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# Magnetism and Thermomechanical Properties in Si Substituted MnCoGe Compounds

Abdul Rashid Abdul Rahman <sup>1,\*</sup>, Muhamad Faiz Md Din <sup>1,2,\*</sup>, Jianli Wang <sup>2,3</sup>, Nur Sabrina Suhaimi <sup>1</sup>, Nurul Hayati Idris <sup>4</sup>, Shi Xue Dou <sup>2</sup>, Mohammad Ismail <sup>4</sup>, Muhammad Zahir Hassan <sup>5</sup> and Mohd Taufik Jusoh <sup>1</sup>

- <sup>1</sup> Department of Electrical & Electronic Engineering, Faculty of Engineering, National Defence University of Malaysia, Kuala Lumpur 57000, Malaysia; sabrina15.eec@gmail.com (N.S.S.); taufik@upnm.edu.my (M.T.J.)
- Institute for Superconductivity and Electronic Materials, University of Wollongong, Wollongong, NSW 2522, Australia; shi\_dou@uow.edu.au
- <sup>3</sup> Bragg Institute, Australian Nuclear Science and Technology Organization, Lucas Heights, NSW 2234, Australia; jianli@uow.edu.au
- <sup>4</sup> School of Ocean Engineering, Universiti Malaysia Terengganu, Kuala Terengganu 21030, Malaysia; nurulhayati@umt.edu.my (N.H.I.); mohammadismail@umt.edu.my (M.I.)
- Fakulti Teknologi Kejuruteraan Mekanikal dan Pembuatan, Universiti Teknikal Malaysia Melaka, Durian Tunggal 76100, Melaka, Malaysia; zahir@utem.edu.my
- \* Correspondence: rashidrahman2201@gmail.com (A.R.A.R.); faizmd@upnm.edu.my (M.F.M.D.)

**Abstract:** MnCoGe-based compounds have been increasingly studied due to their possible large magnetocaloric effect correlated to the magnetostructural coupling. In this research, a comprehensive study of structure, magnetic phase transition, magnetocaloric effect and thermomechanical properties for MnCoGe<sub>1-x</sub>Si<sub>x</sub> is reported. Room temperature X-ray diffraction indicates that the MnCoGe<sub>1-x</sub>Si<sub>x</sub> (x = 0, 0.05, 0.1, 0.15 and 0.2) alloys have a major phase consisting of an orthorhombic TiNiSi-type structure with increasing lattice parameter b and decreasing others (a and c) with increasing Si concentration. Along with M-T and DSC measurements, it is indicated that the  $T_c$  value increased with higher Si concentration and decreased for structural transition temperature  $T_{\text{str.}}$ . The temperature dependence of the magnetization curves overlaps completely, indicating that there is no thermal hysteresis, and it is shown that the transition is the second-order type. It is also shown that the decreased magnetization on the replacement of Si for Ge decreases the value of  $-\Delta$ S<sub>M</sub> from  $-\Delta$ S<sub>M</sub>~8.36 J kg<sup>-1</sup> K<sup>-1</sup> at x = 0 to  $-\Delta$ S<sub>M</sub>~5.49 J kg<sup>-1</sup> K<sup>-1</sup> at x = 0.2 with 5 T applied field. The performed Landau theory has confirmed the second-order transition in this study, which is consistent with the Banerjee criterion. The magnetic measurement and thermomechanical properties revealed the structural transition that takes place with Si substitution of Ge.

**Keywords:** thermomechanical properties; second-order magnetic transition; magnetocaloric effect; MnCoGe; magnetic refrigeration

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# 1. Introduction

New technology of magnetic refrigeration (MR) that was recently developed has been an alternative to conventional gas compression (CGC). This shows that this new cooling technology is more effective and environmentally friendly. MR is based upon the magnetocaloric effect (MCE), where the adiabatic temperature changes the material upon the application of a magnetic field [1]. Highly advanced industries require materials that come with a large MCE near to room temperature. Hence, immense efforts have been made to meet these rapidly growing requirements for the past few decades. Due to the structural behavior and strong magnetic coupling, it is clear that the magnetostructural transition (MST) of the first order produces a large MCE. A lot of the ongoing studies on magnetic refrigeration are mainly focused on perovskite manganites [2], Gds(SixGe1-x)4 [3], MnFeP1-xAsx [4], La(Fe1-xSix)13 [5], NiMn-based Heusler alloys [1], and MM'X alloys (M

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and M' denote 3D transition elements and X denotes main group elements) [6–8]. The intermetallic MnCoGe-based compound is a typical member of the MM'X family that maintains sufficient physical contents and application prospects that are promising. It has a double crystalline structure, TiNiSi-type orthorhombic structure and Ni2In-type hexagonal structure [8]. It was previously found that MnCoGe<sub>1-x</sub>Si<sub>x</sub> offers the best affordable magnetic material compound while still providing a magnetic entropy change and decent refrigeration capacity [9]. This finding is interesting as the compound can provide a pleasing magnetocaloric effect at a lower price for a magnetic refrigeration material.

This work has investigated the thermomechanical properties of  $MnCoGe_{1-x}Si_x$  compounds to understand the essence of the magnetic transition in  $MnCoGe_{1-x}Si_x$ . From the study, the structural change and magnetic phase transition were effectively shifted from a suitable Si concentration into the temperature region of interest. This led to achieving high addition of the GMCE. It is also found that the unit cell volume and lattice parameters of all alloys were found to reduce with the increase in Si content. In addition, the peaks shifted to a higher angle due to a lower atomic radius of Si,  $\sim 1.32$  Å, whereas Ge,  $\sim 1.37$  Å.

## 2. Experimental Details

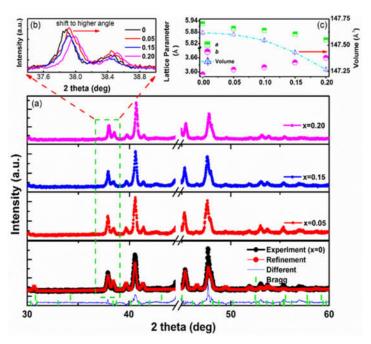
In the preparation of a batch of MnCoGe<sub>1-x</sub>Si<sub>x</sub> compounds (with x = 0-0.2), sufficient amounts of Mn (99.9%), Ge (99.999%) powder, Co (99.9%), and Si (99.999%) chips were arc melted in the vicinity of an argon atmosphere. Approximately 3% of Mn above the stoichiometric amount of the starting materials is needed. This is important for the evaporation that occurs in order to reimburse for the weight lost throughout the melting process. In order to attain a good homogeneity of the samples, the ingots were melted 5 times in a row. They were then closed in a quartz tube and hardened at a temperature of 900 °C for a duration of 120 h. The samples were then characterized by X-ray diffraction (XRD) (Cu K5-007 radiation, 5-007 = 15,418 Å). They were then analyzed by Rietveld refinement analysis by using the Fullprof program. The phase transition temperatures were analyzed using TA instrument Q100 Differential Calorimeter (DSC) (TA Instruments, New Castle, DE, USA) at room temperature. A Quantum Model P650 14 T Physical Property Measurement System (PPMS) (Quantum Design, San Diego, CA, USA) was used to measure magnetization using the option of a vibrating sample magnetometer. This was achieved in a temperature range of 10-330 K and up to 0.01 T of the applied fields. The thermomechanical measurement was conducted in a Perkin Elmer 8000 dynamic mechanical analyzer (DMA) (Perkin Elmer, Waltham, MA, USA) in the tension mode. The heating rate for the measurement was set at 278 K/min.

### 3. Results and Discussion

#### 3.1. Structure Properties

The X-ray diffraction patterns of the MnCoGe<sub>1-x</sub>Si<sub>x</sub> (x = 0, 0.05, 0.1, 0.15, and 0.2) alloys are shown in Figure 1a. The XRD patterns show that the MnCoGe<sub>1-x</sub>Si<sub>x</sub> alloy has a major phase comprised of an orthorhombic TiNiSi-type structure with no reflection to a hexagonal Ni<sub>2</sub>In-type structure. This is observed with an increase in the Si concentration up to x = 0.2. All diffraction angles (20) shifted to the higher angle with the increase in Si concentration. A gradual displacement to the right can be seen for each XRD result (see Figure 1b).

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**Figure 1.** (a) Refined XRD pattern of MnCoGe<sub>1-x</sub>Si<sub>x</sub> (x = 0.05, 0.10, 0.15, and 0.20) measured at room temperature. (b) The inset shows the patterns for the indicated range of the MnCoGe<sub>1-x</sub>Si<sub>x</sub> samples, representing that the Bragg peaks shifted to the higher angles. (c) Rietveld refinement analysis indicating the unit cell volumes and lattice parameters.

The Rietveld refinement analysis indicated in Figure 1c showed that the lattice parameter b increases from 3.592 Å at x = 0 to 3.664 Å at x = 0.2. However, the lattice parameters a and c were found to decrease at a (5.925 Å at x = 0 to 5.843 Å at x = 0.2) and c (6.942 Å at x = 0 to 6.891 Å at x = 0.2), which resulted in the unit cell volume of all alloys decreasing at (147.62 ų at x = 0 to 147.27 ų at x = 0.2). As the Si content increases, the peaks seem to be shifted to the higher angle positions related to the lower atomic radius of Si,  $\sim 1.32$  Å, compared to Ge,  $\sim 1.37$  Å. Their unit cell volumes and lattice parameters are correlated by  $a = a_{\text{ortho}} = c_{\text{hex}}$ ,  $b = b_{\text{ortho}} = a_{\text{hex}}$ ,  $c = c_{\text{ortho}} = \sqrt{3}a_{\text{hex}}$ , and  $V = V_{\text{ortho}} = 2V_{\text{hex}}$  [10]. With the increasing Si concentration, the lattice shrinks along the a ortho axis and expands along the a ortho axis, inducing a decrease in volume of the orth. These phenomena are also related to the rule of the valence electron. According to this, the main-group atoms (Si/Ge) occupy the 2c sites. In contrast, the transition-metal atoms with significantly less (Mn) and higher (Ni) valence electrons tend to occupy the 2a and 2d sites, respectively [11].

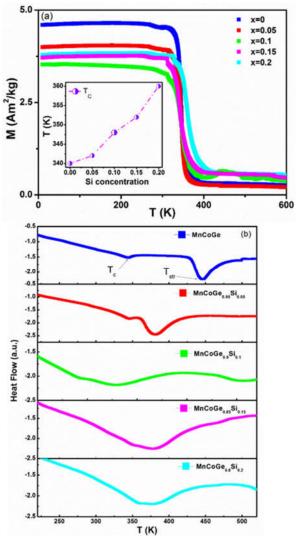
# 3.2. Magnetic Phase Transitions

The change in the distance between the magnetic atoms is well known to directly alter the coupling exchange, resulting in a shift in the temperature transition [12]. The Curie temperature increases and decreases in accordance with variations in the lattice parameter b, suggesting that the interaction of the Mn–Mn moments along this axis is significant in defining the Curie temperature. Figure 2a presents the temperature dependence of the magnetization of MnCoGe<sub>1-x</sub>Si<sub>x</sub> alloys measured in the field of 0.01 T from 10–600 K. In the case of compositions x = 0.00 to 0.2, it was found that the Curie temperature slightly increases (340 K–360 K) with the increase in the Si concentration, as indicated in the inset in Figure 2a. These have shown the same pattern with the previous report on these materials with a different composition, where the maximum Curie temperature reported is  $T_c = 374$  K [9]. In addition, the transition for compounds x > 0.05 was deemed to be incomplete and not reaching the zero moment, as demonstrated by the fact that the structural transition temperatures were higher than 430 K. Consequently, this behavior is thought to be related to the necessity for a lower Si atomic radius in order to change the

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symmetry of the TiNiSi-type orthorhombic structure to the hexagonal Ni<sub>2</sub>In-type structure at a higher temperature.

In addition, the temperature dependence of the magnetization of MnCoGe<sub>1-x</sub>Si<sub>x</sub> was measured for heating and cooling in order to determine the thermal hysteresis characteristic of the first-order transition [13,14]. The cooling and heating curves overlap completely, indicating that there is no thermal hysteresis, and it is shown that the transition is the second-order type.



**Figure 2.** (a) M–T curves measured for  $MnCoGe_{1\rightarrow x}Si_x$  (x = 0.05, 0.1, 0.15, and 0.2). The inset shows the pattern of Curie temperature with an increased Si concentration. (b) DSC curves measured ranging from 300 to 500 K. The arrows indicate the Curie temperature structural transition.

The magnetic moment, structural transition temperature  $T_{\rm str}$ , cell volume, and Curie temperature are all strongly dependent on the concentration of Co vacancy rather than on Ge vacancy [15]. However, Figure 2b shows that the structural transition  $T_{\rm str}$  determined by the DSC measurements decreased from 450 K for x = 0 to 360 K for x = 0.2. This indicates that substituting Ge with a minor percentage of Si (20%) led to a drop in the structural transition temperature and a slight increase in the Curie temperature.

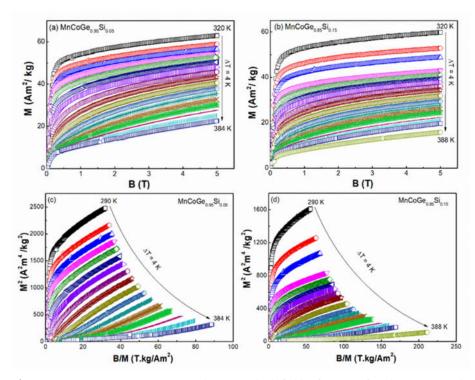
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## 3.3. The Magnetocaloric Effect

Figure 3a,b show the magnetization curves of MnCoGe0.95Si0.05 and MnCoGe0.85Si0.15 for fields within the range B = 0–5 T near the Curie temperatures with 4 K interval steps. It is apparent that the results are overlapped for increasing and decreasing fields of the compounds. This finding suggests that there are no observable magnetic hysteresis loss effects for the increase in Si content in MnCoGe1-xSix compounds. In addition, Figure 3c,d represent the Arrott plots ( $M^2$  versus B/M) for MnCoGe0.95Si0.05 and MnCoGe0.85Si0.15, which were found to exhibit features by the Banerjee criterion of a second-order transition for all samples. Thus, the results show that the ferromagnetic transition at  $T_C$  in MnCoGe1-xSix is a second-order transition to the substitution of Si for Ge for Si content up to x = 0.2.

For the whole range of MnCoGe<sub>1-x</sub>Si<sub>x</sub> compounds (x = 0–0.2) of their magnetization curves, the magnetic entropy change,  $-\Delta$ S<sub>M</sub>, has been calculated to increase and decrease field values as a function of the temperature and magnetic field ( $\Delta$ B = 0–5 T). The increase in magnetic entropy was obtained by applying the standard Maxwell relation [16]:

$$-\Delta S_{\rm M}(T,H) = \int_0^H \left(\frac{\partial M}{\partial T}\right)_H (dH) \tag{1}$$



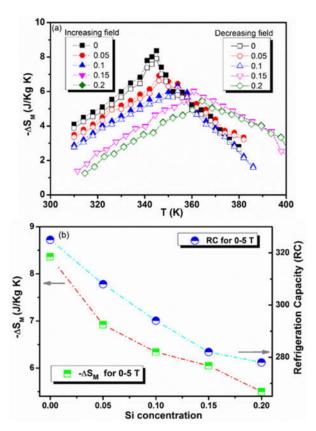
**Figure 3.** Magnetization curves with an applied field of 0–5 T for the compounds of (a) MnCoGeo.95Sio.05 and (b) MnCoGeo.85Sio.15. The second-order transition Arrott plot types for both compounds are shown in (c,d).

Figure 4a shows that the  $-\Delta SM$  peak moderately widens towards the higher temperatures with an increase in the magnetic field. As a result, it is observed that the compounds hold a behavior of a field-induced transition from a paramagnetic to a ferromagnetic state. The variations in magnetic entropy for MnCoGe<sub>1-x</sub>Si<sub>x</sub> alloys (x = 0.05, 0.1, 0.15, and 2.0) around the ferromagnetic temperatures have been presented from increasing and decreasing applied fields (with slightly no hysteresis loss). This is to indicate the feasibility of different experimental and correlated analytical approaches to establish the isothermal entropy change [17]. The values of the entropy at the corresponding Curie temperatures indicated in Figure 4b are slightly decreased, with  $\Delta B = 0.5 \text{ T} : -\Delta S_M \sim 8.36 \text{ J kg}^{-1} \text{ K}^{-1}$  at Tc=

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340 K at MnCoGe;  $-\Delta S_M \sim 6.92$  J kg<sup>-1</sup> K<sup>-1</sup> at Tc = 342 K at MnCoGe<sub>0.95</sub>Si<sub>0.05</sub>;  $-\Delta S_M \sim 6.33$  J kg<sup>-1</sup> K<sup>-1</sup> at Tc = 348 K at MnCoGe<sub>0.90</sub>Si<sub>0.10</sub>;  $-\Delta S_M \sim 6.04$  J kg<sup>-1</sup> K<sup>-1</sup> at Tc = 352 K at MnCoGe<sub>0.85</sub>Si<sub>0.15</sub>;  $-\Delta S_M \sim 5.49$  J kg<sup>-1</sup> K<sup>-1</sup> at Tc = 360 K at MnCoGe<sub>0.80</sub>Si<sub>0.20</sub>. The peak  $|\Delta S_M|$  value of MnCoGe<sub>0.95</sub>Si<sub>0.05</sub> is higher than the  $|\Delta S_M|$  value reported in Mn<sub>0.9</sub>Ti<sub>0.1</sub>CoGe ( $-\Delta S_M \sim 4.2$  J kg<sup>-1</sup> K<sup>-1</sup>) [18] and MnCoGe<sub>0.97</sub>Al<sub>0.03</sub> ( $-\Delta S_M \sim 3.5$  J kg<sup>-1</sup> K<sup>-1</sup>) [19]. Decreased magnetization on the replacement of Si for Ge similarly decreases the value of  $-\Delta S_M$ , which is associated with the interaction between Mn–Mn moments.

Moreover, the fundamental requirement for magnetic refrigeration does not only rely on a large magnetic entropy change, but also refrigeration efficiency evaluation such as the refrigeration capacity (RC) also needs to be considered [20]. The RC indicates the amount of heat transmitted during one thermodynamic cycle. Substituting Ge with Si leads to a decrease in RC from 325 at x = 0 to 278 at x-0.2. This behavior related to replacing Si on Ge leads to a change in the interplanar crystal distance. The generated 3D-band can occasionally minimize the electron splitting exchange interactions such that it decreases the value of  $\Delta$ SM and RC values.



**Figure 4.** (a) Temperature dependence of the isothermal magnetic entropy change,  $\neg \Delta S_M$ , (T, H) for MnCoGe<sub>1-x</sub>Si<sub>x</sub> compounds (x = 0, 0.05, 0.10, 0.15, 0.20) calculated from magnetization isotherms with 0–5 T field change applied. Increasing fields are denoted by closed symbols, while decreasing fields are denoted by open symbols. (b) The RC and  $\neg \Delta S_M$  of MnCoGe<sub>1-x</sub>Si<sub>x</sub> compounds for 0–5 T.

# 3.4. Thermomechanical Testing

Figure 5 depicts the behavior of the storage modulus (E') and the loss modulus (E''), which are the real and imaginary elements of the complex modulus during tension mode testing. The elastic and viscous behavior of a viscoelastic sample is represented by the storage and loss moduli [21]. It was found that the values of E' for all samples exhibited a gradual decrease. The value of E' for MnCoGe undergoes a considerable decrease after 437 K. However, in the case of MnCoGe<sub>1-x</sub>Si<sub>x</sub>(x = 0.05), the values of E' exhibited this nature after 375 K, which is the point of the magnetostructural transformation of this compound. The

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transition is due to the glass transition that occurred at the glass transition temperature (Tg). The glass transition temperature is when amorphous polymers transform from a glassy to a rubbery form due to the increased mobility of molecular segments and an increase in the free volume. The values of E" against T also revealed a similar pattern, with the E" peak recorded at 462 K at the magnetostructural transition region for MnCoGe. In contrast, the E" peak is observed at 400 K for the case of MnCoGe1-xSi $_x$ (x = 0.05).

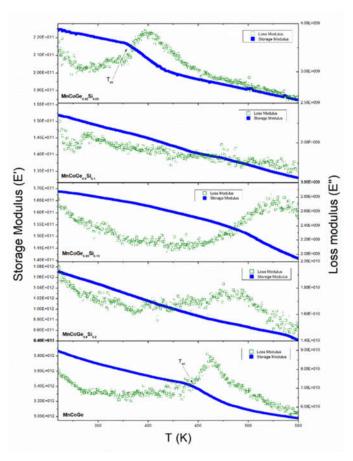
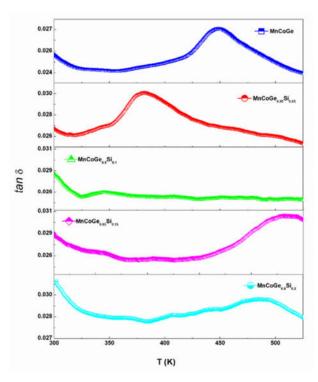


Figure 5. Storage modulus (E') and loss modulus (E") measured for MnCoGe<sub>1-x</sub>Si<sub>x</sub> (x = 0, 0.05, 0.1, 0.15, and 0.2).

The loss ratio (tangent) tanð, which quantifies the phase angle between the sample strain and the stress throughout the periodic excitation of the samples, is shown in Figure 6. The values of tanð are to measure the energy dissipative abilities of a material [22]. It is observed that the values of tanð gradually increase until ~375 K for MnCoGe1-xSix (x = 0.05) and ~450K for MnCoGe. These values of tanð increase to 375 K and 450 K related to the two dissipative mechanisms, which are the dissipative effect due to the twin boundary movement inside the alloy particles and the dissipative effects at the alloy interface. A substantial increase in the value of tan  $\delta$  for MnCoGe1-xSix (x = 0.05) is visible at a temperature above 375 K. Having stated that the value of tanð corresponds to the magnetostructural transition, it overcomes the values arising from the glass transition event of the matrix [21]. This behavior also shows the same scenario as DSC curves, where the substitution of Ge with Si leads to a drop in the structural transition temperature.

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**Figure 6.** Tan delta measured for MnCoGe<sub>1-x</sub>Si<sub>x</sub> (x = 0, 0.05, 0.1, 0.15, and 0.2).

#### 4. Conclusions

To conclude, the Si substitution for Ge atoms in MnCoGe<sub>1-x</sub>Si $_x$  compounds demonstrated the major structure as orthorhombic TiNiSi-type with no reflection on the hexagonal Ni2In-type structure with an increase in Si content up to x = 0.2. Furthermore, the Rietveld refinement analysis revealed that the lattice parameters (a and c) and unit cell volume of the compounds were identified to decrease with the increase in Si content, which is related to the small atomic radius of Si compared to Ge.

The magnetization and DSC measurements specified that with the increase in Si content, the  $T_{\rm C}$  value is slightly increased, even though the structural transition temperature,  $T_{\rm str}$ , is reduced. In addition, the structural transition temperatures for all compounds were significantly higher than 430 K, indicating that the transition was incomplete and did not achieve the zero moment.

The magnetic entropy change,  $-\Delta S_M$ , shows decrement behavior with increasing Si concentration up to x = 0.2 ( $\Delta B = 0.5$  T:  $-\Delta S_M \sim 8.36$  J kg<sup>-1</sup> K<sup>-1</sup> at MnCoGe;  $-\Delta S_M \sim 6.92$  J kg<sup>-1</sup> K<sup>-1</sup> at MnCoGe<sub>0.95</sub>Si<sub>0.05</sub>;  $-\Delta S_M \sim 6.33$  J kg<sup>-1</sup> K<sup>-1</sup> at MnCoGe<sub>0.90</sub>Si<sub>0.10</sub>;  $-\Delta S_M \sim 6.04$  J kg<sup>-1</sup> K<sup>-1</sup> at MnCoGe<sub>0.85</sub>Si<sub>0.15</sub>;  $-\Delta S_M \sim 5.49$  J kg<sup>-1</sup> K<sup>-1</sup> at MnCoGe<sub>0.80</sub>Si<sub>0.20</sub>) followed by the absence of any magnetic hysteresis loss, which is a second-order magnetic transition behavior and particularly advantageous to the magnetic refrigerator application. Moreover, substituting Ge with Si leads to a decrease in RC from 325 at x = 0 to 278 at x–0.2 as related to the replacement of Si on Ge, leading to a change in the interplanar crystal distance and the resulting 3D-band minimizing the electron splitting exchange interactions.

On top of that, the thermomechanical measurement shows that the values of  $\tan\delta$  increase to 375 K and 450 K with the increases in Si content related to the two dissipative mechanisms, which are the dissipative effect due to the twin boundary movement inside the alloy particles and the dissipative effects at the alloy interface.

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**Author Contributions:** A.R.A.R.: Writing—Original Draft, Visualization. M.F.M.D.: Writing—Review and Editing, Data Curation. J.L.W.: Conceptualization, Supervision. N.S.S.: Visualization. N.H.I.: Project administration. M.I.: Validation. M.Z.H.: Resources. S.X.D.: Resources. M.T.J.: Funding acquisition. All authors have read and agreed to the published version of the manuscript.

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