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Material perspectives of HfO₂-based ferroelectric films for device applications

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Abstract— Ferroelectric HfO₂ attracts a huge amount of attention not only for memory and negative capacitance, but also for programmable logic including memory-in-logic and neuromorphic applications. However, the understanding of material fundamentals still needs to be improved. This paper gives material fundamentals and new insights to this ferroelectric material for future device applications. In particular, the key role of dopants, effects of the interface on the ferroelectric phase, and a detailed discussion of the switching kinetics are of central focus. Based on material properties newly obtained, we discuss opportunities of ferroelectric HfO₂ for device applications.

INTRODUCTION

Since the first publication on ferroelectric HfO₂, it has been reported that various dopants can stabilize the ferroelectric (FE) properties in HfO₂ films [1-5]. Although oxygen vacancies (Vo's) formed during processing are often discussed as a possible origin for FE orthorhombic phase formation of polymorphic HfO₂, it has not been clearly understood how dopants and other process conditions induce the FE phase in HfO₂. And, it was reported that even un-doped HfO₂ exhibited FE behaviors [6-8]. Besides, we previously reported an impact of thermal annealing transient on the resultant HfO₂ phase. A significant effect of the ramp-down speed on the structural phase stabilization of the high-temperature phase was demonstrated [9]. Since it is inferred empirically that the FE phase in HfO₂ is formed at the boundary between monoclinic and tetragonal phases [10], the FE-phase formation should also be sensitive to the post deposition anneal (PDA) transients. Furthermore, the polarization switching kinetics in FE is critically important for applications in terms of speed and power supply voltage. Therefore, the objective of this work is to clarify (i) key roles of doping as well as (ii) of interface electrode in FE phase HfO₂ formation. Furthermore, (iii) coercive field (*E_c*) and the polarization switching kinetics are discussed. Finally, the power supply voltage in FE-HfO₂ devices is discussed from device application viewpoints.

KEY ROLE OF DOPANTS IN FE-HfO₂

The impact of various dopants on the FE-phase formation was studied systematically (Fig. 1) [10]. More recently, a universal relation of the remanent polarization (*P_r*) versus normalized dopant concentration in HfO₂ has been characterized (Fig. 2), and a normalization factor α was determined (see the caption for the normalization) as a function of the ionic radius, as shown in Fig. 3. A clear difference is visible for tetravalent dopants like Si, Ge or Zr in contrast to trivalent dopants (e.g. Y or Sc). Both are expected to be positioned in Hf sites within HfO₂ unit cell. Trivalent dopants are expected to generate Vo's in HfO₂ to compensate the charge imbalance. Results distinguish the simple Vo effect from tetravalent dopant effect on FE phase formation. Accordingly, we cannot say simply everything comes from Vo

in HfO₂, but should take account of the bonding orbital-induced internal stress/strain. For the device fabrication viewpoint, the dopant size impacts the HfO₂ crystallization temperature. Hence, dopants with an atomic radius close to Hf need to be selected for a low temperature e.g. capacitor application and other dopants for devices with high thermal budget as in gate-first FE-FETs.

Next, effects of PDA are discussed to get more insights to the impact of Vo on FE phase formation. Y-doped HfO₂ was deposited on TiN electrode. Before top electrode deposition, different PDAs were carried out in N₂ using a rapid-thermal-annealing (RTA) furnace, having a sample stage designed with a very small heat capacity. The temperature was monitored by the thermocouple directly touching the sample. The peak temperature was set to be 600°C. During PDA, both the ramp-up time up to 600°C and ramp-down time back to 200°C could be varied. Samples were characterized by XRD and polarization versus voltage (P-V) hysteresis. The PDA process can be divided into a ramp-up, holding and ramp-down phases. Since as-deposited films are initially amorphous, a tetragonal HfO₂ should be formed at the initial nucleation step. Therefore, the ramp-up time ($\tau\uparrow$) was set to be rather fast in this study (2.5 sec/100K). Since the holding-time should be related to the growth period or the phase transition process, it was fixed to 10 s. The up and down rate $\tau\uparrow\downarrow$ was defined by (sec/100°C).

Fig. 4(a) and (b) show XRD patterns for un-doped HfO₂ and 1.7 % (Y/Hf ratio) Y-HfO₂ as a function of $\tau\downarrow$ in PDA. It is merely dependent on $\tau\downarrow$ whether un-doped HfO₂ shows FE or not (data not shown). In case of (b) 1.7 % Y-HfO₂, only a slight structural phase transformation is seen in a long $\tau\downarrow$. In any $\tau\downarrow$, FE features were clearly observed. To verify this trend, $2P_r$ is plotted as a function of the monoclinic phase portion r_m in Fig. 5, with r_m derived from the XRD peak intensity at 2 θ ~30° as

$$r_m = \frac{I_{m(-111)} + I_{m(111)}}{I_{m(-111)} + I_{m(111)} + I_{o/t/c(111)}}$$

P_r is simply determined by r_m irrespective of doping concentration or $\tau\downarrow$. In addition, $2P_r$ can be plotted as a function of $\tau\downarrow$ for three HfO₂ films with different Y content (un-doped, 1.0 %, and 1.7 % Y/Hf ratio). With the increase in Y%, less dependence of P_r on $\tau\downarrow$ is observed in Fig. 6. More interestingly, an extrapolated value of P_r to $\tau\downarrow=0$ seems to be a given value irrespective of Y%. Namely, P_r value is the same for different doped case, if the perfect quenching could be carried out. With the increase in $\tau\downarrow$, P_r in Y:1.7 % HfO₂ does not change in the experimental time range, while that in un-doped case is significantly degraded. This fact suggests that amount of Vo's in HfO₂ at 600°C is the same both doped and un-doped HfO₂ films. The dopant, Y in the present case, just stabilizes the FE phase structure in PDA even with a long $\tau\downarrow$, while un-doped HfO₂ cannot maintain FE phase and relaxes to monoclinic one in a long $\tau\downarrow$ at room temperature (Fig. 7). Fig. 8 shows the energy landscape in which the energy barrier in the structural transformation from FE to monoclinic phase is decreased in the long $\tau\downarrow$. The impact of dopants and quenching is indicated by the red dotted line. Vo's in undoped HfO₂ can

diffuse and interact with the film, causing a structural relaxation, while dopants are at a stable position within HfO₂ and stabilizing Vo. Thus, a dopant serves as the Vo stabilizer.

INTERFACIAL EFFECTS ON FE-HfO₂

It is naturally expected that electrode may affect the FE formation of HfO₂. The bottom and top electrodes are important for FE-capacitor and for FE-FET applications, respectively. **Fig. 9** gives an example of bottom electrode effects (Pt, TiN, and p⁺-Ge) on HfO₂ with top Au. In this study, un-doped HfO₂ films were used to highlight possible electrode effects. No FE properties are observed on Pt, while clear FE on TiN and p⁺-Ge. In addition, top electrode difference between nitrides and oxides on structural phase of HfO₂ were compared in XRD. There was very little difference of HfO₂ phase in PMA among Al₂O₃, TiO₂ and SiO₂, and AlN, TiN and TaN (data not shown), in spite of a big difference from HfO₂ with no top-layer in PDA. Hence, a chemical reaction at the top interface is primary not a matter, but that suppressions of the structural relaxation and the out- or in-diffusion of oxygen in PMA are important for keeping FE properties stable.

For particular application such as ferroelectric tunnel junctions (FTJs), the electrode selection is essential. HfO₂ on highly-doped Ge shows clear FE properties (**Fig. 9**), and no interlayer is detected in XTEM (**Fig. 10**). In case of Si electrode, SiO_x interlayer is inevitably formed due to its thermodynamic stability. This means a metal/FE-HfO₂/p⁺-Ge may be the most viable choice for FTJs. In fact, very low voltage FTJ operation was demonstrated [11], as shown in **Fig. 11**. The on/off ratio of 10 at $V_{\text{read}}=0.2$ V and $V_{\text{write}}=1.6$ V in 2.7-nm-thick FE-HfO₂ is very promising. The electrode on HfO₂ should be selected for applications.

COERCIVE FIELD AND POLARIZATION SWITCHING KINETICS IN FE-HfO₂

For most FE-HfO₂ based devices currently in discussion: FE capacitors, FE-FETs including NC-FETs, and FTJs, a detailed understanding of FE-switching is mandatory. The most significant factor is the coercive field (E_C).

First, E_C is plotted for different dopants in HfO₂, again as a function of the monoclinic portion in **Fig. 12**. Higher E_C values are present for trivalent dopants in contrast to the tetravalent dopants like Ge, Si and Zr. This fact makes tetravalent dopants more favorable for a device application. In addition, it is interesting to remember that those doped-HfO₂ films exhibit anti-ferroelectric properties.

Next, E_C is plotted for different thickness in Ge-doped HfO₂ in contrast to PZT (as reported) in **Fig. 13**. At a physical thickness of 10 nm, E_C of HfO₂ is about a factor of 10 higher than that of PZT, resulting in a better retention behavior of HfO₂, but also a higher field necessary for polarization reversal. Here, an absolute value of E_C is reconsidered. According to the simple Landau formula, E_C is described as [14],

$$E_C = \pm \frac{1}{3\sqrt{3}} \left(\frac{P}{k\epsilon_0} \right)_{E=0}$$

By assuming $P_s \sim 30$ $\mu\text{C}/\text{cm}^2$ and $k \sim 30$, E_C is roughly 2 MV/cm, which is very reasonable for the experimental result. Since E_C value generally disagrees with the Landau theory, the sidewise domain growth was proposed and well fitted to experiments. In addition, note that a dramatic difference of the thickness (d) dependence is observed between HfO₂ and PZT. In HfO₂, rather no thickness dependence of E_C [6, 12], while $E_C \sim d^{-2/3}$ has been often reported in PZT [13]. This flat trend in HfO₂ has also been reported in literatures [6, 12]. Above results strongly suggest that the switching kinetics in HfO₂ may not be

described by the domain growth model (KAI:Kolmogorov-Avrami-Ishibashi model) [15]. Hence, another one, NLS (nucleation-limited-switching) model [16] is a candidate. **Fig. 14** schematically compares NLS one, in which the lateral domain growth (sidewise growth) is not relevant, with KAI model. Here, to further investigate the switching kinetics, the pulse I - V measurement for characterizing the switching kinetics was carried out by changing the frequency and amplitude [17]. **Fig. 15** represents the experimental results which are on a straight line in the semi-log plot both for HfO₂ and PZT. The results can be expressed as follows.

$$\frac{1}{\tau} = \frac{1}{\tau_0} \exp\left(-\frac{\gamma}{E_C^2}\right) \quad \gamma: \text{constant.}$$

It means that for both FE-HfO₂ and thin poly-PZT, NLS behavior can be strongly suggested [18]. From the result in **Fig. 15**, a supply voltage of ~ 2.6 V is necessary for HfO₂ to switch $\sim 90\%$ of the FE domains in 10 ns. For longer switching times, lower voltages can be utilized. Thus, NLS and high E_C in FE-HfO₂, are for and against the fast switching under a power supply voltage, respectively. NLS also implies that a voltage below E_C would finally lead to a stochastic switching event in scaled devices [19]. This is beneficial for neuromorphic FE-FET applications, while it needs to be taken into account for read/write disturbs in a memory array.

Almost no interlayers are present in FE capacitors and the supply voltages below 3V are possible. On the other hand, for MIS or SIS based FE-FETs, supply voltages should be enhanced due to an additional voltage drop at the interlayer(s). **Fig. 16(a)** shows an allowable voltage area for FE switching below the breakdown field (E_{BD}) and above E_C in case of no interlayer, and **(b)** displays an overview of reported and measured switching voltages for FE-FETs. Thinner IL is obviously more needed for a low power supply voltage operation. Thus, in FE-FET applications, the interlayer engineering is especially a key to the low power operation, in addition to FE properties.

CONCLUSION

Material fundamentals of FE-HfO₂ and polarization switching kinetics in FE-HfO₂-based devices have been discussed. FE-HfO₂ is the most promising for new memory and logic applications. However, high E_C is intrinsically challenging to high-speed operation under a low supply voltage, while NLS kinetics is favorable. Thus, the dopant selection should be made in terms of lowering E_C . On the other hand, the minor-loop application in FE properties will be attractive for neuromorphic applications, in addition to the fact that novel FE memory technologies are charming toward the future.

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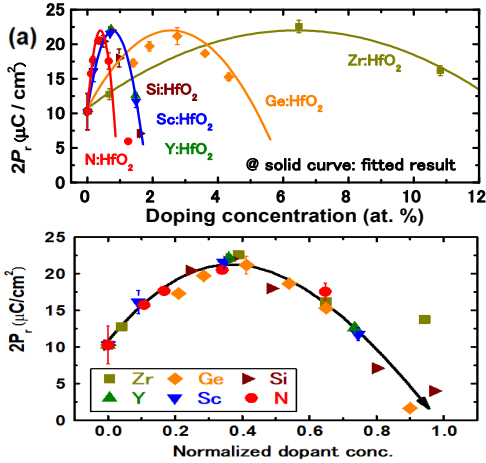


Fig.1 $2P_r$ as a function of the concentration of each dopant, in which it should be noted the concentration is taken for the total atom number. Although this result has been reported in ref. 10, it is put here because of the starting point of this work.

Fig.2 $2P_r$ as a function of normalized concentration to fit a universal parabolic line. The normalization factor, α , for the dopant concentration is fixed to be 0.5 for Sc, because Sc³⁺ has almost the same ionic radius as Hf⁴⁺.

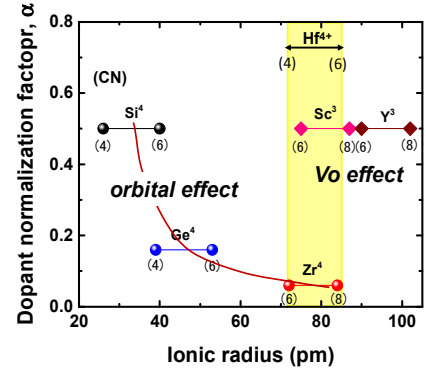


Fig.3 α is plotted as a function of ionic radius. Two kinds of trends are seen for FE stabilization. The number in (n) represents the coordination number, n , of each atom.

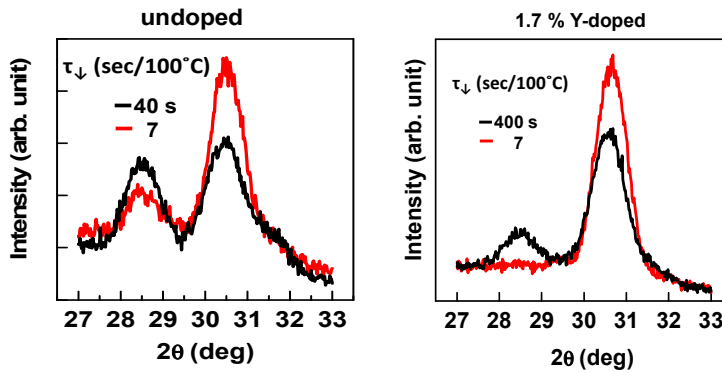


Fig.4(a) XRD patterns in un-doped HfO₂ films with two typical τ_{\downarrow} . In the un-doped case, the peak intensity changes dramatically by changing τ_{\downarrow} from 7 (red) to 40 s (black).

Fig.4(b) XRD patterns in Y(1.7%)-doped HfO₂ samples with two typical τ_{\downarrow} . The monoclinic phase gradually but slowly increases with the increase in τ_{\downarrow} . Note that τ_{\downarrow} (black) is 10x longer than that in undoped case.

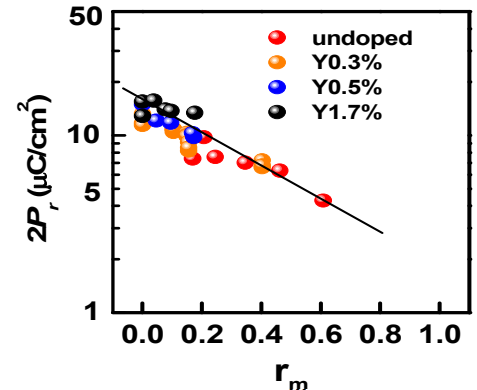


Fig.5 $2P_r$ - r_m relationship in both un-doped and Y-doped HfO₂ for various τ_{\downarrow} . In any cases, $2P_r$ is well described by r_m . Although this trend is not physically clarified yet, it empirically holds good for samples evaluated so far. ($1-r_m$) corresponds to the symmetric phase, which includes orthorhombic phase.

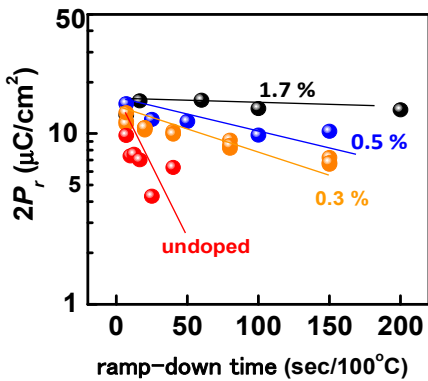


Fig.6 $2P_r$ as a function of τ_{\downarrow} defined by sec/100 C for three kinds of Y-doped and un-doped HfO₂. Note that extrapolated $2P_r$ values converge onto one point irrespective of dopant concentration including un-doped case. This fact indicates that dopant can stabilize the FE phase structure, which is quenched by quick ramp-down process as well.

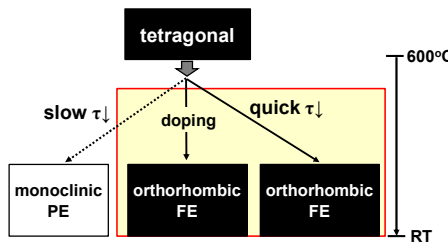


Fig.7 Schematic view of structural phase transformation in HfO₂ under the ramp-down process in PDA. Tetragonal phase at 600°C is not formed by the doping but by intrinsic properties in thin HfO₂. Since the tetragonal-monoclinic transformation is associated with a shear deformation, the dopant suppresses the deformation with the help of Vo or bonding-orbital force effect.

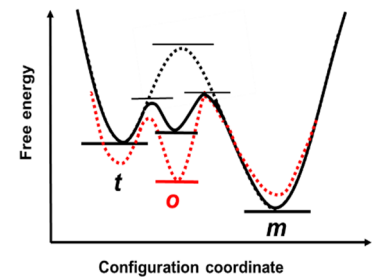


Fig.8 Schematically depicted energy landscape in structural phase transformation process of HfO₂. A solid-black line denotes un-doped HfO₂ case with the normal PDA, while in red-dotted one shows doped or quick ramp-down case. A broken-black line shows the classical picture of the transition from tetragonal to monoclinic. It is inferred that FE-phase is at metastable state with a finite energy barrier from FE- to ground state.

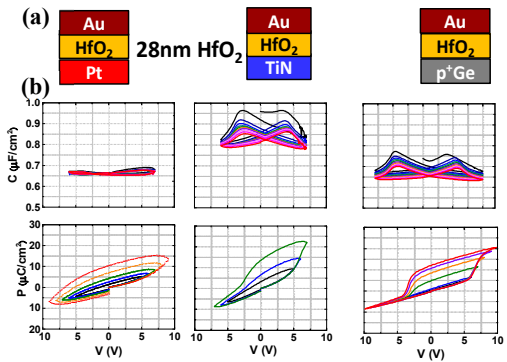


Fig.9 (a) Schematic views of three kinds of bottom electrode structures (Pt, TiN and p⁺-Ge) in un-doped HfO₂. (b) C-V and P-V characteristics in each stack. PDA was at 600°C. Repetitive C-V and V-swing increase in P-V are demonstrated.

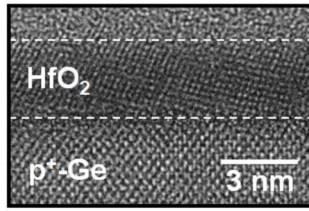


Fig.10 X-TEM image of p⁺-Ge/HfO₂ after PDA at 600°C. No interlayer is formed on p⁺-Ge. All the results in Fig. 1 was obtained on p⁺-Ge. In case of Si, SiO_x interlayer is inevitably formed. This will be a big advantage of Ge devices

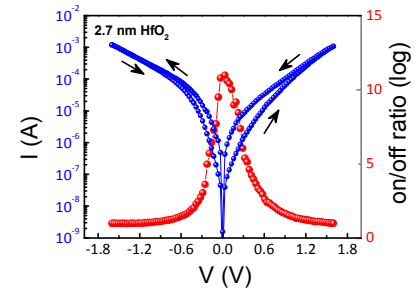


Fig.11 I-V characteristics in TiN/Y-HfO₂/p⁺-Ge junction. HfO₂ thickness was 2.7 nm and P_r was ~4 μC/cm². 1.6 V writing, 0.2 V reading proved the on/off ratio ~10. No area dependence was observed. In Ge electrode, a big difference of electron density between top and bottom electrodes is very advantageous for FTJs, in addition to no interlayer formation.

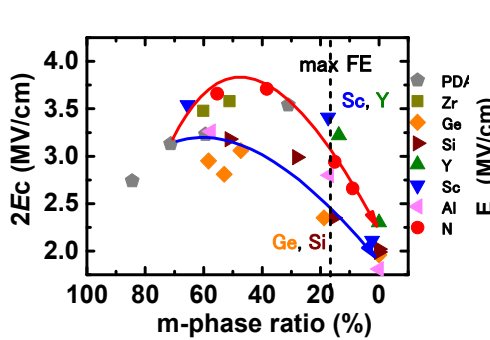


Fig. 12 2E_c is plotted as a function of r_m x100 % for many dopants. Two trends are seen. The results may suggest that one (red line) is Vo-driven FE, while the other (blue line) is bonding orbital-driven one.

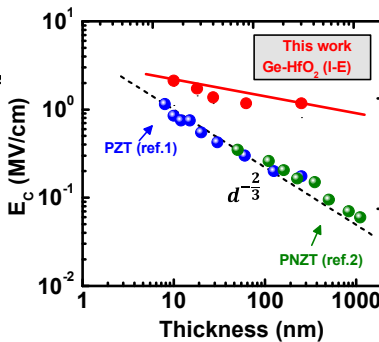


Fig. 13 Thickness dependence of E_c both for Ge-doped HfO₂ and PZT (ref.1, N. Pertsev et al., *APL* **83**, 3356 (2003); ref.2, C. Bjormander et al., *APL* **64**, 2493 (1994)). E_c in PZT is on a line with the power law of oxide thickness (d), while E_c in HfO₂ looks rather flat.

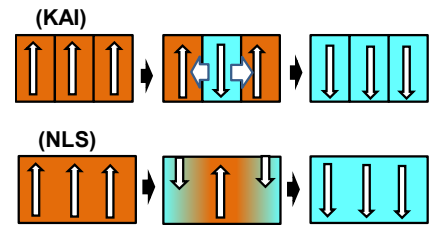


Fig. 14 Two kinds of polarization switching model. The upper is KAI model, in which the lateral domain-wall growth eventually changes the polarization direction through the film, while the lower is NLS one, in which the domain growth is not involved. In KAI model, a rather long time to switch the polarization is a matter of concern.

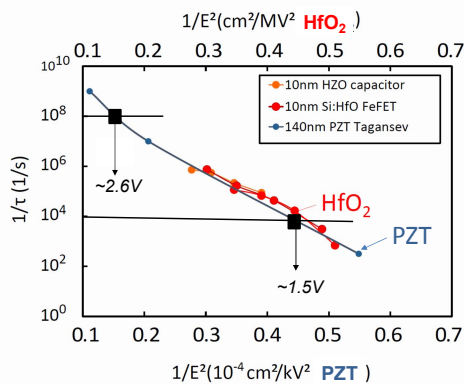


Fig. 15 Polarization switching analysis using the pulse I-V measurement for two kinds of HfO₂ samples, and poly-PZT. Experimental details are found in ref. [17]. A linear relationship between 1/τ and 1/E² suggests that NLS is a proper model, in which τ and E are the pulse width and applied electric field, respectively. Note a unit difference of the horizontal axis E between HfO₂ and poly-PZT (a scale difference between the top and bottom axes). From this trend, ~2.6 V and ~1.5 V are needed for 10 ns and 100 μs polarization switching.

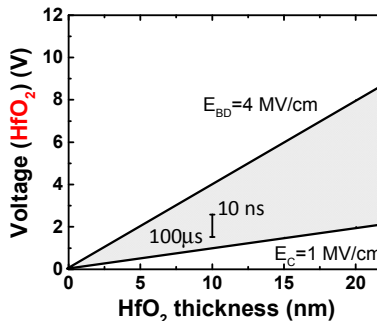


Fig.16 Thickness-voltage relationship for HfO₂ FE-FETs (for 1Tr memory). (a) Two fixed oxide electric field lines (1 and 4 MV/cm) are shown for E_c and E_{BD}. Voltage applied on HfO₂ should be in this triangle area. Reliability, memory window and actual voltage swing are estimated in this figure. 2.6 V (10 ns) and 1.5 V(100 μs) operating points from Fig. 15 are shown in the plot. (b) Data points are from literatures, which include the finite interlayers (IL). (GF IEDM 2015, UCB EDL 2017, IMEC IEDM 2018, Xiangtan JEDS 2019). (#) denotes IL thickness. Thinner IL allows a lower total voltage, while thicker HfO₂ is better for achieving a larger memory window. Thus, when the power supply voltage is fixed to be low, the interlayer engineering is critical for keeping the high-speed polarization switching.