# The Yamato nakhlite consortium

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*Abstract:* Among 3550 meteorite samples found near the Yamato Mountains area during the field season of 2000–2001, three meteorites, Yamato (Y) 000593, Y000749, and Y000802 of total weight of about 15 kg, are identified as new paired nakhlites. In ordered to improve our knowledge of the nakhlite parent body, possibly Mars, a coordinated consortium study of the Yamato nakhlites was organized. We here present an outline of the Yamato nakhlite consortium and implications for Martian geological history on the basis of ongoing studies.

key words: nakhlites, Martian meteorites, Y000593, Y000749, Y000802

### 1. Introduction

The wintering party of the 41st Japanese Antarctic Research Expedition conducted a meteorite search during the Antarctic summer field season of 2000–2001 in the bare ice field near the Yamato Mountains area (Imae *et al.*, 2002a). Among 3550 meteorites found, Yamato 000593 (hereafter Y000593), Y000749, and Y000802 were identified as nakhlites (Kojima and Imae, 2001; Imae *et al.*, 2002b; Kojima *et al.*, 2002). Nakhlites are rare clinopyroxene-rich cumulate rocks of probable Martian origin (*e.g.*, McSween, 1994). A total of eight meteorites, Nakhla, Lafayette, Governador Valadares, North West Africa (hereafter NWA) 817 (Sautter *et al.*, 2002), Yamato (Y000593, Y000749, and Y000802), and NWA998 (Irving *et al.*, 2002; Russell, 2003), have thus far been identified as nakhlites.

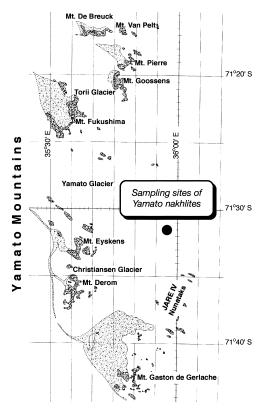
After the announcement of the new Antarctic nakhlites (Kojima and Imae, 2001), more than twenty sample requests for investigation were received. At a meeting held on 19 March 2002 the Committee on Antarctic Meteorite Research recognized the need for a coordinated consortium study of the Yamato nakhlites and approved H. Kojima (NIPR) and N. Nakamura (Kobe Univ.) to lead this consortium.

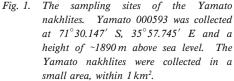
The goals of this consortium: first, to identify similarities and differences between Yamato and non-Antarctic nakhlites; second, to constrain chemical and isotopic signatures of the nakhlite source(s); third, to evaluate mechanisms of low-temperature aqueous alteration on Martian surface, and last, to identify the nature of nakhlite-Chassigny source crater. Results from some of the consortium studies were presented at the 27th NIPR Symposium on Antarctic Meteorites and are reported in the present volume.

## 2. Yamato nakhlites overview

On 29th November 2000, Y000593 was collected in the bare ice field ~10 km north from the northern end of the JARE IV Nunataks (Fig. 1). Four days later both Y000749 and Y000802 were found. These specimens were collected in a small area, within 1 km<sup>2</sup>, suggesting they are paired. The largest, Y000593, is 13.7 kg in weight and is the heaviest achondritic sample so far recovered in Antarctica (Figs. 2 and 3a). The second specimen, Y000749 (Fig. 3b), is 1.28 kg in weight and shows similar macroscopic texture as Y000593. A small 22-g meteorite, Y000802 (Fig. 3c), was later identified as the third Yamato nakhlite (Kojima *et al.*, 2002).

All three Yamato nakhlites are half covered with shiny black-colored fusion crust with ablation features. However, these stones have experienced some surficial weathering and in some places are deeply eroded. This weathering features may be due to the heavy katabatic wind blowing predominantly from east on the sampling site, because the physically eroded surface of Y000593 turned to the east in the field. Pyroxene and olivine crystals could be easily identified with the naked eye. All the specimens are quite fragile. We cannot determine the spatial relationships of these three meteorites to one another, suggesting that they may represent individual stones of a meteorite shower.

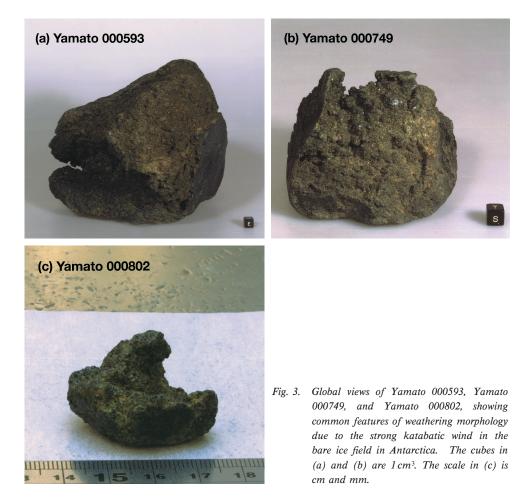




Yamato nakhlite consortium



Fig. 2. Occurrence of Yamato 000593 in the bare ice field of the Yamato Mountains area. The specimen is partly covered with black fusion crust, showing green colored interior.



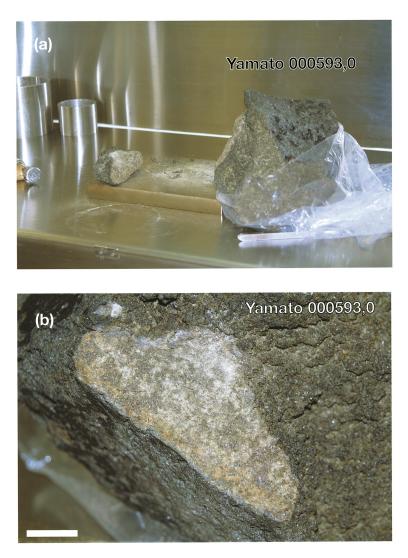
## 3. Curatorial processing of Yamato nakhlites

Two 2 g-sized fragments, Y000593,64 (2.180 g) and Y000749,47 (2.014 g), were used for wet chemical analysis at NIPR. After the identification of these nakhlites, two 90 mg-sized samples (Y000593,70 and Y000749,70) were allocated to R. Clayton (Univ. Chicago) for measurements of oxygen isotopes. Even before the consortium commenced, Y000749,40 (205 mg) was allocated to R. Okazaki (Univ. Tokyo) for rare gas study that confirmed Y000749 is a nakhlite. In addition to this sample, K. Nagao (Univ. Tokyo) received a 506 mg sample (Y000593,65) for rare gas geochemistry. A mass of 227 mg (Y000802,60) was allocated to R. Okazaki (Univ. Tokyo) for rare gas study.

Initial sample processing for the consortium took place in the Antarctic Meteorite Research Center at NIPR, using stainless steel tools. Samples of Y000593 for polished thin sections (PTSs) and for chemical analyses were taken from a limited interior portion of the main mass (Fig. 4). Most samples were sent to investigators in April 2002. After initial processing, a homogenized powder sample weighing about 15 g was prepared from six randomly selected chips from a large area. These chips of Y000593,40-45 ranged in weight from 2 to 3 g. Using 200 mg-sized aliquots (Y000593,40,1, 41,1, 42,1, 43,1, 44,1, 45,1), M. Ebihara (Tokyo Metropolitan Univ.) investigated heterogeneity of the powdered sample. For bulk analysis this sample is available upon request.

A mass of 4.9 g (Y000593,54) was allocated to M. Grady's consortium of UK scientists. The samples will be used for mineralogy and petrology (M. Grady, NHM, London), for noble gas chemistry and <sup>129</sup>I-<sup>129</sup>Xe, <sup>40</sup>K-<sup>40</sup>Ar, and <sup>39</sup>Ar-<sup>40</sup>Ar dating (G. Turner, Univ. Manchester), and for light element geochemistry and oxygen isotope compositions (C. Pillinger, Open Univ.). Two large fragments, Y000593,66 (1.211 g) and Y000749,46 (3.663 g), were allocated to M. Ebihara for nondestructive neutron induced prompt gamma-ray analysis (PGA). An offcut (Y000593,00-01) was allocated to A. El Goresy (MPI, Mainz) to estimate the temperature-fO<sub>2</sub> conditions using Fe-Ti partitioning between titanomagnetite and ilmenite. T. Mikouchi (Univ. Tokyo) received a 439-mg sample (Y000593,88) for transmitted electron microscopy (TEM) of augite and symplectite in olivine. D. Stöffler (Humboldt Univ.) received Y000593,75 (1.102 g) to study shock effects on the Yamato nakhlites. T. Hiroi (Brown Univ.) and M. Miyamoto (Univ. Tokyo) received 62-mg (Y000593,90) and 651-mg (Y000593,86) samples, respectively, for reflectance spectroscopy. N. Imae (NIPR) received Y000593,57 and ,58 (3.171 g and 0.796 g, respectively), and Y000749,49 (1.190 g) to investigate absorption spectra of OH using Fourier transform infrared (FTIR) spectroscopy. Fragments weighing a total of 4.256 g (Y000592,55) were allocated to N. Nakamura (Kobe Univ.) for the Rb-Sr, Sm-Nd, and U-Pb isotope systematics, and for the determination of chlorine isotopic compositions and lithophile trace element abundances by thermal ionization mass spectrometry (TIMS). G. Dreibus (MPI, Mainz) received a 1.371-g sample (Y000593,81) for the determination of Rb-Sr, Sm-Nd and U-Pb ages and for the determination of trace element abundances with instrumental neutron activation analysis (INAA). The NASA-JSC group led by L. Nyquist received fragments weighing a total of 2.265 g. Together with JSC personnel D. Bogard

#### Yamato nakhlite consortium



- Fig. 4. Sample processing of Yamato 000593.
  - (a) Many allocated samples were taken from a limited interior portion of the main mass, Yamato 000593,0.
  - (b) Fresh broken surface of Yamato 000593,0. Scale bar is 2 cm.

and C.-Y. Shih (Lockheed-Martin), he will determine crystallization age by the Rb-Sr, Sm-Nd, and <sup>39</sup>Ar-<sup>40</sup>Ar systems, mantle differentiated age by the <sup>146</sup>Sm-<sup>142</sup>Nd system, and neutron capture effects by <sup>149</sup>Sm and <sup>150</sup>Sm isotopes. Two chips (Y000593,87 and Y000749,52) weighing 487 mg and 616 mg, respectively, were allocated to P. Warren (UCLA) for study of bulk compositions especially focused on siderophile element abundances using INAA and radiochemical neutron activation analysis (RNAA) as clues to Martian core-mantle differentiation. K. Marti (UCSD) received Y000593,85 (685 mg) and Y000749,51 (462 mg) for identification of fission, spallation, and indige-

Yamato 000593 (13.713 kg)						
Subnumber	Weight (gram)	Investigator	Institution	Type of investigation		
35	3.227	McKay, G.	NASA-JSC	Re-Os		
40	2.447	57	NIPR	bulk chemistry (powdered)		
41	3.591		NIPR	bulk chemistry (powdered)		
42	2.000		NIPR	bulk chemistry (powdered)		
43	2.591		NIPR	bulk chemistry (powdered)		
44	2.109		NIPR	bulk chemistry (powdered)		
45	2.577		NIPR	bulk chemistry (powdered)		
54	4.922	Grady, M.	NHM, London	UK consortium		
55	4.256	Nakamura, N.	Kobe Univ.	chronology & chemistry		
56	2.256	Nyquist, L.	NASA-JSC	chronology		
57	3.171	Imae, N.	NIPR	FTIR		
58	0.796	Imae, N.	NIPR	FTIR		
59	1.172	Nishiizumi, K.	Space Sci. Lab., UCB	CRE		
60	1.060	Nishiizumi, K.	Space Sci. Lab., UCB	CRE		
61	0.607		NIPR	PTSs		
62	2.248		NIPR	PTSs		
63	1.884		NIPR	PTSs		
64	2.180		NIPR	PTSs		
65	0.506	Nagao, K.	Univ. Tokyo	rare gas		
66	1.212	Ebihara, M.	Tokyo Metropolitan Univ.	PGA		
67	1.011	Bonnara, m.	NIPR	PTSs		
68	2.230		NIPR	PTSs		
69	0.227	Hidaka, H.	Hiroshima Univ.	neutron exposure effects		
70	0.093	Clayton, R.	Univ. Chicago	oxygen isotope		
70 71	0.871	Funaki, M.	NIPR	rock magnetism		
72	0.716	Funaki, M.	NIPR	rock magnetism		
73	0.918	Okazaki, R.	Univ. Tokyo	rare gas		
73 74	1.212	Nakamuta, Y.	Kyushu Univ.	alteration products		
75	1.102	Stöffler, D.	Humboldt Univ.	shock effects		
76	0.719	Noguchi, T.	Ibaraki Univ.	TEM		
70 79	0.507	Mikouchi, T.	Univ. Tokyo	shock experiment		
81	1.371	Dreibus, G	MPI, Mainz	chemistry & chronology		
82	0.938	Shinoda, K.	Osaka City Univ.	FTIR		
82 84	0.796	Murty, S.	Phys. Res. Lab., Ahmedabad	CRE, rare gas & nitrogen isotop		
85	0.685	Marti, K.	UCSD	nitrogen, argon & xenon isotope		
86	0.651	Miyamoto, M.	Univ. Tokyo	reflectance spectroscopy		
87	0.487	Warren, P.	UCLA	INAA & RNAA		
88	0.493	Mikouchi, T.	Univ. Tokyo	TEM		
90	0.062	Hiroi, T.	Brown Univ.	reflectance spectroscopy		
91	0.086	Murty, S.	Phys. Res. Lab., Ahmedabad	CRE, rare gas & nitrogen isotop		
92	0.156	Murty, S.	Phys. Res. Lab., Ahmedabad	CRE, rare gas & nitrogen isotop		
93	0.130	Murty, S.	Phys. Res. Lab., Ahmedabad	CRE, rare gas & nitrogen isotop		
93 94	0.127	Murty, S.	Phys. Res. Lab., Ahmedabad	CRE, rare gas & nitrogen isotop		
94 95	0.238	Murty, S. Murty, S.	Phys. Res. Lab., Ahmedabad Phys. Res. Lab., Ahmedabad	CRE, rare gas & nitrogen isotop		
102	0.222	Herzog, G.	Rutgers Univ.	CRE, fare gas & muogen isotop CRE		
102	0.338	Eugster, O	Univ. Bern	CRE		
103	0.333	Halliday, A.	ETH	Hf-W		
	offcut	El Goresy, A.	MPI, Mainz	temperature & oxygen fugacity		
00-01	oncut	El Golesy, A.	WIF I, MAINZ	temperature & oxygen fugacity		

Table 1. Distribution of samples for the Yamato nakhlite consortium, including PTS samples.

PTSs; Y000593,61-1 & ,67-1: Imae, N. (NIPR), Y000593,62-1: Terada, K. (Hiroshima Univ.), Y000593,62-2: Goodrich, C. (Univ. Hawaii), Y000593,62-3 ,62-6 & ,68-1: Mikouchi, T. (Univ. Tokyo), Y000593,62-4: McCoy, T. (USNM), Y000593,63-1: Warren, P. (UCLA), Y000593,63-2: Ikeda, Y. (Ibaraki Univ.), Y000593,63-4 & ,63-6: Nakamura, N. (Kobe Univ.), Y000593,63-5: Wadhwa, M. (FMNH), Y000593,67-2: McKay, G. (NASA-JSC), Y000593,68-2: Shinoda, K. (Osaka City Univ.).

Yamato 000749 (1.283 kg)				
Subnumber	Weight (gram)	Investigator	Institution	Type of investigation
1	1.415		NIPR	PTSs
5	5.401		NIPR	PTSs
40	0.205	Okazaki, R.	Univ. Tokyo	rare gas
46	3.663	Ebihara, M.	Tokyo Metropolitan Univ.	PGA
47	2.014		NIPR	PTSs
48	1.048	Dreibus, G.	MPI, Mainz	chemistry & chronology
49	1.190	Imae, N.	NIPR	FTIR
50	0.996	Nishiizumi, K.	Space Sci. Lab., UCB	CRE
51	0.462	Marti, K.	UCSD	nitrogen, argon & xenon isotopes
52	0.616	Warren, P.	UCLA	INAA & RNAA
53	0.678	Okazaki, R.	Univ. Tokyo	rare gas
54	0.289	Herzog, G.	Rutgers Univ.	CRE
55	0.234	Hidaka, H.	Hiroshima Univ.	neutron exposure effects
56	0.466	Nakamuta, Y.	Kyushu Univ.	alteration products
57	0.891	Noguchi, T.	Ibaraki Univ.	alteration products
58	0.938	Shinoda, K.	Osaka City Univ.	FTIR
61	0.419	Eugster, O	Univ. Bern	CRE
70	0.090	Clayton, R.	Univ. Chicago	oxygen isotope
71	0.082	Halliday, A.	ETH	Hf-W

Table 1 (continued).

**PTSs**; Y000749,1-1: Imae, N. (NIPR), Y000749,1-3: Ikeda, Y. (Ibaraki Univ.), Y000749,1-6: Shinoda, K. (Osaka City Univ.), Y000749,5-1: Goodrich, C. (Univ. Hawaii), Y000749,5-2: Warren, P. (UCLA), Y000749,5-3: McCoy, T. (USNM), Y000749,5-4: Terada, K. (Hiroshima Univ.), Y000749,5-5: Wadhwa, M. (FMNH).

amato 000802	(22.306 g)			
Subnumber	Weight (gram)	Investigator	Institution	Type of investigation
20	1.593		NIPR	PTSs
60	0.227	Okazaki, R.	Univ. Tokyo	rare gas
61	0.215	Nishiizumi, K.	Space Sci. Lab., UCB	CRE

PTS; Y000802,20-1: Imae, N. (NIPR).

nous components of nitrogen, argon, and xenon in the source region of nakhlites. An interior chip, Y000593,84 (796 mg), and six samples of near surface material (Y000593,90-95) of a total weight of 911 mg were allocated to S. Murty (Phys. Res. Lab., Ahmedabad) to investigate cosmic-ray exposure ages (CRE), trapped noble gases, and nitrogen components. In order to study in detail the solar cosmic-ray effects and pre-atmospheric sizes of the Yamato nakhlites using <sup>41</sup>Ca, <sup>36</sup>Cl, <sup>26</sup>Al, <sup>10</sup>Be, and <sup>53</sup>Mn, K. Nishiizumi (Space Sci. Lab., UCB) was allocated a 1.172 g-sized interior sample (Y000593,59), a 1.060 g-sized exterior sample (Y000593,60), and a 996 mg-sized Y000749 chip (.50). Later he received a 215 mg-sized Y000802 chip (.61) adjacent to the sample allocated to R. Okazaki for rare gas studies. He split the allocated samples and sent aliquots of chips to A. Jull (Arizona Univ.) for <sup>14</sup>C measurement. H. Hidaka (Hiroshima Univ.) was allocated ~230 mg-sized samples (Y000593,69 and Y000749,55) to study neutron capture effects on samarium and gadolinium isotopes by TIMS. G. Herzog (Rutgers Univ.) received interior and fusion crust samples of Y000593,102 (338 mg) and Y000749,54 (289 mg) to study CRE by measuring nuclides  ${}^{36}$ Cl,  ${}^{26}$ Al,  ${}^{10}$ Be, and <sup>53</sup>Mn using accelerating mass spectrometry (AMS). In order to study magnetic properties of nakhlites, M. Funaki (NIPR) received Y000593,71 (871 mg) and Y000593,72 (716 mg).

Apart from those used for the initial classification, PTSs were produced for the Yamato nakhlite consortium, from four separate parent chips (Y000593,62, Y000593,63, Y000593,67, Y000593,68) and from one separate parent chip (Y000749,5) (*see* Table 1). We will produce PTSs using the samples allocated for PGA (Y000593,66 and Y000749,46). Five PTSs has been prepared from a parent chip of Y000802,20. Thus far, PTSs have been loaned to C. Goodrich (Univ. Hawaii), Y. Ikeda (Ibaraki Univ.), N. Imae (NIPR), T. McCoy (USNM), T. Mikouchi (Univ. Tokyo), N. Nakamura (Kobe Univ.), K. Terada (Hiroshima Univ.), M. Wadhwa (FMNH), and P. Warren (UCLA) for analysis of mineralogy and petrology. Four PTSs, Y000593,62-1, Y000749,5-4 (both loaned to K. Terada), Y000593,63-5, and Y000749,5-5 (both loaned to M. Wadhwa) will be used for ion microprobe analysis.

#### 4. Results and discussion

### 4.1. Mineralogy and petrology

Imae et al. (2003) and Mikouchi et al. (2003) reported mineralogy and petrology of the Yamato nakhlites and concluded that they are most similar to Nakhla (Fig. 5). The Yamato nakhlites, unbrecciated cumulate rocks, are composed principally of augite and olivine. Modal abundances of augite, olivine, and mesostasis vary slightly between different sections and are roughly 70-85%, 10-20%, and 5-10%, respectively. The chemical composition of the augite core in the Yamato nakhlites is nearly identical to those of other nakhlites  $(En_{40}Fs_{20}Wo_{40})$ . Olivine in the Yamato nakhlites shows extensive chemical zoning ( $Fa_{60-80}$ ). Olivine in the Yamato nakhlites contains magnetite-augite symplectic exolutions. Nearly identical symplectites have been found in Nakhla, Governador Valarades, and Chassigny (Yamada et al., 1997; Greshake et al., 2000; Mikouchi et al., 2000) but have not been found in Lafayette. These inclusions may constrain the cooling rate of olivine under the relatively higher oxygen fugacity. Mesostasis predominantly consists of plagioclase with minor K-feldspar, augite, pigeonite, olivine, titanomagnetite, apatite, and silica. Preterrestrial aqueous alteration products (Bunch and Reid, 1975; Gooding et al., 1991; Treiman et al., 1993; Gillet et al., 2002) are found in the Yamato nakhlites. They are reddish in color and may be composed of montmorillonite, goethite, saponite, and ferrihydrite (Treiman and Goodrich, 2002; Imae et al., 2003).

The reflectance spectrum of Y000593 is similar to that of Nakhla, and is also similar to Lafayette except for differences in slope from  $0.6-0.8 \mu m$  in wavelength (Ueda *et al.*, 2003)

#### 4.2. Bulk chemical composition

The bulk compositions of Y000593 and Y000749 (see Table 1 of Imae *et al.*, 2003) are comparable to those of non-Antarctic nakhlites: enriched in iron, magnesium, and calcium. Using bulk powdered samples of Y000593, Oura *et al.* (2003) performed PGA, and examined whether or not chemical heterogeneities exist in a 2 g-sized sample of the Yamato nakhlites. The Ti/Ca ratios, which are one of the key elemental ratios for nakhlites, of six powdered samples from Y000593 display nearly constant values of

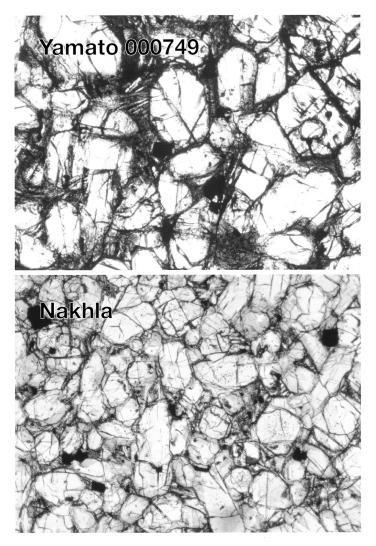


Fig. 5. Photomicrographs thin sections of Yamato 000749,49 and Nakhla. Transmitted light. A field of view is 2.6 mm.

40, confirming literature values of other nakhlites (see Lodders, 1998). Trace element abundances of Y000593 and Y000749 are similar to those of Nakhla. Rare earth element abundance patterns of Y000593 and Y000749 are light REE-rich: CI-chondrite-normalized La/Lu ratio is approximately 4.5 (Oura *et al.*, 2003; Nakamura *et al.*, 2002).

# 4.3. Oxygen isotopic signature

Oxygen isotopic compositions of Y000593 ( $\delta^{18}O = +4.84\%$  relative to SMOW,  $\delta^{17}O = +2.72\%$ ,  $\Delta^{17}O = +0.203\%$ ) and Y000749 ( $\delta^{18}O = +4.66\%$ ,  $\delta^{17}O = +2.57\%$ ,  $\Delta^{17}O = +0.147\%$ ) are in agreement with a Martian origin of these meteorites (R.N. Clayton and T.K. Mayeda, unpublished data, 2001).

## 4.4. Rare gases and cosmic-ray exposure age

Elemental ratios of trapped  ${}^{36}$ Ar/ ${}^{84}$ Kr/ ${}^{132}$ Xe of the bulk samples of Y000593, Y000749, and Y000802 are identical to those of Nakhla (Okazaki *et al.*, 2003). An averaged CRE age for the three Yamato nakhlites based on cosmogenic  ${}^{21}$ Ne is calculated to be  $12.1\pm0.7$  Ma (Okazaki *et al.*, 2003). The ejection ages of Nakhla, Lafayette, Governador Varadares, NWA817 (Marty *et al.*, 2001), Yamato and the dunite Chassigny are ~ 10 Myr and are the same within experimental errors, suggesting that the all nakhlites and Chassigny were ejected from Mars in a single impact cratering event.

# 4.5. Rubidium-strontium and samarium-neodymium systematics

Data points of separated mineral and whole-rock fractions from Y000593 define a linear array with a slope corresponding to an Rb-Sr age of ~1.3 Ga (Nakamura *et al.*, 2002; Shih *et al.*, 2002; Misawa *et al.*, 2003). An initial <sup>87</sup>Sr/<sup>86</sup>Sr ratio of 0.702525 $\pm$ 27 for Y000593 is in good agreement with previous results of Lafayette (Shih *et al.*, 1998) and Chassigny (Nakamura *et al.*, 1982b) but slightly different from those of other nakhlites (Papanastassiou and Wasserburg, 1974; Gale *et al.*, 1975; Shih *et al.*, 1999). The Sm-Nd isotopic system in Y000593 was not disturbed in the six samples (leachates and residues of whole-rock and augite, and one untreated whole-rock sample) analyzed so far. The data define an isochron corresponding to an Sm-Nd age of  $1.31\pm0.03$  Ga with an initial  $\varepsilon^{143}$ Nd of  $\pm 16.0\pm0.2$  (Shih *et al.*, 2002; Misawa *et al.*, 2003). The Yamato nakhlites so far analyzed and Chassigny are identical (Nakamura *et al.*, 1982a; Jagoutz, 1996; Shih *et al.*, 1999; Jagoutz, 2000; see Nyquist *et al.*, 2001).

# 5. Summary

On the basis of texture, mineralogy, petrology, chemistry, and CRE ages, we conclude that three Yamato nakhlites are paired specimens, bringing the total of identified nakhlites to six. The Yamato nakhlites have many similarities to Nakhla, and, apparently, have escaped from heavy terrestrial weathering, which is sometimes indicated for shergottites and nakhlites found in hot deserts. The Rb-Sr and Sm-Nd ages of Y000593 are  $\sim 1.3$  Ga and are concordant, showing similar isotopic signatures to those of non-Antarctic nakhlites and Chassigny. These facts strongly suggest that nakhlites and Chassigny crystallized within a short period of time  $\sim 1.3$  Ga. The ejection ages of nakhlites and Chassigny are  $\sim 10$  Myr, suggesting that all of the nakhlites and Chassigny were ejected from Mars in a single impact cratering event.

## Acknowledgments

We thank Y. Ikeda and an anonymous reviewer for their helpful comments. We are grateful to T. Mikouchi for providing thin sections of non-Antarctic nakhlites.

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(Received November 6, 2002; Revised manuscript accepted February 10, 2003)