

CHEMISTRY

A European Journal

Supporting Information

© Copyright Wiley-VCH Verlag GmbH & Co. KGaA, 69451 Weinheim, 2014

N-Heterocyclic Carbene Formation Induced Fluorescent and Colorimetric Sensing of Fluoride Using Perimidinium Derivatives

Dawei Zhang,^[a, b] Haiqiang Yang,^[a] Alexandre Martinez,*^[b] Kelsey Jamieson,^[b]
Jean-Pierre Dutasta,^[b] and Guohua Gao^{*[a]}

chem_201404806_sm_miscellaneous_information.pdf

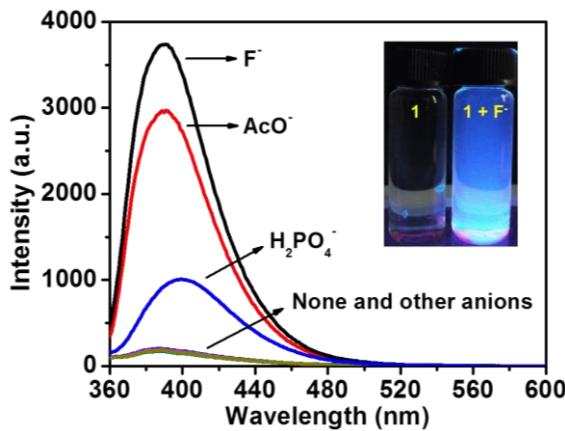


Figure S1 Fluorescence spectra of 5 μM chemodosisensor **1** excited at 353 nm upon addition of 3 equiv. of different anions (TBA salts) in DMSO. Inset: fluorescent color of chemodosisensor **1** (5 μM) in the absence and presence of TBAF under UV lamp excited at 365 nm.

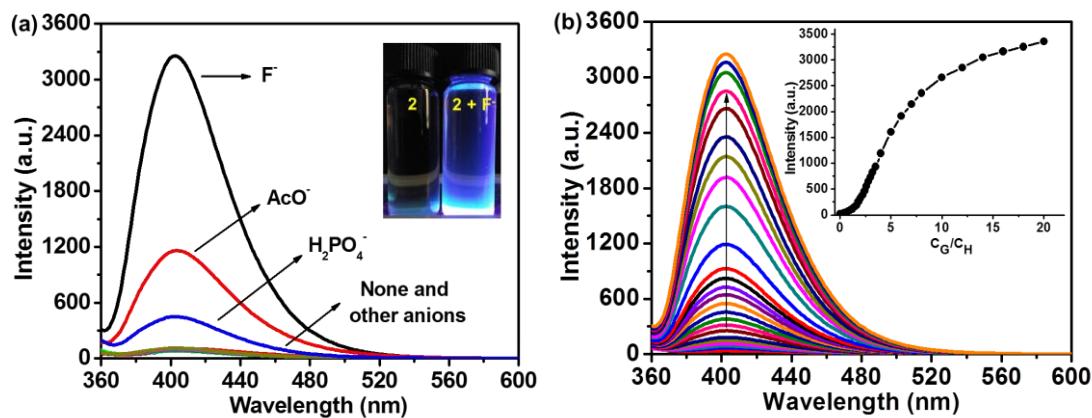


Figure S2 (a) Fluorescence spectra of 5 μM chemodosisensor **2** excited at 357 nm upon addition of 20 equiv. of different anions (TBA salts) in DMSO. Inset: fluorescent color of chemodosisensor **2** (5 μM) in the absence and presence of TBAF under UV lamp excited at 365 nm. (b) Fluorescence titrations of 5 μM chemodosisensor **2** with TBAF in DMSO. Inset: the fluorescence intensity at 402 nm as a function of the added TBAF.

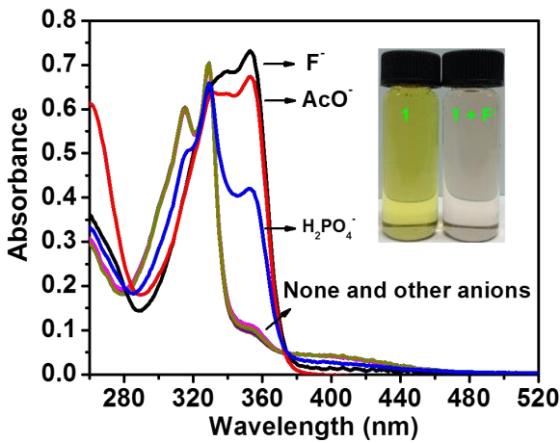


Figure S3 (a) Absorption spectra of 50 μ M chemodosisensor **1** upon addition of 3 equiv. of different anions (TBA salts) in DMSO. Inset: solution color of the chemodosisensor **1** in the absence and presence of TBAF.

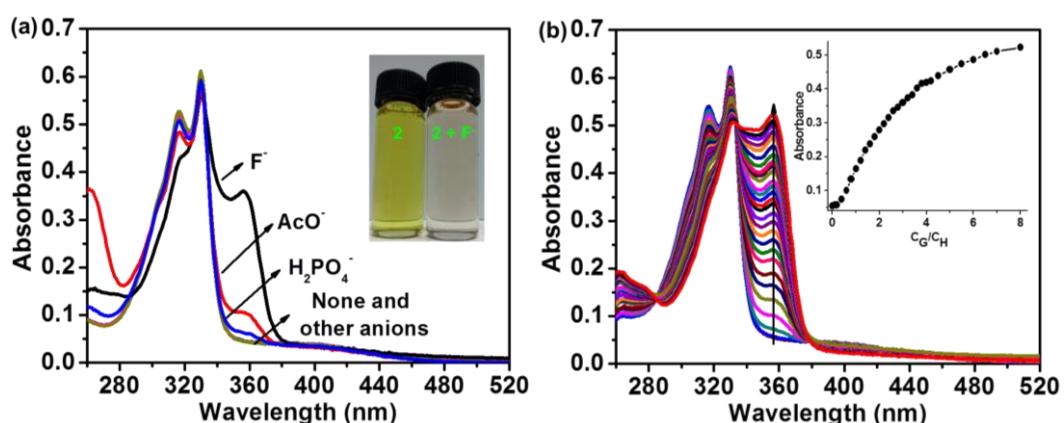


Figure S4 (a) Absorption spectra of 50 μ M chemodosisensor **2** upon addition of 3 equiv. of different TBA anions in DMSO. Inset: solution color of the chemodosisensor **2** in the absence and presence of TBAF. (b) Absorption titrations of 50 μ M chemodosisensor **2** with TBAF in DMSO. Inset: the absorbance at 357 nm as a function of added TBAF.

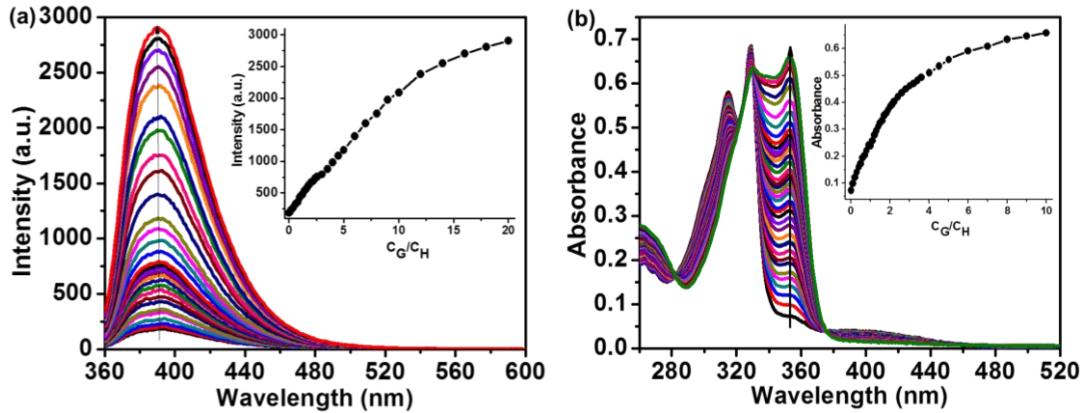


Figure S5 (a) Fluorescence titrations of 5 μM chemodosisensor **1** with TBAF excited at 353 nm in DMSO containing 10% water. Inset: the fluorescence intensity at 400 nm as a function of added TBAF. (b) Absorption titrations of 50 μM **1** with TBAF in DMSO containing 10% water. Inset: the absorbance at 353 nm as a function of added TBAF.

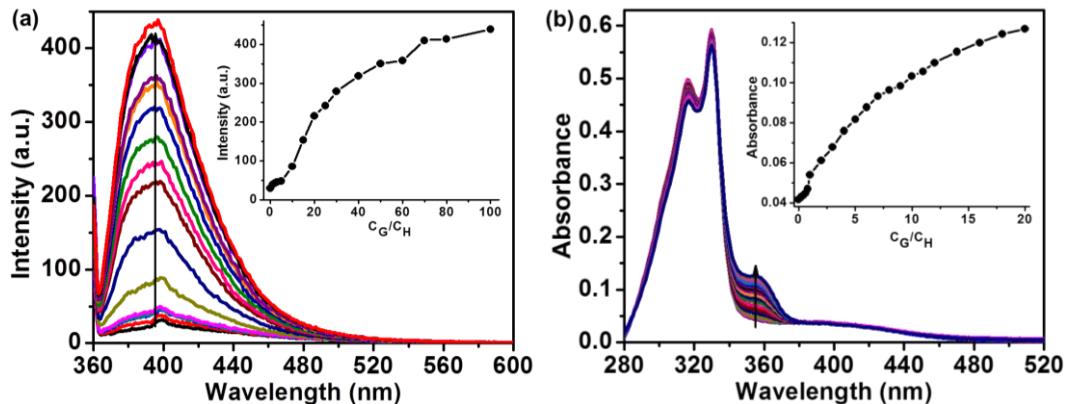


Figure S6 (a) Fluorescence titrations of 5 μM chemodosisensor **2** with TBAF excited at 357 nm in DMSO containing 10% water. Inset: the fluorescence intensity at 402 nm as a function of added TBAF. (b) Absorption titrations of 50 μM **2** with TBAF in DMSO containing 10% water. Inset: the absorbance at 357 nm as a function of added TBAF.

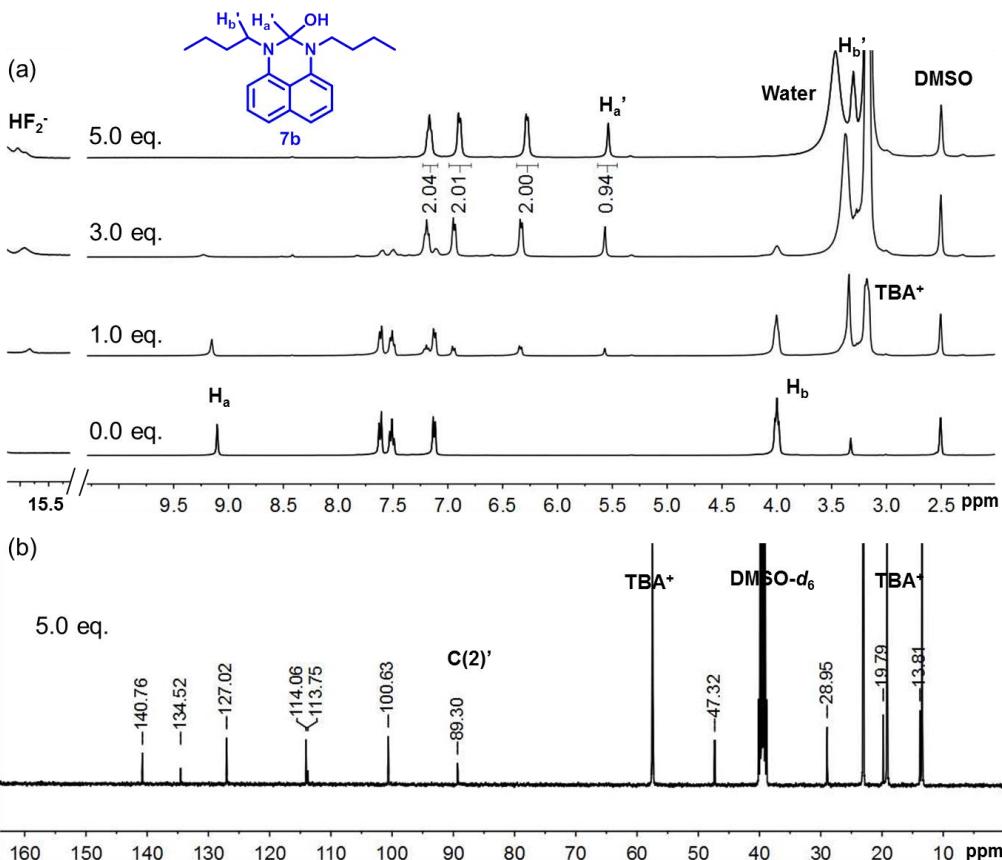


Figure S7 (a) ¹H NMR titrations of chemodosisensor **2** (5 mM) with TBAF in DMSO-*d*₆. (b) ¹³C NMR spectra of chemodosisensor **2** (5 mM) when in presence of 5.0 equiv. of TBAF in DMSO-*d*₆.

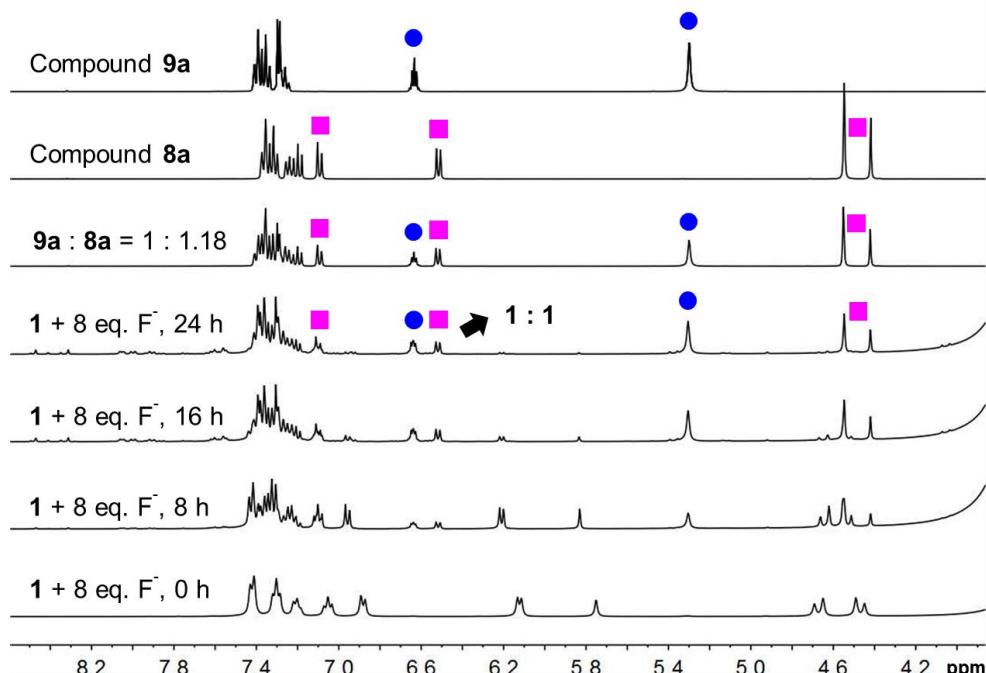
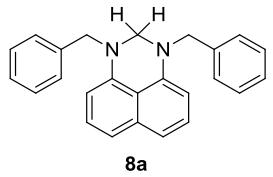


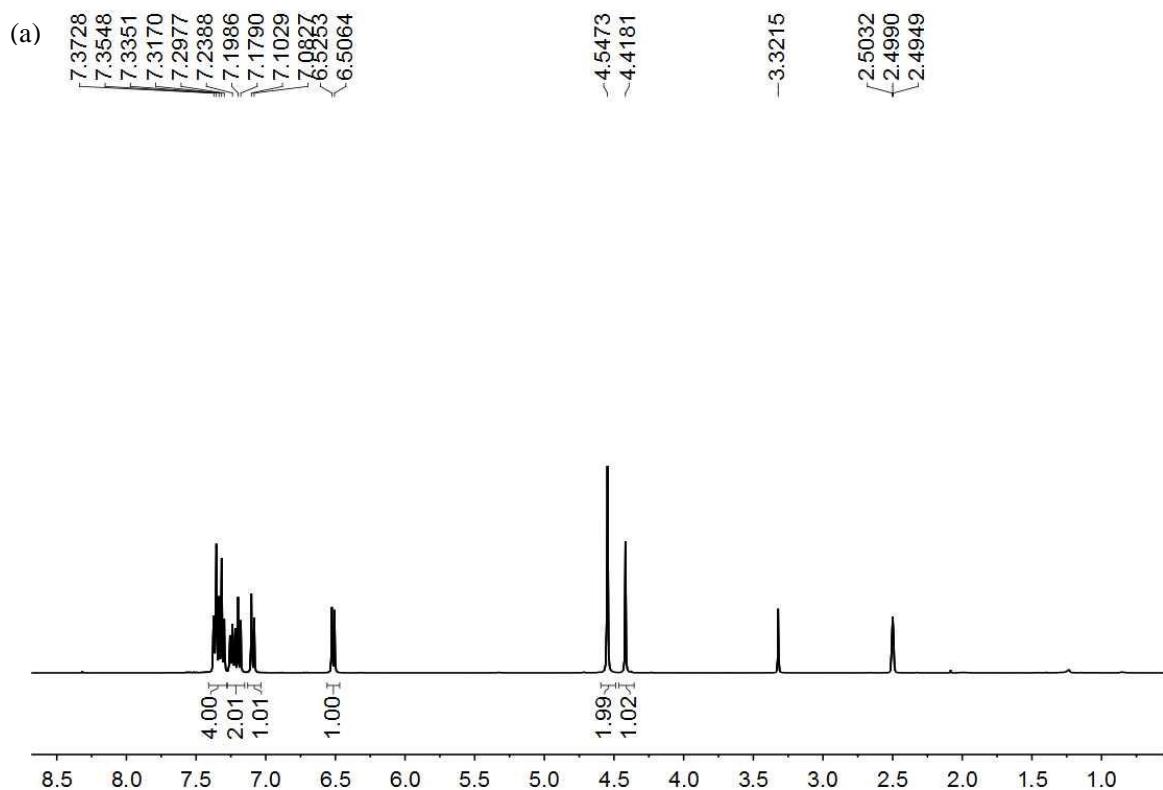
Figure S8 ¹H NMR spectra of chemodosisensor **1** unupon addition of 8.0 equiv. of TBAF in DMSO-*d*₆ after different time, and the spectra of compound **8a**, **9a** and the 1 : 1.18 mixture of **9a** and **8a**.



¹H NMR (400 MHz, DMSO-d₆) δ 4.42 (s, 2H), 4.55 (s, 4H), 6.51 (d, *J* = 7.6 Hz, 2H), 7.09 (d, *J* = 8.1 Hz, 2H), 7.18-7.26 (m, 4H), 7.30-7.37 (m, 8H).

¹³C NMR (100 MHz, DMSO-d₆) δ 53.34, 66.47, 104.03, 115.13, 116.98, 127.18, 127.43, 127.68, 128.95, 135.21, 138.50, 143.81.

HR MS: C₂₅H₂₂N₂, requires 350.1783, found [C₂₅H₂₂N₂Na]⁺ 373.1686.



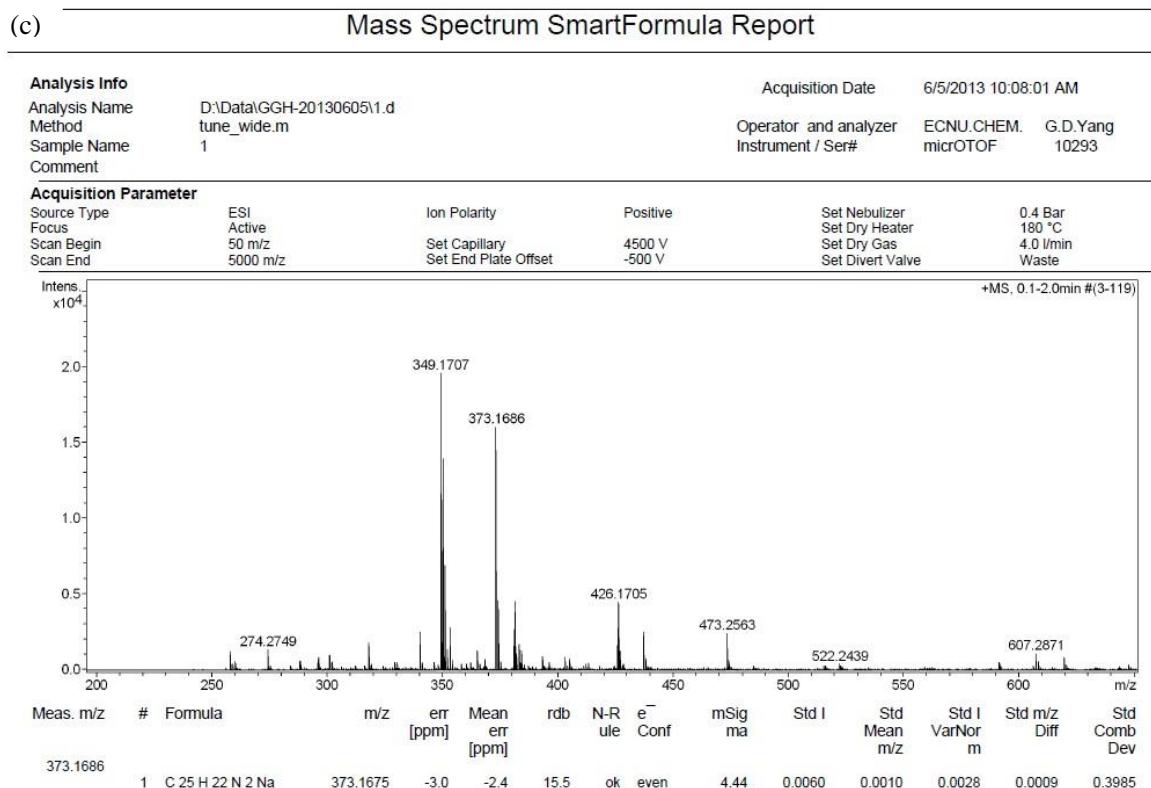
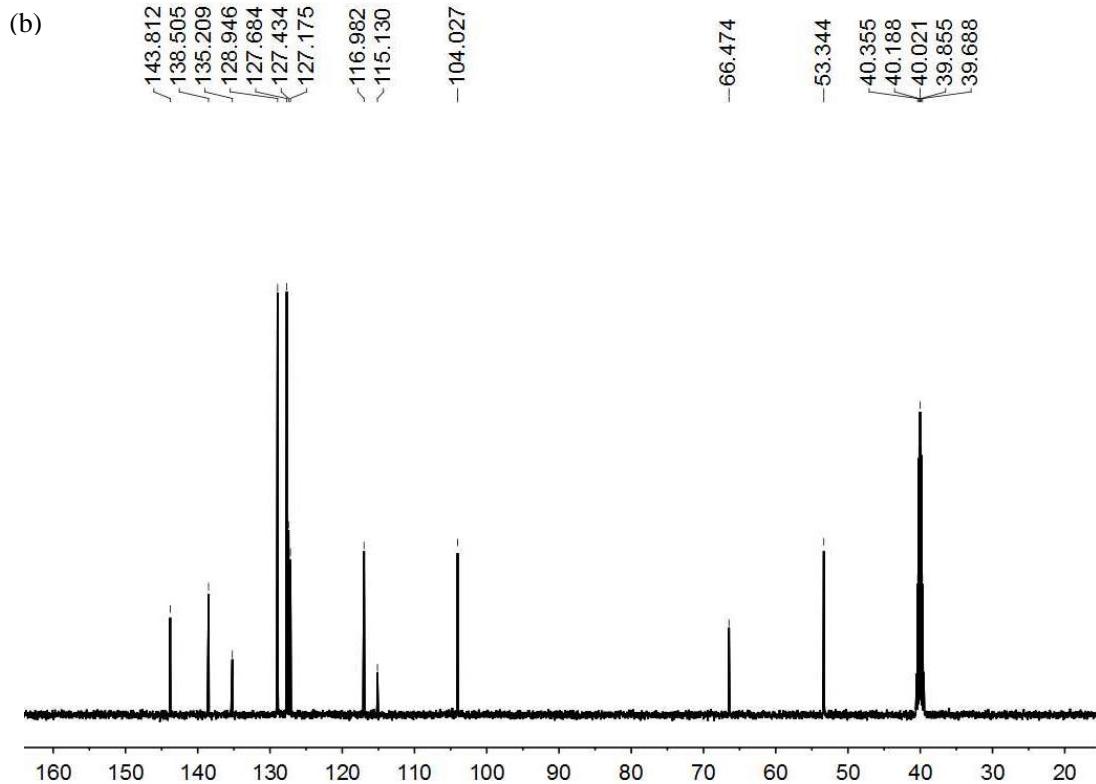
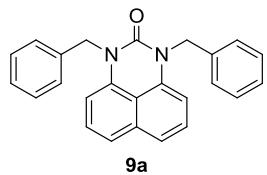


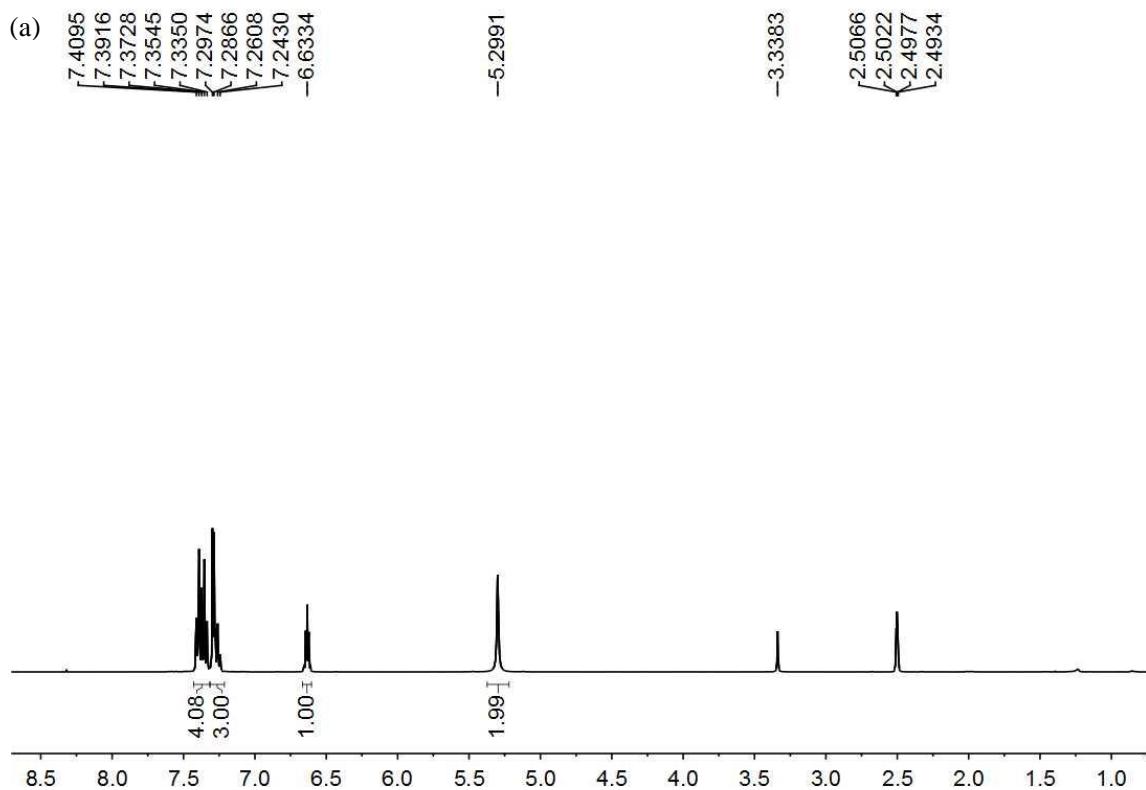
Figure S9 ^1H NMR (a), ^{13}C NMR (b) and HR MS (c) spectra of compound **8a**.



¹H NMR (400 MHz, DMSO-d₆) δ 5.30 (s, 4H), 6.61-6.66 (m, 2H), 7.24-7.41 (m, 14H).

¹³C NMR (100 MHz, DMSO-d₆) δ 46.96, 105.97, 114.85, 119.66, 126.93, 127.56, 128.28, 129.16, 134.51, 136.75, 136.95, 151.59.

HR MS: C₂₅H₂₀N₂O, requires 364.1576, found [C₂₅H₂₀N₂ONa]⁺ 387.1495.



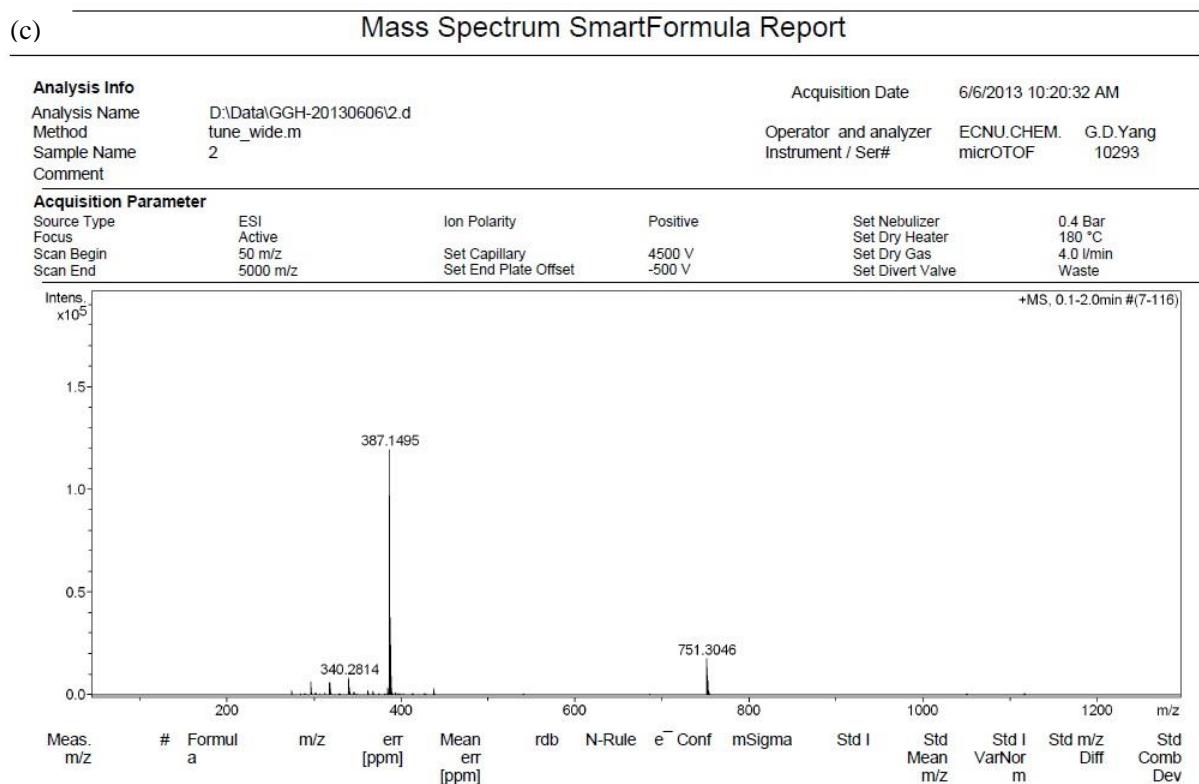
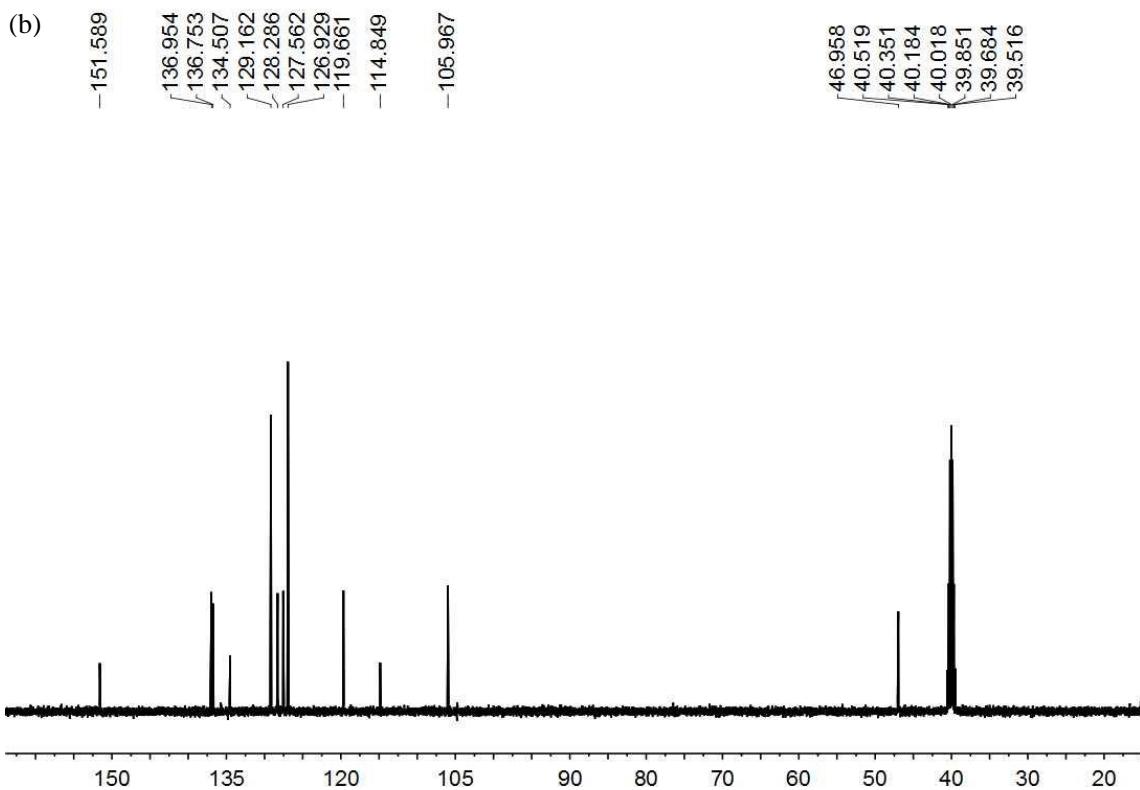


Figure S10 ¹H NMR (a), ¹³C NMR (b) and HR MS (c) spectra of compound 9a.

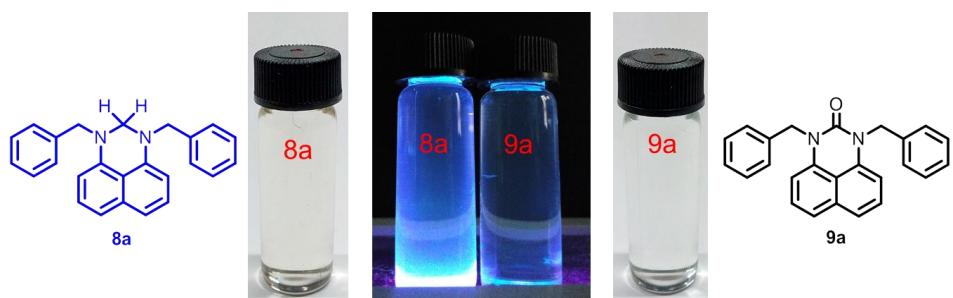
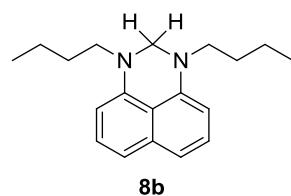


Figure S11 Solution colors and fluorescent colors of compounds **8a** and **9a** (5 μM).

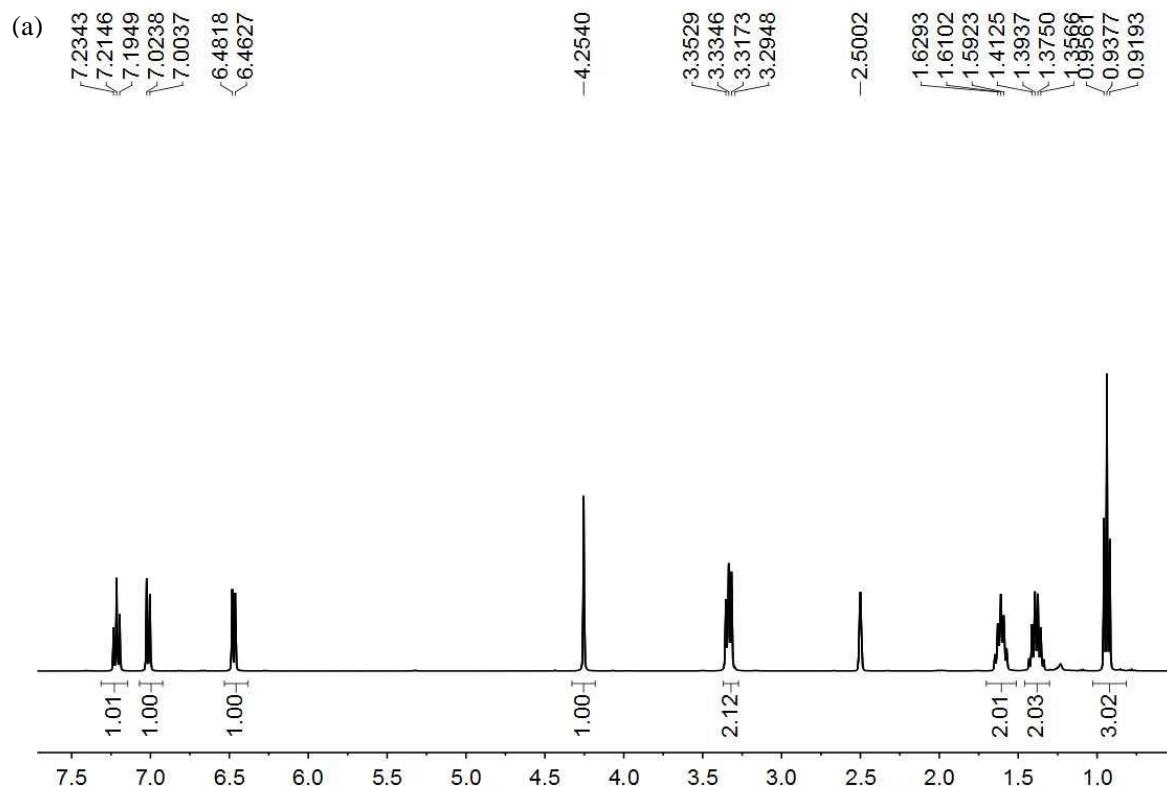


8b

¹H NMR (400 MHz, DMSO-d₆) δ 0.94 (t, *J* = 7.4 Hz, 6H), 1.35-1.41 (m, 4H), 1.57-1.63 (m, 4H), 3.33 (t, *J* = 7.4 Hz, 4H), 4.25 (s, 2H), 6.47 (d, *J* = 7.6 Hz, 2H), 7.01 (d, *J* = 8.0 Hz, 2H), 7.20 (d, *J* = 8.0 Hz, 2H).

¹³C NMR (100 MHz, DMSO-d₆) δ 13.80, 19.87, 27.52, 48.45, 65.03, 102.12, 114.62, 115.63, 126.71, 134.91, 143.30.

HR MS: C₁₉H₂₆N₂, requires 282.2096, found [C₁₉H₂₇N₂]⁺ 283.2179.



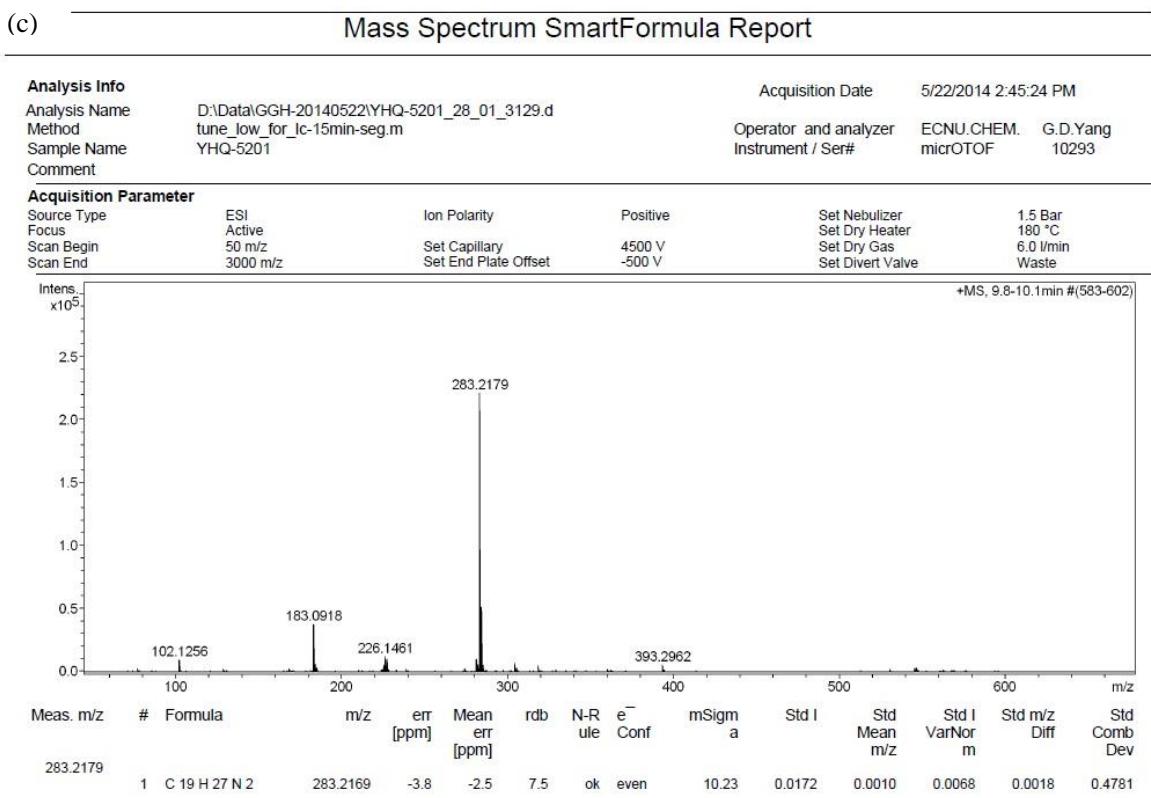
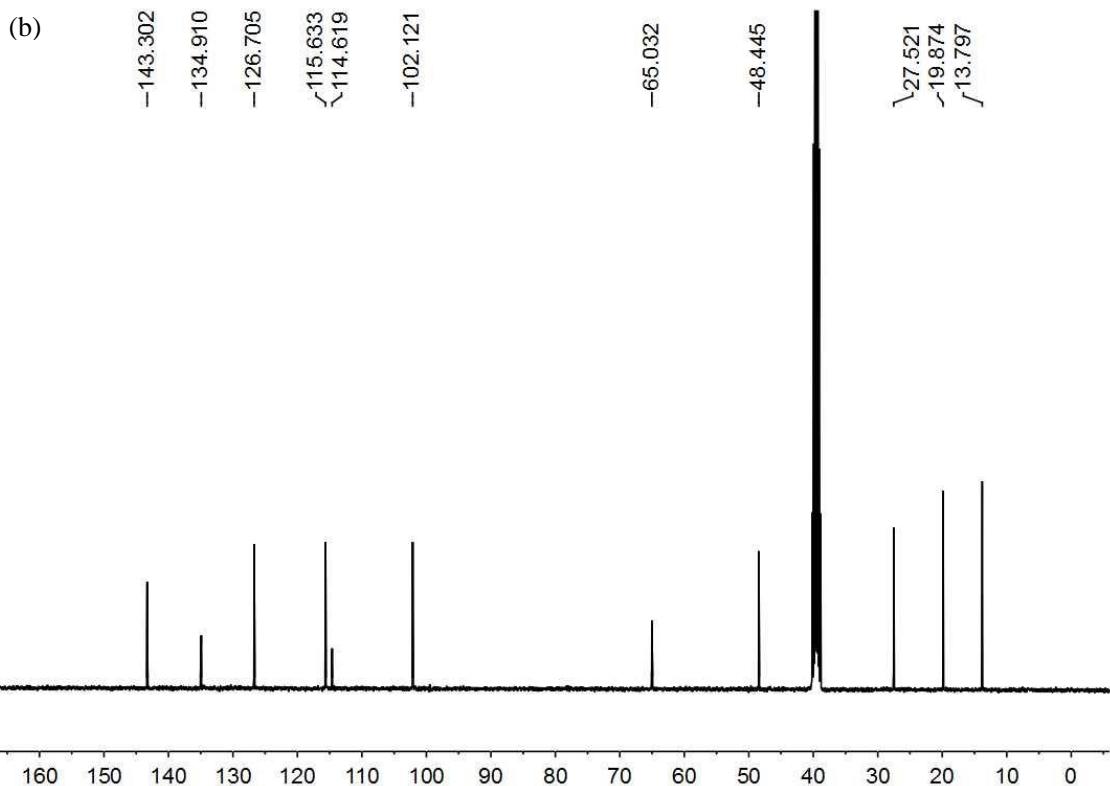
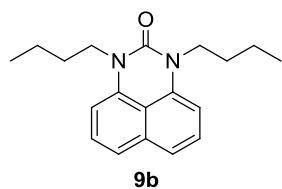


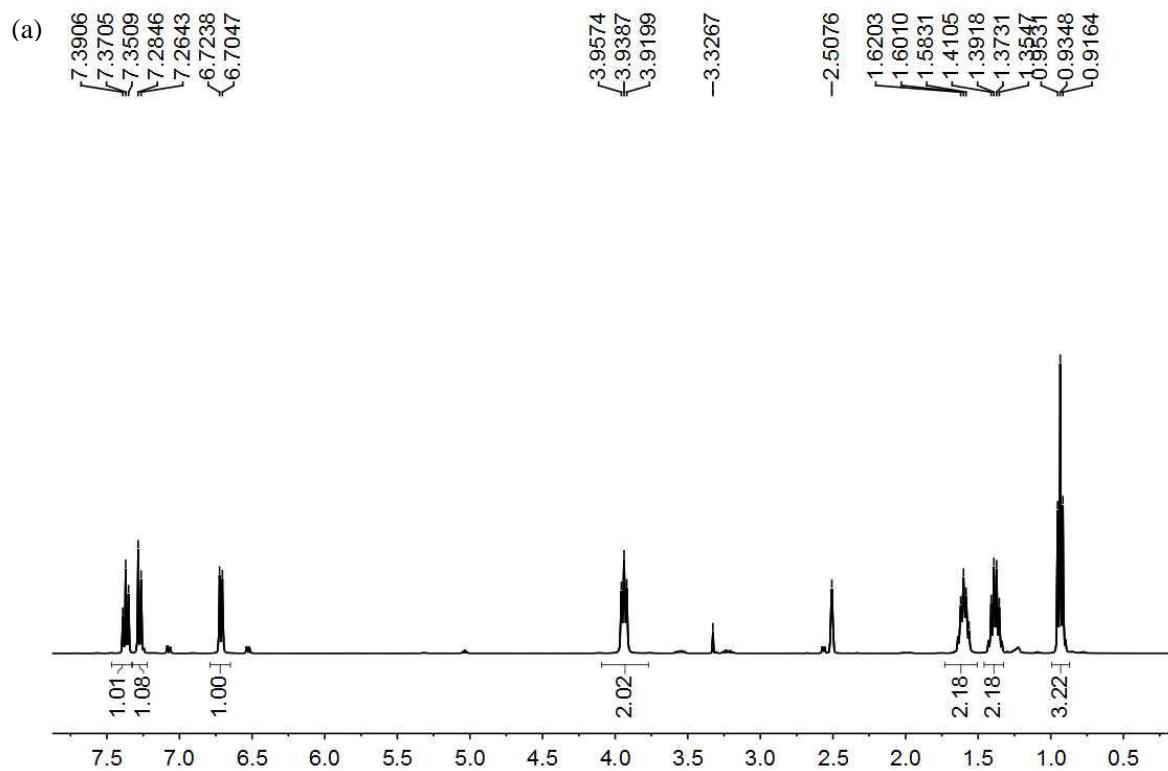
Figure S12 ¹H NMR (a), ¹³C NMR (b) and HR MS spectra of compound 8b.



¹H NMR (400 MHz, DMSO-d₆): δ 0.90-0.95 (m, 6H), 1.35-1.41 (m, 4H), 1.56-1.64 (m, 4H), 3.94 (t, J = 7.4 Hz, 4H), 6.72 (d, J = 7.6 Hz, 2H), 7.28 (d, J = 8.4 Hz, 2H), 7.37 (t, J = 8.0 Hz, 2H).

¹³C NMR (100 MHz, DMSO-d₆) δ 13.71, 19.51, 27.48, 42.40, 104.10, 114.43, 118.58, 127.93, 134.34, 136.42, 149.72.

HR MS: C₁₉H₂₄N₂O, requires 296.1889, found [C₁₉H₂₄N₂ONa]⁺ 319.1792.



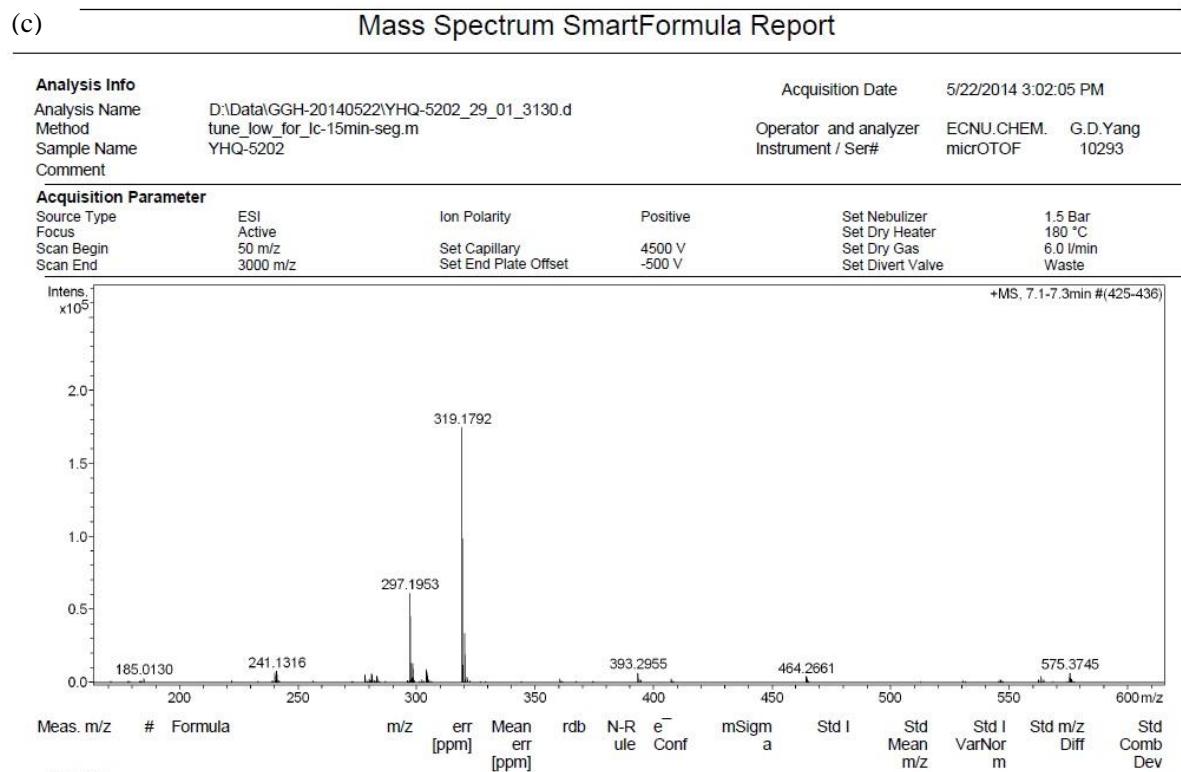
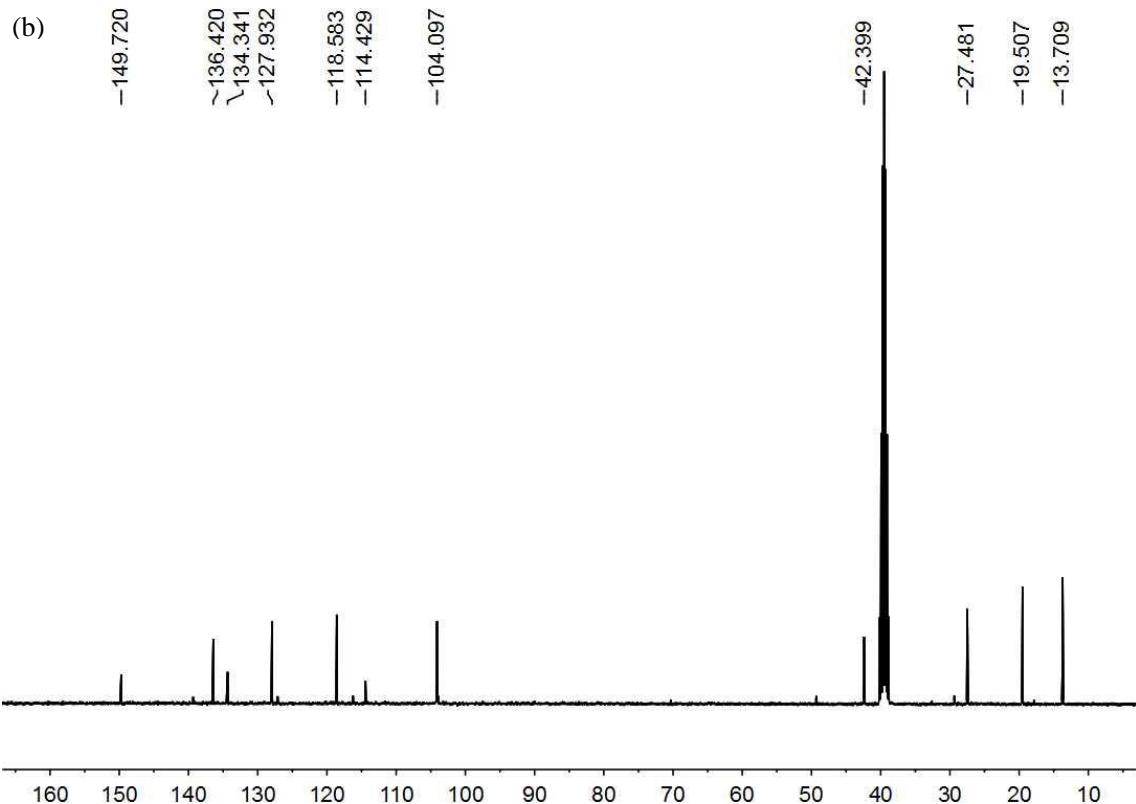


Figure S13 ^1H NMR (a), ^{13}C NMR (b) and HR MS spectra of compound **9b**.

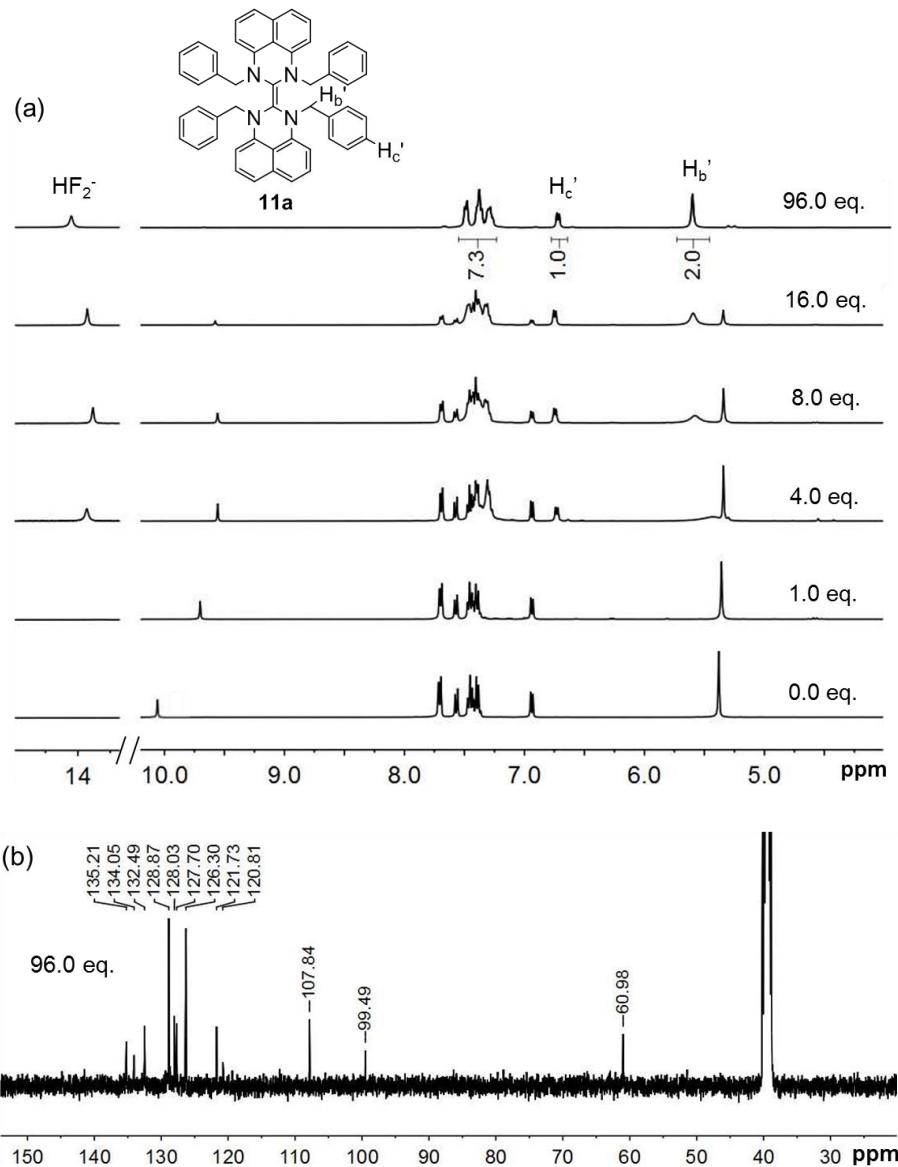
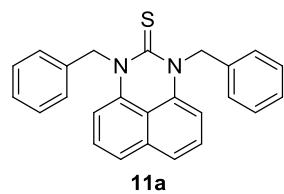


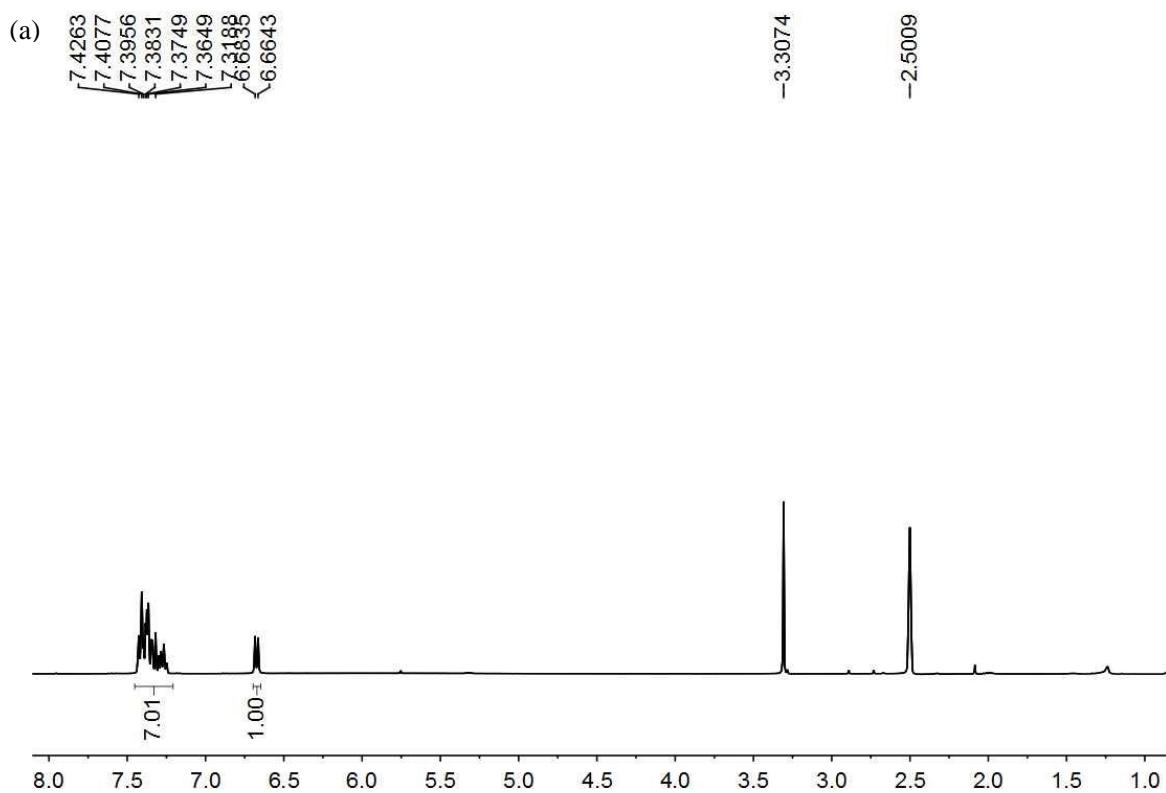
Figure S14 (a) ^1H NMR titrations of chemodosisensor **1** (5 mM) with AgF in DMSO-d_6 . (b) ^{13}C NMR spectra of chemodosisensor **1** (5 mM) when in presence of 96.0 equiv. of AgF in DMSO-d_6 .



¹H NMR (400 MHz, DMSO-d₆) δ 6.67 (d, *J* = 8.0 Hz, 2H), 7.27-7.42 (m, 14H).

¹³C NMR (100 MHz, DMSO-d₆) δ 54.37, 107.33, 116.45, 120.52, 126.03, 126.96, 127.97, 128.65, 133.68, 134.28, 135.29, 178.23.

HRMS: C₂₅H₂₀N₂S, requires 380.1347, found [C₂₅H₂₀N₂SNa]⁺ 403.1253.



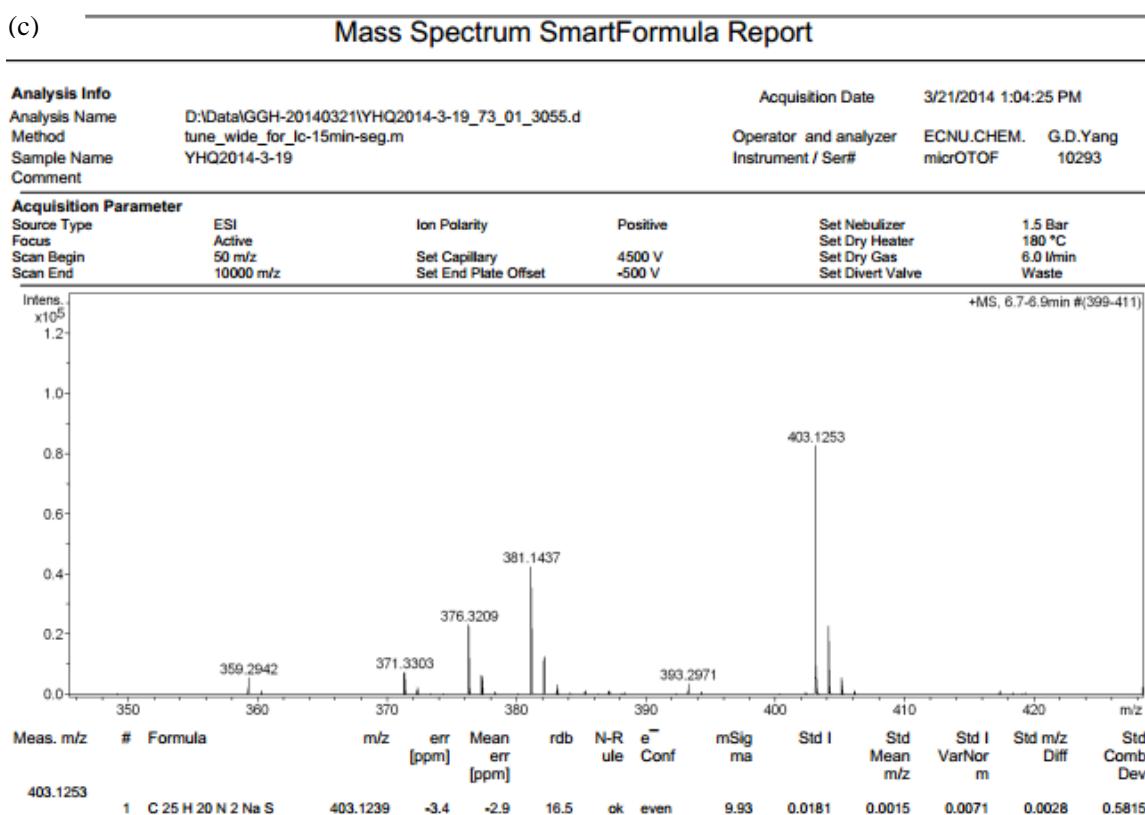
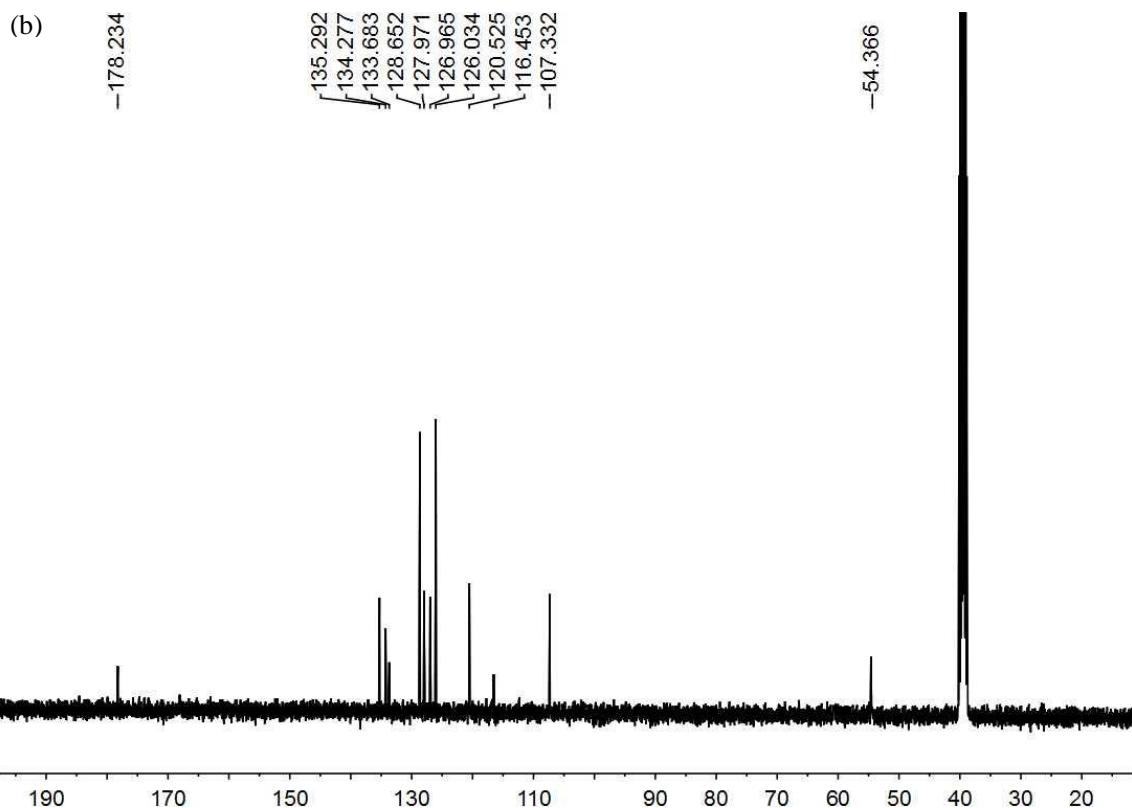
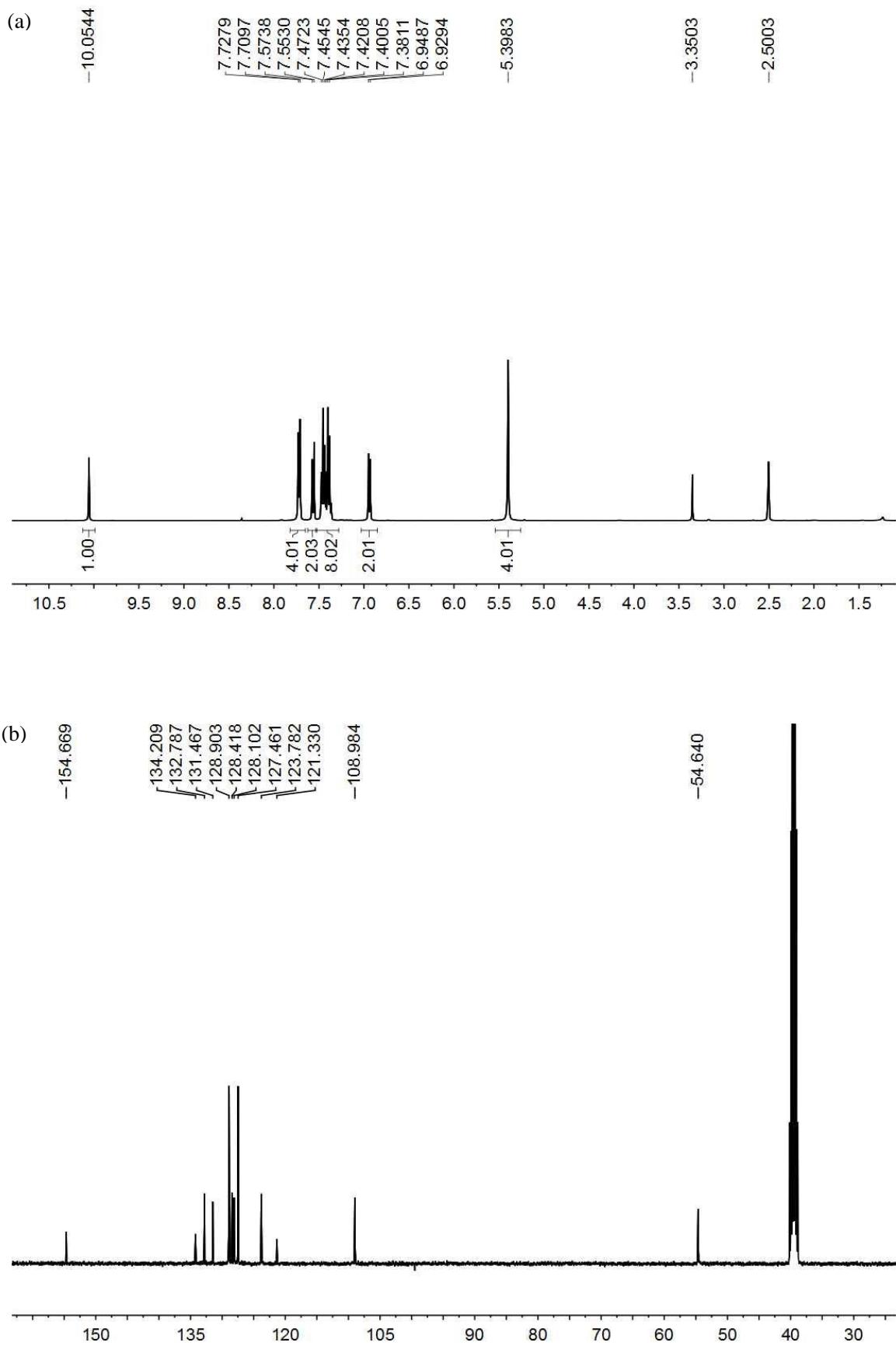
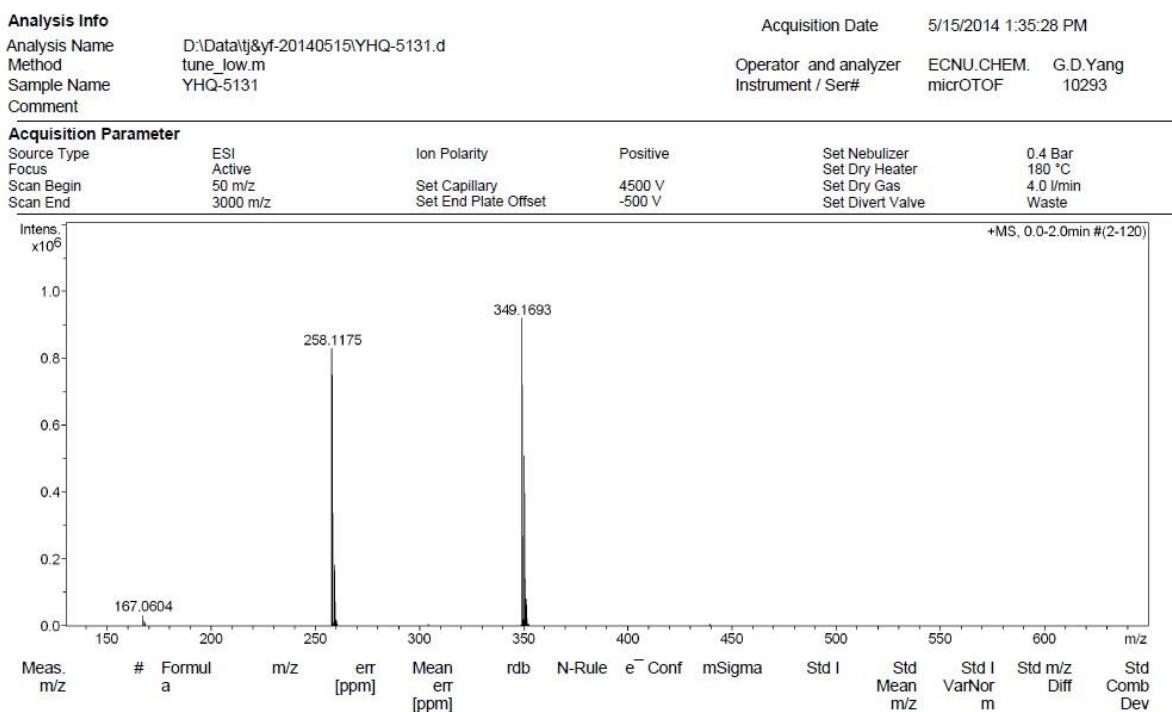
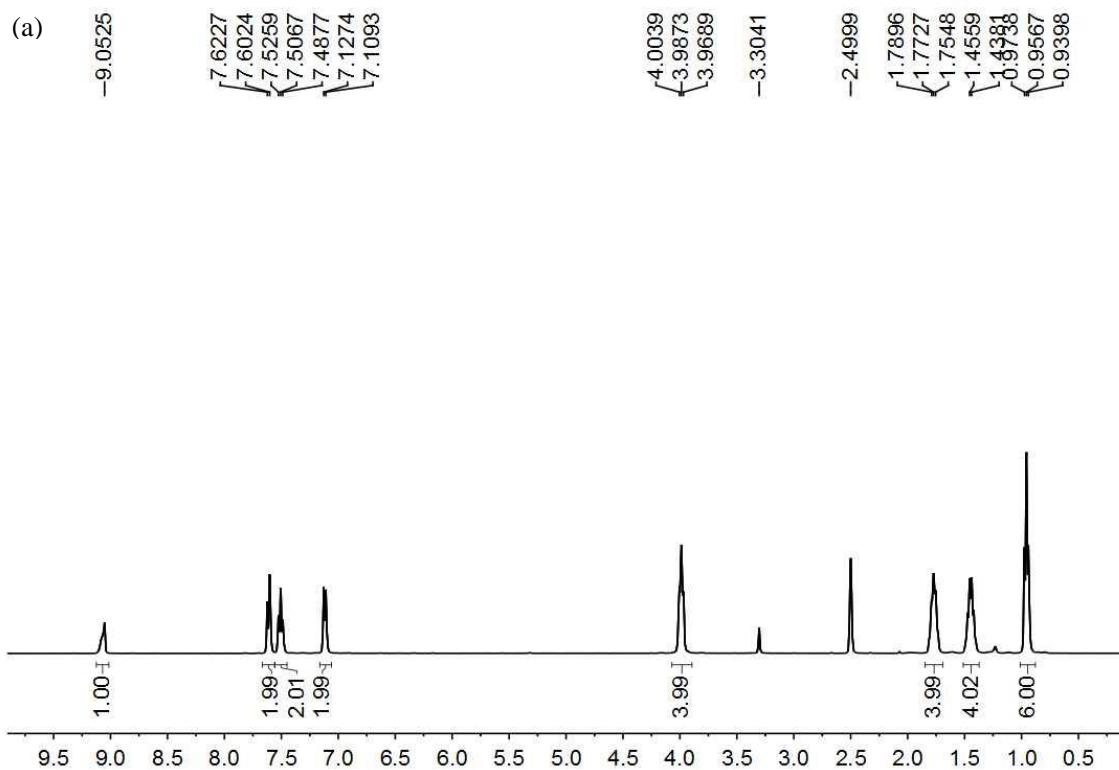


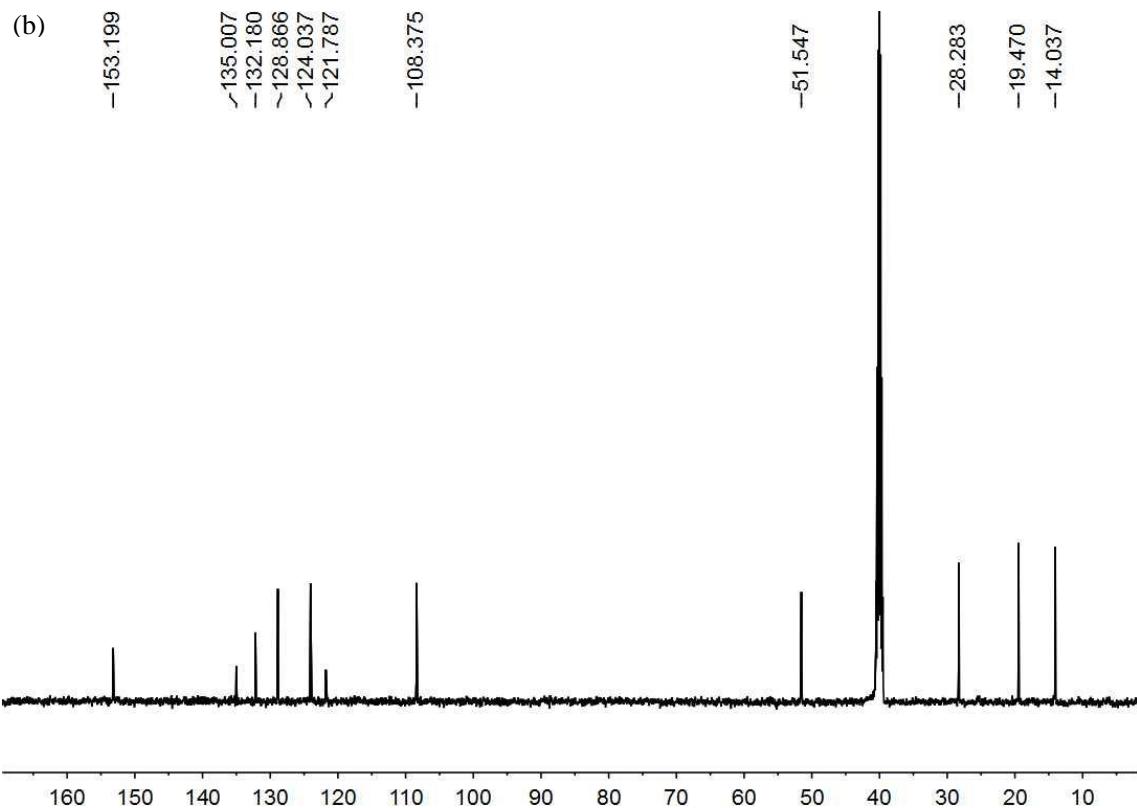
Figure S15 ¹H NMR (a), ¹³C NMR (b) and HR MS (c) spectra of compound 11a.



(c)

Mass Spectrum SmartFormula Report

**Figure S16** ^1H NMR (a), ^{13}C NMR (b) and HR MS (c) spectra of chemodosisensor **1**.



(c) Mass Spectrum SmartFormula Report

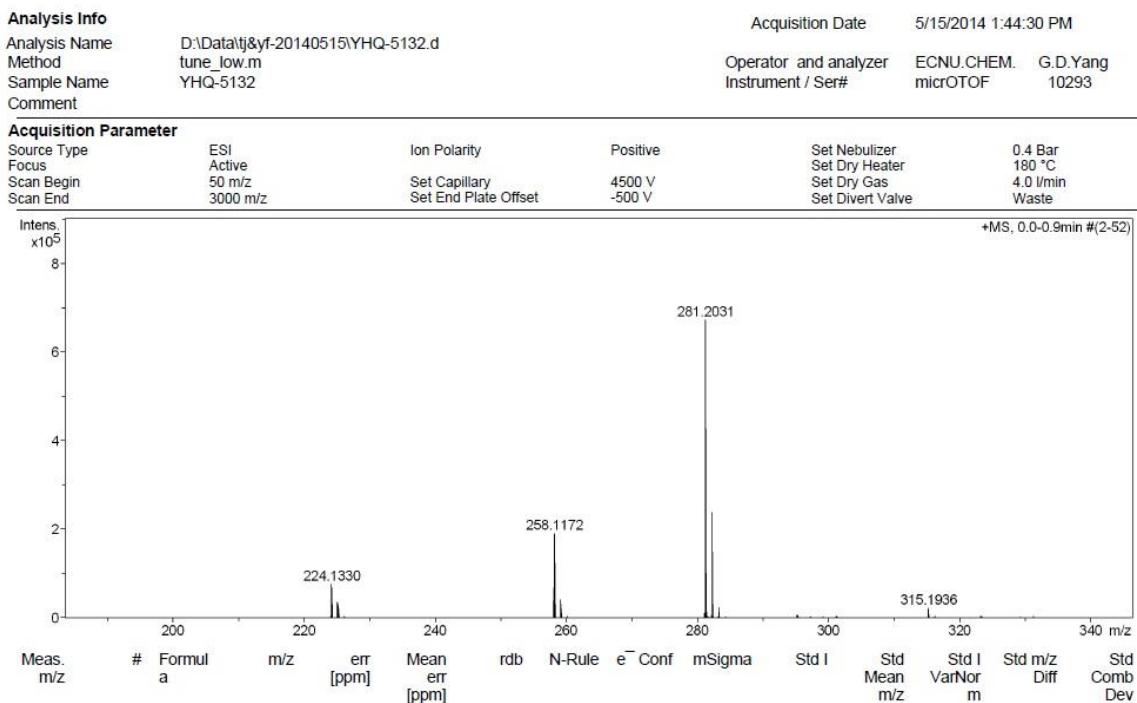


Figure S17 ^1H NMR (a), ^{13}C NMR (b) and HR MS (c) spectra of chemodosisensor 2.

Table S1. Crystal data and refinement of compounds **1**, **2** and **9a**.

Compound	1	2	9a
CCDC	1005672	1005673	1005674
Empirical formula	C25H21ClN2	C19H25BrN2O	C25H20N2O
M_r	384.90	377.32	364.43
Temperature (K)	296(2)	296(2)	296(2)
Crystal system	Orthorhombic	Triclinic	Monoclinic
Space group	P2 ₁ 2 ₁ 2 ₁	P-1	C2/c
$a/\text{\AA}$	7.2746(10)	10.4898(12)	30.3297(13)
$b/\text{\AA}$	14.2457(19)	11.5739(13)	8.7540(4)
$c/\text{\AA}$	18.631(3)	16.4456(19)	14.8727(6)
$\alpha/^\circ$	90.00	93.603(4)	90.00
$\beta/^\circ$	90.00	98.485(3)	105.795(2)
$\gamma/^\circ$	90.00	106.291(4)	90.00
$V/\text{\AA}^3$	1930.8(5)	1884.0(4)	3799.7(3)
Z	4	4	8
Crystal size (mm ³)	0.32×0.21×0.12	0.48×0.36×0.22	0.33×0.28×0.14
$D_c/\text{g cm}^{-3}$	1.324	1.330	1.274
μ/mm^{-1}	0.21	2.188	0.078
$F(000)$	808.0	784	1536
θ range (°)	1.80-25.01	1.26-25.01	2.43-25.00
Reflections collected	22626	22167	21508
Unique reflections	3397	6613	3349
GOF on F^2	1.029	1.036	1.028
R_{int}	0.0795	0.0240	0.0531
R_1 [$I > 2\sigma(I)$]	0.0418	0.0518	0.0420
wR ₂ (all data)	0.0973	0.1536	0.1124