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The Role of Spontaneous Cap Domain Mutations in Haloalkane Dehalogenase Specificity and Evolution*

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The first step in the utilization of the xenobiotic chlorinated hydrocarbon 1,2-dichloroethane by Xanthobacter autotrophicus is catalyzed by haloalkane dehalogenase (DhlA). The enzyme hydrolyses 1-haloalkanes to the corresponding alcohols. This allows the organism to grow also on short-chain (C2-C4) 1-chloro-nalkanes. We have expressed DhlA in a strain of Pseudomonas that grows on long-chain alcohols and have selected 12 independent mutants that utilize 1-chlorohexane. Six different mutant enzymes with improved K_m or V_{max} values with 1-chlorohexane were obtained. The sequences of the mutated dhlA genes showed that several mutants had the same 11-amino acid deletion, two mutants carried a different point mutation, and three mutants had different tandem repeats. All mutations occurred in a region encoding the N-terminal part of the cap domain of DhlA, and it is concluded that this part of the protein is involved in the evolution of activity toward xenobiotic substrates.

Some bacterial strains of the species Xanthobacter autotrophicus (1, 2) and Ancylobacter aquaticus (3) are capable of growing on the synthetic chlorinated hydrocarbon 1,2-dichloroethane. Degradation proceeds via 2-chloroethanol, chloroacetaldehyde, and chloroacetic acid to glycolate (1). Haloalkane dehalogenase (DhlA),¹ a 310-amino acid cytoplasmic protein, catalyzes the first hydrolytic reaction step. The gene has been cloned and sequenced (4), and identical dehalogenases were detected in different strains that grow on 1,2-dichloroethane (3).

The three-dimensional structure of DhIA was recently solved by x-ray crystallography (5, 6). DhlA is composed of a globular main domain with an α/β -hydrolase fold structure and a separate cap domain (5-7). The active site cavity is located between the two domains. X-ray crystallographic studies indicated that the enzyme catalyzes cleavage of halogenated compounds by nucleophilic displacement of the halogen by Asp¹²⁴ (8). It was shown that during incubations of DhlA with substrate in H₂¹⁸O incorporation of 18 O occurred both in Asp 124 and product (9). This indicates that the covalent alkyl-enzyme intermediate, formed by nucleophilic displacement of the halogen by Asp¹²⁴, is hydrolyzed by nucleophilic attack of a water molecule on the carbonyl carbon atom. His²⁸⁹ probably activates this water molecule by subtracting a proton, with assistance of $\operatorname{Asp}^{260}\left(8\right).$ Two tryptophans, Trp¹²⁵ and Trp¹⁷⁵, are involved in substrate and halide binding (8, 10).

The α/β -hydrolase fold domain of DhlA is a structurally conserved fold shared by several hydrolytic proteins (7). The overall topology of DhlA consists of an eight-stranded β -sheet, with connecting α -helices. The nucleophilic residues, either Ser, Cys, or Asp, are positioned on a sharp bend between the fifth β -strand and the following α -helix. The positions of the other two active site residues, His²⁸⁹ and Asp²⁶⁰, are conserved as well. The proteins differ by the presence, position, and sequences of cap domains, and therefore this part of the proteins was suggested to influence substrate specificity (7).

Haloalkane dehalogenase is active with several 1-chloro-*n*alkanes and α,ω -dichloro-*n*-alkanes, but there is hardly any activity with 1-chlorohexane or chloroalkanes of longer chain length (1). The enzyme has some activity with long-chain bromoalkanes (C₅-C₁₀), suggesting that the active site can accommodate longer substrates and that differences in binding affinities or reaction rates are responsible for the lower activity with long-chain chloroalkanes.

Since 1,2-dichloroethane is not known to occur or to be produced naturally, it is likely that an enzyme capable of hydrolyzing this compound evolved to its present form after industrial 1,2-dichloroethane production and emission started in 1922 (11, 12). No sequences that are similar to DhlA and from which the present enzyme could be derived are known, however. We decided to investigate how DhlA can evolve to convert new substrates by selecting spontaneous mutants that degrade 1-chlorohexane, an approach termed experimental enzyme evolution (13–15). The sequences and activity of the mutant enzymes that were obtained indicate that short direct repeats in the N-terminal part of the cap domain play an important role in the evolution of haloalkane dehalogenase specificity.

EXPERIMENTAL PROCEDURES

Strains and Plasmids—Pseudomonas GJ31 is a chlorobenzene utilizing organism described earlier (4, 16). Escherichia coli JM101 was used as host strain for isolation of single strand DNA and sequencing. Plasmid pPJ20 is a broad host range plasmid (4) derived from pLAFR1 (17). It contains the *dhlA* gene with its own promoter and constitutively produces DhlA in several Gram-negative bacteria (4).

Selection of Mutants-Selection of spontaneous mutant enzymes was achieved with the recombinant bacterial strain Pseudomonas GJ31(pPJ20). Pseudomonas GJ31 is capable of growing on long-chain alcohols. The plasmid was introduced by triparental mating (17) and allowed the organism to grow on 1-chlorobutane by expression of the dhlA gene up to levels of 25% of the total cellular protein (4). Mutants of Pseudomonas GJ31(pPJ20) that could utilize 1-chlorohexane as a growth substrate were selected (see Fig. 1) at 22 °C by cultivation in 1-liter serum flasks containing 400 ml of synthetic medium (1). The medium contained 2 mm 1-chlorobutane, which did support growth of the recombinant, and 2 mm 1-chlorohexane, which is hardly hydrolyzed by the wild type enzyme and did not support growth. After 4 weeks of incubation, some cultures showed higher turbidities than a control containing no 1-chlorohexane. After three serial transfers in medium containing 2 mm 1-chlorohexane as the sole carbon source, mutants were purified on nutrient broth agar plates containing tetracyclin (12 µg/ml).

Sequencing—The dhlA gene of mutant $\nabla 152-153$ was determined by cloning AluI-HindIII and Sau3A DNA fragments in phage M13mp18

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 $^{^{\}rm l}$ The abbreviations used are: DhlA, haloalkane dehalogenase; bp, base pair.

and M13mp19 (18), followed by dideoxy sequencing (19). The other mutant dhlA genes were amplified by the polymerase chain reaction (20), followed by dideoxy sequencing of the double-stranded polymerase chain reaction DNA. The primers used for sequencing are as follows (listed from 3' to 5' with the starting nucleotide number (4) given and c and n indicating coding and non-coding strand, respectively): 999 ATGATAAATGCAATTCGC (c), 1303 ACCGCAACTTCCTGGCTGC (c), 1590 CTGACCGAAGCTGAGGCC (c), 1928 TTCTGTCTCGGCAAAG-TG (n), 1628 GAAAGGCGCAGCATACGC (n), 1322 TGCAAGCAGGA-AGTTGCG (n).

Purification of Haloalkane Dehalogenase—Mutants of Pseudomonas GJ31(pPJ20) were grown at 30 °C in a mineral medium (1) containing 4 mm 1-chlorohexane as the sole carbon source. After 4 days, cells were harvested by centrifugation, washed in 10 mm Tris·H₂SO₄ buffer, pH 7.5, containing 1 mm EDTA and 1 mm β -mercaptoethanol, and sonicated after resuspension in this buffer. Mutant haloalkane dehalogenases were purified as described before (1).

Determination of Dehalogenase Activities—The activities of the mutant enzymes were tested with a variety of substrates by spectrophotometric measurement of the rates of halide production (1) using purified enzymes (wild type and mutants P168S, $\Delta 164-174$, and $\nabla 152-153$) or crude extracts (mutants D170H, $\nabla 172-174$, and $\nabla 145-154$). The concentration of 1-chlorohexane in the assays was 3 mM, the other substrates were used at 5 mM. The dehalogenase content of cell-free extracts was determined by Coomassie Brilliant Blue staining of the protein bands of SDS-polyacrylamide gels, and subsequent peak integration of scans of the gels was taken with a desktop scanner. Accuracy was better than 10%.

For the determination of K_m , V_{max} , and k_{cat}/K_m , alcohol production was measured in 4.5-ml incubations containing 0.01-4 mM substrate in 50 mM Tris·H₂SO₄ buffer, pH 8.2, and 10 milliunits of 1,2-dibromoethane dehalogenase or 100 milliunits of 1-chlorohexane or 1,2-dichloroethane dehalogenase. Samples were incubated for 15 min at 30 °C, and the amount of alcohol produced was determined on a Chrompack 438S gas chromatograph with a CPWax 52 CB column, using an ECD detector for 2-bromoethanol and a FID detector for 2-chloroethanol and 1-hexanol. The carrier gas was nitrogen (60 kPa), and the temperature program was 3-min isothermal at 45 °C followed by an increase to 200 °C at 10 °C/min. K_m and V_{max} values were calculated from the rates of alcohol production by nonlinear regression analysis using the Michaelis-Menten equation and the Enzfitter program of Leatherbarrow (21).

RESULTS

Isolation of 1-Chlorohexane-utilizing Mutants—The dehalogenase-expressing strain Pseudomonas GJ31(pPJ20) was found to be capable of growth on 1-chlorobutane, but not on 1-chlorohexane. The specific activity of purified haloalkane dehalogenase with 1-chlorohexane was only 0.17 unit/mg of protein, which apparently was not sufficient for growth, although the enzyme was expressed at a high level (4). Mutants of strain GJ31(pPJ20) that utilized 1-chlorohexane as sole carbon source were easily obtained in batch incubations that contained a mixture of 1-chlorobutane and 1-chlorohexane as sole carbon sources (Fig. 1). Twelve independent mutants were purified and analyzed for the presence of mutations in the dhlA gene by DNA sequencing (Table I) and determining dehalogenase activities in crude extracts (Table II).

Sequences of Mutated Dehalogenases—From sequencing the dhlA genes of the 1-chlorohexane utilizing mutants, six different mutants were identified. Only two mutants carried a single base substitution, leading to the mutations $Asp^{170} \rightarrow His$, and $Pro^{168} \rightarrow Ser$. The other mutants had much larger changes, being deletions or insertions. A large in frame deletion of 33 bp, causing an 11-amino acid deletion of Phe^{164} —Ala¹⁷⁴, was detected in six independently selected mutants. The duplications that were found were in frame direct repeats, varying in length from 6 to 30 bp. A mutant that carried a 6-bp duplication leading to a tandem repeat of Met¹⁵²—Thr¹⁵³ was found twice. The other duplications were detected only once, and caused tandem repeats of Phe^{172} —Ala¹⁷⁴ and Ile^{145} —Asp¹⁵⁴, respectively. The 9-bp duplication in mutant $\nabla 172$ —174 encompassed the last 9 bp of the 33-bp deletion in mutant $\Delta 164$ —174 (see also Fig. 2).

The deletions and duplications caused slight changes in elec-



FIG. 1. Selection of mutants of haloalkane dehalogenase. Plasmid pPJ20 is a broad host range vector, expressing the *dhlA* gene product. It was introduced in *Pseudomonas* GJ31 by conjugation (4). Strains were cultivated as described under "Experimental Procedures."

TABLE I Mutants of haloalkane dehalogenase selected on 1-chlorohexane

Mutant	Mutation	No. of mutants found	
D170H	$G \rightarrow C$ substitution (Asp ¹⁷⁰ \rightarrow His)	1	
P168S	$C \rightarrow T$ substitution (Pro ¹⁶⁸ \rightarrow Ser)	1	
$\Delta 164 - 174$	33 bp deletion (Phe ¹⁶⁴ –Ala ¹⁷⁴)	6	
V172–174	9 bp tandem duplication (Phe ¹⁷² -Ala ¹⁷⁴)	1	
∇ 152–153	6 bp tandem duplication (Met ¹⁵² -Thr ¹⁵³)	2	
∇145–154	30 bp tandem duplication (Ile ¹⁴⁵ –Asp ¹⁵⁴)	1	

trophoretic mobility of some mutant dehalogenases on SDSpolyacrylamide gels (Fig. 3). The dehalogenase carrying the 11-amino acid deletion clearly had a higher mobility, whereas mutants $\nabla 172-174$ and $\nabla 145-154$ produced dehalogenases that migrated somewhat slower than the wild type.

No plasmids could be isolated from the D170H and $\nabla 145-154$ mutants suggesting that the plasmid had integrated into the chromosome of *Pseudomonas* GJ31. These mutants still produced haloalkane dehalogenase, but the D170H mutant expressed the enzyme at a lower level than the wild type and the other mutants (Fig. 3).

Activities of in Vivo Mutants—All mutants produced a dehalogenase with increased activity toward 1-chlorohexane and other C_5 and C_6 1-chloro-*n*-alkanes (Table II). With all mutant enzymes, both the K_m and V_{max} values for 1-chlorohexane had improved, whereas these values were worse for 1,2-dichloroethane. The K_m value for 1,2-dibromoethane, which is the best substrate for the wild type enzyme, was also higher with all mutant enzymes (Table III). The individual mutants showed varying activities with several other substrates of the enzyme.

Mutant P168S had the highest V_{max} for 1-chlorohexane and 1,2-dibromoethane whereas the K_m for 1-chlorohexane was only slightly improved. The $\Delta 164-174$ deletion enzyme had lower activities than the wild type for all chlorinated compounds mentioned in Table II, except 1-chlorohexane, 1-chloropentane, and 1,6-dichlorohexane. The duplication mutant enzyme $\nabla 172$ -174, which has changes in the same part of the protein as the $\Delta 164-174$ deletion, had similar activities, except that the specificity constant k_{cat}/K_m for 1-chlorohexane was only slightly improved and the activity for 1,2-dichloroethane was less reduced. The other two duplication mutant enzymes had the largest increase of the specificity constant for 1-chlorohexane, about 20 fold. Both V_{max} and K_m values were improved. They had a broad activity with long-chain (C_4-C_8) chloroalkanes. Summarizing, the general substrate range of the mutant enzymes thus had shifted from short chain substrates to 1-chloro-n-alkanes of longer chain length. This modification of specificity was not

Spontaneous Cap Domain Mutations in DhlA

TABLE II Activities of the mutant enzymes debalarconces (wild time (WT))

Halide production in incubations containing purified dehalogenase (wild type (WT), mutants P168S, $\Delta 164-174$ and $\nabla 152-153$) or cell extract (the other mutants). Specific activities represent the amount of halide liberated (micromoles) per min per mg of dehalogenase protein added.

Substrate				Specific activ	ities			
Substrate	WT	D170H	P168S	$\Delta 164 - 174$	⊽172–174	7152-153	⊽145–154	
		units/mg of protein						
Dichloromethane	0.40	0.15	<0.1	< 0.1	< 0.1	0.06	< 0.1	
1,2-Dichloroethane	6.60	1.22	4.69	0.11	1.03	0.15	0.43	
1-Chlorobutane	2.05	0.71	1.21	0.34	1.07	1.37	0.92	
1-Chloropentane	0.40	0.56	0.48	0.48	0.63	1.80	1.00	
1-Chlorohexane	0.17	0.46	3.70	0.64	0.40	1.16	1.00	
1,6-Dichlorohexane	0.28	0.38	0.57	0.66	0.44	0.95	1.57	
1-Chlorooctane	0.26	0.17	0.21	< 0.2	< 0.2	0.39	0.81	
1,2-Dibromoethane	6.00	4.33	8.05	4.25	3.96	4.30	5.82	
1-Bromopropane	1.43	0.51	1.25	1.12	0.87	0.56	1.22	
1-Bromobutane	1.53	0.35	1.53	1.92	1.23	0.77	1.63	
1-Bromopentane	1.94	0.64	2.13	1.52	1.50	0.72	1.75	
1-Bromohexane	1.08	0.88	2.50	1.47	1.82	0.97	1.28	
1-Bromooctane	1.21	0.37	1.37	0.84	1.23	0.72	1.22	
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FIG. 2. Repeats and mutations in the *dhlA* gene. The part of the gene that encodes the N-terminal region of the cap domain is shown, together with the amino acid sequence. Residues forming the active site cavity are *encircled*. The *boxes* above the protein sequence indicate the position of strand 6 of the β -sheet and α -helices 4 and 5. The *large solid arrows* mark the 15-bp direct repeat. One copy of the repeat with flanking nucleotides is deleted in the $\Delta 164-174$ mutant enzymes (*dashed box*), the formation of which is explained by misalignment of the 9-bp imperfect direct repeat at the ends of the deleted sequence (*open arrows*). Of this 9-bp repeat, the right-hand copy is again duplicated in mutant $\nabla 172-174$ (*closed box*). The other repeats detected in the mutants are located in the N-terminal part of the sequence encoding the cap domain (*closed boxes*). The bases printed in *bold characters* mark the position of the mutations in D170H and P168S, causing the substitutions indicated.

observed with bromoalkanes, as illustrated by the high activity of both the wild type and the mutant enzymes with 1,2-dibromoethane (Table III).

Location of Mutations—Attempts to crystallize purified mutant enzymes have failed so far, probably due to disturbance of the crystal contacts of the DhlA molecule at Gln^{167} and Ala^{169} by the mutations (6). Examination of the amino acids that are changed using the three-dimensional structure of the wild type enzyme (5) showed that none of the mutations directly affects active site residues, with the exception of the 11-amino acid deletion, in which the active site cavity forming residues Phe¹⁶⁴ and Phe¹⁷² are lost. The other mutations thus have indirect effects.

The mutations were all located in a segment of the dhlA gene that encodes the N-terminal part of the cap domain (Fig. 2). In the mutant enzymes D170H, P168S, Δ 164–174, and ∇ 172–174, the changes all affect the structurally important salt bridge Asp¹⁷⁰-Lys²⁶¹ (Fig. 4). Although the Asp¹⁷⁰ Oδ1 is 15.6 Å away from the active site $Asp^{124} O\delta 1$, the loss of the Asp^{170} -Lys²⁶¹ salt bridge appears to change the active site cavity in such a way that improved 1-chlorohexane conversion is achieved. Probably, the active site cavity residues Phe¹⁶⁴ and Phe¹⁷² are more mobile or have a slightly different position in the mutants (Fig. 4). In the P168S enzyme, the proline which is located in a cis proline turn between helices 4 and 5, is substituted. This allows rotation around the C α -N bond of residue 168 and a change of the cis bond in a trans bond (22), which could lead to disturbance of the Asp¹⁷⁰-Lys²⁶¹ salt bridge that is in its proximity. In the $\nabla 172-174$ mutant enzyme. Phe¹⁷² is preceded by an Ala instead of Gly¹⁷¹, which has unusual $\phi\psi$ -angles for Ala in the

structure of wild type DhlA and forms the N-cap of helix 5, where Ala is less favored (23, 24). This will affect the cavity geometry via Phe¹⁷² or via the salt bridge by increasing the size of the *cis* proline turn. In the $\Delta 164-174$ mutant enzyme, Phe¹⁶⁴ and Phe¹⁷², which point to the cavity, are deleted, as is the salt bridge residue Asp¹⁷⁰. In the other two mutant enzymes, $\nabla 152-$ 153 and $\nabla 145-154$, cavity size may be influenced, but this is less clear since the amino acids surrounding the duplication are not in contact with the active site.

DISCUSSION

Structure-Activity Relationships of Mutant Enzymes-The mutants of haloalkane dehalogenase described here show that the cap domain of DhlA acts as an activity-modifying domain that can accommodate spontaneous mutations which can be selected in vivo and lead to alterations of substrate specificity. Since all mutants showed improved kinetics for 1-chlorohexane, the substrate binding site must have been altered in these enzymes such that improved binding and/or conversion is achieved. It is known that two tryptophans are involved in binding of the halide moiety of the substrate (10), but other residues are probably responsible for determining the substrate specificity. Candidates for residues that interact with the R-group of the substrate are Phe¹²⁸, Phe¹⁶⁴, Phe¹⁷², Phe²²², Pro²²³, Val²²⁶, Leu²⁶², and Leu²⁶³ (5, 6). All mutations that influence substrate range were located in the part of the cap domain that is N-terminal of Trp¹⁷⁵. This part is mainly stabilized by a surface located salt bridge between Asp¹⁷⁰ and Lys²⁶¹ of the main domain (Fig. 4) and hydrophobic interactions of 4 buried residues (Phe¹⁶¹, Phe¹⁶⁴, Phe¹⁷², and Gly¹⁷¹). Since buried resi-



FIG. 3. SDS-polyacrylamide gel electrophoresis of crude extracts from *Pseudomonas* GJ31(pPJ20) and its 1-chlorohexane utilizing mutants. See Table II for properties of the mutants. The *arrow* indicates the position of the haloalkane dehalogenase protein. Note the differences in mobility of mutant proteins carrying the deletion and duplications. The different lanes indicate crude extracts from: 1, 2, and 3, three independent mutants $\Delta 164-174$; 4 and 5, two independent mutants $\nabla 152-153$; 6, mutant P168S; 7, mutant $\nabla 172-174$; 8, mutant $\nabla 145-154$; 9, mutant D170H; 10, purified wild type haloalkane dehalogenase.

TABLE III Kinetic constants of dehalogenase mutants

Kinetic constants were determined by measuring alcohol production. 1,2-dibromoethane is included since it is the best substrate known and the $V_{\rm max}$ values are similar for the different mutants and the wild type enzyme.

	Strain	1,2-D	bibromoeth	1-Chlorohexane			
		$V_{ m max}$	K_m	$k_{\rm cat}/K_m$	$V_{ m max}$	K_m	$k_{\rm cat}/K_m$
		units/mg	тм	$s^{-1}mM^{-1}$	units/mg	тм	$s^{-1}M^{-1}$
	Wild type	6.0	0.0072	483	0.25	1.38	0.11
	D170H	4.4	0.087	30	0.59	0.88	0.40
	P168S	8.1	0.031	152	5.51	1.17	1.8
	$\Delta 164 - 174$	4.5	0.35	7.5	0.85	1.0	0.5
	$\nabla 172 - 174$	4.06	0.092	26	0.46	0.46	0.21
	$\nabla 152 - 153$	4.5	0.18	14.5	1.29	0.34	2.2
	$\nabla 145 - 154$	6.4	0.51	7.3	1.06	0.26	2.3

dues serve as structural anchors and strongly resist translocation (25) and because of the low number of hydrogen bonds and salt bridges, structural changes will be easily accommodated by the N-terminal part of the cap domain. If the mutations influence the Asp¹⁷⁰–Lys²⁶¹ salt bridge, as indicated by examination of the position and role of the residues influenced by the mutations, then this part of the cap domain could have undergone significant changes that allow improved binding or conversion of long chain chloroalkanes.

The mutations all influence the binding and/or reaction rate of various substrates (Tables II and III). The higher $V_{\rm max}$ for 1-chlorohexane shows that the reaction of bound substrate is faster in the mutants. Since this effect is observed with chlorinated but not with brominated substrates, formation of the covalent alkyl-enzyme intermediate must be the rate-limiting step in the wild type that is accelerated in the mutants. This may be caused by a better average positioning of the C α of the bound substrate for nucleophilic attack by Asp¹²⁴ and stabilization of the reaction intermediate. The lower K_m could indicate that substrate binding is improved in the mutants, but it can also be explained by a higher rate of formation of the covalent intermediate. How 1-chlorohexane is bound is unknown, but a larger or more flexible hydrophobic cavity that increases the binding energy of the larger substrate is likely. The wild type enzyme also must have considerable flexibility to allow substrate binding, since the active site cavity (5, 6) is too small for binding 1-bromohexane, which is a substrate for wild type haloalkane dehalogenase. Furthermore, the cavity is buried which requires some cap domain residues to move for substrate entrance. Large substrate-induced structural changes are observed in the mobile surface loops of lipase of *Rhizomucor miehei* (26), human pancreatic lipase (27), and lipoprotein lipase (28, 29), which are also α/β -hydrolase fold excursions (7).

Mechanism of in Vivo Mutagenesis-Of the six mutations found, only two appeared to be substitutions. The high frequency with which the 11-amino acid deletion mutant was found (6 out of 12 mutants) can be explained by a strand slippage mechanism involving the 9-bp imperfect repeat that flanks in the wild type sequence the segment that is deleted in the $\Delta 164$ -174 mutant. This can be followed by excision during mismatch repair or replication (Figs. 2 and 5). Such a process has been observed in β -galactosidase of Lactobacillus bulgaricus (15) and bacteriophage T4 (30) and was also proposed for the evolutionary divergence of benzoate dioxygenases (31). A role for replication was suggested (32, 33). Plasmid pPJ20 is derived from plasmid RK2, which has an unidirectional mode of replication (34). The orientation of the *dhlA* gene in pPJ20 is such that the direction of transcription is the same as the direction of replication (4). All deletion mutants still contained the proximal copy of the 9-bp imperfect repeat encoding FSA, not FTA. This means that if the strand slippage occurred during leading strand synthesis (Fig. 5A), excision and mismatch repair occurred only in the template strand, but if strand slippage occurred during lagging strand synthesis (Fig. 5B), excision and mismatch repair took place in both strands, which seems less likely.

The occurrence of three different mutations that are direct tandem repeats of varying length, rather than insertions of random sequence, suggests a specific mechanism for their origin. Formation of direct repeats has been found to occur at sites flanked by short repeats that create DNA structures that are easily misaligned, recombined and extended (30, 35, 36). The tandem repeats in the mutated *dhlA* genes are not flanked by repeated sequences, however, indicating that specific sequences are not required for their generation. The results suggest that an aspecific stuttering type of process can lead to the formation of short repeats, irrespective of existing repetitions.

Evolution of Haloalkane Dehalogenase-1.2-Dichloroethane is a synthetic compound. Although traces of it may be formed by haloperoxidases (37), it is unlikely that sufficient selective pressure to generate a dehalogenase for this chlorinated hydrocarbon existed until its industrial production and emission started in 1922. DhlA could have adapted to 1,2-dichloroethane by similar modifications of the cap domain as found here in the spontaneous mutants. Our results present experimental evidence that the cap domain, which is an excursion on the α/β hydrolase fold structure (7), is involved in the evolutionary modification of specificity, and that generation of repeats is an important mutational event during this process. The DNA coding for the N-terminal part of the cap domain of the wild type enzyme already contains two in frame direct repeats: a 15-bp perfect direct repeat encoding the sequence Val-Thr-Gln-Pro-Ala and a 9-bp repeat with one mismatch encoding Phe-Ser-Ala and Phe-Thr-Ala, respectively (Fig. 2). The two large repeats are in the same reading frame, but they are not tandem. It is highly unlikely that the repeats are required for the optimal catalytic performance of the enzyme. The direct repeats may well have been generated from an older dehalogenase during a



FIG. 4. Part of the structure of haloalkane dehalogenase with the mutations indicated. Stereoscopic depiction of the active site of wild type haloalkane dehalogenase. Asp¹²⁴ acts as the nucleophile (9), His²⁸⁹ and Asp²⁶⁰ are probably involved in the general-base catalyzed hydrolysis of the covalent intermediate, and Trp¹²⁵ and Trp¹⁷⁵ play a role in substrate binding and halide release (10). The two spheres represent water Side chains of the residues Asp^{124} , Trp^{125} , Phe^{128} , Trp^{175} , Asp^{260} , Leu^{262} , Leu^{263} , and His^{289} line the active site cavity. In mutants D170H, P168S, and $\nabla 172-174$ the salt bridge between Asp^{170} and Lys^{261} can be disturbed such that the active site cavity is larger or more flexible, allowing improved conversion of 1-chlorohexane. Probably, the active site cavity residues Phe^{124} and Phe^{172} have become more mobile or have slightly different positions in the mutants. These residues are both deleted in the $\Delta 164-174$ mutant. The $\nabla 145-154$ mutant is indicated by the *thick line* on the *right* side of the structure, and the $\nabla 152$ -153 mutant by the intervening thin part in the sequence.



FIG. 5. Proposed models for the formation of the A164-174 deletion during replication. Replication of the plasmid and transcription of dhlA are in the same direction (4, 34). The two direct repeats which are present in wild type dhlA are indicated by lines. The broken line indicates the 9-bp imperfect repeat that can be involved in strand slippage. A loop of DNA is excised and a g (bold) is mismatch repaired to c. The result is a deletion of 33 bases. In A, leading strand synthesis during replication is indicated by the arrow. Mismatch repair occurs only in the template strand. In B, the arrow indicates lagging strand synthesis, and repair must occur in both strands. The slipped DNA is shown in italics.

process similar to the selection of the mutations that arose during cultivation on 1-chlorohexane, with a few additional fine tuning modifications leading to the present wild type DhlA. Thus, we hypothesize that the direct repeats in the DNA sequence encoding the N-terminal part of the cap domain are of recent evolutionary origin and were selected during the adaptation of an older dehalogating enzyme to industrially produced 1.2-dichloroethane.

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