#### ORIGINAL ARTICLE



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# Cold sintered LiMgPO<sub>4</sub> based composites for low temperature co-fired ceramic (LTCC) applications

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### Abstract

Cold sintered,  $\text{Li}_2\text{MoO}_4$ -based ceramics have recently been touted as candidates for electronic packaging and low temperature co-fired ceramic (LTCC) technology but  $\text{MoO}_3$  is an expensive and endangered raw material, not suited for large scale commercialization. Here, we present cold sintered temperature-stable composites based on  $\text{LiMgPO}_4$  (LMP) in which the Mo (and Li) concentration has been reduced, thereby significantly decreasing raw material costs. Optimum compositions,  $0.5\text{LMP}-0.1\text{CaTiO}_3-0.4\text{K}_2\text{MoO}_4$  (LMP-CTO-KMO), achieved 97% density at <300°C and 600 MPa for 60 minutes. Raman spectroscopy, X-ray diffraction, scanning electron microscopy, and energy dispersive X-ray mapping confirmed the coexistence of end-members, LMP, CTO, and KMO, with no interdiffusion and parasitic phases. Composites exhibited temperature coefficient of resonant frequency ~ -6 ppm/°C, relative permittivity ~9.1, and  $Q \times f$  values ~8500 GHz, properties suitable for LTCC technology and competitive with commercial incumbents.

### KEYWORDS

cold sintering, LTCC, microwave dielectric ceramics, phosphates

### 1 | INTRODUCTION

With the rapid development of wireless communications such as Wi-Fi, "Internet of Things", and 5G, the use of microwave (MW) dielectric ceramics in the manufacture of radio frequency (RF) components, such as filters, resonators,

antennas, and substrates,  $^{1-5}$  has dramatically increased. Miniaturization and integration of RF components for 5th generation (5G) technology in particular requires new materials with short delay times ( $\varepsilon_r < 15$ ), lower dielectric loss (high MW quality factor,  $Q \times f$ ), and near-zero temperature coefficient of resonant frequency (TCF).  $^{1-5}$ 

Dawei Wang and Jinrong Chen contributed equally to this work.

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Traditional MW ceramics are sintered at high temperature (T > 1000°C) to impart strength, integrity, and to optimize the required physical properties, <sup>6–9</sup> but energy consumption and associated carbon emissions are substantial and high T limits the integration of low cost metal electrodes (Ag) and polymers. These limitations have led to the development of low temperature cofired ceramics (LTCCs, 700°C-950°C) and ultralow temperature cofired ceramics (ULTCCs, 400 ~ 700°C), <sup>10–13</sup> with many potential new materials for LTCC and ULTCC technology reported in systems based on molybdates, borates, phosphates, and tungstates. <sup>14–24</sup>

Cold sintering is a radical departure in sintering technology in comparison with LTCC and ULTCC and employs an aqueous phase and uniaxial pressure to densify ceramics at <300°C for shorter times (<2 hours) than conventional sintering. 25-43 The energy consumed by cold sintering is <50% of that used by its conventional counterpart, 6 facilitating integration with polymers and base metals and liberating RF design space for ceramics, previously forbidden by their high sintering temperatures. To date, there have been numerous cold sintered materials touted to show promise as potential replacements of LTCC and ULTCC but almost all are based on  $MoO_4^{2-}$ , eg refs 25, 26. Mo is an expensive and endangered raw material<sup>44,45</sup> not suited for mass production in the electronics sector, and new ceramics and composites are required that either eradicate or reduce the concentration of Mo if cold sintering is to compete for applications in MW technology.

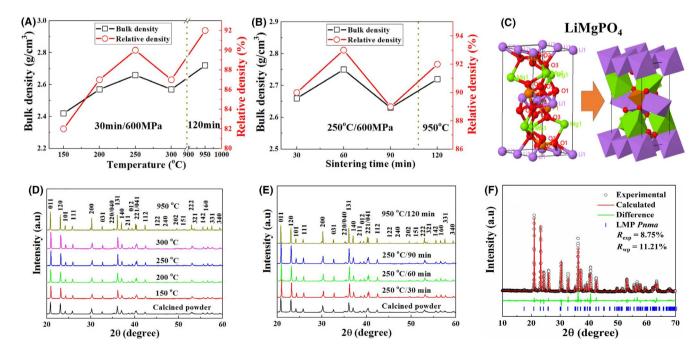
Phosphates (PO<sub>4</sub><sup>3-</sup>) are low cost in comparison to molybdates, <sup>46,47</sup> exhibit high  $Q \times f$  (25 000~80 000) and low  $\varepsilon_r$  (<20), <sup>20,21,22</sup> when conventionally sintered at 800°C-1200°C

and offer partial solubility in water, essential for cold sintering. <sup>35,39</sup> Although cold sintered phosphates have been reported, most have relatively low density and none have near-zero TCF. <sup>48</sup> In the present work, LiMgPO<sub>4</sub> (LMP,  $\varepsilon_{\rm r}=6.6$ ,  $Q\times f=79\,100$ , TCF = -55 ppm/°C) is selected as a base to realize low cost, cold sintered (<300°C) composite ceramics to develop materials suitable for LTCC applications. 0.5LMP-0.1CaTiO<sub>3</sub>-0.4K<sub>2</sub>MoO<sub>4</sub> (LMP-CTO-KMO) composites densified at 250°C, exhibited TCF ~ -6 ppm/°C,  $\varepsilon_{\rm r}$  ~ 9.1, and  $Q\times f$  ~ 8500 GHz, properties superior to commercial LTCC incumbents based on glass-ceramic composites. <sup>49</sup>

## 2 | EXPERIMENTAL PROCEDURES

LMP powder was synthesized by conventional solid-state processing. Raw chemicals, Li<sub>2</sub>CO<sub>3</sub> (99.9%, Sigma-Aldrich), MgO (99.9%, Sigma-Aldrich), and NH<sub>6</sub>PO<sub>4</sub> (>99%, Sigma-Aldrich) were weighed according to the stoichiometric compositions and ball-milled 24 hours in isopropanol. The mixed powders were dried at 80°C and calcined 4 hours at 900°C. LMP powders were ball-milled to reduce particle size. KMO (95%) and CTO (>99%) powders were purchased from Sigma-Aldrich. For comparison, LMP was uniaxially pressed and conventionally sintered 2 hours at 950°C to obtain dense bulk ceramics.

1-2g of LMP, KMO, and CTO powders were weighed according to the formula of 50wt%LMP-10wt%CTO-40wt%KMO and then sufficiently mixed with 5-10wt%



**FIGURE 1** Bulk and relative density of LMP ceramics as a function of (A) sintering temperature and (B) sintering time. C, Schematic of the crystal structure of LMP. XRD patterns of LMP ceramics at different (D) sintering temperatures and (E) sintering times. F, Rietveld refinement of LMP ceramics cold sintered 60 min at 250°C [Color figure can be viewed at wileyonlinelibrary.com]

Composition	Sintering condition	ρ (g/ cm <sup>3</sup> )	$ ho_r$ (%)	$arepsilon_r$	tanδ	$Q \times f$ (GHz)
LMP	950°C/120min	2.72	92	6.4	0.00035	31,000
	150°C/30min/600MPa	2.42	82	5.4	0.0016	6,900
	200°C/30min/600MPa	2.57	87	5.8	0.0007	16,000
	250°C/30min/600MPa	2.66	90	6.1	0.0008	14,000
	300°C/30min/600MPa	2.57	87	6.0	0.001	11,000
	250°C/60min/600MPa	2.75	93	6.5	0.0007	16,000
	250°C/90min/600MPa	2.63	89	6.2	0.003	3,600
LMP-CTO-KMO	200°C/60min/600MPa	2.84	94.5	8.59	0.001	8,480
	250°C/60min/600MPa	2.92	97	9.1	0.001	8,500
	300°C/60min/600MPa	2.87	95.5	9.01	0.001	8,000

deionized water by using a pestle and mortar, after which cold sintered ceramics and composites were prepared by pressing (Atlas Heated Platens, Specac) 30-90 minutes at 600 MPa and 150°C-300°C depending on composition.

The geometric method was used to calculate the bulk density  $(\rho)$  of samples, as reported elsewhere. <sup>25,26</sup>The ceramic microstructure and grain morphology were examined by scanning electron microscopy (SEM, FEI Inspect F-50). The crystal structure and phase assemblage were determined by X-ray powder diffraction (XRD, Bruker D2 Phaser). Raman spectra were obtained using a Renishaw inVia Raman spectroscopy. The measurement of MW properties ( $\varepsilon_r$ , TCF,  $\tan\delta$ 

and  $Q \times f$ ) was conducted by the TE<sub>018</sub> mode with a vector network analyser (Advantest R3767CH), following previously defined protocols. The cavity was heated by a Peltier device and the resonant frequency (f) was measured from 25°C to 85°C. The corresponding TCF values were obtained using the formula:

$$TCF = \frac{f_T - f_{T_0}}{f_{T_0} \times (T - T_0)} \times 10^6$$
 (1)

where the  $f_T$  and  $f_{T_0}$  are the TE<sub>018</sub> resonant frequencies at temperatures, T and  $T_0$  respectively.

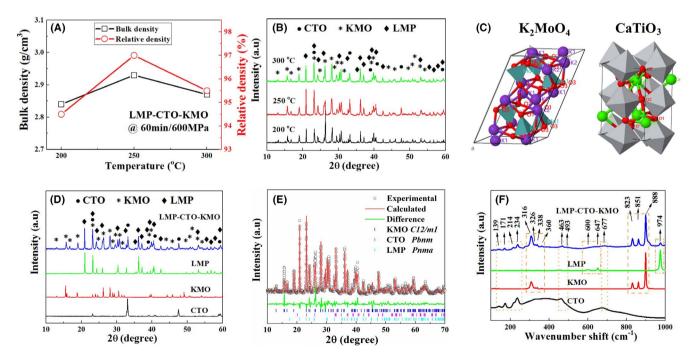


FIGURE 2 A, Bulk and relative density of cold sintered LMP-CTO-KMO composites as a function of sintering temperature. B, XRD patterns of cold sintered LMP-CTO-KMO composites at different sintering temperatures. C, Schematic of the crystal structures of KMO and CTO. D, XRD patterns of cold-sintered LMP-CTO-KMO, LMP, and KMO and commercial CTO powder. E, Rietveld refinement of cold sintered LMP-CTO-KMO composites. F, Raman spectra of cold-sintered LMP-CTO-KMO, LMP, and KMO and commercial CTO powder [Color figure can be viewed at wileyonlinelibrary.com]

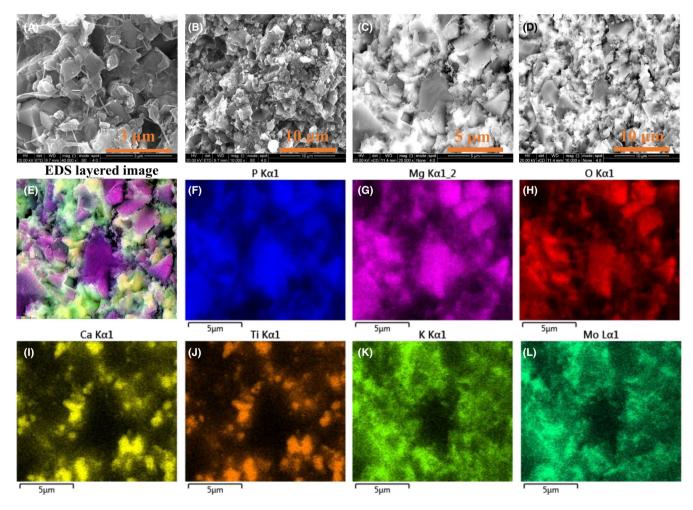
### 3 | RESULTS AND DISCUSSION

The values of density  $(\rho)$  and relative density  $(\rho_r)$  for cold-sintered LMP ceramics are plotted in Figure 1A,B and listed in Table 1. As sintering temperature increases, the values of  $\rho$  and  $\rho_r$  both increase from 2.42 g/cm³ (82%) at 150°C to 2.66 g/cm³ (90%) at 250°C, followed by a decrease for higher sintering temperatures (Figure 1A). The optimum cold sintering temperature is therefore, 250°C. Sintering time at 250°C is subsequently increased and  $\rho$  and  $\rho_r$  is further optimized to 2.73 g/cm³ (93%) for samples cold-sintered 60 minutes (Figure 1B). The maximum values of  $\rho \sim 2.73$  g/cm³ and  $\rho_r \sim 93\%$  for cold-sintered LMP are higher than those reported for conventionally sintered LMP, (2.72 g/cm³ and 92%) (Figure 1B), illustrating the great promise of cold sintered LMP as a base to develop new materials.

Room-temperature XRD patterns of LMP calcined powders, conventionally sintered ceramics, and cold-sintered samples are shown in Figure 1D,E. LMP has a Mg<sub>2</sub>SiO<sub>4</sub>-type olivine structure (space group: *Pnma*, ISCD collection code: 201 138) and is composed of PO<sub>4</sub> tetrahedra and Li/MgO<sub>6</sub>

octahedra (Figure 1C). Only diffraction peaks associated with olivine-structured LMP are detected in the XRD patterns (Figure 1D,E) with no impurity phases. Full-pattern Rietveld refinement of XRD data for LMP ceramics cold-sintered 60 minutes at 250°C was conducted using a Topas 5 software. The calculated pattern is in good agreement, with low values of  $R_{\rm exp} = 8.75\%$  and  $R_{\rm wp} = 11.21\%$ , Figure 1F. The calculated lattice parameters are a = 10.7418 Å, b = 5.9070 Å, c = 4.6909 Å for LMP, which agree with those previously reported. <sup>20</sup>

The data presented above confirms that LMP may be cold sintered to moderately high density (93%) but further improvements were not obtained in the current study. Moreover, LMP has a large negative TCF (–55 ppm/°C) that falls outside excepted values for LTCC and ULTCC (+/– 30 ppm/°C) applications. To adjust TCF closer to zero and to improve density, a bespoke cold sintering flux, 0.8KMO-0.2CTO (TCF = +70 ppm/°C) was developed based on KMO (TCF = -70 ppm/°C) but with a large positive TCF, adjusted through adding CTO (+850 ppm/°C). Forming composites in this manner is typically used in commercial LTCC which



**FIGURE 3** A and B, SEM images of LMP ceramics cold-sintered 60min at 250°C/600MPa. C and D, BSE images of LMP-CTO-KMO composites cold-sintered 60min at 250°C/600MPa. EDS mapping of LMP-CTO-KMO: (E) layered image, (F) P, (g) Mg, (H) O, (I) Ca, (J) T, (K) K, (L) Mo [Color figure can be viewed at wileyonlinelibrary.com]

are composed of a low melting temperature glass matrix with negative TCF and high  $\varepsilon_r$ , positive TCF phases such as  ${\rm TiO_2.}^{49}$ 

 $\rho$  and  $\rho_{\rm r}$  for cold-sintered LMP-CTO-KMO composites as a function of sintering temperature are plotted in Figure 2A and listed in Table 1. As sintering temperature increases to 200°C and 250°C,  $\rho$  and  $\rho_{\rm r}$  increase to 2.84 g/cm<sup>3</sup> (94.5%) and 2.92 g/cm<sup>3</sup> (97%), respectively, followed by a slight decrease at higher sintering temperatures.

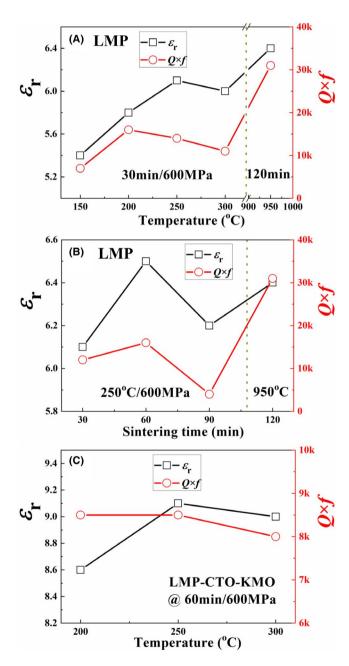
Room-temperature XRD patterns of cold-sintered LMP-CTO-KMO composites are shown in Figure 2B,D which reveal little change in phase assemblage as a function of sintering temperature. Only peaks from each end member are present: CTO exhibits an orthorhombic, perovskite structure (*Pbnm*, ISCD collection code: 62149); KMO is monoclinic (*C12/m1*, ISCD collection code: 16154) and LMP is as described in Figure 1. There is no evidence of secondary phases. Rietveld refinement was performed using a three-phase mix of LMP (*Pnma*), KMO (*C12/m1*), and CTO (*Pbnm*). The calculated pattern matches well with experimental data ( $R_p = 9.65\%$  and  $R_{wp} = 12.92\%$ ), where the weight fractions (LMP ~ 49.9%, CTO ~ 10.6%, KMO ~ 39.5%) are close to the nominal compositions, as shown in Figure 2E.

To confirm further the coexistence of three phases in composites, Raman spectra of cold-sintered LMP-CTO-KMO samples are shown in Figure 2F. As reported previously, 36, 10, and 39 Raman bands are commonly observed in LMP, CTO, and KMO respectively. 25,50,51 For LMP, there are 18 external modes (<400 cm<sup>-1</sup>, 12 translations of Li, Ni, PO<sub>4</sub> tetrahedra, six vibrations of PO<sub>4</sub> tetrahedra), and 18 internal modes of PO<sub>4</sub> tetrahedra (>400 cm<sup>-1</sup>, P-O stretching:  $\nu_1 = 974$  cm<sup>-1</sup>,  $\nu_3 = 1020\text{-}1080 \text{ cm}^{-1}$ , O-P-O bending and P vibration:  $\nu_2 = 416-468 \text{ cm}^{-1}$ ,  $\nu_4 = 590-650 \text{ cm}^{-1}$ ), Figure S1. For CTO, Raman bands of 630 and 678 cm<sup>-1</sup> are related to symmetric stretching of Ti-O. The 463 and 493 cm<sup>-1</sup> bands are the torsional modes of Ti-O. The bands of 171, 214, 234, 275, and 323 cm<sup>-1</sup> are related to the bending of O-Ti-O. The 139 cm<sup>-1</sup> band belongs to the Ca ions motion. For KMO, Raman bands in the range of 100 ~ 160 cm<sup>-1</sup> correspond to a combination of the translations and vibrations of MoO<sub>4</sub> tetrahedra and translations of K ions. The 310 ~ 370 cm<sup>-1</sup> bands are related to bending modes of MoO<sub>4</sub> tetrahedra. The 820 ~ 890 cm<sup>-1</sup> bands are related to stretching modes of MoO<sub>4</sub> tetrahedra. Raman spectra of LMP-CTO-KMO composites therefore, represent an overlay of Raman bands from individual phases, confirming the existence of LMP, KMO and CTO but without a significant volume fraction of interaction (Figure 2F).

SEM and BSE images of cold-sintered LMP ceramics and cold-sintered LMP-CTO-KMO composites are shown in Figure 3A,B and Figure 3C,D respectively. A denser microstructure in LMP-CTO-KMO composites than in LMP ceramics is evident, coincident with the higher density listed in Table 1. The variations in contrast in BSE images

of LMP-CTO-KMO suggest that there are three chemically discrete CTO, KMO, and LMP rich phases, confirmed by the EDS mapping (Figure 3E-L).

The MW dielectric properties of LMP ceramics and LMP-CTO-KMO composites as a function of sintering temperature and time are presented in Figure 4 and listed in Table 1. As sintering temperature and time increase,  $\varepsilon_{\rm r}$  and  $Q \times f$  of LMP increases initially before decreasing. The highest values of  $\varepsilon_{\rm r} \sim 6.5$  and  $Q \times f \sim 16\,000$  are achieved for LMP cold sintered 60 min at 250°C under a uniaxial pressure of 600 MPa



**FIGURE 4** The microwave dielectric properties of LMP as a function of (A) sintering temperature (B) sintering time; (C) The microwave dielectric properties of LMP-CTO-KMO as a function of sintering temperature [Color figure can be viewed at wileyonlinelibrary.com]

Compounds	ST (°C)	$oldsymbol{arepsilon}_r$	$Q \times f$ (GHz)	TCF (ppm/°C)	Ref.
CaZrO <sub>3</sub> +glass (Murata)	<1000	25	3500	<+/-30	49
Nd <sub>2</sub> O <sub>3</sub> -TiO <sub>2</sub> -SiO <sub>2</sub> +glass (NEC)	<1000	18	1200	<+/-30	49
Al <sub>2</sub> O <sub>3</sub> -CaZrO <sub>3</sub> +glass (Dupont)	<1000	7.8	900	<+/-30	49
SiO <sub>2</sub> -Al <sub>2</sub> O <sub>3</sub> -cordierite+glass (Kyocera)	<1000	5	8,000	<+/-30	49
LMP	950	6.6	79 100	-55	20
$(Bi_{0.05}Ce_{0.95})VO_4$	950	11.9	22 360	+6.6	19
$Bi_2(Li_{0.5}Ta_{1.5})O_7 + 2\% Bi_2O_3$	920	64.1	11 200	-19	12
$Bi(Sc_{1/3}Mo_{2/3})O_4$	915	24.4	48 100	-68	15
$0.45 \mathrm{BiVO_4}$ - $0.55 \mathrm{TiO_2}$	900	86	9500	-8	10
$Ca_{0.66}Bi_{0.34}Mo_{0.66}V_{0.34}O_4$	870	21.9	18 150	+0.1	18
$\text{Li}_6\text{B}_4\text{O}_9$	640	5.95	41 800	-72	17
$K_{0.5}(Nd_{0.6}Bi_{0.4})_{0.5}MoO_4$	620	17.3	13 050	+4	16
$(Na_{1.2}Ag_{0.8})MoO_4$	410	8.1	44 800	-82	14
LMP	250	6.5	16 000	-58	This work
LMP-CTO-KMO	250	9.1	8,500	-6	This work

**TABLE 2** Comparison of sintering temperature (ST) and MV properties of LMP ceramics and LMP-CTO-KMO composites with other LTCCs

(Figure 4A,B). The same conditions resulted in optimized values of cold sintered LMP-CTO-KMO composites with  $\varepsilon_{\rm r} \sim 9.1$ ,  $Q \times f \sim 8500$  and a near zero TCF  $\sim -6$  ppm/°C (Figure 4C).

The comparison of sintering temperature and MW properties for recently reported and commercial LTCCs are listed in Table 2. LMP-CTO-KMO composites exhibit the lowest sintering temperature (250°C), reducing energy costs in manufacture and ensuring compatibility with all low-cost electrode systems.  $Q \times f$  of cold-sintered LMP-CTO-KMO composites are is superior to commercial LTCCs,  $\varepsilon_r$  is ideal and they are temperature stable (<+/-30 ppm/°C). 49 Cold sintering does not result in lateral shrinkage and hence issues relating to dissimilar shrinkage and thermal expansion between electrode and substrate are alleviated. Many materials are reported with higher  $Q \times f$  and  $\varepsilon_r$  but these have not been commercialized due to either TCF> +/-30 ppm/°C, sintering temperature >900°C, high cost and environmental issues (Mo and V based systems) or they are over designed for the application; metallized LTCCs do not require ultra-high  $Q \times f$  as losses are dominated by the metal/ceramic interface. Cold sintered LMP-CTO-KMO, therefore, satisfies the criteria for LTCC applications but we note that the production of cold sintered LTCC requires a radical rethink of ceramic processing and scale-up.

### 4 | CONCLUSIONS

Microwave LMP ceramics and LMP-CTO-KMO ceramic composites were successfully fabricated by cold sintering. As sintering temperature and time increase,  $\rho$  and  $\rho_r$  for LMP increased from 2.42 g/cm<sup>3</sup> (82%) at 150°C/30 min to 2.73 g/

cm³ (93%) at 250°C/60 min for which  $\varepsilon_{\rm r}$  and  $Q \times f$  was ~ 6.5 and ~ 16,000 GHz respectively. At the same sintering conditions,  $\rho$  ~ 2.92 g/cm³ (97%),  $\varepsilon_{\rm r}$  ~ 9.1,  $Q \times f$  ~ 8500, and near zero TCF ~ -6 ppm/°C were obtained for cold sintered LMP-CTO-KMO composites. LMP-CTO-KMO cold sintered composites were therefore considered suitable for LTCC applications.

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### SUPPORTING INFORMATION

Additional supporting information may be found online in the Supporting Information section.

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