Biospecific Affinity Chromatography: Computational Modelling *via* Lattice Boltzmann Method and Influence of Lattice-Based Dimensionless Parameters

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Abstract: Based on a dynamic (i.e. time-dependent) one-dimensional approach, this work applied lattice Boltzmann method (LBM) to computationally model biospecific affinity chromatography (BAC). With governing equations expressed in lattice-based dimensionless form, LBM was implemented in D1Q2 lattice by assigning particle distribution functions to adsorbate concentration in both fluid and solid phases. The LBM simulator was firstly tested in view of a classic BAC work on lysozyme and the streaming step relating to adsorbate concentration in the solid-phase was suppressed from the LBM code with no loss of functionality. Expected behaviour of breakthrough curves was numerically reproduced and the influence of lattice-based dimensionless parameters was examined. The LBM simulator was next applied so as to assess lattice-based dimensionless parameters regarding an experimental BAC work on lipase.

Keywords: Biospecific affinity chromatography, phenomenological modelling, numerical simulation, lattice Boltzmann method.

1. INTRODUCTION

Innovative design, scale-up procedures and optimal operation can be rapidly achieved via numerical simulation [1], thus reducing the number of tests and saving valuable material and human resources required in bioseparation processes. Given the costs of industrial-scale experimentation, distinct design parameters and operation scenarios can be numerically tested for process viability [2] and such has been the case of bioreactor engineering [3-5].

Bioseparation models are prone to be complex enough to justify computational modelling [6-8] and biospecific affinity chromatography (BAC) may benefit from it if comprehensive knowledge is required [9]. Use of computational fluid dynamics (CFD) towards bioprocesses has increased as its importance has been recognized while suitable techniques have been developed [10].

Distinct approaches can be followed. By relying on continuum concept, macroscale simulation applies basic conservation principles to obtain differential equations to observable properties, which are numerically solved. Finite differences method (FDM), finite elements method (FEM) and finite volumes method (FVM) have been among widespread discretization techniques for food and bioprocesses [11-16].

Microscale models individually identify constituent particles in conjunction with their mutual interactions. In molecular dynamics (MD) simulation, Newtonian mechanics is then applied to predict space-time evolution of such (colossal!) particle collection while statistical mechanics is applied to retrieve observable properties. The task is to simulate macroscopic behaviour from microscopic modelling, requiring huge computational effort and memory resources. MD has been used in pharmacological and emulsion research [17-20].

Between those two scales, mesoscopic models treat bulk media as cellular automata, i.e. as systems where quantities may only assume discrete values [21]. From the mathematical viewpoint, a particle distribution function is introduced in order to describe the behaviour of constituent particle collections [22]. Based on that function one implements lattice Boltzmann method (LBM), regarded as an extension of its predecessor method, namely lattice-gas cellular automata [23].

With pioneering ideas launched in [24] and more recent if compared to long-standing methods (e.g. FDM, FEM or FVM), LBM has become an alternative technique to simulate food and bioprocesses [25]. Interesting applications in view of biosystems have comprised flow through fibrous materials [26], particle suspensions and particle-fluid interactions [27], solute transport in porous media [28, 29], fluid dynamics of blood [30, 31], liquid-vapour interface [32] and chromatography [33, 34].

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As part of on-going research on LBM simulation of food and bioprocesses [35], LBM has been applied to biospecific affinity chromatography (BAC). Based on a dynamic one-dimensional reaction-diffusive model with Langmuir kinetics [36, 37], this work aimed at the modelling computational of one-component breakthrough curves via LBM. Governing differential equations were cast in lattice-based dimensionless form together with initial and boundary conditions. Trial LBM simulations were performed in view of a classic BAC work on lysozyme [38] and the LBM simulator was next applied towards an existing experimental work on lipase bioseparation [39].

While there is no doubt about the efficiency of methods like FDM, FEM or FVM to perform equivalent simulations, comparisons in terms of computational effort and memory use fall beyond the purpose of this work. Rather, its goal is to present LBM as an alternative route for BAC simulation. By rendering relatively simpler codes [40], LBM can easily deal with moving or free boundaries while being able to simulate fluid flow without directly solving Navier-Stokes equations. Such features are particularly appealing for those who have programmed their own computational fluid dynamics codes *via* FDM, FEM or FVM.

2. THEORY

2.1. Mathematical Models for Biospecific Affinity Chromatography (BAC)

BAC models have typically invoked 2nd-order adsorption and 1st-order desorption kinetics, uniform fluid flow and adsorbate transport by either convection or diffusion [7, 9, 16, 38, 41-43]. Additional assumptions have included uniform porosity ε over the chromatographic column and constant volumetric flow rate \dot{V} of the percolating solution. Hence, interstitial fluid velocity $v = \dot{V}/(\epsilon A)$ results uniform, being A the cross-sectional area of the column. In model equations one may use superficial velocity $\overline{v} = \dot{V}/A$, also referred to as seepage velocity, which is related to interstitial velocity via Dupuit-Forchheimer relation $\overline{v} = \varepsilon v$ [44]. Those models are dynamic with 1st-order spatial dependence so that adsorbate concentrations are allowed to vary along a coordinate z (usually the column axis) besides depending on time t.

This work modelled chromatographic columns as stratified cylindrical fixed-beds with inlet at z = 0 and outlet at z = L (= column length). Adsorbate concentrations in fluid and solid phases were identified respectively as c = c(z,t) and q = q(z,t) while

corresponding governing partial differential equations (PDEs) were put forward as:

$$\frac{\partial c}{\partial t} + v \frac{\partial c}{\partial z} = D \frac{\partial^2 c}{\partial z^2} - \frac{1 - \varepsilon}{\varepsilon} \dot{r}$$
(1)

$$\dot{r} = \frac{\partial q}{\partial t} = k_{ads} c (q_{max} - q) - k_{des} q$$
⁽²⁾

where *D* is axial diffusivity in the fluid phase, \dot{r} is the instantaneous rate at which adsorbate is transferred from fluid to solid phase, k_{ads} and k_{des} are respectively adsorption and desorption coefficients, and q_{max} is the maximum adsorption capacity (i.e. saturation) of the column.

Initial conditions were prescribed as:

$$c(z,0) = 0$$
 and $q(z,0) = 0$, for $0 \le z \le L$ (3)

In the proposed model framework, the governing equation for solid-phase concentration lacks partial derivatives with respect to coordinate z so that boundary conditions were only required for the fluid-phase PDE. Being $c_{in} \neq 0$ the adsorbate concentration in feed solution, one may impose Dirichlet condition at column inlet (z = 0) [7, 41, 45] and null Neumann condition at column exit (z = L), namely:

$$c(0,t) = c_{\text{in}} \text{ and } \left. \frac{\partial c}{\partial z} \right|_{z=L} = 0$$
, for $t > 0$ (4)

At inlet, one may alternatively prescribe Danckwerts condition [16, 34, 42], namely:

$$vc_{\rm in} = vc(0,t) - D \frac{\partial c}{\partial z}\Big|_{z=0}$$
, for $t > 0$ (5)

which simplifies to Dirichlet condition if D = 0.

In this work, LBM simulations use process parameters as suggested in [41], where Dirichlet inlet condition is adopted (thus Danckwerts condition was disregarded). Moreover, a numerical study has shown that no practical effect is introduced when one boundary condition is replaced by the other for low mass diffusivities, $D < 10^{-8}$ m²/s [46]. Differences become noticeable for $D > 10^{-8}$ m²/s but, in this case, simulated breakthrough curves deviate from experimental data, regardless of the boundary condition type imposed at column inlet.

2.2. Rationale of Lattice-Boltzmann Method (LBM)

Inspired on kinetic gas theory, LBM is a bottom-up technique different not only from top-down methods like

FDM, FEM or FVM but also from MD simulation, which is another bottom-up approach [22]. Contrast from MD relies on the fact that LBM treats the macroscale medium, whether solid or fluid, as a set of fictitious particles in a discrete space, namely a fictitious lattice. During discrete time steps and according to their speeds, such particles travel (stream) between adjacent sites along pre-defined directions (lattice links). As particles arrive at lattice sites, they mutually collide and their velocities become rearranged. By imposing mass and momentum conservations to such dynamics, referred to as streaming and collision, macroscopic behaviour can be simulated [23].

LBM mathematically relies on a particle distribution function $f(\vec{r}, \vec{u}, t)$ giving the probability of finding, at time *t*, fictitious particles about position \vec{r} with velocities between \vec{u} and $\vec{u} + d\vec{u}$. Observable properties (e.g. species concentration, bulk flow velocity or temperature) can be retrieved through moments of function *f* [25, 40].

LBM is implemented to numerically obtain function *f*, which is ruled by Boltzmann transport equation. In the absence of external forces, such governing equation is written as:

$$\frac{\partial f}{\partial t} + \vec{u} \cdot \vec{\nabla} f = \Omega(f)$$

$$\xrightarrow{BGK \text{ approximation for } \Omega} \frac{\partial f}{\partial t} + \vec{u} \cdot \vec{\nabla} f = \frac{f^{eq} - f}{\Delta t_{relax}}$$
(6)

where the collision operator $\Omega = \Omega(f)$ gives the variation rate of function *f* due to collisions between particles. In Equation (6), BGK approximation (after Bhatnagar-Gross-Krook) has been invoked to linearize such operator as $\Omega(f) = (f^{eq} - f)/\Delta t_{relax}$ [47], meaning that particles tend to local equilibrium values f^{eq} at a rate controlled by a relaxation time Δt_{relax} [48, 49].

In LBM, Equation (6) is written for each link k in the fictitious lattice so that it becomes referred to as lattice-Boltzmann equation:

$$\frac{\partial f_k}{\partial t} + \vec{u}_k \cdot \vec{\nabla} f_k = \frac{f_k^{\text{eq}} - f_k}{\Delta t_{\text{relax}}}$$
(7)

Distinct lattices are identified as DnQm, where *n* is the problem dimensionality (e.g., n = 1 = 1-D = one dimensional) and *m* refers to the speed model (= number of particle distribution functions to be solved for each observable property). Typical lattices for LBM simulations are depicted elsewhere [23, 25, 40]. Space-time discretisation of Equation (7) written for a given macroscopic property yields an algebraic equation whose numerical evolution is accomplished in two steps. During collision (time evolution), particle distribution functions f_k are updated from instant t to $t + \Delta t$ at all lattice sites, being Δt the advancing time step. During streaming (spatial evolution), collision updates are transferred to adjacent sites.

The connection between (LBM mesoscale simulation) and macroscale (observable properties) is established by means of the equilibrium distribution function f^{eq} together with the relaxation parameter $\omega =$ $\Delta t_{\rm relax} / \Delta t$. The former dictates the transport phenomenon (i.e. momentum, heat or mass transfer) while the later sets the related transport coefficient (i.e. kinematic viscosity, thermal diffusivity or mass diffusivity).

3. NUMERICAL METHOD

3.1. BAC Model Equations in Lattice-Based Dimensionless Form

The BAC model described in section 2.1 was cast in dimensionless form as an attempt to deal with concurrent effects while aiming at fewer parameters. Being $\dot{r}_{\rm ref} \neq 0$ a reference value for adsorbate transfer rate defined ahead, dimensionless variables were based on LBM parameters Δz and Δt as well as on BAC parameters $c_{\rm in}$ and $q_{\rm max}$ as follows:

$$Z = \frac{z}{\Delta z} , \quad \tau = \frac{t}{\Delta t} , \quad C = \frac{c}{c_{\rm in}} ,$$

$$Q = \frac{q}{q_{\rm max}} , \quad \dot{R} = \frac{\dot{r}}{\dot{r}_{\rm ref}}$$
(8)

Accordingly, lattice-based dimensionless forms of Equations (1) and (2) resulted as:

$$\frac{\partial C}{\partial \tau} + Ma \frac{\partial C}{\partial Z} = \frac{1}{Pe_m} \frac{\partial^2 C}{\partial Z^2} - \frac{1 - \varepsilon}{\varepsilon} P_{max} \dot{R}$$
(9)

$$\dot{R} = \frac{\partial Q}{\partial \tau} = P_{ads} C (1 - Q) - P_{des} Q$$
(10)

Lattice-based Mach number (Ma) and mass-transfer Peclet number (Pe_m) were defined as:

Ma =
$$\frac{v \Delta t}{\Delta z}$$
 and Pe_m = $\frac{(\Delta z)^2}{D \Delta t}$ (11)

By identifying $\dot{R} = \partial Q / \partial \tau$ as claimed in Equation (10), reference value for adsorbate transfer rate

resulted as $\dot{r}_{ref} = q_{max}/\Delta t$ so that the remaining dimensionless parameters became:

$$P_{\max} = \frac{\dot{r}_{ref} \Delta t}{c_{in}} = \frac{q_{\max}}{c_{in}} , \quad P_{ads} = k_{ads} c_{in} \Delta t , \quad P_{des} = k_{des} \Delta t$$
(12)

By introducing $N_z = L/\Delta z$ so that $N_z + 1$ is the number of lattice sites in *z* axis including end points, initial and boundary conditions, Equations (3) and (4), respectively became:

$$C(Z,0) = 0$$
 and $Q(Z,0) = 0$, for $0 \le Z \le N_z$ (13)

$$C(0,\tau) = 1 \text{ and } \left. \frac{\partial C}{\partial Z} \right|_{Z=N_{\tau}} = 0 \text{ , for } \tau > 0 \tag{14}$$

3.2. LBM Implementation for BAC Simulation

Based on BGK approach, LBM was programmed in Fortran 90/95 to simulate BAC ruled by Equations (9) to (14), with code lines following [40]. As the proposed model deals with adsorbate concentrations in fluid and solid phases, two particle distribution functions were required. Functions $f_k = f_k(Z,\tau)$ were assigned to dimensionless fluid-phase concentration $C = C(Z,\tau)$ whereas functions $s_k = s_k(Z,\tau)$ referred to solid-phase counterpart $Q = Q(Z,\tau)$.

At any dimensionless instant τ and position Z within the chromatographic column, dimensionless adsorbate concentrations were retrieved as:

$$C(Z,\tau) = f_1(Z,\tau) + f_2(Z,\tau)$$
 and $Q(Z,\tau) = s_1(Z,\tau) + s_2(Z,\tau)$

(15)

where k = 1 and k = 2 refer to forward and backward streaming directions, respectively. LBM was implemented in D1Q2 lattice so that functions f_0 and s_0 were disregarded.

The underlying physics at each phase dictates the equilibrium distribution functions f_k^{eq} and s_k^{eq} as well as the relaxation factors ω_f and ω_s . Provided that the solid matrix remains stationary while diffusive-convective transport takes place in the fluid phase, the equilibrium distribution functions were adopted as [40]:

$$f_k^{\text{eq}}(Z,\tau) = w_k C(Z,\tau) [1 \pm \text{Ma}] \text{ and } s_k^{\text{eq}}(Z,\tau) = w_k Q(Z,\tau)$$
(16)

Fulfilling the condition $\sum w_k = 1$, weighting factors w_k are the same for f_k^{eq} and s_k^{eq} , namely $w_0 = 0$ (i.e. f_0 and s_0 are disregarded) and $w_1 = w_2 = 1/2$ for D1Q2 lattice. The sign before Mach number depends on streaming

direction, being +Ma for forward (k = 1) and -Ma for backward (k = 2) streaming.

For governing equations invoking diffusive mass transport, relaxation factor ω refers to diffusivity *D*. Consistent with Pe_m definition in Equation (11), the following expressions were applied [35]:

$$\omega_{\rm f} = \left(\frac{1}{{\rm Pe}_{\rm m}} + \frac{1}{2}\right)^{-1}$$
 and $\omega_{\rm s} = 2$ (17)

In LBM, eventual source or sink terms are introduced in the collision step [40]. In view of Equations (9) and (10), the following expressions were computationally implemented:

$$f_{k}(Z, \tau + \Delta \tau) = [1 - \omega_{f}] f_{k}(Z, \tau) + \omega_{f} f_{k}^{eq}(Z, \tau) - w_{k} \left(\frac{1 - \varepsilon}{\varepsilon} P_{max} \dot{R}\right) \Delta \tau$$
(18)

$$s_k(Z,\tau + \Delta \tau) = [1 - \omega_s] s_k(Z,\tau) + \omega_s s_k^{eq}(Z,\tau) + w_k \dot{R} \Delta \tau \quad (19)$$

where $\Delta \tau$ is the dimensionless time step and $\dot{R} = P_{ads} C (1-Q) - P_{des} Q$.

At this point, it is worth recalling the absence of partial derivatives with respect to coordinate Z in Equation (10). As far as the streaming step is concerned in such a case, one may implement LBM by either imposing periodic boundary conditions or simply suppressing the streaming step itself [50]. The later approach was adopted so that streaming was solely implemented for fluid-phase particle distribution functions as:

$$f_k(Z + \Delta Z_k, \tau + \Delta \tau) = f_k(Z, \tau + \Delta \tau)$$
(21)

being ΔZ_k the dimensionless separation distances between lattice sites. Yet, systems modeled by zeroorder equations with respect to space may still "perceive" external influence through source and/or sink terms [35].

With C(Z,0) and Q(Z,0) provided by Equation (13), initial conditions for particle distribution functions were imposed as:

$$f_k(Z,0) = w_k C(Z,0)$$
 and $s_k(Z,0) = w_k Q(Z,0)$ (21)

At inlet (Z = 0), $f_2(0,\tau)$ was obtained *via* streaming from the adjacent site at Z = 1 so that $f_1(0,\tau)$ was the only unknown. As Equation (14) provides $C(0,\tau) = 1$, flux conservation leads to the following condition [40]:

$$f_1(0,\tau) = C(0,\tau) - f_2(0,\tau) \implies f_1(0,\tau) = 1 - f_2(0,\tau)$$
 (22)

At outlet $(Z = N_z)$, $f_1(N_z, \tau)$ was obtained *via* streaming from the adjacent site at $Z = N_z - 1$ so that $f_2(N_z, \tau)$ was unknown. By approximating the null Neumann condition in Equation (14) by first-order finite-differences [40], the following condition was obtained:

$$f_2(N_z, \tau) = f_2(N_z - 1, \tau)$$
 (23)

4. RESULTS AND DISCUSSION

4.1. Trial LBM Simulations Against a Classic BAC Work

Implementation of the LBM simulator was tested against a classic work on lysozyme bioseparation [38], whose parameters concerning а given chromatographic column are shown in Table 1a together with LBM parameters Δz and Δt . The later were adopted so as to ensure low lattice-based Mach number [40] whereas the former were used in simulations performed in [41]. Bed porosity was set as ϵ = 0.5 so that Equations (1) and (2) could match model equations in [41] as $(1 - \varepsilon)/\varepsilon = 1$. For LBM simulations. Table 1b shows lattice-based dimensionless parameters obtained by means of Equations (11) and (12).

Table 1(a) Process Parameters Concerning LysozymeBioseparation [38] and LBM Parameters Δz and Δt , (b) Resulting Lattice-Based DimensionlessParameters for LBM Simulations

(a) BAC and LBM parameters	(b) Lattice-based dimensionless parameters
$L = 0.104 \text{ m}$ $c_{in} = 0.0071 \text{ mol/m}^3$ $q_{max} = 0.875 \text{ mol/m}^3$ $k_{ads} = 0.286 \text{ m}^3/(\text{mol} \cdot \text{s})$ $k_{des} = 0.0005 \text{ s}^{-1}$ $v = 0.000224 \text{ m/s}$ $\Delta z = 0.0001 \text{ m}$ $\Delta t = 0.05 \text{ s}$	$N_{z} = 1040$ $P_{max} = 123.24$ $P_{ads} = 1.0153 \times 10^{-4}$ $P_{des} = 2.5 \times 10^{-5}$ Ma = 0.112 $\Delta Z = 1$ $\Delta \tau = 1$

For comprehensiveness towards large-scale BAC, axial diffusion was considered in the fluid phase. Functionality of the LBM simulator was tested for distinct lattice-based mass-transfer Peclet numbers, namely $Pe_m = 0.15, 0.30, 1.00$ and ∞ , the latter being implemented by setting $\omega_f = 2$ in agreement with Equation (17). For each testing Pe_m , Figure **1a** compares experimental data [38] with breakthrough

curves $C_{\text{exit}}(\tau) = C(N_z, \tau)$ simulated using parameters from Table **1**.

Despite breakthrough curves are shifted to right with respect to time, their expected shape was reproduced. It is worth noting that curve slope reduces as Pe_m decreases, i.e. as mass diffusion becomes more influential. This is because diffusion "spreads" species to both forward and backward directions in 1-D transport. Hence, saturation front becomes smoother than if transported solely by convection.

Additional simulations were performed in [41] by adopting a slightly lower maximum adsorption capacity, namely $q'_{max} = 0.845 \text{ mol/m}^3$, for which Equation (12) provides $P'_{max} = 119.01$. For each trial Pe_m as in prior LBM simulations, Figure **1b** shows that numerically simulated breakthrough curves are closer to experimental data, i.e. saturation arrives earlier at the column exit as expected. Moreover, as parameters Pe_m were kept the same, the slope of each corresponding breakthrough curve remains apparently unchanged.

With regard to the original q_{max} value, two lower maximum adsorption capacities were tested, namely $q''_{\text{max}} = 0.95 q_{\text{max}}$ and $q'''_{\text{max}} = 0.90 q_{\text{max}}$, respectively rendering $P''_{\text{max}} = 117.08$ and $P'''_{\text{max}} = 110.92$. By keeping Pe_m = 0.30 (as this value yields a slope close to experimental data), Figure **1c** shows the corresponding simulated breakthrough curves, where good match is observed for $P''_{\text{max}} = 117.08$.

It is worth mentioning that some simulations were carried out in [41] by disregarding diffusive transport in fluid phase, which is mathematically equivalent to assume D = 0 in Equation (1) or $Pe_m \rightarrow \infty$ in Equation (9). In this case, fluid-phase concentration becomes ruled by a first-order PDE with respect to spatial dependence and adsorbate transport in fluid phase becomes convection-dominant. In doing so, one may claim the advantage of being able to apply marching numerical methods for initial-value problems (e.g., Runge-Kutta method) while dismissing the additional boundary condition at column exit.

Nonetheless, discretisation schemes of convective terms (e.g. upwind scheme) might yield numerical dispersion [51], also referred to as false diffusion, which is not the case of LBM [25, 40]. In FDM, for instance, numerical dispersion due to upwind schemes becomes evident for convection prevailing over diffusion [52], i.e. for higher Pe_m . For that reason, slope of breakthrough curves simulated in [41] with parameters from Table **1a** could be "artificially

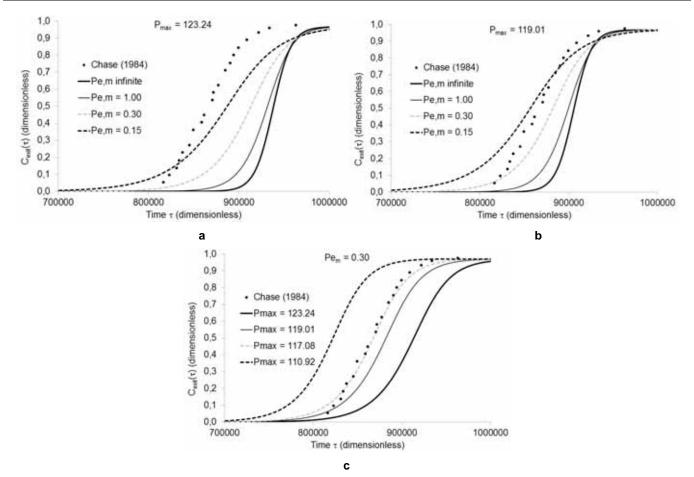


Figure 1: Comparison between experimental data [38] and breakthrough curves simulated with lattice-based dimensionless parameters as assessed from (**a**) model parameters in [41] and distinct mass-transfer Peclet numbers Pe_m , (**b**) lower maximum adsorption capacity as suggested in [41] and distinct mass-transfer Peclet numbers Pe_m , (**c**) trial higher values of maximum adsorption capacity and $Pe_m = 0.30$.

smoother", which may suggest a re-evaluation of some (if not all) parameters. As depicted in Figures **1a** and **1b**, curves simulated *via* LBM with $Pe_m \rightarrow \infty$, i.e. which mimic numerical simulations in [41], seem steeper than experimental data.

4.2. LBM Simulation of BAC for Lipase Bioseparation

Thanks to industrial and medical applications, there has been a growing interest in microbial lipases, whose high-degree purification can be achieved through BAC [53]. After trial tests as addressed in section 4.1, the LBM simulator was applied to lipase bioseparation. Along with LBM parameters Δz and Δt , Table **2a** shows parameters concerning a given column studied in [39]. Adsorbate concentrations are here expressed in enzymatic activity units defined as 1 U = 1 µmol of fatty acids released per minute. However, lattice-based dimensionless parameters are insensitive to units and Table **2b** shows their values as assessed by Equations (11) and (12). Bed porosity was initially set at $\varepsilon = 0.5$.

Table 2: (a) Process Parameters Concerning Lipase
Bioseparation [39] and LBM Parameters Δz and
 Δt , (b) Resulting Lattice-Based Dimensionless
Parameters for LBM Simulations

(a) BAC and LBM parameters	(b) Lattice-based dimensionless parameters
$L = 0.032 \text{ m}$ $c_{in} = 9.52 \times 10^{6} \text{ U/m}^{3}$ $q_{max} = 9.89 \times 10^{7} \text{ U/m}^{3}$ $k_{ads} = 1.00 \times 10^{-9} \text{ m}^{3}/(\text{U} \cdot \text{s})$ $k_{des} = 0.00233 \text{ s}^{-1}$ $v = 0.00025 \text{ m/s}$ $D = 3.06 \times 10^{-8} \text{ m}^{2}/\text{s}$ $\Delta z = 0.0001 \text{ m}$ $\Delta t = 0.05 \text{ s}$	$N_z = 320$ $P_{max} = 10.389$ $P_{ads} = 4.76 \times 10^{-4}$ $P_{des} = 1.17 \times 10^{-4}$ Ma = 0.125 Pe _m = 6.536 $\Delta Z = 1$ $\Delta \tau = 1$

Recalling that values in Table **2a** are tentative, LBM simulations were also performed for different values of dimensionless parameters P_{max} , Pe_m and ε (i.e. variations in adsorption-desorption parameters P_{ads} and

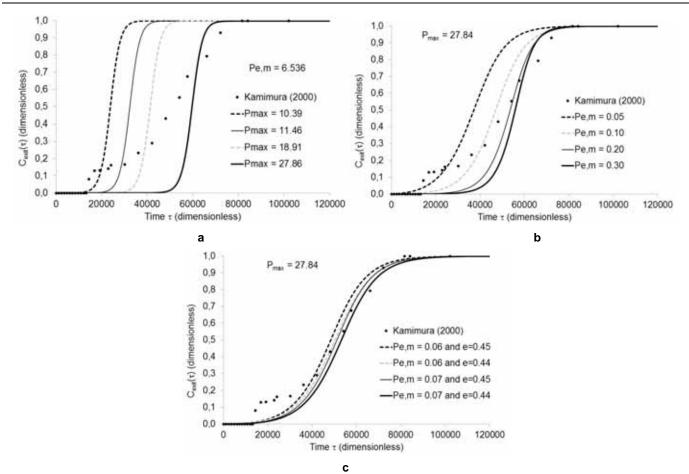


Figure 2: Comparison between experimental data [39] and breakthrough curves simulated with lattice-based dimensionless parameters as assessed from (**a**) distinct maximum adsorption capacities as suggested in [39] and mass-transfer Peclet number $Pe_m = 6.536$, (**b**) the highest maximum adsorption capacity in [39] and distinct Pe_m , (**c**) the highest maximum adsorption capacity and combinations of $Pe_m = 0.06$ or 0.07 with trial lower porosities $\varepsilon = 0.44$ or 0.45.

 P_{des} were not tested). In Figure **2a**, experimental data [39] are compared with breakthrough curves as simulated with parameters from Table **2b** as well as exploratory higher $P_{max} = 11.46$, 18.91 and 27.86. Such values are related *via* Equation (12) to higher q_{max} equally tested in [39]. One notes that saturation arrives at proper time at column exit but slopes need some adjustment.

Accordingly, Figure **2b** shows breakthrough curves as simulated with tentative Pe_m , namely 0.05, 0.10, 0.20 and 0.30, while keeping $P_{max} = 27.86$. As expected and as pointed in section 4.1, curve slopes are reduced inasmuch as Pe_m decreases. Yet, numerically simulated curves remain distant from experimental data, thus claiming for further tests.

Figure **2c** shows breakthrough curves simulated with $P_{max} = 27.86$ while combining either $Pe_m = 0.06$ or 0.07 with porosity $\varepsilon = 0.44$ or 0.45. Any combination of those values of diffusivity *D* (for Pe_m calculations) or bed porosity ε yields reasonable curves. It should be noted that experimental assessment of aforementioned parameters is not straightforward.

5. CONCLUSION AND FUTURE WORK

Computational modelling of biospecific affinity chromatography (BAC) requires the solution of coupled partial differential equations and lattice-Boltzmann method (LBM) comes forward as an interesting numerical technique. This work applied LBM following a dynamic one-dimensional BAC model cast in dimensionless form so that lattice-based Mach number (Ma) and mass-transfer Peclet number (Pe_m) arose as model parameters.

As the differential equation for adsorbate concentration in solid phase lacked partial derivatives with respect to spatial coordinate, the related streaming step was simply suppressed in the LBM code. No loss of functionality was observed as the LBM simulator was able to reproduce the expected shape of breakthrough curves. Influence of Pe_m (sometimes neglected in BAC models) and dimensionless parameter P_{max} related to maximum adsorption capacity of the column was examined by performing LBM simulations of a classic work on lysozyme bioseparation. While Pe_m influenced the slope of the breakthrough curve, P_{max} affected the time lag for column saturation. Differences between simulated breakthrough curves and experimental data were assigned to the absence of numerical diffusion in LBM. Those discrepancies were mitigated by reevaluating the original model parameters.

Numerical simulations of lipase bioseparation proved that bed porosity are influential in BAC models, apart from lattice-based dimensionless parameters Pe_m and P_{max} . In view of an accurate assessment of model parameters *via* best-fit against experimental data, the numerical simulator is currently being extended in order to combine the LBM solution procedure with optimisation routines (to minimise the sum of squared differences between numerical and experimental breakthrough curves).

At its current development stage the LBM simulator is limited to 1-D modelling and extension to 2-D models implies in simulating the downstream solution as well. Accordingly, upcoming versions of the LBM simulator will cope with bed hydrodynamics by the use of D2Q9 lattices not only for flow simulation but also mass transfer as already accounted for.

The LBM simulator will be equally extended towards multi-component bioseparation. As pointed in [23], LBM may tackle multiphase flows by relying on a set of multicomponent distribution functions f_{ki} with index *i* running over as many species as necessary, e.g. other enzymes or protein products. In view of that, corresponding adsorption-desorption kinetics should be properly modelled and numerically implemented *via* source or sink terms R_{ki} related to the variation rate of species *i* for each lattice link *k*.

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APPENDIX: NOMENCLATURE

Latin Symbols

A = cross-sectional area of the chromatographic column, m²

- C = dimensionless adsorbate concentration in fluid phase
 - adsorbate concentration in fluid phase, mol·m⁻³
- D = adsorbate axial diffusivity, m²·s⁻¹

С

- *f* = particle distribution function related to fluidphase concentration, dimensionless
- *k* = index for lattice (streaming) link, dimensionless
- k_{ads} = adsorption kinetic constant, m³·mol⁻¹·s⁻¹
- k_{des} = desorption kinetic constant, s⁻¹
- *L* = chromatographic column length, m
- Ma = lattice-based Mach number, dimensionless
- N_z = index of the last lattice site, dimensionless
- Pe_m = lattice-based mass-transfer Peclet number, dimensionless
- P_{ads} = adsorption-related parameter, dimensionless
- P_{des} = desorption-related parameter, dimensionless
- *P*_{max} = inlet concentration and maximum adsorption capacity parameter, dimensionless
- Q = dimensionless adsorbate concentration in solid phase
- q = adsorbate concentration in solid phase, mol·m⁻³
- \dot{R} = dimensionless adsorbate transfer rate from fluid to solid phase
 - adsorbate transfer rate from fluid to solid phase, mol·m⁻³·s⁻¹
 - particle position, m
 - particle distribution function related to solidphase concentration, dimensionless
 - = time, s

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- \vec{u} = particle velocity, m·s⁻¹
 - volumetric flow rate of the percolating solution, m³·s⁻¹

v	= interstitial velocity of the percolating solution,	FEM = finite elements method
	m⋅s ⁻¹	FVM = finite volumes method
W	 weighting factors, dimensionless 	LBM = lattice Boltzmann method
Ζ	= dimensionless axial coordinate	MD = molecular dynamics
Z	= axial coordinate, m	PDE = partial differential equation
Greel	k Symbols	1-D = one dimensional
ε	= bed porosity, dimensionless	REFERENCES
τ	= dimensionless time	
Ω	= collision operator, s^{-1}	 Souza-Santos ML. Solid Fuels Combustion and Gasification: Modeling, Simulation, and Equipment Operation. 2nd ed. New York: CRC Press; 2010. Print ISBN 978-1-4200-4749-3, eBook ISBN: 978-1-4200-4750-9.
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