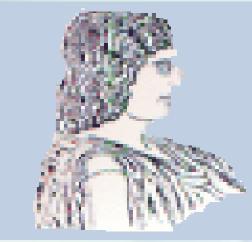
# POTENT AND SELECTIVE INHIBITION OF HUMAN Ca<sup>2+</sup>-INDEPENDENT PHOSPHOLIPASE A<sub>2</sub> BY FLUOROKETONES



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# **Introduction**

The superfamily of phospholipases  $A_2(PLA_2)$  consists of hydrolytic enzymes that act upon the *sn-2* ester bond of phospholipids, releasing free fatty acids and lysophospholipids. The main representative of these fatty acids is arachidonic acid, which can be transformed into eicosanoids (prostaglandins, leukotriens, etc) by the action of other enzymes. Lysophospholipids are precursors for other bioactive compounds, such as platelet-activating factor (PAF). PAF and eicosanoids constitute basic mediators of inflammation. The three predominant groups of phospholipase A<sub>2</sub> found in human tissues are the cytosolic PLA<sub>2</sub> (cPLA<sub>2</sub>), the calcium-independent PLA<sub>2</sub> (iPLA<sub>2</sub>) and the secreted PLA<sub>2</sub> (sPLA<sub>2</sub>). iPLA<sub>2</sub> has been proposed as a homeostatic enzyme involved in basal metabolism within the cell, but has also been found to be involved in diseases of the brain, the heart and the central nervous system, which makes this enzyme an attractive drug target. We have recently demonstrated that  $Ca^{2+}$ -independent phospholipase  $A_2$  (GVIA iPLA<sub>2</sub>) plays a key-role in experimental autoimmune encephalomyelitis and that GVIA iPLA<sub>2</sub> is a novel target for the development of new therapies for multiple sclerosis.<sup>1</sup> A series of fluoroketones has been presented as iPLA<sub>2</sub> inhibitors and the structureactivity relationship has been evaluated.<sup>2</sup> To extend this research, we synthesized a variety of polyfluoroketones containing an aromatic ring and a four carbon atom chain between the ring and the polyfluoroketone group.

#### **In vitro results**

All synthesized compounds were tested for inhibition on human GVIA iPLA<sub>2</sub>, GIVA cPLA<sub>2</sub> and GV sPLA<sub>2</sub>. (Table 1) The percentage of inhibition for each enzyme was determined at 0.091 mol fraction, and  $X_{I}(50)$  values were measured for inhibitors which showed more than 93% inhibition for an enzyme. When compared to the known inhibitor FKGK18, changing the position of the carbon chain on the naphthalene ring, leads to less potency, but higher selectivity. Increasing the number of fluorine atoms, **GK172** and **GK173** become less potent inhibitors of iPLA<sub>2</sub>. When a methoxy group is added to the naphthalene ring, inhibitors GK213 and GK214 lose selectivity towards iPLA<sub>2</sub>. The biphenyl group instead of the naphthalene shows a similar inhibition for both iPLA<sub>2</sub> and cPLA<sub>2</sub> (GK174, GK175). The insertion of a trifluoromethyl group on the aromatic ring presented excellent potency (GK178 and GK189), but poor selectivity. In the case of GK176 and GK188 the fluorine on the aromatic ring shows both great potency and selectivity towards iPLA<sub>2</sub>. When the fluorine atom is replaced by a methoxy group, compounds GK177 and GK187 are currently the most potent and selective inhibitors for iPLA<sub>2</sub>. Inhibitors GK176, GK177, GK187 and GK188 can be used to study the role of GVIA iPLA<sub>2</sub> in neurological disorders.

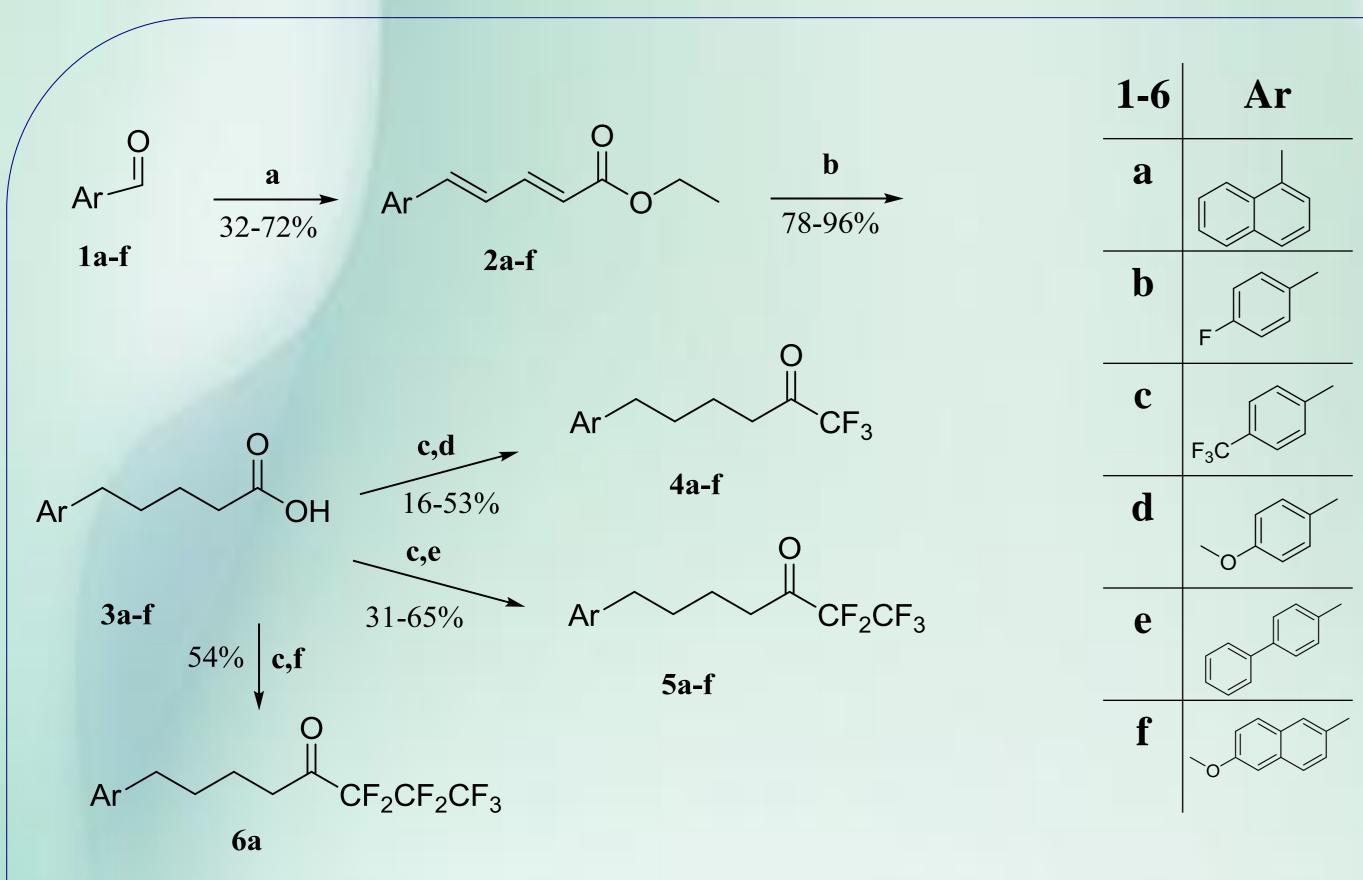
#### **Synthesis**

Commercially available aromatic aldehydes **1a-f** were used as starting materials, to prepare the desired fluoroketones (**4a-f**, **5a-f**, **6a**). Each aldehyde underwent a Horner – Wadsworth – Emmons reaction with triethyl-4-phosphonocrotonate in the presence of LiOH to produce the corresponding unsaturated ester **2a-f**. After catalytic hydrogenation and saponification with 1 N NaOH in ethanol we acquired the corresponding carboxylic acids **3a-f** which were converted to acyl chlorides by the oxalyl chloride/DMF method. In situ, acyl chlorides were treated with pyridine and trifluoroacetic anhydride or pentafluoropropionic anhydride or heptafluorobutanoic anhydride to provide trifluoromethyl ketones **4a-f**, pentafluoroethyl ketones **5a-f** and the heptafluoropropyl ketone **6a**, respectively (**Scheme 1**).

# **Table 1:** Inhibition of PLA<sub>2</sub> by fluoroketones

$\frac{9}{10}$ $X_{f}(50)$ $\frac{9}{10}$ $X_{f}(50)$	No	Structure	iPLA <sub>2</sub>		cPLA <sub>2</sub>		sPLA <sub>2</sub>
Image: Constraint of Constr				<i>X</i> <sub>I</sub> (50)		<i>X</i> <sub>I</sub> (50)	% inhibition
Image: Constraint of the sector of the se	FKGK11	O $C_2F_5$	99.4		N.D.		28
<b>Aabb</b>	FKGK18	CF3	99.9		93.1		36.8
GK172, Ga $f = 1$ $0.14$ $0.0002$ $04.3$ $27.4$ GK173, Ga $f = 1$ $0.002$ $0.0002$ $04.3$ $04.3$ $27.4$ GK173, Ga $f = 1$ $0.002$ $0.0002$ $00002$ $00002$ $00002$ $00002$ GK174, 4e $f = 1$ $f = 1$ $0.002$ $0.0189 \pm 0.0002$ $96.5$ $0.0074 \pm 0.0074$ $40.5$ GK175, 5e $f = 1$ $f = 1$ $0.00134 \pm 0.0003$ $72.6$ $0.0003$ $38.3$ GK175, 4b $f = 1$ $f = 1$ $98.5$ $0.0002 \pm 0.0001 \pm 0.0003$ $41.4$ N.D.GK177, 		CF <sub>3</sub>	98.7		77.4		31.9
6a10.1710.171000400040004GK174, 4e $ff(r) = 10^{-1}$ 97.9 $0.0189 \pm 0.0045$ 96.5 $0.0074 \pm 0.0003$ 40.5GK175, 5e $ff(r) = 10^{-1}$ 94.6 $0.0134 \pm 0.0017$ 72.638.3GK176, 4b $ff(r) = 10^{-1}$ 98.5 $0.0002 \pm 0.0002 \pm 0.0002 \pm 0.00002 \pm 0.00002 \pm 0.00001 \pm 0.00000$ 41.4N.D.GK177, 4d $ff(r) = 10^{-1}$ 99.8 $0.0001 \pm 0.0000 \pm 0.00002 \pm 0.00002 \pm 0.00000 \pm 0.00000$ 83.930.4GK178, 4c $ff(r) = 10^{-1}$ 99.8 $0.0001 \pm 0.00001 \pm 0.00002 \pm 0.00000 \pm 0.000000 \pm 0.000000 \pm 0.00000000$			96.4		64.0		29.4
4e $f = 0.0045$ $f = 0.0003$ GK175, 5e $f = 0.003$ $0.0013 \pm 0.0003$ $72.6$ $38.3$ GK176, 4b $f = 0.0005$ $98.5$ $0.0002 \pm 0.0002 \pm 0.0000$ $41.4$ N.D.GK177, 4d $f = 0.0005$ $99.8$ $0.0001 \pm 0.0000$ $54.3$ N.D.GK178, 4c $f = 0.0005$ $99.9$ $0.0002 \pm 0.0002 \pm 0.0000$ $83.9$ $30.4$ GK187, 5d $f = 0.0005$ $99.8$ $0.0001 \pm 0.0000$ $12.9$ $32.8$		C <sub>3</sub> F <sub>7</sub> O	76.7		60.4		57.8
GK175, 5e $f_{C_2F_6}$ 94.6 $0.0134 \pm 0.0017$ 72.638.3GK176, 4b $f_{C_2F_6}$ 98.5 $0.0002 \pm 0.0002 \pm 0.0000$ 41.4N.D.GK177, 4d $f_{C} \leftarrow f_{-3}$ 99.8 $0.0001 \pm 0.0000$ 54.3N.D.GK178, 		CF <sub>3</sub>	97.9		96.5	±	40.5
4b $f = f + f + f + f + f + f + f + f + f + $			94.6		72.6		38.3
4d $\int_{0}^{+} \int_{CF_{3}}^{+} CF_{3}$ 0.000010.0000GK178, 4c $\int_{F_{3}C}^{+} \int_{CF_{3}}^{+} CF_{3}$ 99.9 $0.0002 \pm 0.0002 \pm 0.0000$ 83.930.4GK187, 5d $\int_{0}^{+} \int_{C_{2}F_{5}}^{+} C_{2}F_{5}$ 99.8 $0.0001 \pm 0.0001 \pm 0.0000$ 12.932.8			98.5		41.4		N.D.
4c $GF_3$ 0.0000       0.0000       0.0000         GK187, $GC_2F_5$ 99.8       0.0001 ±       12.9       32.8         5d $GC_2F_5$ 99.8       0.0000       12.9       32.8		CF <sub>3</sub>	99.8		54.3		N.D.
5d C <sub>2</sub> F <sub>5</sub> 0.0000			99.9		83.9		30.4
			99.8		12.9		32.8

# Scheme 1: Synthesis of polyfluoroketones



**Reagents and conditions**: a)  $C_2H_5OOCCH=CHCH_2P(=O)(OC_2H_5)_2$ , LiOH, THF, 77 °C, b) i)  $H_2$ , 10% Pd/C, EtOH, ii) NaOH 1N, EtOH, c) (COCl)\_2, DMF,  $CH_2Cl_2$ , d) pyridine,  $(CF_3CO)_2O$ ,  $CH_2Cl_2$ , 0 °C to r.t., e) pyridine,  $(CF_3CF_2CO)_2O$ ,  $CH_2Cl_2$ , 0 °C to r.t., f) pyridine,  $(CF_3CF_2CF_2CO)_2O$ ,  $CH_2Cl_2$ , 0 °C to r.t.

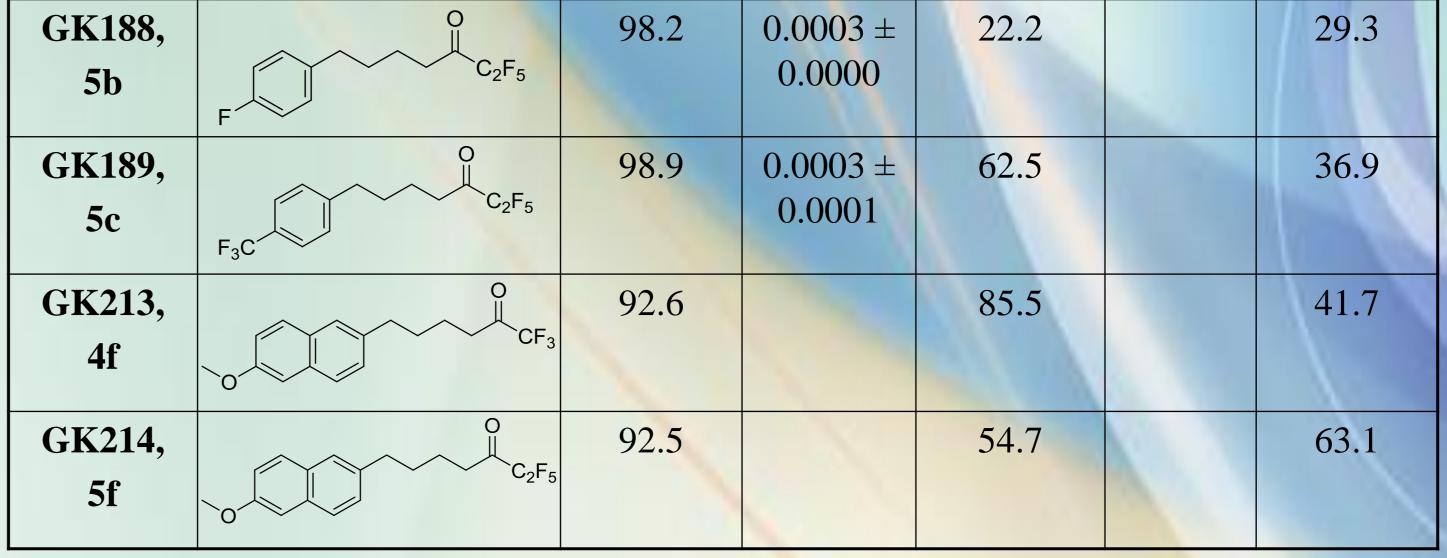
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N.D. signifies compounds with less than 25% inhibition (or no detectable inhibition).