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Short title: Conjugation of para-aminobenzoic acid derivatives with serum proteins

Biomolecular study and conjugation of two para-aminobenzoic acid derivatives with serum proteins: drug binding efficacy and protein structural analysis

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Abstract

Two aminobenzoc acid viriatives DAB-0 and DAB-1 showed distinct biological properties on nurin bladder cancer (BCa) cell line MB49-I. In contrast to DAB-1, DAB-0 does not posses) any anti-inflammatory activity and is less toxic. Furthermore, DAB-0 does not nurifiere with INF γ -induced STAT1 activation and TNF α -induced IkB phosphorylation, while DAB-1 does. In order to rationalize these results, the binding efficacy of DAB-0 and DAB-1 with serum proteins such a human serum albumin (HSA), bovine serum albumin (BSA) and beta-lactoglobulin (β -LG) was investigated in aqueous solution at physiological pH. Multiple spectroscopic methods and thermodynamic analysis were used to determine the binding efficacy of DAB-0 and DAB-1 with serum proteins. Drug-protein conjugation was observed *via* through ionic contacts. DAB-1 forms stronger adducts than DAB-0, while β -LG shows more affinity with the order of stability β -LG > BSA > HSA. The stronger complexation of DAB-1 with serum proteins might account for its biological potential and transport in the blood. The binding efficacy ranged from 40 to 60%. Dajor Alterations of protein secondary structures were detected upon drug complexation. Shum proteins are capable of delivering DAB-1 *in vitro*.

Keywords: DAB-0, DAB-1, serum protein, delivery, binding efficacy, thermodynamic analysis

<u>Abbreviations</u>: BSA, bovine serum albumin; (DAF40), N'-[4-(2,5-dioxo-pyrrolidin-1-yl)benzoyl]-hydrazine carboxylic acid *tert*-butyl ester (DAB-1), N'-[4-(2,5-dioxo-2,5-dihydropyrrol-1-yl)-benzoyl]-hydrazine carboxylic acid *tert*-butyl est ;HSA. Human serum albumin; FTIR, Fourier transform infrared; β LG, beta-lactoglobulin.

Introduction

Inflammation plays an import ant role in the initiation and progression of several types of carcer, prostate cancer and breast cancer (Colotta et al., 2009; Sui cancers including bladd the e is a renewed scientific interest in the development of novel antiet al., 2017). Hend atrient strategies able either to circumvent cancer inflammation and/or to inflammato ative impact of known drugs (Zhu et al., 2012; Nakamura et al., 2017; Dai et enb nce the (17). Recently, we studied the anti-inflammatory and anticancer/antimetastatic al.. properties of small aminobenzoic acid derivative namely, DAB-1 (Hamelin-Morrissette et al., 2015). In that paper, we reported that both portions of the molecules are necessary for its biological action. Indeed, the maleimide as well as the *tert*-butoxycarbonyl moieties of DAB-1 are critical for its anti-inflammatory activity and its anticancer properties. Furthermore, we showed that DAB-1 (Scheme 1) efficiently inhibits cytokine-induced expression of pro-inflammatory genes and motility in human M1 macrophages (Hamelin-Morrissette et al., 2015). We disclosed that DAB-1 could inhibit the production of NO, a pro-tumor inflammatory mediator, in murine urothelial bladder carcine lls and ha macrophages. Because the maleimide portion is required for its anti-affagmatory properties, DAB-1 was hydrogenated (Berube & Reyes-Moreno, 20 7). The hydrogenation and the loss of double bound in the maleimide moiety w confirmed by standard spectrometric analysis (Berube & Reyes-Moreno, 2017). The and-inflammatory properties of DAB-0 was investigated by its ability to inhibit No production in murine macrophage cells (Berube & Reyes-Moreno, 2017). However, once the maleimide portion is hydrogenated the molecule, called DAB lose its anti-inflammatory properties (Berube & Reves-Moreno, 2017). Therefore, in order to inderstand the possible role and implication of the maleimide portion of DAB-1, the comparison with DAB-0 needed to be investigated. In this study, to further investigat the anticancer potential of these molecules, the antiand inflammatory activity and cell signaling pathways were proliferative activity, determined in the adder cancer (BCa) cell line MB49-I, a highly invasive and odel (Hamelin-Morrissette et al., 2015). As well, the interactions of tumorigenic B The serum proteins were investigated to provide additional insight into its these mole cules transport in the blood. mð

Secum proteins contain multiple binding sites with different affinities and can transport diverse drugs, fatty acids, steroid hormones and many other bioactive materials (Yang et al., 2014; Ghuman et al., 2005; Kratochwil et al., 2002; Ahmed-Ouameur et al., 2006). Serum albumins are emerging as versatile protein carriers for drug targeting systems and for improving the pharmacokinetic profile of peptides or protein-based drugs (Elsadek & Kratz, 2012; Kratz & Elsadek, 2012; Kratz, 2008). In order to evaluate the potential application of serum proteins in delivery of DAB-1, it was of interest to compare the binding pricacy of this drug to serum proteins in aqueous solution. The reference inactive analyte DAB-0 was also assessed to contrast the results obtained with DAB-1. Receipt statices showed encapsulation of testosterone and its dimeric derivatives by serum proteint (Chanphai et al., 2015a; Chanphai et al., 2015b). Carrier proteins such as He' DFA and β -LG, show different hydrophobicity (Akdogan et al., 2012; Jamson et al., 2012); Chanphai et al., 2018) and exhibit different affinity towards drug interactions

The biological activity and conjugation of DAB/0 and DAB-1 (Scheme 1) with human and bovine serum albumins and folactographian was investigated, using multiple spectroscopic methods and thermodynamic analysis. Furthermore, the possibility of delivery of DAB-0 and DAB-1 by serum processes discussed herein.



Biological evaluation

Cell culture

Biological assays were performed using the murine bladder cancer (BCa) cell line MB49-I is I, as we previously described (Hamelin-Morrissette et al., 2015). The cell line MB49-I is a highly invasive and tumorigenic BCa cell model that was developed by successive *in vivo* passages of MB49 primary tumors (Fabris et al., 2012). The cells were maintained in RPMI medium supplemented with 10% heat-inactivated fetal bovine serum (FBS) and containing 1 mM sodium pyruvate, 10 mM 4-(2-hydroxyethyl) prperazue-1ethanesulfonic acid (HEPES) and 50 μ g/mL gentamycin (referred s 10) FBS RPMI-1640). The cells were maintained at 37 °C in a moisture-saturate lampsphere containing 5% CO₂.

Evaluation of cell proliferation by the MTT assay

To evaluate and compare the anti-proliferation between DAB-0 and DAB-1, cell acti viability/proliferation MTT assays were orn ed as previously described (Carmicheal et al., 1987; Dallagi et al., 2015; Duresne et al., 2011; Dumas et al., 2013; Leduc et al., $\frac{3}{\text{cel}}$ s/well) were plated in 96-well plates in 100 µL 2012). Briefly, MB49-I cells (5 ed or 24 h at 37°C and 5% CO₂. Cells were pretreated 10% FBS RPMI-1640 and call vehicle (DMSO) or compounds DAB-0 and DAB-1 at for a period of 30 mi Wh figure $(15, 30, 60 \text{ and } 90 \mu \text{M})$, and then activated for 24 h different increasin with 5 ng/mL IFN and 25 ng/mL TNF α . At the end of the culture period, the culture each experimental point was transferred into new 96-well plates to assess NO media for on (see next sub-section), and the well containing the cells were replenished with prod 100 µL phosphate-buffered solution (PBS) to assess MTT assays. For this purpose, 10 µL of 5 mg/mL methylthiazolyldiphenyl-tetrazolium bromide (MTT) solution was added to each well containing 100 µL PBS. After a 3-h incubation period with MTT reagent, 100

µL of MTT solubilization buffer (10% SDS in 10 mM HCl) was added and plates were placed overnight in the cell incubator before absorbance measure. The optical density was read at 580 nm using the Microplate Reader Manager (from Bio-Rad Laboratories).

Evaluation of NO production by the Griess reagent method

NO production was measured using the Griess reagent method as previ ibed [26]. This method involves the detection of nitrite ions (NO₂) formed by the spontaneous oxidation of NO under physiological conditions. According to the manufacturer protocol (Life Technologies; # G-7921), equal volumes of sulfacilic and N-(1-naphthyl) ethylenediamine are mixed together to form the Grie s reasent. In contact with NO_2^{-} , present in the supernatant from the above pretreat doubd activated MB49-I cells, the sulfanilic acid is converted to a diazonium al, which in turn is coupled to N-(1ink coloration that is measured with a naphthyl) ethylenediamine to produce a HT) 548 nm. spectrophotometer (Biotek, synergy

Cell signaling studies

MB49-I cells ($25 \times 10^{\circ}$ pell//well) were pretreated for 30 min with vehicle (DMSO) or compounds DAP-trane DAB-1 at different increasing concentrations (15, 30 and 60 μ M), and then remed and recovered after 15 min of activation with 5 ng/mL IFN γ or after remin or activation with 25 ng/mL TNF α . Cell lysates were prepared and analyzed by immunoblotting as described (Dallagi et al., 2015; Dufresne et al., 2011; Dumas et al., 2013; Leduc et al., 2012). Briefly, protein samples were resolved by SDS-PAGE under reducing conditions and transferred onto a PVDF membrane. Blots were first probed with rabbit polyclonal antibodies against phospho-STAT1 (pSTAT1) and p-I κ B (both at 1:2000) overnight at 4°C. Blots were then incubated with HRP-conjugated goat antirabbit IgG Ab (1:3000) for 1 h at room temperature. The same blots were stripped and then probed with anti-STAT1 (at 1:1000) and anti- β -actine (1:4000) antibodies. It both cases, probed molecules were visualized using an enhancement chemiltanines ence detection kit (Thermo Fisher Scientific).

Chemistry

Materials

HSA, BSA fraction V and β -lactoglobulin (A variant pure 90) were purchased from Sigma Chemical Company (St-Louis, MO) and used as supplied. DAB-0 and DAB-1 were synthesized in our laboratory using a three way (DAB-1) or four-step (DAB-0) reaction sequence from *para*-aminobenzoic acid a terperiod earlier (Leduc et al., 2012; Belgorosky et al., 2014). The compounds were fully characterized by IR, ¹H-NMR, ¹³C-NMR spectroscopy and by mass spectrometry Other chemicals were of reagent grade and used without further purification

Preparation of DAB-0 and AB-1-protein complexes

Appropriate amounts of protein (BSA, HSA or β -LG) were dissolved in aqueous solution (125 μ M) containing 10 mM Tris-HCl (pH 7.2). BSA and HSA concentrations were determined spectrophotometrically using the extinction coefficient of 36500 M⁻¹ cm⁻¹ at 280 nm Paratetet al., 1998). β -LG concentration was determined spectrophotometrically using the extinction coefficient of 17600 M⁻¹ cm⁻¹ at 280 nm (Collini et al., 2000). DAB-0 and DAB-1 (200 μ M) were dissolved in water-ethanol (50/50%) and then diluted by serial dilution to different concentrations in Tris-HCl (pH 7.2).

UV spectroscopy

The UV-Vis spectra were recorded on a Cary 60 UV-Visible spectrophotometer with a slit of 2 nm and scan speed of 400 nm min⁻¹. Quartz cuvettes of 1 cm were used. The absorbance measurements were performed at pH 7.2 by keeping the concentration of protect constant (60 μ M), while increasing drug concentrations (1 μ M to 60 μ M). The binding contants of drug-protein complexes were determined as reported (Zhong et al., 2004).

FTIR spectroscopy

Infrared spectra were recorded on a FTIR spectromete (Impact 420 model), equipped with deuterated triglycine sulphate (DTGS) detector and KBirbeam splitter, using AgBr windows. Solution of DAB drug was added dropy he rotein solution with constant stirring to e to ensure the formation of a homogeneous solution and to reach the target drug concentrations of 15, 30 and 60 µM with a final proton concentration of 60 µM. Interferograms were accumulated over the spectral large $1000-600 \text{ cm}^{-1}$ with a nominal resolution of 2 cm⁻¹ and (protein solution + drug solution) – (protein solution)] 100 scans. The differen ombination mode around 2300 cm^{-1} , as standard (Dousseau et were generated usig ing difference spectra, this band was adjusted to the baseline level, al., 1989). WI normalize difference spectra. Analysis of the secondary structures of protein and in order to complexes were carried out as reported (Byler & Susi, 1986; Beauchemin et al., the The curve-fitting analysis was performed using the GRAMS/AI Version 7.01 2007)software of the Galactic Industries Corporation.

Molecular modeling

The structure of free HSA (PDB id:1AO6, chain A) obtained by X-ray crystallography was used as a template (Sugio et al., 1999). The structure of BSA was predicted by automated homology modeling using SWISS-MODEL Workspace from the amino acid sequence NP-851335 [Arnol et al., 2006; Rost, 1999; Bourassa et al., 2011). The two proteins state 78.1% of sequence identity, which is sufficient to obtain reliable sequence alignment. The β LG structure was obtained from the literature report (Loch et al., 2015). The docking studies were performed with ArgusLab 4.0.1 software (Mark A. Thompson, Tanaria Software LLC, Seattle, WA, http://www.arguslab.com). Three-dimensional structure of DAB-0 and DAB-1 were generated from PM3 semi-empirical calculations, using Chan3D Ultra 6.0.

Statistical analyses

For all biological assays, data were presented as much \pm SEM from three independent experiments. Data were analyzed by one-way WOVA followed by Bonferonni post-test using Prism software, version 3.03 (Graphi d, San Diego, CA). *p* values of ≤ 0.05 were considered to indicate statistical significance.

Results and discussion

Anti-proliferative activity of PAB-0 and DAB-1

The anti-proliferative activity of the molecules was evaluated using the MTT assay on murine BCa cell her MI49-I (Hamelin-Morrissette et al., 2015). The results from the anti-proliferative activity assay are shown in **Fig. 1A**. DAB-0 is essentially non-toxic up to the highest concentration used; 90 μ M. However, DAB-1 displays significant cytotoxicity at concentrations of 60 and 90 μ M, but remains not toxic up to a concentration of 30 μ M. The anti-proliferative result clearly contrast the discrepancy between both DAB molecules; DAB-1 is more cytotoxic than DAB-0, the latter being nontoxic towards BCa cells.

Anti-inflammatory activity

The anti-inflammatory activity was evaluated by measuring NO production in murine BCa cell line MB49-I stimulated by two powerful pro-inflammatory cytokines, INF γ and TNF α , and using the Griess reaction methodology (Belgorosky et al., 2014). Figure 1B presents the results obtained from this study. DAB-0 did not decrease NO production by the calls at any of the concentration tested. Contrarily, DAB-1 significantly decreased 10 production starting at a concentration as low as 30 μ M. At 15 μ M no effect was detected for both compounds. This assay confirms that DAB-1 possesses significant inti-inflammatory activity in comparison to DAB-0 that displays none.



Figure 1. Graphical representation of relative cell viability (A) and NO production (B) in the murine BCa cell line M8(9-1) which were pretreated for 30 min with vehicle (DMSO) or the compounds D4B-1 and DAB-0 at different concentrations, and activated for 24 h with the provinflammatory cytokines IFN γ and TNF α . Mean ± SEM was obtained from three independent measurements. ** *p* < 0.01 denote significant difference compared to control (LAMSO) for each DAB-1 or DAB-0 dose. *Cell signaling surfles*

To tudy in anti-cancer potential of DABs, we investigated the activation status of two major pro-inflammatory and pro-tumor signaling pathways in MB49-I cells; IFN γ /STAT3 and TNF α /NF κ B. In order to test the first signaling pathway, cells were pretreated with DMSO (control) and activated for a period of 15 minutes with INF γ triggering STAT3

activation. Naturally, PBS stimulation did not activate STAT-1. In contrast, in response to INF γ activation, pretreatment with DAB-1 at 30 and 60 μ M drastically decreases the formation of p-STAT1 in comparison to DAB-0, which displays no effect at any of the concentrations tested (**Figs 2A and 2B**).

For the second signaling pathway, the cells were activated with TNF α for 5 minutes initiating the production of p-I κ B. Control cells were treated with PBS without effect on NF κ B activation. As shown in figure 2C and 2D, DAB-1 inhibits NF κ B activation at concentration as low as 15 μ M. In comparison, DAB-0 only slightly inhibited the production of p-I κ B at the highest concentration tested (60 μ M).

Taken together, these results suggest that DAB-1 but not DAB-0, effectively inhibited IFN γ /STAT3 and TNF α /NF κ B signaling provage activation in MB49-I cells. Hence, DAB-1 is an anti-inflammatory molecule of a b inhibition of NO production and has the ability to inhibit pro-inflammatory are pro-tumor pathways by the inhibition of STAT-1 and I κ B phosphorylation/activation



Figure 2. Representative images (A) and graphical analysis (B) of pl sphorylated STAT1 (p-STAT1) in MB49-I cells pretreated for 30 min with vehicle (D) (ISO) or the compounds DAB-1 and DAB-0 at different concentrations, and accurated for 15 minutes with PBS (-) or 5 ng/mL IFN γ (+). Representative images (U) and graphical analysis (D) of phosphorylated I κ B (p-I κ B) in MB49-I cells for 30 min with vehicle (DMSO) rea or the compounds DAB-1 and DAB-0 fferent concentrations, and activated for 5 minutes with PBS (-) or 25 ng/mL TNF α (-). Mean \pm SEM was obtained from three 0.07 and ** p < 0.05 denote significant difference independent measurements. р compared to control (DMSO) AB-1 or DAB-0 dose.

Stability of DAB-0 and DAL1-periode complexes by UV spectroscopy

The DAB-0 and DAB1 complexation with serum proteins induces major changes in the absorption spectra of plotent and this can be used to calculate the drug-protein binding constants. The UV pectra of drugs with HSA, BSA and β -LG complexes are presented in **Figs 3** and **4**. Urug conjugation with protein occurred with an increase in the intensity of protein band at 280 nm.

The drug-protein binding constants were calculated as described in materials and methods (Zhong et al., 2004), using plots of $1/(A-A_0)$ vs (1/drug concentrations) (**Figs 3** and **4**), where A₀ is the initial absorbance of the free protein at 280 nm and A is the recorded

absorbance of complexes at different drug concentrations (**Figs 3, 4** and **Table 1**). The results showed that strong drug-protein interactions were observed with the order of stability β -LG > drug-BSA > drug-HSA with DAB-1 forming stronger conjugates than DAB-0 (**Figs 3, 4** and **Table 1**). In part, this could explain the observed biological activity of α AB-1 in comparison to DAB-0, the inactive analog. Experimental evidence regarding hydrophobic, hydrophilic or H-bonding contacts comes from the thermodynamic analysis or drug-protein complexes discussed below.

Complexes	Temperature	$constant K(mol/L)^{-1}$
	(K)	
DAB - 0 – HSA	298.15	$1.18 \ge 10^5$
	308.1	8.52x 10 ⁴
	318.1.	6.23×10^4
DAB - 1 – HSA	28 15	$1.30 \ge 10^5$
	30.15	$9.15 \ge 10^4$
	318.1	$7.38 \ge 10^4$
DAB - 0 – BSA	398.15	9.32×10^4
	3 8.15	$7.50 \ge 10^4$
	318.15	$5.60 \ge 10^4$
	298.15	9.59×10^4
DAB - 1 – BSA	308.15	8.93×10^4
	318.15	$8.42 \ge 10^4$
	298.15	$4.00 \ge 10^5$
DAB - 0	308.15	$3.68 \ge 10^5$
	318.15	3.35×10^5
DATE - LG	298.15	5.95×10^5
	308.15	$4.21 \ge 10^5$
	318.15	3.39×10^5

Table 1. Variations of the binding constants for DAB -0 and DAB 1 yith serum proteins at different temperatures



Figure 3. UV-Visible spectra of DAB-0 and its conjugates with HSA (**A**), BSA (**B**) and β -DF (Cast 60 μ M (a) and its DAB-0 complexes (b-i) for DAB-1 at 1, 5, 10, 20, 30, 40, 50 and 60 μ M; Inset: plot of 1/(A-A₀) vs (1/ drug concentration) and binding constants (*K*) for drug-protein conjugates.





Figure 4. UV-visible spectra of DAB-1 and its conjugates with HSA (A), BSA (B) and β -LG C at 60 μ M (a) and its DAB-1 complexes (b-i) for DAB-1 at 1, 5, 10, 20, 30, 40, 50 and 60 μ M; Inset: plot of 1/(A-A₀) vs (1/ drug concentration) and binding constants (*K*) for drug-protein conjugates.

Thermodynamic analysis of DAB-0 and DAB-1-protein adducts

DAB-0 and DAB-1 complexes with serum proteins were further characterized by thermodynamic parameters. Based on the data of ΔH^0 and ΔS^0 , the drug-protein interaction can be elucidated (Ross & Subramanian, 1981; Bose, 2016). The thermodynamic trameters for the interaction of drug and protein complexes at 298.15, 308.15 and 318.1 **K** are presented in Figs 5, 6 and Table 2. The negative sign of ΔG^0 shows that the binding process between drug and protein is spontaneous. Furthermore, all the drugrotein complexes have negative ΔH^0 , which means the complex formation between profein and drug is an exothermic reaction. The negative ΔH^0 and positive ΔS^0 for drug protein conjugates indicate that ionic contacts are observed in the drug-protein topple ation (Ross & Subramanian, 1981; Bose, 2016). A detailed thermodynam c analysis of drug-protein interactions shows ΔH ΔS^0 and ΔG^0 in determining what type of the importance of the binding constant, interaction is predominant in these drug-provin conjugates (Ross & Subramanian, 1981; Bose, 2016; Bekale et al., 2015 Bekke et al., 2014). For drug-protein, the enthalpy provides more contribution to V t an entropy, which indicates that the binding process is enthalpy driven (Table The binding efficag B-0 and DAB-1 complexes by serum proteins was determined al., 2011). The binding efficacy was estimated to be 40%-50% for as reported (C for drug-BSA and 45-60% for drug- β -LG systems. The loading efficacy drug-HSA DAB-1-protein than DAB-0-protein complexes. Hence, a greater amount of can be transported by serum proteins. DAB



Figure 5. lnK vs 1/T for DAB-0 with HSA (A), BSA (B) and β -LG (C) conjugates.



Figure 6. lnk vs 1/T for DAB-1 with HSA (A), BSA (B) and β -LG (C) conjugates.

	Thermodynamic parameters			
Complexes		ΔS^0	ΤΔS ⁰	ΔG^0
	(kJ. mol ⁻¹)	(J. mol⁻¹. K⁻¹)	(kJ. mol ⁻¹)	(kJ mol
DAB - 0 – HSA	-10.92	5.55	1.65	-12.5 X (258.15 K)
			1.71	- 2.63 (308.15 K)
			1.76	-12.68 (318.15 K)
DAB - 1 – HSA	-9.77	9.70	2.8	-12.66 (298.15 K)
			2.99	-12.76 (308.15 K)
			3.09	-12.85 (318.15 K)
	>			
DAB - 0 – BSA	-8.71	12.14	3.62	-12.33 (298.15 K)
	XO		3.74	-12.46 (308.15 K)
	Q		3.86	-12.58 (318.15 K)
	2.23	33.02	10.11	12 35 (208 15 K)
	-2.23	55.92	10.11	-12.33 (298.13 K)
			10.45	-12.69 (308.15 K)
			10.79	-13.03 (318.15 K)
$DAB - 0 - \beta - LG$	-3.03	36.41	10.85	-13.89 (298.15 K)
			11.22	-14.25 (308.15 K)

Table 2. Thermodynamic parameters for DAB -0 and DAB -1 with serum proteins

			11.58	-14.62 (318.15 K)
DAB - 1 – β - LG	-9.65	15.58	4.64	-14.29 (298.15 K)
			4.80	-14.45 (198.15 K)
			4.96	◆14•1 (318.15 K)
				\sim
			ſ	

Structural analysis of DAB-0 and DAB-1-protein by FTIR spectroscopy

The conjugation of DAB-0 and DAB-1 with HSA, BSA and Lywas analyzed by infrared spectroscopy and its derivative methods. Drug interactions alter protein roten amide I band at 1659-1657 cm⁻¹ conformation and induce spectral change ein 1546 (mainly C=O stretch) and amide II band 1545 cm⁻¹ (C-N stretching coupled with N-H bending modes) (Krimm & Bartlekar, 1955; Tian et al., 2005; Grdadolnik, 2011). The intensity variations of protein amile I and amide II bands obtained by difference spectra protein solution)] are shown in **Figs 7, 8** and **9**. [(protein solution + drug s ptrations (15 to 60 µM), major intensity changes were observed At low and high drug c amide II, in the difference spectra of the drug-HSA and drugfor the protein am inplexes (Figs 7, 8, 9 and diffs 15 and 60 μ M). The positive features BSA and drug re due to the increase in intensity of amide I and amide II bands located in the observed spectra at 1720-1500 cm⁻¹ (Figs 7, 8, 9 and diffs 15 and 60 μ M). As drug diff. ren. ration increased (60 μ M), more intensity variations of protein amide I and amide II conce were observed with stronger features around 1700-1500 cm⁻¹ (Figs 7 and 8, diffs 60 μ M). As drug concentration increased more perturbations of the amide I and amide II were observed (diffs 60 μ M). The spectral alterations observed are due to changes in the intensity of the amide I and amide II band, upon drug interactions with protein C-O, C-N and NH groups (hydrophilic contacts) and linked to variations of protein conformation discussed below.

The secondary structures of the free HSA, BSA and β -LG and their drug onjugates are shown in **Table 3**. The free HSA has 57 % α -helix (1656 cm⁻¹), β -showt 14 628 and 1617 cm⁻¹), turn structure 13 % (1679 cm⁻¹), β -antiparallel 4 % (1679 cm⁻¹), β -antiparallel 89 cm and random coil 12 % (1637 cm⁻¹) (**Table 3**). The free BSA contains α -heli (1656 cm⁻¹), β -sheet 16% (1612 and 1626 cm⁻¹), turn 12% (1678 cm⁻¹), β -antipallel 3° (1691 cm⁻¹) and random coil 6% (1638 cm⁻¹) (**Table 3**). The free β -LG has maker β -seet 58% (1640, 1623), α -helix 11% (1655 cm⁻¹), turn 14% (1667 cm⁻¹) and β -anti-ballel 17% (1679) (**Table 3**). Upon DAB-0 and DAB-1 interactions, a decre e of other from 57% (free HSA) to 37-34% with an increase in random and turn structures (drug-HSA) were observed (**Table 3**). Similarly, for BSA a decrease of α -helix from 63% free BSA) to 41-38% with an increase of turn and random coil structures (drug BSA) were observed, upon drug complexation (Table 3). set β -sheet from 58% (free β -LG) to 46-32% with an increase However, for β -LG a de of α -helix, turn and structures (drug- β -LG) were observed, upon drug conjugation (Table 3). The nowed that major conformational changes occurred for HSA, BSA inplexes, leading to a partial protein destabilization (Essemine et al., 2011; and β -LG лug 2014; Yeggoni et al., 2014). Similar protein conformational changes were Go d for HSA, BSA and β -LG upon various ligands complexation (Nerusu et al., 2017; obser Hasni).

	Amide I (cm ⁻¹) components				
Sample	α-helix (±2 1654-1600	β-helix (±1 1614-1637	random coi (±1) 1638-1648	turn (±2) 1670-1678	β-antiparallel (±1) 1680-1691
Free HSA (250 µm)	57	14	12	13	·.0*
DAB - 0 - HSA (250 μm)	37	14	26	14	9
DAB - 1 - HSA (250 µm)	34	13	23	20	10
Free BSA (250 µm)	63	16	6		3
DAB - 0 - BSA (250 µm)	41	11	19	16	13
DAB - 1 - BSA (250 μm)	38	10	NO	23	7
Free β – LG (250 μm)	11		0	14	17
DAB -0-β - LG (250 μm)	32	Q Q	6	12	4
DAB - 1 - β – LG (250 μm)		32	9	13	9
PC C	5				

Table 3. Percentage changes in secondary structure contents for serum proteins and theirDAB-0 and DAB-1 complexes in hydrated film at pH 7.



Figure 7. FTIR spectra in the region of 1800-600 cm⁻¹ of hydrated films (pH 7.2) for free HSA for with DAB-0 (**A**) and DAB-1 (**B**) (drug 60 μ M) and the difference spectra (diff.) of the protein conjugates (bottom two curves) obtained at different drug concentrations (indicated on the figure).



Figure 8. TTIR spectra in the region of 1800-600 cm⁻¹ of hydrated films (pH 7.2) for free BSA with LAB-0 (**A**) and DAB-1 (**B**)(drug 60 μ M) and the difference spectra (diff.) of drug-protein conjugates (bottom two curves) obtained at different drug concentrations (indicated on the figure).



Figure 9. FTIR spectra in the region of 1800-600 cm⁻¹ of hydrated films (pH 7.2) for free β -L f wet DAB-0 (A) and DAB-1 (**B**) (drug 60 μ M) and the difference spectra (diff.) of drug-protein conjugates (bottom two curves) obtained at different drug concentrations (indicated on the figure).

Location of DAB-0 and DAB-1 binding sites on HSA, BSA and b-LG by docking

Docking results showed that in the DAB-0 and DAB-1 binding HSA via different amino acids in drug-HSA adducts, DAB-0 is closed to Arg-186, Ile-142, Leu-182, Leu-185, Lys-137, Met-123, Phe-134, Phe-165, Tyr-138, Tyr-161 with the free binding energy of -9.7 kcal/mol, while DAB-1 is located in the vicinity of Ala-210, Glu-479, Leu-81, vs-199. Phe-206, Phe-211, Ser-202, *Ser-480, Trp-214, *Val-482 (H-bonding) vith binding energy of -9.61 kcal/mol (Fig. 10A and Table 4). In drug-BSA complexes, DAB-0 binds *Arg-209 (H-bonding), Asp-142, Cys-147, *Glu-206, Ile-16 Let 139, Leu-162, Leu-202, Phe-150, Phe-141, Trp158, Tyr-161 with the free energy of -9.57 kcal/mol, whereas in DAB-1-BSA, drug is surrounded by*Asp-142 (H-B nding), Cys-147, Ile-165, Leu-139, Leu-146, Leu-162, Lys-140, Phe-60, Phe-160, **Trp-158, Tyr-161** with the free binding energy of -11.14 kcal/mol (Fig. 10B and Table 4). Similarly, in the drug- β -LG adducts, DAB-0 is located near Asn-90, In 56, Ile-71, Ile-84, Leu-46, Leu-54, Leu-58, Leu-103, Leu-122, Met-107, Phe-105, WI-41, Val-43, Val-92 with the free binding energy of 11.05 kcal/mol, while DAH 1 binds to Asn-90, Asp-85, Ile-56, Ile-71, Ile-84, Leu-46, Ph. 105, Val-41, Val-43, Val-92 with the free binding energy of Leu-54, Leu-58, Leu-12 -10.77 kcal/mol (Fi d Table 4). As it is shown in Fig. 10 and Table 4, DAB-0 and by both hydrophilic and hydrophobic amino acids residues. It is DAB-1 are s noning that drugs bind to either cysteine or methionine containing sulfur groups worth mer (Tabl It is interesting to note that H-bonding systems are established in the drug-HSA g-BSA complexes that are not present in drug- β -LG adducts (Fig. 10 and Table 4). and d The free binding energy showed that β -LG forms more stable conjugates than HSA and BSA which is in agreement with our experimental results (Tables 1 and 4).



Figure 10. Best docked conformations of drug–HSA (A), drug–BSA (B) and drug β -LG

conjugates (C) with the free binding energy.



Table 4. Amino acid residues involved in DAB-0 and DAB-1 with serum proteins interactions and free binding energy for the best selected docking positions.

Complexes	Amino acid residues involved in the interaction	
DAB - 0 – HSA	Arg-186, Ile-142, Leu-182, Leu-185, Lys-137, Met-123, Phe-134, Phe-165, Tyr-138, Tyr-161	- 9.47
DAB - 1 – HSA	Ala-210, Glu-479, Leu-481, Lys-199, Ptc-206, Phe-211, Ser-202, *Ser-480, Trp-210, *Yar-482	- 9.61
DAB - 0 – BSA	*Arg-209, Asp-142, Cys-141, *Clu-206, Ile- 165, Leu-139, Leu-162, Leu-202, Phe-150, Phe- 141, Trp158, Tyr-161	- 9.57
DAB - 1 – BSA	*Asp-142, Cys-147, In-100, Leu-139, Leu-146, Leu-162, Lys 140, Pie-60, Phe-150, Phe-157, Trp-158, Tyr-161	- 11.14
DAB - 0 – β-LG	Asn-90, Jul 56, Ile-71, Ile-84, Leu-46, Leu-54, Leu-58, Leu-108, Leu-122, Met-107, Phe-105, Val-41, Val-3, Val-92	- 11.05
DAB - 1 – β-LG	Vsn-2 Asp-85, Ile-56, Ile-71, Ile-84, Leu-46, Leu 54, Leu-58, Leu-122, Phe-105, Val-41, Val- 4. Val-92	- 10.77

Concluding remarks

The distinct biological properties shown by DAB-0 and DAB-1 (anti-proliferative, antiinflammative and regulation of cell signaling pathways) prompt us to investigate the intervation of these molecules with transport proteins. It is well known that serum proteins are widely used in clinical settings as a drug delivery system due to their potential for improving targeting while decreasing the side effects of drugs. DAB-0 and DAB-1 bind serum proteins via ionic contacts with β -LG forming more stable conjugates than HSA and BSA. DAB-1 forms stronger protein adducts than DAB-0, which could at least partly explain the overall inactivity of the latter drug. The binding efficacies were 40% to 50% for DAB-0 and 45% to 60% for DAB-1-protein adducts. Drug complexation induced major perturbations of protein conformations leading to structural destabilization. Server proteins can be used for the delivery of DAB-1 *in vitro*.

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