Investigation of nitride layers deposited on annealed AISI H13 steel by die-sinking electrical discharge machining

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Abstract

The performance of electrical discharge machining is dependent mainly on the dielectric fluid and process parameters. Significant attention has been drawn to introducing alternative dielectric fluids in order to achieve superior surface integrity and higher productivity. However, there is a challenge in producing more uniform surface layers in the emerging nitriding process by electrical discharges. Thus, this work studies the relationship between process parameters and the features of the layers produced using surface modification by die-sinking electrical discharge machining employing urea [(NH₂)₂CO] in deionized water as dielectric fluid. The influence of pulse-on time and pulse-off time on the uniformity and thickness of the recast layer, heat-affected zone, and depth of hardened layer was investigated. The experimental work employed annealed AISI H13 tool steel as base material and electrolytic copper as electrode. In order to characterize the samples and investigate the modifications produced by the process, chemical analysis, optical microscopy, scanning electron microscopy, X-ray diffraction, and microhardness measurements were used. The results indicated an evident influence of the process parameters on uniformity and thickness of the layers. The thickest altered metal zone layer obtained was $53.2 \pm 5.5 \,\mu$ m when both pulse-on and -off times of 500 μ s were used, while the smallest thickness of $31.0 \pm 4.6 \,\mu$ m was obtained for both pulse times of 100 μ s. The investigated process increased the surface hardness about three times compared with the non-processed AISI H13 steel. This can be attributed to the iron nitrides such as Fe₁₁N found in the samples.

Keywords AISI H13 tool steel · EDM · Microhardness · Nitriding · Surface modification · XRD

Nomenclat	ture	EDS	Energy-dispersive spectroscopy		
α	Significance level	HAZ	Heat-affected zone		
AMZ	Altered metal zone	ICSD	Inorganic Crystal Structure Database		
ANOVA	Analysis of variance	Ip	Peak current [A]		
COD	Crystallography open database	ÓES	Optical emission spectrometry		
CI	Confidence interval	RL	Recast layer		
D_{T}	Duty factor [%]	p value	Probability value		
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Surface modification by die-sinking
electrical discharge machining
Scanning electron microscopy
Thickness of altered metal zone
Thickness of heat-affected zone
Pulse-off time [µs]
Pulse-on time [µs]
Thickness of recast layer
Discharge voltage [V]
X-ray diffraction

1 Introduction

Electrical discharge machining (EDM) is applied for electrically conductive materials with the thermoelectric working principle in which the thermal energy of a spark is used to remove material from the workpiece [1]. The EDM input parameters [e.g., peak current (I_p) , pulse-on time (T_{ON}) , pulse-off time (T_{OFF}) , discharge voltage (U), working time, nozzle flushing] affect the process aiming at improving its performance, e.g., material removal rate, surface finishing, tool wear rate [2], and corrosion resistance [3, 4]. Nevertheless, a rather limited number of EDM machining parameters and few settings have been tried by researchers and most of them show the remarkable influence of I_p and $T_{\rm ON}$ on the EDM performance [5]. Therefore, there is a need to independently study the effect of T_{OFF} as this is an important input process parameter on the phenomenon of surface modification [6].

The EDM process changes not only the surface of the workpiece metal, but alters also the metallurgical structure and characteristics of subsurface [7] once the discharge temperature reaches up to 12,000 °C [8]. On the top of the unaffected base metal, a molten and resolidified layer is observed, named as recast layer (RL). This layer is formed by the unexpelled molten metal solidifying in the crater [6] from the dielectric fluid cooled at very high cooling rate. Adjacent to the RL is the heat-affected zone (HAZ) or annealed layer, which was not melted but influenced by high temperatures [7]. The depth of the RL and the HAZ is determined by the heat sinking ability of the material and the power applied (pulse energy and duration). With this regard, the discharge energy—determined by the I_p and T_{ON} [9], assuming a constant discharge voltage [10]-is the main responsible for the depth of hardened layer.

Klocke et al. [9] studied the relationship between discharge energy and the thickness of recast layer (t_{RL}) and stated that a higher discharge energy results in a thicker RL. This is caused by a larger amount of metal which melts and resolidifies on the surface, as also observed by Guu et al. [11] and Rebelo et al. [12]. Similarly, the thickness of the heat-affected zone (t_{HAZ}) is influenced by discharge energy, i.e., the higher average discharge energy, the thicker HAZ [9]. This means that the more energy dissipates into the workpiece, the deeper the temperature reaches the recrystallization level and recrystallization takes place [9].

The field of surface modification using EDM is at the experimental stage. Some topics of interest are (i) workpiece surface modification using chemical elements present in the dielectric fluid [6, 13–15], transfer of chemical elements from powder metallurgy (PM) electrodes (e.g., WC-Co) to the workpiece surface (e.g., AISI H13) [16, 17], and alloyed workpiece surface layer (e.g., the formation of a TiC layer in a high speed steel substrate) [18]. In some cases, powders are added and mixed in the dielectric fluid (e.g., Al, Cr, graphite, SiC, Ti, etc.), named after powder mixed EDM. The feasibility of the surface modification process is established, although many issues need to be addressed before the method can be formally accepted by industry [6]. For instance, one of the issues to be addressed is the non-uniformity of the layers formed in the case of surface modification by die-sinking electrical discharge machining (SM-SEDM) using deionized water as dielectric, as indicated by [19, 20]. For Marashi et al. [4], among various types of dielectric used in powder mixed EDM, hydrocarbon oil, such as mineral oil or kerosene, is still the most common and efficient die-sinking EDM dielectric fluid. Nevertheless, water has found increasing application in recent years due to (i) healthier and safer operational environment; (ii) deionized water can be a favorable choice in some EDM applications owing to its lower viscosity, higher thermal conductivity, and higher flow rate compared with hydrocarbon oil dielectrics. Most of the published research works on powder mixed EDM have studied its effects on material remove rate, surface finishing, and tool wear rate. However, very few works have taken up the study of the effects of powder mixed EDM method on surface modification [6], and only a few investigations have considered the addition of powder to water as dielectric fluid.

The hardness of the machined surface is also affected by SM-SEDM, by powder mixed EDM or by using PM electrodes. Yan, Tsai, and Huang [21] employed distilled water and distilled water with urea (10 g L¹) as dielectric fluids during EDM of pure titanium. The authors obtained higher cross section microhardness values of the samples produced with urea solution in comparison with distilled water. In addition to that, the hardened layer obtained with urea solution was approximately 80 µm thick, while the hardened layer produced with distilled water was 50 µm thick. This effect was explained by the decomposition of nitrogen from the urea solution dielectric, with nitrogen enrichment evenly distributed on the machined surface that resulted in a TiN hard layer produced on the titanium surface, which exhibited improved friction and wear characteristics [21]. Santos et al. [22] found that surface hardening occurs in both cases: with deionized water only and with a solution of urea in deionized water as dielectric fluids. However, the microhardness values on the RL and HAZ were higher for the samples machined with a solution of urea as dielectric fluid in comparison with deionized water. Furthermore, the hardness of the RL of the samples machined with a solution of urea was higher than that of the HAZ. It is suggested that the amount of nitrides is larger in the RL than in the HAZ, which is reasonable since the nitrogen concentration decreases with the distance from the surface. This phenomenon was suggested by [22], who detected highest concentration of nitrogen on the surface, which reduced gradually below the machined surface. The presence of urea in the dielectric fluid did not alter the typical morphology of the machined surfaces in comparison with deionized water [22]. Çoğun, Özerkan, and Karaçay [5] studied the variation of surface hardness with different T_{ON} (25 and 100 µs) and found that increasing $T_{\rm ON}$ increases the hardness of the machined surface. This is mainly due to thicker resolidified layer formed by the higher discharge energy. The hardness values for both powders (H₃BO₃ and graphite) used during machining were higher than for the pure kerosene used as base dielectric. Aspinwall et al. [17] obtained an increase in hardness of up to 100% of AISI H13 steel processed by die-sinking EDM, which can be attributed to factors as I_{p} , T_{ON} , and the presence of elements from PM electrode on the machined surface.

With regard to steel grades and their applications, highalloy steels as AISI H13 hot work die steel provide a compromise between resistance to thermal softening (wear) and reasonable toughness. AISI H13 steel is especially suitable for press dies as well as for hammer applications in which the amount of parts to be forged is large and wear resistance becomes an important economic requirement [23]. Some of the chemical elements for AISI H13 (e.g., Cr, Mo, V) form hard carbides, which contribute to wear resistance. Nitriding on parts manufactured with AISI H13 steel offers additional wear resistance [24].

In this work, the surface modification by die-sinking electrical discharge machining (SM-SEDM) of AISI H13 steel using electrolytic copper as electrode and a solution of urea [(NH₂)₂CO] in deionized water as dielectric fluid (concentration of 30 g L¹) is proposed. According to [25, 26], the solubility of urea (minimum purity of 99.98% certified ACS-grade) in double-distilled water is 121.0 g in 100 g of H_2O (1206.3 g L⁻¹) at 25.5 °C. The influence of T_{ON} and T_{OFF} on the formation of RL and HAZ was investigated, aimed at identifying the conditions related to the set of parameters, defined experimentally based on preliminary tests, responsible for better uniformity of the RL and HAZ, as well as thicker hardened layer after nitriding by SM-SEDM. Furthermore, nitride formation simultaneously to electrical discharge machining of AISI H13 steel was evaluated. Prior to the experimental work, the electrical discharge machine was adapted (development of a new system with an auxiliary tank to

deliver the dielectric fluid) in order to enable the application of SM-SEDM.

2 Materials and methods

Annealed AISI H13 tool steel samples (25.40 mm diameter × 12.00 mm thickness) with an average hardness of 214 ± 2 HV_{0.5} were used as base material and C11000 UNS class electrolytic copper (31.75 mm diameter × 40.00 mm length) was employed as electrode.

An Engemaq EDM 200 NC L series die-sinking electrical discharge machine (maximum electric current of 60 A) was adapted to perform SM-SEDM, as follows: in order to avoid contamination of the usual EDM dielectric fluid, a tank (capacity for 62 L) was designed and built to work with aqueous fluids. It is placed inside the original machine tank, as shown in Fig. 1. An open flushing system composed of a motorpump (power of 120 W, flow rate of 62 L min⁻¹, and pressure of 0.7 bar) and dielectric fluid orienting pipe was used to supply the dielectric fluid during SM-SEDM treatment.

The process parameters used in this work are shown in Table 1. Six sets of operational parameters (I to VI), defined experimentally based on preliminary tests, were selected and the main goal was to investigate the influence of different $T_{\rm ON}$ and T_{OFF} on the formation of RL and HAZ. For all tests, the white crystalline granular urea (nitrogen source) dissolved in deionized water was used as a dielectric fluid at a concentration of 30 g L¹, defined based on preliminary tests that resulted in good agreement with hardness, electrical conductivity of the solution, and consumption of urea. Urea contains 46.6% of nitrogen in its composition, with 25.5% nitrogen in the saturated solution (1206.3 g L¹ at a temperature of 25.5 °C). However, according to [20, 21], the increase in the amount of urea (higher than 10 g L⁻¹) in pure water can influence the electrical conductivity, which affects the performance of the process (e.g., formation of the plasma channel and material removal rate). Therefore, using deionized water $(0 \ \mu S \ cm^{-1})$ as a solvent, it was observed in this work that, when adding 30 g L 1 of high purity urea (99.5%), the average electrical conductivity remained low $(13 \pm 1 \ \mu S \ cm^{-1} \ at the$ beginning of the test and $21 \pm 1 \ \mu\text{S cm}^{-1}$ at the end of the experiments with SM-SEDM). Based on that, and the fact that variations in the urea content did not significantly change the thickness of the nitride layer and the hardness of the surface, the concentration of 30 g L¹ was defined for this work. Furthermore, if the urea concentration increases (e.g., doubling from 30 to 60 g L¹), there will be a difference of only approximately 0.65% of nitrogen in the solution, which does not contribute significantly to higher nitrogen migration to the surface and subsurface of the samples. For the tests, the conductivity of the dielectric fluid was monitored using a portable

Fig. 1 Schematics of SM SEDM system used in the experimental work



conductivity meter (range 0–1999 μ S cm⁻¹, temperature 1–80 °C, and accuracy of ± 1 digit).

The duration of each test was 15 min for roughing (51.45 A) and 20 min for finishing (6.86 A), determined in preliminary tests and aiming at ensuring spark erosion of the entire workpiece surface. Each test was performed three times and the average values are report.

Chemical analysis of AISI H13 steel and electrolytic copper electrode samples was carried out by optical emission spectrometry (OES) in a SPECTRO SPECTROMAXx LMX equipment. Vickers microhardness tests were conducted on cross sections of AISI H13 steel samples using a Shimadzu HMV-2TE automatic microtester with a load of 25 gf and 20 s dwell time. The phase compositions on AISI

 Table 1
 Experimental work parameters used for SM SEDM

Fixed parameters			Descriptio	n		
Dielectric fluid	Urea in de	Urea in deionized water				
Dielectric concentration	30 [g L ¹]					
Tool electrode	Electrolyti	c copper				
Tool electrode polarity			Positive [+	·]		
Erosion time			0.8 [s]			
Periodic electrode retraction			2 [mm]			
Distance between the electrode and	workpiece (gap)		15			
Sensitivity of servomechanism			15			
Roughing operation						
Peak current, I _p			51.45 [A]			
Test duration			15 [min]			
Set of parameters	Ι	II	III	IV	V	VI
Pulse on time, T_{ON} [µs]	100	100	100	100	500	500
Pulse off time, T_{OFF} [µs]	43	100	150	203	500	1015
Duty cycle, $D_{\rm T} \left[\%\right]^{\rm a}$	70	50	40	33	50	33
Finishing operation						
Peak current, I _p			6.86 [A]			
Test duration			20 [min]			
Set of parameters	Ι	II	III	IV	V	VI
Pulse on time, T_{ON} [µs]	10	10	10	10	50	50
Pulse off time, T_{OFF} [µs]	4	10	15	20	50	102
Duty cycle, $D_{\rm T}$ [%]	70	50	40	33	50	33

^a Duty cycle [($T_{\rm ON}$ / ($T_{\rm ON}$ + $T_{\rm OFF}$)) × 100], $D_{\rm T}$ [%]

H13 steel samples before and after the SM-SEDM were obtained with a Shimadzu XRD-7000 diffractometer using Cu K α 1 radiation (λ = 1.540562 Å). Data were collected at a tube voltage of 40 kV and a current of 30 mA at a step size of 0.02° 2 θ for a scan step time of 5 s over an angular range of 30° to 110° 2 θ . The obtained diffractograms were analyzed with the DIFFRAC.EVA 5.2 software by Bruker AXS, Germany [27]. The XRD patterns were interpreted by comparison with the cards of the Crystallography Open Database (COD) [28] and the Powder Diffraction Files (PDF) from the Inorganic Crystal Structure Database (ICSD).

3 Results and discussion

3.1 Material characterization

Tables 2 and 3 show, respectively, the chemical compositions of AISI H13 steel and electrolytic copper samples used in the SM-SEDM experiments. The presence of chemical elements such as Cr, Mo, and V, which favor nitriding, can be noticed in AISI H13 steel (Table 2) and the corresponding amounts are in accordance with ASTM standard [29].

Table 3 shows that the composition of electrolytic copper electrodes is in accordance with the literature [30]: 99.90% Cu (min) and 0.04% O for UNS C11000 class electrolytic copper.

Figure 2 presents results concerned with AISI H13 steel cross section. Figure 2 a shows a scanning electron microscopy (SEM) image of the microstructure etched in 2% Nital and obtained by secondary electrons. Figure 2 b shows the semiquantitative chemical composition analysis (wt.%) by energydispersive spectroscopy (EDS), performed in six different regions of the SEM image and Fig. 2c shows the X-ray diffractogram. It can be noted in Fig. 2a that the microstructure of AISI H13 steel consists of spheroidized carbides uniformly dispersed in a ferrite matrix. This microstructure is typical of annealed AISI H13 steel. The semi-quantitative analysis by EDS, see Fig. 2b, indicates the presence of Cr, V, Si, and Fe, and Fig. 2c indicates the presence of ferrite (α -Fe), (Cr, $Fe_{7}C_{3}$, and $Cr_{3}C_{2}$ carbides. These results corroborate with Roberts, Krauss, and Kennedy [31], who state that the equilibrium structure at room temperature of the annealed AISI H13 steel is M₇C₃ carbides in a ferrite matrix (M denotes a

 Table 2
 AISI H13 steel chemical composition (wt.%)

Element	Concentration	Element	Concentration
Fe	91.145	Mn	0.367
Cr	4.738	V	0.802
Мо	1.226	Cu	0.058
С	0.380	Others	0.350
Si	0.935		

 Table 3
 Electrolytic copper chemical composition (wt.%)

Element	Concentration	Element	Concentration
Cu	99.902	Ni	0.006
Zn	0.022	С	0.005
Те	0.020	Al	0.004
Со	0.014	S	0.003
Bi	0.009	Others	0.015

transition metal, e.g., Cr and Fe). These data are in agreement with results from the Fe-Cr-C system alloys containing 5% chromium, as AISI H13 steel [31].

The average Vickers microhardness of AISI H13 as received steel is 214 ± 2 HV_{0.5}, with a 95% confidence interval (CI). This result agrees with data from the manufacturer which indicates a maximum hardness of 207 HB (approximately 218 HV [32]).

3.2 Influence of SM-SEDM parameters on the deposited nitride layer

Figure 3 shows the microhardness profile beneath the surface of AISI H13 steel after SM-SEDM. The higher hardness values (above 800 $HV_{0.025}$) are observed near the surface and are related to the RL. These values decrease rapidly, indicating the transition zone between the RL and HAZ. However, the decrease in hardness is steeper for the samples obtained using $T_{ON} = 100 \ \mu s$ and resulted thinner hardened layers. In contrast, thicker hardened layers were obtained after SM-SEDM with $T_{ON} = 500 \ \mu s$. This behavior can be explained by to the higher discharge energy $(I_p \times T_{ON} \times U)$ applied to the samples produced using $T_{\rm ON} = 500 \ \mu s$, which promotes a deeper modification beneath the surface. In general, the hardness values stabilize at approximately 285 $HV_{0.025}$ at distance of 61.4 µm from the surface. Thus, the hardness limit (H limit) of 340 HV_{0.025} was considered as the baseline to define the depth of the hardened layer after nitriding, according to DIN [33] and ISO [34] criteria (H limit = core actual hardness + 50 HV, rounded by 10 HV).

Based on the microhardness profiles of each set of parameters (I to VI) studied in this work, Table 4 shows the values of the depth of hardened layer after nitriding using SM-SEDM. The depth of hardened layer with $T_{\rm ON} = 500 \ \mu s$ is larger than with $T_{\rm ON} = 100 \ \mu s$. This corroborates with the data by [9, 11, 12, 35], who observed that higher discharge energy ($I_{\rm p} \times T_{\rm ON} \times U$) results in thicker RL and thicker HAZ. $T_{\rm ON}$ is directly proportional to the thickness of the HAZ [9].

Table 5 shows the analysis of variance (ANOVA) for the depth of the hardened layer after nitriding with 95% confidence level ($\alpha = 0.05$). The factor represented by samples with different T_{ON} and T_{OFF} presented a probability value (*p* value) of 0.000, which leads to considering that the obtained results

Fig. 2 AISI H13 steel: **a** SEM microstructure image, **b** semi quantitative chemical composition (wt. and atomic %) by EDS, and **c** X ray diffractogram







Fig. 3 Microhardness profile of AISI H13 after SM SEDM with different $T_{\rm ON}$ and $T_{\rm OFF}$

Table 4Depth of the hardened layer after nitriding with different T_{ON} and T_{OFF}

Set of parameters	Ι	II	III	IV	V	VI
Roughing						
Pulse on time, $T_{\rm ON}$ [µs]	100	100	100	100	500	500
Pulse off time, T_{OFF} [µs]	43	100	150	203	500	1015
Finishing						
Pulse on time, $T_{\rm ON}$ [µs]	10	10	10	10	50	50
Pulse off time, T_{OFF} [µs]	4	10	15	20	50	102
Nitride depth [μ m] at H limit of 340 HV _{0.025}	40.4 ± 3.8	31.0 ± 4.6	40.3 ± 1.7	35.6 ± 0.6	53.2 ± 5.5	47.2 ± 1.4

are statistically significant [36]. Therefore, the null hypothesis (H_0) can be rejected in favor of the alternative hypothesis (H_1) , which indicates that there is a relation between T_{ON} and T_{OFF} with the depth of the hardened layer produced by nitriding via SM-SEDM.

Based on the Tukey test shown in Table 6, the selected sets of parameters (I to VI) resulted in significant differences in relation to the thickness of the hardened lay, represented by the response data "thickness of the hardened layer after nitriding". The sets of parameters V and VI employed a $T_{\rm ON}$ of 500 µs for roughing followed by 50 µs for finishing, which resulted in highest thickness mean values, while the remaining sets of parameters used a $T_{\rm ON}$ of 100 µs for roughing and 10 µs for finishing. This indicates that $T_{\rm ON}$ influences the thickness of the hardened layer after nitriding by SM-SEDM.

Figure 4 presents the boxplot of depth of the hardened layer after nitriding by SM-SEDM, based on the hardness limit (H limit) of 340 HV_{0.025}, versus the set of parameters (I to VI) with different pulse-on times and pulse-off times (T_{ON}/T_{OFF}). The samples with $T_{ON} = 100 \ \mu$ s showed lower nitride layer depth, while the samples with T_{ON}/T_{OFF} of 500/500 μ s showed higher average value (53.2 μ m), followed by T_{ON}/T_{OFF} of 500/1015 μ s (47.2 μ m). This can be attributed to the higher discharge energy during nitriding of the samples with $T_{ON} = 500 \ \mu$ s. Since during the processing of the samples with the highest T_{OFF} (1015 μ s) in combination with the other parameters used on this work (according to Table 1), it was noticed that there was arcing effect and instability. These results corroborate the work by Chounde and Pawar [2], who state that the higher I_p will result in increasing pulse discharge energy. Moreover, an inappropriate combination of $T_{\rm ON}$, I_p , and duty cycle ($D_{\rm T}$), which is related to $T_{\rm OFF}$, may cause arcing effect or instability. Thus, despite the fact that the depth of the hardened layer after nitriding of the samples produced with $T_{\rm ON}/T_{\rm OFF}$ of 500/500 µs against 500/1015 µs is statistically similar. The results obtained under the former condition are considered the most appropriate to produce the thickest hardened layer by SM-SEDM with urea solution, as it does not promote arcing effect.

Figure 5 shows the X-ray diffractograms (θ -2 θ configuration) performed after SM-SEDM with different $T_{\rm ON}$ and $T_{\rm OFF}$ compared with the AISI H13 steel base material. Some peaks for SM-SEDM samples at 43.0°, 50.7°, 74.5°, and 90.2° 2 θ , as well as change of peaks with 2 θ of 44.6°, 64.7°, 82.1°, and 98.6°, were recorded. There is an appreciable similarity for all XRD spectra of SM-SEDM samples. The major phase of the surface layer after SM-SEDM is iron nitride (Fe₁₁N), originated from the FeN_{0.0950} phase [37], space group Fm-3 m (225). The presence of iron nitride justifies the increase in hardness (shown in Fig. 3) on the surface and subsurface after the SM-SEDM treatment.

In order to perform further characterization of the layers produced by SM-SEDM, it was used optical microscopy analysis and Vickers microhardness measurements. Figure 6 presents images of AISI H13 sample cross section after SM-SEDM. The analysis of the images indicates a lighter layer on the surface and a second layer just below it. These layers were produced during the SM-SEDM and they represent the RL and the HAZ, respectively. The formation of a more

Table 5 Analysis of variance (ANOVA) for hardened layer thickness

Source of variation	Degrees of freedom	Sum of squares	Contribution ratio [%]	Mean square	F value	p value
Samples	5	1297.00	90.19	259.40	22.05	0.000
Error	12	141.10	9.81	11.76		
Total	17	1438.20	100.00			
R^2	90.19%					

 $H_0: \beta_1 = 0$ $H_1: \beta_1 \neq 0$

Set of parameters	$T_{\rm ON}/T_{\rm OFF}$ parameter		Thickness of the hardened layer after nitriding [µ		
	Roughing	Finishing			
V	500/500	50/50	$53.2^{a} \pm 5.5$		
VI	500/1015	50/102	$47.2^{a} \pm 1.4$		
III	100/150	10/15	$40.3^{b} \pm 1.7$		
Ι	100/43	10/4	$40.4^{b} \pm 3.8$		
IV	100/203	10/20	$35.6^{b} \pm 0.6$		
II	100/100	10/10	$31.0^{b} \pm 4.6$		

Table 6	Tukey test for	thickness	of the	hardened	layer	after	nitriding
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*Means that do not share a letter (^{a, b}) are significantly different by the Tukey method and 95% confidence

irregular RL and HAZ thickness is observed in Fig. 6 a, b, c and d in comparison to Fig. 6 e and f. Irregularities of melted and resolidified material on the surface of samples in Fig. 6 a, b, c, d and f are observed. That indicates that the T_{ON} and T_{OFF} are not adequate to meet the proposed goal of obtaining better uniformity of these layers. The uniform and thickest layer is obtained using $T_{\rm ON}$ and $T_{\rm OFF}$ both of 500 µs, as shown in Fig. 6e. Moreover, it suggests that the combination of high $T_{\rm ON}$ with high $T_{\rm OFF}$ contributes to a higher electrical discharge energy while it provides favorable conditions for the working gap (e.g., off time for adequate flushing) and consequently contributed to less accumulation of particles and molten material, deposited irregularly on the workpiece surface. The test configuration with $T_{ON} = 500$ and $T_{OFF} = 1015 \ \mu s$ shown in Fig. 6f presents thicker layer, though not uniform. This suggests that arcing effect and instability occur during the SM-SEDM of those samples. The distance between electrode and the workpiece became unstable for T_{OFF} values greater than $T_{\rm ON}$ combined with the $I_{\rm p}$ used, thus confirming the work by Chounde and Pawar [2]. When the electrical discharge becomes unstable, it is necessary to reduce I_p . Furthermore, as a reduction on $D_{\rm T}$ means an increasing on the $T_{\rm OFF}$, resulting in lower operation efficiency, since the workpiece material



Fig. 4 Boxplot of depth of the hardened layer for samples with different $T_{\rm ON}/T_{\rm OFF}$ (I 100/43; II 100/100; III 100/150; IV 100/203; V 500/500; and VI 500/1015)

removal rate is reduced [2, 38]. Thus, tests with $D_{\rm T}$ lower than 33% were not performed in this work.

It can also be noticed in the images of Fig. 6 that at the top, the layer of the sample cross section has a "white" aspect, that layer is related to the RL, while below the RL, the HAZ is present and the aspect is darker than the AISI H13 steel base material. The new phase found after SM-SEDM, $Fe_{11}N$ shown in Fig. 5, was obtained from these regions. However, the limits where there is no further formation of the new phase are not well established and, therefore, the microhardness profile was used to determine the depth of hardened layer after nitriding by SM-SEDM.

Figure 7 shows the results of the microhardness measurements performed on each cross section of the samples after SM-SEDM with different T_{ON} and T_{OFF} . The microhardness of the AISI H13 steel samples core was 285.7 ± 2.4 HV_{0.025}, as represented by the dashed red line in Fig. 7. However, the hardness values of the RL and HAZ layers are significantly higher than those of the core of AISI H13. The microhardness of RL is nearly 298% higher than that of the AISI H13 core. HAZ has a microhardness of approximately 198% higher than the AISI H13 core. These results indicate that surface hardening may be due to enrichment by nitrides on AISI H13 steel. In all investigated situations, the microhardness of RL is higher than that of HAZ, which is higher than that of the AISI H13 steel base material. These results suggest that the nitrogen concentration is higher on the surface than on the subsurface of AISI H13 steel after SM-SEDM.

The thicknesses of the recast layer ($t_{\rm RL}$) and the altered metal zone ($t_{\rm AMZ}$) were evaluated by optical microscopy. Figure 8 shows the average results of 20 measurements for each test condition (I to VI), represented by different $T_{\rm ON}$ and $T_{\rm OFF}$ for roughing. It is observed, as previously discussed, that the samples with $T_{\rm ON}/T_{\rm OFF}$ of 500/500 µs and 500/1015 µs provide the highest layer thickness. According to ASM HANDBOOK [23], the thicker the nitride layer, the higher the wear resistance because the resistance against abrasion increases, thus increasing the component's life. This behavior is essential for tool steel applications subjected to abrasion **Fig. 5** XRD patterns of AISI H13 base material and after SM SEDM with different T_{ON} and T_{OFF}



wear. In addition to that, similar behavior is expected against fatigue, since the increase in nitriding depth allows higher surface pressures and constant wear resistance, therefore recommending $T_{\rm ON}/T_{\rm OFF}$ of 500/500 µs as the parameter responsible for the thickest AMZ (53.2 µm). The 500/500 µs $T_{\rm ON}/T_{\rm OFF}$ working condition is more stable compared to $T_{\rm ON}/T_{\rm OFF}$ of 500/1015 µs, as the latter showed arcing effect and instability.

4 Conclusions

This work has studied the effects of process parameters [(pulse-on time (T_{ON}) and pulse-off time (T_{OFF})] on surface

modification by die-sinking EDM of annealed AISI H13 tool steel using a solution of urea in deionized water as dielectric fluid at 30 g L⁻¹ concentration. Six different sets of samples were produced and compared regarding depth of hardened layer and heat-affected zone (HAZ) uniformity and hardness, thicknesses of the recast layer ($t_{\rm RL}$), and the altered metal zone ($t_{\rm AMZ}$). Concluding remarks of the experimental work can be stated as follows:

- T_{ON} and T_{OFF} parameters influence the results of RL and AMZ thicknesses and uniformity of AISI H13 steel processed by SM-SEDM.
- The most favorable test configuration for thicker and uniform RL and HAZ studied in this work were $T_{\rm ON}/T_{\rm OFF}$ of



Fig. 6 AISI H13 steel cross section after SM SEDM treatment with different $T_{\rm ON}$ and $T_{\rm OFF}$

Fig. 7 Effect of T_{ON} and T_{OFF} on microhardness



500/500 μs and 500/1015 μs. The $T_{\rm ON}/T_{\rm OFF}$ of 500/ 500 μs sample configuration presented the thickest hardened layer (53.2 μm at the hardness limit of 340 HV_{0.025}), followed by $T_{\rm ON}/T_{\rm OFF}$ of 500/1015 μs (47.2 μm). This can be attributed to the higher discharge energy with $T_{\rm ON}$ = 500 μs.

- SM-SEDM increased the surface hardness approximately three times compared with the non-processed AISI H13 steel, which can be attributed to the presence of iron nitrides, such as Fe₁₁N, found in the samples.
- The presence of nitrides on the AISI H13 steel samples after SM-SEDM is evident. The main nitride identified in this study is the Fe₁₁N type, inferring dissociation of urea that introduces atomic nitrogen into the steel surface and subsurface.

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Fig. 8 Effect of T_{ON} and T_{OFF} on RL and AMZ thickness

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Authors' contributions Sinval Pedroso has provided the conceptualiza tion, methodology, performed the experiments, and writing the manu script. Alexandre Abrão, Ernane Silva, Marcelo Câmara, and Peter Weidler provided the review and contributed to the technical discussion. Peter Weidler contributed to the XRD investigation and analysis. Ernane Silva and Marcelo Câmara provided the supervision.

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Compliance with ethical standards

Conflict of interest The authors declare that they have no conflict of interest.



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