**Date:** April 11, 2022 **To:** American Institute of Chemical Engineers **From:** Kelsey DePrang, Karissa Huffman, and Aubrey Norton

To whom it may concern,

The AIChE 2021-2022 Student Design Competition was assigned in January 2022 and was submitted April 2022. The purpose of the project was to design three different-sized modular processes that convert natural gas into hydrocarbons that can be repurposed instead of releasing harmful gases into the atmosphere. Once the modular units were designed, the next step was to optimize the deployment of the modules to various wellheads over the 20-year timeframe. The goal was to reduce environmental impacts while making profit off the new hydrocarbon products.

This modular gas-to-liquids synthesis consisted of a Syngas Unit that takes in natural gas, steam, CO<sub>2</sub>, and O<sub>2</sub> to form syngas. The syngas was then sent to a Fischer-Tropsch Reactor that converts the syngas into multiple hydrocarbon chains. The reactor was designed as a fixed bed tubular reactor with a Cobalt-based catalyst. These hydrocarbons are sent on to the separation unit to be separated into their respective product streams. The heavy distillates/waxes are sent on to the Hydro-isomerization unit to be converted into lighter products to be sold or repurposed. The products made include tail gas, LPG, naphtha, water, paraffin, and diesel fuel. The products that will be sold are LPG, naphtha, and diesel fuel.

The results of the project show that the present worth cost (PWC) is approximately \$4.69 B and the net present value (NPV) is approximately \$4.35 B. The discounted cash flow rate of return (DCFROR) is 9.10% with a minimum rate of return of 8%. This indicates that the project is successful but has chances for error. The close range indicates that the project has a potential to have the DCFROR decrease below 8%, however it reduces carbon footprint of harmful gases and should still be pursued by the company.

Sincerely,

Kelsey DePrang, Karissa Huffman, and Aubrey Norton

# 2021-2022 AIChE Design Competition Report

## Modular Distributed Gas-to-Liquids Synthesis

April 11, 2022

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#### **Executive Summary**

AIChE has challenged teams with the task of designing three sizes of modular Gas-To-Liquids systems with the purpose of reducing harmful gas emissions into the atmosphere. This task was accomplished by using Fischer-Tropsch synthesis which turns syngas into hydrocarbon chains that can be repurposed and sold for a profit. These products include LPG, naphtha, and diesel fuel. The project also included optimizing the deployment of the different sizes of the modular units. The solution was designed to maximize the NPV and DCFROR as well as mitigate risks to the environment and to a person's health and safety.

Three sizes of modular units are predefined to have natural gas feeds of 500 MSCF/day, 2,500 MSCF/day, and 5,000 MSCF/day. It was assumed that a Syngas Unit, an Air Separation Plant, a Steam Plant, a CO<sub>2</sub> Recovery System were already designed for each modular size and only needed estimates of capital cost and utility costs for the total cost estimate. The focus for this preliminary design was a vertical Fischer-Tropsch reactor and a separations unit including one two-phase separator and two three-phase separators.

This design poses a small risk for explosion or damage to surrounding areas because of its modular size. Units will be transported between multiple well sites and therefore caution should be taken to not improperly assemble units after transportation. Hydrocarbons are fire hazards therefore safety measures were put in place to mitigate the risks.

It was determined that the optimal number of units needed for the project would be 13 small units, 9 medium units, and 5 large units. The capital costs of each small, medium, and large unit were \$13.3 MM, \$36.9 MM, and \$ 50.91 MM, respectively. Annual utilities were estimated to be \$1.45 MM, \$8.60 MM, and \$14.8 MM for the small, medium, and large units, respectively. Annual revenue for the small, medium, and large units were calculated to be \$0.54 MM, \$3.66 MM, and \$5.79 MM, respectively. Four five-year truck contracts will be needed for the duration of the 20-year project life consisting of 10, 9, 8, and 6 trucks for the individual contracts. It was estimated that the modular units would be built January of 2024 and deployed December of 2024 to begin operating in January of 2025. Estimating a build date of January 2022 allowed for 6 to 12 months of detailed design as well as 6 months for bidding of equipment and an additional 6 months to obtain required permits before units are built. A positive net present value of \$4.35 B was achieved for this project. An internal rate of return was calculated to be 9.1% which is greater than the 8% hurdle rate given for the project. The payback period was calculated to be 6 years and 9 months which would occur September of 2030. The project NPV and DCFROR show that the project design is economically attractive.

#### **Introduction**

Oil production has historically had catastrophic environmental impacts, with flaring of natural gas being one. Natural gas is mainly composed of methane and, when flared, is converted to carbon dioxide and released into the atmosphere. This process occurs, since methane has almost 25 times the greenhouse contribution than carbon dioxide. However, this process has resulted in the accumulation of carbon dioxide in the Earth's atmosphere with associated environmental impacts. The U.S. Energy Information Administration (EIA) reported that approximately 430,000 oil wells produced about 6.5 trillion cubic feet (tcf) of gases in 2019. The flaring of natural gas has contributed to about 111,000 metric tons of atmospheric carbon dioxide per year.

However, an alternative method for natural gas is currently being considered. To minimize the environmental impact, it is possible to convert the natural gases into liquids using Fischer-Tropsch Liquid (FTL) technology. The scope of this project included the conversion of methane into longer hydrocarbon liquids to prevent flaring of natural gas and produce additional profit. A modular approach was taken into consideration to make the system scalable and easy to move. The system consisted of three module sizes capable of a feed of 500 MSCF/day (small unit), 2,500 MSCF/day (medium unit), and 5,000 MSCF/day (large unit) of natural gas. The Syngas Unit, the Hydro-isomerization Unit (HIU), the Air Separation Unit, the CO<sub>2</sub> Separation System, and the Steam System have already been designed. The overall goal of the project included two parts. The first was to optimize the design of the Fischer-Tropsch reactor (FTR) and the separation equipment The second was the optimization of equipment deployed over the project lifetime at various wellhead sites.

#### **Summary**

This report covers topics ranging from design specification and practices to a sensitivity analysis of the various costs needed to make this project successful as specified in the preliminary design statement. The main driving factors of this project were to reduce the carbon footprint of  $CO_2$  by converting natural gas into diesel and naphtha that could be sold for a profit instead of being flared. Optimizing time and location of different modules deployed at the wellheads was another driving factor for increasing the project's net present value (NPV). The project timeline spans 20 years, concluding in the year 2044.

Safety and environmental analyses were done to make sure the designed process minimized as much risk as possible. Inherently safe decisions were made throughout the preliminary design process to reduce costs and to minimize risks. A Process Hazard Analysis (PHA) was done which includes: the worst-case scenario of damage, a P&ID of the separation unit, and potential process hazards of the materials being used.

Based on the preliminary design economic results, this project generates a NPV of \$4.35 B, breaking even in September of 2030. As such, it should be pursued. The sources of errors were analyzed for the design to clarify potential downfalls to the economic analysis. This included performing two sensitivity analyses with the Monte-Carlo Method and a Tornado Chart.

#### **Project Premises**

The entire Fischer-Tropsch (FT) synthesis process requires many specifications to run properly. Each unit had its own set of specifications that needed to be followed as given by AIChE design statement. The goal of the upstream process was to produce syngas. The syngas reactor required a temperature range of 1600-1950 °F and a pressure range of 300-500 psig. The steam and natural gas entering the process needed to be at a minimum of 0.5 mol/mol ratio to prevent coking in the feed preheater. The syngas product was specified to have H<sub>2</sub> and CO at a 2:1 ratio to equal the consumption ratio in the FTR. The maximum feed preheat temperature was set to be no more than 1000 °F. The Air Separation Plant also provided O<sub>2</sub> for the Syngas Unit. The O<sub>2</sub> was supplied at 99% purity with 1% mol nitrogen.

The design portion of the project included the FTR and the separation unit. It was assumed that the vapor phase contained 100% of the H<sub>2</sub>, CO, H<sub>2</sub>O, CH<sub>4</sub>, N<sub>2</sub>, and 0.7 moles of C2+ hydrocarbons for every mole of methane made. The maximum allowable pressure drop per reactor was 50 psi. The separator unit could not have the temperature of the distillate and heavier boiling fractions be lower than 250 °F because it could lead to wax crystallization.

The downstream process included the Hydro-Isomerization (HI) Unit where the wax was converted to lighter products and the paraffins are isomerized. This process converted 100% of the hydrocarbons with a boiling point greater than 700 °F to lighter materials with the given overall selectivities of 1.0 wt% methane, 0.5 wt% ethane, 3.5 wt% propane, 3.5 wt% butane, 25 wt% naphtha, and the balanced diesel. The free water could not be above the solubility limit that was fed to the HI unit. The makeup gas could not have a CO content higher than 0.1 mol%.

The FTR units will be deployed to wellheads located various distances from the central plant where the products will be transported for further processing and be blended into finished fuels. The amount of gas produced by each well and the distances from the central plant are listed in Table 1.

Associated	Initial Associated	Distance from
Gas Well ID	<b>Gas Production</b>	<b>Central Plant</b>
	(MSCF/day)	(Miles)
1A	4,494	62.0
1B	2,989	21.0
1C	8,365	17.0
1D	7,447	73.4
1E	3,290	6.2
1F	14,737	91.0
1G	11,155	59.0
1H	7,081	89.3
2A	9,182	85.8
2B	12,258	84.7
2C	8,742	2.4
2D	3,840	42.0
2E	7,939	98.8
2F	2,874	13.0
2G	6,076	87.0
2H	6,468	36.0

Table 1: Wellhead Information

#### **Discussion**

#### a. Design Philosophy

Driving factors for the design of this project included a positive NPV and DCFROR above the 8% hurdle rate, creating modular units that meet specifications, optimize wellhead placement, and a process that can be operated safely as well as sustainably. Major assumptions and key design elements implemented to reach objectives are listed below:

- Maximize the amount of diesel produced
- Optimize product of carbon chains of twenty-one carbons and above out of the FTR
- Optimize the inlet flowrate of carbon chains into the HIU
- Minimize amount of catalyst needed in the FTR
- Minimize equipment size to fit shipping containers for easy transfer
- Remove excess water and carbon dioxide

#### b. Description of the Process

Natural gas (assumed to be pure methane) is mixed with  $CO_2$  and high-pressure steam. A minimum 2:1 mole ratio is kept between natural gas and steam. The mixture is then sent to a preheater and heated to a temperature of 1000 °F.  $O_2$  from the Air Separation Plant is added after the preheater and supplied at a rate that allows for a 2:1 ratio of hydrogen and carbon dioxide to be produced from the syngas reactors. Next, the stream enters the Syngas Reactor System by first entering the Steam Methane Reforming (SMR) reactor and then into the Water Gas Shift (WGS) reactor. For the purposes of this project, the SMR and WGS reactors were modeled as one reactor. The Syngas Unit operated close to 1600 °F and 450 psi. After the Syngas Unit, the

stream was cooled to 212 °F and entered a separator to remove water from the mixture. The stream was then sent to an absorber to remove excess carbon dioxide before being sent to the FTR unit. Removing water and carbon dioxide allows for a smaller FTR to be designed.

The FTR was designed to be a fixed bed tubular reactor that uses a Cobalt-based catalyst. The FTR unit is ran at the max temperature of 450 °F to optimize conversion and the amount of higher carbon chains in the product stream. The product stream from the FTR is then sent to a cooler. The product stream is cooled to 275 °F before entering a two-phase separator. The bottom stream of the two-phase separator is heated to 420 °F by a heater before entering a three-phase separator. This three-phase separator removes wastewater from the bottom and LPG lighter gases out the top so that a heavy carbon chain stream can be sent to the HIU. The LPG product stream is then cooled down to forma liquid. The top vapor stream from the two-phase separator stream are separated.

#### c. Design Practices

#### i. Syngas Unit

The Syngas Unit was modeled using Aspen HYSYS. The fluid package used for simulation was CPA. Each size of unit was simulated then optimized using the given specifications. The given natural gas rates of 500 MSCF/day, 2500 MSCF/day, and 5000 MSCF/day were converted to lbmole/hr then put into a stream at the given temperatures and pressures. The natural gas stream was mixed with high pressure (HP) steam and CO<sub>2</sub> which was then heated before adding the purified air stream. The incoming steam and natural gas were kept at a minimum of 0.5 molar ratio to prevent coking in the syngas feed preheater. The mixed stream after the preheater was sent into the Syngas Reactor System which was modeled as a Gibbs reactor in Aspen HYSYS for simplicity [1]. The steam reforming and water gas shift reactions are in equilibrium while the partial oxidation reaction goes to completion. A Ln(Keq) equation was inputted into the reaction setup for the steam reforming and water gas shift reactions [2]. The partial oxidation reaction was modeled as if it went to equilibrium by using an extremely high Keq value (100,000) to favor a forward reaction.

After determining the amount of methane and steam entering the reactor, the  $O_2$  stream flow rate was adjusted to achieve the temperature required for the overall composition of the synthesis gas. The  $O_2$  for the process is purchased from a third-party packaged  $O_2$  plant. The Air Separation Plant provides a 99% pure supply of  $O_2$  with 1 mol% of nitrogen. The  $O_2$  enters the syngas system at 500 psig and 75°F. The flowrate of CO<sub>2</sub> was then supplemented to the mixed stream to meet the specification of a 2:1 molar ratio of H<sub>2</sub> and CO that would be entering the FTR. The flows of  $O_2$  and CO<sub>2</sub> were optimized by using a case study in HYSYS. The molar flow rate of CO<sub>2</sub> and O<sub>2</sub> were set to be the independent variables while the molar flow of H<sub>2</sub> and CO in the syngas were the dependent variables. The results of each case study were exported to Excel and organized to find the cases where the molar ratio of H<sub>2</sub>/CO would be 2. The flowrates that had the lowest O<sub>2</sub> rate with the temperature needed (~1600°F) were ultimately chosen as the initial flowrates of O<sub>2</sub> and CO<sub>2</sub>. The lower O<sub>2</sub> rate minimize the cost since it was the more expensive reactant. The inlet flowrates to the syngas reactor are listed in Table 2.

Component	Small Unit Molar Flowrate (lbmol/hr)	Medium Unit Molar Flowrate (lbmol/hr)	Large Unit Molar Flowrate (lbmol/hr)
Natural Gas (Methane)	54.02	270.12	540.20
Steam	27.01	135.05	270.10
Carbon Dioxide	11.00	45.00	105.00
Oxygen	20.79	123.75	216.81
Nitrogen	0.21	1.25	2.19

Table 2: Inlet Flowrates and into Syngas Reactor

lists the outlet flowrates of the syngas reactor that were modeled in Aspen HYSYS. Approximately 64.8%, 77.5%, and 67.7% of methane was converted in the syngas reactor for the small, medium, and large units, respectively. The  $H_2$  to CO molar ratio was approximately 2:1 for the small, medium, and large units.

Table 3 lists the outlet flowrates of the syngas reactor that were modeled in Aspen HYSYS. Approximately 64.8%, 77.5%, and 67.7% of methane was converted in the syngas reactor for the small, medium, and large units, respectively. The  $H_2$  to CO molar ratio was approximately 2:1 for the small, medium, and large units.

Component	Small Unit	Medium Unit	Large Unit
	Molar Flowrate	<b>Molar Flowrate</b>	Molar Flowrate
	(lbmol/hr)	(lbmol/hr)	(lbmol/hr)
Natural Gas (Methane)	18.99	60.66	174.29
Carbon Monoxide	32.86	196.76	343.33
Hydrogen	65.67	393.57	686.71
Steam	31.39	160.40	315.22
Carbon Dioxide	13.17	57.69	127.59
Nitrogen	0.21	1.25	2.19

 Table 3: Outlet Flowrates of the Syngas Reactor

After the Gibbs reactor, the syngas stream was approximately  $1600^{\circ}F$  and was sent to a cooler and cooled down to the boiling point of water ( $212^{\circ}F$ ). Most of the water was removed from the stream by being sent to a two-phase separator. The syngas stream also needed to have CO<sub>2</sub> removed before being sent on to the FTR to reduce the size, so a splitter in HYSYS was used to simulate the CO<sub>2</sub> separation unit removing 95% of the CO<sub>2</sub>. The removed CO<sub>2</sub> syngas stream was heated up to  $450^{\circ}F$  and sent to the FTR unit.

ii. Fischer-Tropsch Reactor

The Fischer-Tropsch reactor converts syngas into hydrocarbons and water. The reactants are hydrogen and carbon monoxide and can be described by Equation 1.

$$xH_2 + CO \rightarrow H_2O + (CH_2)_nH_2$$
 (Equation 1)

It has been determined that the reaction can be modeled by a Langmuir-Hinshelwood form of the overall rate equation. It has been modeled as the rate of conversion of CO and is expressed by Equation 2.

$$-r_{CO} = \frac{(k * T_1 * P_{H_2} * P_{CO})}{(1 + k_2 + T_2 * P_{CO})^2}$$
(Equation 2)

An ultra-stable Cobalt-based FTR catalyst was specifically designed for this process with a low deactivation rate. The average activity for the life of the catalyst is implemented in the rate of conversion of CO.

The Fischer-Tropsch reactor was modeled in Polymath with the Ordinary Differential Equations Solver as a vertical Packed-Bed Reactor (PBR). A vertical reactor was used because the orientation allows the reactor to fill completely. Furthermore, the catalyst can be installed and replaced with ease. It also takes advantage of gravity-controlled flow of all the fluids.

The reactor was modeled based on the conversion of CO as a differential of weight of catalyst. The rate of methane is dependent on temperature over the range 390 F° – 450 F° and varies slightly. The reactor was designed to operate at a temperature of 450 F° since this allowed for the greatest rate of conversion and more diesel and naphtha to be produced. Ethane, propane, and butane selectivities were then obtained as 4% of methane's selectivity. The formation of water is proportional to the conversion of CO. Hydrogen was estimated to be an average mole ratio of 2.06 for the multiple reactions. CO<sub>2</sub> and N<sub>2</sub> were considered inert gases.

Pressure drop was also considered based on conversion and varied based on the catalyst weight. The partial pressure of CO and  $H_2$  was related to conversion with the ideal gas law and then corrected to consider the mole balance. The vapor phase components were estimated to contain 100%  $H_2$ , CO,  $H_2O$ , CH<sub>4</sub>,  $N_2$  components, and 0.7 moles of C2+ hydrocarbons for every mol of CH<sub>4</sub> made.

Before the reaction was optimized, a Levenspiel Plot was made to determine the required volume of the reactor by taking the integral at desired conversion [3]. Figure 1 shows how conversion is affected by the rate of conversion of CO. Since there is a large jump from 80% to 90% of conversion, the most efficient conversion would be 80% resulting in a smaller reactor. When sized and optimized, the reactor with 80% conversion fit within the size restrictions, however the conversion was increased to 90% to obtain more heavy hydrocarbons and result in a higher profit and still fit within the restrictions.

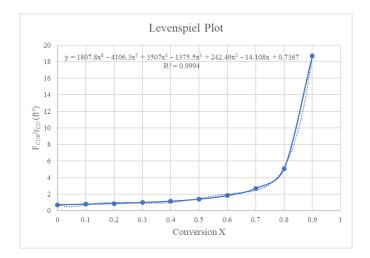


Figure 1: Levenspiel Plot for Fischer-Tropsch Reactors

The Fischer-Tropsch reaction is highly exothermic but was modeled as isothermal and then coolant was provided at a sufficient rate to cool the reactor down to prevent a runaway reaction. The heat of reaction of 70,200 Btu/lbmol of CO converted was multiplied by the amount of CO converted. This calculation provided the total duty occurring within the reactor at 450°F. The duty was then divided by the latent heat of MP pressure steam to calculate the mass flowrate of medium pressure steam needed to maintain a temperature of 450°F.

The Anderson-Schulz Flory (ASF) probability distribution was implemented to determine the hydrocarbon distribution of carbon chains longer than C5. Only carbon chains up to C30 were modeled to be able to replicate the stream into Aspen HYSYS. The relative mass fraction of the C5+ products are listed in Table 4. Based on this distribution, the percentage of the individual C5+ products were determined, and the mass flowrate of these products were obtained by multiplying by the summation of the C1-C5 and water produced during the reaction. A material balance was performed on each of the units and can be found in the Appendix.

Component	Relative Mass Frac.	Component	Relative Mass Frac.	Component	Relative Mass Frac.
C5	0.0332	C14	0.0357	C23	0.0226
C6	0.0359	C15	0.0344	C24	0.0212
C7	0.0376	C16	0.0330	C25	0.0198
C8	0.0386	C17	0.0316	C26	0.0185
C9	0.0391	C18	0.0300	C27	0.0173
C10	0.0391	C19	0.0285	C28	0.0161
C11	0.0386	C20	0.0270	C29	0.0150
C12	0.0379	C21	0.0255	C30	0.0140
C13	0.0369	C22	0.0240		

Table 4: Relative Mass Fractions of C5+ Products

The Fischer-Tropsch reactor was optimized with four variables in mind: number of tubes (n), catalyst weight (W), temperature (T), and tube diameters (D). These four variables directly affect the products produced and size of the reactor. The products were optimized by maximizing Page 8 of 48

the C5+ components to achieve a greater conversion of heavier carbon chains that result in more revenue sold as naphtha and diesel. The reactors were sized to maintain the constraints of a conventional intermodal shipping container with the dimensions of 8 ft wide x 8 ft 6 in tall x 40 ft long.

After simulating the system in Polymath, the code was translated to MATLAB using the ODE45 differential equation solver. This allows use of MATLAB's built-in optimization toolbox which solves multivariate systems simultaneously. The only requirements for this optimization program are a cost function and constraint functions. The cost function, described in Equation 3, is dependent on the monetary cost of the catalyst and the amount of tubes in each reactor. The "a" term in Equation 3 is the scaling factor for the cost function and was determined to be on the order of  $10^{-3}$  for all reactors. This adjusts the order of magnitude to be a constant offset because of the price fluctuations. The "n" term is the number of tubes in the reactor and the "W<sub>f</sub>" term is the catalyst weight in pounds. The cost function accounts for the fact that the catalyst would need to be changed at least every 3 years for the 20-year project life and the catalyst costs \$10 per pound. Equation 3 expresses the resulting cost function and was squared to obtain a parabolic curve to determine the minimum.

$$f = a * \left(\frac{20}{3} * 10 * n * W_f\right)^2$$
 (Equation 3)

The constraints implemented in the solver are listed in Table 5. The length of the reactor was determined based on the size restrictions of the shipping containers to not exceed a height of 8 feet to allow for some clearance in the container. The inequality function for the number of tubes was determined by an L/D ratio of 3. The maximum number of tubes was determined by calculating how many tubes could be arranged in a circular bundle meeting the ratio constraints [4]. The temperature constraint maintained the reactor temperature such that the selectivity of methane was applicable. Since the reactor was modeled as a tube and shell heat exchanger for the heat transfer, the tube diameters were limited to 2 inches which is equivalent to 0.1667 feet. The overall diameter of the reactor was determined through Engineering Toolbox [4] for sizing the tubes inside of the shell and assuming that the total area of the inner tubes utilizes approximately 72% of the available surface area. Thus, the diameter of the reactor can be expressed by Equation 4.

$$D_{overall} = 2 \sqrt{\frac{n*A}{0.72*\pi}}$$
 (Equation 4)

From there, the L/D ratio was implemented again to restrict the total diameter of the reactor. The pressure drop ratio throughout reactor was limited by the operating pressure of the reactor and a maximum pressure drop of 50 psi. Conversion was set to 94% for the small reactors since the reactors will be operating for the longest period and the medium and large units were set to 90%.

Variables	Constraints
Length (ft)	L < 8
Number of Tubes	1 < n < 1795
Temperature (°F)	390 < T < 450
Diameter of Tubes (ft)	0 < D < 0.1667
Diameter of Reactor (ft)	$D_{Reactor} < L / 3$
Pressure Drop Ratio	$p_{Total} < 0.88$
Conversion	Xf = 0.9

Table 5: Constraint Functions for Optimization

MATLAB used the constraints given and the cost function to perform various iterations until a local minimum was found satisfying the constraints. The optimization completed the iterations when the objective function stopped decreasing in a feasible direction. The results were then tabulated in Table 6 with the overall diameter rounded up to a quarter foot for ease of construction and the length adjusted to desired L/D ratio.

Variables	Small Unit	Medium Unit	Large Unit
Number of Tubes	14.00	29.00	49.00
Tube Diameter (in)	2.00	2.00	2.00
Overall Diameter (ft)	1.00	1.25	1.50
Length (ft)	3.00	3.75	4.50
Volume (ft <sup>3</sup> )	2.36	4.60	7.95
Weight of Catalyst (lb)	2.51	3.61	4.43
Conversion	0.94	0.90	0.90

Table 6: Optimization Results for the FTR

#### iii. Separation Vessels

After using Polymath, MATLAB, and Excel to find the mass flow rates of the components exiting the FTR, Aspen HYSYS was used to simulate the separation process. The fluid package CPA was once again used for this simulation. A process stream was made with the component mass flow rates leaving the FTR. The process stream leaving the FTR was at 450°F, so it needed to be cooled down for the lighter products to not be vaporized. The stream was sent through a cooler and cooled to a temperature of 275°F. Any process stream prior to isomerization containing C11+ material should not be cooled under 250°F to prevent wax crystallization. The cooled stream was sent to a two-phase separator and split into lighter and heavy products. The lighter product stream was cooled down to a temperature of 250°F with a heat exchanger to allow for even better separation. This stream was sent to a three-phase separator to get the tail gas, naphtha, and wastewater. The bottom stream with the heavier components. The streams exiting the three-phase separator included LPG, the feed to the HIU unit with the heavy distillates and waxes, and wastewater. The LPG stream needed to be cooled down to 250°F to condense the C3 and C4 hydrocarbons to sell as liquid. Since LPG can be sold no matter what

the composition of the stream is, there are other components mixed into the LPG product stream. The naphtha and LPG were stored and transported without temperature concern. The HIU feed stream had to have CO below 1 mol% and water below the solubility limit. The wastewater from each three-phase separator was combined into one stream and then sent to a Wastewater Treatment Plant.

Each separator was designed after the separation unit was completely simulated. The two-phase separator (V-102) was designed as a horizontal separator because they are preferred for separating vapor from mixtures with a lower vapor/liquid ratio and are less expensive. The separator K value was found with an equation based on a pressure range so the range between 40 and 5,500 psia was applicable. Holdup and surge times were chosen to be 2 min and 1 min, respectively, due to this separator being a feed to another drum or tankage [5]. The diameter was found using an initial L/D estimate in the range from 3-4 because the pressure of the vessel was between 250 and 500 psig. The minimum length required to accommodate the liquid holdup and surge was calculated after the total area, vapor disengagement area, and the low liquid area were found. Both the diameter and length found were rounded up by 0.5 ft. The volume of the vessel was then calculated and was optimized by varying the initial L/D estimate. The final L/D values for the small, medium, and large units were 2, 3, and 3, respectively.

The three-phase separators were designed similarly to the two-phase separator. Many of the same assumptions and correlations were made, including the separator K value and the holdup and surge times. The three-phase separators are horizontal and controlled with a weir. This design maximizes the volume of heavy liquid being separated. A  $k_s$  value of 0.163 was chosen due to hydrocarbons being in the light phase and the heavy phase is water [6]. The diameters were found based on an initial L/D estimate and were later optimized to achieve a minimum volume. Length was calculated by adding the minimum length of the light liquid compartment to accommodate holdup and surge with the minimum length to facilitate liquid-liquid separation. The final L/D values for the three-phase separator of the lighter components (V-103) were 2, 3, and 3 for the small, medium, and large units, respectively. For the three-phase separator of the heavy products (V-104), the final L/D values were 2, 3, and 4 for the small, medium, and large unit, respectively.

#### iv. Heat Exchangers

For each unit size, four heat exchangers were designed to either heat up or cool down streams to optimize the amount of product being separated. The heat exchangers were modeled in Aspen HYSYS as coolers and heaters. For each unit size, the streams being cooled or heated change by the same temperature difference. The location of the first heat exchanger (E-103) is before V-102, and it was cooled from 450°F to 275°F. The next heat exchanger (E-104) is on the light stream separated by V-102 and it was cooled even further to 250°F. On the bottom stream from V-102 with the heavy products, there is a heat exchanger (E-105) that heats up the stream from 275°F to 420°F. The final heat exchanger (E-106) is on the LPG product stream to cool it down from 420°F to 250°F to liquify the stream.

The design portion consisted of taking the heat duty given by HYSYS and using it for finding the needed area of the heat exchanger. The LMTD method was used to find  $\Delta T_{lm}$  assuming counter-current heat exchangers for the best heat transfer. A 20-degree approach temperature was assumed for all heat exchangers. There were no entering and returning

temperatures given for the process cooling water so a heuristic of 90°F and 120°F entering and returning temperatures were used. The overall heat transfer coefficients (U<sub>o</sub>) were taken from the GPSA based off the contents of process stream whether process cooling water or HP steam was used. For E-103, a Uo of 160 Btu/(hr•ft<sup>2</sup>•°F) was used because the process stream consisted mainly of water and the cooler was using process cooling water. However, for E-104 a Uo of 130 Btu/(hr•ft<sup>2</sup>•°F) was used because the stream consisted mainly of hydrocarbons and was cooling with process cooling water. For E-105, Uo was assumed to be 130 Btu/(hr•ft<sup>2</sup>•°F) because the heater was using HP steam and the process stream consisted of mainly hydrocarbons. The LMTD correction factor (F) of 0.9 was used because the heat exchangers were designed as 1-shell pass with 2 or more tube passes. After all the necessary variables were found, the area of each heat exchanger was calculated [7].

Due to the modular approach to this project, the designed heat exchangers did not fall within many of the size ranges of various heat exchangers. All four of the heat exchangers for the small and medium units were designed to be Teflon tube heat exchangers because they fell within the size range given [8]. For the large unit, E-103 was designed to be a U-Tube heat exchanger because it was more cost-effective and was the largest of the heat exchangers. The remaining heat exchangers in the large unit were designed to be Teflon tube exchangers.

## v. Hydro-Isomerization Unit

The Hydro-isomerization Unit did not need to be designed for the purpose of this project, but conversions were given to find the flow rates of the product streams exiting the unit. The distillate and heavier boiling fractions were sent from the separation unit on to this unit to be converted into lighter products to sell. The HIU converted 100% of the material with a boiling point greater than 700°F to material with a boiling point less than 700°F. The selectivity of the feed that will convert was given as 1.0 wt% methane, 0.5 wt% ethane, 3.5 wt% propane, 3.5 wt% butane, 25 wt% naphtha, and the remaining would be diesel. Any material that was below 700°F was assumed to pass through the unit unconverted. The initial and final product flowrates can be seen for the small, medium, and large units in Table 7, Table 8, and Table 9 below, respectively.

Component	Inlet Flowrate (lb/hr)	<b>Converted Product</b> <b>Flowrate (lb/hr)</b>	Outlet Flowrate (lb/hr)
Methane	0.24	0.18	0.42
Ethane	0.01	0.09	0.10
Propane	0.05	0.63	0.68
Butane	0.16	0.63	0.79
Naphtha	68.74	4.53	73.27
Diesel	120.74	12.05	132.79
C11+	18.12	0.00	0.00

Component	Inlet Flowrate (lb/hr)	<b>Converted Product</b> <b>Flowrate (lb/hr)</b>	Outlet Flowrate (lb/hr)
Methane	1.33	1.08	2.41
Ethane	0.09	0.54	0.63
Propane	0.38	3.77	4.15
Butane	1.33	3.77	5.10
Naphtha	488.16	26.91	515.07
Diesel	729.71	71.58	801.29
C11+	107.64	0.00	0.00

#### Table 8: HIU Medium Unit Product Selectivity

 Table 9: HIU Large Unit Product Selectivity

Component	Inlet Flowrate (lb/hr)	Converted Product Flowrate (lb/hr)	Outlet Flowrate (lb/hr)
Methane	2.35	1.88	4.23
Ethane	0.12	0.94	1.06
Propane	0.51	6.59	7.10
Butane	1.81	6.59	8.40
Naphtha	744.53	47.05	791.58
Diesel	1260.3	125.15	1385.45
C11+	188.19	0.00	0.00

Inert material that passed through the HIU are not shown in the table since they did not convert but they include  $N_2$ ,  $H_2$ , CO, CO<sub>2</sub>, and water. Everything except the water had negligible amounts. The catalyst in the HIU was said to be sensitive to water and CO, so it was made sure that there was no free water above the solubility limit in the feed. The specification for CO was that the content in the HIU could be no greater than 0.1 mol%. The small unit HIU feed contains 0.04 mol% CO, the medium unit contains 0.05 mol% CO, and the large unit contains 0.05 mol% CO so the design is within that specification.

#### **Sources of Error**

There are numerous potential errors when the preliminary design was done for this process. Some could stem from assumptions made, the differences between modular and regular plants, or from human error. The heuristics used for the sake of this project were designed for conventional equipment and provided uncertainty for the module units.

When modeling the Syngas Unit in Aspen HYSYS, a singular Gibbs reactor was used to simulate the syngas being made. Since the steam reforming and water gas shift reactions are equilibrium and the partial oxidation reaction goes to completion, there were some difficulties during simulation. To get the simulation to work properly, the partial oxidation reaction was modeled as an equilibrium reaction using an extremely high Keq value to simulate it as if it was going to completion. This assumption could lead to some potential error in the output of this simulated reactor.

The Fischer-Tropsch reactor was designed using Polymath, MATLAB, and Excel. Once completely modeled using Polymath and MATLAB, the ASF distribution was used in Excel to determine the relative weight fraction and the relative mole fractions of the C5+ products. There was a percent error in the mass balance due to this distribution. From calculations, there was more material leaving the FTR than there was entering and that is due to this error.

In the designed separation unit, a two-phase separator was used to make an initial separation of products. The two separated streams were then sent to two separate three-phase separators. This gives decent separation of products but could potentially be better with a distillation column.

#### Heat and Material Balance

For the heat and material balance, a stream table was put together to show the molar, mass, and actual volume flow rate as well as the mass enthalpy of each stream. There was a percent error for the material balance between what is entering the FTR and what is leaving the FTR. This error is due to the ASF probability distribution which was not exact to the mass flow rate coming in. The small unit had a percent error of 7.54%, the medium unit had a percent error of 7.76%, and the large unit had a percent error of 1.52%. The stream tables are based off the simplified process flow diagram of the whole unit from the Syngas Unit to the HIU. The stream table for the small unit can be seen in Table 10 below. The stream tables for the medium and large unit can be found in the appendix.

#### **Overall Process Flow Diagram**

The overall process flow diagram (PFD) spans from the Syngas Unit all the way through the HIU. The stream table shown in Table 10 below gives a complete summary of each labeled stream in this PFD. For each control valve, a pressure drop of 10 psi was assumed. For each heat exchanger, a pressure drop of 4 psi was assumed [8]. The overall PFD for the small unit can be seen in Figure 2 below. The PFDs for the medium and large units can be found in the appendix.

## Table 10: Small Unit Stream Table

	Stream Number		1	2	2	4	с	c	-	0	0	10	1	1 12	12	14	10	16	17	18	19	20	21	22	23	24	25	26	27	28	20	20	21	22
	Stream Number		1	2	3	4	5	H-101	,	0	3	CO2	V-101	1 12	15	14	15	10	17	10	19	20	Ton	22	25	24	23	20	Bottom	Top & Bottom	-	50	51	52
			Oxygen	Natural	HP Steam	Carbon	H-101		R-101 Feed	E-101 Feed	V-101	Separation	Waste	Removed	E-102 Feed	R-102 Feed	-103 Feed	V-102	E 104 Food	V-103 Feed	Tail Gas	Тор	Top Water	E-105	V 104 Food	E-106 Feed	Cooled LBG	LILL Eood		Water	Waster	Paraffin	Bottom	Diesel
	Description		Oxygen	Gas	nr steam	Dioxide	Feed	Stream	K-101 Feeu	E-101 Feeu	Feed	Unit Feed	Waster	CO2	E-102 Feeu	K-102 Feeu	-105 Feeu	Feed	E-104 Feeu	v-105 Feeu	Tall Gas	Nanntna	Product	Feed	v-104 reeu	E-100 Feeu (	COOIEU LFG	nio reeu	Product	Products	Disposal	Falaliii	Naphtha	Fuel
	Description							Stream				onicreed	water										TTOULCE						TTOULCE	Troducts	Disposal			
	Vapor Fraction		1.00	1.00	0.00	1.00	0.74	1.00	1.00	1.00	0.83	1.00	0.0	0 1.00	1.00	1.00	1.00	0.52	1.00	0.96	1.00	0.00	0.00	0.00	0.04	1.00	0.11	0.00	0.00	0.00	0.00	0.21	0.00	0.00
	Temperature	°F	75.00	100.00	490.00		-	1000.00	890.84	1606.73		212.00	212.0			450.00	450.00		275.00		250.00		250.00	275.00	420.00		250.00			412.41				
	Pressure	psia	514.70	514.70	614.70	514.70		500.70	500.70	500.70		496.70	496.7	-	-	472.70	445.00		431.00		417.00		417.00		417.00		433.00			407.00				407.00
	Total Molar Flow	lbmole/hr	21.00	54.02	27.01	11.00	92.03	92.03	113.03	162.30	162.30	135.47	26.8	3 12.50	122.97	122.97	66.64	66.64	34.47	34.47	32.95	0.09	1.43	32.17	32.17	1.29	1.29	1.53	29.35	30.79	57.62	0.307	0.113898	0.06057
	Total Mass Flow	lb/hr	671.16	866.64	486.59	484.11	1837.33	1837.33	2508.49	2508.49	2508.49	2024.84	483.6	5 549.91	1474.93	1474.93	1535.46	1535.46	752.06	752.06	715.78	10.45	25.83	783.41	783.41	41.93	41.93	212.55	528.93	554.76	1038.40	6.08	73.27	132.78
	Actual Vol. Flow	barrel/day	970.66	2524.52	42.22	446.06	4099.64	12099.84	13744.58	30313.09	8267.78	8232.90	34.8	9 657.99	7574.99	10296.68	5915.98	2634.30	2566.81	2395.69	2392.63	1.16	1.90	67.49	170.98	100.65	13.46	27.17	43.15	44.99	79.21	0.42	5.10	9.24
	Mass Enthalpy	BTU/lb	-4.85	-2010.79	-6371.27	-3857.52	-3652.18	-2993.28	-2193.71	-2193.70	-3182.26	-2345.61	-6685.0	2 -3833.73	-1790.78	-1643.13	-3027.61	-3456.82	-1967.78	-2014.18	-1864.21	-849.21	-6640.86	-4886.27	-4730.07	-2771.92	-3272.32	-789.65	-6468.78	-6476.79	-6570.26	-5210.93	-859.82	-767.18
	Oxygen	lb/hr	665.28	0.00	0.00	0.00	0.00	0.00	665.28	0.00	0.00	0.00	0.0	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
	Water	lb/hr	0.00	0.00	486.59	0.00	486.59	486.59	486.59	565.58	565.58	82.61	482.9	7 0.00	82.61	82.61	620.15	620.15	69.88	69.88	44.04	0.03	25.81	550.27	550.27	17.13	17.13	4.44	528.70	554.52	1037.49	4.44	0.00	0.00
	Nitrogen	lb/hr	5.88	0.00	0.00	0.00	0.00	0.00	5.88	5.88		5.88	0.0	0.00	5.88	5.88	2.94	2.94	2.93		2.93	0.00	0.00	0.01	0.01	0.01	0.01	0.00	0.00	0.00	0.00	0.00	0.00	0.00
	Carbon Monoxide	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	920.46	920.46	920.42	0.0	4 0.00	920.42	920.42	84.89	84.89	84.65	84.65	84.64	0.01	0.00	0.24	0.24	0.22	0.22	0.02	0.00	0.01		0.02	0.00	0.00
	Carbon Dioxide	lb/hr	0.00	0.00			484.11	484.11	484.11	579.49		578.86	0.6		28.94	28.94	28.94		28.64		28.62	0.01	0.00	0.30	0.30		0.24		0.02	0.02			0.00	0.00
	Hydrogen	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	132.39		132.39	0.0			132.39	4.26	-	4.25		4.25	0.00	0.00	0.01	0.01	0.01	0.01	0.00		0.00			0.00	0.00
	Methane	lb/hr	0.00	866.64	0.00	0.00	866.64	866.64	866.64	304.68		304.68	0.0			304.68	359.91		357.87		357.76	0.09	0.02	2.04	2.04		1.72	-		0.10			0.00	0.00
	Ethane	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00		0.00	0.0			0.00	4.15	-	4.10		4.09		0.00	0.05	0.05		0.04		0.00	0.00			0.00	0.00
	Propane	lb/hr	0.00	0.00		0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.0			0.00	6.08		5.92		5.91	0.01	0.00	0.16	0.16	0.11	0.11	0.05	0.00	0.00			0.00	0.00
	n-Butane	lb/hr	0.00	0.00			0.00	0.00	0.00			0.00	0.0			0.00	8.01		7.61		7.58		0.00	0.40	0.40		0.24		0.00	0.00			0.00	0.00
	n-Pentane	lb/hr	0.00	0.00			0.00	0.00	0.00			0.00	0.0			0.00	55.86		50.65		50.29	0.36	0.00	5.21	5.21	2.50	2.50			0.03			-	0.00
	n-Hexane	lb/hr	0.00	0.00		0.00	0.00	0.00	0.00	0.00		0.00	0.0	-		0.00	51.93		43.16		42.54		0.00	8.77	8.77	3.23	3.23	1	0.03	0.03			H	0.00
Naphtha	n-Heptane	lb/hr lb/hr	0.00	0.00		0.00	0.00	0.00	0.00			0.00	0.0			0.00	46.93		33.99 24.21		33.04	0.95	0.00	12.94	12.94 17.34		3.61	9.31	0.02	0.02			73.27	0.00
	n-Octane n-Nonane	lb/hr	0.00	0.00		0.00	0.00	0.00	0.00	0.00		0.00	0.0			0.00	41.55 36.21	41.55	24.21		22.92 13.99		0.00	17.34 20.71	20.71	3.56	3.56		0.02	0.02			-	0.00
		lb/hr	0.00			0.00	0.00	0.00	0.00	0.00		0.00	0.0			0.00		30.21	9.06			1.52			20.71	2.30	2.30	-	0.01				ŀ	0.00
	n-Decane n-C11	lb/hr	0.00	0.00		0.00	0.00	0.00	0.00			0.00	0.0			0.00	31.17 26.56		9.06		7.51		0.00	22.11 21.74	22.11 21.74		2.30		0.01	0.01			0.00	0.00
	n-C12	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.0			0.00	20.30	20.30	2.44		1.40	1.04	0.00	20.00	21.74		0.98		0.00	0.00			0.00	
	n-C13	lb/hr	0.00	0.00		0.00	0.00	0.00	0.00			0.00	0.0	-		0.00	18.84		1.24		0.54		0.00	17.60	17.60		0.58		0.00	0.00			0.00	
	n-C14	lb/hr	0.00	0.00		0.00	0.00	0.00	0.00		0.00	0.00	0.0	_		0.00	15.71	15.71	0.57		0.16		0.00	15.14	15.14		0.35		0.00	0.00			0.00	
	n-C15	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.0			0.00	13.04		0.28		0.05	0.23	0.00	12.77	12.77	0.33	0.33	12.56	0.00	0.00			0.00	
Diesel	n-C16	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.0			0.00	10.78	10.78	0.13		0.01	0.11	0.00	10.65	10.65		0.12	10.53	0.00	0.00			0.00	132.78
	n-C17	lb/hr	0.00	0.00		0.00	0.00	0.00	0.00			0.00	0.0			0.00	8.87	8.87	0.13		0.01	0.07	0.00	8.80	8.80	-	0.08		0.00	0.00			0.00	
	n-C18	lb/hr	0.00	0.00		0.00	0.00	0.00	0.00			0.00	0.0	-		0.00	7.28		0.03		0.00		0.00	7.25	7.25		0.04		0.00	0.00			0.00	
	n-C19	lb/hr	0.00	0.00		0.00	0.00	0.00	0.00			0.00	0.0	-		0.00	5.95		0.01		0.00		0.00	5.94	5.94		0.02		0.00	0.00			0.00	
	n-C20	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00		0.00	0.0		0.00	0.00	4.85		0.01		0.00		0.00	4.84	4.84		0.01	1	0.00	0.00			0.00	
	n-C21	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.0	0.00	0.00	0.00	3.95	3.95	0.00	0.00	0.00	0.00	0.00	3.94	3.94	0.01	0.01	3.93	0.00	0.00	0.00	0.00	0.00	0.00
	n-C22	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.0	0.00	0.00	0.00	3.20	3.20	0.00	0.00	0.00	0.00	0.00	3.20	3.20	0.01	0.01	3.20	0.00	0.00	0.00	0.00	0.00	0.00
	n-C23	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.0	0.00	0.00	0.00	2.59	2.59	0.00	0.00	0.00	0.00	0.00	2.59	2.59	0.00	0.00	2.59	0.00	0.00	0.00	0.00	0.00	0.00
	n-C24	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.0	0.00	0.00	0.00	2.10	2.10	0.00	0.00	0.00	0.00	0.00	2.10	2.10	0.00	0.00	2.09	0.00	0.00	0.00	0.00	0.00	0.00
	n-C25	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.0	0.00	0.00	0.00	1.69	1.69	0.00		0.00	0.00	0.00	1.69	1.69	0.00	0.00	1.69	0.00	0.00	0.00	0.00	0.00	0.00
	n-C26	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00		0.00	0.0	0.00		0.00	1.36	1.36	0.00		0.00	0.00	0.00	1.36	1.36	0.00	0.00	1.36	0.00	0.00		0.00	0.00	0.00
	n-C27	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00		0.00	0.0			0.00	1.10	1.10	0.00		0.00	0.00	0.00	1.10	1.10		0.00	-		0.00			0.00	0.00
	n-C28	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00		0.00	0.0			0.00	0.88		0.00		0.00	0.00	0.00	0.88	0.88		0.00			0.00			0.00	0.00
	n-C29	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.0			0.00	0.71	0.71	0.00		0.00	0.00	0.00	0.71	0.71	0.00	0.00	-	0.00	0.00			0.00	0.00
	n-C30	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.0	0.00	0.00	0.00	0.57	0.57	0.00	0.00	0.00	0.00	0.00	0.57	0.57	0.00	0.00	0.57	0.00	0.00	0.00	0.00	0.00	0.00

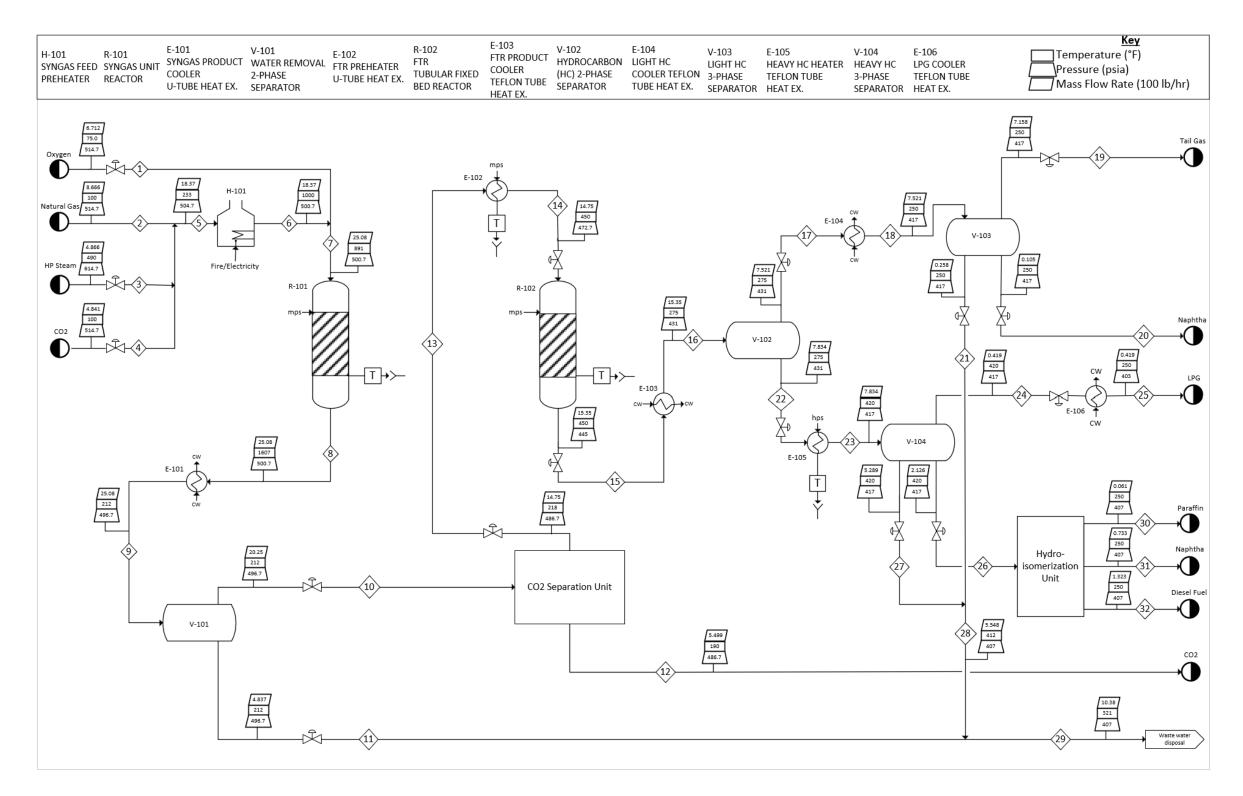


Figure 2: Small Unit Overall PFD

#### **Safety/Environmental Summary**

#### a. Inherent Safety Evaluation

Being inherently safe in the preliminary design stage was extremely important to the overall value of the project. There are numerous features of this project that were designed to be inherently safe including the materials of construction and the size of equipment. One feature that will make this process inherently safe will be making incorrect assembly impossible for dismantling and reattaching pieces of equipment during maintenance. This will be done by having specific labels for the pieces of equipment that will need to be replaced or maintained.

An example of minimization in the design was using a continuous-flowing reactor instead of a batch reactor. The FTR is a vertical fixed bed tubular reactor, which means it has more uniform mixing and less material in the reactor at a given time. It was determined to be a vertical reactor for ease of catalyst replacement. Another example of minimization is the trucks taking product back to the central plant periodically. The storage tanks will not hold more than 5,000 gallons of liquid at a given time. This reduces the amount of product that is kept at the wellheads. Also, since this whole process is modular, the pieces of equipment are smaller than normal equipment. This minimizes risk of potential large-scale damage.

For heat exchangers and the FTR, medium or high-pressure steam was substituted in place of fuel gas. This reduces risk of fire and the probability of an incident. The inherent safety practice of simplification was used in the separation system by using less pieces of equipment than what could be required. For completely perfect separation of each stream, multiple distillation columns would need to be used. The designed separator system only consists of three separators and four heat exchangers. All equipment was designed using stainless steel to have stronger vessels to handle the operating conditions. Stainless steel has an acceptable corrosion characteristic for gasoline while carbon steel is cautionary. Although gasoline is not a product, the intermediates contain a similar composition that could potentially result in corrosive behaviors. Even though carbon steel is significantly cheaper than stainless steel, a substitution of stainless steel will provide an inherently safer design.

#### b. Process Safety Management

This process was designed to mitigate risk and prioritize safety so that minimal accidents occur. If an accident occurs, it would be because the operating conditions have deviated outside the permitted range. During more detailed design, a more in-depth process hazard analysis should be done. Safety culture will also be important to maintain this process. Operators and other staff will be encouraged to say something or act if something seems off about the process. If workers remain proactive, the risk of hazards should decrease significantly. Proper training on how to operate the pieces of equipment and how they work will be vital to making sure that accidents are limited. Lockout tagout procedures should be implemented during maintenance or servicing to prevent mistakes from occurring during that time. Audits will also occur periodically to ensure that safety practices are maintained. An emergency response plan should be made in case of various emergencies including fires and loss of containment. Implementing a management of change plan will need to be created as well to prepare for various changes with the internal and external processes.

#### i. Process Hazards

There are numerous process hazards for this GTL process. Many of the components required for this process to work are flammable and could lead to fires and explosions if ignited. The components that mix to make syngas have many hazards that need to be monitored as seen in Figure 3 obtained from Cameo Chemicals [9].

	METHANE					
WATER	Compatible	WATER				
CARBON MONOXIDE	Compatible 🔲	Compatible 🗖	CARBON MONOXIDE			
HYDROGEN	Compatible 🗖	Compatible	Compatible 🗖	HYDROGEN		
NITROGEN	Compatible 🔲	Compatible	Compatible 🔲	Compatible 🗖	NITROGEN	
OXYGEN	Incompatible Flammable Flammable Generates gas Generates heat Intense or explosive reaction Toxic	Incompatible Corrosive Generates gas Generates heat Toxic	Incompatible Explosive Explosive Generates gas Generates heat Intense or explosive reaction Unstable when heated	Incompatible Explosive Explosive Generates gas Generates heat Intense or explosive reaction Unstable when heated	Compatible 🗖	OXYGEN
CARBON DIOXIDE	Compatible 🔲	Caution Corrosive Generates heat	Compatible 🔲	Compatible 🔲	Compatible	Compatible

Figure 3: Material Compatibility Chart

The lighter hydrocarbons that are being formed in the FTR are flammable and need to be kept safe from potential ignition sources. If something were to happen which caused the FTR to run outside of operating conditions, a runaway reaction could possibly occur. All the hydrocarbons formed are compatible together as opposed to the  $O_2$  in the Syngas Unit which could lead to potential fires or explosions.

High pressures and temperatures are used throughout the entire process, so overheating or overpressure events are possible. There is potential for overpressure in the FTR as well as the separator equipment. It will be important that the FTR has a rupture disk as well as a pressure relief valve on it to indicate an overpressure event. Pressure relief valves will also be on every separator being used in the units. Steam and cooling water are used to cool down or heat up different parts of the process. Medium pressure (MP) steam cools down the extremely exothermic reaction happening in the FTR and HP steam is used as a utility for one of the heat exchangers in the separator unit. Cooling water is used for three heat exchangers in each unit. These utilities are essential for ensuring temperatures do not fall out of their setpoint.

Human error is also a potential process hazard if proper procedures are not done correctly. If proper lockout tagout procedures are not followed, it could lead to failures in the system and potential injuries to people in the plant. Extensive training will be done on how to operate the equipment

## ii. P&ID of the Major Fractionator

Many safety features were taken into consideration when designing the controls system of the major fractionator. The major fractionator was assumed to be the separation unit part of the process, so control systems were only put on that section of the process. There were pressure relief valves and control loops put in place to prevent overpressure or other potential hazards. The full P&ID of the small unit can be found in Figure 4 below. The P&IDs for the medium and large unit can be found in the Appendix.

To prevent overpressure, pressure relief valves were placed in various places in the separation unit. The first valve is on the cooling water stream out of E-103. This was put in place in case the cooling water is heated too high and causes the tubing in the heat exchanger to expand and burst [8]. Other pressure relief valves were put on the cooling water streams coming out of E-104 and E-106 for the same exact reason. Pressure relief valves were also put on each separator because they are pressure vessels and could easily overpressure.

There were 11 different control loops designed to control different parts of the process to minimize the risks of this process as much as possible. On each of the process streams from the heat exchangers, a temperature control loop was put in place to control either the temperature of the cooling water entering or the HP steam entering the exchanger. This was done to ensure that the exiting process stream was not too hot or too cold for the next part of the unit. On V-102, a pressure control loop was needed to ensure the exiting vapor stream's pressure was not too high. Also, on V-102, there was a level control loop designed to ensure the liquid level in the vessel did not get too low or too high. On the three-phase separators, V-103 and V-104, two level control loops were designed. Since there were three process streams leaving the vessel, two of them being heavy and light liquids, two liquid level controls were needed to control loop to control loop to control the vapor pressure.

For the fail open and fail closed positions of the control valves, a decision was made based on what the safest position would be for that valve. The valves on the cooling water streams into the heat exchangers were made to be fail closed to ensure that the cooling water does not keep cooling the process stream for the risk of the temperature falling below 250°F. The control valve on the HP steam inlet to E-105 was chosen as fail closed to prevent the process stream from overheating and potentially causing a runaway reaction. For the pressure control loops on the vapor outlet of each separator, the position chosen was fail open to ensure there would not be a gas buildup and possibly overpressure the vessel. For the level control loops on the light and heavy liquid outlets of the separators, the fail closed position was chosen for the valve so that the liquid leaving does not continue moving on to the next part of the process. This helps stop the process in case a potential hazard arises.

Pipe diameters were calculated using heuristics given in the 1997 ASHRAE Fundamentals Handbook for gas flow [10]. For liquid and two-phase flow streams, equation heuristics that were used can be found in the Appendix [11]. It was assumed that for the modular system there would be 20 feet between each equipment. This is based on the heuristic that 50 feet is between equipment for conventional process plants [8].

E-103	V-102	E-104	V-103	E-105	V-104	E-106
FTR COOLER	HYDROCARBON	LIGHT HC COOLER	LIGHT HC	HEAVY HC HEATER	HEAVY HC	LPG COOLER
TEFLON TUBE	(HC) 2-PHASE	TEFLON TUBE	3-PHASE	TEFLON TUBE	3-PHASE	TEFLON TUB
HEAT EX.	SEPARATOR	U-TUBE HEAT EX.	SEPARATOR	HEAT EX.	SEPARATOR	HEAT EX.

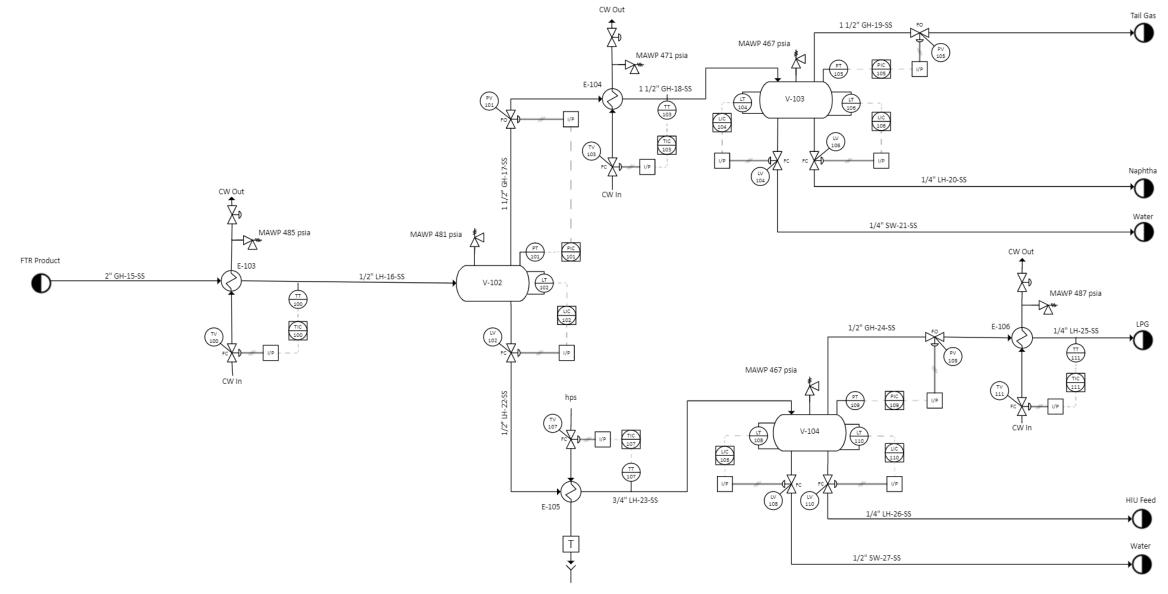
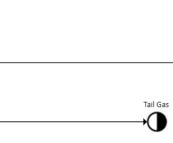


Figure 4: Small Unit P&ID



#### iii. Uncongested Vapor Cloud Deflagration

In a deflagration, the pressure front moves away from the reaction front, which is moving at less than the speed of sound. The potential explosions of this process were measured on an equivalent TNT basis. Based on that equivalent TNT value, the overpressure of the potential leak from the equipment was found and potential damage determined [12]. Due to this being a modular process with smaller pieces of equipment, the damage from a potential explosion is significantly less than a regular-sized process. All scenarios were calculated at 10 m (32.81 ft) from the explosion source. An explosion of efficiency of 2% was assumed and all required values for calculation were taken from Aspen HYSYS.

The small unit's worst-case scenario would occur at the FTR where the hydrocarbons are being formed. The damage for the small unit would be occasionally breaking of large glass windows already under strain. The worst-case scenario for the medium unit would also occur at the FTR. The damage for this unit would be a loud noise (143 dB), sonic boom, and/or glass failure. The worst-case scenario that could occur in the large unit would be in V-103, the first three-phase separator. The damage would be the breakage of small windows under strain. Table 11 below shows the information needed to calculate the overpressure to determine the damage that could potentially happen.

	Small	Medium	Large
	R-102	R-102	V-103
Mass of vapor released (lb)	0.10	0.18	0.33
Heat of combustion (Btu/lb)	10315.89	9276.43	10066.33
Mass of TNT (lb)	0.01	0.02	0.04
Distance (ft)	32.81	32.81	32.81
Overpressure (kPa)	0.24	0.38	0.81

#### c. Potential Health Impacts

Based on the entire safety analysis, it was also important to assess and mitigate potential health impacts. High temperatures are used throughout the process, so hot steam or surfaces could lead to burns. If a fire or explosion occurs, severe burns or death could be a result. The cobalt catalyst that is used for the FTR has the potential to be toxic to workers who replace it so it will be important to mitigate the risk involved in switching out the catalyst [13]. A summary of potential health risks and how they are mitigated can be seen below in Table 12.

Risk	Health Impact	Mitigation
Steam	Severe burns	Proper training and storage
Runaway reactions	Chemical burn; death	Strong vessels and relief systems
Flammable chemicals	Severe burns	Away from ignition sources
Toxic chemicals	Poisoning; death	Wear proper PPE; proper training; shower & eyewash stations
Cobalt catalyst	Harmful to eyes, skin, heart, and lungs	Wear proper PPE; store safely
Hot surfaces	Skin burn	Wear proper PPE; proper training

#### Table 12: Health Risks and Mitigation

## d. Safety Summary

The main risks of this GTL process are the flammability of the lighter hydrocarbons, the mixture of  $O_2$  in the Syngas Unit, the high temperatures and pressures, the potential for loss of containment, vapor cloud deflagration of the units, and lack of understanding of the process. The best way to prevent safety risks is to design the system with inherent safety practices and have proper training on how to operate the equipment. An additional safety feature was making the design pressures of each piece of equipment 50 psi more than the operating pressure [8]. The safety features that were put in place significantly minimize the environmental and human health risks.

## e. Environmental Summary

The main goal for this project was minimizing the amount of greenhouse gases produced during oil extraction. This was successfully accomplished by converting methane into naphtha and diesel. Table 13 illustrates the methane that has been converted due to the Fischer-Tropsch synthesis. The small unit converts 58.8% of the methane, medium unit 70.2%, and the large unit by 61.3%. The remaining methane in the tail gas will be flared resulting in a much smaller environmental impact than conventional methods of flaring the natural gas initially.

	Small	Medium	Large
Starting Methane (lb/hr)	864.3	4,321.7	8,643.4
Methane in Tail Gas (lb/hr)	357.8	1,289.2	3,347.8
Converted Methane (lb/hr)	508	3,032.5	5,295.6

Table 13: Converted Methane	Table	13:	Converted	Methane
-----------------------------	-------	-----	-----------	---------

As the tail gas will be flared later in the process, Table 14 lists the mass flows and quantity of gas that will be flared. This information is required in order to obtain operating

permits from the regulatory authorities. The application for Best Available Control Technology (BACT) will be utilized to treat waste prior to discharging into the environment [14].

	Mass Flow (lb/hr)	Mass Flow (lb/hr)	Mass Flow (lb/hr)
	Small Unit	Medium Unit	Large Unit
Water	44.0	187.3	429.1
Nitrogen	2.9	17.4	30.6
Carbon Monoxide	84.6	543.3	933.9
Carbon Dioxide	28.6	124.6	277.1
Hydrogen	4.3	28.6	48.7
Methane	357.8	1289.2	3347.8
Ethane	4.1	24.1	42.4
Propane	5.9	34.6	61.3
Butane	7.6	45.6	78.4
Pentane	50.3	281.1	517.2
Hexane	42.5	226.8	433.9
Heptane	33.0	165.1	333.2
Octane	22.9	105.4	227.5
Nonane	14.0	58.9	136.3
Decane	7.5	29.1	71.9
C11	3.5	12.4	32.6
C12	1.4	4.7	13.0
C13	0.5	1.7	5.0
C14	0.2	0.5	1.5
C15		0.1	0.4
C16	—	_	0.1
Total Mass Flow (lb/hr)	715.8	3178.7	7022.0

Table 14: Tail Gas Composition and Quantity

The Fischer-Tropsch synthesis results in by-product of water that is separated out of the main product streams through multiple three-phase separators. The resulting wastewater is combined into a singular stream and sent to the wastewater treatment plant. Table 15 shows the wastewater mass flow rates of each unit.

Table 15:	Wastewater	Mass	Flow	Rates
-----------	------------	------	------	-------

	Small	Medium	Large		
Wastewater (lb/hr)	1038.4	5729.2	10606.0		

Since the catalyst is replaced almost every three years, it will need to be properly disposed. For this project, the CRI-MET process will be incorporated. Table 16 shows the total amount of catalyst that will need to be recycled over the course of the project life [15].

## Table 16: Total Amount of Catalyst

	Small	Medium	Large
Catalyst (lb)	125.3	144.4	97.4

### **Equipment Information Summary**

Below are the Equipment Summary Tables for the small, medium, and large units:

<u>R-102</u> **Reactor** Fixed Bed Туре Tubular Reactor Pressure (psig) 430.30 450.00 Temperature (°F) Orientation Vertical Number of Tubes 14.00

Table 17: Small Unit FTR and Separation Equipment List

Number of Tubes	14.00			
Diameter of Tubes	0.17			
(ft)				
Diameter (ft)	1.00			
Height (ft)	2.50			
Volume (ft <sup>3</sup> )	1.96			
MOC	SS			
Vessel	<u>V-102</u>	<u>V-103</u>	<u>V-104</u>	
Туре	Two-Phase	Three-Phase	Three-Phase	
Pressure (psig)	416.30	402.00	402.00	
Temperature (°F)	275.00	250.00	420.00	
Orientation	Horizontal	Horizontal	Horizontal	
Diameter (ft)	1.50	0.50	1.50	
Length (ft)	2.00	0.50	1.50	
Volume (ft <sup>3</sup> )	3.53	0.10	2.65	
MOC	SS	SS	SS	
Heat Exchanger	<u>E-103</u>	<u>E-104</u>	<u>E-105</u>	<u>E-106</u>
Туре	<u>E-103</u> Teflon Tube	<u>E-104</u> Teflon Tube	<u>E-105</u> Teflon Tube	<u>E-106</u> Teflon Tube
Туре	Teflon Tube	Teflon Tube	Teflon Tube	Teflon Tube
Type Area (ft <sup>2</sup> )	Teflon Tube 19.48	Teflon Tube 1.54	Teflon Tube 7.28	Teflon Tube 0.81
Type Area (ft <sup>2</sup> ) Duty (Btu/hr)	Teflon Tube 19.48	Teflon Tube 1.54	Teflon Tube 7.28	Teflon Tube 0.81
Type Area (ft <sup>2</sup> ) Duty (Btu/hr) <b>Shell</b>	Teflon         Tube           19.48         659,039.02	Teflon Tube 1.54 34,891.13	Teflon Tube 7.28 122,369.43	Teflon         Tube           0.81         20,980.00
Type Area (ft <sup>2</sup> ) Duty (Btu/hr) Shell Temp. In (°F)	Teflon Tube           19.48           659,039.02           450.00	Teflon Tube 1.54 34,891.13 275.00	Teflon Tube 7.28 122,369.43 275.00	Teflon Tube           0.81           20,980.00           420.00
Type Area (ft <sup>2</sup> ) Duty (Btu/hr) Shell Temp. In (°F) Temp. Out (°F)	Teflon Tube           19.48           659,039.02           450.00           275.00	Teflon Tube 1.54 34,891.13 275.00 250.00	Teflon Tube           7.28           122,369.43           275.00           420.00	Teflon Tube           0.81           20,980.00           420.00           250.00
Type Area (ft <sup>2</sup> ) Duty (Btu/hr) Shell Temp. In (°F) Temp. Out (°F) Pressure (psig)	Teflon Tube           19.48           659,039.02           450.00           275.00           420.30	Teflon Tube 1.54 34,891.13 275.00 250.00 406.30	Teflon Tube 7.28 122,369.43 275.00 420.00 406.30	Teflon Tube           0.81           20,980.00           420.00           250.00           392.30
Type Area (ft <sup>2</sup> ) Duty (Btu/hr) Shell Temp. In (°F) Temp. Out (°F) Pressure (psig) Phase	Teflon Tube           19.48           659,039.02           450.00           275.00           420.30           Vapor	Teflon Tube 1.54 34,891.13 275.00 250.00 406.30 Vapor	Teflon Tube 7.28 122,369.43 275.00 420.00 406.30 Liquid	Teflon Tube           0.81           20,980.00           420.00           250.00           392.30           Vapor
Type Area (ft <sup>2</sup> ) Duty (Btu/hr) Shell Temp. In (°F) Temp. Out (°F) Pressure (psig) Phase MOC	Teflon Tube           19.48           659,039.02           450.00           275.00           420.30           Vapor	Teflon Tube 1.54 34,891.13 275.00 250.00 406.30 Vapor	Teflon Tube 7.28 122,369.43 275.00 420.00 406.30 Liquid	Teflon Tube           0.81           20,980.00           420.00           250.00           392.30           Vapor

Pressure (psig)	35.30	35.30	600.00	35.30
Phase	Liquid	Liquid	Vapor	Liquid
MOC	SS	SS	SS	SS

## Table 18: Medium Unit FTR and Separation Equipment List

Reactors	R-102			
	Fixed Bed			
Туре	Tubular Reactor			
Pressure (psig)	430.30			
Temperature (°F)	450.00			
Orientation	Vertical			
Number of Tubes	29.00			
Diameter of Tubes	0.17			
(ft)				
Diameter (ft)	1.25			
Height (ft)	3.50			
Volume (ft <sup>3</sup> )	4.29			
MOC	SS			
Vessel	<u>V-102</u>	<u>V-103</u>	<u>V-104</u>	
Туре	Two-Phase	Three-Phase	Three-Phase	
Pressure (psig)	416.30	402.00	402.00	
Temperature (°F)	275.00	250.00	420.00	
Orientation	Horizontal	Horizontal	Horizontal	
Diameter (ft)	2.00	1.00	2.00	
Length (ft)	4.50	1.00	2.00	
Volume (ft <sup>3</sup> )	14.14	0.79	6.28	
MOC	SS	SS	SS	
Heat Exchanger	<u>E-103</u>	<u>E-104</u>	<u>E-105</u>	<u>E-106</u>
Туре	Teflon Tube	Teflon Tube	Teflon Tube	Teflon Tube
Area (ft <sup>2</sup> )	114.68	6.65	47.34	6.87
Duty (Btu/hr)	3,879,048.97	150,701.92	795,251.66	179,000.00
Shell				
Temp. In (°F)	450.00	275.00	275.00	420.00
Temp. Out (°F)	275.00	250.00	420.00	250.00
Pressure (psig)	420.30	406.30	406.30	392.30
Phase	Vapor	Vapor	Liquid	Vapor
MOC	SS	SS	SS	SS
Tube	. <u></u>			
Temp. In (°F)	90.00	90.00	490.00	90.00
Temp. Out (°F)	120.00	120.00	490.00	120.00
Pressure (psig)	35.30	35.30	600.00	35.30
Phase	Liquid	Liquid	Vapor	Liquid
MOC	SS	SS	SS	SS

<b>Reactors</b>	<u>R-102</u>			
Tuno	Fixed Bed			
Туре	<b>Tubular Reactor</b>			
Pressure (psig)	430.30			
Temperature (°F) 450.0				
Orientation	Vertical			
Number of Tubes	49.00			
Diameter of Tubes	0.17			
(ft)				
Diameter (ft)	1.50			
Height (ft)	4.25			
Volume (ft <sup>3</sup> )	7.51			
MOC	SS			
Vessel	<u>V-102</u>	<u>V-103</u>	<u>V-104</u>	
Туре	Two-Phase	Three-Phase	Three-Phase	
Pressure (psig)	416.30	402.00	402.00	
Temperature (°F)	275.00	250.00	420.00	
Orientation	Horizontal	Horizontal	Horizontal	
Diameter (ft)	2.00	1.00	2.00	
Length (ft)	7.00	4.00	3.00	
Volume (ft <sup>3</sup> )	21.99	3.14	9.42	
MOC	SS	SS	SS	
Heat Exchangers	<u>E-103</u>	<u>E-104</u>	<u>E-105</u>	<u>E-106</u>
Туре	U-Tube	Teflon Tube	Teflon Tube	Teflon Tube
Area (ft <sup>2</sup> )	181.73	13.52	85.47	8.93
Duty (Btu/hr)	6,829,817.44	340,712.46	1,292,192.59	232,700.00
Shell				
Temp. In (°F)	450.00	275.00	275.00	420.00
Temp. Out (°F)	275.00	250.00	420.00	250.00
Pressure (psig)	420.30	406.30	406.30	392.30
Phase	Vapor	Vapor	Liquid	Vapor
MOC	SS	SS	SS	SS
Tube				
Temp. In (°F)	90.00	90.00	490.00	90.00
Temp. Out (°F)	120.00	120.00	490.00	120.00
Pressure (psig)	35.30	35.30	600.00	35.30
Phase	Liquid	Liquid	Vapor	Liquid
MOC	SS	SS	SS	SS

Table 19: Large Unit FTR and Separation Equipment List

## **Unit Control and Instrumentation Description**

Even though the FTR was not included in the P&ID, it will require different controls and pressure relief instruments. For pressure relief, a rupture disk in series with a pressure relief

valve was designed to prevent overpressure. The rupture disk would be in place to protect against toxicity or corrosion and the spring-loaded relief valve would close and minimize losses [12]. To control the pressure of the reactor, a pressure control loop will be needed. A pressure transmitter would be directly attached to the reactor and send a signal to the feed stream to manipulate the flowrate of that stream. A temperature control loop would be needed for the FTR product stream due to the highly exothermic nature of the reaction. The flowrate of the medium-pressure steam entering the reactor would be manipulated to ensure the temperature does not go above 450°F. A simplified P&ID of what the FTR should look like can be seen in Figure 5 below.

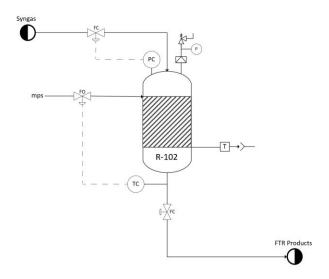


Figure 5: Simplified P&ID of FTR

For the separation unit, there were three kinds of control loops added: temperature, pressure, and level. The separation unit P&ID can be seen above in Figure 4. The temperature controls were necessary for the heat exchangers that either cooled or heated up process streams. The pressure control loops were needed for the outlet vapor streams from the separators. Level control loops were put in place to adjust the heavy and light liquid levels in the separation vessels. Table 20 below shows a brief description of what each control loop's controlled and manipulated variables are.

Loop Number	Controlled Variable	Manipulated Variable
100	E-103 outlet stream temperature	Cooling water flowrate into E-103
101	V-102 pressure	Outlet vapor flowrate from V-102
102	V-102 liquid level	Outlet liquid flowrate from V-102
103	E-104 outlet stream temperature	Cooling water flowrate into E-104
104	V-103 heavy liquid level	Outlet heavy liquid flowrate from V-103
105	V-103 pressure	Outlet vapor flowrate from V-103
106	V-103 light liquid level	Outlet light liquid flowrate of V-103
107	E-105 outlet stream temperature	High-pressure steam flowrate into E-105
108	V-104 heavy liquid level	Outlet heavy liquid level from V-104
109	V-104 pressure	Outlet vapor flowrate from V-104
110	V-104 heavy liquid level	Outlet heavy liquid flowrate from V-104
111	E-106 outlet stream temperature	Cooling water flowrate into E-106

Table 20: P&ID Controlled and Manipulated Variables

## Expected Plot Layout for the Wellhead Site System

The wellheads are located at various distances from the central plant with varying production rates of natural gas. A parallel modular manufacturing approach was taken to maximize the process without bottlenecking the wellhead. The small, medium, and large units were utilized throughout the field. The following cartesian coordinate plot, Figure 6, allowed access to visually observe the distances between each wellhead from the central plant and from each other.

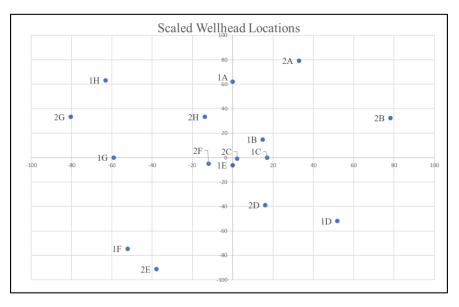


Figure 6: Plot Layout of the Wellhead Site System

#### Wellsite Plan and Modular Unit Deployment

Since each well will have a sharp decline of production of natural gas after the first two years by 35%, the units will be moved between wellheads to ensure maximum efficiency of the units. Figure 7 illustrates how each wellhead will decrease in production when put into operation. The central plant can also only process 30,000 MSCF of natural gas per day feed equivalent from the wells regardless of the products formed. This emphasizes the efficiency of exchanging units and turning on the wellheads at various times throughout the project life.

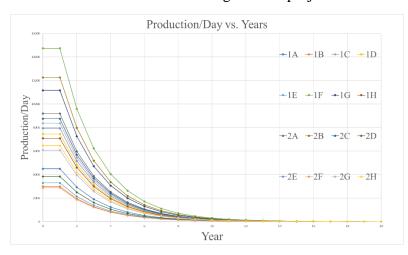


Figure 7: Declining Production Rate of Wellheads

The modular network optimization was approached on three premises. Firstly, each unit will be transported from the central plant to the desired wellhead. Secondly, the well sites will be turned on at different years throughout the project life to maximize the product that the central plant can process. Thirdly, when a unit is no longer being utilized it will be transported back to the central plant to be decommissioned.

Based on the three premises stated above, the modular network optimization began by starting the process at the well sites farthest from the central plant transferring units to the well sites closest to the central plant to eliminate the cost of decommission transportation costs. It also consisted of moving the units to close by well sites instead of transporting new units to a well site to only transport back at the end of its life. All these conditions were optimized while ensuring the largest possible gas production rate of 30,000 MSF of natural gas per day and striving to maximize gas production rate at the beginning of the project life. This was done to obtain the greatest revenue in the first five years since time value of money will increase. Table 21 depicts the gas production rate of all the well sites at a given year.

Year	2024	2025	2026	2027	2028	2029	2030
Gas Production	0	27,199	27,199	29,937	29,914	29,966	29,478
Rate (MSCF) Year	2031	2032	2033	2034	2035	2036	2037
Gas Production Rate (MSCF)		29,275	25,929	29,159	25,754	17,562	26,061
Year	2038	2039	2040	2041	2042	2043	2044
Gas Production Rate (MSCF)	22,029	22,258	17,177	10,996	7,061	4,510	2,760

Table 21: Gas production rate per year of deployment plan

The cost of redeploying a module will be \$3.00 per MCF capacity per mile transported and will require a month of downtime. This results in a loss of production during redeployment. Also, the catalyst in the FTR will require to be changed every three years and result in a downtime of one month. To limit the loss of production from these downtimes, the catalyst will sometimes be exchanged a year before requirement and changed the same month as the redeployment. The downtime will also be scheduled to occur in January of every year.

The number of units needed at a given period was determined based on the capacity of the wells. Since wells with the highest production rate declined the fastest, large units were used sparingly and replaced with medium units when needed to lengthen their capabilities. For this project, 13 small units, 9 medium units and 5 large units will be used. Table 22 shows the sizes and number of units needed for each well site. At the bottom of Table 22, the number of units need to be deployed in the field are listed and will not begin to be manufactured until one year before required. The numbers marked red indicate the units that are at the well site but not operating at the time.

## Table 22: Unit Distribution between Wellheads

	2024	2025	2026	2027	2028	2029	2030	2031	2032	2033	2034	2035	2036	2037	2038	2039	2040	2041	2042	2043	2044
Well ID	Gas Production	Gas Production	Gas Production	Gas Production	Gas Production	Gas Production	Gas Production	Gas Production	Gas Production	Gas Production	Gas Production	Gas Production	Gas Production	Gas Production	Gas Production	Gas Production	Gas Production	Gas Production	Gas Production	Gas Production	Gas Production
	MSCF/day 4494	MSCF/day	MSCF/day	MSCF/day	MSCF/day	MSCF/day	MSCF/day	MSCF/day	MSCF/day	MSCF/day	MSCF/day	MSCF/day	MSCF/day	MSCF/day	MSCF/day	MSCF/day	MSCF/day	MSCF/day	MSCF/day	MSCF/day	MSCF/day
1A mall	4494	4494	2921 1		1234 L 3		521 <b>1,2</b>			143											
/ledium			1	1	L 1	,	_,_														(
arge	1	1	1	1	L 1	0															
1B mall											2989	2989	1943	1263	821		347				
/ledium											1	1	1	1	1		1	L 1	1,1	1	
arge																					(
1C mall					8365	8356	5431	3530		5 1492		630	410 <b>1,1</b>		173		-		1		
nan Aedium					2	2	2					2	1,1	1			<b>1</b>	L 1	1		
arge					1	1	1	0													(
1D	7447	7447	4841	3146	5 2045	1329	864														
mall Aedium	1	1	1	1	L 1	3	2				1,1	2	2	2	C						
arge	1	1					0														
1E				3290	3290	2139	1390														
mall Aedium				-	2 2	1,1	1,1	2		2 <u>1</u> ,1		1	1	1	1	1	1	u 1	1	1	1
arge				4		1,1	1,1	- <sup>2</sup>	· · · ·	1											
1F													14737	14737	9579	6226	4047	7 2631	1710	1111	72
mall Aodium														2	-		-	-			
Aedium arge													2				2		1,1	1,1	
1G										11155	5 11155	7251				-	841	L 547		231	150
mall						]											2	2 2	1,1	1,1	<b>1</b> ,1
Aedium arge										1		3	2		1	1	1	u 1	1	1	
1H						7081	7081	4603	2993	-	-	822			226	5 147					
mall									:	L 1		2	2	<mark>1</mark> ,1			. 0	)			
Nedium						1	1					1	1	1	. 1	1	1	L 1	1	1	1
2A	9182	9182	5968	3879	2522	1639	1065														- · · ·
mall			2				2														(
/ledium								-											-		
2B	2	2	1 12258		L 1 3 7968		1 3366	-	142	2 924	1 601	1 391	<b>1</b> 254	1 165	1	1	1	u 1	1	1	1
mall					1		1			1		1,1			. 0	)					(
/ledium			1	1	L 1		1	1		L (		]									
arge 2C			2	2	2 1	1	1	0 8742		2 5682	3693	2401	1561	1014	659	429	279	9 181	118		•
mall								0/42		5082		2401	1501	1014	2						2
/ledium										1		1	1		. 1	-	. 1	l 1	1	1	
2D								2	384	_/-			2 1055	-			2 188		2	2	2
mall									384	3840	2496	1022	1022	2			188			1	1
/ledium									:	2 2	2 <b>1</b> ,1	1,1	<mark>1</mark> ,1		2		2		2	2	2
arge																					(
2E mall															7939		5160 1	) 3354 L 1	2180	1417	92:
/ledium															1	-	1	<u>ا</u>	1	1	
arge															1	1	1	1	1	1	1
2F mall				2874	1 2874 L 1		1214 1			333 2 1,1				1	1	1	1	1	1	1	1
/ledium				1	L 1		1				, <b>1</b> , 1	1,1									
arge							_														(
2G mall	6076	6076	3949	2567			705			3 L <b>(</b>	)					-					-
/ledium	3	3	<mark>1</mark> ,2				0		· · · ·	1	-				1	1		1			
arge	-		,	,																	(
2H mall							6468	6468	4204										-		
mall Nedium							1	1	:	1 L 1		1			. <u>1,1</u>	<b>1,1</b>	1,1	2	2	2	
arge							1			1		1	0								
	27207	27207	29948	29930	29988	29498	28126	29298	2595	29175	5 25770	17580	25975	22052	22174	17196	11017	7 7082	4527	2781	1815
ummation																					
ifference	2793	2793	52		) 12 Yr 5	502 173778	1874	702	405	9 825 Yr 10	4230 312096		4025		7826 Yr 15	5 12804 417073	18983	3 22918		27219 Yr 20	28185
					11.5	1/3//8				11 10	312096				11 15	41/0/3				11 20	444294
mall	0	0	÷	5	5 8	11	12					13	13	13	13		13		13	13	
/ledium			6		n 9	. 9	9	ı 9	1 1	71 5	'I 5	9	y	9	9 9	n 9	9	יו 9	. 9	1 9	7 S

After determining which well sites would be operating at a given period, the redeployment of the units was determined by proximity. Since the redeployment of the wells was used in the optimization of the initial plan, it allowed easy transition to determining where to send the units no longer needed at a certain well. Each unit was broken down by number to assign how the unit would be transferred from wells.

Table 23 and Table 24 illustrate the units' deployment for each well site broken down by the first 10 years and the last 10 years. The tables also analyze the redeployment downtimes and catalyst downtimes to minimize the total amount of downtime required for the entire project. The blue boxes depict when the units will be required to shut down for redeployment. The red boxes represent the years needed to replace the catalyst in the FTR. The purple boxes represent the years that the catalyst will be replaced during redeployment. The grey boxes represented the period of time that the units will not be in service. Minimizing the shutdown periods is critical to maximize the revenue. The wellhead will not need to be shut down for a month if a unit is being disconnected from the process or if the unit was previously connected and is still on site.

Small units										
	2025	2026	2027	2028	2029	2030	2031	2032	2033	2034
1	-	-	1A	1A	1A	1A	1A	1E	1E	1E
2	-	-	2A	2B						
3	-	-	2A							
4	-	-	-	2F						
5	-	-	-	2G	2G	2G	2G	2G	1H	1H
6	-	-	-	-	2B	2B	2B	2B	2B	2B
7	-	-	-	-	1A	1A	1A	1A	1A	1A
8	-	-	-	-	1A	1A	1A	2F	2F	2F
9	-	-	-	-	-	1D	1D	1D	1D	1D
10	-	-	-	-	-	1D	1D	1D	1D	1D
11	-	-	-	-	-	1D	1C	1C	1C	1C
12	-	-	-	-	-	-	2G	2G	2G	2H
13	-	-	-	-	-	-	-	1E	1E	1E
Medium units										
	2025	2026	2027	2028	2029	2030	2031	2032	2033	2034
1	1D	2D	2D							
2	2G	2G	2G	2G	2G	2G	2H			2H
3	2G	2G	2G	2G	1C	1C	1C	1C	2D	2D
4	2G	2G	2G	2G	1C	1C	1C	1C	1C	1C
5	-	-	1A	1A	1A	1H	1H	1H	1H	
6	-	-	2B	2C						
7	-	-	-	1E	1E	1E	1E	1E	1E	
8	-	-	-	1E	1E	1E	1E	1E	1E	1G
9	-	-	-	2F	2F	2F	2F	2F	2F	1G
Large units										
	2025	2026	2027	2028	2029	2030	2031	2032		
1	1A	1A	1A	1A	1A	1H	1H	1H		
2	1D	1D	1D	1D	1D	1D	2H			2H
3	2A	2A	2B	2B	1C	1C	1C	2C		2C
4	2A		2A							
5	-	-	2B	2B	2B	2B	2B	2C	2C	2C

Small units										
	2036	2037	2038	2039	2040	2041	2042	2043	2044	2045
1	1E									
2	2B	2B	2D	2C						
3	1H	1H	1H	1H	1H	1G		1G	1G	1G
4	2F	2H								
5	1H	1H	1H	1H	1H	1G			1G	1G
6	2B	2B	2B	1B						
7	1B									
8	2F									
9	1D	1D	1D	2E						
10		1D	1D	2C						
11		1C	2D							
12	2H									
13	1C									
Medium units										
	2036	2037	2038	2039	2040	2041	2042	2043	2044	2045
1	2D									
2	2H	1F								
3	2D									
4	1B									
5	1H									
6	2C									
7	1G	1F								
8	1G									
9	1G	1G	1G	2E						
Large units										
	2036	2037	2038	2039	2040	2041	2042	2043	2044	2045
1	1G	1F								
2	2H	1F	1F	2E						
3										
4										2A
	2A	ZA								
3	2C									

Table 24: Last 10 Years of Redeployment and Shutdown Periods

As mentioned before, the cost of redeployment will cost \$3 per MSCF per mile traveled. With transferring units to close by well sites, this minimizes the cost. The initial costs are accounted for by the initial deployment of the equipment into the field from the central plant. The distance between wells was calculated by the distance in miles of each well from the central plant and the angle between each well. Each ring of wells has a 45° angle between the well sites and to relate the two ring of wells it was determined that the inner ring would be half of the angle of the outer ring. For example, well 2A and 2B have a 45° angle from the central plant, while well 1B has a 22.5° angle from 2A and 2B in respect to the central plant. The total distance between each wellsite was calculated using the law of cosines. These distances were then used to calculate the cost of transportation of redeployment. Table 25 also includes the transportation cost of transferring the units back to the central plant for decommissioning before being removed.

								Cost o	of Transportatio	n of Units Betw	een Wells										
per MSCF per mile																					
\$3																					
Small units	0	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	)
500							0				10			10		10	10	1,	10		
1	_	-	\$93,000		_			\$102,300	_		-	-	\$9,300	_		_					-
2	-	_	\$128,700	_	-	_		\$102,500	_	\$130,891.26	_	_		\$141,812.21	\$60,507.99	\$3,600	_	_	_		
2	-	-		-	-	-	-	-	-	\$150,891.20		\$145,985.80		\$141,012.21	\$00,307.99	\$5,000	- \$94.920.77	-	\$88,500	-	
3	-	-	\$128,700			-	-	-	-	-		\$145,985.80	-	-	-	-	1			-	-
4	-	-	-	\$19,500	-	-	-	-	-	-	-	-	\$57,412.98	-	-	-	-	\$54,000	-	-	-
5	-	-	-	\$130,500	-	-	-	-	\$51,702.48	-	-	-	-	-	-	-	\$94,920.77	-	-	-	\$88
6	-	-	-	-	\$127,050	-	-	-	-	-	-	-	-	-	\$98,686.79	-	-	-	-	-	\$31
7	-	-	-	-	\$93,000	-	-	-	-	-	\$74,150.60	-	-	-	-	-	\$31,500	-	-	-	-
8	-	-	-	-	\$93,000	-	-	\$102,064.90	-	-	-	-	\$19,500	-	-	-	-	-	-	-	-
9	-	-	-	-	-	\$110,100	-	-	-	-	-	-	-	-	\$146,958.83	-	-	\$148,200	-	-	-
10	-	-	-	-	-	\$110,100	-	-	-	-	-	-	-	-	\$106,782.92	-	-	-	-	-	\$3,60
11	-	-	-	-	-	\$110,100	\$93,817.83	-	-	-	-	-	-	\$58,221.03	-	-	-	-	\$63,000	-	-
12	-	-	-	-	-	-	\$130,500	-	-	\$99,901.39	-	-	-	_	-	-	-	\$54,000	-	-	-
13	-	-	-	-	-	-	-	\$9,300	-	-	\$27,142.95	-	-	-	-	-	-	-	-	-	\$25
Medium units	0	1	2	2		5	6	7	0	0	10	11	12	13	5 14	15	16	17	18	19	<u> </u>
2500	0		. 4	. 3	-		0	1	0		10	11	12	15	14	15	10	17	10	17	
2500	\$550,500								¢294 752 15		¢215.000										
1		-	-	-	-	-	-	-	\$284,752.15	-	\$315,000	-	-	-	-	-	-	-	-	-	-
2	\$652,500	-	-	-	-	-	\$499,506.93	-	-	-	-	-	\$824,466.03	-	-	-	-	-	-	-	\$682
3	\$652,500	-	-	-	\$771,838.39	-	-	-	\$291,105.14	-	-	-	-	\$315,000	-	-	-	-	-	-	-
4	\$652,500	-	-	-	\$771,838.39	-	-	-	-	-	\$112,531.44	-	-	-	\$157,500	-	-	-	-	-	-
5	-	-	\$465,000	-	-	\$473,663	-	-	-	-	-	\$669,750	-	-		-	-	-	-	-	-
6	-	-	\$635,250	-	-	-	-	-	-	\$622,652.18	-	-	-	-	\$18,000	-	-	-	-	-	-
7	-	-	-	\$46,500	-	-	-	-	-	\$444,936.51	-	-	-	-	-	-	-	-	\$442,500	-	-
8	-	-	-	\$46,500	-	-	-	-	-	\$444,936.51	-	-	-	-	-	-	\$442,500	-	-	-	-
9	-	-	-	\$97,500	-	-	-	-	-	\$444,936.51	-	-	-	-	\$702,800.96	-	-	-	-	-	\$741
Large	0	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	)
5000	5		2	5		5	0	1		,	10	11	12	15	14	15	10	17	10		
1	\$930,000	-	_	_	_	\$947,325.14	_	-					\$1,912,455.29	_	\$1,365,000	_		_	_	_	-
1	\$930,000		-	-	-		\$1,610,210.33		-	-	-	-	\$1,648,932.05	-	\$1,565,000		-	-	\$1,482,000	-	-
2	. , ,	-	- ¢070.021.50	-	-	-		- #222.177.80	-	-	-	-				-	-	-	. , ,	-	
3	\$1,287,000	-	\$978,831.59		\$1,039,501.27	-	-	\$222,167.89	-	\$36,000	-	-	-	-	-	-	-	-	-	-	-
4	\$1,287,000	-	-	-	-	-	-	\$1,287,000	-	-	-	-	-	-	-	-	-	-	-	-	-
5	-	-	\$1,270,500	-	-	-	-	\$1,245,304.36	-	-	-	-	-	\$36,000	-	-	-	-	-	-	-
Total																					
Transportation	\$7,113,000	\$0	\$3,699,982	\$340,500	\$2,896,228	\$1,751,288	\$2,334,035	\$2,968,137	\$627,560	\$2,224,254	\$528,825	\$815,736	\$4,472,066	\$551,033	\$3,223,390	\$3,600	\$663,842	\$256 200	\$2,076,000	\$0	\$1,572
Cost	$\psi_{1,113,000}$																				

# Table 25: Transportation Cost of Redeployment

The products from the units will be collected from each well site and transported to the central plant for further processing. Each truck has a maximum capacity of 5,000 gallons of liquid. LPG, naphtha, and diesel will require separate trucks for each trip with cleaning in between routes to eliminate cross contamination. Table 26 shows the number of trips that will be required for the trucks to take per year for each product. This was determined by calculating the gallons produced in a year given the mass flow rate and density and then dividing them by the maximum capacity of the trucks. Each product will be collected at least once from a given well per year to empty the storage tank and collect a revenue for that product in that year regardless of the quantity in the tank. Some trucks will be carrying a smaller quantity of product for each route, while others are carrying the maximum capacity.

Year	2024	2025	2026	2027	2028	2029	2030
LPG	0	14	14	15	18	19	18
Naphtha	0	1238	1238	1272	1348	1360	1315
Diesel	0	1567	1567	1715	1750	1774	1706
Year	2031	2032	2033	2034	2035	2036	2037
LPG	19	22	21	22	21	16	19
Naphtha	1250	1331	1195	1313	1202	842	1169
Diesel	1646	1712	1528	1734	1508	1065	1543
Year	2038	2039	2040	2041	2042	2043	2044
LPG	16	19	15	10	8	7	3
Naphtha	1032	1115	802	402	313	226	139
Diesel	1295	1448	1027	506	409	282	174

Table 26: Number of Trips per Year Required per Product

The total number of trucks required was calculated by summing the number of trucks needed for all products in one year. The trucks will be required to make one trip per day considering the service factor of 80%. From this assumption, the total number of trips for each well can be divided by the days that the process is operating. Table 27 shows the total number of trucks that will be making trips throughout the year.

Table 27: Number of Truck	ks Required per Year
---------------------------	----------------------

Year	2024	2025	2026	2027	2028	2029	2030
Trucks Required	0	10	8	9	9	9	9
Year	2031	2032	2033	2034	2035	2036	2037
Trucks Required	8	9	8	9	8	6	8
Year	2038	2039	2040	2041	2042	2043	2044
Trucks Required	7	8	6	3	2	2	1

The vehicles and drivers will be supplied on a 5-year contract. Due to this, the project life was divided into four timeframes of 5 years. In those five years, the maximum number of trucks needed at one time from that period will be the number of trucks contracted for the entire five years. Table 28 illustrates the number of trucks that will be contracted and the period that the contracts will begin and end.

Contract	Trucks
Period	Contracted
2025 - 2029	10
2030 - 2024	9
2035 - 2039	8
2040 - 2044	6

Table 28: Number of Trucks Contracted

The contract with the drivers states that it will cost \$1.25/mile driven for the vehicle and driver for transportation of product to the central plant. Each truck will travel from the central plant to a desired well site to collect the product from the storage tanks on site. After the product is loaded onto the trucks, they will then travel directly back to the central plant for the product to be further processed. The total cost of transportation was calculated using the number of trips that a well site will require and the distance to and from the central plant. Table 29 and Table 30 show the transportation cost of the product from the well sites to the central plant broken down in the first 10 years and the last 10 years.

	Transportation Cost of Product																
Well ID	distance (mile)	2024		2025	2026		2027		2028		2029		2030	2031	2032	2033	2034
1A	62.0	\$		\$ 67,890.00	\$ 67,890.00	\$	49,600.00	\$	34,875.00	\$	17,205.00	\$	11,780.00	\$ 7,750.00	\$ 4,960.00	\$ 3,410.00	\$ 2,325.00
1B	21.0	\$		\$ -	\$-	\$	-	\$	-	\$	-	\$	-	\$ -	\$ -	\$ -	\$ -
1C	17.0	\$		\$ -	\$-	\$	-	\$	-	\$	36,040.00	\$	37,485.00	\$ 25,670.00	\$ 17,637.50	\$ 11,517.50	\$ 7,182.50
1D	73.4	Ş		\$ 142,029.00	\$ 142,029.00	\$	86,428.50	\$	54,316.00	\$	44,223.50	\$	21,836.50	\$ 14,863.50	\$ 9,725.50	\$ 6,422.50	\$ 4,404.00
1E	6.2	\$		\$ -	\$ -	\$	-	\$	5,766.00	\$	6,014.00	\$	3,906.00	\$ 2,449.00	\$ 1,271.00	\$ 852.50	\$ 573.50
1F	91.0	\$	-	\$ -	\$ -	\$	-	\$	-	\$	-	\$	-	\$ 	\$ -	\$ -	\$ -
1G	59.0	\$	-	\$-	\$ -	\$	-	\$	-	\$	-	\$	-	\$ 	\$ 	\$ -	\$ 171,690.00
1H	89.3	\$		\$ -	\$-	\$	-	\$	-	\$	-	\$	157,168.00	\$ 163,195.75	\$ 120,555.00	\$ 73,002.75	\$ 51,347.50
2A	85.8	Ş		\$ 191,548.50	\$ 191,548.50	\$	119,047.50	\$	81,081.00	\$	52,767.00	\$	33,033.00	\$ 22,522.50	\$ 14,157.00	\$ 9,223.50	\$ 6,006.00
2B	84.7	\$		\$ -	\$ -	\$	252,194.25	\$	261,934.75	\$	167,917.75	\$	106,510.25	\$ 69,454.00	\$ 54,631.50	\$ 35,574.00	\$ 17,787.00
2C	2.4	\$	-	\$-	\$ -	\$		\$	-	\$	-	\$	-	\$ 	\$ 4,920.00	\$ 5,100.00	\$ 3,174.00
2D	42.0	\$		\$ -	\$-	\$	-	\$	-	\$	-	\$	-	\$ -	\$ -	\$ 45,675.00	\$ 47,460.00
2E	98.8	\$		\$ -	\$-	\$	-	\$	-	\$	-	\$	-	\$ 	\$ -	\$ -	\$ -
2F	13.0	Ş		\$ -	\$ -	\$	-	\$	10,302.50	\$	10,692.50	\$	7,182.50	\$ 4,485.00	\$ 2,405.00	\$ 1,592.50	\$ 1,072.50
2G	87.0	\$	-	\$ 155,295.00	\$ 155,295.00	\$	101,137.50	\$	63,292.50	\$	42,847.50	\$	27,840.00	\$ 14,137.50	\$ 9,570.00	\$ 6,307.50	\$ -
2H	36.0	\$	-	\$-	\$ -	\$		\$	-	\$	-	\$	-	\$ 58,950.00	\$ 61,290.00	\$ 41,400.00	\$ 27,360.00
Total																	
Transportation																	
Cost		\$		\$ 556,762.50	\$ 556,762.50	\$	608,407.75	\$	511,567.75	\$	377,707.25	\$	406,741.25	\$ 383,477.25	\$ 301,122.50	\$ 240,077.75	\$ 340,382.00

Table 29: Product Transportation Cost for the First 10 Years

					Transportation	Cos	t of Product					
Well ID	distance (mile)	2035	2036	2037	2038		2039	2040	2041	2042	2043	2044
1A	62.0	\$ -	\$ -	\$ -	\$ -	\$	-	\$ -	\$ -	\$ -	\$ -	\$ -
1B	21.0	\$ 14,910.00	\$ 15,487.50	\$ 11,602.50	\$ 7,875.00	\$	3,937.50	\$ 2,677.50	\$ 1,785.00	\$ 1,155.00	\$ 787.50	\$ -
1C	17.0	\$ 3,697.50	\$ 2,550.00	\$ 1,657.50	\$ 1,062.50	\$	765.00	\$ 510.00	\$ -	\$ -	\$ -	\$ -
1D	73.4	\$ 3,119.50	\$ -	\$ -	\$ -	\$	-	\$ -	\$ -	\$ -	\$ -	\$ -
1E	6.2	\$ 372.00	\$ 263.50	\$ -	\$ -	\$	-	\$ -	\$ -	\$ -	\$ -	\$ -
1F	91.0	\$ -	\$ -	\$ 334,652.50	\$ 347,392.50	\$	299,390.00	\$ 159,932.50	\$ 108,290.00	\$ 70,297.50	\$ 44,362.50	\$ 30,030.00
1G	59.0	\$ 178,180.00	\$ 120,802.50	\$ 81,715.00	\$ 53,100.00	\$	33,482.50	\$ 22,567.50	\$ 11,357.50	\$ 7,670.00	\$ 5,015.00	\$ 3,392.50
1H	89.3	\$ 33,487.50	\$ 16,743.75	\$ 11,385.75	\$ 7,590.50	\$	4,911.50	\$ 3,348.75	\$ -	\$ -	\$ -	\$ -
2A	85.8	\$ 4,075.50	\$ -	\$ -	\$ -	\$	-	\$ -	\$ -	\$ -	\$ -	\$ -
2B	84.7	\$ 12,069.75	\$ 8,046.50	\$ 5,293.75	\$ 3,599.75	\$	-	\$ -	\$ -	\$ -	\$ -	\$ -
2C	2.4	\$ 2,160.00	\$ 1,698.00	\$ 1,062.00	\$ 720.00	\$	360.00	\$ 246.00	\$ 156.00	\$ 114.00	\$ 78.00	\$ -
2D	42.0	\$ 30,870.00	\$ 19,320.00	\$ 13,125.00	\$ 6,615.00	\$	4,515.00	\$ 3,045.00	\$ 1,890.00	\$ 1,365.00	\$ -	\$ -
2E	98.8	\$ -	\$ -	\$ -	\$ -	\$	195,130.00	\$ 202,540.00	\$ 67,184.00	\$ 77,805.00	\$ 63,726.00	\$ 39,767.00
2F	13.0	\$ 682.50	\$ 487.50	\$ -	\$ -	\$	-	\$ -	\$ -	\$ -	\$ -	\$ -
2G	87.0	\$ -	\$ -	\$ -	\$ -	\$	-	\$ -	\$ -	\$ -	\$ -	\$ -
2H	36.0	\$ 18,900.00	\$ 12,330.00	\$ 6,120.00	\$ 4,230.00	\$	2,790.00	\$ 1,800.00	\$ 1,350.00	\$ -	\$ -	\$ -
Total												
Transportation												
Cost		\$ 302,524.25	\$ 197,729.25	\$ 466,614.00	\$ 432,185.25	\$	545,281.50	\$ 396,667.25	\$ 192,012.50	\$ 158,406.50	\$ 113,969.00	\$ 73,189.50

Table 30: Product Transportation Cost for the Last 10 Years

Once the units are no longer needed, they will be transported back to the central plant. The equipment will be retired at the central plant to be decommissioned at the end of the project life. At this time, the equipment will be salvaged with and removal cost of \$25,000 per unit.

# **Economics**

An important objective of the design was to maximize the NPV. Maximizing the NPV was done by optimizing the product revenue while keeping capital cost, utility cost, and other manufacturing cost at a minimum. There is a tradeoff between the cost of equipment and manufacturing cost, as well as the revenue made from the product. The design reflects optimization and balance of these variables. It was estimated that the modular units would be built January of 2024 and deployed December of 2024 to begin operating in January of 2025. Estimating a built date of January 2022 allowed for 6 to 12 months of detailed design as well as 6 months for bidding of equipment and an additional 6 months to obtain required permits before units are built. A positive net present value of \$4.35 B was achieved for this project. The DCFROR was calculated to be 9.1% which is greater than the 8% hurdle rate given for the project. The project NPV and IRR show that the project design is economically attractive [16].

# a. Capital Cost Estimates

The capital cost was estimated for the fractionator and FTR by using the module costing technique. Since the equipment is modular there is expected error within the cost estimate since modular variables such as reactor or heat exchanger area do not fall within the module costing correlations. To limit the modular bare module cost error, the purchase cost of base conditions was multiplied by 4.8 instead of calculating for pressure factor and material of construction factor. The 4.8 multiplier is given in the AIChE design problem statement and accounts for the direct costs, indirect costs, and working capital for the modular equipment. The cost estimate equation constants are from a 2001 price index, therefore a 2001 CEPCI value of 397 and 2022 CEPCI value of 776.9 were applied to account for the inflation between years 2001 and 2022 [17]. The capital costs were then escalated again to year 2024 with an inflation rate of 3% for two years. Guthrie's Method was used to estimate the total modular cost to include 18% of contingency and fees. The Syngas Unit, CO<sub>2</sub> Recovery System, Air Separation Plant, and HIU

capital cost were estimated by equipment cost attributes and given scaling factors. Table 31 shows the bare module cost of the FTR and separator, the capital cost of the Syngas Unit, CO<sub>2</sub> Recovery System, and HIU, and the total capital cost for each small, medium, and large unit.

	Sm	all Unit	Medi	um Unit	Larg	e Unit
	Bare Module Cost	Capital Cost	Bare Module Cost	Capital Cost	Bare Module Cost	Capital Cost
FTR	\$1,140	\$1,350	\$1,480	\$1,740	\$1,830	\$2,160
Separator	\$214,000	\$252,000	\$487,000	\$574,000	\$650,000	\$768,000
Syngas Unit	_	\$1,980,000	_	\$5,890,000	_	\$9,290,000
CO <sub>2</sub> Recovery System	_	\$398,000	_	\$897,000	_	\$1,390,000
HI Unit	_	\$10,700,000	_	\$29,600,000	_	\$39,000,000
Total						
Capital Cost		\$13.3 MM		\$37.0 MM		\$50.5 MM

Table 31: Capital Cost of Units

Another economic factor considered in the design is the cost of modular manufacturing. There is a cost reduction for every unit made after the first unit built. This first-of-a-kind (FOAK) concept is specified in the project statement. The economy of mass production decreases the unit cost by a "p" factor when number of units are doubled as well as 20% learning rate which decreases the cost by 20% when number of units are doubled. A summary of the FOAK cost is given in Table 32, Table 33, and Table 34 for the small, medium, and large units, respectively.

Unit Number	FOAK Cost
Sma	all Unit
1	\$13.3 MM
2	\$10.6 MM
3	\$9.3 MM
4	\$8.5 MM
5	\$7.9 MM
6	\$7.4 MM
7	\$7.1 MM
8	\$6.8 MM
9	\$6.6 MM
10	\$6.3 MM
11	\$6.2 MM
12	\$6.0 MM
13	\$5.8 MM

Unit Number	FOAK Cost
Medi	um Unit
1	\$37.0 MM
2	\$29.6 MM
3	\$26.0 MM
4	\$23.7 MM
5	\$22.0 MM
6	\$20.8 MM
7	\$19.8 MM
8	\$18.9 MM
9	\$18.2 MM

Table 33: Medium Unit FOAK Cost

Table 34: Large Unit FOAK Cost

Unit Number	FOAK Cost								
Large Unit									
1	\$50.5 MM								
2	\$40.4 MM								
3	\$35.5 MM								
4	\$32.3 MM								
5	\$30.0 MM								

# b. Revenue Estimates

Revenue includes the LPG, naphtha, and diesel produced and maximum revenue made for each is given in Table 35. A 11/12 factor was also used to account for the one-month turnaround that occurs when needed. The turnaround period is marked in green.

# Table 35: Product Revenue

1A LPG Naphtha Diesel 1B LPG Naphtha Diesel 1C LPG	distance (mile) 62	2025 \$4,494 \$3,026	2026 \$4,494	2027 \$2,921	2028 \$1,899	2029	2030	2031	2032	2033	2034	2035	2036	2037	2020	2039	2040	2041	2042	2043	2044	204
LPG Naphtha Diesel 1B LPG Naphtha Diesel 1C	62		\$4,494				\$802	\$521	\$339		\$143	2055	2050	2057	2038	2059	2040	2041	2042	2045	2011	204
Naphtha Diesel 1B LPG Naphtha Diesel 1C			\$3,026	\$2,895	\$2,006	\$1,234 \$1,472	\$802	\$315	\$205	\$220 \$133	\$143											
1B LPG Naphtha Diesel 1C		\$5,751	\$5,751	\$4,570	\$3,089	\$745	\$957	\$622	\$404	\$263	\$171											
LPG Naphtha Diesel 1C		\$9,039	\$9,039	\$6,628	\$4,417	\$2,379	\$1,546	\$1,005	\$653	\$425	\$276											
Naphtha Diesel 1C	21											\$2,989	\$2,989	\$1,943	\$1,263	\$821	\$534	\$347	\$225	\$147		
Diesel 1C												\$2,936	\$2,936	\$2,052	\$1,334	\$496	\$322	\$209	\$136	\$88		
1C												\$4,651	\$4,651	\$3,161	\$2,055	\$979 \$956	\$636	\$414 \$404	\$269	\$175		
	17					\$8,365	\$8,356	\$5,431	\$3,530	\$2,295	\$1,492	\$6,400 \$970	\$6,400 \$630	\$4,520 \$410	\$2,938 \$266	\$956	\$621 \$112	\$404	\$262	\$171		
	1/					\$6,921	\$6,912	\$5,542	\$3,729	\$2,424	\$1,576	\$586	\$381	\$247	\$161	\$105	\$68					
Naphtha						\$11,873	\$11,858	\$8,649	\$5,744	\$3,733	\$2,427	\$1,157	\$752	\$489	\$318	\$206	\$134					
Diesel						\$17,885	\$17,864	\$12,464	\$8,214	\$5,339	\$3,470	\$1,869	\$1,215	\$790	\$513	\$334	\$217					
1D	73	\$7,447	\$7,447	\$4,841	\$3,146	\$2,045	\$1,329	\$864	\$562	\$365	\$237	\$154										
LPG		\$5,952	\$5,952	\$3,259	\$2,119	\$2,160	\$803	\$522	\$339	\$220	\$143	\$93										
Naphtha		\$10,379	\$10,379	\$6,194	\$4,026	\$3,327	\$1,586	\$1,031	\$670	\$435	\$283	\$184										
Diesel 1E	6	\$15,750	\$15,750	\$6,556	\$4,261 \$3,290	\$4,758 \$3,290	\$2,563 \$2,139	\$1,666 \$1,390	\$1,083 \$904	\$704 \$587	\$457 \$382	\$297 \$248	\$161									
LPG	0				\$3,476	\$3,476	\$2,259	\$1,350	\$546	\$355	\$231	\$150	\$101									
Naphtha					\$5,353	\$5,353	\$3,479	\$2,261	\$1,078	\$701	\$455	\$150	\$192									-
Diesel					\$7,654	\$7,654	\$4,975	\$3,234	\$1,742	\$1,132	\$736	\$478	\$311							1		
1F	91						÷.,=.3			, _,	+·5	÷		\$14,737	\$14,737	\$9,579	\$6,226	\$4,047	\$2,631	\$1,710	\$1,111	\$7.
LPG														\$11,738	\$11,738	\$8,204	\$6,577	\$4,275	\$2,779	\$1,806	\$1,174	\$7
Naphtha														\$20,503	\$20,503	\$13,848	\$10,130	\$6,584	\$4,280	\$2,782	\$1,808	\$1,1
Diesel	<b>↓</b>													\$31,134	\$31,134	\$20,710	\$14,486	\$9,416	\$6,120	\$3,978	\$2,586	\$1,6
1G	59										\$11,155	\$11,155	\$7,251	\$4,713	\$3,063	\$1,991	\$1,294	\$841	\$547	\$355	\$231	\$1
LPG	<u>├</u> ───										\$9,869	\$9,869	\$7,660	\$4,979	\$3,236	\$2,104	\$1,367	\$508	\$330	\$215	\$140	\$9
Naphtha Diesel	<u>                                      </u>										\$16,412 \$24,376	\$16,412 \$24,376	\$11,797 \$16,869	\$7,668 \$10,965	\$4,984 \$7,127	\$3,240 \$4,633	\$2,106 \$3,011	\$1,004 \$1,622	\$652 \$1,054	\$424 \$685	\$276 \$445	\$1 \$2
1H	89						\$7,081	\$7,081	\$4,603	\$2,992	\$1,945	\$1,264	\$10,805	\$534	\$347	\$226	\$147	\$1,022	Ş1,054	2005	Ş44J	<u></u>
LPG	05						\$5,565	\$5,565	\$4,862	\$2,938	\$2,054	\$1,335	\$496	\$323	\$210	\$136	\$89					
Naphtha							\$9,784	\$9,784	\$7,488	\$4,654	\$3,164	\$2,056	\$980	\$637	\$414	\$269	\$175					
Diesel							\$14,898	\$14,898	\$10,708	\$6,764	\$4,524	\$2,941	\$1,584	\$1,029	\$669	\$435	\$283					
2A	86	\$9,182	\$9,182	\$5,968	\$3,879	\$2,522	\$1,639	\$1,065	\$692	\$450	\$293	\$190										
LPG		\$6,183	\$6,183	\$3,952	\$2,612	\$1,698	\$1,104	\$717	\$418	\$272	\$177	\$115										
Naphtha		\$11,750	\$11,750	\$7,553	\$4,964	\$3,227	\$2,097	\$1,363	\$826	\$537	\$349	\$227										
Diesel	05	\$18,468	\$18,468	\$11,923	\$7,803	\$5,072	\$3,297	\$2,143	\$1,335	\$868	\$564	\$367	6204	625.4	64.65							
2B LPG	85			\$12,258 \$9,119	\$12,258 \$9,119	\$7,968 \$6,290	\$5,179 \$3,367	\$3,366 \$2,267	\$2,188 \$2,312	\$1,422 \$1,502	\$924 \$558	\$601 \$363	\$391 \$236	\$254 \$153	\$165 \$100							-
Naphtha				\$16,470	\$16,470	\$11,023	\$6,398	\$2,901	\$3,761	\$1,502	\$666	\$433	\$281	\$153	\$100							
Diesel				\$25,366	\$25,366	\$16,774	\$10,057	\$5,834	\$8,749	\$5,687	\$1,284	\$834	\$542	\$353	\$229							
2C	2			+==,===	+,	<i>q = = q</i> ,	+==,==:	<i>+=/==</i>	\$8,742	\$8,742	\$5,682	\$3,693	\$2,401	\$1,561	\$1,014	\$659	\$429	\$279	\$181	\$118		
LPG									\$5,886	\$5,886	\$3,779	\$2,487	\$2,536	\$1,648	\$1,072	\$398	\$259	\$168	\$109	\$71		-
Naphtha									\$11,187	\$11,187	\$7,212	\$4,726	\$3,906	\$2,539	\$1,650	\$786	\$511	\$332	\$216	\$140		
Diesel									\$17,583	\$17,583	\$11,372	\$7,429	\$5,585	\$3,631	\$2,360	\$1,271	\$826	\$537	\$349	\$227		
2D	42									\$3,840	\$3,840	\$2,496	\$1,622	\$1,055	\$685	\$446	\$290	\$188	\$122			
LPG										\$4,057	\$4,057	\$2,637	\$1,714	\$1,114	\$414	\$269	\$175	\$114	\$74			
Naphtha Diesel	+									\$6,247 \$8.934	\$6,247 \$8,934	\$4,061 \$5,807	\$2,640 \$3,775	\$1,716 \$2,453	\$818 \$1,321	\$531 \$859	\$345 \$558	\$225 \$363	\$146 \$236			-
2E	99											\$3,007	ر ، , , ۍ ې		41,521	\$7,939	\$7,939	\$5,160	\$3,354	\$2,180	\$1,417	\$93
LPG																\$6,273	\$6,273	\$1,954	\$2,259	\$2,303	\$1,497	\$9
Naphtha																\$10,989	\$10,989	\$3,607	\$4,292	\$3,547	\$2,306	\$1,49
Diesel																\$16,719	\$16,719	\$5,546	\$6,746	\$5,072	\$3,297	\$2,14
2F	13				\$2,874	\$2,874	\$1,868	\$1,214	\$789	\$513	\$333	\$217	\$141									
LPG	ļ				\$2,867	\$2,867	\$1,973	\$1,283	\$477	\$310	\$201	\$131	\$85									
Naphtha	<u>├</u>				\$4,513	\$4,513	\$3,039	\$1,976	\$941	\$612	\$398	\$259	\$168									
Diesel 2G	87	\$6,076	\$6,076	\$3,949	\$6,537 \$2,567	\$6,537 \$1,669	\$4,346 \$1,085	\$2,825 \$705	\$1,521 \$458	\$989 \$298	\$643	\$418	\$272									
LPG	07	\$6,419	\$6,419	\$4,172	\$2,681	\$1,763	\$1,085	\$426	\$277	\$180												
Naphtha		\$9,885	\$9,885	\$6,425	\$4,284	\$2,715	\$1,765	\$841	\$547	\$355												·
Diesel		\$14,136	\$14,136	\$9,188	\$9,255	\$3,882	\$2,523	\$1,359	\$883	\$574												
2H	36							\$6,468	\$6,468	\$4,204	\$2,733	\$1,776	\$1,155	\$750	\$488	\$317	\$206	\$134				
LPG								\$5,313	\$5,313	\$3,788	\$3,989	\$1,876	\$1,220	\$453	\$295		\$124	\$81				
Naphtha								\$9,145	\$9,145	\$6,248	\$6,731	\$2,890	\$1,878	\$895	\$582	\$378	\$246	\$160				
Diesel								\$13,797	\$13,797	\$3,428	\$10,120	\$4,133	\$2,686	\$1,447	\$940	\$611	\$397	\$258				
read wate Day 100																						
roduct Revenue per LPG	er uay	\$21,579	\$21,579	\$23,397	\$24,879	\$26,647	\$23,613	\$23,418	\$24,364	\$22,066	\$26,720	\$22,578	\$17,361	\$22,708	\$18,558	\$18,176	\$15,254	\$7,309	\$5,687	\$4,484	\$2,811	\$1,8
Naphtha	+	\$21,579 \$37,765	\$21,579	\$23,397 \$41,213	\$24,879 \$42,700	\$26,647 \$42,777	\$23,613 \$40,963	\$23,418 \$38,573	\$24,364 \$41,790	\$22,066	\$44,515	\$22,578 \$37,351	\$17,361 \$27,245	\$22,708	\$18,558 \$31,442	\$18,176 \$31,228	\$15,254 \$25,273	\$7,309 \$12,326	\$5,687 \$9,855	\$4,484 \$7,068	\$2,811 \$4,389	\$1,8 \$2,8
Diesel		\$57,392	\$57,392	\$59,662	\$65,294	\$64,942	\$62,070	\$59,225	\$66,269	\$52,426	\$66,757	\$55,349	\$39,238	\$56,321	\$47,232	\$46,527	\$37,119	\$18,146	\$14,768	\$10,133	\$6,328	\$4,1
All Product	t per day	\$116,736	\$116,736	\$124,271	\$132,873	\$134,366	\$126,646	\$121,216	\$132,423	\$111,909	\$137,992	\$115,277	\$83,844	\$116,819	\$97,232	\$95,930	\$77,646	\$37,781	\$30,311	\$21,685	\$13,528	\$8,7
All Product	t per year	\$34,086,983	\$34,086,983	\$34,134,264	\$37,411,251	\$37,401,047	\$35,965,974	\$33,677,487	\$37,639,998	\$31,826,785	\$37,740,287	\$33,190,643	\$23,326,265	\$32,057,221	\$28,305,636	\$26,802,686	\$21,894,971	\$10,913,504	\$8,510,887	\$6,123,642	\$3,756,616	\$2,567,7

# c. Operating Expense Estimates

The utility cost of the heat exchangers, the FTR unit, and the Syngas Unit are shown in Table 36. Fuel gas was needed for the syngas feed preheat furnace and accounted for 80% of the furnace duty. The remanding amount of furnace duty was split between electricity and 125 psig steam. A service factor of 0.8 was applied to the calculation.

Equipment		Utility Cost (\$/yr)									
	Small Unit	<b>Medium Unit</b>	Large Unit								
E-103	\$77,100	\$454,000	\$799,000								
E-104	\$4,530	\$17,600	\$39,900								
E-105	\$5,890	\$38,300	\$62,200								
E-106	\$2,460	\$20,900	\$27,200								
FTR Unit	\$70,200	\$502,000	\$589,000								
Syngas Unit	\$98,500	\$507,000	\$990,000								

Table 36: Utility Cost of Heat Exchanger, FTR Unit, and Syngas Unit

An additional utility of cooling water was required for the Air Separation Plant at a rate of 400 gpm per short ton/day. For the Air Separation Plant, it was found that the electricity option was significantly less expensive per year than the steam option when combined with the utility cost of  $O_2$  feed. Wastewater treatment was also included in manufacturing cost and includes the water stream removed from the Syngas Unit and the water streams removed from the separator system. A summary of the wastewater treatment cost is shown in Table 37.

Table 37:	Wastewater	Treatment Cost
-----------	------------	----------------

	Small	Medium	Large
Cost (\$/yr)	\$5,240	\$28,900	\$53,500

Reactant costs are shown in Table 38 and include the  $H_2$  for the HIU and the  $O_2$ , and MP steam supplied to the Syngas Unit. The  $CO_2$  is recycled from the absorber to the syngas feed after the FTR and will not require a supplemental  $CO_2$  stream when operating at steady state.

Table 38:	Reactant	Costs
-----------	----------	-------

Reactant	Small	Medium	Large
MP Steam	\$3,900	\$19,200	\$38,800
Oxygen	\$235,000	\$1,400,000	\$2,450,000
Hydrogen	\$24,000	\$153,000	\$254,000

Another manufacturing cost includes the cost of catalyst for each of the small, medium, and large FTR unit shown in Table 39.

#### Table 39: Cost of Catalyst

	Small	Medium	Large
Catalyst Weight (lb)	\$25	\$36	\$44

Transportation costs were an operating cost considered in the design and are summarized in Table 25. Other yearly operating costs were found by multiplying the utilities cost by 3% and estimates extra costs such as plant overhead cost, administrative costs, distribution and marketing, as well as research and development. A removal cost of units also had to be implemented which is equivalent to \$25,000 for every unit removed.

#### d. Discounted Cash Flow

To calculate the PWC, NPV, and IRR a cash flow table was constructed seen in Table 40 and Table 41. The cash flow is modeled as an expense project over a 20-year project life. Fixed capital cost was depreciated with a 7-year straight line depreciation method.

# Table 40: Cash Flow Table

Project Title	Modula	r Distributed Gas-to	-Liquids (GTL) Synthe	esis					
Corporate Financial Situation	Expense								
Minimum ROR	8%	or	0.08						
Inflation	3%	or	0.03						
Real ROR	5%		0.05						
\$1=\$1000									
Actual Year	2024	2025	2026	2027	2028	2029	2030	2031	2032
End of Year	0	1	2	3	4	5	6	7	8
+Revenue		34,087	34,087	34,134	37,411	37,401	35,966	33,677	37,640
+Salvage Value									
Net Revenue	0	34,087	34,087	34,134	37,411	37,401	35,966	33,677	37,640
-Utilities		(84,757)	(84,757)	(87,469)	(91,377)	(91,499)	(89,753)	(82,695)	(90,046)
-Transportation		(303)	(812)	(1,111)	(1,172)	(1,198)	(903)	(764)	(725)
-Yearly Expenses (Fixed Charges)		(2,543)	(2,543)	(2,624)	(2,741)	(2,745)	(2,693)	(2,481)	(2,701)
-Removal Cost									
-Depreciation		(114,655,840)	(114,655,840)	(114,766,854)	(114,845,866)	(114,869,609)	(114,891,380)	(114,898,418)	(242,578)
Taxable Income	0	(114,709,356)	(114,709,865)	(114,823,924)	(114,903,744)	(114,927,651)	(114,948,763)	(114,950,681)	(298,411)
-Tax @ 20%	0	22,941,871	22,941,973	22,964,785	22,980,749	22,985,530	22,989,753	22,990,136	59,682
Net Income	0	(91,767,485)	(91,767,892)	(91,859,139)	(91,922,996)	(91,942,121)	(91,959,010)	(91,960,545)	(238,729)
+Depreciation		114,655,840	114,655,840	114,766,854	114,845,866	114,869,609	114,891,380	114,898,418	242,578
+Writeoff									
-Fixed Capital (Equipment)	(116,317,519)		(112,623)	(80,157)	(24,088)	(22,086)	(7,140)	0	0
Cash Flow	(116,317,519)	22,888,355	22,775,326	22,827,558	22,898,782	22,905,403	22,925,229	22,937,874	3,850
Discount factor (P/F)	1.00	0.93	0.86	0.79	0.74	0.68	0.63	0.58	0.54
Discounted Cash Flow	(116,317,519)	21,192,922	19,526,171	18,121,252	16,831,289	15,589,032	14,446,783	13,384,029	2,080
PWC	4,694,170								
NPV	4,346,454								
DCFROR	9.096%								

# Table 41: Continued Cash Flow Table

2033	2034	2035	2036	2037	2038	2039	2040	2041	2042	2043	2044
9	10	11	12	13	14	15	16	17	18	19	20
31,827	37,740	33,191	23,326	32,057	28,306	26,803	21,895	10,914	8,511	6,124	3,757
											11,656,361
31,827	37,740	33,191	23,326	32,057	28,306	26,803	21,895	10,914	8,511	6,124	11,660,118
(80,672)	(93,279)	(81,771)	(56,260)	(78,450)	(69,786)	(66,783)	(53,737)	(26,851)	(21,243)	(14,692)	(8,908)
(581)	(464)	(582)	(543)	(373)	(924)	(898)	(1,190)	(909)	(462)	(400)	(302)
(2,420)	(2,798)	(2,453)	(1,688)	(2,353)	(2,094)	(2,003)	(1,612)	(806)	(637)	(441)	(267)
											(675)
(242,578)	(131,564)	(52,553)	(28,809)	(7,038)	0	0	0	0	0	0	0
(294,424)	(190,366)	(104,168)	(63,974)	(56,157)	(44,498)	(42,882)	(34,645)	(17,652)	(13,832)	(9,409)	11,649,965
58,885	38,073	20,834	12,795	11,231	8,900	8,576	6,929	3,530	2,766	1,882	(2,329,993)
(235,539)	(152,293)	(83,334)	(51,179)	(44,926)	(35,598)	(34,306)	(27,716)	(14,122)	(11,066)	(7,527)	9,319,972
242,578	131,564	52,553	28,809	7,038	0	0	0	0	0	0	0
0	0	0	0	0	0	0	0	0	0	0	0
7,039	(20,728)	(30,782)	(22,370)	(37,887)	(35,598)	(34,306)	(27,716)	(14,122)	(11,066)	(7,527)	9,319,972
0.50	0.46	0.43	0.40	0.37	0.34	0.32	0.29	0.27	0.25	0.23	0.21
3,521	(9,601)	(13,202)	(8,884)	(13,931)	(12,120)	(10,815)	(8,090)	(3,817)	(2,769)	(1,744)	1,999,583

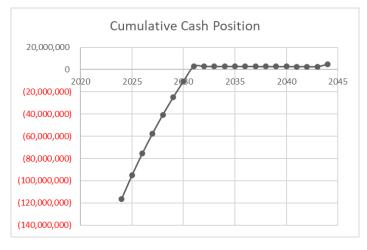
Units were built a year before they were put in operation, so units were depreciated the year after they were built. Units were also inflated to the year they were built by using the 3% inflation rate given in the project statement. A salvage value was calculated with the AIChE method in the project statement where salvage values is assumed to be equivalent to 10% of the fixed capital cost. Working capital or write-off was not needed because it was assumed to be accounted for in the 4.8 multiplier for fixed capital costs. A tax rate of 20% was given and no loss forward was required for the expense cash flow and was assumed to be absorbed through other projects. No amortization or depletion was needed since no minerals, timbers, or patents were needed. All costs were brought to the present at an 8% hurdle rate.

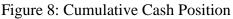
## **NPV and Sensitivity Analysis**

An NPV of \$4.35 B was achieved for this project. An internal rate of return was calculated to be 9.1%, which is greater than the 8% hurdle rate given for the project. The project NPV and IRR show that the project designed is economically attractive. However, with any preliminary design cost estimates there are errors and uncertainties present. Since the design cost estimates were based on heuristic methods for conventional process plants, there are uncertainties existing for the estimated modular processing plant.

# a. Payback Period

The cumulative cash position is shown in Figure 8 and a payback period was determined to be 6 years and 9 months which would occur September of 2030.





# b. Monte-Carlo Sensitivity Analysis

A propagation of uncertainty analysis was accomplished using the Monte-Carlo method. The Monte-Carlo method determines the probability of a parameter to occur. For the designed project, the probability of NPV was determined by varying the revenue, utilities, and fixed capital costs and applying a function of dependent of the variables. A distribution is then found by measuring the frequency of each NPV occurring out of 22,000 iterations. The Monte-Carlo

Method Analysis NPV frequency is shown in Figure 9. The probability of NPV occurring is then calculated from the frequencies and is shown in Figure 10.

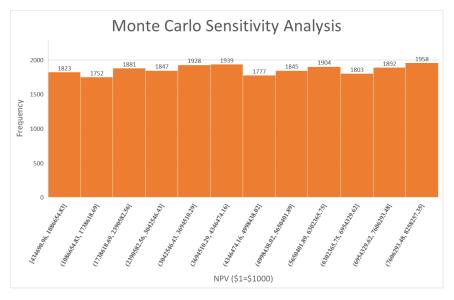


Figure 9: Monte-Carlo Sensitivity Analysis

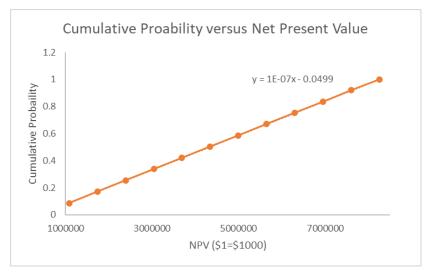


Figure 10: Monte-Carlo Cumulative Probability

The cumulative probability distribution in Figure 10 is interpreted as a uniform probability function [8]. The probability does not increase or decrease with increasing or decreasing NPV. The cumulative probability also shows that there is a 30% probability of the NPV to be less than or equal to the NPV calculated for the design project.

# c. Single-Variable Sensitivity Analysis

A tornado chart was constructed for a single-variable analysis. Revenue, utility cost, and transportation costs were parameters varied because of their impact on the NPV as well as the uncertainties possible with these parameters. Other operating costs were not varied because they

were directly calculated from the cost of utilities or fixed capital cost. Best-case and worst-case predictions used for the single-variable analysis shown in Table 42 were from *Table 10.1 Range of Variation of Factors Affecting the Profitability of a Chemical Process* in Turton, et.al [8]. However, these variables are from a 10-year project life, and it should be noted that the change in predictions could be minimal to the actual risks present. Figure 11 shows the tornado chart results.

Variable	Prediction
Utility Cost	-10% to +100%
Revenue	+5 to -20%
Transportation	-10% to =100%

Table 42: Single Va	ariable Predictions
---------------------	---------------------

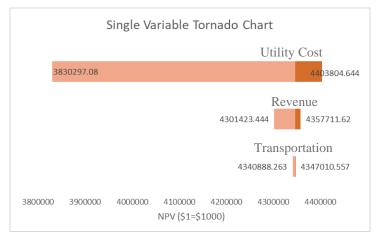


Figure 11: Single Variable Tornado Chart

It can be seen from the tornado chart that utility costs have the biggest impact on the project NPV compared to the other two variables and impose a larger risk on the success of the project. Capital costs were also varied but not shown because it had the biggest impact on NPV than other variables. However, it made utilities, transportation, and revenue almost negligible and their effects on NPV could not easily analyzed.

# **Conclusions**

The conversion of methane into higher molecular weight hydrocarbons was designed for a Gas-To-Liquid plant by utilizing Fischer-Tropsch Synthesis with three modular sized units capable of handling 500 MMSF/day, 2500 MMSF/day, and 5,000 MMSF/day: small, medium, and large units, respectively. The modular designs are a more efficient design than conventional units since they can be redeployed on multiple wells. In total, 13 small units, 9 medium units, and 5 large units were required throughout the project life to maximize the product. A different number of trucks were contracted for each quarter of the project life, 10 during the first, 9 during the second, 8 for the third, and 6 in the last quarter. The results of the project show that the PWC is approximately \$4.69 B and the net NPV is approximately \$4.35 B. The DCFROR is 9.10% with a hurdle rate of 8%. Based on the uncertainty analysis provided by the Monte Carlo Method the project has a 70% chance of producing a higher profit, however, it has a 30% probability of the NPV to be less than or equal to the NPV calculated. If the project were to decrease in profit, there could be a chance that the company would lose money through this project. However, the environmental impacts are less intense than flaring the methane into the atmosphere and therefore this project should be pursued.

#### **Recommendations**

Even though the conclusions of the project results in a reduced carbon footprint, there is still natural gas produced. A recycle stream of methane back to the Syngas Unit would allow the environmental hazards to be reduced even further.

Another recommendation would be to design a modular distillation column to further improve separation and maximize the products. This could potentially increase the NPV and make the project more desirable. A further way of saving money would be through heat integration. Heat integration using process streams for heat exchangers would cause there to be less cost of utilities. For the future of modular manufacturing, heuristics for modular designs would make the preliminary design stage even more accurate and cause less sources of error.

Additionally, the separation process could be improved if additional research was performed to find out if the heavier hydrocarbons form wax crystallization when cooled below 250 °F. It is standard practice to cool naphtha to 100 °F but this design limited that cooling due to the 250 °F restriction. The amount of heavier hydrocarbons that are in this stream are very little and could have the potential to have little to no affect. If the stream was cooled to 100 °F, this would result in an increase in Naphtha production, thus increasing revenue.

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# **Calculations and Simulation Outputs**

# **Small Unit MATLAB Simulation**

# FTR Small Unit Optimization Program

This script optimizes the FTR differential program with respect to a constructed objective function

#### **Include Parameters or Data**

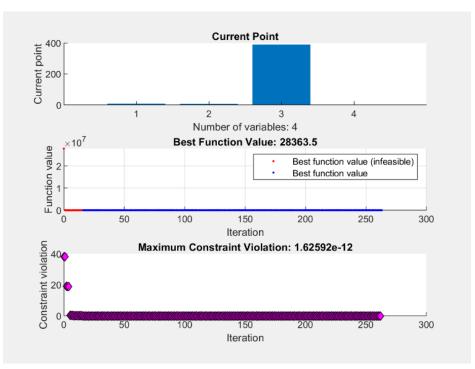
Set the initial point x0 and scale a for the optimization.

```
clear all;
close all;
x0 = [50;50;390;0.1667]; % n; W; T; D
a = 0.001; % Scaling factor for cost of n
```

Place the x0 value and any other problem data into the workspace by running this section before proceeding.

#### **Optimize Live Editor Task**

```
% Pass fixed parameters to objfun
objfun = @(x)objectiveFcn(x,a);
% Set nondefault solver options
options = optimoptions('fmincon','PlotFcn',{'optimplotx',...
'optimplotfvalconstr','optimplotconstrviolation'});
% Solve
[solution,objectiveValue] = fmincon(objfun,x0,[],[],[],[],[],[],[],...
@constraintFcn,options);
```



<stopping criteria details>

#### % Clear variables clearvars objfun options

## **Results**

View these variables.

<pre>sol.n = solution(1);</pre>	
sol.w = solution(2);	
<pre>sol.T = solution(3);</pre>	
sol.D = solution(4);	
disp(sol)	

n: 9.3047
w: 8.5855
T: 390
D: 0.1667

objectiveValue

objectiveValue = 2.8363e+04

View the nonlinear constraint function values at the solution.

[ccons,ceqcons] = constraintFcn(solution)

 $ccons = 1 \times 9$   $10^{3} \times -0.0083 -1.7857 \qquad 0 \qquad 0 \quad -0.0001 \quad -0.0000 \quad -0.0002 \cdots$  ceqcons = 1.6259e-12 D = 0.1667;

D = 0.1667; Ac = 3.14/4\*D^2; V = solution(2)\*454/0.8/28217; L = V/Ac

```
L = 7.9155
```

[w,X] = FTR\_eq(solution);7
Xf = X(end,1)

Xf = 0.9000

[w,X] = FTR\_eq(solution); Pf = X(end,2)

Pf = 1.0000

solution(1)\*solution(2)

ans = 79.8861

#### <u>Helper Functions — Local Functions</u>

The following code creates the objective function. Modify this code for your problem.

```
function f = objectiveFcn(x,a)
% Run simulation
[w, X] = FTR_eq(x);
% We care about w, n, &, Xf
Xf = X(end,1);
Wf = w(end);
f = a*((20/3*10*x(1)*Wf)^2); %-(10000*Xf)^2);
end
```

The following code creates the constraint function. Modify this code for your problem.

```
function [c,ceq] = constraintFcn(x)
[w,X] = FTR_eq(x);
Wf = w(end);
Xf = X(end, 1);
Pf = X (end, 2);
D = 0.1667;
Ac = 3.14/4*D^{2};
V = Wf*454/0.8/28217;
L = V/Ac;
PT = 450*Pf;
Total P = 450 - PT;
Allowable_P = 1.5*Total_P;
bigPipeDiameter = 2*sqrt(x(1)*Ac/(.65*pi));
c(1) = 1 - x(1);
                                      % n > 1
c(2) = x(1)-1795; & \% n < 1795 \\ \% c(3) = x(2)-100; & \% Wf < 100 \\ c(3) = x(3)-390; & \% T < 450 \\ c(4) = 390-x(3); & \% 390 < T \\ c(5) = L - 8; & \% L < 8ft \\ c(6) = x(4) - 0.1667; & \% D < 0.1667 ft \\ c(7) = 0 - x(4); & \% D > 0
c(7) = 0 - x(4);
                            % D > 0
c(8) = (bigPipeDiameter*3) - L;
c(9) = Allowable_P - 0.88; % pT < 0.88 for PT/P0 and
ceq = [Xf - .9]; % Xf = 0.75 (conversion ratio)
                                                           % pT < 0.88 for PT/P0 and liquid
end
```

```
function dXdW = FTR_differential(~, y, params)
% Extract Parameters
X = y(1);
p = y(2);
n = params(1);
T = params(2);
k = params(3);
k2 = params(3);
k2 = params(4);
delta = params(5);
P0 = params(6);
```

```
D
      = params(7);
rho0 = params(8);
mu
       = params(9);
phi
       = params(10);
rhob = params(11);
       = params(12);
gc
Dp
        = params(13);
% Explicit Equations
FT0 = 122.97/n;
FC00 = 32.86/n;
yCOO = FCOO/FTO;
e = yCOO*delta;
PCO0 = P0*yCO0;
Ac = 3.14/4*D^{2};
G = FTO/Ac;
rhoc = rhob/(1-phi);
B0 = (G*(1-phi))/(rho0*gc*Dp*phi^3)*(((150*(1-
phi)*mu)/Dp)+1.75*G)/144^2/3600^2;
alpha = (2*B0)/(Ac*rhoc*(1-phi)*P0);
PCO = PCOO^{*}((1-X)/(1+e^{X}))^{*}p;
PH2 = PCO0*((2-2.06*X)/(1+e*X))*p;
PCOSI = PCO/14.7;
PH2SI = PH2/14.7;
TK = (T-32)*(5/9)+273;
T1 = \exp(-4492*(1/TK-1/473));
T2 = \exp(8237*(1/TK-1/473));
T3 = exp(-10000*(1/TK-1/473));
rCOSI = -(k*T1*PH2SI*PCOSI)/(1+k2+T2*PCOSI)^2;
rC0 = rC0SI*28316.8/454;
rH20 = -rC0;
rCH4 = -rC0*0.03*T3;
rH2 = 2.06*rC0;
rC2 = 0.04*rCH4;
rC3 = 0.04*rCH4;
rC4 = 0.04*rCH4;
% ODEs
dXdW = [-rCO / FCOO;
                                     % dX/dW
```

(-alpha/(2*p))*(1+e*X);	% dp/dW
rCH4;	% dFCH4/dW
rC2;	% dFC2/dW
rC3;	% dFC3/dW
rC4;	% dFC4/dW
rH20;	% dFH2O/dW
rH2;	% dFH2/dW
];	
end	

```
function [w,X] = FTR_eq(x)
% Setup parameters
      = x(1);
n
Т
       = x(3);
k
      = 0.0173;
k2 = 4.512;
delta = -2;
P0
      = 450;
D
      = x(4);
rho0 = 0.5919;
mu
     = 0.04196;
phi = 0.4;
rhob = 49.9424;
     = 32.2;
gc
Dp = 0.006667;
% Initial Conditions
X0
     = 0;
p0
     = 1;
FCH40 = 18.99/n;
FC20 = 0/n;
FC30 = 0/n;
FC40 = 0/n;
FH200 = 4.59/n;
FH20 = 65.67/n;
IC = [X0 p0 FCH40 FC20 FC30 FC40 FH200 FH20];
params = [n T k k2 delta P0 D rho0 mu phi rhob gc Dp];
% Span of catalyst (low high)
Wspan = [0 x(2)];
```

```
% Run simulation
[w, X] = ode45(@(t,y) FTR_differential(t,y,params), Wspan, IC);
end
```

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# **Small Unit Polymath Report**

POLYMATH Report	No Title
Ordinary Differential Equations	28-Mar-2022

# Calculated values of DEQ variables

	Variable	Initial value	Minimal value	Maximal value	Final value
1	Ac	0.0218143	0.0218143	0.0218143	0.0218143
2	Allowablepressuredrop	0	0	0.0018409	0.0018409
3	alpha	9.47E-07	9.47E-07	9.47E-07	9.47E-07
4	B0	0.0002321	0.0002321	0.0002321	0.0002321
5	D	0.1667	0.1667	0.1667	0.1667
6	delta	-2.	-2.	-2.	-2.
7	Dp	0.006667	0.006667	0.006667	0.006667
8	e	-0.5344393	-0.5344393	-0.5344393	-0.5344393
9	FC2	0	0	0.0034218	0.0034218
10	FC3	0	0	0.0034218	0.0034218
11	FC4	0	0	0.0034218	0.0034218
12	FCH4	3.286	3.286	3.371546	3.371546
13	FCH40	3.286	3.286	3.286	3.286
14	FCO	3.286	0.2889359	3.286	0.2889359
15	FCO0	3.286	3.286	3.286	3.286

<mark>16</mark>	FCO20	0.06576	0.06576	0.06576	0.06576
17	FH2	6.5669	0.3929479	6.5669	0.3929479
18	FH20	6.5669	6.5669	6.5669	6.5669
19	FH2O	0.4585	0.4585	3.455564	3.455564
20	FH2O0	0.4585	0.4585	0.4585	0.4585
21	FN20	0.021	0.021	0.021	0.021
22	FT0	12.297	12.297	12.297	12.297
23	G	563.7133	563.7133	563.7133	563.7133
24	gc	32.2	32.2	32.2	32.2
25	k	0.0173	0.0173	0.0173	0.0173
26	k2	4.512	4.512	4.512	4.512
27	mu	0.04186	0.04186	0.04186	0.04186
28	n	10.	10.	10.	10.
29	р	1.	0.9999973	1.	0.9999973
30	PO	450.	450.	450.	450.
31	РСО	120.2488	20.62882	120.2488	20.62882
32	PCO0	120.2488	120.2488	120.2488	120.2488
33	PCOSI	8.180193	1.403321	8.180193	1.403321
34	PH2	240.4977	28.41896	240.4977	28.41896
35	PH2SI	16.36039	1.933263	16.36039	1.933263
<mark>36</mark>	phi	0.4	0.4	0.4	0.4
37	РТ	450.	449.9988	450.	449.9988
38	rC2	0.0008185	6.72E-05	0.0008185	6.72E-05
<mark>39</mark>	rC3	0.0008185	6.72E-05	0.0008185	6.72E-05
40	rC4	0.0008185	6.72E-05	0.0008185	6.72E-05

41	rCH4	0.0204636	0.00168	0.0204636	0.00168
42	rCO	-0.716937	-0.716937	-0.0588572	-0.0588572
43	rCO2	0	0	0	0
44	rCOSI	-0.0114946	-0.0114946	-0.0009437	-0.0009437
45	rH2	-1.47689	-1.47689	-0.1212459	-0.1212459
<mark>46</mark>	rH2O	0.716937	0.0588572	0.716937	0.0588572
47	rho0	0.6123	0.6123	0.6123	0.6123
48	rhob	49.9424	49.9424	49.9424	49.9424
<mark>49</mark>	rhoc	83.23733	83.23733	83.23733	83.23733
50	rN2	0	0	0	0
51	Т	390.	390.	390.	390.
52	T1	0.9778869	0.9778869	0.9778869	0.9778869
53	T2	1.041856	1.041856	1.041856	1.041856
54	Т3	0.9514385	0.9514385	0.9514385	0.9514385
55	T4	1.001245	1.001245	1.001245	1.001245
56	ТК	471.8889	471.8889	471.8889	471.8889
57	Totalpressuredrop	0	0	0.0012273	0.0012273
58	TR	849.4	849.4	849.4	849.4
<mark>59</mark>	W	0	0	8.5855	8.5855
60	Х	0	0	0.9120706	0.9120706
61	yCO0	0.2672196	0.2672196	0.2672196	0.2672196

Differential equations

1 d(X)/d(W) = -rCO / FCOO

2 d(p)/d(W) = (-alpha/(2\*p))\*((1+e\*X))

3	d(FCH4)/d(W) = rCH4
4	d(FC2)/d(W) = rC2
5	d(FC3)/d(W) = rC3
6	d(FC4)/d(W) = rC4
7	d(FH2O)/d(W) = rH2O
8	d(FH2)/d(W) = rH2

Explicit	equations
----------	-----------

	Explicit equations
1	n = 10
	number of tubes
2	FT0 = 122.97/n
3	FCO0 = 32.86/n
4	k2 = 4.512
5	k = 0.0173
6	delta = -2
7	yCO0 = FCO0/FT0
8	P0 = 450
9	e = yCO0*delta
10	PCO0 = P0*yCO0
11	PT = p*P0
12	Total pressured rop = PO-PT
13	Allowable pressured rop = 1.5*Total pressured rop
14	D = .1667
	ft
15	$Ac = 3.14/4 * D^2$

ft

16	G = FT0/Ac
1	superficial mass velocity
17	rho0 = 0.6123
	aspen lb/ft^3
18	mu = 0.04186
	aspen lb/ft*hr
19	phi = 0.4
20	rhob = 49.9424
	lb/ft^3 based on the bulk density 0.8 g/cm^3 converted
21	rhoc = (rhob)/(1-phi)
22	gc = 32.2
	lb*ft/lbf*s^2
23	Dp = 0.006667
	ft based on catalyst diameter of 1/16 in
24	$B0 = (G^{*}(1-phi))/(rho0^{*}gc^{*}Dp^{*}phi^{3})^{*}(((150^{*}(1-phi)^{*}mu)/Dp) + 1.75^{*}G)/144^{*}2/3600^{*}2)^{*}$
25	alpha = (2*B0)/(Ac*rhoc*(1-phi)*P0)
	rhoc is density of catalyst
26	PCO = PCO0*((1-X)/(1+e*X))*p
27	PH2 = PCO0*((2-2.06*X)/(1+e*X))*p
28	PCOSI = PCO/14.7
29	PH2SI = PH2/14.7
30	T = 390
31	$TK = (T-32)^*(5/9) + 273$
32	$T1 = \exp(-4492^*(1/TK - 1/473))$
33	$T2 = \exp(8237 * (1/TK - 1/473))$
34	$T3 = \exp(-10000*(1/TK-1/473))$

35	$T4 = \exp(250*(1/TK-1/473))$
36	rCOSI = -(k*T1*PH2SI*PCOSI)/(1+k2+T2*PCOSI)^2
37	rCO = rCOSI*28316.8/(454)
38	TR = TK*1.8
39	FCH40 = 3.286
40	FH2O0 = 0.4585
41	FH20 = 6.5669
42	FCO20 = 0.6576/n
43	FN20 = 0.21/n
44	FCO = FCO0 - FCO0*X
45	rH2O = -rCO
46	rCH4 = -rCO*0.03*T3
47	rN2 = 0
48	rCO2 = 0
49	rH2 = 2.06*rCO
50	rC2 = 0.04*rCH4
51	rC3 = 0.04 * rCH4
52	rC4 = 0.04 * rCH4

# General

Total number of equations	60
Number of differential equations	8
Number of explicit equations	52
Elapsed time	1.157 sec
Solution method	RKF_45
Step size guess. h	0.000001

# Medium Unit MATLAB Simulation

# FTR Medium Unit Optimization Program

This script optimizes the FTR differential program with respect to a constructed objective function

#### **Include Parameters or Data**

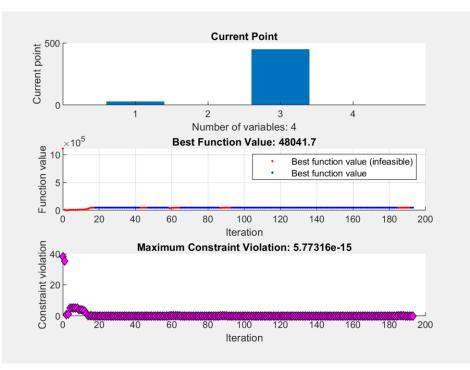
Set the initial point x0 and scale a for the optimization.

```
clear all;
close all;
x0 = [10;50;450;0.1667]; % n; W; T; D
a = 0.001; % Scaling factor for cost of n
```

Place the x0 value and any other problem data into the workspace by running this section before proceeding.

#### **Optimize Live Editor Task**

```
% Pass fixed parameters to objfun
objfun = @(x)objectiveFcn(x,a);
% Set nondefault solver options
options = optimoptions('fmincon','PlotFcn',{'optimplotx',...
    'optimplotfvalconstr','optimplotconstrviolation'});
% Solve
[solution,objectiveValue] = fmincon(objfun,x0,[],[],[],[],[],[],[],...
    @constraintFcn,options);
```



<stopping criteria details>

#### % Clear variables clearvars objfun options

#### **Results**

View these variables.

```
sol.n = solution(1);
sol.w = solution(2);
sol.T = solution(3);
sol.D = solution(4);
disp(sol)
```

n: 28.8016
w: 3.6098
T: 450.0000
D: 0.1667

objectiveValue

objectiveValue = 4.8042e+04

View the nonlinear constraint function values at the solution.

[ccons,ceqcons] = constraintFcn(solution)

```
ccons = 1 \times 9
```

 $10^{3}$  ×

 $-0.0278 -1.7662 -0.0000 -0.0600 -0.0047 -0.0000 -0.0002 \cdots$ ceqcons = -5.7732e-15

D = 0.1667; Ac = 3.14/4\*D^2; V = solution(2)\*454/0.8/28217; L = V/Ac

```
L = 3.3281
```

[w,X] = FTR\_eq(solution); Xf = X(end,1)

Xf = 0.9000

[w,X] = FTR\_eq(solution); Pf = X(end,2)

Pf = 1.0000

solution(1)\*solution(2)

ans = 103.9682

#### <u>Helper Functions — Local Functions</u>

The following code creates the objective function. Modify this code for your problem.

```
function f = objectiveFcn(x,a)
% Run simulation
[w, X] = FTR_eq(x);
% We care about w, n, &, Xf
Xf = X(end,1);
Wf = w(end);
f = a*((20/3*10*x(1)*Wf)^2); %-(10000*Xf)^2);
end
```

The following code creates the constraint function. Modify this code for your problem.

```
function [c,ceq] = constraintFcn(x)
[w,X] = FTR_eq(x);
Wf = w(end);
Xf = X(end, 1);
Pf = X (end, 2);
D = 0.1667;
Ac = 3.14/4*D^{2};
V = Wf*454/0.8/28217;
L = V/Ac;
PT = 450*Pf;
Total P = 450 - PT;
Allowable_P = 1.5*Total_P;
bigPipeDiameter = 2*sqrt(x(1)*Ac/(.65*pi));
c(1) = 1 - x(1);
                                      % n > 1
c(2) = x(1)-1795; & \% n < 1795 \\ \% c(3) = x(2)-100; & \% Wf < 100 \\ c(3) = x(3)-450; & \% T < 450 \\ c(4) = 390-x(3); & \% 390 < T \\ c(5) = L - 8; & \% L < 8ft \\ c(6) = x(4) - 0.1667; & \% D < 0.1667 ft \\ c(7) = 0 - x(4); & \% D > 0
                            % D > 0
c(7) = 0 - x(4);
c(8) = (bigPipeDiameter*3) - L;
c(9) = Allowable_P - 0.88; % pT < 0.88 for PT/P0 and
ceq = [Xf - .9]; % Xf = 0.75 (conversion ratio)
                                                           % pT < 0.88 for PT/P0 and liquid
end
```

```
function dXdW = FTR_differential(~, y, params)
% Extract Parameters
X = y(1);
p = y(2);
n = params(1);
T = params(2);
k = params(3);
k2 = params(3);
k2 = params(4);
delta = params(5);
P0 = params(6);
```

```
D
      = params(7);
rho0 = params(8);
mu
       = params(9);
phi
       = params(10);
rhob = params(11);
       = params(12);
gc
        = params(13);
Dp
% Explicit Equations
FT0 = 679.9355/n;
FC00 = 196.757/n;
yCOO = FCOO/FTO;
e = yCOO*delta;
PCO0 = P0*yCO0;
Ac = 3.14/4*D^{2};
G = FTO/Ac;
rhoc = rhob/(1-phi);
B0 = (G*(1-phi))/(rho0*gc*Dp*phi^3)*(((150*(1-
phi)*mu)/Dp)+1.75*G)/144^2/3600^2;
alpha = (2*B0)/(Ac*rhoc*(1-phi)*P0);
PCO = PCOO^{*}((1-X)/(1+e^{X}))^{*}p;
PH2 = PCO0*((2-2.06*X)/(1+e*X))*p;
PCOSI = PCO/14.7;
PH2SI = PH2/14.7;
TK = (T-32)*(5/9)+273;
T1 = \exp(-4492*(1/TK-1/473));
T2 = \exp(8237*(1/TK-1/473));
T3 = exp(-10000*(1/TK-1/473));
rCOSI = -(k*T1*PH2SI*PCOSI)/(1+k2+T2*PCOSI)^2;
rC0 = rC0SI*28316.8/454;
rH20 = -rC0;
rCH4 = -rC0*0.03*T3;
rH2 = 2.06*rC0;
rC2 = 0.04*rCH4;
rC3 = 0.04*rCH4;
rC4 = 0.04*rCH4;
% ODEs
dXdW = [-rCO / FCOO;
                                     % dX/dW
```

(-alpha/(2*p))*(1+e*X);	% dp/dW
rCH4;	% dFCH4/dW
rC2;	% dFC2/dW
rC3;	% dFC3/dW
rC4;	% dFC4/dW
rH20;	% dFH2O/dW
rH2;	% dFH2/dW
];	
end	

```
function [w,X] = FTR_eq(x)
% Setup parameters
      = x(1);
n
Т
       = x(3);
k
      = 0.0173;
k2 = 4.512;
delta = -2;
P0
      = 450;
D
      = x(4);
rho0 = 0.5919;
mu
     = 0.04196;
phi = 0.4;
rhob = 49.9424;
    = 32.2;
gc
Dp = 0.006667;
% Initial Conditions
X0 = 0;
p0
     = 1;
FCH40 = 60.6622/n;
FC20 = 0/n;
FC30 = 0/n;
FC40 = 0/n;
FH200 = 24.8352/n;
FH20 = 393.5494/n;
IC = [X0 p0 FCH40 FC20 FC30 FC40 FH200 FH20];
params = [n T k k2 delta P0 D rho0 mu phi rhob gc Dp];
% Span of catalyst (low high)
Wspan = [0 x(2)];
```

```
% Run simulation
[w, X] = ode45(@(t,y) FTR_differential(t,y,params), Wspan, IC);
end
```

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# **Medium Unit Polymath Report**

	Calculated values of DEQ valuables								
	Variable	Initial value	Minimal value	Maximal value	Final value				
1	Ac	0.0218143	0.0218143	0.0218143	0.0218143				
2	Allowablepressuredrop	0	0	0.0022624	0.0022624				
3	alpha	2.946E-06	2.946E-06	2.946E-06	2.946E-06				
4	B0	0.0007222	0.0007222	0.0007222	0.0007222				
5	D	0.1667	0.1667	0.1667	0.1667				
6	delta	-2.	-2.	-2.	-2.				
7	Dp	0.006667	0.006667	0.006667	0.006667				
8	e	-0.578752	-0.578752	-0.578752	-0.578752				
9	FC2	0	0	0.0282493	0.0282493				
10	FC3	0	0	0.0282493	0.0282493				
11	FC4	0	0	0.0282493	0.0282493				
12	FC5up	0	0	3.883317	3.883317				
13	FCH4	2.0918	2.0918	2.798032	2.798032				
14	FCH40	2.0918	2.0918	2.0918	2.0918				
15	FCO	6.784724	0.6720297	6.784724	0.6720297				
16	FCO0	6.784724	6.784724	6.784724	6.784724				
17	FCO20	0.099369	0.099369	0.099369	0.099369				
18	FH2	13.5707	0.9785495	13.5707	0.9785495				

# Calculated values of DEQ variables

<u>19</u>	FH20	13.5707	13.5707	13.5707	13.5707
20	FH2O	0.8564	0.8564	6.969094	6.969094
21	FH2O0	0.8564	0.8564	0.8564	0.8564
22	FN20	0.0431	0.0431	0.0431	0.0431
23	Fproducts	0	0	10.78699	10.78699
24	FT	23.44609	15.52824	23.44609	15.52824
25	FT0	23.44605	23.44605	23.44605	23.44605
26	G	1074.803	1074.803	1074.803	1074.803
27	gc	32.2	32.2	32.2	32.2
28	k	0.0173	0.0173	0.0173	0.0173
29	k2	4.512	4.512	4.512	4.512
30	mu	0.04196	0.04196	0.04196	0.04196
31	n	29.	29.	29.	29.
32	р	1.	0.9999966	1.	0.9999966
33	PO	450.	450.	450.	450.
34	РСО	130.2192	26.95138	130.2192	26.95138
35	PCO0	130.2192	130.2192	130.2192	130.2192
36	PCOSI	8.858448	1.833427	8.858448	1.833427
37	PH2	260.4384	39.19399	260.4384	39.19399
38	PH2SI	17.7169	2.666258	17.7169	2.666258
39	phi	0.4	0.4	0.4	0.4
40	РТ	450.	449.9985	450.	449.9985
41	rC2	0.0201838	0.0011943	0.0201838	0.0011943
42	rC3	0.0201838	0.0011943	0.0201838	0.0011943
43	rC4	0.0201838	0.0011943	0.0201838	0.0011943

<mark>44</mark>	rCH4	0.5045961	0.0298575	0.5045961	0.0298575
45	rCO	-4.367463	-4.367463	-0.2584276	-0.2584276
46	rCO2	0	0	0	0
47	rCOSI	-0.070023	-0.070023	-0.0041433	-0.0041433
48	rH2	-8.996973	-8.996973	-0.5323608	-0.5323608
<mark>49</mark>	rH2O	4.367463	0.2584276	4.367463	0.2584276
50	rho0	0.5919	0.5919	0.5919	0.5919
51	rhob	49.9424	49.9424	49.9424	49.9424
52	rhoc	83.23733	83.23733	83.23733	83.23733
53	rN2	0	0	0	0
54	Т	450.	450.	450.	450.
55	T1	1.832519	1.832519	1.832519	1.832519
<mark>56</mark>	T2	0.3293417	0.3293417	0.3293417	0.3293417
57	T3	3.851177	3.851177	3.851177	3.851177
58	T4	0.9668524	0.9668524	0.9668524	0.9668524
<mark>59</mark>	ТК	505.2222	505.2222	505.2222	505.2222
60	Totalpressuredrop	0	0	0.0015083	0.0015083
61	TR	909.4	909.4	909.4	909.4
62	W	0	0	3.6098	3.6098
63	Х	0	0	0.9009496	0.9009496
64	yC5up	0	0	0.2500809	0.2500809
65	yCO0	0.289376	0.289376	0.289376	0.289376

Differential equations

1 d(X)/d(W) = -rCO / FCOO

2 d(p)/d(W) = (-alpha/(2*p))*((1+e*X))
--

3	d(FCH4)/d(W) = rCH4
4	d(FC2)/d(W) = rC2
5	d(FC3)/d(W) = rC3
6	d(FC4)/d(W) = rC4
7	d(FH2O)/d(W) = rH2O
8	d(FH2)/d(W) = rH2

Explicit equations

1	n = 29
	number of tubes
2	FT0 = 679.9355/n
3	FCO0 = 196.757/n
4	k2 = 4.512
5	k = 0.0173
6	delta = -2
7	yCO0 = FCO0/FT0
8	P0 = 450
9	e = yCO0*delta
10	PCO0 = P0*yCO0
11	PT = p*P0
12	Total pressured rop = PO-PT
13	Allowable pressured rop = 1.5*Total pressured rop
14	D = .1667
	ft
15	Ac = 3.14/4 *D^2

	ft
16	G = FT0/Ac
	superficial mass velocity
17	rho0 = 0.5919
	aspen lb/ft^3
18	mu = 0.04196
	aspen lb/ft*hr
19	phi = 0.4
20	rhob = 49.9424
	lb/ft^3 based on the bulk density 0.8 g/cm^3 converted
21	rhoc = (rhob)/(1-phi)
22	gc = 32.2
	lb*ft/lbf*s^2
23	Dp = 0.006667
	ft based on catalyst diameter of 1/16 in
24	$B0 = (G^{*}(1-phi))/(rho0^{*}gc^{*}Dp^{*}phi^{3})^{*}(((150^{*}(1-phi)^{*}mu)/Dp) + 1.75^{*}G)/144^{2}/3600^{*}2)^{*}$
25	alpha = (2*B0)/(Ac*rhoc*(1-phi)*P0)
	rhoc is density of catalyst
26	PCO = PCO0*((1-X)/(1+e*X))*p
27	PH2 = PCO0*((2-2.06*X)/(1+e*X))*p
28	PCOSI = PCO/14.7
29	PH2SI = PH2/14.7
30	T = 450
31	TK = (T-32)*(5/9)+273
32	$T1 = \exp(-4492^*(1/TK - 1/473))$
33	$T2 = \exp(8237 * (1/TK - 1/473))$

ft

34	$T3 = \exp(-10000*(1/TK-1/473))$
35	$T4 = \exp(250^{*}(1/TK - 1/473))$
36	rCOSI = -(k*T1*PH2SI*PCOSI)/(1+k2+T2*PCOSI)^2
37	rCO = rCOSI*28316.8/(454)
38	TR = TK*1.8
39	FCH40 = 2.0918
40	FH2O0 = 0.8564
41	FH20 = 13.5707
42	FCO20 = 2.8817/n
43	FN20 = 1.2499/n
44	FCO = FCO0 - FCO0*X
45	rH2O = -rCO
46	rCH4 = -rCO*0.03*T3
47	rN2 = 0
48	rCO2 = 0
49	rH2 = 2.06*rCO
50	rC2 = 0.04 * rCH4
51	rC3 = 0.04 * rCH4
52	rC4 = 0.04 * rCH4
53	Fproducts = ((FH2O-FH2O0)+(FCH4-FCH40)+FC2+FC3+FC4)/.64
54	$FC5up = Fproducts^*.360$
55	FT = FH2+FCO+FH2O+FCH4+FN20+FC2+FC3+FC4+FC5up+FCO20
56	yC5up = FC5up/FT

Large Unit MATLAB Simulation

FTR Large Unit Optimization Program

This script optimizes the FTR differential program with respect to a constructed objective function

#### **Include Parameters or Data**

Set the initial point x0 and scale a for the optimization.

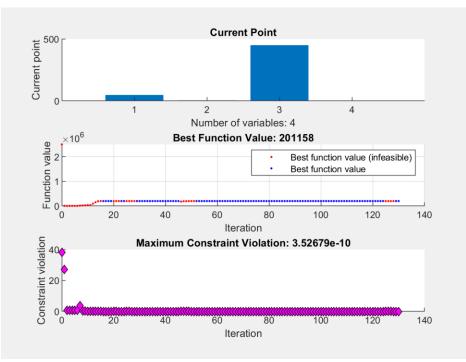
```
clear all;
close all;
x0 = [15;50;450;0.1667]; % n; W; T; D
a = 0.001; % Scaling factor for cost of n
```

Place the x0 value and any other problem data into the workspace by running this section before proceeding.

#### **Optimize Live Editor Task**

#### % Solve

```
[solution,objectiveValue] = fmincon(objfun,x0,[],[],[],[],[],[],...
@constraintFcn,options);
```



Appendix xxv

Local minimum found that satisfies the constraints.

Optimization completed because the objective function is non-decreasing in

feasible directions, to within the value of the optimality tolerance,

and constraints are satisfied to within the value of the constraint tolerance.

<stopping criteria details>

% Clear variables clearvars objfun options

#### **Results**

View these variables.

```
sol.n = solution(1);
sol.w = solution(2);
sol.T = solution(3);
sol.D = solution(4);
disp(sol)
```

n: 48.0313
w: 4.4293
T: 450.0000
D: 0.1667

objectiveValue

objectiveValue = 2.0116e+05

View the nonlinear constraint function values at the solution.

#### L = 4.0837

[w,X] = FTR\_eq(solution); Xf = X(end,1)

```
Xf = 0.9000
```

[w,X] = FTR\_eq(solution); Pf = X(end,2)

```
Pf = 1.0000
```

solution(1)\*solution(2)

ans = 212.7452

#### <u>Helper Functions — Local Functions</u>

The following code creates the objective function. Modify this code for your problem.

```
function f = objectiveFcn(x,a)
% Run simulation
[w, X] = FTR_eq(x);
% We care about w, n, &, Xf
Xf = X(end,1);
Wf = w(end);
f = a*((20/3*10*x(1)*Wf)^2); %-(10000*Xf)^2);
end
```

The following code creates the constraint function. Modify this code for your problem.

```
function [c,ceq] = constraintFcn(x)
[w,X] = FTR_eq(x);
Wf = w(end);
Xf = X(end,1);
Pf = X (end,2);
```

```
D = 0.1667;
   Ac = 3.14/4*D^{2};
   V = Wf^*454/0.8/28217;
    L = V/Ac;
   PT = 450*Pf;
   Total P = 450 - PT;
   Allowable_P = 1.5*Total_P;
   bigPipeDiameter = 2*sqrt(x(1)*Ac/(.72*pi));
   c(1) = 1 - x(1);
                                                                                                                                                                                                                                  % n > 1
c(1) = 1^{-}x(1), \qquad x \text{ if } y \text{
   c(8) = (bigPipeDiameter*3) - L;
   c(9) = Allowable_P - 0.88;
                                                                                                                                                                                                                                                                                                                                                          % pT < 0.88 for PT/P0 and liquid
   ceq = [Xf - .9];
                                                                                                                                                                                                                                                                                                       % Xf = 0.75 (conversion ratio)
    end
```

```
function dXdW = FTR_differential(~, y, params)
% Extract Parameters
Х
      = y(1);
      = y(2);
р
n
     = params(1);
Т
     = params(2);
k
     = params(3);
    = params(4);
k2
delta = params(5);
P0
    = params(6);
     = params(7);
D
rho0 = params(8);
mu = params(9);
phi = params(10);
rhob = params(11);
gc = params(12);
Dp
     = params(13);
```

```
% Explicit Equations
FT0 = 1259.56/n;
FC00 = 343.3128/n;
yCOO = FCOO/FTO;
e = yCO0*delta;
PCO0 = P0*yCO0;
Ac = 3.14/4*D^{2};
G = FT0/Ac;
rhoc = rhob/(1-phi);
B0 = (G*(1-phi))/(rho0*gc*Dp*phi^3)*(((150*(1-
phi)*mu)/Dp)+1.75*G)/144^2/3600^2;
alpha = (2*B0)/(Ac*rhoc*(1-phi)*P0);
PCO = PCO0*((1-X)/(1+e*X))*p;
PH2 = PC00*((2-2.06*X)/(1+e*X))*p;
PCOSI = PCO/14.7;
PH2SI = PH2/14.7;
TK = (T-32)*(5/9)+273;
T1 = \exp(-4492*(1/TK-1/473));
T2 = \exp(8237*(1/TK-1/473));
T3 = exp(-10000*(1/TK-1/473));
rCOSI = -(k*T1*PH2SI*PCOSI)/(1+k2+T2*PCOSI)^2;
rCO = rCOSI*28316.8/454;
rH20 = -rC0;
rCH4 = -rC0*0.03*T3;
rH2 = 2.06*rC0;
rC2 = 0.04*rCH4;
rC3 = 0.04*rCH4;
rC4 = 0.04*rCH4;
% ODEs
                                    % dX/dW
dXdW = [-rCO / FCOO;
                                   % dp/dW
         (-alpha/(2*p))*(1+e*X);
         rCH4;
                                     % dFCH4/dW
                                     % dFC2/dW
         rC2;
                                     % dFC3/dW
         rC3;
                                     % dFC4/dW
         rC4;
                                     % dFH2O/dW
         rH20;
                                     % dFH2/dW
         rH2;
         ];
```

end

```
function [w,X] = FTR_eq(x)
% Setup parameters
      = x(1);
n
Т
       = x(3);
k
      = 0.0173;
k2
      = 4.512;
delta = -2;
P0
     = 450;
D
      = x(4);
rho0 = 0.6017;
     = 0.04186;
mu
phi = 0.4;
rhob = 49.9424;
     = 32.2;
= 0.006667;
gc
Dp
% Initial Conditions
X0 = 0;
p0 = 1;
FCH40 = 174.2847/n;
FC20 = 0/n;
FC30 = 0/n;
FC40 = 0/n;
FH200 = 46.7159/n;
FH20 = 686.6826/n;
IC = [X0 p0 FCH40 FC20 FC30 FC40 FH200 FH20];
params = [n T k k2 delta P0 D rho0 mu phi rhob gc Dp];
% Span of catalyst (low high)
Wspan = [0 x(2)];
% Run simulation
[w, X] = ode45(@(t,y) FTR_differential(t,y,params), Wspan, IC);
end
```

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# Large Unit Polymath Report

### Calculated values of DEQ variables

	Variable	Initial value	Minimal value	Maximal value	Final value
1	Ac	0.0218143	0.0218143	0.0218143	0.0218143
2	Allowablepressuredrop	0	0	0.003285	0.003285
3	alpha	3.411E-06	3.411E-06	3.411E-06	3.411E-06
4	BO	0.0008361	0.0008361	0.0008361	0.0008361
5	D	0.1667	0.1667	0.1667	0.1667
6	delta	-2.	-2.	-2.	-2.
7	Dp	0.006667	0.006667	0.006667	0.006667
8	e	-0.5451313	-0.5451313	-0.5451313	-0.5451313
9	FC2	0	0	0.0292248	0.0292248
10	FC3	0	0	0.0292248	0.0292248
11	FC4	0	0	0.0292248	0.0292248
12	FC5up	0	0	4.017417	4.017417
13	FCH4	3.5568	3.5568	4.28742	4.28742
14	FCH40	3.5568	3.5568	3.5568	3.5568
15	FCO	7.006384	0.6826033	7.006384	0.6826033
16	FCO0	7.006384	7.006384	7.006384	7.006384
17	FCO20	0.1300531	0.1300531	0.1300531	0.1300531
18	FH2	14.0139	0.9869124	14.0139	0.9869124
19	FH20	14.0139	14.0139	14.0139	14.0139
20	FH2O	0.9534	0.9534	7.27718	7.27718
21	FH2O0	0.9534	0.9534	0.9534	0.9534

22	FN20	0.0446918	0.0446918	0.0446918	0.0446918
23	Fproducts	0	0	11.15949	11.15949
24	FT	25.70523	17.51395	25.70523	17.51395
25	FT0	25.70531	25.70531	25.70531	25.70531
26	G	1178.371	1178.371	1178.371	1178.371
27	gc	32.2	32.2	32.2	32.2
28	k	0.0173	0.0173	0.0173	0.0173
29	k2	4.512	4.512	4.512	4.512
30	mu	0.04186	0.04186	0.04186	0.04186
31	n	49.	49.	49.	49.
32	р	1.	0.9999951	1.	0.9999951
33	P0	450.	450.	450.	450.
34	РСО	122.6545	23.52397	122.6545	23.52397
35	PCO0	122.6545	122.6545	122.6545	122.6545
36	PCOSI	8.343847	1.60027	8.343847	1.60027
37	PH2	245.3091	33.97208	245.3091	33.97208
38	PH2SI	16.68769	2.311026	16.68769	2.311026
<mark>39</mark>	phi	0.4	0.4	0.4	0.4
40	PT	450.	449.9978	450.	449.9978
41	rC2	0.0186493	0.0009267	0.0186493	0.0009267
42	rC3	0.0186493	0.0009267	0.0186493	0.0009267
43	rC4	0.0186493	0.0009267	0.0186493	0.0009267
44	rCH4	0.4662327	0.0231665	0.4662327	0.0231665
45	rCO	-4.035413	-4.035413	-0.2005145	-0.2005145
<mark>46</mark>	rCO2	0	0	0	0

47	rCOSI	-0.0646993	-0.0646993	-0.0032148	-0.0032148
<mark>48</mark>	rH2	-8.312951	-8.312951	-0.4130599	-0.4130599
<mark>49</mark>	rH2O	4.035413	0.2005145	4.035413	0.2005145
50	rho0	0.6017	0.6017	0.6017	0.6017
51	rhob	49.9424	49.9424	49.9424	49.9424
52	rhoc	83.23733	83.23733	83.23733	83.23733
53	rN2	0	0	0	0
54	Т	450.	450.	450.	450.
55	T1	1.832519	1.832519	1.832519	1.832519
<mark>56</mark>	T2	0.3293417	0.3293417	0.3293417	0.3293417
57	T3	3.851177	3.851177	3.851177	3.851177
58	T4	0.9668524	0.9668524	0.9668524	0.9668524
<mark>59</mark>	ТК	505.2222	505.2222	505.2222	505.2222
60	Totalpressuredrop	0	0	0.00219	0.00219
61	TR	909.4	909.4	909.4	909.4
62	W	0	0	4.4293	4.4293
63	Х	0	0	0.9025741	0.9025741
64	yC5up	0	0	0.2293838	0.2293838
65	yCO0	0.2725657	0.2725657	0.2725657	0.2725657

# Differential equations

$$d(X)/d(W) = -rCO / FCO0$$
  

$$d(p)/d(W) = (-alpha/(2*p))*((1+e*X))$$
  

$$d(FCH4)/d(W) = rCH4$$
  

$$d(FC2)/d(W) = rC2$$

5	d(FC3)/d(W) = rC3
6	d(FC4)/d(W) = rC4
7	d(FH2O)/d(W) = rH2O
8	d(FH2)/d(W) = rH2

**Explicit** equations n = 49 1 number of tubes 2 FT0 = 1259.56/n3 FCO0 = 343.3128/n 4 k2 = 4.5125 k = 0.01736 delta = -27 yCO0 = FCO0/FT08 P0 = 4509 e = yCO0\*delta10 PCO0 = P0\*yCO0PT = p\*P011 12 Totalpressuredrop = P0-PT 13 Allowable pressured rop = 1.5\*Total pressured rop D = .1667 14 ft 15  $Ac = 3.14/4 * D^2$ ft

16

superficial mass velocity

G = FT0/Ac

17	rho0 = 0.6017
1	aspen lb/ft^3
18	mu = 0.04186
	aspen lb/ft*hr
19	phi = 0.4
20	rhob = 49.9424
	lb/ft^3 based on the bulk density 0.8 g/cm^3 converted
21	rhoc = (rhob)/(1-phi)
22	gc = 32.2
	lb*ft/lbf*s^2
23	Dp = 0.006667
	ft based on catalyst diameter of 1/16 in
24	$B0 = (G^{*}(1-phi))/(rho0^{*}gc^{*}Dp^{*}phi^{3})^{*}(((150^{*}(1-phi)^{*}mu)/Dp) + 1.75^{*}G)/144^{2}/3600^{*}2)^{*}$
25	alpha = (2*B0)/(Ac*rhoc*(1-phi)*P0)
	rhoc is density of catalyst
26	$PCO = PCO0*((1-X)/(1+e^{X}))*p$
27	PH2 = PCO0*((2-2.06*X)/(1+e*X))*p
28	PCOSI = PCO/14.7
29	PH2SI = PH2/14.7
30	T = 450
31	$TK = (T-32)^*(5/9) + 273$
32	$T1 = \exp(-4492*(1/TK-1/473))$
33	$T2 = \exp(8237^*(1/TK - 1/473))$
34	$T3 = \exp(-10000*(1/TK-1/473))$
35	$T4 = \exp(250^{*}(1/TK - 1/473))$
36	rCOSI = -(k*T1*PH2SI*PCOSI)/(1+k2+T2*PCOSI)^2

Appendix xxxv

37	rCO = rCOSI*28316.8/(454)
38	TR = TK*1.8
39	FCH40 = 3.5568
40	FH2O0 = 0.9534
41	FH20 = 14.0139
42	FCO20 = 6.3726/n
43	FN20 = 2.1899/n
44	FCO = FCO0 - FCO0*X
45	rH2O = -rCO
46	rCH4 = -rCO*0.03*T3
47	rN2 = 0
48	rCO2 = 0
49	rH2 = 2.06*rCO
50	rC2 = 0.04 * rCH4
51	rC3 = 0.04 * rCH4
52	rC4 = 0.04 * rCH4
53	Fproducts = ((FH2O-FH2O0)+(FCH4-FCH40)+FC2+FC3+FC4)/.64
54	$FC5up = Fproducts^*.360$
55	FT = FH2+FCO+FH2O+FCH4+FN20+FC2+FC3+FC4+FC5up+FCO20
56	yC5up = FC5up/FT

# Appendix

#### Medium Unit Stream Table

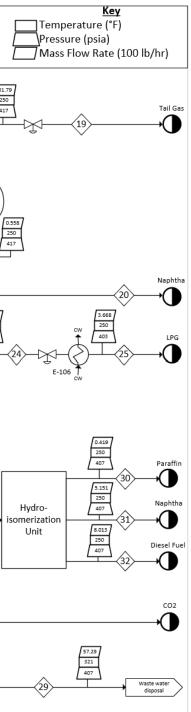
	Stream Number		1	2	3	4	4 5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	5 27	28	29	30	31	32
	bucanna			-				H-101				CO2	V-101		10		15	10	1,	10	10		p						Bottom	Top & Bottom	Waste			
			Oxygen	Natural	HP Steam	Carbon	H-101	Heated	R-101 Feed	F-101 Feed	V-101	Separation	Waste	Removed	E-102 Feed	R-102 Feed	E-103 Feed	V-102	F-104 Feed	V-103 Feed	Tail Gas	lop wa	iter	E-105	V-104 Feed	LPG	Cooled LPG	HIU Feed	Water	Water		araffin	Bottom	Diesel
	Description		owigen.	Gas	in occum	Dioxide	Feed	Stream		2 101 . ccu	Feed	Unit Feed	Water	CO2	2 202 7 000		2 200 7 000	Feed	2 20 11 000	1 100 1000	i un ous	Naphtha Pro		Feed	1011000	2. 0	coolica zi o		Product	Products	Disposal	N	Naphtha	Fuel
	Description							Stream				onicreed	water									110							TTOULCE	Troducts	Disposal			
	Vapor Fraction		1.00	1.00	1.00	1.00	0.74	1.00	1.00	1.00	0.84	1.00	0.00	1.00	1.00	1.00	1.00	0.43	1.00	0.96	1.00	0.00	0.00	0.00	0.05	1.00	0.08	0.00	0.00	0.00	0.00	0.19	0.00	0.00
	Vapor Fraction	0r	75.00			100.00			1.00			1.00	0.00	192.10		1.00		275.00	275.00		250.00			275.00	0.05	420.00								250.00
	Temperature	F	514.70						870.40	1699.00		212.00	212.00	486.70	216.78 486.70						417.00				420.00					414.69 407.00		250.00 407.00	250.00 407.00	407.00
	Pressure	psia	125.00			514.70	-		500.70	500.70		496.70 734.69	496.70	486.70		472.70		431.00	431.00					431.00	196.94	417.00		1			317.91			0.365511
	Total Molar Flow			4334.00		45.00 1980.00			575.20	870.30			135.65		679.94 7887.18	7887.18		343.53	146.59 3344.53		140.00 3178.66		10.04	196.94	4899.21	366.84	. 10.71	1		182.26			0.800722 (	
	Total Mass Flow Actual Vol. Flow								12740.00 68872.12	169796.10	12740.00	10296.83 44762.12	2445.03 176.37	2409.65 2896.72	41865.88	56955.63			10928.58	3344.53 10195.30	10180.96			4899.21	1266.42	832.62				3284.16 274.68		41.88 2.91	515.07 35.85	801.29 55.77
	Mass Enthalpy	barrel/day BTU/lb			-6371.00		1 20012.75 -3642.00		-2035.00		-3085.00	-2230.07		-3833.20	-1740.29	-1591.53			-1883.54		-1784.31		41.05 -		-4610.47	-2653.33						-	-859.82	-767.18
		lb/hr	3995.00			-3858.00			3960.00	-2035.00		-2230.07	-6685.68 0.00	0.00	-1740.29	0.00		0.00	-1883.54		0.00		+1.05 - 0.00	0.00	-4610.47	0.00				-0403.19	0.00	0.00	0.00	0.00
	Oxygen Water	lb/hr	0.00		2433.00	0.00			2432.94		2889.61	447.41	2442.20	0.00	447.41	447.41			297.42		187.32			3344.49	3344.49	142.29		1		3282.78		29.37	0.00	0.00
		lb/hr	0.00			0.00			35.02	35.02		35.01	0.00	0.00	35.01	35.01	17.51	17.51	17.42		187.32		0.00	0.09	0.09	0.08		-		0.00		0.01	0.00	0.00
	Nitrogen Carbon Monoxid		0.00			0.00			0.00		5511.55	5511.34	0.00	0.00	5511.34	5511.34		545.88	543.42		543.32		0.00	2.46	2.46	2.27	2.2			0.00		0.01	0.00	0.00
	Carbon Dioxide	lb/hr	0.00			45.00					2539.06		2.59	2409.65	126.82	126.82		126.82	124.71	124.71	124.62		0.01	2.40	2.40	1.75	-			0.04		0.15	0.00	0.00
	Hydrogen	lb/hr	0.00			45.00			0.00	793.43		793.40	0.03	0.00	793.40	793.40		28.66	28.57	28.57	28.57	0.08	0.02	0.09	0.09	0.08				0.10		0.20	0.00	0.00
	Ethane	lb/hr	0.00			0.00			0.00	0.00		0.00	0.00	0.00	0.00	0.00		24.63	24.12		28.57		0.00	0.05	0.51	0.00				0.01		0.62	0.00	0.00
	Propane	lb/hr	0.00			0.00	-		0.00	0.00		0.00	0.00	0.00	0.00	0.00		36.13	34.63		34.56		0.00	1.50	1.50	1.10				0.01		4.14	0.00	0.00
	Methane	lb/hr		4334.00		0.00			4333.51	973.20		973.20	0.00	0.00	973.20	973.20		1301.53	1289.71	1289.71	1289.21	0.45	0.06	11.82	11.82	10.34	_			0.40		2.21	0.00	0.00
	n-Butane	lb/hr	0.00			0.00			4333.31	0.00		0.00	0.00	0.00	0.00	0.00		47.61	43.79	43.79	43.59		0.00	3.82	3.82	2.45		-		0.40	0.40	5.10	0.00	0.00
	n-Pentane	lb/hr	0.00			0.00			0.00	0.00			0.00	0.00	0.00	0.00	-	332.00	283.79	283.79	281.11	2.67	0.00	48.21	48.21	26.03	-			0.20		0.00	0.00	0.00
	n-Hexane	lb/hr	0.00			0.00			0.00	0.00			0.00	0.00	0.00	0.00		308.63	231.25	231.25	226.83	4.42	0.00	77.38	77.38	32.98				0.19	0.20	0.00	- F	0.00
	n-Heptane	lb/hr	0.00			0.00			0.00	0.00			0.00	0.00	0.00	0.00		278.93	171.41	171.41	165.14	6.28		107.52	107.52	35.64				0.14		0.00	- F	0.00
Nap	htha n-Octane	lb/hr	0.00			0.00			0.00	0.00			0.00	0.00	0.00	0.00		246.94	113.29		105.44			133.66	133.66	33.30		1		0.10	0.14	0.00	515.07	0.00
	n-Nonane	lb/hr	0.00			0.00			0.00	0.00			0.00	0.00	0.00	0.00		215.21	67.30		58.86			147.91	147.91	26.70			0.10	0.07	0.10	0.00	- F	0.00
	n-Decane	lb/hr	0.00			0.00			0.00	0.00			0.00	0.00	0.00	0.00		185.24	36.96		29.06	-		148.27	148.27	19.40		1		0.04		0.00	- F	0.00
	n-C11	lb/hr	0.00			0.00			0.00	0.00			0.00	0.00	0.00	0.00		157.85	18.79		12.40			139.05	139.05	12.53				0.02		0.00	0.00	0.00
	n-C12	lb/hr	0.00			0.00			0.00	0.00			0.00	0.00	0.00	0.00		133.39	9.24		4.70			124.15	124.15	7.82			0.01	0.01		0.00	0.00	
	n-C13	lb/hr	0.00			0.00			0.00	0.00			0.00	0.00	0.00	0.00		111.95	4.61		1.71	-		107.34	107.34	4.96				0.00		0.00	0.00	
	n-C14	lb/hr	0.00			0.00			0.00	0.00			0.00	0.00	0.00	0.00		93.39	2.12	-	0.49		0.00	91.27	91.27	2.78		-		0.00		0.00	0.00	
	n-C15	lb/hr	0.00						0.00	0.00			0.00	0.00	0.00	0.00		77.51	1.01		0.15	-	0.00	76.50	76.50	1.63				0.00		0.00	0.00	
Die	esel n-C16	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00		64.05	0.47	0.47	0.04	0.43	0.00	63.58	63.58	0.92	0.92	62.66	0.00	0.00		0.00	0.00	801.29
	n-C17	lb/hr	0.00						0.00	0.00			0.00	0.00	0.00	0.00		52.72	0.27	-	0.02		0.00	52.45	52.45	0.61				0.00		0.00	0.00	
	n-C18	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00		43.24	0.11	0.11	0.00		0.00	43.13	43.13	0.32	0.32	42.82	0.00	0.00	0.00	0.00	0.00	
	n-C19	lb/hr	0.00			0.00			0.00	0.00			0.00	0.00	0.00	0.00	-	35.36	0.05		0.00	-	0.00	35.31	35.31	0.18		-		0.00		0.00	0.00	
	n-C20	lb/hr	0.00			0.00	0.00	0.00	0.00	0.00			0.00	0.00	0.00	0.00		28.83	0.03		0.00		0.00	28.81	28.81	0.10				0.00		0.00	0.00	
	n-C21	lb/hr	0.00	0.00		0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	23.45	23.45	0.01	0.01	0.00		0.00	23.44	23.44	0.07	0.07		0.00	0.00	0.00	0.00	0.00	0.00
	n-C22	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	19.03	19.03	0.01	0.01	0.00	0.01	0.00	19.03	19.03	0.04	0.04	18.98	0.00	0.00	0.00	0.00	0.00	0.00
	n-C23	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	15.41	15.41	0.00	0.00	0.00	0.00	0.00	15.41	15.41	0.02	0.02	15.39	0.00	0.00	0.00	0.00	0.00	0.00
	n-C24	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	12.46	12.46	0.00	0.00	0.00	0.00	0.00	12.46	12.46	0.01	0.01	. 12.44	0.00	0.00	0.00	0.00	0.00	0.00
	n-C25	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	10.05	10.05	0.00	0.00	0.00	0.00	0.00	10.05	10.05	0.01	0.01	. 10.05	0.00	0.00	0.00	0.00	0.00	0.00
	n-C26	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	8.10	8.10	0.00	0.00	0.00	0.00	0.00	8.10	8.10	0.00	0.00	8.10	0.00	0.00	0.00	0.00	0.00	0.00
	n-C27	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	6.52	6.52	0.00	0.00	0.00	0.00	0.00	6.52	6.52	0.00	0.00	6.51	0.00	0.00	0.00	0.00	0.00	0.00
	n-C28	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	5.23	5.23	0.00	0.00	0.00	0.00	0.00	5.23	5.23	0.00	0.00	5.23	0.00	0.00	0.00	0.00	0.00	0.00
	n-C29	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	4.20	4.20	0.00	0.00	0.00	0.00	0.00	4.20	4.20	0.00	0.00	4.20	0.00	0.00	0.00	0.00	0.00	0.00
	n-C30	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	3.37	3.37	0.00	0.00	0.00	0.00	0.00	3.37	3.37	0.00	0.00	3.37	0.00	0.00	0.00	0.00	0.00	0.00

# Large Unit Stream Table

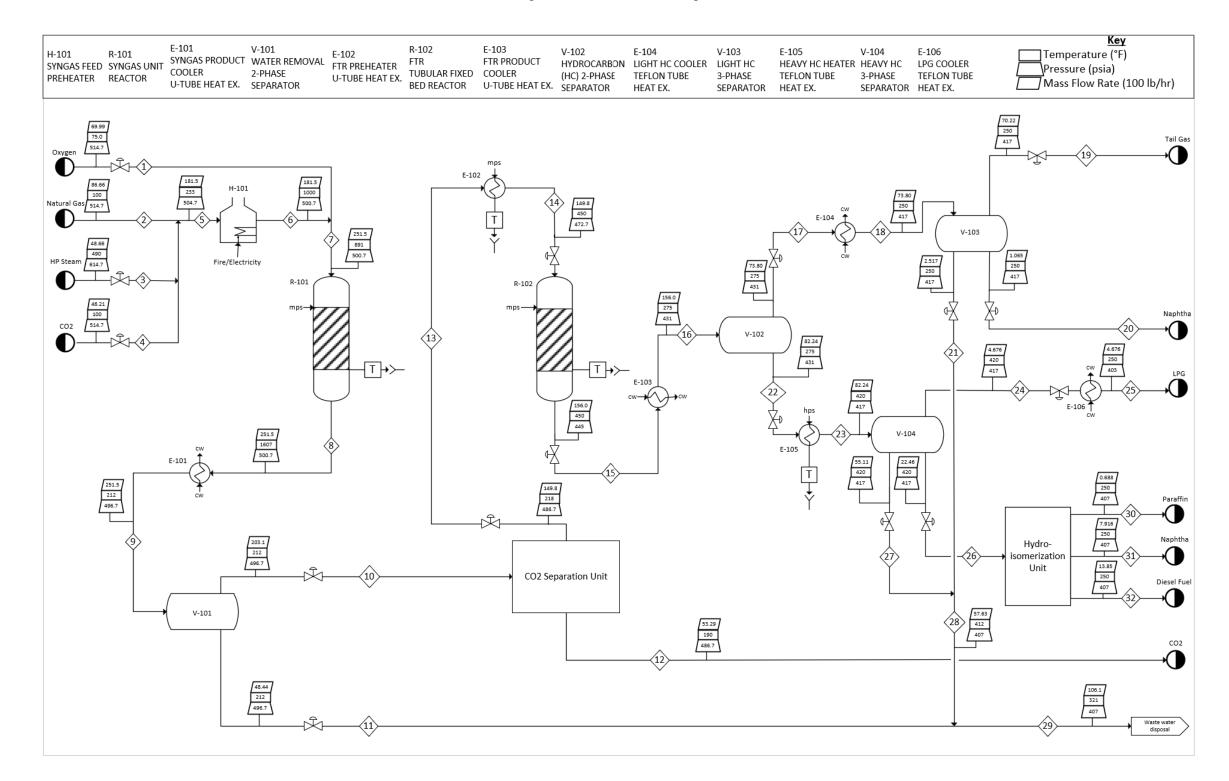
	Stream Number		1	n	2		1 -	c	-		0	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	20	21	22
	Stream Number		1	2	3	4	+ 3		/	•	9	-	11 V 101	12	15	14	15	10	17	10	19	20		22	25	24	25	20			-	50	51	52
			0.0.00	Natural	HP Steam	Carbon	H-101	H-101	R-101 Feed	E 101 Food	V-101	CO2	V-101	Removed	E-102 Feed	D 102 Food	F 102 Food	V-102	E-104 Feed	( 102 Food	Tail Cas	Тор	Top	E-105	V-104 Feed	LPG	Cooled LDC	LILL Food	Bottom	Top & Bottom	Waste	Paraffin	Bottom	Diesel
			Oxygen	Gas	TP Steam	Dioxide	Feed	Heated	K-101 Feed	E-101 Feed	Feed	Separation	Waste	CO2	E-102 Feeu	R-102 Feed	E-105 Feed	Feed	E-104 Feed	V-105 Feed	Tail Gas	Naphtha	Water	Feed	v-104 Feeu	LPG	Cooled LPG	nio reeu	Water	Water	Water	Paratilit	Naphtha	Fuel
	Description							Stream				Unit Feed	Water										Product						Product	Products	Disposal			
	Vener Frenting		1.00	1.00	0.00	1.00	0.74	1.00	1.00	1.00	0.04	1.00	0.00	1.00	1.00	1.00	1.00	0.50	1.00	0.00	1.00	0.00	0.00	0.00	0.04	1.00	0.10	0.00	0.00	0.00	0.00	0.21	0.00	0.00
	Vapor Fraction	95	1.00			1.00			1.00		0.84	1.00	0.00	1.00	1.00	1.00		0.50	1.00	0.96	1.00			0.00	0.04	1.00	0.10			0.00		0.21	0.00	
	Temperature	-F	75.00						886.18	1626.92	212.00	212.00	212.00	190.46	217.67	450.00			275.00	250.00	250.00				420.00	420.00	250.00			413.08				
	Pressure	psia	514.70	514.70	614.70		-		500.70	500.70	496.70	496.70	496.70	486.70	486.70	472.70		431.00	431.00	417.00	417.00	417.00		431.00	417.00	417.00	433.00			407.00				
		Ibmole/hr	219.00		270.10					1649.32	1649.32	1380.64	268.68	121.08	1259.56	1259.56	-		335.83	335.83	320.96	0.90		336.29	336.29	14.18	14.18	_		319.81			1.230576	
	Total Mass Flow	lb/hr	6999.27	8666.37	4865.88		2 18153.27			25152.47		20308.94	4843.54	5328.66	14980.28	14980.28			7380.20	7380.20	7022.01	106.46			8224.20	467.64		2245.80			10606.03	68.76	791.58	
	Actual Vol. Flow				422.19		40753.58			311045.78		83961.90	349.37	6382.53	77580.39		59504.81			23354.71	23324.36	11.87		709.94	1846.12	1108.99	143.67				820.39	4.79		96.43
	Mass Enthalpy	BTU/lb		-2010.79						-2156.16	-3158.72	-2317.69	-6685.19	-3833.61	-1778.45	-1630.49			-1948.93	-1995.09	-1845.89		-6640.92	-4863.78	-4706.66	-2745.52	-3243.13				-6565.59	-4926.10	-859.82	-767.18
	Oxygen	lb/hr	6937.92	0.00		0.00			6937.92	0.00	0.00	0.00	0.00	0.00	0.00	0.00		0.00	0.00	0.00	0.00	0.00		0.00	0.00	0.00	0.00			0.00		0.00	0.00	0.00
	Water	lb/hr	0.00	0.00		0.00			4865.88	5678.66	5678.66	841.59	4837.07	0.00	841.59	841.59			680.96	680.96	429.15	0.27		5744.64	5744.64	188.89	188.89				10597.07	47.27	0.00	
	Nitrogen	lb/hr	61.35	0.00		0.00			61.35	61.35	61.35	61.35	0.00	0.00	61.35	61.35		30.68	30.58	30.58	30.57	0.00		0.10	0.10	0.09	0.09			0.00		0.01	0.00	
	Carbon Monoxide Carbon Dioxide		0.00						0.00	9616.89	9616.89	9616.50	0.39	0.00	9616.50	9616.50		936.87	934.00	934.00	933.85	0.12		2.87	2.87 3.22	2.62	2.62		0.05	0.07		0.21	0.00	0.00
		lb/hr	0.00				-		4621.02	5615.13	5615.13	5609.12	6.01	5328.66	280.46	280.46		280.46	277.24	277.24	277.06	0.14		3.22	3.22	2.58	2.58		-	0.20			0.00	
	Hydrogen Ethane	lb/hr lb/hr	0.00			0.00		0.00	0.00	1384.41 0.00	1384.41	1384.35 0.00	0.06	0.00	1384.35 0.00	1384.35		48.84	48.74 42.46	48.74 42.46	48.73 42.42	0.00	0.00	0.10	0.10	0.10	0.10		0.00	0.00			0.00	0.00
		lb/hr	0.00			0.00			0.00		0.00	0.00	0.00	0.00	0.00	0.00		43.06 63.15	61.37	42.46	42.42	0.03	0.00	1.78	1.78	0.47	0.47		0.01	0.01			0.00	0.00
	Propane								8666.37					0.00					3348.84		3347.77	0.11		-	20.90				0.03	0.03				
	Methane	lb/hr lb/hr		8666.37 0.00	0.00	0.00				2796.04		2796.03 0.00	0.00		2796.03 0.00	2796.03		3369.74 83.23	78.65	3348.84	78.35	0.93	0.14	20.90	20.90	17.77 2.73	1.24		0.78	0.92			0.00	0.00
	n-Butane		0.00			0.00	-		0.00	0.00	0.00		0.00	0.00		0.00				78.65							2.73						0.00	
	n-Pentane	lb/hr	0.00			0.00			0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00		580.32	521.15	521.15	517.22	3.92 6.76			59.18 98.76	28.96	28.96 37.24		0.29	0.29			-	0.00
	n-Hexane	lb/hr	0.00			0.00			0.00				0.00	0.00		0.00		539.46	440.71	440.71	433.95					37.24	-		0.28	0.28			-	0.00
Naphth	a n-Heptane	lb/hr	0.00			0.00			0.00		0.00	0.00	0.00	0.00	0.00	0.00		487.55 431.64	343.30	343.30	333.17 227.50	10.13 13.56	0.00		144.25 190.58	41.43 40.38	41.43		0.22	0.22			791.58	0.00
	n-Octane	lb/hr	0.00			0.00						0.00	0.00						241.06	241.06									-		-		-	0.00
	n-Nonane	lb/hr	0.00			0.00			0.00		0.00	0.00	0.00	0.00	0.00	0.00		376.17	152.02	152.02	136.32		0.00		224.16	33.85	33.85		0.11	0.11				0.00
	n-Decane	lb/hr	0.00			0.00			0.00		0.00	0.00	0.00	0.00	0.00	0.00		323.79	87.65	87.65	71.94	-	0.00		236.14	25.54	25.54		0.07	0.07				0.00
	n-C11	lb/hr	0.00			0.00	-		0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00		275.91	46.16	46.16	32.62		0.00		229.75	16.93	16.93		0.04	0.04				
	n-C12 n-C13	lb/hr	0.00			0.00			0.00		0.00	0.00	0.00	0.00	0.00	0.00		233.16	23.20	23.20	13.03		0.00		209.97	10.72	10.72		0.02	0.02				
	n-C13	lb/hr	0.00			0.00			0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00		195.67	11.71	11.71	4.95 1.47	6.76 3.94	0.00		183.97 157.83	6.84 3.85	6.84		0.00	0.00				
		lb/hr	0.00			0.00							0.00	0.00		0.00		163.24	5.41	5.41			0.00				3.85			0.00				
Diesel	n-C15	lb/hr	0.00			0.00			0.00		0.00	0.00	0.00	0.00	0.00	0.00		135.49	2.60	2.60	0.45	2.15	0.00		132.89	2.25	2.25		0.00	0.00				1385.45
	n-C16	lb/hr	0.00			0.00	-		0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00		111.96	1.21	1.21	0.12	1.09	0.00		110.74	1.28	1.28		0.00	0.00				
	n-C17	lb/hr	0.00			0.00			0.00		0.00	0.00	0.00	0.00	0.00	0.00		92.15	0.69	0.69 0.28	0.05	0.64	0.00		91.46 75.30	0.85	0.85		0.00	0.00				
	n-C18 n-C19	lb/hr lb/hr	0.00			0.00			0.00				0.00	0.00		0.00		75.58	0.28	0.28			0.00			0.43	0.43		0.00	0.00				
	n-C20	lb/hr	0.00			0.00			0.00		0.00	0.00	0.00	0.00	0.00	0.00		61.80 50.40	0.13	0.13	0.00	0.13	0.00		61.67 50.33	0.25	0.25		0.00	0.00			0.00	
	n-C20	lb/hr	0.00			0.00	1		0.00		0.00	0.00	0.00	0.00	0.00	0.00		40.99	0.06	0.06	0.00		0.00		40.96	0.14	0.14		0.00	0.00				
	n-C21	lb/hr									0.00	0.00		0.00	0.00	0.00				0.03	0.00	0.03	0.00		33.25				0.00	0.00				0.00
	n-C23	lb/hr	0.00			0.00			0.00		0.00	0.00	0.00	0.00	0.00	0.00		33.27 26.94	0.02	0.02	0.00	0.02	0.00		26.93	0.06	0.06		0.00	0.00				0.00
		- 7																																
	n-C24 n-C25	lb/hr lb/hr	0.00			0.00			0.00		0.00	0.00	0.00	0.00	0.00	0.00		21.78 17.57	0.00	0.00	0.00	0.00	0.00		21.77 17.57	0.02	0.02		0.00	0.00				0.00
																	-	-						-	-									
	n-C26 n-C27	lb/hr lb/hr	0.00			0.00			0.00		0.00	0.00	0.00	0.00	0.00	0.00		14.16 11.39	0.00	0.00	0.00	0.00	0.00		14.16 11.39	0.01	0.01	14.15	0.00	0.00			0.00	0.00
	n-C27 n-C28	lb/hr							0.00			0.00	0.00					9.15		0.00			0.00						0.00	0.00				
		lb/hr	0.00			0.00			0.00		0.00	0.00	0.00	0.00	0.00	0.00		9.15	0.00	0.00	0.00	0.00	0.00	9.15 7.34	9.15 7.34	0.00	0.00							0.00
	n-C29		0.00			0.00			0.00		0.00	0.00	0.00		0.00			-	0.00	0.00	0.00	0.00	0.00	-	7.34 5.88	0.00	0.00	-		0.00				0.00
	n-C30	lb/hr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	5.88	5.88	0.00	0.00	0.00	0.00	0.00	5.88	5.68	0.00	0.00	5.88	0.00	0.00	0.00	0.00	0.00	0.00

# Medium Unit Process Flow Diagram

	R-101 EED SYNGAS UNIT R REACTOR	E-101 SYNGAS PRODUCT COOLER U-TUBE HEAT EX.	V-101 WATER REMOVAL 2-PHASE SEPARATOR	E-102 FTR PREHEATER U-TUBE HEAT EX.	R-102 FTR TUBULAR FIXED BED REACTOR	E-103 FTR PRODUCT COOLER TEFLON TUBE HEAT EX.	(HC) 2-PHASE	E-104 LIGHT HC COOLER TEFLON TUBE HEAT EX.	V-103 LIGHT HC 3-PHASE SEPARATOR	E-105 HEAVY HC HEATER TEFLON TUBE HEAT EX.	V-104 HEAVY HC 3-PHASE SEPARATOR	E-106 LPG COOLER TEFLON TUBE HEAT EX.	
Oxygen	39.95       75.0       514.7       43.34       100       514.7       2       43.34       400       514.7       100       514.7       100       514.7       100       514.7       100       514.7       100       514.7       120       121       406       1212       406.7       9	E-101 V-101	6		E-102		14 78.87 450 472.7	cw + CW	24.10 190 485.7			21 V-104	

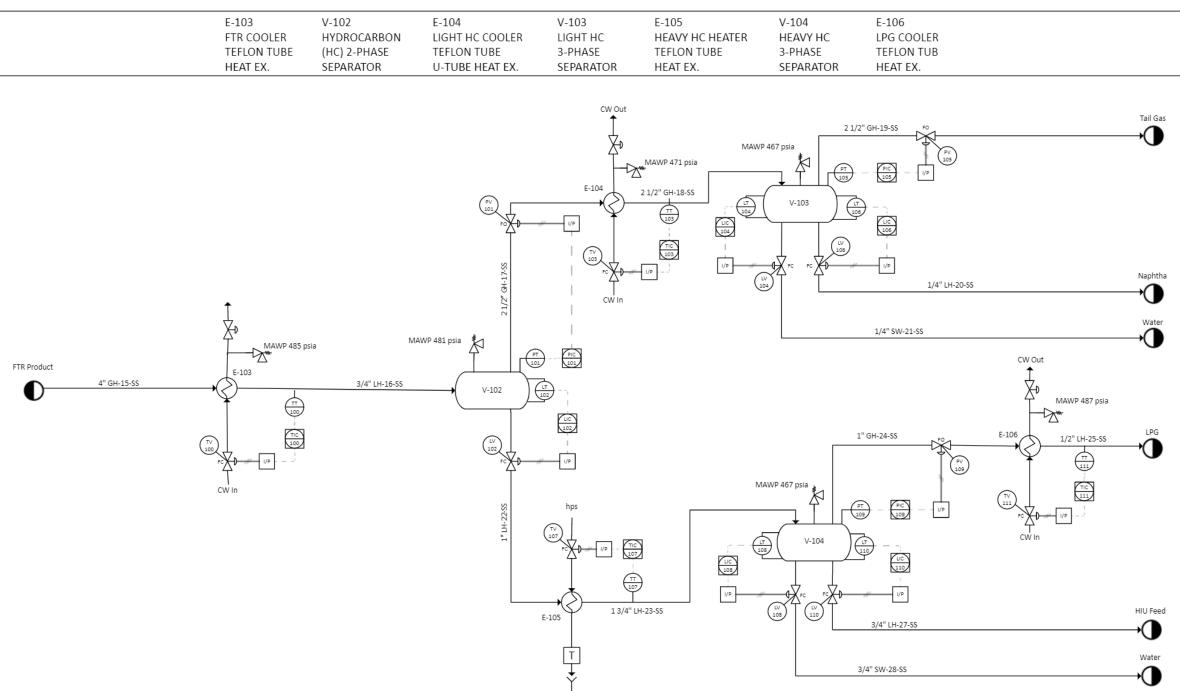


### Large Unit Process Flow Diagram



# Medium Unit P&ID

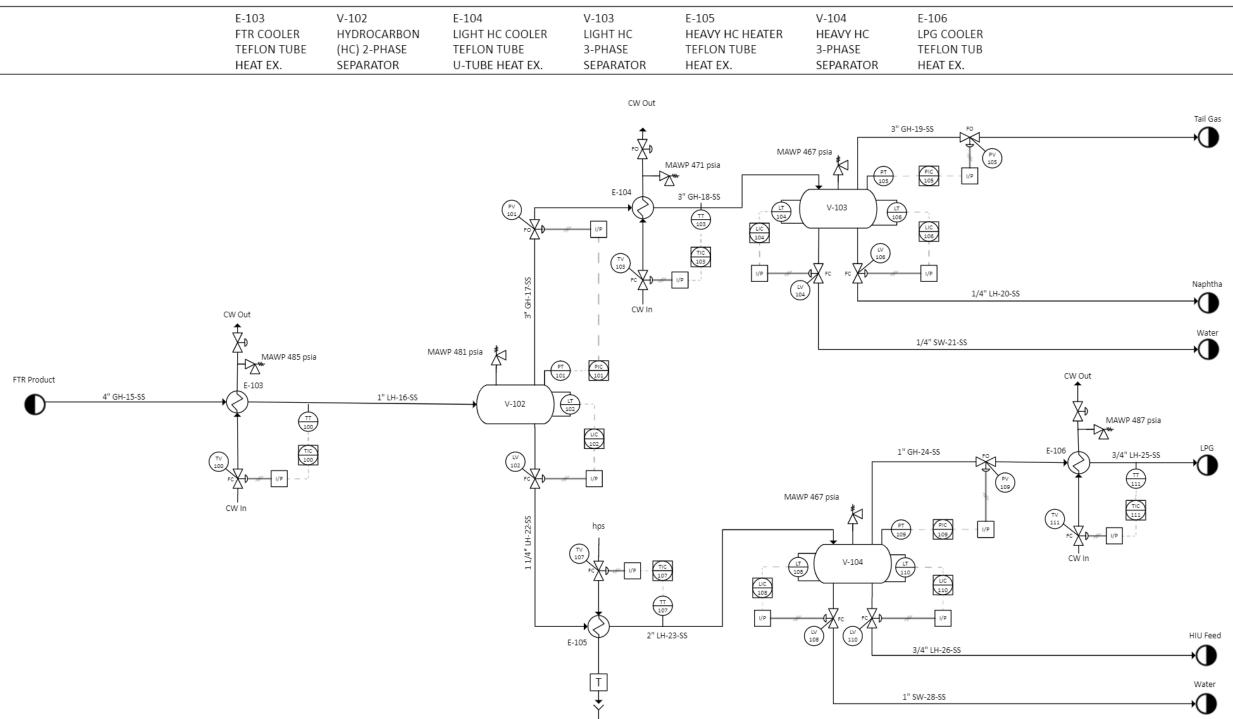
E-103	V-102	E-104	V-103	E-105	V-104	E-106
FTR COOLER	HYDROCARBON	LIGHT HC COOLER	LIGHT HC	HEAVY HC HEATER	HEAVY HC	LPG COOLER
TEFLON TUBE	(HC) 2-PHASE	TEFLON TUBE	3-PHASE	TEFLON TUBE	3-PHASE	TEFLON TUB
HEAT EX.	SEPARATOR	U-TUBE HEAT EX.	SEPARATOR	HEAT EX.	SEPARATOR	HEAT EX.



### Appendix xli

# Large Unit P&ID

E-103	V-102	E-104	V-103	E-105	V-104	E-106
FTR COOLER	HYDROCARBON	LIGHT HC COOLER	LIGHT HC	HEAVY HC HEATER	HEAVY HC	LPG COOLE
TEFLON TUBE	(HC) 2-PHASE	TEFLON TUBE	3-PHASE	TEFLON TUBE	3-PHASE	TEFLON TU
HEAT EX.	SEPARATOR	U-TUBE HEAT EX.	SEPARATOR	HEAT EX.	SEPARATOR	HEAT EX.



Appendix xlii

### Syngas Reactions

		Reaction 1 (Equili	brium)	
	CH4	H2O	CO	H2
Stoich	1	1	1	3
		Reaction 2 (Comp	letion)	
	CH4	02	CO	H2O
Stoich	1	1.5	1	2
		Reaction 3 (Equili	brium)	
	CO	H2O	CO2	H2
Stoich	1	1	1	1

### Flow Rate of Natural Gas into the System

	Molar Flow of Meth	ane Into Syngas Ur	nit			CO2	Steam	02	CO2+Steam+O2	Total
MSCF/day	lbmole/hr	lb/hour	kg/hr	kg/day	kg/s	kg/hr	kg/hr	kg/hr	kg/s	kg/s
500	54.0211472	864.3383552	392	9409.4	0.108905	219.6	220.7	304.4	0.206861111	0.315766
2500	270.105736	4321.691776	1960	47046.8	0.544524	898.3	1104	1812	1.059527778	1.604051
5000	540.211472	8643.383552	3921	94093.7	1.089047	2096	2207	3175	2.077222222	3.166269

### Capital Cost of Syngas Reactor

Capital	Capital Cost of Unit										
Ca	21620000	\$									
scaling factor	0.67										
Base Capacity	12.2	kg/s									
Small Unit	1868827.58	\$									
Medium Unit	5552529.06	\$									
Large Unit	8757127.37	\$									

### Utility Cost of Syngas Reactor

Tot	tal Utilty Cost of Sy	yngas
Small	14.0496415	\$/hr
Medium	72.32035452	\$/hr
Large	141.277803	\$/hr

### Cost of Steam into Syngas Reactor

Reactant Cost		
	Steam (kg/hr)	lb/hr
Small	220.7	486.5606095
Medium	1104	2433.90536
Large	2207	4865.606095

Small Unit		
Sman Omt		
	1211000	
Total RequiredDuty	1211000	-
Total Actual Duty	1424705.88	Btu/hr
Duty from Fire	1139764.71	Btu/hr
	1.13976471	MBtu/hr
Cost from Fire	3.41929412	\$/hr
Duty from Electricity	142470.588	Btu/hr
	41.7539978	kW
Cost of Electricity	1.67015991	\$/hr
Duty from 125 psig Steam	142470.588	Btu/hr
mass flowrate of 125psig stear	139.267437	lb/hr
	0.13926744	klb/hr
Cost of 125 psig Steam	0.55706975	\$/hr
Total PreHeater Utility	5.64652378	\$/hr
Cooler		
Duty	2480000	Btu/hr
mass flowrate of PCW	82749.4161	lb/hr
	9.92199234	kgal/hr
Utility Cost of Cooler	4.96099617	\$/hr
Small Unit		
Duty	217800	Btu/hr
mass flowrate of HP Steam	299.175824	lb/hr
	0.29917582	-
Cost of HP Steam	1.49587912	\$/hr

# Utilities of Small Unit Syngas Reactor

Medium Unit		
	5950000	
Total RequiredDuty	5950000	
Total Actual Duty	7000000	
Duty from Fire	5600000	
	5.6	MBtu/hr
Cost from Fire	16.8	\$/hr
Duty from Electricity	700000	Btu/hr
	205.1497	kW
Cost of Electricity	8.205988	\$/hr
Duty from 125 psig Steam	700000	Btu/hr
mass flowrate of 125psig stear	684.261975	lb/hr
	0.68426197	klb/hr
Cost of 125 psig Steam	2.7370479	\$/hr
Total Preheater Utility	27.7430359	\$/hr
Cooler		
Duty	13390000	Btu/hr
mass flowrate of PCW	446780.113	lb/hr
	53.570757	kgal/hr
Utility Cost of Cooler	26.7853785	\$/hr
Medium Unit		
Duty	1173000	Btu/hr
mass flowrate of HP Steam	1611.26374	lb/hr
	1.61126374	
Cost of HP Steam	8.05631868	\$/hr

# Utilities of Medium Unit Syngas Reactor

		[]
Large Unit		
	12050000	
Total RequiredDuty	12050000	Btu/hr
Total Actual Duty	14176470.6	Btu/hr
Duty from Fire	11341176.5	Btu/hr
	11.3411765	MBtu/hr
Cost from Fire	34.0235294	\$/hr
Duty from Electricity	1417647.06	Btu/hr
	415.471241	kW
Cost of Electricity	16.6188496	\$/hr
Duty from 125 psig Steam	1417647.06	Btu/hr
mass flowrate of 125psig stear	1385.77425	lb/hr
	1.38577425	klb/hr
Cost of 125 psig Steam	5.543097	\$/hr
Total PreHeater Utility	56.1854761	\$/hr
Cooler		
Duty	25200000	Btu/hr
mass flowrate of PCW	840840.841	lb/hr
	100.820245	kgal/hr
Utility Cost of Cooler	50.4101224	\$/hr
Large Unit		
Duty	2216000	Btu/hr
mass flowrate of HP Steam	3043.95604	lb/hr
	3.04395604	
Cost of HP Steam	15.2197802	\$/hr

# Utilities of Large Unit Syngas Reactor

# CO2 Recovery Capital Cost

CO2 Recovery Captital Cost					
Base Cost	5300000	\$			
Base Capacity	8.54	kg/s			
scaling factor	0.55				
	CO2 kg/hr	CO2 kg/s	\$		
Small Unit	249.4	0.069278	375232.7		
Medium Unit	1093	0.303611	845763.1		
Large Unit	2417	0.671389	1308609		

Product Selectiv	/ity	450	(F)	25-C	25	0.019830378	0.007867067
	Ť	505.2222222		26-C	26	0.018544172	0.007073852
	T1	1.832519396	(К)	27-C	27	0.017315736	0.006360614
	T2	0.329341661	. ,	28-C	28	0.016146498	0.005719291
	T3	3.851176787		29-C	29	0.015037008	0.00514263
	-			30-C	30	0.013987104	0.004624113
	T4	0.966852364	(K)	31-C	31	0.012996049	0.004157876
	alpha	0.899172699		32-C	32	0.01206265	0.003738648
				33-C	33	0.011185356	0.003361691
Hydrocarbon	#C	Wn	Mn	34-C	34	0.010362342	0.00302274
Pentane	5			35-C	35	0.00959158	0.002717966
Hexane	6			36-C	36	0.008870901	0.00244392
Heptane	7	0.037610837		37-C	37	0.00819804	0.002197507
Octane	8			38-C	38	0.007570682	0.001975938
Nonane	9	0.039097023		39-C	39	0.006986492	0.001776709
Decane	10			40-C	40	0.006443141	0.001597569
11-C	11			41-C	41	0.005938334	0.00143649
12-C	12			42-C	42	0.005469822	0.001291653
13-C	13			43-C	43	0.005035417	0.001161419
14-C	14			44-C	44	0.004633005	0.001044316
15-C	15			45-C	45	0.004260551	0.000939021
16-C	16			46-C	46		
17-C	17	0.031556948	0.0184106	47-C	47	0.003597802	0.000759209
18-C	18			48-C+	48		0.00068266
19-C	19				Sum	0.88177875	0.647602988
20-C	20		0.01338435		Jun	0.00177075	0.352397012
21-C	21	0.010.01100		C21_C25	C26 C20	C20 C25	
22-C	22	0.024004039		C21-C25	C26-C29	C30-C35	C36-C47
23-C	23			0.049202846	0.024296	0.007342078	0.017468092
24-C	24	0.021171865	0.008749228				

# Anderson-Schulz-Flory Probability Distribution

# Levenspiel Plot Data

	Graph ind	epentent o	of tempera	ture
Х	Rate of CC	FCO0/rCO	PCO (atm)	PH2 (atm)
0	3.269415	0.717898	7.154	14.308
0.1	3.0043	0.781249	6.754	13.464
0.2	2.714725	0.864584	6.313	12.532
0.3	2.398904	0.978408	5.825	11.499
0.4	2.054881	1.142211	5.28	10.348
0.5	1.682093	1.395349	4.668	9.056
0.6	1.283392	1.828831	3.976	7.596
0.7	0.868768	2.701651	3.189	5.933
0.8	0.463055	5.068749	2.285	4.022
0.9	0.125686	18.67434	1.235	1.803

### Flow Rate into Small Unit FTR

	Small Unit		Number of Tubes	14
Out o	of Syngas Reactor		Flow rate per cha	nnel
Molar Flow Overall	1.0000	122.97	8.7838	
Molar Fraction H2O	0.0373	4.59	0.3275	
Molar Fraction Nitrogen	0.0017	0.21	0.0150	
Molar Fraction CO	0.2672	32.86	2.3471	
Molar Fraction CO2	0.0053	0.66	0.0470	
Molar Fraction Hydrogen	0.5340	65.67	4.6906	
Molar Fraction Methane	0.1544	18.99	1.3566	

### Variables for Small Unit FTR

Diamter of each tube	0.1667	ft
Number of tubes	14	
Overall minimum diameter	1	ft
Length	3	ft
Volume	2.35619449	ft3
Weight of catalyst	2.505	lb
Overall pressure drop	0.0001357	psi
Conversion	0.94	

### Mass Balance of Small FTR

Product Flow Rates from FTR	Dolotivo Mess Fre	Molor Flow for and table	Total Malar Flam		Total Inlat Mass Flow
		Molar Flow for each tube		Mass Flows	Total Inlet Mass Flow
Molar Flow Overall	1475	0.0102204	1475	1586.258691	1475
FC2		0.0102284		4.305951832	
FC3		0.0102284		6.31501416	Carbon Flow Rate
FC4		0.0102284		8.322644512	635.3965
FCH4		1.61231		362.0603336	
FCO		0.1338822	1.8743508	52.50056591	Percent error
FCO2		0.0469714	0.6575996	28.9409584	-7.542962074
FH2		0.130683	1.829562	1.84785762	
FH2O		2.543261	35.605654	641.6138851	
FN2		0.015	0.21	2.9421	
FC5	0.033227582			21.11269024	
FC6	0.035852801			22.7807456	
FC7	0.037610837			23.89779525	
FC8	0.038649872			24.55799435	
FC9	0.039097023			24.84211282	
FC10	0.039061084			24.81927736	
FC11	0.038634927			24.54849827	
FC12	0.037897605			24.08000667	
FC13	0.036916199			23.45642497	
FC14	0.035747426			22.71379056	
FC15	0.034439046			21.88245038	
FC16	0.033031093			20.9878421	
FC17	0.031556948			20.05117554	
FC18	0.030044273			19.09002666	
FC19	0.028515823			18.11885473	
FC20	0.026990157			17.14945211	
FC21	0.025482253			16.1913351	
FC22	0.024004039			15.25208298	
FC23	0.022564857			14.33763192	
FC24	0.021171865			13.45252924	
FC25	0.019830378			12.60015315	
FC26	0.018544172			11.78290226	
FC27	0.017315736			11.0023588	
FC28	0.016146498			10.25942883	
FC29	0.015037008			9.554462531	
FC30	0.013987104			8.887357097	

#### Flow Rate into Medium Unit FTR

	Medium Unit		Number of Tubes	29
Out	of Syngas Reactor			Flow rate per channel
Molar Flow Overall	1.0000	679.94		23.4461
Molar Fraction H2O	0.0365	24.84		0.8564
Molar Fraction Nitrogen	0.0018	1.25		0.0431
Molar Fraction CO	0.2894	196.76		6.7847
Molar Fraction CO2	0.0042	2.88		0.0994
Molar Fraction Hydrogen	0.5788	393.55		13.5707
Molar Fraction Methane	0.0892	60.66		2.0918

### Variables for Medium Unit FTR

Diamter of each tube	0.1667	ft
Number of tubes	29	
Overall minimum diameter	1.25	ft
Length	3.75	ft
Volume	4.601942364	ft3
Weight of catalyst	3.6098	lb
Overall pressure drop	0.0022624	psi
Conversion	0.9009	

Mass Balance of FTR

	Relative Mass Fra	Molar Flow for each tube	Total Molar Flow	Mass Flows	Total Inlet	Mass Flow
Molar Flow Overall (Ibmol/hr			7887		7887	
FC2		0.0282493	0.8192297	24.63423708		
FC3		0.0282493	0.8192297	36.12802977	Product Fl	ow Rate
FC4		0.0282493	0.8192297	47.61363016	3631.211	
FCH4		2.798032	81.142928	1301.532565		
FCO		0.6720297	19.4888613	545.883005	Percent er	ror
FCO2		0.099369	2.881701	126.823661	-7.76	
FH2		0.9785495	28.3779355	28.66171486		
FH2O		6.969094	202.103726	3641.909143		
FN2		0.0431	1.2499	17.511099		
FC5	0.033227582			120.6563512		
FC6	0.035852801			130.1890764		
FC7	0.037610837			136.5728737		
FC8	0.038649872			140.3458279		
FC9	0.039097023			141.969529		
FC10	0.039061084			141.8390273		
FC11	0.038634927			140.2915591		
FC12	0.037897605			137.6141889		
FC13	0.036916199			134.0504984		
FC14	0.035747426			129.8064368		
FC15	0.034439046			125.055433		
FC16	0.033031093			119.9428599		
FC17	0.031556948			114.5899291		
FC18	0.030044273			109.097085		
FC19	0.028515823			103.5469604		
FC20	0.026990157			98.00694721		
FC21	0.025482253			92.53142979		
FC22	0.024004039			87.16372286		
FC23	0.022564857			81.93775084		
FC24	0.021171865			76.87950111		
FC25	0.019830378			72.00827969		
FC26	0.018544172			67.33779436		
FC27	0.017315736			62.8770873		
FC28	0.016146498			58.6313366		
FC29	0.015037008			54.60254349		
FC30	0.013987104			50.79012041		

# Flow Rate into Large Unit FTR

	Number of Tubes	49			
Out o	f Syngas Reactor			Flow rate p	per channel
Molar Flow Overall	1.0000	1259.5600		25.7053	
Molar Fraction H2O	0.0371	46.7159		0.9534	
Molar Fraction Nitrogen	0.0017	2.1899		0.0447	
Molar Fraction CO	0.2726	343.3128		7.0064	
Molar Fraction CO2	0.0051	6.3726		0.1301	
Molar Fraction Hydrogen	0.5452	686.6826		14.0139	
Molar Fraction Methane	0.1384	174.2847		3.5568	

### Variables for Large Unit FTR

Diamter of each tube	0.1667	ft
Number of tubes	49	
Overall minimum diameter	1.5	ft
Length	4.5	ft
Volume	7.952156404	ft3
Weight of catalyst	4.4293	lb
Overall pressure drop	0.003285	psi
Conversion	0.9026	

Mass Balance of FTR

Product Flow Rates from FTR						
	<b>Relative Mas</b>	Molar Flow for each tube	Total Molar Flow		Mass Flows	Total Mass Flow Rate
Molar Flow Overall (Ibmol/hr	14980		14980		15207.44	14980.2767
FC2		0.0292248	1.4320152	FC2	43.0607	
FC3		0.0292248	1.4320152	FC3	63.15187	<b>Product Flow Rate</b>
FC4		0.0292248	1.4320152	FC4	83.22872	6347.16191
FCH4		4.28742	210.08358	FCH4	3369.741	
FCO		0.06826033	3.34475617	FCO	93.68662	Percent error
FCO2		0.1300531	6.3726019	FCO2	280.4582	-1.5163888
FH2		0.9869124	48.3587076	FH2	48.84229	
FH2O		7.27718	356.58182	FH2O	6425.604	
FN2		0.0446918	2.1898982	FN2	30.68047	
FC5	0.033227582			FC5	210.9008	
FC6	0.035852801			FC6	227.5635	
FC7	0.037610837			FC7	238.7221	
FC8	0.038649872			FC8	245.317	
FC9	0.039097023			FC9	248.1551	
FC10	0.039061084			FC10	247.927	
FC11	0.038634927			FC11	245.2221	
FC12	0.037897605			FC12	240.5422	
FC13	0.036916199			FC13	234.3131	
FC14	0.035747426			FC14	226.8947	
FC15	0.034439046			FC15	218.5902	
FC16	0.033031093			FC16	209.6537	
FC17	0.031556948			FC17	200.2971	
FC18	0.030044273			FC18	190.6959	
FC19	0.028515823			FC19	180.9945	
FC20	0.026990157			FC20	171.3109	
FC21	0.025482253			FC21	161.74	
FC22	0.024004039			FC22	152.3575	
FC23	0.022564857			FC23	143.2228	
FC24	0.021171865			FC24	134.3813	
FC25	0.019830378			FC25	125.8666	
FC26	0.018544172			FC26	117.7029	
FC27	0.017315736			FC27	109.9058	
FC28	0.016146498			FC28	102.4844	
FC29	0.015037008			FC29	95.44232	
FC30	0.013987104			FC30	88.77841	

# Sensitivity Analysis for Tornado Chart

	Transportation	Revenue	Utility Cost
Worst NPV	4340888.263	4301423.444	3830297.08
Best NPV	4347010.557	4357711.62	4403804.644

#### Monte Carlo Variables

Probability	Frequency	NPV
0.086797011	1940	1086713
0.171938616	3843	1738621
0.25520111	5704	2390529
0.337971455	7554	3042437
0.420965505	9409	3694345
0.504541184	11277	4346253
0.58668516	13113	4998161
0.671513579	15009	5650068
0.754418147	16862	6301976
0.835577826	18676	6953884
0.920316764	20570	7605792
0.999910519	22349	8257700

# Small Unit FTR Sizing

				SMALL	UNIT			
		FTR Sizing						
A (ft^2)	L	D	n	A (m^2)				
0.785398163	2.5	1	14	0.072965877				
				Utilite	s			
Steam Flow (Ibmole/hr)	Steam Flow Ib/hr	Steam Flow klb/hr	Cost of Steam/hour	Cost of Steam/year	P/A .08, 20	PWC of Utilities	F/A .08, 20	FWC
	2505.94179	2.50594179	\$10.02	\$70,246.56	9.818147407	\$689,691.08	45.7619643	\$3,214,620.58
	Cataly	st Cost			Total Cost	\$549.58		
Price of C	Catalyst (\$/lb)	Amount (lb)	Cost (\$)					
	10	2.505	25.05					

Purchase Cost						
C <sub>p</sub>	192.3849314					
	Purchase Cost of E	Base Conditions				
	$C_p^0$	109.2763864				
		<b>K</b> <sub>1</sub>	2.7652	Range 10-100 m <sup>2</sup>		
		K <sub>2</sub>	0.7282	Assumed MultiPipe **		
		K <sub>3</sub>	0.0783			
		$A(m^2)$	0.07296588	Area of Reboiler		
	Pressure Factor					
	F <sub>p</sub>	0.978075145				
		P (barg)	33.6734694	Design Pressure		
		C <sub>1</sub>	0.6072	Pressure Range:		
		C <sub>2</sub>	-0.912	40 < P < 100 barg		
		C <sub>3</sub>	0.3327			
	Material of Constru		tor			
	F <sub>m</sub>	1.8		1- A 2/E A 19		
		Assumed 55-She.	wss-tube, Tab	ek A.3/Figure A.18		
Installed Cost						
C <sub>BM, Heat Exchanger</sub>	\$524.53					
	<b>B</b> <sub>1</sub>	1.74	Heat Exchang	er Multiple Pipe, Table A	A.4	
	B <sub>2</sub>	1.55				

### Purchasing Cost of the Small Unit

### Medium Unit FTR Sizing

				MEDI	UM UNIT			
				FT	R Sizing			
A (ft^2)	L	D	n	A (m^2)				
1.227185	2.5	1.25	14	0.114009183				
				U	tilites			
Steam Flow (Ibmole/ hr)	Steam Flow Ib/hr	Steam Flow klb/hr	Cost of Steam/hour	Cost of Steam/year	P/A .08, 20	PWC of Utilities	F/A .08, 20	FWC
	14336.66283	14.33666283	\$57.35	\$502,356.67	9.818147407	\$4,932,211.79	45.7619643	\$22,988,827.79
			Cataly	st Cost	-		Total Cost	\$711.16
		Price of Cat	alyst (\$/lb)	Amount (lb)	Cost (\$)			
			10	3.6098	36.098			

Purchase Cost						
C <sub>p</sub>	247.9576053					
	Purchase Cost of B	ase Conditions				
	$C_p^{0}$	140.637348				
		<b>K</b> <sub>1</sub>	2.7652	Range 10-100 m <sup>2</sup>		
		K <sub>2</sub>	0.7282	Assumed MultiPipe **		
		K <sub>3</sub>	0.0783			
		$A(m^2)$	0.11400918	Area of Reboiler		
	Pressure Factor					
	F <sub>p</sub>	0.979499593				
		P (barg)	34.137931	Design Pressure		
		<b>C</b> <sub>1</sub>	0.6072	Pressure Range:		
		C <sub>2</sub>	-0.912	40 < P < 100 barg		
		C <sub>3</sub>	0.3327			
	Matarial of Co					
	Material of Constru	1.8 1.8	tor			
	F <sub>m</sub>		II/SS tube Teb	le A.3/Figure A.18		
			<u>140</u>	R. 7.17 Iguie A. 18		
Installed Cost						
C <sub>BM, Heat Exchanger</sub>	\$675.06					
	B <sub>1</sub>	1.74	Heat Exchang	er Multiple Pipe, Table A	A.4	
	B <sub>2</sub>	1.55				

# Purchasing Cost of the Medium Unit

#### Large Unit FTR Sizing

				LAR	GE UNIT				
	FTR Sizing								
A (ft^2)	L	D	n	A (m^2)					
1.767146	4.5	1.5	14	0.164173223					
				U	tilites				
Steam Flow (Ibmole/ hr)	Steam Flow Ib/hr	Steam Flow klb/hr	Cost of Steam/hour	Cost of Steam/year	P/A .08, 20	PWC of Utilities	F/A .08, 20	FWC	
	16813.52112	16.81352112	\$ 67.25	\$589,145.78	9.818147407	\$ 5,784,320.11	45.7619643	\$ 26,960,468.16	
	Price of Cata	Catalyst Co		Cost (\$)		Total Cost	\$ 882.28		
	The of cata	10							

Purcha	se Cost				
C <sub>p</sub>	307.8032334				
	Purchase Cost of Bas	e Conditions			
	$C_p^{0}$	174.5807732			
		$\mathbf{K}_1$	2.7652	Range 10-100 m <sup>2</sup>	
		K <sub>2</sub>	0.7282	Assumed MultiPipe **	
		K <sub>3</sub>	0.0783		
		$A(m^2)$	0.164173223	Area of Reboiler	
	Pressure Factor				
	F <sub>p</sub>	0.979499593			
		P (barg)	34.13793103	Design Pressure	
		C <sub>1</sub>	0.6072	Pressure Range:	
		C <sub>2</sub>	-0.912	40 < P < 100 barg	
		C <sub>3</sub>	0.3327		
	Material of Construct	ion (MOC) Factor			
	F <sub>m</sub>	1.8			
		Assumed SS-Shel	VSS-tube, Table A	.3/Figure A.18	
Installe	d Cost				
И, Heat E					
	B <sub>1</sub>	1.74	Heat Exchanger M	Iultiple Pipe, Table A.4	
	B <sub>2</sub>	1.55			

#### Purchasing Cost of the Large Unit

Capital	Cost	for	the	Small	Unit
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Small Unit				
	Capital Cost			
Syngas	\$1,868,827.58			
CO2 Recovery Syster	\$375,232.73			
FTR	\$549.58			
Separation	\$102,977.17			
Air Separation				
HIU	\$10,061,097.19			
Total	\$12,408,684.26			
CEPCI 2001	397			
CEPCI 2022	776.9			
Scaled Capital Cost	\$13,308,161.69			
Total Capital Invest.	\$13,308,161.69			
Capital Cost	t in 2024			
	Capital Cost			
Syngas	\$1,982,639.18			
CO2 Recovery Syster	\$398,084.40			
HIU	\$10,673,818.01			

# Capital Cost for the Medium Unit

Medium Unit				
	Capital Cost			
Syngas	\$5,552,529.06			
CO2 Recovery Syste	\$845,763.14			
FTR	\$711.16			
Separation	\$234,346.19			
Air Sepration				
HIU	\$27,905,278.66			
Total	\$34,538,628.21			
CEPCI 2001	397			
CEPCI 2022	776.9			
Scaled Capital Cost	\$36,968,502.40			
Total Capital Invest.	\$36,968,502.40			
Capital Co	st in 2024			
	Capital Cost			
Syngas	\$5,890,678.07			
CO2 Recovery Syste	\$897,270.12			
HIU	\$29,604,710.13			

Large Unit					
	Capital Cost				
Syngas	\$8,757,127.37				
CO2 Recovery Syste	\$1,308,608.76				
FTR	\$882.28				
Separation	\$313,299.68				
Air Separation Unit					
HIU	\$36,802,218.56				
Total	\$47,182,136.65				
CEPCI 2001	397				
CEPCI 2022	776.9				
Scaled Capital Cost	\$50,491,896.88				
Total Capital Invest.	\$50,491,896.88				
Capital Cos	st in 2024				
	Capital Cost				
Syngas	9290436.428				
CO2 Recovery Syste	1388303.033				
HIU	39043473.67				

Capital Cost for the Large Unit

# HIU Costing

Co			
Base HIU	ase HIU 8290000 \$		
Base Capacity	1.13	kg/s	
scaling factor 0.55			
	kg/s	\$	bbl/day of feed
Small Unit	1.606833	10061097	873.2175067
Medium Unit	10.26833	27905279	5580.223067
Large Unit	16.98333	36802219	9229.422667

Oxygen									
	O2 mole flowrate (lbrlb/hour short ton/hr				\$/yr				
Small Unit	21	669.48	0.33474	33.474	234585.8				
Medium Unit	125	3985	1.9925	199.25	1396344				
Large Unit	219	6981.72	3.49086	349.086	2446395				
	Electricity Option								
	Short Ton O2/hr	kWh/hr	\$/hr						
Small Unit	0.33474	3347.4	100.422						
Medium Unit	1.9925	19925	597.75						
Large Unit	3.49086	34908.6	1047.258						
	Stear	n Option		<u> </u>					
	Short Ton O2/hr	lb 600 psig stea	klb/hr	\$/hr	\$/yr				
Small Unit	0.33474	3347.4	3.3474	16.737	117292.9				
Medium Unit	1.9925	19925	19.925	99.625	698172				
Large Unit	3.49086	34908.6	34.9086	174.543	1223197				
	Cooling Water Requirement								
	Short Ton O2/hr	Short Ton O2/c		\$/hr	\$/yr				
Small Unit	0.33474	8.03376	3213.504						
Medium Unit	1.9925	47.82	19128	573.84	4021471				
Large Unit	3.49086	83.78064	33512.256	1005.36768	7045617				

Product Formation of Small HIU

Product Formation of Small HIU						
Component	Inlet (lb/hr)		Product (lb/hr)	Outlet (lb/hr)		
Methane	0.236		0.181165	0.417165		
Ethane	0.0106		0.0905825	0.1011825		
Propane	0.046		0.6340775	0.6800775		
Butane	0.1622		0.6340775	0.7962775		
Naphtha	68.7368		4.529125	73.265925		
Diesel	120.737		12.0474725	132.7844725		
C-11+	18.1165		0	0		
Oxygen	0			0		
Nitrogen	0.0007			0.0007		
Hydrogen	0.0005			0.0005		
Carbon Monoxide	0.0178			0.0178		
Carbon Dioxide	0.0457			0.0457		
Water	4.4395			4.4395		

Product Formation of Medium HIU						
Component Inlet (Ib/hr) Product (Ib/hr) C				Outlet (lb/hr)		
Methane	1.1336		1.07638	2.20998		
Ethane	0.0861		0.53819	0.62429		
Propane	0.3763		3.76733	4.14363		
Butane	1.3347		3.76733	5.10203		
Naphtha	488.163		26.9095	515.0725		
Diesel	729.709		71.57927	801.28827		
C-11+	107.638		0	0		
Oxygen	0			0		
Nitrogen	0.0057			0.0057		
Hydrogen	0.0041			0.0041		
Carbon Monoxide	0.1505			0.1505		
Carbon Dioxide	0.2641			0.2641		
Water	29.3745			29.3745		

#### Product Formation of Medium HIU

Product Formation of Large HIU

	Product Formation of Large HIU									
Component	Inlet (Ib/hr)		Product (lb/hr)	Outlet (lb/hr)						
Methane	2.354		1.88189	4.23589						
Ethane	0.1183		0.940945	1.059245						
Propane	0.5128		6.586615	7.099415						
Butane	1.8104		6.586615	8.397015						
Naphtha	744.533		47.04725	791.58025						
Diesel	1260.3		125.145685	1385.445685						
C-11+	188.189		0	0						
Oxygen	0			0						
Nitrogen	0.008			0.008						
Hydrogen	0.0056			0.0056						
Carbon Monoxide	0.2087			0.2087						
Carbon Dioxide	0.4714			0.4714						
Water	47.2738			47.2738						

	ι	Jtilities of	HIU for Sm	all Unit						
Utilities			Co	ost						
Fuel Gas	0.08	MBTU/bbl	209.5722	\$/day	8.732175	\$/hr				
Hydrogen	300	SCF/bbl	82.20	\$/day	3.425196	\$/hr				
MP Steam	10	lb/bbl	34.9287	\$/day	1.455363	\$/hr				
Electricity	2.5	kWh/bbl	87.32175	\$/day	3.638406	\$/hr				
Cooling Wate	300	gal/bbl	130.9826	\$/day	5.457609	\$/hr				
Utilities of HIU for Medium Unit										
Utilities	Cost									
Fuel Gas	0.08	MBTU/bbl	1339.254	\$/day	55.80223	\$/hr				
Hydrogen	300	SCF/bbl	525.32	\$/day	21.88842	\$/hr				
MP Steam	10	lb/bbl	223.2089	\$/day	9.300372	\$/hr				
Electricity	2.5	kWh/bbl	558.0223	\$/day	23.25093	\$/hr				
Cooling Wate	300	gal/bbl	837.0335	\$/day	34.87639	\$/hr				
	ι	Jtilities of	HIU for Lar	ge Unit						
Utilities			Co	ost						
Fuel Gas	0.08	MBTU/bbl	2215.061	\$/day	92.29423	\$/hr				
Hydrogen	300	SCF/bbl	868.86	\$/day	36.20241	\$/hr				
MP Steam	10	lb/bbl	369.1769	\$/day	15.38237	\$/hr				
Electricity	2.5	kWh/bbl	922.9423	\$/day	38.45593	\$/hr				
Cooling Wate	300	gal/bbl	1384.413	\$/day	57.68389	\$/hr				

#### Utilities for Each Unit of the HIU

				SMA	ALL UNIT SEPA	RATOR SIZING			
Con	nstants	V-10	)2	Consta	nts	V-10	03	Consta	nts
P (psia)	481.00	Q <sub>v</sub> (ft <sup>3</sup> /s)	0.17	ρ <sub>v</sub> (lb/ft3)	1.28	Q <sub>v</sub> (ft <sup>3</sup> /s)	0.16	ρ <sub>v</sub>	1.78
ρ <sub>v</sub> (lb/ft3)	1.25	Q <sub>L</sub> (ft <sup>3</sup> /s)	0.01	ρ <sub>L</sub> (lb/ft3)	38.38	$Q_L(ft^3/s)$	0.00	ρι	33.44
ρ <sub>L</sub> (lb/ft3)	37.54	U <sub>T</sub> (ft/s)	1.55	ρ <sub>H</sub>	58.21	Q <sub>LL</sub> (ft <sup>3</sup> /s)	0.00	ρ <sub>H</sub>	52.39
к	0.29	U <sub>v</sub> (ft/s)	1.16	P (psia)	467.00	Q <sub>HL</sub> (ft <sup>3</sup> /s)	0.00	P (psia)	467.00
m <sub>v</sub> (lb/hr)	752.05	Т <sub>н</sub> (s)	120.00	μ <sub>L</sub> (cP)	0.25	U <sub>T</sub> (ft/s)	1.55	μ <sub>L</sub> (cP)	0.19
m <sub>∟</sub> (lb/hr)	783.41	T <sub>s</sub> (s)	60.00	μ <sub>н</sub> (сР)	0.23	U <sub>v</sub> (ft/s)	1.17	μ <sub>н</sub> (сР)	0.12
		V <sub>H</sub> (ft <sup>3</sup> )	0.70	к	0.29	T <sub>H</sub> (min)	120.00	К	0.29
H/D	to A/At	V <sub>s</sub> (ft <sup>3</sup> )	0.35	ks	0.16	T <sub>s</sub> (min)	60.00	ks	0.16
а	4.75593E-05	L/D (est.)	2.00	m <sub>v</sub> (lb/hr)	715.78	V <sub>H</sub> (ft <sup>3</sup> )	0.01	m <sub>v</sub> (lb/hr)	41.94
b	3.924091	D (ft)	1.50	m <sub>L</sub> (lb/hr)	10.45	V <sub>s</sub> (ft <sup>3</sup> )	0.00	m <sub>L</sub> (lb/hr)	212.55
с	0.174875	A <sub>T</sub> (ft <sup>2</sup> )	1.77	m <sub>⊔</sub> (Ib/hr)	10.45	L/D (est)	2.00	m <sub>LL</sub> (lb/hr)	212.55
d	-6.358805	H <sub>LLL</sub> (ft)	0.75	m <sub>HL</sub> (lb/hr)	25.83	D (ft)	0.50	m <sub>HL</sub> (lb/hr)	528.93
e	5.668973	H <sub>LLL</sub> /D	0.50			A <sub>T</sub> (ft <sup>2</sup> )	0.20		
f	4.018448	A <sub>LLL</sub> /A <sub>T</sub>	0.50			H <sub>v</sub> (ft)	0.10		
g	-4.916411	A <sub>LLL</sub> (ft <sup>2</sup> )	0.88			H <sub>v</sub> /D	0.20		
h	-1.801705	H <sub>v</sub> (ft)	0.30			A <sub>v</sub> /A <sub>t</sub>	0.14		
i	-0.145348	H <sub>v</sub> /D	0.20			A <sub>v</sub> (ft <sup>2</sup> )	0.03		
		A <sub>V</sub> /A <sub>T</sub>	0.14			H <sub>LLL</sub> (ft)	0.75		
A/At	t to H/D	A <sub>v</sub> (ft <sup>2</sup> )	0.25			H <sub>LLL</sub> /D	1.50		
а	0.00153756	L <sub>min</sub> (ft)	2.00			A <sub>LLL</sub> /A <sub>T</sub>	1.45		
b	26.787101	r (ft)	0.75			A <sub>LLL</sub> (ft <sup>2</sup> )	0.28		
с	3.299201	V (ft³)	3.53			H <sub>w</sub> (ft)	0.40		
d	-22.923932					L <sub>2</sub> (ft)	0.00		
e	24.353518					H <sub>w</sub> /2 (ft)	0.20		
f	-14.844824					H <sub>HL</sub> (ft)	0.20		
g	-36.999376					H <sub>LL</sub> (ft)	0.20		
h	10.529572					H <sub>HL</sub> /D	0.40		
i	9.892851					A <sub>HL</sub> /A <sub>T</sub>	0.37		
						A <sub>HL</sub> (ft <sup>2</sup> )	0.07		
						$A_{LL}$ (ft <sup>2</sup> )	0.10		
						U <sub>HL</sub> (ft/s)	12.78		
						U <sub>LH</sub> (ft/s)	14.19		
						t <sub>HL</sub> (s)	0.70		
						t <sub>LH</sub> (s)	0.17		
						L <sub>1</sub> (ft)	0.50		
						L min (ft)	0.50		
						φ (s)	0.09		
						U <sub>VA</sub> (ft/s)	5.56		
						L <sub>min</sub> (ft)	0.50		
						H <sub>HLL</sub> (ft)	0.40		
						r (ft) V (ft <sup>3</sup> )	0.25		
						v (it )	0.10		

# Small Unit Separator Sizing

١	/-104
Q <sub>v</sub> (ft <sup>3</sup> /s)	0.01
Q <sub>L</sub> (ft <sup>3</sup> /s)	0.00
Q <sub>LL</sub> (ft <sup>3</sup> /s)	0.00
Q <sub>HL</sub> (ft <sup>3</sup> /s)	0.00
U <sub>T</sub> (ft/s)	1.22
U <sub>v</sub> (ft/s)	0.91
T <sub>H</sub> (min)	120.00
T <sub>s</sub> (min)	60.00
V <sub>H</sub> (ft <sup>3</sup> )	0.21
V <sub>s</sub> (ft <sup>3</sup> )	0.11
L/D (est)	2.00
D (ft)	1.50
A <sub>T</sub> (ft <sup>2</sup> )	1.77
H <sub>v</sub> (ft)	0.30
H <sub>v</sub> /D	0.20
A <sub>v</sub> /A <sub>T</sub>	0.14
A <sub>v</sub> (ft <sup>2</sup> )	0.25
H <sub>LLL</sub> (ft)	0.75
H <sub>LLL</sub> /D	0.50
A <sub>LLL</sub> /A <sub>T</sub>	0.50
A <sub>LLL</sub> (ft <sup>2</sup> )	0.88
H <sub>w</sub> (ft)	1.20
L <sub>2</sub> (ft)	1.00
H <sub>w</sub> /2 (ft)	0.60
H <sub>HL</sub> (ft)	0.60
H <sub>LL</sub> (ft)	0.60
H <sub>HL</sub> /D	0.40
A <sub>HL</sub> /A <sub>T</sub>	0.37
A <sub>HL</sub> (ft <sup>2</sup> )	0.66
A <sub>LL</sub> (ft <sup>2</sup> )	0.86
U <sub>HL</sub> (ft/s)	15.88
U <sub>LH</sub> (ft/s)	24.93
t <sub>HL</sub> (s)	0.57
t <sub>LH</sub> (s)	0.29
L <sub>1</sub> (ft)	0.50
Lmin (ft)	1.50
φ (s)	0.33
U <sub>VA</sub> (ft/s)	0.03
L <sub>min</sub> (ft)	0.50
H <sub>HLL</sub> (ft)	1.20
r (ft) V (ft <sup>3</sup> )	0.75
v (ft )	2.65

	S	MALL UNIT SEPAR	ATOR COSTING		
	V-102		V-103	V-104	
К1	3.5565	К1	3.5565	К1	3.5565
К2	0.3776	К2	0.3776	К2	0.3776
К3	0.0905	КЗ	0.0905	КЗ	0.0905
V (ft <sup>3</sup> )	3.534291735	V (ft³)	0.09817477	V (ft <sup>3</sup> )	2.6507188
V (m³)	0.100079052	V (m <sup>3</sup> )	0.002779974	V (m³)	0.07505929
Cp°	\$1,859.81	Cp°	\$1,522.65	Cp°	\$1,763.23
P (psi)	481	P (psi)	467	P (psi)	467
P (bar)	33.16326531	P (bar)	32.19801434	P (bar)	32.1980143
D (ft)	1.5	D (ft)	0.5	D (ft)	1.5
D (m)	0.45717769	D (m)	0.152392563	D (m)	0.45717769
Fp	1.949573655	Fp	0.968800365	Fp	1.9064011
Fm	3.1	Fm	3.1	Fm	3.1
B1	1.49	B1	1.49	B1	1.49
B2	1.52	B2	1.52	B2	1.52
Cbm <sub>0</sub>	\$8,927.07	Cbm <sub>o</sub>	\$7,308.71	Cbm <sub>0</sub>	\$8,463.51
		Total	\$24,699.29		

# Small Unit Separator Sizing

# Small Unit Heat Exchanger Costing

		SM	ALL UNIT HE	AT EXCHANGE	R COSTING		
	E-103		E-104		E-105		E-106
К1	3.8062	К1	3.8062	К1	3.8062	К1	3.8062
К2	0.8924	К2	0.8924	К2	0.8924	К2	0.8924
КЗ	-0.1671	К3	-0.1671	кз	-0.1671	КЗ	-0.1671
A (ft <sup>2</sup> )	19.484554	A (ft <sup>2</sup> )	1.53853862	A (ft <sup>2</sup> )	7.28466728	A (ft <sup>2</sup> )	0.805141006
A (m²)	1.8101761	A (m²)	0.14293505	A (m²)	0.67676839	A (m²)	0.074800119
Cp°	\$10,594.76	Cp°	\$856.98	Cp°	\$4,467.66	Cp°	\$ 388.48
P (psi)	485	P (psi)	471	P (psi)	471	P (psi)	487
P (bar)	33.439512	P (bar)	32.4742481	P (bar)	32.4742481	P (bar)	33.57925946
C1	0	C1	0	C1	0	C1	0
C2	0	C2	0	C2	0	C2	0
С3	0	С3	0	С3	0	С3	0
Fp	1	Fp	1	Fp	1	Fp	1
Fm	2.7	Fm	2.7	Fm	2.7	Fm	2.7
B1	1.63	B1	1.63	B1	1.63	B1	1.63
B2	1.66	B2	1.66	B2	1.66	B2	1.66
Cbm <sub>0</sub>	\$50,854.87	Cbm <sub>0</sub>	\$4,113.51	Cbm <sub>0</sub>	\$21,444.79	Cbm₀	\$ 1,864.72
		Total	<b>\$78,277.88</b>				

	SMALL UNIT HEAT EXCHANGER UTILITY COSTING										
	E	E-103		E-104			E-105			E-106	
	lb/hr	22011.991		lb/hr	1292.264		lb/hr	168.0899		lb/hr	700.7348
	klb/hr	22.011991		klb/hr	1.292264		klb/hr	0.16809		klb/hr	0.700735
PCW	\$/hr	11.005996	PCW	\$/hr	0.646132	HP Steam	\$/hr	0.840449	PCW	\$/hr	0.350367
	\$/yr	77130.018		\$/yr	4528.093		\$/yr	5889.869		\$/yr	2455.375
						Total Utili	12.84294442				

# Small Unit Heat Exchanger Utility Costing

#### Small Unit Heat Exchanger Sizing

	SMALL UNIT HEAT EXCHANGER SIZING										
E-103		E-104	4	E-105		E-106					
Duty (Btu/hr)	659039.02	Duty (Btu/hr)	34891.13	Duty (Btu/hr)	122369.43	Duty (Btu/hr)	20980.00				
Thi (F)	450.00	Thi (F)	275.00	Thi (F)	490.00	Thi (F)	420.00				
Tci (F)	90.00	Tci (F)	90.00	Tci (F)	275.00	Tci (F)	90.00				
Tho (F)	275.00	Tho (F)	250.00	Tho (F)	490.00	Tho (F)	250.00				
Tco (F)	120.00	Tco (F)	120.00	Tco (F)	420.00	Tco (F)	120.00				
ΔTlm (F)	250.55	ΔTlm (F)	157.49	ΔTIm (F)	129.22	ΔTlm (F)	222.71				
U (Btu/hr*ft <sup>2</sup> *°i	150.00	U (Btu/hr*ft <sup>2</sup> *°F)	160.00	U (Btu/hr*ft <sup>2</sup> *°F)	130.00	U (Btu/hr*ft <sup>2</sup> *°F)	130.00				
A (ft <sup>2</sup> )	19.48	A (ft <sup>2</sup> )	1.54	A (ft <sup>2</sup> )	7.28	A (ft <sup>2</sup> )	0.81				
Cv (Btu/lbm °F)	1.00	Cv (Btu/lbm °F)	1.00	λ (Btu/lb)	728.00	Cv (Btu/lbm °F)	1.00				
m (lb/hr)	22011.99	m (lb/hr)	1292.26	m (lb/hr)	168.09	m (lb/hr)	700.73				
F (h-ft2-°F/Btu)	0.90	F (h-ft2-°F/Btu)	0.90	F (h-ft2-°F/Btu)	0.90	F (h-ft2-°F/Btu)	0.90				
cooler		cooler		heater		cooler					

	í	1	ī		IUM UNIT SEP				1
	nstants	V-10		Consta		V-10		Consta	
P (psia)	481.00	$Q_v$ (ft <sup>3</sup> /s)	0.71	ρ <sub>v</sub> (lb/ft3)	1.33	Q <sub>v</sub> (ft <sup>3</sup> /s)	0.66	ρ <sub>ν</sub>	1.88
o <sub>v</sub> (lb/ft3)	1.31	$Q_L(ft^3/s)$	0.04	ρ <sub>L</sub> (lb/ft3)	38.13	$Q_L(ft^3/s)$	0.00	ρι	33.20
o <sub>L</sub> (lb/ft3)	37.42	U <sub>T</sub> (ft/s)	1.51	ρ <sub>Η</sub>	58.21	Q <sub>LL</sub> (ft <sup>3</sup> /s)	0.00	ρ <sub>H</sub>	52.40
к	0.29	U <sub>v</sub> (ft/s)	1.13	P (psia)	467.00	Q <sub>HL</sub> (ft <sup>3</sup> /s)	0.00	P (psia)	467.00
m <sub>v</sub> (Ib/hr)	3344.53	T <sub>H</sub> (s)	120.00	μ <sub>L</sub> (cP)	0.23	U <sub>T</sub> (ft/s)	1.52	μ <sub>L</sub> (cP)	0.19
m <sub>L</sub> (Ib/hr)	4899.21	T <sub>s</sub> (s)	60.00	μ <sub>н</sub> (сР)	0.23	U <sub>v</sub> (ft/s)	1.14	μ <sub>н</sub> (сР)	0.12
		V <sub>H</sub> (ft <sup>3</sup> )	4.36	К	0.29	T <sub>H</sub> (min)	120.00	К	0.29
H/D	to A/At	V <sub>s</sub> (ft <sup>3</sup> )	2.18	ks	0.16	T <sub>s</sub> (min)	60.00	ks	0.16
а	-4.75593E-05	L/D (est.)	3.00	m <sub>v</sub> (lb/hr)	3178.66	V <sub>H</sub> (ft <sup>3</sup> )	0.05	m <sub>v</sub> (lb/hr)	366.85
b	3.924091	D (ft)	2.00	m <sub>⊾</sub> (lb/hr)	55.84	V <sub>s</sub> (ft <sup>3</sup> )	0.02	m <sub>⊾</sub> (lb/hr)	1358.24
с	0.174875	$A_{T}$ (ft <sup>2</sup> )	3.14	m <sub>⊔</sub> (lb/hr)	55.84	L/D (est)	3.00	m <sub>LL</sub> (lb/hr)	1358.24
d	-6.358805	H <sub>LLL</sub> (ft)	0.75	m <sub>HL</sub> (lb/hr)	110.03	D (ft)	1.00	m <sub>HL</sub> (lb/hr)	3174.13
e	5.668973	H <sub>LLL</sub> /D	0.38			A <sub>T</sub> (ft <sup>2</sup> )	0.79		
f	4.018448	A <sub>LLL</sub> /A <sub>T</sub>	0.34			H <sub>v</sub> (ft)	0.20		
g	-4.916411	A <sub>LLL</sub> (ft <sup>2</sup> )	1.08			H <sub>v</sub> /D	0.20		
h	-1.801705	H <sub>v</sub> (ft)	0.40			A <sub>V</sub> /A <sub>T</sub>	0.14		
i	-0.145348	H <sub>v</sub> /D	0.20			A <sub>v</sub> (ft <sup>2</sup> )	0.11		
		A <sub>V</sub> /A <sub>T</sub>	0.14			H <sub>LLL</sub> (ft)	0.07		
A/At	t to H/D	A <sub>v</sub> (ft <sup>2</sup> )	0.45			H <sub>LLL</sub> /D	0.07		
а	0.00153756	L <sub>min</sub> (ft)	4.50			A <sub>LLL</sub> /A <sub>T</sub>	0.03		
b	26.787101	r (ft)	1.00			A <sub>LLL</sub> (ft <sup>2</sup> )	0.03		
с	3.299201	V (ft <sup>3</sup> )	14.14			H <sub>w</sub> (ft)	0.80		
d	-22.923932					L <sub>2</sub> (ft)	0.50		
e	24.353518					H <sub>w</sub> /2 (ft)	0.40		
f	-14.844824					H <sub>HL</sub> (ft)	0.40		
g	-36.999376					H <sub>LL</sub> (ft)	0.40		
h	10.529572					H <sub>HL</sub> /D	0.40		
i	9.892851					A <sub>HL</sub> /A <sub>T</sub>	0.37		
						A <sub>HL</sub> (ft <sup>2</sup> )	0.29		
						A <sub>LL</sub> (ft <sup>2</sup> )	0.38		
						U <sub>HL</sub> (ft/s)	14.01		
						U <sub>LH</sub> (ft/s)	14.38		
						t <sub>HL</sub> (s)	0.06		
						t <sub>LH</sub> (s)	0.33		
						L <sub>1</sub> (ft)	0.50		
						L min (ft)	1.00		
						φ(s)	0.18		
						U <sub>VA</sub> (ft/s)	5.92		
						L <sub>min</sub> (ft)	1.50		
						H <sub>HLL</sub> (ft) r (ft)	0.80		
						V (ft <sup>3</sup> )	0.30		

Medium Unit Separator Sizing

1	/-104
Q <sub>v</sub> (ft <sup>3</sup> /s)	0.05
Q <sub>L</sub> (ft <sup>3</sup> /s)	0.01
Q <sub>LL</sub> (ft <sup>3</sup> /s)	0.01
Q <sub>HL</sub> (ft <sup>3</sup> /s)	0.02
U <sub>T</sub> (ft/s)	1.18
U <sub>v</sub> (ft/s)	0.88
T <sub>H</sub> (min)	120.00
T <sub>s</sub> (min)	60.00
V <sub>H</sub> (ft <sup>3</sup> )	1.36
V <sub>s</sub> (ft <sup>3</sup> )	0.68
L/D (est)	3.00
D (ft)	2.00
A <sub>T</sub> (ft <sup>2</sup> )	3.14
H <sub>v</sub> (ft)	0.40
H <sub>v</sub> /D	0.20
A <sub>V</sub> /A <sub>T</sub>	0.14
A <sub>v</sub> (ft <sup>2</sup> )	0.45
H <sub>LLL</sub> (ft)	0.75
H <sub>LLL</sub> /D	0.38
A <sub>LLL</sub> /A <sub>T</sub>	0.34
A <sub>LLL</sub> (ft <sup>2</sup> )	1.08
H <sub>w</sub> (ft)	1.60
L <sub>2</sub> (ft)	1.50
H <sub>w</sub> /2 (ft)	0.80
H <sub>HL</sub> (ft)	0.80
H <sub>LL</sub> (ft)	0.80
H <sub>HL</sub> /D	0.40
A <sub>HL</sub> /A <sub>T</sub>	0.37
A <sub>HL</sub> (ft <sup>2</sup> )	1.17
A <sub>LL</sub> (ft <sup>2</sup> )	1.52
U <sub>HL</sub> (ft/s)	16.82
U <sub>LH</sub> (ft/s)	25.23
t <sub>HL</sub> (s)	0.54
t <sub>LH</sub> (s)	0.38
L <sub>1</sub> (ft)	0.50
Lmin (ft)	2.00
φ(s)	0.45
U <sub>VA</sub> (ft/s)	0.12 0.50
L <sub>min</sub> (ft)	1.60
H <sub>HLL</sub> (ft) r (ft)	1.00
V (ft <sup>3</sup> )	6.28
- ()	0.20

		DIUM UNIT SEPA	RATOR COSTING		
	V-102		V-103	V-104	
К1	3.5565	К1	3.5565	K1	3.5565
К2	0.3776	К2	0.3776	К2	0.3776
К3	0.0905	КЗ	0.0905	КЗ	0.0905
V (ft³)	14.13716694	V (ft <sup>3</sup> )	0.785398163	V (ft <sup>3</sup> )	6.28318531
V (m³)	0.40031621	V (m³)	0.022239789	V (m³)	0.17791832
Cp°	\$2,634.35	Cp°	\$1,512.24	Cp°	\$2,109.89
P (psi)	481	P (psi)	467	P (psi)	467
P (bar)	33.16326531	P (bar)	32.19801434	P (bar)	32.1980143
D (ft)	2	D (ft)	1	D (ft)	2
D (m)	0.609570253	D (m)	0.304785126	D (m)	0.60957025
Fp	2.432764873	Fp	1.43760073	Fp	2.37520146
Fm	3.1	Fm	3.1	Fm	3.1
B1	1.49	B1	1.49	B1	1.49
B2	1.52	B2	1.52	B2	1.52
Cbm₀	\$12,644.89	Cbm <sub>0</sub>	\$7,258.75	Cbm <sub>0</sub>	\$10,127.47
		Total	\$30,031.11		

# Medium Unit Separator Costing

# Medium Unit Heat Exchanger Costing

		MEI	DIUM UNIT HEA	T EXCHANGER	COSTING		
I	E-103		E-104	E-105			E-106
К1	4.188	К1	3.8062	К1	3.8062	К1	3.8062
К2	-0.2503	К2	0.8924	К2	0.8924	K2	0.8924
К3	0.1974	К3	-0.1671	КЗ	-0.1671	КЗ	-0.1671
A (ft²)	114.68447	A (ft <sup>2</sup> )	6.645263725	A (ft <sup>2</sup> )	47.341428	A (ft <sup>2</sup> )	6.869410869
A (m²)	10.654546	A (m <sup>2</sup> )	0.617365799	A (m²)	4.39816683	A (m²)	0.63818977
Cp°	\$13,779.34	Cp°	\$4,092.13	Cp°	\$20,469.63	Cp°	4224.545471
P (psi)	485	P (psi)	471	P (psi)	471	P (psi)	487
P (bar)	33.439512	P (bar)	32.47424813	P (bar)	32.4742481	P (bar)	33.5774073
C1	0.03881	C1	0	C1	0	C1	0
C2	-0.11272	C2	0	C2	0	C2	0
С3	0.08183	С3	0	C3	0	С3	0
Fp	1.1405561	Fp	1	Fp	1	Fp	1
Fm	2.7	Fm	2.7	Fm	2.7	Fm	2.7
B1	1.63	B1	1.63	B1	1.63	B1	1.63
B2	1.66	B2	1.66	B2	1.66	B2	1.66
Cbm <sub>0</sub>	\$66,140.84	Cbm <sub>0</sub>	\$19,642.21	Cbm <sub>0</sub>	\$98,254.21	Cbm <sub>0</sub>	\$20,277.82
		Total	\$204,315.08				

		n	MEDIUN	1 UNIT F	IEAT EXCI	HANGER	UTILIT		G		
	E	-103		E	-104		E	-105		E	-106
	lb/hr	129560.8		lb/hr	5033.464		lb/hr	1092.379		lb/hr	5978.624
	klb/hr	129.5608		klb/hr	5.033464		klb/hr	1.092379		klb/hr	5.978624
PCW	\$/hr	64.78038	PCW	\$/hr	2.516732	HP Steam	\$/hr	5.461893	PWC	\$/hr	2.989312
	\$/yr	453980.9		\$/yr	17637.26		\$/yr	38276.95		\$/yr	20949.1
					Total Utili	75.74831					

# Medium Unit Heat Exchanger Utility Costing

#### Medium Unit Heat Exchanger Sizing

		MEDIUM (	JNIT HEAT EXCH	ANGER SIZING				
E-10	)3	E-10	4	E-105		E-106		
Duty (Btu/hr)	3879048.97	Duty (Btu/hr)	150701.92	Duty (Btu/hr)	795251.66	Duty (Btu/hr)	179000.00	
Thi (F)	450.00	Thi (F)	275.00	Thi (F)	490.00	Thi (F)	420.00	
Tci (F)	90.00	Tci (F)	90.00	Tci (F)	275.00	Tci (F)	90.00	
Tho (F)	275.00	Tho (F)	250.00	Tho (F)	490.00	Tho (F)	250.00	
Tco (F)	120.00	Tco (F)	120.00	Tco (F)	420.00	Tco (F)	120.00	
ΔTlm (F)	250.55	ΔTIm (F)	157.49	ΔTlm (F)	129.22	ΔTIm (F)	222.71	
U (Btu/hr*ft <sup>2</sup> *°F	150.00	U (Btu/hr*ft <sup>2</sup> *°F)	160.00	U (Btu/hr*ft <sup>2</sup> *°F)	130.00	U (Btu/hr*ft <sup>2</sup> *°F)	130.00	
A (ft <sup>2</sup> )	114.68	A (ft <sup>2</sup> )	6.65	A (ft <sup>2</sup> )	47.34	A (ft <sup>2</sup> )	6.87	
Cv (Btu/lbm °F)	1.00	Cv (Btu/lbm °F)	1.00	λ (Btu/lb)	728.00	Cv (Btu/lbm °F)	1.00	
m (lb/hr)	129560.75	m (lb/hr)	5033.46	m (lb/hr)	1092.38	m (lb/hr)	5978.62	
F (h-ft2-°F/Btu)	0.90	F (h-ft2-°F/Btu)	0.90	F (h-ft2-°F/Btu)	0.90	F (h-ft2-°F/Btu)	0.90	
cooler		cooler		heater		cooler		

			LARGE UNIT SEPARATORS										
Cor	istants	V-10	2	Consta	ants	V-10	)3	μL (CP)         μ           μH (CP)         μ           K         μ           ks         μ           mL (lb/hr)         2           mL (lb/hr)         2	ants				
P (psia)	481.00	Q <sub>v</sub> (ft <sup>3</sup> /s)	1.63	ρ <sub>v</sub> (lb/ft3)	1.29	Q <sub>v</sub> (ft <sup>3</sup> /s)	1.52	ρν	1.80				
o <sub>v</sub> (lb/ft3)	1.26	$Q_{L}(ft^{3}/s)$	0.06	ρ <sub>L</sub> (lb/ft3)	38.34	$Q_L(ft^3/s)$	0.00	ρ <sub>ι</sub>	33.39				
o <sub>L</sub> (lb/ft3)	37.52	U <sub>T</sub> (ft/s)	1.54	ρ <sub>H</sub>	58.21	Q <sub>LL</sub> (ft <sup>3</sup> /s)	0.00	ρ <sub>H</sub>	52.40				
<b>‹</b>	0.29	U <sub>v</sub> (ft/s)	1.16	P (psia)	467.00	Q <sub>HL</sub> (ft <sup>3</sup> /s)	0.00	P (psia)	467.00				
m <sub>v</sub> (lb/hr)	7380.21	Т <sub>н</sub> (s)	120.00	μ <sub>L</sub> (cP)	0.25	U <sub>T</sub> (ft/s)	1.55	μ <sub>L</sub> (cP)	0.19				
m <sub>L</sub> (lb/hr)	8224.19	T <sub>s</sub> (s)	60.00	μ <sub>н</sub> (сР)	0.23	U <sub>v</sub> (ft/s)	1.16	μ <sub>н</sub> (cP)	0.12				
		V <sub>H</sub> (ft <sup>3</sup> )	7.31	К	0.29	T <sub>H</sub> (min)	120.00	К	0.29				
H/D	to A/At	V <sub>s</sub> (ft <sup>3</sup> )	3.65	ks	0.16	T <sub>s</sub> (min)	60.00	ks	0.16				
a	-4.75593E-05	L/D (est.)	3.00	m <sub>v</sub> (lb/hr)	7022.01	V <sub>H</sub> (ft <sup>3</sup> )	0.09	m <sub>v</sub> (lb/hr)	467.64				
ט	3.924091	D (ft)	2.00	m <sub>L</sub> (lb/hr)	106.46	V <sub>s</sub> (ft <sup>3</sup> )	0.05	m <sub>L</sub> (lb/hr)	2245.79				
2	0.174875	A <sub>T</sub> (ft <sup>2</sup> )	3.14	m <sub>LL</sub> (lb/hr)	106.46	L/D (est)	3.00	m <sub>LL</sub> (lb/hr)	2245.79				
ł	-6.358805	H <sub>LLL</sub> (ft)	0.75	m <sub>HL</sub> (lb/hr)	251.75	D (ft)	1.00		5510.76				
2	5.668973	H <sub>LLL</sub> /D	0.38			A <sub>T</sub> (ft <sup>2</sup> )	0.79						
	4.018448	A <sub>LLL</sub> /A <sub>T</sub>	0.34			$H_{v}$ (ft)	0.20						
:	-4.916411	A <sub>LLL</sub> (ft <sup>2</sup> )	1.08			H <sub>v</sub> /D	0.20						
ı	-1.801705	H <sub>v</sub> (ft)	0.40			A <sub>v</sub> /A <sub>T</sub>	0.14						
	-0.145348	H <sub>v</sub> /D	0.20			A <sub>v</sub> (ft <sup>2</sup> )	0.11						
		A <sub>V</sub> /A <sub>T</sub>	0.14			H <sub>LLL</sub> (ft)	0.75						
A/At	to H/D	A <sub>v</sub> (ft <sup>2</sup> )	0.45				0.75						
1	0.00153756	L <sub>min</sub> (ft)	7.00				0.80						
)	26.787101	r (ft)	1.00				0.63						
:	3.299201	V (ft <sup>3</sup> )	21.99				0.80						
1	-22.923932												
5	24.353518					H <sub>LL</sub> /D         0.75           A <sub>LL</sub> /A <sub>T</sub> 0.80							
:	-14.844824					H <sub>HL</sub> (ft)	0.40						
5	-36.999376					H <sub>LL</sub> (ft)	0.40						
ı	10.529572					H <sub>HL</sub> /D	0.40						
	9.892851					A <sub>HL</sub> /A <sub>T</sub>	0.37						
						A <sub>HL</sub> (ft <sup>2</sup> )	0.29						
						A <sub>LL</sub> (ft <sup>2</sup> )	0.38						
						U <sub>HL</sub> (ft/s)	13.01						
						U <sub>LH</sub> (ft/s)	14.22						
						t <sub>HL</sub> (s)	0.69						
						t <sub>LH</sub> (s)	0.34						
						L <sub>1</sub> (ft)	0.50						
						L min (ft)	4.00						
						φ (s)	0.17						
						U <sub>VA</sub> (ft/s)	13.55						
						L <sub>min</sub> (ft)	2.50						
						H <sub>HLL</sub> (ft)	0.80						
						r (ft)	0.50						
						V (ft <sup>3</sup> )	3.14						

Large Unit Separator Sizing

1	/-104
Q <sub>v</sub> (ft <sup>3</sup> /s)	0.07
$Q_{L}(ft^{3}/s)$	0.02
Q <sub>LL</sub> (ft <sup>3</sup> /s)	0.02
Q <sub>HL</sub> (ft <sup>3</sup> /s)	0.03
U <sub>T</sub> (ft/s)	1.21
U <sub>v</sub> (ft/s)	0.91
T <sub>H</sub> (min)	120.00
T <sub>s</sub> (min)	60.00
V <sub>H</sub> (ft <sup>3</sup> )	2.24
V <sub>s</sub> (ft <sup>3</sup> )	1.12
L/D (est)	4.00
D (ft)	2.00
A <sub>τ</sub> (ft <sup>2</sup> )	3.14
H <sub>v</sub> (ft)	0.40
H <sub>v</sub> /D	0.20
A <sub>V</sub> /A <sub>T</sub>	0.14
A <sub>v</sub> (ft <sup>2</sup> )	0.45
H <sub>LLL</sub> (ft)	0.75
H <sub>LLL</sub> /D	0.38
A <sub>LLL</sub> /A <sub>T</sub>	0.34
A <sub>LLL</sub> (ft <sup>2</sup> )	1.08
H <sub>w</sub> (ft)	1.60
L <sub>2</sub> (ft)	2.50
H <sub>w</sub> /2 (ft)	0.80
H <sub>HL</sub> (ft)	0.80
H <sub>LL</sub> (ft)	0.80
H <sub>HL</sub> /D	0.40
A <sub>HL</sub> /A <sub>T</sub>	0.37
A <sub>HL</sub> (ft <sup>2</sup> )	1.17
A <sub>LL</sub> (ft <sup>2</sup> )	1.52
U <sub>HL</sub> (ft/s)	16.08
U <sub>LH</sub> (ft/s)	24.99
t <sub>HL</sub> (s)	0.56
t <sub>∟н</sub> (s)	0.38
L <sub>1</sub> (ft)	0.50
Lmin (ft)	3.00
φ(s)	0.44
U <sub>VA</sub> (ft/s)	0.16
L <sub>min</sub> (ft) ⊌ (ft)	0.50
H <sub>HLL</sub> (ft) r (ft)	1.60 1.00
V (ft <sup>3</sup> )	9.42
- ()	5.72

		LARGE UNIT	COSTING		
	V-102		V-103		V-104
К1	3.5565	K1	3.5565	К1	3.5565
К2	0.3776	K2	0.3776	К2	0.3776
К3	0.0905	КЗ	0.0905	КЗ	0.0905
V (ft³)	21.99114858	V (ft <sup>3</sup> )	3.141592654	V (ft <sup>3</sup> )	9.42477796
V (m³)	0.622714104	V (m³)	0.088959158	V (m³)	0.26687747
Cp°	\$3,038.46	Cp°	\$1,818.21	Cp°	\$2,342.40
P (psi)	481	P (psi)	467	P (psi)	467
P (bar)	33.16326531	P (bar)	32.19801434	P (bar)	32.1980143
D (ft)	2	D (ft)	1	D (ft)	2
D (m)	0.609570253	D (m)	0.304785126	D (m)	0.60957025
Fp	2.432764873	Fp	1.43760073	Fp	2.37520146
Fm	3.1	Fm	3.1	Fm	3.1
B1	1.49	B1	1.49	B1	1.49
B2	1.52	B2	1.52	B2	1.52
Cbm <sub>0</sub>	14,584.59	Cbm <sub>0</sub>	8,727.42	Cbm <sub>0</sub>	11,243.51
		Total	\$34,555.52		

# Large Unit Separator Costing

#### Large Unit Heat Exchanger Costing

		LAR	GE UNIT HEAT E	XCHANGER CO	OSTING		-
	E-103		E-104		E-105		E-106
К1	4.188	К1	3.8062	К1	3.8062	К1	3.8062
К2	-0.2503	К2	0.8924	К2	0.8924	К2	0.8924
кз	0.1974	К3	-0.1671	КЗ	-0.1671	КЗ	-0.1671
A (ft <sup>2</sup> )	181.731811	A (ft <sup>2</sup> )	13.52147142	A (ft <sup>2</sup> )	85.47153497	A (ft <sup>2</sup> )	8.93023413
A (m²)	16.88345404	A (m <sup>2</sup> )	1.256187016	A (m²)	7.940573117	A (m²)	0.829646701
Cp°	\$15,072.36	Cp°	\$7,815.50	Cp°	\$29,779.77	Cp°	5404.071011
P (psi)	485	P (psi)	471	P (psi)	471	P (psi)	487
P (bar)	33.4395124	P (bar)	32.47424813	P (bar)	32.47424813	P (bar)	33.5774073
C1	0.03881	C1	0	C1	0	C1	0
C2	-0.11272	C2	0	C2	0	C2	0
С3	0.08183	С3	0	C3	0	С3	0
Fp	1.140556102	Fp	1	Fp	1	Fp	1
Fm	2.7	Fm	2.7	Fm	2.7	Fm	2.7
B1	1.63	B1	1.63	B1	1.63	B1	1.63
B2	1.66	B2	1.66	B2	1.66	B2	1.66
Cbm <sub>0</sub>	\$72,347.35	Cbm <sub>0</sub>	\$37,514.38	Cbm <sub>0</sub>	\$142,942.90	Cbm <sub>0</sub>	\$ 25,939.54
		Total	\$278,744.16				

			LAR	GE UNIT HEAT	<b>FEXCHAN</b>	GER UTI	LITY COSTI	NG			
		E-103	E-104				E-105			E-106	
	lb/hr	228116.81		lb/hr	11379.84		lb/hr	1774.99		lb/hr	7772.211
	klb/hr	228.11681		klb/hr	11.37984		klb/hr	1.77499		klb/hr	7.772211
PCW	\$/hr	114.05841	PCW	\$/hr	5.689921	HP Steam	\$/hr	8.874949	PCW	\$/hr	3.886106
	\$/yr	799321.32		\$/yr	39874.97		\$/yr	62195.64		\$/yr	27233.83
				Total Utility	132.5094						

#### Large Unit Heat Exchanger Utility Costing

#### Large Unit Heat Exchanger Sizing

		LARGE U	JNIT HEAT EXCHA	ANGER SIZING				
	E-103	E-1	.04	E-105		E-106		
Duty (Btu/hr)	6829817.44	Duty (Btu/hr)	340712.46	Duty (Btu/hr)	1292192.59	Duty (Btu/hr)	232700.00	
Thi (F)	450.00	Thi (F)	275.00	Thi (F)	490.00	Thi (F)	420.00	
Tci (F)	90.00	Tci (F)	90.00	Tci (F)	275.00	Tci (F)	90.00	
Tho (F)	275.00	Tho (F)	250.00	Tho (F)	490.00	Tho (F)	250.00	
Tco (F)	120.00	Tco (F)	120.00	Tco (F)	420.00	Tco (F)	120.00	
ΔTlm (F)	250.55	ΔTIm (F)	157.49	ΔTlm (F)	129.22	ΔTIm (F)	222.71	
U (Btu/hr*ft <sup>2</sup> *°F	150.00	U (Btu/hr*ft <sup>2</sup> *°F)	160.00	U (Btu/hr*ft <sup>2</sup> *°F)	130.00	U (Btu/hr*ft <sup>2</sup> *°F)	130.00	
A (ft <sup>2</sup> )	181.73	A (ft <sup>2</sup> )	13.52	A (ft <sup>2</sup> )	85.47	A (ft <sup>2</sup> )	8.93	
Cv (Btu/lbm °F)	1.00	Cv (Btu/lbm °F)	1.00	λ (Btu/lb)	728.00	Cv (Btu/lbm °F)	1.00	
m (lb/hr)	228116.81	m (lb/hr)	11379.84	m (lb/hr)	1774.99	m (lb/hr)	7772.21	
F (h-ft2-°F/Btu)	0.90	F (h-ft2-°F/Btu)	0.90	F (h-ft2-°F/Btu)	0.90	F (h-ft2-°F/Btu)	0.90	
cooler		cooler		heater		cooler		

#### Small Unit Pipe Diameter Sizing

			Small Un	it			
Stream	vapor frac	mass Flowrate (lb/hr)	density (lb/ft^3)	volumetric (ft^3/hr	)	D (in)	D (in)
15	0.9989	1535	1.109	1384.129847	172.5548542	2	2
16	0.5173	1535	2.492	615.9711075	76.79106474	0.4006499	0.5
17	1	752.1	1.252	600.7188498	74.88961661	1.5	1.5
18	0.9559	752.1	1.342	560.4321908	69.86721311	1.5	1.5
19	1	715.8	1.279	559.6559812	69.77044566	1.5	1.5
20	0	10.45	38.38	0.272277228	0.033943894	0.05	0.25
21	0	25.83	58.21	0.443738189	0.055319361	0.06	0.25
22	0	783.4	49.62	15.78798871	1.968235926	0.35	0.5
23	0.0399	783.4	19.59	39.98979071	4.985393908	0.56	0.75
24	1	41.93	1.781	23.5429534	2.935021523	0.375	0.5
25	0.1053	41.93	13.32	3.147897898	0.392437938	0.16	0.25
26	0	212.5	33.44	6.354665072	0.792214912	0.22	0.25
27	0	528.9	52.39	10.09543806	1.258564612	0.28	0.5
28	0	554.8	52.7	10.52751423	1.312430108	0.29	0.5
29	0	1038	56.03	18.52578976	2.309548456	0.38	0.5
11	0	483.6	59.37	8.145528044	1.01547583	0.25	0.5

			Medium U	Init			
Stream	vapor frac	mass Flowrate (lb/hr)	density (lb/ft^3)	volumetric (ft^3/h	gal/min	D (in)	D (in)
15	1	8224	1.174	7005.110733	875.6388416	4	4
16	0.4267	8244	3.103	2656.783758	332.0979697	0.7188015	0.75
17	1	3345	1.308	2557.33945	319.6674312	2.5	2.5
18	0.955	3345	1.402	2385.877318	298.2346648	2.5	2.5
19	1	3179	1.335	2381.273408	297.659176	2.5	2.5
20	0	55.84	38.13	1.464463677	0.18305796	0.1069632	0.25
21	0	110	58.21	1.889709672	0.236213709	0.1215046	0.25
22	0	4899	49.1	99.77596741	12.47199593	0.8828928	1
23	0.0544	4899	16.54	296.191052	37.0238815	1.5211813	1.75
24	1	366.8	1.883	194.795539	24.34944238	1	1
25	0.0785	366.8	16.61	22.08308248	2.76038531	0.4153602	0.5
26	0	1358	33.2	40.90361446	5.112951807	0.5652959	0.75
27	0	3174	52.4	60.57251908	7.571564885	0.6879119	0.75
28	0	3284	51.11	64.25357073	8.031696341	0.7085062	0.75
29	0	5729	55.1	103.9745917	12.99682396	0.9012777	1
11	0	2445	58.29	41.94544519	5.243180648	0.5724498	0.75

# Medium Unit Pipe Diameter Sizing

# Large Unit Pipe Diameter Sizing

			Large Un	it			
Stream	vapor frac	mass Flowrate (lb/hr)	density (lb/ft^3)	volumetric (ft^3/h	gal/min	D (in)	D (in)
15	0.9998	15600	1.121	13916.1463	1739.518287	4	4
16	0.4887	15600	2.592	6018.518519	752.3148148	0.9939834	1
17	1	7380	1.261	5852.498017	731.5622522	3	3
18	0.9557	7380	1.351	5462.620281	682.8275352	3	3
19	1	7022	1.287	5456.099456	682.012432	3	3
20	0	106.5	38.24	2.785041841	0.34813023	0.1475064	0.25
21	0	251.7	58.21	4.323999313	0.540499914	0.1837967	0.25
22	0	8224	49.52	166.0743134	20.75928918	1.1390591	1.25
23	0.0422	8224	19.04	431.9327731	53.99159664	1.8369744	2
24	1	467.6	1.803	259.3455352	32.4181919	1	1
25	0.0998	467.6	13.91	33.61610352	4.20201294	0.5124703	0.75
26	0	2246	33.39	67.2656484	8.40820605	0.7249227	0.75
27	0	5511	52.4	105.1717557	13.14646947	0.9064515	1
28	0	5762	50.95	113.0912659	14.13640824	0.9399604	1
29	0	10610	55.26	192.0014477	24.00018096	1.2247495	1.25
12	0	4844	59.26	81.74147823	10.21768478	0.7991278	1

# FTR Material Compatibility Chart

METHANE																					
Compatible 🛄	WATER																				
Compatible 🔲	Compatible	CARBON MONOXIDE																			
Compatible	Compatible	Compatible	HYDROGEN																		
Compatible	Compatible	Compatible	Compatible	ETHANE																	
Compatible	Compatible	Compatible	Compatible	Compatible	PROPANE																
Compatible 🔲	Compatible	Compatible	Compatible 🔲	Compatible	Compatible	BUTANE															
Compatible	Compatible	Compatible 🔲	Compatible	Compatible	Compatible	Compatible	N-PENTANE														
Compatible	Compatible	Compatible 🔲	Compatible	Compatible 🔲	Compatible	Compatible	Compatible	N-HEXANE													
Compatible	Compatible	Compatible 🔲	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	N-HEPTANE												
Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	N-OCTANE											
Compatible 🗖	Compatible	Compatible 🔲	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	N-NONANE										
Compatible	Compatible	Compatible 🔲	Compatible	Compatible	Compatible	Compatible 🔲	Compatible	Compatible	Compatible 🔲	Compatible	Compatible	N-DECANE	í								
Compatible	Compatible	Compatible	Compatible 🔲	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible 🔲	Compatible	UNDECANE								
Compatible	Compatible 🖬	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible 🔲	Compatible 🔲	Compatible	Compatible	Compatible 🔲	Compatible	Compatible	N-DODECANE	Ĩ						
Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	TRIDECANE						
Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible 🛙	Compatible	N-TETRADECANE					
Compatible	Compatible	Compatible 🔲	Compatible	Compatible	Compatible 🔲	Compatible	Compatible	Compatible	Compatible	Compatible 📴	Compatible	Compatible	Compatible	Compatible 🛙	Compatible 🛙	Compatible	N-PENTADECANE				
Compatible 🔲	Compatible	Compatible 🛄	Compatible	Compatible	Compatible	Compatible 🔲	Compatible 🔲	Compatible	Compatible 🔲	Compatible 🛙	Compatible 🔲	Compatible 🔲	Compatible	Compatible 🛄	Compatible 🛙	Compatible 🔲	Compatible 🔲	N-HEXADECANE			
Compatible	Compatible 🛄	Compatible	Compatible	Compatible	Compatible 🔲	Compatible	Compatible 🔲	Compatible	Compatible	Compatible 🖪	Compatible 🔲	Compatible	Compatible	Compatible 🛄	Compatible 🛄	Compatible	Compatible	Compatible	N-HEPTADECANE		
Compatible 🔲	Compatible	Compatible 🔲	Compatible	Compatible 🔲	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible 🔲	Compatible 🔲	Compatible 🔲	Compatible 🔲	Compatible	Compatible	Compatible 🔲	N-OCTADECANE	
Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible 🗖	Compatible	Compatible	Compatible	Compatible 🛄	Compatible	Compatible 🔲	Compatible	Compatible	Compatible	Compatible	N-NONADECANE
Compatible 🔲	Compatible	Compatible	Compatible	Compatible	Compatible	Compatible 🔲	Compatible 🔲	Compatible	Compatible	Compatible 🔲	Compatible 🔲	Compatible	Compatible	Compatible 🛙	Compatible 🔲	Compatible	Compatible 🔲	Compatible	Compatible	Compatible	Compatible
	Compatible II Co	Compatible     WATER.       Compatible     Compatible       Compatible	Compatible II       WATER         Compatible II       Compatible II       Compatible II         Compatible II       Compatibl	Compatible       WATER         Compatible       Compatible       Compatible       HYDROGEN         Compatible       Compatible       Compatible       HYDROGEN         Compatible       Compatible       Compatible       Compatible       HYDROGEN         Compatible       Compatible       Compatible       Compatible       Compatible         Compatible       Compatible       Compatible	Compatible III       NATER         Compatible III       Compatible III       Compatible IIII         Compatible IIII       Compatible IIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIII	Compatible II       NATER         Compatible II       Compatible II       CARDAN MONICOLDE         Compatible II       Compatible II       Compatible II       PORDOBII         Compatible II       Compatible II       Compatible II       PORDOBII         Compatible II       Compatible III       Compatible III       Compatible III       PORDOBII         Compatible III       Compatible III       Compatible III       Compatible IIII       Compatible IIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIII	Compatible III       Compatible III       Compatible III       Compatible IIII       Compatible IIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIII	Compatibility       NUTE         Compatibility       Comp	Conguth III Conguth IIIIConguth IIIIIConguth IIIIIIConguth IIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIIII	InstalInstalComputed IComputed IIComputed IISector Sector	konstallkonstallConstallConstallConstallMaccanConstallConstallConstallMaccanConstallConstallConstallMaccanConstallConstallConstallMaccanConstallConstallConstallConstallParaConstall<	oquithe oparities oparities oparitiesoparities oparities oparitiesoparities oparities oparitiesoparities oparities oparitiesoparities oparities oparities oparitiesoparities oparities oparities oparities oparitiesoparities oparities oparities oparitiesoparities oparities oparities oparitiesoparities oparities oparities oparities oparitiesoparities oparities oparit	ional and anal anal anal anal anal anal an	Interpretation        Interpretation        Interpretation	ional       ional <t< td=""><td>ional </td><td>initial       initial       initial</td><td>image       image       <th< td=""><td>image       image       <t< td=""><td>image       image       <t< td=""><td>image       image       <t< td=""></t<></td></t<></td></t<></td></th<></td></t<>	ional 	initial       initial	image       image <th< td=""><td>image       image       <t< td=""><td>image       image       <t< td=""><td>image       image       <t< td=""></t<></td></t<></td></t<></td></th<>	image       image <t< td=""><td>image       image       <t< td=""><td>image       image       <t< td=""></t<></td></t<></td></t<>	image       image <t< td=""><td>image       image       <t< td=""></t<></td></t<>	image       image <t< td=""></t<>

			Small	Unit	
R-102		V-102		V-103	
Pressure (psia)	445.00	Pressure (psia)	441.00	Pressure (psia)	437.00
Volume (ft3)	2.36	Volume (ft3)	3.53	Volume (ft3)	0.10
R (ft <sup>3</sup> *psi/R*lbmol)	10.73	R (ft3*psi/R*lbmol)	10.73	R (ft3*psi/R*lbmol)	10.73
Temp (F)	450.00	Temp (F)	275.00	Temp (F)	250.00
Temp (°R)	909.67	Temp (°R)	734.67	Temp (°R)	709.67
n (Ibmol)	0.11	n (Ibmol)	0.20	n (Ibmol)	0.01
MW (lb/lbmol)	23.04	MW (lb/lbmol)	23.04	MW (lb/lbmol)	21.82
m (lb)	2.47	m (lb)	4.56	m (lb)	0.12
Molar Enthalpy of Vapor (Btu/Ibmo	-69128.24	Molar Enthalpy of Vapor (Btu/Ibmol	-69127.45	Molar Enthalpy of Vapor (Btu/Ibm	ol -42933.84
Molar Enthalpy of Liquid (Btu/Ibmo	-90204.45	Molar Enthalpy of Liquid (Btu/Ibmol	-90208.19	Molar Enthalpy of Liquid (Btu/lbm	<b>ol</b> -56009.49
λ	21076.21	λ	21080.74	λ	13075.65
Cp (BTU/lbmol*F)	12.71	Cp (BTU/lbmol*F)	11.54	Cp (BTU/lbmol*F)	11.35
T1 (F)	450.00	T1 (F)	275.00	T1 (F)	250.00
Tb (F)	382.67	Tb (F)	382.67	Tb (F)	186.14
mv (lb)	0.10	mv (lb)	-0.27	mv (lb)	0.01
η	0.02	η	0.02	η	0.02
ΔH <sub>c</sub> (Btu/lb)	10315.89	ΔH <sub>c</sub> (Btu/lb)	10315.89	ΔH <sub>c</sub> (Btu/lb)	16237.14
E <sub>TNT</sub> (Btu/lb)	2014.62	E <sub>TNT</sub> (Btu/lb)	2014.62	E <sub>TNT</sub> (Btu/lb)	2014.62
m <sub>TNT</sub> (kg)	0.00	m <sub>TNT</sub> (lb)	-0.012482	m <sub>TNT</sub> (lb)	0.000499
m <sub>TNT</sub> (g)	4.67				
r <sub>1</sub> (m)	10.00	r <sub>1</sub> (m)	10.00	r <sub>1</sub> (m)	10.00
z <sub>e</sub> (m/kg <sup>1/3</sup> )	713.59	z <sub>e</sub> (m/kg <sup>1/3</sup> )	-267.0536	z <sub>e</sub> (m/kg <sup>1/3</sup> )	6682.771
p <sub>s</sub>	0.00	ps	0.01	p <sub>s</sub>	0.00
p <sub>o</sub> (kPa)	0.24	p <sub>o</sub> (kPa)	0.63	p <sub>o</sub> (kPa)	0.03

# Small Unit Vapor Deflagration Analysis

V-104	
Pressure (psia)	437.00
Volume (ft3)	2.65
R (ft3*psi/R*lbmol)	10.73
Temp (F)	420.00
Temp (°R)	879.67
n (Ibmol)	0.12
MW (lb/lbmol)	24.35
m (lb)	2.99
Molar Enthalpy of Vapor (Btu/Ibmole	-99961.72
Molar Enthalpy of Liquid (Btu/lbmole	-119104.05
λ	19142.33
Cp (BTU/lbmol*F)	12.25
T1 (F)	420.00
Tb (F)	406.29
mv (lb)	0.03
η	0.02
ΔH <sub>c</sub> (Btu/lb)	4631.63
E <sub>TNT</sub> (Btu/lb)	2014.62
m <sub>TNT</sub> (lb)	0.0005472
r <sub>1</sub> (m)	10.00
z <sub>e</sub> (m/kg <sup>1/3</sup> )	6091.9951
p <sub>s</sub>	0.00
p <sub>o</sub> (kPa)	0.03

			Mediu	m Unit		
R-102		V-102		V-103		
Pressure (psia)	445.00	Pressure (psia)	441.00	Pressure (psia)	437.00	
Volume (ft3)	4.60	Volume (ft3)	14.14	Volume (ft3)	0.79	
R (ft <sup>3</sup> *psi/R*lbmol)	10.73	R (ft3*psi/R*lbmol)	10.73	R (ft3*psi/R*lbmol)	10.73	
Temp (F)	450.00	Temp (F)	275.00	Temp (F)	250.00	
Temp (°R)	909.67	Temp (°R)	734.67	Temp (°R)	709.67	
n (Ibmol)	0.21	n (Ibmol)	0.79	n (Ibmol)	0.05	
MW (lb/lbmol)	24.00	MW (lb/lbmol)	24.00	MW (lb/lbmol)	22.82	
m (lb)	5.03	m (lb)	18.98	m (lb)	1.03	
Molar Enthalpy of Vapor (Btu/Ibmo	-74323.89	Molar Enthalpy of Vapor (Btu/Ibmol	-74322.27	Molar Enthalpy of Vapor (Btu/Ib	<b>mol</b> -42974.84	
Molar Enthalpy of Liquid (Btu/lbmo	-97033.83	Molar Enthalpy of Liquid (Btu/Ibmol	-97037.20	Molar Enthalpy of Liquid (Btu/Ib	<b>mol</b> -56282.41	
λ	22709.94	λ	22714.93	λ	13307.57	
Cp (BTU/lbmol*F)	12.93	Cp (BTU/lbmol*F)	11.79	Cp (BTU/lbmol*F)	11.58	
T1 (F)	450.00	T1 (F)	275.00	T1 (F)	250.00	
Tb (F)	387.14	Тb (F)	387.14	Tb (F)	183.83	
mv (lb)	0.18	mv (lb)	-1.10	mv (lb)	0.06	
η	0.02	η	0.02	η	0.02	
ΔH <sub>c</sub> (Btu/lb)	9276.43	ΔH <sub>c</sub> (Btu/lb)	9276.4328	ΔH <sub>c</sub> (Btu/lb)	15466.01	
E <sub>TNT</sub> (Btu/lb)	2014.62	E <sub>TNT</sub> (Btu/lb)	2014.62	E <sub>TNT</sub> (Btu/lb)	2014.62	
m <sub>TNT</sub> (kg)	0.01	m <sub>TNT</sub> (lb)	-0.046172	m <sub>TNT</sub> (lb)	0.004126	
m <sub>tnt</sub> (g)	7.53					
r <sub>1</sub> (m)	10.00	r <sub>1</sub> (m)	10.00	r <sub>1</sub> (m)	10.00	
z <sub>e</sub> (m/kg <sup>1/3</sup> )	442.42	z <sub>e</sub> (m/kg <sup>1/3</sup> )	-72.19423	z <sub>e</sub> (m/kg <sup>1/3</sup> )	807.9643	
p <sub>s</sub>	0.00	p <sub>s</sub>	0.02	p <sub>s</sub>	0.00	
p <sub>o</sub> (kPa)	0.38	p <sub>o</sub> (kPa)	2.34	p <sub>o</sub> (kPa)	0.21	

# Medium Unit Vapor Deflagration Analysis

V-104	
Pressure (psia)	437.00
Volume (ft3)	6.28
R (ft3*psi/R*lbmol)	10.73
Temp (F)	420.00
Temp (°R)	879.67
n (Ibmol)	0.29
MW (lb/lbmol)	24.88
m (lb)	7.24
Molar Enthalpy of Vapor (Btu/Ibmole	-99681.57
Molar Enthalpy of Liquid (Btu/Ibmole	-118848.41
λ	19166.83
Cp (BTU/lbmol*F)	12.57
T1 (F)	420.00
Tb (F)	405.31
mv (lb)	0.07
η	0.02
ΔH <sub>c</sub> (Btu/lb)	5051.01
E <sub>TNT</sub> (Btu/lb)	2014.62
m <sub>TNT</sub> (lb)	0.0015863
r <sub>1</sub> (m)	10.00
z <sub>e</sub> (m/kg <sup>1/3</sup> )	2101.2652
p <sub>s</sub>	0.00
p <sub>o</sub> (kPa)	0.08

			Large	e Unit		
R-102		V-102		V-103		
Pressure (psia)	445.00	Pressure (psia)	441.00	Pressure (psia)	437.00	
Volume (ft3)	7.95	Volume (ft3)	21.99	Volume (ft3)	3.14	
R (ft <sup>3</sup> *psi/R*lbmol)	10.73	R (ft3*psi/R*lbmol)	10.73	R (ft3*psi/R*lbmol)	10.73	
Temp (F)	450.00	Temp (F)	275.00	Temp (F)	250.00	
Temp (°R)	909.67	Temp (°R)	734.67	Temp (°R)	709.67	
n (Ibmol)	0.36	n (lbmol)	1.23	n (Ibmol)	0.18	
MW (lb/lbmol)	23.22	MW (lb/lbmol)	23.22	MW (lb/lbmol)	21.98	
m (lb)	8.42	m (lb)	28.56	m (lb)	3.96	
Molar Enthalpy of Vapor (Btu/Ibmo	-70089.92	Molar Enthalpy of Vapor (Btu/Ibmol	-70089.05	Molar Enthalpy of Vapor (Btu/Ibmol	-42831.40	
Molar Enthalpy of Liquid (Btu/lbmo	-91518.06	Molar Enthalpy of Liquid (Btu/Ibmol	-91521.66	Molar Enthalpy of Liquid (Btu/Ibmol	-55916.30	
λ	21428.15	λ	21432.61	λ	13084.90	
Cp (BTU/lbmol*F)	12.73	Cp (BTU/lbmol*F)	11.57	Cp (BTU/lbmol*F)	11.37	
T1 (F)	450.00	T1 (F)	275.00	T1 (F)	250.00	
Tb (F)	383.59	Тb (F)	383.60	Tb (F)	185.57	
mv (lb)	0.33	mv (lb)	-1.67	mv (lb)	0.22	
η	0.02	η	0.02	η	0.02	
ΔH <sub>c</sub> (Btu/lb)	10066.33	$\Delta H_{c}$ (Btu/lb)	10066.332	ΔH <sub>c</sub> (Btu/lb)	16030.14	
E <sub>TNT</sub> (Btu/lb)	2014.62	E <sub>TNT</sub> (Btu/lb)	2014.62	E <sub>TNT</sub> (Btu/lb)	2014.62	
m <sub>TNT</sub> (kg)	0.02	m <sub>TNT</sub> (lb)	-0.075988	m <sub>tnt</sub> (kg)	0.016029	
m <sub>TNT</sub> (g)				m <sub>tnt</sub> (g)	16.02935	
r <sub>1</sub> (m)	10.00	r <sub>1</sub> (m)	10.00	r <sub>1</sub> (m)	10.00	
z <sub>e</sub> (m/kg <sup>1/3</sup> )	221.22	z <sub>e</sub> (m/kg <sup>1/3</sup> )	-43.86662	z <sub>e</sub> (m/kg <sup>1/3</sup> )	207.9519	
p <sub>s</sub>	0.01	p <sub>s</sub>	0.04	p <sub>s</sub>	0.01	
p <sub>o</sub> (kPa)	0.76	p <sub>o</sub> (kPa)	3.88	p <sub>o</sub> (kPa)	0.81	

# Large Unit Vapor Deflagration Analysis

V-104	
Pressure (psia)	437.00
Volume (ft3)	9.42
R (ft3*psi/R*lbmol)	10.73
Temp (F)	420.00
Temp (°R)	879.67
n (Ibmol)	0.44
MW (lb/lbmol)	24.46
m (lb)	10.67
Molar Enthalpy of Vapor (Btu/Ibmole	-99911.10
Molar Enthalpy of Liquid (Btu/Ibmole	-119059.82
λ	19148.72
Cp (BTU/lbmol*F)	12.31
T1 (F)	420.00
Tb (F)	406.08
mv (lb)	0.10
η	0.02
ΔH <sub>c</sub> (Btu/lb)	4714.55
E <sub>TNT</sub> (Btu/lb)	2014.62
m <sub>TNT</sub> (lb)	0.0020284
r <sub>1</sub> (m)	10.00
z <sub>e</sub> (m/kg <sup>1/3</sup> )	1643.2973
p <sub>s</sub>	0.00
p <sub>o</sub> (kPa)	0.10

#### Small Unit Revenue

	Small Unit	:	Revenue of Small Unit				÷		
				Flow (lb/hr)	Density (lb/ft^3)	Vol. Flow (ft^3/hr)	Vol. Flow (bbl/hr)	Vol. Flow (gal/hr)	Revenue (\$/hr)
5	Separation	S							
LPG	41.94	lb/hr	LPG	0.21	13.3	0.016	0.0028	0.1183	12.58
Naphtha	10.45	lb/hr	Naphtha	83.72	45.0	1.860	0.3313	13.9164	24.85
			Diesel	132.78	53.0	2.505	0.4462	18.7415	40.16
	HIU		Total Reve	enue					77.59
Naphtha	73.27	lb/hr							
Diesel	132.78	lb/hr							

#### Medium Unit Revenue

M	ledium Un	it		Revenue of Medium Unit					
				Flow (lb/hr)	Density (lb/ft^3)	Vol. Flow (ft^3/hr)	Vol. Flow (bbl/hr)	Vol. Flow (gal/hr)	Revenue (\$/hr)
S	eparation	s							
LPG	366.8	lb/hr	LPG	2.53	16.6	0.152	0.0272	1.1406	110.04
Naphtha	55.84	lb/hr	Naphtha	570.91	45.0	12.687	2.2596	94.9049	169.47
			Diesel	801.29	53.0	15.119	2.6927	113.0953	242.35
	HIU		Total Re	venue					521.86
Naphtha	515.07	lb/hr							
Diesel	801.29	lb/hr							

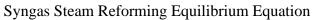
# Large Unit Revenue

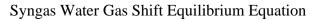
	Large Unit			Revenue of Large Unit					-
				Flow (lb/hr)	Density (lb/ft^3)	Vol. Flow (ft^3/hr)	Vol. Flow (bbl/hr)	Vol. Flow (gal/hr)	Revenue (\$/hr)
S	Separation	s							
LPG	467.6	lb/hr	LPG	2.52	13.9	0.181	0.0322	1.3537	140.28
Naphtha	106.5	lb/hr	Napht	ha 898.08	45.0	19.957	3.5546	149.2913	266.59
			Diese	1385.45	53.0	26.140	4.6558	195.5444	419.02
	HIU		Total	Revenue					685.62
Naphtha	791.58	lb/hr							
Diesel	1385.45	lb/hr							

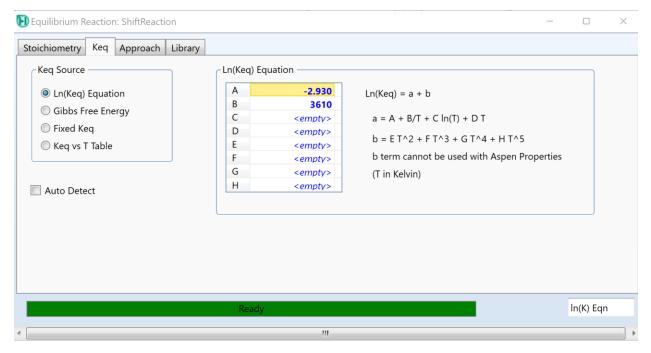
#### Wastewater Treatment Cost

	(lb/hr)	(gal/hr)	(kgal/hr)	\$/hr	
small	1038.40	124.5088	0.124509	0.747053	5235.346
medium	5729.20	686.9542	0.686954	4.121725	28885.05
large	10606.03	1271.707	1.271707	7.63024	53472.73

Equilibrium Reaction: SteamReformi	ng	- (	
Stoichiometry Keq Approach Lik Keq Source Ln(Keq) Equation Gibbs Free Energy Fixed Keq Keq vs T Table Auto Detect	Ln(Keq) Equation A 30.53 B -4.850e+004 C <empty> D <empty> E <empty> F <empty> G <empty> H <empty></empty></empty></empty></empty></empty></empty>	Ln(Keq) = a + b a = A + B/T + C ln(T) + D T b = E T^2 + F T^3 + G T^4 + H T^5 b term cannot be used with Aspen Properties (T in Kelvin)	
	Ready	In(K	) Eqn







#### Equilibrium Reaction: PartialOxidation-1 $\times$ Stoichiometry Keq Approach Library Keq Source -Ln(Keq) Equation А 1.000e+005 Ln(Keq) = a + bLn(Keq) Equation В <empty> Gibbs Free Energy С $a = A + B/T + C \ln(T) + D T$ <empty> Fixed Keq D <empty> b = E T^2 + F T^3 + G T^4 + H T^5 Е 🔘 Keq vs T Table <empty> b term cannot be used with Aspen Properties F <empty> G <empty> (T in Kelvin) Н <empty> Auto Detect ln(K) Eqn 4 .... •

#### Syngas Partial Oxidation Equilibrium Equation

#### Small Unit Syngas Gibbs Reactor

Worksheet	Name	SyngasFeed2	GibbsLiq	Syngas	
Conditions	Vapour	1.0000	0.0000	1.0000	
Properties	Temperature [F]	890.8	1607	1607	
Composition	Pressure [psia]	510.7	510.7	510.7	
PF Specs	Molar Flow [lbmole/hr]	113.0	0.0000	162.3	
	Mass Flow [lb/hr]	2508	0.0000	2508	
	Std Ideal Liq Vol Flow [barrel/day]	312.3	0.0000	365.7	
	Molar Enthalpy [Btu/lbmole]	-4.869e+004	-3.391e+004	-3.391e+004	
	Molar Entropy [Btu/lbmole-F]	45.60	43.36	43.36	
	Heat Flow [Btu/hr]	-5.503e+006	0.0000	-5.503e+006	

#### Medium Unit Syngas Gibbs Reactor

Worksheet	Name	SyngasFeed2	GibbsLiq	Syngas	
Conditions	Vapour	1.0000	0.0000	1.0000	
Properties	Temperature [F]	870.4	1699	1699	
Composition PF Specs	Pressure [psia]	510.7	510.7	510.7	
	Molar Flow [lbmole/hr]	575.2	0.0000	870.3	
	Mass Flow [lb/hr]	1.274e+004	0.0000	1.274e+004	
	Std Ideal Liq Vol Flow [barrel/day]	1564	0.0000	1884	
	Molar Enthalpy [Btu/lbmole]	-4.507e+004	-2.979e+004	-2.979e+004	
	Molar Entropy [Btu/lbmole-F]	45.09	42.37	42.37	
	Heat Flow [Btu/hr]	-2.592e+007	0.0000	-2.592e+007	

#### Large Unit Syngas Gibbs Reactor

 Oibbs Reactor: GBR-100 - Set-1

 Design Reactions Rating Worksheet Dynamics

 Worksheet Conditions

 Name
 SyngasFeed2

 Vapour
 1.0000

 Property of Lamorature [5]
 886.2

Properties	Temperature [F]	886.2	1627	1627	
Composition	Pressure [psia]	510.7	510.7	510.7	
PF Specs	Molar Flow [lbmole/hr]	1134	0.0000	1649	
	Mass Flow [lb/hr]	2.515e+004	0.0000	2.515e+004	
	Std Ideal Liq Vol Flow [barrel/day]	3122	0.0000	3681	
	Molar Enthalpy [Btu/lbmole]	-4.781e+004	-3.288e+004	-3.288e+004	
	Molar Entropy [Btu/lbmole-F]	45.48	43.12	43.12	
	Heat Flow [Btu/hr]	-5.423e+007	0.0000	-5.423e+007	

GibbsLiq

0.0000

Syngas

1.0000

#### Small Unit Syngas Separator

Worksheet	Name	CooledSyngas	Water	Syngaswowater	
Conditions	Vapour	0.8347	0.0000	1.0000	
Properties			212.0	212.0	
Composition	Pressure [psia]	506.7	506.7	506.7	
PF Specs	Molar Flow [lbmole/hr]	162.3	26.83	135.5	
	Mass Flow [lb/hr]	2508	483.6	2025	
	Std Ideal Liq Vol Flow [barrel/day]	365.7	33.20	332.5	
	Molar Enthalpy [Btu/lbmole]	-4.919e+004	-1.205e+005	-3.506e+004	
	Molar Entropy [Btu/lbmole-F]	29.72	17.02	32.24	
	Heat Flow [Btu/hr]	-7.983e+006	-3.233e+006	-4.750e+006	

 $\times$ 

# Medium Unit Syngas Separator

Design React	ons Rating Worksheet Dynamics				
Worksheet	Name	CooledSyngas	Water	Syngaswowater	
Conditions	Vapour	0.8441	0.0000	1.0000	
Properties	Temperature [F]	212.0	212.0	212.0	
Composition	Pressure [psia]	506.7	506.7	506.7	
PF Specs	Molar Flow [lbmole/hr]	870.3	135.6	734.7	
	Mass Flow [lb/hr]	1.274e+004	2445	1.030e+004	
	Std Ideal Liq Vol Flow [barrel/day]	1884	167.8	1716	
	Molar Enthalpy [Btu/lbmole]	-4.517e+004	-1.205e+005	-3.125e+004	
	Molar Entropy [Btu/lbmole-F]	28.97	17.02	31.17	
	Heat Flow [Btu/hr]	-3.931e+007	-1.635e+007	-2.296e+007	

# Large Unit Syngas Separator

Design	Reaction	8 Rating	Worksheet	Dynamics						
Worksh	eet	Name			CooledSyngas	Water	Syngaswowater			_
Conditio	ins	Vapour			0.8371	0.0000	1.0000			
Properti	es	Temperature [F] Pressure [psia] Molar Flow [Ibmole/hr]			212.0	212.0				
Compos	ition				506.7	506.7	506.7			
PF Specs	5				1649	268.7	1381			
		Mass Flow [lb/hr]			2.515e+004	4844	2.031e+004			
		Std Ideal Liq Vol Flow [barrel/day]			3681	332.5	3348			
		Molar Enthalpy [Btu/lbmole]			-4.817e+004	-1.205e+005	-3.409e+004			
		Molar Entropy [Btu/lbmole-F]			29.55	17.02	31.98			
		Heat Flow [	[Btu/hr]		-7.945e+007	-3.238e+007	-4.707e+007			

# Syngas CO<sub>2</sub> Splitter

Design	Rating	Worksheet	Dynamics						
Desi	gn	Split Fraction	ons (Overheads	/Bottoms) ——					
Connect		Com	ponents	Basis	Туре	Syngas1	CO2removed		
Parameters Splits TBP Cut Point			Oxygen	Molar	FeedFrac. to Products	1.000	0.0000	*	
			H2O	Molar	FeedFrac. to Products	1.000	0.0000		
	ser Variables			Nitrogen	Molar	FeedFrac. to Products	1.000	0.0000	Ξ
Votes			СО	Molar	FeedFrac. to Products	1.000	0.0000		
			CO2	Molar	FeedFrac. to Products	5.000e-002	0.9500		
			Hydrogen	Molar	FeedFrac. to Products	1.000	0.0000		
			Ethane	Molar	FeedFrac. to Products	1.000	0.0000		
			Propane	Molar	FeedFrac. to Products	1.000	0.0000		
				Methane	Molar	FeedFrac. to Products	1.000	0.0000	
			n-Butane	Molar	FeedFrac. to Products	1.000	0.0000		
			n-Pentane	Molar	FeedFrac. to Products	1.000	0.0000		
			n-Hexane	Molar	FeedFrac. to Products	1.000	0.0000	*	
		Se	et to 1.0000	Set t	o 0.0000				
	Delete				ОК			Ignore	