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Optical Properties of Silicon Nanowires Fabricated by Environment-Friendly Chemistry

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Abstract

Silicon nanowires (SiNWs) were fabricated by metal-assisted chemical etching (MACE) where hydrofluoric acid (HF), which is typically used in this method, was changed into ammonium fluoride (NH₄F). The structure and optical properties of the obtained SiNWs were investigated in details. The length of the SiNW arrays is about 2 μ m for 5 min of etching, and the mean diameter of the SiNWs is between 50 and 200 nm. The formed SiNWs demonstrate a strong decrease of the total reflectance near 5–15 % in the spectral region λ < 1 μ m in comparison to crystalline silicon (c-Si) substrate. The interband photoluminescence (PL) and Raman scattering intensities increase strongly for SiNWs in comparison with the corresponding values of the c-Si substrate. These effects can be interpreted as an increase of the excitation intensity of SiNWs due to the strong light scattering and the partial light localization in an inhomogeneous optical medium. Along with the interband PL was also detected the PL of SiNWs in the spectral region of 500–1100 nm with a maximum at 750 nm, which can be explained by the radiative recombination of excitons in small Si nanocrystals at nanowire sidewalls in terms of a quantum confinement model. So SiNWs, which are fabricated by environment-friendly chemistry, have a great potential for use in photovoltaic and photonics applications.

Keywords: Silicon nanowires, Environment-friendly chemistry, Total reflectance, Photoluminescence, Raman scattering

Background

In the last years, silicon nanowires (SiNWs) are of great interest because of their potential applications in microand optoelectronics [1, 2], photonics [3], photovoltaics [4, 5], and sensorics [6–8]. Initially, SiNWs were obtained by vapor-liquid-solid method with the help of a noble metal (mostly gold, Au) which was first proposed by Wagner and Ellis in 1964 [9]. An alternative method was metal-assisted chemical etching (MACE) by which initially it was believed that the porous silicon (PSi) was obtained [10, 11]. MACE is based on the anisotropic etching of crystalline silicon (c-Si) in aqueous solutions which are usually based on hydrofluoric acid (HF) [10–15]. The reaction is catalyzed by metal nanoparticles such as Au [10, 12], Ag [13], or Pt [10, 11] at the substrate surface, and the oxidizing agents are H₂O₂ [10–13], KMnO₄ [14],

Ammonium fluoride (NH₄F) has long been used for SiO₂ dissolution. It has been shown that the SiO₂ etch rate in aqueous NH₄F solutions depends on the pH and on the NH₄F concentration [26]. Then, aqueous NH₄F solution was proposed as an alternative of HF in the electrochemical method of obtaining PSi films [27–29]. Furthermore, the microstructure properties of the fabricated PSi have shown a strong dependency from the pH of the electrolyte.

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or Fe(NO₃)₃ [15]. It has been shown that SiNWs, which were obtained by MACE, are found to possess such remarkable optical properties as visible photoluminescence (PL) [16], extremely low total reflection [17, 18], enhancement of Raman scattering [17, 19–21], coherent anti-Stokes light scattering [22], interband PL [17, 19–21] and efficiency of generation of third harmonics [23] in comparison with the corresponding intensities for c-Si, and sensitivity of visible PL to molecular surroundings [24]. However, HF is rather toxic and may also result in hypocalcemia and hypomagnesemia [25]. Therefore, there is great interest in the study of possible modifications of the MACE method by using environment-friendly chemistry.

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A pebble-like surface structure was formed during anodization at pH 4.5, while nanoporous Si layers were formed at lower pH [27]. Also, the usage of NH_4F as the etchant allows it to survive the aluminum contact during the etching process, because the aluminum etch rate in 4.5 % ammonium fluoride is five times lower than that of 5 % HF [28, 29]. It is very important because aluminum is usually used as an interconnect layer in sensors [29].

The possibilities of usage of NH₄F in the MACE process are shown in [30], where n-type Si wafers were covered with sparse Ag particles and then etched in solutions with different NH₄F concentrations. Examination by scanning electron microscope (SEM) shows that the etching of the smooth surface of n-Si (100) in 1.0 M $NH_4F + 5.0$ M H_2O_2 for 1 h is leading to the formation of a PSi-like surface, where the pores were 50-150 nm in diameter and 200-300 nm in depth. In contrast, only a few shallow pores on the Si (100) surface could be formed after the etching conducted in 11.0 M $NH_4F + 5.0$ M H_2O_2 for 1 h [30]. In [31], the Si (100) surfaces, covered with sparse Ag particles, were put in 1.0 M $\text{NH}_4\text{F} + 5.0$ M H_2O_2 to investigate their dark etching. It has been shown, that the morphology on the surface etched for 1 h revealed a sparse distribution of nanopores (10~40 nm in diameter) according to the locations of Ag particles. However, it exhibited porous surface consisting of micropores (1.5~3.1 µm in diameter with 15~20 µm in depth) where nanopores (100~150 nm in diameter) were embedded inside for the etching duration prolonged for 5 h. In ref. [32], it has been shown that the replacement of HF on the NH₄F aqueous solution in the first MACE step, and also the usage of NH₄F solution for subsequent deposition of silver particles on SiNW's sidewall, leads to a good coating of SiNWs without formation of silver dendrites and etching pits. Such silver-coated SiNWs are proposed for surfaceenhanced Raman scattering application [32]. It was also showed that additional etching of SiNWs in NH₄F leads to their surface coverage mainly with $Si-(O^-)_x$ species (x = 1-3) [33]. This, in its turn, leads to the greater stability of the samples, which is important for creating SiNW-based sensors [34].

In this study, SiNW arrays were prepared by MACE method, where HF is changed on NH_4F in both etching steps: the decoration of the c-Si layer with Ag nanoparticles was done by using of 0.02~M of $AgNO_3$ and 5~M of NH_4F , and the etching was done in the solution containing 5~M of NH_4F and 30~% of H_2O_2 . The structure and optical properties (total reflectance, Raman scattering, and PL) of SiNWs are investigated in detail.

Methods

SiNWs were fabricated by MACE of p-type (100)-oriented c-Si wafer with specific resistivity of 10–20 Ω cm. The difference between the standard MACE method and

the new approach was in the changing of HF on NH₄F. The pH value of the NH₄F solution was changed to 2.0 by adding H₂SO₄ droplets. It was made to have H⁺ ions in the reactions. The pH value was controlled with a calibrated pH meter Checker1 (Hanna Instruments). Prior the MACE procedure, the c-Si substrate was rinsed in 2 % HF solution for 1 min to remove native oxide. Then, in the first step of MACE, thin (~100 nm) layers of Ag nanoparticles of different morphology were deposited on the substrates by immersing them in aqueous solution of 0.02 M of silver nitrate (AgNO₃) and 5 M of NH₄F in the volume ratio of 1:1 for 30 s. In the second step, the c-Si substrates covered with Ag nanoparticles were immersed in the solution containing 5 M of NH₄F and 30 % H₂O₂ in the volume ratio of 10:1 in a Teflon vessel for 5 min. All the etching steps were performed at room temperature. Then, the sample were rinsed several times in de-ionized water and dried at room temperature. Finally, SiNW arrays were immersed in concentrated (65 %) nitric acid (HNO₃) for 15 min to remove residual Ag nanoparticles from the SiNWs. Figure 1a shows the step-by-step formation of SiNWs. The main net etching reaction described in ref. [35] is as follows:

$$Si + H2O2 + 6F- + 4H+$$
= SiF₆²⁻ + 4H₂O, (1)

where ions F^- and H^+ were obtained not from the dissociation of HF as in the standard MACE method but from the dissociation of NH₄F and H₂SO₄. Ag was a catalyst.

The structure properties of SiNWs were investigated by using a SEM of Lyra3 Tescan and a TEM of LEO912 AB Omega. The total reflectance spectra in the region from 250 to 1500 nm were studied with a Perkin Elmer Lambda 950 spectrometer equipped with an integrating sphere. The interband PL and Raman spectra under excitation with a cw Nd:YAG laser at 1.064 µm (excitation intensity ~100 mW; spot size ~2 mm) were measured in a backscattering geometry with a Fourier transform infrared (FTIR) spectrometer of Bruker IFS 66v/S equipped with a FRA-106 unit. PL of SiNWs in the spectral region of 500-1100 nm under excitation with ultraviolet (UV) radiation of an N2 laser (wavelength 337 nm, pulse duration 10 ns, repetition rate 100 Hz) was detected by using a grating spectrometer (MS 750, SOLAR TII) equipped with a CCD unit. All measurements were carried out at room temperature in air.

Results and Discussion

Figure 1b shows a TEM micrograph of a single SiNW. The diameter of the SiNW is about 80 nm, and it is nearly constant through the whole SiNW length. The sidewall surface of the SiNW is porous and contains silicon nanocrystals with the diameter of a few nanometers.

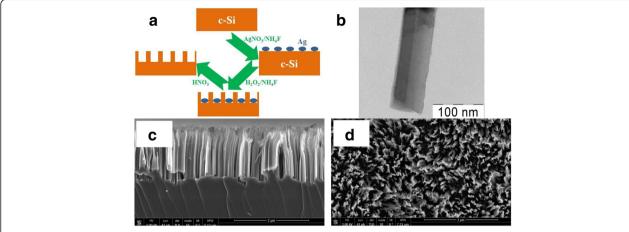


Fig. 1 Typical TEM and SEM micrographs of SiNWs. a Schematic step-by-step representation of SiNWs preparation method. b TEM micrograph of a single SiNW. c SEM cross-sectional micrograph of SiNWs. d SEM micrograph of SiNWs (view from above)

Thus, SiNWs are structured as a single-crystal core that replicates the crystallographic orientation of the substrate and are covered by a thin (<10 nm) nanostructured layer. The same structure of SiNWs was observed for SiNWs fabricated on the lightly doped (10 Ω cm) substrate by standard MACE method [24].

A typical large-scale cross-sectional SEM micrograph of a sample with SiNWs is shown in Fig. 1c. One can see that SiNWs look like quasi-ordered arrays with preferential orientation along the [100] crystallographic direction. The length of SiNW arrays is about 2 μ m. From the SEM micrograph (view from above) is seen the high density of SiNW arrays (Fig. 1d).

Figure 2 shows the total reflectance spectra of SiNWs and the c-Si substrate. The total reflectance of c-Si demonstrates the well-known behavior of the wafer, the reflection value of ~ 30 % for the range of the strong absorption ($\lambda < 1 \mu m$), and for the transparency region ($\lambda > 1 \mu m$), the

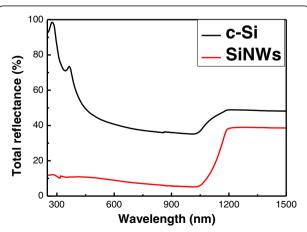


Fig. 2 Total reflectance spectra. Total reflectance spectra of c-Si (black curve) and SiNWs (red curve)

reflection value increases because of both side contributions. SiNWs exhibit a strong decrease of the total reflectance to 5–15 % in the spectral region λ < 1 µm in comparison to the c-Si substrate. In the visible spectral region, the sample with SiNWs looks similar to "black silicon" and can be used as antireflection coating in photovoltaic applications. The same property was observed for samples with SiNWs fabricated by standard MACE method [17, 18], and it can be explained by the strong light scattering and absorption, which results in partial localization (trapping) of the excitation light in SiNW arrays.

The effect of light localization can be confirmed by analyzing the Raman spectra of SiNWs. Figure 3 shows the typical spectra of the interband PL (broadband) and

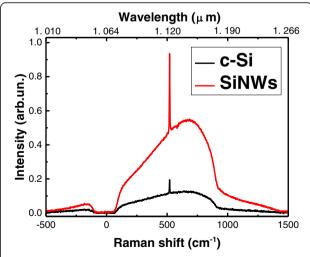


Fig. 3 Interband photoluminescence and Raman spectra. Interband photoluminescence (broadband) and Raman spectra (sharp peak at 520 cm⁻¹) of c-Si (*black curve*) and SiNWs (*red curve*)

Raman scattering (sharp peak at 520 cm⁻¹) of SiNWs and c-Si substrate for comparison. The PL band and Raman peak position and shape for SiNWs are similar to the c-Si substrate because the SiNW's diameter is about 50–200 nm and it is far from the quantum confinement regime. The PL and Raman intensities increase strongly for the SiNWs in comparison with the corresponding value of the c-Si substrate. The same results were observed for SiNWs fabricated by standard MACE method [17, 19–21] and can be interpreted as an increase of the excitation intensity of SiNWs because of the strong light scattering and the partial light localization in inhomogeneous optical medium.

Along with the interband PL, the PL of SiNWs in the spectral region of 500–1100 nm with maximum at 750 nm (photon energy of 1.65 eV) was also detected, as it is shown in Fig. 4. This PL can be explained by the radiative recombination of excitons in small Si nanocrystals at the nanowire sidewalls (Fig. 1b) in terms of a quantum confinement model. The nanocrystal size was estimated from the spectral position of the PL peak by using the equation [36]

$$h\nu_{\rm PL} = E_g + \frac{3.73}{d^{1.39}},$$
 (2)

where $h\nu_{\rm PL}$ —PL peak position (eV), $E_{\rm g}$ —band gap of c-Si (1.12 eV), and d—nanocrystal size (nm). With the photon energy of 1.65 eV, the size of nanocrystals was about 4 nm. Same nanocrystals were observed on nanowire sidewalls in [18, 24]. The presence of the PL in the spectral region of 500–1100 nm will allow the use of SiNWs as luminescent labels for living cells as was shown for SiNWs fabricated by the standard MACE method [18].

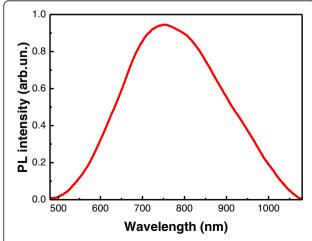


Fig. 4 Photoluminescence spectra. PL in the spectral region of 500–1100 nm of SiNWs

Conclusions

Total reflectance, interband PL, Raman scattering, and PL in the spectral region of 500-1100 nm were investigated. SiNWs exhibit a strong decrease of the total reflectance to 5–15 % in the spectral region λ < 1 μ m in comparison to the c-Si substrate. The PL and Raman intensities increase strongly for the SiNWs in comparison with the corresponding value of the c-Si substrate. The nanocrystal size at nanowire sidewalls was estimated from the spectral position of the PL and was about 4 nm. It was shown that optical properties of SiNWs formed by MACE using NH₄F were not much different from optical properties of SiNWs formed by the standard MACE technique with HF. It gives an opportunity to obtain high-quality SiNWs for various applications in photonics, photovoltaics, and sensorics using environment-friendly chemistry.

Abbreviations

c-Si, crystalline silicon; FTIR, Fourier transform infrared; MACE, metal-assisted chemical etching; PL, photoluminescence; PSi, porous silicon; SEM, scanning electron microscope; SiNWs, silicon nanowires; TEM, transmission electron microscope; UV, ultraviolet

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Authors' Contributions

KAG and AAZ performed the optical measurements and data analysis. LAO performed the SiNW fabrication. AS performed the SEM measurements. KAG, LAO, and VS performed the general data analysis and discussion of the obtained data. All authors read and approved the final manuscript.

Competing Interests

The authors declare that they have no competing interests.

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