

# HIGH-TEMPERATURE SUPERCONDUCTORS

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The following tables list selected physical properties of most known high-temperature superconductors as of the date of compilation in 2020. The classification as a “high-temperature” superconductor is open to some degree of ambiguity. Naming was driven by excitement at the prospect of operation above the boiling point of liquid nitrogen ( $\sim 77$  K). However, the reality of superconductor applications today is that few can operate in this regime, while it is clearly illogical to exclude closely related materials of the same family that fall below such an arbitrary cut-off. Instead, we adopt here the classification that a “high-temperature” superconductor is a non-BCS type superconductor that is thereby able to operate beyond the BCS temperature limit of  $\sim 30$  K. Until relatively recently, this limited the known materials to cuprate compounds; however, this has now been extended first by magnesium diboride and then by the extensive families of iron-based superconductors. We do not, however, include all unconventional superconductors, intentionally excluding those isolated examples and families where no related material is able to superconduct at an elevated temperature. Notably, this excludes all the so-called “heavy fermion” superconductors thus far discovered. For the same reason, we also exclude the organic superconductors, which are featured elsewhere in this volume.

The high-temperature superconductors are presented here in a series of tables, one for each distinctly identifiable family. Table 1 lists the detailed superconducting properties of a select number of materials drawn from families that have been studied in depth due to either importance or accessibility. Table 2 contains the “214” phases including  $(\text{La,Ba})_2\text{CuO}_4$ , which was the original 1986 Bednorz and Müller discovery of high-temperature superconductivity. Here are also found the rare examples of electron-doped cuprates. Table 3 presents the rare-earth barium cuprates, including the most famous of the high-temperature superconductors,  $\text{YBa}_2\text{Cu}_3\text{O}_7$ . These materials are the core constituents of so-called second-generation commercial high-temperature superconducting wires. Table 4 supplements these materials with derivatives having missing rare-earth planes, while Tables 5 and 6 feature two distinct homologous series of cuprates based on Hg, Tl, Pb, Bi, and others, of which  $\text{Bi}_2\text{Ba}_2\text{Ca}_2\text{Cu}_3\text{O}_{10}$  is the material used in the production of first-generation commercial high-temperature superconducting wires. Tables 7 and 8 move from the cuprates to the

chalcogenide and pnictide families of iron-based superconductors, respectively, and Table 9 contains the few other non-cuprate high-temperature superconductors known, including  $\text{MgB}_2$ .

In compiling and modernizing these data tables, we have sought to be comprehensive in our coverage of the known materials, but selective in the properties to include. In guiding the latter selection, we aim to feature only those properties that are of genuine stand-alone use as figures of merit or in a comparative sense. Because no mere table of data can hope to adequately convey the nuances of materials as complex in structure and properties as the high-temperature superconductors, we avoid qualifications and list in each case the best value to which we have access. This matter of judgment must be supplemented by the reader referring in each case to a primary source to validate the entry. Lattice parameters have been rounded to four decimal places as a compromise between accuracy and generality. Critical temperatures are given to two significant figures for the same reason. The data given in Table 1 are ranges or averages of the most reliable values found in the literature.

These tables are current through 2020 and constitute a complete revision and update of previous data. The data listed here are compiled from a large number of sources spanning the primary literature. Individual references for each datum are too numerous to list. Secondary sources, as listed under the references, have been used to ensure the accuracy and completeness of the work.

## References

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TABLE 1. Detailed Superconducting Properties of Selected High-Temperature Superconductors

Mol. form.	$T_c^{\text{max}}/\text{K}$	Coherence lengths		Penetration depths		Critical fields			
		$\xi_{\text{ab}}/\text{nm}$	$\xi_c/\text{nm}$	$\lambda_{\text{ab}}/\text{nm}$	$\lambda_c/\mu\text{m}$	$B_{\text{c1}}^{\text{ab}}/\text{mT}$	$B_{\text{c1}}^{\text{c}}/\text{mT}$	$B_{\text{c2}}^{\text{ab}}/\text{T}$	$B_{\text{c2}}^{\text{c}}/\text{T}$
$\text{La}_{1.85}\text{Sr}_{0.15}\text{CuO}_4$	39	$2.2 \pm 0.1$	0.06-0.3	$219 \pm 10$	<5	7	30	$84 \pm 6$	$70 \pm 10$
$\text{Nd}_{1.85}\text{Ce}_{0.15}\text{CuO}_{4-\delta}$	30	$7.0 \pm 1.0$	0.23	$76 \pm 5$				>100	$7 \pm 1$
$\text{YBa}_2\text{Cu}_3\text{O}_{7-\delta}$	93	1.5	0.14-0.3	$90 \pm 10$	0.64	3-5	30-50	$240 \pm 25$	$150 \pm 30$
$\text{YBa}_2\text{Cu}_4\text{O}_8$	81	1.9	0.2-1.1	130-200	0.16	34	17-32		90
$(\text{Cu}_{0.5}\text{C}_{0.5})\text{Ba}_2\text{Ca}_3\text{Cu}_4\text{O}_{11+\delta}$	117	1.6	1.0	120	0.22	26	63	195	121
$\text{HgBa}_2\text{CuO}_{4+\delta}$	97	2.1	1.2	120-200	0.45	8.2	12.9	125	70-100
$\text{HgBa}_2\text{CaCu}_2\text{O}_{6+\delta}$	127	$1.5 \pm 0.1$	0.4	190	0.83	21	50		110-170
$\text{HgBa}_2\text{Ca}_2\text{Cu}_3\text{O}_{8+\delta}$	135	$1.3 \pm 0.1$		$180 \pm 30$	3.5	5-10	$35 \pm 10$	$350 \pm 50$	100-200

Mol. form.	$T_c^{\max}/\text{K}$	Coherence lengths		Penetration depths		Critical fields			
		$\xi_{ab}/\text{nm}$	$\xi_c/\text{nm}$	$\lambda_{ab}/\text{nm}$	$\lambda_c/\mu\text{m}$	$B_{c1}^{ab}/\text{mT}$	$B_{c1}^c/\text{mT}$	$B_{c2}^{ab}/\text{T}$	$B_{c2}^c/\text{T}$
$\text{Ti}_2\text{Ba}_2\text{CuO}_{6+\delta}$	92	5	0.2	170±10	2.0		6	300	65
$\text{Ti}_2\text{Ba}_2\text{CaCu}_2\text{O}_{8+\delta}$	119	3	0.7	180±40	>25	60	28	>120	>100
$\text{Ti}_2\text{Ba}_2\text{Ca}_2\text{Cu}_3\text{O}_{10+\delta}$	128	1-3	<0.09	196±10	>20			200	>75
$\text{Bi}_2\text{Sr}_2\text{CaCu}_2\text{O}_{8+\delta}$	96	0.8-0.9	≤0.05	150±20	10-40	0.25-0.5	6-10	>250	220±30
$\text{Bi}_2\text{Sr}_2\text{Ca}_2\text{Cu}_3\text{O}_{10+\delta}$	110	0.6-0.9	0.02-0.09	120±10	1.0	0.94	54	>250	184
$\text{Fe}_{1+\delta}\text{Se}$	8.5	4.7	2.8	400			2.5	25	15
$\text{FeSe}_{0.5}\text{Te}_{0.5}$	15	2.8	3.0	460±100	1.1±0.3	2.2	4.5	42	45
$(\text{Li},\text{Fe})\text{OHFeSe}$	42	2.7	0.24	200	1.8		4.5	98	67
$\text{LiFeAs}$	18	4.4±0.4	1.8±0.2	200	0.25	16	19	30	24
$\text{LaFeAsO}_{1-x}\text{F}_x$	26	2.8	1.0±0.1	310			6	63	42
$\text{NdFeAsO}_{1-x}\text{F}_x$	47	2.2±0.2	0.4±0.1	200		18	25	130	70
$\text{Ba}_{0.6}\text{K}_{0.4}\text{Fe}_2\text{As}_2$	38	1.6	0.75	200				90	75
$\text{Ba}(\text{Fe}_{0.92}\text{Co}_{0.08})_2\text{As}_2$	22	2.45	1.4±0.1	200			20	65	55
$\text{BaFe}_2(\text{As}_{0.7}\text{P}_{0.3})_2$	31	2.3	1.3±0.2	300			63	90	60
$\text{CaKFe}_4\text{As}_4$	35	1.83±0.03	1.87±0.06	100		120	25	95	102
$\text{MgB}_2$	39	10±1	2.0±0.2	85±6	0.09±0.01	35±5		17±2	3.0±0.5

TABLE 2. Crystal Structures and Critical Temperatures of the Infinite Layer and “214” High-Temperature Superconductors

Mol. form.	Structure (lattice parameters in nm)	$T_c^{\max}/\text{K}$	Comments
<b>0011, i.e., infinite layer structures P4/mmm (BaCuO<sub>2</sub> structure)</b>			
$\text{Sr}_{1-x}\text{CuO}_2$	$a = 0.3927, c = 0.3435$	60	
$\text{Sr}_{1-x}\text{Ca}_x\text{CuO}_2$	$a = 0.3902, c = 0.3350$	110	$x_{\max} = 0.3$
$\text{Sr}_{1-x}\text{Ba}_x\text{CuO}_2$	$a = 0.3922, c = 0.343$	90	$x_{\max} = 0.1$
$\text{Sr}_{1-x}\text{La}_x\text{CuO}_2$	$a = 0.3951, c = 0.3409$	42	$x_{\max} = 0.1$
$\text{Sr}_{1-x}\text{Pr}_x\text{CuO}_2$	$a = 0.3942, c = 0.3391$	43	$x_{\max} = 0.12$
$\text{Sr}_{1-x}\text{Nd}_x\text{CuO}_2$	$a = 0.3935, c = 0.3413$	45	$x_{\max} = 0.12$
$\text{Sr}_{1-x}\text{Sm}_x\text{CuO}_2$	$a = 0.3942, c = 0.3391$	42	$x_{\max} = 0.1$
<b>0201, i.e., “214” T structures I4/mmm (K<sub>2</sub>NiF<sub>4</sub> structure)</b>			
$\text{La}_2\text{CuO}_{4+\delta}$	$a = 0.379, c = 1.319$	0	
$\text{La}_2\text{CuO}_{4+\delta}$	Bmab; $a = 0.5345, b = 0.5433, c = 1.3252$	42	
$\text{La}_2\text{CuO}_4\text{F}_x$	Phase uncertain	35	
$(\text{Ca}_{1-x}\text{Na}_x)_2\text{CuO}_2\text{Cl}_2$	$a = 0.3855, c = 1.510$	26	$x_{\max} = 0.04$
$(\text{La}_{1-x}\text{Na}_x)_2\text{CuO}_4$	$a = 0.3775, c = 1.3170$	30	$x_{\max} = 0.3-0.5$
$(\text{La}_{1-x}\text{K}_x)_2\text{CuO}_4$	$a = 3.7683, c = 1.3259$	41	
$(\text{La}_{1-x}\text{Rb}_x)_2\text{CuO}_4$	$a = 0.3878, c = 1.3276$	22	
$(\text{La}_{1-x}\text{Ca}_x)_2\text{CuO}_4$	Cmca; $a = 0.5341, b = 0.5359, c = 1.3170$	24	$x_{\max} = 0.05$
$(\text{La}_{1-x}\text{Sr}_x)_2\text{CuO}_4$	$a = 0.3779, c = 1.3226$	39	$x_{\max} = 0.075$
$(\text{La}_{1-x}\text{Ba}_x)_2\text{CuO}_4$	$a = 0.3779, c = 1.323$	28	
<b>0021, i.e., “214” T' structures I4/mmm (Nd<sub>2</sub>CuO<sub>4</sub> structure)</b>			
$(\text{La}_{1-x}\text{RE}_x)_2\text{CuO}_{4+\delta}$		up to 21	Thin films; RE = Y, Lu, Sm, Eu, Gd, Tb
$(\text{La}_{1-x}\text{Ce}_x)_2\text{CuO}_{4-\delta}$	$a = 0.4007, c = 1.244$	30	Electron-doped, $x_{\max} = 0.065$
$(\text{Pr}_{1-x}\text{Ce}_x)_2\text{CuO}_{4-\delta}$		17	Electron-doped
$\text{Pr}_{1-x}\text{Ce}_{0.15}\text{Sr}_x\text{CuO}_{3.94}$	$a = 0.396, c = 1.216$	21	Electron-doped, $x_{\max} = 0.06$
$(\text{Nd}_{1-x}\text{Ce}_x)_2\text{CuO}_{4-\delta}$	$a = 0.3947, c = 1.2078$	30	Electron-doped, $x_{\max} = 0.075$
$\text{Nd}_2\text{CuO}_{4-\delta}\text{F}_y$	$a = 0.3951, c = 1.2115$	25	Electron-doped, $x_{\max} = 0.2$
$(\text{Sm}_{1-x}\text{Ce}_x)_2\text{CuO}_{4-\delta}$		15	Electron-doped
$(\text{Eu}_{1-x}\text{Ce}_x)_2\text{CuO}_{4-\delta}$		13	Electron-doped, $x_{\max} = 0.075-0.085$
<b>0222, i.e., “214” T* structures P4/nmm (alternating T and T')</b>			
$\text{La}_{1-x/2}\text{Eu}_{1-x/2}\text{Sr}_x\text{CuO}_4$	$a = 0.3871, c = 1.2597$	25	$x_{\max} = 0.14$
$(\text{Nd},\text{Sr},\text{Ce})_2\text{CuO}_4$	$a = 0.3856, c = 1.2484$	25	
$\text{SmLa}_{1-x}\text{Sr}_x\text{CuO}_{4-\delta}$		33	$x_{\max} = 0.15$

**TABLE 3. Crystal Structures and Critical Temperatures of the  $RE_2Ba_4Cu_{5+n}O_{13+n}$  Series of High-Temperature Superconductors**

Mol. form.	Structure (lattice parameters in nm)	$T_c^{\max}/K$	Comments
<b>123 (<math>n = 1</math>)</b>			
<b>Pmmm</b>			
YBa <sub>2</sub> Cu <sub>3</sub> O <sub>7-<math>\delta</math></sub>	$a = 0.3820, b = 0.3885, c = 1.1676$	93	
YSr <sub>2</sub> Cu <sub>3</sub> O <sub>7-<math>\delta</math></sub>	P4/mmm, $a = 0.3786, c = 1.1386$	63	
YCa <sub>2</sub> Cu <sub>3</sub> O <sub>7-<math>\delta</math></sub>	$a = 0.3643, b = 0.3838, c = 1.1759$	84	
La <sub>1+x</sub> Ba <sub>2-x</sub> Cu <sub>3</sub> O <sub>7-<math>\delta</math></sub>	$a = 0.3885, b = 0.3938, c = 1.1817$	93	$x > 0.25$ for phase stability
CeBa <sub>2</sub> Cu <sub>3</sub> O <sub>7-<math>\delta</math></sub>	Forms BaCeO <sub>3</sub> due to tetravalent Ce	—	
Pr <sub>1+x</sub> Ba <sub>2-x</sub> Cu <sub>3</sub> O <sub>7-<math>\delta</math></sub>	$a = 0.3866, b = 0.3933, c = 1.1724$	0	$x > 0.15$ for phase stability
Nd <sub>1+x</sub> Ba <sub>2-x</sub> Cu <sub>3</sub> O <sub>7-<math>\delta</math></sub>	$a = 0.3878, b = 0.3913, c = 1.1753$	94	$x > 0.1$ for phase stability
PmBa <sub>2</sub> Cu <sub>3</sub> O <sub>7-<math>\delta</math></sub>	Not synthesized due to radioactive Pm	—	
SmBa <sub>2</sub> Cu <sub>3</sub> O <sub>7-<math>\delta</math></sub>	$a = 0.3902, b = 0.3844, c = 1.1725$	95	
EuBa <sub>2</sub> Cu <sub>3</sub> O <sub>7-<math>\delta</math></sub>	$a = 0.3897, b = 0.3838, c = 1.1707$	95	
GdBa <sub>2</sub> Cu <sub>3</sub> O <sub>7-<math>\delta</math></sub>	$a = 0.3895, b = 0.3835, c = 1.1699$	95	
TbBa <sub>2</sub> Cu <sub>3</sub> O <sub>7-<math>\delta</math></sub>	Forms BaTbO <sub>3</sub> by Ba-Tb intermixing	—	
TbSr <sub>2</sub> Cu <sub>2.7</sub> Mo <sub>0.3</sub> O <sub>7-<math>\delta</math></sub>	P4/mmm, $a = 0.3871, c = 1.15784$	80	Mo necessary for phase stability
DyBa <sub>2</sub> Cu <sub>3</sub> O <sub>7-<math>\delta</math></sub>	$a = 0.3887, b = 0.3825, c = 1.1686$	93	
HoBa <sub>2</sub> Cu <sub>3</sub> O <sub>7-<math>\delta</math></sub>	$a = 0.3885, b = 0.3819, c = 1.1677$	93	
ErBa <sub>2</sub> Cu <sub>3</sub> O <sub>7-<math>\delta</math></sub>	$a = 0.3878, b = 0.3813, c = 1.1664$	93	
TmBa <sub>2</sub> Cu <sub>3</sub> O <sub>7-<math>\delta</math></sub>	$a = 0.3875, b = 0.3809, c = 1.1666$	90	
YbBa <sub>2</sub> Cu <sub>3</sub> O <sub>7-<math>\delta</math></sub>	$a = 0.3871, b = 0.3802, c = 1.1658$	90	Multiphase due to small Yb ion
LuBa <sub>2</sub> Cu <sub>3</sub> O <sub>7-<math>\delta</math></sub>	$a = 0.387, b = 0.380, c = 1.169$	93	Mostly multiphase due to small Lu ion
<b>247 (<math>n = 2</math>)</b>			
<b>Ammm</b>			
Y <sub>2</sub> Ba <sub>4</sub> Cu <sub>7</sub> O <sub>15-<math>\delta</math></sub>	$a = 0.3854, b = 0.3874, c = 5.040$	55	
Pr <sub>2</sub> Ba <sub>4</sub> Cu <sub>7</sub> O <sub>15-<math>\delta</math></sub>	$a = 0.3892, b = 0.3902, c = 5.081$	10	
Nd <sub>2</sub> Ba <sub>4</sub> Cu <sub>7</sub> O <sub>15-<math>\delta</math></sub>	$a = 0.3894, b = 0.3901, c = 5.079$	40	
Eu <sub>2</sub> Ba <sub>4</sub> Cu <sub>7</sub> O <sub>15-<math>\delta</math></sub>	$a = 0.3879, b = 0.3886, c = 5.039$	45	
Gd <sub>2</sub> Ba <sub>4</sub> Cu <sub>7</sub> O <sub>15-<math>\delta</math></sub>	$a = 0.3872, b = 0.3879, c = 5.036$	45	
Dy <sub>2</sub> Ba <sub>4</sub> Cu <sub>7</sub> O <sub>15-<math>\delta</math></sub>	$a = 0.3864, b = 0.3879, c = 5.039$	50	
Ho <sub>2</sub> Ba <sub>4</sub> Cu <sub>7</sub> O <sub>15-<math>\delta</math></sub>	$a = 0.3857, b = 0.3879, c = 5.040$	55	
Er <sub>2</sub> Ba <sub>4</sub> Cu <sub>7</sub> O <sub>15-<math>\delta</math></sub>	$a = 0.3847, b = 0.3873, c = 5.044$	55	
Yb <sub>2</sub> Ba <sub>4</sub> Cu <sub>7</sub> O <sub>15-<math>\delta</math></sub>	$a = 0.381, b = 0.386, c = 5.045$	86	
<b>124 (<math>n = 3</math>)</b>			
<b>Ammm</b>			
YBa <sub>2</sub> Cu <sub>4</sub> O <sub>8</sub>	$a = 0.3840, b = 0.3870, c = 2.7231$	81	
LaBa <sub>2</sub> Cu <sub>4</sub> O <sub>8</sub>	$a = 0.3854, b = 0.3874, c = 2.710$	?	Superconductivity not reported
CeBa <sub>2</sub> Cu <sub>4</sub> O <sub>8</sub>	No reports on phase formation	—	
PrBa <sub>2</sub> Cu <sub>4</sub> O <sub>8</sub>	$a = 0.3884, b = 0.3903, c = 2.7293$	0	
NdBa <sub>2</sub> Cu <sub>4</sub> O <sub>8</sub>	Small volume fraction, multiphase	~57	
PmBa <sub>2</sub> Cu <sub>4</sub> O <sub>8</sub>	Not synthesized due to radioactive Pm	?	
SmBa <sub>2</sub> Cu <sub>4</sub> O <sub>8</sub>	$a = 0.3872, b = 0.3886, c = 2.7308$	69	
EuBa <sub>2</sub> Cu <sub>4</sub> O <sub>8</sub>	$a = 0.3865, b = 0.3884, c = 2.7279$	78	
GdBa <sub>2</sub> Cu <sub>4</sub> O <sub>8</sub>	$a = 0.3863, b = 0.3881, c = 2.7259$	74	
TbBa <sub>2</sub> Cu <sub>4</sub> O <sub>8</sub>	Single synthesis attempt	?	Superconductivity not reported
DyBa <sub>2</sub> Cu <sub>4</sub> O <sub>8</sub>	$a = 0.3846, b = 0.3873, c = 2.7237$	77	
HoBa <sub>2</sub> Cu <sub>4</sub> O <sub>8</sub>	$a = 0.3840, b = 0.3870, c = 2.7221$	80	
ErBa <sub>2</sub> Cu <sub>4</sub> O <sub>8</sub>	$a = 0.3837, b = 0.3869, c = 2.7230$	78	
TmBa <sub>2</sub> Cu <sub>4</sub> O <sub>8</sub>	$a = 0.3827, b = 0.3864, c = 2.718$	79	
YbBa <sub>2</sub> Cu <sub>4</sub> O <sub>8</sub>	$a = 0.3846, b = 0.3871, c = 2.7231$	80	
Yb(Ba <sub>0.8</sub> Sr <sub>0.2</sub> ) <sub>2</sub> Cu <sub>4</sub> O <sub>8</sub>	$a = 0.3805, b = 0.3855, c = 2.7072$	83	Sr improves phase stability
(Yb <sub>0.95</sub> Ca <sub>0.05</sub> )Ba <sub>2</sub> Cu <sub>4</sub> O <sub>8</sub>	$a = 0.3822, b = 0.3853, c = 2.7175$	85	Ca improves phase stability
LuBa <sub>2</sub> Cu <sub>4</sub> O <sub>8</sub>	$a = 0.3844, b = 0.3871, c = 2.7225$	82	

**TABLE 4. Crystal Structures and Critical Temperatures of the  $RE_nBa_mCu_{n+m}O_{2(n+m+1)}$  Series of High-Temperature Superconductors**

Mol. form.	Structure (lattice parameters in nm)	$T_c^{\max}/K$	Comments
<b><math>n = 2</math></b>			
<b>Pmm2</b>			
$YBa_2Cu_3O_{7-\delta}$	$a = 0.3820, b = 0.3885, c = 1.1676$	93	cf. Table 3 (123 compounds)
<b>Pmmm</b>			
$YBa_3Cu_4O_{9-\delta}$	$a = 0.3802, b = 0.3885, c = 1.5256$	88	
$YBa_4Cu_5O_{11-\delta}$	$a = 0.3802, b = 0.3865, c = 1.9382$	91	
$YBa_5Cu_6O_{13-\delta}$	$a = 0.3801, b = 0.3891, c = 2.2944$	85	
<b><math>n = 3</math></b>			
<b>Pmm2</b>			
$Y_3Ba_5Cu_8O_{18}$	$a = 0.3888, b = 0.3823, c = 3.1013$	102	
$Nd_3Ba_5Cu_8O_{18}$	$a = 0.3922, b = 0.3862, c = 3.5211$	95	
$Sm_3Ba_5Cu_8O_{18}$	$a = 0.3899, b = 0.3852, c = 3.5146$	97	
$Eu_3Ba_5Cu_8O_{18}$	$a = 0.3864, b = 0.3888, c = 3.113$	66	
$Gd_3Ba_5Cu_8O_{18}$	$a = 0.3851, b = 0.3877, c = 3.112$	97	
$Dy_3Ba_5Cu_8O_{18}$	$a = 0.3833, b = 0.3867, c = 3.1025$	81	
$Ho_3Ba_5Cu_8O_{18}$	$a = 0.3832, b = 0.3875, c = 3.101$	84	
$(Yb_{0.6}Sm_{0.4})_3Ba_5Cu_8O_{18}$	$a = 0.3891, b = 0.3829, c = 3.1209$	88	Sm required for phase stability
$Y_3Ba_8Cu_{11}O_{24}$	$a = 0.3814, b = 0.3885, c = 4.2699$	91	
<b><math>n = 4</math></b>			
<b>Pmm2</b>			
$Y_2Ba_3Cu_5O_{11-\delta}$	$a = 0.3820, b = 0.3868, c = 1.8930$	92	
$Gd_2Ba_3Cu_5O_{11-\delta}$	$a = 0.3882, b = 0.3874, c = 1.9355$	16	
$Y_2Ba_5Cu_7O_{15-\delta}$	$a = 0.3832, b = 0.3851, c = 2.8683$	98	
$Y_2Ba_5Cu_8O_{17-\delta}$	$a = 0.3871, b = 0.3848, c = 2.7160$	105	Extra $CuO_2$ layer
$Y_2Ba_5Cu_9O_{19-\delta}$	$a = 0.3821, b = 0.3898, c = 2.3320$	94	Extra $CuO_2$ layers
<b><math>n &gt; 4</math></b>			
<b>Pmm2</b>			
$Gd_2Ba_7Cu_{12}O_{26-\delta}$	$a = 0.3868, b = 0.3876, c = 4.6507$	22	
$Y_2Ba_8Cu_{13}O_{28-\delta}$	$a = 0.3819, b = 0.3897, c = 5.0461$	91	
$Y_7Ba_{11}Cu_{18}O_{38-\delta}$	$a = 0.3824, b = 0.3880, c = 6.9870$	91	
$Y_{13}Ba_{20}Cu_{33}O_{68-\delta}$	$a = 0.3815, b = 0.3878, c = 12.8116$	89	

**TABLE 5. Crystal Structures and Critical Temperatures of the  $M_mAE_2Ca_{n-1}Cu_nO_{2+m+2n}$  ( $M = B, Cu, Au, Hg, Tl, Pb, Bi$ ;  $AE = Ca, Sr, Ba, La$ ) Series of High-Temperature Superconductors**

Mol. form.	Structure (lattice parameters in nm)	$T_c^{\max}/K$	Comments
<b><math>02(n-1)n</math> (<math>m = 0</math>)</b>			
<b>I4/mmm</b>			
$Ba_2CuO_4$	$a = 0.385, c = 1.46$	85	Thin film; cf. Table 2 (T structures)
$Ba_2CaCu_2O_6$	$a = 0.385, c = 2.20$	103	
$Ba_2Ca_2Cu_3O_8$	$a = 0.385, c = 2.83$	126	
$Ba_2Ca_3Cu_4O_{10}$	$a = 0.385, c = 4.04$	117	
$Ba_2CaCu_2O_4(O_{1-x}F_x)_2$	$a = 0.3879, c = 1.471$	90	
$Ba_2Ca_2Cu_3O_6(O_{1-x}F_x)_2$	$a = 0.3861, c = 2.108$	120	
$Ba_2Ca_3Cu_4O_8(O_{1-x}F_x)_2$	$a = 0.3856, c = 2.745$	105	
$Ba_2Ca_4Cu_5O_{10}(O_{1-x}F_x)_2$	$a = 0.3860, c = 3.382$	90	
$Sr_2CuO_{3+\delta}$	$a = 0.3764, c = 1.2548$	70	
$Sr_2CuO_{4-\delta}$	$a = 0.3795, c = 1.2507$	95	cf. Table 2 (T structures)
$Sr_2CuO_2F_{2+\delta}$	Fmmm; $a = 0.5394, b = 0.5513, c = 1.3468$	46	
$Sr_2CaCu_2O_6$	$a = 0.386, c = 2.04$	70	
$Sr_3Cu_2O_{5+\delta}$ (i.e. $Sr_2SrCu_2O_{5+\delta}$ )	$a = 0.3902, c = 2.1085$	100	
$(Sr,Ca)_2CaCu_2O_4Cl_2$	$a = 0.387, c = 2.216$	80	
$Sr_2Ca_2Cu_3O_8$	$a = 0.386, c = 2.72$	90	
$Sr_2Ca_3Cu_4O_{10}$	$a = 0.386, c = 3.40$	70	
$(La_{1-x}Ca_x)_2CaCu_2O_{6+\delta}$	$a = 0.3821, c = 1.953$	60	$x_{\max} = 0.075$
$(La_{1-x}Sr_x)_2CaCu_2O_{6-\delta}$	$a = 0.3821, c = 1.9599$	60	$x_{\max} = 0.2$
$(La_{1-x}Sr_x)_2CaCu_2O_4Cl_2$	$a = 0.3827, c = 1.942$	45	$x_{\max} = 0.2$

Mol. form.	Structure (lattice parameters in nm)	$T_c^{\max}/\text{K}$	Comments
<b>12(n-1)n (m = 1)</b>	<b>P4/mmm</b>		
BSr <sub>2</sub> Ca <sub>3</sub> Cu <sub>3</sub> O <sub>9</sub>	$a = 0.3821, c = 1.3854$	75	
BSr <sub>2</sub> Ca <sub>3</sub> Cu <sub>4</sub> O <sub>11</sub>	$a = 0.3836, c = 1.7082$	110	
BSr <sub>2</sub> Ca <sub>4</sub> Cu <sub>5</sub> O <sub>13</sub>	$a = 0.3837, c = 2.022$	85	
CuBa <sub>2</sub> Ca <sub>2</sub> Cu <sub>3</sub> O <sub>9</sub>	Not synthesized in pure form	–	
(Cu <sub>0.5</sub> C <sub>0.5</sub> )Ba <sub>2</sub> Ca <sub>2</sub> Cu <sub>3</sub> O <sub>9</sub>	$a = 0.3859, c = 1.4766$	118	
CuBa <sub>2</sub> Ca <sub>3</sub> Cu <sub>4</sub> O <sub>11+δ</sub>	Not synthesized in pure form	–	
(Cu <sub>0.5</sub> C <sub>0.5</sub> )Ba <sub>2</sub> Ca <sub>3</sub> Cu <sub>4</sub> O <sub>11+δ</sub>	$a = 0.3855, c = 1.7930$	117	
Au(Ba,La)CuO <sub>5+δ</sub>	Pmmm; $a = 0.3798, b = 0.3851, c = 0.8575$	19	
AuBa <sub>2</sub> (Y <sub>1-x</sub> Ca <sub>x</sub> )Cu <sub>2</sub> O <sub>7</sub>	Pmmm; $a = 0.3826, b = 0.3850, c = 1.2075$	84	$x_{\max} = 0.4$
AuBa <sub>2</sub> Ca <sub>3</sub> Cu <sub>3</sub> O <sub>9</sub>	Pmmm; $a = 0.3812, b = 0.3856, c = 1.5443$	30	
AuBa <sub>2</sub> Ca <sub>3</sub> Cu <sub>4</sub> O <sub>11</sub>	Pmmm; $a = 0.3827, b = 0.3851, c = 1.8494$	99	
HgSr <sub>2</sub> CuO <sub>4+δ</sub>	Not synthesized in pure form	–	
(Hg <sub>1-x</sub> Mo <sub>x</sub> )Sr <sub>2</sub> CuO <sub>x</sub>	$a = 0.3787, c = 0.8844$	78	$x_{\max} = 0.15$
(Hg <sub>0.9</sub> Re <sub>0.1</sub> )Sr <sub>2</sub> CuO <sub>x</sub>	$a = 0.378, c = 0.884$	66	$x_{\max} = 0.15$
HgBa <sub>2</sub> CuO <sub>4+δ</sub>	$a = 0.3883, c = 0.9513$	97	
HgBa <sub>2</sub> CaCu <sub>2</sub> O <sub>6+δ</sub>	$a = 0.3853, c = 1.2637$	127	
HgBa <sub>2</sub> Ca <sub>2</sub> Cu <sub>3</sub> O <sub>8+δ</sub>	$a = 0.3850, c = 1.5784$	135	
HgBa <sub>2</sub> Ca <sub>3</sub> Cu <sub>4</sub> O <sub>10+δ</sub>	$a = 0.3854, c = 1.9006$	127	
HgBa <sub>2</sub> Ca <sub>4</sub> Cu <sub>5</sub> O <sub>12+δ</sub>	$a = 0.3851, c = 2.2136$	110	
HgBa <sub>2</sub> Ca <sub>5</sub> Cu <sub>6</sub> O <sub>14+δ</sub>	$a = 0.3851, c = 2.5251$	107	
HgBa <sub>2</sub> Ca <sub>6</sub> Cu <sub>7</sub> O <sub>16+δ</sub>	$a = 0.3851, c = 2.8406$	89	
HgBa <sub>2</sub> Ca <sub>7</sub> Cu <sub>8</sub> O <sub>18+δ</sub>	$a = 0.3847, c = 3.1583$	<90	
TlSr <sub>2</sub> CuO <sub>5-δ</sub>	Pmmm; $a = 0.3661, b = 0.3793, c = 0.899$	0	
(Tl <sub>0.5</sub> Pb <sub>0.5</sub> )Sr <sub>2</sub> CuO <sub>5±δ</sub>	$a = 0.3736, c = 0.9022$	60	
Tl(Sr,Ba) <sub>2</sub> CuO <sub>5-δ</sub>	$a = 0.3805, c = 0.9120$	43	
Tl(Sr,La) <sub>2</sub> CuO <sub>5</sub>	$a = 0.37, c = 0.9$	40	
TlSr <sub>2</sub> CaCu <sub>2</sub> O <sub>7</sub>	$a = 0.3797, c = 1.2092$	55	
(Tl <sub>0.5</sub> Pb <sub>0.5</sub> )Sr <sub>2</sub> CaCu <sub>2</sub> O <sub>7</sub>	$a = 0.3802, c = 1.2107$	85	
TlSr <sub>2</sub> (Ca <sub>0.5</sub> Y <sub>0.5</sub> )Cu <sub>2</sub> O <sub>7</sub>	$a = 0.380, c = 1.210$	90	
(Tl <sub>0.5</sub> Pb <sub>0.5</sub> )Sr <sub>2</sub> (Ca <sub>0.8</sub> Y <sub>0.2</sub> )Cu <sub>2</sub> O <sub>7</sub>	$a = 0.3808, c = 1.2014$	107	
TlBa <sub>2</sub> CuO <sub>5-δ</sub>	$a = 0.3859, c = 0.9261$	9.5	
Tl(Ba <sub>0.6</sub> La <sub>0.4</sub> ) <sub>2</sub> CuO <sub>5</sub>	$a = 0.3848, c = 0.9091$	52	
TlBa <sub>2</sub> CaCu <sub>2</sub> O <sub>7</sub>	$a = 0.3857, c = 1.2754$	103	
TlBa <sub>2</sub> Ca <sub>1-x</sub> Y <sub>x</sub> Cu <sub>2</sub> O <sub>7</sub>	$a = 0.3850, c = 1.265$	86	$x_{\max} = 0.25$
TlBa <sub>2</sub> Ca <sub>2</sub> Cu <sub>3</sub> O <sub>9</sub>	$a = 0.385, c = 1.59$	130	
TlBa <sub>2</sub> Ca <sub>3</sub> Cu <sub>4</sub> O <sub>11</sub>	$a = 0.385, c = 1.91$	122	
TlBa <sub>2</sub> Ca <sub>4</sub> Cu <sub>5</sub> O <sub>13</sub>	$a = 0.385, c = 2.23$	117	
PbSr <sub>2</sub> CuO <sub>5-δ</sub>	Tetragonal; $a = 0.381, c = 0.881$	40	Thin film
PbSr <sub>2</sub> CaCu <sub>2</sub> O <sub>7</sub>	Orthorhombic; $a = 0.381, b = 0.383, c = 1.21$	70	
PbSr <sub>2</sub> Ca <sub>2</sub> Cu <sub>3</sub> O <sub>9</sub>	Pseudotetragonal; $a = 0.3834, c = 1.529$	122	
PbSr <sub>2</sub> Ca <sub>3</sub> Cu <sub>4</sub> O <sub>11</sub>	Not synthesized in pure form	–	
(Pb <sub>0.6</sub> Sr <sub>0.3</sub> Cu <sub>0.1</sub> )Sr <sub>2</sub> (Ca,Sr) <sub>3</sub> Cu <sub>4</sub> O <sub>11</sub>	$a = 0.3833, c = 1.8442$	107	
<b>22(n-1)n (m = 2)</b>	<b>I4/mmm</b>		
Cu <sub>2</sub> Ba <sub>2</sub> Ca <sub>2</sub> Cu <sub>3</sub> O <sub>11</sub>	Not synthesized in pure form	–	
(Cu <sub>0.5</sub> C <sub>0.5</sub> ) <sub>2</sub> Ba <sub>3</sub> Ca <sub>3</sub> Cu <sub>4</sub> O <sub>13</sub>	P4/mmm; $a = 0.3855, c = 2.187$	113	Extra BaO layer
(Cu <sub>0.5</sub> C <sub>0.5</sub> ) <sub>2</sub> Ba <sub>3</sub> Ca <sub>4</sub> Cu <sub>5</sub> O <sub>15</sub>	P4/mmm; $a = 0.3857, c = 2.5067$	~110	Extra BaO layer
Hg <sub>2</sub> Ba <sub>2</sub> CaCu <sub>2</sub> O <sub>8</sub>	Unstable	–	
Hg <sub>2</sub> Ba <sub>2</sub> (Y <sub>0.6</sub> Ca <sub>0.4</sub> )Cu <sub>2</sub> O <sub>8-δ</sub>	$a = 0.3855, c = 2.894$	45	
(Hg <sub>0.7</sub> Tl <sub>0.3</sub> ) <sub>2</sub> Ba <sub>2</sub> (Y <sub>1-x</sub> Ca <sub>x</sub> )Cu <sub>2</sub> O <sub>8-δ</sub>	$a = 0.386, c = 2.90$	84	$x_{\max} = 0.4; T_c = 12 \text{ K for } x = 0$
(Hg <sub>0.6</sub> Tl <sub>0.4</sub> ) <sub>2</sub> Ba <sub>2</sub> Ca <sub>2</sub> Cu <sub>3</sub> O <sub>10</sub>	$a = 0.3840, c = 3.569$	45	
(Hg <sub>0.6</sub> Tl <sub>0.4</sub> ) <sub>2</sub> Ba <sub>2</sub> Ca <sub>3</sub> Cu <sub>4</sub> O <sub>12</sub>	$a = 0.3845, c = 4.206$	114	
Tl <sub>2</sub> Ba <sub>2</sub> CuO <sub>6+δ</sub>	$a = 0.3866, c = 2.3239$	92	
Tl <sub>2</sub> Ba <sub>2</sub> CaCu <sub>2</sub> O <sub>8+δ</sub>	$a = 0.3855, c = 2.9318$	119	
Tl <sub>2</sub> Ba <sub>2</sub> Ca <sub>2</sub> Cu <sub>3</sub> O <sub>10+δ</sub>	$a = 0.3850, c = 3.588$	128	
Tl <sub>2</sub> Ba <sub>2</sub> Ca <sub>3</sub> Cu <sub>4</sub> O <sub>12+δ</sub>	$a = 0.3581, c = 4.199$	119	

Mol. form.	Structure (lattice parameters in nm)	$T_c^{\max}/K$	Comments
$Pb_2(Sr,La)_2Cu_2O_6$	P222 <sub>1</sub> ; $a = 0.5333, b = 0.5421, c = 1.2609$	32	
$Pb_2Sr_2(Y_{1-x}Ca_x)Cu_3O_8$	Cmmm; $a = 0.5393, b = 0.5431, c = 1.5733$	80	
$Pb_2Sr_2La_{0.5}Ca_{0.5}Cu_3O_8$	Cmmm; $a = 0.5435, b = 0.5463, c = 1.5817$	70	
$Bi_2Sr_2CuO_{6+\delta}$	Cmmm; $a = 0.5361, b = 0.5370, c = 2.4369$	9	
$Bi_2(Sr_{1-x}La_x)_2CuO_{6+\delta}$	Cmmm; $a = 0.5417, b = 0.5381, c = 2.439$	33	$x_{\max} \approx 0.2$
$Bi_{1.6}Pb_{0.4}Sr_{1.6}La_{0.4}CuO_{6+\delta}$		39	
$Bi_2Sr_2CaCu_2O_{8+\delta}$	Fmmm; $a = 0.5413, b = 0.5411, c = 3.091$	96	
$Bi_2Sr_2Ca_2Cu_3O_{10+\delta}$	Fmmm; $a = 0.539, b = 0.539, c = 3.71$	110	
$(Bi_{0.8}Pb_{0.2})_2Sr_2Ca_2Cu_3O_{10+\delta}$	Fmmm; $a = 0.5413, b = 0.5413, c = 3.710$	110	Pb stabilizes phase formation
$Bi_2Sr_2Ca_3Cu_4O_{12+\delta}$		47	Thin film
$Bi_2Sr_2(Ln_{1-x}Ce_x)_2Cu_2O_{10+y}$	Cmma; $a = 0.55, b = 0.55, c = 1.79$	$\sim 25$	$Ln = Sm, Eu, Gd; x_{\max} = 0.15$

**TABLE 6. Crystal Structures and Critical Temperatures of the  $M_mAE_2RE_sCu_2O_{4+m+2s}$  ( $M = Fe, Co, Cu, Ga, Nb, Ru, Tl, Pb, Bi, AE = Sr, Ba, RE = Rare-Earth or Ca, Sr$ ) Series of High-Temperature Superconductors**

Mol. form.	Structure (lattice parameters in nm)	$T_c^{\max}/K$	Comments
<b>12s2</b>			
$FeSr_2YCu_2O_{6+\delta}$	Ortho; $a = 0.5409, b = 0.5458, c = 2.292$	60	
$FeSr_2NdCu_2O_{6+\delta}$	P4/mmm; $a = 0.3843, c = 1.1458$	50	
$CoSr_2(Y_{1-x}Ca_x)Cu_2O_{7-\delta}$	Ima2; $a = 0.540, b = 0.541, c = 2.253$	40	
$CuSr_2YCu_2O_{7-\delta}$	P4/mmm; $a = 0.3786, c = 1.1386$	63	$YSr_2Cu_3O_{7-\delta}$ , cf. Table 3 (123 compounds)
$(Cu_{1-x}Au_x)Sr_2YCu_2O_{7-\delta}$	P4/mmm; $a = 0.394, c = 1.201$	80	
$(Cu_{0.75}Mo_{0.25})Sr_2YCu_2O_{7+\delta}$	P4/mmm; $a = 0.3815, c = 1.15$	88	
$(Cu_{0.75}Mo_{0.25})Sr_2(Ce_{0.50}Y_{0.50})_2Cu_2O_{9+\delta}$	I4/mmm; $a = 0.3825, c = 1.40$	58	
$(Cu_{0.75}Mo_{0.25})Sr_2(Ce_{0.67}Y_{0.33})_3Cu_2O_{11+\delta}$	P4/mmm; $a = 0.3829, c = 1.68$	55	
$(Cu_{0.75}Mo_{0.25})Sr_2(Ce_{0.75}Y_{0.25})_4Cu_2O_{13+\delta}$	I4/mmm; $a = 0.3828, c = 1.98$	55	
$Cu(Ba, Eu)_2(Ce, Eu)_2Cu_2O_9$	I4/mmm; $a = 0.386, c = 2.848$	62	
$GaSr_2(Y_{1-x}Ca_x)Cu_2O_{7-\delta}$	Ima2; $a = 0.539, b = 0.548, c = 2.275$	70	
$(Ga_{1-y}Cu_y)Sr_2(Y_{1-x}Ca_x)Cu_2O_7$	Pmmm; $a = 0.539, b = 0.548, c = 2.281$	50	
$(Nb_{1-x}Cd_x)Sr_2EuCu_2O_{8-\delta}$	P4/mmm; $a = 0.3874, c = 1.1629$	43	$x_{\max} = 0.2$
$RuSr_2YCu_2O_{8-\delta}$	P4/mbm; $a = 0.3820, c = 1.1518$	39	Ferromagnetic superconductor
$RuSr_2SmCu_2O_{8-\delta}$	P4/mbm; $a = 0.3852, c = 1.156$	12	Ferromagnetic superconductor
$RuSr_2EuCu_2O_{8-\delta}$	P4/mbm; $a = 0.3843, c = 1.155$	36	Ferromagnetic superconductor
$RuSr_2GdCu_2O_{8-\delta}$	P4/mbm; $a = 0.3838, c = 1.153$	45	Ferromagnetic superconductor
$RuSr_2[(Eu, Gd)_{0.7}Ce_{0.3}]_2Cu_2O_{10-\delta}$	I4/mmm; $a = 0.3844, c = 2.8615$	35	Ferromagnetic superconductor
$TaSr_2(Gd_{1+x}Ce_{1-x})_2Cu_2O_9$	I4/mmm; $a = 0.3858, c = 2.881$	30	$x_{\max} = 0.6$
$TlBa_2(Eu_{1-x}Ce_x)_2Cu_2O_9$	I4/mmm; $c = 3.05$	40	
$(Pb, Cu)(Sr, Eu)_2(Eu, Ce)_2Cu_2O_9$	I4/mmm; $a = 0.3837, c = 2.901$	25	
$(Bi_{0.4}Pb_{0.35}Cu_{0.05})Sr_2(Y_{0.5}Ca_{0.5})Cu_2O_{7+\delta}$	P4/mmm; $a = 0.3819, c = 1.181$	102	
<b>22s2</b>			
	<b>P4/nmm</b>		cf. Table 5 (2212 compounds)
$Bi_2Sr_2(Y_{1-x}Ce_x)_2Cu_2O_{10}$	$a = 0.3836, c = 1.785$	20	$x = 0.18$
$Bi_2Sr_2(Nd_{1-x}Ce_x)_2Cu_2O_{10}$	$a = 0.3881, c = 1.793$	14	$x = 0.18$
$Bi_2Sr_2(Sm_{1-x}Ce_x)_2Cu_2O_{10}$	$a = 0.3863, c = 1.790$	16	$x = 0.18$
$Bi_2Sr_2(Eu_{1-x}Ce_x)_2Cu_2O_{10}$	$a = 0.3854, c = 1.788$	27	$x = 0.18$
$Bi_2Sr_2(Gd_{1-x}Ce_x)_2Cu_2O_{10}$	$a = 0.3851, c = 1.788$	34	$x = 0.18$
$Bi_2Sr_2(Dy_{1-x}Ce_x)_2Cu_2O_{10}$	$a = 0.3844, c = 1.787$	27	$x = 0.18$
$Bi_2Sr_2(Ho_{1-x}Ce_x)_2Cu_2O_{10}$	$a = 0.3840, c = 1.786$	24	$x = 0.18$

**TABLE 7. Crystal Structures and Critical Temperatures of the Chalcogenide Fe-Based Superconductors**

Mol. form.	Structure (lattice parameters in nm)	$T_c^{\max}/K$	Comments
<b>11</b>			
	<b>P4/nmm (PbO structure)</b>		
FeS	$a = 0.3680, c = 0.5031$	4.5	
$FeSe_{1-x}S_x$	$a = 0.377, c = 0.552$	11	
$Fe_{1+\delta}Se$	$a = 0.3762, c = 0.5502$	8.5	$\beta$ -FeSe
$Fe_{1-x}Cr_xSe$	$a = 0.3773, c = 0.5524$	12	$x_{\max} = 0.02$

Mol. form.	Structure (lattice parameters in nm)	$T_c^{\max}/K$	Comments
$Fe_{1-x}Nb_xSe$		14	$x_{\max} = 0.04$
$FeSe_{1-x}Te_x$	$a = 0.3798, c = 0.6038$	15	$x_{\max} = 0.4-0.6$
FeTe	$a = 0.3825, c = 0.6291$	0	Antiferromagnetic
$FeTe_{1-x}S_x$	$a = 0.3812, c = 0.6244$	9.4	$x_{\max} = 0.06-0.12$
<b>122</b>	<b>I4/mmm (ThCr<sub>2</sub>Si<sub>2</sub> structure)</b>		
$Li_xFe_2Se_2$	$a = 0.3775, c = 1.704$	44	
$Na_xFe_2Se_{2-\delta}$	$a = 0.3785, c = 1.7432$	46	
$K_xFe_2Se_2$	Multiphase	40	
$(Tl_{1-x}K_x)Fe_{2-y}Se_2$	$a = 0.388, c = 1.405$	31	$x = 0.25-0.46, y_{\max} = 0.12-0.22$
$Rb_xFe_{2-y}Se_2$		33	
$Cs_xFe_{2-y}Se_2$	$a = 0.3850, c = 1.5647$	30	
$Ca_xFe_2Se_2$	Multiphase	~40	
$Sr_xFe_2Se_2$	Multiphase	38	
$Ba_xFe_2Se_{2-\delta}$	$a = 0.3778, c = 1.6843$	40	
$Eu_xFe_2Se_2$	Multiphase	40	
$Yb_xFe_2Se_2$	Multiphase	42	
<b>Intercalated FeSe</b>			
$(C_2H_8N_2)_xFeSe$	Amma; $a = 0.3865, b = 0.3897, c = 2.1700$	30	
$(C_4H_4N_2O_2S)_{0.3}FeSe$	$c = 1.55$	48	
$(C_{40}H_{54}O_{27})_{0.3}FeSe$	$c = 1.45$	45	
$(Li,Fe)OHFeSe$	P4/nmm; $a = 0.378, c = 0.930$	42	
$A_x(NH_3)_yFe_2Se_2$	I4/mmm	up to 44	$A = Li, Na, K, Cs$
$Li_x(NH_2)_y(NH_3)_{1-y}Fe_2Se_2$	I4/mmm; $a = 0.3825, c = 1.6527$	43	$x_{\max} = 0.6, y_{\max} = 0.2$
$Li_x[(CH_2)_n(NH_2)_2]_yFe_2Se_2$	I4/mmm	up to 41	$n = 1,2,3$
$Na_x[(CH_2)_n(NH_2)_2]_yFe_2Se_2$	I4/mmm	up to 47	$n = 1,2,3$
$Sr_x[(CH_2)_n(NH_2)_2]_yFe_2Se_2$	I4/mmm	up to 37	$n = 1,2,3$

TABLE 8. Crystal Structures and Critical Temperatures of the Pnictide Fe-Based Superconductors

Mol. form.	Structure (lattice parameters in nm)	$T_c^{\max}/K$	Comments
<b>111</b>	<b>P4/nmm (anti-PbFCl structure)</b>		
LiFeP	$a = 0.3692, c = 0.6031$	~6	
LiFeAs	$a = 0.3772, c = 0.6357$	18	
$Na_{1-\delta}FeAs$	$a = 0.3949, c = 0.7040$	15	
$Na(Fe_{1-x}Co_x)As$	$c = 0.704$	20	$x_{\max} = 0.3$
<b>112</b>	<b>P21/m</b>		
CaFeAs <sub>2</sub>		0	
$(Ca_{1-x}RE_x)FeAs_2$		up to 43	$RE = La, Pr, Nd, Sm, Eu, Gd$
$(Ca_{1-x}RE_x)Fe(As_{1-y}Sb_y)_2$		up to 47	$RE = La, Ce, Pr, Nd$
EuFeAs <sub>2</sub>		0	
$Eu(Fe_{1-x}Ni_x)As_2$	$a = 0.3987, b = 0.3908, c = 1.0645$	18	$x_{\max} = 0.4$
$(Eu_{1-x}La_x)FeAs_2$	$a = 0.3980, b = 0.3900, c = 1.0643$	11	$x_{\max} = 0.15$
<b>1111</b>	<b>P4/nmm (ZrCuSiAs structure)</b>		
LaOFeP	$a = 0.3964, c = 0.8512$	~4	
$LaOFeAs_{1-x}P_x$		11	$x_{\max} = 0.25$ and 0.7
$La_{1-x}OFeP$	$a = 0.3961, c = 0.8506$	~7	$x_{\max} = 0.1$
$LnOFeP$		~4	$Ln = Pr, Nd$
$(Sr_{1-x}RE_x)FeAsF$		up to 56	$RE = La, Nd, Sm$
$Sr(Fe_{1-x}Co_x)AsF$	$a = 0.4002, c = 0.8943$	~4	$x_{\max} = 0.125$
$(Ca_{1-x}RE_x)FeAsH$		up to 47	$RE = La, Sm$
$Ca(Fe_{1-x}Co_x)AsH$	$a = 0.382, c = 0.822$	24	$x_{\max} = 0.15$
$Ca(Fe_{1-x}Co_x)AsF$	$a = 0.3880, c = 0.8552$	22	$x_{\max} = 0.1$
$Ca(Fe_{1-x}Ni_x)AsF$	$a = 0.3879, c = 0.8578$	15	$x_{\max} = 0.05$
$CaFeAs(F_{1-x}H_x)$	$a = 0.3896, c = 0.8669$	29	$x_{\max} \approx 0.3$

Mol. form.	Structure (lattice parameters in nm)	$T_c^{\max}/\text{K}$	Comments
YFeAsO <sub>1-δ</sub>	$a = 0.3842, c = 0.8303$	47	
LaFeAsO <sub>1-δ</sub>	$a = 0.4026, c = 0.8719$	28	
CeFeAsO <sub>1-δ</sub>	$a = 0.3995, c = 0.8631$	39	
PrFeAsO <sub>1-δ</sub>	$a = 0.3963, c = 0.8572$	48	
NdFeAsO <sub>1-δ</sub>	$a = 0.3943, c = 0.8529$	53	
PmFeAsO <sub>1-δ</sub>	Not synthesized due to radioactive Pm	–	
SmFeAsO <sub>1-δ</sub>	$a = 0.3922, c = 0.8452$	53	
EuFeAsO <sub>1-δ</sub>	Not reported	–	
GdFeAsO <sub>1-δ</sub>	$a = 0.3891, c = 0.8393$	54	
TbFeAsO <sub>1-δ</sub>	$a = 0.3878, c = 0.8353$	52	
DyFeAsO <sub>1-δ</sub>	$a = 0.3863, c = 0.8322$	51	
HoFeAsO <sub>1-δ</sub>	$a = 0.3846, c = 0.8295$	50	
YFeAsO <sub>1-x</sub> F <sub>x</sub>		10	$x_{\max} = 0.1$
LaFeAsO <sub>1-x</sub> F <sub>x</sub>	$a = 0.4036, c = 0.8739$	26	
CeFeAsO <sub>1-x</sub> F <sub>x</sub>	$a = 0.3989, c = 0.8631$	41	$x_{\max} = 0.16$
PrFeAsO <sub>1-x</sub> F <sub>x</sub>	$a = 0.3967, c = 0.8561$	47	$x_{\max} = 0.15$
NdFeAsO <sub>1-x</sub> F <sub>x</sub>	$a = 0.3954, c = 0.8540$	47	$x_{\max} = 0.15$
PmFeAsO <sub>1-x</sub> F <sub>x</sub>	Not synthesized due to radioactive Pm	–	
SmFeAsO <sub>1-x</sub> F <sub>x</sub>	$a = 0.3934, c = 0.8468$	58	$x_{\max} = 0.2$
SmFeAsF	$a = 0.3940, c = 0.8503$	56	
EuFeAsO <sub>1-x</sub> F <sub>x</sub>		11	$x_{\max} = 0.15$
GdFeAsO <sub>1-x</sub> F <sub>x</sub>	$a = 0.3915, c = 0.8457$	40	$x_{\max} = 0.25$
TbFeAsO <sub>1-x</sub> F <sub>x</sub>	$a = 0.3860, c = 0.8332$	51	$x_{\max} \approx 0.1$
DyFeAsO <sub>1-x</sub> F <sub>x</sub>	$a = 0.3843, c = 0.8284$	46	$x_{\max} \approx 0.1$
HoFeAsO <sub>1-x</sub> F <sub>x</sub>	$a = 0.3830, c = 0.8270$	36	$x_{\max} \approx 0.1$
Gd <sub>1-x</sub> Th <sub>x</sub> FeAsO	$a = 0.3916, c = 0.8439$	56	$x_{\max} = 0.2$
Tb <sub>1-x</sub> Th <sub>x</sub> FeAsO	$a = 0.3903, c = 0.8413$	52	$x_{\max} = 0.2$
La(Fe <sub>1-x</sub> Co <sub>x</sub> )AsO	$a = 0.4035, c = 0.8724$	13	$x_{\max} = 0.06$
Sm(Fe <sub>1-x</sub> Co <sub>x</sub> )AsO	$a = 0.3939, c = 0.8467$	17	$x_{\max} = 0.1$
LaFeAsO <sub>1-x</sub> H <sub>x</sub>	$a = 0.399, c = 0.865$	33	$x_{\max} = 0.36$
CeFeAsO <sub>1-x</sub> H <sub>x</sub>	$a = 0.397, c = 0.861$	48	$x_{\max} = 0.25$
NdFeAsO <sub>1-x</sub> H <sub>x</sub>		54	
SmFeAsO <sub>1-x</sub> H <sub>x</sub>	$a = 0.391, c = 0.845$	56	$x_{\max} = 0.22$
GdFeAsO <sub>1-x</sub> H <sub>x</sub>	$a = 0.389, c = 0.840$	54	$x_{\max} = 0.1$
DyFeAsO <sub>1-x</sub> H <sub>x</sub>		52	$x_{\max} = 0.17$
ErFeAsO <sub>1-x</sub> H <sub>x</sub>	$a = 0.3822, c = 0.8281$	41	$x_{\max} = 0.05$
ThFeAsN	$a = 0.4037, c = 0.8526$	30	Undoped superconductor
<b>122</b>	<b>I4/mmm (ThCr<sub>2</sub>Si<sub>2</sub> structure)</b>		
NaFe <sub>2</sub> As <sub>2</sub>	Metastable $a = 0.3809, c = 1.2441$	25	
(Ce <sub>0.5-x</sub> Na <sub>0.5+x</sub> )Fe <sub>2</sub> As <sub>2</sub>	$a = 0.3841, c = 1.2239$	26	$x_{\max} = 0.3$
(Pr <sub>0.5-x</sub> Na <sub>0.5+x</sub> )Fe <sub>2</sub> As <sub>2</sub>	$a = 0.3839, c = 1.2193$	25	$x_{\max} = 0.25$
AFe <sub>2</sub> As <sub>2</sub>		up to 3.8	$A = \text{K, Rb, Cs}$
(Ca <sub>1-x</sub> Na <sub>x</sub> )Fe <sub>2</sub> As <sub>2</sub>	$a = 0.3841, c = 1.22$	33	$x_{\max} = 0.66$
(Ca <sub>1-x</sub> RE <sub>x</sub> )Fe <sub>2</sub> As <sub>2</sub>		up to 49	$RE = \text{La, Ce, Pr}$
Ca(Fe <sub>1-x</sub> M <sub>x</sub> ) <sub>2</sub> As <sub>2</sub>		up to 20	$M = \text{Co, Ni, Rh, Pd, Ir}$
CaFe <sub>2</sub> (As <sub>1-x</sub> P <sub>x</sub> ) <sub>2</sub>	$a = 0.390, c = 1.165$	15	$x_{\max} = 0.05$
(Sr <sub>1-x</sub> A <sub>x</sub> )Fe <sub>2</sub> As <sub>2</sub>		up to 37	$A = \text{Na, K, Cs}$
(Sr <sub>1-x</sub> La <sub>x</sub> )Fe <sub>2</sub> As <sub>2</sub>	$a = 0.395, c = 1.21$	22	$x_{\max} = 0.3$ . Thin film; metastable
Sr(Fe <sub>1-x</sub> M <sub>x</sub> ) <sub>2</sub> As <sub>2</sub>		up to 24	$M = \text{Co, Ni, Ru, Rh, Pd, Ir, Pt}$
SrFe <sub>2</sub> (As <sub>1-x</sub> P <sub>x</sub> ) <sub>2</sub>	$a = 0.390, c = 1.21$	33	$x_{\max} = 0.35$
(Ba <sub>1-x</sub> A <sub>x</sub> )Fe <sub>2</sub> As <sub>2</sub>		up to 38	$A = \text{Na, K, Rb}$
(Ba <sub>1-x</sub> RE <sub>x</sub> )Fe <sub>2</sub> As <sub>2</sub>		up to 22	$RE = \text{La, Ce, Pr, Nd}$
Ba(Fe <sub>1-x</sub> M <sub>x</sub> ) <sub>2</sub> As <sub>2</sub>		up to 24	$M = \text{Co, Ni, Ru, Rh, Pd, Ir, Pt}$
BaFe <sub>2</sub> (As <sub>1-x</sub> P <sub>x</sub> ) <sub>2</sub>		31	$x_{\max} = 0.32$
Ba <sub>2</sub> Ti <sub>2</sub> Fe <sub>2</sub> As <sub>4</sub> O	$a = 0.4028, c = 2.7344$	21	Intergrowth of BaFe <sub>2</sub> As <sub>2</sub> and BaTi <sub>2</sub> As <sub>2</sub> O
LaFe <sub>2</sub> As <sub>2</sub>	$a = 0.3938, c = 1.1714$	12	
(La <sub>0.5-x</sub> Na <sub>x</sub> )Fe <sub>2</sub> As <sub>2</sub>	$a = 0.3841, c = 1.2325$	27	$x_{\max} = 0.3$
(La <sub>0.5-x</sub> Na <sub>x</sub> K <sub>0.5</sub> )Fe <sub>2</sub> As <sub>2</sub>	$a = 0.3850, c = 1.321$	23	$x_{\max} = 0.25$



Mol. form.	Structure (lattice parameters in nm)	$T_c^{\max}/K$	Comments
$(Eu_{1-x}A_x)Fe_2As_2$		up to 35	$A = Na, K, x_{\max} = 0.5$ , antiferromagnetic
$(Eu_{0.78}La_{0.27})Fe_2As_2$		13	Antiferromagnetic
$Eu(Fe_{1-x}M_x)_2As_2$			$M = Co, Ir, Ru, Rh$ , antiferromagnetic
$EuFe_2(As_{1-x}P_x)_2$	$a = 0.389, c = 1.1835$	30	$x_{\max} = 0.3$ , antiferromagnetic
<b>1144 (2×122)</b>	<b>P4/mmm</b>		
$(La,Na)RbFe_4As_4$	$a = 0.3861, c = 1.326$	26	
$(La,Na)CsFe_4As_4$	$a = 0.3880, c = 1.360$	24	
$CaKFe_4As_4$	$a = 0.3866, c = 1.2817$	35	
$CaRbFe_4As_4$	$a = 0.3876, c = 1.3104$	35	
$CaCsFe_4As_4$	$a = 0.3891, c = 1.3414$	32	
$SrRbFe_4As_4$	$a = 0.3897, c = 1.3417$	35	
$SrCsFe_4As_4$	$a = 0.3910, c = 1.3729$	37	
$BaCsFe_4As_4$	Possibly 122 structure	26	
$EuRbFe_4As_4$	$a = 0.3889, c = 1.330$	36	
$EuCsFe_4As_4$	$a = 0.3901, c = 1.361$	35	
<b>12442 (122+2×1111)</b>	<b>I4/mmm</b>		
$KGd_2Fe_4As_4O_2$	$a = 0.3897, c = 3.0670$	37	
$KTb_2Fe_4As_4O_2$	$a = 0.3886, c = 3.0621$	37	
$KDy_2Fe_4As_4O_2$	$a = 0.3874, c = 3.0598$	37	
$KHo_2Fe_4As_4O_2$	$a = 0.3866, c = 3.0597$	36	
$RbSm_2Fe_4As_4O_2$	$a = 0.3921, c = 3.1381$	36	
$RbGd_2Fe_4As_4O_2$	$a = 0.3901, c = 3.1343$	35	
$RbTb_2Fe_4As_4O_2$	$a = 0.3890, c = 3.1277$	35	
$RbDy_2Fe_4As_4O_2$	$a = 0.3879, c = 3.1265$	34	
$RbHo_2Fe_4As_4O_2$	$a = 0.3869, c = 3.1242$	34	
$CsNd_2Fe_4As_4O_2$	$a = 0.3949, c = 3.223$	35	
$CsSm_2Fe_4As_4O_2$	$a = 0.3926, c = 3.2124$	35	
$CsGd_2Fe_4As_4O_2$	$a = 0.3907, c = 3.2051$	33	
$CsTb_2Fe_4As_4O_2$	$a = 0.3895, c = 3.1982$	33	
$CsDy_2Fe_4As_4O_2$	$a = 0.3888, c = 3.1961$	33	
$CsHo_2Fe_4As_4O_2$	$a = 0.3876, c = 3.1949$	33	
$KCa_2Fe_4As_4F_2$	$a = 0.3868, c = 3.1007$	33	
$RbCa_2Fe_4As_4F_2$	$a = 0.3872, c = 3.1667$	30	
$CsCa_2Fe_4As_4F_2$	$a = 0.3881, c = 3.2363$	28	
$BaTh_2Fe_4As_4(N_{0.7}O_{0.3})_2$	$a = 0.3989, c = 2.9853$	30	
<b>42214 ("221"+2×1111)</b>	<b>I4/mmm</b>		
$Pr_4Fe_2As_2Te_{0.88}O_4$	$a = 0.4016, c = 2.9857$	25	
$Sm_4Fe_2As_2Te_{0.92}O_4$	$a = 0.3964, c = 2.9509$	25	
$Sm_4Fe_2As_2Te_{0.72}O_{4-y}F_y$	$a = 0.3960, c = 2.9268$	40	
$Gd_4Fe_2As_2Te_{0.90}O_4$	$a = 0.3935, c = 2.9369$	25	
$Gd_4Fe_2As_2Te_{0.92}O_{4-y}F_y$	$a = 0.3936, c = 2.9350$	45	
<b>10-3-8 and 10-4-8</b>	<b><math>\bar{P}1</math></b>		
$(Ca_{1-x}RE_xFeAs)_{10}Pt_3As_8$	$a = 0.8749, b = 0.8753, c = 1.0714$	up to 33	$x_{\max} = 0.14$
$(CaFe_{1-x}M_xAs)_{10}Pt_3As_8$		up to 15	$M = Co, Ni, Pd, Pt$
$(CaFeAs)_{10}Pt_4As_8$	$a = 0.8755, b = 0.8764, c = 1.0690$	35	
$(CaFe_{1-x}Ir_xAs)_{10}Ir_4As_8$	$P4/n; a = 0.8732, c = 1.0391$	16	
<b>32522</b>	<b>I4/mmm</b>		
$Ca_3Al_2O_5Fe_2P_2$	$a = 0.3715, c = 2.5236$	16	
$Ca_3Al_2O_5Fe_2As_2$	$a = 0.3742, c = 2.6078$	30	
$Ca_{n+1}(Mg,Ti)nO_yFe_2As_2$		up to 47	$n = 3, 4; y \approx 3n-1$
$Ca_{n+1}(Sc,Ti)nO_yFe_2As_2$		up to 42	$n = 3, 4, 5; y \approx 3n-1$
$Sr_3Sc_2O_5Fe_2As_2$	$a = 0.4069, c = 2.6876$	0	
$Sr_4(Sc,Ti)_3O_8Fe_2As_2$		28	
$Ba_3Sc_2O_5Fe_2As_2$	$a = 0.4133, c = 2.8355$	0	
$Ba_4Sc_3O_{7.5}Fe_2As_2$	$a = 0.4123, c = 3.7565$	11	

Mol. form.	Structure (lattice parameters in nm)	$T_c^{\max}/\text{K}$	Comments
<b>42622</b>	<b>P4/mmm</b>		
$\text{Ca}_4\text{Al}_2\text{O}_6\text{Fe}_2\text{P}_2$	$a = 0.3693, c = 1.4927$	17	
$\text{Ca}_4\text{Al}_2\text{O}_6\text{Fe}_2\text{As}_2$	$a = 0.3713, c = 1.5404$	28	
$\text{Ca}_5(\text{Al,Ti})_3\text{O}_9\text{Fe}_2\text{As}_2$		39	
$\text{Ca}_6(\text{Al,Ti})_4\text{O}_{12}\text{Fe}_2\text{As}_2$		36	
$\text{Ca}_8(\text{Mg,Ti})_6\text{O}_{18}\text{Fe}_2\text{As}_2$		40	
$\text{Sr}_4\text{Sc}_2\text{O}_6\text{Fe}_2\text{P}_2$	$a = 0.4016, c = 1.5543$	17	
$\text{Sr}_4\text{V}_2\text{O}_6\text{Fe}_2\text{As}_2$	$a = 0.3930, c = 1.5673$	37	
$\text{Sr}_4(\text{Mg,Ti})_2\text{O}_6\text{Fe}_2\text{As}_2$	$a = 0.3935, c = 1.5952$	39	

**TABLE 9. Crystal Structures and Critical Temperatures of Other Non-Cuprate High-Temperature Superconductors**

Mol. form.	Structure (lattice parameters in nm)	$T_c^{\max}/\text{K}$
$\text{MgB}_2$	P6/mmm; $a = 0.3074, c = 0.3534$	39
$\text{Ba}_{0.6}\text{K}_{0.4}\text{BiO}_3$	$\text{Pm}\bar{3}\text{m}; a = 0.4287$	32