

promoting access to White Rose research papers



Universities of Leeds, Sheffield and York
<http://eprints.whiterose.ac.uk/>

This is an author produced version of a paper published in **Combustion Theory and Modelling**.

White Rose Research Online URL for this paper:

<http://eprints.whiterose.ac.uk/10273/>

Published paper

Sharpe, G.J. (2008) *Effect of thermal expansion on the linear stability of planar premixed flames for a simple chain-branching model: The high activation energy asymptotic limit*. *Combustion Theory and Modelling*, 12 (4). pp. 717-738.

<http://dx.doi.org/10.1080/13647830802032849>

Effect of thermal expansion on the linear stability of planar premixed flames for a simple chain-branching model: the high activation energy asymptotic limit.

G. J. Sharpe*

School of Mechanical Engineering, University of Leeds, Leeds, LS2 9JT, UK

(Received 00 Month 200x; In final form 00 Month 200x)

The linear stability of freely propagating, adiabatic, planar premixed flames is investigated in the context of a simple chain-branching chemistry model consisting of a chain-branching reaction step and a completion reaction step. The role of chain-branching is governed by a crossover temperature. Hydrodynamic effects, induced by thermal expansion, are taken into account and the results compared and contrasted with those from a previous purely thermal-diffusive constant density linear stability study. It is shown that when thermal expansion is properly accounted for, a region of stable flames predicted by the constant density model disappears, and instead the flame is unstable to a long-wavelength cellular instability. For a pulsating mode, however, thermal expansion is shown to have only a weak effect on the critical fuel Lewis number required for instability. These effects of thermal expansion on the two-step chain-branching flame are shown to be qualitatively similar to those on the standard one-step reaction model. Indeed, as found by constant density studies, in the limit that the chain-branching crossover temperature tends to the adiabatic flame temperature, the two-step model can be described to leading order by the one-step model with a suitably defined effective activation energy.

1 Introduction

Freely propagating premixed flames may, in principal, exist as planar and steady waves. However, experiments show that in many cases the flame is actually wrinkled and time-dependent [1–4]. These ‘cellular’ or ‘pulsating’ flames may be regarded as the outcome of thermal-diffusive and/or hydrodynamic instabilities. Hence a first step in understanding their origins and onset is a linear stability analysis of the underlying steady, planar flame.

Much of the fundamental theory of combustion employs a standard, exothermic, one-step chemistry model [1, 4], $F \rightarrow P$, where F denotes the fuel and P denotes the combustion products. This model predicts an adiabatic flame structure consisting of a diffusive pre-heat zone, followed by a thin reaction zone at the downstream end of the flame, where the fuel is consumed and all the heat is released. In the asymptotic limit of high activation energy, the reaction zone can be replaced to leading order by a reaction sheet or surface across which suitable jump conditions apply, and such that no chemical reaction or heat release occurs on either side of this reaction sheet. This asymptotic limit of the one-step model has been successfully employed in a number of linear stability studies of premixed flames. Sivashinsky [5] first employed a constant density approximation in order to study purely thermal-diffusive effects on the flame stability. The constant density model (CDM) used by Sivashinsky [5] ignores any hydrodynamic effects, and is formally valid only in the limit of small heat of reaction. Several workers then independently obtained long-wavelength asymptotic solutions of the linear stability problem, in the context of the Reactive Navier-Stokes equations [6–8]. These ‘slowly varying flame’ analyses capture mainly hydrodynamic effects on the flame stability. Jackson and Kapila [9] then solved numerically the one-step high-activation energy asymptotic linear stability problem for arbitrary perturbation wavelengths, and showed how the combination of both thermal-diffusive and hydrodynamic effects are important in determining the stability of premixed flames. These asymptotic one-step reaction studies also entail a near-equidiffusional flames (NEF) approximation, i.e they are valid for Lewis numbers asymptotically close to unity.

Finite activation energy numerical linear stability studies, which do not invoke the asymptotic limit, have also been performed for the one-step reaction model, both in terms of the CDM [10] and the Reactive Navier-Stokes equations [11, 12]. These studies show that, when the Lewis number is sufficiently far from unity, very high activation energies may be required for the asymptotic results to be quantitatively predic-

*Corresponding author. E-mail: mengjs@leeds.ac.uk

tive. However, these finite activation energy studies do not reveal any qualitatively different instabilities than those predicted by the asymptotic analyses, even when the activation energy is quite moderate. The main use of such finite activation energy calculations is hence to provide quantitative test problems for numerical codes designed to simulate the fully non-linear flame evolutions [10, 13].

Although the one-step model has been very successful in describing and predicting many aspects of flame phenomena [1, 4], it fails to capture some aspects of hydrocarbon and hydrogen flame structures [4, 14]. In these flames, there are many intermediate steps between the conversion of fuel into products. These include chain-branching reactions, which produce a net increase in intermediate species such as radicals. The chain-branching reactions tend to have high-activation energy, and are hence active in the high-temperature regions of the flame, where they convert the fuel into intermediate species [14]. The intermediate molecules produced may then diffuse forwards and backwards over the entire flame structure, so that the ‘pre-heat’ zone is in reality also chemically active [4, 14]. Completion reactions, which remove the intermediates and convert them into products, tend to be temperature insensitive but highly exothermic, so that heat release occurs throughout the flame [4], in contrast to in a narrow region at the downstream end of the flame structure as predicted by the one-step model. Indeed, the exothermic completion reactions continue even after the fuel has been completely converted into intermediates, so that the fuel is exhausted interior to the flame, i.e. before the adiabatic flame temperature is reached at the downstream end [14].

This discrepancy between the one-step model and real flame structures motivates the need for a chemistry model which better mimics the effects of chain-branching outlined above. For the purposes of the mathematical theory of flames, however, any improved model should still be sufficiently simple and generic such that transparent, fundamental insights can be obtained and that some analytical or asymptotic progress is still possible. In this spirit, Dold and co-workers [14–16] have suggested a two-step chemistry model, consisting of a single chain-branching step, $F+Y\rightarrow 2Y$, and a single completion reaction step, $Y+M\rightarrow P+M$, where Y represents a lumped or ‘pooled’ amalgam of intermediate species, and M is any species required to trigger the completion reactions, but is unchanged in the process. In the simplest version of this model, the branching reaction is assumed to have a high activation temperature but is thermally neutral, while the completion reaction is assumed to be temperature insensitive but releases all the heat. Dold [14] gives a very detailed discussion which puts the two-step model and its assumptions into the context of hydrogen and hydrocarbon oxidation, including how the model parameters can be fitted to hydrogen and hydrocarbon flame structures as predicted from detailed chemistry calculations.

Fundamental to the two-step model described above is the concept of a chain-branching crossover temperature, T_c , which, in regards to flame structure, is the temperature at which the rate of chain-branching balances the rate of removal of intermediates by diffusion [14]. Thus above this temperature a chain-branching explosion occurs in which the fuel is converted rapidly into intermediates. For large activation energies of the branching step, the reaction is then active only in a narrow region occurring at temperatures close to T_c . As for the one-step reaction, in the asymptotic limit of infinite activation energy, the branching reaction can be replaced to leading order by a reaction sheet across which appropriate jump conditions apply. For the two-step model, the reaction sheet occurs at the crossover temperature, where the fuel is consumed.

As well as being able to mimic the main features of real flames which the one-step model can not, the two-step model also has mathematical advantages over the one-step model. Firstly, the simple chain-branching model does not suffer from the well known ‘cold-boundary difficulty’ inherent in the one-step model [17]. Secondly, in the large activation energy asymptotic limit, the jump conditions across the reaction sheet are linear in the variables [14], as compared to the more complex jump conditions of the one-step model, which are non-linear in the temperature at which the reaction sheet occurs [1]. Further, in terms of the asymptotic flame stability, the two-step model does not require the NEF approximation and hence is valid for arbitrary values of the Lewis numbers [14].

Dold and co-workers [14–16] have applied the high-activation energy asymptotic limit of the two-step chemistry model to a number of flame problems, in the context of the purely thermal-diffusive constant density model. These include studies of the structure and stability of flame balls [16] and of premixed flames [14]. In this paper, we extend the asymptotic linear stability analysis of the two-step chain-branching premixed flames to include hydrodynamic effects induced by thermal expansion, by considering the Re-

active Navier-Stokes equations. The main purpose of the paper is hence to compare and contrast with both the two-step CDM [14] and the one-step chemistry Reactive Navier-Stokes predictions [9]. Indeed, the paper should be viewed as the two-step analog of the one-step linear stability study of Jackson and Kapila [9].

2 The Model

The governing equations of the model are the quasi-isobaric Navier-Stokes equations coupled to the two-step chemistry scheme $F+Y\rightarrow 2Y$, $Y+M\rightarrow P+M$. The **non-dimensional** versions of these equations are, in **two spatial dimensions, x and y** ,

$$\frac{\partial \rho}{\partial t} + \frac{\partial(\rho u)}{\partial x} + \frac{\partial(\rho v)}{\partial y} = 0, \quad (1)$$

$$\rho \frac{\partial u}{\partial t} + \rho u \frac{\partial u}{\partial x} + \rho v \frac{\partial u}{\partial y} + \frac{\partial P}{\partial x} = Pr \left(\frac{4}{3} \frac{\partial^2 u}{\partial x^2} + \frac{\partial^2 u}{\partial y^2} + \frac{1}{3} \frac{\partial^2 v}{\partial x \partial y} \right), \quad (2)$$

$$\rho \frac{\partial v}{\partial t} + \rho u \frac{\partial v}{\partial x} + \rho v \frac{\partial v}{\partial y} + \frac{\partial P}{\partial y} = Pr \left(\frac{4}{3} \frac{\partial^2 v}{\partial y^2} + \frac{\partial^2 v}{\partial x^2} + \frac{1}{3} \frac{\partial^2 u}{\partial x \partial y} \right), \quad (3)$$

$$\rho \frac{\partial T}{\partial t} + \rho u \frac{\partial T}{\partial x} + \rho v \frac{\partial T}{\partial y} = \frac{\partial^2 T}{\partial x^2} + \frac{\partial^2 T}{\partial y^2} + QW_C, \quad (4)$$

$$\rho \frac{\partial F}{\partial t} + \rho u \frac{\partial F}{\partial x} + \rho v \frac{\partial F}{\partial y} = \frac{1}{Le_F} \left(\frac{\partial^2 F}{\partial x^2} + \frac{\partial^2 F}{\partial y^2} \right) - W_B, \quad (5)$$

$$\rho \frac{\partial Y}{\partial t} + \rho u \frac{\partial Y}{\partial x} + \rho v \frac{\partial Y}{\partial y} = \frac{1}{Le_Y} \left(\frac{\partial^2 Y}{\partial x^2} + \frac{\partial^2 Y}{\partial y^2} \right) + W_B - W_C, \quad (6)$$

$$\rho T = 1, \quad (7)$$

where ρ is the density, u and v are the x - and y -components of the fluid velocity, respectively, T the temperature, F and Y the mass fraction of fuel and intermediates. These equations have been **non-dimensionalized** using the standard scales employed in previous linear stability analysis of the one-step model [9, 12], with which we seek to compare. Thus

$$\rho = \frac{\bar{\rho}}{\bar{\rho}_f}, \quad u = \frac{\bar{u}}{\bar{V}_f}, \quad v = \frac{\bar{v}}{\bar{V}_f}, \quad p = \frac{\bar{p}}{\bar{p}_f}, \quad T = \frac{\bar{p}}{\bar{\rho}} = \frac{\bar{T}}{\bar{T}_f},$$

$$x = \frac{\bar{\rho}_f \bar{V}_f \bar{c}_p}{\bar{\kappa}} \bar{x}, \quad y = \frac{\bar{\rho}_f \bar{V}_f \bar{c}_p}{\bar{\kappa}} \bar{y}, \quad t = \frac{\bar{\rho}_f \bar{V}_f^2 \bar{c}_p}{\bar{\kappa}} \bar{t},$$

where a bar ($\bar{\quad}$) denotes dimensional quantities, an ‘ f ’ subscript denotes quantities in the fresh, unburnt gas upstream of the flame (and a ‘ b ’ subscript will be used to denote quantities in the completely burnt

state downstream of the flame). Thus note that the **non-dimensional** temperature and density are unity in the fresh gas. Here p is the pressure, \bar{V}_f is the speed of the steady, planar flame, \bar{c}_p is the specific heat at constant pressure, $\bar{\kappa}$ is the co-efficient of thermal conductivity and $\bar{\rho}_f \bar{V}_f \bar{c}_p / \bar{\kappa}$ is a heat conduction lengthscale (characteristic of the ‘preheat zone’ of the flame in the one-step model). The quantity P appearing in equations (2) and (3) is the $O(M_f^2)$ deviation of the pressure from the upstream value, i.e. $p = 1 + \gamma M_f^2 P$, where $M_f = \bar{V}_f (\bar{\rho}_f / (\gamma \bar{p}_f))^{1/2}$ is the Mach number of the flame, and γ is the ratio of specific heat. The **non-dimensional** parameters appearing in equations (2)-(6) are the Prandtl number, Pr , the heat of reaction of the completion step, Q , and the Lewis numbers of the fuel and of the intermediates, Le_F and Le_Y , respectively.

Note that Dold [14] used alternative scalings for the non-dimensionalization. However, his **non-dimensional** quantities (denoted by a ‘D’ subscript) are simply related to those used here by

$$T_D = \frac{T}{T_c}, \quad F_D = F, \quad Y_D = \frac{Le_F}{Le_Y} Y, \quad x_D = \sqrt{Le_Y \Lambda} x, \quad t_D = Le_Y \Lambda t, \quad Q_D = \frac{Q}{T_c - 1},$$

where Λ and T_c are defined below, and Q_D is an alternately defined ‘heat of reaction’ on which the steady, planar flame structure and CDM flame stability are found to depend [14].

The **non-dimensional** branching and completion reaction rates are assumed to be of the form

$$W_B = \Lambda_B \rho^2 F Y \exp\left(\theta \left[\frac{1}{T_c} - \frac{1}{T}\right]\right), \quad W_C = \Lambda \rho^2 T^n Y,$$

where Λ_B and Λ are **non-dimensional** rate constants, θ is the **non-dimensional** activation energy of the branching step, and a weak power law temperature dependence of the completion step is assumed for later analytical convenience. Here T_c is the ‘inhomogeneous’ crossover temperature defined by Dold [14], i.e. the temperature at which the chain-branching rate becomes equal to the rate of removal of intermediates by molecular diffusion. It can be seen that for large activation energy θ , the branching rate quickly becomes very small compared to diffusion as the temperature decreases below T_c , while it becomes very large as T increases above T_c . Thus while the reaction will be essentially frozen for temperatures below the crossover temperature, the fuel will rapidly be exhausted (and the branching-reaction complete) once T is above T_c due to the very high reaction rate. Hence, as its activation energy increases, the branching step will only be active in an increasingly narrow region around the crossover temperature.

In the asymptotic limit, $\theta \rightarrow \infty$, it can be shown that the branching reaction can be replaced to leading order by an infinitesimally thin reaction sheet in which the fuel is consumed. For the chain-branching model, this surface occurs at $T = T_c$, and suitable jump conditions need to be applied across it [14, 15]. An analysis of the inner branching reaction zone structure [1, 14, 15] determines the outer jump conditions to be applied at the reaction sheet:

$$[T] = [F] = F = [Y] = [u] = [v] = [P] = 0, \quad (8)$$

$$[T, n] = [F, n] + \frac{Le_F}{Le_Y} [Y, n] = [u, n] = [v, n] = 0, \quad (9)$$

$$T = T_c, \quad (10)$$

where $[\cdot]$ denotes the jump in a quantity across the reaction sheet and \cdot, n denotes the derivative of a quantity in the direction normal to the reaction sheet (hence $[\cdot, n]$ denotes the jump in the normal gradient across the sheet). Note that the jump conditions for the two-step model are linear in the variables and their derivatives and are independent of θ to leading order [14]. Note also that, as $\theta \rightarrow \infty$, the inhomogeneous crossover temperature, T_c , is related to the homogeneous crossover temperature, T_X (which

is the temperature where the branching and completion reaction rates balance) by

$$T_X = T_c - 2T_c^2\theta^{-1}\ln(\theta/T_c),$$

and hence, these inhomogeneous and homogeneous crossover temperatures are the same to leading order [14].

3 Steady, planar flames

In the laboratory frame, the steady, planar flame is assumed to travel in the negative x -direction at unit **non-dimensional** speed, so that the fresh, unburnt fuel is approached as $x \rightarrow -\infty$ and the completely burnt state approached as $x \rightarrow \infty$. Here we work in the rest frame of the flame, such that the flow is steady (independent of t) and the upstream fuel is oncoming at unit speed. Denoting the steady flame solution by a zero subscript, after integrating once with respect to x and employing the boundary conditions $T_0 = \rho_0 = u_0 = F_0 = 1$, $P_0 = Y_0 = 0$ and $dq_0/dx = 0$ as $x \rightarrow -\infty$ (where q denotes any of the dependent variables), the governing equations (1)-(7) can be reduced to

$$\frac{dT_0}{dx} = T_0 - 1 + Q(X_0 + Z_0 - 1), \quad \frac{dF_0}{dx} = Le_F(F_0 - X_0), \quad \frac{dY_0}{dx} = Le_Y(Y_0 - Z_0), \quad (11)$$

$$\frac{dX_0}{dx} = 0, \quad \frac{dZ_0}{dx} = -\Lambda T^{n-2}Y_0, \quad (12)$$

where X_0 and Z_0 are defined by the second and third of equations (11), cf. [18], (**hence note $X_0 = 1$ in the region $x < 0$, and $Z_0 = 0$ at the cold boundary**), together with

$$u_0 = \frac{1}{\rho_0} = T_0, \quad P_0 = \frac{4Pr}{3} \frac{dT_0}{dx} - (T_0 - 1) \quad (13)$$

In the fully burnt state $x \rightarrow \infty$, the boundary conditions $dq_0/dx = 0$ give $T_0 = u_0 = 1/\rho_0 = 1 + Q = T_{\text{ad}}$, where T_{ad} denotes the adiabatic flame temperature, and $P_0 = -Q$, $F_0 = Y_0 = X_0 = Z_0 = 0$. **Hence note that $X_0 = 0$ in the region $x > 0$ by equation (12).**

Since, apart from for the pressure, the steady, flame solution is independent of the Prandtl number, and Pr is known to have only a very weak effect on flame stability [6, 12], throughout this paper we set $Pr = 0.75$. Here will consider the case $n = 2$ in order to retain analytical simplicity of the steady flame structure (the equation for Y_0 is then linear and decoupled from the T_0 and F_0 equations), and such that it has precisely the same structure as in the CDM considered previously [14], with which we seek to compare. Note that many of the main exothermic completion reaction rates given in the example of methane-oxidation in [14] are temperature insensitive, while others have positive or negative values of n . As a lumped or pooled reaction step, it is unclear what value of n in the two-step model would give a best fit to a given real flame speed and structure. While general values of $n \neq 2$ could also be considered, the steady flame structure would then need to be determined numerically. Furthermore, any weak temperature dependence in the completion reaction rate will not qualitatively effect the leading order high-activation energy results (which are themselves approximate) studied here. Hence it is hence worth retaining the mathematical simplicity of the flame structure equations (by considering $n = 2$) which was one of the original motivations for the two-step model [14, 15].

The spatial origin is chosen to occur at the point where the temperature equals the crossover temperature, $T_0 = T_c$. The jump conditions (8)-(10) are thus to be applied at $x = 0$, and for the steady flame give

$$[T_{0,x}] = [T_0] = [Y_0] = [F_0] = F_0 = [X_0 + Z_0] = 0, \quad T_0 = T_c.$$

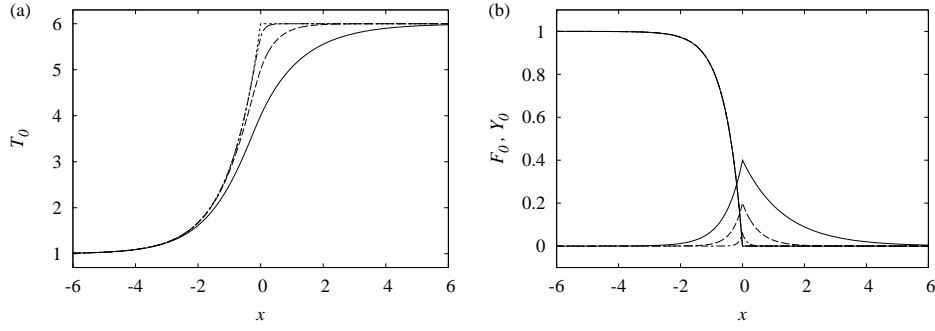


Figure 1. Profiles of (a) temperature and (b) fuel and intermediates mass fractions, in the steady, planar flame, for $Q = 5$, $Le_F = Le_Y = 1$ and $T_c = 4$ ($Q_D = 1.667$) (solid lines), $T_c = 5$ ($Q_D = 1.25$) (dashed lines) and $T_c = 5.7$ ($Q_D = 1.064$) (dot-dashed lines). Also shown as a dotted line in (a) is the temperature profile of the one-step model.

In order to satisfy these and both sets of boundary conditions as $x \rightarrow -\infty$ and $x \rightarrow \infty$ simultaneously, Λ , which is related to the flame speed, must have a specific value. Hence Λ is an eigenvalue of equations (11)-(12) which needs to be determined, for other parameters fixed.

The solution which satisfies the boundary conditions at $x \rightarrow \pm\infty$ and the jump conditions at $x = 0$ is

$$F_0 = \begin{cases} 1 - \exp(Le_F x) & x \leq 0 \\ 0 & x > 0 \end{cases}, \quad (14)$$

$$Y_0 = \frac{Le_Y}{h_1 - h_2} \begin{cases} \exp(h_1 x) & x \leq 0 \\ \exp(h_2 x) & x > 0 \end{cases}, \quad (15)$$

$$T_0 = \begin{cases} 1 - \frac{Q\Lambda Le_Y}{(h_1 - 1)(h_2 - 1)} \exp(x) - \frac{Q\Lambda Le_Y}{h_1(h_1 - 1)(h_1 - h_2)} \exp(h_1 x) & x \leq 0 \\ 1 + Q - \frac{Q\Lambda Le_Y}{h_2(h_2 - 1)(h_1 - h_2)} \exp(h_2 x) & x > 0 \end{cases}, \quad (16)$$

cf. [14], where h_1 and h_2 are the positive and negative roots, respectively, of $h^2 - Le_Y h - Le_Y \Lambda = 0$. The eigenvalue Λ (and hence the flame speed) is then determined by setting $T_0 = T_c$ at $x = 0$, giving

$$Q\Lambda Le_Y = (1 + Q - T_c)h_2(h_2 - 1)(h_1 - h_2), \quad (17)$$

which can be straightforwardly solved by Newton-Raphson iteration for Λ . Note that for $Le_Y = 1$ equation (17) has the analytical solution

$$\Lambda = \frac{Q^2}{4(1 + Q - T_c)^2} - \frac{1}{4},$$

which can be used as an initial guess for non-unity values of Le_Y . Note that in any case Λ depends on Q and T_c only through the combination

$$Q/(1 + Q - T_c) = Q_D/(Q_D - 1) = (T_{ad} - 1)/(T_{ad} - T_c)$$

and does not depend on Le_F [14].

Figure 1 shows some example spatial profiles of the steady, planar flame for $Q = 5$, $Le_F = Le_Y = 1$ and various values of the crossover temperature. Note, the peak in the intermediates occurs at the reaction

sheet. The value of Y_0 at the peak is connected to the crossover temperature, T_c , since $T_{\text{ad}} - T_c$ determines the fraction of the heat still to be released after the fuel has been consumed in the chain-branching reaction sheet, while similarly the peak value of Y_0 represents the amount of intermediates still available for conversion into products (and hence heat). Thus the lower T_c is below the adiabatic flame temperature, the higher is the peak in the intermediates at $x = 0$, and hence the further the intermediates are able to diffuse from the reaction sheet. This results in the heat release and hence the temperature rise being spread over a larger distance for lower T_c . On the other hand, as $T_c \rightarrow T_{\text{ad}}$ (or equivalently as $Q_D \rightarrow 1$), the peak in the radicals becomes very small and thus the fuel is largely converted directly to products. Indeed, it can be shown that in the limit $Q_D \rightarrow 1$, the planar flame structure tends to that of the one-step model (represented as a dotted line in figure 1a) in which intermediates have no role [14].

4 Linearized equations

We now suppose that the steady, planar flame is slightly perturbed such that the perturbed flame position with respect to that of the underlying steady, planar flame is

$$x = X(y, t), \quad |X(y, t)| \ll 1.$$

Since the jump conditions (8)-(10) are to be applied on a surface whose position is perturbed, we transform to a frame moving with this perturbed reaction sheet, cf. [9, 19], via

$$x_p = x - X(y, t), \quad y_p = y, \quad t_p = t, \quad u_p = u - V(y, t), \quad (18)$$

where the ‘ p ’ subscript denotes the perturbed flame co-ordinate system, and

$$V(y, t) = \frac{\partial X}{\partial t}$$

is the speed of the perturbed reaction sheet in the x -direction. Hence in this new co-ordinate system, the perturbed reaction sheet remains stationary at $x_p = 0$. Note that, under this transformation, $\partial \cdot / \partial x$, $\partial \cdot / \partial y$, $\partial \cdot / \partial t$ are replaced in the governing equations (1)-(6) by

$$\frac{\partial \cdot}{\partial x_p}, \quad \frac{\partial \cdot}{\partial y_p} - \frac{\partial X}{\partial y_p} \frac{\partial \cdot}{\partial x_p}, \quad \frac{\partial \cdot}{\partial t_p} - \frac{\partial X}{\partial t_p} \frac{\partial \cdot}{\partial x_p}, \quad (19)$$

respectively, so that ∇^2 becomes

$$\left(1 + \left[\frac{\partial X}{\partial y_p} \right]^2 \right) \frac{\partial^2 \cdot}{\partial x_p^2} + \frac{\partial^2 \cdot}{\partial y_p^2} - 2 \frac{\partial X}{\partial y_p} \frac{\partial^2 \cdot}{\partial x_p \partial y_p} - \frac{\partial^2 X}{\partial y_p^2} \frac{\partial \cdot}{\partial x_p}. \quad (20)$$

Dold [14] chose to Taylor expand the jump conditions around the $x = 0$ instead of transforming to the frame of the perturbed flame, the two approaches being equivalent. Henceforth, for convenience we drop the ‘ p ’ subscript notation.

A normal modes form of the perturbation is then assumed, such that

$$X(y, t) = \epsilon e^{\sigma t} e^{iky}, \quad \epsilon \ll 1, \quad (21)$$

and the perturbed variables are hence of the form

$$q(x, y, t) = q_0(x) + \epsilon q_1(x) e^{\sigma t} e^{iky}, \quad (22)$$

where q represents any of the dependent variables. Here σ is the (complex) growth rate and k is the wavenumber of the disturbance in the y -direction. For convenience, we define the following quantities in terms of the first derivatives of the perturbations with respect to x :

$$\tau_1 = \frac{dT_1}{dx}, \quad U_1 = \frac{du_1}{dx}, \quad V_1 = \frac{dv_1}{dx}, \quad X_1 = F_1 - \frac{1}{Le_F} \frac{dF_1}{dx}, \quad Z_1 = Y_1 - \frac{1}{Le_Y} \frac{dY_1}{dx}. \quad (23)$$

Substituting equations (21)-(23) into the governing equations and linearizing in ϵ , gives

$$\frac{d\mathbf{u}}{dx} = \mathbf{A}\mathbf{u} + \mathbf{s}, \quad (24)$$

where $\mathbf{u} = (T_1, u_1, v_1, P_1, F_1, Y_1, U_1, V_1, X_1, Z_1)^T$ and

$$\mathbf{A} = \begin{pmatrix} \frac{(T'_0 - \sigma)}{T_0} & \frac{-T'_0}{T_0} & ik & 0 & 0 \\ 0 & 0 & 0 & 0 & 0 \\ 0 & 0 & 0 & 0 & 0 \\ A_{41} & A_{42} & \frac{4ikPr(\sigma + T_0 - T'_0)}{3T_0} & 0 & 0 \\ 0 & 0 & 0 & 0 & Le_F \\ 0 & 0 & 0 & 0 & 0 \\ A_{71} & \frac{-\sigma T'_0 + T_0 T''_0}{T_0^2} & \frac{ik(\sigma + T_0 - T'_0)}{T_0} & 0 & 0 \\ 0 & 0 & \frac{3\sigma + 4Prk^2 T_0}{3PrT_0} & \frac{ik}{Pr} & 0 \\ \frac{F'_0}{T_0} & \frac{-F'_0}{T_0} & 0 & 0 & -\frac{Le_F \sigma + T_0 k^2}{T_0 Le_F} \\ \frac{Y'_0}{T_0} & \frac{-Y'_0}{T_0} & 0 & 0 & 0 \\ 0 & 0 & 0 & 0 & 0 \\ 0 & 0 & 0 & 0 & 0 \\ 0 & 0 & 0 & 0 & 0 \\ -\frac{4PrQ\Lambda}{3} & \frac{4Pr(\sigma + T_0) - 3T_0}{3T_0} & -ikPr & 0 & 0 \\ 0 & 0 & 0 & -Le_F & 0 \\ Le_Y & 0 & 0 & 0 & -Le_Y \\ -Q\Lambda & \frac{\sigma + T_0}{T_0} & -ik & 0 & 0 \\ 0 & \frac{ik}{3} & \frac{1}{Pr} & 0 & 0 \\ 0 & 0 & 0 & 0 & 0 \\ \frac{-Le_Y \sigma - k^2 T_0 - Le_Y \Lambda T_0}{Le_Y T_0} & 0 & 0 & 0 & 0 \end{pmatrix},$$

where

$$A_{41} = \frac{4Pr(-\sigma^2 + T'_0 \sigma + k^2 T_0^2 - T_0 T''_0)}{3T_0^2} + \frac{T'_0}{T_0},$$

$$A_{42} = \frac{-(4PrT'_0 + 3T_0)\sigma}{3T_0^2} + \frac{Pr(4T''_0 - 3T_0k^2)}{3T_0} - \frac{T'_0}{T_0},$$

$$A_{71} = \frac{(-\sigma^2 + T'_0\sigma + k^2T_0^2 - T_0T''_0)}{T_0^2},$$

and

$$\mathbf{s} = \left(0, 0, 0, -\frac{3\sigma^2 + 3P_rT_0k^2\sigma + P_rT_0k^2T'_0}{3T_0}, 0, 0, -k^2T'_0, \frac{ik}{P_r}(-P_rT''_0 + T'_0), \frac{k^2F'_0}{Le_F}, \frac{k^2Y'_0}{Le_Y} \right)^T,$$

where henceforth a prime denotes differentiation with respect to x . Note that, since $\rho = 1/T$, density has been eliminated from equations (1)-(6). Equation (1) then contains only first x -derivatives of the remaining variables, hence its linearized form has been used to eliminate τ_1 in terms of the other perturbed quantities:

$$\tau_1 = \frac{(T'_0 - \sigma)T_1}{T_0} - \frac{T'_0u_1}{T_0} + ikv_1 + U_1, \quad (25)$$

Equation (24) is subject to boundedness conditions as $x \rightarrow \pm\infty$. We hence now seek asymptotic solutions to equation (24) valid near the fresh or burnt states, as $x \rightarrow \pm\infty$, which can then be used as initial conditions for numerical integration of the equation. In order to obtain higher order terms in these asymptotic solutions, which are necessary for implementation of straightforward shooting methods [12, 19], it is beneficial to use one of the steady state solution variables as the independent variable [12, 19]. Given the structure of the steady solution in equations (14)-(16), and given that $Y_0 \rightarrow 0$ as $x \rightarrow \pm\infty$, here we choose Y_0 as the independent variable for both the fresh and burnt state analyses.

4.1 Asymptotic solution as $x \rightarrow -\infty$

Transforming to Y_0 as the independent variable by employing equation (15), then in the region $x < 0$ equation (24) becomes

$$h_1Y_0 \frac{d\mathbf{u}}{dY_0} = \mathbf{A}\mathbf{u} + \mathbf{s}, \quad (26)$$

and the other steady variables can be written in terms of Y_0 as

$$T_0 = 1 + \alpha_1Y_0 + \beta_1Y_0^{1/h_1}, \quad F_0 = 1 + \gamma_1Y_0^{Le_F/h_1},$$

where

$$\alpha_1 = -\frac{Q\Lambda}{h_1(h_1 - 1)}, \quad \beta_1 = -\frac{Q\Lambda Le_Y}{(h_1 - 1)(h_2 - 1)} \left(\frac{h_1 - h_2}{Le_Y} \right)^{1/h_1}, \quad \gamma_1 = -\left(\frac{h_1 - h_2}{Le_Y} \right)^{Le_F/h_1}.$$

As $Y_0 \rightarrow 0$, corresponding to $x \rightarrow -\infty$, we can thus expand equation (26) in the form

$$h_1Y_0 \frac{d\mathbf{u}}{dY_0} = (\mathbf{A}_0 + \mathbf{A}_1Y_0 + \mathbf{A}_2Y_0^{1/h_1} + \mathbf{A}_3Y_0^{Le_F/h_1} + \dots)\mathbf{u} + \mathbf{s}_0 + \mathbf{s}_1Y_0 + \mathbf{s}_2Y_0^{1/h_1} + \mathbf{s}_3Y_0^{Le_F/h_1} + \dots, \quad (27)$$

where the co-efficient matrices, \mathbf{A}_0 , etc, depend only on the parameters Q , P_r , Le_Y , Le_F and T_c , and on σ and k . The homogeneous part of equation (27) has 10 independent asymptotic solutions of the form

$$\mathbf{u} = \tilde{\mathbf{u}}_i = Y_0^{\lambda_i} \left(\mathbf{a}_0^i + \mathbf{a}_1^i Y_0 + \mathbf{a}_2^i Y_0^{1/h_1} + \mathbf{a}_3^i Y_0^{Le_F/h_1} + \dots \right), \quad i = 1, \dots, 10, \quad (28)$$

where $h_1 \lambda_i$ and \mathbf{a}_0^i are the eigenvalues and eigenvectors, respectively, of \mathbf{A}_0 , and the \mathbf{a}_1^i , \mathbf{a}_2^i , etc., are found by substituting (28) into equation (27) and equating powers of Y_0 . The eigenvalues of \mathbf{A}_0 are

$$\pm k, \quad \frac{1 \pm [1 + 4(\sigma + k^2)]^{\frac{1}{2}}}{2}, \quad \frac{1 \pm [1 + 4Pr(\sigma + Prk^2)]^{\frac{1}{2}}}{2Pr}, \quad \frac{Le_F \pm [Le_F^2 + 4(\sigma Le_F + k^2)]^{\frac{1}{2}}}{2},$$

$$\frac{Le_Y \pm [Le_Y^2 + 4(\sigma Le_Y + k^2 + Le_Y \Lambda)]^{\frac{1}{2}}}{2}.$$

For $Re(\sigma) > 0$, the solutions corresponding to the eigenvalues with a negative sign are unbounded as $Y_0 \rightarrow 0$ ($x \rightarrow -\infty$), and hence must be discarded. We are left with 5 linearly independent, bounded, asymptotic solutions, denoted by $i = 1, \dots, 5$, say.

Equation (26) has an exact particular integral, \mathbf{u}_L^p , valid in the entire region to the left of $x = 0$, of the form

$$\mathbf{u}_L^p = \mathbf{u}_0^p + \mathbf{u}_1^p Y_0 + \mathbf{u}_2^p Y_0^{1/h_1} + \mathbf{u}_3^p Y_0^{Le_F/h_1}.$$

We now therefore have the asymptotic general solution of equation (26), denoted by $\tilde{\mathbf{u}}$, valid as $Y_0 \rightarrow 0$ or $x \rightarrow -\infty$, of the form

$$\tilde{\mathbf{u}} = \sum_{i=1}^5 a_i \tilde{\mathbf{u}}_i + \mathbf{u}_L^p, \quad (29)$$

where a_i are (complex) constants of integration, to be determined.

4.2 Asymptotic solution as $x \rightarrow \infty$

In the region $x > 0$ equation (24) can be rewritten using Y_0 as the independent variable as

$$h_2 Y_0 \frac{d\mathbf{u}}{dY_0} = \mathbf{A}\mathbf{u} + \mathbf{s}, \quad (30)$$

and in this region, in terms of Y_0 , the other steady variables are, from equations (14)-(16),

$$T_0 = 1 + Q + \alpha_2 Y_0, \quad F_0 = 0,$$

where

$$\alpha_2 = -\frac{Q\Lambda}{h_2(h_2 - 1)},$$

As $x \rightarrow \infty$, corresponding to $Y_0 \rightarrow 0$, we can therefore expand equation (30) in the form

$$h_2 Y_0 \frac{d\mathbf{u}}{dY_0} = (\mathbf{A}_0^* + \mathbf{A}_1^* Y_0 + \dots)\mathbf{u} + \mathbf{s}_0^* + \mathbf{s}_1^* Y_0 + \dots \quad (31)$$

where again the co-efficients matrices, \mathbf{A}_0^* , etc, depend only on the parameters and on σ and k . Noting that in the region $x > 0$, we must have $F_1 = X_1 \equiv 0$, the homogeneous part of equation (31) has 8 independent asymptotic solutions of the form

$$\mathbf{u} = \hat{\mathbf{u}}_i = Y_0^{\lambda_i} (\mathbf{b}_0^i + \mathbf{b}_1^i Y_0 + \dots), \quad i = 1, \dots, 8, \quad (32)$$

where $h_2 \lambda_i$ and \mathbf{b}_0^i are the eigenvalues and eigenvectors, respectively, of \mathbf{A}_0^* , and the co-efficients \mathbf{b}_1^i , etc., are found by substituting (32) into equation (31) and equating powers of Y_0 . The eigenvalues of \mathbf{A}_0^* are

$$\pm k, \quad \frac{1 \pm [1 + 4(\sigma\phi + k^2)]^{\frac{1}{2}}}{2}, \quad \frac{1 \pm [1 + 4Pr(\sigma\phi + Prk^2)]^{\frac{1}{2}}}{2Pr}, \quad \frac{Le_Y \pm [Le_Y^2 + 4(\sigma\phi Le_Y + k^2 + Le_Y \Lambda)]^{\frac{1}{2}}}{2}.$$

where $\phi = 1/(1 + Q) = 1/T_{ad}$. For $Re(\sigma) > 0$, the solutions corresponding to eigenvalues with a positive sign are unbounded as $Y_0 \rightarrow 0$ ($x \rightarrow \infty$), and again must be discarded. We are thus left with 4 linearly independent, bounded asymptotic solutions.

Equation (30) has an exact particular integral valid in the region **to the right of $x = 0$** , \mathbf{u}_R^p , of the form

$$\mathbf{u}_R^p = \mathbf{b}_0^p + \mathbf{b}_1^p Y_0.$$

We thus now have the **asymptotic** general solution, **denoted by $\hat{\mathbf{u}}$** , valid as $Y_0 \rightarrow 0$ or $x \rightarrow \infty$, of the form

$$\hat{\mathbf{u}} = \sum_{i=1}^4 b_i \hat{\mathbf{u}}_i + \mathbf{u}_R^p, \quad (33)$$

where the b_i are (complex) constants of integration, to be determined.

4.3 *Jump conditions at $x = 0$*

Upon linearization in ϵ , the jump conditions (8)-(10) to be applied at the reaction sheet ($x = 0$), become

$$[T_1] = [u_1] = [v_1] = [P_1] = [F_1] = F_1 = [Y_1] = 0, \quad (34)$$

$$[U_1] = [V_1] = [X_1 + Z_1] = 0, \quad (35)$$

$$T_1 = 0. \quad (36)$$

4.4 *Numerical solution of linearized equations*

In summary, the linearized problem consists of equation (24) subject to $\mathbf{u} \rightarrow \tilde{\mathbf{u}}$ as $x \rightarrow -\infty$ and $\mathbf{u} \rightarrow \hat{\mathbf{u}}$ as $x \rightarrow \infty$, where $\tilde{\mathbf{u}}$ and $\hat{\mathbf{u}}$ are given by equations (29) and (33), respectively, and to the jump conditions (34)-(36) at $x = 0$.

The numerical solution of the problem is achieved as follows. Firstly, each of the five asymptotic linearly independent solutions, $\hat{\mathbf{u}}_i$, $i = 1, \dots, 5$ which are valid as $x \rightarrow -\infty$, are used as initial conditions in the numerical integration of the homogeneous form of equation (24), starting from a sufficiently large negative value of x (corresponding to a small value of Y_0), up to $x = 0$. A fourth-order Runge-Kutta routine with adaptive step size was used for the numerical integrations. These integrations thus give numerically the five bounded independent solutions of equation (24), denoted by \mathbf{u}_L^i , which are now valid throughout the

region left of $x = 0$. Hence the general solution in this region, \mathbf{u}_L , is of the form

$$\mathbf{u}_L = \sum_{i=1}^5 a_i \mathbf{u}_L^i + \mathbf{u}_L^p.$$

In particular, we can now evaluate numerically the general solution at $x = 0$ reached from the left, i.e. $\mathbf{u}_L(x = 0)$.

Similarly, the four asymptotic linearly independent solutions, $\hat{\mathbf{u}}_i$, $i = 1, \dots, 4$, can be used as initial conditions for numerical integration of the homogeneous part of equation (24) back to $x = 0$ starting from a sufficiently large value of x . This gives numerically the four bounded linearly independent solutions in the entire region to the right of $x = 0$, denoted by \mathbf{u}_R^i , $i = 1, \dots, 4$. The general solution for $x > 0$ is hence of the form

$$\mathbf{u} = \mathbf{u}_R = \sum_{i=1}^4 b_i \mathbf{u}_R^i + \mathbf{u}_R^p,$$

from which $\mathbf{u}_R(x = 0)$ can be evaluated.

It remains to determine the (complex) constants of integration, a_i and b_i . These are chosen so as to satisfy the jump conditions (34)-(35). Given the form of these jump conditions, we define a new reduced set of dependent variables by $\mathbf{q} = (T_1, u_1, v_1, P_1, F_1, Y_1, U_1, V_1, X_1 + Z_1)^T$, which is hence straightforwardly constructed by adding the ninth and tenth entries of \mathbf{u} . The jump conditions (34) and (35) can thus be written in the form

$$\mathbf{q}_L(x = 0) = \mathbf{q}_R(x = 0), \quad (37)$$

where \mathbf{q}_L and \mathbf{q}_R are just the reduced versions of \mathbf{u}_L and \mathbf{u}_R , respectively. This is a system of 9 simultaneous linear equations for the 9 constants of integration, a_1 to a_5 and b_1 to b_4 . These can hence be straightforwardly determined for given parameters such that the conditions (37) are satisfied.

However, determining the constants of integration in this manner also fixes the value of T_1 at $x = 0$. Only for certain discrete values of σ (for fixed k) will the condition (36) be simultaneously satisfied. The eigenvalues are hence found by Newton-Raphson iteration on the condition $T_1 = 0$. Note that, for fixed parameters, equation (24) and the boundary and jump condition do not have any leading order dependence on the activation energy of the chain-branching step, unlike for the one-step model when Le_F is fixed [9].

5 Linear stability results

As for the one-step Reactive Navier-Stokes model [9] and for the two-step CDM [14], we find there are only two possible distinct modes. The first of these is a cellular mode for which σ is real, and the second a pulsating mode corresponding to a complex conjugate pair of eigenvalues. In this section, the linear neutral stability boundaries and linear dispersion relations of the two-step branching flame are examined for both these cellular and pulsating branches, when thermal expansion is taken into account. The effect of the various model parameters is explored, and the results compared and contrasted with both the two-step CDM predictions and the Reactive Navier-Stokes one-step chemistry results.

5.1 Effect of thermal expansion and comparison with CDM

We first examine the effect of varying thermal expansion (through the heat of reaction Q) on the stability of the flame. For the two-step CDM, in which hydrodynamical effects induced by thermal expansion are completely ignored, the linear stability is independent of Q for fixed values of $Q_D = Q/(T_c - 1)$ [14]. Since the steady, planar flame speed also depends on Q and T_c only through the combination Q_D (see §3), it is

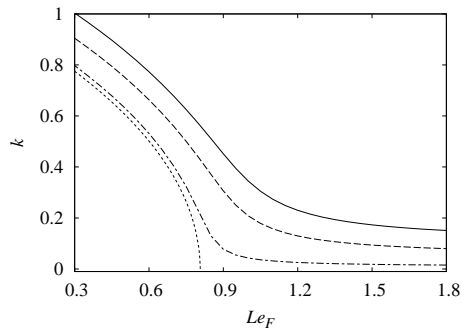


Figure 2. Neutral stability boundary of the cellular branch in the $Le_F - k$ plane for $Le_Y = 1$ and $Q_D = 1.25$ ($T_c = 1 + Q/1.25$), when $Q = 5$ ($T_c = 5$) (solid line), $Q = 1$ ($T_c = 1.8$) (dashed line) and $Q = 0.1$ ($T_c = 1.08$) (dot-dashed line). Dotted line is the CDM prediction for $Q_D = 1.25$.

consistent to first examine the stability effect of varying Q while keeping Q_D fixed (and hence varying T_c appropriately).

5.1.1 Cellular branch. Figure 2 shows the neutral stability boundaries in the $Le_F - k$ plane for various values of Q when $Q_D = 5/4$ (hence $T_c = 1 + 4Q/5$) and $Le_Y = 1$. These neutral stability boundaries correspond to the loci on which $\sigma = 0$, and the flame is unstable to wavenumbers below the curve and stable to those above it. The dotted curve in figure 2 is the neutral stability boundary for $Q_D = 5/4$ as predicted by the two-step CDM. The CDM results predicts the flame is stable to all wavenumbers above a critical fuel Lewis number, while as Le_F decreases below this critical value the flame becomes unstable to an increasingly wide band of wavenumbers [14]. Since the CDM ignores hydrodynamic effects, the instability predicted by the CDM is of a purely thermal-diffusive nature. However, the two-step Reactive Navier-Stokes model studied here simultaneously captures hydrodynamic effects, and figure 2 shows that the stability results are then qualitatively different to the predictions of the CDM, even when Q is quite small. In particular the flame is unstable for all values of Le_F , at least for sufficiently long wavelength perturbations. While the CDM is valid in the asymptotic limit $Q \rightarrow 0$, figure 2 also shows that even for a physically very small heat release of $Q = 0.1$, this hydrodynamical effect is still important. Hence, while the $Q = 0.1$ stability boundary follows closely that of the CDM for fuel Lewis numbers sufficiently below the critical Le_F predicted by the CDM, it then curves sharply to the right near this critical value. Hence the concept of an Le_F stability boundary predicted by the CDM is not physically valid for gaseous flames.

A second difference is that, while the CDM results are independent of Q for fixed Q_D , figure 2 shows that when hydrodynamic effects are taken into account, the stability does depend on Q (with Q_D fixed). Indeed, the higher the value of Q , the wider the range of unstable wavenumbers for any fixed Le_F , so that the neutral stability curve for a given Q lies above those corresponding to lower values in the $Le_F - k$ plane.

For the cellular instability, a larger value of the neutrally stable wavenumber also corresponds to a more unstable flame, in that the maximum linear growth rate is also larger. This is demonstrated in figure 3, which shows the dispersion relations (growth rate, σ , as a function of wavenumber, k) for various values of Q when $Le_F = 0.6$ and when $Le_F = 1.2$. Note, for the cellular mode, σ is also zero at $k = 0$. As Q increases, not only does the band of unstable wavenumbers increase for fixed Le_F , but so does the wavenumber with the maximum value of the growth rate, and the value of this maximum also rapidly increases. Thus weak perturbations to the planar flame will grow much more rapidly and become non-linear more quickly for larger Q . One would expect that cells would first appear on the flame with a wavenumber close to that with maximum linear growth rate. However, it should be noted that a linear analysis only gives information about stability boundaries, onset and initial growth stage of the instability. It is not relevant to the fully developed non-linear cells, which may be of a quite different characteristic wavelength to that predicted by the linear analysis [13].

The dependence of the cellular instability on Q , with fixed Q_D , described above is qualitatively the same as the effect of thermal expansion on the one-step model. The differences between the two-step CDM

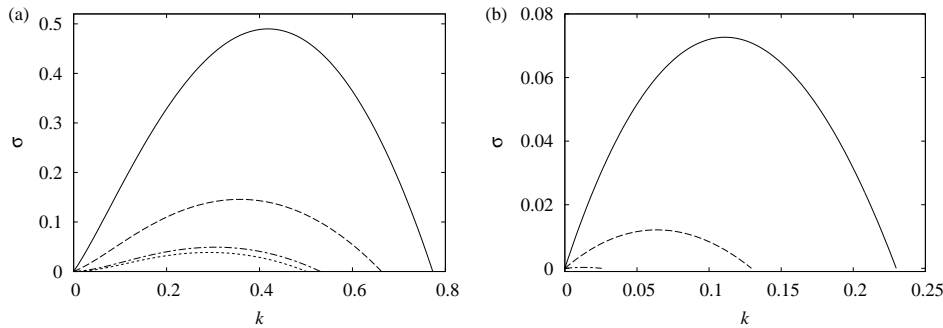


Figure 3. Dispersion relations of the cellular instability for $Le_Y = 1$, $Q_D = 1.25$ and (a) $Le_F = 0.6$ and (b) $Le_F = 1.2$, when $Q = 5$ ($T_c = 5$) (solid line), $Q = 1$ ($T_c = 1.8$) (dashed line) and $Q = 0.1$ ($T_c = 1.08$) (dot-dashed line). Dotted line in (a) is the CDM prediction for $Q_D = 1.25$.

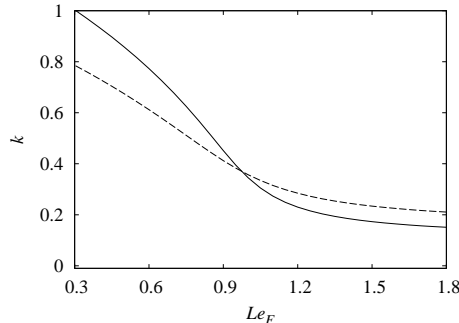


Figure 4. Neutral stability boundary of the cellular branch in the $Le_F - k$ plane for $Le_Y = 1$, $T_c = 5$ and $Q = 5$ ($Q_D = 1.25$) (solid line) and $Q = 6$ ($Q_D = 1.5$) (dashed line).

and Reactive Navier-Stokes model predictions are also qualitatively the same in the one-step case, cf. Jackson and Kapila [9]. For the two-step model, however, the effect of varying Q but keeping the crossover temperature, T_c , fixed is somewhat more complicated. Figure 4 shows the neutral stability boundaries for $T_c = 5$ and heats of reaction $Q = 5$ and 6 (corresponding to $Q_D = 1.25$ and 1.5 , respectively). The stability response of the flame with the larger thermal expansion is less sensitive to the fuel Lewis number than for the smaller Q case. The neutral stability curves for the two thermal expansions cross in the $Le_F - k$ plane, in this case at $Le_F = 0.97$, and the flame is more (less) unstable for the smaller degree of thermal expansion when Le_F is above (below) this value. Thus, for Le_F sufficiently below unity, the stabilizing effect of decreasing thermal expansion is outweighed by a destabilizing thermal-diffusive effect as Q_D is lowered by the increase in Q (cf. §5.2.1).

5.1.2 Pulsating branch. Recall that the pulsating instability branch corresponds to a complex conjugate pair of eigenvalues. For the CDM limit, this instability is found to occur only provided Q_D is below a finite value, and then only if Le_F is above a critical value which is greater than unity [14]. This critical value of the fuel Lewis number decreases and tends to unity as $Q_D \rightarrow 1$. However, $Q_D - 1$ still has to be sufficiently small for the pulsating instability to occur for values of Le_F realistic to gaseous flames. For example, figure 5 shows the CDM prediction of the neutral stability boundary when $Q_D - 1 = 0.1$ (for $Le_Y = 1$), in which case Le_F needs to be above 1.772 for the flame to be unstable to the pulsating mode. Note that the CDM predicts that the pulsating instability first appears at a non-zero wavenumber [14]. For example, when $Q_D = 1.1$, the wavenumber is 0.252 at the turning point of the CDM neutral stability curve in figure 5.

Figure 5 also shows the neutral stability boundaries, on which $Re(\sigma) = 0$, predicted by the Reactive Navier-Stokes model with $O(1)$ values of the thermal expansion, but with $Q_D = 1.1$ fixed. As for the CDM prediction, when hydrodynamic effects are taken into account, the pulsating instability still only occurs above a critical value of Le_F . Increasing thermal expansion lowers this critical value somewhat. For example, the instability occurs when $Le_F > 1.732$ for $Q = 1$ and when $Le_F > 1.683$ for $Q = 5$ in figure

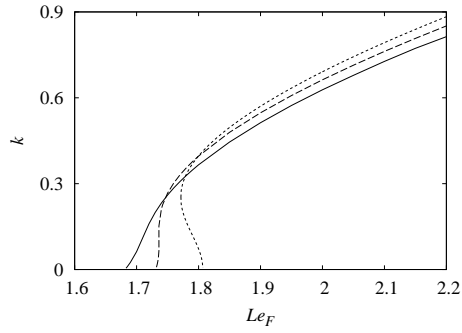


Figure 5. Neutral stability boundary of the pulsating branch in the $Le_F - k$ plane for $Le_Y = 1$ and $Q_D = 1.1$ ($T_c = 1 + Q/1.1$) when $Q = 5$ ($T_c = 5.545$) (solid line) and $Q = 1$ ($T_c = 1.909$) (dashed line). Dotted line is the CDM prediction for $Q_D = 1.1$.

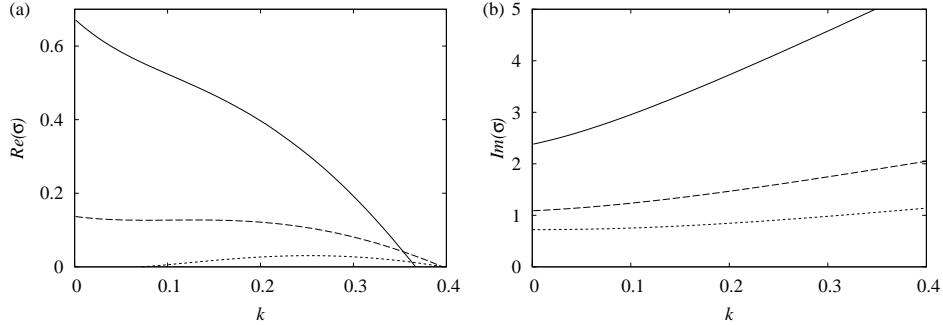


Figure 6. Dispersion relation ((a) growth rate and (b) frequency) of the pulsating instability for $Le_Y = 1$, $Le_F = 1.8$, $Q_D = 1.1$, $Q = 5$ ($T_c = 5.545$) (solid line) and $Q = 1$ ($T_c = 1.909$) (dashed line). Dotted line is the CDM prediction for $Q_D = 1.25$.

5, as compared to the CDM prediction of $Le_F > 1.772$. Also, in contrast to the CDM prediction, there is no turning point in the neutral stability boundary when hydrodynamic effects are taken into account, i.e. the instability first occurs at zero wavenumber.

Figure 5 also shows that for Lewis numbers of the fuel above the CDM critical value, thermal expansion only has a weak effect on the neutral stability boundary. In this region, the boundary for larger values of Q actually lie below those corresponding to lower values, and hence also below the CDM prediction. This weak dependence of the pulsating instability boundary on thermal expansion, for fixed Q_D , is again qualitatively the same as the effect of Q on the boundaries in the one-step model [9].

It is important to note, however, that for the pulsating branch, the range of unstable wavenumbers does not relate to the degree of instability of the wave. Hence, although thermal expansion has only a weak effect on the stability boundaries, it has a large effect on the actual dispersion relations (growth rate, $Re(\sigma)$, and frequency, $Im(\sigma)$, versus wavenumber) as shown, for example, in figure 6. Furthermore, while the CDM predicts that the maximum growth rate occurs at a non-zero wavenumber (hence the turning point in the CDM neutral stability curve), for $O(1)$ values of Q , the maximum growth rate occurs at $k = 0$ (infinite wavelength). This corresponds to a purely one-dimensional pulsation of the planar flame being the linearly most unstable mode. The growth rate at $k = 0$, and hence the degree of instability, also increases quite rapidly with Q , even though the range of unstable wavenumbers decreases. The frequency of the pulsation also increases with Q , and hence the period, given by $2\pi/Im(\sigma)$, decreases.

5.2 Effect of crossover temperature and comparison with one-step model

The two-step CDM results predict that as $T_c \rightarrow T_{ad}$, the flame becomes increasingly unstable in that the critical values of Le_F at which the cellular and pulsating instability occurs both tend to unity [14]. In this section we examine the effect of T_c , for fixed Q , when realistic thermal expansion is taken into account.

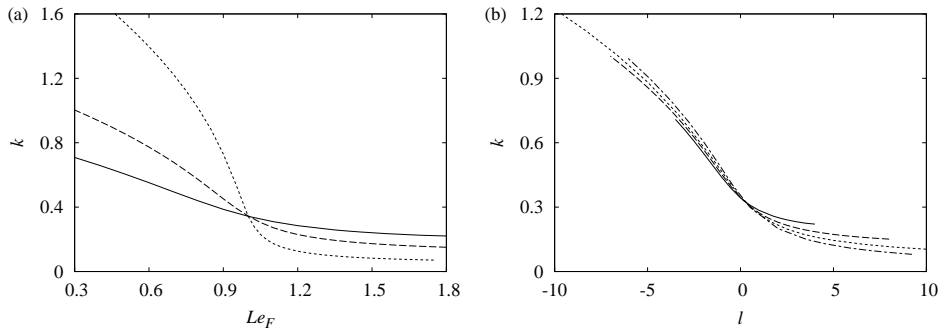


Figure 7. Neutral stability boundary of the cellular branch (a) in the $Le_F - k$ plane and (b) in the $l - k$ plane ($l = \beta_{\text{eff}}(Le_F - 1)$, $\beta_{\text{eff}} = 2Q_D/(Q_D - 1)$), for $Q = 5$, $Le_Y = 1$ and $T_c = 4$ ($Q_D = 5/3$, $\beta_{\text{eff}} = 5$) (solid lines), $T_c = 5$ ($Q_D = 1.25$, $\beta_{\text{eff}} = 10$) (dashed lines) and $T_c = 5.7$ ($Q_D = 1.064$, $\beta_{\text{eff}} = 33.333$) (dotted lines). Also shown in (b) is the high-activation energy one-step model result from Jackson and Kapila [9] (dot-dashed line)

5.2.1 Cellular branch. Figure 7(a) shows the neutral stability boundary when $Q = 5$ for various T_c (and hence varying Q_D also), when $Le_Y = 1$. Again, when thermal expansion is taken into account, the flame is unstable to the cellular instability for all values of Le_F , and there is no critical Lewis number as in the CDM prediction. For unit values of Le_F and Le_Y and realistic thermal expansions, the cellular instability is insensitive to T_c . Thus the neutral stability curves for the different values of T_c cross at $Le_F = 1$ in figure 7(a). As $T_c \rightarrow T_{\text{ad}}$ (or equivalently as $Q_D \rightarrow 1$) the flame becomes increasingly sensitive to thermal-diffusive effects associated with the Lewis number of the fuel. These effects are destabilizing when $Le_F < 1$, but stabilizing for $Le_F > 1$, and appear to have an enhanced role as $Q_D \rightarrow 1$. Thus, for fixed values of Le_F which are less than unity, the flame becomes increasingly unstable (the neutrally stable wavenumber becomes larger) as T_c increases. Conversely, for fixed values of $Le_F > 1$, the thermal-diffusive effect becomes more strongly stabilizing, and hence the neutrally stable wavenumber decreases.

These dependences of the cellular mode stability boundaries in the $Le_F - k$ plane on T_c are in fact qualitatively similar to the dependence of the one-step model boundaries on activation energy [12]. Indeed, in the high-activation energy asymptotic limit, the one-step flame stability depends on Le_F and the activation energy only through a reduced Lewis number, l , defined by $l = \beta(Le_F - 1)$ [9], such that $l = O(1)$ as $\beta \rightarrow \infty$ (i.e. the NEF approximation), where β is the Zel'dovich number (the **non-dimensional** activation energy of the one-step reaction rate). Dold [14] showed that, for the constant density models, in the limit $T_c \rightarrow T_{\text{ad}}$ there is a direct correspondence between the flame structure and stability of the one-step and two-step chemistry models. Indeed, under the asymptotic limit $Q_D \rightarrow 1$, the two-step model can be described to leading order by a one-step model which has an *effective* Zel'dovich number defined by $\beta_{\text{eff}} = 2Q_D/(Q_D - 1) = 2(T_{\text{ad}} - 1)/(T_{\text{ad}} - T_c)$. In other words, in the limit, the two-step CDM linear stability depends to leading order on Le_F only through a reduced Lewis number defined by $l = \beta_{\text{eff}}(Le_F - 1)$, and this leading order dependence is then the same to that of the one-step model with Zel'dovich number $\beta = \beta_{\text{eff}}$ [14].

Since the analysis in Dold [14] is based on comparing the two-step jump conditions in the temperature and fuel in the limit $Q_D \rightarrow 1$ with those of the one-step model, and these jump conditions are unchanged when thermal expansion is taken into account, one would expect the analogy between the two models to still hold in the Reactive Navier-Stokes formulation. However, it remains to determine how predictive the one-step model is of the two-step results for non-zero values of $Q_D - 1$ (finite β_{eff}) when thermal expansion is taken account of. Figure 7(b) shows the neutral stability boundaries in the reduced Lewis number-wavenumber ($l - k$) plane, when $Q = 5$, for $T_c = 4$ (corresponding to $Q_D = 1.667$, $\beta_{\text{eff}} = 5$), $T_c = 5$ ($Q_D = 1.25$, $\beta_{\text{eff}} = 10$) and $T_c = 5.7$ ($Q_D = 1.064$, $\beta_{\text{eff}} = 33.333$). Also shown is the one-step model high activation energy result for $Q = 5$ from Jackson and Kapila [9]. It can be seen that, in this plane, the stability boundary is only weakly dependent on T_c , and the chain-branching model results do converge to those of one-step model as $Q_D \rightarrow 1$. Furthermore, the prediction of the one-step model remains quite good even when $Q_D - 1$ is of the order of unity, or β_{eff} is moderately large. However, it is important to note that the effective activation energy, β_{eff} , with which the one-step model describes to leading order the two-step model, has no correspondence to any real activation energy (e.g. the activation energy of the

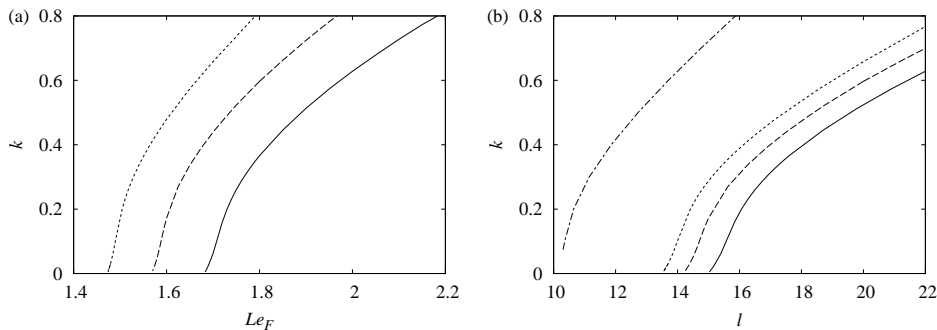


Figure 8. Neutral stability boundary of the pulsating branch (a) in the $Le_F - k$ plane and (b) in the $l - k$ plane for $Q = 5$, $Le_Y = 1$ and $T_c = 5.545$ ($Q_D = 1.1$, $\beta_{\text{eff}} = 22$) (solid lines), $T_c = 5.6$ ($Q_D = 1.087$, $\beta_{\text{eff}} = 25$) (dashed lines) and $T_c = 5.65$ ($Q_D = 1.075$, $\beta_{\text{eff}} = 28.57$) (dotted lines). Also shown in (b) is the high-activation energy one-step model result from Jackson and Kapila [9] (dot-dashed line)

chain-branching step), but is a function of the the crossover temperature [14]. In particular, β_{eff} can be arbitrarily large compared to any physical Zel'dovich number.

5.2.2 Pulsating branch. Figure 8(a) shows the neutral stability curves of the pulsating branch for various values of T_c , when $Q = 5$ and $Le_Y = 1$. Unsurprisingly, since thermal expansion has a weak effect on the pulsating instability boundary, the dependence on T_c (and hence on Q_D) follows the trends predicted by the CDM [14], i.e. the critical value of Le_F decreases towards unity with Q_D . The dependence of the neutral stability curves in the $l - k$ plane is more interesting. Figure 8(b) shows the neutral stability boundaries in this plane, together with the one-step boundary for the pulsating branch when $Q = 5$ from Jackson and Kapila [9]. The values of T_c used in figure 8 correspond to $\beta_{\text{eff}} = 22$, 25 and 28.57. The stability of the pulsating branch in the $l - k$ plane can be seen to be much more sensitive to β_{eff} than is the cellular instability (cf figure 7(b)). Furthermore, while the two-step model results do tend to those of the one-step model as $\beta_{\text{eff}} \rightarrow \infty$ ($T_c \rightarrow T_{\text{ad}}$), it is apparent that a very high β_{eff} would be required for the one-step and two-step model results to quantitatively agree.

Interestingly, these effects of finite β_{eff} on the two-step model pulsating instability boundary in the $l - k$ plane are actually very similar to the role of finite activation energy in the one-step model. Lasseigne *et al.* [10] performed a numerical linear stability analysis of the one-step reaction CDM with finite activation energy (i.e. without invoking the high activation energy asymptotic limit). They also found that, for the pulsating instability, the results are sensitive to β , with the finite activation energy results lying further to the right in the $l - k$ plane the lower the Zel'dovich number, and that a very large β would be required for the high-activation energy asymptotic one-step theory to be quantitatively predictive.

5.3 Effect of Lewis number of intermediates.

Figure 9 shows the stability boundaries of the cellular mode for various values of the intermediates Lewis number for $Q = 5$ when $T_c = 5$ and when $T_c = 5.7$. Increasing Le_Y can be seen to have a destabilizing effect on the cellular mode, in that the neutral stability curves for higher Le_Y lie entirely above those of lower values. However, figure 9 also shows that the destabilizing effect becomes weaker as T_c increases, and the results becomes independent of Le_Y in the limit $Q_D \rightarrow 1$. This is to be expected, since as T_c tends to T_{ad} , the amount of intermediates, and hence the role of their diffusion in the flame structure, decreases (see figure 1). For the pulsating branch, however, increasing Le_Y is found to have a stabilizing effect, in that the critical fuel Lewis number for this instability becomes larger as Le_Y increases, cf. the CDM prediction [14]. The dependence of the pulsating branch critical fuel Lewis number on Le_Y again becomes weaker as T_c tends to T_{ad} .

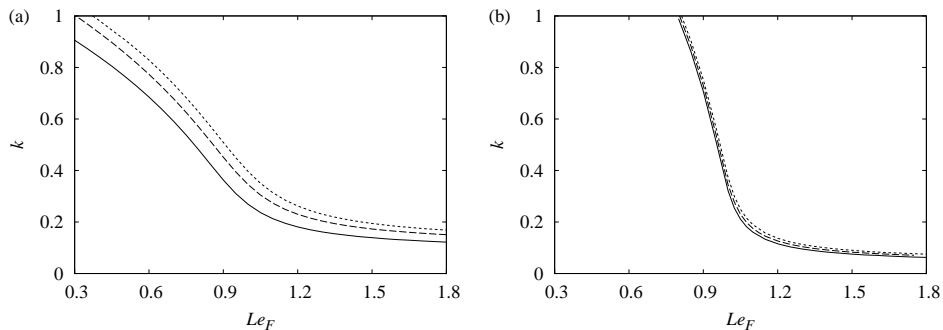


Figure 9. Neutral stability boundary of the cellular branch in the $Le_F - k$ plane for $Q = 5$ and (a) $T_c = 5$ and (b) $T_c = 5.7$, when $Le_Y = 0.3$ (solid lines), $Le_Y = 1$ (dashed lines) and $Le_Y = 1.8$ (dotted lines)

6 Conclusions

A leading order, high-activation energy asymptotic linear stability analysis of premixed flames has been performed in the context of a two-step chain-branching chemistry, Reactive Navier-Stokes model. The main purpose was to extend the results of a previous study, which employed a constant density model [14], to take into account hydrodynamic effects induced by thermal-expansion. The main difference is that, as found previously for the standard one-step chemistry model, when thermal expansion is taken into account, the flame always has a band of perturbation wavelengths which are unstable to a cellular mode. Hence the region of stable fuel Lewis numbers predicted by the constant density study does not exist for realistic thermal expansions. On the other hand, a sufficiently large, critical fuel Lewis number is still required for the flame to be unstable to a pulsating mode, as correctly predicted by the constant density model. For this mode, thermal expansion has a weak effect on the stability boundaries, but a large effect on the maximum linear growth rate.

Comparing the results with those of a previous one-step reaction Reactive Navier-Stokes study [9], the two-step behaviour is found to be broadly similar, and hence no qualitatively new behaviour is revealed by using the more realistic two-step model. Furthermore, in the limit that the branching cross-over temperature, T_c , tends to the adiabatic flame temperature, T_{ad} , the two-step model results can be identified to leading order with those of a one-step model with a suitably defined effective activation energy. This effective activation energy tends to infinity in the limit $T_c \rightarrow T_{ad}$, and hence unlike a physical activation energy, can be arbitrarily large. The one-step model approximation is found to be quantitatively good for the cellular instability even when the effective activation energy is moderately large. For the pulsating instability, however, it is found that a crossover temperature very close to the adiabatic temperature (corresponding to a very large effective activation energy) would be required for the effective one-step results to be quantitatively predictive of the two-step model results.

Here we have considered adiabatic, freely propagating premixed flames. Secondary effects, such as buoyancy, heat loss, endothermic branching reaction, finite activation energy effects, etc., could also be included in future linear stability studies. However, calculations of the fully non-linear stages of the evolution, and studies of how these compare and contrast to the fully non-linear one-step model results, would perhaps be a more important next step. Direct numerical simulations using the two-step model, along the lines of the one-step computations in Sharpe and Falle [13], will be presented in a future article.

Acknowledgments

The author was supported by an EPSRC Advanced Fellowship. The author is also grateful to Ted Bearman, Rio Tinto Technology & Innovation for research sponsorship. Thanks also go to Tom Jackson for previously providing the one-step model data used in Jackson and Kapila [9].

References

- [1] Buckmaster, J. D. and Ludford, G. S. S., 1982, *Theory of Laminar Flames*. Cambridge University Press.
- [2] Sivashinsky, G. I., 1983, Instabilities, pattern formation, and turbulence in flames. *Ann. Rev. Fluid Mech.*, **15**, 179–199.
- [3] Strehlow, R. A., 1985, *Combustion Fundamentals*. McGraw-Hill.
- [4] Law, C. K., 2006, *Combustion Physics*. Cambridge University Press.
- [5] Sivashinsky, G. I., 1977, Diffusional-thermal theory of cellular flames. *Combust. Sci. Tech.*, **15**, 137–146.
- [6] Frankel, M. L. and Sivashinsky, G. I., 1982, The effect of viscosity on hydrodynamic stability of a plane flame front. *Combust. Science Tech.*, **29**, 207–224.
- [7] Matalon, M. and Matkowsky, B. J., 1982, Flames as gasdynamic discontinuities. *J. Fluid Mech.*, **124**, 239–259.
- [8] Pelce, P. and Clavin, P., 1982, Influence of hydrodynamics and diffusion upon the stability limits of laminar premixed flames. *J. Fluid Mech.*, **124**, 219–237.
- [9] Jackson, T. L. and Kapila, A. K., 1984, Effect of thermal expansion on the stability of plane, freely propagating flame. *Combust. Science Tech.*, **41**, 191–201.
- [10] Lasseigne, D. S., Jackson, T. L. and Jameson, L., 1999, Stability of freely propagating flames revisited. *Combust. Theory Model.*, **3**, 591–611.
- [11] Liberman, M. A., Bychkov, V. V., Goldberg, S. M. and Book, D. L., 1994, Stability of a planar flame front in the slow-combustion regime. *Physical Review E*, **49**, 445–453.
- [12] Sharpe, G. J., 2003, Linear stability of planar premixed flames: Reactive Navier-Stokes equations with finite activation energy and arbitrary Lewis number. *Combust. Theory Model.*, **7**, 45–65.
- [13] Sharpe, G. J. and Falle, S. A. E. G., 2006, Nonlinear cellular instability of planar premixed flames: numerical simulations of the Reactive Navier-Stokes equations. *Combust. Theory Model.*, **10**, 483–514.
- [14] Dold, J. W., 2007, Premixed flames modelled with thermally sensitive intermediate branching kinetics *Combust. Theory Model.*, in press (available in online edition at <http://www.tandf.co.uk/journals/titles/13647830.asp>).
- [15] Dold, J. W., Thatcher, R. W., Omon-Arancibia, A. and Redman, J., 2003, From one-step to chain-branching premixed-flame asymptotics. *Proc. Combust. Inst.*, **29** 1519–1526.
- [16] Dold, J. W., Weber, R.O., Thatcher, R. W., Shah, A. A., 2003, Flame balls with thermally sensitive intermediate kinetics. *Combust. Theory Model.*, **7**, 175–203.
- [17] Williams, F. A., 1985, *Combustion Theory*, 2nd Ed. Addison-Wesley Publishing.
- [18] Gasser, I. and Szmolyan, P., 1995, Detonation and deflagration waves with multistep reaction schemes. *SIAM J. Appl. Math.*, **55**, 175–191.
- [19] Sharpe, G. J., 1997, Linear stability of idealized detonations. *Proc. Roy. Soc. Lond. A*, **453**, 2603–2625.