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## Supporting Information

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Escherichia coli $\boldsymbol{\beta}$-Galactosidase Inhibitors through Modifications at the Aglyconic Moiety: Experimental Evidence of Conformational Distortion in the Molecular Recognition Process

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chem_201203673_sm_miscellaneous_information.pdf


Figure S1: Coordinate scans of compound 1 (above) and compound 2 (below) fixing for the alternative chair conformation for the pentopyranose ring. Calculations were performed using the force field $O P L S \_2005$.

PAGE S2


Figure S2: Representation of proton distances in the coordinate scan of compound $\mathbf{2}$ with fixed ${ }^{4} \mathrm{C}_{1}$ conformation for the pentopyranose. $A, \mathrm{H}-1^{\prime}$ and $\mathrm{H}-4$ pentose; $B, \mathrm{H}-2^{\prime}$ and $\mathrm{H}-4$ pentose; $C, \mathrm{H}-1^{\prime} \mathrm{Gal}$ and $\mathrm{H}-5 \mathrm{eq}$ pentose; $D, \mathrm{H}-1^{\prime}$ Gal and $\mathrm{H}-3$ eq pentose; $E, \mathrm{H}-1^{\prime} \mathrm{Gal}$ and $\mathrm{H}-3 \mathrm{ax}$ pentose.

PAGE S3


Figure S3: Representation of proton distances of the coordinate scan of compound $\mathbf{2}$ with fixed ${ }^{1} \mathrm{C}_{4}$ conformation for the pentopyranose ring. Distances between: $A, \mathrm{H}-1^{\prime}$ Gal and H-4 pentose; $B, \mathrm{H}-2^{\prime}$ Gal and H-4 pentose; C, H-1' and H-5 pentose; $D, \mathrm{H}-1$ ' Gal and H-3ax pentose.

Table S1. Absolute STD fractions for compounds $\mathbf{1}$ or $\mathbf{2}$ with ligand:enzyme 100:1 molar ratio (20 millimolar phosphate buffer with $\mathrm{MgCl}_{2}$ at pH 7.2 at 298 K ). These values were employed to build the saturation curves of Figures 4B and 5B.

| Compound 1 | 0.5 second | 1.0 second | 1.5 seconds | 2.0 seconds | 2.5 seconds |
| :---: | :---: | :---: | :---: | :---: | :---: |
| aromatic | 0.21 | 0.37 | 0.45 | 0.54 | 0.54 |
| CH2 | not measured | not measured | not measured | not measured | not measured |
| CH2' | not measured | not measured | not measured | not measured | not measured |
| H1 ${ }^{\text {, }}$ | 0.23 | 0.37 | 0.45 | 0.5 | 0.51 |
| H1 | 0.25 | 0.35 | 0.47 | 0.49 | 0.52 |
| H5a | 0.27 | 0.38 | 0.47 | 0.52 | 0.51 |
| H4' | 0.33 | 0.48 | 0.54 | 0.56 | 0.56 |
| H5' ${ }^{\text {+ H6's }}$ | 0.27 | 0.4 | 0.44 | 0.47 | 0.48 |
| H3' | 0.12 | 0.16 | 0.17 | 0.17 | 0.14 |
| H2 | not measured | not measured | not measured | not measured | not measured |
| H2' | 0.16 | 0.21 | 0.24 | 0.25 | 0.24 |
| H5b | 0.24 | 0.33 | 0.4 | 0.42 | 0.41 |
| H4 | 0.27 | 0.38 | 0.4 | 0.45 | 0.45 |
| H3a | 0.33 | 0.41 | 0.45 | 0.48 | 0.49 |
| H3b | 0.36 | 0.4 | 0.45 | 0.49 | 0.51 |

PAGE S5

| Compound 2 | 0.5 second | 1.0 second | 1.5 seconds | 2.0 seconds | 2.5 seconds |
| :---: | :---: | :---: | :---: | :---: | :---: |
| aromatic | 0.19 | 0.22 | 0.3 | 0.36 | 0.4 |
| CH2 | not measured | not measured | not measured | not measured | not measured |
| H1 | not measured | not measured | not measured | not measured | not measured |
| CH2 | 0.11 | 0.13 | 0.2 | 0.19 | 0.24 |
| H1' | 0.11 | 0.14 | 0.21 | 0.21 | 0.24 |
| H5a | 0.09 | 0.14 | 0.16 | 0.2 | 0.2 |
| H2 | 0.08 | 0.16 | 0.18 | 0.22 | 0.26 |
| H4 ${ }^{\text {, }}$ | 0.1 | 0.16 | 0.22 | 0.25 | 0.27 |
| H5' + H6's | 0.07 | 0.1 | 0.17 | 0.18 | 0.18 |
| H3' | 0.07 | 0.11 | 0.13 | 0.13 | 0.15 |
| H5b | not measured | 0.1 | 0.13 | 0.16 | 0.17 |
| H2' | not measured | 0.17 | 0.22 | 0.27 | 0.29 |
| H4 | not measured | 0.18 | 0.21 | 0.2 | 0.2 |
| H3a | not measured | 0.1 | 0.16 | 0.17 | 0.18 |
| H3b | 0.15 | 0.15 | 0.17 | 0.18 | 0.18 |

PAGE S6


Figure S4. Comparison of STD pattern of non-reducing galactose from several compounds. STD values are normalized to the highest saturated proton of galactose residue. Compound $\mathrm{A}, \mathrm{B}$ and C are thiodisaccharides, with $\beta$-galactose in their non-reducing end, whose glycosidic bond is $\beta$-(1-4). Compound A - (2-Propyl 3-deoxy-4-S-( $\beta$-D-Galactopyranosyl)-4-thio- $\alpha-D-l y x o-h e x o p y r a n o s i d e) . ~$ Compound B - (2-Propyl 3-deoxy-4-S-( $\beta$-D-Galactopyranosyl)-4-thio- $\alpha-\mathrm{D}-\mathrm{xylo}$-hexopyranoside). Compound C - (2-Propyl 4-S-( $\beta$-D-Galactopyranosyl)-4-thio- $\alpha-D-$ gulopyranoside). ${ }^{[25]}$

PAGE S7


Figure S5: Trajectory of distance between galactose centroid and $\mathrm{Trp}^{567}$ indol centroid for compounds $\mathbf{1}$ and 2, with the pentopyranose in both alternative conformations ( 5000 steps of 2 picoseconds each).

PAGE S8


Figure S6: Trajectory of distance between protons H3a and H5a along the dynamic runs for compounds $\mathbf{1}$ and $\mathbf{2}$ with the conformation of the pentopyranose fixed as ${ }^{1} \mathrm{C}_{4}$ or ${ }^{4} \mathrm{C}_{1}(5000$ steps of 2 picoseconds each).

PAGE S9


Figure S7: Trajectory of the distances between pentopyranose protons in the ${ }^{1} \mathrm{C}_{4}$ conformation of $\mathbf{2}$ to the $\operatorname{Trp}{ }^{999}$ of $\beta$-galactosidase from E.coli, and molecular representation of the pentose residue (5000 steps of 2 picoseconds each).


Figure S8: Trajectory of the distances between pentopyranose protons in the ${ }^{4} C_{1}$ conformation of 2 to the $\operatorname{Trp}^{999}$ of $\beta$-galactosidase from E.coli, and molecular representation of the pentose residue (5000 steps of 2 picoseconds each).


Figure S9: Trajectory of distance between aromatic protons of compound $\mathbf{2}$ to the side chain of Phe ${ }^{512}$ and Phe ${ }^{601}$ (5000 steps of 2 picoseconds each). Aromatic ring tends to get closer to Phe ${ }^{512} ;{ }^{1} \mathrm{C}_{4}$ conformation keeps close and ${ }^{4} \mathrm{C}_{1}$ conformation moves away from $\mathrm{Phe}{ }^{601}$ and comes close to Phe ${ }^{512}$.

PAGE S11


Figure S10: View of the structure of compound $\mathbf{2}$ in the catalytic site of the $\beta$-galactosidase from $E$. coli at the end of the molecular dynamics. $A$ : the pentose is in the ${ }^{1} \mathrm{C}_{4}$ conformation; $B$ : the pentose is in the ${ }^{4} \mathrm{C}_{1}$.

