## https://doi.org/10.51152/jbarbiomed.v7i1.213



# *Original Article* Fabrication and Release Kinetics of Liposomes Containing Leuprolide Acetate

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Received: 20-7-2021 Revised: 14-8-2021 Published: 28-8-2021

Keywords: Liposomes, Leuprolide acetate, Cholesterol, Lecithin, Drug Release Kinetics, Prostate Cancer, Endometriosis. **Abstract**: The present work is focused on the preparation and *in vitro* release kinetics of liposomal formulation of Leuprolide Acetate. In this work, "Thin Lipid Film Hydration Method" was used for preparation of Leuprolide Acetate loaded liposomes. Prepared liposomal formulations of Leuprolide acetate was evaluated by drug entrapment study, *in-vitro* drug release kinetics and stability studies. The percentage drug entrapment of Leuprolide acetate for F1 and F2 formulations were found to be 78.14  $\pm$  0.67 and 66.70  $\pm$  0.81% respectively. *In-vitro* drug release study of liposomal formulations had shown zero order release pattern. Regression co-efficient (R<sup>2</sup>) value of Zero order kinetics for F1 and F2 formulations were 0.9912 and 0.9676 respectively. After storing formulations for 1 month, stability testing was done at 4°C.It was found that all batches were stable. These liposomal formulations of Leuprolide acetate can be formulated for parenteral application to treat prostate cancer and in women, to treat symptoms of endometriosis (overgrowth of uterine lining outside of the uterus) or uterine fibroids.

Cite this article as: Das, S., Samanta, S., Bankura, K., Roy, R. and Nayak, A.K. (2021). Fabrication and Release Kinetics of Liposomes Containing Leuprolide Acetate. Journal of basic and applied Research in Biomedicine, 7(1): 35-38 https://doi.org/10.51152/jbarbiomed.v7i1.213 This work is licensed under a Creative Commons Attribution 4.0 License. You are free to copy, distribute and perform the work. You must attribute the work in the manner specified by the author or licensor.

### INTRODUCTION

Liposomes are nowadays very effective and useful delivery system for transmission of drug, enzymes, vaccine etc, in human or animal body (Senthilkumar et al., 2021; Bhattacharjee et al., 2019; Budai et al., 2007). Researches are taking place to develop vesicular formulations of different drugs (Das et al., 2017; Malakar et al., 2012; Malakar et al., 2014). Liposomes are comprised of at least one phospholipid layer which has an inner aqueous compartment (Bhattacharjee et al., 2020). Liposome structure consists of polar head group (hydrophilic) and nonpolar tail groups (lipophilic). The lipophilic group is long hydrocarbon chain (Singha Roy et al., 2018). Liposomes are eliminated by simple degradation without creating any toxic effect and also provide significant advances as compared to other Nano Drug Delivery Systems (Moghimipour et al., 2015; Rathod et al., 2010). Both kinds of drugs like hydrophilic and lipophilic can be entrapped within liposomes for treating different ranges of diseases (Bozzutoet et al., 2015; Xing et al., 2016; Lee et al., 2017; Pal et al., 2010). Drug entrapment takes place within aqueous or lipid bilayer (Amin et al., 2018). Cancer is still a complex and difficult disease to treat (Das et al., 2020; Panigrahi et al., 2020). According to WHO, cancer is the second leading death causing disease globally (Quagliarini et al., 2020). In recent years, prostate cancer has become most common type worldwide (Bray et al., 2018; Pikala et al., 2019). According to study in the year 2018, 1.3 Million cases of prostate cancer reports were detected from which 3, 59,000 deaths were observed (Das et al., 2020). This study is focused on formulation and evaluation of liposomal formulation of Leuprolide Acetate for parenteral administration.

Leuprolide Acetate, a nonapeptide, is a strong agonist for receptor that releases gonadotropin hormone. It is applied in the treatment of prostate cancer as well as other diseases like Alzheimer's disease, polycystic ovary syndrome, etc (Wilson *et al.*, 2007). Leuprolide Acetate is mostly preferred for androgen deprivation therapy (ADT) and has been on market for more than 30 years (Hoda *et al.*, 2017). Suppression of testosterone levels in serum is worthwhile for most of the instances (Berges *et al.*, 2006). Leuprolide Acetate also used in women to treat symptoms of endometriosis (overgrowth of uterine lining outside of the uterus) or uterine fibroids.

In the current research work, formulation of liposomes of Leuprolide acetate was prepared by using soya lecithin (L- $\alpha$ -Phosphatidylcholine) and cholesterol. Lecithin can be prepared

from egg yolk and various oils seeds, such as flax, cotton seed, corn germ, sunflower seed, rapeseed and soybeans (Mourad *et al.*, 2010; Ghosh *et al.*, 2020). Hydration of Phospholipids in aqueous solutions is important to form the closed liposome structure (Amin *et al.*, 2018). Another excipient cholesterol increases rigidity of phospholipid bilayer by increasing the viscosity of lipid bilayer. It can also reduce the permeability of membrane and stabilizes the lipid bilayer by improving the rigidity of liposomes (Lee *et al.*, 2017; Pal *et al.*, 2010).

## MATERIALS AND METHODS:

#### Materials:

Leuprolide Acetate was received as a gift from Sun Pharma, India. L- $\alpha$ -Phosphatidylcholine (soy-lecithin) and cholesterol, these two excipients were procured form Yarrow Chem, India. From Loba Chemie Pvt. Ltd. Chloroform was procured.

#### Preparation of Liposomes containing Leuprolide Acetate by "Thin Film Hydration Method:

Thin Lipid Film Hydration Method was employed for preparing liposomal formulation containing Leuprolide Acetate. For preparing formulation of F1 and F2, drug (Leuprolide Acetate) and lipid were used at 1:5 and 1:7molar ratios respectively. Mixtures containing Leuprolide Acetate, specific molar ratio of Cholesterol-Lecithin were dissolved in 5 ml of chloroform with sonication (Sun et al., 2018). The freshly prepared organic solution was transferred into a round bottom flask and rotated on water bath at approximately 40°C, until the mixture had been completely dried till the smell of chloroform disappeared. Depending on the lecithin concentration, drying time may vary. It took almost 40 minutes for completion of drying. Phosphate buffer of pH 6.8 was incorporated in the round bottom flask followed by vigorous shaking formed a yellowish-white suspension. Throughout the process temperature was maintained at 38ºC.The suspension was sonicated by using a Sonicator for 1 hour. The suspension was cooled to room temperature and permitted to stand overnight (Das et al., 2021). On the next day, suspension was taken out for centrifugation process to separate the liposome precipitates. Liposomes were dried under rotary vacuum dryer and examined.

### Determination of drug content:

In a centrifuge tube 5 ml of liposomal suspension of Leuprolide Acetate was taken and centrifuged (Remi Motors, India) at 3500 RPM for 1 hour (Ghosh *et al.*, 2020). After completion of centrifugation process, the supernatant solution (A) was

separated and the pellets were dissolved in 3 ml of ethanol by sonicating for 15 minute. This process was required for dissolving excess lecithin. Phosphate buffer pH 5.8, 7 ml, was added to the centrifuge tube followed by centrifugation at 4000 RPM for 1 hour. The clear supernatant solution (B) was taken out and measured by UV-Spectrophometer (ELICO SA165) at 283 nm by using phosphate buffer pH 5.8 as a blank solution. Dilution was done for solution A with phosphate buffer (pH 5.8) and absorbance was measured by UV-Vis Spectrophotometer (ELICO SA165) at 283 nm. (Das *et al.*, 2020) The total amount of drug present in the liposomal suspension (mg/ml) is the sum of amount of the drug present in solution A and amount of drug entrapped in liposome structure.

## Percentage of drug entrapment efficiency and Particle size

Percentage of drug entrapment was determined by using the ratio of entrapped drug amount (mg) in formulation and the total amount of drug (mg) used in the formulation, which may be expressed by the following formula.

Entrapment efficiency (%) <u>Amountofdrugentrapmentinilposome</u> <u>Tot alamountofdruginnitiallytaken</u> × 100 %

### Determination of particle size

Particle sizes were measured by optical microscopy method. One drop of the prepared liposome was homogeneously spread onto a glass slide. The samples were observed under an optical microscope of 100X magnification fitted with an eye-piece micrometer.

### In vitro drug release study:

Liposomal suspension of Leuprolide Acetate was kept for overnight in refrigerator at 4°C. The liposomal suspension (0.5 ml) was taken in a dialysis membrane (Himedia Laboratories Pvt. Ltd.) that separated the liposomal suspension from the receptor medium. The temperature during this procedure was maintained at  $37\pm0.5^{\circ}$ C. Drug release study was continued for 3 hours. Samples were withdrawn from the receptor chamber at every 15 minutes intervals followed by measuring the absorbance for each samples at 283 nm against receptor medium taken as blank.

#### In vitro Drug Release Kinetics:

Various kinetic models were applied to analyze *in vitro* drug release kinetics (Sunitha *et al.*, 2014; Haghiralsadat *et al.*, 2018; Das *et al.*, 2013; Hasnain *et al.*, 2020). In case of zero order kinetics,  $F = K_0 t$ , equation represents F is the fraction of drug release in time t, where  $K_0$  is the zero order rate constant.

Formula for second order release kinetics,  $ln(1 - F) = -K_1t$ , where F represents the fraction of drug released in time t and  $K_1$  denoted as first order rate constant.

According to Higuchi model, the equation,  $F = K_H t^{1/2}$ , here F represents the fraction of drug released in time t and  $K_H$  denoted as the Higuchi dissolution constant.

In case of Hixon-Crowell model, equation is  $Q^{1/3} = Kt + Q_0^{1/3}$ . Where, Q is the drug release amount in time t and  $Q_0$  is the starting value of Q and K represents rate constant.

Another equation represents Korsmeyer-Peppas model is  $F=K_pt^n.$  Here F denotes the fraction of drug released in time t, Korsmeyer-Peppas rate constant is  $K_p$  and n is the release exponent.

#### Stability Testing:

In Eppendorf's tube the liposomal formulation F1of Leuprolide Acetate was taken and placed at  $4^{\circ}$ C in the refrigerator for 1 month. (Roy *et al.*, 2019) After 1 month, the liposomal formulation was analyzed by drug entrapment study, *in-vitro* drug release and colour inspection.

#### Statistical Analysis:

All the data obtained were expressed as mean  $\pm$  standard deviation (S.D). Each observation was done in triplicate (n= 3). MedCalc software version 11.6.1.0 was used for simple statistical analysis.

### **RESULTS AND DISCUSSION:**

#### Drug Content

The drug contents of Leuprolide acetate liposomal formulations were determined, where the % of drug contents were within the range  $75.55\pm1.10$  to  $81.41\pm0.85$ . This result might be attributed to the fact that the viscosity increased with increasing ratio of lecithin to cholesterol used in the preparation.

### Drug Entrapment Study:

Drug entrapment for liposomal formulations of Leuprolide Acetate was determined. According to the study of drug entrapment, values for F1 and F2 were  $39.07 \pm 0.33$  and  $33.35 \pm 0.41$  mg/ml respectively. Drug entrapment was decreased with increasing the molar ratio of lecithin to cholesterol. But as a result of increasing molar ratio of lecithin to cholesterol, formulations were also become viscous.

Table 1: Results of Drug entrapment study for formulations of Leuprolide Acetate.									
Sl. No.	Formulation	Drug Entrapment	% Drug Entrapment						
	s	(mg/ml)	Efficiency						
1	F1	39.07±0.33	78.14±0.67						
2	E2	33 35+0 41	66 70+0 81						

Mean ± SD; n=3

#### Particle Size

The mean particle sizes (z-average) of liposomes in liposomal suspension of Leuprolide acetate were within the range  $4.95\pm1.51$  to  $47.85\pm2.31\mu$ m. The smaller mean particle size  $(4.95\pm1.51\mu$ m) was noticed in case of formulation F1. With the increment of lecithin amount used in the preparation of liposomal formulations of Leuprolide acetate, the mean particle sizes of liposomes were found to be increased, which might be attributed to the fact of viscosity increment with increasing ratio of lecithin to cholesterol amount used in the preparation.

#### In-vitro drug release:

Drug release immediately started after coming in contact with the dissolution medium. The result of *invitro* release study showed 3 hours of sustained release pattern. (Fig.1.) Completion of 3 hours study it was observed that F1 had released a maximum % release of  $51.89\pm0.78\%$  and F2 showed lowest % release at 180 minutes ( $41.56\pm1.5\%$ ).

#### In vitro drug release kinetics:

In vitro drug release study reports were plotted against various kinetics models (zero order, first order, Higuchi, Hixon Crowell and Korsmeyer-Peppas model equations) for comparison and to analyze the release kinetics (Guo *et al.*, 2016; Hasnain *et al.*, 2018). The results included in various kinetics models are presented in Table 3.Correlation coefficients of kinetic model equations were compared for liposomal formulation of Leuprolide Acetate. It was observed that all the two formulations were best fitted to zero order model ( $R^2$  Value of F1 and F2 was 0.9912 and 0.9676 respectively) over 3 hours. The data were shown in the Table 2.

Table 2: Results of curve fitting of in vitro Leuprolide acetate releasing from the liposomal formulations.

Kinetic Model	Regression Co-efficient			
Kinetic Woder	F1	F2		
Zero order model	0.9912	0.9676		
First order model	0.9703	0.9451		
Higuchi model	0.9416	0.8950		
Hixon-Crowell model	0.9792	0.9536		
Korsmeyer-peppas model	0.7741	0.7242		

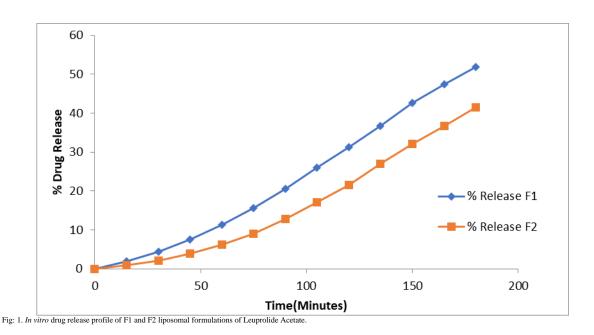


Table 3. Results of stability testing

Table 5: Results of stability testing.									
Time B	Batch	Colour	Drug content	Percentage of drug entrapment	Percentage of drug	Changes in physical			
	Datell			Efficiency	released in 3hrs.	properties			
1 Month	F1	Yellowish white	38.74±0.33	77.48±0.53	50.85±1.13	No Changes			
Mean $\pm$ SD: n=3									

#### Stability:

Formulation F1 prepared with 1:5 of drug-lipid molar ratio was selected for stability study because of better drug entrapment and drug release pattern and stored for 1 month at  $4^{\circ}$ C in refrigerator then further analyzed for colour changes, drug entrapment study and *in vitro* drug release study. The colour of the suspension was same as that at the starting, yellowish white. There was no alteration observed for drug entrapment study. *In vitro* release study for 3 hours duration was performed and the percentage release at 180 minutes was 50.85 %. So, there was no significant change occurred during 1 month storage of liposomal formulations of Leuprolide Acetate. Therefore, results revealed that liposomal formulations of Leuprolide Acetate were physically stable. The data shown in the table 3.

#### **CONCLUSION:**

In this study, liposomal formulations of Leuprolide Acetate were prepared by thin lipid film hydration method by using lecithin  $(L-\alpha$ -Phosphatidylcholine) and cholesterol. The percentage drug entrapment of Leuprolide Acetate for F1 and F2 were 78.14  $\pm$ 0.67 and 66.70  $\pm$  0.81 % respectively. The highest percentages of drug release (at 180 minutes) achieved by F1 was  $51.89 \pm 0.78$ %. It was detected that high lecithin: cholesterol ratio makes suspension more viscous so percentage drug entrapment and percentage drug release were decreased. During drug release study, sustained release patterns were observed till 3 hours. Liposomal formulations of Leuprolide Acetate best fitted to zero order release kinetics where the R<sup>2</sup> Value of F1 and F2 were 0.9912 and 0.9676 over 3 hours respectively. After one month stability study at 4°C of F1 batch was found stable enough. However, long term stability study is required for the future development of these liposomal formulations.

#### **DECLARATION OF COMPETING INTEREST:** No conflict of interest declared by authors.

#### **ACKNOWLEDGEMENT:**

The authors are thankful for the authority of Netaji Subhas Chandra Bose Institute of Pharmacy, Chakdah-741222, Nadia, West Bengal, India, for facilitating necessary provisions to perform the present study.

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