrusive complex which creates a large positive gravity anomaly. This complex emerges at various places at the bottom of the Salazie and Cilaos Cirques as gabbro outcrops; some syenites and micro-syenites are also observed (Berthod et al., 2020). Various models (Gailler and Lénat, 2012) suggest that the volcanic-subvolcanic complex may extend down to the base of the edifice and it is overlaid by lava flows, pyroclastics, landslide deposits and sills and dykes in the central part (Chevallier and Vatin-Pérignon, 1982; Maillot, 1999; Rançon, 1982).

### **2.3 Les Alizés**

Les Alizés, in the south-west of the island, is poorly known, as it constitutes the basement of the youngest Piton de la Fournaise volcano. The discovery of a large positive gravity anomaly near the eastern coast of Piton de la Fournaise (Gérard et al., 1980; Rousset et al., 1989), in the area called Grand Brûlé, produced a debate in the scientific community (Lebas, 2012; Salvany et al., 2012). A drill hole provided the proof that the structure responsible for the gravity anomaly was a hypovolcanic intrusive complex, mostly composed of gabbros and other cumulates (Rançon et al., 1989), encountered between about 1000 and 3000 m b.s.l. (bottom of the drill hole). The geothermal gradient and the relatively low temperature at the bottom of the drill hole ( $\approx 140^\circ$  C) indicated that the complex was old and inactive (Rançon et al., 1989). The 5-8 km wide complex is elongated in a north-south direction in plan view (15-18 km). Subsequent gravity measurements and modelling showed that the dense structure was independent from the structures of Piton de la Fournaise volcano (Gailler et al., 2009). Les Alizés volcano could have emerged at about the same period as Piton des Neiges (around 2 Ma), considering the age of about 3.5 Ma for pillows lava near 2000 m b.s.l. (Bachelery et al., 2015). Large flank landslides probably were responsible for the massive destruction of Les Alizés volcano (Merle et al., 2010; Oehler et al., 2008; Fig. 2.3.1). The reconstruction of the history of building and erosion/burial of Les Alizés volcano is one of the main issues in the understanding of the evolution of La Réunion, but large uncertainties remain.



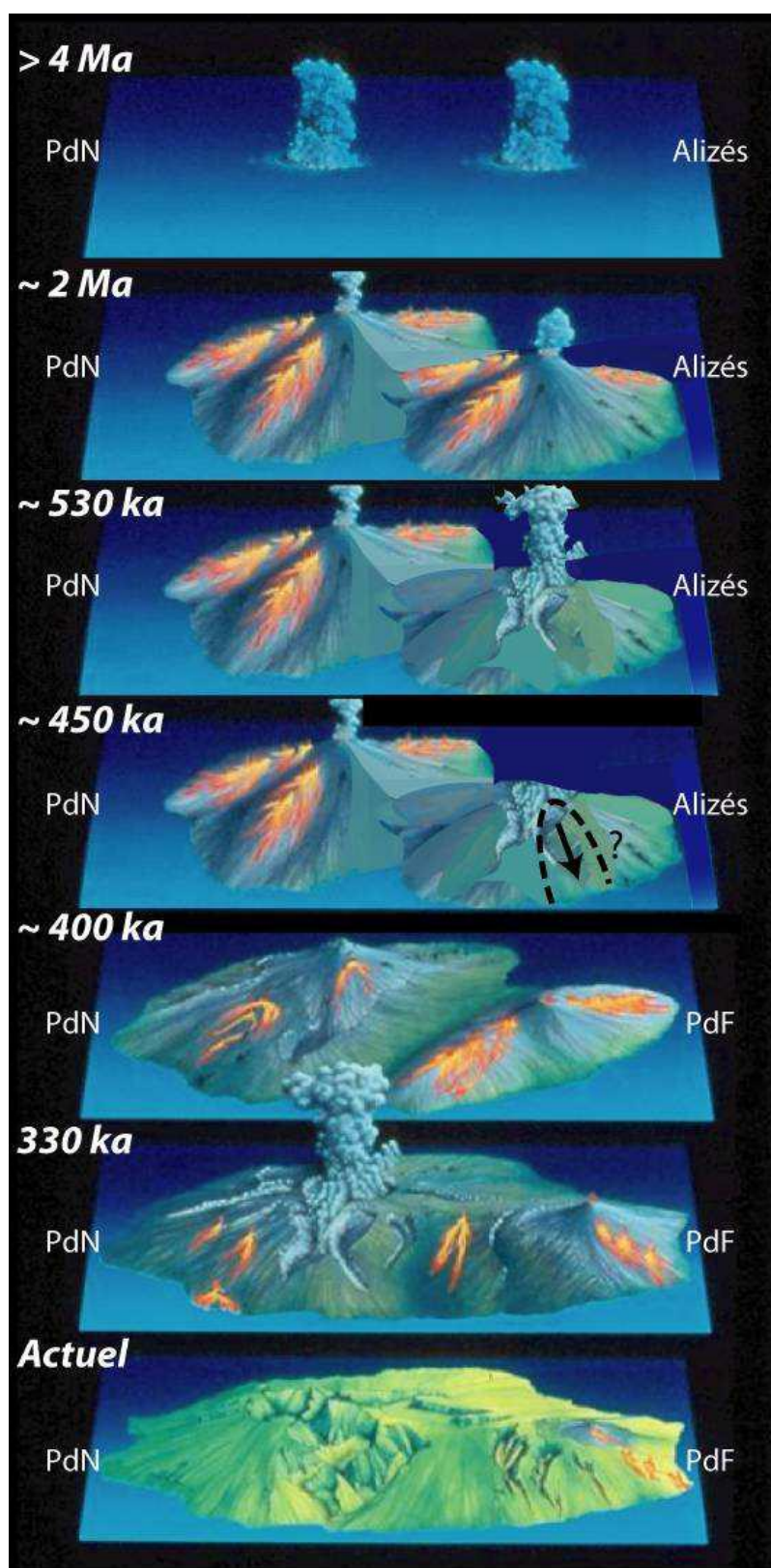


Fig. 2.3.1 - Iconography representing the different stages of construction of La Réunion island (Bachelery and Mairine, 1991; Smietana, 2011).

## 2.4 Piton de la Fournaise

Piton de la Fournaise occupies the south-eastern part of the island, and is a highly active basaltic shield volcano, with most of its activity restricted to effusions from its Central Cone and along rift zone (Michon et al., 2016). Deep valleys have been incised into the areas not covered by young volcanic deposits, providing exposures that enable the reconstruction of certain parts of the volcano's evolution. Piton de la Fournaise has been extensively studied using geological, petrological, geochemical and geophysical monitoring approaches (Fig. 2.4.1). Les Alizés volcano may have produced a large explosive activity occurred around 450 ka, causing a period of quiescence of Piton des Neiges and the birth of Piton de la Fournaise (Smietana, 2011).

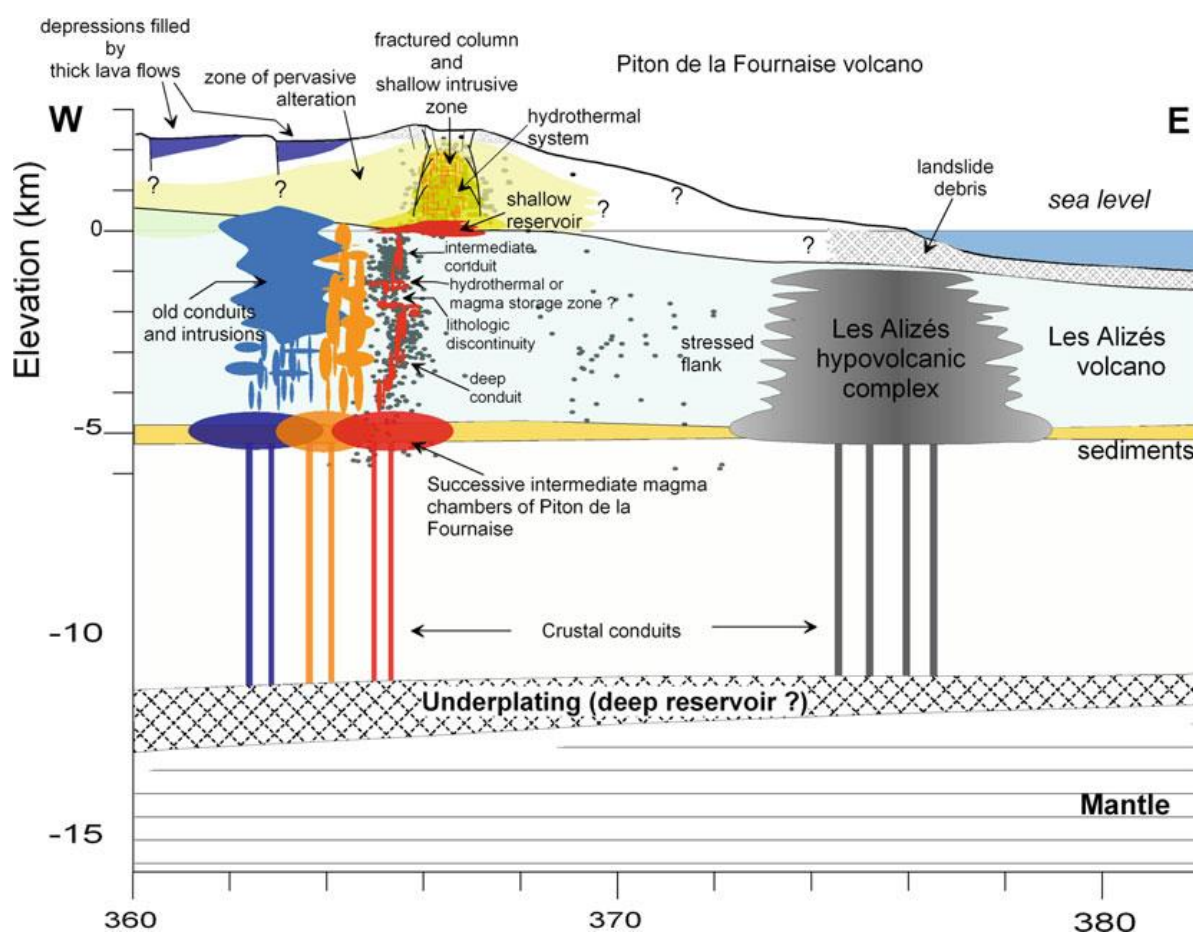


Fig. 2.4.1 - Interpreted W-E geological section of Piton de la Fournaise (Lénat et al., 2012). Horizontal coordinates: UTM km WGS84.

The oldest dated rocks of the Piton de la Fournaise volcano (up to 535 ka; Gillot and Nativel, 1989; Kluska, 1997) are found at the bottom of deep valleys (Rivière des Remparts, Rivière Langevin, Rivière de L'Est) and in La Plaine des Palmistes. Up to 400-450 ka the lavas are plagioclase-rich basalts (locally called "Roches Pintades Series"; Albarède et al., 1997), and they are significantly different from the subsequently emitted basaltic lavas of Piton de la Fournaise. Similar plagioclase-rich basalts are found in Piton

des Neiges, but they are about 200 ka younger than the Piton de la Fournaise ones (Bachelery et al., 2015 and reference therein).

The oldest on-land outcrops are dated at about 530 ka (Gillot and Nativel, 1989) and have been grouped into a single Pintades lava unit (Bachelery, 1981; Bachelery and Mairine, 1990; Mairine and Bachelery, 1997; Merle et al., 2010), observed at the base of all lava sequences in the deep valleys incising Piton de la Fournaise (Rivière de Remparts, Rivière de l'Est and Rivière Langevin). The pintades lava unit, consisting of feldspar-rich basalt to mugearite, does not belong to the Piton de la Fournaise series *stricto sensu* since it includes differentiated rocks, richer in alkali in bulk composition than typical Piton de la Fournaise lavas (Albarède et al., 1997; Luais, 2004; Smietana, 2011). It can represent the upper part of its basement, possibly related to the old Les Alizés volcano (Merle et al., 2010).

Chevallier and Bachelery (1981) proposed two coeval volcanic centres for this old phase of Piton de la Fournaise, one located north of the Plaine des Sables and the other one close to the current location of the summit of Piton de la Fournaise. The Ancient Piton de la Fournaise then experienced alternating episodes of volcanic destruction and reconstruction, destructive episodes of the edifice being characterized by giant landslides and summit caldera collapses (Merle et al., 2010). The oldest recognized landslide occurred at about 290 ka.

## **2.5 The Enclos Fouqué caldera and Bellecombe Ash Member**

The main eruptive centre was located at the same place of the today's Central Cone, but some researchers suggests the possibility of the occurrence of an intermediate location before activity attained the present one (Letourneur et al., 2008). The age of the eastward migration of the eruptive centre remains controversial (Merle et al., 2010). The eastward migration could have taken place at about 150 ka, concurrently with the formation of the Morne Langevin caldera (Bachelery and Mairine, 1990). Several collapse-related structures can be identified in the morphology of Piton de la Fournaise and have been interpreted as the scars of huge landslides or the scarps of collapse calderas. The youngest one corresponds to the polylobate U-shaped Enclos Fouqué caldera (which is opened in the east and connected to the Grand Brûlé depression). The horseshoe-shaped geometry was interpreted as the result of a huge flank landslide (Fig.2.5.1a; Bachelery, 1981; Chevallier and Bachelery, 1981; Duffield et al., 1982; Lénat et al., 1989; Labazuy, 1996; Merle and Lénat, 2003; Oehler et al., 2004, 2008; Michon and Saint-Ange, 2008). Another interpretation was that the Enclos Fouqué caldera results from coalescent caldera collapses and the Grandes Pentes correspond to the head of an eastwards directed flank landslide (Fig.2.5.1b; Bachelery, 1981). Two additional interpretations interpreted the Enclos Fouqué caldera as an hydrothermal collapse caldera triggered

by the lateral deformation of the hydrothermal system caused by a landslide (Fig.2.5.1c; Merle and Lénat, 2003 ) or a vertical collapse of the Grand Brûlé (Fig.2.5.1d; Michon and Saint-Ange, 2008). A second possibility is that it occurred after the Plaine des Sables caldera formed at about 65 and/or 40 ka (Bachèlery and Mairine, 1990; Gillot and Nativel, 1989; Merle et al., 2010; Staudacher and Allègre, 1993). A new edifice built east of the ancient volcanic centre after the collapse of the Plaine des Sables. The formation of the most recent 8 km-wide caldera, termed Enclos Fouqué caldera, stopped its activity. This event marked the most recent major volcano-tectonic event in the history of Piton de la Fournaise whose timing and dynamics remains unclear. Caldera formation has been initially associated with the emplacement of the Bellecombe Ash Member, a sequence of ash deposits cropping along the western area of the Enclos Fouqué caldera (Bachèlery, 1981; Mohamed-Abchir, 1996; Ort et al., 2014). An age of  $4745 \pm 130$  yr BP is usually assumed as an older limit for this caldera collapse event and the beginning of the Bellecombe paroxysmal explosive events (Mohamed-Abchir, 1996; Staudacher and Allègre, 1993). This age corresponds to a radiocarbon dating of the uppermost lava flows of the Bellecombe scarp (Bachèlery, 1981). The recent reappraisal of the existing radiocarbon data, enriched by new dating, suggests that Bellecombe ashes groups several explosive events, whose age ranges between  $4880 \pm 35$  yr BP and  $2340 \pm 30$  yr BP (Morandi et al., 2016; Bachèlery et al., 2015). Seismological data indicate that above 7.5 km below sea level, the plumbing system is almost exclusively restricted to the Enclos Fouqué caldera (Nercessian et al., 1996; Aki and Ferrazzini, 2000; Battaglia et al., 2005; Prôno et al., 2009). The magmatic activity related to this part of the volcanic system progressively led to the building, on the floor of the Enclos Fouqué caldera, of the 400-m-high steep Central Cone capped by the Bory crater and the Dolomieu caldera. This cone and to a lesser extent the Enclos Fouqué caldera are cut by hundreds of eruptive fissures that formed during the recent history of Piton de la Fournaise.

A recent reappraisal of historical reports combined with new geological observations suggests that the historical activity of the Central Cone was characterized by a lava lake activity centred on the current inactive Bory crater (Michon et al., 2013). This lava lake fed the largest recent pahoehoe lava field, the Enclos Fouqué lava field (Lénat et al., 2001), between 1733 and 1750 CE.

The main tephra identified all around the Enclos Fouqué is the "Bellecombe Ash unit". It corresponds to a complex sequence of ash, crystal and lithic-rich deposits (Bachèlery, 1981; Mohamed-Abchir, 1996). This deposit is thought to be emplaced during violent phreatic/phreatomagmatic events related to the flashing of an extensive hydrothermal system during the collapse of the Enclos Fouqué caldera (Bachèlery, 1981; Fontaine et al., 2002; Upton et al., 2000). The ages of the caldera formation and related Bellecombe Ash unit are poorly constrained and are likely comprised in a time

span ranging between the emplacement of the lavas forming the uppermost part of the Enclos Fouqué walls that underlay Bellecombe and Partage ashes ( $4745 \pm 130$  BP,  $3340 \pm 1012$ ) and that of the crystal and lithic-rich Bellecombe and Partage ashes themselves (several ages between  $4175 \pm 145$  and  $2320 \pm 90$  BP). That ashes with age of  $2140 \pm 80$  BP drape the lower and inner wall of Bellecombe cliff (Vergniolle and Bachèlery, 1982). This possibly represents the upper age for a series of caldera-forming events. Most of the eruptive products, lavas and tephras, cropping out inside the Enclos Fouqué caldera are much younger than 2320 BP, actually younger than 1750 CE, final age of the widespread pahoehoe lava field named "champ de lave de l'Enclose Fouqué" (CLEF) by Lénat et al. (2001).

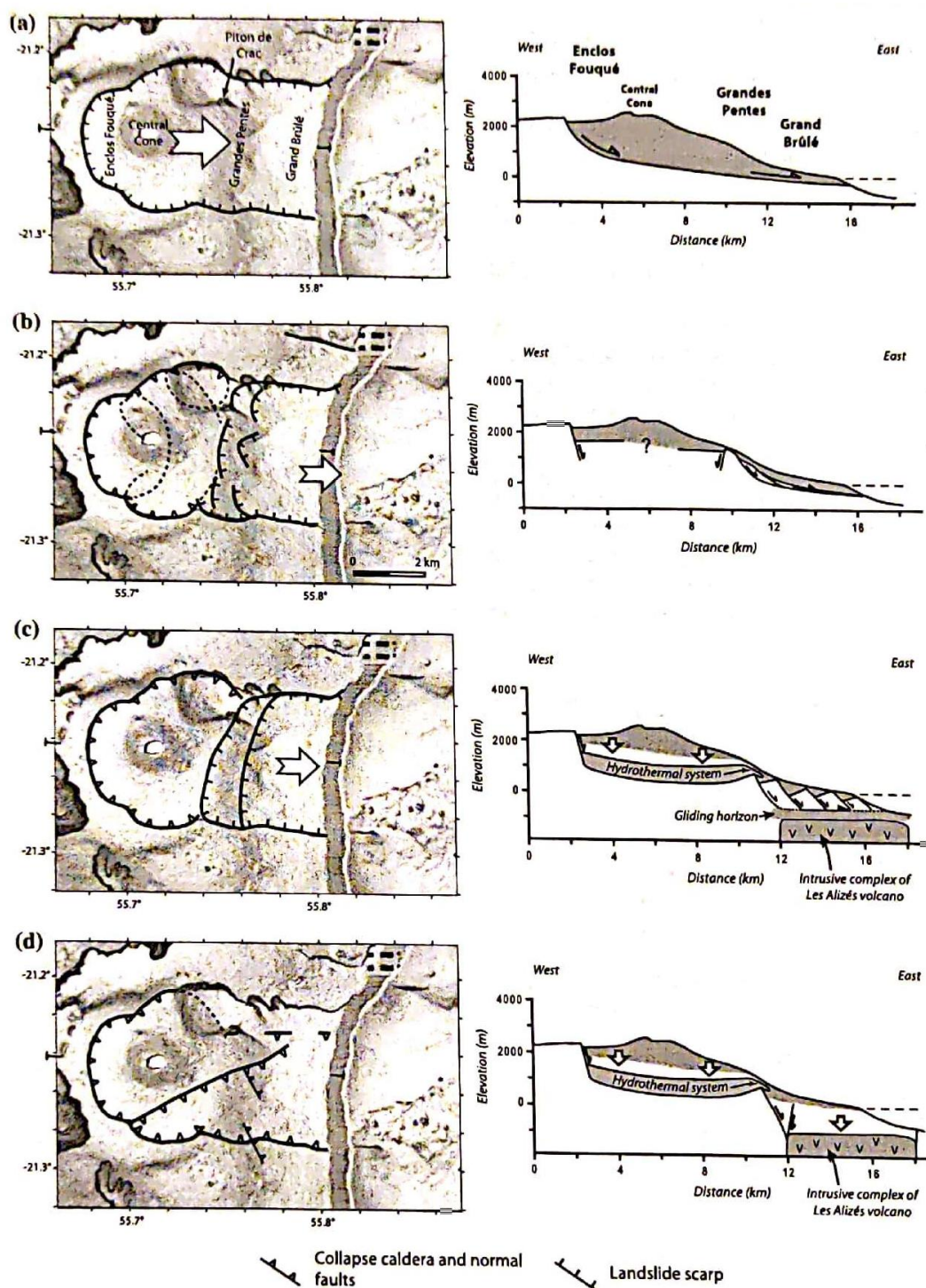


Fig. 2.5.1 - Interpretations of the horseshoe-shaped structure composed of the Enclos Fouqué, the Grand Brûlé and the Grandes Pentes areas (Michon et al., 2016). a) The U-shaped structure as the result of a huge flank landslide; b) The Enclos Fouqué caldera results from coalescent caldera collapses and the Grandes Pentes correspond to the head of an eastwards directed flank landslide; the Enclos Fouqué caldera as an hydrothermal collapse caldera triggered by the lateral deformation of the hydrothermal system caused by c) a landslide or d) a vertical collapse of the Grand Brûlé.

The Plaine des Sables is another larger structure well developed west of the Enclos Fouqué, limited by a north-south trending scarp.

This structure would result from a vertical collapse of the edifice (Bachelery, 1981), a large flank landslide toward the east (Duffield et al., 1982; Gillot et al., 1994; Oehler et al., 2004, 2008), or the deformation of the hydrothermal system in a way similar to the one proposed for the Enclos Fouqué (Merle and Lénat, 2003).

Geological data suggest the occurrence of older collapse structures related to the Ancient Piton de la Fournaise (older than 60 ka). Evidenced, also, a twofold construction of Piton de la Fournaise. Between 530 and 60 kyrs, the volcanic centre located in the current Plaine des Sables led to the building of the western part of the massif. The volcanic centre migrated eastwards to its current location, possibly at 60-40 kyrs (Bachelery and Mairine, 1990; Gillot and Nativel, 1989; Merle et al., 2010; Staudacher and Allègre, 1993). Then Piton de la Fournaise experienced caldera collapses and recurrent phreatomagmatic eruptions especially between 4880 and 2340 yr BP as evidence by the Bellecombe ash deposits (Michon et al., 2016). Most of the recent volcanic activity is now currently focused restricted inside the Enclos Fouqué caldera where lava flow accumulation and rare explosive events built the 400-m-high Central Cone.

### **3 The erupted magmas of Piton de la Fournaise**

#### **3.1 Magmas series and magma types**

The presence of a geothermal well in the eastern flank of the Piton de la Fournaise in the Grand Brulé area ( $x = 187.823$ ,  $y = 38.646$ ,  $z = 172$  m) located on an oceanite flow erupted in 1961 allowed the study of Piton de la Fournaise products for the first  $\sim 4$  km. The exploration geothermal well was drilled in 1985 by the French Company for Geothermal and Renewable Energy Development (CFG) on the eastern flank of the Piton de la Fournaise volcano. The study of the section represent the sequence of rocks intersected through the entire depth of the borehole, from ground level (0 m, 172 m a.s.l.), to the bottom (3003.5 m). Twelve principal units have been distinguished representing the essential lithofacies of the succession (Rancon et al., 1985):

- 0-108 m: subaerial lava flows, massive to scoriaceous olivine basalt and oceanite. The vesicles in some flows are occupied by zeolites and calcite.
- 108-390 m: subaerial lava flows cut by intrusions. Olivine basalt and rare oceanite cut by dykes and/or sill of the same composition.
- 390-440 m: subaerial flows, represented by scoriaceous to massive olivine basalt and aphyric basalt, little altered except for the presence of quartz, calcite and zeolites in the vacuoles.

- 440-556 m: subaerial-submarine transition zone. Subaerial flows of olivine basalt and aphyric basalt and submarine flows of vitric olivine basalt.
- 556-616 m: submarine basalt flows. Olivine basalt and aphyric basalt with thick argillaceous intervals and fragments of vitric basalt most probably representing altered hyaloclastic breccia.
- 616-890 m: subaerial flows, comprising various scoriaceous olivine basalt and aphyric basalt.
- 890-1010 m: submarine flows, predominantly microlitic and vitric rocks.
- 1010-1129 m: lavas intruded by gabbro. Sheets of leucogabbro and biotite melagabbro have been injected into aphyric, locally vitric, basalt.
- 1129-1920 m: layered gabbro: alternations several meters thick of melagabbro and microleucogabbro.
- 1920-2468 m: submarine lavas with basic to ultrabasic intrusions, essentially sheets of melagabbro and microgabbro injected into vitric basalt and hyaloclastite, with intervals several meters thick of peridotite (Ol+Cpx+Fe-Ti and Cr oxides).
- 2468-2820 m: wherlite and dunite. A massive intrusive ultramafic sequence with sparse intervals of several decimetres thick of altered microgabbro.
- 2820-3003.5 m: massive dunite with layers of unaltered, relatively plagioclase-rich gabbro. The fractures in the gabbros are filled with calcite-zeolite-quartz-chlorite assemblages and veinlets of serpentine.

Lavas of Piton de la Fournaise are subdivided into four different compositional groups, which correspond to the activity inside and outside the Enclos Fouqué caldera:

The Steady State Basalt (SSB) is the most abundant group in the recent activity of the volcano. These lavas are transitional basalts with a narrow range of compositions, of MgO = 5-8 wt% and CaO/Al<sub>2</sub>O<sub>3</sub> = 0.75-0.85 (Albarede et al., 1997; Famin et al., 2009; Villemant et al., 2009; Salaun et al., 2010; Fig. 3.1.1). SSB lavas host olivine, clinopyroxene, plagioclase and spinel in variable but generally low proportions (the latter mainly as inclusions in olivine; Pichavant et al., 2016).

Picritic lavas, also termed oceanites (Lacroix, 1936), are characteristic of the volcanic activity of la Réunion Island, being found at Piton de la Fournaise but also at Piton des Neiges. The composition of these lavas range from MgO = 8-28 wt% and CaO/Al<sub>2</sub>O<sub>3</sub> = 0.75-0.85. They contain millimetric to centimetric crystals of olivine embedded in a cryptocrystalline to glassy groundmass. Oceanites of La Réunion Island are much studied but their significance is still not completely clear. Oceanite whole rock



compositions clearly plot on olivine control lines (Salaun et al., 2010) suggesting that their chemistry reflects the proportion of olivine crystals present in the rock. An important observation is that the composition of matrix glasses in oceanites is nearly constant and similar to SSB bulk rocks (Welsch et al., 2009). Most workers consider oceanites result from a general mechanism of shallow incorporation of olivine crystals into a basaltic parental magma. Di Muro et al. (2014) shows the heterogeneous composition and origin of recycled olivines and constrain the olivine accumulation at shallow depth (>1 kbar).

The Differentiated Lavas group has been defined mainly from differentiated lavas erupted near the end of activity of Piton des Neiges (Upton and Wadsworth, 1966). These lavas are characterized by a strongly decreasing CaO and increasing Al<sub>2</sub>O<sub>3</sub> at decreasing MgO (MgO = 2-6 wt%, CaO/Al<sub>2</sub>O<sub>3</sub> = 0.2-0.8; Fig.3.1.1). Differentiated alkaline lavas are found during the early stages of activity of the Piton de la Fournaise (Albarede et al., 1997) but are uncommon among recent products. This group includes chemically evolved terms such as hawaiites, mugerites and benmoreites and, in the recent activity, rare trachytic pumices (Vlastelic et al., 2011; Vlastelic et al., 2021). Recent study (Smietana, 2011) has suggested that the early-differentiated alkaline lavas (the Pintade unit) do not belong to the Piton de la Fournaise series but rather should be associated with the activity of the older Les Alizés volcano.

The Abnormal Group (AbG) is compositionally different from the three others. It comprises lavas with MgO = 8-11 wt% and CaO/Al<sub>2</sub>O<sub>3</sub> = 0.60-0.80, and shows enrichments in Fe, Mg, Ti, Na, K and depletion in Ca relative to SSB basalts. AbG-type compositions are rarely observed inside the Enclos Fouqué caldera, being mostly found at eccentric vents or at recent vents west to the central cone (Lénat et al., 2012). This group of lavas has got mixed geochemical features, some suggesting primitive character (e.g. high MgO) and others indicating a differentiated evolution similar to that shown by the Differentiated lavas group (e.g. low CaO/Al<sub>2</sub>O<sub>3</sub> and high K<sub>2</sub>O). It seems clear, however, that the Abnormal Group lavas does not evolve directly from Steady State Basalts-like parental liquids (Bachelery et al., 2015 and reference therein).

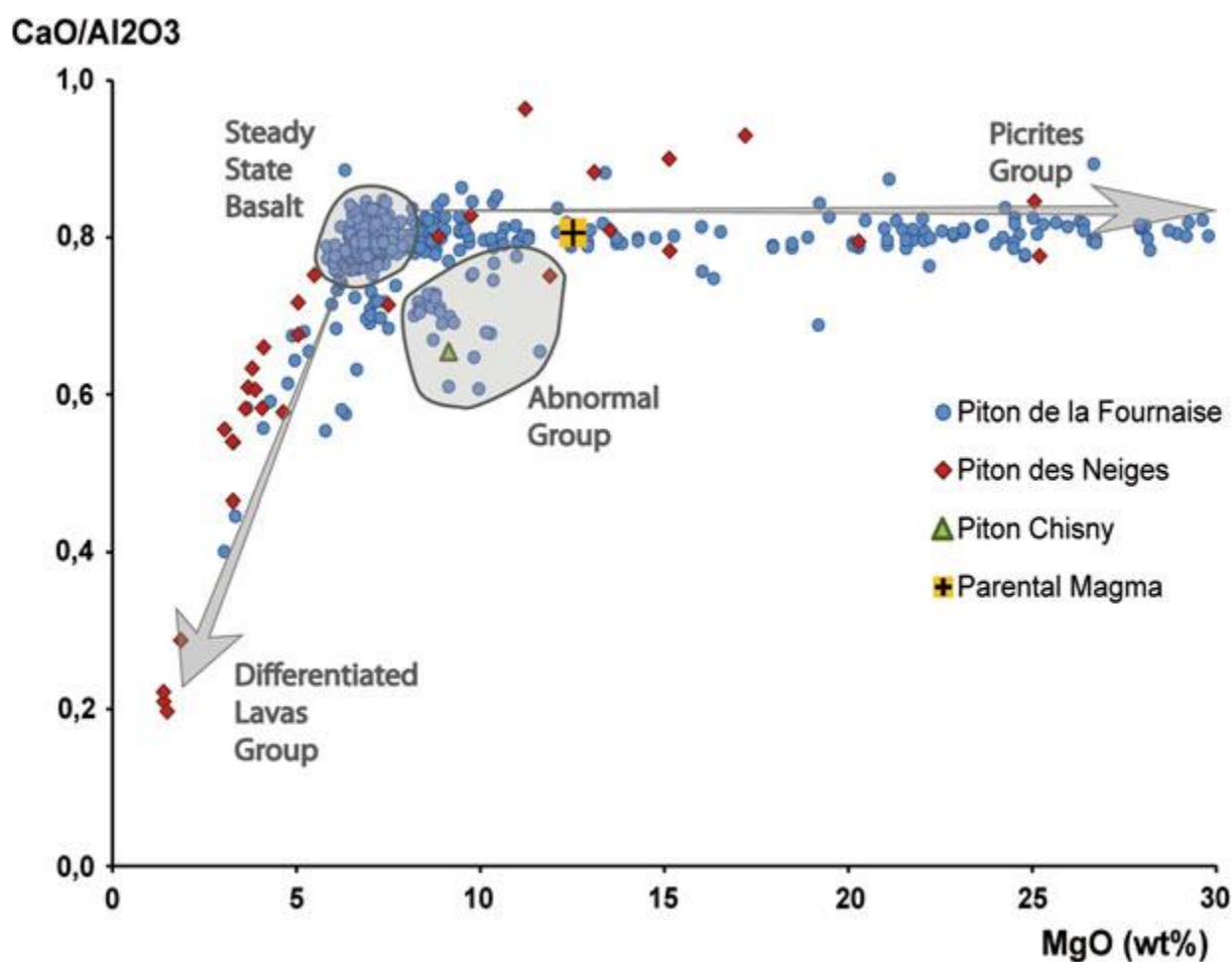


Fig. 3.1.1 -  $\text{CaO}/\text{Al}_2\text{O}_3$  vs.  $\text{MgO}$  plot for Piton de la Fournaise and Piton des Neiges magmas. The four different magma groups and one representative sample of parental magma are shown (Bachelery et al., 2015 and reference therein).

Piton de la Fournaise basalts range from tholeiitic to alkaline series. He, Nd and Sr isotopic homogeneity of La Réunion lavas has been emphasized (Fisk et al., 1988; Graham et al., 1990). Some geochemical variabilities has been interpreted to reflect melting of different source components intrinsic to La Réunion mantle plume, and superimposed crustal processes including magma contamination and fractionation. Clinopyroxene and olivine fractionation ("wherlitic trend") has been proposed to generate Abnormal Group compositions (Kornprobst et al., 1979, 1984) and to drive tholeiitic basalts in the field of alkaline basalts (Albarède et al., 1997).

#### 4 Phase relationships

Olivine, clinopyroxene, plagioclase and Cr-spinel are found systematically in Piton de la Fournaise products. Ti-rich magnetite is also present, together with sulphides. Ilmenite and amphibole are rare (Albarède et al., 1997). The four main crystalline phases are found in a variety of textural types, occurring as phenocrysts, microphenocrysts and microlites (Salaun et al., 2010; Welsch et al.,

2009, 2013). However, the characteristics of the mineral phase assemblage are specific of the distinct lava groups.

SSB lavas host olivine, clinopyroxene, plagioclase and spinel in variable but generally low proportions. Textural evidence suggests that, in SSB, clinopyroxene and plagioclase crystallize together with olivine, this lava group being often designated as "cotectic". The olivine + plagioclase + clinopyroxene + spinel four phase assemblage is also typical of the AbG Group. Early AbG lavas from Piton Chisny are crystal-poor (< 2 vol%) and have only phenocrysts of olivine and Cr-spinel.

Oceanite lavas of the Picrite group contain macrocrysts of olivine, mesocrysts of either olivine or an assemblage of clinopyroxene and plagioclase, and microcrysts of olivine + clinopyroxene + plagioclase + spinel ± Ti-magnetite (Welsch et al., 2009).

Piton de la Fournaise basalts generally contain small amounts of plagioclase. However, plagioclase is the dominant phase in the Differentiated Lava group from Piton des Neiges and early Piton de la Fournaise (Upton and Wadsworth, 1966; Albarede et al., 1997). Olivine compositions at Piton de la Fournaise are rarely highly forsteritic. Histograms of Fo contents yield frequency maxima at 86 (early activity; Albarede et al., 1997), 81 and 84 (1977-1998 activity; Boivin and Bachèlery, 2009; Villemant et al., 2009), 85-86 (Chisny lavas; Brugier, 2012) and 83 and 88 for both the Chisny ultramafic cumulate nodules (Brugier, 2012) and the dunite nodules (Sobolev and Nikogosian, 1994).

Texturally, the occurrence of deformed olivines has led to the inference that many crystals at Piton de la Fournaise are xenocrysts of earlier cumulates (Albarède et al., 1997). Clinopyroxene compositions are mostly augite, some extending into the fields of diopside and salite (Boivin and Bachèlery, 2009; Welsch et al., 2009). Boivin and Bachèlery (2009) distinguished two pyroxene compositional groups, one with an average composition  $\text{En}_{49}\text{Fs}_9\text{Wo}_{42}$  (Mg-augite) and the other  $\text{En}_{48}\text{Fs}_{12}\text{Wo}_{40}$  (augite). They suggested that the former group, with higher Al/Ti ratios, crystallized at higher pressures than latter. Plagioclase in recent (1977-1998) Piton de la Fournaise products ranges from  $\text{An}_{45}$  to  $\text{An}_{80}$  (Boivin and Bachelery, 2009). Spinel (both inclusions in olivine and isolated crystals dispersed in the matrix) are mostly chromites with Cr/(Cr + Al) around 70 and Mg/(Mg +  $\text{Fe}_t$ ) between 50 and 65. They have  $\text{TiO}_2$  contents between 2 and 3 wt%, increasing with decreasing Mg/(Mg +  $\text{Fe}_t$ ), and  $\text{Fe}^{3+}/(\text{Fe}^{3+} + \text{Al} + \text{Cr})$  of 10-15 % (Brugier, 2012). Spinel compositions at Piton de la Fournaise do not show systematic variations with magma types (Sobolev and Nikogosian, 1994; Albarède et al., 1997; Welsch et al., 2009; Brugier, 2012).

Data on matrix glass concerns the SSB and Picrite groups and, to a lesser extent, the AbG group. The composition of the matrix glass in most Piton de la Fournaise products is similar to SSB bulk rocks ( $\text{MgO} = 5.5\text{-}6$  wt%,  $\text{CaO}/\text{Al}_2\text{O}_3 = 0.79\text{-}0.82$ ); it is in equilibrium with

olivines in the range Fo<sub>75-80</sub> (Welsch et al., 2009). Matrix glasses with MgO contents as high as 8–9 wt% have been analysed in 2007 products (Villemant et al., 2009; Di Muro et al., 2016). The 1998–2007 matrix glasses define a unique evolution trend starting from these high-MgO glasses and ending at ~5.5 wt% MgO (Villemant et al., 2009; Salaun et al., 2010).

Melt inclusions are common in olivine crystals from Piton de la Fournaise. Primary inclusions are either glassy or recrystallized. Secondary inclusions are mainly fluid inclusions. These occur along healed fractures in olivine and are associated with glass and crystals in variable proportions (Sobolev and Nikogosian, 1994; Bureau et al., 1998a; Famin et al., 2009). Compositions of primary glassy inclusions (recalculated for post-entrapment crystallization in olivine) range from 8 to 10 wt% MgO (Bureau et al., 1998a), 8–11.5 wt% MgO (Famin et al., 2009) to up to ~14 wt% MgO (Sobolev and Nikogosian, 1994; Di Muro et al., 2016).

H<sub>2</sub>O and CO<sub>2</sub> concentrations in glass inclusions are generally < 1 wt% and < 500 ppm (Bureau et al., 1998b; Famin et al., 2009). However, certain samples have H<sub>2</sub>O and CO<sub>2</sub> concentrations largely exceeding these values, with H<sub>2</sub>O up to ~ 1.6 wt% and CO<sub>2</sub> up to ~ 2800 ppm (Bureau et al., 1998b; 1999). The secondary fluid inclusions contain essentially pure CO<sub>2</sub> with a large range of densities, from 0 to 0.8 g/cm<sup>3</sup> (Bureau et al., 1998; Famin et al., 2009; Di Muro et al., 2016).

## 5 Thermobarometric constraints and oxygen fugacity

Information on temperature of Piton de la Fournaise magmas comes from different methods. First, optical pyrometry and thermocouple measurements in the field provide direct data on lava temperatures, from 872°C and up to 1170°C with a frequency maximum between 1100 and 1150°C. Second, magma temperatures have been computed from olivine-liquid equilibrium. When applied to olivine phenocryst cores and their host bulk lava, grouped values are obtained, close to 1200°C for the 1977–1998 products (Boivin and Bachèlery, 2009). As a variant of this method, the MgO content of the melt (from the analysis of glass inclusions) at equilibrium with olivine yields temperature (e.g. Helz and Thornber, 1987) between 1170 and 1215°C (Bureau et al., 1998a, b) and between 1192 and 1236°C (Famin et al., 2009; Di Muro et al., 2016). It should be noted however that results with methods based on glass inclusion compositions depend on procedures adopted to correct for post-entrapment olivine crystallization. Third, thermometric measurements on melt inclusions have yielded homogenization temperatures from 1170 to about 1215°C (Bureau et al., 1998).

Oxygen fugacities between NNO+0.7 to NNO-1.8 have been computed from the olivine-liquid equilibrium (Boivin and Bachèlery, 2009). Most values are however <NNO, and so Piton de la Fournaise magmas evolve under moderately reducing conditions (Bachelery et al., 2015 and

reference therein). Pichavant et al. (2016) reported a  $f_{O_2}$  value of the Piton de la Fournaise magmas which is reasonably well defined, around NNO-0.5. This makes the average  $Fe_2O_3/FeO$  of basaltic liquids at Piton de la Fournaise slightly higher than commonly assumed (the proportion of  $Fe_2O_3$  is 18% and not 15%; e.g. Fisk et al., 1988). Experiments by Fisk et al. (1988) have provided a framework for interpreting the genesis and evolution of Piton de la Fournaise magmas. The Fisk et al. (1988) data imply minimum pressures of 250-500 MPa for clinopyroxene to crystallize before plagioclase. Fractionation of olivine together with clinopyroxene would also require a minimum of 250-500 MPa from their data. Still higher crystallization pressures (a minimum of 800-1000 MPa for compositions < 11 wt% MgO) would be necessary for fractionation of clinopyroxene alone. Assuming that all Piton de la Fournaise primitive magmas crystallize in the sequence olivine-clinopyroxene-plagioclase (Albarede et al., 1997; Welsch et al., 2009), then the Fisk et al. (1988) experimental results would lead to the conclusion that magma fractionation takes place for the most part at the base of the crust or in the mantle. The crust-mantle boundary beneath La Réunion is located at 12-15 km depths (Gallart et al., 1999; Lénat et al., 2012) or 320-400 MPa. Hence, the 250-500 MPa minimum pressure required by the Fisk et al. (1988) data makes it possible that crystallization of either olivine + clinopyroxene or olivine + clinopyroxene + plagioclase takes place at the base of the crust. It is worth noting that both melt and fluid inclusions are consistent with fluid-melt saturation and fluid entrapment at crustal pressures, and that petrological evidence for crystallization pressures above 400-500 MPa (Bureau et al., 1998, 1999; Di Muro et al., 2016) is presently lacking.

These difficulties lead to an alternative interpretation of the Fisk et al. (1988) data, reminding that their experiments were performed under volatile-free conditions. Numerical runs using the MELTS software have been performed for a parental Piton de la Fournaise magma composition (Famin et al., 2009; Welsch et al., 2009), in presence of 1 wt%  $H_2O$  and 0.1 wt%  $CO_2$  and for  $f_{O_2}$  set at the FMQ buffer (about 0.7 log unit below NNO). They show that clinopyroxene crystallizes second after olivine at 100 MPa (Welsch et al., 2009), at pressures much lower than the 250-500 MPa minimum pressure range found by Fisk et al. (1988). Nevertheless, recent study (Boudoire et al., 2019) individuate in clinopyroxene phenocrysts from "clinopyroxene-bearing" lavas from peripheral eruptions a minimum average pressure of  $3.1 \pm 0.6$  kbar. The highest pressures are recorded in clinopyroxene from dunite-wherlite enclaves erupted as lithics from the polygenetic Chisny cone, with an average of  $5.6 \pm 0.8$  kbar.

## 6 Methods

In this chapter, all the methods used to investigate the mafic and ultramafic ejecta of the Bellecombe Ash Member as well as some quartz xenocrysts were described.

### **6.1 Inductively Coupled Plasma Optical Emission Spectroscopy and Inductively Coupled Plasma Mass Spectroscopy (ICP-OES and ICP-MS)**

Samples were crushed and finely powdered in agate mortar. Major and trace element analysis of bulk-rocks were carried out at the Centre de Recherches Pétrographiques et Géochimiques (CNRS-SARM) in Nancy (France) by Inductively Coupled Plasma - Optical Emission Spectrometry (ICP-OES) and Inductively Coupled Plasma - Mass Spectrometry (ICP-MS; see Carignan *et al.* (2001) for a review of the protocols and of reference materials).

Relative errors for major elements are < 1 % (SiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub>), < 2 % (MgO, FeO and CaO), < 5 % (Na<sub>2</sub>O, K<sub>2</sub>O and TiO<sub>2</sub>). Relative trace element errors are < 5 % with exception for Sc (< 15%) and Th, Cr (< 10%). The analytical results were reported in the appendix (see chapter 17).

### **6.2 Electron Micro Probe Analyses (EMPA)**

An electron probe micro-analyser is a microbeam instrument used primarily for the *in situ* non-destructive chemical analysis of minute solid samples. EMPA is also informally called an electron microprobe, or just probe. Electron micro-probe analyser is an analytical instrument largely diffused to determine the quantitative chemical composition of small volume (few  $\mu^3$ ) of sample in non-destructive form, primarily by wavelength-dispersive spectroscopy. Analyses of single phase minerals (olivine, clinopyroxene, feldspars, oxides, biotite and interstitial or glass inclusion) were performed with a Cameca SX100 microprobe at University of Hannover, operated at accelerating voltage of 15 kV and 15 nA beam current. The beam was focused to 1 $\mu$ . Na and K were analysed first with counting times of 8 s to minimize alkali volatilization. Counting time for other elements was 10 s. Standards used for calibrations were natural mineral standards. During each microprobe session, analytical precision and accuracy was verified by measuring the mineral standards. The analytical results were reported in the appendix (see chapter 17).

### **6.3 Isotopic and elemental composition of light noble gases and CO<sub>2</sub> analysis**

The isotopic and elemental composition of light noble gases (He, Ne, and Ar) and CO<sub>2</sub> was determined in fluid inclusions (FI) hosted in crystals of olivine, clinopyroxene and feldspars at Istituto Nazionale di Geofisica e Vulcanologia, Sezione di Palermo (INGV - Palermo). These crystals were separated by handpicking under a binocular microscope from the 1-0.5 mm-sieved fraction of the mafic and ultramafic subvolcanic ejecta of the Bellecombe Ash Member (Piton de la Fournaise). The crystals were carefully separated (0.4 - 1.9 g), cleaned in an ultrasonic bath following internal protocol (Rizzo *et al.*, 2018). These crystals were split into two aliquots: the first

was loaded into a stainless-steel crusher capable of holding up to six samples simultaneously for noble-gas analysis, and the second was used for determining the concentration and isotope ratio of CO<sub>2</sub>. Noble gases trapped inside primary (when present) and secondary fluid inclusions were released by in-vacuo single-step crushing at about 200 bar. This procedure is the most conservative in order to minimize the contribution of cosmogenic <sup>3</sup>He and radiogenic <sup>4</sup>He possibly grown/trapped in the crystal lattice (Kurz, 1986; Hilton et al., 1993, 2002; Rizzo et al., 2018). The CO<sub>2</sub> measurement was performed during noble-gas extraction at the time of crushing by quantifying the total gas pressure (CO<sub>2</sub> + N<sub>2</sub> + O<sub>2</sub> + noble gases) and subtracting the residual pressure of N<sub>2</sub> + O<sub>2</sub> + noble gases after removing CO<sub>2</sub> using a "cold finger" immersed in liquid N<sub>2</sub> at -196°C. Gases were then cleaned in an ultra-high vacuum (10<sup>-9</sup> - 10<sup>-10</sup> mbar) purification line, and all the species of the gas mixture except noble gases were removed. Helium (<sup>3</sup>He and <sup>4</sup>He) and neon (<sup>20</sup>Ne, <sup>21</sup>Ne and <sup>22</sup>Ne) isotopes were measured separately by two different split-flight-tube mass spectrometers (Helix SFT-Thermo). The values of the <sup>3</sup>He/<sup>4</sup>He ratio are expressed as R/Ra (where Ra is the <sup>3</sup>He/<sup>4</sup>He ratio of air, which is equal to 1.39\*10<sup>-6</sup>). The analytical uncertainty of the He isotopic ratio is generally < 1.2%. The R/Ra values were corrected for atmospheric contamination based on the <sup>4</sup>He/<sup>20</sup>Ne ratio (Sano and Wakita, 1985) and are expressed hereafter as Rc/Ra values. Argon isotopes (<sup>36</sup>Ar, <sup>38</sup>Ar and <sup>40</sup>Ar) were analyzed by a multicollector mass spectrometer (GVI Argus), with an analytical uncertainty generally < 1.1%. The uncertainty in the determinations of the elemental He, Ne, and Ar contents was < 5% after considering natural variability and the assumption of residual (not crushed) crystals for weight normalization of the number of moles of gas. Typical blanks for He, Ne and Ar were < 10<sup>-15</sup>, < 10<sup>-16</sup> and < 10<sup>-15</sup> mol, respectively. <sup>40</sup>Ar values were corrected for atmospheric contamination (<sup>40</sup>Ar\*) according with the following formula:  $^{40}\text{Ar}^* = ^{40}\text{Ar}_m - ((^{40}\text{Ar}/^{36}\text{Ar})_{\text{air}} \times ^{36}\text{Ar}_m)$ , where <sup>40</sup>Ar\* represents the corrected <sup>40</sup>Ar, and the "m" subscript indicates "measured".

Quantities of noble gases are reported either in moles of gas or as cm<sup>3</sup> STP (Standard Temperature and Pressure); it should be noted that although STP is usually defined as 273.15 K and 101.325 kPa (= 1 bar), some definitions use 100.000 kPa (McNaught and Wilkinson, 1997). As a consequence, noble gas concentrations are usually expressed as moles (or cm<sup>3</sup> STP) per gram of rock or fluid. Sometimes these are converted to percent or ppm of the noble gas by weight, as is common for other elements in many branches of geochemistry (Burnard et al., 2013).

#### **6.4 Scanning Electron Microscopy Backscattered Electron Energy Dispersive Spectroscopy (SEM-BSD-EDS)**

The Scanning Electron Microscopy Backscattered system equipped at the Department of "Scienze della Terra e dell'Ambiente" of University of Pavia was only used on wherlitic sample B28 and on layered

ultramafic to mafic sample B108a, for the investigation of the different textures and compositions. Polished thin section B28 and B108a were analysed by a Scanning Electron Microscopy Tescan series 3xMU operating in backscattered Electron imaging mode equipped with EDAX Energy Dispersive Spectrometer (Fe-SEM-EDS) for major elements analyses, using a 20 keV accelerating voltage. Beam current and spot size were 0.4 nA and 3µm respectively. The counting times on peak and background were 100 s. Maps of samples were surveyed using 20 keV, a working distance of 15.8 mm, 32 frame, resolution of 1024x800 and max counts per channel of 256. The analytical results both for samples B108a and B28 were reported in the appendix (see chapter 17).

### **6.5 Laser Ablation Inductively Coupled Plasma Mass Spectroscopy (LA-ICP-MS)**

The Laser Ablation-Inductively Coupled Plasma-Mass Spectrometer (LA-ICP-MS) system equipped at the Department of "Fisica e Geologia" of University of Perugia was only used for the investigation of the quartz xenocrysts. The instrument is composed by a commercial New Wave UP213 (New Wave, UK) LA system coupled with a Thermo Electron X7 (Thermo Electron Corporation, Waltham, USA) ICP-MS. The laser ablation system is a frequency quintupled Nd:YAG laser, whose fundamental wavelength of 1064 nm is converted into 213 nm by means of three harmonic generators. The commercial ablation cell (NewWave, UK), capable to support both thin sections and cylindrical resin chips, has been modified in order to mount together both reference materials and unknown samples. In addition, special adjustments have been developed on the ablation cell for the lodging of 'large' samples (up to c.a. 5 cm). Helium is preferred over argon as a carrier gas to enhance the transport efficiency of ablated aerosol (Eggins et al., 1998). The helium carrier gas exiting the ablation cell is mixed with argon make-up gas before entering the ICP torch; this configuration permits the maintenance of stable and optimum excitation condition. LA-ICP-MS measurements are carried out using time resolved analysis operating in a peak jumping mode. The laser repetition rate and the laser energy density are fixed to 10 Hz and ~10 J/cm<sup>2</sup>, respectively. The LA-ICP-MS system is optimized for dry plasma conditions prior to each analytical session on a continuous linear ablation of NIST SRM 612 glass standard by maximizing the signals for selected masses (La<sup>+</sup> and Th<sup>+</sup>) and reducing oxide formation by minimizing the ThO<sup>+</sup>/Th<sup>+</sup> ratio. For the operating conditions outlined above, background intensities are typically 20-200 cps for Sc, V, Cr and Ga, 10-20 cps for Rb, Sr and Pb and < 10 cps for all other elements. Analyte sensitivities vary as a function of the amount of material reaching the torch, the ionization efficiency, isotope abundance, and mass. For the operating conditions outlined above and for a laser beam of 40 µm sensitivities range between 690 and 6230 cps/µg\*g<sup>-1</sup>.



## 6.6 Raman spectroscopy analyses

This is a versatile non-destructive technique for fluid inclusion analysis. For barometric studies on representative phases of mafic and ultramafic samples, Raman spectra were acquired on polished crystals at the Ludwig Maximilian University of Munich (LMU Lab. of crystallography; Resp. of the Raman lab: Dr. M. Kaliwoda) using an XPlora One spectrometer (Jobin Yvon) equipped with an Olympus microscope at 100x magnification. Analytical conditions were: excitation wavelength: 532 nm, filter: 50%, hole 100, slit 50, gratings 1800, acquisition time: 150s x 2.

Raman spectroscopy was tentatively used on the quartz xenocrysts. It has a wide field of application ranging from qualitative detection of solid, liquid and gaseous components to identification of polyatomic ions in solution. Raman technique is commonly used to calculate the density of CO<sub>2</sub> fluids, the chemistry of aqueous fluids, and the molar proportions of gaseous mixtures present as inclusions. The main advantages of this technique are the minimal sample preparation and the high versatility. The analyzed samples of quartz xenocrysts are both doubled-polished thin sections (Q1, Q13, Q16 and Q19) and crystals *tal quale* (Q14, Q15, Q17 and Q18).

The analyses were performed by the confocal integrated micro/macro-Raman spectrometer "Labram HR800" (HORIBA JOBIN YVON) of the "Centro interdipartimentale G.Scansetti" (Department of "Scienze della Terra", University of Torino). Excitation line at 532 nm (solidstate Nd laser and 180 mW of emission power) was used, with a grating of 600 grooves/mm and a slit ranging from 100 to 300  $\mu\text{m}$ ; the corresponding spectral resolution was 2-4  $\text{cm}^{-1}$ . Each spectrum was collected in confocal setting with a hole of 200  $\mu\text{m}$ . The laser was focussed on the sample using an Olympus BX41 microscope. An objective 100x (spot size resolution of ca. 1  $\times$  1  $\times$  3  $\mu\text{m}$ ) was used on thin sections, whereas an objective 20x (spot size resolution of ca. 5  $\times$  5  $\times$  15  $\mu\text{m}$ ) was used on sample *tal quale*. Each spectrum was collected by 2-5 accumulations of 20-120 s. The calibration of the instrument has been checked daily using the 521  $\text{cm}^{-1}$  silicon peak and the 1332  $\text{cm}^{-1}$  diamond peak (precision  $\leq$  1  $\text{cm}^{-1}$ ). Spectra were acquired using Labspec5 software in the 100- 1800 and 3100-4200 $\text{cm}^{-1}$  domains.

## 6.7 X-ray powder diffraction analyses

X-ray powder diffraction is a very versatile technique. In qualitative and quantitative analyses of unknown materials, the analyst must reduce preferred orientation so that the diffraction pattern can be reduced and matched to standards for identification. Materials with equal crystal shapes and no cleavages that produce flat samples usually yield random samples using any specimen preparation method. If the particles have any crystallographically related shape, the shape will make achieving randomness difficult. As the difference between the maximum and minimum dimensions of the particle increases, it becomes more difficult to make random samples.

The easiest way to reduce particle size effects is to grind the sample in a mortar and pestle and obtain sample powdering. After that, the powder of sample were put into a sample holder, commonly made of aluminium, with a depression or cavity in which to mount the sample. In a diffractometer, the X-rays are generally diffracted from the top of the sample. The cavities are usually designed so the surface of the sample is on the diffractometer axis. It is the half-depth of the penetration of the X-ray beam in the sample that determines the effective diffracting surface and hence the specimen displacement error that shifts the resultant peak position. The X-ray penetration into a specimen will depend on incident angle, absorption, and elemental composition of the sample. Very low average atomic number compounds allow considerable penetration of the X-ray beam, resulting in peak displacement and broadening.

The oscillating electric field of a light wave will interact with the electrons in matter to cause coherent scattering. Each scattering point maybe treated as a new source of spherical waves. Thus, waves scattering from two objects will expand in space round the objects until they interfere with each other. Their interaction will produce constructive interference at certain angles of view and destructive interference at other angles depending on the distance between the scatterers and the wavelengths of the radiation. When a periodic array of objects each scatter radiation coherently, the concerted constructive interference at specific angles is called diffraction (Jenkins and Snider, 1996).

X-ray diffraction analyses were performed at the X-ray laboratory of University of Urbino "Carlo Bo" on some xenocrysts supposed to be quartz. For the X-ray diffraction analyses was used the side-loading technique. The type of holder employed has a rectangular hole punched through it. The first step is to attach a microscope slide to the top surface using scotch tape or clamps. The holder is then turned over and the powder carefully loaded into the cavity. The cavity is filled by applying gentle pressure to the powder as the cavity is filled up. A cover is then placed over the holder again so that the microscope slide is uppermost. When the specimen is to be exposed, the top microscope slide is carefully removed.

## 7 The subvolcanic ejecta of the Bellecombe Ash Member and comparisons with literature

### 7.1 Petrography

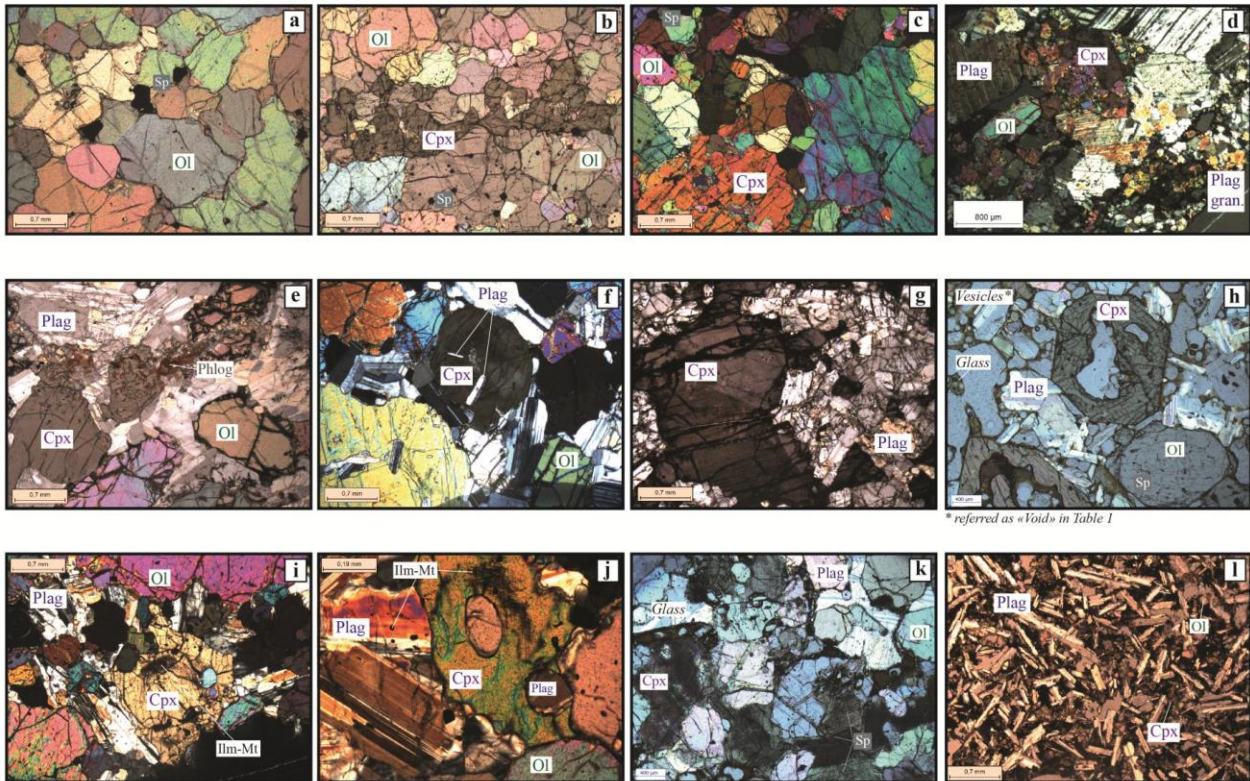
The petrographic study of more than one hundred samples (109), representative of the mafic and ultramafic subvolcanic ejecta of the Piton de la Fournaise emphasizes the presence of several lithotypes (see Table 7.1.1; all tables with the analytical results were reported in the appendix on chapter 17). Cumulate cognates *stricto sensu* are present, often characterized by abundant quenched glass entrapped interstitially and in minerals (melt inclusions), with no disequilibrium textures induced into the subvolcanic crystal frameworks. Wager et al. (1960) recognized that such a mechanism of enlargement of cumulus crystals at constant temperature could only take place at or near the top of any pile of crystals. They called this style of growth *adcumulus* growth. The *adcumulus* process gradually reduced the volume of intercumulus liquid by mechanically pushing it out of the pile and could result in vanishingly small quantities being preserved. Any intercumulus liquid remaining as the result of continued accumulation of crystals was termed trapped liquid which crystallized to the pore material. This liquid, and the subsequently crystallized pore material, necessarily had the composition of the contemporary magma. By contrast, orthocumulates are characterized by zoned cumulus minerals and a variety of postcumulus material, representing crystallization during cooling from original intercumulus liquid trapped interstitial pore spaces.

A wide range of textures in these ejecta results from crystal-liquid fractionation, variable P-T gradients in the subvolcanic environment of crystallization and/or reactions (peritectic?) between continuously rising basaltic magmas and the intrusive piles of products.

These clasts also allow to compare the compositions of the interstitial glass in the subvolcanic ejecta with small volumes of evolved liquids, which are trachytes from petrographic observation, detected as pumiceous components in the products of Bellecombe Ash Member, Langevin caldera and Piton des Neiges.

Other subvolcanic clasts (e.g. doleritic gabbros) may simply represent slowly-cooled equivalents of basalts, rather than cumulates, with no crystal-liquid fractionation.

Brugier (thesis, 2016) investigated some different lithotypes of mafic and ultramafic ejecta at La Réunion Island (Fig. 7.1.1): ol-rich cumulates, Cpx-rich veins in banded wherlite, massive gabbros with phlogopite-bearing veins from Piton des Neiges, gabbroic xenoliths from Piton Chisny.



\* referred as «Void» in Table 1

Fig. 7.1.1 - Photomicrographs of representative thin sections of the different rock types; from Brugier et al. (2016). (a) Cumulative texture of rounded olivine in dunite xenolith. Note the interstitial Cr-spinels (from Plaines des Sables). (b) Clinopyroxene-rich vein in banded wherlite (from Plaines des Sables). (c) Large (cm size) Cpx in wherlite (from Plaines des Sables). (d) Gabbroic xenolith from Piton Chisny. (e) Phlogopite-bearing vein in the Piton des Neiges massive gabbro. (f) Massive gabbro showing zoned Cpx with Plg inclusions (from Salazie Cirque). (g) Gabbroic inclusion from a Piton des Neiges ignimbrite. (h) Gabbroic xenolith from the Bellecombe Tephra. (i) Gabbroic xenolith from Piton Haüy. (j) Plg inclusions in Cpx and the crystallographically-oriented ilmenite exsolutions (from Plaines des Sables). (k) Cpx phenocrysts strongly darkened by Cr-spinel exsolutions in gabbroic xenolith (from Plaines des Sables). (l) Dolerite sample (from Plaines des Sables).

In Piton des Neiges samples from cirque of Salazie eroded intrusive complex (Berthod et al., 2020) were recognized several lithotypes (Fig. 7.1.2 and 7.1.3), both ultramafic (Site 1 and Site 2: dunite, wherlite, plg-wherlite) and mafic (Site 3 and Site 4: olivine gabbros, gabbros, ferrogabbros).

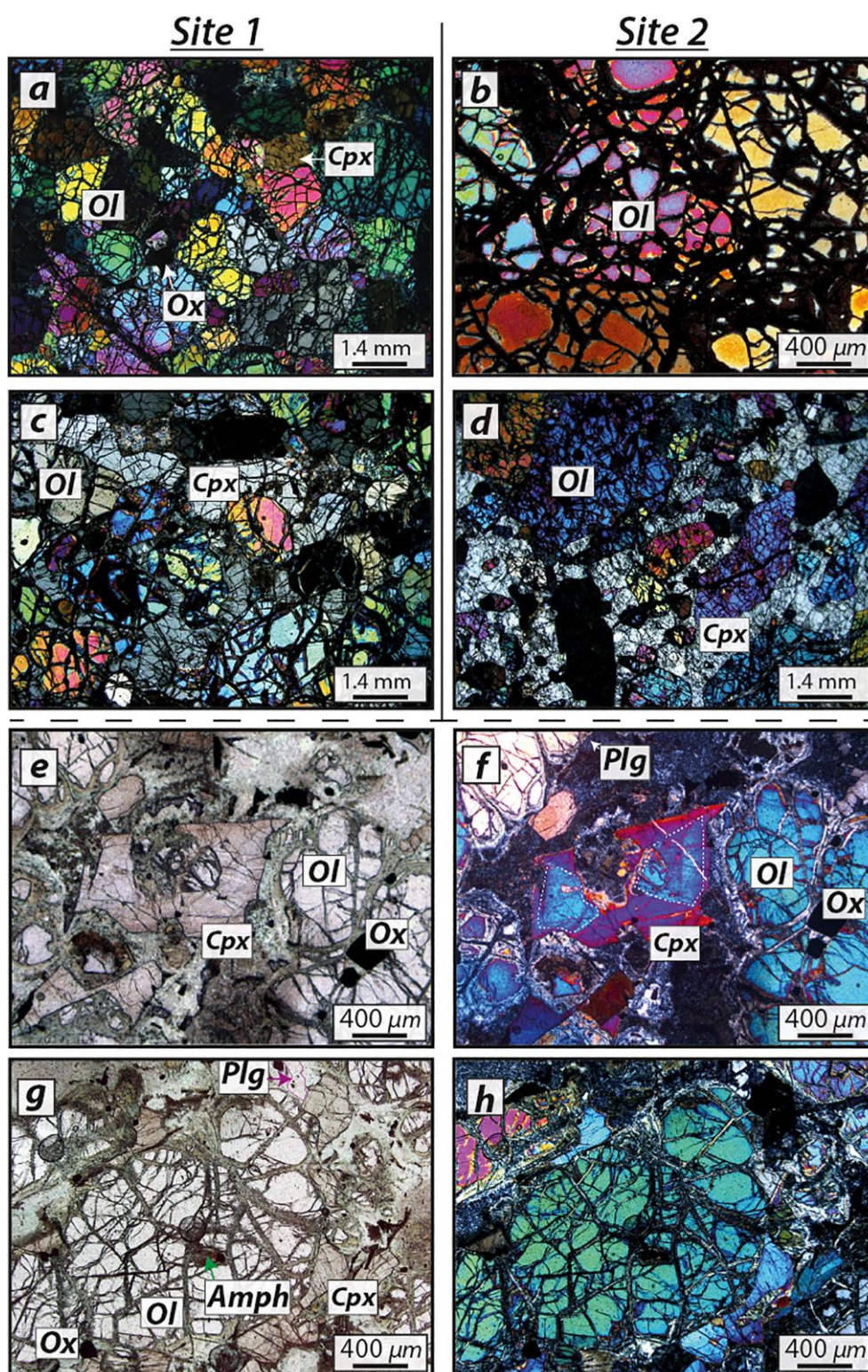


Fig. 7.1.2 - Representative parageneses and textures of samples collected at Cirque of Salazie eroded intrusive complex; from Berthod et al. (2020). Sites 1 and 2. a) Dunite from Site 1 made of anhedral grains of olivine in cumulus position, and anhedral crystals of clinopyroxene and oxides in interstitial position. b) Dunite from Site 2 with only olivine crystals. c) and d) Wherlite samples from Sites 1 and 2, respectively, in which the olivine is euhedral and in cumulus position, and clinopyroxene is poikilitic. e) and f) Wherlite from Site 2 made of rounded grains of olivine and euhedral, sector-zoned crystals of clinopyroxene in cumulus position, and plagioclase in interstitial position. g)

and h) Plagioclase-wherlites from Site 2 displaying a mesocumulate texture with euhedral olivine in cumulus position, and clinopyroxene and plagioclase in interstitial position.

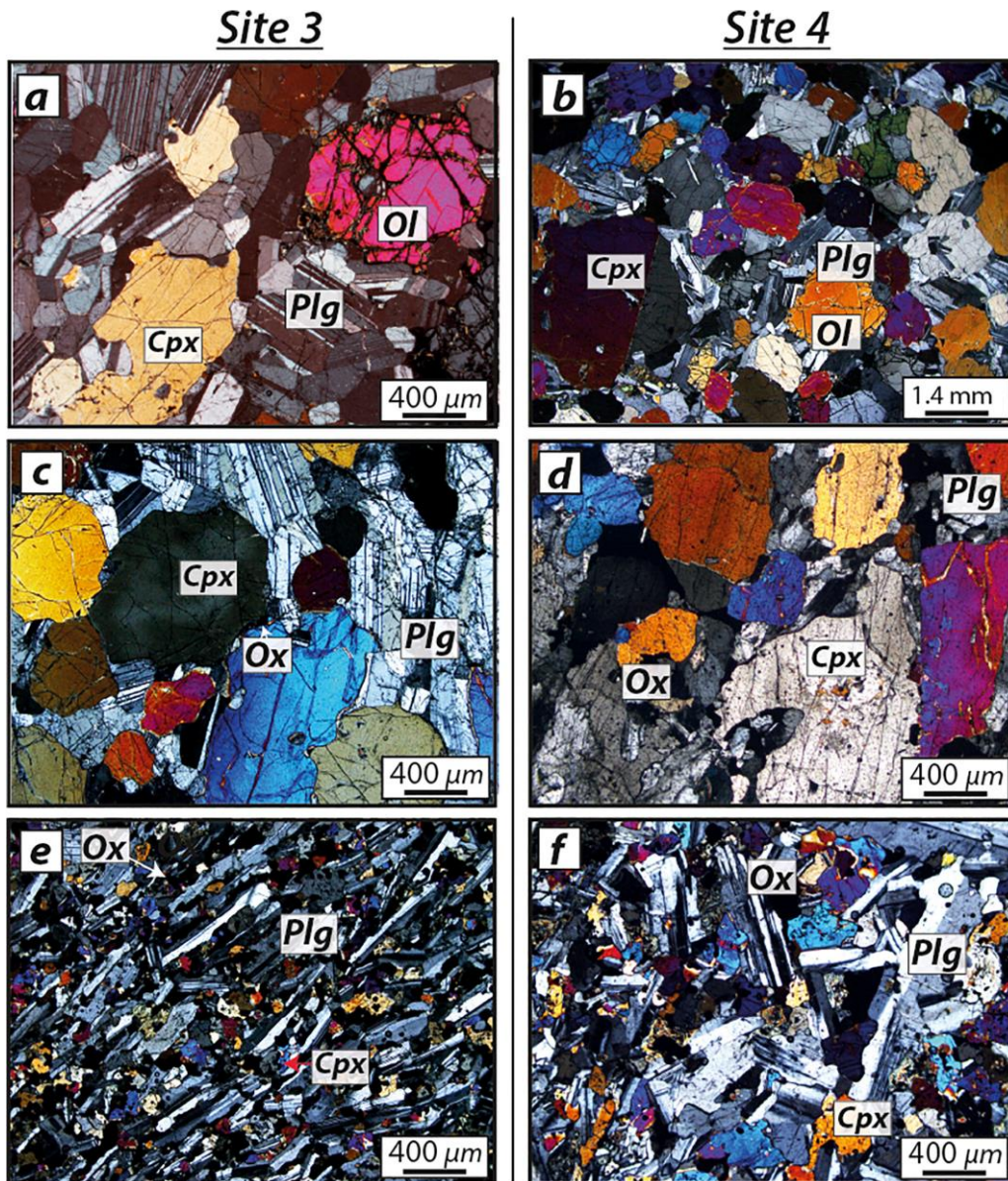


Fig. 7.1.3 - Representative parageneses and textures of samples collected at Sites 3 and 4; from Berthod et al. (2020). a) and b) Olivine gabbros. c) and d) Gabbros made of euhedral crystals of plagioclase and clinopyroxene. e) and f) Strong magmatic fabrics, marked by elongated plagioclases in ferrogabbros.

Samples of the Bellecombe Ash Member studied in the present thesis were subdivided in ultramafic (dunite and wherlite) and mafic (ophitic gabbros, sub-ophitic gabbros, poikilitic gabbros, dolerites, micro-monzogabbros and porphyrogabbros) ejecta. Among the samples sent to Urbino by Andrea Di Muro (Institute de Physique du Globe de Paris - Observatoire Volcanologique du Piton de la Fournaise) some of them (both from Bellecombe and Langevin caldera) are represented by lava lithic clasts and therefore considered as "volcanic" samples in Table 7.1.1. Samples sent to Urbino from La

Réunion Island also comprise several quartz xenocrysts (or supposed to be quartz) from the Langevin caldera and a sedimentary (carbonatic) clast from the Bellecombe Ash Member.

The modal mineralogy of the Bellecombe Ash Member mafic and ultramafic subvolcanic samples is reported in Fig. 7.1.4. Comparing these mineralogical data with literature (Berthod et al., 2020; Fig. 7.1.5), Upton et al. (2000; Fig. 7.1.6) and Brugier et al. (2016; Fig. 7.1.7) the widespread presence of plagioclase in the Bellecombe Ash Member intrusives (gabbroic rocks), can be emphasized if compared to the gabbroic rocks described by Berthod et al. (2020) from Piton des Neiges. The presence of opaque minerals is reported in several samples from Berthod et al. (2020) and Brugier et al. (2016), but are absent in Upton et al. (2000) samples. Only one dunite from Berthod et al. (2020) show amphibole, whereas the samples here investigated show biotite (< 0.1 vol%) in three samples (two porphyrogabbros and one sub-ophitic gabbro).

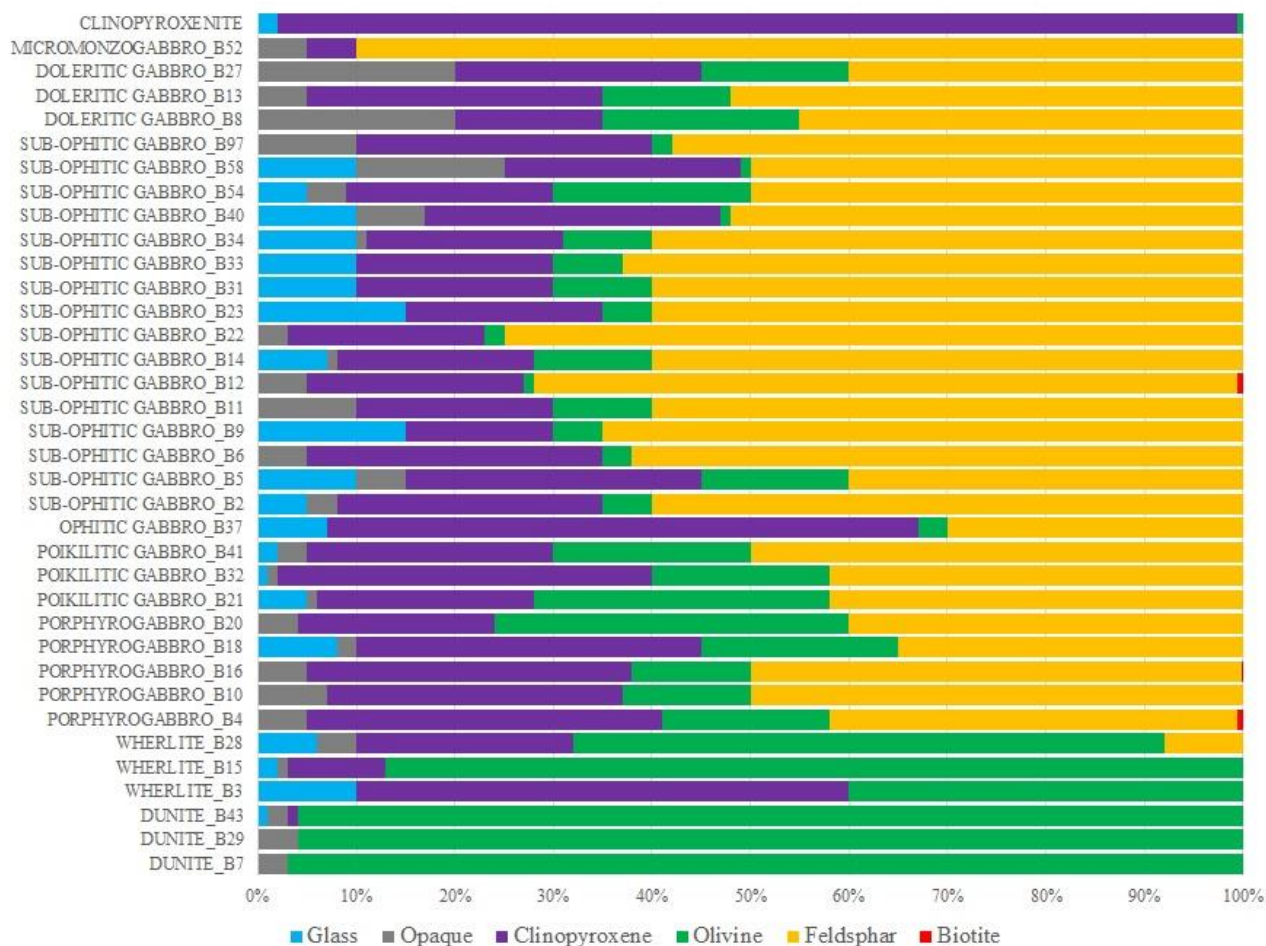


Fig. 7.1.4 - Modal mineralogy of the subvolcanic ejecta of the Bellecombe Ash Member. Biotite is present with low vol% in only three samples, two porphyrogabbros and one sub-ophitic gabbro.

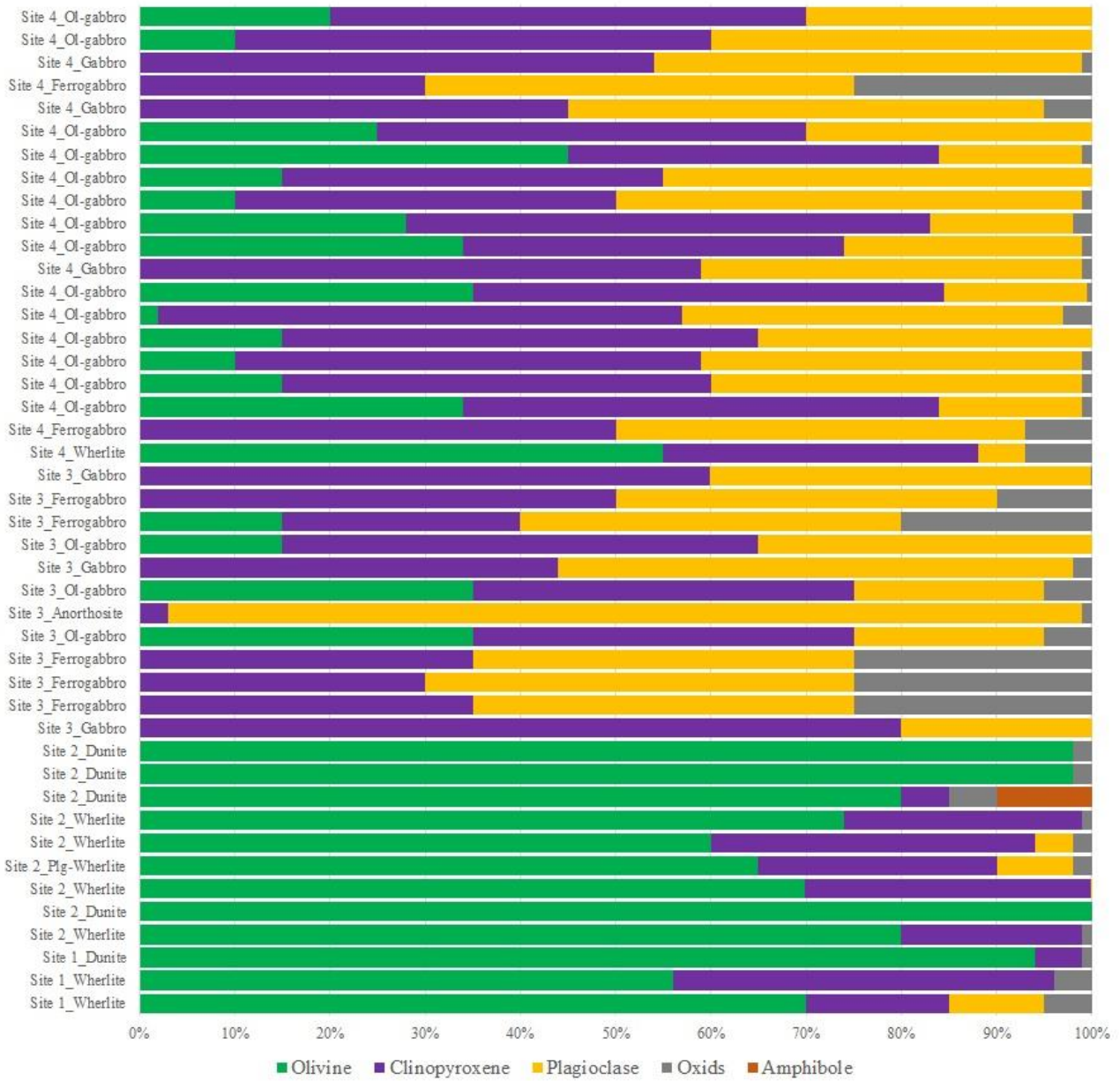


Fig. 7.1.5 - Modal mineralogy of Berthod et al. (2020) samples. Dunites and wherlites containing less than 10 vol% of plagioclase. Gabbroic rocks, including olivine gabbros, gabbros and ferrogabbros containing more than 10 vol% of plagioclase.

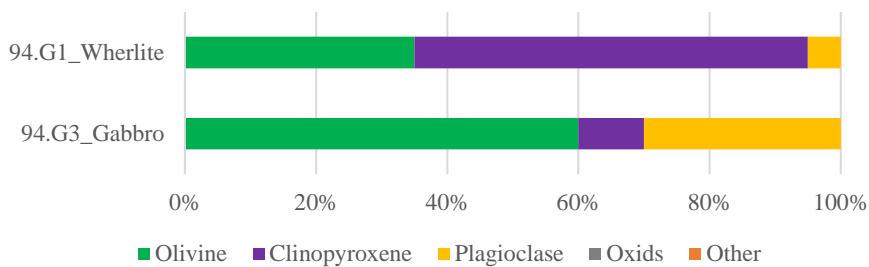


Fig. 7.1.6 - Modal mineralogy of Upton et al. (2000) samples.



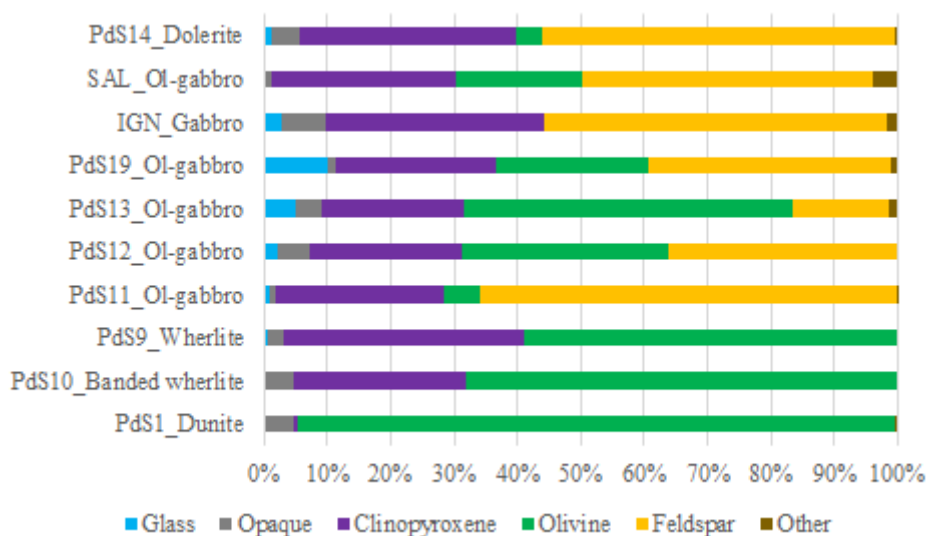


Fig. 7.1.7 - Modal mineralogy of Brugier (thesis, 2016) samples.

## 7.2 The ultramafic lithotypes

The investigated samples representative of the ultramafic group, includes dunites and wherlites (Table 7.2.1).

Table 7.2.1 - The analysed representative samples of ultramafic ejecta.

ROCK TYPE	SAMPLE NUMBER
<b>DUNITES</b>	B7
	B29
	B43
<b>WHERLITES</b>	B3
	B15
	B28

## 7.3 Dunites

Dunites show a cumulitic texture including olivine (> 95 vol%), oxides (2-4 vol%) and minor clinopyroxenes ( $\leq$  1 vol%). Olivine crystals are euhedral to subhedral in shape and 2-8 mm in diameter and form a cumulus framework. Oxide crystals are interstitial in the olivine fabric and include magnetite, ilmenite and Cr-spinel compositions. Samples of dunite differ in grain size (e.g. B7 coarser grained, B29 and B43 finer grained; Fig. 7.3.1) and in oxides content. Furthermore, all dunite samples show the presence of rounded sulphide trails crossing the olivine crystals. Melt and fluid inclusions are evident in some olivine.

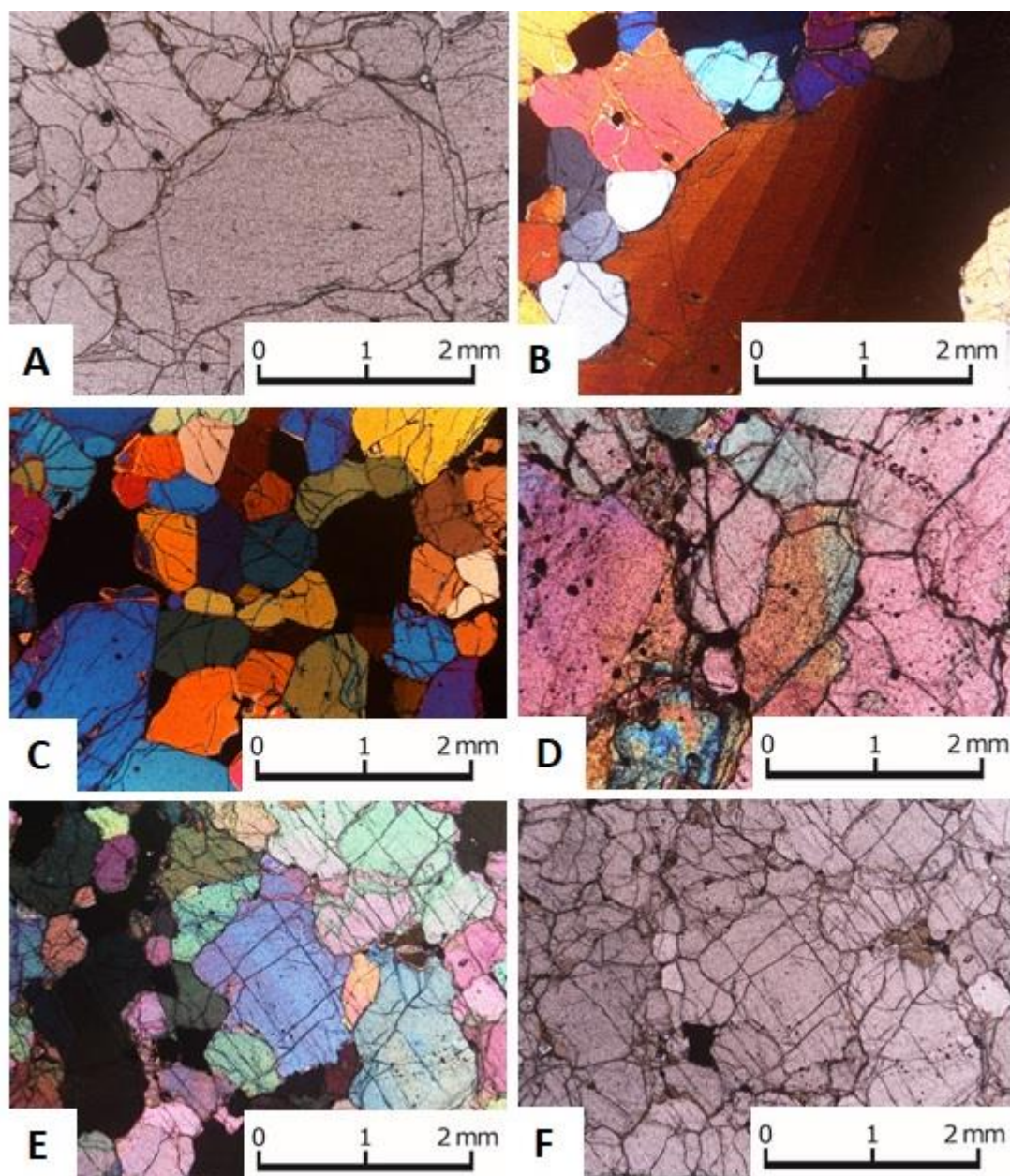


Fig. 7.3.1 - Plane-polarized (A and F) and crossed polars images (B, C, D and E). Dunites showing the presence of oxides (A and F, sample B29) and some sulphides trails (D, sample B7). Dunites samples have both finer (C, sample B43, E, sample 29 F) and coarser grain size (A, B, sample B43, D). Kink banding of the crystals (B) is also present.

#### 7.4 Wherlites

The wherlite samples in the Bellecombe Ash Member shows different modal mineralogy, with olivine ranging from 40 to 87 vol% and clinopyroxene from 10 to 50 vol%. Feldspars are only present in one sample (B28; Fig. 7.4.1) with a small percentage (8 vol%). Interstitial glass is shown in all the samples (2-10 vol%). Oxides are also present but in small amount (0-4 vol%). Melt inclusions are shown in some olivine crystal. Samples of wherlite differ in grain size (e.g. B3 and B15 coarser grained, B28 finer grained). Despite

other wherlites samples, B28 show a poikilitic texture (with clinopyroxene as oikocrysts and olivine as cadacrysts), with small olivines (< 0.5 mm) with rounded margin and larger olivines (2-3 mm) with fritted margin and rounded contours. Poikilitic texture is maybe related to resorbed processes on samples due to a late-stage interactions with successive batches of basaltic magmas. Sample B3 evidence melt inclusions in some olivine crystals, respect other wherlites from Bellecombe.

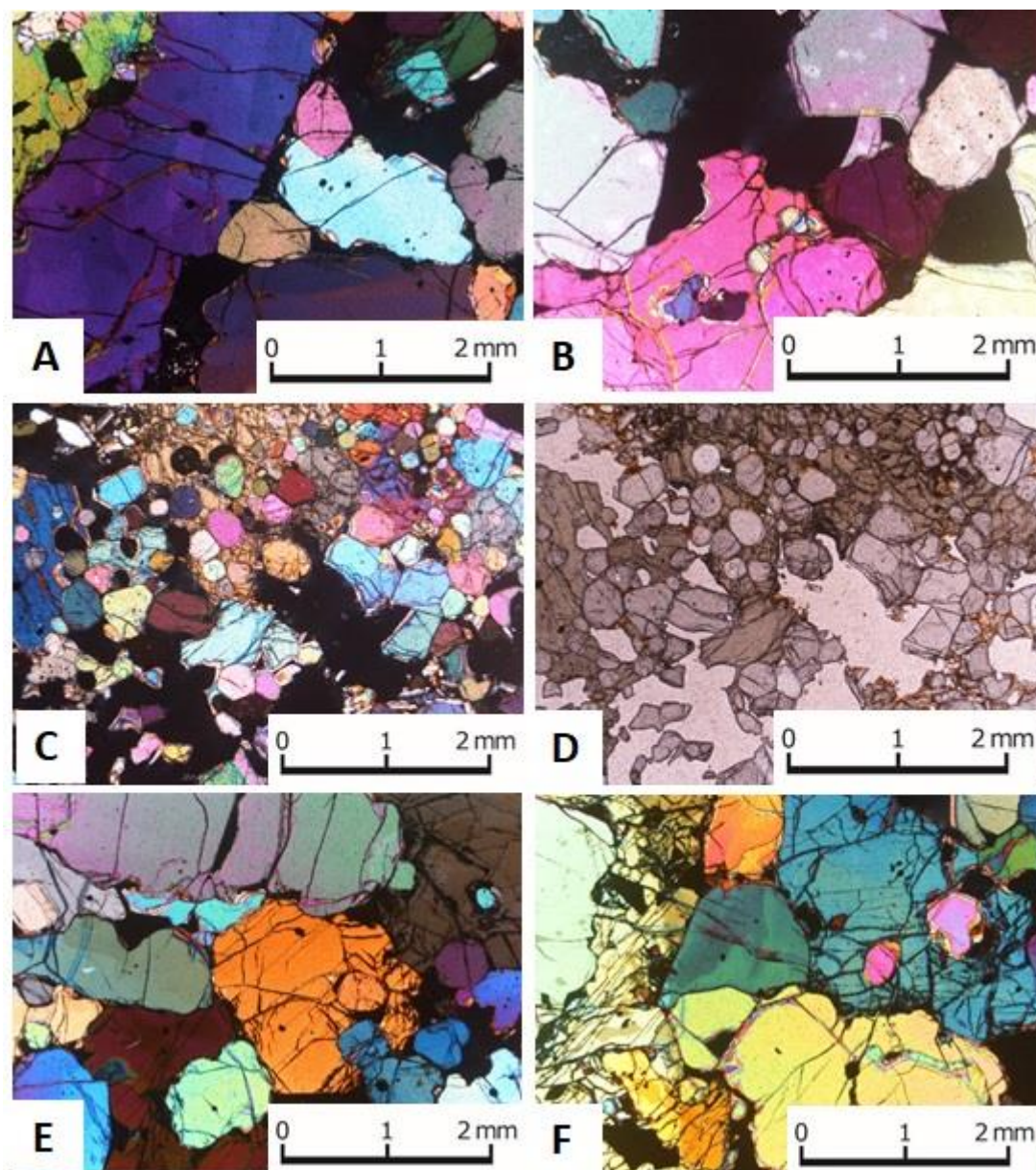


Fig. 7.4.1 - Plane-polarized (D) and crossed polars images (A, B, C, E and F). Wherlite samples of Bellecombe Ash Member show different grain size. Wherlite with finer grain size (C and D, sample B28) show a major olivine content respect to other samples. Some oxides are evident in the framework of the olivine and clinopyroxene (A, B and E, sample B15). Small melt inclusions in single crystals are evident (F, sample B15).

## 7.5 The mafic lithotypes

The mafic lithotypes include ophitic gabbros, sub-ophitic gabbros, poikilitic gabbros, doleritic gabbros, micro-monzogabbros and porphyrogabbros (Table 7.5.1).

Table 7.5.1 - Subdivision of the representative mafic samples analysed.

ROCK TYPE	SAMPLE NUMBER
OPHITIC GABBROS	B37
SUB-OPHITIC GABBROS	B2
	B5
	B6
	B9
	B11
	B12
	B14
	B22
	B23
	B31
	B33
	B34
	B40
	B54
	B58
B97	
POIKILITIC GABBROS	B21
	B32
	B41
DOLERITIC GABBROS	B8
	B13
	B27
MICRO-MONZOGABBRO	B52
PORPHYROGABBROS	B4
	B10
	B16
	B18
	B20

## 7.6 Ophitic gabbros

This group includes only one sample, with an evident ophitic texture, where random feldspar laths (30 vol%) are enclosed by clinopyroxene (60 vol%) or olivine (3 vol%). A small amount of interstitial glass is present (Fig. 7.6.1).

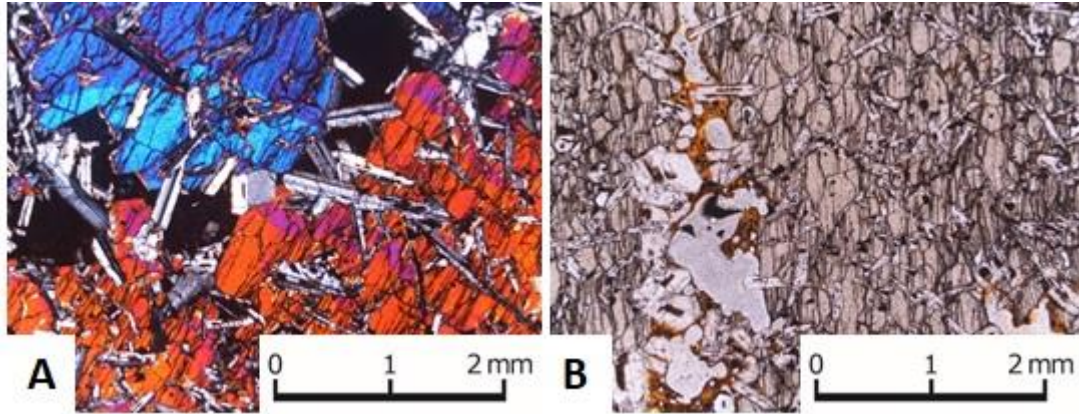


Fig. 7.6.1 - Crossed polars of ophitic texture (A) and plane-polarized of the sample with a small amount of interstitial glass (B).

## 7.7 Sub-ophitic gabbros

The sub-ophitic gabbros is the most represented group in terms of number of samples of mafic igneous subvolcanic clasts in the Bellecombe Ash Member. Clasts have an evident sub-ophitic texture, with a high content of feldspar (from 40 to 75 vol%) often zoned and some sieved plagioclase, a relatively low content in olivine (2-15 vol%) and clinopyroxene (some hourglass zoned) in the range 15-30 vol%. Oxides are not present in all the samples analysed in thin section (0-15 vol%). Interstitial glass is also present (0-15 vol%). The crystals surrounded by the interstitial glass are euhedral to subhedral in shape (Figg. 7.7.1 and 7.7.2).

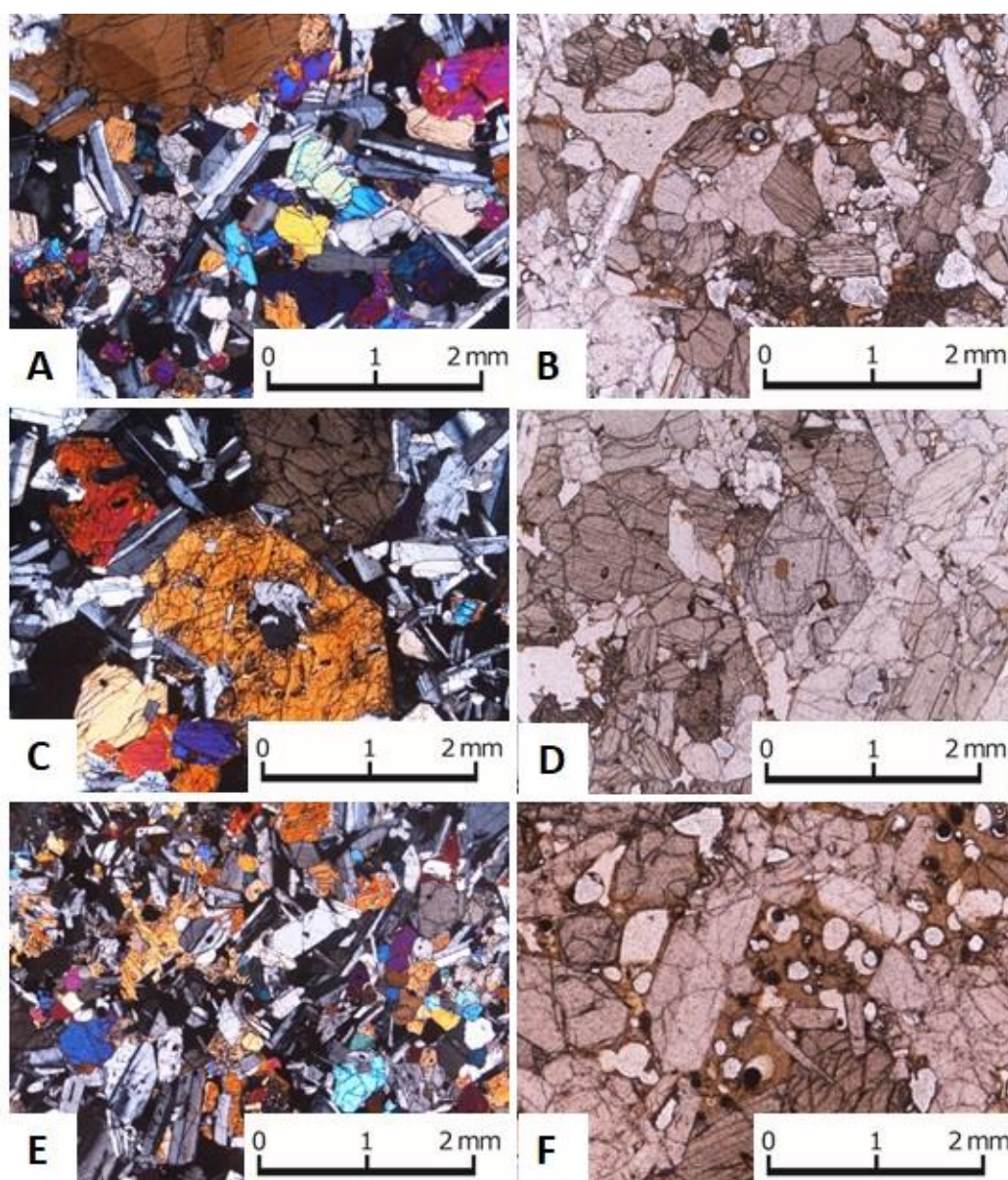


Fig. 7.7.1 - Crossed polars images (A, C and D) and plane-polarized (B, D and F). The sub-ophitic texture is evident in the Bellecombe Ash Member gabbros (A and C, sample B9, E, sample B54). The interstitial glass in these samples is enclosed (entrapped) between euhedral to subhedral crystals (B, sample B14, D, sample B33, F, and sample B9). Melt inclusions occurred in some olivine and clinopyroxene crystals (D).

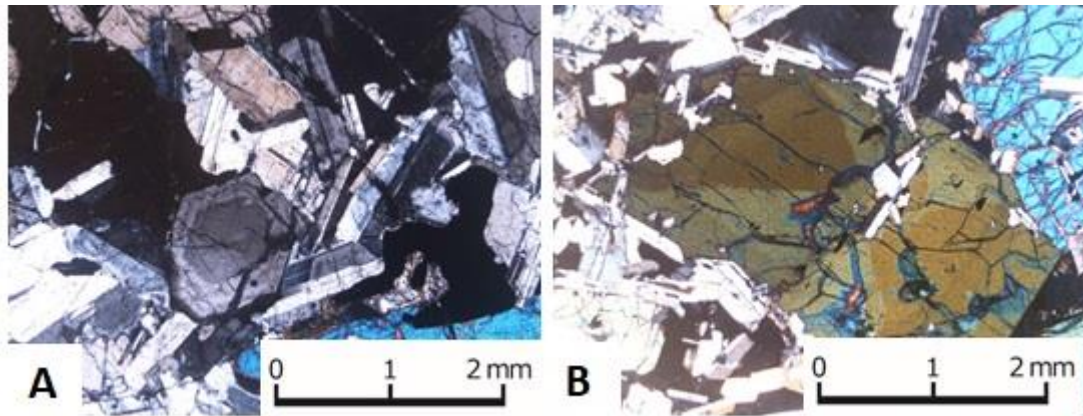
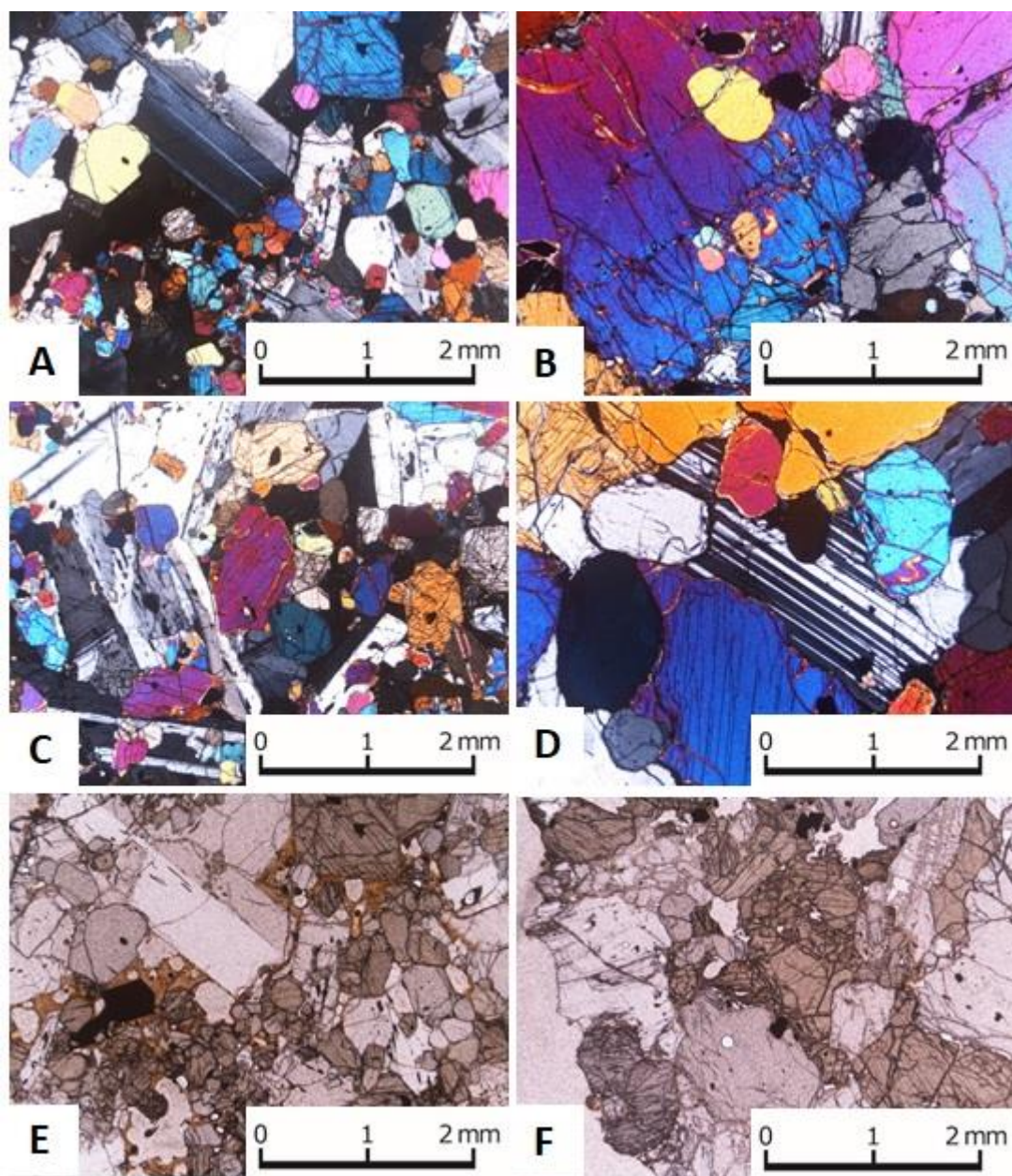


Fig. 7.7.2 - Plane-polarized light A) of sub-ophitic texture showing zoned plagioclase (sample B12), and crossed polars B) of a clinopyroxene with hourglass zoning (sample B23).

### 7.8 Poikilitic gabbros

The samples of this group show a poikilitic texture (olivines as cadacrystals and feldspars and clinopyroxenes as oikocrystals), with feldspar content ranging from 42 to 50 vol%, sometimes zoned and showing a sieved texture. Olivine is present (18-30 vol%) both as small (< 0,5 mm) and slightly larger crystals (> 1 mm). Clinopyroxene (22-38 vol%) show sizes > 3 mm. Interstitial glass is not abundant (1-5 vol%), as well as opaque minerals (1-3 vol%), including titanomagnetite (Fig. 7.8.1).



*Fig. 7.8.1 - Crossed polars (A, B, C and D) and plane-polarized light of poikilitic gabbros. Different olivine size in the samples (A and C, sample B21, B, sample B41); plagioclase are often zoned (D, sample B32, E, sample B21). Crystal rim are euhedral to subhedral in shape, without resorbed margins (E). Some opaque mineral occurs as titanomagnetite (F, sample B41).*

### **7.9 Doleritic gabbros**

Doleritic gabbros are fine-grained (< 2 mm). Feldspar are present in high percentage (40-52 vol%). Also clinopyroxene (15-30 vol%) abundantly occurs. Olivine is present both as subhedral microcrystals and altered crystals (as celadonite, clay minerals or/and serpentine) in all the samples. Opaque minerals are present (5-20 vol%) often as crescumulate crystals (Fig. 7.9.1).



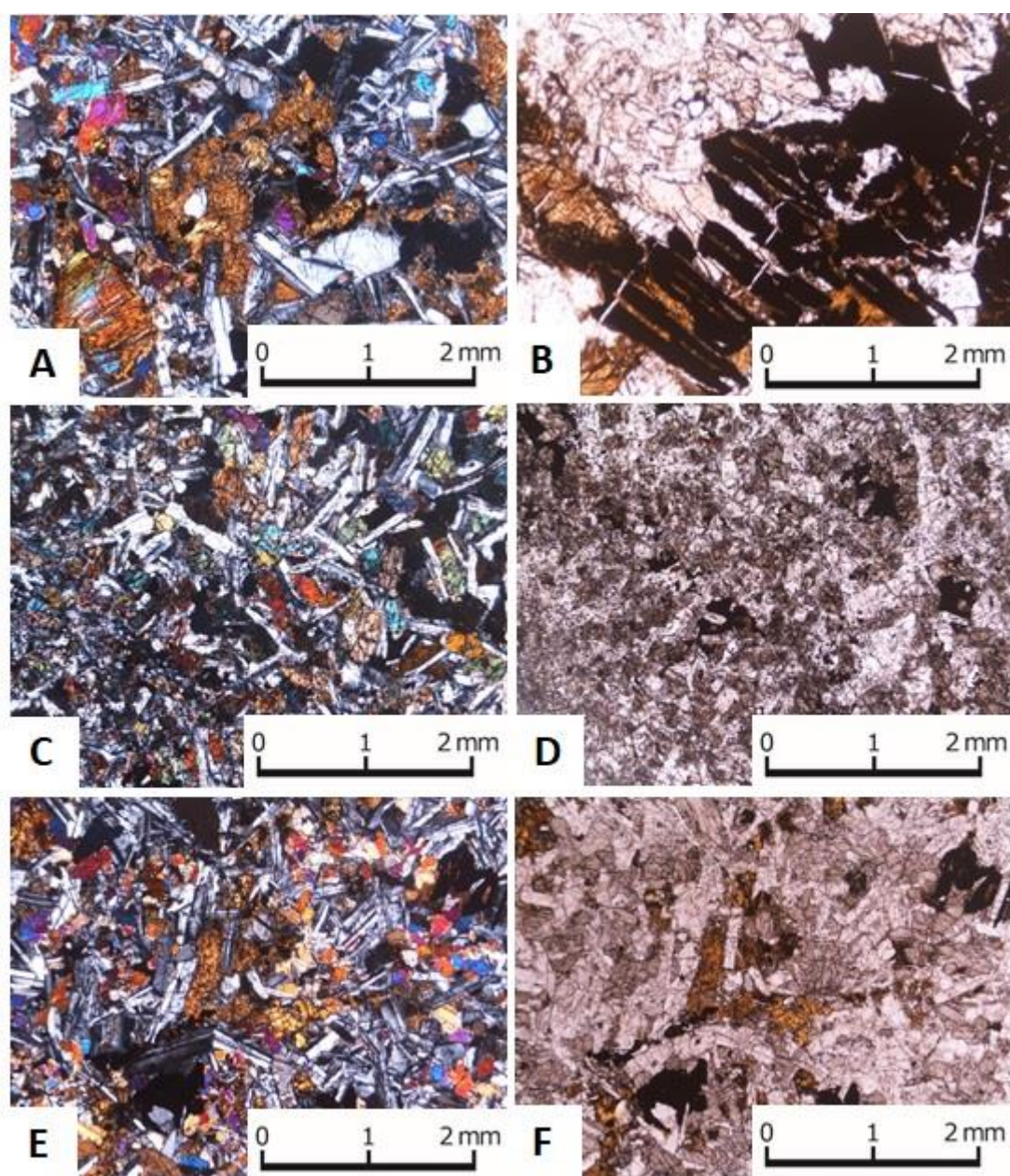


Fig. 7.9.1 - Crossed polars (A, C and D) and plane-polarized light (B, D and F) of doleritic gabbros. Altered olivine is present in all samples. As well as crescumulate opaque crystals (A and B, sample B8; C and D, sample B13, E and F, sample B27).

### 7.10 Micro-monzogabbros

A sample in the Bellecombe Ash Member consists of a micro-monzogabbro (Fig. 7.10.1) included in an aphyric lava. The subvolcanic crystalline part mostly consist of feldspar (90 vol%), and minor part of clinopyroxene (5 vol%) and opaque minerals (5 vol%).

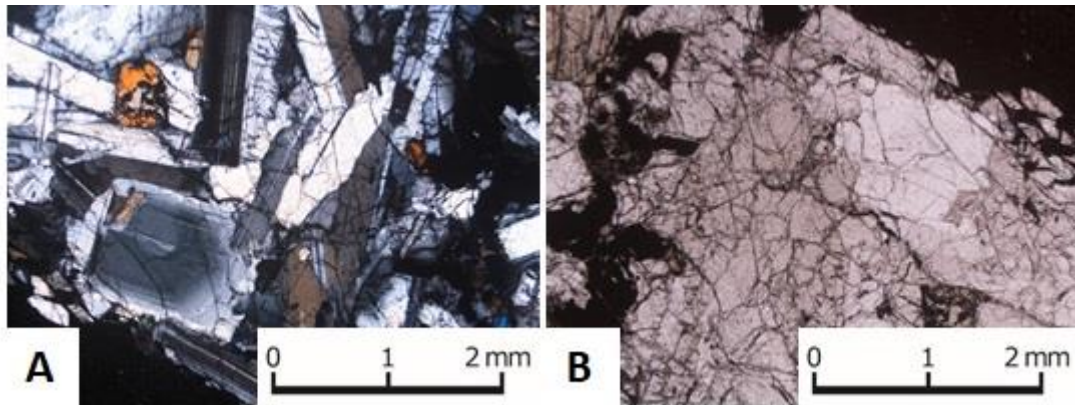


Fig. 7.10.1 - A) Crossed polars of zoned plagioclase on the micromonzogabbro; B) plane-polarized light of the contact between intrusive and aphyric portions of the sample (sample B52).

### 7.11 Porphyrogabbros

Samples of this group display a porphyritic-like texture. Feldspar and mafic minerals in the pseudo-groundmass are very small in size ( $\ll 0.1$  mm; 35-50 vol%), surrounding larger crystals ( $> 0.5$  mm) of clinopyroxene (20-36 vol%) and olivine (12-36 vol%). Opaque minerals are also present (2-7 vol%) both as cromite and titano-magnetite. Only one sample show interstitial glass, altered and in small percentage (8 vol%). These samples also show rare biotite (phlogopite;  $< 1$  vol%), never found in other samples of the Bellecombe Ash Member (Fig. 7.11.1).

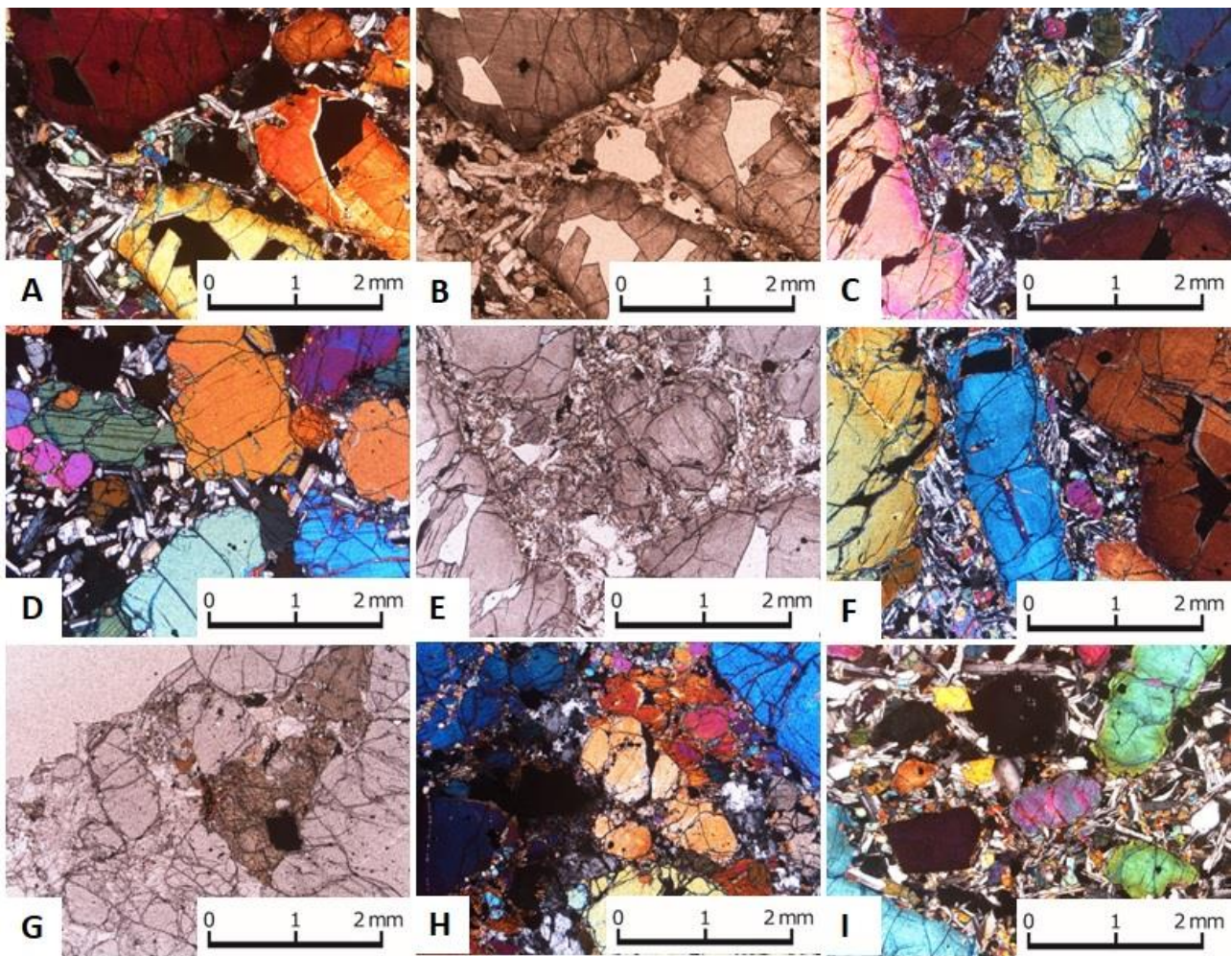


Fig. 7.11.1 - Crossed polars (A, B, C, D, F, H and I) and plane-polarized light (E and G) of porphyrogabbros. Porphyrogabbro samples showing a difference in crystal size (A, B and C, sample B10, E and F, sample B16, D, sample B18, I, sample B10, H, sample B20). Interstitial biotite is present in all the samples in small percentage (G, sample B20).

### 7.12 Samples with an evident magmatic layering: B108a and B108b

These samples show at the macroscopic scale an evident magmatic layering, furtherly investigated at the microscopic scale. In thin section, the samples show a gradual transition from fine grained to coarse grained subvolcanic igneous rock and a gradual transition from ultramafic to mafic lithotypes. It is not evident a crystal orientation. Resorbed margin at the mafic-ultramafic contacts are absent. Rounded crystals of olivine, some of them with fritted margin are evident, both in mafic and ultra-mafic portions. The transition between mafic lithotypes is gradual, whereas wherlites and plg-wherlites are surrounded by mafic lithotypes (fine-grained gabbro and gabbro). In this way, the samples emphasize a transition from wherlite to plg-wherlite to gabbro. Sample B108a also shows a small troctolite portion within the wherlitic domain of the sample (Figg. 7.12.1, 7.12.2, 7.12.3, 7.12.4 and 7.12.5) and rare phlogopite micro crystals (< 0.1 mm).

Fig.7.12.1 - Large (3.5 x 5 cm) thin section of the sample B108a. Ultramafic and mafic portions are evident in the whole sample, with different modal mineralogy and texture. Troctolite portion coexist with wherlite and plg-wherlite.

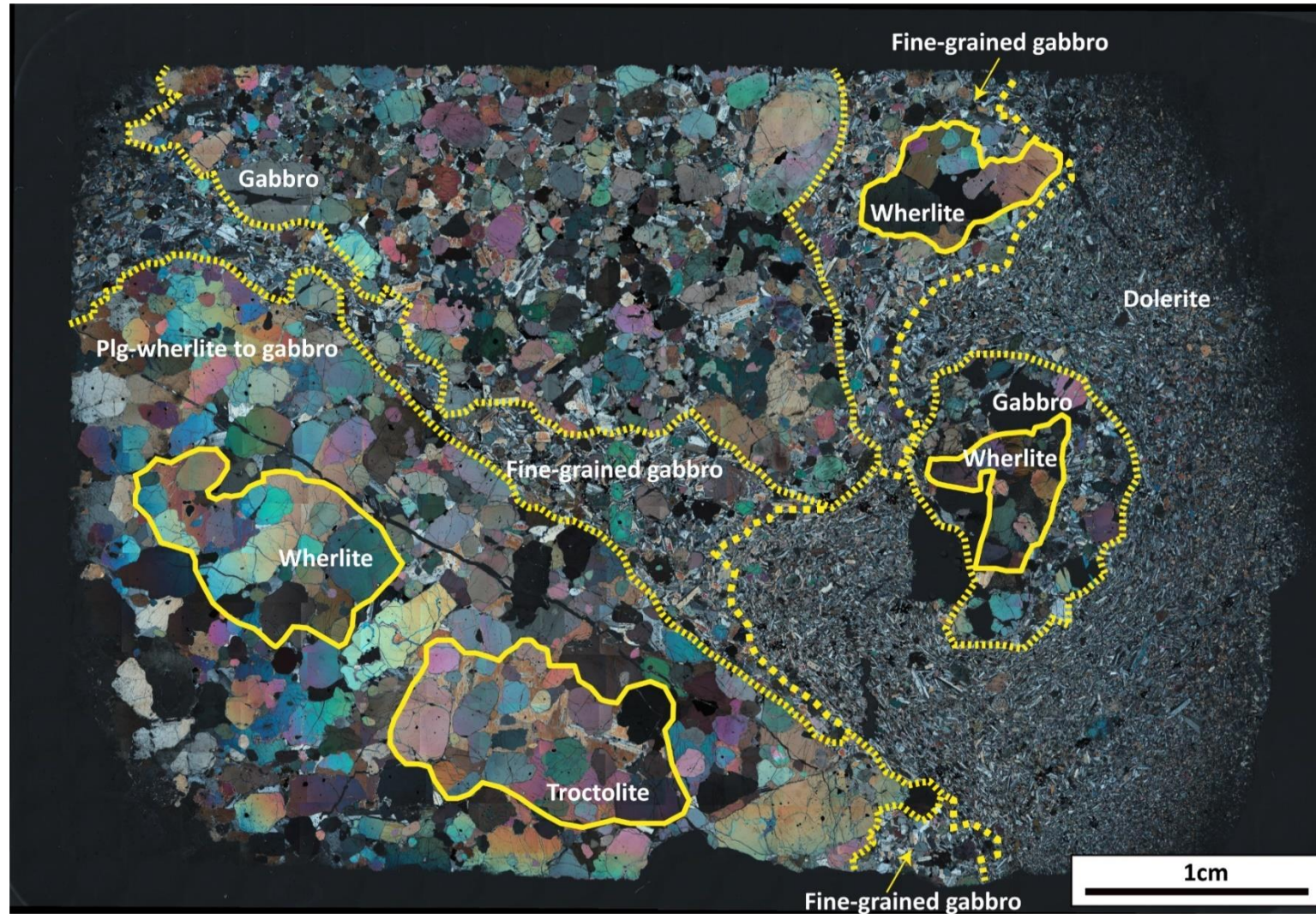


Fig.7.12.2 - Large (3.5 x 5 cm) thin section of the sample B108b. Ultramafic and mafic portions are evident in the whole sample, with different modal mineralogy and texture.

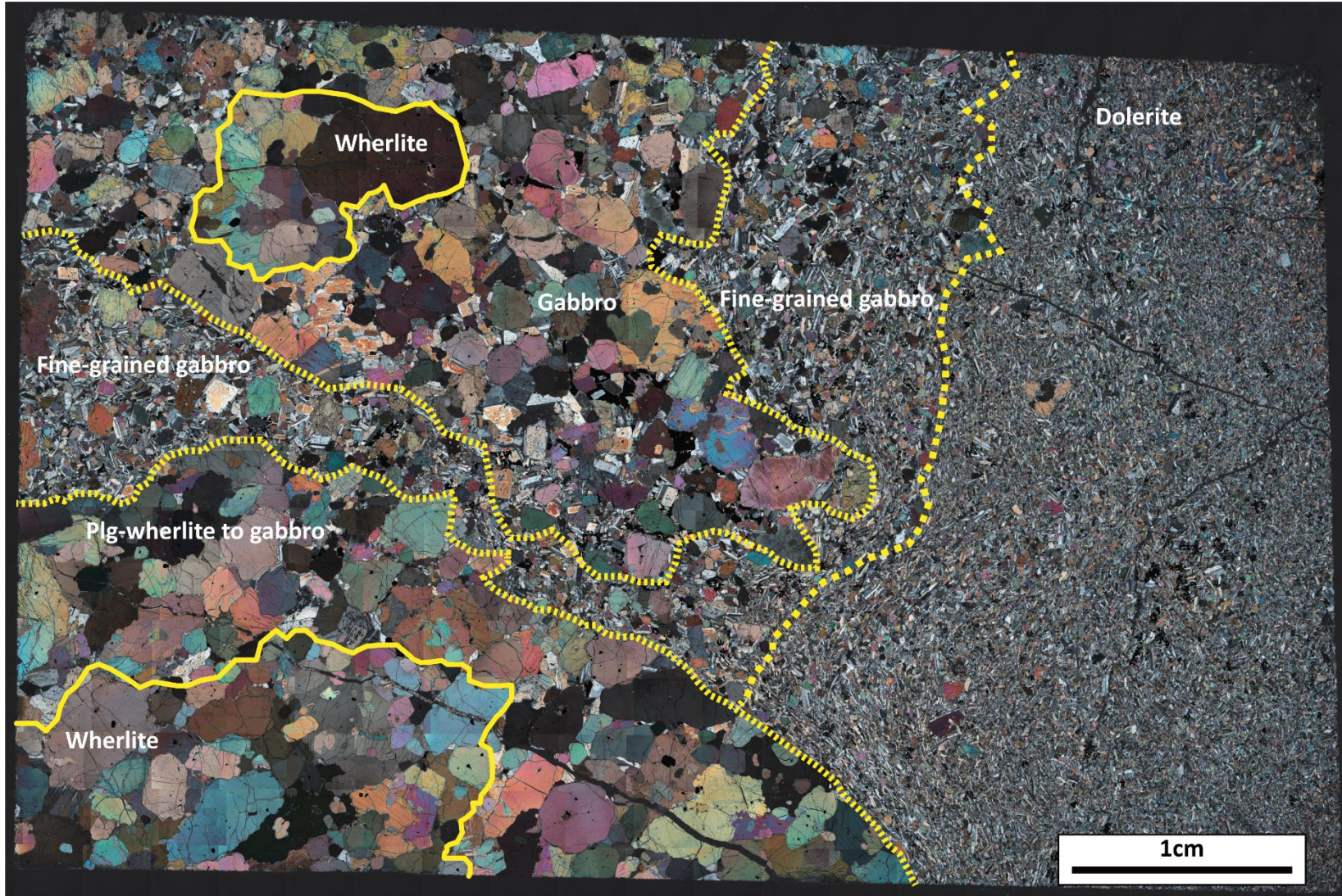


Fig. 7.12.3 - Detail of sample B108a.

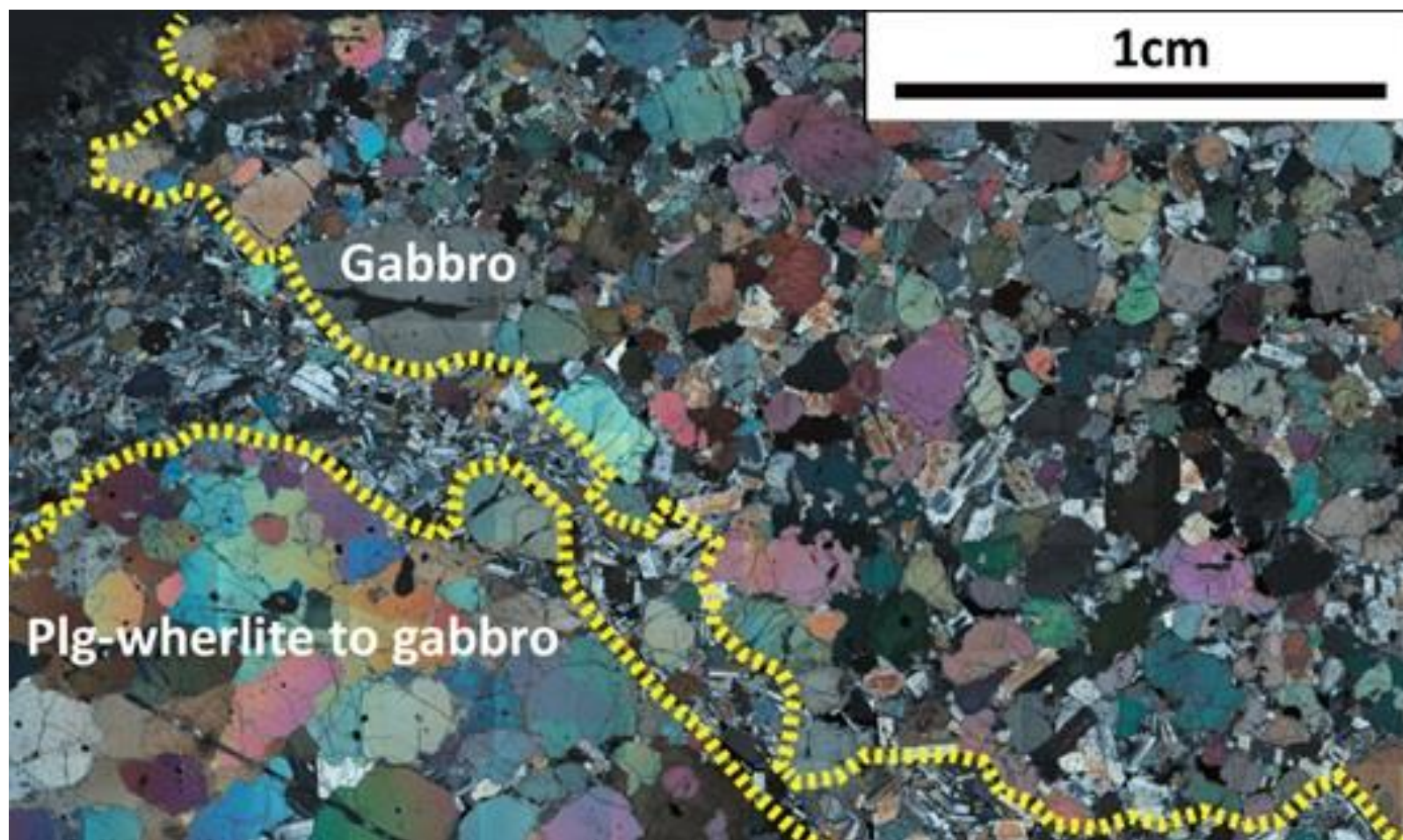
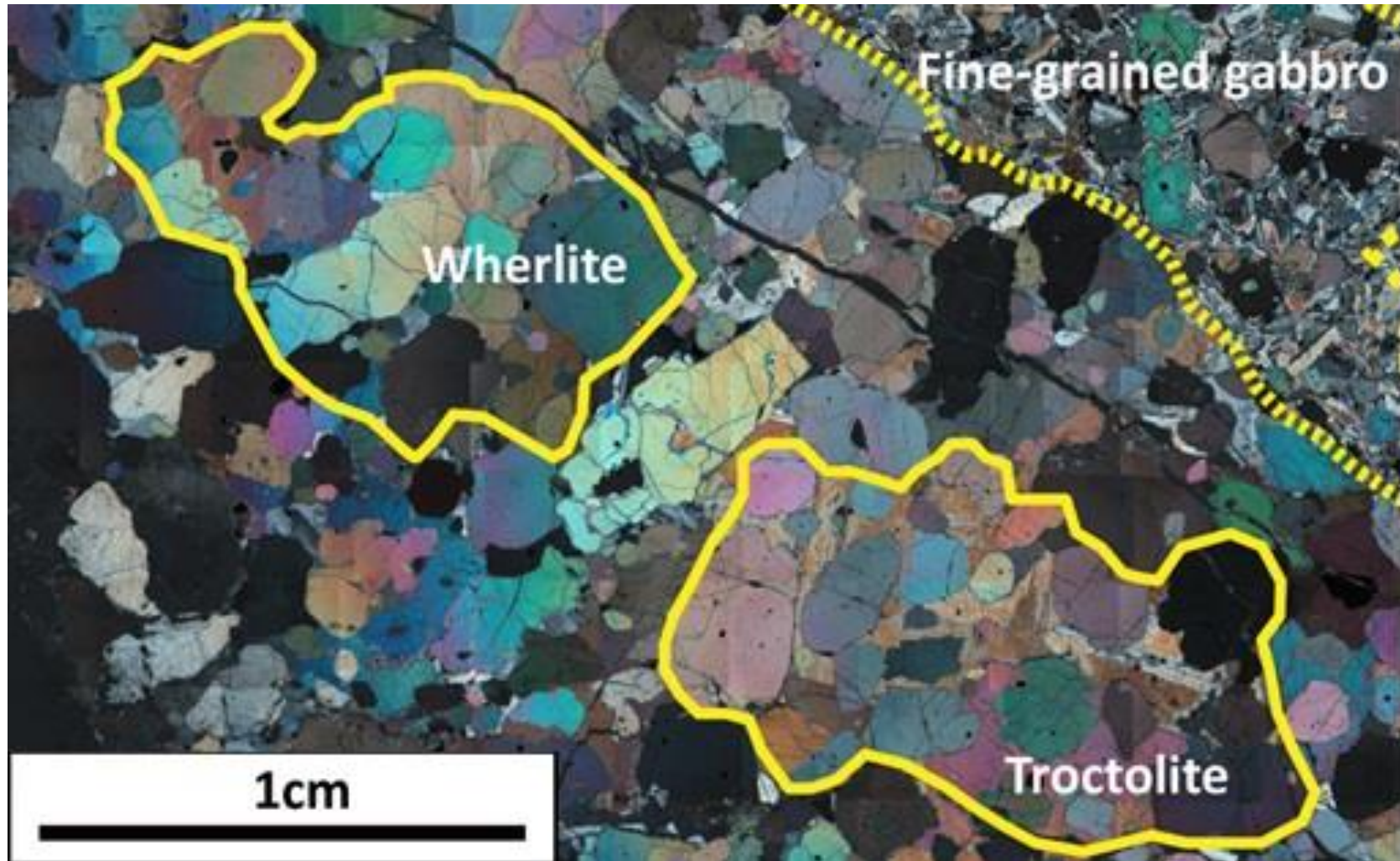


Fig. 7.12.4 - Detail of sample B108a.



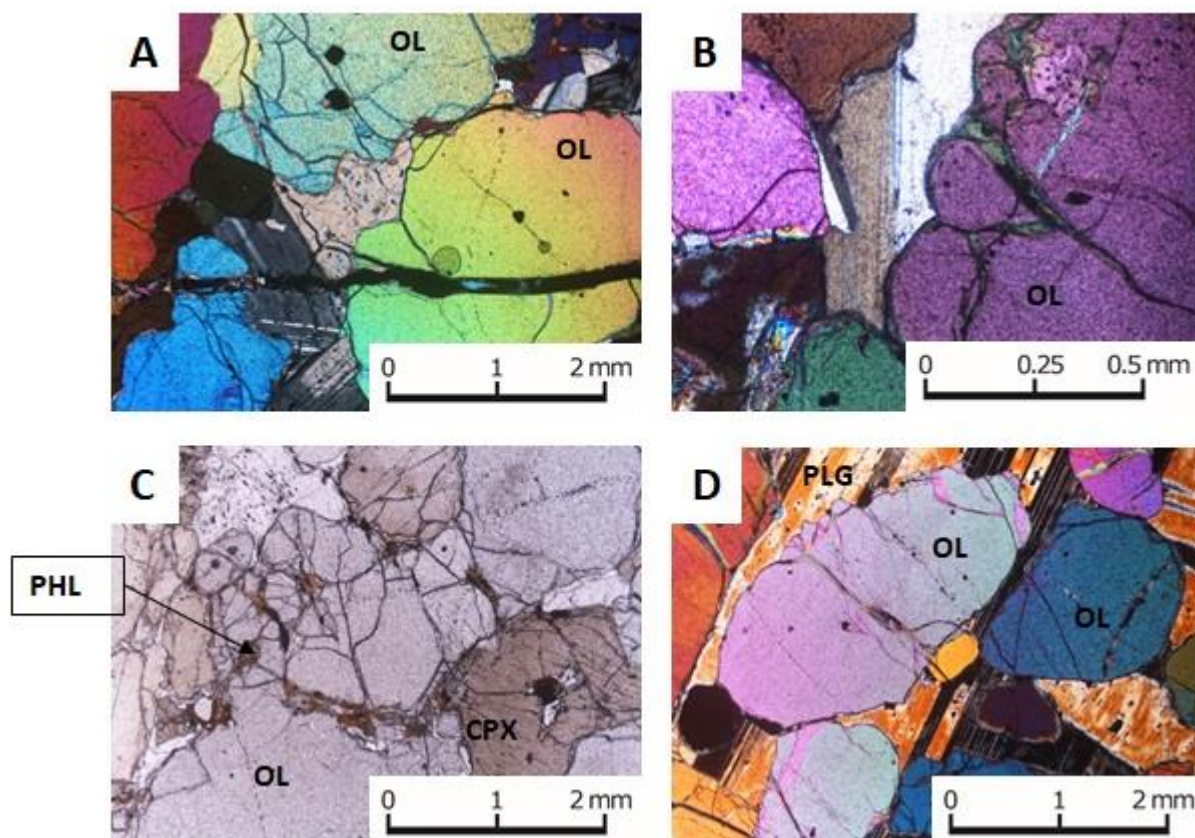


Fig. 7.12.5 - Crossed polars and plane-polarized light images of samples 108a and b. A) Fritted margin olivine on gabbro portions and B) a detailed of fritted margin olivine; C) plagioclase wherlite with phlogopite micro crystals; D) troctolite portion on sample B108a. [OL = olivine, CPX = clinopyroxene, PLG = plagioclase, PHL = phlogopite].

### 7.13 SEM-BSD-EDS analyses on sample B108a

The analyses with SEM-BSD-EDS on sample B108a permitted to investigate in detail the overall composite texture and the composition of single mineral phases (Table 7.13.1). Rounded olivines, often with fritted margins are widespread in the whole sample, in both ultra-mafic and mafic portions. Troctolite portion show olivine with a Fo content of about 79% and plagioclase with An<sub>82</sub> Ab<sub>17</sub> Or<sub>1</sub>. Pyroxene near troctolite show a Fs<sub>9</sub> content (Fig. 7.13.1). Near the troctolite portion, olivine crystals with fritted margins are always characterized by Fo<sub>79</sub> (Fig. 7.13.2).

Plagioclase in gabbro and fine-grained gabbro portions are often zoned, with a higher Ca-content on the core of the crystal. Mafic portions emphasize a high content in opaque minerals which are evident as both inclusions and interstitial crystals (Fig. 7.13.3). Opaque crystals in B108a sample are both Cr-Fe spinel and Ilmenite. In mafic portion, plagioclase appears to be crystallized after rounded olivine and clinopyroxene crystals. Opaque minerals crystallized from late interstitial liquid (Fig. 7.13.4). Rounded olivine crystals in contact with plagioclase show a Fo<sub>67</sub> composition, less forsteritic than other gabbro clasts in the Bellecombe Ash



Member. The presence of ilmenite suggests a relatively low oxygen fugacity environment of crystallization.

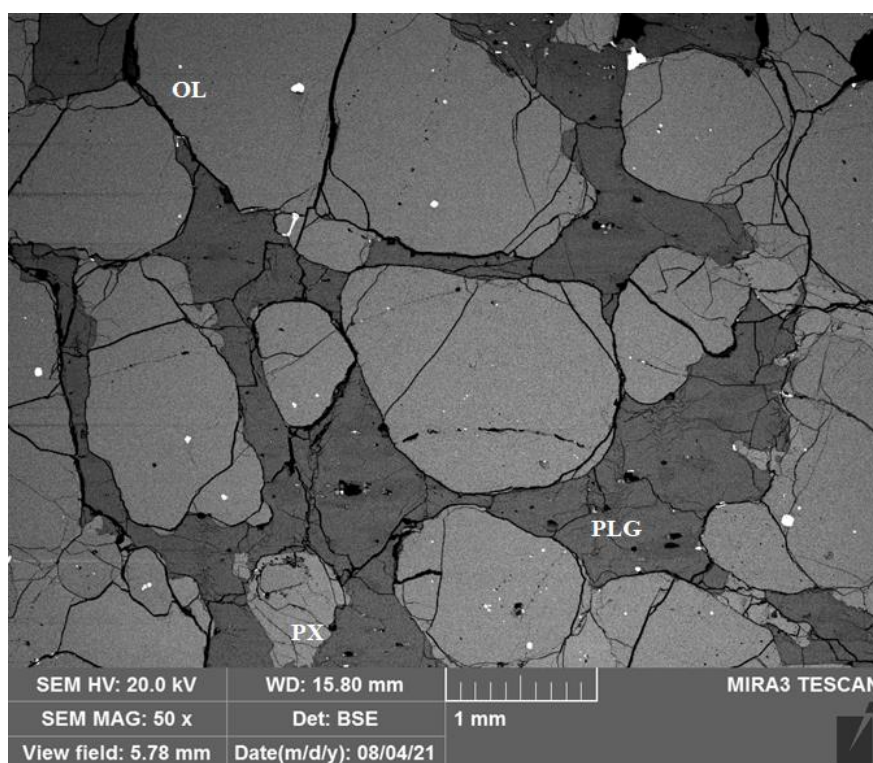


Fig. 7.13.1 - Troctolite portion in the composite sample B108a. Rounded Fo<sub>79</sub> olivine are enclosed by plagioclase (An<sub>82</sub>).

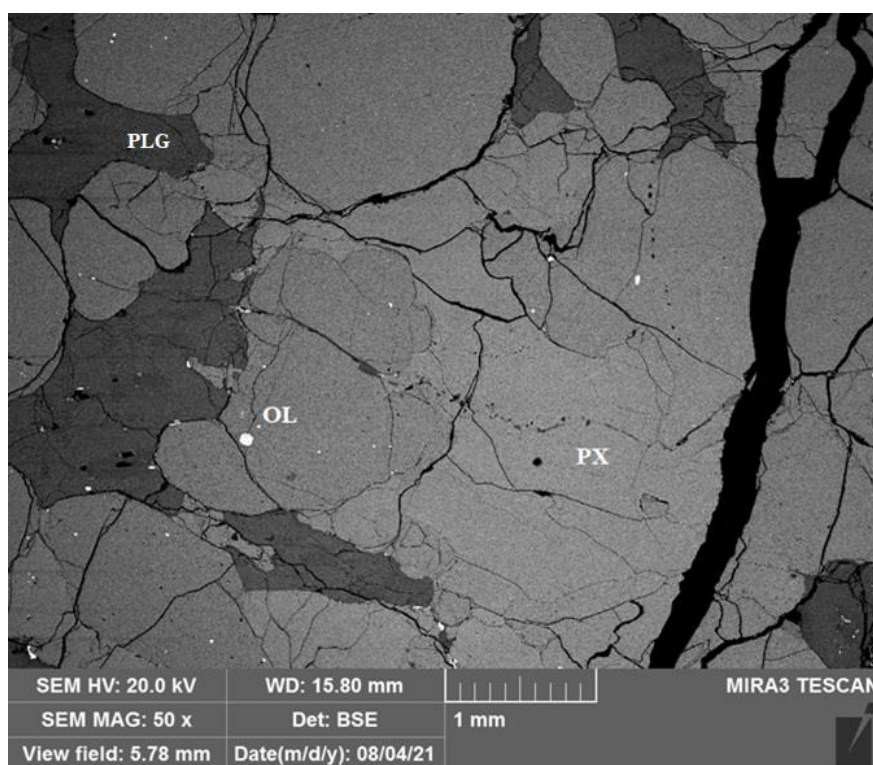


Fig. 7.13.2 - Ultramafic portion (plagioclase wherlite) near troctolite showing crystals of fritted margin olivine and clinopyroxene with rounded margin.

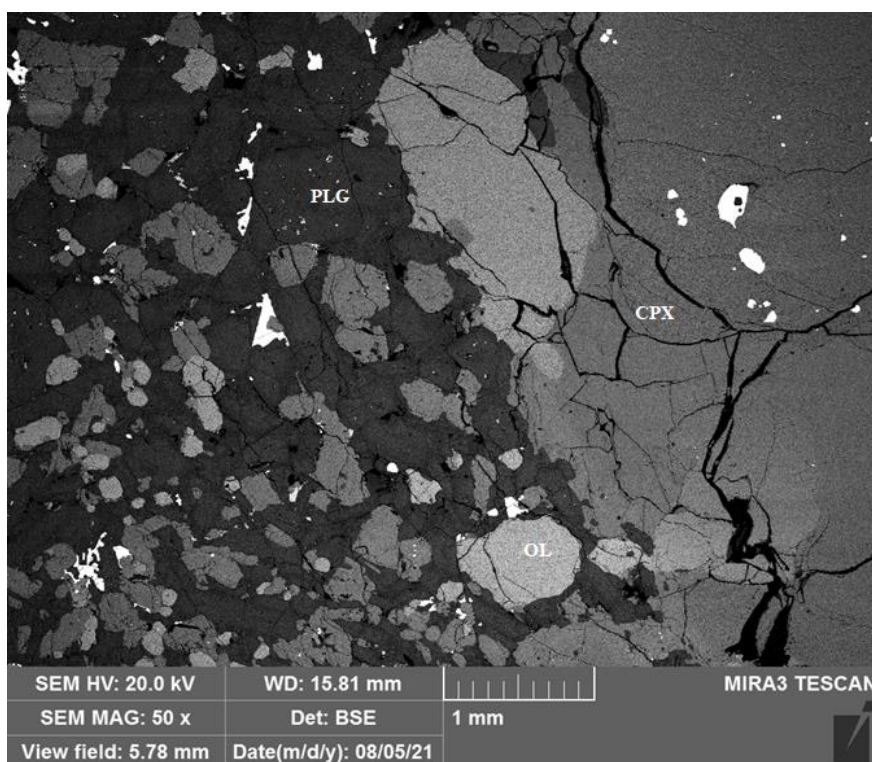


Fig. 7.13.3 - Contact between ultramafic portion and fine-grained gabbro on sample B108a. The massive presence of opaque minerals is evident both on single crystals and interstitially.

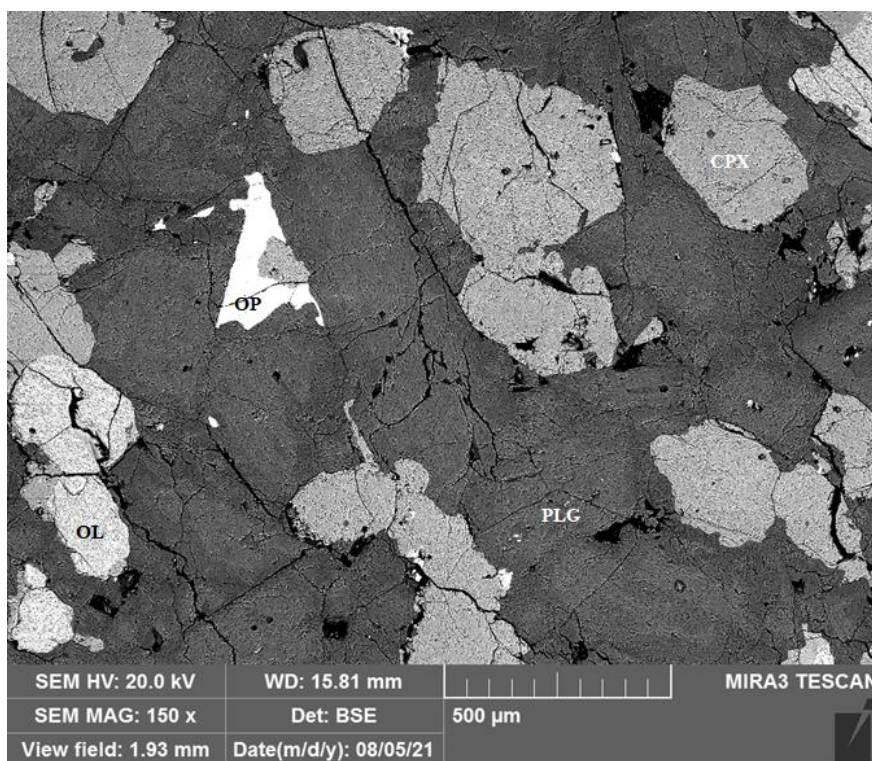
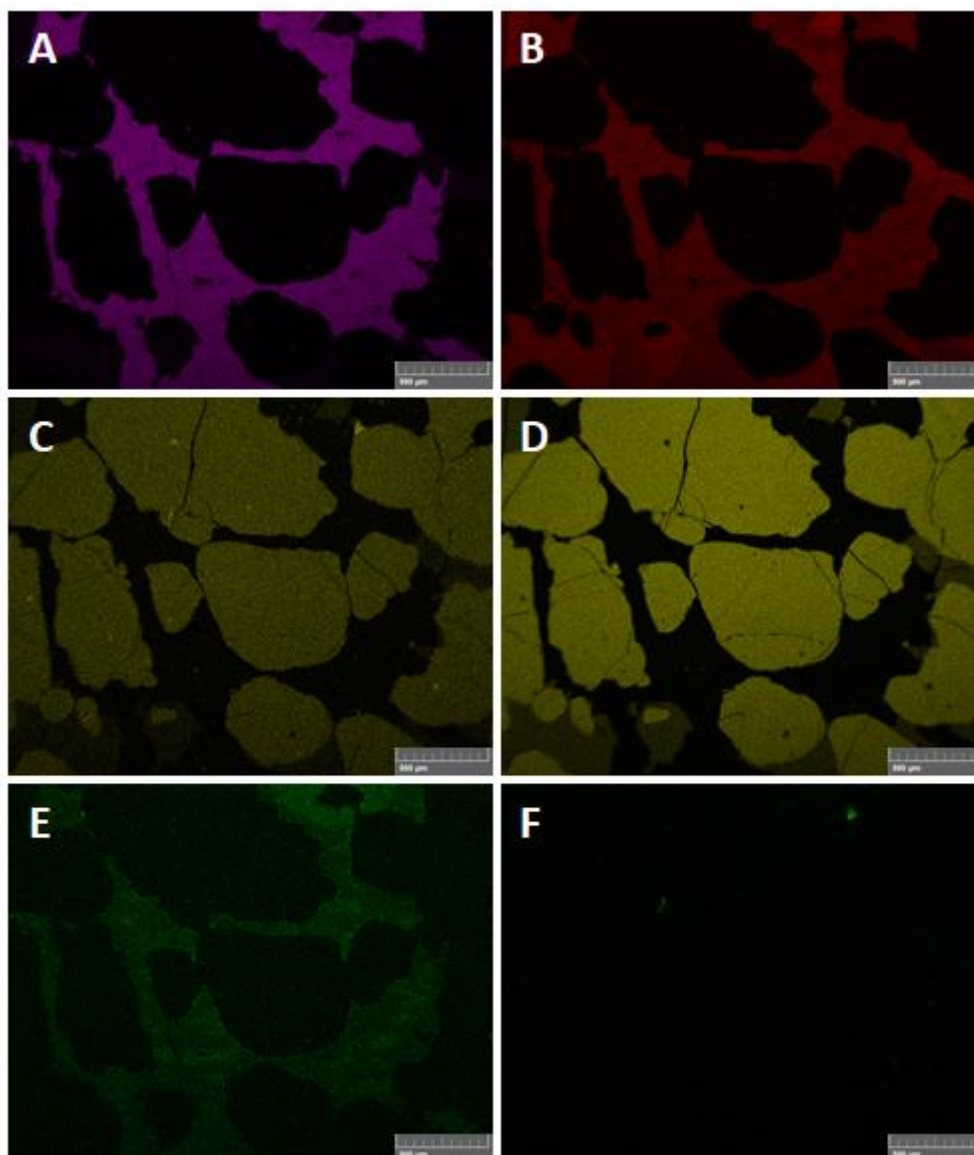


Fig. 7.13.4 - Detail of a portion of fine-grained gabbro. Olivine and clinopyroxene crystals show rounded margin and plagioclase are zoned. Ilmenite crystal (OP) is evident between plagioclase crystals.

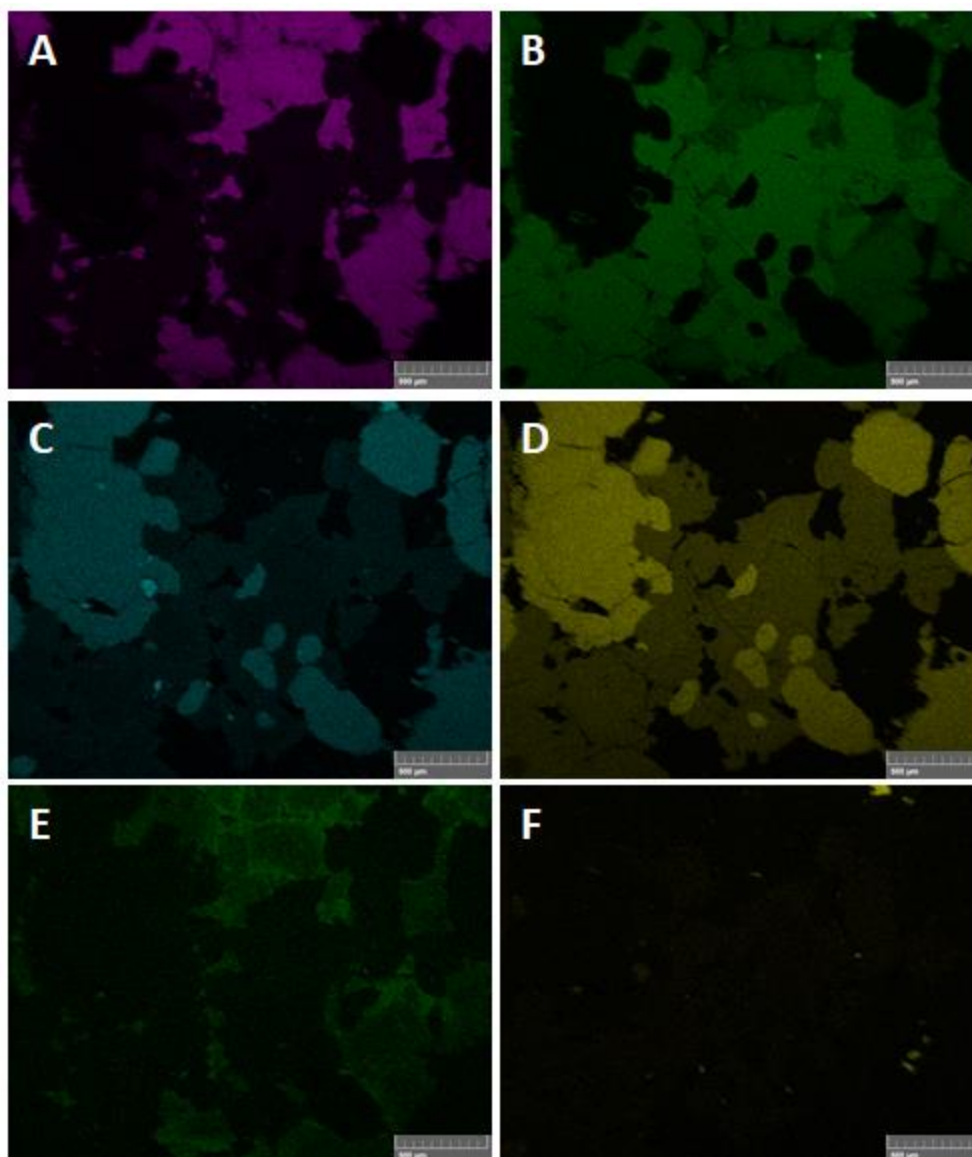
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The EDS map furtherly permit to investigate the composite B108a sample. Troctolite portion emphasizes the presence of fritted margin olivines surrounded by patchy zoned plagioclase (different Ca and Na contents). Ti is poorly present in troctolite portion (Fig. 7.13.5). On the mafic portion, gabbro displays small rounded olivine, surrounded by large anhedral clinopyroxene crystals, and fritted margin olivine. Clinopyroxene crystals are also zoned, with higher concentration of Ca on the core and higher Fe-content on the rim. In addition, also plagioclase is normally zoned, with a more albitic rim. Opaque minerals (both ilmenite and spinel) are present, mainly near the rim of other large crystals (olivine and clinopyroxene; Figg. 7.13.6, 7.13.7 and 7.13.8).

Also on the ultra-mafic portion (wherlite and plagioclase wherlite), single mineral phases show rounded margins, with subhedral crystals, also on plagioclase. Fe-Ti oxides are also present as crescumulates, interstitially and surrounding other crystals. Zoned plagioclase crystals present in sample B108a show a high Ca concentration on the core and high Na concentration on the rim of the crystals. Opaque minerals in the interstice, with regular margins despite the contact with other single mineral phases (olivine, clinopyroxene and plagioclase), suggests that oxides were formed last, as crystallization of an interstitial residual liquid.



*Fig. 7.13.5 - Different element contents in troctolite portion on sample B108a. Rounded margin of olivine is evident (A, B and D). Plagioclase is zoned (E) with lesser Na contents between olivine crystals. Some little clinopyroxene crystals occurs (C and D). Troctolite shows poor content of oxides (F). A) Al, B) Ca, C) Fe, D) Mg, E) Na, F) Ti.*



*Fig. 7.13.6 - Different element contents in gabbro portions. Plagioclase are interstitial (A), and zoned (E), with higher contents of Na close to olivine crystals. Large clinopyroxene crystals surrounding rounded olivines (C and D). Few oxides crystals occurs, mostly spinels. A) Al, B) Ca, C) Fe, D) Mg, E) Na, F) Ti.*

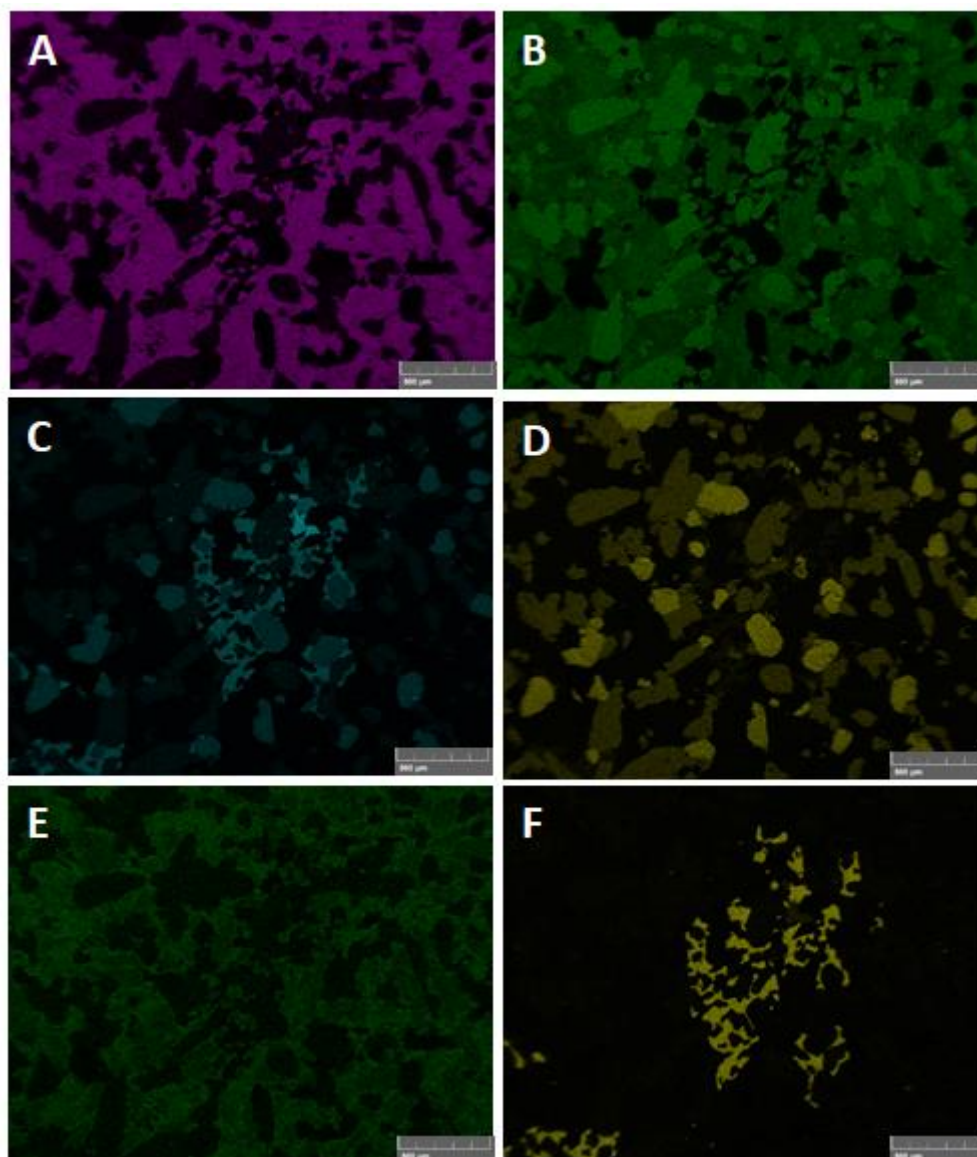


Fig. 7.13.7 - Different element contents in fine-grained gabbro portions. All single mineral phases show rounded margins with sub-hedral aspect (A, B, C and D). Plagioclase are zoned, with higher content of Na close to olivine and clinopyroxene (E). Fe-Ti oxides are massive evident and dendritic, in the interstices and surrounding other crystals. A) Al, B) Ca, C) Fe, D) Mg, E) Na, F) Ti.

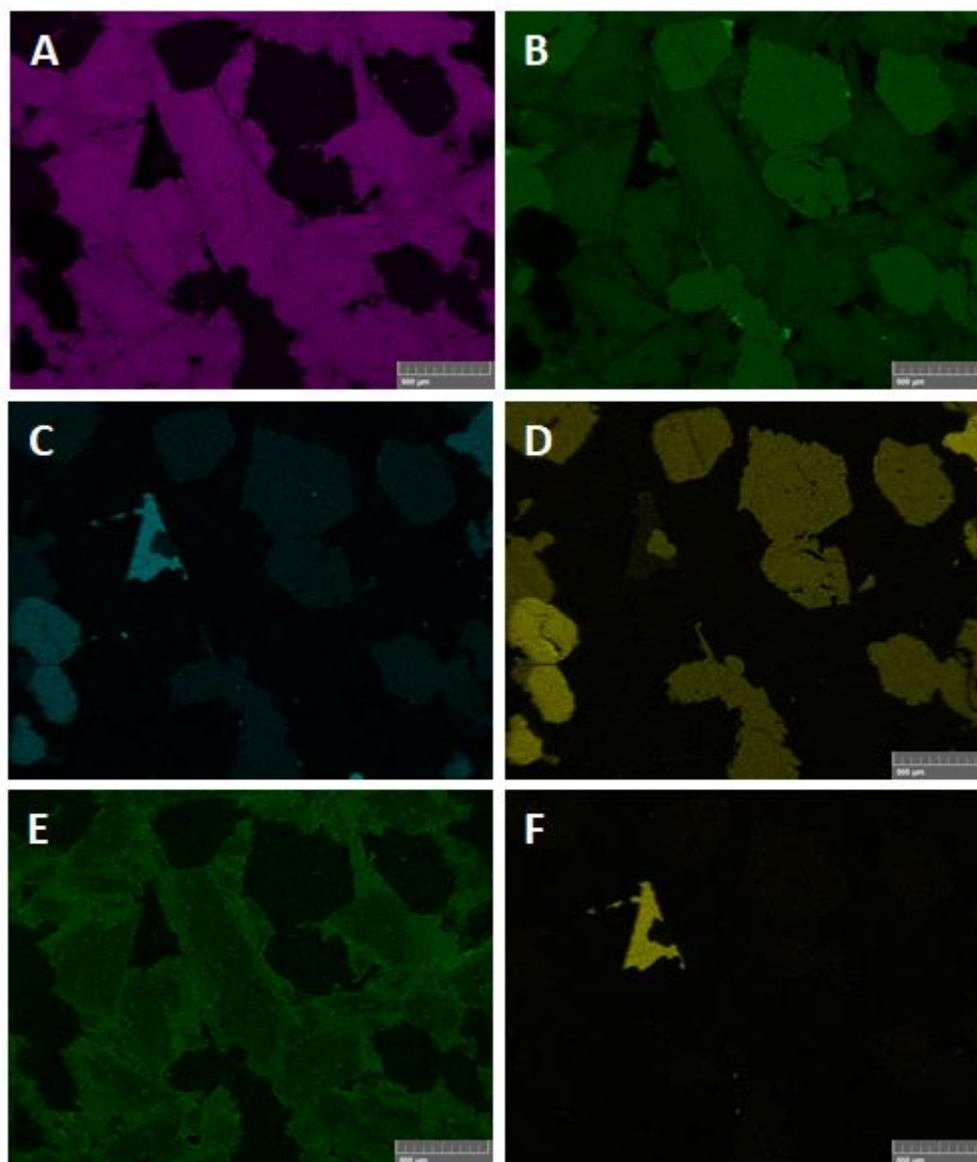


Fig. 7.13.8 - Detail of a portion of fine-grained gabbro. Fe-Ti oxides are interstitial (A, B, C, D, E and F) between plagioclase crystals. Plagioclase are subhedral and zoned, with higher contents of Na next to other single mineral phases and higher contents of Ca at the core of the crystals (B and E). A) Al, B) Ca, C) Fe, D) Mg, E) Na, F) Ti.

#### 7.14 SEM-BSD-EDS analyses on sample B28

The SEM-BSD-EDS analyses on sample B28 permit to improve the study on single mineral phases (Table 7.14.1). This wherlite sample show fritted margin olivine and small rounded olivine surrounded by subhedral clinopyroxene crystals (Figg. 7.14.1 and 7.14.2). The composition of olivine in sample B28 is Fo<sub>79</sub> therefore characterized by a less forsteritic composition with respect olivines in other wherlite samples analysed with EMPA (Fo<sub>80-85</sub>). Zoned clinopyroxene crystal are also evident, showing an increase of Fs from rim (Fs<sub>6</sub>) to the core (Fs<sub>10</sub>).

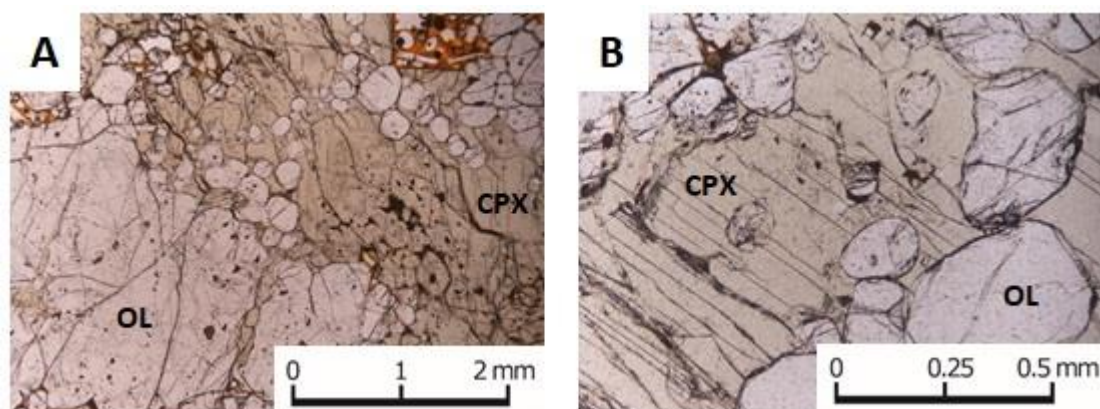


Fig. 7.14.1 - Plane-polarized light images of wherlite B28. A) and B) evidence the presence of rounded olivine and fritted margin olivine. [OL = olivine, CPX = clinopyroxene].

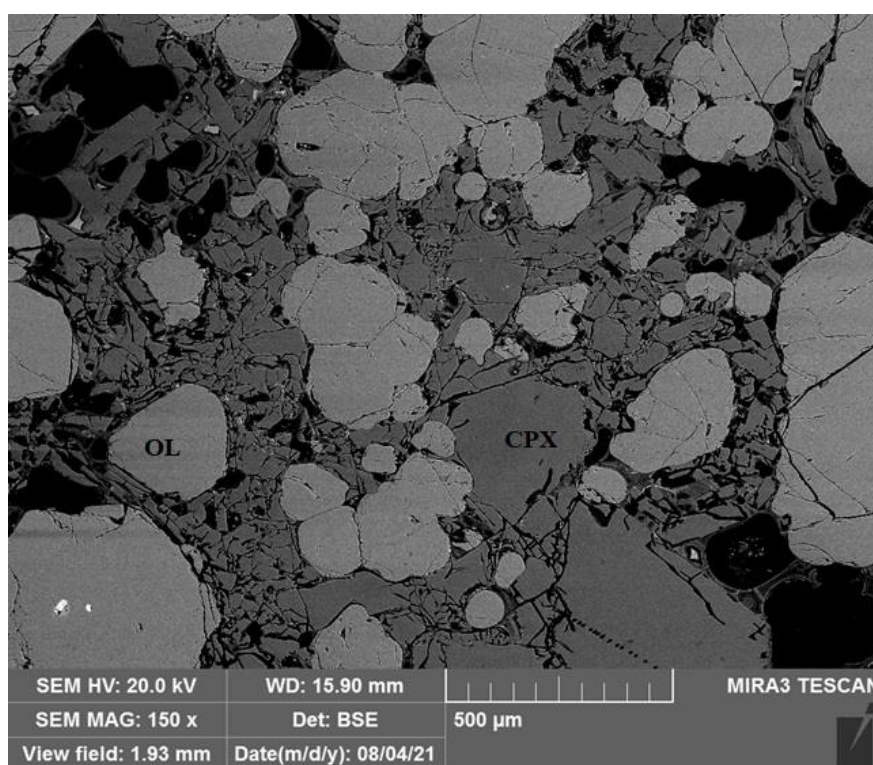


Fig. 7.14.2 - Rounded olivine and fritted margin olivine surrounded by clinopyroxene crystals in wherlitic sample B28. Some olivine crystals show fritted margin.

An EDS map of the sample displays the concentration of several elements, denoted the presence of single mineral phases (olivine, clinopyroxene) and interstitial glass on a particular structure. Al, Ca and Na concentration well evidence subhedral clinopyroxene crystals, surrounded rounded olivines, confirmed by high concentration of Mg and Fe. Interstitial glass is individuated by the presence of Ti (Figg. 7.14.3). The texture of this wherlitic sample suggest a changing on the conditions after the crystallization of the olivine, with consequent rounding of the crystals, and a rapid ascent of the sample, which led to a quenching of interstitial residual liquids forming Ti-rich interstitial glass.



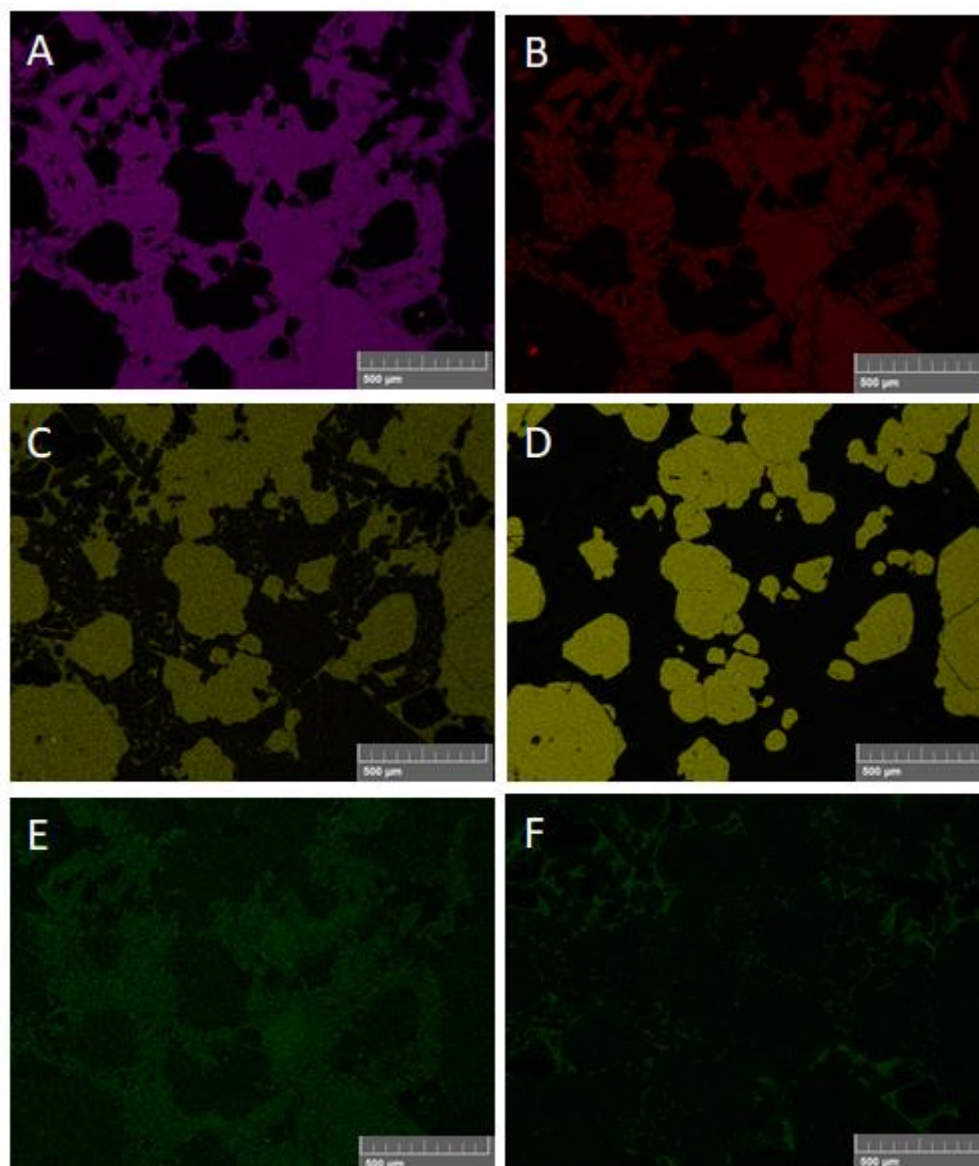


Fig. 7.14.3 - Different element contents on single mineral phase of wherlitic sample B28. Rounded olivine crystals are separated each other from clinopyroxene subhedral crystals (A, B, C and D). Clinopyroxene shows a little zoning (E). B28 sample display a Ti-rich interstitial glass. A) Al, B) Ca, C) Fe, D) Mg, E) Na, F) Ti.

## 8 Geochemistry: major and trace elements

The whole-rock composition of representative subvolcanic ejecta of the Bellecombe Ash Member are reported in Table 8.1. All the samples analysed were compared with literature data. In the legend, recent, Langevin-Bellecombe period and ancient samples are from Observatoire Volcanologique du Piton de la Fournaise (OVPF) database (> 50 ky, 50-3 ky and < 3 ky respectively). Site 1, Site 2, Site 3 and Site 4 are Piton des Neiges samples (Berthod et al., 2020). Other Piton de la Fournaise literature data (Upton et al., 2000) were subdivided in dunite, wherlite and gabbro. Finally, doleritic samples and ballistic block from Plan des Sables (Piton de la Fournaise; Brugier,

thesis, 2016) were compared with other literature data and samples of this PhD thesis.

Whole rock major data were plotted in Fig. 8.1. In addition to the bulk-rock analyses from this study from the Bellecombe Ash Member, data from previous studies (Upton et al., 2000; Berthod et al., 2020; Brugier, thesis, 2016; OVPF literature data subdivided in three categories: Ancient > 50 ky, Langevin-Bellecombe period 50-3 ky, Recent < 3 ky) have been used to provide a comprehensive picture of the whole compositional range. The dunite samples from the Bellecombe Ash Member point out the highest MgO content (46.01 wt%) with respect other dunites from literature (Upton et al., 2000; Berthod et al., 2020). Wherlites and dunites from Cirque of Salazie (Piton des Neiges) show a depletion in Fe<sub>2</sub>O<sub>3</sub> wt% content, indicating a differentiation from a tholeiitic magma, with MgO wt% values ranging between dunites and wherlites from the Bellecombe Ash Member. Wherlites from Bellecombe show a different content in major and trace elements, with sample B3 showing a depletion in MgO wt% and the highest CaO content with respect the other wherlites from Bellecombe. Gabbros from Cirque of Salazie are similar in composition to sub-ophitic gabbros from the Bellecombe Ash Member. Gabbros of Piton de la Fournaise from Upton et al. (2000) can be subdivided in two groups, i) a group with minor MgO wt% content and major and trace elements comparable to mafic ejecta (sub-ophitic, poikilitic and doleritic gabbros) from Bellecombe and ii) a second group showing a higher MgO wt% and major and trace elements comparable to wherlites and porphyrogabbros from the Bellecombe Ash Member. Despite the different textures, porphyrogabbros from Bellecombe are similar in composition to wherlites, with similar MgO wt% content and both major and trace elements. Poikilitic gabbros show some differences in major and trace elements content. In fact, sample B32 show the highest MgO wt% content (25.4 wt%) with respect to other poikilitic samples (MgO 13.6-17.9 wt%), while B21 show the lowest Fe<sub>2</sub>O<sub>3</sub> wt% content. The highest CaO content in mafic and ultramafic samples of the Bellecombe Ash Member compared to other Bellecombe lavas should be the result of higher contribution of pyroxene accumulations ( $\pm$  plagioclase).

Bellecombe Ash Member subvolcanic samples were compared with literature data of intrusive clasts from Piton de la Fournaise (Upton et al., 2000; Brugier et al., 2016; OVPF data) and Piton des Neiges samples (Berthod et al., 2020).

Total alkali vs. silica diagram (Fig. 8.2) displays a Na<sub>2</sub>O + K<sub>2</sub>O wt% content for the investigated Bellecombe Ash Member subvolcanic clasts ranging from 0.04 to 3.5 wt% with a SiO<sub>2</sub> wt% content < 49 wt% (picro-basalt to basalt fields in the TAS diagram). Ancient (> 50 ky) samples are more alkaline (basaltic trachyandesite). Dolerites and some sub-ophitic gabbros are quite close to typical SSB composition.

CaO/Al<sub>2</sub>O<sub>3</sub> vs. MgO variation diagram (Fig. 8.3) of Bellecombe Ash Member ejecta, compared with literature data (Upton et al., 2000;

Berthod et al., 2020; data from OVPF), points out three different trends, interpreted as "fractionation trends": 1) a steep decrease of CaO/Al<sub>2</sub>O<sub>3</sub> ratio correlated with decrease of MgO in the range of ca. 13–6 wt. %: gabbro fractionation; 2) a much less steep decrease of CaO/Al<sub>2</sub>O<sub>3</sub> ratio correlated with decrease of MgO from ca. 36 wt. % downward: wherlite fractionation; 3) a constant CaO/Al<sub>2</sub>O<sub>3</sub> ratio with a marked decrease of MgO wt% from concentration of ca. 46 wt. %: dunite fractionation (or oceanitic trend). Bellecombe Ash Member porphyrogabbro clasts mainly reflect the oceanitic trend, representing a compositional *trait d'union* between ultramafic clasts (dunite and wherlite) and other mafic ejecta. Doleritic gabbros of the Bellecombe Ash Member are similar in compositions to Steady State Basalts (SSB) group.

The trace element composition of the analysed samples is given from Fig. 8.4 to Fig. 8.7. Bellecombe Ash member subvolcanic samples show a REE and incompatible elements depletion with respect to the volcanic samples, especially in comparison with ancient and recent products of Piton de la Fournaise (OVPF data).

CaO/Al<sub>2</sub>O<sub>3</sub> vs. Sc and Th (Fig. 8.8) bulk rock composition (trace element) of Piton de la Fournaise lavas erupted from both peripheral (Boudoire et al., 2019) and central areas (Vlastelic et al., 2007; Peltier, 2007; Boivin and Bachelery, 2009; Salaun et al., 2010; Di Muro et al., 2014, 2015) were compared with Bellecombe Ash Member ejecta. Olivine-bearing basalts from peripheral vents are preferentially depleted in trace elements that are compatible in clinopyroxene (such as Sc). Sub-ophitic gabbros and doleritic gabbros are more similar in composition to peripheral products. Instead, porphyrogabbros of Bellecombe Ash Member show very low Sc content respect to other samples. Olivine-bearing basalts from peripheral vents are preferentially enriched in incompatible trace elements such as Th (Boudoire et al., 2019). Samples of Bellecombe Ash Member reflect peripheral products composition; doleritic gabbros follow a plagioclase-fractionation trend, whereas sub-ophitic gabbros reflect a clinopyroxene fractionation trend.

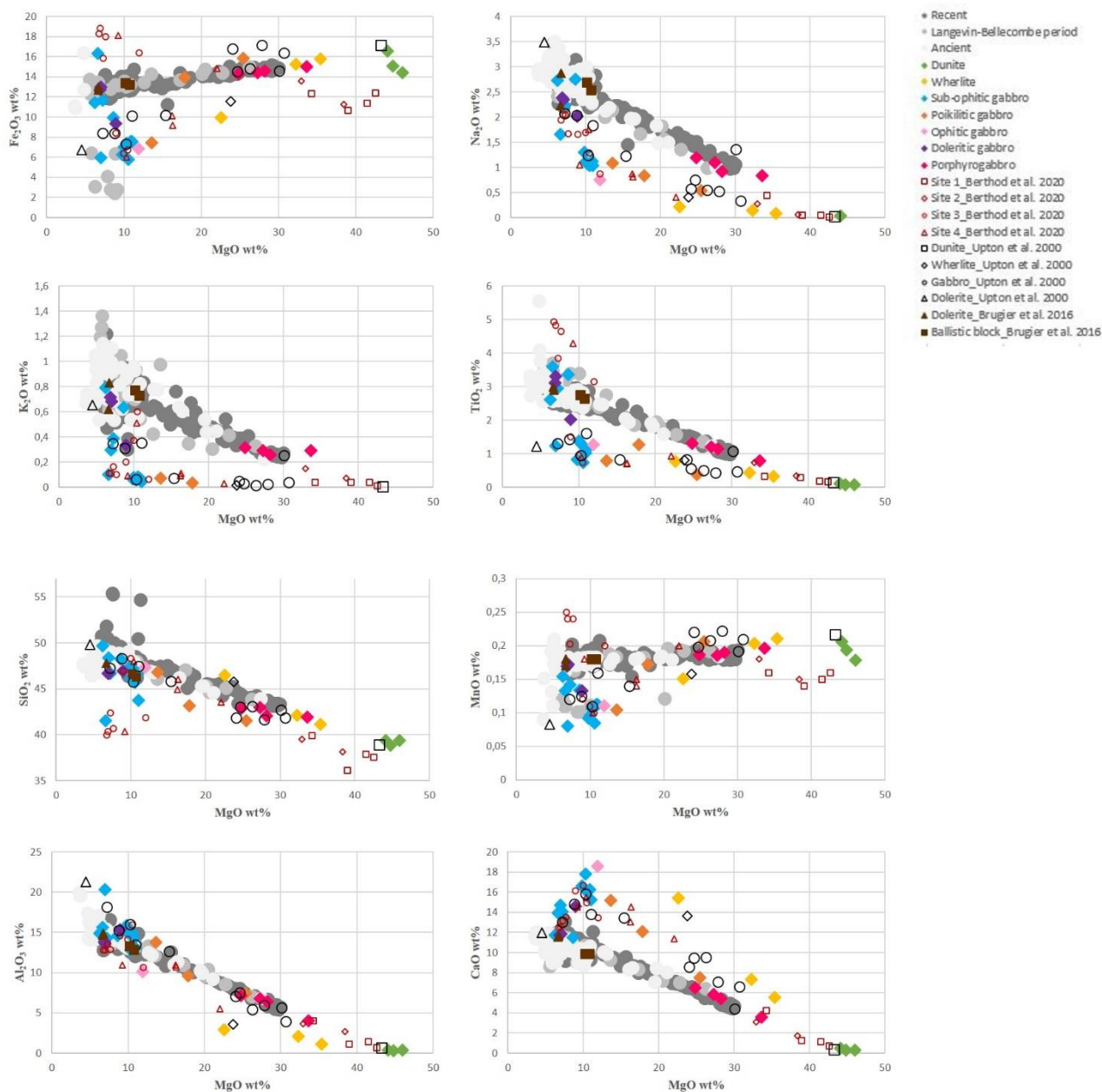


Fig. 8.1 - Major element vs. MgO wt% variation diagrams. All the samples of the Bellecombe Ash Member were compared with literature data.

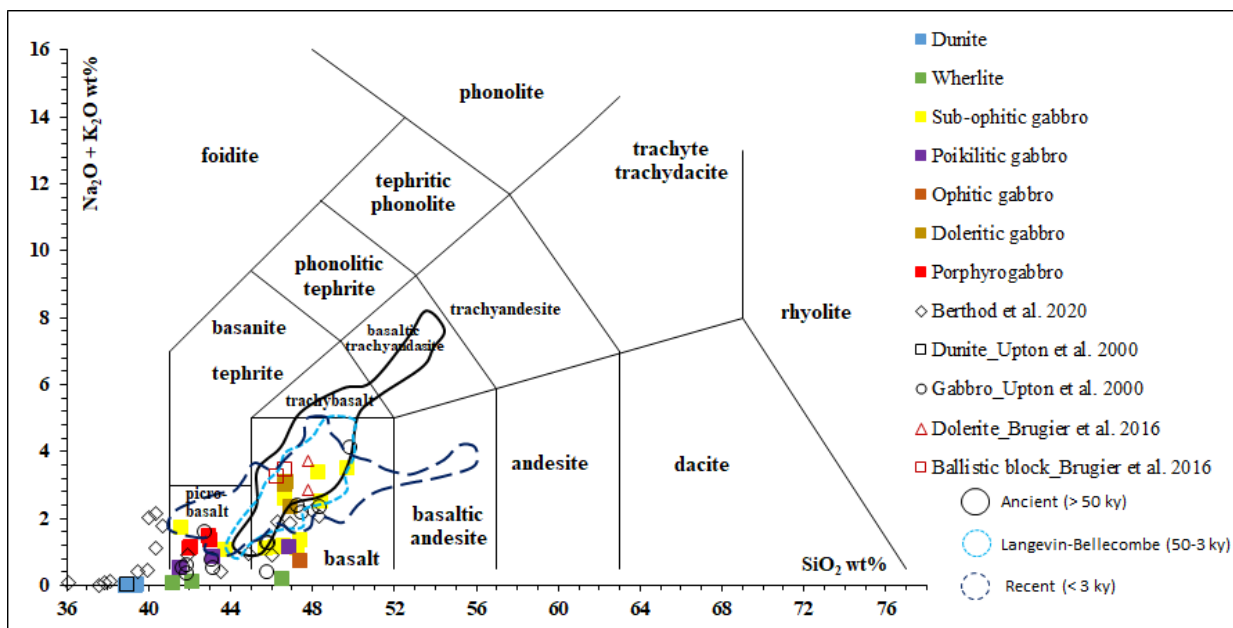


Fig. 8.2 - Total alkali vs. silica graph with analysed representative samples of the Bellecombe Ash Member compared with literature data, from OVPF database, Berthod et al. (2020), Upton et al. (2000) and Brugier (thesis, 2016).

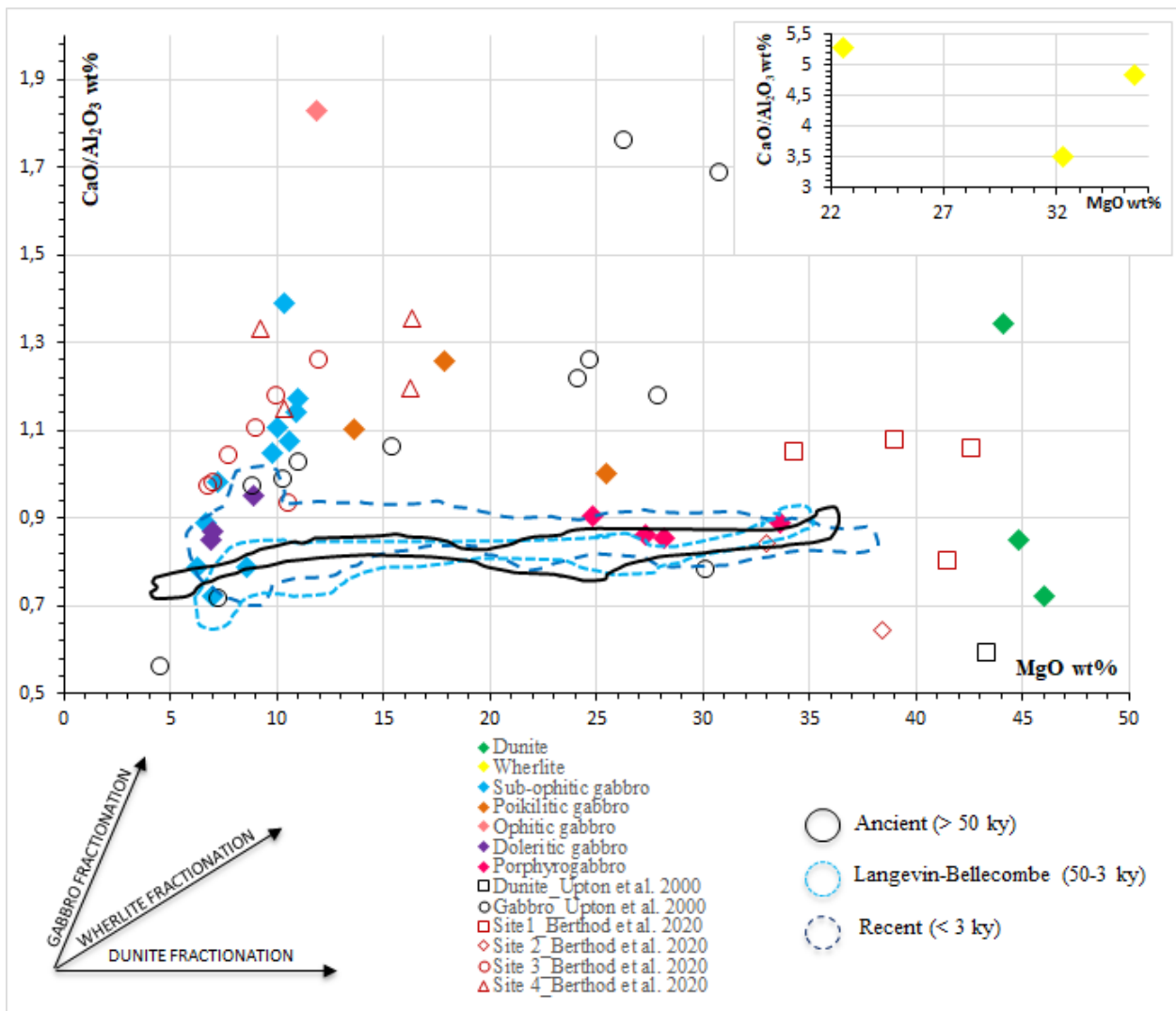


Fig. 8.3 -  $\text{CaO}/\text{Al}_2\text{O}_3$  vs. MgO graph. Bellecombe Ash Member data were compared with literature data (OVPF data; Upton et al., 2000; Berthod et al., 2020).



Fig. 8.4 - Trace element vs. MgO wt% variation diagrams.



Fig. 8.5 - Trace element vs. MgO wt% variation diagrams.



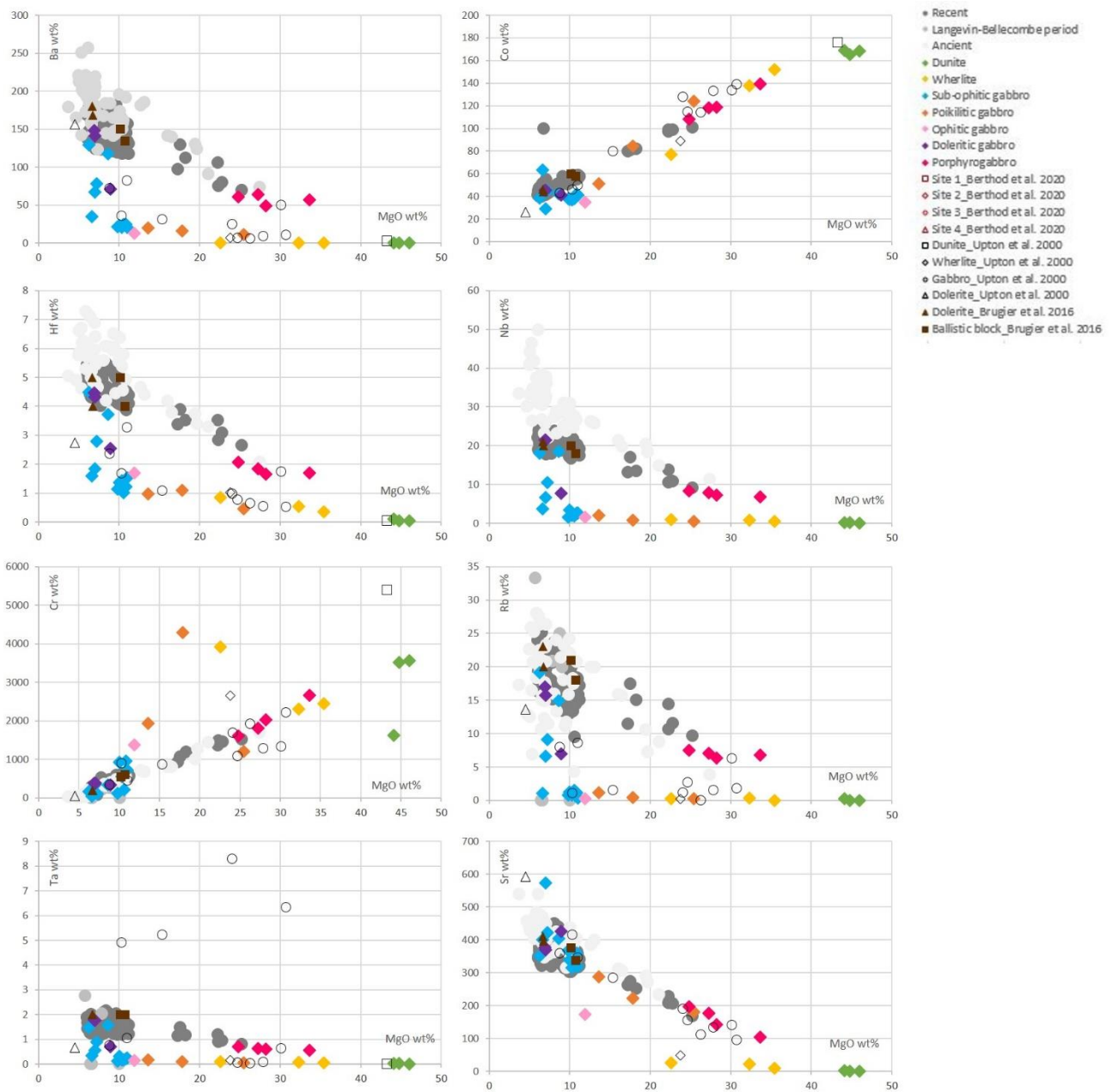


Fig. 8.6 - Trace element vs. MgO wt% variation diagrams.

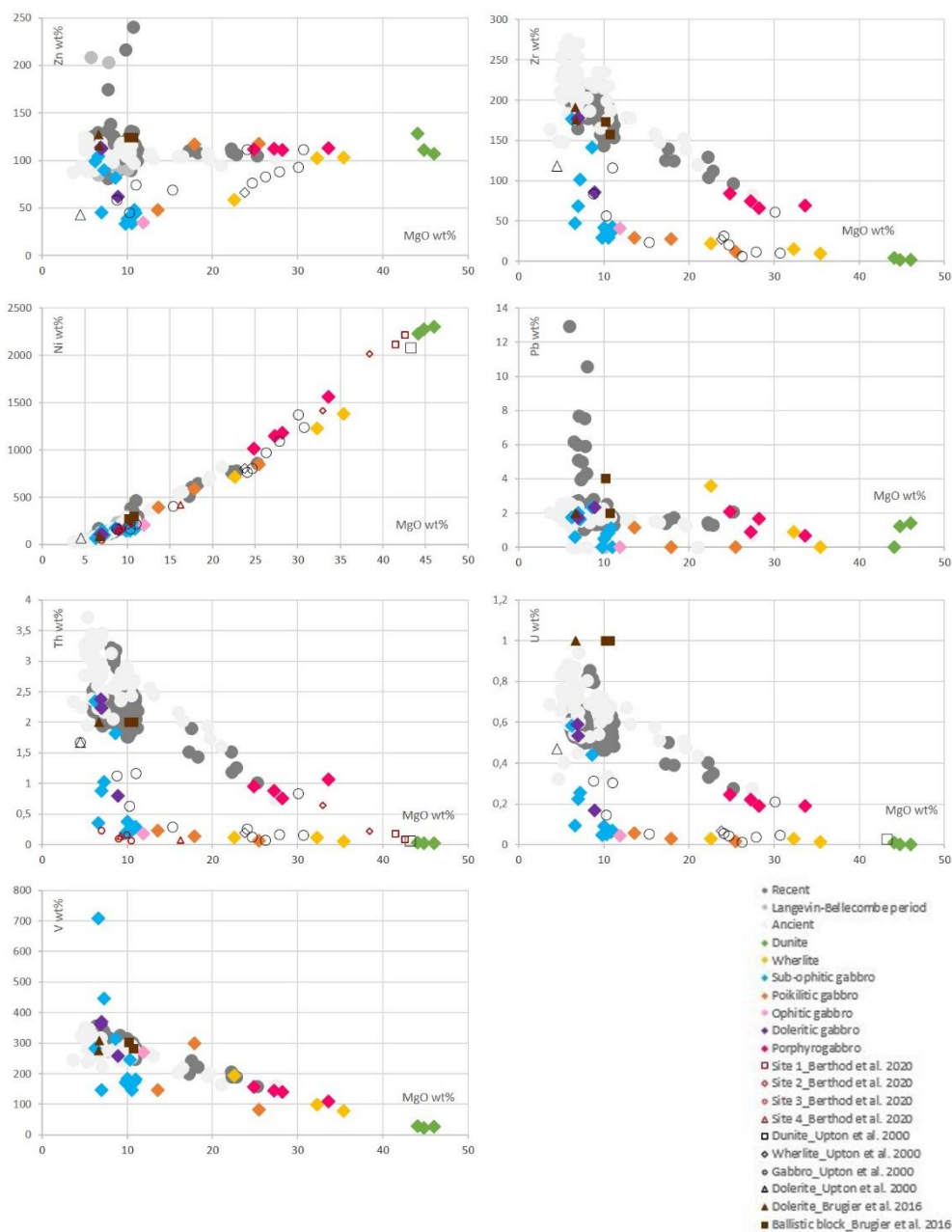


Fig. 8.7 - Trace element vs. MgO wt% variation diagrams.

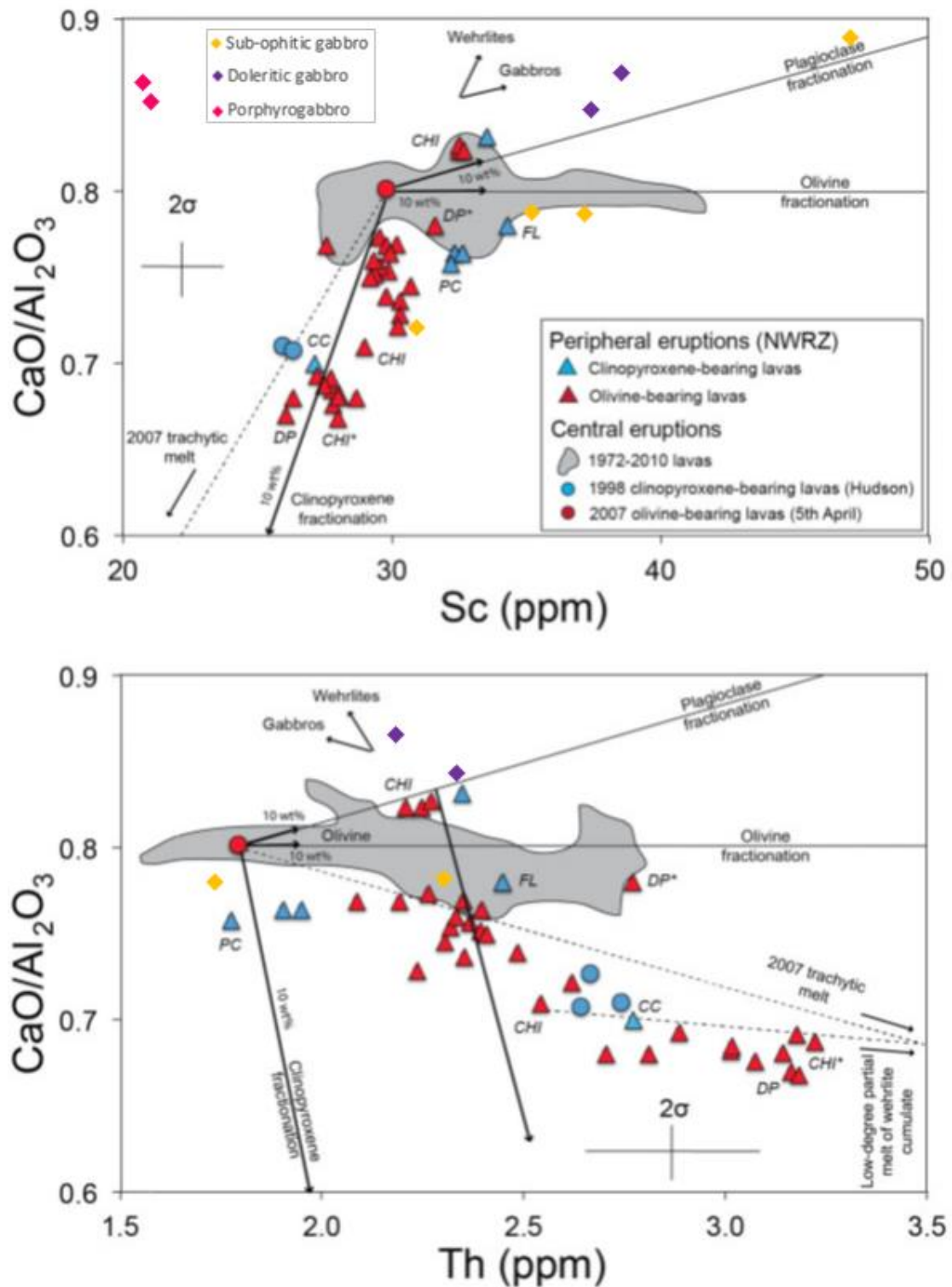


Fig. 8.8 - Bulk-rock composition (trace elements) of Piton de la Fournaise lavas emitted in both peripheral and central areas compared with Bellecombe Ash Member ejecta (Boudoire et al., 2019).

## 9 Mineral phases and glass composition

Representative chemical compositions of the mineral phases, interstitial glass, melt inclusions and trachitic glass of the pumices are given in Table 9.1 (see appendix on chapter 17).

### 9.1 Olivine

Olivine is present in almost all samples analysed. Fo content ranges from 83 to 89 % in the dunites (Fig. 9.1.1) to 29-44 % in the most evolved thrachite samples (pumices of Bellecombe Ash Member; Fig. 9.1.3). Wherlite samples show a Fo content 80-85 %, similar to dunites (Fig. 9.1.2) Sub-ophitic gabbros show a Fo content ranging from 68 to 80 Fo% (Fig. 9.1.4) whereas poikilitic gabbros and porphyrogabbros have a Fo content range from 72-81% and 79-84% respectively (Figg. 9.1.5 and 9.1.6).

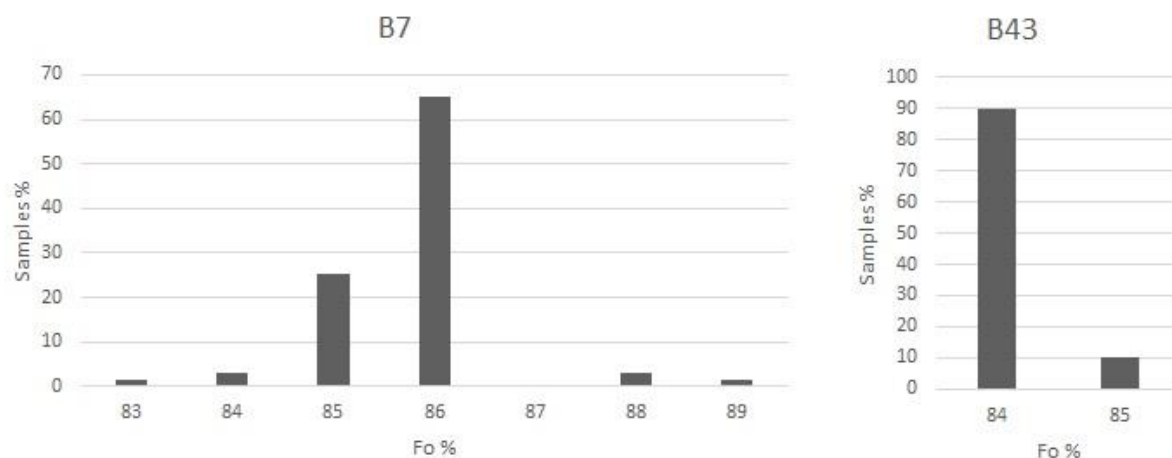


Fig. 9.1.1 - Fo content in different dunite samples showing a compositional range from 83 to 89 %.

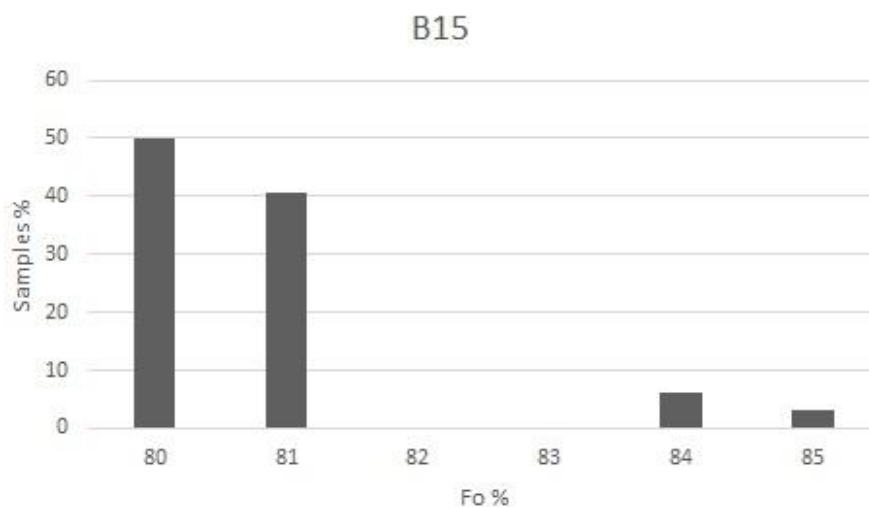


Fig. 9.1.2 - Wherlite sample showing a compositional range of Fo content from 80 to 85%.

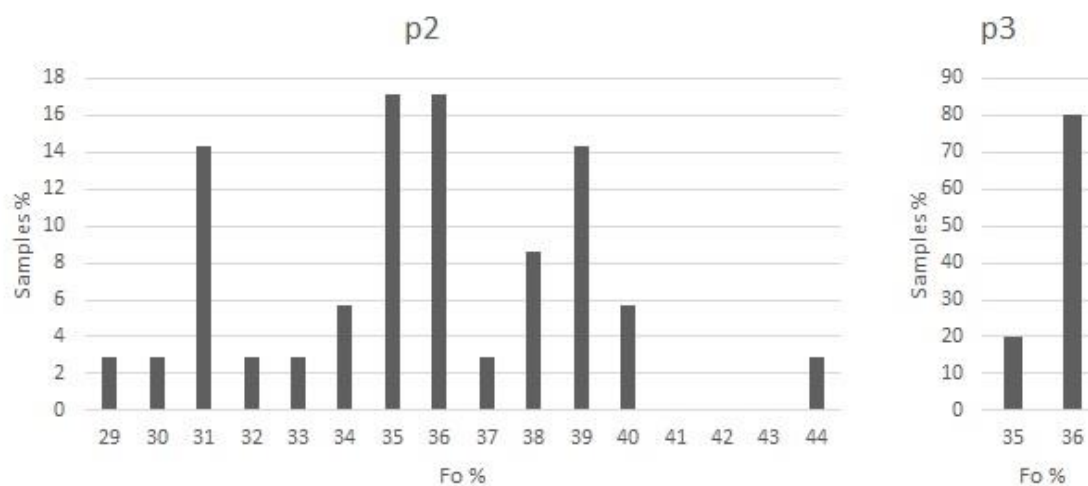


Fig. 9.1.3 - More differentiated samples (trachyte pumices of the Bellecombe Ash Member) showing lower Fo content (29 to 44%).

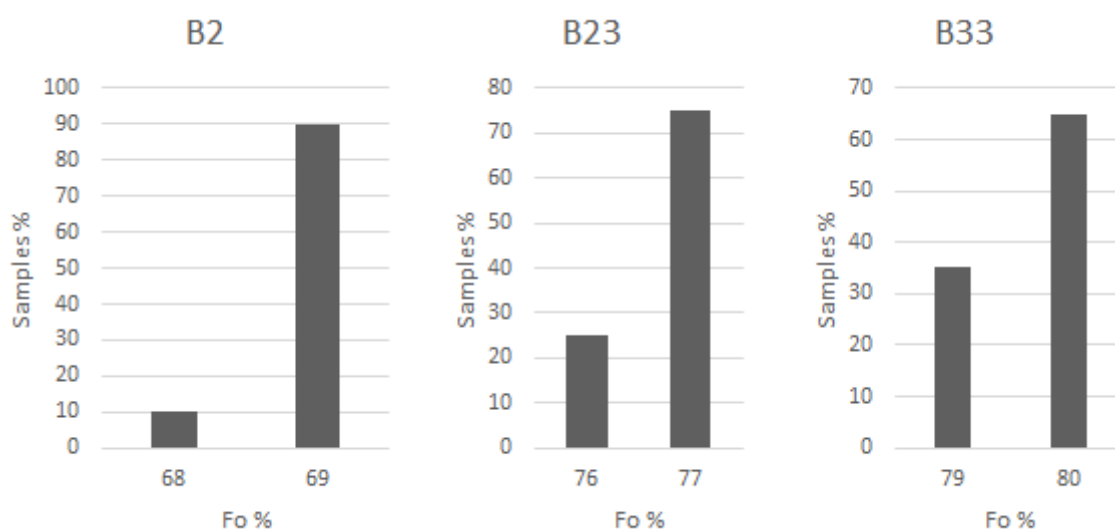


Fig. 9.1.4 - Sub-ophitic gabbro samples showing a wide range of Fo content (from 68 to 80%).

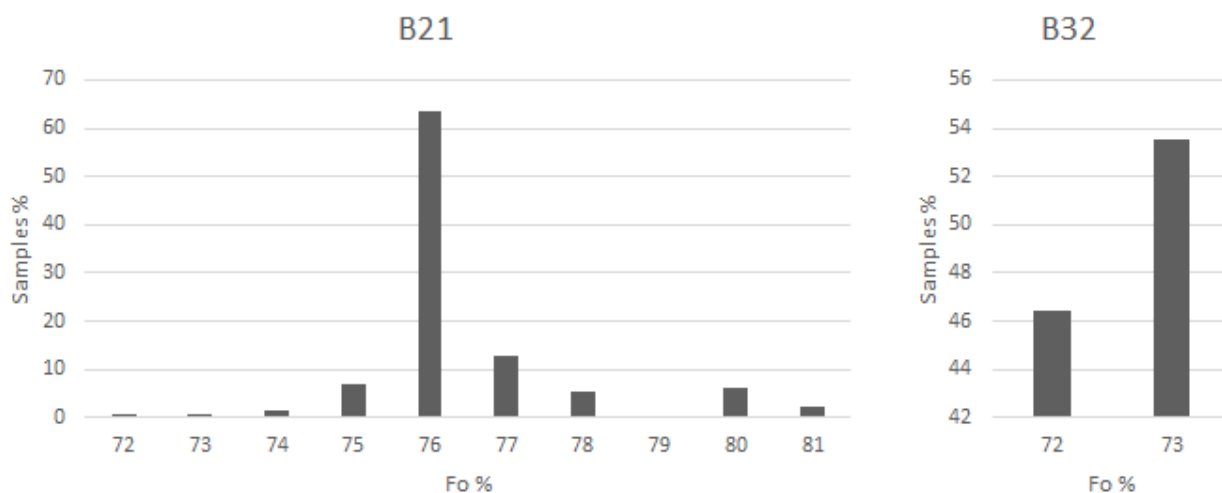


Fig. 9.1.5 - Poikilitic gabbros showing a Fo content from 72 to 81%.

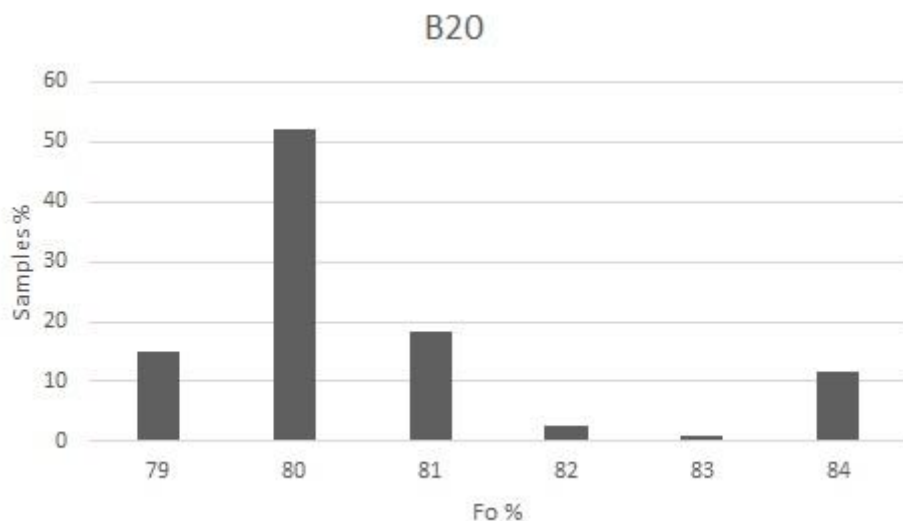


Fig. 9.1.6 - A representative porphyrogabbro sample showing a range of Fo content from 79 to 84%.

To better understand the shallow magmatic system of Piton de la Fournaise, olivine compositions of the Bellecombe Ash Member intrusive ejecta were compared to Boudoire et al. (2018) data from central and peripheral vents in a CaO vs. Fo% diagram (Fig. 9.1.7). Olivines in dunites have a Fo% content comparable to peripheral vents products (Pas des Sables and Chysny enclaves). Olivine found in poikilitic gabbros show Fo% intermediate between olivines in sub-ophitic gabbros and porphyrogabbros at the same range of CaO content (0.15-0.38 wt%). Olivine compositions in wherlitic samples range between those in sub-ophitic gabbros and dunitic samples. Finally, olivines in punices of the Bellecombe Ash Member have the lowest Fo% in the range 0.15-0.30 wt% CaO. In Fig. 9.1.7 is evident that poikilitic gabbro samples show a disequilibrium between crystals and bulk rock. Indeed, these samples are relatively rich in MgO (13.6-25.4 wt%) but show olivine with a Fo% content ranging from 72 to 81

wt%. The same inference can be done for the porphyrogabbro samples, with a MgO from ca. 25 to 34 wt.% and a mean olivine value of 80 Fo% in B20 sample.

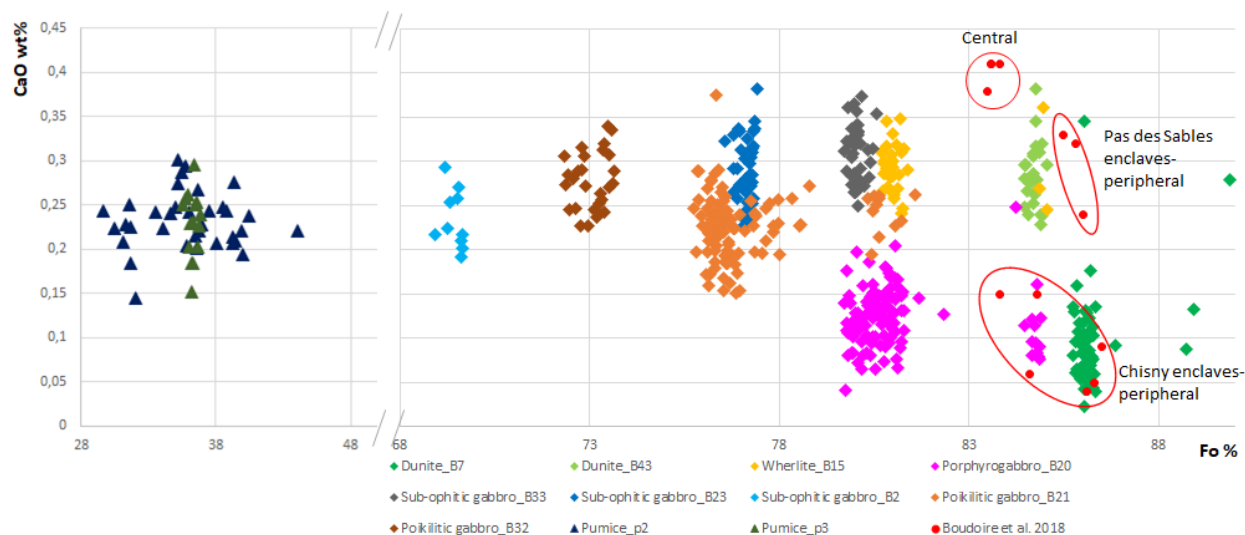


Fig. 9.1.7 - Evolution trend of Bellecombe Ash Member products. Dunites are the most primitive ejecta with the highest Fo%, comparable to products of Piton de la Fournaise peripheral vents.

In Fig. 9.1.8 the olivine compositions in the subvolcanic ejecta of the Bellecombe Ash Member were compared with peripheral-central extrusives and intrusive clasts respectively reported by Boudoire et al. (2018) and Berthod et al. (2020). Berthod et al. (2020) classified the CaO content of olivine into three populations. 1) A low CaO (0.05-0.20 wt%), high Fo<sub>84-89</sub> population in dunites and wherlites from Site 1 and 2; 2) a high CaO (0.20-0.40 wt%), high Fo<sub>80-89</sub> population only found in the dunites and wherlites of Site 2; 3) a low CaO (~0.01-0.12 wt%), low Fo<sub>70-79</sub> population in the wherlite and olivine gabbros of Site 3 and 4. It can be seen that no one Bellecombe Ash Member samples can be included in population 3. A fourth group was recognized, with high CaO content (0.15-0.34 wt%) and Fo<sub>69-79</sub>.

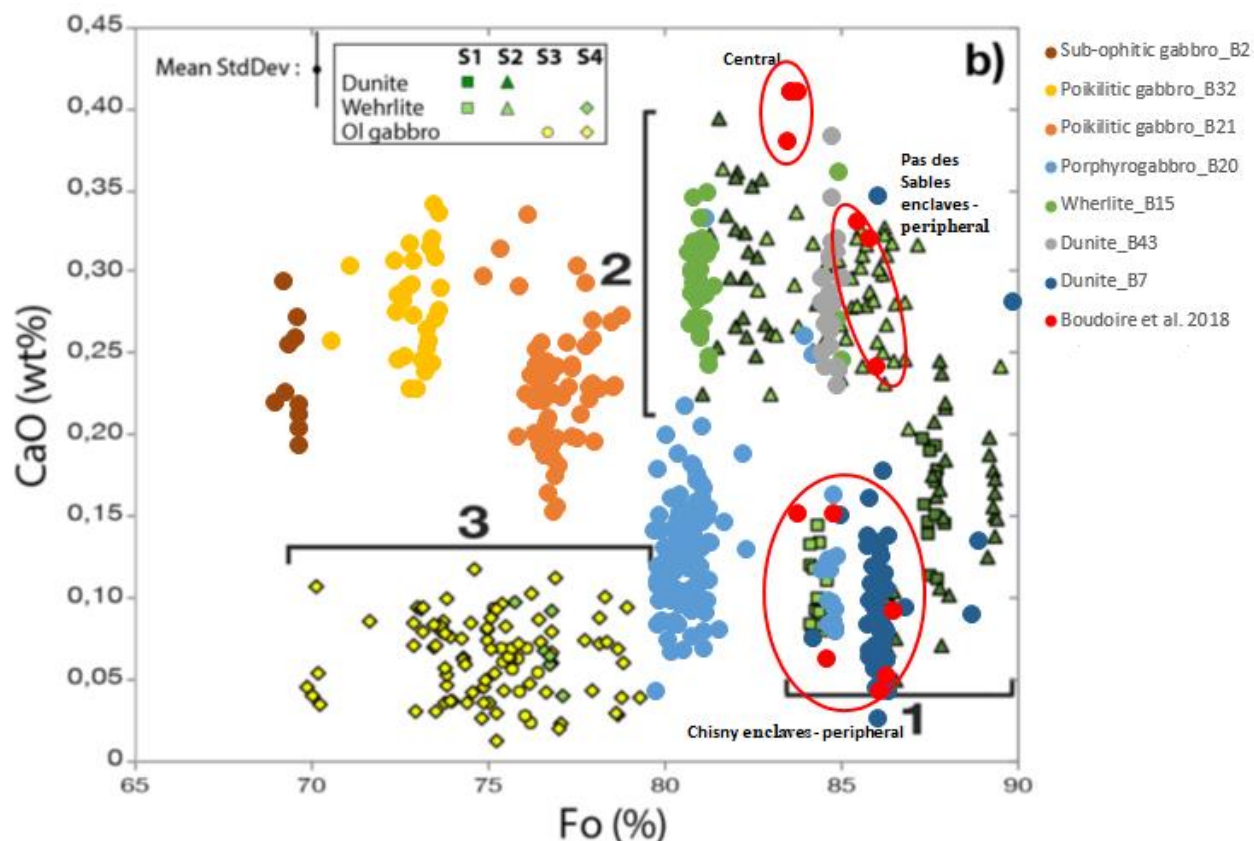


Fig. 9.1.8 - CaO vs. Fo wt%. Bellecombe Ash Member samples were plotted with Boudoire et al. (2018) and Berthod et al. (2020) data. Berthod et al. (2020) subdivided samples in four Site (1-2-3-4) and analysed three lithotypes: dunite, wherlite, ol-gabbro. Boudoire et al. (2018) distinguished samples in peripheral and central origin.

## 9.2 Plagioclase and Na-K feldspar (sanidine and anorthoclase)

Feldspar are present in all mafic samples in different percentages (see chapter 7.5). Triangular phase diagram Ab-An-Or with EMPA analyses on mineral phases (Fig.9.2.1) show the complete miscibility of feldspar on Bellecombe Ash Member samples. Sub-ophitic gabbros and poikilitic gabbros show the presence of plagioclase ranging from Ab<sub>13-42</sub> and An<sub>57-87</sub>. Porphyrogabbros have some sanidine microcrolites (Ab<sub>34-45</sub> An<sub>4-8</sub> Or<sub>49-60</sub>) as well as plagioclase (ranging from Ab<sub>32-58</sub> and An<sub>32-66</sub>). Plagioclases in the pumices of Bellecombe Ash Member show a more albitic content (pumice p2 ranging from Ab<sub>53-76</sub> and An<sub>17-45</sub>, p3 from Ab<sub>41-67</sub> and An<sub>29-57</sub>). It is worth to note that pumices of Piton des Neiges (PDN1; Fig.9.2.1) have some anorthoclase (Ab<sub>66-68</sub> An<sub>11-12</sub> Or<sub>20-23</sub>) as well as plagioclase (ranging from Ab<sub>58-68</sub> and An<sub>26-38</sub>).



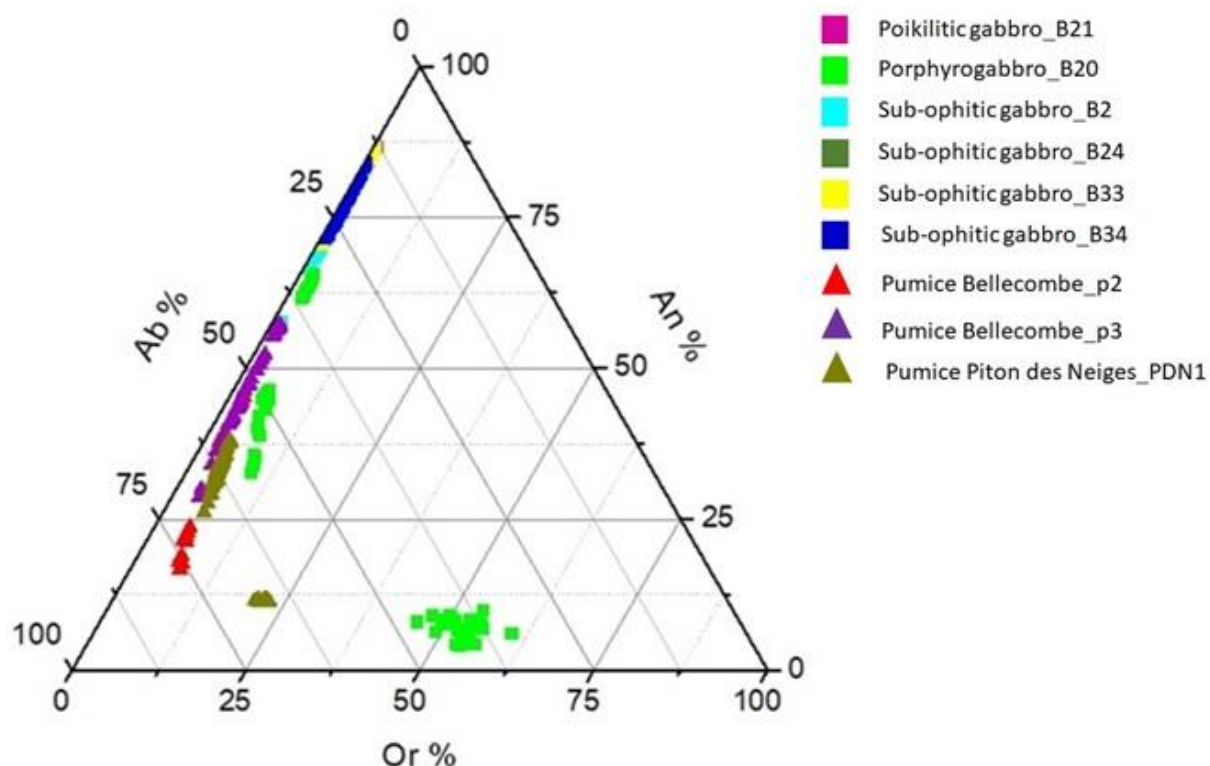


Fig. 9.2.1 - Ternary feldspar diagram (Ab-An-Or, EMPA data).

### 9.3 Clinopyroxene

In almost all samples, analysed clinopyroxene is present with different modal content and composition. Fs content range from 6-11% in few crystals of dunite samples (Fig. 9.3.1) to 17-23% in the trachyte pumices of Bellecombe Ash Member and Piton des Neiges; Fig. 9.3.2). Clinopyroxene compositions of wherlitic samples (Fs 8%; Fig. 9.3.3) are similar to those in the dunites. Clinopyroxene of porphyrogabbros and sub-ophitic gabbros are very similar, in the range  $Fs_{8-13}$  and  $Fs_{9-14}$  respectively (Figg. 9.3.4 and 9.3.5). Clinopyroxene compositions of poikilitic gabbros are also comparable ( $Fs_{8-15}$ ; Fig. 9.3.6).

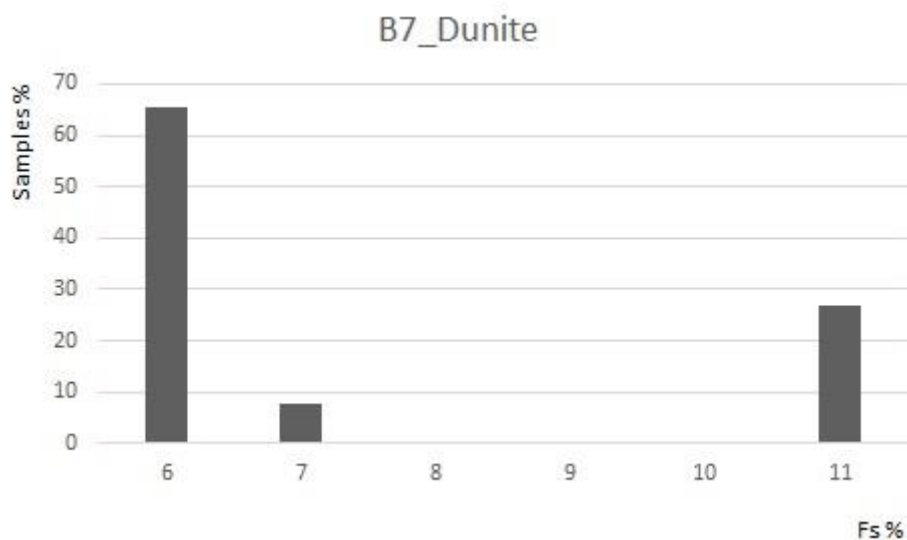


Fig. 9.3.1 - Dunite samples of Bellecombe Ash Member show a limited range of composition for Fs% content of clinopyroxene.

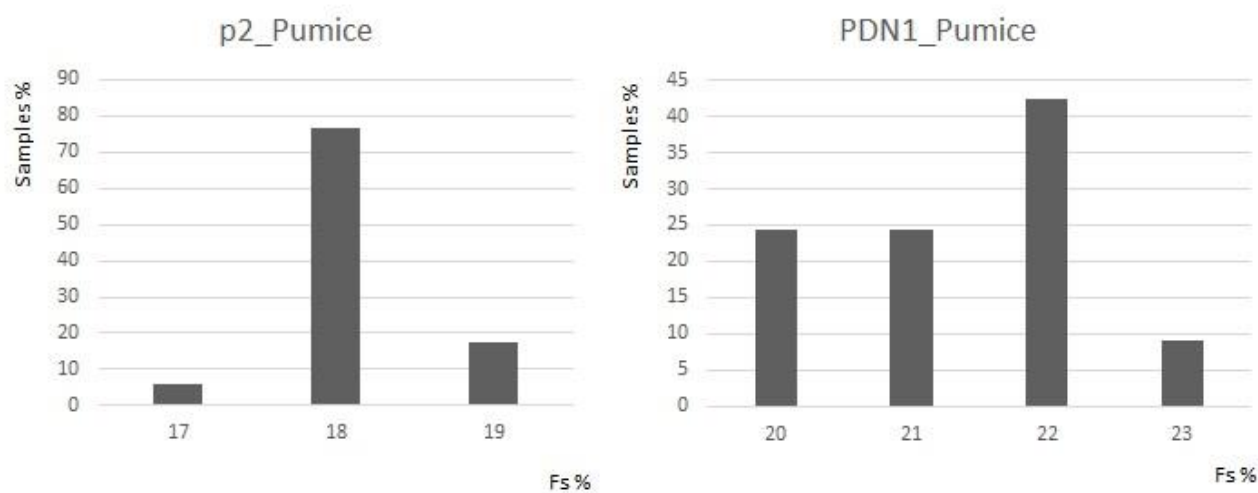


Fig. 9.3.2 - Both pumices of Piton de la Fournaise (p2) and Piton des Neiges (PDN1) show the highest content of Fs% of clinopyroxene.

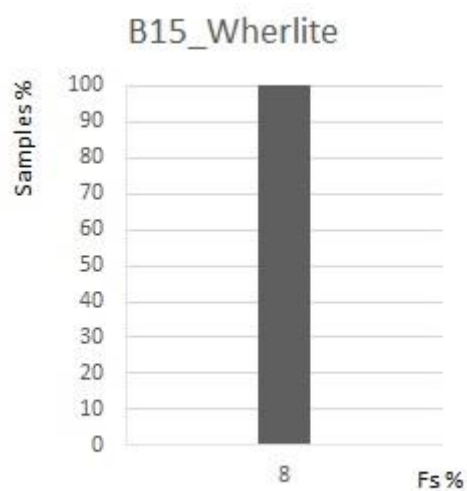


Fig. 9.3.3 - Fs% of clinopyroxene in the wherlite B15 sample.

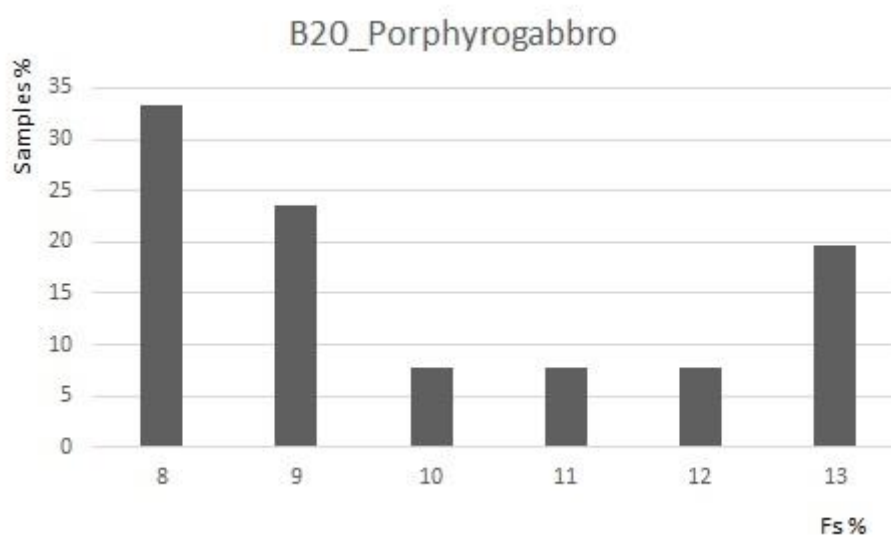


Fig. 9.3.4 - Fs% compositional variation of clinopyroxenes in the porphyrogabbro sample B20.

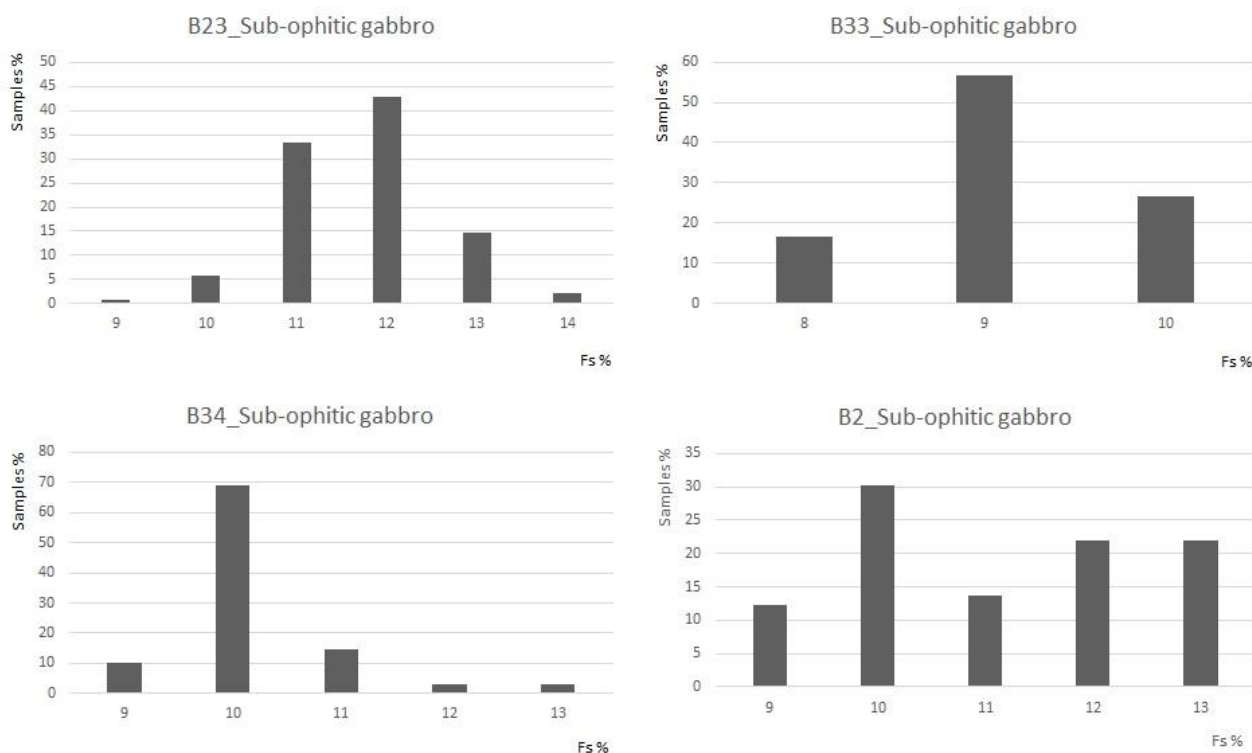


Fig. 9.3.5 - *Fs%* compositional variation of clinopyroxene in sub-ophitic gabbros.

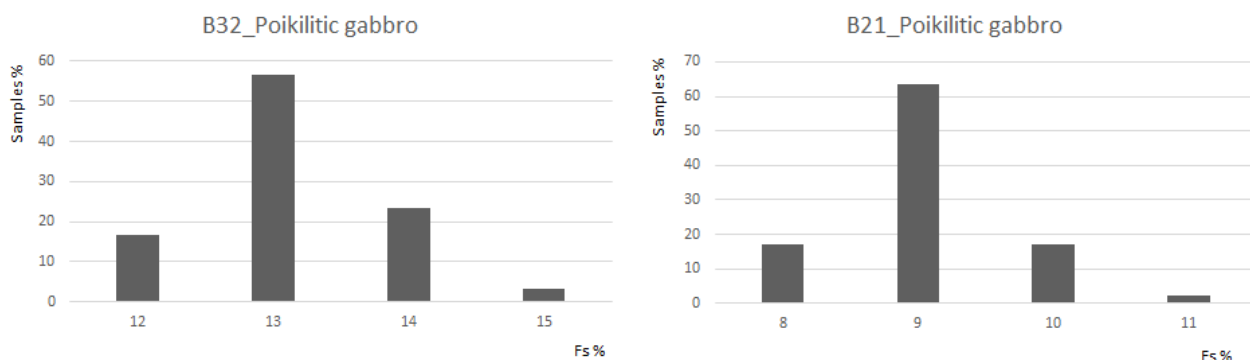


Fig. 9.3.6 - Composition of clinopyroxenes (*Fs%* content) in the poikilitic gabbros

Clinopyroxene of porphyrogabbros show a wide range of  $\text{Al}_2\text{O}_3$  content (1.7–7.5 wt%; Fig. 9.3.7), whereas some clinopyroxenes of the poikilitic gabbros have the highest Mg# 84.7 among the Mg values of the whole suite of the mafic intrusive clasts, whereas dunites show the highest Mg# (86–88) on clinopyroxene between the ultra-mafic intrusive clasts.  $\text{Al}_2\text{O}_3$  vs. Mg# show three range of composition for Mg# content: evolved samples (pumices of Bellecombe and Piton des Neiges), mafic rocks + wherlite and dunite. Clinopyroxene of pumices of Bellecombe and Piton des Neiges show the lowest  $\text{Al}_2\text{O}_3$  content (1.6–3.3 wt%) with the lowest Mg# (55.5–68.0 Mg#). Comparing Bellecombe Ash Member samples with Piton des Neiges data (Fig. 9.3.8; Berthod et al., 2020), it can be noticed that their composition are similar. Comparing all clinopyroxene crystals both mafic and ultramafic rocks, Mg# increases whereas  $\text{TiO}_2$  decrease from mafic to ultramafic rocks (Fig. 9.3.9).

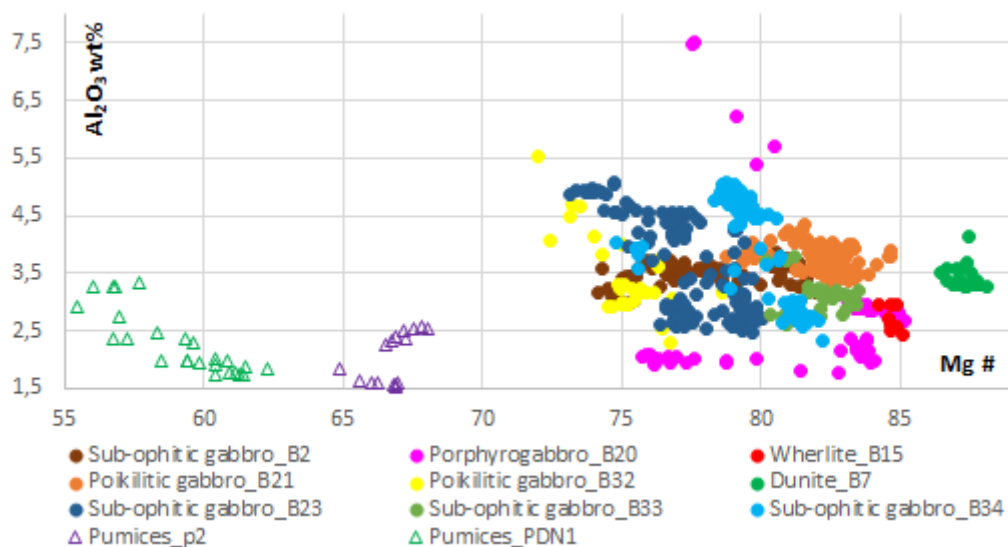


Fig. 9.3.7 -  $Al_2O_3$  vs. Mg# of Bellecombe Ash Member and Piton des Neiges samples (pumices PDN1).

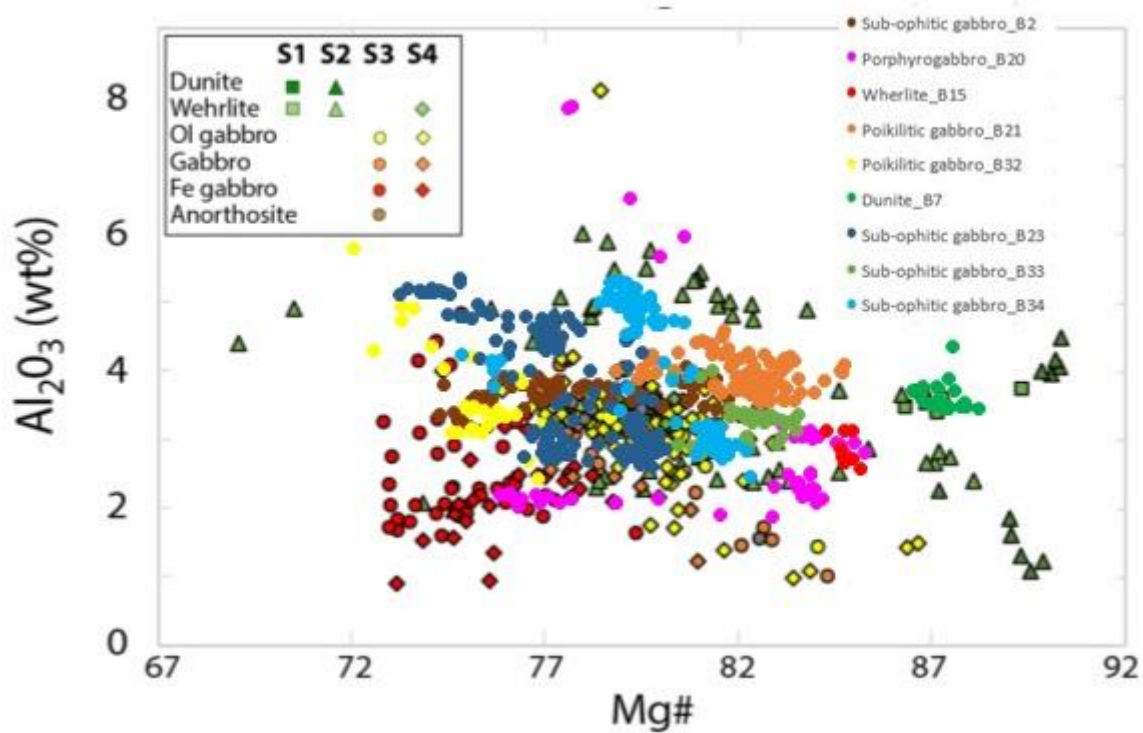


Fig. 9.3.8 - Chemical composition of Bellecombe Ash Member samples compared with samples investigated by Berthod et al. (2020).

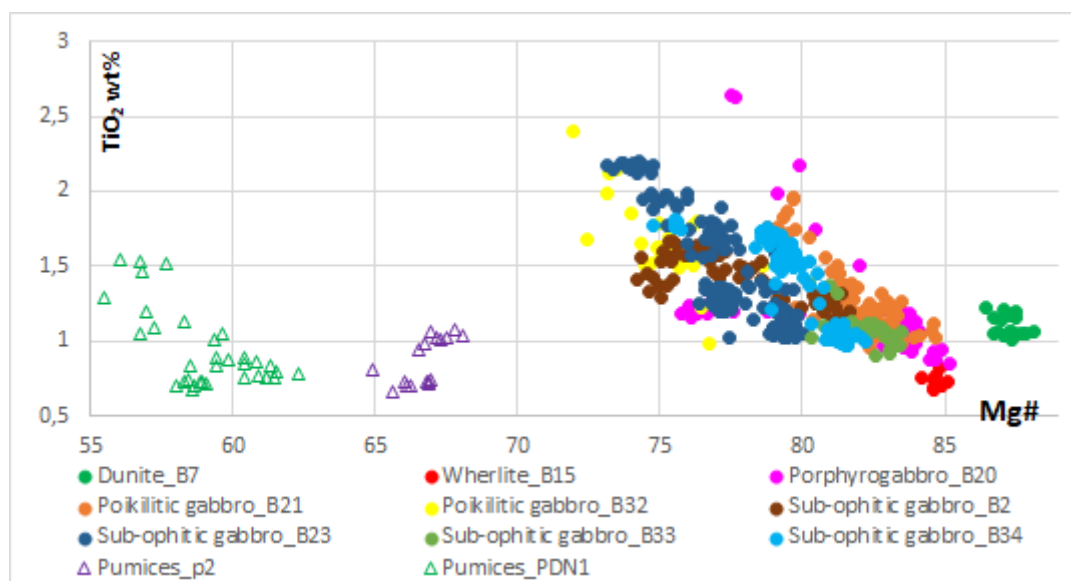


Fig. 9.3.9 -  $TiO_2$  of clinopyroxene decreasing from more evolved samples (mafic rocks) to more primitive one (ultramafic rocks). Pumices show moderate  $TiO_2$  content and the lowest Mg#.

#### 9.4 Opaque minerals (Cr-spinel, magnetite and ilmenite)

Opaque minerals includes Cr-spinel, Ti-magnetite and ilmenite. Not all the mafic and ultramafic groups contain oxides and if present in small percentages. Cr-spinels in dunites show MgO content ranging from 10.5 to 11.5 wt%, high  $Cr_2O_3$  contents (34.9–29.3 wt%), moderate  $Al_2O_3$  content (15.6–18.7 wt%), FeO 28.6–31.2 wt% and  $TiO_2$  2.5–3.4%. Wherlite show Cr-spinels with  $Al_2O_3$  content ranging from 15.0 to 15.5 wt%, MgO 9.5–10.0 wt%,  $Cr_2O_3$  26.7–31.3 wt%, FeO 37.7–39.4 wt% and low  $TiO_2$  (3.6–4.8 wt%). In poikilitic gabbros oxides show low  $Al_2O_3$  (0.5–10.7 wt%) and  $Cr_2O_3$  (0.2–16.2 wt%), with high FeO (40.5–69.5 wt%). MgO ranging from 4.5 to 7.8 wt% and  $TiO_2$  show a wide range of content 4.6–50.0 wt%, therefore implying both magnetite and ilmenite. Opaque minerals in the sub-ophitic gabbros are ilmenites having moderate FeO (43.0–45.9 wt%) and high  $TiO_2$  (47.2–49.4 wt%), with low  $Al_2O_3$  (0.4–0.5 wt%) and MgO (5.2–5.7 wt%) contents. Oxides in porphyrogabbros show a wide range in composition for each major element: MgO 6.3–8.1%,  $Cr_2O_3$  0.4–27.0 wt%,  $Al_2O_3$  0.9–8.2 wt%, FeO 23.3–50.6 wt% and  $TiO_2$  2.6–62.1 wt% therefore emphasizing the presence of magnetite, ilmenite and Cr-spinels.

#### 9.5 Phlogopite

Phlogopite is evident only in porphyrogabbros as interstitial phase, often in association with opaque minerals. The composition is  $SiO_2$  35.1–40.7 wt%,  $Al_2O_3$  11.9–13.8 wt%, MgO 16.5–19.3 wt%, FeO 6.6–7.5 wt% and  $K_2O$  8.0–9.4 wt%.

### 9.6 Interstitial glass

Not all the representative samples analysed show interstitial glass. Glass is common in sub-ophitic gabbros and poikilitic gabbros, with a pale yellow (sideromelane) to pale-brown colour appearance in thin section. They can be also partially oxidised (yellow-orange to tachylite). The composition of the interstitial glass in the sub-ophitic gabbros is SiO<sub>2</sub> 48.5-57.7 wt%, TiO<sub>2</sub> 2.3-3.3 wt%, Al<sub>2</sub>O<sub>3</sub> 14.3-16.3 wt%, Cr<sub>2</sub>O<sub>3</sub> 0-0.1 wt%, FeO<sub>tot</sub> 6.9-12.1 wt%, MnO 0.1-0.2 wt%, MgO 2.3-6.3 wt%, CaO 5.1-12.3 wt%, Na<sub>2</sub>O 1.2-5.8 wt%, K<sub>2</sub>O 0.9-3.0 wt%, P<sub>2</sub>O<sub>5</sub> 0.3-1.0 wt%. In the poikilitic samples the composition of the interstitial glass is SiO<sub>2</sub> 53.2-56.0 wt%, TiO<sub>2</sub> 3.1-3.4 wt%, Al<sub>2</sub>O<sub>3</sub> 15.8-16.4 wt%, Cr<sub>2</sub>O<sub>3</sub> 0-0.7 wt%, FeO<sub>tot</sub> 7.4-9.0 wt%, MnO 0.1-0.2 wt%, MgO 3.3-4.1 wt%, CaO 6.5-7.7 wt%, Na<sub>2</sub>O 3.4-4.7 wt%, K<sub>2</sub>O 1.8-2.2 wt%, P<sub>2</sub>O<sub>5</sub> 0.7-0.9 wt%.

### 9.7 Melt inclusions

Melt inclusions are present in representative analysed samples of the Bellecombe Ash Member. Crystals of clinopyroxene, olivine, plagioclase and oxides show small melt inclusions in sub-ophitic gabbros. The composition of those melt inclusions is SiO<sub>2</sub> 45.2-51.4 wt%, TiO<sub>2</sub> 1.9-4.3 wt%, Al<sub>2</sub>O<sub>3</sub> 13.1-14.9 wt%, Cr<sub>2</sub>O<sub>3</sub> 0-0.1 wt% FeO<sub>tot</sub> 9.5-14.9 wt%, MnO 0.10-0.3 wt%, MgO 3.5-6.2 wt%, CaO 6.4-11.5 wt%, Na<sub>2</sub>O 2.2-4.6 wt%, K<sub>2</sub>O 0.7-1.9 wt%, P<sub>2</sub>O<sub>5</sub> 0.2-0.8 wt%. In addition, samples of pumices from Bellecombe Ash Member (p2 and p3) and Piton des Neiges (PDN1) show some melt inclusions on few crystals too. The composition of those melt inclusions is more evolved than other samples, as evidenced in total alkali vs. silica graph (Fig. 9.8.1), SiO<sub>2</sub> ranging from 61.3 wt% to 67.8 wt%, with Na<sub>2</sub>O + K<sub>2</sub>O 8.5-11.5 wt%.

### 9.8 Trachitic glass of the pumices

Some pumice samples from the Bellecombe Ash Member and Piton des Neiges were also investigated. They show a vesiculated-quenched texture, with a porphyritic index variable from 1 to 3 vol%. Bellecombe Ash Member (p2, p3) pumices also show the presence of few crystals of olivine, feldspar and clinopyroxene, whereas pumice from Piton des Neiges (PDN1) only display feldspars, clinopyroxene and opaque minerals.

Pumices from the Bellecombe Ash Member (p2 and p3) and Piton des Neiges (PDN1) show a more evolved groundmass glass composition than other interstitial glass in the subvolcanic samples of the Bellecombe Ash Member analysed. It is trachytic in composition (PDN1 59.4 wt% SiO<sub>2</sub> and 11.1 wt% alkali; p2 70.4 wt% SiO<sub>2</sub> and 10.6 wt% alkali; Fig. 9.8.1). Comparing all melt inclusions and interstitial glass data with whole rock literature data it is evident the different composition between intrusive ejecta of the Bellecombe Ash Member and the pumices (both Bellecombe Ash Member and Piton des Neiges). Dunites of Bellecombe Ash Member show different composition of interstitial glass. Sample B7 show a basaltic trachyandesite

composition, whereas interstitial glass of dunite B43, the sample with large crystals of olivine presenting kink-banding and rounded olivines, is basaltic. Interstitial glass of the sub-ophitic and poikilitic gabbros has a composition basaltic trachyandesitic, but melt inclusions are less evolved, basaltic with composition similar to the interstitial glass of dunite sample B43.

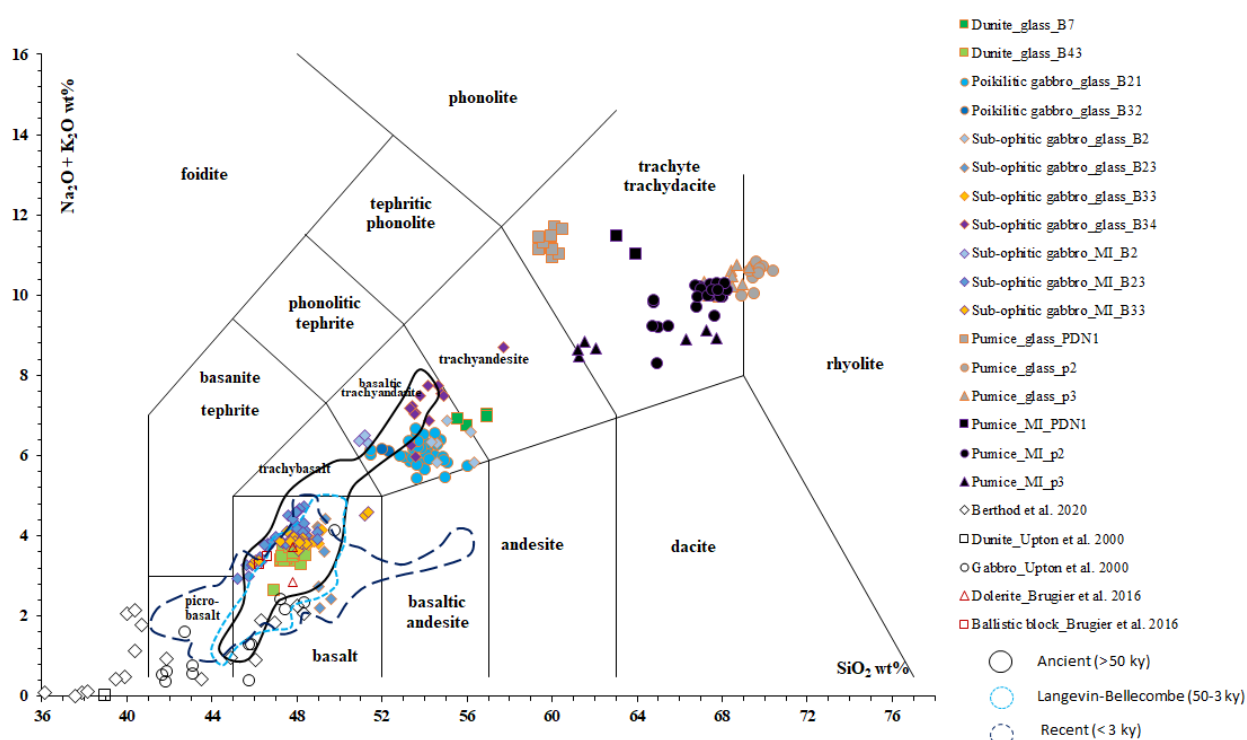


Fig. 9.8.1 - Total alkali vs. silica diagram of interstitial glass and melt inclusions of analysed representative subvolcanic samples of Bellecombe Ash Member (dunites and sub-ophitic gabbros) and some trachyte pumice samples (p2, p3: Bellecombe Ash Member; PDN1: Piton de Neiges) compared with whole rock literature data, from OVPF database, Berthod et al. (2020), Upton et al. (2000) and Brugier (thesis, 2016).

## 10 Fluid inclusions

Fluid inclusions in single mineral phases were analysed using both thin section and single hand-picked mineral phase (olivine, clinopyroxene, feldspar).

### 10.1 Thermobarometric evaluation

A Raman spectroscopy analyses were conducted on single hand-picked mineral phases of representative samples, both mafic and ultramafic, for a thermobarometric evaluation. Ultramafic lithotypes provided valuable results whereas no reliable results were obtained from mafic samples, due to the low CO<sub>2</sub> content. Dunite (B7) had a density > to wherlite (B3). Wherlite had ~86% of the density range from 0.34 to 0.42 g/cm<sup>3</sup>, with only a little percentage (~14%) about 0.26 g/cm<sup>3</sup>.



Dunite show a wider range of CO<sub>2</sub> density, with ~32% from 0.54 to 0.64 g/cm<sup>3</sup> and ~ 68% from 0.34 to 0.5 g/cm<sup>3</sup> (Fig. 10.1.1). Minor pressure registered on fluid inclusions of wherlite could be at least partly related to the fact that fluid inclusions analysed were in clinopyroxene, which conserve worse the barometric data than olivine. Lamadrid (2017) equation for CO<sub>2</sub> Dyad splitting Raman spectroscopy have provided an indicative provenance of both dunite and wherlite samples of Bellecombe. Assuming a temperature of formation range of 1108-1225°C, dunite show a pressure of 309-338 MPa, corresponding to 13 km depth, ~ underplating level; wherlite sample pressure is about 151-166 MPa, and it correspond to 6 km depth, rather close to submarine base of the edifice (Fig. 10.1.2).

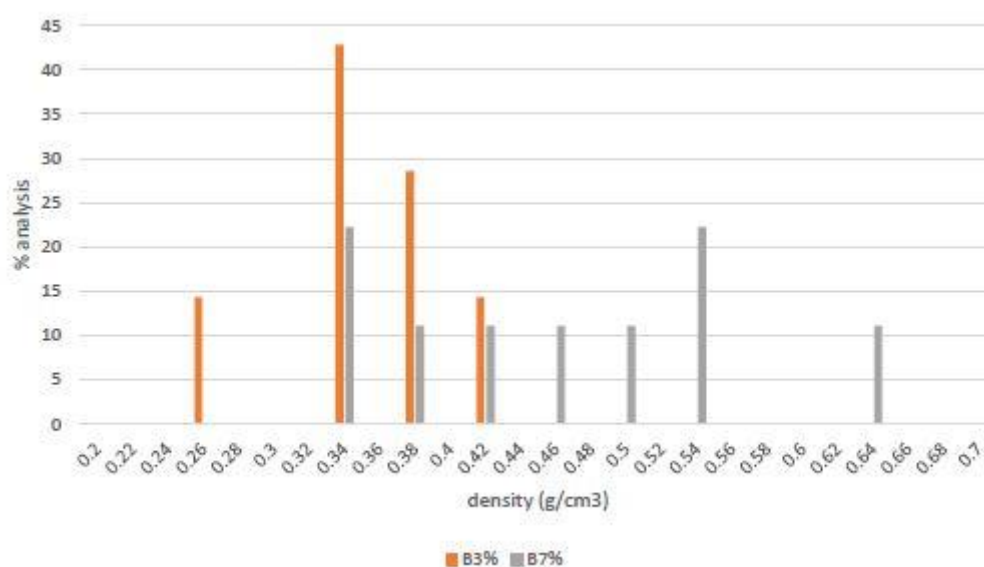


Fig. 10.1.1 - Density content of CO<sub>2</sub> on samples B3 (wherlite) and B7 (dunite).

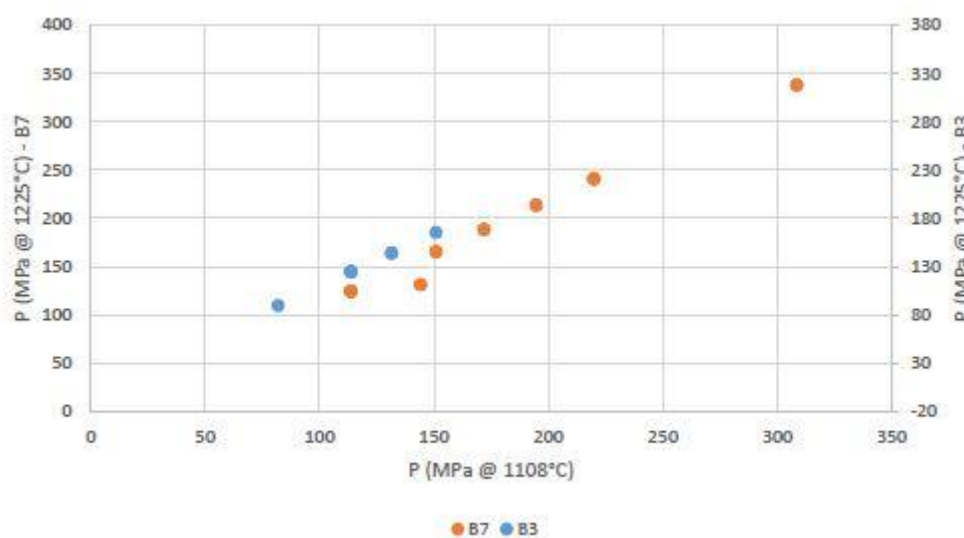


Fig. 10.1.2 - P vs. T°C diagram for samples B7 (dunite) and B3 (wherlite) of Bellecombe Ash Member.

Since dunite (B7) is the deepest sample analysed and wherlite (B3) is most superficial, it has been assuming that also other analysed

samples can be considered close to submarine base (or shallower depths) of the Piton de la Fournaise.

## 10.2 Noble gases and CO<sub>2</sub> analyses

Samples of the Bellecombe Ash Member show single mineral phase with both primary and secondary (along crystal fractures) fluid inclusions. Representative samples were chosen for this analyses, for both ultramafic (dunite and wherlite) and mafic (sub-ophitic gabbros, poikilitic gabbros and porphyrogabbros) samples. Single hand-picked mineral phase of olivine, clinopyroxene and feldspar were analysed for noble gases and CO<sub>2</sub> content.

In the investigated products, noble gases and CO<sub>2</sub> are expected to vary in function of magmatic degassing. Assuming that the various mineral phases containing a comparable amount of fluid inclusions and that the moles of gas mostly depend on the entrapment pressure, high content of gases (both noble gases and CO<sub>2</sub>) indicate a deep provenance of the samples. Noble gases and CO<sub>2</sub> solubility in silicatic melts could be described with the ratio  $N_2 \sim Ar < CO_2 < Ne < He$ . Therefore, high content of CO<sub>2</sub> and Ar could indicate samples formed at high pressure. However, further and more independent information arise from noble gases and CO<sub>2</sub> ratios (e.g. He/CO<sub>2</sub>, He/Ar\*) that are generally considered as tracer of magmatic degassing and depressurization noble gases and CO<sub>2</sub> content in the subvolcanic ejecta of the Bellecombe Ash Member are reported in from Fig. 10.2.1 to Fig. 10.2.9.

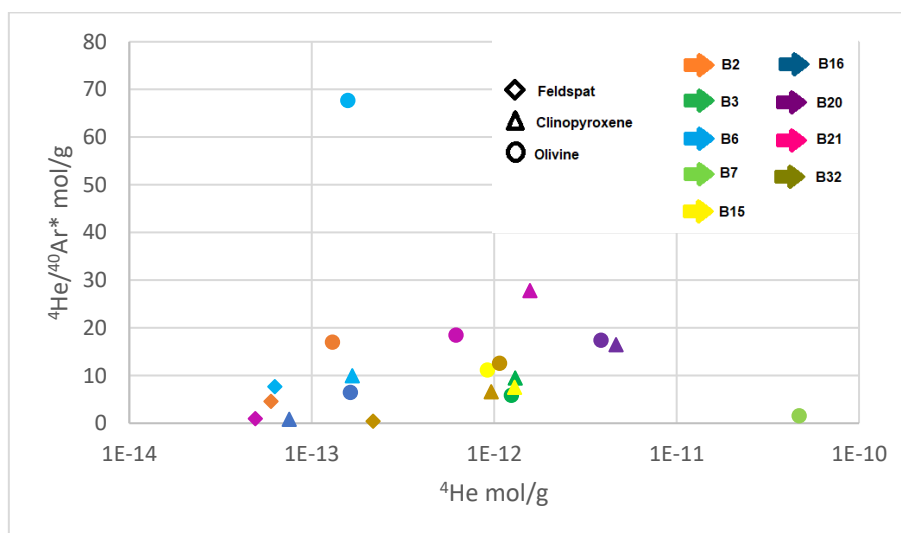


Fig. 10.2.1 - <sup>4</sup>He/<sup>40</sup>Ar\* vs. <sup>4</sup>He. B2: sub-ophitic gabbro; B3: wherlite; B6: sub-ophitic gabbro; B7: dunite; B15: wherlite; B16: porphyrogabbro; B20: porphyrogabbro; B21: poikilitic gabbro; B32: poikilitic gabbro.

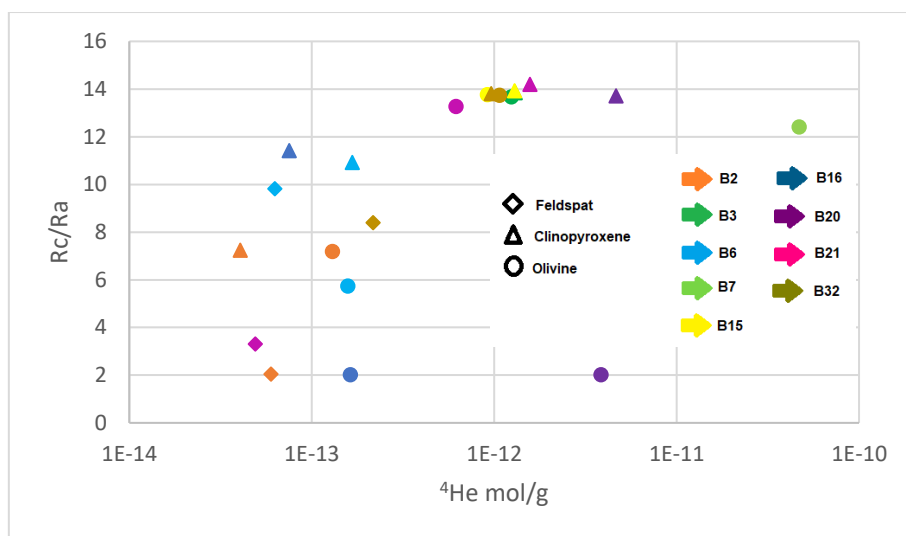


Fig. 10.2.2 -  $R_c/R_a$  vs  $^4\text{He}$ . B2: sub-ophitic gabbro; B3: wherlite; B6: sub-ophitic gabbro; B7: dunite; B15: wherlite; B16: porphyrogabbro; B20: porphyrogabbro; B21: poikilitic gabbro; B32: poikilitic gabbro.

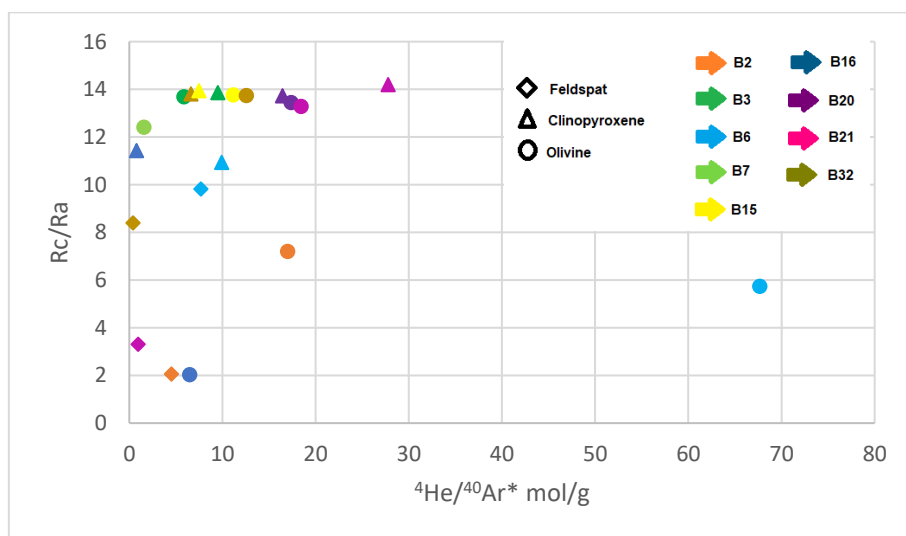


Fig. 10.2.3 -  $R_c/R_a$  vs  $^4\text{He}/^{40}\text{Ar}^*$ . B2: sub-ophitic gabbro; B3: wherlite; B6: sub-ophitic gabbro; B7: dunite; B15: wherlite; B16: porphyrogabbro; B20: porphyrogabbro; B21: poikilitic gabbro; B32: poikilitic gabbro.

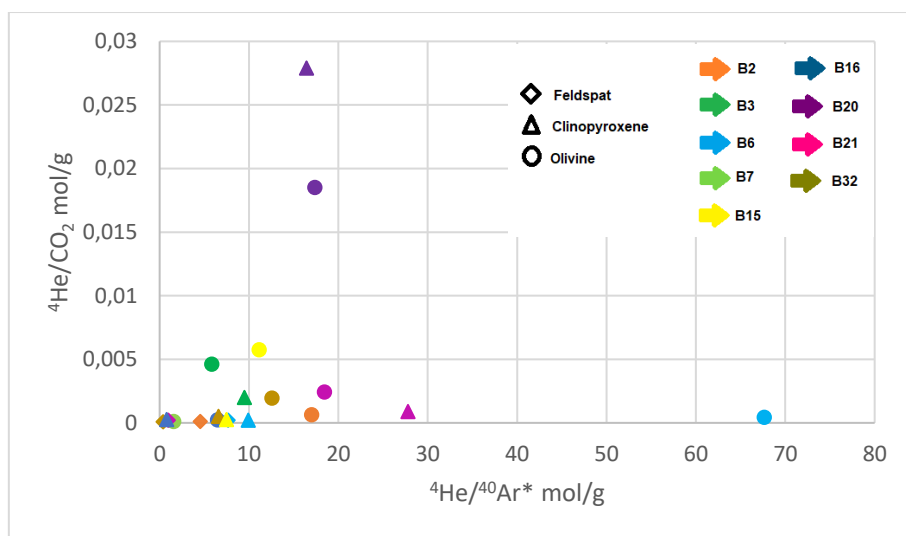


Fig. 10.2.4 -  $^4\text{He}/\text{CO}_2$  vs.  $^4\text{He}/^{40}\text{Ar}^*$ . B2: sub-ophitic gabbro; B3: wherlite; B6: sub-ophitic gabbro; B7: dunite; B15: wherlite; B16: porphyrogabbro; B20: porphyrogabbro; B21: poikilitic gabbro; B32: poikilitic gabbro.

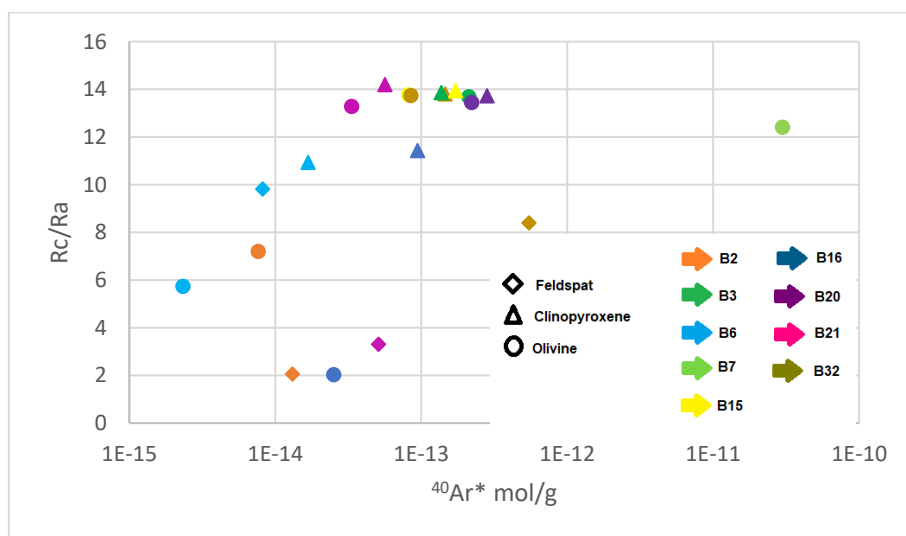


Fig. 10.2.5 -  $R_c/R_a$  vs.  $^{40}\text{Ar}^*$ . B2: sub-ophitic gabbro; B3: wherlite; B6: sub-ophitic gabbro; B7: dunite; B15: wherlite; B16: porphyrogabbro; B20: porphyrogabbro; B21: poikilitic gabbro; B32: poikilitic gabbro.

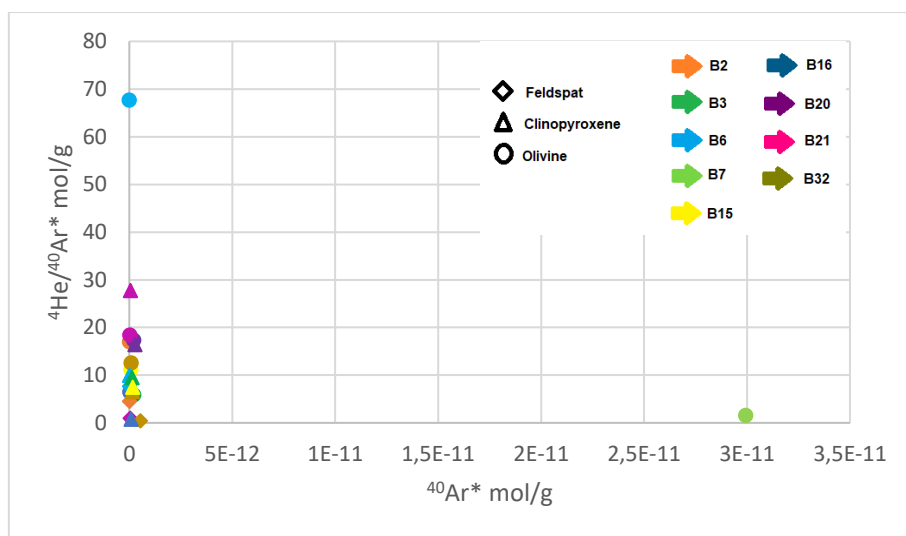


Fig. 10.2.6 -  $^4\text{He}/^{40}\text{Ar}^*$  vs.  $^{40}\text{Ar}^*$ . B2: sub-ophitic gabbro; B3: wherlite; B6: sub-ophitic gabbro; B7: dunite; B15: wherlite; B16: porphyrogabbro; B20: porphyrogabbro; B21: poikilitic gabbro; B32: poikilitic gabbro.

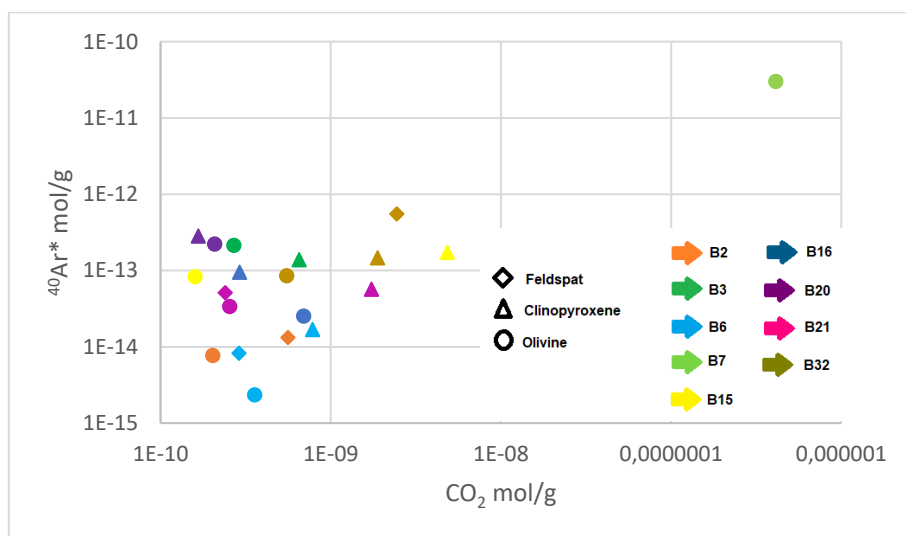


Fig. 10.2.7 -  $^{40}\text{Ar}^*$  vs.  $\text{CO}_2$ . B2: sub-ophitic gabbro; B3: wherlite; B6: sub-ophitic gabbro; B7: dunite; B15: wherlite; B16: porphyrogabbro; B20: porphyrogabbro; B21: poikilitic gabbro; B32: poikilitic gabbro.

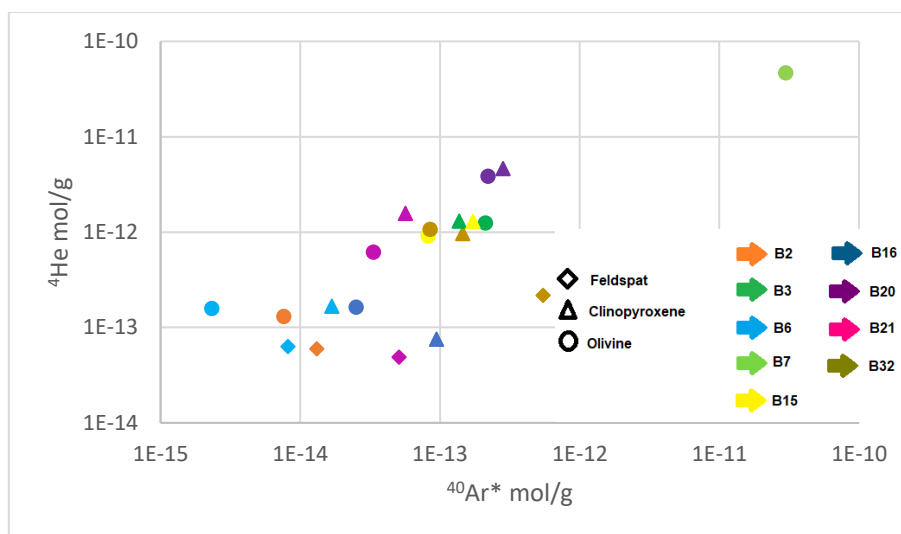


Fig. 10.2.8 -  $^4\text{He}$  vs.  $^{40}\text{Ar}^*$ . B2: sub-ophitic gabbro; B3: wherlite; B6: sub-ophitic gabbro; B7: dunite; B15: wherlite; B16: porphyrogabbro; B20: porphyrogabbro; B21: poikilitic gabbro; B32: poikilitic gabbro.

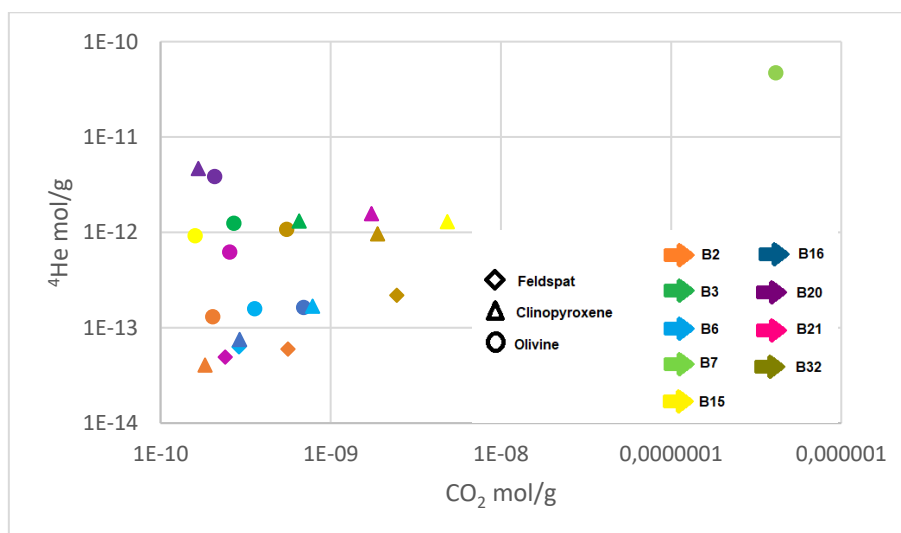


Fig. 10.2.9 -  $^4\text{He}$  vs.  $\text{CO}_2$ . B2: sub-ophitic gabbro; B3: wherlite; B6: sub-ophitic gabbro; B7: dunite; B15: wherlite; B16: porphyrogabbro; B20: porphyrogabbro; B21: poikilitic gabbro; B32: poikilitic gabbro.

## 11 Quartz xenocrysts

Among the Bellecombe Ash Member deposits, loose quartz xenocrysts coming from sites near Langevin Caldera were sampled, from few millimetres to about one centimeter in size. The presence of these quartz xenocrysts might derive from the involvement of the shallow hydrothermal system or circulating hydrothermal fluids when the eruption of the Bellecombe Ash Member occurred. The total number of the quartz xenocrysts is 52. These samples were investigated with a stereomicroscope and were subdivided on the basis of the presence of fluid inclusions. Subsequently, representative samples were prepared for different analyses, in order to try to understand their origin.

### 11.1 X-ray diffraction analyses

The samples analysed by X-ray diffractometric analyses (Q3, Q4, Q5, Q8, Q9; Fig. 11.1.1) were chosen among not containing fluid inclusions (at naked eye). For all the analyses was used the "side loaded" technique, with the exception of smaller samples (Q8 and Q9) where "scratch" technique was used.



Fig. 11.1.1 - Some of the quartz crystals sampled on Langevin caldera.

All the analysed samples resulted in quartz xenocrysts (Figg. from 11.1.3 to 11.1.8), with the exception of a calcium carbonate crystal (Q3; Fig. 11.1.2).

Details on the diffractometric spectra were reported below.

Q3: concretionary aspect with evident lateral depositional bands. The sample, as evidenced by its spectrums, is a calcium carbonate concretion. Only two other samples found in the Langevin caldera were identified as calcium carbonate concretion.

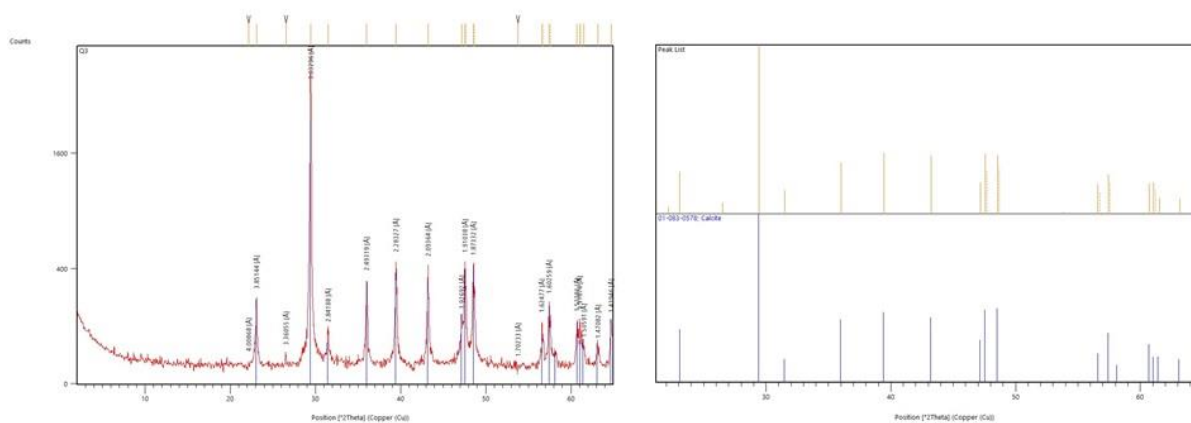


Fig. 11.1.2 - Q3 XRD analyses. Spectrum of analyses and pattern.

Q4: twinned sample.

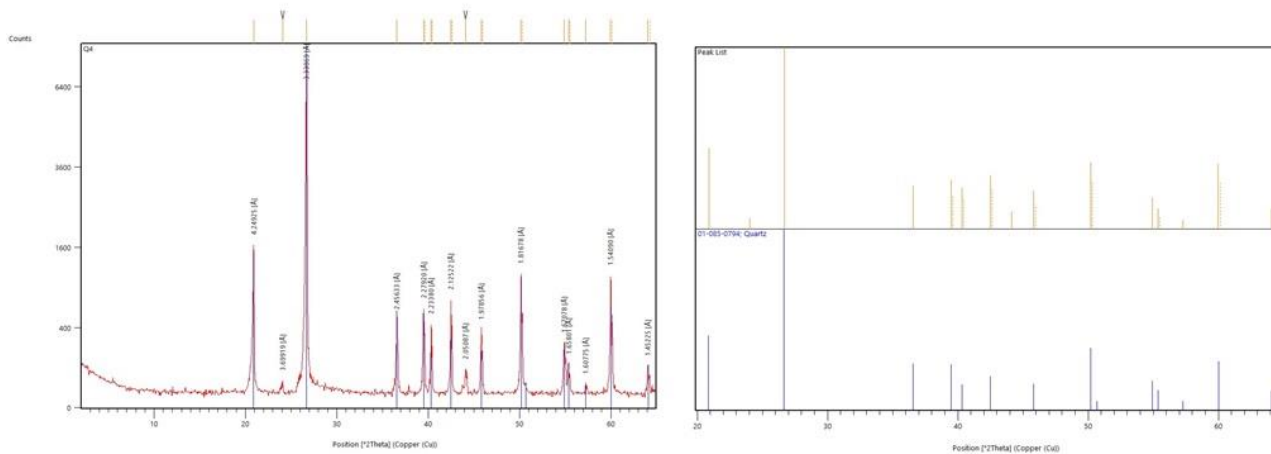


Fig. 11.1.3 - Q4 XRD analyses. Spectrum of analyses and pattern.

Q5: very transparent sample.

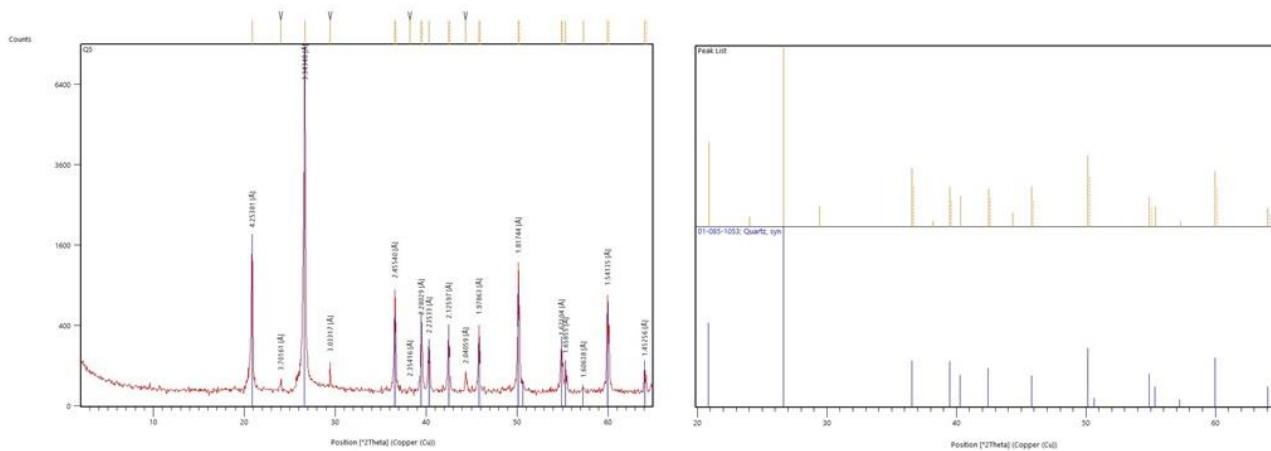


Fig. 11.1.4 - Q5 XRD analyses. Spectrum of analyses and pattern.



Q8: The analyses was also replicated with the "suspension in acetone" technique.

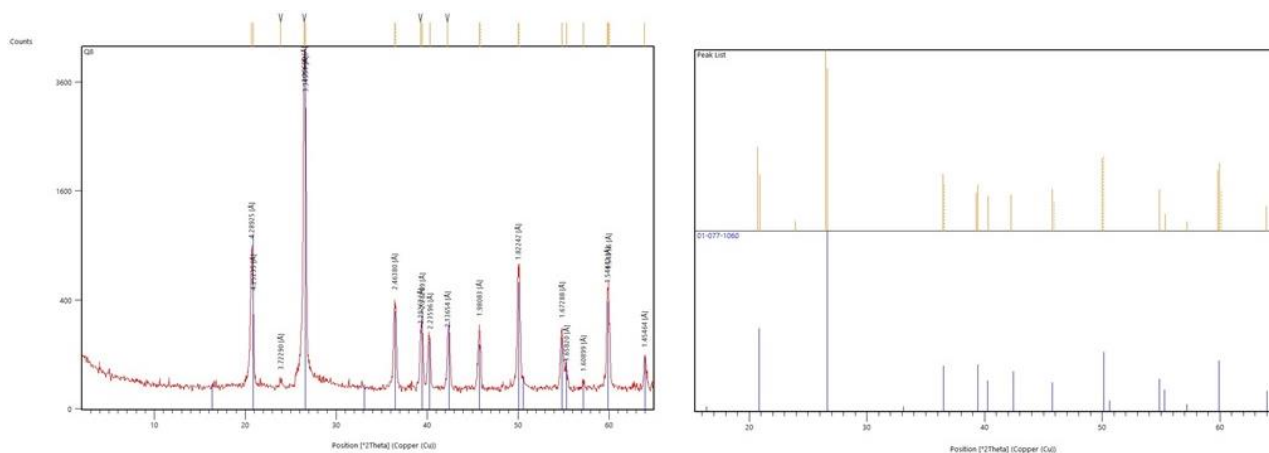


Fig. 11.1.5 - Q8 XRD analyses. Spectrum of analyses and pattern.

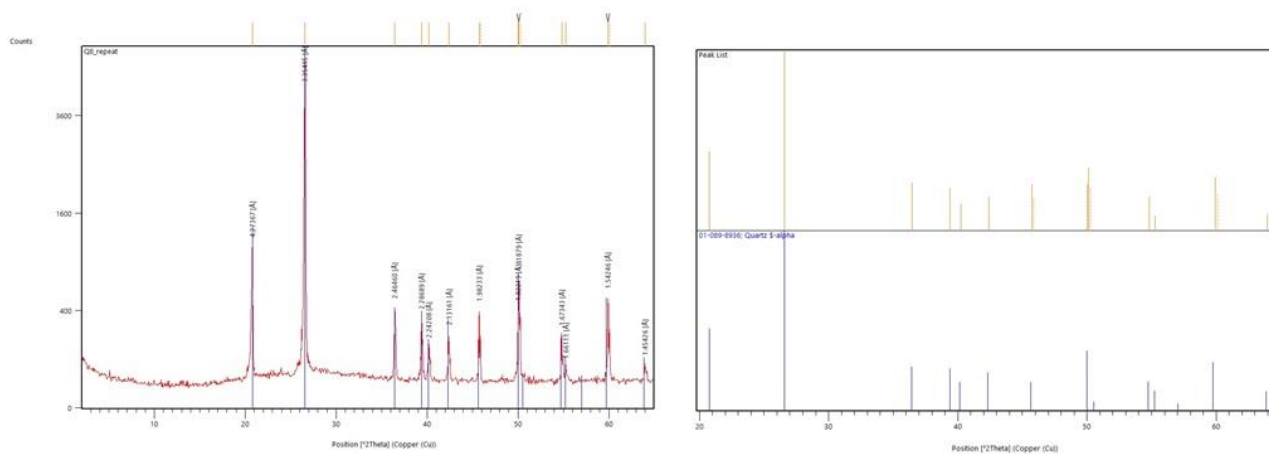


Fig. 11.1.6 - Q8 repetition of the XRD analyses. Spectrum of analyses and pattern.

Q9: The result of the analyses was good but also for this sample the analyses were replicated twice with the "suspension in acetone" technique to try to improve the result.

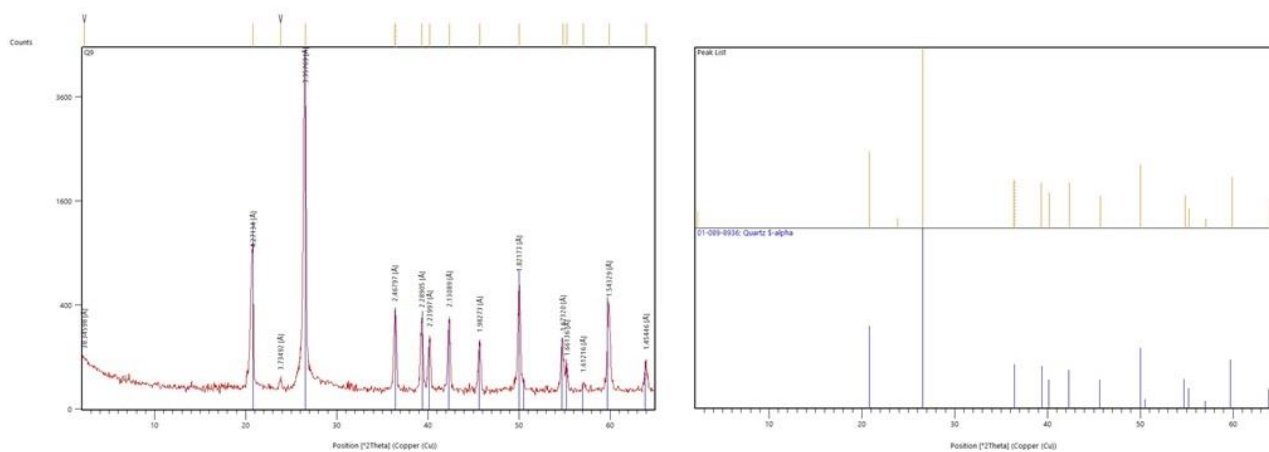


Fig. 11.1.7 - Q9 XRD analyses. Spectrum of analyses and pattern.

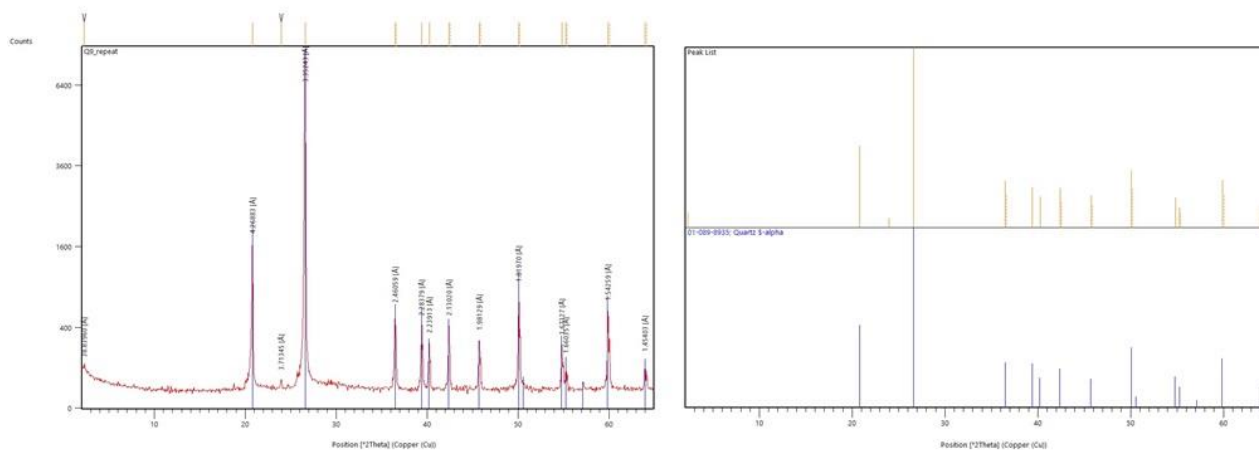


Fig. 11.1.8 - Q9 replication of the XRD analyses. Spectrum of analyses and pattern.

## 11.2 Laser Ablation Inductively Coupled Plasma Mass Spectrometry analyses

Some quartz samples were studied on thin section and display different accretions/layers on the crystal rim (Figg. from 11.2.1 to 11.2.4).

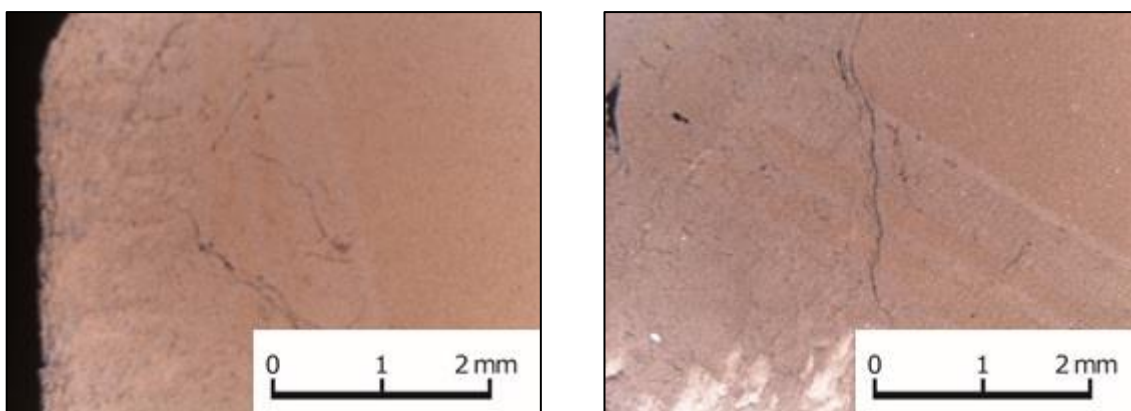


Fig. 11.2.1 - Plane-polarized light of quartz xenocryst Q1.

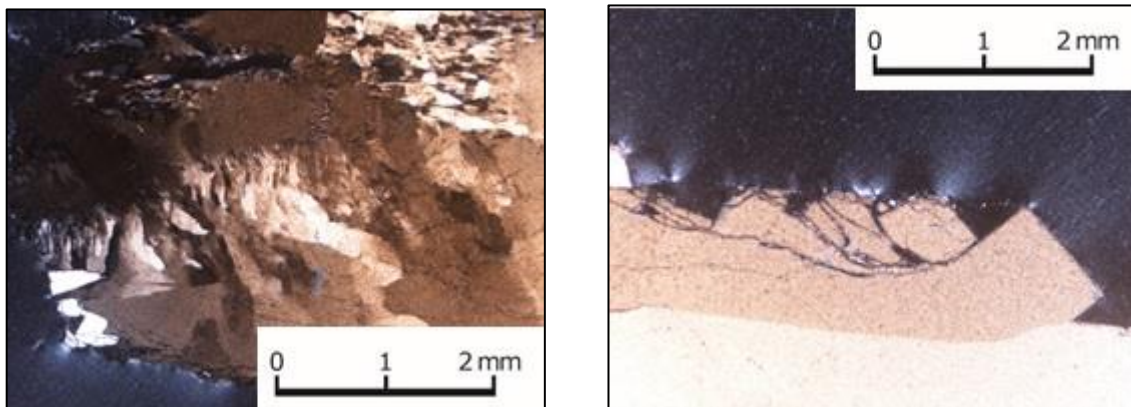


Fig. 11.2.2 - Crossed polars of quartz xenocryst Q13.

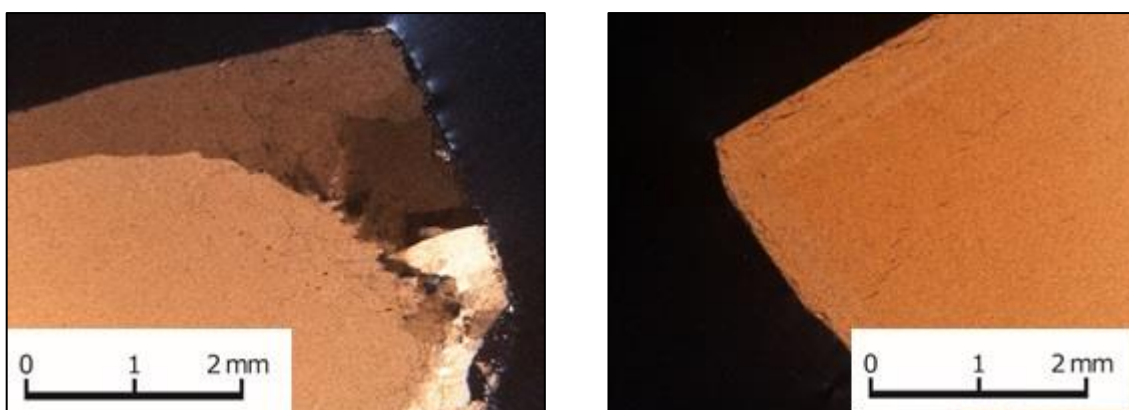


Fig. 11.2.3 - Crossed polars of quartz xenocryst Q16.

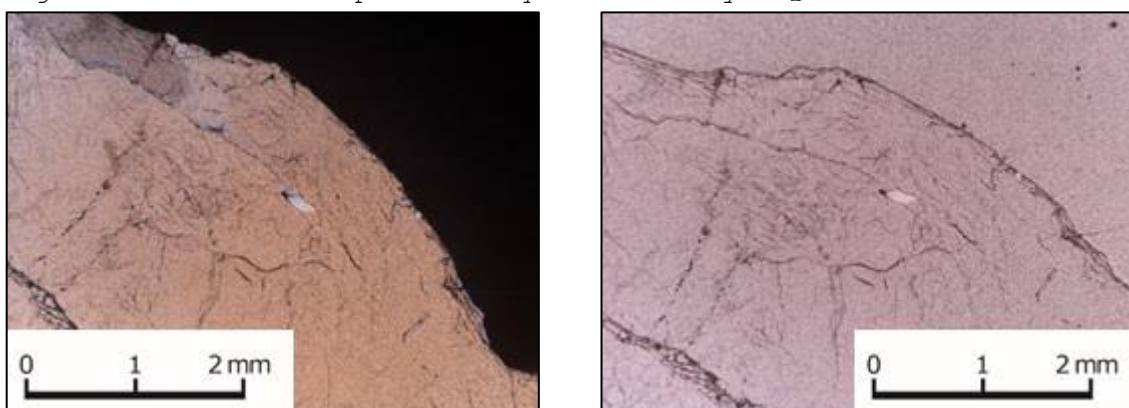


Fig. 11.2.4 - (On the left) Crossed polars and (on the right) plane-polarized light of quartz xenocryst Q19.

To better understand the nature of these rims, some quartz samples without trace of fluid inclusions were analysed with a transect rim-core and analysed through LA-ICP-MS (Figg. 11.2.5, 11.2.6, 11.2.7 and 11.2.8). The analyses were made on samples Q1, Q13, Q16 and Q19.

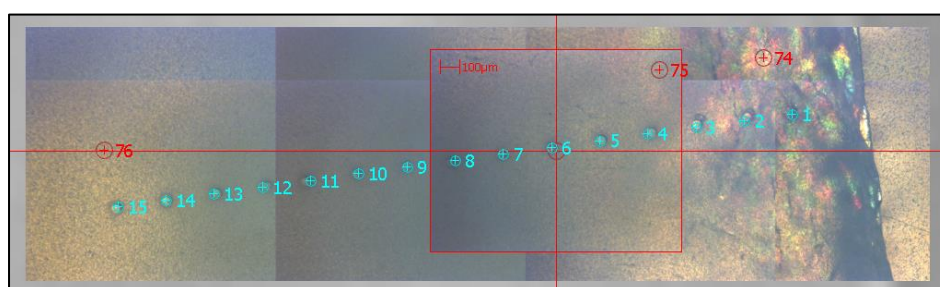


Fig. 11.2.5 - Q1 transect rim-core. The image was take with the LA-ICP-MS integrated camera. The laser spots of the ablation analyses are shown.

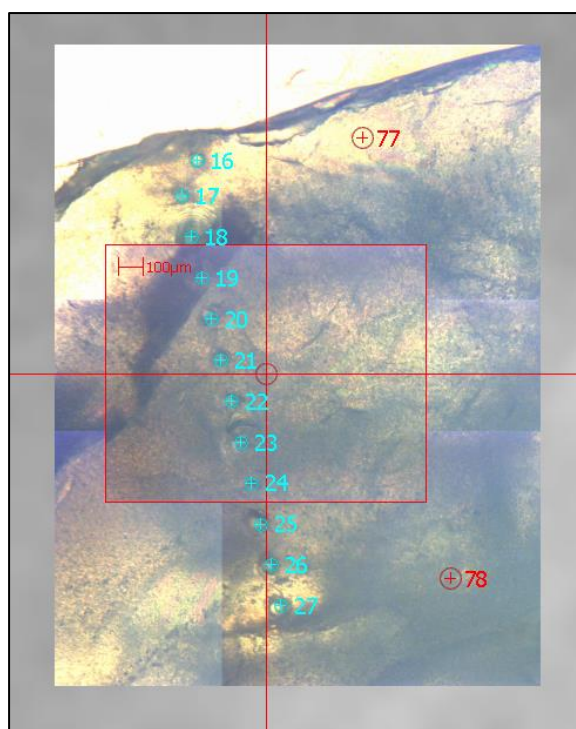


Fig. 11.2.6 - Q13 transect rim-core. The image was taken with the LA-ICP-MS integrated camera. The laser spots of the ablation are shown.

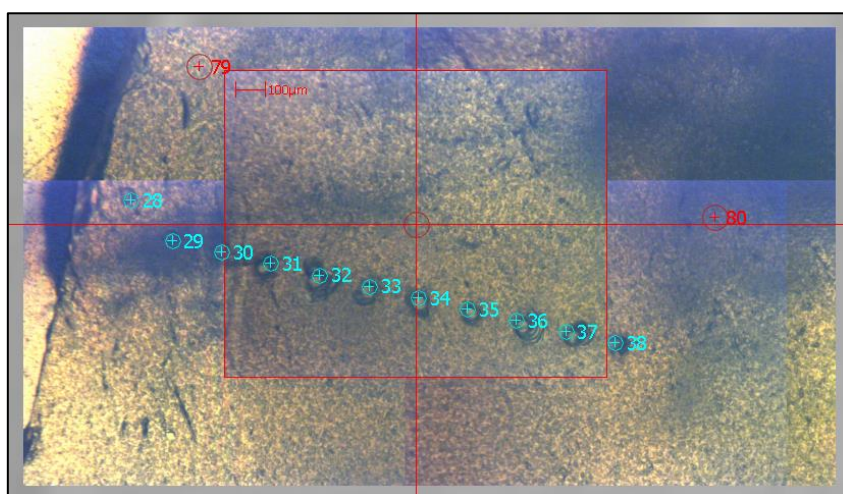


Fig. 11.2.7 - Q16 transect rim-core. The image was taken with the LA-ICP-MS integrated camera. The laser spots of the ablation are shown.

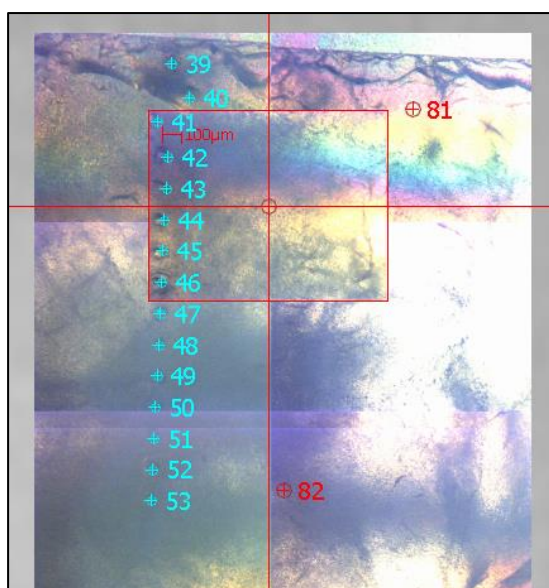


Fig. 11.2.8 - Q19 transect rim-core. The image was take with the LA-ICP-MS integrated camera. The laser spots of the ablation are shown.

Some enrichments in trace elements in quartz xenocrysts are shown from the core to the rim of the crystal (Figg.11.2.9 and 11.2.10).



Fig. 11.2.9 - Transect of quartz xenocrysts samples from the core (right) to the rim (left). Samples shows different content of trace elements through the core-rim transect. Some of them show an enrichment from the core to the rim.

Common trace elements in quartz can be very useful proxy of interaction between the hydrothermal system and the magmatic plumbing system during phreatomagmatic eruptions (e.g. Ti was found to be universally present in magmatic quartz, but at much lower abundance in hydrothermal quartz; Muller et al., 2003). As a matter of fact, in the investigated quartz xenocrysts Ti is very low, around 1 ppm (Limit Of Detection LOD are: Li 0.16-0.22 ppm; B 0.55-0.77

ppm; Al 0.37-0.46 ppm; P 4.4-6.7 ppm; Ti 0.45-0.57 ppm; K 1-1.5 ppm; Fe 2.9-4.3 ppm; Sr 0.0016-0.0041 ppm).

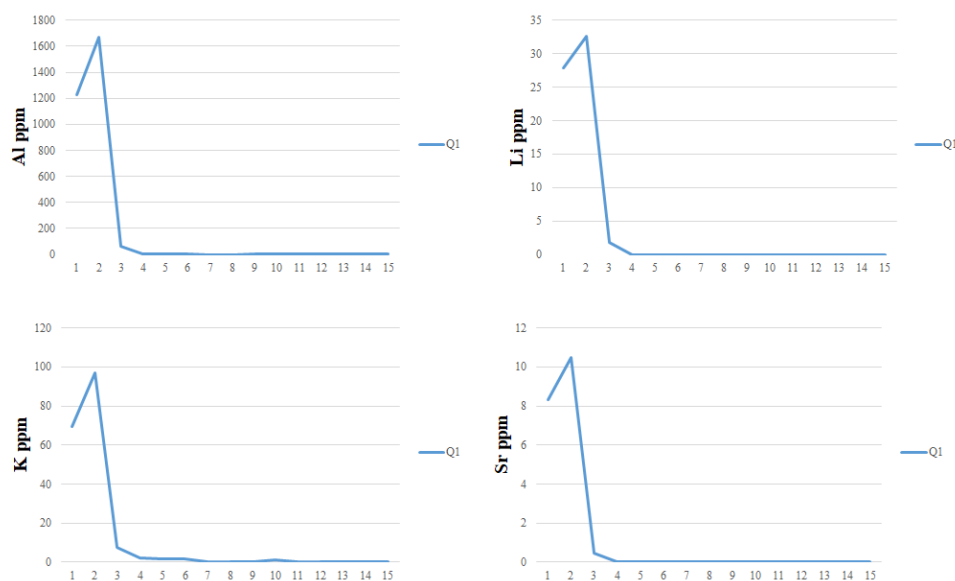


Fig. 11.2.10 - Transect of Q1 quartz sample from the core (right) to the rim (left). Q1 sample show the maximum concentration on the rim respect the other samples analyzed.

The enrichment of different elements is clearly visible in Q1 sample especially for Al, Li, K and Sr. This enrichment on the rim of  $\text{Al}^{3+}$  and  $\text{Li}^+$  can be interpreted as a substitution of  $\text{Si}^{4+}$  in the quartz structural lattice.

Sample Q1 is that better emphasizes this evidence of enrichments of different elements at the rims and therefore was mapped in detail (Fig. 11.2.11).

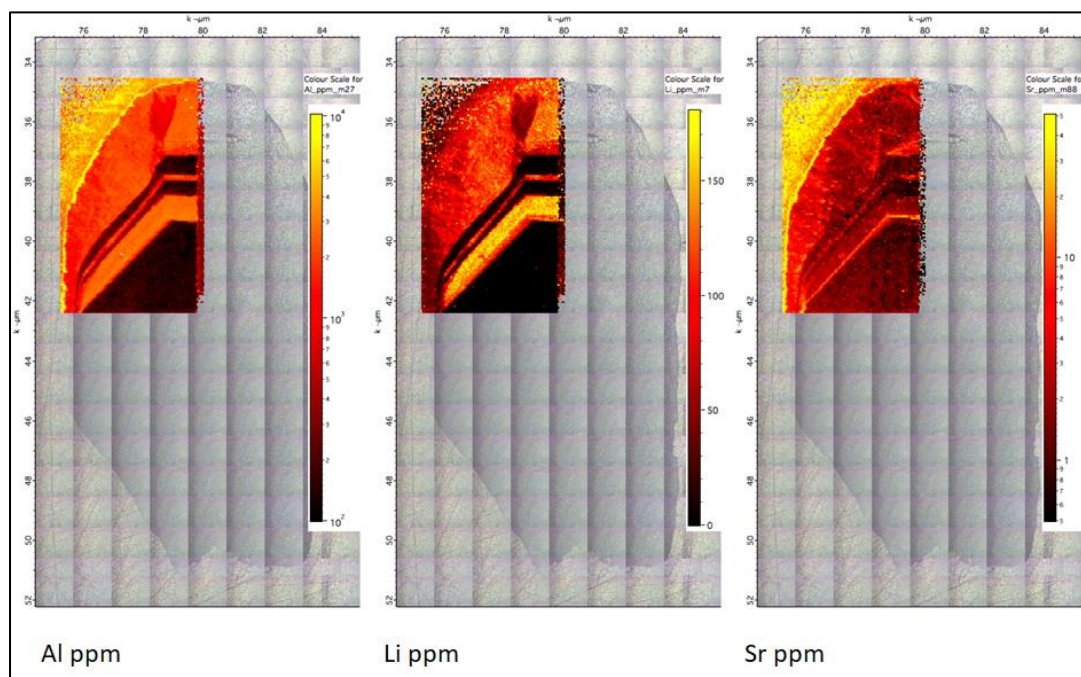


Fig. 11.2.11 -LA-ICP-MS map showing the enrichment in Al, Li and Sr toward the rim of the Q1 xenocryst.

### 11.3 Raman analyses

The analysed samples were both doubled-polished in thin sections (Q1, Q13, Q16 and Q19) or bulk crystals (Q14, Q15, Q17 and Q18). The fluid inclusions occur in different generations and are both biphase (liquid + vapour; Fig. 11.3.1) and monophasic (vapour; Fig. 11.3.2). Raman analyses reveal the dominant presence of low-salinity aqueous inclusions (Fig. 11.3.3) with, locally, trace of CO<sub>2</sub> (Fig. 11.3.4). Characterization of true salinity, kind of dissolved salts and density of the inclusions, such as of immiscibility/boiling processes, would require a more detailed fluid inclusion study.

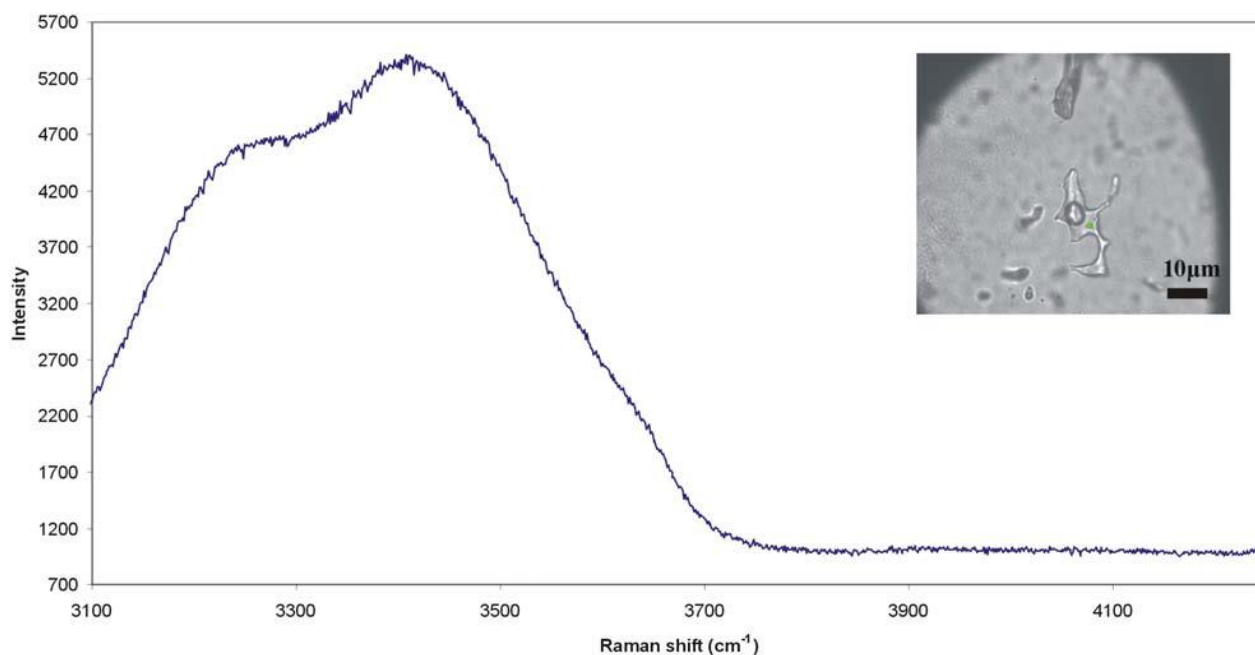


Fig. 11.3.1 - Typical Raman spectrum of the liquid water present in a biphasic (liquid + vapour) fluid inclusion. The evident band at about 3200 cm<sup>-1</sup> is typical of low-salinity water (e.g., Frezzotti et al., 2012). The green dot in the photomicrograph represents the analysed site.

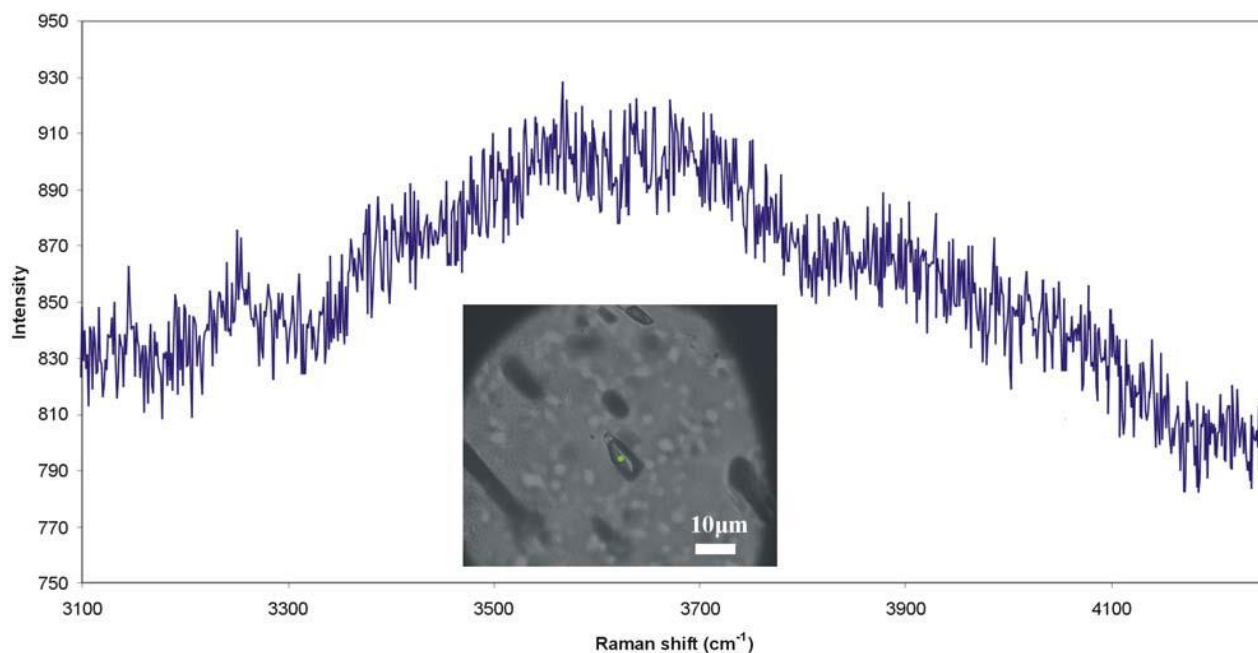


Fig. 11.3.2 - Raman spectrum of the vapour water present in a biphasic (liquid + vapour) fluid inclusion. The two large bands at about 3550 and 3650 cm<sup>-1</sup> are typical of low-salinity water (e.g., Frezzotti et al., 2012). The green dot in the photomicrograph represents the analysed site.

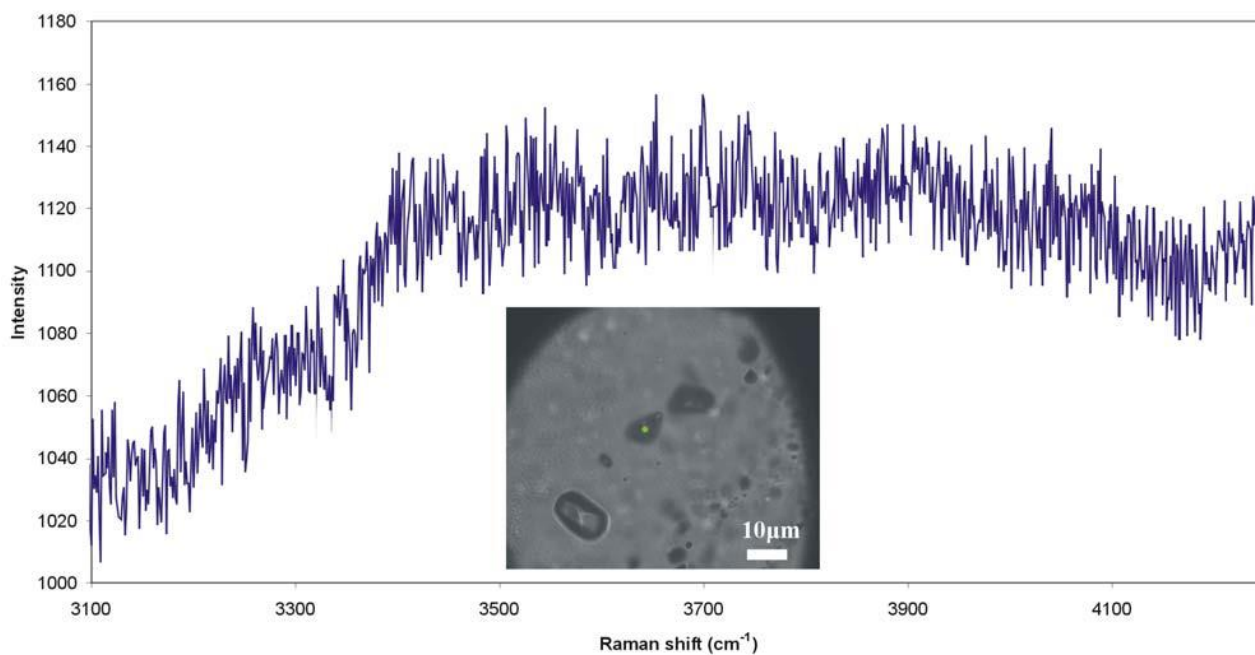


Fig. 11.3.3 - Raman spectrum of the vapour water present in a monophasic (vapour) fluid inclusion also containing trace of CO<sub>2</sub> (Fig. 12.3.4). The green dot in the photomicrograph represents the analysed site.



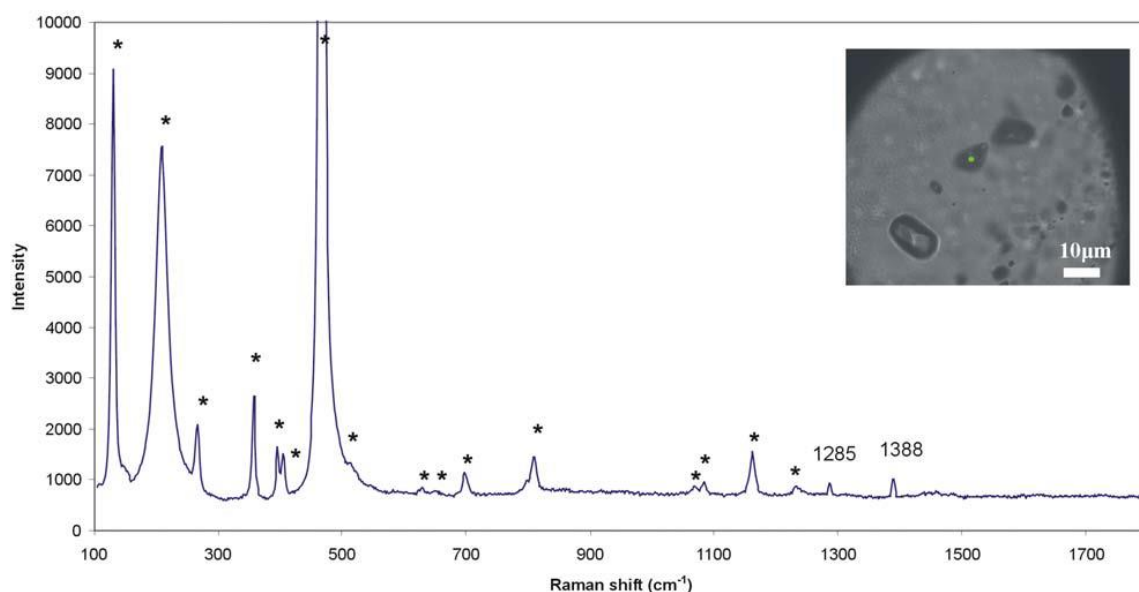


Fig. 11.3.4 - Raman spectrum of  $\text{CO}_2$  in a monophasic vapour inclusion of water (Fig. 12.3.3). The two peaks at  $1285$  and  $1388 \text{ cm}^{-1}$  and their low intensity are typical of low-density vapour  $\text{CO}_2$  (e.g., Frezzotti et al., 2012). The green dot in the photomicrograph represents the analysed site.

## 12 Sedimentary clasts of Langevin Caldera

Some sub-centimeter to centimeter size carbonatic xenoliths (calcite, see XRD analyses in Fig. 11.1.2) from Langevin Caldera were detected. One of these carbonatic samples ( $\approx 4 \text{ cm}$  in size) was used for a thin section investigation. The sample was identified as a sedimentary rock, probably a carbonatic biomicrite. Radiolaria, microforaminifera and broken sponge spicols were recognized, also suggesting an apparent reworking of the deposits (Fig. 12.1). Microforaminifera detected are mainly genera *Hedbergella* (Barremian-Albian L., 110-120 Ma) and *Hedbergella similis*. Some crystals of ematite and glauconite, typically sedimentary minerals, also occurred. Since the oldest sediments of the oceanic crust below the Réunion Island should be not older than 70-80 Ma, the hypothesis of a reworked older sedimentary rocks (from the adjacent Mascarene basin) could be the most reliable. The sediments probably were also subjected to some directional current, as suggest by the orientation of sponge spicols (Fig. 12.2).

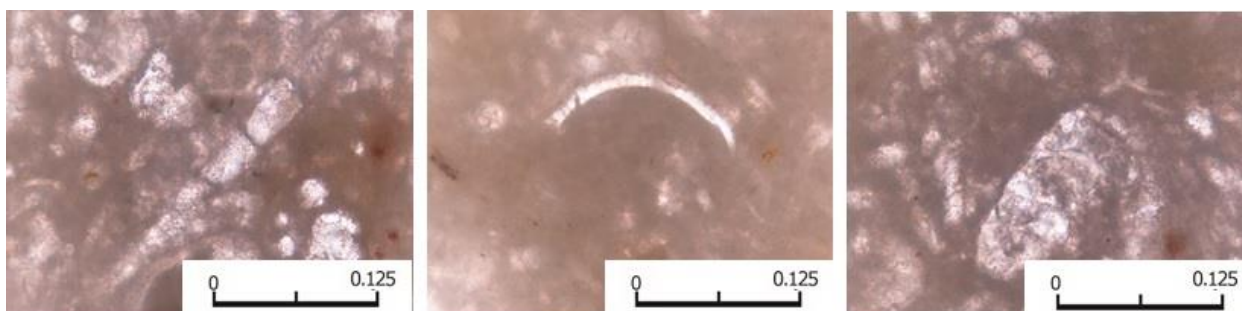
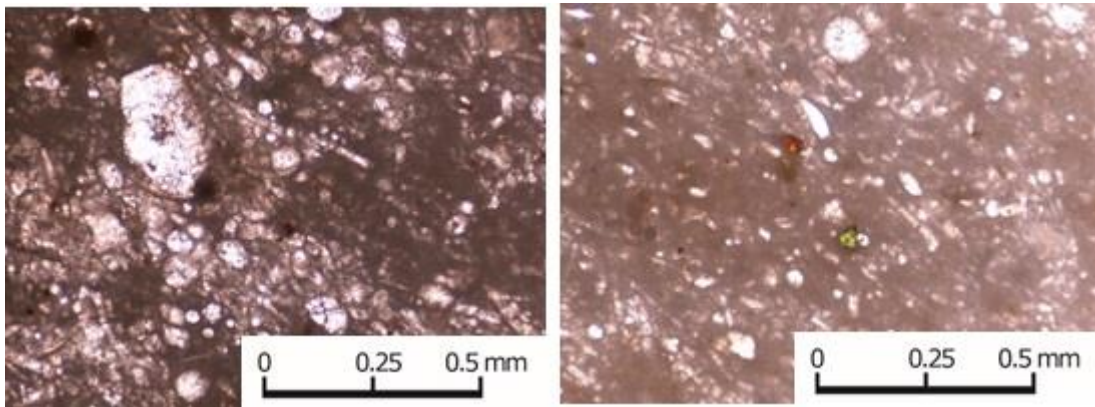


Fig. 12.1 - Plane-polarized light of sample B30. From the left: microforaminifer, a section of valve and a particular of microforaminifer (*Hedbergella*).



*Fig. 12.2 - Plane-polarized light of sample B30. Glauconite and hematite were individuate.*

## 13 Discussion

### 13.1 The subvolcanic crystallization below the Réunion Island: comparison between Piton de la Fournaise and Piton des Neiges mafic and ultramafic intrusive products

The detailed petrographic study by optical microscopy, together with phase and bulk rock chemistry, allowed the recognition of several petrographic groups among the mafic and ultramafic subvolcanic rocks of the Bellecombe Ash Member, unravelling an overall, much wider range of textures and igneous subvolcanic lithotypes than previously reported by the literature.

Comparing textures of Piton de la Fournaise subvolcanic ejecta with those of Piton des Neiges, a longer subsolidus strain history is pointed out for the latter, being an older and eroded magmatic plumbing system. Some differences in modal mineralogy and texture were also recognized. Some dunitic clasts of Piton des Neiges show the presence of amphibole, absent in other samples of Piton de la Fournaise. Wherlitic ejecta of the Bellecombe Ash Member are mainly made of olivine, clinopyroxene, interstitial glass and opaque minerals. Plagioclase are however detected in small amount in one wherlite sample (B28) and in a composite layered ejecta (B108) emphasizing the role of plagioclase crystallization in the transition wherlite-gabbro (through the presence of plagioclase-wherlite and troctolite patches) also at isobaric condition in the Piton de la Fournaise magmatic plumbing system.

The interstitial trapped liquid (as interstitial quenched glass) in the intrusive investigated ejecta range from basalt to basaltic trachyandesite.

Piton des Neiges dunitic samples has a homogeneous Fo% content close to 88, comparable with that found in the dunitic ejecta of the Bellecombe Ash Member which are however characterized by a wider compositional range from Fo<sub>89</sub> down to Fo<sub>80</sub>. Fo% content of Piton des Neiges wherlites is 77-85, whereas Bellecombe wherlitic ejecta are characterized by Fo<sub>80-85</sub>. Wherlites of Piton des Neiges and Piton de la Fournaise are comparable in terms of Fs% content in clinopyroxene, respectively Fs<sub>7-9</sub> and Fs<sub>8</sub>. Also the ferrosilite contents in clinopyroxene of the gabbro samples are comparable in Piton des Neiges and ejecta of the Piton de la Fournaise being respectively FS<sub>10-15</sub> and FS<sub>8-15</sub>.

Most of the mafic and ultramafic subvolcanic ejecta of the Bellecombe Ash Member can be interpreted as cumulate cognates *sensu stricto*: dunites, wherlites, ophitic gabbros, poikilitic gabbros and some sub-ophitic gabbros (those with the higher CaO/Al<sub>2</sub>O<sub>3</sub> ratios). By contrast, other subvolcanic clasts (doleritic gabbros and the sub-ophitic gabbros with the lower CaO/Al<sub>2</sub>O<sub>3</sub> ratios) may simply represent slowly-cooled equivalents (with no crystal-liquid fractionation) of the erupted cotectic basalts (SSB). Finally, the porphyrogabbros may

represent oceanite (picritic) magmas, whose storing in the magmatic plumbing system allowed the fine-grained crystallization of an evolved groundmass represented by sanidine- and biotite-bearing (phlogopite) assemblage, which are instead absent in Piton des Neiges and other Piton de la Fournaise intrusive clasts analysed in previous works (Upton et al., 2000; Berthod et al., 2020; Brugier, thesis, 2016). The presence in the Bellecombe intrusive ejecta of such a hydrate interstitial phase could also indicate a vapor phase crystallization that can be either pneumatolytic within the magmatic plumbing system or linked to hydrothermal fluids.

### **13.2 Subvolcanic crystallization and relation between interstitial glass of Bellecombe Ash Member ejecta and trachitic pumices**

The cumulitic mafic and ultramafic ejecta of the Bellecombe Ash Member can be interpreted as formed by solid-liquid crystal settling or *in situ* crystallization. By contrast, the subvolcanic samples reflecting the slowly-cooled equivalent of the basaltic melt (i.e. doleritic gabbros and some sub-ophitic gabbros) should have formed as an *in situ* crystallization front from the basaltic magma wall-rock inward without solid-liquid separation.

During the process of *in situ* crystallization, some residual liquids may have been interstitially entrapped and finally quenched as glass, so widespread in some of the investigated ejecta. The absence of resorbed margin on the crystal framework (clinopyroxene, feldspar and olivine) in contact with the interstitial glass, suggest a fast ascent of the magma carrying the subvolcanic clasts. The composition ranging from basalt to basaltic trachyandesite of the interstitial glass indicates that the melt was entrapped within the crystal frameworks, with no time to evolve and crystallize because involved in the explosive eruption. The quenched interstitial glass as well as the texture of the mafic and ultramafic ejecta represent a snapshot of the pre-eruptive plumbing system of Piton de la Fournaise. This suggest the clasts were disrupted from the wall rocks and transported very fast to the surface, mostly as cognates. Nevertheless, during the history of a magma plumbing system, liquid entrapped interstitially could also escape from the subvolcanic crystal frameworks and could also form small melt pockets (or larger body) of more evolved silicate magma. This is what happened in the mugearitic portion of a sill of Piton des Neiges studied by Upton et al. (1967).

The trachytic pumices of the Bellecombe Ash Member show a more evolved glass composition than the analysed interstitial glass in the subvolcanic ejecta. Therefore, these pumices likely represent a differentiated liquid, which can be considered akin with that quenched as interstitial glass in the intrusive suite found as ejecta in the Bellecombe Ash Member.

Porphyrogabbro samples show a modal mineralogy characterized by the presence of biotite (phlogopite) and sanidine. This mineral phase

association (also coupled with low-Ca plagioclase) is characteristic of a trachytic composition. Interstitial phlogopite and sanidine in porphyrogabbros can be thus related to the crystallization of the residual trachytic liquids otherwise erupted as quenched pumice glass. Trachytic pumices of the Piton de Neiges are similar to those found in the Bellecombe Ash Member but characterized by the feldspar pair albitic plagioclase-anorthoclase in the glassy groundmass. Different equilibrium of feldspar pairs suggest different temperature of crystallization.

Vlastelic et al. (2021) consider the small volume of trachyte pumices erupted by Piton de la Fournaise in 2007 as a liquid remnant of Piton des Neiges or Les Alizés or a result from partial melting of an old syenite intrusion or remobilization of interstitial melts not fully solidified. Nevertheless, also the sanidine- and phlogopite-bearing microcrystalline groundmass in the porphyrogabbro ejecta seem to be representative of a trachyte liquid. In this way, as the porphyrogabbro ejecta were erupted during the Bellecombe Ash Member (about 5 ka), small fractions of evolved liquids in the Piton de la Fournaise plumbing system should have been not uncommon during the Holocene. The presence of trachyte pumices in the 2007 eruption of Piton de la Fournaise should therefore not be considered an exceptional event.

### **13.3 Thermobarometric evaluation and fluid inclusions of the subvolcanic ejecta in the Bellecombe Ash Member and some isobaric coexistence of ultramafic-mafic lithotypes**

The presence in some Bellecombe Ash Member ejecta of fluid inclusions allowed a thermobarometric evaluation, to better understand the dynamic of the subvolcanic plumbing system of Piton de la Fournaise.

Both mafic and ultramafic analysed ejecta show different content in noble gases and CO<sub>2</sub>. Ultramafic samples gave significant results for CO<sub>2</sub> content on Raman analyses, suggesting the deepest levels of crystallization. Dunite (B7) show a range of crystallization pressure of about 309–338 MPa, suggesting an underplating provenance (~ 13 km b.s.l.). By contrast, the wherlite (B15) pressure of crystallization ranges from 151 to 166 MPa, correspondent to an origin close to the submarine base of the edifice (~ 6 km b.s.l.). Fluid inclusion in mafic ejecta of Bellecombe Ash Member show a CO<sub>2</sub> content below the limit of detection, therefore indicating a possible even shallower depth (above 6 km b.s.l.).

High content of noble gases and CO<sub>2</sub> measured in fluid inclusions coupled to low He/Ar\* and He/CO<sub>2</sub> ratios indicate the deepest provenance of the samples (Boudoire et al., 2018). Noble gases and CO<sub>2</sub> high content together with He/Ar\* and He/CO<sub>2</sub> on the dunite sample (B7) agree with a deep provenance, close to the underplating, while the provenance of the wherlitic sample B15 is just shallower than the dunite one.

Noble gases and CO<sub>2</sub> from fluid inclusions of mafic intrusive ejecta (poikilitic gabbro and sub-ophitic gabbro), also suggest a shallower depth of crystallization than the wherlites, i.e. close to the submarine base of the edifice. Nevertheless, the presence of composite subvolcanic samples (B108a and B108b), displaying the coexistence of plagioclase-wherlite, troctolite and gabbro patchy portions (with modal and grain size variations) surrounding wherlite, could be explained by dissolution process due to percolation of basaltic melt. Also in the B28 wherlite sample, the dissolution of olivine and its progressive rounding, can involve basaltic melt percolation coupled by reaction, thus producing clinopyroxene. In a strongly active basaltic volcano such as Piton de la Fournaise, the transition from ultramafic to mafic intrusive domains (B108) or textural changes in the ultramafic lithotypes, could be therefore driven by the interaction of the persistently rising batches of basaltic magmas interacting with the already crystallized cumulates.

Fluid inclusion and thermobarometric studies should be deepened, together with a detailed petrologic investigation of the composite ejecta (B108a-b), aimed at a more exhaustive and consistent information about the petrogenetic history of the mafic and ultramafic samples of the Bellecombe Ash Member.

#### **13.4 The involvement of the hydrothermal system in the Piton de la Fournaise explosive eruptions**

As already documented by Ort et al. (2016), in the Bellecombe Ash Member deposits, quartz crystals and hydrothermally altered grains were found. This could indicate the involvement of a shallow hydrothermal system, from 0.5 to 2 km depth, in the magma plumbing system of this explosive eruption of the Piton de la Fournaise.

In the deposit at the Langevin Caldera, several loose quartz xenocrysts were found and collected. The analyses of these quartz xenocrysts pointed out the presence of a low-salinity aqueous inclusions (Raman analyses) coupled with trace of CO<sub>2</sub>. A hydrothermal origin for these quartz xenocrysts is very likely. The enrichment, on the rim of some quartz crystals, of some trace element was also considered a very useful proxy of interaction between the hydrothermal system and the magmatic plumbing system during phreatomagmatic eruptions. Some quartz xenocrysts show a Li and Al enrichment on the rim. These enrichments can be interpreted as a substitution of Si in the quartz lattice by circulating fluids. The interaction between shallow hydrothermal and magmatic plumbing systems during the Bellecombe Ash Member eruption is thus strongly suggested. During the interaction, the eruption could have disrupted and transported to the surface not only the subvolcanic mafic and ultramafic suite but also quartz xenocrysts of hydrothermal origin.

## 14 Conclusions

The textural and petrologic studies of the subvolcanic ejecta of Piton de la Fournaise provides new insights on the magmatic plumbing system and the subvolcanic crystallization processes at La Réunion Island. With respect to the available literature data, a wider spectrum of subvolcanic lithotypes, mainly in terms of a great variety of textures, was emphasized. Several mafic and ultramafic lithotypes have been recognized. Based on the texture and modal mineralogy, mafic ejecta can be grouped into doleritic gabbros, poikilitic gabbros, sub-ophitic gabbros, ophitic gabbros, micro-monzogabbros and porphyrogabbros. Ultramafic samples consist of dunites and wherlites, these latter being also represented by composite samples making transition to plagioclase wherlite and troctolite-gabbro patchy microdomains.

A wide spectrum of crystallization processes can be envisaged in this mafic-ultramafic subvolcanic suite: (i) cumulitic dunites, wherlites, poikilitic to ophitic and sub-ophitic gabbros, and (ii) doleritic gabbros and some sub-ophitic gabbros instead representing *in situ* crystallization of cotectic basalt (SSB) in subvolcanic environment and thus retaining a liquid composition (e.g. slowly cooled equivalents of basalts).

Samples of the Bellecombe Ash Member defined as porphyrogabbros could represent oceanite (picritic) magmas stored at crustal levels where the basaltic melts may have evolved to a sanidine- and phlogopite-bearing micro-cryptocrystalline groundmass finally supporting the abundant olivine crystals. This groundmass in the porphyrogabbros could be really considered a trachytic liquid, which may be erupted as glassy pumice if the undercooling would be high enough to preclude groundmass crystallization.

The compositional range from basalt to basaltic trachyandesite of the interstitial glass found in the intrusive clasts indicate that the melt was entrapped within the crystal frameworks and quenched during the eruption. This likely suggests the clast were disrupted from the wall rocks and transported very fast to the surface, mostly as cognates. If not entrapped interstitially during *in situ* crystallization, silicate liquid could be also expelled from the crystal frameworks, forming separated melt pockets or larger body, like a differentiated sill. Trachytic pumices from the Bellecombe Ash Member show a more evolved composition than the quenched interstitial glass in the subvolcanic ejecta, therefore suggesting the liquid may really evolve and escape if not entrapped in the crystal framework of some crystallization front of the magma reservoirs.

Fluid inclusion barometry (Raman analyses) and noble gases and CO<sub>2</sub> from fluid inclusion crushing extraction and mass spectrometer analyses indicate the dunites are the intrusive ejecta from the deepest levels, approximately at the underplating levels of formation (~13 km b.s.l.). Wherlites should have crystallized close

to the submarine base of the edifice (~6 km b.s.l.), whereas the mafic subvolcanic samples are even shallower. Nevertheless, a large composite intrusive ejecta (B108) emphasizes the coexistence of ultramafic-mafic portions (wherlite to plagioclase-wherlite to troctolite and gabbro) at intermediate crustal level of crystallization. In fact, according to the liquidus curves of the La Reunion magmas (Fisk et al., 1988) the presence of troctolite patchy microdomains suggest a pressure of crystallization below 0.3 GPa. Textures present in these composite ejecta can be the result of basaltic melts interacting with wherlite cumulates at relatively low pressure, leading to the instability of clinopyroxene and high dissolution of olivines (highly rounded crystals) with a high loss of temperature of the system. In this way at the same pressure, but lower temperature, plagioclase is the *liquidus phase*, and therefore wherlitic cumulates interacting with basaltic melts at relatively low pressure can produce troctolite patchy microdomains at the mesoscale (Fig. 14.1).

The presence of highly rounded olivines often with fritted margin, surrounded by clinopyroxene in the wherlite sample B28 can be also explained by processes of interaction with rising basaltic magmas and phase relations in the system forsterite-diopside-SiO<sub>2</sub> (Kushiro, 1969, 1972), where the ferromagnesian minerals (olivine, clinopyroxene) are not sensitive to water pressure. The peritectic between forsterite, diopside and enstatite in the anhydrous system at atmospheric pressure changes composition only slightly in the water-saturated system up to 2 GPa (whereas silica strongly reduce its stability). In particular the texture characterizing the wherlite sample B28 should reflect processes occurring in the forsterite-diopside-SiO<sub>2</sub> system along the upper part of the cotectic line Fo-Di that can be also considered a reaction curve towards the reaction point defined by the reaction basaltic liquid + forsterite = diopside + pigeonite. As a matter of fact, it can be predicted that at 0.3 GPa (pressure inferred for the wherlite lithotypes by fluid inclusions) textural and modal modifications of ultramafic cumulate assemblages may occur during percolation and rising of basaltic melts. It is worth to note that at this pressure (about 0.3 GPa) clinopyroxene can be out of its stability curve whereas plagioclase is on the liquidus (Fisk et al., 1988; Albarede et al., 1997). B28 wherlite has got, in fact, a plagioclase modal content of about 8 vol%.



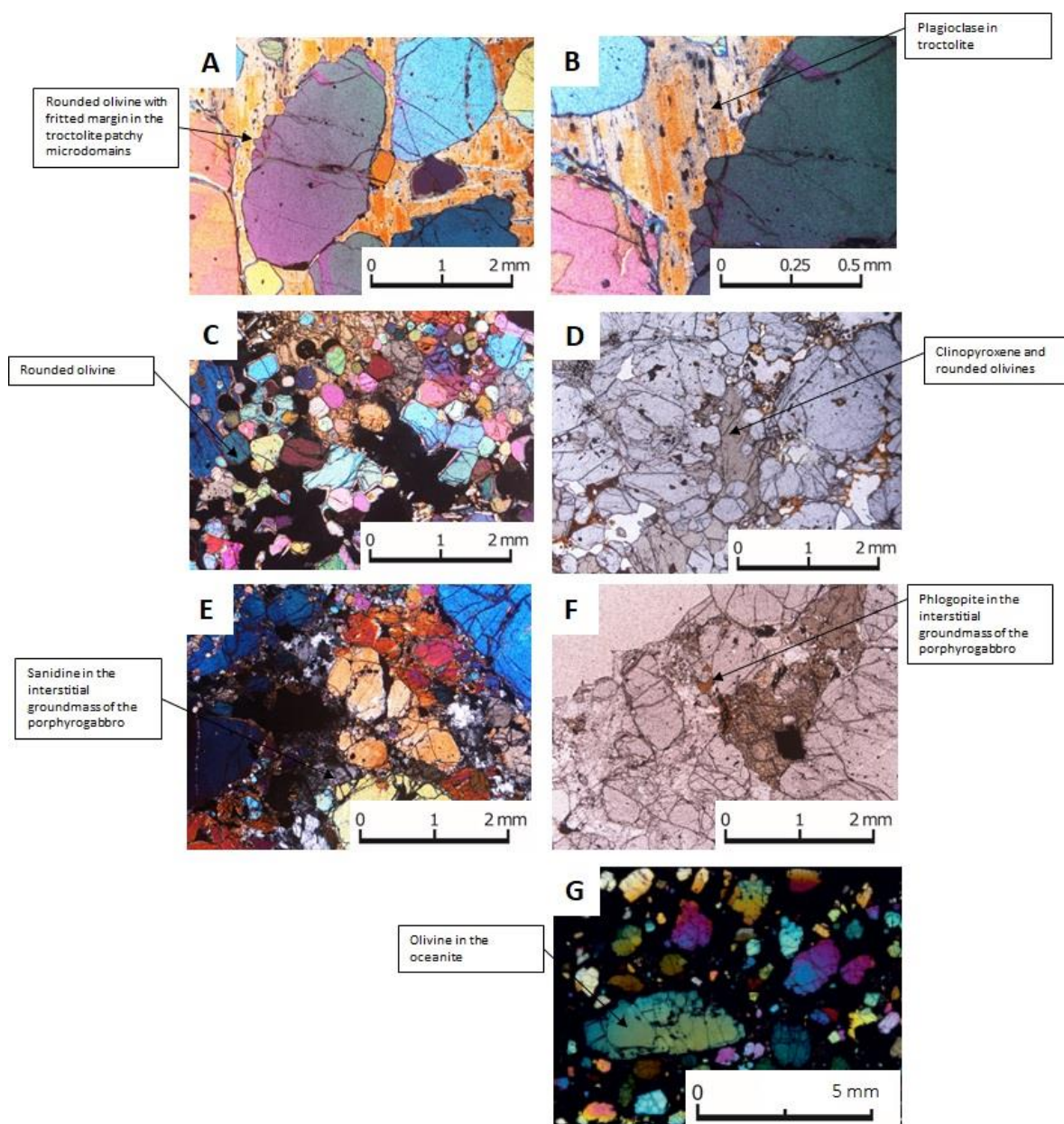


Fig. 14.1 - Thin section photos of some key-samples of the investigated subvolcanic ejecta. A-B: troctolite portions in the composite wherlite-plagioclase wherlite-troctolite-gabbro sample (B108a); B is a close up of A. C-D: wherlite B28 sample; E-F: porphyrogabbro B20 sample. G: oceanite sample (from Di Muro et al., 2014) for comparison.

The overall textural and geochemical data of the investigated mafic-ultramafic ejecta do not only reflect different mode of subvolcanic crystallization (i.e. cumulate *sensu stricto* vs. slowly-cooled equivalents of the steady state basalts) but also (i) the interaction between basaltic liquids and cumulates already formed at crustal level of the Piton de la Fournaise plumbing system and (ii) the possibility that small pockets of liquids may also evolve to trachytic compositions.

1. Some textures and modal mineralogy of composite ultramafic-mafic ejecta or plagioclase-bearing wherlites can be the result

- of complex processes of modification of the pristine ultramafic cumulates by the interactions with high-temperature basalts persistently rising in the plumbing system of the Piton de la Fournaise. These destabilization/disequilibrium processes of cumulate assemblages occur at low-intermediate crustal levels where the liquidus curves of plagioclase and clinopyroxene straddled with stability-in (crystallization) and out of one or the other mineral phase as soon as the melt temperature rapidly decrease for loss of energy due to olivine dissolution. The presence of textures with rounded olivines, even with fritted margins, indicate an interaction of ultramafic cumulates with basaltic melts at crustal levels leading to the transition of wherlite to plagioclase wherlite and troctolite. It is worth to note that similar processes of basaltic percolation into peridotites in mantle environments give rise to dunitization processes (Renna et al., 2011; Drouin et al., 2009, 2010; Basch et al., 2018) rather than ultramafic (wherlite) to mafic (troctolite-gabbro) transition. In addition, the behaviour of part of the olivine-diopside curve not as cotectic but as reaction curve, in the Forsterite-Diopside-Quartz system, may also explain some textures in heterogranular wherlite ejecta.
2. Some ejecta, namely porphyrogabbros, may represent, for texture and modal mineralogy, the "unerupted" counterpart of the oceanite (picritic) magmas. The micro-cryptocrystalline groundmass of the porphyrogabbros, otherwise erupted as basaltic glass in the oceanites, shows a modal mineralogy with sanidine and phlogopite microlites. In this way, an evolution of the basaltic interstitial melt towards trachytic terms may finally support, as groundmass, the olivine phyric magmas of the porphyrogabbros. In these subvolcanic lithotypes, if the sustained heat flux would prevent the crystallization of the evolved interstitial melt, then small pockets of trachyte could have the opportunity to survive and be erupted as very small volumes of pumices.
  3. Trachytic melts effectively erupted as pumices in the Bellecombe Ash Member, formed when interstitial basaltic melt differentiated and quenched to form glass. It is worth to note that in April 2007, Piton de la Fournaise emitted a very small volume of trachytic pumice during its largest historical eruption. The trachyte, genetically related to the evolution of basic magmas of the La Réunion mantle plume, was interpreted by Vlastelic et al. (2021) as a liquid remnant of an extinct volcano, such as Piton des Neiges or Les Alizés or partial melting of old syenite intrusions or remobilization of interstitial melts not fully crystallized. The few trachitic pumices in the Bellecombe Ash Member and the sanidine- and phlogopite-bearing micro-cryptocrystalline groundmass in the porphyrogabbro ejecta agree with the hypothesis that small pockets of trachyte melt could generally form in the Piton de la Fournaise plumbing systems. Solidification and trapping of

these very small volumes of trachytic melts in the subvolcanic crystallization processes could be locally and temporarily prevented by very high heat flux which is not unlikely for La Reunion mantle plume.

4. Finally, the preliminary investigation of some loose quartz xenocrysts on the pyroclastic products highlighted the possible hydrothermal origin of these crystals with implications on the involvement of the hydrothermal system of the Piton de la Fournaise during the most violent and explosive eruptions of the volcano.

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## 16 References

Albarède, F., Luais, B., Fitton, G., Semet, M., Kaminski, E., Upton, B. G. J., & Cheminée, J. L. (1997). The geochemical regimes of Piton de la Fournaise volcano (Réunion) during the last 530 000 years. *Journal of Petrology*, 38(2), 171-201.

- Bachelery, P. (1981). "*Le Piton de la Fournaise, Ile de la Réunion: étude volcanologique, structurale et pétrologique. \_ 63*" (Doctoral dissertation, Impr. UER Sci.).
- Bachèlery, P., & Mairine, P. (1990). Evolution volcano-structurale du Piton de la Fournaise depuis 0, 53 Ma. In *Le volcanisme de la Réunion-Monographie* (pp. 213-242).
- Bachelery, P., Lenat, J. F., Di Muro, A., & Michon, L. (Eds.). (2015). *Active Volcanoes of the Southwest Indian Ocean: Piton de la Fournaise and Karthala*. Springer.
- Bambauer, H.U. (1961): Spurenelementgehalt und -Farbzentren in Quarzen aus Zerrklüften der Schweizer Alpen. *Schweiz. Mineral. Petrogr. Mitt.*, 41, 335-369.
- Basch, V., Rampone, E., Crispini, L., Ferrando, C., Ildefonse, B., & Godard, M. (2018). From mantle peridotites to hybrid troctolites: Textural and chemical evolution during melt-rock interaction history (Mt. Maggiore, Corsica, France). *Lithos*, 323, 4-23.
- Bernard A, Munschy M (2000) Were the Mascarene and Laxmi Basins (western Indian Ocean) formed at the same spreading centre? *C R Acad Sci Paris Sci de la Terre et des Planètes Earth Planet Sci* 330:777-783
- Berthod, C., Michon, L., Famin, V., Welsch, B., Bachèlery, P., & Bascou, J. (2020). Layered gabbros and peridotites from Piton des Neiges volcano, La Réunion Island. *Journal of Volcanology and Geothermal Research*, 107039.
- Boivin, P., & Bachèlery, P. (2009). Petrology of 1977 to 1998 eruptions of Piton de la Fournaise, La Réunion Island. *Journal of volcanology and geothermal research*, 184(1-2), 109-125.
- Boudoire, G., Brugier, Y. A., Di Muro, A., Wörner, G., Arienzo, I., Metrich, M., & Michon, L. (2019). Eruptive activity on the western flank of Piton de la Fournaise (La Réunion Island, Indian Ocean): insights on magma transfer, storage and evolution at an oceanic volcanic island. *J Petrol.* <https://doi.org/10.1093/petrology/egz04>, 5.
- Boudoire, G., Rizzo, A. L., Di Muro, A., Grassa, F., & Liuzzo, M. (2018). Extensive CO<sub>2</sub> degassing in the upper mantle beneath oceanic basaltic volcanoes: First insights from Piton de la Fournaise volcano (La Réunion Island). *Geochimica et Cosmochimica Acta*, 235, 376-401.
- Brugier, Y. (2012). *Pétrologie naturelle et expérimentale de laves du Piton de la Fournaise, La Réunion* (Doctoral dissertation, Msc thesis, Université d'Orléans).
- Brugier, Y. (2016). *Magmatologie du Piton de la Fournaise (Ile de La Réunion). Approche volcanologique, pétrologique et expérimentale.* (Doctoral dissertation, Université d'Orléans).
- Bureau, H., Métrich, N., Pineau, F., & Semet, M. P. (1998). Magma-conduit interaction at Piton de la Fournaise volcano (Réunion

- Island): a melt and fluid inclusion study. *Journal of Volcanology and Geothermal Research*, 84(1-2), 39-60.
- Bureau, H., Métrich, N., Pineau, F., & Semet, M. P. (1998). Magma-conduit interaction at Piton de la Fournaise volcano (Réunion Island): a melt and fluid inclusion study. *Journal of Volcanology and Geothermal Research*, 84(1-2), 39-60.
- Bureau, H., Métrich, N., Semet, M. P., & Staudacher, T. (1999). Fluid-magma decoupling in a hot-spot volcano. *Geophysical Research Letters*, 26(23), 3501-3504.
- Bureau, H., Pineau, F., Métrich, N., Semet, M. P., & Javoy, M. (1998). A melt and fluid inclusion study of the gas phase at Piton de la Fournaise volcano (Réunion Island). *Chemical geology*, 147(1-2), 115-130.
- Burnard, P., Zimmermann, L., & Sano, Y. (2013). The noble gases as geochemical tracers: History and background. In *The Noble Gases as Geochemical Tracers* (pp. 1-15). Springer, Berlin, Heidelberg.
- Carignan, J., Hild, P., Mevelle, G., Morel, J., & Yeghicheyan, D. (2001). Routine analyses of trace elements in geological samples using flow injection and low pressure on-line liquid chromatography coupled to ICP-MS: A study of geochemical reference materials BR, DR-N, UB-N, AN-G and GH. *Geostandards Newsletter*, 25(2-3), 187-198.
- Charvis, P., Laesanpura, A., Gallart, J., Hirn, A., Lépine, J. C., de Voogd, B., ... & Pontoise, B. (1999). Spatial distribution of hotspot material added to the lithosphere under La Réunion, from wide-angle seismic data. *Journal of Geophysical Research: Solid Earth*, 104(B2), 2875-2893.
- Chevallier, L., & Bachelery, P. (1981). Evolution structurale du volcan actif du Piton de la Fournaise, Ile de la Réunion-Océan indien occidental. *Bulletin volcanologique*, 44(4), 723-741.
- Chevallier, L., & Vatin-Perignon, N. (1982). Volcano-structural evolution of Piton des Neiges, Reunion Island, Indian Ocean. *Bulletin volcanologique*, 45(4), 285-298.
- Coffin MF, Rabinowitz PD (1987) Reconstruction of Madagascar and Africa: evidence from the Davie fracture zone and western Somali basin. *J Geophys Res* 92(B9):9385-9406
- D'Lemos, R.S., Kearsley, A.T., Pembroke, J.W., Watt, G.R., Wright, P. (1997): Complex quartz growth histories in granite revealed by scanning cathodoluminescence techniques. *Geol. Mag.*, 134, 459-552.
- Dennen, W. H. (1967). Trace elements in quartz as indicators of provenance. *Geological Society of America Bulletin*, 78(1), 125-130.
- Dennen, W.H. (1966): Stoichiometric substitution in natural quartz. *Geochim. Cosmochim. Acta*, 30, 1235-1241.
- Dennen, W.H., Blackburn, W.H., Quesada, A. (1970): Aluminium in quartz as a geothermometer. *Contrib. Mineral. Petrol.*, 27, 332-342.

- Di Muro, A., Métrich, N., Vergani, D., Rosi, M., Armienti, P., Fougeroux, T., Civetta, L. (2014). The shallow plumbing system of Piton de la Fournaise Volcano (La Reunion Island, Indian Ocean) revealed by the major 2007 caldera-forming eruption. *Journal of Petrology*, 55(7), 1287-1315.
- Di Muro, A., Métrich, N., Allard, P., Aiuppa, A., Burton, M., Galle, B., & Staudacher, T. (2016). Magma degassing at Piton de la Fournaise volcano. In *Active Volcanoes of the Southwest Indian Ocean* (pp. 203-222). Springer, Berlin, Heidelberg.
- Drouin, M., Godard, M., Ildefonse, B., Bruguier, O., & Garrido, C. J. (2009). Geochemical and petrographic evidence for magmatic impregnation in the oceanic lithosphere at Atlantis Massif, Mid-Atlantic Ridge (IODP Hole U1309D, 30 N). *Chemical Geology*, 264(1-4), 71-88.
- Drouin, M., Ildefonse, B., & Godard, M. (2010). A microstructural imprint of melt impregnation in slow spreading lithosphere: Olivine-rich troctolites from the Atlantis Massif, Mid-Atlantic Ridge, 30 N, IODP Hole U1309D. *Geochemistry, Geophysics, Geosystems*, 11(6).
- Duffield, W. A., Stieltjes, L., & Varet, J. (1982). Huge landslide blocks in the growth of Piton de la Fournaise, La Reunion, and Kilauea Volcano, Hawaii. *Journal of Volcanology and Geothermal Research*, 12(1-2), 147-160.
- Duncan RA (1990) The volcanic record of the Reunion hotspot. In: Duncan RA, Backman J, Dunbar RB, Peterson LC (eds) Proceedings of ocean drilling program science results, vol 115, pp 3-10
- Duncan RA, Backman J, Peterson L (1989) Reunion hotspot activity through tertiary time: initial results from the ocean drilling program, Leg 115. *J Volcanol Geoth Res* 36:193-198
- Dyment J (1991a) Structure et évolution de la lithosphere océanique dans l'Océan Indien: apport des anomalies magnétiques. Thesis Thesis, IPGS, Université Louis-Pasteur, Strasbourg, 374 pp
- Eggins, S. M., Kinsley, L. P. J., & Shelley, J. M. G. (1998). Deposition and element fractionation processes during atmospheric pressure laser sampling for analysis by ICP-MS. *Applied Surface Science*, 127, 278-286.
- Famin, V., Welsch, B., Okumura, S., Bachèlery, P., & Nakashima, S. (2009). Three differentiation stages of a single magma at Piton de la Fournaise volcano (Reunion hot spot). *Geochemistry, Geophysics, Geosystems*, 10(1).
- Fisk, M. R., Upton, B. G. J., Ford, C. E., & White, W. M. (1988). Geochemical and experimental study of the genesis of magmas of Reunion Island, Indian Ocean. *Journal of Geophysical Research: Solid Earth*, 93(B5), 4933-4950.
- Fontaine, F. J., Rabinowicz, M., Boulègue, J., & Jouniaux, L. (2002). Constraints on hydrothermal processes on basaltic edifices:

inferences on the conditions leading to hydrovolcanic eruptions at Piton de la Fournaise, Réunion Island, Indian Ocean. *Earth and Planetary Science Letters*, 200(1-2), 1-14.

Frezzotti M.L., Tecce F., Casagli A. (2012): Raman spectroscopy for fluid inclusion analysis. *J. Geochem. Explor.* 112, 1-20.

Gailler, L. S., & Lénat, J. F. (2010). Three-dimensional structure of the submarine flanks of La Réunion inferred from geophysical data. *Journal of Geophysical Research: Solid Earth*, 115(B12).

Gallart, J., Driad, L., Charvis, P., Sapin, M., Hirn, A., Diaz, J., ... & Sachpazi, M. (1999). Perturbation to the lithosphere along the hotspot track of La Réunion from an offshore-onshore seismic transect. *Journal of Geophysical Research: Solid Earth*, 104(B2), 2895-2908.

Georgen JE, Lin J, Dick HJB (2001) Evidence from gravity anomalies for interactions of the Marion and Bouvet hotspots with the Southwest Indian Ridge: effects of transform offsets. *Earth Planet Sci Lett* 187:283-300

Gillot P.-Y., Lefèvre J.-C., Nativel P. E., (1994). Model for the structural evolution of the volcanoes of Réunion island. *Earth Planet. Sci. Lett.* 122, 291-302.

Gillot, P. Y., & Nativel, P. (1989). Eruptive history of the Piton de la Fournaise volcano, Réunion island, Indian Ocean. *Journal of volcanology and geothermal research*, 36(1-3), 53-65.

Götze, J., Plötze, M., & Habermann, D. (2001). Origin, spectral characteristics and practical applications of the cathodoluminescence (CL) of quartz—a review. *Mineralogy and petrology*, 71(3-4), 225-250.

Helz, R. T., & Thornber, C. R. (1987). Geothermometry of Kilauea Iki lava lake, Hawaii. *Bulletin of volcanology*, 49(5), 651-668.

Hilton, D. R., Hammerschmidt, K., Loock, G., & Friedrichsen, H. (1993). Helium and argon isotope systematics of the central Lau Basin and Valu Fa Ridge: Evidence of crust/mantle interactions in a back-arc basin. *Geochimica et Cosmochimica Acta*, 57(12), 2819-2841.

Jenkins, R., & Snyder, R. L. (1996). *Introduction to X-ray powder diffractometry* (Vol. 138, pp. 1-202). New York: Wiley. pp. 47; 244-247.

Kent R.W., Pringle M.S., Müller R.D., Saunders A.D., Ghose N.C., (2002). 40Ar/39Ar geochronology of the rajmahal basalts, India, and their relationship to the kerguelen plateau. *J. Petrol.* 43, 1141-1153.

Kluska, J. M. (1997). *Evolution magmatique et morpho-structurale du Piton des Neiges au cours des derniers 500 000 ans* (Doctoral dissertation, Paris 11).

- Kurz, M. D. (1986). Cosmogenic helium in a terrestrial igneous rock. *Nature*, 320(6061), 435.
- Kushiro, I. (1969). The system forsterite-diopside-silica with and without water at high pressures. *American Journal of Science*, 267(A), 269-294.
- Kushiro, I. (1972). Effect of water on the composition of magmas formed at high pressures. *Journal of Petrology*, 13(2), 311-334.
- Labazuy, P. (1996). Recurrent landslides events on the submarine flank of Piton de la Fournaise volcano (Reunion Island). *Geological Society, London, Special Publications*, 110(1), 295-306.
- Lacroix, A. F. A. (1936). *Le volcan actif de l'île de la Réunion et ses produits*. Gauthier.
- Lamadrid, H. M., Moore, L. R., Moncada, D., Rimstidt, J. D., Burruss, R. C., & Bodnar, R. J. (2017). Reassessment of the Raman CO<sub>2</sub> densimeter. *Chemical Geology*, 450, 210-222.
- Le Friant, A., Lebas, E., Clément, V., Boudon, G., Deplus, C., De Voogd, B., & Bachèlery, P. (2011). A new model for the evolution of La Réunion volcanic complex from complete marine geophysical surveys. *Geophysical Research Letters*, 38(9).
- Lehmann, G. (1975): On the colour centres of iron in amethyst and synthetic quartz: a discussion. *Am. Mineral.*, 60, 335-337.
- Lehmann, G. & Bambauer, H.V. (1973): Quarzkristalle und ihre Farben. *Angew. Chem.*, 7, 281-289.
- Lenat, J. F., Bachelery, P., & Desmulier, F. (2001). Genese du champ de lave de l'Enclos Fouqué; une eruption d'envergure exceptionnelle du Piton de la Fournaise (Reunion) au 18 e siecle. *Bulletin de la Société Géologique de France*, 172(2), 177-188.
- Lénat, J. F., Bachèlery, P., & Merle, O. (2012). Anatomy of Piton de la Fournaise volcano (La Réunion, Indian Ocean). *Bulletin of volcanology*, 74(9), 1945-1961.
- Lénat, J. F., Boivin, P., Deniel, C., Gillot, P. Y., & Bachèlery, P. (2009). Age and nature of deposits on the submarine flanks of Piton de la Fournaise (Reunion Island). *Journal of volcanology and geothermal research*, 184(1-2), 199-207.
- Lénat, J. F., Vincent, P., & Bachèlery, P. (1989). The off-shore continuation of an active basaltic volcano: Piton de la Fournaise (Réunion Island, Indian Ocean); structural and geomorphological interpretation from sea beam mapping. *J. Volcanol. Geotherm. Res*, 36(1-3), 1-36.
- Letourneur, L., Peltier, A., Staudacher, T., & Gudmundsson, A. (2008). The effects of rock heterogeneities on dyke paths and asymmetric ground deformation: the example of Piton de la Fournaise (Réunion Island). *Journal of Volcanology and Geothermal Research*, 173(3-4), 289-302.



- Luais, B. (2004). Temporal changes in Nd isotopic composition of Piton de la Fournaise magmatism (Réunion Island, Indian Ocean). *Geochemistry, Geophysics, Geosystems*, 5(1).
- Mairine, P., & Bachèlery, P. (1997). Un grand épisode érosionnel dans l'histoire ancienne du Piton de la Fournaise (île de la Réunion). *Comptes Rendus de l'Académie des Sciences-Series IIA-Earth and Planetary Science*, 325(4), 243-249.
- Maschmeyer, D. & Lehmann, G. (1983): A trapped-hole centre causing rose coloration of natural quartz. *Z. Kristallogr.*, 163, 181-196.
- McNaught, A. D., & Wilkinson, A. (1997). Compendium of chemical terminology. IUPAC recommendations.
- Merle, O., & Lénat, J. F. (2003). Hybrid collapse mechanism at Piton de la Fournaise volcano, Reunion Island, Indian Ocean. *Journal of Geophysical Research: Solid Earth*, 108(B3).
- Merle, O., Mairine, P., Michon, L., Bachèlery, P., & Smietana, M. (2010). Calderas, landslides and paleo-canyons on Piton de la Fournaise volcano (La Réunion Island, Indian Ocean). *Journal of Volcanology and Geothermal Research*, 189(1-2), 131-142.
- Michon L, Di Muro A, Villeneuve N, Saint-Marc C, Fadda P, Manta F (2013) Explosive activity of the summit cone of Piton de la Fournaise volcano (La Réunion island): a historical and geological review. *J Volcanol Geotherm Res* 264:117-133. doi:10.1016/j.jvolgeores.2013.06.012
- Michon, L., & Saint-Ange, F. (2008). Morphology of Piton de la Fournaise basaltic shield volcano (La Réunion Island): Characterization and implication in the volcano evolution. *Journal of Geophysical Research: Solid Earth*, 113(B3).
- Mohamed-Abchir A (1996) Les Cendres de Bellecombe: un évènement majeur dans le passé récent du Piton de la Fournaise, Ile de la Réunion. PhD thesis, Univ Paris VII 248 pp
- Morandi, A., Di Muro, A., Principe, C., Michon, L., Leroi, G., Norelli, F., & Bachèlery, P. (2016). Pre-historic (< 5 kiloyear) explosive activity at Piton de la Fournaise volcano. In *Active Volcanoes of the Southwest Indian Ocean* (pp. 107-138). Springer, Berlin, Heidelberg.
- Müller, A. (2000): Cathodoluminescence and characterisation of defect structures in quartz with applications to the study of granitic rocks. Unpubl. Ph.D. Thesis, University Göttingen, 230 pp.
- Müller, A., Seltmann, R., Behr, H.-J. (2000): Application of cathodoluminescence to magmatic quartz in a tin granite - case study from the Schellerhau Granite Complex, Eastern Erzgebirge, Germany. *Mineralium Deposita*, 35, 169-189.
- Müller, A., Wiedenbeck, M., KERKHOF, A. M. V. D., Kronz, A., & Simon, K. (2003). Trace elements in quartz—a combined electron microprobe, secondary ion mass spectrometry, laser-ablation ICP-MS, and

- cathodoluminescence study. *European Journal of Mineralogy*, 15(4), 747-763.
- Oehler, J. F., Labazuy, P., & Lénat, J. F. (2004). Recurrence of major flank landslides during the last 2-Ma-history of Reunion Island. *Bulletin of Volcanology*, 66(7), 585-598.
- Oehler, J. F., Lénat, J. F., & Labazuy, P. (2008). Growth and collapse of the Reunion Island volcanoes. *Bulletin of Volcanology*, 70(6), 717-742.
- Ort, M., Di Muro, A., Michon, L., & Bachèlery, P. (2014). The Bellecombe ash: indications of explosive eruptions at Piton de la Fournaise, Reunion Island. In *IAVCEI-5IMC conference. Queretaro, Mexico*.
- Ort, M. H., Di Muro, A., Michon, L., & Bachèlery, P. (2016). Explosive eruptions from the interaction of magmatic and hydrothermal systems during flank extension: the Bellecombe Tephra of Piton de La Fournaise (La Réunion Island). *Bulletin of Volcanology*, 78(1), 5.
- Peltier, A., Staudacher, T., & Bachèlery, P. (2007). Constraints on magma transfers and structures involved in the 2003 activity at Piton de La Fournaise from displacement data. *Journal of Geophysical Research: Solid Earth*, 112(B3).
- Perny, B., Eberhardt, P., Ramseyer, K., Mullis, J., Pankrath, R. (1992): Microdistribution of aluminium, lithium and sodium in a quartz: possible causes and correlation with short lived cathodoluminescence. *Am. Mineral.*, 77, 534-544.
- Petrelli, M., Perugini, D., Alagna, K. E., Poli, G., & Peccerillo, A. (2008). Spatially resolved and bulk trace element analysis by laser ablation-inductively coupled plasma-mass spectrometry (LA-ICP-MS). *Periodico di Mineralogia*, 77(3).
- Pichavant, M., Brugier, Y., & Di Muro, A. (2016). Petrological and experimental constraints on the evolution of Piton de la Fournaise magmas. In *Active Volcanoes of the Southwest Indian Ocean* (pp. 171-184). Springer, Berlin, Heidelberg.
- Putirka, K. D. (2008). Thermometers and barometers for volcanic systems. *Reviews in mineralogy and geochemistry*, 69(1), 61-120.
- Ramseyer, K., & Mullis, J. (1990). Factors influencing short-lived blue cathodoluminescence of alpha-quartz. *American Mineralogist*, 75(7-8), 791-800.
- Ramseyer, K., Baumann, J., Matter, A., & Mullis, J. (1988). Cathodoluminescence colours of  $\alpha$ -quartz. *Mineralogical Magazine*, 52(368), 669-677.
- Rançon, J. P., Lerebour, P., & Augé, T. (1989). The Grand Brule exploration drilling: New data on the deep framework of the Piton de la Fournaise volcano. Part 1: Lithostratigraphic units and

volcanostructural implications. *Journal of Volcanology and Geothermal Research*, 36(1-3), 113-127.

Reed, S. J. B. (2005). *Electron microprobe analysis and scanning electron microscopy in geology*. Cambridge university press.

Renna, M. R., & Tribuzio, R. (2011). Olivine-rich troctolites from Ligurian ophiolites (Italy): evidence for impregnation of replacive mantle conduits by MORB-type melts. *Journal of Petrology*, 52(9), 1763-1790.

Rizzo, A. L., Pelorosso, B., Coltorti, M., Ntaflou, T., Bonadiman, C., Matusiak-Małek, M., ... & Bergonzoni, G. (2018). Geochemistry of noble gases and CO<sub>2</sub> in fluid inclusions from lithospheric mantle beneath Wilcza Góra (Lower Silesia, southwest Poland). *Frontiers in Earth Science*.

Saint-Ange, F., Bachèlery, P., Babonneau, N., Michon, L., & Jorry, S. J. (2013). Volcaniclastic sedimentation on the submarine slopes of a basaltic hotspot volcano: Piton de la Fournaise volcano (La Réunion Island, Indian Ocean). *Marine Geology*, 337, 35-52.

Salaün, A., Villemant, B., Semet, M. P., & Staudacher, T. (2010). Cannibalism of olivine-rich cumulate xenoliths during the 1998 eruption of Piton de la Fournaise (La Réunion hotspot): Implications for the generation of magma diversity. *Journal of Volcanology and Geothermal Research*, 198(1-2), 187-204.

Sano, Y., & Wakita, H. (1985). Distribution of <sup>3</sup>He/<sup>4</sup>He ratios and its implications for geotectonic structure of the Japanese Islands. *J. Geophys. Res*, 23, 109-123.

Schettino A, Scotese CR (2005) Apparent polar wander paths for the major continents (200 Ma to the present day): a palaeomagnetic reference frame for global plate tectonic reconstructions. *Geophys J Int* 163:727- 759. doi:10.1111/j.1365-246X.2005.02638.x

Schneider, N. (1993): Das lumineszenzaktive Strukturinventar von Quarzphänokristen in Rhyolithen. *Göttinger Arb.Geol. Paläont.*, 60, 81 pp.

Smietana, M. (2011). *Pétrologie, géochronologie (K-Ar) et géochimie élémentaire et isotopique (Sr, Nd, Hf, Pb) de laves anciennes de La Réunion: Implications sur la construction de l'édifice volcanique* (Doctoral dissertation, La Réunion).

Smith W, Sandwell D (1997) Global sea floor topography from satellite altimetry and ship depth soundings. *Science* 277:1956-1962

Smith, J.V. & Steele, I.M. (1984): Chemical substitution in silica polymorphs. *N. Jb. Miner. Mh.*, 1984, 137-144.

Sobolev, A. V., & Nikogosian, I. K. (1994). Petrology of long-lived mantle plume magmatism: Hawaii, Pacific and Reunion Island, Indian Ocean. *Petrology*, 2(2), 111-144.

- Staudacher, T., & Allègre, C. J. (1993). Ages of the second caldera of Piton de la Fournaise volcano (Réunion) determined by cosmic ray produced  $^3\text{He}$  and  $^{21}\text{Ne}$ . *Earth and Planetary Science Letters*, 119(3), 395-404.
- Storey, B. C. (1995). The role of mantle plumes in continental breakup: case histories from Gondwanaland. *Nature*, 377(6547), 301-308.
- Upton, B. G. J., & Wadsworth, W. J. (1966). The basalts of Reunion Island, Indian Ocean. *Bulletin Volcanologique*, 29(1), 7-23.
- Upton, B. G. J., & Wadsworth, W. J. (1967). A complex basalt-mugearite sill in Piton des Neiges volcano, Reunion. *American Mineralogist: Journal of Earth and Planetary Materials*, 52(9-10), 1475-1492.
- Upton, B. G. J., Semet, M. P., & Joron, J. L. (2000). Cumulate clasts in the Bellecombe Ash Member, Piton de la Fournaise, Réunion Island, and their bearing on cumulative processes in the petrogenesis of the Réunion lavas. *Journal of volcanology and geothermal research*, 104(1-4), 297-318.
- Villemant, B., Salaün, A., & Staudacher, T. (2009). Evidence for a homogeneous primary magma at Piton de la Fournaise (La Réunion): a geochemical study of matrix glass, melt inclusions and Pélé's hairs of the 1998-2008 eruptive activity. *Journal of volcanology and geothermal research*, 184(1-2), 79-92.
- Vlastélic, I., Peltier, A., & Staudacher, T. (2007). Short-term (1998-2006) fluctuations of Pb isotopes at Piton de la Fournaise volcano (Réunion Island): origins and constraints on the size and shape of the magma reservoir. *Chemical Geology*, 244(1-2), 202-220.
- Vlastélic, I., Staudacher, T., Bachèlery, P., Télouk, P., Neuville, D., & Benbakkar, M. (2011). Lithium isotope fractionation during magma degassing: constraints from silicic differentiates and natural gas condensates from Piton de la Fournaise volcano (Réunion Island). *Chemical Geology*, 284(1-2), 26-34.
- Vlastélic, I., Bachèlery, P., Sigmarsson, O., Koga, K. T., Rose-Koga, E. R., Bindeman, I., ... & Staudacher, T. (2021). Prolonged trachyte storage and unusual remobilization at Piton de la Fournaise, La Réunion Island, Indian Ocean: Li, O, Sr, Nd, Pb and Th Isotope study. *Journal of Petrology*, 62(7), egab048.
- Wager, L. R., Brown, G. M., & Wadsworth, W. J. (1960). Types of igneous cumulates. *Journal of Petrology*, 1(1), 73-85.
- Watt, G. R., Wright, P., Galloway, S., McLean, C. (1997): Cathodoluminescence and trace element zoning in quartz phenocrysts and xenocrysts. *Geochim. Cosmochim. Acta*, 61, 4337-4348.
- Weil, J. A. (1984): A review of electron spin spectroscopy and its application to the study of paramagnetic defects in crystalline quartz. *Phys. Chem. Minerals*, 10, 149-165.

Welsch, B., Faure, F., Bachèlery, P., & Famin, V. (2009). Microcrysts record transient convection at Piton de la Fournaise volcano (La Réunion hotspot). *Journal of Petrology*, 50(12), 2287-2305.

## **17 Appendix**

*Table 7.1.1 - Subdivision of the representative analysed samples of the Bellecombe Ash Member.*

ROCK TYPE	Sample	Thin section	WR EOS/MS	SEM- BSD- EDS	EMPA	Raman	Noble gas and CO <sub>2</sub>	Thermobarometry
DUNITES	B7	X	X		X	X	X	X
	B29	X	X					
	B43	X	X		X			X
WHERLITES	B3	X	X			X	X	X
	B15	X	X		X		X	X
	B28	X	X	X				
OPHITIC GABBROS	B37	X	X					
SUB-OPHITIC GABBROS	B2	X	X		X		X	X
	B5	X	X					
	B6	X	X				X	
	B9	X	X					
	B11	X	X					
	B12	X	X					
	B14	X	X					
	B22	X	X					
	B23	X	X					
	B31	X						
	B33	X						
	B34	X	X					
	B40	X	X					
	B54	X						

	B58	X						
	B97	X						
<b>POIKILITIC GABBROS</b>	B21	X	X		X		X	X
	B32	X	X		X		X	X
	B41	X	X					
<b>DOLERITIC GABBROS</b>	B8	X	X					
	B13	X	X					
	B27	X	X					
<b>MICRO- MONZOGABBROS</b>	B52	X						
<b>PORPHYROGABBROS</b>	B4	X	X					
	B10	X	X					
	B16	X	X				X	
	B18	X						
	B20	X	X		X		X	X
<b>SEDIMENTARY</b>	B30	X						
<b>VOLCANICS (lavas and sub- volcanic products)</b>	B17	X						
	B19	X						
	B46	X						
	B56	X						
	R07-4	X						
	L4	X						
	L7	X						
	PDN1	X			X			



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	p2	X			X			X
	p3	X			X			
<b>Samples with an evident magmatic layering</b>	B108a	X		X				
	B108b	X						

Table 7.13.1 - SEM-BSD-EDS analyses of sample B108a on single mineral phases.

Mineral phase	Olivine										
SiO <sub>2</sub>	37,48	37,81	36,28	36,21	36,36	37,06	36,2	35,86	36,36	37,63	53,74
TiO <sub>2</sub>	0,09	0,06	0,07	0,1	0,11	0,04	0,04	0,03	0	0,02	0,39
Al <sub>2</sub> O <sub>3</sub>	0,05	0	0,06	0,05	0	0,03	0,06	0,07	0,04	0,05	0,85
Cr <sub>2</sub> O <sub>3</sub>	0,04	0,06	0,04	0,05	0,06	0,12	0,04	0	0	0,02	0,06
FeO	20,71	18,93	26,78	26,88	25,37	21,11	26,72	29,26	28,32	19,48	13,68
MnO	0,33	0,3	0,34	0,35	0,39	0,3	0,32	0,38	0,31	0,24	0,26
MgO	40,04	41,62	35,31	35,04	36,45	40,02	35,68	33,49	34,19	41,32	28,99
CaO	0,2	0,21	0,24	0,31	0,32	0,21	0,26	0,18	0,16	0,25	1,19
Na <sub>2</sub> O	0,29	0,28	0,24	0,3	0,24	0,32	0,24	0,27	0,18	0,3	0,25
NiO	0,07	0,04	0,1	0,11	0,1	0,06	0,05	0,05	0,03	0,08	0,05
Total	99,3	99,31	99,46	99,4	99,4	99,27	99,61	99,59	99,59	99,39	99,46
Fo	77,23	79,41	69,88	69,64	71,61	76,91	70,17	66,82	68,04	78,88	78,75
Fa	22,41	20,26	29,73	29,97	27,96	22,76	29,48	32,75	31,61	20,86	20,85
N° Mg	77,51	79,67	70,15	69,91	71,92	77,17	70,42	67,11	68,27	79,08	79,07
Mineral phase	Clinopyroxene										
SiO <sub>2</sub>	51,4	50,22	51,79	50,09	51,06	50,19	51,62				
TiO <sub>2</sub>	0,82	1,18	0,86	1,3	0,83	1,25	0,79				
Al <sub>2</sub> O <sub>3</sub>	2,9	3,06	1,79	3,36	3,45	3,18	2,79				
FeO	5,64	7,99	8,85	7,91	5,81	7,75	5,52				
MnO	0,13	0,13	0,2	0,17	0,06	0,16	0,05				

MgO	16,32	14,96	15,11	15,12	16,09	15,21	16,36
CaO	21,16	21,15	20,29	20,52	21,13	20,71	21,41
Na2O	0,42	0,42	0,56	0,44	0,37	0,41	0,31
K2O	0,11	0,05	0,07	0,12	0,04	0,1	0,09
Cr2O3	0,62	0,31	0,05	0,24	0,67	0,47	0,6
Totale	98,9	99,16	99,52	99,03	98,84	98,96	98,94
Wo	43,15	43,19	41,21	42,28	43,37	42,56	43,66
En	46,31	42,51	42,70	43,35	45,95	43,49	46,42
Fs	8,98	12,74	14,03	12,72	9,31	12,43	8,79
N° Mg	83,76	76,95	75,27	77,31	83,16	77,77	84,08
<b>Mineral phase</b>							
	<b>Clinopyroxene</b>						
SiO2	50,64	50,95	50,97	50,18	50,53	50,39	
TiO2	1,34	1,19	1,08	1,31	1,02	1,6	
Al2O3	2,98	2,29	2,85	3,01	3,83	2,99	
FeO	7,39	9,12	8,22	8,7	5,5	7,21	
MnO	0,17	0,2	0,15	0,17	0,03	0,18	
MgO	15,24	15,46	15,13	14,67	15,76	15,59	
CaO	20,82	19,58	20,49	20,61	21,32	20,8	
Na2O	0,56	0,43	0,45	0,51	0,41	0,51	
K2O	0,08	0,11	0,11	0,09	0,13	0,06	
Cr2O3	0,24	0,18	0,12	0,25	1,09	0,26	
Totale	99,22	99,33	99,45	99,25	98,53	99,33	

Wo	42,66	39,97	42,01	42,29	44,16	42,41	
En	43,45	43,91	43,16	41,88	45,42	44,23	
Fs	11,82	14,53	13,16	13,93	8,89	11,48	
N° Mg	78,61	75,14	76,64	75,04	83,63	79,40	

Mineral phase	Feldspar								
SiO2	48,08	56,25	49,19	46,86	47,76	55,57	47,68	46,69	62,64
TiO2	0	0,13	0	0,06	0,11	0	0,21	0	0,06
Al2O3	32,96	27,25	31,94	33,34	32,71	27,87	32,79	33,73	22,87
Cr2O3	0	0,05	0	0,06	0,05	0	0,04	0	0
FeO	0,66	0,6	0,73	0,74	0,72	0,54	0,85	0,51	0,61
MnO	0	0	0	0	0	0	0	0	0
MgO	0,24	0,21	0,27	0,26	0,27	0,24	0,2	0,37	0,2
CaO	15,43	8,94	14,61	15,97	15,23	9,69	15,34	16,59	4
Na2O	2,35	5,7	2,8	1,99	2,43	5,31	2,37	1,84	8,45
K2O	0,17	0,6	0,26	0,21	0,26	0,51	0,23	0,18	0,78
Totale	99,89	99,73	99,8	99,49	99,54	99,73	99,71	99,91	99,61
Ab %	21,39	51,65	25,35	18,17	22,06	48,27	21,55	16,54	75,62
An %	77,60	44,77	73,10	80,57	76,39	48,68	77,08	82,40	19,78
Or %	1,02	3,58	1,55	1,26	1,55	3,05	1,38	1,06	4,59

Mineral phase	Oxides			
SiO2	0,3	0,41	0,47	17,03
TiO2	4,3	46,25	6,08	0,05
Al2O3	12,78	0,31	3,97	5,94
Cr2O3	29,03	0,05	1,64	0,05
FeO	46,19	48,05	84,68	64,36

MnO	0	0,27	0,17	0,09
MgO	6,63	4,39	2,49	0,4
CaO	0,15	0,11	0,12	0,88
Totale	99,38	99,84	99,62	88,8

Table 7.14.1 - SEM-BSD-EDS analyses of sample B28 on single mineral phases.

Mineral phase	Olivine					Mineral phase	Clinopyroxene					
	SiO <sub>2</sub>	37,55	37,86	37,81	37,64		37,71	SiO <sub>2</sub>	54,36	50,47	52,14	52,03
TiO <sub>2</sub>	0,06	0,08	0,03	0,05	0,05	TiO <sub>2</sub>	0,48	0,93	0,72	0,65	0,98	1,3
Al <sub>2</sub> O <sub>3</sub>	0	0	0,04	0,04	0,08	Al <sub>2</sub> O <sub>3</sub>	3,92	3,49	3,36	3,6	4,09	3,77
Cr <sub>2</sub> O <sub>3</sub>	0,02	0,03	0,04	0,1	0,06	FeO	3,79	6,49	5,29	5,53	5,03	6,85
FeO	19,44	19,31	19,33	19,37	19,08	MnO	0	0,14	0,05	0	0	0,12
MnO	0,28	0,25	0,25	0,26	0,25	MgO	18,94	15,82	16,78	16,39	17,6	15,71
MgO	41,23	41,28	41,25	41,12	41,37	CaO	16,9	20,78	19,69	20,48	18,56	20,84
CaO	0,31	0,29	0,28	0,29	0,34	Na <sub>2</sub> O	0,51	0,43	0,44	0,39	0,48	0,4
Na <sub>2</sub> O	0,28	0,3	0,3	0,28	0,29	K <sub>2</sub> O	0	0,05	0,07	0	0,02	0,08
NiO	0,09	0,07	0,04	0,05	0,06	Cr <sub>2</sub> O <sub>3</sub>	0,5	0,76	0,78	0,71	0,16	0,21
Totale	99,26	99,47	99,37	99,2	99,29	Totale	98,9	98,6	98,54	99,07	99,5	99,28
Fo	78,84	79,00	78,97	78,87	79,23	Wo	35,86	42,73	41,05	42,40	38,79	42,73
Fa	20,85	20,73	20,76	20,84	20,50	En	55,91	45,26	48,68	47,21	51,18	44,82
N° Mg	79,08	79,21	79,18	79,10	79,45	Fs	6,28	10,42	8,61	8,94	8,21	10,96
						N° Mg	89,91	81,29	84,97	84,08	86,18	80,35

Mineral phase	Feldspar	Mineral phase	Oxids
SiO <sub>2</sub>	47,77	SiO <sub>2</sub>	0,32
TiO <sub>2</sub>	0,05	TiO <sub>2</sub>	6,72
Al <sub>2</sub> O <sub>3</sub>	32,77	Al <sub>2</sub> O <sub>3</sub>	12,25

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Cr2O3	0		Cr2O3	19,73
FeO	0,77		FeO	51,53
MnO	0		MnO	0
MgO	0,31		MgO	8,78
CaO	15,51		CaO	0,11
Na2O	2,31		Totale	99,44
K2O	0,18			
Totale	99,67			
Ab %	21,00			
An %	77,92			
Or %	1,08			

Table 8.1 - Whole rock ICP-OES compositions of representative analysed ejecta of the Bellecombe Ash Member.

Rock type	Dunite			Wherlite			Sub-ophitic gabbro								
	Sample	B7	B29	B43	B3	B15	B28	B2	B5	B6	B9	B11	B12	B14	B22
SiO <sub>2</sub>		39.36	39.39	38.89	46.5	41.17	42.09	46.57	43.72	48.24	45.72	49.68	46.65	46.13	48.36
Al <sub>2</sub> O <sub>3</sub>		0.393	0.36	0.329	2.917	1.14	2.078	14.22	12.994	14.613	14.93	14.858	14.293	14.888	20.343
Fe <sub>2</sub> O <sub>3</sub>		14.465	16.6	15.1	9.922	15.86	15.32	7.593	7.556	9.919	5.765	11.445	11.675	7.009	5.941
MnO		0.1783	0.2059	0.1937	0.1507	0.2106	0.2034	0.1123	0.1122	0.134	0.0846	0.1545	0.1416	0.0962	0.0803
MgO		46.015	44.125	44.805	22.59	35.41	32.29	10.86	11	8.612	10.575	6.262	7.241	10.012	6.984
CaO		0.283	0.483	0.28	15.408	5.532	7.296	16.25	15.22	11.51	16.028	11.723	14.028	16.483	14.68
Na <sub>2</sub> O	< L.D.		0.031	< L.D.	0.219	0.081	0.139	1.124	1.031	2.766	1.039	2.733	2.243	1.104	2.232
K <sub>2</sub> O	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	0.067	0.042	0.634	0.082	0.792	0.387	0.076	0.296
TiO <sub>2</sub>		0.07	0.105	0.071	0.765	0.315	0.435	1.028	1.133	3.352	0.731	2.602	2.954	1.371	1.227
P <sub>2</sub> O <sub>5</sub>	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	0.22	< L.D.	0.29	< L.D.	< L.D.	< L.D.
Total		99.6543	100.4899	99.69869	99.1617	99.37861	99.94139	99.55431	99.7682	99.78	99.8446	100.33	99.1826	100.1692	100.1033

Rock type	Sub-ophitic gabbro			Poikilitic gabbro			Ophitic gabbro	Doleritic gabbro			Porphyrogabbro				
	Sample	B23	B34	B40	B21	B32	B41	B37	B8	B13	B27	B4	B10	B16	B20
SiO <sub>2</sub>		47.38	47.22	41.55	46.83	41.5	43.15	47.36	46.65	46.94	46.7	42.91	42.97	42.04	41.94
Al <sub>2</sub> O <sub>3</sub>		15.843	12.816	15.625	13.756	7.475	9.582	10.153	13.646	15.343	13.816	7.158	6.756	6.375	4.047
Fe <sub>2</sub> O <sub>3</sub>		6.233	7.402	16.36	7.462	15.87	13.995	6.862	12.97	9.329	12.755	14.325	14.485	14.625	15.09
MnO		0.0922	0.1064	0.1325	0.1042	0.2059	0.1718	0.1109	0.1723	0.133	0.1691	0.1865	0.1855	0.1902	0.1961
MgO		9.819	10.315	6.625	13.615	25.445	17.865	11.885	7.009	8.915	6.905	24.83	27.275	28.21	33.63
CaO		16.583	17.808	13.905	15.178	7.49	12.065	18.553	11.863	14.575	11.718	6.487	5.834	5.438	3.595
Na <sub>2</sub> O		1.294	1.058	1.65	1.082	0.542	0.831	0.748	2.342	2.016	2.378	1.197	1.095	0.918	0.838
K <sub>2</sub> O		0.067	0.057	0.099	0.071	< L.D.	0.033	< L.D.	0.683	0.332	0.713	0.315	0.293	0.259	0.29
TiO <sub>2</sub>		0.829	1.293	3.584	0.784	0.384	1.269	1.274	3.297	2.021	3.116	1.302	1.198	1.137	0.781

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P205	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.		0.32	< L.D.	0.35		0.13	0.15	0.11	0.11	
Total	99.89021	99.6854	99.77049	98.8022	99.5019	98.6718		99.7159		100.33	99.944	100.4		98.72	99.11	100.01	99.66	



Table 8.1 - Continue. Whole rock ICP-MS compositions of representative analysed ejecta of the Bellecombe Ash Member.

Rock type Sample	Dunite			Wherlite			Sub-ophitic gabbro							
	B7	B29	B43	B3	B15	B28	B2	B5	B6	B9	B11	B12	B14	B22
As	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.
Ba	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	24.533	20.5431	117.6989	24.6039	128.8332	78.2151	23.1044	66.7128
Be	< L.D.	< L.D.	< L.D.	0.1077	< L.D.	0.0701	0.2388	0.2838	0.9873	0.1724	1.0651	0.5794	0.2485	0.5178
Bi	< L.D.	0.0733	< L.D.	0.0974	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	0.0564	0.0634	< L.D.	< L.D.
Cd	0.0298	0.03	0.0299	0.2264	0.0935	0.0904	0.1351	0.3072	0.1358	0.0636	0.1288	0.1332	0.1271	0.0777
Co	168.5371	169.0781	165.3549	77.2169	151.9122	137.8273	42.4135	41.3554	42.3577	38.1329	38.426	44.5825	37.5853	28.7782
Cr	3556.6429	1625.2103	3511.8354	3921.1512	2441.6341	2309.9943	958.5664	723.8789	338.9905	212.223	174.5259	102.1788	927.61	336.09
Cs	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	0.0224	< L.D.	0.2082	0.0327	0.2222	0.12	0.0279	0.1607
Cu	5.0771	15.2647	4.9343	10.5433	7.2828	14.7801	20.3033	27.8878	31.3634	12.85	44.4919	84.6075	15.9179	18.827
Ga	1.6836	1.4296	1.1215	5.547	2.4023	3.723	14.386	14.3737	20.8135	13.8124	21.4588	21.3644	14.5171	20.6735
Ge	0.8494	0.8162	0.8345	1.8046	1.2388	1.3164	1.4463	1.4848	1.4916	1.2589	1.5667	1.529	1.3464	1.1161
Hf	0.0433	0.1046	0.0461	0.8492	0.3454	0.5293	1.2201	1.5371	3.7235	1.0034	4.4844	2.8001	1.3558	1.8496
In	< L.D.	< L.D.	< L.D.	0.0529	0.0309	0.0358	0.0477	0.0577	0.0856	0.0424	0.0829	0.0879	0.0552	0.0474
Mo	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	0.6414	< L.D.	0.9072	0.5656	< L.D.	< L.D.
Nb	0.0731	0.2214	0.1816	0.9714	0.4475	0.7722	2.6552	2.6761	18.4663	1.9308	18.0519	10.5604	3.466	6.5935
Ni	2303.6328	2227.5309	2274.3222	711.7977	1383.0793	1226.8081	168.9093	185.9505	176.1571	195.162	69.7067	102.5265	187.4529	144.1745
Pb	1.4199	< L.D.	1.2323	3.5929	< L.D.	0.8964	1.1208	< L.D.	2.3291	0.8238	1.7251	1.6438	0.4824	1.9972
Rb	< L.D.	0.2335	< L.D.	0.2734	< L.D.	0.3789	1.1177	0.3603	14.9476	1.5033	19.1467	9.1236	1.2523	6.6327
Sb	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.
Sc	4.45	5.32	4.75	69.34	27.55	34.49	52.07	55.73	37.18	48.67	35.2	46.41	52.25	30.89
Sn	< L.D.	< L.D.	< L.D.	0.4907	0.4197	< L.D.	0.4941	0.6016	1.4912	0.424	2.2197	1.1784	0.9638	0.687
Sr	1.3262	3.4768	1.0796	26.8132	10.3074	23.316	360.6072	319.0316	405.2594	328.6908	352.5668	422.0717	338.3764	574.084
Ta	0.0076	0.02	0.0156	0.0862	0.0396	0.0686	0.2284	0.2304	1.5752	0.1638	1.478	0.8852	0.313	0.5593
Th	0.0257	0.0347	0.0231	0.1175	0.0538	0.1138	0.272	0.2864	1.8151	0.2375	2.3488	1.0192	0.3729	0.8835
U	< L.D.	0.0102	< L.D.	0.0295	0.0138	0.0284	0.0623	0.0697	0.4408	0.0614	0.5825	0.255	0.0909	0.2237
V	27.3713	27.6552	22.1252	193.5274	78.8363	97.9822	173.2096	182.4005	314.0072	146.1379	283.2509	446.232	183.4892	147.3704
W	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.
Y	0.4204	0.7596	0.4726	8.0524	3.0918	4.6294	8.9766	10.2831	20.9458	7.6382	25.1473	15.8182	8.9554	9.9845
Zn	106.8647	127.9181	110.905	58.4987	102.8215	102.3621	48.2442	44.3361	81.5842	33.7325	98.9431	89.4978	38.6339	45.2235
Zr	1.9346	3.8184	1.7864	22.4099	9.6336	15.2991	35.6697	42.7871	141.5424	29.2465	176.3469	101.3952	41.7361	68.6569
La	0.118	0.2425	0.3062	1.7738	0.6993	1.1755	3.4898	3.4458	13.7537	3.1178	18.028	7.9136	3.7219	7.2301

Ce	0.3153	0.6476	0.4342	5.1102	1.9232	3.0795	8.4675	9.4019	29.7166	6.9687	38.9158	17.9211	8.7655	15.4561
Pr	0.047	0.1043	0.0792	0.8924	0.3404	0.5621	1.338	1.5763	4.0639	1.1279	5.2605	2.6098	1.373	2.1125
Nd	0.2224	0.5165	0.3434	4.8398	1.9071	2.9281	6.7305	7.978	17.9686	5.742	23.0552	12.4814	6.7644	9.4972
Sm	0.0667	0.1549	0.0867	1.6411	0.6179	0.9531	2.0575	2.4924	4.6287	1.6493	5.6444	3.4659	2.009	2.3503
Eu	0.0207	0.0476	0.0232	0.5749	0.217	0.3313	0.878	0.9664	1.6858	0.6906	1.8028	1.4566	0.8228	1.1289
Gd	0.0654	0.1447	0.0735	1.8387	0.6652	1.0353	2.0731	2.5669	4.5336	1.7523	5.4879	3.526	2.0656	2.292
Tb	0.0114	0.0245	0.0123	0.3018	0.1109	0.1675	0.3278	0.4039	0.7125	0.2782	0.8708	0.5718	0.3373	0.3608
Dy	0.0752	0.1476	0.0779	1.8188	0.663	1.0064	1.9616	2.3667	4.2844	1.6623	5.1057	3.3445	1.9381	2.0902
Ho	0.0169	0.031	0.0171	0.3466	0.1302	0.1917	0.3694	0.444	0.8247	0.3111	0.9961	0.6465	0.3673	0.3991
Er	0.049	0.0854	0.047	0.8137	0.3058	0.4545	0.8539	1.0259	2.0134	0.7247	2.4206	1.537	0.8711	0.9458
Tm	0.0085	0.013	0.0078	0.1054	0.0403	0.059	0.1145	0.1352	0.2805	0.0965	0.346	0.2123	0.1165	0.1315
Yb	0.0637	0.0915	0.0553	0.6093	0.2436	0.3567	0.6683	0.7803	1.6925	0.5488	2.0271	1.2219	0.6584	0.7572
Lu	0.0122	0.0167	0.0087	0.0846	0.0357	0.0513	0.0939	0.1101	0.2462	0.0773	0.2991	0.176	0.0932	0.114

Rock type	Sub-ophitic gabbro	Sub-ophitic gabbro	Sub-ophitic gabbro	Poikilitic gabbro	Poikilitic gabbro	Poikilitic gabbro	Ophitic gabbro	Doleritic gabbro	Doleritic gabbro	Doleritic gabbro	Porphyro gabbro	Porphyro gabbro	Porphyro gabbro	Porphyro gabbro
Samp le	B23	B34	B40	B21	B32	B41	B37	B8	B13	B27	B4	B10	B16	B20
As	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.
Ba	21.4485	20.1253	34.891	19.9142	11.3032	16.0676	12.216	141.1865	70.8336	148.2733	60.633	63.5944	48.7911	56.327
Be	0.1903	0.2092	0.2668	0.2011	0.1221	0.2039	0.2059	1.086	0.5954	1.1105	0.5041	0.4426	0.3574	0.3719
Bi	< L.D.	< L.D.	< L.D.	0.0803	< L.D.	< L.D.	< L.D.	< L.D.	0.082	< L.D.	< L.D.	< L.D.	0.2234	< L.D.
Cd	0.0857	0.1055	0.12	0.1128	0.0607	0.0962	0.2205	0.2159	0.1428	0.2342	0.1001	0.0919	0.1081	0.0653
Co	37.9905	36.8434	63.2784	51.2614	124.0875	84.2485	34.7331	45.5879	41.326	45.4957	108.372	118.0504	118.8594	139.3364
				1933.344	1205.414	4288.197	1381.99				1611.674	1815.506	2029.951	2658.048
Cr	124.8525	550.0253	45.9679	1	5	2	75	379.5522	333.7561	382.6178	5	7	1	7
Cs	< L.D.	< L.D.	0.0362	0.0232	< L.D.	< L.D.	< L.D.	0.2928	0.2148	0.3042	0.1019	0.1172	0.0956	0.0779
Cu	54.6905	23.6939	54.4371	16.9442	16.0156	18.9535	58.0853	123.1429	44.934	118.0934	39.9928	36.7014	20.0561	9.5923
Ga	15.8037	14.4345	23.578	13.1601	7.6131	13.5117	12.9347	21.9058	18.8446	22.4384	10.5525	9.8751	9.1738	7.201
Ge	1.322	1.5031	1.2917	1.3542	1.1071	1.3342	1.7947	1.5565	1.5209	1.6344	1.1959	1.2277	1.2139	1.1097
Hf	1.1399	1.4268	1.5949	0.9735	0.4555	1.088	1.7036	4.3206	2.5495	4.4732	2.0695	1.8466	1.6529	1.6905
In	0.0483	0.0593	0.0723	0.0416	0.0333	0.0563	0.073	0.095	0.1652	0.0964	0.0525	0.0521	0.1644	0.0394
Mo	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	0.8822	0.8755	0.9005	0.5124	< L.D.	< L.D.	< L.D.
Nb	1.4813	2.0137	3.6804	1.9496	0.4893	0.7174	1.4959	21.5047	7.7691	21.382	8.4017	7.8089	7.2888	6.876

							203.337					1012.724	1146.574	1178.815	1562.529
Ni	145.568	146.3144	108.5797	390.4179	850.4694	589.7472	9	106.7598	160.7673	106.2346	3	2	9	5	
Pb	< L.D.	0.7915	0.586	1.1608	< L.D.	< L.D.	< L.D.	1.7155	2.3388	1.7705	2.0904	0.8769	1.6641	0.6597	
Rb	0.8414	0.83	1.1103	1.1441	0.2484	0.4561	0.3021	15.7455	6.9663	16.9948	7.4673	7.0625	6.3108	6.7803	
Sb	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	0.0639	< L.D.	< L.D.	
Sc	55.05	65.89	47.05	45.81	23.31	45.71	84.85	38.53	47.31	37.39	22.21	20.76	21.06	15.08	
Sn	0.5012	0.5627	1.0357	0.8165	0.4221	0.558	0.6813	1.5402	1.6966	1.6554	0.8948	0.9372	0.8571	0.9436	
							172.913								
Sr	367.4287	314.8451	401.5141	287.3118	181.1577	222.3178	5	370.1421	426.9755	376.3446	196.661	176.9609	143.3316	103.856	
Ta	0.1303	0.178	0.33	0.1663	0.0486	0.0869	0.1366	1.7607	0.6989	1.7781	0.6987	0.6353	0.6043	0.5536	
Th	0.1818	0.2113	0.3556	0.2245	0.0596	0.1385	0.1798	2.237	0.7935	2.3793	0.9474	0.878	0.7534	1.0624	
U	0.0459	0.0503	0.0937	0.0552	0.0153	0.0289	0.0444	0.5348	0.169	0.5877	0.2461	0.2217	0.1878	0.1906	
							270.400								
V	171.0035	245.908	709.3181	146.016	83.0706	300.2687	3	369.9093	257.906	361.7314	156.1557	145.2547	139.3597	108.6753	
W	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	< L.D.	
Y	8.7325	10.1781	10.2959	7.5554	3.8437	8.4996	13.418	24.943	14.9248	25.7504	11.1959	10.5551	8.5443	8.9211	
Zn	33.0323	40.5406	104.4528	48.0994	117.5548	116.8877	34.807	112.4472	61.7997	113.6664	111.6817	112.375	111.001	112.7347	
Zr	29.3896	37.3924	47.0865	29.1306	12.2737	27.8801	40.9554	177.9659	85.579	176.1766	83.867	74.5281	66.1044	69.1597	
La	2.7139	2.9859	4.1226	2.8023	1.338	2.5132	2.7933	20.0214	7.5066	21.1138	8.1929	7.7927	5.5522	7.6034	
Ce	6.7707	7.9051	10.079	6.8733	3.288	6.3066	8.207	42.0701	17.0843	44.6289	17.7312	16.9548	11.9997	15.8931	
Pr	1.138	1.3468	1.5576	1.1008	0.5274	1.0873	1.4768	5.8426	2.5791	6.1966	2.4552	2.3912	1.7261	2.1214	
Nd	6.0049	7.1889	7.8154	5.5885	2.6993	5.65	7.9265	25.3643	12.0345	27.1186	10.9067	10.5932	7.7049	9.118	
Sm	1.9033	2.2377	2.3046	1.6631	0.8489	1.8184	2.7121	6.1512	3.3621	6.3703	2.6427	2.5686	1.9745	2.0869	
Eu	0.7884	0.9106	0.9477	0.6905	0.3854	0.7136	0.9976	2.0327	1.401	2.1544	0.9401	0.8865	0.7334	0.6804	
Gd	2.0176	2.3754	2.3766	1.7221	0.8681	1.9529	3.0084	5.8052	3.4866	6.1599	2.5334	2.4468	1.9085	2.0053	
Tb	0.3272	0.3838	0.3816	0.2822	0.14	0.3095	0.498	0.8825	0.5498	0.9307	0.3926	0.3773	0.3081	0.3076	
Dy	1.9189	2.2632	2.2314	1.6441	0.8468	1.8498	2.951	5.2029	3.292	5.4196	2.2698	2.1966	1.8092	1.7899	
Ho	0.3712	0.4263	0.423	0.3128	0.1591	0.3459	0.5552	0.9812	0.6195	1.0306	0.4372	0.4228	0.3443	0.3465	
Er	0.8481	0.9808	0.9795	0.7286	0.3854	0.8279	1.2901	2.4224	1.4648	2.5196	1.0746	1.0295	0.8533	0.8747	
Tm	0.1112	0.1319	0.1326	0.0961	0.0534	0.1109	0.1698	0.3214	0.1969	0.3449	0.1484	0.1407	0.1178	0.1192	
Yb	0.639	0.7156	0.7779	0.5508	0.3272	0.6473	0.9675	1.9211	1.1538	2.0119	0.8708	0.8357	0.717	0.7197	
Lu	0.091	0.1018	0.1085	0.08	0.0483	0.0886	0.1359	0.2732	0.1654	0.285	0.129	0.1205	0.1017	0.104	

Table 9.1 - Single mineral phase EMPA analyses. Sample: sub-ophitic gabbro B2. Clinopyroxenes.

Sam ple	B2_1_Cpx rim detail to glass	B2_1_Cpx rim detail to glass	B2_1_Cpx rim detail to glass	B2_1_Cpx rim detail to glass	B2_1_Cpx rim detail to glass	B2_1_Cpx rim detail1 to glass	B2_1_Cpx rim detail1 to glass	B2_1_C px c-r	B2_1_C px c-r	B2_1_C px c-r	B2_1_C px c-r	B2_1_C px c-r	B2_1_C px c-r	B2_1_C px c-r
SiO2	50.4226412	50.583884	50.6191558	50.8469113	51.333663	51.013193	51.3104844	51.2147	50.6947	51.0202	51.1875	51.2147	51.1280	50.8670
TiO 2	1.60106684	1.52628893	1.53755683	1.45663279	1.40439068	1.36648955	1.32756406	465	3841	4733	3674	4647	7846	6666
Al2 O3	3.43137182	3.39515408	3.30098797	3.21510019	3.17577808	3.02883755	3.10023823	615	2916	728	9375	3894	437	3565
FeO	8.33908075	8.47288481	8.33805149	8.65094714	8.67770795	8.27114946	8.50993824	3.69834	3.62901	3.60935	3.71387	3.84942	3.62591	3.68386
Mn o	0.23222222	0.17444444	0.17777778	0.24444444	0.26555556	0.23111111	0.36111111	827	7178	6121	0161	7974	2801	1179
Mg O	14.1274497	14.2825023	14.30998	14.2039947	13.9753412	13.9007589	14.0440354	6.18586	6.20850	6.31555	6.39377	6.46170	6.46479	6.50699
CaO	21.9667688	21.5686224	21.8465165	21.8738006	21.5797381	22.1153158	21.403907	464	8403	1651	5563	6856	4642	4384
Na2 O	0.37333717	0.38893622	0.36501768	0.37957679	0.3712573	0.38789628	0.40349533	0.13777	0.17888	0.11888	0.17333	0.14888	0.18777	0.15111
K2O	0.02428978	0.01795332	0.02006547	0.02112155	0.03168232	0.02217763	0.04435525	15.1500	14.9831	14.9488	15.0283	15.0037	14.9184	15.2550
Cr2 O3	0	0.06043839	0	0.06043839	0	0.02253635	0	22.2689	22.8772	22.7984	22.6569	22.4083	22.4154	22.7266
Tota l	100.518228	100.41067	100.515109	100.892529	100.815114	100.336929	100.505129	154	5084	2998	5663	6775	4141	8278
Fe2 O3	2.19200245	1.88770406	2.0651084	2.12597057	0.75374397	1.3774674	0.87728182	0.28494	0.26622	0.26622	0.25062	0.28390	0.25686	0.25790
FeO	6.36669281	6.77430791	6.47984421	6.73797541	7.99948083	7.03168912	7.7205504	258	3721	3721	4675	264	4293	4229
Wo	45.0062278	44.2355709	44.6520263	44.5707876	44.5216949	45.4909678	44.2986407	0.01161	0.00739	0.03590	0.01900	0.01689	0.01267	0.00316
En	40.2736604	40.7572085	40.6957353	40.2705405	40.1179595	39.7851904	40.4427015	685	2542	6633	9394	7239	2929	8232
Fs	13.3359194	13.5637219	13.3021517	13.7590378	13.9742624	13.2799441	13.7474485	0.28375	0.16185	0.25302	0.25507	0.12394	0.32575	0.21716
Mg#	75.1239994	75.0303946	75.3654219	74.5342491	74.1658561	74.9742572	74.6310935	312	1961	1736	0495	9919	268	8454
								100.268	100.071	100.404	100.702	100.691	100.257	100.708
								52	3559	1275	8508	733	6209	6878
								0.65865	2.03216	1.43108	1.18500	1.03735	0.78430	2.29625
								289	3277	1828	2598	2013	2773	0333
								5.59320	4.37994	5.02784	5.32749	5.52828	5.75907	4.44080
								151	5509	8464	7177	6091	037	3049
								45.7336	46.6382	46.5183	46.2278	45.8792	46.0429	45.9141
								957	13	611	509	9525	4839	8365
								43.2913	42.5004	42.4402	42.6641	42.7423	42.6372	42.8819
								47	031	3089	9973	5293	6996	3496
								9.91598	9.87924	10.0584	10.1825	10.3264	10.3649	10.2609
								796	3439	0465	8079	7536	9056	9782
								81.3634	81.1391	80.8406	80.7318	80.5413	80.4442	80.6916
								944	5597	3606	8056	5413	4811	9825

Sam ple	B2_1_Cpx c-r							B2_1_C	B2_1_C	B2_1_C	B2_1_C	B2_1_C	B2_1_C	B2_1_C
	B2_1_Cpx c-r	B2_1_Cpx c-r	B2_1_Cpx c-r	B2_1_Cpx c-r	B2_1_Cpx c-r	B2_1_Cpx c-r	B2_1_Cpx c-r	px c-r	px c-r	px c-r	px c-r	px c-r	px c-r	px c-r
SiO2	50.9073774	51.0716435	51.2308707	51.0877678	51.0706357	51.1693969	51.0887755	50.8590	50.6977	51.1180	51.6138	50.7622	51.3860	50.8166
TiO								045	6172	0078	2241	5884	6695	7829
2	1.27532195	1.25278615	1.27122453	1.26098099	1.29376034	1.28044373	1.2333234	1.27737	1.28249	1.29785	1.25790	1.30502	1.24868	1.28351
Al2								066	244	7765	792	8249	8725	6795
O3	3.67040888	3.62694759	3.62384322	3.6952439	3.72835726	3.66006095	3.68386118	3.69524	3.74491	3.65695	3.69627	3.63729	3.64039	3.65385
FeO	6.23732774	6.52860889	6.40509745	6.38142442	6.35878066	6.52346258	6.3279028	6.31658	6.37010	6.33922	6.33099	6.30937	6.48846	6.20850
Mn								091	2537	4677	0581	6079	7668	8403
o	0.13777778	0.14333333	0.19	0.14888889	0.12666667	0.14777778	0.11	0.15444	0.09111	0.14888	0.14333			0.11555
Mg								444	1111	8889	3333	0.16	0.11	5556
O	14.8438319	15.1882842	15.0351943	14.9841643	15.0773922	15.1205714	15.0440264	15.0440	15.1745	15.0116	15.0646	14.8497	15.0793	15.0391
CaO	22.9318191	22.4720308	22.7701353	22.7620511	22.7398196	22.5326622	22.5346832	22.4619	22.4780	22.4669	22.8226	22.8408	22.4487	22.1476
Na2								255	939	7813	8255	7198	887	5258
O	0.29638188	0.30262149	0.24646493	0.26206398	0.30366143	0.24542499	0.25790423	0.27454	0.29326	0.31822	0.29950	0.26726	0.30782	0.30054
K2O	0.01478508	0.00316823	0.01795332	0.00633646	0.02534586	0.02534586	0.00739254	0.00739	0.01056	0.01161	0.02428	0.02217	0.01478	0
Cr2								254	0774	6852	9781	7626	5084	0
O3	0.09321853	0.17414451	0.14136437	0.13316933	0.21204656	0.1946321	0.16390072	0.22536	0.19872	0.17414	0.20487	0.10038	0.21716	0.14853
Tota								349	9623	4515	59	9191	8454	5027
l	100.315032	100.589424	100.790784	100.588922	100.72442	100.705146	100.287869	100.090	100.142	100.369	101.253	100.153	100.724	99.5654
Sam								532	8438	3862	4416	992	3731	2512
ple	B2_1_Cpx c-r	B2_1_Cpx c-r	B2_1_Cpx c-r	B2_1_Cpx c-r	B2_1_Cpx c-r	B2_1_Cpx c-r	B2_1_Cpx c-r	B2_1_C	B2_1_C	B2_1_C	B2_1_C	B2_1_C	B2_1_C	B2_1_C
Fe2								px c-r	px c-r	px c-r	px c-r	px c-r	px c-r	px c-r
O3	1.72237896	1.81331137	1.40628758	1.46896191	1.77847722	1.37153895	1.14480783	1.37728	1.98441	1.25433	1.15369	1.74215	1.10082	0.97577
FeO	4.68751218	4.89697135	5.1397044	5.05963631	4.75848727	5.28933673	5.29779211	173	0571	0192	0114	5927	9558	157
Wo	46.8088644	45.6299108	46.3414849	46.390495	46.189777	45.884491	46.108829	5.07728	4.58450	5.21056	5.29288	4.74176	5.49792	5.33049
En	42.1586333	42.9108298	42.5759432	42.4913998	42.6123985	42.8422738	42.8299553	764	8055	4558	7513	4993	911	84
Fs	9.93771432	10.3472823	10.1748595	10.1515788	10.0816367	10.3688343	10.1062673	46.0084	45.7851	45.9653	46.3243	46.6987	45.7655	45.7001
Mg#	80.924355	80.5714437	80.7114603	80.7161771	80.8675942	80.5137787	80.9085975	247	241	8392	1682	2667	3293	9948
								42.8751	43.0062	42.7331	42.5453	42.2436	42.7739	43.1781
								217	4265	9124	2404	5842	415	1783
								10.0988	10.1276	10.1232	10.0302	10.0687	10.3249	9.99944
								237	7332	6767	6362	8623	0592	77
								80.9362	80.9393	80.8476	80.9222	80.7525	80.5553	81.1961
								44	4329	2415	035	9854	1066	1607



SiO2	50.8338103	50.9940454	50.4307033	50.497216	50.1787614	50.0729458	49.8955788	50.9194706	50.4710140	50.5687674	50.1283730	50.4528742	49.9741846	50.3037246
									2	7	6	50.4528742	2	1
TiO2	1.41975601	1.28863857	1.42282907	1.4996557	1.48941215	1.47814424	1.42282907	1.53243506	1.46790069	1.44434052	1.48736343	1.59184764	1.62155393	1.63896797
									2	7	7	5	9	4
Al2O3	3.10541219	2.98744585	3.56382525	3.55658171	3.58969506	3.55554691	3.60107778	3.70973099	3.62487800	3.56589483	3.59797340	3.47897227	3.57831234	3.59900819
									8	8	4	1	7	7
FeO	8.51302603	8.37613418	7.35201849	7.50537853	7.43127166	7.51258336	7.61756809	7.75240141	7.66594339	7.27482383	7.21306811	7.69579199	7.89032251	7.90576144
									9	8	8	7	6	6
Mno	0.18777778	0.16	0.16888889	0.20555556	0.16777778	0.13333333	0.16777778	0.20777778	0.12444444		0.16777777	0.17666666	0.17333333	0.26333333
									4	0.1	8	7	3	3
MgO	14.1578714	14.1215616	14.4375549	14.7653243	14.7103689	14.3404017	14.3698421	14.3305882	14.3089986	14.5101744	14.5837753	14.1470766	14.4041891	14.2687634
									4	6	8	3	5	7
CaO	22.2759891	21.8202428	22.0779264	22.4467677	22.4538413	22.0152739	22.1324947	21.8303481	22.1638209	22.7206196	22.5255885	22.1496736		22.0880316
									6	4	2	3	22.4275677	7
Na2O	0.37229723	0.359818	0.39309596	0.29950169	0.32965984	0.359818	0.32757997	0.37333717	0.29742181	0.31718060	0.28494257	0.38165666	0.36085793	0.48981004
									3	5	6	2	4	9
K2O	0	0.00316823	0.01372901	0.01795332	0.01584116	0.00739254	0.01795332	0.02640194	0.01478508	0.00316823	0.01372900	0.01584116	0.01795331	0.01795331
									4	2	7	1	6	6
Cr2O3	0	0.03790204	0.02560949	0.08092598	0.03073138	0	0.05531649	0.07785284	0.05019459	0.07273094		0.02560948	0.07785284	0.07068218
									5	4	0.14341313	7	2	5
Total	100.86594	100.111055	99.8605713	100.793934	100.366629	99.4754399	99.5527015	100.682491	100.139207	100.504969	100.002591	100.090400	100.448274	100.575354
Sampl	B2_4_Cpx c-	B2_4_Cpx c-	B2_6_Cpx c-	B2_6_Cpx c-	B2_6_Cpx c-	B2_6_Cpx c-	B2_6_Cpx c-	B2_6_Cpx c-	B2_6_Cpx c-	B2_6_Cpx c-	B2_6_Cpx c-	B2_6_Cpx c-	B2_6_Cpx c-	B2_6_Cpx c-
e	r	r	r	r	r	r	r	r	r	r	r	r	r	r
Fe2O3	2.38294325	1.31341262	2.00963999	2.71908949	3.11664416	2.11373195	2.61148728	1.07486135	1.56867628	2.23612825	2.46812808	1.67694096	3.25965828	2.84121891
									5	9	2	5	9	8
FeO	6.36882745	7.19431104	5.54372229	5.05871186	4.62688093	5.61062408	5.26772308	6.78522933	6.25443120	5.26273100	4.99221929	6.18686205	4.95724609	5.34920137
									5	8	2	2	4	8
Wo	45.1903129	44.8440778	45.4126642	45.446653	45.5279509	45.407757	45.4790064	45.014199	45.6091188	46.2155004	46.0051668		45.5093034	
									7	6	3	45.6398448	3	45.0787932
En	39.96292	40.3811927	41.3202791	41.5950816	41.5013207	41.1545797	41.0850189	41.1153301	40.9701155	41.0668149	41.4430135	40.5596717	40.6684598	
									9	6	8	6	6	40.51842
Fs	13.4800238	13.4365413	11.8038488	11.8609381	11.761127	12.0946584	12.2178641	12.4773818	12.3132019	11.5501684	11.4987026	12.3773734	12.4971537	12.5938152
									9	2	5	6	7	2
Mg#	74.776794	75.033246	77.7806258	77.8117822	77.9185383	77.2867015	77.0784179	76.7181369	76.8910748	78.0485925	78.2804497	76.6186922	76.4939160	76.2882974
									3	4	7	4	5	7

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: sub-ophitic gabbro B2. Clinopyroxenes.

Samp le	B2_6_Cpx c-r	B2_6_Cpx c-r	B2_6_Cpx c-r	B2_6_Cpx c-r	B2_6_Cpx c-r	B2_6_Cpx c-r	B2_6_Cpx c-r	B2_Cpx_ c-r	B2_Cpx_c -r	B2_Cpx_c -r	B2_Cpx_c -r	B2_Cpx_c -r	B2_Cpx_c -r	B2_Cpx_c -r	B2_Cpx_c -r	B2_Cpx_c -r	B2_Cpx_c -r
SiO2	49.85627 58	50.00441 76	49.87139 23	50.11325 65	50.22914 98	50.54558 88	50.10620 22	50.767	50.617	50.501	50.327	50.21	49.917	49.773	48.81	49.688	49.539
TiO2	1.637943 62	1.662528 14	1.614383 45	1.665601 2	1.530386 35	1.410536 81	1.373660 03	1.067	1.16	1.2	1.287	1.28	1.391	1.536	1.573	1.628	1.567
Al2O 3	3.595903 82	3.472763 52	3.572103 59	3.452067 67	3.257526 68	3.010211 28	2.972958 75	3.157	3.244	3.399	3.37	3.484	3.381	3.57	3.448	3.471	3.38
FeO	8.128082 04	8.172340 3	7.938697 83	8.276295 77	8.358636 73	8.299968 79	8.375104 92	6.175	6.266	6.258	6.547	7.019	7.25	7.33	7.811	7.814	7.852
Mno	0.208888 89	0.137777 78	0.198888 89	0.132222 22	0.226666 67	0.211111 11	0.265555 56	0.111	0.093	0.16	0.091	0.133	0.138	0.13	0.159	0.169	0.159
MgO	14.13530 05	14.10684 15	14.15492 74	14.16179 68	14.34236 44	14.29231 58	14.30114 79	15.76	15.593	15.602	15.295	15.104	14.969	15.046	14.746	14.668	14.627
CaO	21.88794 79	22.57106 21	21.68382 21	22.19211 56	21.83540 07	21.70504 31	22.02032 66	22.012	21.847	22.301	22.053	21.891	21.507	21.709	21.539	21.447	21.422
Na2O	0.397255 71	0.385816 41	0.388936 22	0.381656 66	0.353578 38	0.413894 69	0.325500 1	0.254	0.289	0.236	0.27	0.271	0.33	0.294	0.272	0.316	0.347
K2O	0.019009 39	0.019009 39	0.010560 77	0.020065 47	0.008448 62	0.004224 31	0.024289 78	0.01	0.003	0.015	0.008	0.008	0.009	0	0.015	0	0
Cr2O 3	0.005121 9	0.037902 04	0.017414 45	0.032780 14	0.053267 73	0.067609 05	0 0	0.064	0.094	0.051	0	0	0.048	0.119	0	0.043	0.081
Total	99.86660 77	100.5325 57	99.43371 26	100.3950 78	100.1421 58	99.89289 47	99.76474 57	99.313	99.112	99.672	99.248	99.4	98.892	99.388	98.373	99.201	98.893
Fe2O 3	2.651670 75	3.230837 62	2.100449 09	2.747913 07	2.576478 93	2.193492 22	3.091242 69	2.31151 304	2.065974 698	2.712361 224	2.330504 287	2.460923 215	2.457098 631	2.751598 056	3.756437 928	2.405753 975	2.708693 364
FeO	5.742079 49	5.265197 04	6.048690 61	5.803693 33	6.040292 62	6.326240 34	5.593570 71	4.09507 512	4.407013 218	3.817387 503	4.449986 581	4.804634 192	5.039075 596	4.854081 77	4.430915 789	5.649276 083	5.414687 885
Wo	45.01934 92	44.91330 52	45.24619 59	44.60258 63	44.46040 05	44.89813 66	44.7217 276	44.7217 184	44.62494 368	45.21576 441	45.06752 386	44.78911 795	44.26133 142	44.39706 045	44.27739 199	44.19598 774	44.14109 366
En	40.45294 17	39.83292 87	40.79413 47	40.17469 82	40.76341 19	40.73485 91	40.57199 44	44.5518 948	44.31659 978	44.01459 441	43.49070 015	42.99815 133	42.86356 858	42.81405 406	42.17757 668	42.05698 078	41.93621 535
Fs	13.04910 09	12.94512 79	12.83473 18	13.17096 86	13.32700 85	13.27050 73	13.32886 49	9.79251 796	9.990209 404	9.903744 733	10.44327 483	11.20935 176	11.64611 279	11.70083 132	12.53318 587	12.56863 303	12.62878 913
Mg#	75.61008 83	75.47251 89	76.06749 37	75.31014 34	75.36161 04	75.42742 85	75.27151 69	81.9806 352	81.60413 113	81.63195 512	80.63692 721	79.32140 18	78.63478 102	78.53644 699	77.09191 888	76.99131 936	76.85551 526



Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: sub-ophitic gabbro B2. Olivines.

Sample	B2_Ol_c-r	B2_Ol_c-r	B2_Ol_c-r	B2_Ol_c-r	B2_Ol_c-r	B2_Ol_c-r	B2_Ol_c-r	B2_Ol_c-r	B2_Ol_c-r	B2_Ol_c-r
SiO <sub>2</sub>	37.907	37.303	37.77	38.015	37.282	37.171	37.433	37.614	37.452	37.394
TiO <sub>2</sub>	0.005	0.031	0.021	0.025	0.031	0.023	0.017	0.035	0	0.037
Al <sub>2</sub> O <sub>3</sub>	0.01	0.026	0	0.024	0.036	0.06	0.015	0.05	0.016	0.034
Cr <sub>2</sub> O <sub>3</sub>	0.016	0.054	0	0.014	0.061	0	0.028	0	0	0
FeO	27.305	26.751	27.245	26.69	26.509	26.964	27.222	26.879	26.681	27.153
MnO	0.442	0.474	0.419	0.413	0.512	0.404	0.427	0.504	0.471	0.465
MgO	34.526	34.896	34.872	34.706	34.733	35.227	34.926	35.211	34.953	35.003
CaO	0.218	0.271	0.293	0.258	0.211	0.202	0.225	0.192	0.217	0.254
Na <sub>2</sub> O	0.013	0	0	0.002	0	0	0.025	0	0.003	0.003
NiO										
Total	100.442	99.806	100.62	100.147	99.375	100.051	100.318	100.485	99.793	100.343
Fo	68.92147	69.55261	69.19876	69.53269	69.61229	69.64224	69.2432	69.62006	69.64398	69.31357
Fa	30.57721	29.91061	30.32884	29.99719	29.80468	29.90397	30.27581	29.81375	29.82282	30.16326

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: sub-ophitic gabbro B2. Feldspars.

Sample	B2_2_small Pl	B2_2_small Pl	B2_2_small Pl	B2_2_small Pl	B2_2_small Pl	B2_2_small Pl	B2_5_Pl c-	B2_5_Pl c-	B2_5_Pl c-r	B2_5_Pl c-r	B2_5_Pl c-r	B2_5_Pl c-r	B2_5_Pl c-r	B2_5_Pl c-r
	r-r	r-r	r-r	r-r	r-r	r-r	r	r						
SiO2	49.7665845	49.7182117	49.1206055	48.8797491	49.2334755	49.6446446	48.00097	48.68827	48.640908	47.561589	47.837717	48.181366	46.847081	47.549495
TiO2	0.13521486	0.06453436	0.10448421	0.0932163	0.06043694	0.1024355	0.080924	0.035852	0.0645343	0.0696561	0.0419985	0.0727292	0.0501933	0.0573638
Al2O3	32.2244722	32.5669885	32.63425	32.803956	32.1365148	32.5038662	33.09162	33.49623	33.302725	33.246847	33.574876	33.903940	33.378265	33.435179
Cr2O3	0	0	0	0.00819504	0	0.03892642	0.018438	0.059414	0.0286826	0.0133169			0.0542921	
FeO	0.7719465	0.82546813	0.73077602	0.70607373	0.85737525	0.87898975	0.704015	0.707103	0.8131169	0.6381424	0.8059121	0.6350546	0.7637124	0.6793129
Mno	0.07888889	0.01555556	0	0.03222222	0.00333333	0.01666667	0.018888				0.0355555			
MgO	0.06280611	0.07752629	0.07458226	0.03140306	0.06182477	0.0863584	0.028459	0.065750	0.1118733	0.0618247	0.0716382	0.0834143	0.0618247	0.0765449
CaO	15.4084686	15.7035416	16.0137724	16.2461929	16.0026567	15.7530573	16.53823	16.47659	16.532171	16.941433	17.163748	16.969728	17.177895	17.262779
Na2O	2.93470055	2.82446729	2.58424198	2.74335225	2.77247047	2.86606474	2.385614	2.119390	2.3190581	2.1401891	2.0892322	1.9457210	1.9488408	1.9717194
K2O	0.12039283	0.10983205	0.11088813	0.11405636	0.09293481	0.11088813	0.107719	0.090822	0.0834301	0.0844861	0.0823740	0.0876544	0.0908226	0.0707571
SrO							9	66	17	94	39	26	59	88
BaO														
Total	101.503475	101.906125	101.373601	101.658417	101.221023	102.001898	100.9748	101.7394	101.89650	100.75748	101.70305	101.87960	100.37292	101.10315
Ab %	25.4557333	24.402461	22.4591285	23.256529	23.7436244	24.6137254	20.57357	18.78205	20.148836	18.517555	17.967056	17.096410	16.944802	17.059797
An %	73.8571436	74.9731743	76.9067727	76.1072692	75.73269	74.7596773	78.81517	80.68835	79.374212	81.001462	81.566828	82.396820	82.535600	82.537382
Or %	0.68712308	0.62436462	0.63409873	0.63620187	0.52368559	0.6265973	0.611248	0.529588	0.4769504	0.4809825	0.4661145	0.5067693	0.5195967	0.4028203
Sample							B2_5_Pl c-	B2_5_Pl c-						
	B2_5_Pl c-r	B2_5_Pl c-r	B2_5_Pl c-r	B2_5_Pl c-r	B2_5_Pl c-r	B2_5_Pl c-r	r	r	B2_5_Pl c-r	B2_1_pl r-r	B2_1_pl r-r	B2_1_pl r-r	B2_1_pl r-r	B2_1_pl r-r
SiO2	47.9515952	47.9949292	49.6103805	48.8001354	48.389974	49.3030114	49.15083	49.64262	53.502378	47.332825	47.530348	49.731312	48.505867	51.103892
TiO2	0.05326646	0.05736388	0.06453436	0.04814468	0.0727292	0.05326646	0.087070	0.083997	0.1188251	0.0542908	0.0758022	0.0573638	0.0542908	0.0778509

Al2O3	33.6545554	33.4610492	32.2575855	32.2824206	32.3807258	32.0371747	32.22654 18	32.39831 73	29.479167 77	34.277500 44	33.833574 47	34.059159 23	33.211664 2	31.551857 08
Cr2O3	0	0.10551109	0.02868263	0.0747797	0.05429211	0	0	0	0.0102437 95	0.0133169 33	0.0133169 33	0	0.0030731 38	0
FeO	0.57124041	0.69578111	0.6906348	0.68240071	0.65872768	0.67210809	0.793561 0.0111111	31 0.018888	18 0.0433333	0.750332 0.0333333	39 0.0155555	86 0.0144444	24 0.0077777	28
Mno	0.01333333	0.04888889	0.02444444	0.03555556	0.02222222	0	11	89	33	33	56	44	78	0
MgO	0.0647688	0.06182477	0.08733975	0.07752629	0.09519051	0.10009724	0.092246 48	0.080470 33	0.0971532 04	0.0441605 47	0.0824330 21	0.0804703 3	0.0510299 66	0.0951905 13
CaO	17.3173483	16.8171389	15.9784041	15.5620682	16.1986983	15.8935201	15.65099 43	15.68737 32	12.498159 74	17.827662 85	16.599876 17.390106	16.796928 3	14.502028 46	66
Na2O	1.93324178	2.08923225	2.81302799	2.77663021	2.60712058	2.42305183	2.671596 63	2.907662 2	4.8939407 4	1.8178088 42	2.0653137 08	2.3408968 56	2.3388169 83	3.6189787 02
K2O	0.08131796	0.08448619	0.1077199	0.12356106	0.13940222	0.13200968	0.109832 05	0.120392 83	0.2376174 21	0.0665328 78	0.0749814 97	0.0813179 62	0.0792058 07	0.1594676 91
SrO														
BaO														
Total	101.640668	101.416205	101.662754	100.463222	100.619083	100.61424	100.7937 92	101.6952 09	101.61983 01	102.21776 45	101.93366 07	103.76972 45	101.80104 5	101.85342 21
Ab %	16.7288685	18.2657442	24.0151039	24.2340949	22.3782337	21.4568411	23.45026 75.91540	01 74.37415	09 57.762126	64 84.106884	43 81.963728	62 79.300607	03 79.517273	35 68.274066
An %	82.8081336	81.2482421	75.3798082	75.0563233	76.834455	77.7739913	61 0.634333	86 0.679631	81 1.3076041	26 0.3737440	78 0.4207987	99 0.4625503	81 0.4464671	62 0.8939240
Or %	0.46299786	0.48601369	0.60508788	0.70958183	0.7873113	0.76916756	98	25	05	91	88	81	66	32

Sample	B2_1_pl r-r	B2_1_pl r-r	B2_1_pl r-r	B2_1_pl r-r	B2_1_pl r-r	B2_1_pl r-r	B2_1_pl1 r-r	B2_1_pl1 r-r	B2_1_pl1 r-r	B2_1_pl1 r-r	B2_1_pl1 r-r	B2_1_pl1 r-r	B2_1_pl1 r-r	B2_1_pl1 r-r
SiO2	51.1643581	49.3402988	50.6292335	47.7308941	48.5401314	47.8377174	47.2290258	46.4994021	48.62377612	48.90796656	48.41113716	48.19043607	48.53307704	48.72556064
TiO2	0.09628937	0.09526501	0.0727292	0.05838823	0.08297275	0.03994984	0.07068049	0.05019339	0.060436944	0.066583074	0.058388234	0.055315169	0.090143239	0.053266459
Al2O3	31.2890198	30.53983	31.2196887	33.2561603	33.2416732	33.5945374	33.8687574	32.968488	33.02747114	32.91364397	33.06161929	32.64252834	32.97987068	33.23649922
Cr2O3	0	0	0.01229255	0.02765825	0.00819504	0	0.01741445	0.03585328	0	0.007170656	0	0.038926421	0.071706565	0.028682626
FeO	0.66078621	0.77606355	0.70401521	0.8532582	0.69475185	0.80591215	0.64946432	0.76474167	0.624762036	0.711220044	0.681371446	0.705044472	0.692693328	0.665932516
Mno	0.04555556	0.04666667	0.00333333	0.01888889	0.03333333	0.02666667	0.00777778	0	0	0	0.015555556	0.002222222	0.013333333	0.041111111
MgO	0.08537706	0.10500397	0.08733975	0.04023517	0.053974	0.05004862	0.04514189	0.04121651	0.101078586	0.082433021	0.077526294	0.078507639	0.048085929	0.067712839
CaO	14.7890175	14.6687651	14.8102385	16.8949493	16.7534759	17.2233696	17.2557063	17.0394542	16.20577198	16.32299275	16.62008678	16.56349744	16.71103393	16.38362418
Na2O	3.48586684	3.37979333	3.31635721	2.16826741	2.32425787	2.10899104	1.96963956	1.96859962	2.378334567	2.418892087	2.453209988	2.338816983	2.362735521	2.294099718

K2O	0.16263592	0.16580416	0.14362653	0.09399089	0.0982152	0.06442072	0.06442072	0.05914034	0.088710504	0.108775975	0.081317962	0.097159123	0.087654426	0.095046968
SrO														
BaO														
Total	101.778906	99.1174907	100.998854	101.142691	101.830981	101.751613	101.178029	99.4270891	101.1103419	101.5396781	101.4602127	100.7124539	101.590334	101.5915363
Ab %	29.6283842	29.1490786	28.6016036	18.7464747	19.9560913	18.0734207	17.0567928	17.2328602	20.87719681	21.01455277	20.98377959	20.23943702	20.27240467	20.10562857
An %	69.4620675	69.9100241	70.5833622	80.7188322	79.4890505	81.5633312	82.5761377	82.4264997	78.61043001	78.36365005	78.55855554	79.20734297	79.23274244	79.34627657
Or %	0.90954828	0.94089726	0.81503424	0.53469317	0.55485817	0.36324808	0.36706951	0.34064017	0.512373178	0.621797178	0.457664873	0.55322001	0.494852892	0.548094869
Sample	B2_1_pl1 r-r	B2_1_pl1 r-r	B2_1_pl1 r-r	B2_1_pl1 r-r	B2_1_pl1 r-r	B2_2_PIC-r	B2_2_PIC-r	B2_2_PIC-r	B2_2_PIC-r	B2_2_PIC-r	B2_2_PIC-r	B2_2_PIC-r	B2_2_PIC-r	B2_2_PIC-r
SiO2	48.8283529	48.4716032	47.6674047	47.0889461	48.186405	48.5925353	47.47996	48.1914438	48.27811185	48.88680344	48.67920333	48.27307301	47.71476979	48.83439953
TiO2	0.05531517	0.08502146	0.06453436	0.05531517	0.0819484	0.07068049	0.06658307	0.06760743	0.072729204	0.071704849	0.074777914	0.089118884	0.061461299	0.074777914
Al2O3	33.1557854	33.3751614	33.2892736	34.2930223	33.217873	34.085029	33.4175879	33.1516462	33.64938141	33.0605845	32.60320623	32.92916585	32.93020065	32.79671242
Cr2O3	0	0	0.06863343	0.04609708	0	0.04917022	0.01843883	0	0	0	0.003073138	0	0	0
FeO	0.74415643	0.70092742	0.75239052	0.74827348	0.72562971	0.7420979	0.70916152	0.69681038	0.726658974	0.685488494	0.728717498	0.76577093	0.699898162	0.727688236
Mno	0	0	0.00222222	0.02111111	0	0.00555556	0.02666667	0.02444444	0.008888889	0.018888889	0.003333333	0.008888889	0.007777778	0.013333333
MgO	0.09420917	0.07458226	0.05986208	0.07850764	0.07065688	0.06575015	0.06673149	0.08243302	0.066731494	0.099115895	0.073600912	0.093227822	0.087339749	0.068694185
CaO	16.4260662	16.9363808	16.8757493	17.7983577	16.3846347	17.1657697	16.6291815	16.6221078	17.03339106	16.68577083	16.52610806	17.01621215	16.60492892	16.63928673
Na2O	2.30449908	2.207785	2.16410767	1.87916509	2.20882493	1.96132007	2.09859167	2.28890004	2.160987857	2.302419209	2.378334567	2.437610942	2.361695584	2.534325028
K2O	0.06864503	0.08659835	0.0559721	0.06758896	0.10243951	0.0749815	0.06864503	0.07286934	0.092934814	0.087654426	0.098215201	0.074981497	0.096103046	0.099271278
SrO														
BaO														
Total	101.677029	101.93806	101.00015	102.076385	100.978412	102.81289	100.581548	101.198263	102.0898155	101.8984305	101.1685702	101.68805	100.564175	101.7884887
Ab %	20.1676391	18.9936333	18.7750921	15.9806306	19.4947049	17.060197	18.5174461	19.865078	18.57349147	19.88159236	20.5466479	20.50115843	20.35800545	21.48730906
An %	79.4370855	80.516167	80.9053961	83.6411735	79.9104074	82.5106602	81.0840117	79.7188003	80.90093709	79.62038083	78.89506339	79.08390642	79.09691415	77.95888777
Or %	0.39527536	0.49019964	0.31951183	0.3781959	0.59488767	0.42914275	0.39854225	0.41612174	0.525571443	0.498026808	0.558288707	0.414935147	0.545080399	0.553803167
Sample	B2_2_PIC-r	B2_2_PIC-r	B2_2_PIC-r	B2_4_PIC-r	B2_4_PIC-r	B2_4_PIC-r	B2_4_PIC-r	B2_4_PIC-r	B2_4_PIC-r	B2_4_PIC-r	B2_4_PIC-r	B2_4_PIC-r	B2_4_PIC-r	B2_4_PIC-r
SiO2	49.1377376	50.8660589	51.5211078	49.2929338	48.8726947	49.0419996	49.309058	47.7772514	47.75508049	48.54920132	48.81222865	48.82331409	48.24787882	48.29625166



BaO														
Total	102.567826	101.951276	101.500255	100.206694	100.106743	100.311932	100.450117	101.387974	101.7356385	101.3834062	101.6808628	101.5796766	101.7380666	101.6019709
Ab %	21.4868119	17.2217856	16.6063672	17.3755191	16.7049075	18.301261	18.7632233	18.6092587	19.07547361	20.84813202	21.39078563	21.49966917	22.33256621	21.15975258
An %	77.947615	82.2892966	82.8519174	82.0759686	82.7726121	81.1528126	80.7136321	80.8904661	80.25368042	78.65260879	78.04320819	77.90398792	76.84173367	78.27278512
Or %	0.56557313	0.48891775	0.54171542	0.54851233	0.52248037	0.54592638	0.52314457	0.50027525	0.67084596	0.499259182	0.56600618	0.596342906	0.82570012	0.567462298
Sample	B2_5_Pl c-r	B2_5_Pl c-r	B2_5_Pl c-r	B2_5_Pl c-r	B2_5_Pl c-r	B2_5_Pl c-r	B2_5_Pl c-r	B2_5_Pl c-r	B2_5_Pl c-r	B2_5_Pl c-r	B2_5_Pl c-r	B2_5_Pl c-r	B2_5_Pl c-r	B2_6_Pl r-r
SiO2	49.0147899	49.6607689	49.1447919	48.9795181	49.0863414	48.5794343	48.8011432	49.0520773	49.32719786	49.44611443	49.42293578	50.2916314	51.07063571	48.54113918
TiO2	0.10550856	0.08911888	0.0932163	0.06248565	0.07170485	0.09833808	0.06658307	0.05633952	0.104484208	0.087070174	0.061461299	0.074777914	0.075802269	0.085021464
Al2O3	32.6249369	32.3972825	32.5576754	32.4065957	32.4624745	32.8370693	32.8691479	32.1965328	32.44281339	32.43867422	32.12306252	31.85815565	31.36352485	32.11064501
Cr2O3	0.02356073	0	0.02048759	0.03278014	0.01024379	0	0	0.02253635	0	0	0	0	0	0
FeO	0.59903049	0.76268314	0.80385362	0.73077602	0.78944396	0.75959536	0.72562971	0.75753683	0.72974676	0.720483402	0.71430783	0.629908346	0.655639896	0.754449048
Mno	0.04333333	0.02444444	0.05111111	0	0.02222222	0.10444444	0	0.03555556	0.003333333	0.037777778	0	0	0	0.011111111
MgO	0.10892935	0.06182477	0.09224648	0.05888073	0.11481742	0.07163822	0.10402262	0.09813455	0.063787457	0.084395712	0.075563603	0.08047033	0.052992657	0.074582257
CaO	15.8935201	15.8632044	15.9359621	15.910699	16.1037091	16.4311188	16.0511618	16.0673302	15.63684701	15.63078387	15.70253107	14.97596437	14.5333549	16.05924601
Na2O	2.64871803	2.67575638	2.56344325	2.41785215	2.4792084	2.40537291	2.60920045	2.60712058	2.236903215	2.858785188	2.706954472	3.168686237	3.530584108	2.665357016
K2O	0.10666382	0.1214489	0.12778537	0.12356106	0.10455167	0.10032736	0.0887105	0.10560774	0.101383433	0.122504981	0.100327356	0.124617136	0.158411614	0.122504981
SrO														
BaO														
Total	101.168991	101.656532	101.390573	100.723148	101.244717	101.387339	101.315599	100.998771	100.6464967	101.4265898	100.9071439	101.2042114	101.440946	100.4240561
Ab %	23.0289936	23.2236639	22.3808349	21.4131963	21.6583056	20.8235231	22.6150554	22.5618818	20.43853183	24.69372836	23.6411252	27.49067021	30.26399906	22.93705835
An %	76.3608117	76.0827668	76.885081	77.8667822	77.740722	78.604994	76.8790299	76.8367755	78.9519571	74.61001236	75.78234912	71.79795911	68.8425332	76.3692792
Or %	0.61019475	0.69356932	0.73408409	0.72002153	0.60097236	0.57148293	0.50591477	0.60134273	0.609511071	0.696259279	0.576525685	0.71137068	0.893467739	0.693662446
Sample	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r
SiO2	49.2929338	47.0516588	47.4829833	47.1030549	47.5454648	47.9012068	48.3758653	48.5250149	48.7981199	48.81928302	48.89990442	48.79106553	49.12866765	48.77594901
TiO2	0.09219195	0.04507162	0.05941259	0.07887533	0.0727292	0.03994984	0.06453436	0.06453436	0.077850979	0.061461299	0.060436944	0.055315169	0.058388234	0.110630338
Al2O3	32.7056507	32.2141243	33.0222972	32.7584251	33.297552	33.1847596	32.4169436	32.549397	32.5928583	32.52145762	32.74600759	32.18411528	32.37865626	32.0164789

Cr2O3	0.0747797	0	0.02253635	0.01843883	0	0	0	0	0	0	0	0	0	0
FeO	0.65975694	0.73592233	0.69269333	0.68857628	0.69681038	0.6175572	0.64123023	0.6690203	0.686517756	0.70916152	0.73489307	0.632996132	0.71945414	0.680342184
Mno	0.01888889	0	0.06888889	0.00333333	0.03555556	0.01111111	0.00333333	0.08222222	0.04888889	0	0	0.03	0.06	0.01111111
MgO	0.07065688	0.08733975	0.0755636	0.08047033	0.05691804	0.07948898	0.09813455	0.08439571	0.067712839	0.06967553	0.091265131	0.084395712	0.065750148	0.057899384
CaO	16.2593297	16.4250557	17.1061488	17.0162122	17.0384437	16.693855	16.5503606	16.3260243	16.43415038	16.40888728	16.15423526	16.03297239	16.08046701	15.98143567
Na2O	2.48024833	2.77975002	2.05907409	2.13602938	1.94884083	2.28058054	2.37001508	2.54368446	2.459449607	2.583202039	2.636238796	2.608160513	2.662237206	2.572802675
K2O	0.11722459	0.10877597	0.08448619	0.10877597	0.07920581	0.11616852	0.11088813	0.10032736	0.109832052	0.099271278	0.105607743	0.103495588	0.126729291	0.129897523
SrO														
BaO														
Total	101.771661	99.4476984	100.674084	99.9921916	100.77152	100.924678	100.631305	100.944621	101.2753807	101.2723996	101.4285889	100.5225163	101.2803499	100.3365468
Ab %	21.4883518	23.3047714	17.8004658	18.3969694	17.0705258	19.6906496	20.451063	21.8689934	21.17798842	22.04835044	22.66257783	22.608713	22.88778331	22.3924076
An %	77.8433985	76.0951831	81.7189635	80.9866	82.4729754	79.6493941	78.919341	77.5634659	78.19972852	77.39413855	76.74006728	76.80098335	76.3953373	76.86370486
Or %	0.66824971	0.60004551	0.48057065	0.61643055	0.45649882	0.6599563	0.62959597	0.56754072	0.622283063	0.557511006	0.597354893	0.590303645	0.716879389	0.743887545
Sample	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r	B2_6_Pl r-r
SiO2	49.0530851	49.6587534	48.7537781	49.3292134	49.4249513	49.4461144	48.3980362	48.2257079	49.33324446	48.70338975	48.88378014	48.10175253	48.59958969	48.81122088
TiO2	0.0727292	0.07477791	0.07170485	0.0409742	0.06965614	0.07580227	0.09116759	0.06863178	0.081948399	0.074777914	0.050193394	0.061461299	0.077850979	0.089118884
Al2O3	32.5173185	32.7884341	32.3258818	32.0692533	32.5287012	32.3465777	32.1520367	31.7257022	32.4562657	32.195498	32.2906989	32.07442728	32.32174268	32.63942396
Cr2O3	0.0901454	0.04097518	0.03892642	0	0.00819504	0.0051219	0.02765825	0.03278014	0.003073138	0	0.002048759	0	0.043023939	0
FeO	0.6906348	0.69372259	0.67931292	0.67622514	0.74724421	0.64431801	0.64843506	0.73901012	0.681371446	0.64225949	0.821351078	0.712249306	0.816204768	0.869726393
Mno	0.01333333	0	0.02222222	0.02444444	0	0.04111111	0	0	0.037777778	0.035555556	0.014444444	0	0.035555556	0.035555556
MgO	0.07360091	0.06967553	0.06967553	0.08439571	0.08341437	0.08341437	0.09911589	0.09028379	0.093227822	0.103041277	0.102059931	0.10892935	0.06967553	0.079488985
CaO	16.1299827	15.9986146	15.8338992	16.0349934	16.0521723	15.5741945	15.7672046	15.910699	16.12695111	15.40038442	16.09461435	15.83187813	16.19263516	16.39473995
Na2O	2.59672121	2.46880903	2.66951676	2.68927555	2.74751199	2.72151358	2.84734589	2.70175479	2.662237206	2.739192501	2.696555108	2.678876189	2.633118987	2.624799496
K2O	0.12989752	0.14151438	0.12250498	0.11933675	0.11300028	0.12250498	0.13200968	0.14151438	0.108775975	0.133065756	0.124617136	0.114056362	0.138346143	0.103495588
SrO														
BaO														
Total	101.367449	101.935277	100.587423	101.068112	101.774847	101.060673	100.16301	99.6360841	101.584873	100.0271647	101.0803632	99.68363044	100.9277434	101.6475697
Ab %	22.3939712	21.6509708	23.2133067	23.1261115	23.4984713	23.8552947	24.4466771	23.3168648	22.86051541	24.16151608	23.10189492	23.28925873	22.55888258	22.33370569

An %	76.8689417	77.532442	76.0857705	76.1986556	75.8656246	75.4381598	74.8075657	75.8795409	76.52489619	75.06619458	76.19563444	76.05831019	76.66124016	77.08686749
Or %	0.73708702	0.81658714	0.70092284	0.67523294	0.63590414	0.7065455	0.74575725	0.80359437	0.614588398	0.772289333	0.702470636	0.652431082	0.779877262	0.579426821

Sample	B2_P1_c-r	B2_P1_c-r	B2_P1_c-r	B2_P1_c-r	B2_P1_c-r	B2_P1_c-r	B2_P1_c-r	B2_P1_c-r	B2_P1_c-r	B2_P1_c-r	B2_P1_c-r	B2_P1_c-r	B2_P1_c-r	B2_P1_c-r
SiO2	47.371	47.879	47.755	48.581	46.881	47.486	47.095	47.666	47.662	47.464	46.356	46.403	46.361	46.814
TiO2	0.069	0.086	0.046	0.076	0.052	0.068	0.061	0.054	0.052	0.051	0.065	0.051	0.036	0.056
Al2O3	32.043	32.278	32.368	31.918	32.326	32.101	32.075	32.127	32.226	31.825	32.555	32.69	33.009	32.713
Cr2O3	0.008	0	0.013	0.003	0.091	0	0.057	0	0.031	0	0.026	0.01	0	0.05
FeO	0.72	0.631	0.747	0.708	0.601	0.682	0.772	0.68	0.718	0.722	0.651	0.689	0.67	0.748
Mno	0	0	0	0	0	0.046	0.01	0	0	0.022	0	0.014	0.012	0.026
MgO	0.076	0.057	0.071	0.076	0.059	0.092	0.051	0.097	0.083	0.05	0.08	0.07	0.088	0.073
CaO	16.474	15.961	16.15	15.832	16.291	16.373	16.013	16.021	16.242	16.618	16.675	16.586	17.084	16.853
Na2O	2.33	2.343	2.258	2.493	2.149	2.274	2.222	2.186	2.226	2.149	1.791	1.79	1.79	1.763
K2O	0.054	0.079	0.074	0.084	0.064	0.077	0.082	0.073	0.08	0.071	0.061	0.052	0.068	0.074
SrO														
BaO														
Total	99.145	99.314	99.482	99.771	98.514	99.199	98.438	98.904	99.32	98.972	98.26	98.355	99.118	99.17

Ab %	20.3155091	20.8915965	20.1047312	22.0677273	19.1986007	19.995795	19.9734919	19.7163519	19.77967899	18.88594824	16.21438296	16.28805955	15.87531312	15.84766775
An %	79.3746936	78.6449154	79.4617398	77.4430273	80.4251937	79.5587016	79.5415147	79.8504253	79.75259	80.70349536	83.42224896	83.40060272	83.72786987	83.71465209
Or %	0.30979732	0.46348804	0.433529	0.48924536	0.37620559	0.44550342	0.48499348	0.43322286	0.467731016	0.410556405	0.363368084	0.311337737	0.396817015	0.437680168

Sample	B2_P1_c-r	B2_P1_c-r	B2_P1_c-r	B2_P1_c-r	B2_P1_c-r
SiO2	46.247	47.267	47.413	47.788	52.964
TiO2	0.064	0.056	0.064	0.069	0.109
Al2O3	32.721	32.468	32.317	31.737	28.415
Cr2O3	0.01	0	0	0.016	0
FeO	0.686	0.748	0.697	0.67	0.67
Mno	0.024	0.005	0.026	0.01	0.033



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MgO	0.059	0.052	0.077	0.059	0.088
CaO	16.426	16.451	16.061	15.563	11.506
Na2O	2.003	2.176	2.237	2.468	4.657
K2O	0.075	0.064	0.066	0.085	0.242
SrO					
BaO					
Total	98.315	99.287	98.958	98.465	98.684
Ab %	17.9974363	19.241414	20.0524538	22.1861671	41.6754241
An %	81.5591562	80.3862199	79.5582703	77.3110643	56.8996206
Or %	0.44340753	0.37236613	0.38927596	0.50276861	1.42495536

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: sub-ophitic gabbro B2. Opaque minerals.

Sam ple	B2_2_Oxide of MI	B2_2_Oxide of MI	B2_2_Oxide of MI	B2_3_O x C-r	B2_3_O x C-r	B2_3_O x C-r	B2_3_O x C-r	B2_3_O x C-r	B2_3_O x C-r	B2_3_O x C-r	B2_3_O x C-r	B2_3_O x C-r	B2_3_O x C-r	B2_3_O x C-r	
SiO2	0.03527186	0.06046605	0.00100777	0.04736 507	0	0	0	0	0	0	0	0	0	0.00403 107	0.01813 9816
TiO2	49.3862025	48.8801711	49.2069404	48.8945	49.1086	49.4241	49.3800	49.4363	48.5339	49.3841	48.9867	49.2878	49.4404	49.1526	49.1526
Al2O 3	0.51739623	0.4532391	0.49152642	0.47186	0.51843	0.50083	0.45427	0.39736	0.47496	0.44806	0.44909	0.43771	0.45013	0.47600	0.47600
Cr2 O3	0	0.00409752	0.00921942	0.49222	0.03380	0.00717	0.01536	0.01126	0.01536		0.02868	0.10243	0.02458		
FeO	45.0734416	45.337962	45.1475485	45.4079	45.8772	45.2864	45.0260	45.2690	44.7461	45.6735	44.6555	45.4182	45.7373	44.6977	44.6977
Mno	0.48444444	0.39777778	0.37	0.49222	0.56111	0.47555	0.51222	0.32222	0.53777	0.47111		0.48555			
MgO	5.26295588	5.36010908	5.16482133	5.40034	5.55147	5.33950	5.45333	5.62310	5.35912	5.61427	5.54656	5.50829	5.54165	5.37090	5.37090
CaO	0.08993663	0.08993663	0.07073667	0.00909	0.00505	0.06467	0.05052	0.00808	0.02223	0.00606	0.02526	0.04749	0.04951	0.01616	0.01616
Tota l	100.849649	100.583759	100.4618	100.771	101.586	101.183	100.855	101.257	99.4739	101.663	100.162	101.212	101.733	100.191	100.191
	100.849649	100.583759	100.4618	133	879	898	21	442	9238	8388	9868	0509	2891	6272	6272
Sam ple	B2_3_Ox C-r	B2_3_Ox C-r	B2_3_Ox C-r	B2_3_Ox C-r	B2_Opq_core zone	B2_Opq_core zone	B2_Opq_core zone	B2_Opq_core zone	B2_Opq_rim zone	B2_Opq_rim zone	B2_Opq_rim zone	B2_Opq_rim zone	B2_Opq_rim zone	B2_Opq_rim zone	
SiO2	0	0	0	0.013100 98	0.023	0.011	0.044	0.027	0.017	0.025	0.024				
TiO2	49.15162	49.08401	49.20796	49.31142	47.315	47.232	47.693	47.475	47.405	47.048	47.33				
Al2O 3	0.412882	0.490491	0.490491	0.477039	0.43	0.482	0.431	0.417	0.415	0.481	0.421				
Cr2O 3	0.033804	0.028682													
	52	63	0	0	0	0.036	0	0.009	0	0	0.038				

FeO	45.64262	45.09917	44.85523	44.81715								
	35	32	81	54	43.967	43.574	44.001	43.816	43.995	43.973	43.041	
Mno	0.558888	0.551111	0.526666									
	89	11	67	0.51	0.412	0.36	0.481	0.426	0.491	0.43	0.484	
MgO	5.615258	5.548527	5.669232	5.650587								
	91	42	91	35	5.642	5.56	5.595	5.73	5.508	5.505	5.558	
CaO	0.036378	0.039410	0.028294	0.066694								
	86	43	67	58	0.035	0.013	0.044	0.044	0.145	0.21	0.232	
Total	101.4514	100.8414	100.7778	100.8460								
	62	14	89	02	97.824	97.268	98.289	97.944	97.976	97.672	97.128	

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: sub-ophitic gabbro B2. Glass.

Sample	B2_1_Glass	B2_1_Glass	B2_1_Glass	B2_1_Glass	B2_1_glass of Cpx rim detail1 to glass	B2_2_Glass	B2_2_Glass	B2_3_Glass	B2_3_Glass	B2_5_Glass	B2_5_Glass
SiO2	54.6738168	54.8487131	54.8719652	55.1803085	55.5503205	54.6343893	54.6738168	52.9632697	53.06840966	56.22058808	55.66557013
TiO2	2.61247924	2.58858778	2.50237077	2.52626223	2.40368865	2.79322333	2.83581246	2.97188991	3.083037134	2.560541285	2.493021941
Al2O3	15.0303952	15.0529679	15.2130295	14.853917	15.0468117	15.3115289	14.9493383	14.8580212	14.86417739	15.14325907	15.69834432
Cr2O3	0.011	0	0.011	0.037	0	0.081	0	0.029	0.022	0.088	0.004
FeOtot	10.1942058	9.99472267	10.3268518	10.4502435	10.0451076	10.6528114	10.8913686	12.4265662	11.4137265	9.594728109	9.463110362
MnO	0.19638682	0.17357421	0.2638328	0.16266383	0.18250089	0.14976975	0.26680836	0.27871059	0.194403117	0.147786043	0.182500886
MgO	3.12260676	3.26692892	3.20940191	3.21747587	3.32041895	3.18518001	3.32849292	3.46575036	3.309317249	3.031774632	2.92176683
CaO	6.33021533	6.44312887	6.05498858	6.13261664	6.6981925	6.34332137	6.36852529	6.91091354	6.772796085	6.223350734	6.027768352
Na2O	4.50207189	4.28819674	4.30102925	4.14596976	4.13634538	3.68399944	3.82836517	3.85082206	4.087154098	4.193022297	3.948135251
K2O	2.13767372	2.11790934	2.16888063	2.10750703	2.07525989	2.0825415	2.05965643	1.8370471	1.955633376	2.24585769	2.234415155
P2O5	0.94774347	0.81793133	0.78691427	1.02586051	0.78117037	0.83516303	0.88226301	0.76049233	0.845502051	0.908684949	0.860436191
F											
Cl											
Total	99.758595	99.5926609	99.7102648	99.8398249	100.239816	99.752928	100.084447	100.352483	99.61615667	100.3575929	99.49906942

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: sub-ophitic gabbro B2. Melt Inclusions (MI).

Sample	B2_2_MI in Oxide	B2_2_MI in Oxide	B2_2_MI in Oxide
SiO <sub>2</sub>	50.9352806	51.3710051	51.1940868
TiO <sub>2</sub>	3.65435466	3.50788962	3.66162598
Al <sub>2</sub> O <sub>3</sub>	14.4989087	14.7903028	14.6876993
Cr <sub>2</sub> O <sub>3</sub>	0	0.011	0.007
FeO <sub>tot</sub>	12.6435298	12.7124235	12.9550937
MnO	0.25490613	0.19539497	0.22118314
MgO	3.80889396	3.58080439	3.54951777
CaO	6.4471615	6.43607178	6.43909625
Na <sub>2</sub> O	4.57692819	4.43470122	4.64857636
K <sub>2</sub> O	1.78503557	1.88281724	1.85681148
P <sub>2</sub> O <sub>5</sub>	0.74785575	0.76508745	0.79725329
F			
Cl			
Total	99.3528549	99.6874981	100.017944

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: sub-ophitic gabbro B23. Clinopyroxenes.

Data set	B23_1_Cpx r to MI	B23_1_Cpx r to MI	B23_1_Cpx r to MI	B23_1_Cpx r to MI	B23_1_Cpx r to MI	B23_1_Cpx r to MI	B23_1_Cpx r to MI	B23_1_Cpx r to MI	B23_1_Cpx r to MI	B23_1_Cpx r to MI	B23_1_Cpx r to MI	B23_1_Cpx r to MI	B23_1_Cpx r to MI	B23_1_Cpx r to MI
SiO									49.3709246		49.4655087	49.2982442	50.3924332	51.1371588
2	49.7731561	50.2470724	49.6586595	49.3669422	49.48343	49.4864169	49.5043381	49.5700491	6	49.9762631	9	3	7	4
TiO									1.71311441	1.75988193			1.37049153	1.35015783
2	1.67753044	1.66329685	1.61347928	1.74564834	1.61754602	1.78834912	1.73853155	1.6734637	8	3	1.68363055	1.73141475	4	2
Al2									4.22133647		4.38687908		3.33946444	3.49172276
O3	4.08849365	4.16309001	4.37155107	4.37870414	4.18659297	4.46249731	4.54935609	4.22848955	9	4.31534833	6	4.37461667	3	6
FeO	7.61632184	7.6112239	7.41138466	7.66832082	7.50314758	7.68769299	7.64690948	7.64690948	7	9	7.87631676	7.77741673	3	5
Mn									0.20060331	0.12880844	0.07496229	0.10452488	0.11191553	0.17315233
o	0.26606335	0.12986425	0.14781297	0.13303167	0.24494721	0.19215686	0.17737557	0.19215686	8	6	3	7	5	8
Mg									14.3329328	14.3613848	14.4634195		14.7420527	14.7665803
O	14.5213046	14.3643282	14.5811519	14.3015376	14.2760289	14.4133833	14.458514	14.3456872	6	4	5	14.3810069	8	6
CaO	22.2545501	21.7674217	21.5158389	21.6962569	21.6060479	21.5218529	21.5890084	21.4206184	5	2	2	7	7	1
Na2										0.37673388	0.31256492	0.18422700		
O	0.31877482	0.32498472	0.34257944	0.37776887	0.34154445	0.4067484	0.31359991	0.34878934	0.32394974	3	5	9	0.34982432	0.37880385
K2O	0.025	0.0125	0.021875	0.02708333	0.0125	0.009375	0.003125	0	0.01875	7	3	3	0.034375	3
Cr2									0.16509475	0.04253975		0.06482247	0.06988673	0.02734698
O3	0.07798954	0	0.09014376	0	0.05469397	0.01215421	0	0.13470921	1	2	0	9	5	3
Tota									99.5279392	100.697126	100.030016	99.4317534	99.5210398	100.378893
le	100.541195	100.283782	99.6643328	99.6952938	99.271785	99.9684727	99.9807581	99.4261637	9	1	1	8	5	6
Fe2									2.27449806	2.35877638		1.64849449	2.09636582	1.21563340
O3	3.06331585	1.335302	2.08791495	2.58534122	2.04280765	2.52493974	2.1823651	1.77803478	2	1	2.69174554	4	8	1
FeO	4.8599165	6.40970446	5.5326558	5.34200232	5.66500676	5.41572441	5.68319333	6.0470142	5	8	3	2	6	5
Wo	45.4348881	45.083081	44.6304593	44.9528626	45.0649112	44.5464888	44.7507416	44.643869	44.8943836	45.3581499	44.7738704	44.9550299	44.7370940	44.5909700
En	41.2503063	41.3945164	42.0838829	41.2293196	41.4306266	41.5098013	41.7005804	41.6008377	8	3	4	5	8	7
Fs	12.1370818	12.3043717	11.9997106	12.4014127	12.2153248	12.4201888	12.3723397	12.4398206	41.3903647	40.9874132	41.4113651	41.6992935	42.1162685	42.2058154
N°									12.4985621	12.2560518	12.6508348	12.6509232	11.8468275	11.7950783
Mg	77.2660132	77.0863567	77.8126603	76.8762943	77.2297359	76.9697921	77.1191574	76.9806272	9	1	3	5	2	9
										76.9811153	76.5994820	76.7233251	78.0464273	78.1576237
									76.8068082	3	2	3	3	7
Data set	B23_4_Cpx of MI	B23_4_Cpx of MI	B23_4_Cpx of MI	B23_6_Cpx of MI	B23_6_Cpx of MI	B23_6_Cpx of MI	B23_6_Cpx of MI1	B23_6_Cpx of MI1	B23_6_Cpx of MI1	B23_8_Cpx of MI	B23_8_Cpx of MI	B23_8_Cpx of MI	B23_8_Cpx of MI1	B23_8_Cpx of MI1

SiO										48.8731134	48.6630371	48.4668995	50.8563933	50.7687785
2	50.3326959	50.2211862	51.1112727	48.7397	48.9019865	48.9557501	50.772761	50.511908	50.3625646	5	2	1	2	5
TiO									1.46910999	1.92763497	1.91441807	1.87680072	1.58806214	1.33694092
2	1.34914115	1.33490756	1.39895872	1.97541918	1.97033575	1.98151929	1.35727463	1.34100767	1	7	1	2	9	6
Al2									3.80645809	4.52278752	4.57899025	4.56059663	3.86368269	3.41917014
O3	3.39566718	3.28326171	3.28939291	4.72205177	4.64438981	4.54833422	3.58777835	3.54690364	3	2	9	6	8	2
									7.41240425	8.26274058	8.10062610	8.30250451	6.95766803	6.93727627
FeO	6.78739685	7.15037015	7.18707532	8.20666325	8.01600031	8.40752207	7.01680413	7.10041034	1	5	4	4	5	6
Mn									0.09185520	0.12141779	0.21855203		0.16681749	0.11825037
o	0.21644042	0.10241327	0.12880845	0.13936652	0.10769231	0.15837104	0.12669683	0.16470588	4	8	6	0.15520362	6	7
Mg									14.7862024	13.9287184	14.0062255		14.7234118	15.0255915
O	14.9108025	14.8117111	14.7911079	13.9287184	13.6903104	13.9414728	14.8313332	14.7842402	2	5	8	13.8119672	3	3
									22.1964154	22.0791437	22.0210091	21.8275610	21.9779092	22.4008890
CaO	22.3237103	22.3317289	22.1342715	21.8035053	21.709287	22.2515431	22.2755988	21.9087491	3	8	2	2	8	7
Na2									0.38811869	0.36431408	0.39639856		0.25046593	0.34464940
O	0.29600519	0.28669035	0.26185075	0.28979529	0.34050947	0.36534907	0.26288573	0.29497021	8	5	4	0.32394974	3	4
									0.01041666	0.00416666			0.01041666	0.01666666
K2O	0	0	0.00833333	0	0.02604167	0.01875	0.02083333	0	7	7	0.00625	0	7	7
Cr2									0.09824656	0.05064256		0.03038553	0.12660640	0.07089958
O3	0.06786103	0.05773252	0.04253975	0.04253975	0.04760401	0	0.01519277	0.12154215	9	2	0	7	4	6
Tota									100.523545	100.084037	99.9055068	99.3254829	100.394827	100.368212
le	99.6118595	99.5222692	100.311072	99.8052198	99.4065533	100.628612	100.251966	99.652895	3	3	5	5	4	9
Fe2									2.79961386		3.83147205	3.20402896	0.43627860	2.47770756
O3	2.34689829	2.55944269	0.97335194	2.63005065	1.91111494	3.73107664	1.62343843	1.51731994	6	3.32684165	1	6	5	4
									4.89328086	5.26921183	4.65302536	5.41948396	6.56509974	4.70780772
FeO	4.67563192	4.84735543	6.31124253	5.84011471	6.29635804	5.05025823	5.55601636	5.73510915	7	5	3	9	1	7
									45.0615850	45.4600520	45.3578829	45.3680417	45.4577045	
Wo	45.6479502	45.5371475	45.3587313	45.3163695	45.578076	45.5290752	45.5864433	45.1210736	1	8	9	8	8	45.3955594
									41.7667963		40.1409433	39.9439730	42.3721482	42.3672810
En	42.4235315	42.0242147	42.1742405	40.2801285	39.9921379	39.6906533	42.2315877	42.365443	3	39.9033869	2	1	8	2
									11.7457583	13.2791466	13.0236433		11.2326745	10.9732545
Fs	10.8331933	11.380736	11.4959856	13.3135462	13.1361016	13.4274968	11.2084103	11.4141516	7	1	1	13.4695253	4	6
N°										75.0310003	75.5031607	74.7825442		
Mg	79.6585439	78.689736	78.5803296	75.1583629	75.2747282	74.7214525	79.0261775	78.7760549	8	4	8	6	79.0454031	79.4279258
Dat														
a														
set	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc
SiO									49.3320953	48.6142516	48.5475449	44.8697156	51.0176841	50.9519730
2	49.6218215	49.4864169	48.8999953	49.3470297	49.4455963	49.5033424	49.7363181	48.4768557	9	2	2	9	5	7
TiO									1.63584634	1.98355265	1.93983519	1.88798425	1.23120567	1.29627352
2	1.60026237	1.67854712	1.79444923	1.7212479	1.60737917	1.56671176	1.55654491	1.95101873	9	9	9	8	4	1

Al2O3	4.25914559	4.44716929	4.54015928	4.43286314	4.26936427	4.13856518	3.80850183	4.55037796	4.23257702	4.43081940	4.59023080	4.22031461	2.87247079	2.75495597
FeO	6.84959171	7.34409186	7.64385071	7.66424247	7.59185173	7.79067137	7.93647245	7.87427758	6.84755253	7.76008373	8.34634679	8.06697970	7.92831574	7.77741673
MnO	0.18159879	0.15309201	0.15098039	0.23016591	0.15625943	0.1066365	0.1361991	0.11719457	0.12247360	0.09502262	0.15625942	0.13197586	0.20165912	0.16892911
MgO	14.5262101	14.3221407	14.2171627	14.3986668	14.5242479	14.4732306	14.4889282	13.9296995	14.4771549	13.7354411	13.5882757	14.0307531	14.8136733	15.1521538
CaO	22.1843876	21.2983351	21.6862336	21.6070502	21.7253242	22.0911716	21.914763	21.6792174	22.1222436	22.0871623	21.6100571	21.2522283	21.8105215	21.6321082
Na2O	0.35603422	0.38294378	0.32187977	0.32084479	0.37466392	0.30325008	0.33947449	0.33843951	0.33740452	0.32705469	0.34050947	0.49265200	0.32601970	0.30739000
K2O	0.01770833	0.02395833	0.01770833	0.01145833	0.02708333	0.01041667	0.01666667	0.01666667	0.0125	0.03333333	0.0125	0.05208333	0.01354166	0.02395833
Cr2O3	0.09419516	0.04557831	0.00506426	0.05064256	0.08913091	0	0.02734698	0.08609235	0.02532128	0.10128512	0.03241123	0.00506425	0.01519276	0.05064256
Totale	99.5967602	99.1366951	99.2724193	99.7335693	99.7217703	99.9839962	99.9338688	98.9337477	99.1198480	99.0667215	99.1315595	95.0046869	100.215091	100.065158
Fe2O3	2.45881629	1.61030333	2.64489025	2.42018016	2.95654815	3.15111018	3.01321869	2.70722554	2.49439740	2.52211882	2.45705070	8.36541815	2.12480506	2.27046002
FeO	4.63712174	5.89512319	5.26394934	5.48653771	4.93151715	4.95526774	5.22514509	5.43828623	4.60306628	5.49065344	6.13546551	0.53968433	6.01639275	5.73443179
Wo	45.8480029	44.6960475	45.1638455	44.8239945	44.7593211	45.2142056	44.8304024	45.3353889	45.8766787	46.1615291	45.3538915	44.3098794	44.3297869	43.8414756
En	41.7711276	41.8198334	41.1974763	41.5612167	41.6353075	41.2166352	41.2404141	40.5309135	39.9423648	39.6802033	40.7031460	41.8931728	42.7279034	41.7731207
Fs	11.0493317	12.0298416	12.4255987	12.4103108	12.2085276	12.4459883	12.672483	12.8529498	11.0840000	12.6591637	13.6726725	13.1282067	12.5779214	12.3032551
N°	79.0813412	77.6603264	76.8278885	77.0058189	77.3260438	76.8069702	76.4945241	75.92353	79.0302614	75.9338481	74.3731292	75.6123410	76.9089980	77.6431107
Mg														
Data set	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc	B23_1Cpx r-r with Pl inc
SiO2	51.0724434	50.2779367	50.6174439	50.4999605	50.4581443	50.488013	50.9041832	50.532816	50.4880130	50.1106721	50.2251687	51.1968961	55.1754033	51.5433727
TiO2	1.2637396	1.32779076	1.35829131	1.3796417	1.33389087	1.36032468	1.34812446	1.31864059	1.31864059	1.27187307	1.31050711	1.14478743	1.02176854	1.03803550
Al2O3	2.95728583	2.76824026	2.78867762	2.89188628	3.05845076	2.93173913	2.94911089	2.94502342	2.94706715	3.00429175	3.11260975	2.76721839	2.83874915	2.61087259
FeO	7.75906415	7.855925	7.9323941	8.02415701	7.94258998	7.95176627	8.12407663	7.83247448	7.92729615	7.70196722	7.62345895	7.46950117	7.17789902	7.34511144





CaO	21.3795232	21.8756725	21.8606376	21.8075146	21.8726655	21.5759783	21.4757461	21.5068181	21.9889348	21.6802197	22.0671159	21.7924797	21.6220850	21.6210827
Na2O	0.31566987	0.2835854	0.24218607	0.31980981	0.25771082	0.32394974	0.26702566	0.30842499	0.33947448	0.26288573	0.30842499	0.31877482	0.24736098	0.29600519
K2O	0.01458333	0.01875	0	0.0125	0.00625	0.009375	0.01875	0	0.00833333	0	0.00833333	0.009375	0.01145833	0.00520833
Cr2O3	0.04051405	0	0	0.04557831	0	0.04760401	0.02025702	0.09925942	0	0.04355260	0.04051404	0.05368111	0.02025702	0.11445218
Totale	99.1296696	100.32843	99.9114256	99.4353476	98.9963925	98.9301531	99.2747047	99.2484828	100.267687	99.9777614	100.187083	99.6589251	100.003994	99.1156613
Fe2O3	1.31761624	1.52892707	1.70156748	3.00560789	2.43742325	1.73416822	1.74560606	1.57463344	3.23438312	1.14666458	2.95025995	2.71481601	1.42300329	1.85628624
FeO	6.06774293	5.90615159	5.59990883	4.44996943	4.94899317	5.43191037	5.5113422	5.4673852	4.28693764	6.09513821	4.53953603	4.63209930	6.02389389	5.44131900
Wo	43.9160542	44.2866635	44.3115482	43.9915818	44.5024107	44.113461	43.9227287	43.8741434	43.9371615	44.1057864	44.2583318	44.0314075	43.9395950	44.1844612
En	43.281012	43.1676151	43.5176526	43.5757786	43.2061169	43.529065	43.7832482	44.025304	43.6101339	43.6093885	43.3598585	43.6453185	43.5646511	43.3770551
Fs	11.6295307	11.5067964	11.2824329	11.2651743	11.34261	11.1588939	11.3057366	10.9619535	11.2251982	11.3170204	11.2624031	11.1577297	11.5860950	11.3438211
N°	78.820951	78.9539639	79.4116509	79.4584636	79.2064625	79.5953366	79.477319	80.0645568	79.5292600	79.3960307	79.3812947	79.6403118	79.2696647	
Mg									7	5	5	4	78.9919523	1
Data set	B23_3Cpxr-r	B23_3Cpxr-r	B23_3Cpxr-r	B23_3Cpxr-r	B23_3Cpxr-r	B23_3Cpxr-r	B23_3Cpxr-r	B23_3Cpxr-r	B23_3Cpxr-r	B23_3Cpxr-r	B23_3Cpxr-r	B23_3Cpxr-r	B23_3Cpxr-r	B23_3Cpxr-r
SiO2	51.4886136	51.3243359	51.3223446	50.9410212	50.8663495	50.8962182	50.4850262	51.1152551	50.9290737	51.0704521	50.3824770	50.2032650	50.7020718	50.6751900
TiO2	1.01566843	1.04413561	1.02075186	1.120387	1.06853606	1.1061534	1.09496987	1.16308777	1.11428688	1.17630467	1.16410445	1.17732136	1.18748821	1.17528799
Al2O3	2.82137739	2.76313092	2.65992226	2.70284071	2.6905783	2.65379105	2.86940519	2.79991817	2.93480473	2.88677694	2.90517056		2.91538924	3.07275690
FeO	6.97296185	7.15444851	7.11672375	6.96276597	7.20134955	7.01680413	7.12182169	6.92096287	7.10346910		6.90464946	7.14119386	7.03821548	7.05452888
MnO	0.16259427	0.18582202	0.13514329	0.17104072	0.13831071	0.15309201	0.10769231	0.11508296	0.13725490	0.19110105			0.16576168	0.15098039
MgO	15.4366737	15.6534974	15.4768989	15.2904893	15.4347115	15.5318406	15.2551696	15.3532799	15.2139633	15.1148719			15.1178152	15.2394719
CaO	22.1412877	21.5779829	22.1583272	21.4105952	21.6270967	21.9478396	21.8095192	21.889705	21.5659550	21.8295656	22.0580950	21.7433659	22.0811484	21.7032731
Na2O	0.30221509	0.27323556	0.30118011	0.28565536	0.34154445	0.2835854	0.31049496	0.34050947	0.22459135	0.32187977	0.34464940		0.35913916	0.34671937
O									3	4	4	0.32394974	9	1

K2O	0.00625	0.00416667	0.00104167	0.009375	0.01979167	0.00729167	0.00833333	0.01458333	0	0.01875	0.01354166	0.0125	0.0125	0.01458333
Cr2									0.04861685	0.09115661	0.06887388		0.05064256	0.02329557
O3	0	0.09419516	0.02329558	0.02329558	0.06077107	0	0	0.0790024	9	1	4	0	2	8
Tota									99.2233990	99.5449392	99.3066499	98.9135740	99.5795292	99.4327920
le	100.347642	99.9807555	100.192334	98.8941705	99.3882684	99.5966162	99.0624323	99.7123846	8	5	8	1	7	2
Fe2									1.22011510	3.33651820	2.93965132	2.68535828	2.40237265	
O3	2.17847595	1.993052	2.5858203	1.37661138	2.64374206	2.73392308	2.7542972	2.0559285	2	1.55430234	4	8	8	5
FeO	5.01274521	5.36107837	4.78997418	5.7240759	4.82248133	4.55679004	4.64347474	5.07101577	5	5.3665873	3.90241364	4.49606322	4.62190052	4.89284751
Wo	44.6312052	43.6474074	44.5010835	44.0208332	43.8310538	44.2953046	44.367703	44.4275842	8	44.6813542	44.6413202	44.3656308	44.8260999	44.2551517
En	43.2952608	44.0564898	43.2482419	43.7423187	43.5244312	43.6153878	43.1806119	43.3575623	8	43.0463767	43.1893661	43.0648502	42.7020995	43.2373846
En									6	5	2	3	7	4
Fs	10.9711303	11.2959332	11.1560915	11.1740256	11.3918939	11.0535967	11.3086378	10.9642179	11.3885425	8	11.0800289	10.9070928	11.3733670	11.1524479
N°									79.2440640	79.8376954	79.1077525	79.2915390	79.3849753	
Mg	79.7828267	79.5927032	79.4941123	79.6526412	79.2559064	79.7808633	79.246112	79.8161661	9	79.5293466	4	7	8	1
Dat														
a														
set	B23_3Cpx-r-r	B23_3Cpx-r-r	B23_5Cpx r-r	B23_5Cpx r-r	B23_5Cpx r-r	B23_5Cpx r-r	B23_5Cpx r-r	B23_5Cpx r-r	B23_5Cpx r-r	B23_5Cpx r-r	B23_5Cpx r-r	B23_5Cpx r-r	B23_5Cpx r-r	B23_5Cpx r-r
SiO									47.6664190	48.1960902	47.4404127	47.4025791	47.6942965	
2	50.8663495	50.7379143	49.4983643	49.1220191	48.2408932	47.9531583	48.0756199	47.54794	8	47.6853359	1	9	4	
TiO									2.14215553	2.19197310	2.16960603	2.20213995	2.14825564	
2	1.21392203	1.21697208	1.63584635	1.89306768	1.9601689	2.18688968	2.16655598	2.18993974	6	7	5	8	7	
Al2									4.95503766	4.89270371	5.03167775	4.88963811	4.96423447	
O3	3.11567536	3.16268129	4.04761893	4.43490688	4.5554873	4.90905361	4.86920076	4.93664404	3	8	9	4	4	
FeO									8.61857677	8.45952105	8.44524682	8.25662305	8.35552308	
Mn	7.06676394	7.01884331	6.75477004	7.40220837	8.14854674	8.48297158	8.6165376	8.51865716	3	6	5	7	7	
o									0.06968325	0.16365007	0.17948717	0.15098039	0.12880844	
Mg	0.11930618	0.17843137	0.14464555	0.1372549	0.17209653	0.22699849	0.16892911	0.18793363	8	5	9	2	6	
O	15.2993193	15.049138	14.6321693	14.0121122	13.5480505	13.5735592	13.1673826	13.3547732	13.3184724	13.3027747	14.0072066	13.3655653	13.8443435	
CaO									2	7	8	6	13.3135669	
Na2	21.5238575	21.9779093	22.1052041	21.8476075	21.942828	21.5629481	21.8435982	21.8546237	21.8506144	22.0791437	21.5398946	21.5388923	21.8205447	
O	0.30325008	0.31463489	0.28772533	0.35499924	0.34878934	0.42434311	0.35396425	0.34775435	0.36327910	0.36017415	0.42537809	0.34154445	0.42951802	
K2O	0.00625	0	0.01458333	0.01666667	0.009375	0.02291667	0.00729167	0.00520833	0	0.009375	0.009375	7	7	
Cr2									0.06279677	0.05064256	0.02532128	0.09824656	0.07089958	
O3	0.00303855	0.01316707	0.04861686	0.05874537	0.13167066	0.11647789	0.06583533	0.05773252	6	2	1	0.06583533	9	



FeO	5.34697401	4.84299922	4.85702326	7.68124992	6.44943668	4.89368923	5.06610713	4.66835798	5.0167222	9	4.96354395	8	6.03275094	2	4.9251202
Wo	46.0585126	45.7154391	46.3012829	44.8949156	44.6403466	44.822978	45.4502052	45.2120688	45.6022557	45.5116109	9	44.9147122	44.8567863	9	45.6651518
En	38.7571008	39.300888	38.5676495	40.0343672	41.8247614	41.7920022	41.0650739	40.8604023	41.0725239	42.0983999	40.8328780	40.8616237	41.2603115	40.0018047	
Fs	13.500445	13.5002856	13.7496863	13.5440821	12.3183549	12.1706659	12.3258139	12.6035014	11.9291341	11.1532946	12.8920787	12.9395867	12.9809220	13.1662375	
N°									1	5	5	9	8	9	
Mg	74.1655587	74.4318456	73.7186803	74.7210264	77.2485299	77.4461376	76.9140121	76.4261483	77.4929039	79.0555122	76.0035567	75.9492646	76.0681658	75.2365575	
									2	6	7	2	3	1	
Dat															
a															
set	B23_7Cpx	B23_7Cpx	B23_7Cpx	B23_7Cpx	B23_7Cpx	B23_7Cpx	B23_7Cpx	B23_7Cpx	B23_7Cpx	B23_7Cpx	B23_7Cpx	B23_7Cpx	B23_7Cpx	B23_7Cpx	B23_7Cpx
	class c-r	class c-r	class c-r	class c-r	class c-r	class c-r	class c-r	class c-r	class c-r	class c-r	class c-r	class c-r	class c-r	class c-r	class c-r
SiO									49.9205082	49.8617665	50.2460767	50.0091186	51.3542045		
2	50.0638779	50.1972913	50.1106721	49.9284732	50.3565909	49.9434076	49.9961755	49.7771386	4	2	7	4	3	50.1763832	
TiO									1.19155495		1.23323904	1.25865617	1.22408887		
2	1.30644037	1.23120567	1.22612225	1.2362891	1.26068954	1.24645595	1.2362891	1.23933915	4	1.19663838	4	2	8	1.23832247	
Al2									2.69262203	2.66196599	2.64663797	2.70590631	2.86531771	2.63335369	
O3	2.72327807	2.56386667	2.56693228	2.59656645	2.64459424	2.60678513	2.6650316	2.67933775	4	5	6	7	6	3	
FeO	7.88039511	7.89874769	7.91404151	7.89874769	8.08023435	8.09348899	7.86510129	7.93443327	8.01090236		7.89263016	7.84470953	7.76620126	7.76212291	
Mn									6	7.90078687	6	4	3	1	
o	0.14253394	0.15625943	0.1984917	0.11719457	0.17104072	0.21749623	0.17420814	0.22383107	0.18687782	0.08657616	0.18371040	0.12352941	0.18371040	0.23227752	
Mg									8	9	7	2	7	6	
O	14.9716309	14.7812969	15.0412892	14.8941237	15.020686	14.6890732	14.8852938	14.792089	15.1992467	14.9117835			16.0282787	15.0167616	
CaO	21.4456764	21.3274025	21.5589388	21.2592446	21.3895464	21.2873096	21.478753	21.7503822	5	9	14.848993	14.8480119	4	1	
Na2										21.4316439		21.4887762	20.1256186	21.2812956	
O	0.37052398	0.35189429	0.33222961	0.34878934	0.37155897	0.32084479	0.32291476	0.32808967	0.29807516	0.35499923	0.31773984	0.34154445	0.48644210	0.33947448	
K2O	0.01979167	0.00625	0.00104167	0.01354167	0.01354167	0.0125	0.00833333	0.00416667	0.00729166	0.01354166	0.00833333				
Cr2									7	7	3	0.025	0.059375	0.021875	
O3	0.03848835	0.02835983	0.05570682	0.09115661	0.07596384	0.09318231	0.05064256	0	0.08102809	0.03848834	0.02228272		0.07292528	0.04861685	
Tota											98.4197023	98.7388419	98.6452526	100.093237	98.7018665
le	98.9241483	98.5142144	98.9497591	98.2929704	99.3084828	98.4173615	98.6321006	98.7288075	98.7994002		5	4	9	3	6
Fe2									3.50658575	3.32122271	2.45301287		2.42818715	2.80553332	
O3	3.22293637	2.47729276	3.36619884	2.87118736	3.10573369	2.65511008	2.98397421	3.65945734	6	4	9	2.99754153	3	7	
FeO	4.98036148	5.66965239	4.88509873	5.31522169	5.28566097	5.7043917	5.18008844	4.64161326	4.85563789	4.9123141	6	8	5	1	
Wo	43.6787325	43.7968718	43.7604401	43.5824342	43.4141101	43.7798059	43.9116856	44.2810405	43.2482613	43.7477017	43.8113535	43.9537928	40.7653185	43.5658443	
									7	2	2	4	6	9	

En	42.427778	42.2346176	42.4805627	42.4844021	42.4199346	42.0337706	42.3427785	41.9016702	42.9554831	42.3743650	42.2575075	45.1730469	42.7735459
Fs	12.5278521	12.6608166	12.5386556	12.6392191	12.801228	12.9923342	12.5508665	12.6085511	12.7006348	12.6350045	12.5244852	12.2785925	12.4030053
N°									3	6	9	7	9
Mg	77.2036967	76.9364853	77.2104076	77.0711379	76.818257	76.3887809	77.1360301	76.8693819	77.1801640	77.0875127	77.0311773	78.6279509	77.5212384
									2	9	7	9	5
									12.7006348	12.6350045	12.5244852	12.2785925	12.4030053
									3	6	9	7	9
									77.1801640	77.0875127	77.0311773	78.6279509	77.5212384
									1	7	4	8	8
Data set	B23_7Cpx class c-r	B23_7Cpx class c-r	B23_7Cpx class c-r	B23_7Cpx class c-r	B23_7Cpx class c-r	B23_7Cpx class c-r	B23_7Cpx class c-r	B23_7Cpx class c-r	B23_7Cpx class c-r	B23_7Cpx class c-r	B23_7Cpx class c-r	B23_7Cpx class c-r	B23_7Cpx class c-r
SiO2	50.0638779	50.1972913	50.1106721	49.9284732	50.3565909	49.9434076	49.9961755	49.7771386	49.9205082	49.8617665	50.2460767	50.0091186	51.3542045
TiO2	1.30644037	1.23120567	1.22612225	1.2362891	1.26068954	1.24645595	1.2362891	1.23933915	1.19155495	1.19663838	1.23323904	1.25865617	1.22408887
Al2O3	2.72327807	2.56386667	2.56693228	2.59656645	2.64459424	2.60678513	2.6650316	2.67933775	2.69262203	2.66196599	2.64663797	2.70590631	2.86531771
FeO	7.88039511	7.89874769	7.91404151	7.89874769	8.08023435	8.09348899	7.86510129	7.93443327	8.01090236	7.90078687	7.89263016	7.84470953	7.76620126
Mno	0.14253394	0.15625943	0.1984917	0.11719457	0.17104072	0.21749623	0.17420814	0.22383107	0.18687782	0.08657616	0.18371040	0.12352941	0.18371040
MgO	14.9716309	14.7812969	15.0412892	14.8941237	15.020686	14.6890732	14.8852938	14.792089	15.1992467	14.9117835	14.848993	14.8480119	16.0282787
CaO	21.4456764	21.3274025	21.5589388	21.2592446	21.3895464	21.2873096	21.478753	21.7503822	21.2923212	21.4316439	21.3614814	21.4887762	20.1256186
Na2O	0.37052398	0.35189429	0.33222961	0.34878934	0.37155897	0.32084479	0.32291476	0.32808967	0.29807516	0.35499923	0.31773984	0.34154445	0.48644210
K2O	0.01979167	0.00625	0.00104167	0.01354167	0.01354167	0.0125	0.00833333	0.00416667	0.00729166	0.01354166	0.00833333	0.025	0.059375
Cr2O3	0.03848835	0.02835983	0.05570682	0.09115661	0.07596384	0.09318231	0.05064256	0	0.08102809	0.03848834	0.02228272	0.07292528	0.04861685
Total	98.9241483	98.5142144	98.9497591	98.2929704	99.3084828	98.4173615	98.6321006	98.7288075	98.7994002	98.4197023	98.7388419	98.6452526	100.093237
Fe2O3	3.22293637	2.47729276	3.36619884	2.87118736	3.10573369	2.65511008	2.98397421	3.65945734	3.50658575	3.32122271	2.45301287	2.99754153	2.42818715
FeO	4.98036148	5.66965239	4.88509873	5.31522169	5.28566097	5.7043917	5.18008844	4.64161326	4.85563789	4.9123141	5.68538216	5.14748865	5.58129172
Wo	43.6787325	43.7968718	43.7604401	43.5824342	43.4141101	43.7798059	43.9116856	44.2810405	43.2482613	43.7477017	43.8113535	43.9537928	40.7653185
En	42.427778	42.2346176	42.4805627	42.4844021	42.4199346	42.0337706	42.3427785	41.9016702	42.9554831	42.3743650	42.2575075	45.1730469	42.7735459
Fs	12.5278521	12.6608166	12.5386556	12.6392191	12.801228	12.9923342	12.5508665	12.6085511	12.7006348	12.6350045	12.5244852	12.2785925	12.4030053
N°									3	6	9	7	9
Mg	77.2036967	76.9364853	77.2104076	77.0711379	76.818257	76.3887809	77.1360301	76.8693819	77.1801640	77.0875127	77.0311773	78.6279509	77.5212384
									2	9	7	9	5
									12.7006348	12.6350045	12.5244852	12.2785925	12.4030053
									3	6	9	7	9
									77.1801640	77.0875127	77.0311773	78.6279509	77.5212384
									1	7	4	8	8

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: sub-ophitic gabbro B23. Olivines.

Data set	B23_2_OI								B23_9_OI			B23_2OI		B23_2OI from Mi to r
	of MI	of MI	of MI	of MI1	of MI1	of MI1	of MI	of MI	of MI	of MI	of MI	from Mi to r	from Mi to r	
SiO2	38.8740775	38.4648767	38.8641212	39.0254121	39.0144602	38.8810468	38.9626879	38.9945478	38.4818022	38.0964963	38.5355658	38.8890118	38.8561562	38.7346903
TiO2	0.02338376	0.03863403	0.054901	0.07015127	0.06710122	0.05388431	0.04575083	0.0477842	0.06506784	0.04880088	0.03151723	0.06201779	0.03761734	0.02745049
Al2O3	0.04087472	0.03065604	0	0.04087472	0.05518087	0.04291845	0.04189659	0.02145923	0.07561822	0.05415900		0.02145922	0.04087471	0.06131207
Cr2O3	0	0	0.05368112	0.07495099	0.03241124	0.03038554	0	0.02126988	0.03342409		0.02633413		0.00506425	0.01417991
FeO	20.5263442	20.4457968	20.1572534	20.4926978	20.5222659	20.4804628	20.4121504	20.8740237	20.3856410	20.2388204		20.4692472	20.4009348	20.5548926
MnO	0.2280543	0.31568627	0.3653092	0.22594268	0.28717949	0.22488688	0.26923077	0.28612368	0.28401206	0.24600301	0.34208144	0.26395173	0.23333333	0.32941176
MgO	39.8013841	39.7484045	39.4501492	39.2794373	39.2912106	39.7042549	39.6022202	39.3088704	39.4982232	39.1214797	39.4687901	39.5472784	39.371661	39.4579980
CaO	0.34580101	0.33477547	0.38188459	0.32575457	0.30370349	0.33778243	0.25759669	0.32976386	0.25258508	0.31472903	0.27563848	0.30470581		0.27563848
Na2O	0.01448976	0.02380461	0.01448976	0.03208448	0.00517492	0	0.03829438	0	0.02380461	0.01034983		0.01345478		0.01034983
NiO														
Total	99.8544093	99.4026344	99.3417895	99.5673059	99.5786879	99.7556221	99.6298277	99.8838427	99.1001785	98.1470439	99.1706748	99.5711268	99.2561757	99.4659236
Fo	77.3656209	77.3353593	77.4055595	77.1640476	77.0915333	77.3642969	77.3390252	76.8031158	77.3026999	77.2925299	77.1844213	77.2708297		77.1024764
Fa	22.3825156	22.3156681	22.1871903	22.583764	22.5883252	22.3867344	22.3622432	22.8792559	22.3814863	22.4354920	22.4361479	22.4628040	22.5318028	22.5318028
									1	22.4313233	7	6	3	2
Data set	B23_2OI								B23_2OI			B23_2OI		B23_2OI from Mi to r
	from Mi to r	from Mi to r	from Mi to r	from Mi to r	from Mi to r	from Mi to r	from Mi to r	from Mi to r	from Mi to r	from Mi to r	from Mi to r	from Mi to r	from Mi to r	
SiO2	38.6998435	38.6430931	38.825292	38.6709705	38.5893295	38.7287166	38.8730818	38.7615721	38.6898873	38.7874583	38.8969767	41.6677939	38.7635633	38.5276008
TiO2	0.01728365	0.01525028	0.03050055	0.01423359	0	0.03050055	0.0203337	0.03456729	0.03558397	0.01525027	0.01321690	0.03761734	0.01321690	0.02541712
Al2O3	0.07459636	0.02759043	0.0510934	0.02350296	0.02759043	0.00102187	0.02656857	0.02656857	0.06335581	0.04189658		0.06131207	0.01124054	
Cr2O3	0.03139839	0.02430843	0.03544979	0.01620562	0	0.01215421	0.06178393	0	0.04051404		0.03848834			0.02126987
									9	0	7	0	0	6

FeO	20.5324617	20.7659474	20.9963742	20.3723864	20.5926174	20.7017133	20.8495536	20.4498751	20.3713668	20.5915978	20.9668061	20.4937173	20.9107288	20.5365400	
Mn									4	4	9	9	6	9	
o									0.30723981	0.27450980	0.28295625	0.29034690	0.29351432	0.31463046	
Mg	0.23016591	0.37586727	0.33680241	0.34524887	0.23544495	0.32730015	0.31779789	0.27239819	9	4	9	8	9	8	
O	39.7101415	39.2460798	39.8641747	39.4207161	39.5953525	39.7601778	39.5973147	39.248042	39.4942988	39.7788187	39.4678090	38.1217358	39.6758029		
CaO									7	2	9	5	1	39.6061446	
Na2	0.31673368	0.29167563	0.29167563	0.2856617	0.29768956	0.26561527	0.27062688	0.25458973	0.31071974	0.27162919	0.26561526	0.32374992	0.33778243	0.31071974	
O	0.00517492	0.02069966	0.0124198	0.02587458	0.03932936	0.00206997	0	0.00517492	6	8	7	9	4	6	
NiO									0.04553926	0.02380461	0.01862969	0.05071417	0.00206996		
Tota									1	4	8	7	6	0	
le	99.6177997	99.410512	100.443782	99.1748004	99.3773537	99.8292697	100.017061	99.052788	99.3585057	99.7849653	99.9504985	101.046987	100.007919	99.3872849	
Fo	77.3184328	76.789102	76.9072631	77.2268068	77.2122593	77.1153013	76.926734	77.1463577	77.2931053	77.2610331	76.7997603	76.5753965	76.9312396		
Fa	22.4269429	22.7930532	22.723558	22.3889085	22.5268801	22.524024	22.722482	22.5494285	22.3652607	22.4360358	22.8874060	23.0932351	22.7454024	22.4548841	
									3	6	9	6	5	8	
Dat	B23_2OI	B23_2OI	B23_2OI	B23_2OI	B23_2OI	B23_2OI	B23_2OI	B23_2OI	B23_2OI	B23_2OI	B23_2OI	B23_2OI	B23_2OI	B23_2OI	B23_2OI
a	from Mi to	from Mi to	from Mi to	from Mi to	from Mi to	from Mi to	from Mi to	from Mi to	from Mi to	from Mi to	from Mi to	from Mi to	from Mi to	from Mi to	from Mi to
set	r	r	r	r	r	r	r	r	r	r	r	r	r	r	r
SiO2	38.2896471	38.8322613	38.3623275	38.7824802	38.8890118	38.3862225	38.7307078	38.7974145	38.5843513	38.7745152	38.7207516	38.6809267	38.3862224	38.7595808	38.9358060
TiO2	0.0203337	0.04270077	0.02135039	0.02338376	0.02846718	0.02541713	0.0406674	0.02440044	0.01830033	0.03558397	0.01626696	0.02338375	0.02236707	0.02846718	0.02033370
Al2									2	9	2	8	3	3	2
O3	0.08583691	0.07459636	0.03372164	0.01839362	0.00204374	0.03474351	0.04394032	0.03065604	0.01430615	0.09196811	0.00408747	0.00817494	0.02656856	0.06642141	
Cr2									0.02963417	1	5	2	4	7	6
O3	0	0.00506426	0.02633413	0.04557831	0.02633413	0	0	0.06887388	0.02633413	0.01417991	0.02126987		0.05266826	0.06887388	
FeO	20.4498751	20.3254854	20.5977154	20.5732453	20.6405381	20.8464948	20.6293226	20.4294834	20.6782628	20.5569318	20.7139483	20.5763040	20.6170875	20.8169267	
Mn									1	5	9	2	20.7384185	3	
o									0.27345399	0.33996983	0.32202111	0.30090497		0.32307692	
Mg	0.33574661	0.36003017	0.29351433	0.33046757	0.28506787	0.29984917	0.29984917	0.28612368	7	4	6	7	0.29668175	0.29668175	3
O	39.3402657	39.5080343	39.6826706	39.6463698	39.5835792	39.6316533	39.4629036	39.3795098	39.4236594	39.3186814	39.4118862	39.4884122	39.2686451	39.3873586	39.6885572
CaO	0.28065009	0.27664081	0.23354097	0.30671046	0.24356419	0.33377315	0.27664081	0.24556883	6	6	2	5	9	5	5
Na2									0.27463616	0.23654793	0.27363384	0.30871510	0.25859901	0.22652471	
O	0.0310495	0.00310495	0.00724488	0.02069966	0.02173465	0.05071418	0.06416896	0	3	1	1	3	5	8	0.31272439
									0.02069966	0.01345478	0.01655973	0.01655973	0.00827986	0.04760922	
									4	2	0	1	1	6	7





Data Set	B23_1Pl r-r	B23_1Pl r-r	B23_1Pl r-r	B23_1Pl r-r	B23_1Pl r-r	B23_1Pl r-r	B23_1Pl r-r	B23_1Pl r-r	B23_1Pl r-r	B23_3Pl r-r	B23_3Pl r-r	B23_3Pl r-r	B23_3Pl r-r	B23_3Pl r-r	B23_3Pl r-r
SiO2	48.2797225	48.283705	48.2010683	48.2936612	48.6729933	47.9690883	48.1662215	47.9790445	45.9798347	46.0983137	45.8294957	46.6528754	46.9635096	47.7610032	
TiO2	0.05795105	0.05286763	0.06303448	0.05591768	0.05896774	0.04371746	0.07930144	0.07320133	0.07116795	0.05286762	0.03253392	0.05693436	0.07320132	0.06303447	
Al2O3	32.3615359	32.12855	32.2276712	32.4800726	31.8700175	32.0815441	32.3257706	32.4575915	32.2051901	32.2378899	32.6578776	32.0672379	32.1285500	31.4111987	
Cr2O3	0.03139839	0	0	0.05165541	0	0.1266064	0.06178393	0.02532128	0.61990946	0	0	0	0	0.01316706	
FeO	0.6250074	0.67394762	0.66579092	0.64641875	0.69128062	0.66375174	0.66273215	0.74226001	0.66477133	0.55771459	0.64030122	0.66783009	0.60665481	0.60665481	
Mno	0	0.03589744	0.0158371	0	0.05806938	0.04751131	0.01794872	0	0.03800905	0.04012066	0.05490196	0.02956259	0.02956259	0.02956259	
MgO	0.15010875	0.11478904	0.13539221	0.07358272	0.15991978	0.11675125	0.13048669	0.11184574	0.13343	0.11577014	0.10203470	0.11577014	0.10497801	0.12852448	
CaO	16.4070049	16.4491024	16.2095475	16.2656776	16.2757008	16.4791721	16.4581233	16.0692225	17.4283708	17.2529644	17.2649923	16.8319893	15.8016025	15.8016025	
Na2O	2.34216697	2.32250229	2.31939734	2.56054843	2.61022762	2.34009701	2.37528643	2.35769172	1.80708066	1.94783837	2.10929575	2.16932477	2.52949893	2.52949893	
K2O	0.08541667	0.10104167	0.09583333	0.096875	0.09895833	0.09479167	0.084375	0.08229167	0.05520833	0.07604166	0.06979166	0.09479166	0.09791666	0.09791666	
SrO															
BaO															
Totale	100.340313	100.162403	99.9335725	100.524409	100.496135	99.9630313	100.36203	99.8984703	98.3382011	98.4464573	98.4651402	98.4787612	99.0504669	98.4593532	
Ab %	20.4290451	20.2330302	20.4534983	22.0495331	22.3683322	20.3329949	20.6088426	20.8798705	15.7488969	16.6178594	18.5144208	18.8062670	22.3351115	22.3351115	
An %	79.0807418	79.1877847	78.9904424	77.4015707	77.0736864	79.1250669	78.909472	78.6406072	83.9345187	82.6752069	82.9827242	80.9381159	77.1020581	77.1020581	
Or %	0.49021313	0.5791851	0.55605934	0.5488962	0.55798137	0.54193819	0.48168542	0.47952223	0.9	0.43387445	0.39941627	0.54746318	0.55853031	0.56283026	
Data Set	B23_3Pl r-r	B23_3Pl r-r	B23_3Pl r-r	B23_3Pl r-r	B23_3Pl r-r	B23_3Pl r-r	B23_3Pl r-r	B23_3Pl r-r	B23_Salt Pl r-r	B23_Salt Pl r-r	B23_Salt Pl r-r	B23_Salt Pl r-r	B23_Salt Pl r-r	B23_Salt Pl r-r	B23_Salt Pl r-r
SiO2	48.8312973	49.0503342	47.6355548	48.0377862	48.5664617	49.2604106	47.8068018	46.7663764	48.4390220	47.1467041	48.8611659	45.7000647	47.3089906	47.3697236	

TiO2	0.06506785	0.09150166	0.0406674	0.04981757	0.05998442	0.08031812	0.05693437	0.05388431	0.06201779	0.05591768	0.05083425	0.05693436	0.06405116	0.05185094
									2	1	6	6	2	1
Al2O3	31.2425905	30.8644994	31.9170234	31.6717751	30.9697518	31.107704	32.3492735	32.3605141	33.7471888	32.1561404	31.0586543	30.3086032	32.3676671	32.1816871
									7	8	1	3	4	8
Cr2O3	0	0.03646264	0	0	0	0	0.01519277	0	0	0.04659115	0.02532128	0.01012851	0.00303855	
									0	7	1	2	4	0
FeO	0.64641875	0.66171257	0.7096332	0.65457545	0.74124043	0.58116512	0.70249608	0.69331979	0.65865380	0.58422388	0.69026102	0.69637855	0.71473113	
									3	4	9	6	9	0.60563523
Mno	0.0306184	0.0147813	0.0158371	0	0	0.01055807	0	0.04223228	0	0.00211161	0.00527903	0.01266968		0.04012066
									0	4	5	3	0	4
MgO	0.1667875	0.15403316	0.15893868	0.13048669	0.1344111	0.17169301	0.12263787	0.10203471	0.11577014	0.06965830			0.11871345	0.13637330
									7	9	0.15697647	0.1667875	6	9
CaO	15.4948921	14.9817034	16.1243502	15.4517923	15.5951243	15.503913	16.2446288	16.693669	16.9372331	16.8730845	15.3164788	16.0431621		16.3518772
									4	5	5	3	16.9031542	3
Na2O	2.83378399	3.13082417	2.44773526	2.7344256	2.85344867	3.06458524	2.41668576	2.1310304	1.93955850	2.06996639	2.85758860	2.37632141	2.16932477	2.39081118
									9	2	4	8	8	2
K2O	0.10520833	0.11979167	0.09583333	0.125	0.11875	0.12083333	0.09375	0.078125	0.07083333		0.13020833		0.08854166	0.11666666
									3	0.0875	3	0.134375	7	7
SrO														
BaO														
Totale	99.4166648	99.1056442	99.1455733	98.8556589	99.0391725	99.9011805	99.808401	98.9211859	101.970277	99.0918982	99.1527681	95.5054251	99.7382127	
									7	3	5	9	3	99.244746
Ab %	24.7157627	27.2517145	21.4316677	24.0805262	24.7063471	26.1671824	21.0969184	18.6811826	17.0951166	18.0757624	25.0508426	20.9733487	18.7523642	20.7831028
									82.4940942	81.4214859	74.1980992		80.7440293	
An %	74.6804701	72.0622069	78.0162292	75.1951679	74.6171274	73.1539515	78.3645859	80.8681908	5	4	8	78.2462945	2	78.5495927
									0.41078915		0.75105807	0.78035672	0.50360641	0.66730442
Or %	0.60376721	0.68607859	0.55210315	0.72430583	0.67652554	0.67886609	0.53849568	0.45062663	1	0.50275163	1	6	1	1
Data Set	B23_5alt PI r-r	B23_5alt PI r-r	B23_5alt PI r-r	B23_5alt PI r-r	B23_5alt PI r-r	B23_5alt PI r-r	B23_5alt PI r-r	B23_5alt PI r-r	B23_5alt PI r-r	B23_5alt PI r-r	B23_5alt PI r-r	B23_5alt PI r-r	B23_5alt PI r-r	B23_5alt PI r-r
SiO2	47.3508068	47.1705991	48.1273923	47.5917474	47.8824691	47.511102	47.5369882	47.2850957	47.7440776	48.2408932	48.6729933	47.6823490		
									3	2	5	4		
TiO2	0.08133481	0.03355061	0.02643381	0.08336818	0.07015127	0.03761735	0.05591768	0.03965072	0.06811790	0.06201779	0.07523469	0.06913458		
									3	2	8	8		
Al2O3	32.1418343	32.9378695	32.1837309	32.329858	31.7780493	31.9732261	32.2205182	32.647659	31.7300215	31.8976078	31.8730830	32.5863468		
									4	9	6	8		
Cr2O3	0.04861686	0	0	0	0	0.02329558	0.00810281	0.01519277	0.03342409	0.02835983	0.02025702	0.03646264		
									1	4	5	4		
FeO	0.65559504	0.60767441	0.60563523	0.67496721	0.67190845	0.69841773	0.62092905	0.61787029	0.74735795	0.62500740	0.60665481	0.73206413		
									3	1	8	4		

Mno	0	0.05067873	0.03589744	0.00527903	0	0.02217195	0	0.0453997	0.05490196	0.02745098	0.04012066	0.02533936
									1	4	4	7
MgO	0.0990914	0.1000725	0.12460007	0.11773235	0.10301581	0.14814654	0.13833551	0.14814654	0.14422213	0.12558117	0.11871345	0.12067566
									2	6	6	2
CaO	16.6365366	16.9673028	16.344861	16.3739283	15.9389207	16.2917379	16.4450932	16.8319894	16.3608981	15.8026049	15.9699926	16.0762387
									3	4	4	5
Na2O	2.25315842	2.15483501	2.43842041	2.30387259	2.58538802	2.42393064	2.20761916	2.1620799	2.38667125	2.67750152	2.68060647	2.28420791
									8	7	7	3
K2O	0.09479167	0.09583333	0.10208333	0.09375	0.09583333	0.08333333	0.075	0.0875	0.11770833	0.10833333	0.115625	0.09583333
									3	3	3	3
SrO												
BaO												
Totale	99.3617659	100.118416	99.9890544	99.5745031	99.125736	99.2129791	99.3085037	99.8805839	99.3874009	99.5953580	100.173281	99.7086523
									2	5	2	2
Ab %	19.5775536	18.5857292	21.1342506	20.1849782	22.5673938	21.1114665	19.4597624	18.7663972	20.7443550	23.3205255	23.1451701	20.3384187
									5	3	7	1
An %	79.8805092	80.8704019	78.2835856	79.2745747	76.8821987	78.4109723	80.1052402	80.7338798	78.5824720	76.0586300	76.1979420	79.1001323
									2	1	8	7
Or %	0.54193719	0.54386897	0.58216382	0.54044712	0.55040754	0.47756124	0.43499748	0.49972298	0.67317293	0.62084446	0.65688775	0.56144892
									9	8	3	8

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: sub-ophitic gabbro B23. Glass.

Data set	B23_1_Glass	B23_1_Glass	B23_1_Glass	B23_1_Glass	B23_5_Glass	B23_5_Glass	B23_5_Glass
SiO <sub>2</sub>	49.614913	49.2602139	49.051391	49.0304088	49.3231605	48.94648	49.0364037
TiO <sub>2</sub>	3.26352035	3.25846376	3.2311582	3.31610884	3.25441849	3.26958825	3.22205634
Al <sub>2</sub> O <sub>3</sub>	14.6177439	14.3343874	14.1180614	14.2978253	14.3770432	14.6279001	14.4339177
Cr <sub>2</sub> O <sub>3</sub>	0.051	0	0.044	0	0.066	0	0.015
FeO <sub>tot</sub>	11.716904	12.0686156	11.9909118	12.3119509	11.5134429	11.7250833	11.8109664
MnO	0.19176707	0.20883534	0.17570281	0.20180723	0.16666667	0.25502008	0.20180723
MgO	5.47619798	5.81068308	5.54986434	5.57176515	5.42542792	5.56280573	5.37863983
CaO	10.5526848	10.2742365	10.6519137	10.6620391	10.7531676	10.3025876	10.429155
Na <sub>2</sub> O	1.39121129	2.60204199	1.17040056	1.68147893	3.43448809	3.25514416	3.25618083
K <sub>2</sub> O	1.02739808	0.99525949	1.02739808	1.05020611	0.9776351	0.98281875	0.95171689
P <sub>2</sub> O <sub>5</sub>	0.41370031	0.43810398	0.40788991	0.50550459	0.34165138	0.40905199	0.42299694
F							
Cl							
Total	98.3170408	99.250841	97.4186919	98.6290949	99.6331018	99.3364799	99.1588408

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: sub-ophitic gabbro B23. Melt inclusions (MI).

Data set	B23_1_MI in Cpx	B23_1_MI in Cpx	B23_1_MI in Cpx	B23_1_MI in Cpx	B23_1_MI in Cpx	B23_2_MI in Cpx	B23_2_MI in Cpx	B23_2_MI in Cpx	B23_2_MI in Cpx	B23_2_MI in Cpx	B23_4_MI in Ol	B23_4_MI in Ol	B23_4_MI in Ol	B23_4_MI1 in Ol
SiO2	46.639437	46.6334421	46.4695811	46.8282768	47.0251098	48.1791309	48.4059385	48.0622301	48.36697158	47.79645553	48.29803006	48.0162690	48.4109342	
TiO2	3.85210691	3.87840116	3.97043101	3.94312545	4.0938117	3.20385264	3.22711293	3.13710571	3.194750783	3.20284132	3.18160366	7	9	47.56964793
Al2O3	14.2145449	14.0906398	14.1922013	13.924079	14.0500153	14.5212605	14.3638402	14.0266561	14.22876346	14.0530621	14.39430868	3.20789790	3.25441849	3.764122321
Cr2O3	0.058	0	0	0.094	0	0.011	0	0.029	0	0.08	0.007	14.2815754	3	14.47962032
FeOtot	13.6267798	13.6922145	13.4366101	13.5184035	13.3558391	11.6708952	11.5635414	11.4848152	11.37234929	11.36416995	11.59625871	11.1924038	11.5318464	11.29464556
MnO	0.19578313	0.16064257	0.18473896	0.24598394	0.21084337	0.19678715	0.22188755	0.19879518	0.257028112	0.128514056	0.146586345	0.26104417	0.15963855	0.206827309
MgO	5.16859115	5.30596896	5.14669034	5.06306907	5.04216375	5.34379763	5.50506723	5.3816263	5.414477517	5.345788613	5.359725492	5.38162630	5.38162630	5.142708375
CaO	10.1496942	9.81758127	9.8287192	10.0342647	10.1253932	10.3775155	10.2742365	10.2701863	10.26512365	10.47674436	10.28841205	10.2033587	10.8280955	9.598872772
Na2O	3.01256335	2.93584976	2.99286662	3.01671003	3.16599053	3.05506683	3.19087061	2.85395283	2.925483063	3.216787365	3.151477148	3.01360002	3.06439685	3.270694211
K2O	0.80553815	0.79828105	0.79413414	0.83352982	0.81072179	0.94549652	0.95690053	0.96415763	0.985928934	0.975561647	0.918541572	0.93201904	0.92890885	1.237853992
P2O5	0.33119266	0.33932722	0.3346789	0.35327217	0.30911315	0.33351682	0.44740061	0.40324159	0.373027523	0.373027523	0.42648318	0.39278287	0.35094801	0.399755352
F														
Cl														
Total	98.0542311	97.6523484	97.3506517	97.8547145	98.1890018	97.8383197	98.1567961	96.811767	97.38390391	97.01295247	97.7684269	96.7982813	98.1923887	97.04874814
												6	2	
Data set	B23_4_MI1 in Ol	B23_4_MI1 in Ol	B23_4_MI in Cpx	B23_4_MI in Cpx	B23_4_MI in Cpx	B23_6_MI in Cpx	B23_6_MI in Cpx	B23_6_MI in Cpx	B23_6_MI1 in Cpx	B23_6_MI1 in Cpx	B23_6_MI1 in Cpx	B23_8_MI in Cpx	B23_8_MI in Cpx	B23_8_MI1 in Cpx
SiO2	48.3399945	47.9663114	45.1926643	45.7062286	45.6453056	46.3177099	46.2497675	45.9010633	48.31101904	47.98529535	48.26379253	47.3058716	47.5057021	48.96146729
TiO2	3.7509752	3.7985071	4.24348664	4.27281483	4.22839339	4.04729111	4.17673971	4.00178184	3.565904166	3.454659282	3.680400115	8	7	48.96146729
Al2O3	14.3181376	14.4257927	13.061822	13.2842417	13.4610664	13.6671285	13.8946262	13.7747836	14.44915188	14.5608695	14.81197586	3.64680953	3.61950397	3.301950397
Cr2O3	0	0.007	0.004	0.062	0	0	0	0	0.12	0	0.033424091	13.9454069	13.9088448	14.44610503
												3	3	

FeOt													13.3323235	13.2525749			
ot	11.3549682	11.3641699	15.0980385	14.5275295	14.9022972	14.249432	14.2555665	13.8936307	13.13295214	13.1482884	13.7440453		4	8	12.42339443		
MnO	0.18273092	0.20381526	0.31526104	0.23293173	0.29457014	0.22791165	0.17771084	0.20582329	0.240963855	0.16064257	0.249170437		0.17068273	0.21385542	0.201807229		
MgO	5.21139728	5.22832063	5.42940989	5.29601404	5.35584095	5.09890675	5.30397798	5.37465786	4.492652516	4.84704744	4.201082791		4.95057854	5.21537924	4.967501895		
CaO	9.89352171	10.0535029	10.1122302	10.3663776	10.454216	10.1010923	10.2438603	10.2570233	9.505719154	9.279922885	8.848496517		9.94111106	9.56748405	9.474330435		
Na2O	3.52364172	3.09238695	2.17908058	2.30555433	2.5377788	2.65491216	2.76790921	2.59996865	3.383691254	3.282097582	3.129789184		2.97005987	2.97213321	3.116230364		
K2O	1.21815615	1.18083392	0.75577518	0.67283689	0.728125	0.69460819	0.69046127	0.66661651	0.915431386	0.894696814	0.964583333		0.86877859	0.83456655	0.962084175		
P2O5	0.53920489	0.53688073	0.30330275	0.33932722		0.20801223	0.24752294	0.22079511	0.327706422	0.231253823			0.29633027	0.31492354	0.413700306		
F																	
Cl																	
Total	98.3327281	97.8575217	96.6950711	97.0658565	97.6075934	97.2670048	98.0081425	96.8961443	98.44519181	97.84477365	97.92676015		97.4279527	97.4089679	98.26857154		

Data set	B23_8_MI in Cpx	B23_8_MI in Cpx	B23_9_MI in Ol	B23_9_MI in Ol	B23_9_MI in Ol	B23_9_MI in Ol
SiO2	48.9684614	48.3539826	48.0902064	47.8014513	47.9952869	47.9703081
TiO2	3.28779196	3.19778473	3.70445461	3.65186612	3.74389598	3.63467373
Al2O3	14.1779827	14.2419665	14.2886847	14.2693881	14.3861838	14.441027
Cr2O3	0	0	0	0.033	0.022	0
FeOt	12.0246516	12.1391624	11.263973	11.5001515	11.4899273	11.2118297
MnO	0.24297189	0.1686747	0.15863454	0.21485944	0.21987952	0.23493976
MgO	5.18252803	5.44931971	5.00035311	5.05211866	5.0939293	4.68577784
CaO	9.99072549	9.79733048	10.0757788	9.73759066	9.98768787	10.0443901
Na2O	2.97835324	3.02189338	3.38680126	3.26862087	3.2986843	3.40338799
K2O	0.93927614	0.86670514	1.27724968	1.16528299	1.28243332	1.17772373
P2O5	0.34281346	0.47993884	0.55431193	0.53688073	0.56012232	0.51828746
F						

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Cl						
Total	98.135559	97.7167584	97.8004481	97.2312103	98.0800305	97.3223453

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: sub-ophitic gabbro B33. Clinopyroxenes.

Data Set	B33_5_Cpx of MI	B33_5_Cpx of MI	B33_5_Cpx of MI	B33_10Cpx c-r	B33_10Cpx c-r	B33_10Cpx c-r	B33_10Cpx c-r	B33_10Cpx c-r	B33_10Cpx c-r	B33_10Cpx c-r	B33_10Cpx c-r	B33_10Cpx c-r	B33_6Cpc r-r	B33_6Cpc r-r	B33_6Cpc r-r	B33_6Cpc r-r
SiO2	50.6572688	50.3814814	51.0226623	51.714619 8	51.809204 8	51.6509 51.6509	8	51.356195 8	51.833098 9	50.025048 6	51.761414 1	51.845046 38	50.524851 05	50.797651 6	50.804620 95	50.871327 65
TiO2	1.37455827	1.32067396	1.37760833	1.0258352 8	1.0227852 3	1.1000532 9	1.0136350 6	1.0614192 1.0380355	1.0583692 6	1.0492190 04	1.0675193 38	1.0258352 7	1.0878530 81	1.0705694 73	1.0705694 26	
Al2O3	3.72266492	3.80237062	3.72164305	2.7794808 1	2.8377272 8	2.7283874 1	2.8254648 7	2.6190475 4	2.7723277 33	2.8745145 28	2.7610871 86	3.1810749 11	3.1350908 54	3.1310033 82	3.1902717 23	
FeO	6.31634723	6.26128948	6.57838132	6.8128865 5	6.5977535 5	6.6436349 5	6.5426957 5	6.5121081 1	6.0736852 99	6.0685873 6	6.1267038 72	5.5832635 05	5.7178491 12	5.9350213 41	5.8728264 77	
Mno	0.1668175	0.15098039	0.12352941	0.1815987 9	0.1868778 3	0.1499245 9	0.1604826 5	0.1562594 3	0.1171945 7	0.1699849 17	0.1921568 63	0.1657616 89	0.0971342 38	0.1076923 08	0.0918552 04	
MgO	15.0422703	15.1953223	14.9706498	15.615234 4	15.751607 7	15.627988 7	15.667232 9	15.506332 15.506332	16.539433 37	15.666251 75	15.817341 61	15.839906 97	15.578933 59	15.303243 66	15.289508 22	
CaO	22.5402118	22.3487683	22.1432924	22.089167 0.2846203	5 0.3115299	3 0.2701306	1 0.246326	4 0.2587458	71 63	19 86	5 13	22 5	69 12	95 68	11 53	
Na2O	0.26806065	0.22355637	0.28048045	8 0.0041666	4 0.015625	1 0	0.246326 0	0.2587458 0	63 0.0125	86 0	13 0	5 0	12 0.003125	68 0.009375	53 33	0.0020833
K2O	0.00416667	0.00833333	0	7 0.0587453	0.015625 0.1114136	0 0.1114136	0 0.0182313	0 0.0688738	0.0125 0.0688738	0 0.0688738	0 0.0941951	0 0.0638096	0 0.1630690	0.003125 0.1377477	0.009375 0.0911566	33 0.0587453
Cr2O3	0.0810281	0.04051405	0.05368112	7 0.0435526	4 0.0435526	2 0.0435526	8 0.0435526	84 0.0435526	64 0.0435526	28 0.0435526	48 0.0435526	67 0.0435526	11 0.0435526	71 0.0435526		
Totale	100.092366	99.6927763	100.218247	100.50761	100.61125	100.14592	100.03250	100.18218	98.33074	99.842848	100.37356	99.692271	98.869666	99.435843	99.356716	
Fe2O3	1.73926725	1.92795705	0.8575335	1.8469754 2	1.9351449 6	1.3659952 1	2.1218647 5	1.1845372 7	4.0766992 84	0.5774230 46	1.5474362 24	3.5808681 68	1.8077817 34	2.1240840 11	1.7473816 6	
FeO	4.7513354	4.52649249	5.80676324	5.1509577 3	4.8564887 5	5.4144974 2	4.6334184 8	5.4462484 3	2.4054263 93	5.5490157 44	4.7343037 17	2.3611588 97	4.0911872 01	4.0237471 58	4.3005132 18	
Wo	46.1084003	45.8123694	45.5476219	44.490680 9	44.415247 1	44.488897 2	44.824248 1	45.071273 5	43.150441 61	44.961878 77	44.974635 59	46.226176 5	45.479623 86	46.391868 96	46.408210 17	
En	42.8140536	43.3401066	42.8464714	43.761189 7	44.090533 8	44.022902 7	43.974706 2	43.688059 3	46.295186 34	44.518636 54	44.374840 18	44.141488 83	44.344196 16	43.297087 33	43.408865 8	
Fs	10.0852439	10.0182367	10.5618721	10.710733 6	10.360101 3	10.498537 3	10.301840 5	10.292543 8	9.5370694 2	9.6741275 73	9.6422335 34	8.7283052 08	9.1301770 58	9.4198702 68	9.3536160 57	
N° Mg	80.9350136	81.2246106	80.2243032	80.337148 1	80.973405 9	81.019720 80.744204	80.932884 1	82.918351 7	82.148672 96	82.149655 92	83.490941 89	82.926069 54	82.131233 99	82.272221 1	82.272221 23	
Data Set	B33_6Cpc r-r	B33_6Cpc r-r	B33_6Cpc r-r	B33_6Cpc r-r	B33_6Cpc r-r	B33_6Cpc r-r	B33_6Cpc r-r	B33_6Cpc r-r	B33_6Cpc r-r	B33_6Cpc r-r	B33_6Cpc r-r	B33_6Cpc r-r	B33_6Cpc r-r	B33_6Cpc r-r	B33_6Cpc r-r	B33_6Cpc r-r
SiO2	50.7289536	50.4840305	50.8972138	50.360573 4	50.286897 3	50.891240 1	50.826524 6	50.595540 2	50.802629 71	50.933056 24	51.078417 12	50.469096 2	50.765791 68	50.705058 71	50.746874 85	



TiO2	1.09801992	1.06040257	1.120387	1.1234370	1.0888697	1.1112368	1.0675193	1.0898864	0.9414504	0.9699175	0.9689009	0.9201000	0.9841511	0.9068831	1.0421022
				5	6	3	7	4	16	99	14	29	91	22	42
				3.2638462	3.2270589		3.1228284	3.1912935	3.1115878	3.0942161	2.9664826	2.8918862	3.1023910	3.0625382	3.0819537
Al2O3	3.21377469	3.15450634	3.0666257	1	7	3.1432658	4	9	91	36	43	83	8	3	21
						5.8351017	5.8809831	5.8208274	5.7025552	5.6230274	5.5914202	5.6627913	5.6169099	5.7667893	5.6505563
FeO	6.09305747	6.12772346	5.89219865	6.0400389	5.9217667	2	8	9	93	34	08	63	07	32	08
				0.1150829	0.1425339	0.1245852	0.1319758	0.0960784	0.1171945	0.0960784	0.1636500	0.1562594	0.0770739	0.1625942	0.0897435
Mno	0.14042232	0.17420814	0.14464555	6	4	2	7	3	7	31	75	27	06	68	9
				15.132531	15.191397	15.426862	15.572065	15.665270	15.670176	15.823228	15.758475	15.589725	15.719231	15.363090	15.594631
MgO	15.4062595	15.3670154	15.3218846	8	9	6	9	6	16	22	43	72	31	94	24
				22.572286	22.848926	22.627413		22.356786	22.624406	22.616388	22.515153	22.509139	22.533195	22.478067	22.567274
CaO	23.1646582	22.4008891	22.4991166	1	9	8	22.547228	9	82	24	75	82	54	84	48
				0.2918652	0.2918652	0.2515009	0.1976817		0.2411510	0.2918652	0.2401161	0.2401161	0.2380461	0.2339062	0.2556408
Na2O	0.26909563	0.27634051	0.27530553	6	6	2	9	0.2339062	85	61	01	01	35	02	49
				0.0104166				0.0177083		0.0052083	0.0020833	0.0072916			0.0020833
K2O	0.00416667	0.00416667	0.00208333	7	0	0	0.003125	3	0	33	33	67	0	0	33
				0.1448377		0.0283598	0.1397734	0.0769766	0.1377477	0.0101285	0.0992594	0.0536811	0.0891309	0.0658353	0.1144521
Cr2O3	0.03342409	0.06380963	0.13268351	3	0.0790024	3	7	9	67	12	21	15	08	3	89
Totale	100.118408	99.0492827	99.2194608	98.910078	98.999316		99.349932	99.067298	99.211151	99.452985	99.284699	98.446406	99.036790	98.678928	99.030860
				2	7	99.411207	2	3	94	91	57	6	75	65	61
Fe2O3	3.2955942	2.58341557	1.66313062	2.3373185	2.8962974	1.9372972	1.9452615		2.4073543	2.7647747	1.9936332	2.6338473	2.2629096	1.8966734	2.1612318
				3	5	7	5	2.3216939	73	85	31	86	21	73	59
				3.9368939	3.3156463	4.0919003	4.1306154	3.7317417	3.5363913	3.1352526	3.7975270	3.2928264	3.5807189	4.0601416	3.7058560
FeO	3.12764552	3.80313769	4.39569541	4	6	1	1	6	19	39	74	83	12	56	83
					46.500867	46.082485		45.514544	45.877023	45.643597	45.725035	45.885312	45.780653	46.081686	45.925644
Wo	46.4745535	45.655033	45.9961065	46.188118	1	3	45.862887	3	29	07	12	84	54	56	63
				43.084118	43.017372	43.714899	44.072264	44.374105	44.212269	44.432689	44.529131	44.218516	44.436659	43.822675	44.157264
En	43.0068302	43.5776523	43.5831575	6	6	6	2	6	3	55	82	77	69	36	1
				9.6470116	9.4068643	9.2757226	9.3371968		9.0258036	8.8577855	8.8633848	9.0103924	8.9074831	9.2278775	8.9756457
Fs	9.5416383	9.74812097	9.40223997	5	8	2	8	9.2496199	45	36	96	22	04	07	48
				81.705281	82.056268	82.495539		82.750881	83.046336	83.378295	83.399574	83.072370	83.301853	82.605501	83.107182
N° Mg	81.8422143	81.7196819	82.2550355	9	4	3	82.51771	6	68	05	62	71	52	72	02

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: sub-ophitic gabbro B33. Olivines.

Data set	B33_3_OI of MI	B33_3_OI of MI	B33_3_OI of MI	B33_4_OI of MI	B33_4_OI of MI	B33_4_OI of MI	B33_8_OI of MI	B33_8_OI of MI	B33_8_OI of MI	B33_4OI c-r	B33_4OI c-r	B33_4OI c-r	B33_4OI c-r	B33_4OI c-r
SiO2	39.4256523	39.6277636	39.5481138	39.7920413	39.4057398	39.7860676	39.3121513	39.4067355	39.43162603	38.95571851	39.1508605	39.4037486	39.55010509	39.37587117
TiO2	0.0274505	0.02643381	0	0.02338376	0.04270077	0.03050055	0.01321691	0.01728365	0.04168409	0.015250277	0.013216906	0.009150166	0	0.003050055
Al2O3	0.01226242	0.0572246	0.04189659	0.00715308	0.0510934	0.04189659	0.06131208	0.00102187	0.04496219	0.018393623	0.017371755	0.003065604	0.059268341	0.028612302
Cr2O3	0.0435526	0.01924417	0	0	0.07495099	0	0	0	0	0.012154215	0	0.002025702	0.055706818	0.069886735
FeO	17.7602022	17.8244362	17.8927486	18.3230147	18.2618394	18.1507043	18.3230147	18.030393	17.95188468	18.14968474	18.30874046	18.27509406	18.40050337	18.11501875
Mno	0.22277526	0.2745098	0.20588235	0.31568627	0.32096531	0.29457014	0.28084465	0.30723982	0.291402715	0.22066365	0.178431373	0.291402715	0.261840121	0.304072398
MgO	41.5722749	41.0650447	41.2141723	41.6438954	41.6438954	41.5389174	41.5860103	41.3436779	41.41726063	41.38880864	41.40352519	41.44375041	41.29560387	41.52223864
CaO	0.31472903	0.37386602	0.2716292	0.27363384	0.28666402	0.26661759	0.3087151	0.28065009	0.276640806	0.334775469	0.364845121	0.304705816	0.322747607	0.249578119
Na2O	0.00310495	0.04967919	0.04243431	0.0186297	0.04346929	0.0248396	0.02173465	0.02276963	0.010349832	0.036224412	0.018629698	0.008279866	0.018629698	0.010349832
NiO														
Totale	99.3820041	99.3182021	99.2168772	100.397438	100.131318	100.134114	99.9069997	99.4097714	99.46581097	99.13167354		99.74122293	99.96440492	99.67867801
Fo	80.4697847	80.1735871	80.2321907	79.9274722	79.9755458	80.0542396	79.9355519	80.0723011	80.18291028	80.06223519	79.96703215	79.91270222	79.7725419	80.0703949
Fa	19.2852113	19.5219082	19.5400907	19.7282748	19.674234	19.6232119	19.7577324	19.5896123	19.49655784	19.6952419	19.83716362	19.76805024	19.94007383	19.59645172
Data set	B33_4OI c-r	B33_4OI c-r	B33_4OI c-r	B33_4OI c-r	B33_4OI c-r	B33_4OI c-r	B33_4OI c-r	B33_4OI c-r	B33_4OI c-r	B33_4OI c-r	B33_4OI c-r	B33_4OI c-r	B33_4OI c-r	B33_4OI c-r
SiO2	39.3141426	39.1528517	39.3828405	39.3400288	39.2544052	39.1438911	39.3718887	37.9421749	39.49534586	38.42803862	39.15185612	38.77252398	39.2175672	39.03735955
TiO2	0.02338376	0.01525028	0.01728365	0.02135039	0.01423359	0.01931702	0.01525028	0.04473415	0.04575083	0.015250277	0.029483868	0.026433813	0.028467183	0.016266962
Al2O3	0.04904966	0.00510934	0.06335581	0.03167791	0.01941549	0.00919681	0.0255467	0.06233394	0.013284283	0.062333945	0.06948702	0.014306151	0.019415491	0.040874718
Cr2O3	0	0.08204095	0.06786103	0	0	0.03848835	0.05975822	0	0.045578305	0	0.010128512		0.05773252	0
FeO	18.0976858	18.0844311	18.3016033	18.2302322	18.2373693	18.5503828	18.2078013	18.1160383	18.05792183	18.39744461	18.47493329	18.31281881	18.21493837	18.21493837

Mno	0.34208145	0.17104072	0.28612368	0.27556561	0.24494721	0.29034691	0.25972851	0.27134238	0.294570136	0.2206636	0.2892911	0.2354449	0.2892911	0.2607843
MgO	41.5075221	41.8940767	41.6370277	41.7380813	41.6252545	41.5173331	41.4486559	42.7368441	42.11678703	5	01	47	01	14
CaO	0.35682655	0.28265474	0.26461295	0.3407894	0.32174529	0.31172207	0.27062688	0.35381958	0.313726712	41.530087	41.645857	41.096439	41.324055	41.634084
Na2O	0.03829438	0	0.01345478	0.00206997	0.01448976	0.03622441	0.00310495	0.02690956	0.00310495	47	62	97	85	38
NiO										0.3367801	0.3608358	0.2886686	0.2836570	0.3327708
Totale	99.7289862	99.6874555	100.034163	99.9797955	99.7318603	99.9169026	99.6623614	99.5541969	100.3860699	12	34	68	59	25
Fo	80.0464057	80.3549674	79.9690478	80.0785171	80.0559147	79.7050752	80.0005874	80.5538624	80.35363891	0.0382943	0.0165597		0.0248395	0.0393293
Fa	19.5787752	19.4586365	19.7187233	19.6210927	19.6764236	19.9782228	19.7145867	19.1555489	19.32704837	78	31	0	97	61
Data set	B33_40I c-r	B33_40I c-r	B33_40I c-r	B33_40I c-r	B33_40I c-r	B33_40I c-r	B33_40I c-r	B33_40I c-r	B33_40I c-r	99.028893	100.04843	98.746636	99.459964	99.576408
SiO2	39.2802914	38.9606966	35.7627574	39.1767467	38.951736	38.5106753	38.9168892	38.8352482	38.68092672	06	31	34	38	48
TiO2	0	0.0071168	0.0406674	0.03151724	0.0203337	0.02846718	0.00813348	0.0274505	0.051850941	79.902335	79.820645	79.793738	79.920211	80.064820
Al2O3	0.054159	0.03269977	0.01737176	0.02656857	0.02145923	0.00919681	0.06029021	0.04700593	0.00510934	92	07	91	52	75
Cr2O3	0.05064256	0.0020257	0.02835983	0.03848835	0	0.08204095	0	0	0	19.856448	19.864322	19.946525	19.761906	19.650240
FeO	18.1884291	18.0966662	18.3933663	18.0915682	18.2587807	18.1517239	17.9692177	18.2394085	18.1027837	91	19	66	47	6
Mno	0.27662142	0.28717949	0.32518854	0.25972851	0.29562594	0.28401207	0.24600302	0.27134238	0.323076923					
MgO	41.4879	41.5987647	42.2462926	41.5379363	41.2671519	41.468278	41.8911334	41.6066135	41.31130151					
CaO	0.26561527	0.32074296	0.33076618	0.29267795	0.32074296	0.34179172	0.29969421	0.31873832	0.328761538					
Na2O	0.01034983	0	0	0	0	0.03208448	0	0	0.008279866					
NiO														
Totale	99.6140086	99.3058922	97.14477	99.4552318	99.1358304	98.9082704	99.3913612	99.3458073	98.81209054					
Fo	80.0176836	80.1304553	80.0887199	80.1356701	79.8544991	80.035431	80.3878451	80.0239112	79.98290342					
Fa	19.6791874	19.5552422	19.5610164	19.5796363	19.8204775	19.6531248	19.3439382	19.67957	19.66170149					

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: sub-ophitic gabbro B33. Feldspars.

Data Set	B33_10_Pl from MI to r	B33_10_Pl from MI to r	B33_10_Pl from MI to r	B33_10_Pl from MI to r	B33_10_Pl from MI to r	B33_10_Pl from MI to r	B33_10_Pl from MI to r	B33_10_Pl from MI to r	B33_10_Pl from MI to r	B33_10_Pl from MI to r	B33_10_Pl from MI to r	B33_10_Pl from MI to r	B33_10_Pl from MI to r	B33_10_Pl from MI to r
SiO2	47.0640675	50.7399055	48.2996349	48.3623592	48.6889233	48.6690109	46.5921425	46.7793195	46.9067591	46.2008628	46.7594070	46.3243200	46.8032144	46.4338385
TiO2	0.054901	0.0955684	0.07320133	0.06608453	0.06405116	0.06608453	0.07930144	0.04270077	0.02948386	0.05591768	0.05795105	0.0477842	0.04168409	0.05998442
Al2O3	33.4958094	31.2323719	32.1980371	32.0876753	31.8403833	32.3911701	33.6971173	33.30472	33.6460239	33.7308389		33.5693838	33.4201911	33.7706918
Cr2O3	0	0	0.02835983	0.05874537	0.02532128	0	0.00810281	0.03848835	0.05671966	0.03646264				
FeO	0.60563523	0.55567542	0.70351567	0.57504759	0.59849811	0.57708677	0.53834243	0.59034141	0.61175275	0.67088885	0.59543935		0.54547954	0.52304860
MnO	0.04117647	0.02006033	0.00527903	0.0306184	0	0.02322775	0.01055807	0	0.05490196	0.09607843	0.03061840	0.04856711	0.00950226	
MgO	0.14029772	0.18640956	0.15599537	0.18052294	0.15108985	0.1658064	0.10792132	0.12852449	0.10203470	0.10301580	0.11184573	0.14618433	0.13441110	
CaO	17.469466	14.6779999	16.4681466	15.9920437	15.7474772	16.0561923	17.5386262	17.4494196	17.5276006	17.9495781	17.7912112	17.5075542	17.5676935	17.7250580
Na2O	1.84847999	3.30780629	2.3835663	2.56882829	2.50258937	2.63713718	1.70151237	1.69323251	1.83502520	1.51521539	1.80501069	1.75015658	1.74084173	1.58869920
K2O	0.07291667	0.1375	0.09166667	0.11770833	0.1	0.10208333	0.08020833	0.065625	0.06354166	0.04791666	0.04895833	0.05520833	0.06666666	0.06666666
SrO														
BaO														
Total	100.79275	100.953297	100.407403	100.039634	99.7183336	100.687799	100.353833	100.092372	100.817204	100.364617	100.909155	99.9535663	100.380522	100.311900
Ab %	16.0040284	28.7401727	20.6473315	22.3697564	22.2048757	22.7791766	14.8653815	14.8803685	15.8703595	13.2152157	15.4688837	15.270321	15.1474725	13.9024956
An %	83.580584	70.4737521	78.8301999	76.9558002	77.2113152	76.6406308	84.6735425	84.7401606	83.7680512	86.5098058	84.2550470	84.4127310	84.4708454	85.7136452
Or %	0.41538755	0.78607522	0.52246865	0.67444338	0.58380909	0.58019268	0.46107603	0.37947093	0.36158918	0.27497840	0.27606918	0.31694798	0.38168204	0.38385918
Data Set	B33_10_Pl from MI to r	B33_11_Pl r	B33_11_Pl r	B33_11_Pl r	B33_11_Pl r	B33_11_Pl r	B33_11_Pl r	B33_11_Pl r	B33_11_Pl r	B33_11_Pl r	B33_11_Pl r	B33_11_Pl r	B33_11_Pl r	B33_11_Pl r
SiO2	46.3502062	46.3044076	47.0809931	47.0103039	47.1725904	47.0541113	46.669801	46.2745389	47.4065616	47.2841000	47.2363101	48.8671397	49.3659465	49.4057714

TiO2	0.05998442	0.03660066	0.05693437	0.03456729	0.04473415	0.00610011	0.04676752	0.04168409	0.05388431	0.06303447	0.06100110	0.06303447	0.06811790	0.04270077
Al2									1	7	7	7	3	5
O3	33.4620877	32.817289	33.3445729	33.7625169	33.4927437	33.4712845	33.7104016	33.6266084	33.746167	33.7778449	33.2945013	32.4841600		32.0764347
Cr2													32.3053332	8
O3	0	0.03646264	0	0	0	0.03848835	0	0	0	0	0.01012851	0.00303855	0	0
FeO	0.58422388	0.57912594	0.53120531	0.55873419	0.56791048	0.49042179	0.53018572	0.51183314	0.51489190	0.51081355	0.56179294	0.62704657	0.59849811	0.58218470
Mn									5	3	9	7	5	8
o	0	0	0	0	0.03589744	0.02322775	0.02217195	0.00316742	0.01478129		0.00950226	0.04012066	0.00527903	0.02533936
Mg									7	0	2	4	5	7
O	0.10694022	0.13735441	0.12852449	0.12460007	0.11282684	0.13146779	0.11773235	0.0990914	0.15108985	0.12165676	0.12950558	0.16776860	0.14618433	0.16188198
CaO	17.7120279	17.4554335	17.0023841	17.210867	16.9302169	16.9061612	17.5476471	17.6779489	17.5095588	17.5015402	17.2930573	16.0992921	15.4036808	15.6001359
Na2									7	9	6	5	4	1
O	1.67874274	1.75015658	1.96853804	1.75222655	1.89815918	1.86917965	1.66011305	1.70254736	1.72117705		1.90954399	2.49534448	2.75719523	2.58331805
K2O	0.05625	0.08020833	0.0625	0.09166667	0.08854167	0.06041667	0.05833333	0.0625	5	1.67770776	6	5	4	7
SrO									0.07083333		0.05833333	0.09791666		0.11354166
BaO									3	0.0625	3	7	0.1125	7
Tota									101.188945	100.999197	100.563676		100.762735	100.591308
le	100.010463	99.1970387	100.175652	100.545483	100.343621	100.050859	100.363154	99.9999197	2	8	7	100.944862	2	7
Ab									15.0405557	14.7294098	16.5988459	21.7815174	24.3068209	22.9044609
%	14.5934127	15.2867966	17.2600253	15.4745581	16.7799563	16.6130337	14.5683601	14.7886682	8	2	3	5	9	9
An									84.5521688	84.9095453	83.0675148	77.6561071	75.0406117	76.4331554
%	85.0848456	84.252235	82.3794046	83.9927802	82.7050309	83.0336475	85.0948168	84.8541231	7	3	3	1	7	2
Or									0.40727534	0.36104484	0.33363923	0.56237544	0.65256724	0.66238359
%	0.3217417	0.46096842	0.36057012	0.53266179	0.51501276	0.35331881	0.33682313	0.35720867	5	9	9	2	3	1
Dat														
a	B33_11_Pl r-	B33_11_Pl r-												
Set	r	r	B33_7Pl r-r	B33_7Pl r-r	B33_7Pl r-r	B33_7Pl r-r	B33_7Pl r-r	B33_7Pl r-r	B33_7Pl r-r	B33_7Pl r-r	B33_7Pl r-r	B33_7Pl r-r	B33_7Pl r-r	B33_7Pl r-r
SiO2	47.2721526	47.0132908	46.4726678	46.6837397	48.1393397	48.3484204	48.1224141	48.1562653	47.8605654	47.503137	47.5977211	47.4742639	47.5409706	47.9929832
TiO2	0.0681179	0.0477842	0.04981757	0.08031812	0.07015127	0.05693437	0.07218464	0.07625138	5	0.07421801	0.07218464	0.07015127	0.05286762	0.08235149
Al2									0.04676751	3	3	3	6	4
O3	33.5182904	34.0527274	33.8943379	33.3445729	32.5219692	32.4116075	32.6466371	32.810136	32.8367045	32.7876548	32.5883906	32.4228480	32.9531974	32.7406489
Cr2									3	7	2	1	7	4
O3	0.03848835	0	0	0	0	0	0.04659116	0.02329558	0.02329557		0.03342409	0.07697669		0.02025702
									8	0	1	4	0	5



CaO	16.5453253	16.5994507	16.7788663	16.749799	16.862059	16.9061612	16.3598958	17.5757121	17.7320743	16.4180304	15.8246559	15.880786	16.2696868	17.4213545
Na2O	2.19002444	2.25522838	2.16311488	2.14241522	2.06168653	2.10412084	1.91161396	1.58352429	1.56696455	2.26557821	2.47567980	2.43428047	2.61022762	1.77810113
K2O	0.12395833	0.09166667	0.07708333	0.090625	0.08958333	0.06979167	0.08125	0.06354167	0.06458333	0.09270833	0.09583333	0.09583333	0.08541666	0.08541666
SrO														
BaO														
Totale	100.194747	100.164712	100.328359	100.5274	100.111707	99.9027222	99.2202637	100.036544	100.193404	99.9157965	100.010158	100.131556	101.060970	100.602596
Ab%	19.1862695	19.6304716	18.8328792	18.6980013	18.0239232	18.3088121	17.3695292	13.9669173	13.7354103	19.8593842	21.9448089	22.3733898	15.5138872	15.5138872
An%	80.0991857	79.8445231	80.7255404	80.7815815	81.4607699	81.2916066	82.1447091	85.6643204	85.8920993	79.5278031	77.5144763	77.8468710	83.9957483	83.9957483
Or%	0.71454472	0.52500537	0.44158034	0.52041723	0.51530697	0.39958131	0.48576176	0.36876232	0.37249034	0.61281270	0.54071473	0.56398235	0.49036435	0.49036435
Data Set	B33_7Pl r-r	B33_7Pl r-r	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r
SiO2	46.5692432	46.7842976	45.5108965	46.0684451	46.2546265	47.9372284	46.3790793	46.242679	45.5328002	46.2436746	46.9814308	46.7295383	46.4238823	46.6628316
TiO2	0.01321691	0.06100111	0.05185094	0.0681179	0.05083426	0.05185094	0.07320133	0.05185094	0.05185094	0.05998442	0.0477842	0.04676751	0.06811790	0.03558397
Al2O3	33.8279164	33.5908431	33.1524617	33.0676467	33.4794595	33.8085009	33.5520121	33.336398	33.6603300	33.1851615	33.6123023	33.2250143	33.2209268	32.8837104
Cr2O3	0.02025702	0	0.06684818	0	0.04659116	0	0.08811806	0	0.03848834	0.03342409	0	0.03342409	0	0
FeO	0.5995177	0.55159707	0.61073317	0.5403816	0.61787029	0.53426408	0.52712696	0.5322249	0.54853830	0.53732283	0.56281253	0.58218470	0.60359605	0.59645893
MnO	0.00950226	0.00316742	0	0.01794872	0	0.02956259	0	0.01266968	0.04012066	0.03061840	0	0.02850678	0.03589743	0
MgO	0.13048669	0.12263787	0.10497801	0.14716544	0.15108985	0.11773235	0.1344111	0.11577015	0.14324102	0.11969455	0.14324102	0.14127882	0.12754338	0.13833551
CaO	17.3712385	17.6939861	18.1129566	17.4043151	17.6258282	17.3993035	17.2659947	17.8683901	17.1366951	17.3722407	17.3722407	17.2970666	17.2379296	17.0244351
Na2O	1.6570081	1.64148335	1.66839291	1.79880079	1.68909258	1.89401925	1.83916514	1.66735793	18.1881307	1.79052092	1.85675985	1.83295524	1.83709517	2.04409181
K2O	0.05	0.071875	0.06041667	0.07291667	0.059375	0.05520833	0.078125	0.05625	1.52970516	0.08645833	0.07708333	0.08645833	0.07083333	0.07083333
SrO									0.0625	0.071875	0.08645833	0.07708333	0.08645833	0.07083333

BaO														
Totale	100.248387	100.520889	99.3395347	99.185738	99.9747673	101.82767	99.9372336	99.8835906	99.7957054 7	99.2089715 4	100.663029 9	99.9938198 8	99.6414471 2	99.4562808 4
Ab %	14.6777379	14.3154951	14.2386257	15.6903038	14.7284888	16.405242	16.0881984	14.4005291	13.1626066 1	15.8347267 3	16.1267290 9	16.0194783 3	16.0871124 3	17.7771713 5
An %	85.030844	85.2720657	85.4221097	83.8912042	84.9308517	83.2801172	83.4621369	85.2798139	86.4835376 2	83.7470375 4	83.3791769 8	83.5372509 8	83.4147315 2	81.8174955 8
Or %	0.29141816	0.41243912	0.33926464	0.41849203	0.34065947	0.31464077	0.44966473	0.31965701	0.35385577 7	0.41823573 1	0.49409392 7	0.44327077 6	0.49815604 4	0.40533307 5
Data Set	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r	B33_6Pl r-r
SiO2	48.2697663	48.2030596	47.9690883	46.7594071	48.3962103	48.3404555	48.7247657	48.1632347	48.8621616 1	48.5883654 4	48.6859364 4			
TiO2	0.06506785	0.04676752	0.03660066	0.06405116	0.06303448	0.07625138	0.05896774	0.05795105	0.07015127 3	0.05490099 6	0.04473414 5			
Al2O3	31.9446138	32.0641724	32.3022676	33.1299806	31.6452065	31.8189241	31.7821368	31.9916197	31.8853454 7	31.5522165 2	31.8894329 4			
Cr2O3	0.03848835	0	0	0	0.03848835	0	0	0	0.00303855 4	0.01316706 6	0.02835983 4			
FeO	0.54445995	0.59238059	0.57606718	0.57300842	0.50571561	0.54344037	0.57096924	0.61277235	0.54649913 1	0.52304860 8	0.54955789 4			
MnO	0.04223228	0	0.00316742	0.00950226	0.0612368	0	0	0	0.01794871 8	0.05279034 7	0.04539969 8			
MgO	0.17954184	0.19720169	0.14127882	0.12263787	0.17267412	0.2011261	0.16090088	0.15893868	0.15599536 8	0.17463632 3	0.15403316 2			
CaO	15.9309021	15.9880344	16.1794779	17.6528909	16.1063084	15.7264285	15.6071522	16.1313664	15.7635143 6	15.6211846 7	16.0090831 9			
Na2O	2.34941185	2.44256034	2.23349374	1.84434005	2.53984876	2.71165597	2.58952796	2.55537351	2.55123357 8	2.70337610 8	2.49844943 5			
K2O	0.121875	0.12604167	0.103125	0.05625	0.10104167	0.11354167	0.11041667	0.10208333	0.12708333 3	0.10208333 3	0.11354166 7			
SrO														
BaO														
Totale	99.4863593	99.6602182	99.5445666	100.212068	99.629765	99.5318235	99.6048372	99.7733398	99.9829713 9	99.3857694 1	100.018528 4			
Ab %	20.9152174	21.5004885	19.8671686	15.849821	22.0728107	23.627293	22.9431833	22.1498712	22.4861965 3	23.7079650 6	21.8782555 6			
An %	78.3708948	77.7695008	79.5292619	83.8321125	77.3494095	75.7217584	76.4131222	77.2679133	76.7768052 1	75.7029814 6	77.4675460 1			



Or										0.73699829	0.58905351	0.65419842
%		0.7138878	0.73001077	0.60356953	0.31806655	0.57777984	0.65094867	0.64369451	0.5822155	1	7	7

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: sub-ophitic gabbro B33. Glass.

Data set	B33_1_Glass	B33_1_Glass	B33_1_Glass	B33_1_Glass	B33_2_Glass	B33_2_Glass	B33_2_Glass	B33_2_Glass
SiO <sub>2</sub>	49.1622969	49.0304088	48.6876995	48.9344902	48.5498165	48.5707987	48.7176741	48.7896131
TiO <sub>2</sub>	3.03698531	2.94798941	2.9773176	3.00765712	2.99450999	3.02990609	2.97630628	3.01170238
Al <sub>2</sub> O <sub>3</sub>	14.7375865	14.3069658	14.4999326	14.6146971	14.5720413	14.3008721	14.3557153	14.5903223
Cr <sub>2</sub> O <sub>3</sub>	0.018	0.015	0.048	0.018	0	0	0.044	0.066
FeO <sub>tot</sub>	10.6771054	10.4276356	10.5401015	10.7210694	11.1024311	10.9286201	10.7875265	11.1433278
MnO	0.23493976	0.25301205	0.16465863	0.20481928	0.21686747	0.22188755	0.25803213	0.13353414
MgO	6.15512309	6.07150182	6.08046124	6.22978494	6.32336113	6.22381199	6.18001037	6.27159558
CaO	11.6847038	11.3212022	11.4184059	11.2766504	11.0022523	11.1298322	11.0913557	11.4771332
Na <sub>2</sub> O	3.22922741	2.89852964	2.99182995	2.99493996	2.83632943	2.99079328	2.93170308	2.97938991
K <sub>2</sub> O	0.91854157	0.89469681	0.88432953	0.88225607	0.94238633	0.95171689	0.9486067	0.89158663
P <sub>2</sub> O <sub>5</sub>	0.34629969	0.38232416	0.42648318	0.45437309	0.4253211	0.46948012	0.35559633	0.39510703
F								
Cl								
Total	100.200809	98.5492662	98.7192197	99.3387375	98.9653166	98.8177191	98.6465265	99.7493121

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: sub-ophitic gabbro B33. Melt inclusions (MI).

Data set	B33_3_ MI in Ol	B33_3_ MI in Ol	B33_3_ MI in Ol	B33_3_ MI in Ol	B33_3_ MI1 in Ol	B33_3_ MI1 in Ol	B33_4_ MI in Ol	B33_4_ MI in Ol	B33_4_ MI in Ol	B33_4_ MI in Ol	B33_5_M I in Cpx	B33_5_M I in Cpx	B33_5_M I in Cpx	B33_8_ MI in Ol	B33_8_ MI in Ol	B33_8_ MI in Ol	B33_10_ MI in Pl	B33_10_ MI in Pl
SiO2	47.8703 928	48.2810 445	48.0952 021	48.0792 157	47.19196 83	47.89736 99	47.5376 751	48.2290 885	47.6825 5216	45.94502 605	46.06292 604	46.18482 264	48.2111 038	48.4538 9784	48.1171 8347	51.20456 453	51.37242 214	
TiO2	2.98540 814	2.97124 97	2.96417 048	2.91866 121	3.103732 24	3.200818 69	3.32217 674	3.24228 269	3.33936 9131	3.786371 298	3.772212 858	3.796484 469	2.97731 7602	2.88832 1695	2.93079 7014	1.943751 505	1.945774 139	
Al2	14.9244	14.5893	14.3943	14.6959	14.53547	14.59844	14.4146	14.5202	14.7182	13.78595	13.80931	13.89868	14.4907	14.5039	14.4176	14.49587	14.10282	
O3	596	067	087	463	91	73	21	449	898	541	455	865	9208	9508	6782	016	723	
Cr2																		
O3	0	0	0.033	0.022	0	0	0.033	0	0	0.004	0.095	0	0	0	0.052	0.007	0.022	
FeO	10.9173	10.4552	10.8304	10.7476	11.08607	11.42347	10.8161	10.8642	10.7128	14.00711	13.81490	13.98871	10.9562	10.9286	11.1310	9.485989	9.904157	
tot	735	408	68	522	24	02	542	078	9003	905	457	554	2539	2012	5877	094	831	
Mn	0.15963	0.16767	0.22289	0.11947	0.106425	0.236947	0.21184	0.23493	0.13453	0.243975	0.168674	0.254016	0.20783	0.16967	0.23293	0.172690	0.144578	
O	855	068	157	791	7	79	739	976	8153	904	699	064	1325	8715	1727	763	313	
Mg	5.84552	6.08046	5.95602	6.00281	5.862448	5.360720	5.72108	5.80968	5.83855	5.719097	5.940096	6.030686	5.97493	6.16308	6.17403	5.685251	5.888331	
O	528	124	482	291	63	98	886	759	6837	873	956	67	9153	7021	7426	167	405	
CaO	11.4295	11.0518	11.1379	11.3232	11.15717	11.67052	10.7126	10.8736	11.3606	10.65292	10.81088	10.80278	11.3485	11.3991	11.4943	10.34815	9.994775	
Na2	439	667	325	272	08	82	66	598	9119	621	234	203	4072	6769	4638	187	647	
O	3.03226	2.98871	2.99286	2.77412	2.914079	2.937923	3.14836	2.85913	2.83010	2.607225	2.708819	2.719185	2.99493	2.95139	3.03433	3.178430	3.294537	
	008	994	662	923	69	1	714	618	9413	338	009	71	9961	9816	3425	571	624	
	0.93823	0.95068	0.85011	0.86048	0.961047	0.923725	0.97037	1.00147	1.02221	0.693571	0.668689	0.673873	0.82005	0.84804	0.81175	1.316645	1.303167	
K2O	942	016	748	477	45	22	8	986	4436	458	971	614	2352	4025	8523	369	896	
P2O	0.38581	0.33351	0.41834	0.46948	0.496207	0.472966	0.43461	0.46715	0.50085	0.389296	0.323058	0.392782	0.38697	0.39626	0.36024	0.251009	0.241712	
S	04	682	862	012	95	36	774	596	6269	636	104	875	2477	9113	4648	174	538	
F																		
Cl																		
Tota	98.4886	97.8697	97.8953	98.0130	97.41463	98.72291	97.3225	98.1018	98.1400	97.83456	98.17457	98.74203	98.3687	98.7024	98.7563	98.08935	98.21428	
l	517	573	31	876	23	77	921	831	6742	523	91	826	1486	811	592	42	476	



SiO 2	51.9604944	51.9272381	51.8173915	49.6456524	49.5237125	49.465262	49.1387453	48.9442462	49.18510264	49.2979726	49.42092024	49.23851432	49.34432991	49.14176863
TiO 2	1.05508563	1.10220596	1.05303692	1.61745652	1.70350234	1.71477024	1.71681895	1.71681895	1.637943619	1.707599758	1.702477983	1.720916373	1.623602649	1.736281698
Al2 O3	2.74220003	2.69873875	2.69873875	4.75487138	4.86145501	5.0404741	4.86145501	5.03633493	4.830411231	4.730036362	4.805576212	4.677261946	4.751767004	4.898707534
Fe O	6.35775139	6.46170686	6.39583409	6.75607579	6.90120173	6.80548036	6.75710505	6.95369409	6.955752617	6.63873992	6.785924387	6.734461286	7.083381106	6.919728447
Mn o	0.10222222	0.15888889	0.12666667	0.19111111	0.18	0.14333333	0.18222222	0.20888889	0.17888889	0.13888889	0.14111111	0.12222222	0.08333333	0.11888889
Mg O	15.7319496	15.6995652	15.6671808	14.551391	14.4611072	14.3884876	14.227547	14.4424616	14.54942828	14.23441638	14.44540566	14.23834176	14.38750628	14.22656562
Ca O	22.3477363	22.2042419	22.4750623	22.0142634	22.3416732	22.2658839	22.6124936	22.6064304	22.31034693	22.45485185	22.16281044	22.17695777	21.92533732	22.42958875
Na 2O	0.23918537	0.23502563	0.28390264	0.26726366	0.30262149	0.27350328	0.28806239	0.37957679	0.290142258	0.258944166	0.330699778	0.32446016	0.297421813	0.311980923
K2 O	0.01478508	0	0.00422431	0.02006547	0.00422431	0.01689724	0.00316823	0.01161685	0.016897239	0.012672929	0	0.005280387	0.005280387	0.003168232
Cr2 O3	0.22741225	0.30116757	0.18233955	0.20897342	0.27146057	0.22741225	0.28682626	0.30424071	0.214095315	0.345215891	0.226387869	0.231509766	0.314484506	0.229461007
Tot ale	100.55141	100.487611	100.522038	99.8181507	100.279498	100.114092	99.7876177	100.300069	99.9549137	99.47412285	99.79492581	99.23841622	99.50195979	99.78667872
Fe2 O3	1.16417213	0.94242895	1.70934507	1.58450255	2.17587565	1.77049273	2.34868907	3.75928963	2.904995817	1.591311537	2.021728127	1.758890436	1.792075137	2.340754257
Fe O	5.31021649	5.61369891	4.85774657	5.33032294	4.94332486	5.21237151	4.64372876	3.57104389	4.341805393	5.206860277	4.966751156	5.151792305	5.470852164	4.813491991
Wo	45.0264466	44.8411764	45.1484882	45.8457699	46.1632078	46.2976509	46.9098318	46.3092901	45.99246324	46.86784148	46.02173916	46.36442278	45.66714201	46.54396174
En	44.102944	44.1143181	43.7909192	42.1648461	41.5750451	41.6279536	41.0673184	41.1650124	41.73276622	41.33855788	41.73676992	41.41838865	41.69595963	41.0764597
Fs N°	9.99852701	10.1856007	10.0285431	10.9821652	11.1302099	11.0452652	10.9414388	11.1185988	11.19238964	10.81555375	10.99880639	10.98965592	11.5158634	11.20803386
Mg	81.5189369	81.2419596	81.3663261	79.3362507	78.8821629	79.0305862	78.9623144	78.734065	78.85242008	79.26231814	79.14347929	79.03059347	78.35844979	78.56336918
Dat a Set	B34_1_Cpx rim datail to glass	B34_1_Cpx rim datail to glass	B34_1_Cpx rim datail to glass	B34_1_Cpx rim datail to glass	B34_1_Cpx rim datail to glass	B34_1_inter stitial Cpx c-r	B34_1_inter stitial Cpx c-r	B34_1_inter stitial Cpx c-r	B34_1_inter stitial Cpx c-r	B34_1_inter stitial Cpx c-r	B34_1_inter stitial Cpx c-r	B34_2_Cpx c-r	B34_2_Cpx c-r	B34_2_Cpx c-r
SiO 2	49.2556464	49.2838639	48.858586	49.0198288	48.7779645	49.5499145	49.9237962	49.6879786	49.89658652	50.17271483	50.16767599	51.87684974	51.88088081	51.96956435
TiO 2	1.73320863	1.76189057	1.67891782	1.68711266	1.61848087	1.69735621	1.71169718	1.6953075	1.671747334	1.575457965	1.648187169	0.998746109	1.057134343	1.017184499
Al2 O3	5.04461327	5.06220474	4.68864466	4.69174904	4.82627206	4.95458633	5.02081305	4.79212391	4.815924137	4.738314702	4.833515609	2.835331357	2.852922829	2.684251657



Ca														
O	21.8869374	21.8515691	21.7778008	21.9293794	21.7778008	21.7596114	21.8768322	21.932411	22.18908406	21.65552743	21.66866424	22.09914743	22.15472625	21.98596875
Na														
2O	0.24958474	0.25374448	0.2599841	0.28182277	0.26726366	0.19862785	0.21110709	0.21318696	0.232945756	0.132071924	0.230865883	0.285982512	0.274543212	0.241265247
K2														
O	0	0.00422431	0.02112155	0.01161685	0.00528039	0	0.00316823	0.00739254	0	0	0	0.001056077	0.001056077	0.005280387
Cr2														
O3	0.31038699	0.19668086	0.30116757	0.28067998	0.21716845	0.27760684	0.15160817	0.19668086	0.212046556	0.199754002	0.332923337	0.177217653	0.286826259	0.219217213
Tot														
ale	100.402809	99.7029734	100.450827	100.331869	99.621922	100.229269	100.513944	100.523377	100.6412571	99.6212201	99.52881984	100.5198289	100.524142	100.6591842
Fe2														
O3	0.30400512	1.01511043	1.1589545	1.5466726	1.37771286	0.39650688	0.36688783	0.8088145	0.816228061	0	0.999248625	1.567464645	1.924869502	0.749442466
Fe														
O	6.34666605	5.55859193	5.56708056	5.10704726	5.17673813	6.1131597	6.09452371	5.83273581	5.825035749	6.456560546	5.506991792	5.039963026	4.731746606	5.790438058
Wo	44.4313053	44.5721532	44.0413219	44.2666839	44.3782745	44.1871403	44.4775955	44.4190366	44.80804753	44.42300821	44.08234497	44.44498948	44.47714196	44.42220616
En	44.1618445	44.1869367	44.5734754	44.4643238	44.430351	44.8277591	44.5502979	44.4286741	44.00154604	44.74866476	44.89525902	44.38834581	44.39674013	44.50017458
Fs	10.4899737	10.3042837	10.4337617	10.239518	10.2058099	10.2551832	10.1954152	10.3709635	10.33915034	10.33805345	10.17247004	10.12584488	10.12871649	10.19547813
N°														
Mg	80.8058103	81.0900111	81.0320201	81.2819033	81.3204117	81.3822886	81.3767789	81.074759	80.97346735	81.23312881	81.52734784	81.42530458	81.4238759	81.35961887
Dat														
a	B34_2_Cpx	B34_2_Cpx	B34_2_Cpx	B34_2_Cpx	B34_2_Cpx	B34_2_Cpx	B34_2_Cpx	B34_2_Cpx	B34_2_Cpx	B34_2_Cpx	B34_2_Cpx	B34_2_Cpx	B34_2_Cpx	B34_2_Cpx
Set	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r
SiO														
2	51.702506	51.0262939	49.3231668	49.7524758	49.9933322	49.7988331	49.9852701	50.2603906	49.91774964	49.77867772	50.64334227	50.17170706	50.63225683	50.80861615
TiO														
2	1.00898966	1.24971308	1.46585198	1.45458408	1.53038635	1.52219151	1.5201428	1.50580183	1.539605541	1.526288926	1.427950847	1.360343419	1.375708743	1.381854873
Al2														
O3	2.64906871	3.73353122	4.34302398	4.46719908	4.51790391	4.45374677	4.50134723	4.31508458	4.435120509	4.447538019	3.929106993	3.796653558	3.652817405	3.554512121
Fe														
O	6.33613689	6.50596512	6.74475391	6.4009804	6.46376538	6.53684298	6.59345239	6.91972845	6.527579624	6.574925676	6.588306082	6.361868441	6.62330099	7.022654647
Mn														
o	0.14111111	0.16111111	0.16111111	0.12111111	0.15444444	0.17888889	0.21	0.11888889	0.173333333	0.13	0.133333333	0.196666667	0.128888889	0.174444444
Mg														
O	15.8163453	15.17062	14.4866222	14.8457946	14.7339212	14.5739619	14.6210665	14.6171411	14.47386468	14.56316712	14.7849512	14.94196648	15.07150409	14.8732723
Ca														
O	22.250726	22.0455897	22.54782	22.1789788	22.2497155	22.7014197	22.3699678	22.6236093	22.17998934	22.77619845	22.65897768	22.43059927	22.37603098	22.38613622
Na														
2O	0.26830359	0.30678124	0.2714234	0.32965984	0.29118219	0.31094099	0.30262149	0.32861991	0.355658252	0.357738125	0.322380287	0.32134035	0.237105501	0.281822767
K2														
O	0.00105608	0	0.01478508	0	0.0095047	0.00422431	0	0.00739254	0	0.002112155	0.001056077	0.006336465	0.009504697	0.005280387

Cr2														
O3	0.30526509	0.37902041	0.34931341	0.38311793	0.26019239	0.29809443	0.35238655	0.26736305	0.30526509	0.246875459	0.375947276	0.2847775	0.18438831	0.017414451
Tot														
ale	100.174243	100.199605	99.3585585	99.5507837	99.9441559	100.08105	100.103868	100.696657	99.60290092	100.1566462	100.4894048	99.58748171	100.1071181	100.4885939
Fe2														
O3	1.64406347	1.07858063	2.64255786	2.06595762	1.58730625	2.34299163	1.71663227	2.20480754	1.420752408	2.7124097	1.665254885	2.249508464	1.671083972	1.895034034
Fe														
O	4.85679048	5.5354464	4.36695125	4.54200898	5.03548974	4.42859331	5.04880777	4.93581835	5.249170963	4.13426956	5.089891399	4.337735955	5.119641231	5.317482156
Wo	44.7802478	45.1871491	46.5285149	45.7976679	46.0440841	46.6679414	46.2115032	46.2105313	46.15296958	46.65462161	46.28383565	45.9944713	45.71637568	45.61872643
En	44.2893537	43.2660701	41.5941891	42.6536463	42.4246842	41.6863773	42.0256739	41.5424788	41.90576012	41.50688719	42.02036542	42.63079291	42.84461375	42.17173235
Fs	9.95325669	10.4088632	10.8637331	10.3168392	10.4407887	10.4889503	10.6315346	11.0323103	10.60202542	10.51241816	10.50415557	10.18234446	10.56237486	11.17027005
N°														
Mg	81.6504836	80.6075899	79.2905769	80.5234196	80.250269	79.8967237	79.8099161	79.0159686	79.80866016	79.79131385	80.00142528	80.72005381	80.22286009	79.05914749
Dat														
a	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx
Set	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r
SiO														
2	52.0834421	51.6128146	51.7801041	51.9887119	51.1230396	51.9151449	52.2970888	51.5432787	51.76095647	49.45619211	49.68999417	49.62348152	49.47634746	49.79278646
TiO														
2	1.01616014	1.03664724	0.97006417	0.98440514	1.04279337	1.02230627	0.98235643	1.02128192	1.005916594	0.991575624	1.536532476	1.60106684	1.539605541	1.563165705
Al2														
O3	2.62423369	2.7339217	2.6780429	2.62837286	2.75875671	2.68735603	2.69563437	2.67390373	2.57663324	2.325178671	4.710375305	4.78384557	4.572747907	4.732105947
Fe														
O	6.44318014	6.42979973	6.34745877	6.53169667	6.38759999	6.33716615	6.38451221	6.23423995	6.389658515	6.346429511	6.785924387	6.598598702	6.793129221	6.764309885
Mn														
o	0.09555556	0.19555556	0.16777778	0.14444444	0.16444444	0.15222222	0.12111111	0.14111111	0.15777778	0.12111111	0.12222222	0.08555556	0.17888889	0.17
Mg														
O	16.0665884	16.0420548	15.8026065	16.0332227	16.1755178	15.8055505	15.8565805	16.0391107	15.80751321	16.46599603	14.50919312	14.49349159	14.41792799	14.3708234
Ca														
O	21.7373799	21.8788532	22.3406626	21.9869793	22.1264316	21.9991056	21.6818011	22.3588521	21.94857937	21.9374636	22.40129408	21.9415057	22.47607285	22.35784155
Na														
2O	0.21942658	0.23398569	0.25582436	0.24230518	0.27246334	0.25062467	0.2828627	0.27038347	0.317180605	0.236065565	0.308861113	0.322380287	0.28390264	0.32134035
K2														
O	0.00316823	0.01372901	0.00422431	0.01478508	0.01267293	0	0.00211215	0	0.010560774	0	0.002112155	0.00422431	0.00422431	0.014785084
Cr2														
O3	0.29604567	0.23253415	0.2345829	0.32780144	0.30321633	0.26531429	0.21716845	0.34111837	0.353410927	0.32575268	0.555213688	0.569555001	0.521409164	0.513214128
Tot														
ale	100.289135	100.177362	100.346765	100.554923	100.06372	100.169476	100.304059	100.282162	99.97477655	97.88001222	100.066509	99.45415007	99.7428468	100.0871584
Fe2														
O3	0.72575892	1.7420931	1.72541887	1.45833024	3.0607997	0.91898219	0.32949096	2.30428111	1.446162424	5.29065878	1.91020182	1.064399378	2.093688792	1.58111864



Fe														
O	5.79013427	4.86224518	4.79490788	5.21947507	3.63345872	5.51025586	6.08803261	4.16082243	5.088385642	1.585836483	5.067103755	5.640840421	4.909204991	5.341601922
Wo	43.8973587	44.0733679	44.9096624	44.1219273	44.1540812	44.5408274	44.033018	44.6899817	44.33653709	43.67484302	46.24069076	45.86282283	46.4830823	46.3714081
En	45.144595	44.9637555	44.2001114	44.7672719	44.9126378	44.5260103	44.8067591	44.605854	44.42933422	45.61249659	41.67209793	42.15200875	41.48854816	41.47180716
Fs	10.1561666	10.1099144	9.95960181	10.2308864	9.94936884	10.0149019	10.1206677	9.7261852	10.07468205	9.862175956	10.93348386	10.76575317	10.96586032	10.95070933
N°														
Mg	81.6346714	81.6429259	81.6106821	81.3977655	81.8647376	81.6378174	81.5744733	82.0986193	81.51570702	82.22220069	79.21611454	79.65569067	79.09449245	79.11067598
Dat														
a	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx	B34_4_Cpx
Set	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	B34_4_Cpx c-r
SiO														
2	49.8895321	49.5861941	49.7595301	49.721235	50.0860468	49.5942563	49.638598	49.8502292	49.90364089	49.85728359	50.18178473	49.98829337	51.53622431	
TiO														
2	1.52526457	1.5723849	1.56828748	1.56521442	1.51399667	1.57955539	1.54984909	1.50682618	1.540629896	1.490436501	1.498631341	1.56828748	1.215909366	
Al2														
O3	4.68243591	4.66898361	4.68967946	4.55722602	4.71347968	4.66380964	4.50238202	4.47961659	4.422703	4.512729944	4.508590774	4.331641263	3.249248343	
Fe														
O	6.57904272	6.80856815	6.65932516	6.77769029	6.78077808	6.54404782	6.69637859	6.80856815	6.739607596	6.536842982	6.553311174	6.71078826	7.048386197	
Mn														
o	0.21	0.18555556	0.14111111	0.16222222	0.15222222	0.12444444	0.13555556	0.15111111	0.18	0.19333333	0.11	0.14111111	0.14	
Mg														
O	14.25895	14.5082118	14.3767115	14.6024209	14.4640512	14.4336295	14.4277414	14.6554136	14.44933104	14.38554359	14.66424571	14.32371882	14.7633616	
Ca														
O	22.5043675	22.4053362	22.4073572	22.484157	22.3174206	22.3164101	22.6862618	22.3265153	22.30125221	22.98638742	22.53670428	22.24870497	22.45990447	
Na														
2O	0.28494258	0.31198092	0.36189787	0.29846175	0.34421895	0.28910232	0.27766302	0.33797933	0.354618315	0.314060795	0.272463339	0.342139078	0.328619905	
K2														
O	0.00105608	0.00422431	0.01689724	0	0.01161685	0	0.00422431	0.01478508	0.014785084	0.006336465	0.016897239	0.021121549	0.008448619	
Cr2														
O3	0.50911661	0.45380012	0.36058158	0.42921501	0.45584888	0.40155676	0.35853282	0.44662946	0.281704362	0.442531943	0.381069173	0.404629902	0.072730944	
Tot														
ale	99.9355915	100.05144	99.9807972	100.168628	100.383831	99.5452555	99.9186539	100.131045	99.90656804	100.2829546	100.3426286	99.6758059	100.7501028	
Fe2														
O3	1.08567835	2.21571996	1.88960169	2.21709739	1.51834254	1.46966219	2.12643908	2.24642854	1.869007155	2.255205011	1.522040597	1.189712496	1.179411997	
Fe														
O	5.60213739	4.81483893	4.95904075	4.78272164	5.41455674	5.2216296	4.78298529	4.78720702	5.057854378	4.50758468	5.183762285	5.640271856	5.987138296	
Wo	46.8888823	46.2242293	46.4267724	46.2329562	46.1501679	46.4630878	46.7851872	45.8977586	46.16443297	47.22613826	46.42081953	46.31682989	45.74517962	
En	41.3372435	41.6469306	41.4465427	41.778305	41.6169392	41.8128901	41.3994604	41.9198298	41.61758694	41.123388	42.02744846	41.48974589	41.83825743	

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Fs	10.6995157	10.9640839	10.7697698	10.8781519	10.9447832	10.6347809	10.7791317	10.9250811	10.88958211	10.48281959	10.53614003	10.90450541	11.20535116
N°													
Mg	79.4385433	79.1600979	79.3747024	79.3412763	79.1772744	79.7230636	79.3418503	79.3261434	79.26077085	79.68690187	79.95544	79.18759188	78.87520956

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: sub-ophitic gabbro B34. Feldspars.

Data Set	B34_2_Pl (confuso per vetro)	B34_2_Pl (confuso per vetro)	B34_2_Pl (confuso per vetro)	B34_1_Pl-c r	B34_1_Pl-c r	B34_1_Pl-c r	B34_1_Pl-c r	B34_1_Pl-c r	B34_1_Pl-c r	B34_1_Pl-c r	B34_1_Pl-c r	B34_1_Pl-c r	B34_1_Pl-c r	B34_1_Pl-c r
SiO2	47.7487295	47.5475481	47.0592536	49.074248 2	49.405803 7	49.280840 6	49.318128 3	50.330934 24	49.650691 1	49.183087 07	49.842167 67	50.034650 67	49.212312 36	50.230157 58
TiO2	0.08102321	0.07582942	0.05609299	0.0901432 4	0.0676074 3	0.0891188 8	0.0737535 6	0.0839971 1	0.0676074 29	0.0758022 69	0.0717048 49	0.0839971 09	0.0850214 64	0.0880945 29
Al2O3	33.081439	33.2353444	33.3358959	32.057870 6	32.562849 3	32.755320 7	32.468683 2	31.574622 5	32.378656 26	32.965383 59	32.216193 84	31.986469 92	32.847417 25	32.131340 86
Cr2O3	0	0	0	0.7225419 0	0.7451856 0	0.7101907 0	0.6782836 8	0.7719465 5	0.7369515 0	0.7019566 85	0.7420979 0	0.7565075 39	0.6813714 46	0.6813714 0
FeO	0.73520851	0.75268899	0.83392182	0.7225419 3	0.7451856 9	0.7101907 8	0.6782836 6	0.7719465 94	0.7369515 86	0.7019566 86	0.7420979 04	0.7565075 72	0.6813714 46	0.6813714 46
MnO	0	0	0.05058448	0.7225419 0	0.7451856 0	0.7101907 3	0.6782836 8	0.7719465 1	0.7369515 0	0.7019566 11	0.7420979 67	0.7565075 78	0.6813714 0.06	0.6813714 44
MgO	0.05853626	0.03633285	0.09587836	0.0922464 8	0.0745822 6	0.0991158 9	0.0912651 3	0.1010785 9	0.0863584 03	0.1079480 04	0.0873397 49	0.0932278 22	0.0893024 4	0.0883210 94
CaO	17.3382773	17.355416	17.4441338	15.962235 7	16.140087 9	16.093603 8	14.972932 8	15.333689 84	16.170403 64	15.726783 64	15.563078 77	15.584299 77	15.542868 29	15.542868 29
Na2O	1.88317069	2.05106268	2.01577328	2.8130279 9	2.6445582 9	2.6653570 2	2.5956812 8	3.1957245 8	2.9128618 81	2.5811221 66	2.7901493 85	3.0033363 48	2.7131940 9	2.8972628 35
K2O	0.06241383	0.07177591	0.05929314	0.1203928 3	0.1193367 5	0.1182806 7	0.1277853 7	0.1467947 6	0.0982152 01	0.0908226 59	0.1203928 27	0.1351779 11	0.1130002 85	0.1415143 75
SrO														
BaO														
Total	100.988798	101.125998	100.950827	100.73262 3	101.58215 9	101.86164 6	101.52593 7	101.27167 9	101.26503 19	101.92836 86	101.57026 69	101.71283 82	101.48871 35	101.82537 55
Ab %	16.3677443	17.5470977	17.2370484	24.245192 1	22.908871 2	22.854662 3	22.428586 2	27.629471 9	25.437866 98	22.295926 74	24.136255 81	25.686128 85	23.801124 61	25.020898 06
An %	83.2753183	82.0488691	82.4293426	75.072053 9	76.410928 5	76.478000 6	76.844900 2	71.535452 3	73.997779 01	77.187866 98	75.178484 54	73.553172 72	75.546634 15	74.174969 36
Or %	0.35693743	0.40403329	0.33360901	0.6827540 6	0.6802003 5	0.6673371 5	0.7265136 3	0.8350758 1	0.5643540 06	0.5162062 79	0.6852596 48	0.7606984 26	0.6522412 43	0.8041325 83
Data Set	B34_1_Pl-c-r	B34_1_Pl-c-r	B34_3_sievePl c-r	B34_3_siev ePl c-r	B34_3_siev ePl c-r	B34_3_siev ePl c-r	B34_3_siev ePl c-r	B34_3_siev ePl c-r	B34_3_siev ePl c-r	B34_3_siev ePl c-r	B34_3_siev ePl c-r	B34_3_siev ePl c-r	B34_3_siev ePl c-r	B34_3_siev ePl c-r
SiO2	50.0336429	50.5062859	47.8185698	47.455773 5	47.731901 8	47.804461 1	48.030201 1	48.90192 2	48.801143 14	48.380904 14	47.646241 6	47.927408 74	47.959657 31	48.105783 6
TiO2	0.09936243	0.08092404	0.08297275	0.0594125 9	0.0594125 9	0.0717048 5	0.0645343 6	0.1116546 9	0.0839971 09	0.0747779 14	0.0481446 84	0.0921919 49	0.0635100 09	0.0450716 19

Al2O3	31.7546764	31.8126248	33.3803354	33.583154	33.597641	33.348256	33.618337	32.488344	32.492483	33.302725	33.610059	33.247881	33.369987	33.211664
Cr2O3	0	0.04097518	0.02765825	0.7369515	0.6268205	0.5918256	0.0020487	0.0460970	0.0122925	0.0122925	0.0338045	0.0307313	0	0
FeO	0.69269333	0.68445923	0.59079639	0.0488888	0.0333333	0.0322222	0.0433333	0.6402009	0.6607862	0.6741666	0.6082938	0.6618154	0.6926933	0.6649032
Mno	0	0.00777778	0.02666667	0.0500486	0.0520113	0.0902837	0.0696755	0.0706568	0.0353284	0.0353284	0.0490672	0.0441605	0.0657501	0.0392538
MgO	0.107948	0.08341437	0.07654495	17.215285	16.972759	16.945475	16.898991	16.133014	16.099666	16.664549	17.061685	16.983875	16.683749	16.978822
CaO	15.25689	15.2013112	17.0091385	2.1339495	2.0715533	2.2088249	2.1672274	2.6653570	2.5676029	2.4448904	2.1360293	2.2857802	2.1682674	2.2712211
Na2O	3.01685552	3.15932681	2.1537083	0.0781497	0.0929348	0.0707571	0.0834301	0.0918787	0.0887105	0.0792058	0.0813179	0.0770936	0.0887105	
K2O	0.12989752	0.14890692	0.08554227	0.0781497	0.0929348	0.0707571	0.0834301	0.0918787	0.0887105	0.0792058	0.0813179	0.0770936	0.0887105	
SrO														
BaO														
Total	101.091966	101.726006	101.251933	101.31272	101.25392	101.16492	101.62230	101.18928	100.84517	101.70834	101.30475	101.38738	101.08070	101.44098
Ab %	26.1578069	27.1011091	18.5516569	18.241154	17.994954	19.009706	18.746811	22.279053	20.874916	18.387986	19.495621	18.955970	19.392098	
An %	73.1011221	72.0584262	80.963514	81.319296	81.473861	80.589614	80.778337	76.597948	77.196385	78.626713	81.163375	80.048025	80.600559	80.109530
Or %	0.74107096	0.84046472	0.48482913	0.4395494	0.5311842	0.4006791	0.4748515	0.5015076	0.5245609	0.4983706	0.4486380	0.4563533	0.4434696	0.4983707
Data Set	B34_3_sievePl c-r	B34_3_sievePl c-r	B34_3_sievePl c-r	B34_3_sievePl c-r	B34_3_sievePl c-r	B34_3_sievePl c-r	B34_3_sievePl c-r	B34_3_sievePl c-r	B34_3_sievePl c-r	B34_3_sievePl c-r	B34_2_PI r-r	B34_2_PI r-r	B34_2_PI r-r	B34_2_PI r-r
SiO2	48.4333081	48.4282692	48.8948656	49.00572	49.369524	49.525728	49.232467	48.542146	48.210591	48.309352	47.602907	48.686257	48.945253	48.693312
TiO2	0.12702002	0.05019339	0.10550856	0.0512177	0.0942406	0.1085816	0.0839971	0.0409742	0.0778509	0.0686317	0.0409741	0.0655587	0.0635100	0.0921919
Al2O3	32.9291659	33.0460974	32.8815654	32.904330	32.367273	32.927096	32.942618	32.932270	33.834609	33.642137	32.90847	31.860225	31.693623	31.609805
Cr2O3	0.03585328	0	0.01536569	0.0327801	0.0071706	0.0133169	0.0133169	0.0542921	0.0071706	0.0071706	0.0542921	0.0081950	0.0122925	
FeO	0.73180528	0.67931292	0.63711318	0.6093231	0.7091615	0.7266589	0.5753574	0.6844592	0.6093231	0.7462149	0.7390101	0.7184248	0.7019566	0.6710788
Mno	0	0.03888889	0	0.0388888	0.0133333	0.0055555	0.0055555	0.0055555	0.0433333	0.0466666	0.0433333	0.0033333	0.0144444	

MgO	0.16094066	0.04121651	0.04612324	0.0745822	0.0893024	0.0912651	0.0736009	0.0667314	0.0618247	0.0392538	0.0559366	0.0736009	0.0794889	0.0726195
				6	4	3	1	9	66	2	93	12	85	67
				16.213856	15.739920	15.612594	15.930909	16.130993	16.773686	16.913138	16.945475	16.210824	15.708594	15.894530
CaO	16.5402554	16.4826555	15.8369307	2	5	4	5	2	41	71	48	6	21	61
Na2				2.6466381	2.7475119	2.8036685	2.5520039	2.5041668	2.2109048	1.9821187	2.2816204	2.7922292	2.7391925	2.7173538
O	2.23378341	2.43657101	2.56448318	6	9	6	5	7	05	95	81	57	01	36
				0.1119442	0.1013834	0.0918787	0.0971591	0.1087759	0.0897665	0.0739254	0.0918787	0.1394022	0.1108881	0.1214489
K2O	0.12567321	0.09187874	0.10560774	1	3	4	2	7	81	2	36	2	3	04
SrO														
BaO														
Total				101.65650	101.26443	101.89464	101.50698	101.01607	101.92284	101.82527		100.64414	100.05403	99.899078
e	101.317805	101.295084	101.087563	1	2	2	6	4	94	8	100.71294	9	65	23
				22.659420		24.397443	22.348047	21.794822	19.160089	17.422206	19.490781		23.834131	23.464729
Ab %	19.4977425	20.9952435	22.524184	3	23.866283	2	3	5	05	35	87	23.578912	85	15
				76.709959	75.554255	75.076485	77.092125	77.582253	80.328047	82.150251	79.992786	75.646529	75.531014	75.845229
An %	79.7804887	78.4838389	76.8654968	3	9	3	6	4	33	07	34	04	03	59
				0.6306203		0.5260714	0.5598271	0.6229241	0.5118636	0.4275425	0.5164317	0.7745589	0.6348541	0.6900412
Or %	0.72176878	0.52091752	0.61031926	1	0.5794611	8	2	4	26	83	93	62	26	55
Data Set	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r
SiO2	48.6207528	49.1125434	49.0853336	48.876725	48.379896	48.928121	48.585480	49.287894	49.032929	48.983549	48.743700	48.709436	49.060139	48.546178
				8	4	9	9	9	74	13	45	36	46	01
				0.0645343	0.0798996	0.0921919	0.0737535	0.0747779	0.0850214	0.0880945	0.1044842	0.0778509	0.0747779	0.0727292
TiO2	0.08297275	0.07887533	0.07375356	6	9	5	6	1	64	29	08	79	14	04
Al2O3	31.7857202	31.8985126	31.9305911	32.034070	31.936799		31.705006	31.833320	32.081670	31.905756	31.728806	31.723632	31.611875	32.046487
				4	9	31.806416	4	6	82	1	59	63	04	88
Cr2O3						0.0081950			0.0153656	0.0440483			0.0122925	
	0	0	0	0	0	4	0	0	92	18	0	0	54	0
				0.7544490	0.6515228	0.6926933	0.7246004	0.7575368	0.6175572	0.6206449	0.7081322	0.6751958	0.6782836	0.7287174
FeO	0.67004956	0.80076584	0.6988689	5	5	3	5	3	02	88	58	74	6	98
							0.0277777		0.0155555	0.0322222	0.0111111	0.0133333		0.0277777
Mno	0.01111111	0	0.01888889	0	0	0	8	0	56	22	11	33	0.03	78
				0.0794889	0.0608434	0.0932278		0.1050039	0.0765449	0.0863584	0.1177614	0.0961718	0.0755636	0.0951905
MgO	0.1099107	0.10304128	0.0863584	8	2	2	0.0755636	7	48	03	59	58	03	13
				15.699499	15.971330	15.837941	16.020846	15.756088	15.613604	15.914741	15.704552	16.011751	15.666152	16.130993
CaO	15.8419834	15.6287628	15.5954155	5	4	3	1	8	96	09	12	38	21	21
Na2				2.8171877	2.7027947	2.8920631	2.7298330	2.8816637	2.8182276	2.8972628	2.8868634	2.9066222	2.8213474	2.6996749
O	2.79950881	2.73607269	2.80054875	3	3	5	7	9	68	35	71	62	77	17
				0.1182806	0.1140563		0.1256732	0.1267292	0.1066638	0.0971591	0.1489069	0.1256732	0.1045516	0.1182806
K2O	0.12356106	0.11405636	0.10032736	7	6	0.1214489	1	9	2	23	17	14	65	72

SrO														
BaO														
Total	100.04557	100.47263	100.390086	100.444236	99.8971437	100.472299	100.068535	100.823016	100.4631419	100.6698367	100.1543186	100.3396679	100.1349836	100.4660297
Ab %	24.0608998	23.9008823	24.385263	24.3478861	23.2927151	24.6678309	23.400609	24.6889634	24.47122752	24.6456336	24.75193452	24.55453928	24.43304166	23.09093041
An %	75.2403471	75.4435494	75.0399383	74.9794915	76.0605316	74.6505702	75.8905548	74.5966248	74.91936341	74.81055566	74.40800631	74.74690988	74.97120838	76.2434044
Or %	0.69875311	0.65556826	0.5747987	0.67262245	0.64675332	0.68159892	0.70883621	0.71441186	0.609409076	0.543810743	0.840059173	0.69855084	0.595749953	0.665665191
Data Set	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r	B34_2_Pl r-r	B34_4_Pl r-r	B34_4_Pl r-r
SiO2	46.9962315	47.2380957	46.6636682	47.0718141	47.0385578	47.3580201	47.7167853	47.6371717	46.98111502	47.3660822	47.89213688	48.47261098	46.43087395	48.33958567
TiO2	0.04916904	0.05121775	0.05429081	0.0522421	0.05429081	0.05429081	0.07068049	0.06658307	0.06351009	0.090143239	0.094240659	0.069656139	0.097313723	0.093216304
Al2O3	32.1478975	32.7294509	33.1361243	33.0191928	32.9136443	32.6528763	32.2399941	32.6549459	33.17855033	33.05437574	32.58768434	32.65287627	32.83086057	32.59389309
Cr2O3	0.03380452	0.06146277	0	0.03073138	0	0.02356073	0.01024379	0	0.06146277	0.023560728	0	0	0	0.025609487
FeO	0.73592233	0.67931292	0.72974676	0.75136126	0.68651776	0.73901012	0.69063487	0.77812207	0.723571188	0.65255211	0.691664066	0.867667869	0.741068642	0.718424878
Mno	0.00222222	0	0.01111111	0.01111111	0.07111111	0	0.04666666	0.01888888	0	0.02666666	0.00777777	0.00333333	0.05666666	0.01888888
MgO	0.09126513	0.10598531	0.06673149	0.05299266	0.08341437	0.06967553	0.06869418	0.08439571	0.071638221	0.068694185	0.059862075	0.073600912	0.073600912	0.068694185
CaO	16.4533503	16.7312444	17.2102327	17.0495594	17.262788	16.5513712	16.1117933	16.58572954	17.05360117	17.18901174	16.33309799	16.57360268	17.3183588	16.58572897
Na2O	2.39289368	2.06947345	2.03619549	2.05803415	2.15890798	2.17138722	2.42305183	2.40017323	1.993558096	2.149548557	2.349216347	2.516646109	2.176586903	2.399133295
K2O	0.09927128	0.08343012	0.07814973	0.08765443	0.08554227	0.07286934	0.08448619	0.08131796	0.07814973	0.07392542	0.09927128	0.098215278	0.053859901	0.082374039
SrO														
BaO														
Total	99.0020276	99.7496734	99.9862507	100.184693	100.354766	99.6930613	99.4630306	100.307327	100.2051574	100.6945606	100.1149514	101.3282095	99.77919011	100.9255488
Ab %	20.7170836	18.2010754	17.556439	17.8380876	18.3664752	19.1048585	21.2884059	20.6573852	17.38243748	18.37716536	20.53472814	21.43678347	18.47358445	20.64904811

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An %	78.7174056	81.3161192	82.0002009	81.662014 1	81.154691 3	80.473286 2	78.223191 1	78.882113	82.169208 19	81.206984 2	78.894317 15	78.012753 34	81.225632 61	78.884455 68
Or %	0.56551084	0.48280543	0.44336004	0.4998983 1	0.4788335 4	0.4218552 9	0.4884029 6	0.4605018 2	0.4483543 25	0.4158504 42	0.5709547 11	0.5504631 92	0.3007829 39	0.4664962 15

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: sub-ophitic gabbro B34. Glass.

Data set	B34_1_Gla ss	B34_1_Gla ss	B34_3_Gla ss	B34_3_Gla ss	B34_3_Gla ss	B34_3_Gla ss	B34_3_Gla ss	B34_3_Gla ss	B34_1_Gla ss	B34_1_Gla ss	B34_1_Gla ss	B34_1_Gla ss	B34_1_Gla ss	B34_1_Dark zone in Glass
SiO2	53.383829 7	53.559737	54.642476 9	53.780126 7	54.145083 8	54.821417 2	54.881063 9	53.594109 7	54.207763 44	53.522331 44	53.429322 97	53.312051 42	57.74006343	
TiO2	2.6571458 9	2.5698901 2	2.6664947 2	2.6893474 2	2.6426032 6	2.7423241 4	2.6072854 5	2.6561071 3	2.6301381 48	2.7173939 16	2.7641380 77	2.6374094 62	2.323704201	
Al2O3	15.968191 7	16.011285 2	15.850197 6	15.591636 6	15.992816 5	15.733229 5	15.865588 1	15.967165 6	15.808130 12	15.700396 39	15.774270 95	16.098498 19	16.32319998	
Cr2O3	0 9.5875302	0 9.9844400	0.092 9.0374092	0.004 9.4240363	0 9.1978183	0 9.1895922	0.037 9.0374092	0.015 10.239449	0 9.8271156	0.114 10.036881	0.059 10.318625	0 10.138679	0.026	
FeOtot	6 0.2360609	3	1 0.1805171	4	4 0.2231668	3 0.2092809	1 0.1785334	4 0.1854764	93 0.1973786	48 0.2023379	72 0.2271342	58 0.2380446	6.875999021	
MnO	3 2.9580996	0.188452 2.9480072	8 2.8975449	0.1299327 2.8450641	4 2.8904802	1 3.0125989	8 2.8571750	4 3.1649950	75 3.0751722	38 3.1599488	54 3.1892169	33 3.1559118	0.239036486	
MgO	8 6.7042414	2 6.1668939	3 6.0328091	4 6.7052495	1 6.1426982	6 6.1043882	9 6.0025644	9 6.7183556	05 6.7929592	59 7.0329004	89 6.7203719	75 6.9270440	2.279886449	
CaO	4 4.0005346	6 3.8903889	4 5.4003475	9 5.1650848	1 5.3907231	5 5.1330035	4 5.0228578	3 4.9276834	17 4.7277101	89 4.9362384	41 5.1490442	46 5.0367597	5.132525285	
Na2O	6 2.2593806	6 2.0939840	1 2.3301163	5 2.3394784	3 2.3550818	8 2.4237371	8 2.4591049	3 2.1407944	7 2.1480760	4 2.1002254	13 2.0783805	6 2.1553576	5.760727139	
K2O	9 1.0189678	3 0.8029971	6 0.8512459	4 0.7731289	9 0.8558410	1 0.9086849	5 0.9109825	1 0.8707752	22 0.8305679	18 0.8788166	77 0.9672727	35 0.8523947	2.936570755	
P2O5	3 F	9	5	1	7	5	1	1	12	7	27	31	0.901792269	
Cl														
Total	98.773982 7	98.216075 7	99.981159 5	99.447085 7	99.836313 3	100.27825 7	99.859565	100.47991 2	100.24501 16	100.40147 1	100.67677 84	100.55215 13	100.539505	



Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: Poikilitic gabbro B21. Clinopyroxenes.

Da a Set	B21_3_s mall Cpx r-r	B21_3_s mall Cpx r-r	B21_3_s mall Cpx r-r	B21_3_s mall Cpx r-r	B21_3_s mall Cpx r-r	B21_3_s mall Cpx r-r	B21_4_Cpx rim detail to glass	B21_4_Cpx rim detail to glass	B21_4_Cpx rim detail to glass	B21_4_Cpx rim detail to glass	B21_4_Cpx rim detail to glass	B21_5_Cpx rim detail to glass	B21_5_Cpx rim detail to glass	B21_5_Cpx rim detail to glass
SiO 2	51.03334 83	51.08071 34	51.12303 96	51.66622 63	51.28932 13	51.43947 86	51.2268397	50.8952842	50.72597921	50.80156177	50.82070936	50.58489175	50.24426632	50.96784342
TiO 2	1.057134 34	1.059183 05	1.104254 67	1.116546 93	1.149326 29	1.162642 91	1.69633185	1.74959831	1.829498001	1.863301716	1.754720087	1.946274469	1.958566729	1.761890572
Al2 O3	3.454137 25	3.513120 42	3.553477 33	3.414815 14	3.775957 71	3.733531 22	3.80803627	3.76871416	3.700417858	3.822523369	3.79768835	4.038794995	3.95290722	4.309910621
FeO	5.832827 77	5.683584 78	5.503463 93	5.580658 58	5.832827 77	5.892524 97	6.40921449	6.66241295	6.805480365	6.732402762	6.987659739	6.665500732	6.66447147	6.924874757
Mn o	0.077777 78	0.134444 44	0.18 0.18	0.185555 56	0.104444 44	0.085555 56	0.14777778	0.12555556	0.164444444	0.203333333	0.208888889	0.112222222	0.187777778	0.121111111
Mg O	15.20202 3	15.26581 05	15.27366 13	15.32076 58	15.14510 5	15.02538 08	14.6259732	14.7349026	14.64265611	14.60045825	14.53568945	14.64363745	14.633824	14.6985928
CaO	22.61855 67	22.69434 6	22.62967 25	22.44070 45	22.55792 53	22.58318 84	22.2608313	22.1042	22.11834738	22.059737	22.16483148	22.15068415	22.37603098	21.32811769
Na2 O	0.280782 83	0.292222 13	0.268303 59	0.262063 98	0.322380 29	0.320300 41	0.37229723	0.36085793	0.399335581	0.446132719	0.366057616	0.379576789	0.365017679	0.544926678
K2O	0.008448 62	0.010560 77	0.020065 47	0.006336 46	0.008448 62	0.008448 62	0.02428978	0.03590663	0.017953316	0.031682323	0.04118702	0.012672929	0.017953316	0.093990891
Cr2 O3	0.611554 56	0.718090 03	0.480433 98	0.500921 57	0.607457 04	0.500921 57	0.41794684	0.4896534	0.380044794	0.266338669	0.235607284	0.311411367	0.382093553	0.182339551
Tot ale	99.56503 67	99.73398 55	99.65593 84	99.99367 33	100.1857 37	100.2510 52	100.571592	100.437432	100.4041123	100.5611332	100.677432	100.5342555	100.4008155	100.7512585
Wo	46.31960 39	46.40946 7	46.50318 88	46.18554 46.18554	46.25658 67	46.40210 25	46.1003578	45.6136402	45.57972507	45.54311105	45.69469725	45.75130996	46.04126131	44.27586179
En	43.31638 77	43.43698 5	43.67154 12	43.87336 16	43.21132 82	42.95649 01	42.1442753	42.3075715	41.98457753	41.94109342	41.69537694	42.08392665	41.89603113	42.45620302
Fs	9.323467 2	9.072136 95	8.827524 61	8.965062 85	9.335808 36	9.450443 27	10.3601507	10.7312379	10.94652211	10.84902724	11.24427519	10.74601682	10.70355926	11.22082436
N° Mg	82.28819 74	82.72274 11	83.18536 82	83.03306 17	82.23345 94	81.96718 89	80.2680432	79.7671968	79.31929965	79.4487546	78.76020196	79.6592309	79.65086956	79.09566735
Da a Set	B21_7_Cp x c-r	B21_7_Cp x c-r	B21_7_Cp x c-r	B21_7_Cp x c-r	B21_7_Cp x c-r	B21_7_Cp x c-r	B21_7_Cpx c-r	B21_7_Cpx c-r	B21_7_Cpx c-r	B21_7_Cpx c-r	B21_7_Cpx c-r	B21_7_Cpx c-r	B21_6_cpx c-r	B21_6_cpx c-r
SiO 2	51.94134 69	51.59568 26	51.73374 67	50.61411 7	51.11699 3	50.34705 86	51.0474571	51.5976981	52.78384053	50.7945074	51.37901257	51.01319296	51.55637966	51.9403391
TiO 2	0.960844 97	1.053036 92	1.201568 4	1.212836 3	1.254834 86	1.213860 66	1.13396097	1.16366726	1.142155807	1.109376448	1.155472422	1.230250336	1.016160144	1.058158698





Totale	100.0710 44	99.98343 93	99.77219 56	100.3365 32	99.71041 07	99.76348 28	99.6626018	101.083727	99.90266077	99.65509661	98.5408484	100.1652246	99.27574903	99.64774926
Wo	46.53700 39	46.14707 22	45.91447 05	46.99034 87	46.44677 55	45.67739 35	45.581942	44.8944517	45.77729946	45.13229364	45.61851666	46.03943113	46.35486358	46.47337943
En	43.31191 1	43.36739 54	43.82180 36	42.82279 03	42.91568 32	43.22441 33	43.0527569	43.8694085	43.04850488	43.91255053	43.26575413	43.03206214	43.17180343	42.71978789
Fs	9.217046 33	9.597719 04	9.240637 98	9.053859 06	9.539059 66	9.939133 85	10.2133596	10.0962024	9.952322676	9.801172644	9.77788241	9.758178427	9.376327064	9.610380557
N°	82.45339	81.87916	82.58535	82.54733	81.81468	81.30460								
Mg	94	87	1	26	61	74	80.8257852	81.2914146	81.22232589	81.75294493	81.56634227	81.51518477	82.15668764	81.63510487
Data Set	B21_4_cp x c-r	B21_4_cp x c-r	B21_4_cp x c-r	B21_4_cp x c-r	B21_4_cp x c-r	B21_4_cp x c-r	B21_4_cpx c-r	B21_4_cpx c-r	B21_Cpx to Pl	B21_Cpx to Pl	B21_Cpx to Pl	B21_Cpx to Pl	B21_Cpx to Pl	B21_Cpx to Pl
SiO2	50.77939 09	51.10893 09	51.07769 01	50.34201 98	51.68537 39	50.55566 65	51.1845134	50.4367499	51.35422468	50.75172164	48.99204637	49.5894607	50.45047012	49.66782681
TiO2	1.107327 74	1.068402 25	1.082743 22	1.11962 96	1.106303 38	1.070450 96	1.1718621	1.71681895	1.035758151	1.116386031	1.446133386	1.377909795	1.344831691	1.355168598
Al2O3	3.770783 75	3.379632 19	3.611425 71	3.599008 2	3.585555 89	3.520363 97	3.57934714	3.7500879	3.654795007	3.896442917	4.071715436	3.99704312	4.034379278	3.984597734
FeO	5.582717 1	5.595068 25	5.829739 99	5.891495 71	5.805037 7	5.855471 54	6.26203003	6.67785188	5.294769556	5.6664032	6.154172358	6.075402183	6.202646312	6.149122988
MnO	0.13 0.13	0.105555 56	0.094444 44	0.107777 78	0.132222 22	0.078888 89	0.17555556	0.20777778	0.110166254	0.103280863	0.15836399	0.100329981	0.129838799	0.101313609
MgO	15.28249 34	15.46404 23	15.35903 83	15.34824 35	15.09309 37	15.01458 6	15.3129151	14.5337268	15.76640437	15.51093254	14.982026	15.20855767	15.05287952	15.661621
CaO	22.48112 55	22.42049 4	22.55388 32	22.72769 33	22.10116 85	22.72769 33	22.376031	22.3588521	22.19276343	22.26363124	22.15530472	21.86170948	21.97914758	21.78274248
Na2O	0.297421 81	0.313020 86	0.283902 64	0.303661 43	0.270383 47	0.263103 91	0.29846175	0.38685634	0.165320473	0.284462043	0	0.156084693	0.220735157	0.291850668
K2O	0.016897 24	0.010560 77	0.016897 24	0.016897 24	0.015841 0	0.015841 16	0.01584116	0.02534586	0.001049007	0	0.005245036	0.015735109	0	0.012588087
Cr2O3	0.847161 84	0.782625 94	0.711943 75	0.789796 59	0.565457 48	0.623847 11	0.32267954	0.34624027	0.967965368	0.798268398	0.347186147	0.308225108	0.361038961	0.384415584
Totale	99.44815 74	99.46570 71	99.90976 48	99.45641 69	99.77913 87	99.10206 63	100.376557	100.094067	99.57525093	99.59326047	97.9650073	98.38223273	99.41492845	99.00683197
Wo	46.21854 4	45.87710 87	46.03910 89	46.16514 76	45.92733 45	46.70184 17	45.5628327	46.1069412	45.69979748	45.63795832	46.3453019	45.50027093	45.63343643	44.543592
En	43.71629 27	44.02757 73	43.62349 98	43.37789 24	43.64004 32	42.92822 27	43.3846836	41.7008299	45.17378193	44.24039688	43.6063195	44.04217833	43.48533432	44.56154981
Fs	8.958641 93	8.936235 32	9.288661 03	9.340772 18	9.415847 09	9.391584 45	9.95270552	10.7486009	8.510367246	9.066424052	10.04837859	9.869683898	10.05188922	9.814862154



SiO	50.83517	49.42865	49.78486	49.39100
2	65	75	71	11
TiO	1.178407	1.112251	1.176340	1.120520
2	48	27	1	79
Al2	3.958669	3.782360	3.861180	3.575974
O3	85	21	99	23
	5.717906	5.724975	5.677511	5.925940
FeO	78	89	81	83
Mn	0.088526	0.118035	0.139675	0.168200
o	45	27	07	26
Mg	14.81836	17.69441	15.72948	15.72349
O	44	84	07	31
	21.28970	20.79666	21.31602	21.99332
CaO	49	74	73	11
Na2	0.271531	0.356501	0.324175	0.229047
O	95	13	9	36
	0.014686	0.025176	0.030421	0.001049
K2O	1	17	21	01
Cr2	0.621645	0.460606	0.435497	0.405194
O3	02	06	84	81
Tot	98.17297	99.03904	98.03968	98.12854
ale	44	33	02	78
	45.43026	41.15677	44.20562	44.96610
Wo	46	84	83	13
	43.99737	48.72308	45.38753	44.72953
En	75	92	36	17
	9.523816	8.843407	9.190257	9.456927
Fs	98	45	78	86
N°	82.20552	84.63792	83.16117	82.54743
Mg	24	66	68	36

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: Poikilitic gabbro B21. Olivines.

Da a set	B21_1_OI rim detail to glass	B21_1_OI rim detail to glass	B21_1_OI rim detail to glass	B21_2_OI rim detail to glass	B21_2_OI rim detail to glass	B21_2_OI rim detail to glass	B21_2_OI rim detail to glass	B21_2_OI rim detail to glass	B21_2_OI rim detail to glass	B21_2_OI rim detail to glass	B21_2_OI rim detail to glass	B21_3_OI rim detail to glass	B21_3_OI rim detail to glass	B21_3_OI rim detail to glass	B21_3_OI rim detail to glass
SiO 2	38.9139359	38.9109125	38.778895	39.7090644	39.4188274	39.3805322	39.4168119	39.4359594	39.58410127	39.85418297	39.01068154	39.21122728	38.97339414	39.281771	
TiO 2	0.06555872	0.07887533	0.07375356	0.05736388	0.02663323	0.06351001	0.05531517	0.06043694	0.044047264	0.136239213	0.033803714	0.037901134	0.057363879	0.059412589	
Al2 O3	0.07864423	0.08485298	0.16349721	0.03828732	0.01966106	0.01966106	0.01862626	0.05173962	0.039322114	0.710902424	0.046565661	0.039322114	0.066226718	0.037252529	
Cr2 O3	0	0	0	0.01946321	0.0399508	0.05429211	0	0.01639007	0	0.009219415	0	0.01946321	0	0.012292554	
Fe O	20.9578329	21.0875199	20.7890339	21.441586	21.2192655	21.5012832	20.9784181	21.2367629	20.85902376	20.62846907	21.09987106	21.30057715	21.0556128	20.73345379	
Mn o	0.32666667	0.32888889	0.30555556	0.35222222	0.34111111	0.40111111	0.33444444	0.35333333	0.31555556	0.30111111	0.33444444	0.33222222	0.29111111	0.26	
Mg O	39.0487185	38.7180051	38.5129039	39.0418491	39.0565693	39.2341928	38.8632442	39.1017112	38.63557207	37.90250699	38.47463141	38.66599378	38.7700164	38.280325	
Ca O	0.25970465	0.28698879	0.27082041	0.15966278	0.20412583	0.21018897	0.21827316	0.21423107	0.264757265	0.375914895	0.171789065	0.268799361	0.240504691	0.231409976	
Na 2O	0	0	0.01039936	0.02703835	0.00207987	0.02703835	0.04783707	0.02079873	0.04055752	0.219426582	0.007279555	0.062396185	0	0.015599046	
NiO															
Tot ale	99.6510615	99.4960436	98.9048589	100.846537	100.328224	100.89181	99.9329703	100.491363	99.78293681	100.1379727	99.17906645	99.93790244	99.45422974	98.91151649	
Fo	76.5792575	76.3148711	76.4922816	76.1490945	76.3510131	76.1475849	76.4697239	76.3468722	76.48130964	76.34592592	76.18548835	76.10824056	76.39834185	76.47025663	
Fa	23.0567551	23.3168112	23.1629093	23.4605787	23.2701138	23.4100985	23.1563794	23.2611538	23.16377811	23.30946881	23.4382422	23.52021704	23.27572846	23.23464455	
Da a set	B21_3_OI rim detail to glass	B21_3_OI rim detail to glass	B21_7_OI c-r	B21_7_OI c-r	B21_7_OI c-r	B21_7_OI c-r	B21_7_OI c-r	B21_7_OI c-r	B21_7_OI c-r	B21_7_OI c-r	B21_7_OI c-r	B21_7_OI c-r	B21_4_ol c-r	B21_4_ol c-r	B21_4_ol c-r
SiO 2	38.935099	39.0913029	39.5236352	39.4964255	38.9260291	39.221305	38.8504465	39.524643	39.82697324	41.26505753	39.06711652	39.60828769	39.25960012	39.28277877	
TiO 2	0.05633952	0.09526501	0.02560887	0.02356016	0.0204871	0.02253581	0.00614613	0.01946274	0.01638968	0.02253581	0.03073065	0.041998554	0.039949844	0.01229226	
Al2 O3	0.06208755	0.39115155	0.02586981	0.05691359	0.00827834	0.03104377	0.04346128	0.03311336	0.026904604	0.020695849	0.065191925	0.013452302	0.035182944	0.032078566	
Cr2 O3	0.05531649	0.01946321	0	0.02356073	0.04814584	0	0.01434131	0	0	0.014341313	0.002048759	0.007170656	0	0.016390072	
Fe O	20.8765212	20.9588622	21.3870352	21.2079436	20.8929894	21.1338367	21.1039881	21.1554512	20.8281459	20.68301995	20.73242453	21.71331122	20.97327184	21.71434048	







SiO														
2	38.9240135	38.5289687	39.3291361	38.6146289	37.9666344	38.908897	39.182002	38.5459463	39.02326715	38.86958816	39.01105425	38.71387369	39.16066227	39.21358483
TiO														
2	0.031755	0.02970629	0.03790113	0.02151145	0.02560887	0.00512177	0.04302291	0.0227412	0.037212868	0.022741197	0.034111795	0.020673815	0.018606434	0.031010723
Al2														
O3	0.03518294	0.00724355	0.03621774	0.01241751	0.07760943	0.03207857	0.05587879	0.06118981	0.031113465	0.05289289	0.046670197	0.120305397	0.044595966	0.021779425
Cr2														
O3	0.03892642	0.01639007	0.03892642	0.02356073	0	0.02868263	0	0.05108225	0.004329004	0	0	0	0.004329004	0
Fe														
O	21.3407184	21.7472769	21.2892553	21.2491141	20.9732718	21.4261471	21.0556128	21.2881446	21.1528215	21.07809082	21.14070301	20.76502987	21.13767339	20.96296518
Mn														
o	0.33222222	0.32666667	0.32222222	0.38888889	0.33888889	0.37555556	0.33	0.23016878	0.328531508	0.275415635	0.323613371	0.297055435	0.27738289	0.307875335
Mg														
O	38.6767886	38.6571617	38.3470565	38.6630497	40.4981658	38.5688406	39.1821815	39.5242871	39.8715691	38.38164938	39.46940055	39.39156148	40.12504507	39.52428708
Ca														
O	0.2576836	0.25869412	0.25060993	0.26172569	0.26273622	0.24454679	0.22736788	0.18526872	0.224752223	0.234876197	0.185268724	0.196405096	0.239938184	0.223739826
Na														
2O	0.02183866	0.02911822	0.03119809	0	0.01039936	0	0.01559905	0	0	0.013853671	0	0.055414684	0	0
NiO														
	0.021	0.003	0.007	0	0.002	0.038	0.001	0.01363709	0.004196029	0	0	0.02307816	0.004196029	0.001049007
Tot														
ale	99.6801293	99.6042261	99.6895234	99.234897	100.155315	99.62787	100.092665	99.9224659	100.6777928	98.92910795	100.2108219	99.58339763	101.0124292	100.2862914
Fo														
	80.6406063	80.3504029	80.5555391	80.6325078	81.5964679	80.4841096	81.0472684	76.601483	76.78752183	76.21073861	76.62059744	76.92296057	76.9556802	76.80730881
Fa														
	18.866072	19.1660252	18.9623832	18.7898781	17.9172464	18.9577468	18.466591	23.1450644	22.85299253	23.47854918	23.02246795	22.74745439	22.74205888	22.85276032
Dat														
a														
set	B21_Ol_c-r	B21_Ol_c-r	B21_Ol_c-r	B21_Ol_c-r	B21_Ol_c-r	B21_Ol_c-r	B21_Ol_c-r	B21_Ol_c-r	B21_Ol_c-r	B21_Ol_c-r	B21_Ol_c-r	B21_Ol_c-r	B21_Ol_c-r	B21_Ol_c-r
SiO														
2	39.8496733	38.8889253	39.3408025	38.6843592	39.1881413	38.586656	38.7881688	38.9316704	37.44373214	38.75356562	38.90520912	38.98357522	38.45434958	39.55249279
TiO														
2	0.0113706	0.0330781	0.03101072	0.01447167	0.04134763	0.03101072	0.02687596	0.02480858	0.033078105	0.024808578	0.009303217	0.012404289	0.027909651	0.034111795
Al2														
O3	0.09645174	0.04978154	0	0.01763096	0.00103712	0.02489077	0.05081866	0.03941039	0	0	0	0.022816541	0.051855775	0.022816541
Cr2														
O3	0	0	0.00952381	0	0	0.00606061	0	0.00779221	0.051082251	0.055411255	0.064935065	0.012121212	0.035497835	0
Fe														
O	20.8084545	21.1629202	21.1588807	21.3770135	21.1932165	21.2669373	21.116466	20.916511	21.21947318	20.80542483	21.1992757	21.05183409	21.02961687	21.02658724
Mn														
o	0.29607181	0.26459574	0.29803906	0.32066249	0.22229976	0.26262848	0.34033503	0.26459574	0.283284653	0.273448381	0.302957199	0.283284653	0.337384153	0.214430744
Mg														
O	37.6661287	39.3636192	39.0931784	39.6480312	39.3785883	39.3316853	40.2348181	39.9763525	39.4274872	39.20594522	39.66998586	39.64603537	39.66499618	39.56520249

Ca	0.22373983	0.18729352	0.23082661	0.22677702	0.19944229	0.23588859	0.1538844	0.22171503	0.193367904	0.18324393	0.208553865	0.192355506	0.192355506	0.150847213
O	0.01200651	0.01200651	0.00461789	0.02031872	0	0	0	0	0	0	0	0.035095966	0.007388624	0.006465046
Na	0.01783312	0	0.00734305	0.00629404	0.01888213	0.00314702	0.00839206	0	0.008392058	0.004196029	0	0	0	0.002098015
2O														
NiO														
Totale	98.9817301	99.9622202	100.174223	100.315559	100.242955	99.7489047	100.719759	100.382856	98.6598975	99.30604384	100.36022	100.2395229	99.80135416	100.5750519
Fo	76.0816985	76.6038843	76.455069	76.5076633	76.6211855	76.5040723	76.968931	77.0844932	76.56990987	76.82508505	76.6799619	76.80861978	76.78995923	76.85170263
Fa	23.578517	23.1035551	23.2137575	23.1407692	23.133058	23.2056852	22.661158	22.625623	23.1175115	22.87047358	22.98731855	22.87955557	22.83893398	22.91164855
Dat				B21_Ol to	B21_Ol to	B21_Ol to	B21_Ol to	B21_Ol to	B21_Ol to	B21_Ol to	B21_Ol to	B21_Ol to	B21_Ol to	B21_Ol to
aset	B21_Ol_c-r	B21_Ol_c-r	B21_Ol_c-r	Opq	Opq	Opq	Opq	Opq	Opq	Opq	Opq	Opq	Opq	Opq
SiO	38.8156479	39.3886364	38.5774963	38.8675527	35.5548037	39.2451348	37.5200628	39.5362089	38.7556011	39.09247358	37.48851277	38.99477038	38.76577851	39.07720745
2														
TiO	0.02894334	0	0.0341118	0.02997703	0.00826953	0.03721287	0.00310107	0.0341118	0.005168454	0.018606434	0.037212868	0.04031394	0.012404289	0.021707506
2														
Al2	0.026965	0.01763096	0.04252174	0.03733616	0.08919193	0.03629904	0	0.01451962	0.030076349	0.06222693	0.060152699	0.049781544	0.02074231	0.046670197
O3														
Cr2	0	0.02597403	0.04329004	0	0.02597403	0	0	0.05021645	0.038095238	0	0	0.025974026	0	0.025974026
O3														
Fe	21.3588358	20.9892219	20.9427677	20.1257796	20.611529	20.6478845	19.9490517	20.4085443	20.32270504	20.62970675	20.19243129	20.30755693	20.16617457	20.46711703
O														
Mn	0.31869523	0.28820279	0.3786965	0.216398	0.23607054	0.32656425	0.2813174	0.29607181	0.265579362	0.291153671	0.30590808	0.30590808	0.254759462	0.243939563
o														
Mg	39.6769714	39.620089	39.288774	40.5162363	42.7675819	40.0931111	42.196762	39.715891	39.47439024	40.20687589	37.75195122	39.9553958	42.03010652	39.4654088
O														
Ca	0.21969024	0.22373983	0.20551667	0.1943803	0.22778942	0.22778942	0.2723349	0.24196298	0.197417493	0.238925787	0.243987774	0.196405096	0.228801813	0.255124145
O														
Na	0	0.00461789	0	0	0.01570083	0.00277073	0	0	0	0	0	0.007388624	0	0.029554498
2O														
NiO	0.00839206	0.00629404	0.01678412	0.00734305	0.00419603	0.00314702	0	0.00104901	0.003147022	0	0.010490073	0.015735109	0.003147022	0.003147022
Totale	100.454141	100.564407	99.5299589	99.9950031	99.5411069	100.619914	100.22263	100.298576	99.0921803	100.539969	96.09064678	99.89922953	101.4819145	99.63585024
Fo	76.5374774	76.845112	76.6574129	78.021846	78.5239673	77.3077355	78.8023131	77.3692768	77.3614519	77.40255251	76.64855586	77.55116061	78.57904892	77.25341005
Fa	23.1132308	22.8372911	22.9227757	21.741389	21.2297655	22.3344985	20.8991943	22.3030216	22.3428281	22.27898784	22.99855953	22.11148917	21.15033488	22.47528327

Data set	B21_Ol to Opq	B21_Ol to Opq	B21_Ol to Opq	B21_Ol to Opq	B21_Ol to Opq	B21_Ol to Opq	B21_Ol to Opq	B21_Ol to Pl							
SiO2	39.1749106	40.9824198	39.0568526	38.7749382	39.2512413	39.1240236	39.1148639	39.3652283	38.56630116	39.30619932	38.63550759	38.75458336	38.56223019	38.75356562	
TiO2	0.01033691	0.01447167	0.01757274	0.03204441	0.02067382	0.02894334	0.01343798	0.01757274	0.022741197	0.042381322	0.022741197	0.019640125	0.04031394	0.029977032	
Al2O3	0.03422481	0.06948674	0.06741251	0.05496712	0.05289289	0	0	0.03941039	0	0	0.026965003	0.012445386	0	0.068449623	
Cr2O3	0	0.13333333	0.04155844	0.00779221	0.02424242	0	0.05800866	0.01385281	0.025974026	0.063203463	0	0.032034632	0.032034632	0.057142857	
FeO	20.0530687	20.0399403	20.2358559	20.2005103	19.8965382	20.1277994	20.0480193	21.3204606	20.91348135	20.81148408	21.34065807	21.37499379	21.50728728	21.38812215	
MnO	0.27148113	0.23016878	0.35508944	0.25082495	0.30590808	0.22623427	0.33246602	0.33443327	0.232136035	0.331482389	0.29902269	0.252792208	0.300989944	0.313777099	
MgO	39.8745629	35.8089682	40.1250451	40.3545705	40.1440059	40.4553622	40.1799316	39.304741	39.48736341	37.17514372	39.2378792	39.57518186	39.56121074	39.22191221	
CaO	0.25309935	0.28954566	0.21057866	0.22070263	0.22981421	0.25714894	0.22778942	0.24196298	0.173119956	0.197417493	0.251074555	0.255124145	0.232851402	0.199442288	
Na2O	0	0	0.04156101	0	0	0	0	0	0	0	0.007388624	0.020318717	0	0.002770734	
NiO	0	0.01049007	0.00629404	0	0	0.00209801	0	0	0.016784116	0.003147022	0.002098015	0	0.005245036	0.013637095	
Total	99.6716844	97.5788245	100.15782	99.8963503	99.9253168	100.22161	99.9745169	100.637662	99.43790125	97.9304588	99.82333494	100.2971142	100.2421632	100.0487967	
Fo	77.7612645	75.8958027	77.6432822	77.8607376	77.9807561	77.9859455	77.8448999	76.3864686	76.89651773	75.80837076	76.36880598	76.53329781	76.37685128	76.30944735	
Fa	21.9379311	23.8270252	21.9663237	21.8642997	21.6816187	21.766269	21.7891301	23.2442493	22.84663894	23.80756651	23.30052629	23.18894304	23.29299108	23.34369763	

Data set	B21_Ol to Pl	B21_Ol to Pl	B21_Ol to Pl
SiO2	38.5459463	38.677235	39.2237622
TiO2	0.01964012	0.02480858	0.03721287
Al2O3	0.07674655	0.0331877	0.03629904

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Cr2			
O3	0.03549784	0.01991342	0.02770563
Fe			
O	21.2154337	21.2012955	21.00437
Mn			
o	0.31180984	0.31476073	0.30000632
Mg			
O	39.6250787	39.841631	39.3396688
Ca			
O	0.18830592	0.2348762	0.16198358
Na			
2O	0.02308945	0.00646505	0
NiO	0.01153908	0	0.00734305
Tot			
ale	100.053088	100.354173	100.138352
Fo	76.6388048	76.7454865	76.6957458
Fa	23.0185492	22.910026	22.9719406

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Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: Poikilitic gabbro B21. Feldspars.

Data Set	B21_7_Pl c-r	B21_7_Pl c-r	B21_7_Pl c-r	B21_7_Pl c-r	B21_7_Pl c-r	B21_7_Pl c-r	B21_7_Pl c-r	B21_7_Pl c-r	B21_7_Pl c-r	B21_6_pl c-r	B21_6_pl c-r	B21_6_pl c-r	B21_5_pl c-r	B21_5_pl c-r	B21_5_pl c-r
SiO2	47.6371717	46.2101729	46.6021944	46.8561518	46.8652218	47.3781754	48.7013742	48.7719179		47.4305793	46.2907942	47.6573270		46.3885477	46.7835925
TiO2	0.05429081	0.01946274	0.031755	0.05121775	0.05121775	0.07989969	0.06965614	0.09014324		0.09526501	0.03994984	0.06760742	0.06351000	0.04404726	0.04916903
Al2O3	33.4807102	33.347222	33.3099695	33.8449572	33.6597293	33.0823151	33.0243668	32.3683083		33.4610491	34.2640481	33.4486316	33.6173028	33.3710222	34.0870986
Cr2O3	0	0	0.03892642	0.01536569	0	0.0051219	0	0					0.01843883	0.06965780	0.00512189
FeO	0.66696178	0.59491344	0.60520606	0.62887908	0.73798086	0.68754702	0.73489307	0.63814244		0.67107882	0.64123022	0.67004956	0.66696177	0.68754701	0.57638672
Mno	0.02111111	0	0	0	0	0.01333333	0.00222222	0.04		0.02111111			0.04333333	0.05111111	
MgO	0.10598531	0.1422951	0.08930244	0.07261957	0.06575015	0.10107859	0.08930244	0.05495535		0.06084342	0.06280611	0.08537705		0.10500396	0.05299265
CaO	17.0040859	17.1122119	17.392127	17.6033265	17.4557901	16.7181076	16.8676651	16.1734352		16.6847603	17.8943574		16.7221496	17.7377262	17.5325898
Na2O	1.89060439	1.60982156	1.75853247	1.76789189	1.7023759	2.10275142	2.27122112	2.50312694		2.22442397		1.98731847	2.09547186	1.76789189	1.44759148
K2O	0.05491603	0.04541133	0.06336465	0.06442072	0.07286934	0.06653288	0.08659835	0.0982152		0.09715912	0.05174779	0.07181326		0.09399089	0.08871050
SrO															
BaO															
Totale	100.915837	99.081511	99.891378	100.90483	100.610935	100.234863	101.847299	100.738245		100.746270	100.857875	100.888126	100.911337	100.316546	100.673253
Ab %	16.6967279	14.5082339	15.4106888	15.3224953	14.93784	18.4695516	19.4966654	21.756488		19.3287690	13.9825112	17.4732077	18.4148970	15.1990288	12.9312999
An %	82.984161	85.2224805	84.223943	84.3101281	84.6414437	81.1459303	80.0142065	77.6818219		80.1157331	85.7223198	82.1113391	81.2064973	84.2692817	86.5472853
Or %	0.3191111	0.2692856	0.36536822	0.3673766	0.42071626	0.38451814	0.48912813	0.56169005		0.29516886	0.41545304	0.37860558	0.53168940	0.52141475	
Data Set	B21_5_pl c-r	B21_5_pl c-r	B21_5_pl c-r	B21_5_pl c-r	B21_5_pl c-r	B21_5_pl c-r	B21_4_pl c-r	B21_4_pl c-r		B21_4_pl c-r	B21_3_pl c-r	B21_3_pl c-r	B21_3_pl c-r	B21_3_pl c-r	B21_3_pl c-r
SiO2	47.3146861	46.5376973	48.9250986	49.4541766	47.3298026	47.9022146	46.1990874	46.7563829		46.3028874	48.9744792	49.1478152	47.8346941	48.7527703	48.1722962
TiO2	0.04712033	0.05941259	0.0819484	0.07477791	0.04302291	0.05736388	0.02765758	0.03585242		0.04507161	0.05941258	0.06760742	0.05838823	0.06351000	0.06248565

Al2O3	33.6742164	33.536589	31.7205283	32.2182634	32.7429032	32.9208875	34.2661177	34.0250111	33.3979268 4	32.3910737 7	32.2534463 7	32.5431882 6	32.7780861 5	32.5773364 2
Cr2O3	0	0.03585328	0	0	0	0	0	0	0	0.10141357	0	0	0	0.02356072 8
FeO	0.65872768	0.56094779	0.67416661	0.65152285	0.68445923	0.60726458	0.64225949	0.59388418	0.66593251 6	0.6474058	0.6988689	0.65769842	0.65152284 8	0.64225949
Mno	0.00333333	0.00777778	0.02888889	0	0.01666667	0.04222222	0.03444444	0.01	0.01666666 7	0	0.03666666 7	0.01555555 6	0	0.06111111 1
MgO	0.08733975	0.06084342	0.09617186	0.10402262	0.10107859	0.10009724	0.0647688	0.07163822	0.11678011 4	0.10991069 5	0.09715320 4	0.06967553 4	0.06084342 1	
CaO	17.2557063	17.2152854	15.3923002	15.3902792	16.398782	16.7019392	17.5113689	17.6690106	0.08047033 17.7033684	15.8298570 8	15.7560888 3	16.3027822 7	16.0370144 8	16.3068243 7
Na2O	1.87708522	1.67741743	2.84006633	2.7506318	1.99147822	2.03827536	1.58070334	1.59838226	1.88436477 3	2.69031548 9	2.69967491 7	2.43865087 8	2.52600553 7	2.32113806 4
K2O	0.07075719	0.08976658	0.13412183	0.11722459	0.08659835	0.08448619	0.05069172	0.05914034	0.07920580 7	0.11300028 5	0.08237403 9	0.10560774 3	0.10243951	
SrO														
BaO														
Totale	100.988972	99.7815906	99.893291	100.760899	99.3947918	100.454751	100.377099	100.819302	100.175894 4	100.925850 1	100.883079 3	100.030485	100.981024 4	100.333463 3
Ab %	16.3806635	14.9108332	24.838549	24.2721984	17.9243996	18.0006166	13.9998648	14.0194237	16.0790629 1	23.3663276 6	23.5145949 6	21.2023861 5	22.0506503 8	20.3576748 6
An %	83.2130521	84.5641326	74.389643	75.0471773	81.5627499	81.5084503	85.704727	85.6392691	83.4762393 9	75.9758303 1	75.8377888 9	78.3263788 1	77.3609576 9	79.0328793 7
Or %	0.40628442	0.52503415	0.77180808	0.68062437	0.51285049	0.49093317	0.29540822	0.34130716	0.44469770 2	0.65784202 2	0.64761615 2	0.47123504	0.58839192 4	0.60944576 6
Data Set	B21_3_pl c- r	B21_3_pl c- r	B21_3_pl c- r	B21_3_pl c- r	B21_2_pl c- r	B21_2_pl c- r	B21_2_pl c- r	B21_2_pl c- r	B21_2_pl c- r	B21_2_pl c- r	B21_2_pl c- r	B21_1_pl c- r	B21_1_pl c- r	B21_1_pl c- r
SiO2	47.8770204	48.0584185	47.0788685	47.3449191	48.3547022	48.4786576	48.4413702	48.2438478	47.8568650 2	47.0113480 5	47.1836763 47.4467036	47.4467036 3	48.9775025 2	47.7107387 2
TiO2	0.07989969	0.05531517	0.04916904	0.05531517	0.06658307	0.0614613	0.06248565	0.0819484	0.04302290 9	0.05326645 9	0.03687677 9	0.05838823 4	0.07170484 9	0.07170484 9
Al2O3	33.0512714	32.8619043	33.8273657	33.1599246	32.7698078	32.0816708	32.4428134	31.7991725	33.0771411 8	33.0305755 2	33.1309503 8	33.4372489 5	31.8146943 7	32.4097000 4
Cr2O3	0.05941401	0.03073138	0	0	0.02356073	0	0.01843883	0	0	0	0.10141357	0	0.03892642 1	0.00307313 8
FeO	0.62990835	0.66181547	0.73798086	0.61858646	0.67622514	0.65049359	0.68240071	0.64123023	0.6988689 0.04555555	4 0.03222222	8 0.04555555	0.60108901 0.04555555	8 0.04555555	0.69372259 0.03777777
Mno	0.02111111	0	0.02111111	0.00222222	0.02333333	0.04555556	0.00333333	0	6	0	2	6	0	8

MgO	0.08243302	0.09519051	0.05691804	0.07850764	0.10009724	0.08243302	0.08243302	0.06967553	0.09519051	0.08733974	0.09420916	0.06673149	0.11874280	0.08733974
CaO	16.8646336	16.6534341	16.9929701	17.0980646	16.0097303	15.6004682	16.2957086	15.9996251	16.3745294	17.2304432	17.0263173	16.8201705	15.1730165	15.7853940
Na2O	2.08091275	2.11315078	1.95716032	2.01019708	2.5249656	2.50104706	2.55408382	2.49272757	2.15058849	2.15474823		1.80012992	2.35025628	2.45113011
K2O	0.09504697	0.08976658	0.06970111	0.08765443	0.11300028	0.11828067	0.09927128	0.11194421	0.10243951	0.09293481	0.09610304	0.06336464	0.11405636	0.09821520
SrO														
BaO														
Totale	100.841651	100.619727	100.791245	100.455391	100.662006	99.6200678	100.682339	99.4401713	100.444201	100.330705	100.431324	100.339382	99.4236418	99.3487962
Ab %	18.153529	18.5772968	17.1781017	17.4552799	22.0592274	22.3313784	21.9717141	21.8511111	19.0882384	18.3578944	17.4398044	16.1640078	21.7416655	21.8095959
An %	81.3008921	80.9034505	82.4193662	82.0439088	77.2912018	76.973726	77.4663788	77.5032162	80.3135040	81.1211305	82.0090338	83.4616193	77.5640951	77.6153984
Or %	0.5455789	0.51925277	0.40253214	0.50081132	0.64957077	0.69489559	0.56190709	0.64567272	0.59825747	0.52097498	0.55116168	0.37437282	0.57500558	0.69423932
Data Set	B21_1_pl c-	B21_1_pl c-	B21_Ol to Pl	B21_Ol to Pl	B21_Ol to Pl	B21_Ol to Pl	B21_Ol to Pl	B21_Ol to Pl	B21_Ol to Pl	B21_Ol to Pl	B21_Ol to Pl	B21_Ol to Pl	B21_Ol to Pl	B21_Ol to Pl
SiO2	48.116869	47.0687908	46.9524922	46.2889246	46.5260584	46.9759002	46.791689	47.7900935	48.5544174	49.2291801	49.1518318	48.824119		
TiO2	0.0409742	0.05736388	0.07545943	0.02894334	0.03928025	0.0330781	0.06305514	0.0444487	0.06925728	0.05168453	0.10336907	0.07029097		
Al2O3	32.7046159	33.4558752	33.8327817	34.0287965	34.2268856	33.4915707	33.5973565	32.6743237	32.8361137	32.353855	32.3673375	31.8747076		
Cr2O3	0.02868263	0	0.06060606	0.02770563	0	0	0.00865801	0.02337662	0.03030303	0	0	0.04761904		
FeO	0.64637654	0.6690203	0.69782296	0.65843787	0.59683555	0.65035888	0.6766156	0.5756282	0.61501328	0.59481580	0.72811917	0.66449711		
Mno	0.01	0	0.0432796	0.04032872	0.02852519	0	0	0	0.01868891	0.01868891	0.13771528	0.12574004		
MgO	0.07850764	0.06967553	0.06087415	0.07484526	0.09380606	0.11077099	0.0838267	0.06985558	0.12374417	0.08582257	6	4		
CaO	16.3896873	17.3244219	17.213793	17.9265208	17.7736488	17.0072639	16.9728424	16.5020776	15.8257961	16.1092674	15.5524488	15.1819114		
Na2O	2.28474029	1.89580407	1.72801455	1.48326637	1.57377702	1.84992686	1.80097722	2.17317918	2.28862643	2.59987224	2.76149840			
K2O	0.08026188	0.05914034	0.04930334	0.03986228	0.0430093	0.07343051	0.06084242	0.05664639	0.10699874	0.10385172	0.10909675	0.12902789		



SrO													
BaO													
Totale	100.380715	100.600092	100.714427	100.597631	100.901826	100.1923	100.055863	99.9096295	100.649763	100.835792	100.749790	99.6892478	
									2	6	7	2	
Ab %	20.051257	16.4735064	15.3290565	12.9932168	13.7762981	16.3761589	16.0511898	19.1815875	22.0112284	20.3271807	23.0772664	24.5773978	
									9	5	9	5	
An %	79.4852681	83.1883598	84.3831662	86.7770246	85.9759804	83.1961343	83.5920162	80.4894295	77.3659496	79.0659040	76.2855620	74.6670110	
									4	1	7	8	
Or %	0.4634749	0.33813373	0.28777727	0.22975858	0.24772155	0.42770678	0.35679405	0.32898294	0.62282187	0.60691523	0.63717144	0.75559107	
									7	3	1	5	

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: Poikilitic gabbro B21. Opaque minerals.

Data set	B21_Opq in Ol_r-r	B21_Opq in Ol_r-r	B21_Opq in Ol_r-r	B21_Opq in Ol_r-r	B21_Opq in Ol_r-r	B21_Opq in Ol_r-r	B21_Opq in Ol_r-r	B21_Opq in Ol_r-r	B21_Opq in Ol_r-r	B21_Opq in Ol_r-r	B21_Opq con MI_c-r	B21_Opq con MI_c-r	B21_Opq con MI_c-r	B21_Opq con MI_c-r	
SiO2	0.04681611	0.0732774	0.07531288	0.04376289	0.05801127	0.2676660 5	0.0630999 8	0.0834548 1	0.0722596 55	0.0936322 3	0.04681611 5	0.01526612 4	0.01933709 1	0.03358547 4	
TiO2	4.65574322	4.65781061	4.72189943	4.78081981	4.71052884	4.7394721 8	4.7187983 6	4.6960571 6	4.7094951 45	4.6578106 06	49.6750436 1	49.6502350 3	49.9613759 5	49.9427695 2	
Al2						10.252923	10.649101	10.708217	10.708217	10.753850	0.57352486	0.53204024	0.56626506	0.59737852	
O3	10.7165144	10.5070171	10.6874752	10.6314709	10.5775409	8	9	5	49	57	9	9	1	5	
Cr2						16.231168	15.907359	15.638095	15.903896	15.817316	0.76450216	0.86580086	0.87359307	0.73246753	
O3	15.8692641	15.9350649	15.7852814	16.0909091	16.0060606	8	3	2	1	02	5	6	4	2	
FeO	55.3289785	56.0793149	55.6420395	55.5673088	55.8480538	55.455212 8	55.705661 5	55.530953 3	55.489548 5	55.172448 05	40.6858050 5	40.6777260 6	40.7443777 4	40.862533 0.34525317	
Mno	0.32361337	0.32853151	0.31672798	0.32164612	0.27246475	0.2783665 2	0.3049244 5	0.2400050 5	0.2803337 71	0.3206624 89	0.34525317 1	0.35508944 4	0.29508818 0.29508818	0.34525317 1	
Mg						6.0245447	6.5275049	6.5295008	6.5195214	6.6292945	7.69908283	7.55737580	7.63621281	7.67712822	
O	6.60235023	6.63628009	6.56642451	6.81091904	6.80293554	9	7	4	74	28	6	3	4	5	
CaO	0.06074384	0	0.00202479	0	0.00506199	0.0192355 5		0.0344215 1	0.0425206 91		0.02024794 8		0.00809917 9	0.06074384 4	
V	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
Ba	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
Ni	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
Zn	0	0	0	0	0	0	0	0	0	0	0	0	0	0	
Tota						93.268590	93.876450	93.460705	93.725792	93.445014	99.8102757		100.104349	100.251859	
l	93.6040238	94.2172966	93.7971856	94.2468367	94.2806577	5	5	5	83	5	6	99.6808683	1	3	
Data set	B21_Opq con MI_c-r	B21_Opq con MI_c-r	B21_Opq con MI_c-r	B21_Opq con MI_c-r	B21_Opq con MI_c-r	B21_Opq con MI_c-r									
SiO2	0	0.0122129	0.01526612	0	0	0.0142483 8									
TiO2	49.9438032	49.9148599	50.048206	49.9882519	49.877647	49.680212 1									
Al2															
O3	0.54655987	0.48951851	0.55589391	0.58285891	0.59115583	0.5755991 0.8138528									
Cr2															
O3	0.81298701	0.78874459	0.73506494	0.75411255	0.84242424	1									

FeO	40.9998759	40.9473624	40.5424029	40.9362538	40.8988885	40.872631	7
						0.2950881	
Mno	0.28623554	0.35312219	0.38558189	0.34426954	0.28721916		8
Mg						7.6082705	
O	7.68311585	7.46556561	7.80686002	7.79488477	7.68211791		8
						0.0556818	
CaO	0.04454549	0.04150829	0.04150829	0.03138432	0.04454549		6
V	0	0	0	0	0	0	0
Ba	0	0	0	0	0	0	0
Ni	0	0	0	0	0	0	0
Zn	0	0	0	0	0	0	0
Tota						99.915584	
I	100.317123	100.012894	100.130784	100.432016	100.223998		7

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: Poikilitic gabbro B21. Glass.

Data set	B21_1_Glass	B21_1_Glass	B21_1_Glass	B21_2_Glass	B21_2_Glass	B21_2_Glass	B21_3_Glass	B21_3_Glass	B21_4_Glass
SiO2	54.9710395	56.0183957	53.4647066	54.0530863	53.849883	53.7053155	54.4665696	54.0854371	54.56362194
TiO2	3.24819984	3.10069604	3.34895947	3.33337809	3.18483553	3.3635021	3.27728509	3.16821538	3.319874218
Al2O3	16.005129	16.3857882	16.1128627	16.0092331	16.1313313	16.4145171	16.122097	16.1241491	16.2257266
Cr2O3	0	0.007	0	0.044	0.011	0.059	0	0.078	0
FeOtot	8.60965153	7.36853731	9.01581567	8.50065559	8.64358423	8.97776992	8.33818993	8.21582656	8.09037839
MnO	0.14282678	0.0753808	0.13489196	0.1676231	0.16068013	0.12794899	0.13885937	0.07835636	0.09025859
MgO	3.76448714	3.2972063	3.80384773	3.6272297	3.7503577	3.64337764	3.52529587	3.66659029	3.453639411
CaO	7.3161925	6.54898532	7.65190865	7.26880914	7.12867537	7.37668189	7.21739315	7.10952039	7.022818923
Na2O	3.41558613	3.47974867	4.17912041	4.18767542	4.32134739	4.49565563	4.64429886	4.36946929	4.273225478
K2O	2.03677136	2.25834046	1.99516214	1.8609724	1.96187476	1.8474494	1.90466208	1.94002992	1.994121905
P2O5	0.76508745	0.77198013	0.80184841	0.6938631	0.85813863	0.73981429	0.81908011	0.81793133	0.844353271
F									
Cl									
Total	100.274971	99.3120589	100.509124	99.7465259	100.001708	100.751033	100.453731	99.6535257	99.87801873

Data set	B21_4_Glass	B21_4_Glass	B21_5_glass	B21_5_Glass	B21_5_Glass	B21_6_glass	B21_6_glass	B21_6_glass	B21_7_glass	B21_7_glass
SiO2	53.638592	53.8266309	53.6547674	54.9134147	54.1905771	53.9388476	53.6780195	53.6982388	53.25745949	53.58905491
TiO2	3.23054093	3.25754867	3.3458432	3.29806028	3.34792072	3.29806028	3.18795181	3.23365721	3.246122319	3.298060276
Al2O3	15.8891869	16.1446698	16.0913159	16.2565077	16.0851597	16.1087585	16.0369361	16.1036284	15.7670887	15.75682834
Cr2O3	0.074	0.022	0.015	0	0.067	0.029	0	0	0.041	0.029
FeOtot	8.59628411	8.67031909	8.98496776	8.40811186	8.65283861	8.29294633	8.56235141	8.38240527	8.525333914	8.927385
MnO	0.11505491	0.11505491	0.1676231	0.0753808	0.1130712	0.11703861	0.10712009	0.12001417	0.124973433	0.249946865
MgO	3.61814649	3.57474892	4.14699133	3.75843167	3.616128	3.65750708	3.65447934	3.739256	3.557591736	3.307298758
CaO	7.20327896	7.14984666	7.49766069	7.26679282	7.03290049	6.99459054	7.2627602	7.28292333	7.197230016	7.727520392
Na2O	4.59938508	4.28284986	3.82943454	3.90215209	3.92995586	4.6336051	4.37909368	4.42186871	4.396203689	4.722363291
K2O	1.92546669	1.90570231	1.94002992	1.98892075	1.98580006	1.91402415	1.85681148	1.9327483	1.957713837	1.943150609
P2O5	0.78346793	0.89719715	0.84665083	0.90983373	0.76049233	0.70879724	0.80069963	0.75015331	0.731772835	0.869626431
F										
Cl										
Total	99.6734041	99.8465682	100.520285	100.777606	99.7818441	99.6931755	99.5262232	99.6648934	98.80248997	100.4202349

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: Poikilitic gabbro B32. Clinopyroxenes.

Data Set	89/1.	89/4.	89/5.	89/6.	89/7.	89/8.	89/9.	89/10.	89/11.	89/12.	89/13.	89/14.	89/15.	89/16.	
SiO <sub>2</sub>	48.264361		48.922839	48.807835	49.352326	49.205772	50.011823	49.805221			51.5659149	49.9884154	50.1726266	50.4331685	
	1	50.498304	9	1	9	1	5	9	50.5552975	51.5771101	4	2	5	1	
TiO <sub>2</sub>	2.4033310	1.5019526	2.1469757	2.1242345	1.9836525	1.8451380		1.6477030	1.48437994	0.98200623	1.22905832	1.50402006	1.52986233	1.62806296	
	4	9	3	3	9	2	1.7800155	9	5	1	5	9	8	2	
Al <sub>2</sub> O <sub>3</sub>	5.5164173	3.1787589	4.6670197	4.7012445	4.4958956	4.1422392	4.0094885	3.8207334		2.29513659	2.54404431	3.15801668	3.15594245	3.14764552	
	2	9	3	4	7	9	1	9	3.24409727	2	1	4	3	9	
FeO			8.5697910	8.5485836	8.5344454	8.4607246	8.4778925		8.14160445	7.98911347	8.09212062	7.89216556	8.21936475	8.32641140	
	8.9060791	8.4122507	5	9	5	5	1	8.481932		5	6	8	9	6	
MnO	0.2144307	0.1790201	0.1603312	0.2508249	0.1937745	0.2104962	0.2114798	0.2193488		0.17902016	0.21836525	0.18492192	0.10623174	0.20557809	
	4	6	4	5	7	4	6	8	0.19082369	3	4	6	5	9	
MgO	12.829476	15.064854	13.326448	13.106902	13.058003	13.492106	14.260517	13.758555		14.7834365		16.2623789	14.5279647	14.5429337	
	2	8	8	7	7	3	7	4	14.207627	7	14.7435191	9	3	9	
CaO	21.579250	21.660242	21.397019	21.771606	21.868796	21.789829		21.525593	22.1178460	21.6845399	21.2330106		21.5691266		
	6	4	1	1	3	3	21.565077	5	2	3	9	19.8672866	3	21.65923	
Na <sub>2</sub> O	0.5365988		0.0840456	0.3999093	0.4460882		0.3906735	0.4008328		0.38236131	0.42761664	0.50704435	0.25767827	0.37127838	
	5	0	0.0840456	0.3999093	0.4460882	0	2	8	0	8	3	6	9	1	
K <sub>2</sub> O	0	0	0	0	0		0.0073430	0.0041960	0.0073430	0.01049007					
	0.0484848	0.0424242	0.0398268	0.0129870	0.0813852	0.0060606	0.0129870	0.0164502			0.06320346		0.09870129	0.01471861	
Cr <sub>2</sub> O <sub>3</sub>	5	4	4	1	8	1	1	2		0	0.05021645	3	0.03982684	9	
Totale	100.24994	100.49538	99.274471	99.711140	99.932983	99.153648	100.71116	99.667264	99.9521659	99.8727243	100.053649	99.3642496	99.5387975	100.314308	
	5	4	1	9	4	9	4	3	6	8	9	1	9	7	
Wo	45.571239	44.036004	45.739088	45.927055	46.033335	46.197504	44.252879	44.830890	45.8484738	44.0904340	43.4771366	40.0772658	44.3214800	44.1437391	
	9	3	7	2	5	1	2	7	8	4	2	9	8	1	
En	37.697652	42.614798	39.636889	38.470612	38.245051		40.717087	39.869975	40.9783529	41.8235084	42.0050407	45.6451359	41.5372145		
	8	6	5	7	8	39.801124	6	4	8	9	4	7	6	41.2409849	
Fs	14.680450	13.349197	14.298906	14.075720	14.022363		13.579280	13.788447	13.1731731	12.6791779	12.9333208	12.4266499	13.1831227	13.2459257	
	4	1	1	9	5	14.001372	8	4	3	8	1	1	3	2	
N° Mg			73.489023	73.212744	73.171882	73.976352	74.990443	74.303293	75.6734960	76.7365999	76.4584883	78.6012265	75.9081844	75.6897104	
	71.972161	76.146812	5	1	7	4	7	6	6	7	1	3	5	8	
Data Set	89/17.	89/19.	89/20.	91/1.	91/3.	91/4.	91/6.	91/7.	91/9.	91/10.	91/11.	91/12.	92/1.	92/2.	92/4.
SiO <sub>2</sub>	48.207367	50.239797	49.909031	50.443345	51.062132	50.762916	50.393476	50.744597	50.1889105		50.2581169	50.8433183	49.9375283		51.2412553
	6	6	6	9	8	8	6	4	2	50.1492186	5	8	4	49.7115897	6
TiO <sub>2</sub>	1.6725116	1.5722436	1.5608730	1.5877490	1.5143569	1.5381318	1.5029863	1.5277949	1.61979343	1.68491595	1.62806296	1.62909665	1.79241979	1.77794812	1.68388226
	6	6	6	2	8	6	8	6	5	4	2	2	4	3	3
Al <sub>2</sub> O <sub>3</sub>	3.0605278	2.9868926	3.1839445	2.9422966	2.9298512	2.9692616	2.9308883	3.0335628	3.28869323	3.23165188	3.30736131		3.62679288	3.72428174	4.07586389
	3	3	7	6	8	6	9	2	7	4	6	3.232689	8	5	8

FeO	8.4284086 8	8.6122057 5	8.2506708 5	8.4273988 1	8.5687811 7	8.5324257 1	8.6061465 1	8.3890235 9	8.57787003 8	8.54454419 4	8.46274439 8	8.65563033 8	7.84066199 3	7.87701745 9	7.90630380 6
Mno	0.1731184 9	0.1475440 9	0.1691838 9	0.1672166 4	0.1819710 4	0.1986927 1	0.1534458 5	0.1632821 3	0.21639799 9	0.18688918 1	0.17705290 8	0.22820152 6	0.20164359 8	0.11606801 8	0.11705164 5
MgO	15.695550 9	14.507008 1	14.380270 1	14.376278 3	14.162719 8	14.374282 5	14.103841 6	14.448129 8	14.3443443 5	14.6507109 7	14.2844681 4	14.5279647 3	14.1587281 14.1587281	14.0958580 8	11.6618900 9
CaO	20.602287 1	21.491172 21.491172	21.224911 5	21.329188 4	21.287680 2	21.723011 21.723011	20.925241 9	21.139870 1	21.0943122 5	21.0862130 7	21.2674322 21.2674322	21.1520189 21.1520189	21.7716061 21.7716061	21.8930938 1	21.3555107 8
Na2O	0.2207351 6	0.3306409 0	0.4386995 5	0.3417238 8	0.2872327 8	0.3666604 0	0.39898572 3	0.40360361 9	0.35095966 3	0.10713505 0.43869958	0.37497269 3	0.07111551 4	0.10713505 5	0.37497269 8	0.07111551 1
K2O	0.0010490 1	0.0167841 2	0.0010490 0	0.0073430 1	0.0199311 5	0.0199311 4	0.0041960 4	0.00209801 3	0.01468610 0	0.01468610 0	0.01468610 5	0.01468610 2	0.00839205 0.01468610	0.01993113 2	0.01993113 8
Cr2O3	0.0337662 3	0.0441558 4	0.0380952 4	0.0545454 5	0.0164502 2	0.0545454 5	0.0709956 7	0.0796536 8	0.06666666 0	0.06666666 0	0.09610389 7	0.02077922 6	0.02077922 1	0.01904761 0	0.01904761 9
Totale	98.061556 3	99.573647 9	99.009526 5	99.713222 4	100.05656 100.05656	100.11865 3	98.923191 1	99.817117 4	99.7293075 5	99.9728434 2	99.7909405 1	100.634565 3	99.4512019 7	99.5792216 7	98.1328044 8
Wo	41.688808 4	44.404723 1	43.969468 7	43.793375 4	44.073546 7	44.897759 4	43.788976 2	43.628307 2	43.0907326 43.5199727	43.8696229 3	43.3821727 2	45.5607516 8	45.2964120 6	48.6660444 5	
En	44.190763 7	41.705930 6	41.449879 3	40.798784 41.070639	41.066020 7	41.488580 41.337263	41.1770134 9	41.6576614 9	40.9981082 7	41.4586492 5	41.2264132 1	40.5787889 3	36.9773621 2	14.0633223 5	
Fs	13.312143 2	13.889346 4	13.341139 9	13.505976 3	13.847362 5	13.764977 6	14.057282 9	13.513747 8	13.8134147 7	13.6292680 9	13.6256872 1	13.8565948 9	12.8071203 5	12.7208701 5	14.0633223 1
N° Mg	76.849617 9	75.017038 8	75.650863 8	75.253180 9	74.659947 2	75.019205 4	74.498475 3	75.430589 7	74.8803287 9	75.3481189 7	75.0554001 2	74.9497717 9	76.2978293 2	76.1332992 2	72.4468383 5

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: Poikilitic gabbro B32. Olivines.

Data Set	90/1.	90/2.	90/3.	90/4.	90/5.	90/6.	90/7.	90/8.	90/9.	90/10.	90/11.	93/5.	93/7.	93/8.
SiO2	38.3871786	38.763743	38.4533318	38.2925287	38.6039576	38.6711285	38.6405963	38.5266092	38.6039576	38.39023186	38.71692692	38.30779478	37.51090309	37.95972715
TiO2	0.01757274	0.0444487	0.02997703	0.02170751	0.00206738	0.03101072	0.01033691	0.04341501	0.027909651	0.015505362	0.028943342	0.029977032	0.011370598	0.039280249
Al2O3	0	0.02177943	0	0.01659385	0	0	0.00622269	0.01970519	0	0	0.006222693	0.035261927	0	0.049781544
Cr2O3	0.02943723	0.00606061	0	0	0	0	0.02510823	0	0	0	0	0	0.015584416	0
FeO	23.7370891	24.2359669	24.0420711	24.1642659	23.9117974	24.4662182	23.8693826	23.7158818	23.87746164	24.10165368	23.70578305	23.99763665	24.15416712	24.25717428
Mno	0.36295846	0.30197357	0.35804033	0.37082748	0.32164612	0.27443201	0.34820405	0.30197357	0.250824953	0.362958462	0.318695235	0.279350144	0.294104553	0.309842589
MgO	36.1632358	36.2191202	36.2111367	36.1782048	36.1821966	36.300951	36.4666086	36.0853967	35.92672472	35.97362775	36.00955348	37.93756748	37.61722975	37.48350621
CaO	0.31586799	0.24500017	0.28144648	0.28448367	0.28853326	0.30574402	0.30574402	0.29055805	0.247024966	0.274359696	0.226777018	0.33510354	0.263223324	0.245000171
Na2O	0	0.00923578	0.00923578	0	0	0	0	0	0	0	0	0	0	0
NiO	0	0.00524504	0.00629404	0	0.00944107	0.00734305	0	0.00314702	0.001049007	0	0	0.004196029	0.002098015	0
Totale	99.01334	99.8525735	99.3915333	99.3286118	99.3196393	100.056828	99.6722034	98.9866866	98.93495253	99.1183368	99.01290173	100.9268876	99.86868087	100.3443122
Fo	72.7842967	72.4576779	72.5649547	72.4367139	72.6856812	72.3386819	72.8535218	72.8099099	72.63204143	72.3809663	72.76259747	73.58162252	73.27888413	73.11381005
Fa	26.8006494	27.1990863	27.0273892	27.1414335	26.9471976	27.3506014	26.7512334	26.8439076	27.07984842	27.20410425	26.87151869	26.11053683	26.39560089	26.54280796
Data Set	93/9.	93/10.	93/11.	93/12.	93/13.	93/14.	93/15.	93/17.	93/18.	93/19.	93/20.	93/21.	93/22.	93/23.
SiO2	37.6208192	37.7409127	37.7215756	37.2137225	37.7022385	35.8163633	37.7093627	37.554666	38.00857875	36.90331134	37.65847563	37.77246269	37.7826401	37.58316275
TiO2	0.0113706	0.02687596	0.04238132	0.02790965	0.02687596	0.03824656	0.03928025	0.01860643	0.042381322	0.046516085	0.036179177	0.023774888	0.034111795	0.015505362
Al2O3	0.02281654	0.01244539	0.03007635	0.02177943	0.27276138	0.02489077	0	0.0134825	0.015556732	0	0	0.002074231	0.03215058	0.035261927
Cr2O3	0.02164502	0	0	0	0	0	0.03896104	0.00779221	0.01991342	0.037229437	0.040692641	0.091774892	0	0.043290043
FeO	24.1814337	24.2056707	23.8299976	23.7421385	23.978449	24.002686	24.059239	24.0945846	23.80576058	23.8168692	24.18345347	24.53589951	24.12084128	23.99460703
Mno	0.35804033	0.32853151	0.32951513	0.33935141	0.36295846	0.32853151	0.34426954	0.37574562	0.408205316	0.362958462	0.416074334	0.335416898	0.258693972	0.331482389
MgO	37.6062524	37.6062524	37.3777249	37.4326114	37.6860874	37.6032586	37.6222194	37.6421782	37.72101518	37.83877173	37.25497867	37.50546082	38.20601249	37.90463556
CaO	0.23690099	0.24297538	0.31991758	0.34016553	0.24297538	0.31384319	0.25613654	0.25208695	0.270310106	0.275372093	0.226777018	0.272334901	0.288533259	0.30776881
Na2O	0.00461789	0	0	0	0.01385367	0.02031872	0	0	0	0	0	0	0	0
NiO	0	0	0.00104901	0	0	0	0.01049007	0.02202915	0.006294044	0.006294044	0.008392058	0	0.008392058	0
Totale	100.063897	100.163664	99.6522375	99.1176785	100.2862	98.1481387	100.079959	99.9811716	100.2980155	99.28732239	99.825023	100.5391988	100.7313755	100.2157139
Fo	73.199543	73.2040592	73.3859228	73.4775447	73.3995737	73.3651276	73.3168271	73.2733727	73.5194922	73.60800614	72.9662613	72.88248305	73.63693462	73.52489632

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Fa | 26.4044915 26.4325873 26.2464967 26.1439854 26.1987774 26.2706905 26.3019885 26.3110589 26.02846977 25.99082969 26.57073347 26.74718534 26.07977783 26.10977838



Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: Poikilitic gabbro B32. Feldspars.

Data Set	82/1.	82/2.	82/3.	82/4.	82/5.	82/6.	82/7.	82/8.	82/9.	82/10.	82/11.	82/12.	82/13.	82/14.
SiO2	47.3494114	47.9590386	46.7794761	47.2099808	45.9632473	47.2791872	47.1285615	46.4537988	47.06546147	47.02169858	46.84461154	46.98404214	47.32396784	47.65066291
TiO2	0.07959419	0.06098776	0.07132466	0.05375192	0.04858347	0.09303217	0.05168454	0.06512252	0.049617157	0.054785611	0.056852992	0.065122518	0.046516085	0.043415012
Al2O3	33.4822367	33.431418	33.2395516	33.4604572	32.6743237	33.3453374	33.1223576	33.5174986	33.77781458	33.66269476	33.42830665	33.72388457	33.52994397	33.5517234
Cr2O3	0	0	0.03636364	0.02164502	0.05541126	0	0	0.004329	0.032034632	0.064069264	0	0	0	0.012987013
FeO	0.57259858	0.53725299	0.54634185	0.56350971	0.56148996	0.5715887	0.58875656	0.54634185	0.513016009	0.579667695	0.618042908	0.607944168	0.642279885	0.629151523
Mno	0.03049245	0	0	0	0	0.03639421	0.04032872	0	0	0	0	0	0.036394209	0.008852645
MgO	0.10478337	0.0648659	0.05987621	0.04490716	0.04889891	0.06686177	0.0459051	0.07983495	0.044907158	0.071851454	0.042911285	0.061872085	0.026944295	0.060874148
CaO	16.6498877	16.6225529	16.8746399	16.6033174	16.8179456	16.4585445	16.513214	17.0315615	17.0619334	16.89286303	17.02751189	16.88780105	16.68025958	16.68937116
Na2O	2.05957908	2.15193689	2.01524733	1.98015137	1.97738063	2.02448311	1.9367432	2.06604413	1.942284665	1.830531719	1.793588597	1.946902555	2.051266876	2.055884767
K2O	0.09336165	0.10175371	0.08496959	0.09231264	0.07552852	0.08811661	0.05454838	0.0723815	0.076577531	0.0576954	0.065038451	0.076577531	0.087067604	0.08496959
SrO														
BaO														
Totale	100.421945	100.929807	99.7077909	100.030033	98.2228093	99.9635457	99.4820995	99.8369128	100.5636466	100.2358575	99.87686431	100.3541466	100.4246403	100.7878922
Ab %	18.1913588	18.8691605	17.6836755	17.654873	17.4669765	18.1121918	17.4515268	17.926136	17.00610462	16.33893124	15.9489743	17.18437202	18.11101427	18.13853818
An %	81.2660568	80.5437756	81.8257333	81.8035759	82.0940376	81.3690956	82.2250618	81.6606388	82.55272593	83.32222486	83.6704928	82.37089134	81.38317353	81.36819814
Or %	0.54258437	0.58706392	0.49059126	0.54155115	0.43898589	0.51871253	0.32341141	0.41322519	0.44116945	0.338843907	0.380532907	0.444736646	0.505812203	0.49326368
Data Set	84/4.	84/5.	84/6.	84/9.	87/7.	87/8.	92/11.	92/12.	92/14.	93/3.				
SiO2	47.4644162	47.5712791	45.8960763	46.6980567	50.9359329	46.0751989	47.4145468	50.1512541	47.29241785	46.87514379				
TiO2	0.09406586	0.05892037	0.07649312	0.04548239	1.92163114	0.0558193	0.06822359	0.0899311	0.056852992	0.075459426				
Al2O3	33.3920076	33.8856746	33.2499228	33.5413522	21.5232579	32.7655899	34.0516131	32.3248158	33.6989938	32.77388678				
Cr2O3	0	0	0	0.02597403	0	0	0	0.00606061	0	0				
FeO	0.63723051	0.59279606	0.58471707	0.45848281	6.397552	0.50291727	0.68570447	0.70994145	0.744277162	0.591786183				
Mno	0	0	0.02852519	0.00196725	0.10918263	0.02360705	0.01475441	0.01475441	0.038361463	0				
MgO	0.0758432	0.04490716	0.04690303	0.06386796	1.78730491	0.05388859	0.08781844	0.06287002	0.075843201	0.074845264				
CaO	17.1975947	17.0912929	17.1732971	16.9333589	10.8771977	16.9313341	17.0801566	14.8447831	17.12166485	16.38058995				
Na2O	1.81298374	1.43339315	1.98107494	1.78527639	4.34543478	1.90995943	1.39829719	2.57678279	1.824066673	2.181491383				

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K2O	0.10280271	0.08496959	0.07343051	0.0870676	0.71437396	0.07657753	0.10490073	0.16469414	0.113292786	0.071332495
SrO										
BaO										
Totale	100.776944	100.763233	99.1104401	99.6408864	98.6118678	98.394892	100.906015	100.945887	100.9657708	99.02453526
Ab %	15.9256847	13.1095554	17.1977837	15.9400442	40.1380538	16.8774901	12.8215669	23.6654346	16.05685406	19.33886674
An %	83.4801314	86.3791181	82.3827856	83.5484474	55.5202364	82.6772668	86.5455372	75.3393275	83.28694913	80.24505307
Or %	0.59418391	0.51132645	0.41943071	0.51150838	4.3417098	0.44524309	0.63289582	0.99523795	0.656196806	0.416080188

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: Poikilitic gabbro B32. Opaque minerals.

Data set	B32_Opq c to big Ol r	B32_Opq c to big Ol r	B32_Opq r-r at big Ol r	B32_Opq r-r at big Ol r	B32_Opq r-r at big Ol r	B32_Opq r-r at big Ol r	B32_Opq 1_r-r	B32_Opq 1_r-r	B32_Opq 1_r-r	B32_Opq 1_r-r	B32_Opq 1_r-r	B32_Opq 1_r-r	B32_Opq 1_r-r	B32_Opq 1_r-r
SiO2	0.07225966	0.04070967	0.15367899	0.06513546	0.14350157	0.40913213	0.1272177	0.08956126	0.11296932	0.08141933	0.059029014	0.103809646	0.07531288	0.08141933
TiO2	6.60528402	6.42645551	13.8152771	14.0003078	13.9992741	13.6860658	17.596518	17.7019544	17.61925916	17.61925916	17.71332502	17.77741385	17.82186255	17.83633422
Al2O3	9.7717022	9.98949645	6.02978949	6.44567281	5.98415641	5.50293482	4.52493491	4.56330818	4.657685691	4.685687809	4.575753567	4.517675099	4.603755685	4.596495877
Cr2O3	12.5125541	11.8805195	6.85627706	7.11428571	6.86406926	6.70995671	0.31948052	0.28398268	0.34978355	0.254545455	0.333333333	0.351515152	0.289177489	0.317748918
FeO	59.5775186	59.4805707	61.4801213	62.1930923	61.866903	59.6280123	69.5480049	69.2369637	68.801708	69.11880845	68.65224665	68.77343153	68.95520886	69.03599878
Mno	0.24393956	0.35607307	0.32558063	0.31771161	0.34230229	0.2921373	0.28328465	0.36197483	0.339351408	0.340335035	0.319678862	0.315744353	0.303940826	0.378696498
MgO	6.24309296	6.16724976	6.70014804	6.33190934	7.26697618	4.53562301	5.62237624	5.69921738	5.736141042	5.564495903	5.591440198	5.394846638	5.566491777	5.542541293
CaO	0.00303719	0	0.00506199	0.01822315	0.07289261	0.12958687	0.02632233	0.00303719	0.00506199	0.013161166	0.041508293	0.032396717	0.014173564	0.008099179
BaO														
V2O3														
NiO														
ZnO														
Total	95.0293883	94.3410746	95.3659346	96.4863382	96.5400754	90.8934489	98.0481392	97.9399997	97.62196016	97.67771231	97.28631494	97.26683298	97.62992363	97.7973341
Data set	B32_Opq1_r-r	B32_Opq1_r-r	B32_Opq1_r-r	B32_Opq1_r-r	B32_Opq1_r-r	B32_Opq1_r-r	B32_Opq 1_r-r	B32_Opq 1_r-r	B32_Opq 1_r-r	B32_Opq 1_r-r	B32_Opq 1_r-r	B32_Opq 1_r-r	B32_Opq 1_r-r	B32_Opq 1_r-r
SiO2	0.09363223	0.10380965	0.07633062	0.0854903	0.1394306	0.30939345	0.09973868	0.15978544	0.088543521	0.134341895	0.084472555	0.083454813		
TiO2	17.8363342	17.7019544	17.6037538	17.7887844	17.6978197	17.3267247	17.5820463	17.7980877	17.71435871	17.35153325	17.46523924	17.51899116		
Al2O3	4.678428	4.53530606	4.29884373	4.53426895	4.60582992	5.34944173	4.58923607	4.49174721	4.780065319	3.636126928	4.566419527	4.613089724		
Cr2O3	0.30562771	0.44415584	0.32034632	0.22770563	0.26147186	0.26753247	0.29350649	0.27532468	0.278787879	0.345454545	0.296969697	0.243290043		
FeO	68.7007206	68.70577	69.1925293	68.8744189	69.314724	67.7564884	68.4876372	69.0087322	69.31169439	67.13137635	69.38844482	69.01782105		
Mno	0.40525443	0.28820279	0.32558063	0.33049876	0.2921373	0.32361337	0.27148113	0.24197231	0.34918768	0.364925717	0.293120926	0.352138562		

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MgO	5.56748971	5.59842576	5.36890028	5.31700756	5.45372491	5.55651241	5.550524 79	5.512603 19	5.678260 705	5.472685 713	5.804998 685	5.571481 461
CaO	0.01619836	0	0.04657028	0	0	0.14679762	0.027334 73	0	0.048595 075	0.079979 395	0.061756 241	0.016198 358
BaO												
V2O3												
NiO												
ZnO												
Total	97.6036853	97.3776245	97.2328549	97.1581746	97.7651383	97.0365041	96.90150 54	97.48825 27	98.24949 328	94.51642 38	97.96142 169	97.41646 517

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: Porphyrogabbro. B20. Clinopyroxenes.

Data set	B20_Cpx-c_Opq_PI	B20_Cpx-c_Opq_PI	B20_Cpx-c_Opq_PI	B20_Cpx-c_Opq_PI	B20_Cpx-c_Opq_PI	B20_Cpx-c_Opq_PI	B20_Cpx-c_Opq_PI	B20_Cpx-c_Opq_PI	B20_Cpx-c_Opq_PI	B20_Cpx-c_Opq_PI	B20_Cpx-c_Opq_PI	B20_Cpx-c_Opq_PI	B20_Cpx-c_Opq_PI	B20_Cpx-c_Opq_PI
SiO2	51.376	51.517	51.363	50.696	51.021	51.346	51.084	51.041	51.362	51.418	51.139	50.81	50.688	47.871
TiO2	0.843	0.88	0.869	0.941	0.977	0.926	0.962	0.932	0.982	0.986	0.96	1.017	1.039	1.751
Al2O3	2.663	2.8	2.651	2.803	2.866	2.762	2.856	2.952	2.889	2.879	2.941	3.048	3.144	5.687
FeO	5.008	5.399	5.289	5.223	5.519	5.347	5.575	5.546	5.675	5.745	5.708	5.762	5.918	6.238
MnO	0.134	0.153	0.086	0.171	0.122	0.116	0.122	0.072	0.054	0.107	0.176	0.121	0.151	0.112
MgO	16.152	16.436	16.405	16.488	16.254	16.454	16.231	16.105	16.207	16.172	16.267	15.843	16.002	14.436
CaO	21.546	21.37	21.343	21.201	21.645	21.468	21.558	21.615	21.36	21.222	21.045	21.609	21.423	21.284
Na2O	0.269	0.294	0.224	0.233	0.264	0.279	0.223	0.262	0.248	0.266	0.255	0.273	0.278	0.33
K2O	0.001	0.003	0.013	0.004	0	0	0	0	0.003	0	0.015	0.013	0.013	0.012
Cr2O3	0.575	0.689	0.623	0.636	0.572	0.638	0.566	0.592	0.568	0.541	0.381	0.449	0.495	0.603
Total	97.992	98.852	98.243	97.76	98.668	98.698	98.611	98.525	98.78	98.795	98.506	98.496	98.656	97.721
Wo	44.502187	43.6254316	43.8231929	43.5868849	44.1317412	43.7760616	44.0887341	44.2690216	43.78058753	43.58661492	43.30633999	44.42235525	43.89017184	45.44326809
En	46.4185855	46.6855248	46.8679205	47.1648397	46.111009	46.6839389	46.1865302	45.8940423	46.22041785	46.21484089	46.57586356	45.31638228	45.61547038	42.88585548
Fs	8.07378844	8.60293988	8.47657611	8.3814249	8.78318869	8.51047169	8.89943386	8.86590137	9.079136583	9.209906283	9.168216538	9.24567263	9.463685534	10.39585045
N° Mg	85.1836361	84.4399009	84.6839765	84.9109117	83.9997868	84.5809175	83.844462	83.8095133	83.58189921	83.38304322	83.55302209	83.05475729	82.8180273	80.48889339
Data set	B20_Cpx-c_Opq_PI	B20_Cpx in smecolato_c -r	B20_Cpx in smecolato_c -r	B20_Cpx in smecolato_c -r	B20_Cpx in smecolato_c -r	B20_Cpx in smecolato_c -r	B20_Cpx in smecolato_c -r	B20_Cpx in smecolato_c -r	B20_Cpx in smecolato_c -r	B20_Cpx in smecolato_c -r	B20_Cpx in smecolato_c -r	B20_Cpx in smecolato_c -r	B20_Cpx in smecolato_c -r	B20_Cpx in smecolato_c -r
SiO2	47.844	50.967	50.466	50.637	50.545	50.683	50.465	50.315	50.818	50.871	50.986	50.627	50.694	50.846
TiO2	1.99	1.15	1.184	1.188	1.215	1.213	1.201	1.197	1.242	1.203	1.268	1.183	1.201	1.238
Al2O3	6.216	2.017	2.057	2.008	2.001	1.981	1.904	2.067	2.09	2.035	1.961	2.057	1.997	1.944

Fe														
O	6.545	8.451	8.547	8.235	8.469	8.414	8.396	8.442	8.471	8.155	8.128	8.278	7.935	8.018
Mn														
o	0.13	0.193	0.245	0.212	0.207	0.276	0.247	0.257	0.264	0.247	0.244	0.242	0.284	0.243
Mg														
O	13.922	15.105	14.971	14.873	15.124	15.117	15.079	14.913	15.057	15.299	15.069	15.26	15.423	15.33
Ca														
O	21.517	20.444	20.658	20.459	20.843	20.552	20.663	20.512	20.451	20.314	20.982	20.674	20.557	20.299
Na														
2O	0.475	0.336	0.349	0.295	0.293	0.277	0.381	0.321	0.323	0.364	0.328	0.396	0.387	0.411
K2														
O	0	0.006	0.008	0	0.001	0.006	0.008	0	0.004	0	0	0.005	0	0
Cr2														
O3	0.711	0	0	0.048	0	0.028	0.008	0.025	0	0.033	0.01	0.063	0	0.053
Tot														
al	98.639	98.669	98.485	97.907	98.698	98.519	98.344	98.024	98.72	98.488	98.966	98.722	98.478	98.329
Wo	45.9212865	42.0092522	42.339359	42.5211552	42.5131856	42.2406811	42.2670361	42.3520305	42.08217903	41.77683902	42.91957311	42.11904859	42.02986154	41.74062478
En	41.341369	43.1867683	42.6931023	43.010068	42.9221024	43.2308059	42.9172472	42.8432241	43.10950334	43.77781395	42.88879475	43.25730636	43.87508594	43.86088141
Fs	10.9028572	13.5545632	13.6731336	13.3592662	13.4832287	13.4982565	13.405382	13.6053564	13.60556869	13.09068705	12.9774855	13.16369494	12.66320102	12.86911555
N°														
Mg	79.1309816	76.1116582	75.7423334	76.3004719	76.0958257	76.2057473	76.1989413	75.8977882	76.01066489	76.98077703	76.77045	76.66880304	77.60243244	77.31514853
B2O														
Cpx in														
PI														
smecolato_c														
-r														
B2O														
Cpx														
brught_dark														
_OI														
SiO														
2	51.048	50.592	51.346	51.698	51.56	51.049	45.361	45.385	47.59	50.068	51.233	50.974	51.069	51.324
TiO														
2	1.215	1.202	1.188	0.994	0.97	1.206	2.64	2.624	2.168	1.503	1.177	1.175	1.105	1.068
Al2														
O3	1.965	1.951	2.011	1.795	1.762	2.167	7.473	7.504	5.389	3.617	2.377	2.235	2.039	2.059
Fe														
O	7.452	7.451	7.007	6.544	6.026	5.866	6.745	6.686	6.25	5.887	5.496	5.603	5.575	5.481
Mn														
o	0.204	0.129	0.134	0.162	0.124	0.121	0.159	0.149	0.097	0.094	0.186	0.092	0.099	0.101
Mg														
O	15.511	15.506	15.592	16.109	16.294	15.892	13.035	13.01	13.91	15.043	15.92	15.846	15.928	15.831
Ca														
O	20.324	20.642	21.038	21.012	21.481	20.97	21.523	21.435	20.96	21.029	21.608	20.989	21.145	21.423
Na														
2O	0.382	0.436	0.493	0.434	0.446	0.508	0.572	0.534	0.642	0.566	0.534	0.512	0.472	0.511

K2	0.01	0.006	0.003	0	0	0	0.001	0.003	0.009	0.013	0	0.007	0	0
O														
Cr2														
O3	0.03	0	0.03	0.117	0.061	0.16	0.544	0.549	0.628	0.456	0.439	0.518	0.427	0.481
Total	98.111	97.915	98.812	98.748	98.663	97.779	97.509	97.33	97.015	97.82	98.531	97.433	97.432	97.798
Wo	41.9797189	42.280261	42.8524144	42.604033	43.2534628	43.1658729	46.8301171	46.8810108	45.2167384	44.19797952	44.0842379	43.42222799	43.58639107	44.03214908
En	44.5780468	44.1912782	44.1899561	45.4467467	45.6504431	45.5167822	39.4624797	39.5914743	41.75283624	43.99153833	45.19212133	45.61323824	45.68307895	45.27399742
Fs	12.0143847	11.9123871	11.1404135	10.3567849	9.47095691	9.42502482	11.4552016	11.4140087	10.52413276	9.657754862	8.752131551	9.047723598	8.969876012	8.79321604
N°														
Mg	78.7703331	78.7671857	79.8656442	81.4406282	82.8180037	82.8454408	77.5025073	77.6219967	79.86851005	81.99835583	83.77559965	83.44755874	83.58757359	83.73650966
Data set	B20_Cpx brught_dark _Ol	B20_Cpx brught_dark _Ol	B20_Cpx brught_dark _Ol	B20_Cpx brught_dark _Ol	B20_Cpx brught_dark _Ol	B20_Cpx brught_dark _Ol	B20_Cpx brught_dark _Ol	B20_Cpx brught_dark _Ol	B20_Cpx brught_dark _Ol	B20_Cpx brught_dark _Ol	B20_Cpx brught_dark _Ol	B20_Cpx brught_dark _Ol	B20_Cpx brught_dark _Ol	B20_Cpx brught_dark _Ol
SiO	51.746	51.46	51.862	51.643	51.078	51.477	51.419	51.28	51.35					
TiO	1.056	1.106	1.129	1.103	1.11	1.135	1.224	1.163	1.218					
Al2	1.996	2.161	2.229	2.153	2.336	1.955	2.21	2.155	2.348					
O3	5.437	5.56	5.584	5.485	5.523	5.438	5.589	5.64	5.657					
Fe	0.168	0.159	0.084	0.089	0.159	0.134	0.137	0.117	0.117					
Mn	16.142	16.099	15.839	16.016	16.023	15.962	15.766	15.953	15.74					
O	21.54	21.232	21.431	21.249	21.274	21.215	20.612	21.071	21.095					
Ca	0.526	0.496	0.493	0.519	0.595	0.489	0.615	0.567	0.56					
Na	0.004	0.016	0.001	0	0	0.005	0.008	0.011	0.027					
2O	0.529	0.491	0.49	0.547	0.518	0.457	0.393	0.323	0.487					
K2														
O														
Cr2														
O3														
Total	98.615	98.289	98.652	98.257	98.098	97.81	97.58	97.957	98.112					
Wo	43.7843927	43.4463209	43.9880577	43.5840313	43.4553236	43.6936328	42.9200754	43.2702425	43.56264312					
En	45.6543193	45.8365476	45.2346263	45.7081726	45.5395321	45.7418077	45.6785977	45.582427	45.22622289					
Fs	8.62643414	8.88045542	8.94614856	8.78140241	8.80576768	8.74203757	9.0839095	9.0402748	9.118412141					

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N°										
Mg	84.107748	83.7702086	83.488334	83.8842524	83.7966343	83.954808	83.4121738	83.4496015	83.22113648	



Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: Porphyrogabbro. B20. Olivines.

Data set	B20_OI	B20_OI	B20_OI	B20_OI	B20_OI	B20_OI	B20_OI-c_PI	B20_OI-c_PI	B20_OI-c_PI	B20_OI-c_PI	B20_OI-c_PI	B20_OI-c_PI	B20_OI-c_PI	B20_OI-c_PI
	-c_PI	-c_PI	-c_PI	-c_PI	-c_PI	-c_PI								
SiO <sub>2</sub>	39.274	39.158	39.094	40.687	39.212	38.877	39.106	39.119	39.328	39.102	39.976	38.91	39.269	39.134
TiO <sub>2</sub>	0.012	0.006	0	0.02	0.02	0.017	0.019	0.026	0	0.007	0.03	0.009	0.006	0.024
Al <sub>2</sub> O <sub>3</sub>	0.005	0	0	0.047	0.036	0.035	0.016	0.012	0.024	0	0.03	0	0.055	0
Cr <sub>2</sub> O <sub>3</sub>	0.06	0.007	0	0	0.026	0.005	0.053	0	0.031	0	0.058	0.084	0	0.005
FeO	17.3	17.384	17.425	17.402	17.498	17.55	17.698	17.397	17.333	17.413	17.747	17.42	17.63	17.739
MnO	0.207	0.273	0.271	0.284	0.292	0.218	0.191	0.259	0.256	0.251	0.237	0.312	0.246	0.272
MgO	42.478	42.736	42.59	39.829	42.764	42.847	42.617	42.844	42.727	42.976	42.799	42.555	42.592	42.805
CaO	0.096	0.089	0.077	0.198	0.144	0.166	0.114	0.132	0.148	0.153	0.131	0.204	0.157	0.174
Na <sub>2</sub> O	0.007	0.003	0	0.018	0	0	0	0	0.001	0.025	0	0.012	0	0
NiO														
Total	99.439	99.656	99.457	98.485	99.992	99.715	99.814	99.789	99.848	99.927	101.008	99.506	99.955	100.153
Fo	81.219	81.180	81.094	80.054	81.075	81.124	80.9382112	81.2200427	81.23634069	81.26011164	80.92174346	81.05013122	80.93976407	80.90035711
Fa	18.556	18.524	18.612	19.621	18.610	18.640	18.8556873	18.500992	18.48711476	18.47023717	18.82365728	18.61224354	18.79462484	18.80756275
	1194	862	4421	4634	0211	5429								
Data set	B20_OI	B20_OI	B20_OI	B20_OI	B20_OI	B20_OI	B20_OI-c_PI	B20_OI-c_PI	B20_OI-c_PI	B20_A_OI_c-r	B20_A_OI_c-r	B20_A_OI_c-r	B20_A_OI_c-r	B20_A_OI_c-r
	-c_PI	-c_PI	-c_PI	-c_PI	-c_PI	-c_PI								
SiO <sub>2</sub>	39.313	40.291	39.041	38.066	39.325	38.784	39.153	39.004	39.086	39.54129763	39.96671363	39.36319285	39.83033626	39.79675078
TiO <sub>2</sub>	0	0	0.025	0	0	0.002	0.029	0.016	0.007	0.007235835	0.029977032	0.005168454	0.032044414	0.035145486
Al <sub>2</sub> O <sub>3</sub>	0.037	0.034	0.035	0.027	0.024	0.02	0.006	0.009	0.039	0	0	0	0.021779425	0
Cr <sub>2</sub> O <sub>3</sub>	0	0.058	0.082	0.05	0	0	0.041	0.014	0	0	0.113419913	0.038095238	0.073593074	0
FeO	17.48	17.262	17.481	17.318	17.742	17.628	17.153	17.754	17.464	17.7071313	17.55666007	17.65461785	17.53242309	17.55666007

Mn	0.256	0.227	0.301	0.259	0.254	0.225	0.233	0.235	0.248	0.260661226	0.240005053	0.250824953	0.210496235	0.307875335
Mg	42.674	41.758	42.359	44.018	42.349	42.84	42.37	42.803	42.371	41.71376054	42.01813128	42.55901305	42.12790433	41.48423506
CaO	0.159	0.147	0.17	0.145	0.134	0.168	0.132	0.1	0.13	0.10123974	0.150847213	0.119462893	0.094152958	0.065805831
Na2O	0.018	0	0	0.013	0.034	0.007	0	0.003	0.01	0	0	0	0	0
NiO										0.009441066	0.013637095	0	0.010490073	0.005245036
Total	99.937	99.777	99.494	99.896	99.862	99.674	99.117	99.938	99.355	99.34076734	100.0893913	99.99037529	99.93321986	99.2517176
Fo	81.090	80.972	80.935	81.696	80.747	81.049								
	146	4723	8488	0456	4726	1821	81.285567	80.9189171	81.00157085	80.53594533	80.79876689	80.9023456	80.88633941	80.53925024
Fa	18.633	18.777	18.737	18.030	18.977	18.708								
	4643	4351	3836	8381	3602	9608	18.4604597	18.8286645	18.72905644	19.17812183	18.93901326	18.8267491	18.88403153	19.12114279
Data set	B20_A_OI_c-r	B20_A_OI_c-r	B20_A_OI_c-r	B20_A_OI_c-r	B20_A_OI_c-r	B20_A_OI_c-r	B20_A_OI_c-r	B20_A_OI_c-r	B20_A_OI_c-r	B20_A_OI_c-r	B20_A_OI1_c-r	B20_A_OI1_c-r	B20_A_OI1_c-r	B20_A_OI1_c-r
SiO2	39.515	39.998	40.082	39.549	39.733	39.870								
	8541	2636	7362	4396	6508	0282	40.0074233	39.6491782	39.57997182	40.00233459	40.87250368	39.74688144	39.46191379	39.709225
TiO2	0.0289	0.0237		0.0082	0.0237	0.0382								
	4334	7489	0	6953	7489	4656	0.02790965	0.00930322	0.017572743	0.012404289	0.028943342	0.020673815	0.034111795	0
Al2O3		0.0114	0.0051	0.0020	0.0114									
	0	0827	8558	7423	0827	0	0.01244539	0	0.003111346	0	0	0	0.054967121	0
Cr2O3				0.0354		0.0138								
	0	0	0	9784	0	5281	0	0.01818182	0.03030303	0	0.05974026	0	0	0.001731602
FeO	17.466	17.250	17.266	17.525	17.626	17.656								
	7813	6682	8262	354	3414	6376	17.6445191	17.47688	17.59503528	17.24258924	18.12622902	18.20903869	17.74348676	17.90405673
MnO	0.2636	0.2636	0.2577	0.2586	0.2527	0.2626								
	1211	1211	1034	9397	9221	2848	0.26066123	0.21738163	0.23410329	0.220332508	0.29508818	0.254759462	0.285251908	0.308858962
MgO	42.530	41.840	42.107	42.657	42.109	42.250								
	0729	4985	9456	8088	9415	6506	41.9422881	41.8225357	41.93530252	41.68182656	41.13695304	41.52914222	41.86644488	42.06503431
CaO	0.1285	0.1103	0.1073	0.1174	0.1488	0.1346								
	7447	5132	1412	381	2242	4885	0.13262406	0.13768605	0.146797623	0.171095161	0.083016587	0.121487688	0.139710841	0.160971187
Na2O					0.0027									
	0	0	0	0	7073	0	0	0	0.019395139	0	0	0	0.00461789	0
NiO		0.0010	0.0052	0.0020		0.0094								
	0	4901	4504	9801	0	4107	0.00524504	0	0.008392058	0	0.017833124	0.009441066	0.01153908	0
Total	99.933	99.499	99.832	100.15	99.909	100.23								
	8382	626	9631	6674	5022	6134	100.033116	99.3311466	99.56998485	99.33058235	100.6203072	99.89142439	99.60204406	100.1498778
Fo	81.043	80.980	81.069	81.042	80.760	80.777								
	0593	216	2464	7747	7442	6301	80.6759151	80.8161398	80.73987266	80.96740363	79.91937505	80.03489163	80.53987225	80.45436973

Fa	18.671	18.729	18.648	18.677	18.963	18.937											
	5353	8997	8497	9843	7971	0863	19.0392158	18.9451957	19.00403713	18.78942119	19.75490155	19.68615352	19.14834606	19.20999639			
Dat	B20_A_	B20_A_	B20_A_	B20_A_	B20_A_	B20_A_	B20_A_OI1_	B20_A_OI1_	B20_A_OI1_	B20_A_OI1_	B20_A_OI1_	B20_A_OI1_	B20_A_OI1_	B20_A_OI1_			
set	OI1_c-r	OI1_c-r	OI1_c-r	OI1_c-r	OI1_c-r	OI1_c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	B20_A_OI1_c-r		
SiO	39.965	41.082	39.821	39.417	39.530	38.545											
2	6959	1585	1766	1332	1025	9463	40.3188522	39.6807282	39.52908473	39.50364119	39.8486556	39.97383782	39.75909434	39.61152181			
TiO	0.0258	0.0082	0.0217	0.0175	0.0382	0.0186											
2	4227	6953	0751	7274	4656	0643	0.01033691	0.04341501	0.008269526	0.024808578	0.035145486	0.029977032	0.012404289	0.007235835			
Al2		0.0466		0.0093		0.0466											
O3	0	702	0	3404	0	702	0	0	0	0	0.010371155	0.046670197	0	0.015556732			
Cr2	0.0155		0.0640		0.0043	0.0121											
O3	8442	0	6926	0	29	2121	0.03030303	0	0.025974026	0.032034632	0.027705628	0.001731602	0.032034632	0.041558442			
	17.852	17.434	17.716	17.688	17.892	17.781											
FeO	5532	4653	2202	9536	9481	862	17.4839491	17.9010271	17.69501281	17.55464032	17.73237815	17.66774621	17.97979729	17.47082077			
Mn	0.2931	0.2832	0.3137	0.2370	0.2970	0.2813											
o	2093	8465	771	5417	5544	174	0.25279221	0.25475946	0.237054172	0.271481126	0.246890444	0.242955935	0.302957199	0.247874072			
Mg	42.341	40.268	41.503	41.562	41.681	47.220											
O	4628	748	1959	0741	8266	3761	41.0391552	41.5441113	41.45729077	41.98320349	41.61197098	42.00316223	41.83950058	42.5899491			
	0.1255	0.1589	0.1305	0.1255	0.1366	0.1275											
CaO	3728	4639	9926	3728	7365	6207	0.10123974	0.10326453	0.123512483	0.103264535	0.152872008	0.125537278	0.127562073	0.145785226			
Na2				0.0018		0.0092											
O	0	0	0	4716	0	3578	0.01200651	0	0.014777249	0	0	0.005541468	0	0			
	0.0031		0.0031	0.0115	0.0062												
NiO	4702	0	4702	3908	9404	0	0.00839206	0.01153908	0.010490073	0	0	0.001049007	0.005245036	0			
Tot	100.62	99.282	99.573	99.071	99.587	104.04											
al	2944	5425	8928	0453	4758	3697	99.2570271	99.5388447	99.10146584	99.47307387	99.66598945	100.0982088	100.0585954	100.130302			
	80.615	80.200	80.401	80.515	80.330	82.329											
Fo	193	5105	5099	5856	0811	3156	80.4832433	80.307996	80.470644	80.75983102	80.48774891	80.6939104	80.30931907	81.0749657			
	19.067	19.478	19.253	19.223	19.344	17.392											
Fa	7225	9296	1229	4943	6472	0088	19.2350819	19.4121981	19.26792243	18.94345591	19.24092412	19.04089573	19.36028254	18.65693977			
Dat	B20_A_	B20_A_	B20_A_	B20_A_	B20_A_	B20_A_	B20_A_OI3_	B20_A_OI3_	B20_A_OI3_	B20_A_OI3_	B20_A_OI3_	B20_A_OI3_	B20_A_OI3_	B20_A_OI3_			
set	OI1_c-r	OI1_c-r	OI1_c-r	OI1_c-r	OI1_c-r	OI3_c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	c-r	B20_A_OI3_c-r		
SiO	40.771	39.784	39.506	40.486	39.307	39.431											
2	7473	5379	6944	7796	2171	3815	39.4710735	39.5087299	39.59829116	39.54333312	39.5087299	40.37279254	39.7458637	39.09043809			
TiO	0.0279	0.0382	0.0423	0.0341	0.0268	0.0217											
2	0965	4656	8132	118	7596	0751	0.03928025	0.05375192	0.041347631	0.019640125	0.015505362	0.033078105	0.005168454	0.024808578			
Al2	0.0248		0.0062														
O3	9077	0	2269	0	0	0	0.00829692	0.03422481		0	0.039410389	0.001037115	0.024890772	0.016593848	0.017630963		

Cr2	0.0580		0.0779			0.0121												
O3	0866	0	2208	0	0	2121	0.04155844	0.01038961	0.073593074	0.01991342	0.015584416	0	0.007792208	0				
	17.298	17.536	17.554	17.469	17.735	18.084												
FeO	1323	4626	6403	8109	4078	8242	17.7687336	18.2363053	17.92526409	18.01009351	17.73843739	17.84548404	17.7808521	17.64956848				
Mn	0.2124	0.3226	0.2449	0.2704	0.2242	0.2547												
o	6349	2974	2319	975	6702	5946	0.26557936	0.24590682	0.231152408	0.275415635	0.302957199	0.344269544	0.187872808	0.217381626				
Mg	41.561	41.910	41.985	40.796	42.451	41.943												
O	0762	3541	1994	6566	2359	286	41.6598719	41.7037812	40.39349008	41.7397069	43.54796848	41.28265182	42.33447726	41.43633409				
	0.1478	0.1255	0.1802	0.1862	0.1791	0.0830												
CaO	1002	3728	0674	8112	9434	1659	0.15084721	0.12857447	0.148822418	0.125537278	0.066818228	0.118450496	0.07390501	0.136673649				
Na2																		
O	0	0	0	0	0	0	0.0212423	0	0	0.002770734	0	0	0	0				
	0.0062				0.0125	0.0041												
NiO	9404	0	0	0	8809	9603	0.01153908	0.00209801	0.003147022	0.005245036	0	0.009441066	0.003147022	0.009441066				
Tot	100.10	99.717	99.598	99.244	99.936	99.835												
al	8332	7681	1901	1375	7861	2925	99.4380226	99.923762	98.41510788	99.78106614	101.1970381	100.0310584	100.1556724	98.58227655				
Fo	80.880	80.703	80.784	80.386	80.816	80.299												
	6281	5054	0174	6123	5979	9604	80.4575913	80.0860697	79.85974482	80.26944929	81.13871701	80.17727761	80.76644168	80.5200338				
	18.884	18.943	18.948	19.310	18.940	19.422												
Fa	4523	5125	2284	5572	8233	9244	19.2509879	19.645625	19.8806029	19.42961989	18.5405684	19.44283094	19.02991139	19.2399599				
Dat																		
a																		
set	B20_A_	B20_A_	B20_A_	B20_A_	B20_A_	B20_A_	B20_A_OI3_	B20_A_OI3_	B20_A_OI3_	B20_A_OI3_	B20_B_OI_c-	B20_B_OI_c-	B20_B_OI_c-	B20_B_OI_c-				
	OI3_c-r	OI3_c-r	OI3_c-r	OI3_c-r	OI3_c-r	OI3_c-r	c-r	c-r	c-r	c-r	r	r	r	r				
SiO	39.566	39.189	39.102	39.371	39.425	39.051												
2	7412	159	651	3348	2751	7639	39.1392897	39.1667687	39.34894447	39.09654454	39.29500416	39.24818805	39.50669442	38.94591878				
TiO	0.0299	0.0248	0.0423	0.0506	0.0144	0.0444												
2	7703	0858	8132	5085	7167	487	0.03721287	0.03824656	0.059954065	0.036179177	0.002067382	0.027909651	0.019640125	0.011370598				
Al2	0.0124	0.0259	0.0134	0.0165		0.0134												
O3	4539	2789	825	9385	0	825	0	0	0.035261927	0.012445386	0.051855775	0	0	0				
Cr2		0.0060	0.0017		0.0077													
O3	0	6061	316	0	9221	0	0	0	0.018181818	0.03030303	0.012121212	0.03982684	0.001731602	0.05974026				
	17.914	17.678	17.778	18.006	17.651	17.860												
FeO	1555	8548	8324	054	5882	6322	17.6768351	17.9525307	17.8394248	17.59705503	17.87679013	17.72429916	17.67885482	18.15248575				
Mn	0.2282	0.2547	0.2734	0.3078	0.3000	0.2813												
o	0153	5946	4838	7533	0632	174	0.23410329	0.24098868	0.230168781	0.311809844	0.224267017	0.307875335	0.225250644	0.275415635				
Mg	41.432	42.053	41.537	41.860	41.977	41.880												
O	3423	0591	1257	4573	2159	416	42.243665	41.9043665	42.28258455	43.63678486	40.60205888	40.65993922	40.37951896	40.62600937				
	0.1144	0.1164	0.0809	0.1417	0.1488	0.0941												
CaO	0091	257	9179	3564	2242	5296	0.09010337	0.14679762	0.092128163	0.108326522	0.118450496	0.108326522	0.071880215	0.139710841				
Na2		0.0101																
O	0	5936	0	0	0	0	0.00184716	0	0.010159359	0	0	0.005541468	0	0				
	0.0083	0.0052	0.0083	0.0010	0.0062													
NiO	9206	4504	9206	4901	9404	0	0.01258809	0.01153908	0.009441066	0.017833124	0	0	0	0				

Tot al	99.306	99.364	98.839	99.755	99.531	99.226										
	6559	4596	0367	7507	4658	2136	99.4356445	99.4612378	99.92624899	100.8472815	98.18261506	98.12190624	97.88357079	98.21065123		
Fo	80.277	80.692	80.395	80.290	80.648	80.446										
	371	3617	3786	2407	0867	8857	80.7825955	80.4116604	80.65975083	81.28218972	79.99144159	80.0741753	80.07839578	79.71249157		
Fa	19.471	19.029	19.303	19.374	19.024	19.246										
	4118	8958	9119	2451	4309	0905	18.9630492	19.3255951	19.0907788	18.38781353	19.75752122	19.58133373	19.66780081	19.98047335		
Dat a set	B20_B_Ol_c-r	B20_B_Ol_c-r	B20_B_Ol_c-r	B20_B_Ol_c-r	B20_B_Ol_c-r	B20_B_Ol_c-r	B20_B_Ol_c-r	B20_B_Ol_c-r	B20_B_Ol_c-r	B20_B_Ol_c-r	B20_B_Ol_c-r	B20_B_Ol_c-r	B20_B_Ol_c-r	B20_B_Ol_c-r	B20_B_Ol_c-r	B20_B_Ol_c-r
SiO 2	39.308	39.335	39.415	39.206	39.160	39.202										
	2348	7138	0977	4606	6623	3897	38.9652559	39.0843316	39.6308589	39.38456542	39.27566707	38.99985909	39.76112982	39.47514443		
TiO 2	0.0299	0.0051	0.0124	0.0351	0.0051	0.0258										
	7703	6845	0429	4549	6845	4227	0.0113706	0.02584227	0.037212868	0.007235835	0.033078105	0.011370598	0.017572743	0.020673815		
Al2 O3	0.0300				0.0093											
	7635	0	0	0	3404	0	0.01555673		0.026965003	0.004148462	0.013482501		0.017630963			
Cr2 O3	0.0320		0.0303		0.0199											
	3463	0	0303	0	1342	0	0.02164502	0.004329		0.01991342		0.035497835	0.025974026	0.051948052		
FeO	18.087	17.872	17.831	17.992	18.009	18.052										
	8538	7506	3458	9256	0836	5082	17.8343754	18.0969427	17.60311427	17.87073089	18.03332061	17.90001724	17.94445169	18.2241868		
Mn o	0.2803	0.2803	0.3019	0.2252	0.2301	0.2124										
	3377	3377	7357	5064	6878	6349	0.23705417	0.20754535	0.249841326	0.247874072	0.265579362	0.353122189	0.209512608	0.230168781		
Mg O	40.563	40.218	40.436	40.412	40.766	40.699										
	1393	8511	4014	4509	7185	8567	40.5132425	40.5351971	40.51723425	40.66093716	40.6349908	40.96131615	40.59108158	40.87249977		
CaO	0.0404	0.1174	0.1174	0.0809	0.0830	0.1397										
	959	381	381	9179	1659	1084	0.11946289	0.10832652	0.064793434	0.107314125	0.10528933	0.12857447	0.10123974	0.177169545		
Na2 O		0	0	3578	8862	0										
		0.0020		0.0062	0.0020	0.0020	0.00554147	0.00461789	0.017547983		0	0	0	0.017547983		
NiO	0	9801	0	9404	9801	9801	0.00419603		0.007343051	0.014686102		0.015735109	0.008392058	0.005245036		
Tot al	98.372	97.832	98.144	97.968	98.293	98.334										
	1456	3539	9638	7549	5523	8692	97.7066025	98.0836128	98.11501599	98.35777001	98.35207374	98.41897518	98.65935427	99.09221518		
Fo	79.739	79.792	79.896	79.812	79.934	79.885										
	8545	3652	4628	7101	3808	6103	79.9824599	79.7856185	80.17787553	79.99886013	79.82950845	79.99709835	79.94052658	79.78752371		
Fa	19.947	19.891	19.764	19.934	19.809	19.877										
	0358	6369	5361	5347	1994	4503	19.7516375	19.9822775	19.54122157	19.72405379	19.87405291	19.61106727	19.82503772	19.95719006		
Dat a set	B20_B_Ol_c-r	B20_B_Ol_c-r	B20_Fe ld in Ol	B20_Fe ld in Ol	B20_Fe ld in Ol	B20_Fe ld in Ol	B20_Cpx brught_dark_Ol	B20_Cpx brught_dark_Ol	B20_Cpx brught_dark_Ol	B20_Cpx brught_dark_Ol	B20_Cpx brught_dark_Ol	B20_Cpx brught_dark_Ol	B20_Cpx brught_dark_Ol	B20_Cpx brught_dark_Ol	B20_Cpx brught_dark_Ol	B20_Cpx brught_dark_Ol
SiO 2	39.425	38.982														
	2751	5575	42.145	39.017	39.045	38.87	38.658	39.006	38.668	39.02	38.75	38.962	39.223	38.872	39.115	39.196

TiO	0.0072	0.0206															
2	3584	7382	0.13	0.038	0.017	0.029	0	0.023	0.016	0.028	0.006	0.02	0.023	0.015	0.003	0.007	
Al2	0.0435																
O3	5885	0	2.214	0	0.029	0	0.036	0	0.028	0.022	0.029	0.01	0.009	0.017	0	0.015	
Cr2	0.0458	0.0277															
O3	8745	0563	0.007	0.031	0.055	0.017	0	0	0	0.007	0.036	0.036	0.024	0.058	0	0.031	
	17.665	17.834															
FeO	7265	3754	16.213	17.717	17.806	17.609	18.037	17.665	17.782	17.612	17.501	17.421	17.651	17.544	17.638	17.75	
Mn	0.2596	0.2222															
o	776	9976	0.223	0.318	0.226	0.29	0.225	0.256	0.272	0.243	0.254	0.253	0.223	0.236	0.258	0.256	
Mg	40.791	40.963															
O	6669	312	37.563	42.737	42.574	42.917	42.466	42.81	42.877	42.535	42.381	42.402	42.911	42.627	42.893	42.522	
	0.0809	0.0961															
CaO	9179	7775	0.248	0.081	0.096	0.079	0.115	0.083	0.115	0.095	0.161	0.091	0.123	0.077	0.079	0.121	
Na2		0.0193															
O	0	9514	0.446	0.018	0.03	0	0	0.019	0.015	0.013	0	0	0.021	0	0.033	0.003	
	0.0020	0.0125															
NiO	9801	8809	0.736	0.012	0.022	0.021	0	0	0.001	0.009	0	0	0.009	0.004	0	0	
Tot	98.322	98.179															
al	118	0851	99.925	99.969	99.9	99.832	99.537	99.862	99.774	99.584	99.118	99.195	100.217	99.45	100.019	99.901	
	80.220	80.171	84.229	84.668	84.665	84.833											
Fo	6601	909	1833	7304	8184	2937	84.469129	84.8019376	84.71984085	84.77226889	84.79188708	84.85704623	84.88168296	84.85722651	84.84370306	84.65395287	
	19.489	19.580	15.414	14.882	15.014	14.758											
Fa	1873	8939	6894	5817	0929	4498	15.2121304	14.8369032	14.89739802	14.88281632	14.84619091	14.7823583	14.80415885	14.8081825	14.79284158	14.98307632	

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: Porphyrogabbro. B20. Feldspars.

Data set	B20_Pl smecolato_t ransect	B20_Pl smecolato_t ransect	B20_Pl smecolato_t ransect	B20_Pl smecolato_t ransect	B20_Pl smecolato_t ransect	B20_Pl smecolato_t ransect	B20_Pl smecolato_t ransect	B20_Pl smecolato_t ransect	B20_Pl smecolato_t ransect	B20_Pl smecolato_t ransect	B20_Pl smecolato_t ransect	B20_Pl smecolato_t ransect	B20_Pl smecolato_t ransect	B20_Pl smecolato_t ransect	
SiO 2	56.016	57.033	58.676	58.341	50.956	50.719	51.062	52.125	51.886	50.87	51.132	51.423	51.65	51.601	
TiO 2	0.154	0.155	0.136	0.177	0.109	0.12	0.151	0.122	0.111	0.115	0.134	0.121	0.114	0.137	
Al2 O3	26.507	25.273	24.634	25.03	30.379	30.122	30.035	30.125	29.796	29.835	29.677	29.626	29.886	29.752	
Cr2 O3	0.026	0.064	0.026	0	0	0.049	0	0	0.018	0	0.09	0	0	0	
Fe O	0.442	0.479	0.36	0.402	0.336	0.359	0.396	0.398	0.423	0.431	0.426	0.388	0.34	0.367	
Mn o	0.035	0	0.032	0.052	0.022	0.024	0.039	0	0.024	0.015	0	0.014	0.002	0	
Mg O	0.053	0.072	0.022	0.065	0.047	0.044	0.038	0.049	0.034	0.003	0.036	0.009	0.016	0.02	
Ca O	8.758	8.207	6.756	7.199	13.46	13.056	13.094	12.998	12.847	13.088	12.86	12.588	13.302	12.973	
Na 2O	6.023	6.012	6.607	6.359	3.673	3.669	3.79	3.926	4.136	3.696	3.923	3.991	3.825	3.858	
K2 O	1.01	1.149	1.633	1.512	0.366	0.338	0.376	0.377	0.397	0.346	0.365	0.41	0.366	0.387	
SrO Ba O															
Tot al	99.024	98.444	98.882	99.137	99.348	98.5	98.981	100.12	99.672	98.399	98.643	98.57	99.501	99.095	
Ab %	52.2501784	53.1884322	57.8807516	56.1152528	32.3560394	33.0357563	33.6197069	34.5697237	35.97624394	33.12987185	34.81045188	35.58022301	33.50410605	34.19750989	
An %	41.9847011	40.1230428	32.7062466	35.1055259	65.5225359	64.9617768	64.1856967	63.2460427	61.75160134	64.82944304	63.05848496	62.01473369	64.38648986	63.54536657	
Or %	5.76512053	6.68852493	9.41300184	8.77922125	2.12142468	2.0024669	2.19459639	2.18423352	2.272154715	2.040685112	2.131063158	2.405043294	2.109404093	2.257123536	
Data set	B20_Pl smecolato_t ransect	B20_Pl smecolato_t ransect	B20_Pl smecolato_t ransect	B20_Feld in Ol	B20_Feld in Ol	B20_Feld in Ol	B20_Feld in Ol	B20_Feld in Ol	B20_Feld in Ol	B20_Feld in Cpx	B20_Feld in Cpx	B20_A_An_c -r	B20_A_An_c -r	B20_A_An_c -r	B20_A_An_c -r
SiO 2	51.354	51.603	51.508	65.589	66.425	66.17	66.954	66.248	67.057	67.338	56.30655543	56.80626657	57.51766796	56.80728431	
TiO 2	0.129	0.111	0.107	1.036	1.091	1.1	1.087	1.102	1.05	1.065	0.16125576	0.163323142	0.158154688	0.175727431	

Al2														
O3	29.769	29.709	29.911	16.915	16.864	16.812	17.122	16.65	16.486	16.207	27.36221812	26.80632422	26.87995942	26.63416304
Cr2														
O3	0.026	0	0.046	0.005	0.031	0	0	0	0.038	0.069	0	0	0	0
Fe														
O	0.389	0.328	0.363	2.481	2.475	2.486	2.606	2.382	1.865	1.897	0.525134497	0.554420844	0.594815806	0.562499836
Mn														
o	0.008	0	0.005	0	0.025	0.059	0.04	0.072	0.029	0.047	0.011803527	0	0.018688918	0
Mg														
O	0.044	0.04	0.016	0.802	0.823	0.789	0.774	0.817	0.667	0.673	0.042911285	0.056882401	0.042911285	0.042911285
Ca														
O	12.764	12.984	12.638	1.302	1.345	1.183	1.245	1.281	1.232	1.161	9.425419803	8.753187929	8.932382269	8.609427498
Na														
2O	3.791	4.008	4.091	4.195	3.377	3.589	2.509	3.602	3.396	3.522	5.811153168	5.341051934	5.755738484	5.574717184
K2														
O	0.388	0.388	0.384	6.259	6.214	6.267	5.75	6.261	6.435	6.384	0.869627033	1.010194009	1.030125147	1.064742387
SrO														
Ba														
O														
Tot	98.662	99.171	99.069	98.584	98.67	98.455	98.087	98.415	98.255	98.363	100.5160786	99.49165104	100.930444	99.47147297
Ab														
%	34.1541142	35.0402954	36.115877	46.4419747	41.1381277	42.8983899	35.9436669	42.7309447	40.86200408	42.10849648	50.13141056	49.25918292	50.62403139	50.52826554
An														
%	63.5458581	62.7277577	61.6535727	7.96527744	9.05413481	7.81382156	9.856036	8.39767623	8.191697374	7.670502462	44.93238955	44.61057966	43.414443816	43.12181527
Or														
%	2.30002773	2.23194699	2.23055027	45.5927479	49.8077375	49.2877885	54.2002971	48.871379	50.94629855	50.22100106	4.936199887	6.130237424	5.961530448	6.349919188
Dat														
a	B20_A_An_c	B20_A_An_c	B20_A_Fel_c	B20_A_Fel_c	B20_A_Fel_c	B20_A_Fel_c	B20_A_Fel_c	B20_A_Fel_c	B20_A_Fel_c	B20_A_Fel_c	B20_A_Fel_c	B20_A_Fel_c	B20_A_Fel_c	B20_A_Fel_c
set	-r	-r	-r	-r	-r	-r	-r	-r	-r	-r	-r	-r	-r	-r
SiO														
2	58.0601243	58.9547191	65.5160994	65.2525043	65.0978076	64.5166771	65.0428495	68.3321905	68.57950169	69.2105015	68.76676615	68.41869852	68.79220969	68.45431947
TiO														
2	0.14471671	0.13851456	0.16125576	0.14885147	0.15918838	0.15608731	0.13334611	1.04816244	1.045061368	1.0078485	1.032657078	1.018185408	1.005781118	1.041960295
Al2														
O3	26.2338365	25.5638599	20.2818306	20.060925	19.9758816	19.8752814	20.1335231	16.5378437	16.48598792	16.81682776	16.36257117	16.67681717	16.51295292	16.50880446
Cr2														
O3	0.02164502	0.07012987	0.01471861	0.03809524	0	0.02164502	0	0.04502165	0	0.012987013	0	0.027705628	0.012987013	0.027705628
Fe														
O	0.55442084	0.52614437	0.16763909	0.25145863	0.2140933	0.26155737	0.26357712	1.70870686	1.710726612	1.64003543	1.736983337	1.813733764	1.737993211	1.741022833
Mn														
o	0	0.02655794	0.00196725	0	0	0.00688539	0.0216398	0	0.035410582	0.033443327	0.040328718	0	0.058034009	0.041312345





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SrO				
Ba				
O				
Total				
al	99.2270374	99.3535605	98.0488457	98.674366
Ab				
%	41.4025073	40.2733024	38.7714444	39.5248545
An				
%	6.49087436	6.81907241	7.6842099	6.90267958
Or				
%	52.1066184	52.9076251	53.5443457	53.5724659

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: Porphyrogabbro. B20. Biotite (phlogopite).

Data set	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r
SiO2	39.28991	39.14844	39.62271	38.96729	39.06601	39.26040	39.24106	38.92251	39.07211	40.44505	38.36377	39.08738	38.98764	38.94998
	55	94	7	14	23	09	39	07	874	22	057	487	619	975
TiO2	8.491769	8.452489	8.415276	8.566195	8.549656	8.450422	8.451455	8.405973	8.401838	8.564128	8.518645	8.475230	8.521746	8.440085
	67	42	55	4	35	04	73	34	572	023	629	617	701	131
Al2O3	13.37049	13.31137	13.39123	13.27092	13.24085	13.25226	13.18692	13.22322	13.14336	13.89838	13.13403	13.29582	13.22633	13.27507
	3	74	53	99	35	18	35	26	468	476	064	065	392	834
Cr2O3	0.097835	0.136796	0.180952	0.103896	0.056277	0.139393	0.145454	0.136796	0.118614	0.228571	0.120346	0.195670	0.259740	0.155844
	5	54	38	1	06	94	55	54	719	429	32	996	26	156
FeO	7.463978	7.215549	7.425603	7.296339	7.278162	7.205451	7.339764	7.315527	7.432672	7.435702	7.411465	7.364001	7.334715	7.361981
	98	97	77	89	16	23	47	5	884	506	529	45	103	702
Mno	0.004918	0.006885	0.029508	0.071804	0.073772	0.044263	0.012787	0.073772	0.060984	0.052132	0.073772	0.078690	0.096395	0.026557
	14	39	82	79	05	23	15	05	891	245	045	181	472	936
MgO	18.53268	17.97284	17.89899	18.02972	17.68044	18.02573	17.60460	17.71138	17.94490	17.12759	18.53567	18.12353	17.91296	17.74331
	53	28	54	52	73	34	41	33	053	024	914	122	655	728
CaO	0.028347		0.015185	0.057706	0.026322	0.026322	0.024297	0.093140	0.070867	0.050619	0.066818	0.002024	0.048595	0.004049
	13	0	96	65	33	33	54	56	818	87	228	795	075	59
Na2O	0.902335	0.780423	0.767493	0.780423	0.609561	0.807207	0.752716	0.615102	0.760104	0.590166	0.611408	0.734244	0.711155	0.678829
	77	46	37	46	52	23	12	99	745	382	677	559	108	876
K2O	9.241754	9.353997	9.389664	9.320429	9.336164	9.245950	9.251195	8.942787	9.291057	9.189303	9.184058	9.394909	9.279518	9.318331
	12	9	14	66	77	15	18	04	46	754	718	181	38	65
BaO	0	0	0	0	0	0	0	0	0	0	0	0	0	0
F	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Cl	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Total	97.42403	96.37881	97.13663	96.46474	95.91722	96.45740	96.01026	95.44021	96.29652	97.58165	96.01999	96.75150	96.37881	95.95406
	31	22	27	24	93	63	22	66	504	14	55	852	275	541
Data set	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt _c-r	B20_A_Bt1 _c-r
SiO2	39.29703	38.78511	39.20238	38.86857	38.82684	38.94490	39.14844	40.71271	39.47717	37.33178	38.77188	39.29703	37.90069	39.51483
	96	56	97	04	3	1	94	82	991	056	496	964	813	635
TiO2	8.475230	8.512443	8.467994	8.411141	8.522780	8.513477	8.399771	8.474196	8.369794	8.403905	8.538285	7.155207	7.196555	7.262711
	62	48	78	79	39	18	19	93	158	954	754	505	135	345
Al2O3	13.33523	13.29789	13.13299	13.06558	13.34249	13.31345	13.26055	13.79259	13.48664	12.52939	13.21077	13.05832	12.24211	12.95149
	1	49	35	1	09	16	87	9	99	23	718	121	131	831
Cr2O3	0.174891	0.130735	0.147186	0.228571	0.157575	0.188744	0.170562	0.180952	0.203463	0.226839	0.276190	0.066666	0.010389	0.106493
	77	93	15	43	76	59	77	38	203	827	476	667	61	506
FeO	7.241806	7.210500	7.443781	7.215549	7.153947	7.303409	7.306438	7.379149	7.484176	7.234737	7.420554	6.569230	6.656079	6.586398
	69	6	5	97	65	01	63	56	46	573	396	586	752	444



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Cl	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Total	95.04262 03	95.66373 4	94.57638 51	95.16621 24	95.15295 89	95.24541 78	95.24904 16	94.86565 35	96.68972 332	94.95339 903	96.11720 816	94.94665 028	95.38630 184	95.80173 213	94.81911 121	86.23921 438	94.99628 468

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: Porphyrogabbro. B20. Opaque minerals.

Data set	B20_Opq in Ol_c-r	B20_Opq in Ol_c-r	B20_Opq in Ol_c-r	B20_Opq in Ol_c-r	B20_Opq in Ol_c-r	B20_Opq in Ol_c-r	B20_Opq in Ol_c-r	B20_Opq in Ol_c-r	B20_Opq in Ol_c-r	B20_Opq in Ol_c-r	B20_Opq in Ol_c-r	B20_A_Op q_c-r	B20_A_Op q_c-r	B20_A_Op q_c-r	B20_A_Op q_c-r
SiO2	0.024	0.031	0.034	0.017	0.028	0.054	0.068	0.033	0.049	0.041	0.0488515	0.0895612	0.0824370	0.0396919	
TiO2	2.595	2.547	2.57	2.62	2.565	2.614	2.543	2.515	2.581	2.519	98	63	72	23	
Al2O3	8.453	8.182	8.349	8.351	8.267	8.198	8.213	8.255	8.397	8.286	61.159348	60.830634	61.742349	61.583161	
Cr2O3	27.451	27.541	27.531	27.535	27.364	27.332	27.805	27.049	27.073	26.402	1.1906085	1.2383159	1.2040910	1.1812745	
FeO	50.041	49.602	50.254	49.504	50.316	49.467	50.053	49.512	50.17	50.638	89	02	9	5	
Mno	0.292	0.301	0.329	0.314	0.293	0.381	0.287	0.292	0.35	0.382	0.6415584	0.5445887	0.4588744	0.6164502	
MgO	6.508	6.281	6.676	6.54	6.485	6.686	6.632	6.525	6.459	6.587	26.872748	26.408205	26.697029	26.834372	
CaO	0	0.007	0.03	0.038	0	0.033	0	0.058	0.038	0.085	02	97	94	81	
V	0	0	0	0	0	0	0	0	0	0	0.0934445	0.0531158			
Ba	0	0	0	0	0	0	0	0	0	0	9	72	0.0649194	0.0973791	
Ni	0	0	0	0	0	0	0	0	0	0	6.5464657	6.5414760	6.8707952	7.0025229	
Zn	0	0	0	0	0	0	0	0	0	0	69	84	47	11	
Total	95.364	94.492	95.773	94.919	95.318	94.765	95.601	94.239	95.117	94.94	0.0688430	0.0688430	0.0080991		
Data set	B20_A_Opq_c-r	B20_A_Opq_c-r	B20_A_Opq_c-r	B20_A_Opq_c-r	B20_A_Opq_c-r	B20_A_Opq_c-r	B20_A_Opq_c-r	B20_A_Opq_c-r	B20_A_Opq_c-r	B20_A_Opq_c-r	B20_A_Opq_c-r	B20_A_Op q_c-r	B20_A_Op q_c-r	B20_A_Op q_c-r	B20_A_Op q_c-r
SiO2	0.04681611	0.04070967	0.01628387	0	0.03358547	0.04376289	0.02035483	0.090579	0.069206431	0.044780632	96.621868	95.774741	97.128596	97.354852	
TiO2	61.7671582	61.5201061	61.8302134	62.0834676	61.7795625	61.2441107	61.4022654	61.6968673	61.48909541	61.22550426	08	24	03	78	
Al2O3	1.20409109	1.17608897	1.20616532	1.20201686	1.20097974	1.16675493	1.18023743	1.18749724	1.207202437	1.21342513	0702241	0.0620822	0.0356209	0.0559757	
Cr2O3	0.64935065	0.72987013	0.61904762	0.69004329	0.67792208	0.56969697	0.71601732	0.72813853	0.674458874	0.719480519	72	39	57	89	
FeO	26.7455039	26.9010245	26.6576449	26.6445165	26.5940228	26.4849564	26.5011144	26.4698083	27.01110076	26.87476777	61.569723	61.707204	61.696867	61.753720	
Mno	0.05114862	0.10032998	0.09639547	0.05705038	0.108199	0.14360958	0.10426449	0.08360832	0.059017636	0.060001263	29	16	25	24	
											1.1781632	1.0983053	1.2424643	1.1885343	
											03	1	64	58	
											0.5705627	0.6216450	0.6441558	0.6199134	
											71	22	44	2	
											26.434462	27.035337	26.814175	26.540499	
											69	74	33	47	
											0.0029508	0.0698375	0.0452468		
											82	36	54	0.0757393	

MgO	7.03445689	6.87179318	6.90871685	7.15820106	6.83187571	6.88177255	6.87478699	6.93865495	7.011504343	7.042440386	6.917698279	6.928675584	6.975578616	7.166184556
CaO	0	0.00809918	0.01518596	0.00303719	0.02530994	0.02632233	0.0384711	0.01113637	0.017210756	0	0.012148769	0	0.037458704	0.014173564
V	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Ba	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Ni	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Zn	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Total	97.4985255	97.3480217	97.3496533	97.8383329	97.2514572	96.5609863	96.8375119	97.20629	97.53879665	97.18039996	96.75593405	97.52308759	97.49156792	97.4147407
Data set	B20_A_Opq_c-r	B20_A_Opq_c-r	B20_B_Opq_Bt_c-r	B20_B_Opq_Bt_c-r	B20_B_Opq_Bt_c-r	B20_B_Opq_Bt_c-r	B20_B_Opq_Bt_c-r	B20_B_Opq_Bt_c-r	B20_B_Opq_Bt_c-r	B20_B_Opq_Bt_c-r	B20_B_Opq_Bt_c-r			
SiO2	0.08345481	0.35926279	0.03460322	0.04986934	0.04681611	0.10075642	0.05597579	0.05292256	0.04986934	4.705019535				
TiO2	61.3175027	58.4211012	61.6079698	61.3981306	61.6379469	61.6813619	61.7175411	61.2420433	60.85957773	55.36034284				
Al2O3	1.077563	1.10867646	1.02259588	1.04022684	1.08793415	1.05474646	1.09934243	0.95310914	1.092082617	2.339732558				
Cr2O3	0.68484848	0.54632035	0.61038961	0.58268398	0.63030303	0.47965368	0.54805195	0.53593074	0.435497835	0.571428571				
FeO	26.656635	25.3316802	27.3706159	27.4938206	27.2039867	27.2706384	27.3393098	26.905064	26.9515182	23.34828765				
Mno	0.06787028	0.04032872	0.09442822	0.08557557	0.108199	0.07278842	0.06000126	0.09934635	0.10426449	0.101313609				
MgO	8.0962617	6.69715423	7.07138055	6.77599125	6.84684476	6.9805683	6.89973542	6.88776017	6.910712721	8.125201873				
CaO	0.04657028	1.88305917	0.04353309	0.01214877	0.04960747	0.0404959	0.01214877	0.02328514	0.005061987	0.12452488				
V	0	0	0	0	0	0	0	0	0	0				
Ba	0	0	0	0	0	0	0	0	0	0				
Ni	0	0	0	0	0	0	0	0	0	0				
Zn	0	0	0	0	0	0	0	0	0	0				
Total	98.0307063	94.3875832	97.8555163	97.4384469	97.6116381	97.6810095	97.7321065	96.6994614	96.40858492	94.67585151				

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: Wherlite B15. Clinopyroxenes.

Data set	B15_Cpx-c to Ol_r	B15_Cpx-c to Ol_r	B15_Cpx-c to Ol_r	B15_Cpx-c to Ol_r	B15_Cpx-c to Ol_r	B15_Cpx-c to Ol_r	B15_Cpx-c to Ol_r	B15_Cpx-c to Ol_r	B15_Cpx-c to Ol_r
SiO2	51.637	49.48	50.884	51.087	51.749	51.335	51.352	51.3	
TiO2	0.761	0.742	0.815	0.77	0.705	0.688	0.729	0.677	
Al2O3	2.969	2.722	2.965	2.962	2.566	2.604	2.441	2.515	
FeO	5.346	5.188	5.161	5.235	5.129	5.262	5.127	5.351	
Mno	0.071	0.123	0.148	0.142	0.158	0.143	0.072	0.103	
MgO	15.995	15.968	16.213	16.196	16.174	16.259	16.41	16.547	
CaO	22.304	21.935	22.016	21.813	22.155	22.197	22.03	22.011	
Na2O	0.248	0.201	0.3	0.233	0.244	0.239	0.242	0.231	
K2O	0.008	0.003	0	0	0	0.008	0.006	0.006	
Cr2O3	0.968	1.001	1.046	1.038	0.966	0.91	0.914	0.799	
Total	99.339	96.362	98.502	98.438	98.88	98.735	98.409	98.741	
Wo	45.3508734	45.1653189	44.7971394	44.6476664	45.1181433	44.9701145	44.6832387	44.350941	
En	45.2520274	45.7476578	45.9014227	46.1256003	45.8297831	45.8325942	46.311583	46.3909462	
Fs	8.48457709	8.33807421	8.19679496	8.36369905	8.15286987	8.32106475	8.11693019	8.41581854	
N° Mg	84.2108053	84.5835974	84.8483087	84.6507495	84.8972412	84.6343444	85.0869889	84.6445624	



Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: Wherlite B15. Olivines.

Data set	B15_Ol_c-r	B15_Ol_c-r	B15_Ol_c-r	B15_Ol_c-r	B15_Ol_c-r	B15_Ol_c-r	B15_Ol_c-r	B15_Ol_c-r	B15_Ol_c-r	B15_Ol_c-r	B15_Ol_c-r	B15_Ol_c-r	B15_Ol in Cpx_c-r	B15_Ol in Cpx_c-r	B15_Ol in Cpx_c-r	B15_Ol in Cpx_c-r	
	SiO2	39.428	38.904	39.001	38.959	39.077	39.217	39.257	39.203	39.042	39.066	39.476	39.393	39.412	39.139	38.914	39.296
TiO2	0.008	0.014	0.027	0.034	0.02	0.018	0.011	0.027	0.026	0.015	0.022	0.023	0.016	0.005	0.018	0.018	
Al2O3	0.065	0.01	0.055	0.051	0.035	0.054	0.04	0.02	0.033	0.021	0.04	0.016	0.047	0.019	0.025	0.035	
Cr2O3	0	0	0.034	0.094	0	0.01	0.058	0.094	0.014	0	0	0.01	0.044	0.051	0.031	0.048	
FeO	17.859	17.803	17.581	17.629	17.631	17.65	17.737	17.603	17.206	17.627	17.616	17.337	17.599	17.554	17.416	17.506	
Mno	0.257	0.277	0.206	0.305	0.242	0.264	0.243	0.279	0.27	0.234	0.211	0.173	0.293	0.292	0.18	0.2	
MgO	42.694	42.633	42.478	42.782	42.752	42.227	42.546	42.885	42.894	42.368	42.659	42.542	42.782	42.617	42.561	42.888	
CaO	0.286	0.286	0.307	0.267	0.332	0.3	0.345	0.258	0.29	0.271	0.3	0.241	0.294	0.3	0.285	0.348	
Na2O	0.001	0.001	0.024	0.006	0.027	0.01	0.016	0	0.007	0.004	0.027	0	0.031	0.004	0.024	0.004	
NiO																	
Total	100.598	99.928	99.713	100.127	100.116	99.75	100.253	100.369	99.782	99.606	100.351	99.735	100.518	99.981	99.454	100.343	
Fo	80.7702627	80.7787	80.9758	80.9578	81.0001	80.7734	80.8334	81.0397	81.3933	80.8714	81.0066	81.2397	80.9940530	80.9741911	81.1716987	81.1932097	
Fa	18.953492	18.9230	18.8010	18.7142	18.7393	18.9395	18.9042	18.6606	18.3155	18.8748	18.7656	18.5725	18.6907827	18.7105815	18.6332527	18.5916651	
Data set	B15_Ol in Cpx_c-r	B15_Ol1_c-r	B15_Ol1_c-r	B15_Ol1_c-r	B15_Ol1_c-r	B15_Ol1_c-r	B15_Ol1_c-r	B15_Ol1_c-r	B15_Ol1_c-r	B15_Ol1_c-r	B15_Ol1_c-r	B15_Ol1_c-r	B15_Ol1_c-r	B15_Ol1_c-r	B15_Cpx-c to Ol_r	B15_Cpx-c to Ol_r	B15_Cpx-c to Ol_r
SiO2	39.484	39.349	39.067	39.259	39.291	39.374	38.935	39.434	39.239	39.306	38.756	39.218	39.313	39.383	39.176	39.279	
TiO2	0.014	0.027	0.016	0.014	0.012	0	0.006	0.032	0.013	0	0.033	0.018	0.015	0.015	0.016	0.005	
Al2O3	0.018	0.041	0.031	0.019	0	0.016	0.024	0.036	0.041	0.039	0.021	0.031	0.026	0.07	0.072	0.025	
Cr2O3	0.051	0.036	0	0	0.048	0.046	0.029	0.123	0.043	0.099	0	0	0.053	0	0.015	0.024	
FeO	17.435	17.838	17.56	17.441	17.401	17.576	17.765	17.805	18.096	17.359	17.715	17.523	17.311	17.271	17.491	17.484	
Mno	0.258	0.296	0.274	0.224	0.211	0.249	0.294	0.254	0.239	0.144	0.249	0.256	0.206	0.227	0.256	0.198	
MgO	42.555	42.52	42.717	42.573	42.752	42.472	42.689	42.716	43.17	42.691	42.24	42.467	42.621	42.263	42.643	43.006	
CaO	0.319	0.311	0.262	0.27	0.31	0.282	0.292	0.317	0.298	0.314	0.267	0.283	0.247	0.361	0.27	0.245	

Na2O	0.015	0.021	0	0	0.007	0.028	0.01	0.034	0.024	0.01	0	0.01	0.015	0	0	0
NiO														0	0	0.002
Total	100.149	100.439	99.927	99.8	100.032	100.043	100.044	100.751	101.163	99.962	99.281	99.806	99.807	99.59	99.939	100.268
Fo	81.084518	80.6908	81.0208	81.1156	81.2258	80.9402	80.8169	80.8271	80.7560	81.2993	80.7351	80.9782	81.2615352	84.9548072	84.8759463	85.0588554
Fa	18.6361735	18.9899	18.6839	18.6418	18.5463	18.7901	18.8668	18.8997	18.9899	18.5448	18.9944	18.7444	14.7202164	14.7611636	14.6622411	
		879	124	593	914	153	34	572	1906	3359	7128	4037	18.5153102	7	2	2

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: Wherlite B15. Opaque minerals.

Data set	B15_Opq in Ol_c-r	B15_Opq in Ol_c-r	B15_Opq in Ol_c-r	B15_Opq in Ol_c-r	B15_Opq in Ol_c-r	B15_Opq in Ol_c-r	B15_Opq in Ol_c-r	B15_Opq in Ol_c-r	B15_Opq in Ol_c-r	B15_Opq in Ol_c-r
SiO2	0.06	0.045	0.048	0.031	0.081	0.069	0.08	0.097	0.085	
TiO2	3.68	3.614	3.569	3.589	3.644	4.822	4.708	4.816	4.77	
Al2O3	15.189	15.109	14.998	15.261	15.381	15.304	15.421	15.5	15.483	
Cr2O3	30.911	31.062	30.446	31.333	30.319	26.923	26.704	26.705	26.9	
FeO	38.059	37.746	37.957	37.714	37.944	39.321	38.866	39.412	39.383	
Mno	0.252	0.354	0.38	0.331	0.3	0.312	0.268	0.289	0.309	
MgO	9.716	9.8	9.718	9.626	9.7	9.95	9.533	9.869	9.959	
CaO	0	0.003	0	0.039	0.068	0.003	0.031	0.014	0.06	
BaO										
V2O3										
NiO										
ZnO										
Total	97.867	97.733	97.116	97.924	97.437	96.704	95.611	96.702	96.949	

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: Dunite B7. Clinopyroxenes.

Data set	B7_Cpx in Ol_c-r_continua	B7_Cpx in Ol_c-r_continua	B7_Cpx in Ol_c-r_continua	B7_Cpx in Ol_c-r_continua	B7_Cpx in Ol_c-r_continua	B7_Cpx in Ol_c-r_continua	B7_Cpx in Ol_c-r_continua	B7_Cpx1 in Ol	B7_Cpx1 in Ol	B7_Cpx1 in Ol
SiO2	50.333	50.853	51.194	50.885	51.063	51.181	51.407	51.165	51.286	51.779
TiO2	1.161	1.173	1.16	1.211	1.193	1.148	1.218	1.068	1.056	1.029
Al2O3	4.138	3.678	3.57	3.552	3.518	3.57	3.519	3.314	3.278	3.283
FeO	4.182	4.143	4.333	4.171	4.107	4.265	4.49	4.373	4.053	4.169
Mno	0.11	0.147	0.108	0.108	0.088	0.098	0.073	0.113	0.174	0.081
MgO	16.375	16.1	15.841	15.785	16.162	16.06	16.077	16.24	16.892	16.205
CaO	21.307	21.8	21.887	21.726	21.976	21.78	21.678	21.826	21.806	21.175
Na2O	0.649	0.67	0.589	0.625	0.68	0.679	0.608	0.649	0.696	0.681
K2O	0.002	0	0	0.008	0	0	0	0	0	0.005
Cr2O3	0.909	1.036	0.934	0.984	1.069	1.073	0.927	0.956	0.77	0.907
Total	98.257	98.564	98.682	98.071	98.787	98.781	99.07	98.748	99.241	98.407
Wo	43.9049761	44.8119541	45.243429	45.1910739	44.9406984	44.7378668	44.5569123	44.533721	43.846561	43.923306
En	46.9487081	46.0483596	45.5619986	45.6844469	45.9872216	45.9001216	45.9781809	46.105396	47.259758	46.770448
Fs	6.72626253	6.64738249	6.99127379	6.77191383	6.55562677	6.83808947	7.20345028	6.9645435	6.3611330	6.7499700
N° Mg	87.4685306	87.3853518	86.696787	87.0903857	87.5232748	87.0338995	86.4550031	86.876669	88.136838	87.388047
Data set	B7_Cpx1 in Ol	B7_Cpx1 in Ol	B7_Cpx1 in Ol	B7_Cpx1 in Ol	B7_Cpx1 in Ol	B7_Cpx1 in Ol	B7_Cpx1 in Ol	B7_Cpx1 in Ol	B7_Cpx1 in Ol	B7_Cpx1 in Ol
SiO2	51.272	51.372	51.959	51.284	51.346	51.837	51.921	51.317	51.351	
TiO2	1.043	1.059	1.008	1.068	1.048	1.05	1.029	1.038	1.048	
Al2O3	3.295	3.31	3.301	3.259	3.362	3.371	3.302	3.39	3.304	
FeO	4.138	4.248	4.193	4.282	4.13	4.48	4.293	4.33	4.125	
Mno	0.145	0.105	0.131	0.094	0.138	0.165	0.083	0.101	0.135	
MgO	16.479	16.503	16.215	16.509	16.652	16.383	16.295	16.412	16.73	
CaO	21.676	21.99	21.477	21.705	21.78	22.082	21.85	22.001	21.752	
Na2O	0.703	0.658	0.667	0.662	0.691	0.697	0.645	0.669	0.692	



Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: Dunite B7. Olivines.

Data set	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r
SiO2	40.114	40.347	40.481	40.095	40.219	39.97	40.272	40.201	40.283	40.603	40.315	40.034	40.347	40.207
TiO2	0.011	0	0.022	0	0.004	0.012	0.013	0	0	0.022	0	0.013	0.016	0.008
Al2O3	0.034	0.012	0.023	0	0.015	0.011	0.029	0.017	0.046	0.024	0.033	0.04	0	0.025
Cr2O3	0.054	0	0.005	0.025	0.071	0	0.049	0.037	0.005	0	0.015	0.005	0	0.012
FeO	13.031	13.237	13.105	12.964	13.359	13.241	13.119	13.174	13.262	12.993	13.039	13.078	13.285	13.416
Mno	0.178	0.21	0.173	0.151	0.213	0.171	0.128	0.254	0.204	0.168	0.172	0.178	0.195	0.209
MgO	46.607	46.599	46.282	46.274	47.019	46.333	46.644	46.798	46.618	46.723	46.733	46.369	46.519	46.626
CaO	0.044	0.059	0.067	0.077	0.132	0.096	0.113	0.099	0.067	0.136	0.059	0.046	0.069	0.065
Na2O	0	0.004	0	0	0.011	0.009	0.001	0.003	0.021	0.021	0.003	0.027	0.027	0.034
NiO														
Total	100.073	100.468	100.158	99.586	101.043	99.843	100.368	100.583	100.506	100.69	100.369	99.79	100.458	100.602
Fo	86.2800749	86.0649239	86.1349923	86.2800328	86.0615521	86.0279233	86.2560236	86.1323454	86.0529465	86.3525417	86.3102830	86.1771097	86.0150224	85.9135147
Fa	13.5327032	13.7147094	13.6820753	13.5600013	13.7169383	13.7916826	13.6094893	13.6020413	13.7330999	13.4710451	13.5092305	13.6349319	13.7801182	13.8676804
									4	3	4	6	4	8
									3	5	5	9	3	4
Data set	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r	B7_Ol_c-r
SiO2	39.947	39.776	40.169	40.05	40.324	40.188	41.212	40.363	40.334	40.883	40.215	40.521	40.21	40.162
TiO2	0	0.006	0.003	0.011	0.012	0	0	0.003	0.006	0.023	0.019	0.018	0.017	0.015
Al2O3	0.042	0	0.009	0.018	0	0.041	0.023	0.015	0.02	0.027	0.016	0.015	0.017	0.022
Cr2O3	0.091	0.002	0	0.034	0.032	0	0.015	0	0.015	0.02	0.044	0.007	0.039	0.032
FeO	13.105	13.365	13.294	13.139	13.183	13.351	13.252	13.053	12.869	13.222	13.107	13.262	13.38	13.369
Mno	0.113	0.151	0.118	0.209	0.128	0.118	0.204	0.145	0.257	0.173	0.245	0.193	0.213	0.199
MgO	46.541	46.802	46.78	46.821	46.586	46.648	45.7	46.172	46.595	45.249	46.639	46.223	46.41	46.141
CaO	0.103	0.067	0.059	0.176	0.107	0.084	0.096	0.069	0.04	0.081	0.069	0.092	0.107	0.065
Na2O	0.003	0.013	0.011	0.033	0	0.009	0.02	0.016	0	0.014	0.014	0.024	0	0

NiO														
Total	99.945	100.182	100.443	100.491	100.372	100.439	100.522	99.836	100.136	99.692	100.368	100.355	100.393	100.005
Fo	86.2559665	86.0563545	86.1435357	86.2099051	86.1839851	86.0589753	85.8215728	86.1789998	86.3505653	85.7564050	86.1595539	85.9604963	85.8857025	85.8376390
Fa	13.6250439	13.7858943	13.7330056	13.5714497	13.6814728	13.8173382	13.9607627	13.6672314	13.3788294	1	2	2	7	3
Data set	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r
SiO2	40.675	39.793	40.275	39.698	39.928	40.172	40.115	38.57	39.945	39.914	39.741	40.213	39.926	40.027
TiO2	0	0.007	0.007	0.015	0.02	0	0.034	0	0.001	0	0	0.005	0.014	0.004
Al2O3	0.013	0	0.015	0.034	0	0.015	0.055	0.011	0.01	0.034	0.01	0	0.013	0.018
Cr2O3	0.039	0.02	0.047	0.042	0.005	0	0.027	0	0	0	0.027	0	0	0.02
FeO	13.404	12.95	13.188	13.227	13.132	13.035	13.45	13.074	13.211	13.106	13.237	13.364	13.284	13.335
Mno	0.288	0.187	0.24	0.158	0.206	0.209	0.166	0.199	0.198	0.216	0.163	0.158	0.223	0.146
MgO	46.312	46.37	46.91	46.115	46.6	46.442	46.293	49.173	46.437	46.295	46.392	46.514	46.405	46.48
CaO	0.13	0.103	0.086	0.096	0.123	0.069	0.159	0.092	0.105	0.053	0.023	0.084	0.057	0.061
Na2O	0	0.008	0.01	0.017	0	0.028	0	0	0.001	0.016	0.024	0	0.01	0.014
NiO														
Total	100.861	99.438	100.778	99.402	100.014	99.97	100.299	101.119	99.908	99.634	99.617	100.338	99.932	100.105
Fo	85.7708893	86.2843119	86.1610323	85.9955717	86.1625317	86.2062521	85.8350745	86.8469141	86.0572258	86.0983089	86.0540485	85.9766927	85.9610354	86.0045980
Fa	13.9260594	13.517985	13.5885103	13.8370234	13.6210585	13.5733277	13.9900477	12.9533952	1	4	8	9	85.9610354	9
Data set	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r	B7_OI_c-r
SiO2	39.998	40.052	39.924	39.985	40.104	40.001	40.25	40.203	39.837	40.143	39.941	39.924	39.86	40.347
TiO2	0	0	0	0.01	0.032	0.021	0.02	0.005	0	0.001	0.021	0.006	0	0
Al2O3	0	0.026	0.012	0.01	0.013	0	0	0.004	0.047	0.019	0.013	0.021	0.023	0.003
Cr2O3	0.017	0	0	0.064	0.049	0.061	0	0	0.061	0.047	0.054	0	0	0







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Zn	0	0	0	0	0	0	0	0	0	0
Total	98.197	98.421	97.753	98.281	97.595	98.091	97.875	97.533	98.129	97.699

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: Dunite B43. Olivines.

Data set	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r
SiO2	40.4918683	40.1997765	40.0939313	40.5773586	40.3025684	39.7031186	40.0857894	40.4888151	40.22623775	39.88732979	40.134641	40.47253122	40.18756357	40.18552808
TiO2	0.02480858	0	0.01653905	0.01240429	0.02170751	0.0227412	0.02170751	0	0.042381322	0.038246558	0.024808578	0.009303217	0.024808578	0.054785611
Al2O3	0.06741251	0.01659385	0.05081866	0.0736352	0.02903923	0.0466702	0.04770731	0.01763096	0.057041352	0.014519617	0.060152699	0	0.059115583	0.001037115
Cr2O3	0.01645022	0.04848485	0.03030303	0.00606061	0.01991342	0	0.06320346	0.01818182	0.001731602	0.058874459	0.007792208	0.052813853	0.025974026	0.03030303
FeO	13.86961	14.3482903	14.0513873	14.4664455	14.2432634	14.2281153	14.0362392	14.3674779	14.0241207	14.12813772	14.09683163	14.34324089	13.73731647	14.29577681
Mno	0.18688918	0.2704975	0.17508565	0.19279094	0.20656173	0.20951261	0.20262722	0.20164359	0.177052908	0.170167517	0.196725454	0.219348881	0.23410329	0.236070544
MgO	44.2704748	44.5748455	44.4221612	44.7424989	44.5059879	44.8303173	44.4740539	44.5818311	44.34532002	44.42216116	44.42216116	45.23049002	44.58682074	44.53293215
CaO	0.23993818	0.24904976	0.26626052	0.28043408	0.27638449	0.25309935	0.38268622	0.29562004	0.345227514	0.286508464	0.306756412	0.308781207	0.295620041	0.279421683
Na2O	0	0	0	0	0	0.02031872	0.0083122	0	0	0.02863092	0	0	0	0
NiO	0	0.01573511	0.00629404	0.00839206	0	0.00314702	0.01153908	0	0.010490073	0.006294044	0	0	0	0.002098015
Total	99.1674517	99.7232733	99.1127808	100.36002	99.605426	99.3170402	99.3338655	99.9712004	99.22960324	99.04087025	99.24986914	100.6365093	99.1513223	99.61795305
Fo	84.8788769	84.4578243	84.768342	84.4717282	84.5903744	84.6956964	84.7718545	84.5052788	84.7689432	84.70326608	84.70725996	84.69908788	85.04684227	84.52403913
Fa	14.9175376	15.2509763	15.0418294	15.3214697	15.1865616	15.0794101	15.0087034	15.2775573	15.03876125	15.11237917	15.07960312	15.06753405	14.69944843	15.22138459
Data set	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r	B43_Ol_c-r
SiO2	40.6893102	39.5972734	40.1203926	40.4847441	40.2700006	40.1122507	40.3096926	39.9636604	40.70661179	40.30053289	40.4623538			
TiO2	0.01653905	0.04134763	0.01550536	0.00723584	0.02480858	0.02480858	0.01550536	0.01240429	0.015505362	0.01343798	0.017572743			
Al2O3	0.0404475	0.02385366	0	0.02385366	0.0404475	0.01970519	0.02903923	0.00622269	0.024890772	0.045633082	0.045633082			
Cr2O3	0.04675325	0.03290043	0.02857143	0	0.07705628	0.0017316	0.03030303	0	0.126406926	0	0.097835498			
FeO	14.398784	14.0402787	14.1230884	14.2442732	14.4371592	14.4028235	13.9776665	14.0513873	14.01503183	14.09380201	13.77872131			
Mno	0.25082495	0.17705291	0.1839383	0.15541311	0.18688918	0.20164359	0.13278968	0.19180732	0.194758199	0.231152408	0.23410329			
MgO	44.9610471	44.9949769	44.7524783	44.6407093	44.6696495	44.9231255	44.5598764	44.4171715	44.7345154	44.56885788	44.13475535			
CaO	0.26626052	0.23791339	0.27942168	0.27739689	0.29562004	0.23993818	0.31991758	0.31789278	0.310806002	0.262210927	0.228801813			
Na2O	0	0	0	0	0	0	0	0	0.012930093	0	0.02401303			
NiO	0.00734305	0	0	0	0.00839206	0.00104901	0.01153908	0.00419603	0.016784116	0.001049007	0.001049007			
Total	100.67731	99.145597	99.503396	99.8336262	100.010023	99.9270758	99.3863295	98.9647423	100.1582405	99.51667618	99.02483892			
Fo	84.5434579	84.9412344	84.7909886	84.6754728	84.4819375	84.5732919	84.9139384	84.7515249	84.873432	84.72114529	84.87884886			

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Fa | 15.1885688 14.8688613 15.0110037 15.1570362 15.3172399 15.2110207 14.9422889 15.0405347 14.91662473 15.02920221 14.86534942

Table 9.1 - Continue. Single mineral phase EMPA analyses. Sample: Dunite B43. Opaque minerals.

Data set	B43_Opq- Ol_c-r	B43_Opq1_c -r	B43_Opq1_c -r	B43_Opq1_c -r	B43_Opq1_c -r	B43_Opq1_c -r	B43_Opq1_c -r	B43_Opq1_c -r	B43_Opq1_c -r	B43_Opq1_c -r	B43_Opq1_c -r	B43_Opq in Ol_r-r	B43_Opq in Ol_r-r	B43_Opq in Ol_r-r
SiO2	0.06920643	0.02544354	0.08447255	0.99840453	0.08854352	0.03765644	0.05394031	0.08447255	0.09261448	0.14350156	0.05190482	0.08752578	0.09363223	0.09770319
TiO2	2.44571236	2.97496203	2.90363737	2.74031423	2.98736632	2.99666954	2.91500797	2.90983952	2.91810904	2.89536784	2.83127901	2.26171540	2.39506151	2.41780270
Al2O3	17.0190653	16.9060197	16.0130633	12.8654177	17.2617503	16.8054195	16.8749062	16.7193389	16.6342954	16.3895361	16.6705944	16.7037821	16.7286729	16.7286729
Cr2O3	33.1428571	32.830303	32.2623377	29.2502165	32.9090909	33.1445887	32.791342	32.791342	32.7393939	32.6649350	32.0883116	32.8831168	33.1861471	33.8848484
FeO	29.2843272	29.2439323	28.7935284	26.2839915	29.4418676	29.2873568	29.0651845	29.1207276	29.3489591	28.4451218	29.0894215	28.3451443	28.8218049	28.7450544
Mno	0.24984133	0.21443074	0.25869397	0.25771034	0.26066123	0.35607307	0.22820153	0.27639926	0.22131613	0.25574309	0.31377709	0.27344838	0.35213856	0.21147986
MgO	12.2077615	12.2816089	11.6549045	10.8216273	12.3774108	12.2985738	12.5271013	12.0540793	12.2786150	11.7067972	12.3953736	12.1319183	12.3305077	12.1229369
CaO	0	0	0.06276864	1.28169511	0.03138432	0	0	0.01619836	0	0.04657028	0.01518596	0	0.05163226	0.02632233
V	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Ba	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Ni	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Zn	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Total	94.4187713	94.4767002	92.0334064	84.4993771	95.358075	94.9263379	94.4556839	93.9723975	94.2333032	91.7458828	93.1747899	92.6534636	93.9347066	94.2348209
Data set	B43_Opq in Ol_r-r	B43_Opq in Ol_r-r	B43_Opq in Ol_r-r	B43_Opq in Ol_r-r	B43_Opq c to Ol c	B43_Opq c to Ol c	B43_Opq c to Ol c	B43_Opq c to Ol c	B43_Opq c to Ol c	B43_Opq1 in Ol_r-r	B43_Opq1 in Ol_r-r	B43_Opq1 in Ol_r-r	B43_Opq1 in Ol_r-r	B43_Opq1 in Ol_r-r
SiO2	0.06208224	0.07938385	0.0854903	0.08650804	0.08345481	0.10889835	0.04681611	0.06309998	0.03562095	0.08549029	0.04274514	0.03053224	0.02544354	0.04070966
TiO2	2.37748877	2.3516465	2.33407376	2.30099565	2.97806311	2.92017642	2.86952558	2.89743523	2.86332343	2.43744283	2.50980118	2.55011512	2.52220547	2.46638617
Al2O3	16.7701576	16.8894258	16.7058564	16.7628978	17.5977757	16.9402445	16.9329847	16.9506157	17.2244141	16.5150271	16.4735425	16.6425923	16.4756167	16.4859879
Cr2O3	34.0683983	33.3489177	33.3272727	33.3047619	32.7601732	32.8813853	32.7229437	33.0467532	32.7454545	34.3471861	34.1359307	34.1272727	34.0822510	33.9151515
FeO	28.6531559	28.9207726	28.364332	28.564287	28.9349108	29.0661944	29.3519888	28.9641971	29.3489591	28.9793452	28.8096864	28.5430796	28.3653418	29.0853820
Mno	0.21443074	0.25082495	0.23410329	0.25574309	0.25869397	0.22525064	0.26656299	0.24689044	0.20164359	0.29017004	0.19967633	0.26754661	0.30787533	0.25082495

MgO	11.9852216	12.2546646	12.1818152	12.120941	11.6369417	12.1339142	12.1319183	12.137906	11.7856342	12.3005696	12.0401081	12.0041824	11.9183598	12.1818151
									6	7	6	3	6	9
CaO	0.01518596	0	0.00303719	0.04150829	0.00809918	0.02024795	0.02328514	0.01518596	0	0	0.02834712	0	0.01822315	0
V	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Ba	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Ni	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Zn	0	0	0	0	0	0	0	0	0	0	0	0	0	0
Total	94.1461211	94.095636	93.2359808	93.4376428	94.2581124	94.2963118	94.3460254	94.3220836	94.2050500	94.9552314	94.2398376	94.1653211	93.7153170	94.4262574
									8		4	8	5	4
Data set	B43_Opq1 in Ol_r-r	B43_Opq1 in Ol_r-r	B43_Opq1 in Ol_r-r	B43_Opq1 in Ol_r-r	B43_Opq1 in Ol_r-r	B43_Opq1 in Ol_r-r	B43_Opq1 in Ol_r-r	B43_Opq1 in Ol_r-r	B43_Opq1 in Ol_r-r	B43_Opq1 in Ol_r-r	B43_Opq1 in Ol_r-r	B43_Opq1 in Ol_r-r	B43_Opq1 in Ol_r-r	
SiO2	0.02646128	0.06513546	0.05902901	0.03765644	0.03460322	0.02544354	0.0488516	0.04783386	0.07327739	0.09464997	0.18115800			
									7	1	9			
TiO2	2.5191044	2.51600333	2.4860263	2.51703702	2.49119475	2.49016106	2.49843059	2.48292523	2.52427285	2.52323916	2.45191450			
									8	8	5			
Al2O3	16.3221237	16.2598967	16.4756168	16.5803654	16.3262721	16.5171014	16.5720685	16.6705945	16.5067302	16.5482148	16.5876252			
									3	5	4			
Cr2O3	34.2181818	34.434632	34.3601732	34.0753247	33.9922078	33.412987	34.2891775	34.4337662	33.7610389	34.3922077	33.4311688			
									6	9	3			
FeO	28.8632097	28.757173	28.8106963	28.8349333	28.5632772	28.8823973	29.2722087	28.4814774	28.7541433	28.9682366	29.2671593			
									5	4	5			
Mno	0.19770908	0.26951387	0.26557936	0.23705417	0.27246475	0.29312093	0.23705417	0.26951387	0.22820152	0.25869397	0.16820026			
									6	2	3			
MgO	12.2776171	12.0949947	11.9622691	12.2127512	11.6010159	12.3015676	12.0490896	12.0760339	12.2067636	12.2237285	13.6238339			
									1	4	4			
CaO	0.02126035	0	0.01619836	0	0.02834713	0.00202479	0.03644631	0	0.02632233	0.04150829				
									2	3	0			
V	0	0	0	0	0	0	0	0	0	0	0			
Ba	0	0	0	0	0	0	0	0	0	0	0			
Ni	0	0	0	0	0	0	0	0	0	0	0			
Zn	0	0	0	0	0	0	0	0	0	0	0			
Total	94.4456675	94.3973491	94.4355883	94.4951222	93.3093829	93.9248037	95.003327	94.4621449	94.0807502	95.0504792	95.7110601			
									6	2	4			

